# Supporting Information 

# Optimization of Signal Amplification by Reversible Exchange for polarization of Tridentate Chelating Bis[(2-pyridyl)alkyl]amine 

Sein Min ${ }^{\text {a }}$, Heelim Chae ${ }^{\text {a }}$, Hye Jin Jeong ${ }^{\text {b }}$, Kiwoong Kim ${ }^{\text {c }}$, Sung Keon Namgoong ${ }^{\text {a* }}$ and Keunhong Jeongb ${ }^{\text {* }}$<br>${ }^{a}$ Department of Chemistry, Seoul Women's University, Seoul 01797, South Korea<br>${ }^{\text {b }}$ Department of Chemistry, Korea Military Academy, Seoul 01805, South Korea<br>${ }^{\text {cK Korea Research Institute of Standards and Science, Daejeon 34113, South Korea }}$<br>* Sung Keon Namgoong<br>Department of Chemistry, Seoul Women's University, Seoul 01797, South Korea, sknam@swu.ac.kr<br>* Keunhong Jeong<br>Department of Chemistry, Korea Military Academy, Seoul 01805, South Korea. +82-2-2197-2823<br>doas1mind@kma.ac.kr/<br>doas1mind@gmail.com

## Table of Contents

## 1. SABRE experimental details

Figure S1. Prediction of coordination with catalyst and Sub (a) Proposed binding compounds (Ir-BPA) (b) ${ }^{1} \mathrm{H}$ NMR spectrum in $\mathrm{CD}_{3} \mathrm{OD}$. The lower is iridium catalyst spectrum, the middle is $\mathrm{Sub}(\mathrm{BPA})$ spectrum, and the upper is the reaction of iridium catalyst ( 1 eq .) and BPA ( 1 eq .) at 60 min . (c) ${ }^{1} \mathrm{H}$ NMR spectrum in $\mathrm{CD}_{3} \mathrm{OD}$ enlarge the spectral region of $\delta 8.65-6.45 \mathrm{ppm}(\mathbf{d})$ Proposed binding compounds (Ir-BPMA) (e) ${ }^{1} \mathrm{H}$ NMR spectrum in $\mathrm{CD}_{3} \mathrm{OD}$. The lower is iridium catalyst spectrum, the middle is Sub (BPMA) spectrum, and the upper is the reaction of iridium catalyst ( 1 eq.) and BPMA ( 1 eq. ) at 30 min . (f) ${ }^{1} \mathrm{H}$ NMR spectrum in $\mathrm{CD}_{3} \mathrm{OD}$ enlarge the spectral region of $\delta 8.70-6.70 \mathrm{ppm}(\mathbf{g}){ }^{1} \mathrm{H}$ NMR spectrum in $\mathrm{CD}_{3} \mathrm{OD}$ enlarge the spectral region of $\delta$ $5.00-3.00 \mathrm{ppm}$ Sub (h) Proposed binding compounds (Ir-BPEA) (i) ${ }^{1} \mathrm{H}$ NMR spectrum in $\mathrm{CD}_{3} \mathrm{OD}$. The bottom spectrum corresponds to the Ir-catalyst, the second-lowest one is the Sub (BPEA) spectrum, the third-lowest is the reaction of Ir-catalyst ( 0.5 eq .) with BPEA ( 1 eq .) at 30 min , and the top one relates to the reaction of Ir-catalyst ( 1 eq .) with BPEA ( $1 \mathrm{eq}$. .) at 30 min . (j) ${ }^{1} \mathrm{H}$ NMR spectrum in $\mathrm{CD}_{3} \mathrm{OD}$ with the enlarged spectral region of $\delta 8.80-6.80 \mathrm{ppm}$. This spectrum shows that the proton is shielding as the concentration increases $(\mathbf{k}){ }^{1} \mathrm{H}$ NMR spectrum in $\mathrm{CD}_{3} \mathrm{OD}$ with the enlarged spectral region between $\delta 4.20-2.80 \mathrm{ppm}$. Especially, the H-8 proton was shielding more than the $\mathrm{H}-7$ proton
Figrure S2. Fully activated iridium catalyst using 30 min . bubbling was used to see the chemical shift with different concentration of BPEA. (a) ${ }^{1} \mathrm{H}$ NMR spectrum in $\mathrm{CD}_{3} \mathrm{OD}$. The bottom spectrum corresponds to the Ir-catalyst, the second-lowest one is the activated iridium catalyst using parahydrgoen, the third-lowest one is the Sub (BPEA) spectrum, the fourth-lowest is the reaction of Ir-catalyst ( $1 \mathrm{eq}$. .) with BPEA ( 0.5 eq. ), and the top spectrum is related to the reaction of Ir-catalyst (1 eq.) with BPEA (1 eq.). (b) ${ }^{1} \mathrm{H}$ NMR spectrum in $\mathrm{CD}_{3} \mathrm{OD}$ with the enlarged spectral region of $9.00-6.00 \mathrm{ppm}$. (c) ${ }^{1} \mathrm{H}$ NMR spectrum in $\mathrm{CD}_{3} \mathrm{OD}$ with the enlarged spectral region between $4.00-2.00 \mathrm{ppm}$
Figure S3. $\mathrm{T}_{1}$ relaxation measurement of hyperpolarized BPEA of 70 G magnetic field. (a) $\mathrm{H}-$ 3,5 (b) H-4, (c) H-6, (d) H-7, (e) H-8
Figure S4. ${ }^{1} \mathrm{H}$ NMR spectra of SABRE hyperpolarization in $\mathrm{CD}_{3} \mathrm{OD}$, where each Sub is (a) BPA and (b) BPMA. The bottom spectrum (black) is a thermally polarized sample spectrum provided for reference and the upper spectrum (red) is a hyperpolarized sample spectrum recorded after $p-\mathrm{H}_{2}$ bubbling.
Figure S5. ${ }^{1} \mathrm{H}$ NMR signal enhancement and Ir-HH peaks of (a) BPA, (b) BPMA, and (c) BPEA dependence in the various concentration at 70 G magnetic field. Notice the different vertical scale (fold)

