

# Application of the electropolymerized poly(3,4-ethylenedioxythiophene) sorbent for solid-phase microextraction of bisphenols

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## Supplement

Table S1. Elemental analysis data for lignosulfonate LS DP 841.

| Lignosulfonate<br>symbol   | Source                       | Mn / Da | Mw / Da | % organic S | % phenolic OH |
|----------------------------|------------------------------|---------|---------|-------------|---------------|
| LS DP 841<br>(sodium salt) | softwood<br>(ultrafiltrated) | 4800    | 47900   | 6.0         | 2.2           |

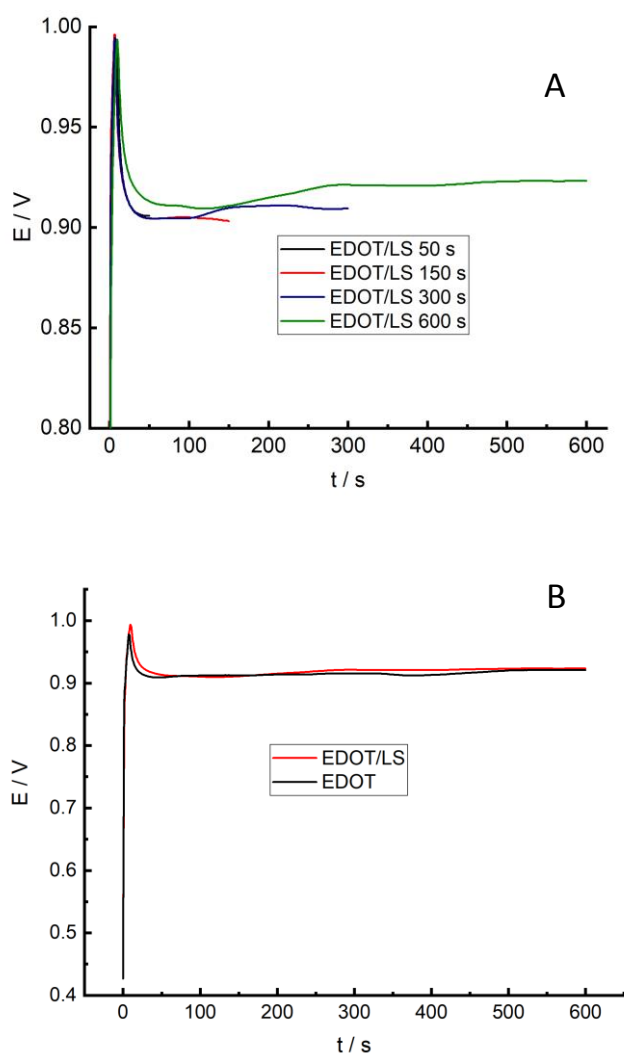


Figure S1. (A) Galvanostatic polymerization profiles for  $1 \text{ mg mL}^{-1}$  EDOT in presence of  $1 \text{ mg mL}^{-1}$  LS recorded at different times. Supporting electrolyte:  $0.1 \text{ M HClO}_4/\text{acetonitrile}$  (9:1 v/v). Applied current  $1.4 \text{ mA}$ . (B) Galvanostatic curves obtained for  $1 \text{ mg mL}^{-1}$  EDOT in the absence and presence of  $1 \text{ mg mL}^{-1}$  LS.

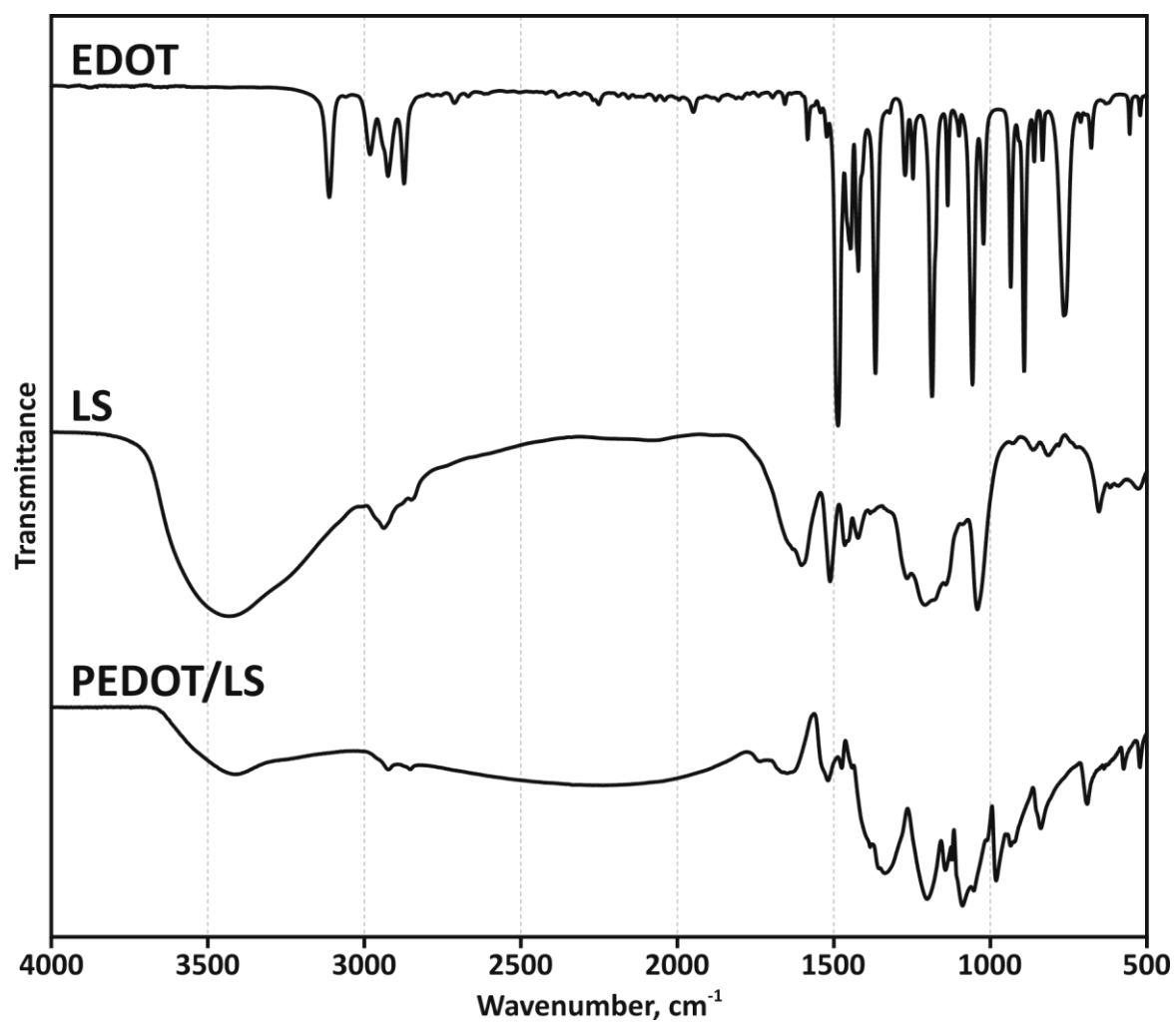


Figure S2. Fourier transform infrared spectra of EDOT, LS, and PEDOT/LS.