## 2. Synthetic details

## 3. Structural Coordination after theoretical calculation

## 4. References

## 1. SABRE experimental details

## Preparation of $\boldsymbol{p}-\mathbf{H}_{\mathbf{2}}$ gas

The $p-\mathrm{H}_{2}$ generator was prepared in-house. ${ }^{[1]} 99.9 \%$ pure $\mathrm{H}_{2}$ gas was passed through various area copper tubes used in the heat-exchanger design. The copper tube was sunk into liquid $\mathrm{N}_{2}$ Dewar. The catalytic amount of $\mathrm{FeO}(\mathrm{OH})$ was used to convert $o-\mathrm{H}_{2}$ into $p-\mathrm{H}_{2}$ through symmetry breaking at low temperatures. This preparation designed and developed the system an $\mathrm{H}_{2}$ enriched with $50 \%$ para-isomer to collect various magnetic field data. The power supply used for this application is GPS-1850D (Bench Power Supply, Linear DC). The shielding coil wound around the copper wire and the outer shielding coil consist of 20 mm diameters and 190 mm lengths. The magnetic field through the shielding coil is adjusted by setting current in the range $0-5 \mathrm{~A}$.

## SABRE methods in High-Field (300 MHz) NMR Detection

SABRE hyperpolarization can be successfully hyperpolarized in very different molecules. NMR data were reported as signal enhancement of concerning the signal of the thermally polarized sample in the Bruker Avance III 300 MHz (7T) magnet. Samples for these SABRE hyperpolarization experiments were conducted using a solution of $[\operatorname{Ir}(\mathrm{COD})(\mathrm{IMes})] \mathrm{Cl}\left(2 \mathrm{mg}, 3.12 \mu_{\mathrm{mol}}\right)$ and Sub in $\mathrm{CD}_{3} \mathrm{OD}$ volume of 0.9 $\mu_{\mathrm{L}}$. The $p-\mathrm{H}_{2}$ gas was bubbled through solution $1 / 16 \mathrm{in}$. PTFE tubing inside the 5 mm medium-wall precision NMR tube. The completion of the catalyst activation usually takes 10-20 minutes. The activation was checked by a hydride signal around $-40 \sim 0 \mathrm{ppm}$. Activation by bubbling $p-\mathrm{H}_{2}$ occurred outside the NMR magnet, and the signal was acquired ( $6 \pm 2 s$ ) after the bubbling was stopped. The pulse program was "zg30", which is provided that the 30 -degree pulse. All SABRE hyperpolarization experiments were conducted with 1 scan acquisition at 298 K on 7 T . The NMR data were analyzed using Mnova software (Mestrelab Research, S. L.).

## Equation of enhancement factor

The polarization enhancement value of ${ }^{1} \mathrm{H}$ NMR signals $\varepsilon$ was calculated according to Equation (1):

$$
\begin{equation*}
\varepsilon=\left(\frac{I_{H P}}{I_{\text {therm }}}\right) \cdot\left(\frac{C_{\text {therm }}}{C_{H P}}\right) \tag{1}
\end{equation*}
$$

The ${ }^{1} \mathrm{H}$ signal enhancement $\varepsilon$ was calculated by comparing the integral values derived from the NMR spectra of the hyperpolarization (HP) sample and the corresponding thermally polarized sample of substrates. In which $\mathrm{I}_{\mathrm{HP}}$ and $\mathrm{I}_{\text {therm }}$ are the integrals of HP and the thermally polarized samples, respectively; $\mathrm{C}_{\mathrm{HP}}$ and $\mathrm{C}_{\text {therm }}$ are the integrals of the HP and the thermal solvent peak of the sample, respectively. The thermally and polarized spectra were recorded using identical acquisition parameters and receiver gain.

## Prediction of the coordinated structure with NMR study

We studied the possible stable structure between Ir-catalyst ([ $\operatorname{Ir}(\mathrm{COD})(\mathrm{IMes}) \mathrm{Cl}])$ and substrates using SABRE hyperpolarization. The two functional groups, pyridine and secondary amine, are well-known ligands forming complexes with Ir-catalysts. ${ }^{[2-6]}$ Moreover, there are three open coordination sites on the

Ir- catalyst. Therefore, it is expected that all substrates, which have three binding sites, bind with the Ircatalysts. However, it is not obvious that a stable structure is formed with BPA and BPMA. leading to large angle strain in the structure, whereas, six- (Fig. S1(a)) and five-membered ring (Fig. S1(d)) formation within the equatorial position can result in a stable structure.

Because BPEA can form two six-membered rings after binding with the Ir-catalyst (1:1), the structure is less strained, and it would be an available structure during SABRE. This hypothesis was proven by ${ }^{1} \mathrm{H}$ NMR without hyperpolarization. With respect to BPA and BPMA, NMR signals of pyridine moieties were detected, which implied pyridine moieties with different chelate structures. Otherwise, split pyridine signals of the Ir-catalyst and the BPEA complex were not detected here. Figure S1(h)-(k) shows that as the amount of Ir-catalyst increased to $100 \mathrm{~mol} \%$, all proton signals were gradually shifted to the upfield because of the direct coordination with the iridium center.

d)
e)


h)
i)



Figure S1. Prediction of coordination with catalyst and Sub (a) Proposed binding compounds (Ir-BPA) (b) ${ }^{1} \mathrm{H}$ NMR spectrum in $\mathrm{CD}_{3} \mathrm{OD}$. The lower is iridium catalyst spectrum, the middle is $\mathrm{Sub}(\mathrm{BPA})$ spectrum, and the upper is the reaction of iridium catalyst ( 1 eq.) and BPA (1 eq.) at 60 min . (c) ${ }^{1} \mathrm{H}$ NMR spectrum in $\mathrm{CD}_{3} \mathrm{OD}$ enlarge the spectral region of $\delta 8.65$ -6.45 ppm (d) Proposed binding compounds (Ir-BPMA) (e) ${ }^{1} \mathrm{H}$ NMR spectrum in $\mathrm{CD}_{3} \mathrm{OD}$. The lower is iridium catalyst spectrum, the middle is Sub (BPMA) spectrum, and the upper is the reaction of iridium catalyst (1 eq.) and BPMA (1 eq.) at 30 min . (f) ${ }^{1} \mathrm{H}$ NMR spectrum in $\mathrm{CD}_{3} \mathrm{OD}$ enlarge the spectral region of $\delta 8.70-6.70 \mathrm{ppm}(\mathrm{g}){ }^{1} \mathrm{H}$ NMR spectrum in $\mathrm{CD}_{3} \mathrm{OD}$
enlarge the spectral region of $\delta 5.00-3.00 \mathrm{ppm} \mathrm{Sub}$ (h) Proposed binding compounds (Ir-BPEA) (i) ${ }^{1} \mathrm{H}$ NMR spectrum in $\mathrm{CD}_{3} \mathrm{OD}$. The bottom spectrum corresponds to the Ir-catalyst, the second-lowest one is the Sub (BPEA) spectrum, the thirdlowest is the reaction of Ir-catalyst ( 0.5 eq .) with BPEA ( 1 eq .) at 30 min , and the top one relates to the reaction of Ir-catalyst (1 eq.) with BPEA ( 1 eq .) at 30 min . (j) ${ }^{1} \mathrm{H}$ NMR spectrum in $\mathrm{CD}_{3} \mathrm{OD}$ with the enlarged spectral region of $\delta 8.80-6.80 \mathrm{ppm}$. This spectrum shows that the proton is shielding as the concentration increases $(\mathrm{k}){ }^{1} \mathrm{H}$ NMR spectrum in $\mathrm{CD}_{3} \mathrm{OD}$ with the enlarged spectral region between $\delta 4.20-2.80 \mathrm{ppm}$. Especially, the H-8 proton was shielding more than the $\mathrm{H}-7$ proton
(a)

(b)

(c)


Figure S2. Fully activated iridium catalyst using 30 min . bubbling was used to see the chemical shift with different concentration of BPEA. (a) ${ }^{1} \mathrm{H}$ NMR spectrum in $\mathrm{CD}_{3} \mathrm{OD}$. The bottom spectrum corresponds to the Ir-catalyst, the secondlowest one is the activated iridium catalyst using parahydrgoen, the third-lowest one is the Sub (BPEA) spectrum, the fourth-lowest is the reaction of Ir-catalyst ( 1 eq .) with BPEA ( 0.5 eq.), and the top spectrum is related to the reaction of Ircatalyst (1 eq.) with BPEA (1 eq.). (b) ${ }^{1} \mathrm{H}$ NMR spectrum in $\mathrm{CD}_{3} \mathrm{OD}$ with the enlarged spectral region of $9.00-6.00 \mathrm{ppm}$. (c) ${ }^{1} \mathrm{H}$ NMR spectrum in $\mathrm{CD}_{3} \mathrm{OD}$ with the enlarged spectral region between $4.00-2.00 \mathrm{ppm}$

## $T_{1}$ experiments

$\mathrm{T}_{1}$ relaxation (longitudinal relaxation) is the means by which the net magnetization indicates the equilibrium over time (sec) and can be calculated mathematically, $\mathrm{y}=\mathrm{A}_{1} * \exp \left(-\mathrm{x} / \mathrm{T}_{1}\right)+\mathrm{y}_{0}$


Figure S3. $\mathrm{T}_{1}$ relaxation measurement of hyperpolarized BPEA of 70 G magnetic field. (a) $\mathrm{H}-3,5$ (b) H-4, (c) H-6, (d) H-7, (e) $\mathrm{H}-8$

## Further spectrum with hyperpolarization

a)

b)


Figure S4. ${ }^{1} \mathrm{H}$ NMR spectra of SABRE hyperpolarization in $\mathrm{CD}_{3} \mathrm{OD}$, where each Sub is (a) BPA and (b) BPMA. The bottom spectrum (black) is a thermally polarized sample spectrum provided for reference and the upper spectrum (red) is a hyperpolarized sample spectrum recorded after $p-\mathrm{H}_{2}$ bubbling.

## Further SABRE experiments

a)



b)



c)


Figure S5. ${ }^{1} \mathrm{H}$ NMR signal enhancement and Ir-HH peaks of (a) BPA, (b) BPMA, and (c) BPEA dependence in the various concentration at 70G magnetic field. Notice the different vertical scale (fold)

## 2. Synthetic details

Most of the reagents used in this work were purchased from Sigma-Aldrich, Alfa aesar, Tokyo chemical industry and companies. In anhydrous conditions, the reaction glassware was flame-dried for at least 1 hour and purged with $\mathrm{N}_{2}$. The THF solvent was distilled from Na /benzophenone under $\mathrm{N}_{2}$ atmospheric condition. All the synthetic compounds were characterized by ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectroscopic analysis. The NMR spectra were recorded on Bruker Avance III 300 MHz . The $\mathrm{CD}_{3} \mathrm{OD}$ was used as solvent unless otherwise stated $\left(\mathrm{CD}_{3} \mathrm{OD}\right.$ peak ${ }^{1} \mathrm{H}$ NMR : $3.31\left(\mathrm{CH}_{3}\right), 4.87(\mathrm{OH}) \mathrm{ppm},{ }^{13} \mathrm{C}$ NMR : 49.0 ppm$)$. The following abbreviations are used. (brs = broad singlet, $\mathrm{s}=$ singlet, $\mathrm{d}=$ doublet, $\mathrm{t}=$ triplet, $\mathrm{q}=$ quartet, qt $=$ quintet, $\mathrm{st}=$ sextet, $\mathrm{spt}=$ septet, and $\mathrm{m}=$ multiplet) Thin layer chromatography was performed with Silica gel 60 F254 and compounds were visualized under UV light at 254 nm .
Chloro( $\eta^{4}$ 1,5-cyclooctadiene)(1,3-bis(2,4,6-trimethylphenyl)-imidazol2-ylidene)iridium( I ), ${ }^{[6,7]}$

## [ $\operatorname{Ir}(\mathbf{C O D})(\mathbf{I M e s})] \mathbf{C l}$

1,3-bis(2,4,6-trimethylphenyl)- 1,3-dihydro-2H-imidazol-2-ylidene ( $0.48 \mathrm{~g}, 1.5 \mathrm{mmol}$ ) in THF ( 3 mL ) was added dropwise to a stirred solution of bis(1,5-cyclooctadiene)diiridium( I ) dichloride ( $0.5 \mathrm{~g}, 0.72$ $\mathrm{mmol})$ in dry THF $(2 \mathrm{~mL})$ at room temperature under $\mathrm{N}_{2}$. The mixture was stirred for 12 hours and quenched with distilled water. THF was evaporated under reduced pressure and $\mathrm{CH}_{2} \mathrm{Cl}_{2}(30 \mathrm{~mL})$ was added to resulting mixture. The organic layer was washed with $\mathrm{H}_{2} \mathrm{O}$, dried over $\mathrm{MgSO}_{4}$, and concentrated under reduced pressure to provide compound $(0.95 \mathrm{~g}, 98 \%)$ as a yellow solid. : ${ }^{1} \mathrm{H}$ NMR ( 300 MHz , MeOD- $\mathrm{d}_{4}$ ): $\delta 7.23(\mathrm{~s}, 2 \mathrm{H}), 7.02(\mathrm{~s}, 4 \mathrm{H}), 4.00-3.98(\mathrm{t}, 2 \mathrm{H}, \mathrm{J}=3 \mathrm{~Hz}$ ), 3.09-3.07(t,2H, J=3 Hz), $2.36(\mathrm{~s}, 6 \mathrm{H})$, $2.26(\mathrm{~s}, 6 \mathrm{H}), 2.18(\mathrm{~s}, 6 \mathrm{H}), 1.77-1.42(\mathrm{~m}, 4 \mathrm{H}), 1.39-1.22(\mathrm{~m}, 4 \mathrm{H}),{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{MeOD}-\mathrm{d}_{4}$ ): $\delta 140.07$, $137.73,130.48,129.49,125.42,82.96,78.37,53.35,34.53,29.96,21.33,203.14$

## Bis[(2-pyridyl)ethyl]amine ${ }^{[8]}$, BPEA

A solution of ammonium chloride $(9.95 \mathrm{~g}, 184 \mathrm{mmol})$ in distilled water ( 28 mL ) and methanol ( 4 mL ) was added to 2-vinylpyridine ( $10 \mathrm{~mL}, 92 \mathrm{mmol}$ ). The solution was heated for 18 hours under reflux, and the resulting mixture was cooled at $0^{\circ} \mathrm{C}$ and basified with $30 \% \mathrm{w} / \mathrm{w}$ aqueous NaOH solution until pH 10 . $\mathrm{CH}_{2} \mathrm{Cl}_{2}(300 \mathrm{~mL})$ was added and the organic layer was dried over $\mathrm{MgSO}_{4}$ and concentrated under the reduced pressure to provide BPEA and PEA mixtures. The crude oily residue ( $10 \mathrm{~g}, 56 \mathrm{mmol}$ ) was dissolved in CH2Cl2 ( 20 mL ) and di-tert-butyl dicarbonate ( $60 \mathrm{~mL}, 254 \mathrm{mmol}$ ). After stirring for 6 hours, the solution concentrated under reduced pressure to obtain protecting compounds, Boc-BPEA and BocPEA. The resulting residue was purified by flash column chromatography on silica gel (dichloromethane:methanol,100:0-30:1) as an eluent to provide Boc-BPEA as a yellow oil. Boc-BPEA $(5 \mathrm{~g})$ was deprotected with TFA $(10 \mathrm{~mL})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. After stirring for 1 hour, the solution was evaporated under reduced pressure, BPEA ( $2.33 \mathrm{~g}, 68 \%$ )was obtained as pale-yellow oil. : ${ }^{1} \mathrm{H}$ NMR ( 300 MHz , MeOD- $\mathrm{d}_{4}$ ): $\delta$ 8.77-8.75 (dt, 2H), 8.46-8.40 (t, 2H), 7.96-7.93 (d, 2H), 7.90-7.85 (t, 2H), 3.65-3.59 (td, $4 \mathrm{H}, \mathrm{J}=6.9 \mathrm{~Hz}, 7.8 \mathrm{~Hz}, 0.9 \mathrm{~Hz}, 0.45 \mathrm{~Hz}$ ), $3.53-3.48(\mathrm{td}, 4 \mathrm{H}, \mathrm{J}=6.9 \mathrm{~Hz}, 7.8 \mathrm{~Hz}, 0.9 \mathrm{~Hz}, 0.45 \mathrm{~Hz}),{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{MeOD}-\mathrm{d} 4$ ): $\delta 154.42,146.51,144.83,128.39,126.63,47.34,31.54$

2-(Pyridin-2-yl)ethanamine, PEA

This product was prepared from 2-vinylpyridine as described above obtained. ${ }^{1} \mathrm{H}$ NMR ( 300 MHz , MeOD- $\mathrm{d}_{4}$ ): $\delta$ 8.62-8.60 (d, 1H), 8.02-7.97 (t, 1H), 7.54-7.47 (m, 2H), 3.40-3.35 (t, 2H), 3.24-3.20 (t, 2H), ${ }^{13} \mathrm{C}$ NMR ( 75 MHz, , MeOD-d4): $\delta$ 157.56, 148.80, 141.17, 126.01, 124.65, 39.85, 34.36 tert-Butyl bis[2-(pyridin-2-yl)ethyl]carbamate Boc-BPEA

This product was protected from 2-vinylpyridine as described above. ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta$ 8.39-8.38 (d, 2H), 7.47-7.41 (t, 2H), 7.06-7.03 (d, 2H), 6.99-6.96 (t, 2H), 3.54-3.48 (m, 4H), 2.97-2.93 $(\mathrm{m}, 4 \mathrm{H}), 1.40(\mathrm{~s}, 9 \mathrm{H}){ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 159.28,155.19,149.25,136.21,123.35,121.23$, 79.17, 47.72, 37.45, 28.27
tert-Butyl 2-(pyridine-2-yl)ethylcarbamate Boc-PEA
This product was protected from 4 b as described above. ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 8.54-8.53$ (d, $1 \mathrm{H}), 7.66-7.60(\mathrm{t}, 1 \mathrm{H}), 7.20-7.14(\mathrm{~m}, 2 \mathrm{H}), 5.14(\mathrm{brs}, 1 \mathrm{H}), 3.58-3.51(\mathrm{q}, 2 \mathrm{H}), 3.01-2.97(\mathrm{t}, 2 \mathrm{H}), 1.42(\mathrm{~s}$, $9 \mathrm{H}){ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta 159.39,155.95,148.98,136.82,123.62,121.58,79.12,39.89,37.71$, 28.42

## 3. Structural Coordination after theoretical calculation

## Theoretical calculation

Since the BP86/LANL2DZ level of theory provided optimized structures with geometrical parameters closest to the experimental X-ray crystal structure, this level of theory was applied to obtain stable structures of Ir-catalyst complexes with BPA, BPMA, BPEA, and pyridine. ${ }^{[6]}$ A reduced catalyst model (dimethyl benzenes are omitted) was used to represent each complex in order to improve the calculation efficiency and had less of an impact on the optimized iridium complex structure. All calculations were performed using the Gaussian 16 W package. Atoms in molecule (AIM) analysis for calculating the Shannon aromaticity index was carried out using the Multiwfn program. ${ }^{[9]}$

## Pyridine-Ir-catalyst

| Ir | 0.25513700 | 0.61120800 | -0.35175100 |
| :--- | :---: | :---: | :---: |
| C | 0.45300300 | 5.33168200 | -1.87091000 |
| C | 0.15995900 | 4.27527100 | -2.76732100 |
| C | 0.10329000 | 2.95315400 | -2.28853900 |
| N | 0.32431600 | 2.63644000 | -0.96262200 |
| C | 0.60686800 | 3.67045000 | -0.09420200 |
| C | 0.67852400 | 5.01463600 | -0.51143100 |
| H | 0.50311500 | 6.36721700 | -2.22168900 |
| H | -0.02343200 | 4.46619800 | -3.82874300 |
| H | 0.77309300 | 3.38876500 | 0.94818900 |
| H | 0.90725200 | 5.79224200 | 0.22349500 |
| C | -3.05513500 | 1.65830400 | 3.31263600 |
| C | -2.99446300 | 2.43090500 | 2.12891800 |
| C | -2.03607500 | 2.11723700 | 1.14467500 |
| N | -1.13722200 | 1.08067200 | 1.27810000 |
| C | -1.21842900 | 0.32254800 | 2.42748900 |
| C | -2.14973900 | 0.58015200 | 3.45298200 |
| H | -3.79018300 | 1.88081400 | 4.09240300 |
| H | -3.68311800 | 3.26391500 | 1.95877700 |
| H | -1.98083200 | 2.68000000 | 0.21082300 |
| H | -2.16311200 | -0.06071700 | 4.33994000 |
| H | -0.99045400 | 0.40494900 | -1.34400000 |
| H | 1.17439500 | 0.27816200 | -1.63024400 |
| C | -0.62513300 | -3.53649600 | 0.25198600 |
| C | 0.74819800 | -3.60946300 | 0.18271700 |
| H | -1.36889400 | -4.31929900 | 0.36974400 |
| H | 1.41295000 | -4.46698400 | 0.23287400 |
| C | 0.16747800 | -1.36614100 | -0.03293100 |
| N | -0.97317300 | -2.17679100 | 0.11567500 |
| N | 1.22160300 | -2.29369000 | 0.00527600 |
| C | -2.37856900 | -1.73445900 | 0.02909100 |
| C | 2.64768300 | -1.98246500 | -0.20210900 |
| C | 4.50901200 | 1.62716300 | 2.17793300 |
| C | 4.38579900 | 1.83988300 | 0.78397100 |
| C | 3.17207000 | 1.52595200 | 0.14153400 |
| N | 2.08084300 | 1.01544200 | 0.81336900 |
| C | 2.20924200 | 0.81789600 | 2.16849100 |
| C | 3.39404200 | 1.10746800 | 2.87660000 |
| H | 5.21669800 | 2.23932900 | 0.19473800 |
| H | 3.03914500 | 1.66420700 | -0.93408900 |
| H | 3.43452700 | 0.92572200 | 3.95497300 |
| H | 5.44214300 | 1.85797200 | 2.70146500 |
|  |  |  |  |
| H |  |  |  |
| H |  |  |  |


| H | -2.39940100 | -0.70703200 | -0.35971900 |
| :--- | ---: | :---: | :---: |
| H | -2.92725800 | -2.39248200 | -0.66634100 |
| H | -2.86629700 | -1.76798300 | 1.01928600 |
| H | 3.14307500 | -2.85413700 | -0.66063700 |
| H | 2.72760100 | -1.12259900 | -0.88299500 |
| H | 3.15000000 | -1.74332500 | 0.75116400 |
| H | -0.11629700 | 2.10743600 | -2.94227700 |
| H | 1.33440100 | 0.42084200 | 2.68543000 |
| H | -0.53830000 | -0.53118700 | 2.48541700 |

## BPA-Ir-catalyst

| Ir | 0.23651800 | 0.57086200 | -0.44218200 |
| :---: | :---: | :---: | :---: |
| C | 0.16627100 | 5.33573500 | -1.91008300 |
| C | 0.00182100 | 4.32319000 | -2.86447500 |
| C | 0.06472700 | 2.94818000 | -2.47536100 |
| N | 0.24494800 | 2.60251600 | -1.13225500 |
| C | 0.40256600 | 3.62370000 | -0.21760700 |
| C | 0.38946800 | 4.98333500 | -0.55300700 |
| H | 0.10566400 | 6.38620000 | -2.21280800 |
| H | -0.22463800 | 4.57607400 | -3.90166900 |
| H | 0.54036700 | 3.30556100 | 0.81747200 |
| H | 0.52946200 | 5.73908700 | 0.22403400 |
| C | -2.88593200 | 1.60960300 | 3.30131300 |
| C | -3.25375300 | 1.86109100 | 1.95369300 |
| C | -2.30954400 | 1.62137800 | 0.94445000 |
| N | -1.03871300 | 1.14401300 | 1.19694400 |
| C | -0.66202100 | 0.97191600 | 2.50699300 |
| C | -1.57849700 | 1.17467600 | 3.57909000 |
| H | -3.59696800 | 1.77172500 | 4.11727800 |
| H | -4.25000700 | 2.22612300 | 1.69106900 |
| H | -2.53860300 | 1.77525800 | -0.11239500 |
| H | -1.25230200 | 1.01604600 | 4.61319100 |
| H | -1.03074500 | 0.24388900 | -1.38823100 |
| H | 1.17750100 | 0.15349600 | -1.68133300 |
| C | -0.53091200 | -3.53331300 | 0.52343400 |
| C | 0.83377400 | -3.59770300 | 0.35216700 |
| H | -1.25072300 | -4.31202900 | 0.75805200 |
| H | 1.51393900 | -4.44237000 | 0.41120000 |
| C | 0.20103600 | -1.38495300 | 0.01519500 |
| N | -0.90842300 | -2.19161800 | 0.31268700 |
| N | 1.26991100 | -2.29426100 | 0.03917700 |
| C | -2.31452000 | -1.75135100 | 0.34480500 |
| C | 2.67374500 | -1.98544800 | -0.28477700 |
| C | 4.27894500 | 1.27714900 | 2.37543900 |
| C | 4.31484200 | 1.50857800 | 0.97544500 |
| C | 3.12857000 | 1.36612200 | 0.24102300 |
| N | 1.92473100 | 1.00292300 | 0.81250100 |
| C | 1.87747100 | 0.85291400 | 2.17798700 |
| C | 3.05041700 | 0.96036600 | 2.98010100 |
| H | 5.24101900 | 1.78461500 | 0.46478600 |
| H | 3.09433400 | 1.50984300 | -0.84112900 |
| H | 2.98328500 | 0.82161300 | 4.06506400 |
| H | 5.18492700 | 1.36524100 | 2.98294900 |
| N | 0.65496000 | 0.57594600 | 2.82736000 |
| H | 0.77026600 | 0.31700600 | 3.81052100 |
| H | -2.44326300 | -0.91911000 | -0.36243300 |


| H | -2.96216100 | -2.58869000 | 0.03642000 |
| :--- | ---: | :---: | :---: |
| H | -2.60937900 | -1.42288700 | 1.35695900 |
| H | 3.14206000 | -2.87145600 | -0.74467900 |
| H | 2.69634100 | -1.15300100 | -1.00302900 |
| H | 3.24297100 | -1.70688400 | 0.61953900 |
| N | -0.09524800 | 1.89727300 | -3.35675400 |
| H | -0.26769200 | 0.97907700 | -2.90683700 |
| C | -0.12006200 | 1.86734300 | -4.77299700 |
| C | 0.52401900 | 2.81121400 | -5.62113900 |
| N | -0.77850600 | 0.76700200 | -5.25863900 |
| C | 0.43264700 | 2.62240400 | -7.01816900 |
| H | 1.12072700 | 3.63239500 | -5.21560400 |
| C | -0.84231600 | 0.59960400 | -6.60984400 |
| C | -0.27180600 | 1.50778200 | -7.53332100 |
| H | 0.92225400 | 3.33160200 | -7.69428200 |
| H | -1.37155300 | -0.29722900 | -6.95093200 |
| H | -0.36179500 | 1.33559000 | -8.60969900 |

## BPMA-Ir-catalyst

| Ir | -0.17747800 | 0.77390400 | -0.52777500 |
| :--- | :---: | :---: | :---: |
| C | -0.19778100 | 5.20953600 | -2.85279800 |
| C | -0.78466200 | 4.03826500 | -3.40426100 |
| C | -0.77921700 | 2.82785900 | -2.68805600 |
| N | -0.20666600 | 2.70099700 | -1.43858700 |
| C | 0.37445100 | 3.83409900 | -0.90358800 |
| C | 0.38965200 | 5.07256800 | -1.56661000 |
| H | -1.22961400 | 4.05460900 | -4.40535800 |
| H | -1.21447600 | 1.91484500 | -3.09710200 |
| H | 0.84875200 | 3.72032800 | 0.07263800 |
| H | 0.86616300 | 5.92997600 | -1.07750200 |
| H | -1.46228600 | 0.42534800 | -1.43639000 |
| H | 0.62857300 | 0.32076700 | -1.83549100 |
| C | -0.96970000 | -3.27695300 | 0.61316900 |
| C | 0.39843600 | -3.34623200 | 0.47135600 |
| H | -1.69316900 | -4.04672700 | 0.86605200 |
| H | 1.07712300 | -4.18701500 | 0.58230000 |
| C | -0.22430200 | -1.14669800 | 0.02707800 |
| N | -1.34339900 | -1.94392300 | 0.34087300 |
| N | 0.84127100 | -2.05635000 | 0.11539600 |
| C | -2.75058600 | -1.50775700 | 0.30406200 |
| C | 2.24856400 | -1.74370800 | -0.19028900 |
| C | 3.46414100 | 2.30556600 | 2.49611000 |
| C | 3.74669800 | 2.14259900 | 1.11850400 |
| C | 2.73125500 | 1.67411300 | 0.26157900 |
| N | 1.47272200 | 1.35444200 | 0.71561200 |
| C | 1.19173300 | 1.51620200 | 2.06213700 |
| C | 2.16575100 | 1.99524200 | 2.96504000 |
| H | 4.73319200 | 2.37762500 | 0.70807000 |
| H | 2.89331700 | 1.54523100 | -0.81164800 |
| H | 1.90355000 | 2.12412800 | 4.02005000 |
| H | -0.23105300 | 2.67396400 | 3.18469600 |
| C | -0.21203800 | 1.10583800 | 2.50488000 |
| H | -1.21190900 | 0.01136800 | 2.67227800 |
| H | 1.39482800 | 1.40961400 |  |
| N |  |  | 3.4437300 |
|  |  |  |  |


| H | -1.99610300 | 0.73194300 | 1.51120500 |
| :--- | ---: | ---: | ---: |
| C | -1.78643100 | 2.80164700 | 1.48278600 |
| H | -2.26068100 | 3.01021700 | 0.51002500 |
| H | -0.93877000 | 3.49565500 | 1.60929200 |
| C | -2.78503800 | 2.99859500 | 2.61725200 |
| C | -4.16743900 | 3.18790000 | 2.36141100 |
| N | -2.27259900 | 2.99051700 | 3.89500900 |
| C | -5.05040400 | 3.38349800 | 3.45092300 |
| H | -4.54023900 | 3.19665000 | 1.33094100 |
| C | -3.13684000 | 3.17063200 | 4.93710900 |
| C | -4.52745700 | 3.37621500 | 4.76482500 |
| H | -6.11992600 | 3.54246000 | 3.27720500 |
| H | -2.69026500 | 3.15794600 | 5.93790600 |
| H | -5.17307300 | 3.52937100 | 5.63512400 |
| H | -2.80029200 | -0.54939300 | -0.23701800 |
| H | -3.35637300 | -2.24953500 | -0.24398100 |
| H | -3.16560700 | -1.40374200 | 1.32472100 |
| H | 2.74197800 | -2.64414800 | -0.59219700 |
| H | 2.27291500 | -0.94809200 | -0.94974100 |
| H | 2.78915100 | -1.40758000 | 0.71154400 |
| C | -0.27193000 | 6.57061600 | -3.55741300 |
| H | -0.32675600 | 6.42248700 | -4.65351000 |
| H | 0.64795300 | 7.14923100 | -3.35183700 |
| N | -1.42449500 | 7.37579800 | -3.12419300 |
| H | -1.41506200 | 7.74303000 | -2.16077100 |
| C | -2.77407000 | 6.96615700 | -3.52200600 |
| H | -2.94093900 | 7.16952200 | -4.60039700 |
| H | -2.96404200 | 5.86992200 | -3.37614700 |
| C | -3.81373000 | 7.72424100 | -2.69564200 |
| C | -5.15021700 | 7.88190800 | -3.14149600 |
| N | -3.38367500 | 8.21846900 | -1.49260200 |
| C | -6.07156500 | 8.56759800 | -2.31798900 |
| H | -5.45509200 | 7.48579400 | -4.11686200 |
| C | -4.28162900 | 8.88290400 | -0.70432900 |
| H | -5.63198200 | 9.07977100 | -1.07218300 |
| H | -7.10762600 | 8.70955100 | -2.64400100 |
| H | -3.89234700 | 9.27017700 | 0.24441600 |
|  | -6.31108100 | 9.62540800 | -0.40946000 |
|  |  |  |  |
| H |  |  |  |

## BPEA-Ir-catalyst

| Ir | -0.77855700 | 0.12434300 | -1.46069700 |
| :--- | :---: | :---: | :---: |
| H | -1.97822300 | -0.36817400 | -2.40639100 |
| H | 0.14731900 | -0.75785400 | -2.43288800 |
| C | -2.22495000 | -3.34757300 | 0.59770300 |
| C | -0.87322700 | -3.56648600 | 0.73380300 |
| H | -3.07280800 | -3.94546600 | 0.91938000 |
| H | -0.33531000 | -4.38724300 | 1.19934000 |
| C | -1.12399900 | -1.55196300 | -0.41688500 |
| N | -2.37499300 | -2.13171500 | -0.10105300 |
| N | -0.21693700 | -2.48333300 | 0.11702300 |
| C | -3.69949300 | -1.64061300 | -0.52087600 |
| C | 1.24790000 | -2.40948100 | -0.00631300 |
| C | 3.42688900 | 1.86795700 | 0.77201000 |
| C | 3.43277000 | 1.06554200 | -0.39329400 |
| C | 2.20882000 | 0.59720100 | -0.89537300 |


| N | 0.97741900 | 0.88509900 | -0.33053200 |
| :--- | ---: | ---: | ---: |
| C | 0.96997600 | 1.67776500 | 0.80146300 |
| C | 2.18383600 | 2.16179700 | 1.36275600 |
| H | 4.36341800 | 0.79661500 | -0.90141500 |
| H | 2.16970300 | -0.04952400 | -1.77338800 |
| H | 2.13442500 | 2.77905900 | 2.26654000 |
| H | 4.35787700 | 2.24712800 | 1.20479700 |
| C | -0.31786900 | 2.13955800 | 1.48146200 |
| H | -0.12289500 | 2.16084300 | 2.57101600 |
| H | -0.48082100 | 3.20213900 | 1.20687200 |
| C | -1.62391200 | 1.35103300 | 1.29024000 |
| H | -2.39120100 | 1.80678500 | 1.95413400 |
| H | -1.48736400 | 0.29892200 | 1.59372000 |
| N | -2.12637400 | 1.32875000 | -0.13207700 |
| C | -2.51192400 | 2.68593200 | -0.66086400 |
| H | -3.39424600 | 3.06942000 | -0.10663200 |
| H | -1.67707200 | 3.38469100 | -0.48695000 |
| C | -2.82196800 | 2.61887800 | -2.17514100 |
| H | -3.47261100 | 3.46978600 | -2.44377600 |
| H | -3.39143300 | 1.69229700 | -2.39175800 |
| C | -1.58245700 | 2.66353200 | -3.04915000 |
| C | -1.46939100 | 3.63702300 | -4.07064800 |
| N | -0.57676100 | 1.73342000 | -2.83517400 |
| C | -0.32729000 | 3.68455800 | -4.89826300 |
| H | -2.28735200 | 4.35047100 | -4.21259800 |
| C | 0.53500300 | 1.78844700 | -3.64991600 |
| C | 0.69581500 | 2.73574900 | -4.67547100 |
| H | -0.24196400 | 4.43197600 | -5.69315600 |
| H | 1.28700400 | 1.02009500 | -3.47225100 |
| H | 1.60071100 | 2.71396000 | -5.28934300 |
| H | -3.56595500 | -0.96611200 | -1.38173100 |
| H | -4.32315600 | -2.49110300 | -0.84378800 |
| H | -4.22273600 | -1.12621200 | 0.30839800 |
| H | 1.50742200 | -2.10569200 | -1.03170300 |
| H | 1.67395800 | -1.68580500 | 0.70784300 |
| H | 1.67589800 | -3.40526500 | 0.19308100 |
| H | -2.98114100 |  |  |
|  |  |  | 0.74857000 |

## 4. References

[1] K. Jeong, S. Min, H. Chae, S. K. Namgoong, Magn. Reson. Chem. 2018, 56, 1089-1093.
[2] A. W. J. Logan, T. Theis, J. F. P. Colell, W. S. Warren, S. J. Malcolmson, Chem. - A Eur. J. 2016, 22, 10777-10781.
[3] H. Zeng, J. Xu, J. Gillen, M. T. McMahon, D. Artemov, J.-M. Tyburn, J. A. B. Lohman, R. E. Mewis, K. D. Atkinson, G. G. R. Green, S. B. Duckett, P. C. M. van Zijl, J. Magn. Reson. 2013, 237, 73-78.
[4] S. Glöggler, R. Müller, J. Colell, M. Emondts, M. Dabrowski, B. Blümich, S. Appelt, Phys. Chem. Chem. Phys. 2011, 13, 13759.
[5] R. W. Adams, J. A. Aguilar, K. D. Atkinson, M. J. Cowley, P. I. P. Elliott, S. B. Duckett, G. G. R. Green, i. G. Khazal, J. Lopez-Serrano, D. C. Williamson, Science (80-. ). 2009, 323, 1708-1711.
[6] M. J. Cowley, R. W. Adams, K. D. Atkinson, M. C. R. Cockett, S. B. Duckett, G. G. R. Green, J. A. B. Lohman, R. Kerssebaum, D. Kilgour, R. E. Mewis, J. Am. Chem. Soc. 2011, 133, 6134-6137.
[7] P. J. Rayner, P. Norcott, K. M. Appleby, W. Iali, R. O. John, S. J. Hart, A. C. Whitwood, S. B. Duckett, Nat. Commun. 2018, 9, 1-11.
[8] K. Ladomenou, G. Charalambidis, A. G. Coutsolelos, Tetrahedron 2007, 63, 2882-2887.
[9] T. Lu, F. Chen, J. Comput. Chem. 2012, 33, 580-592.

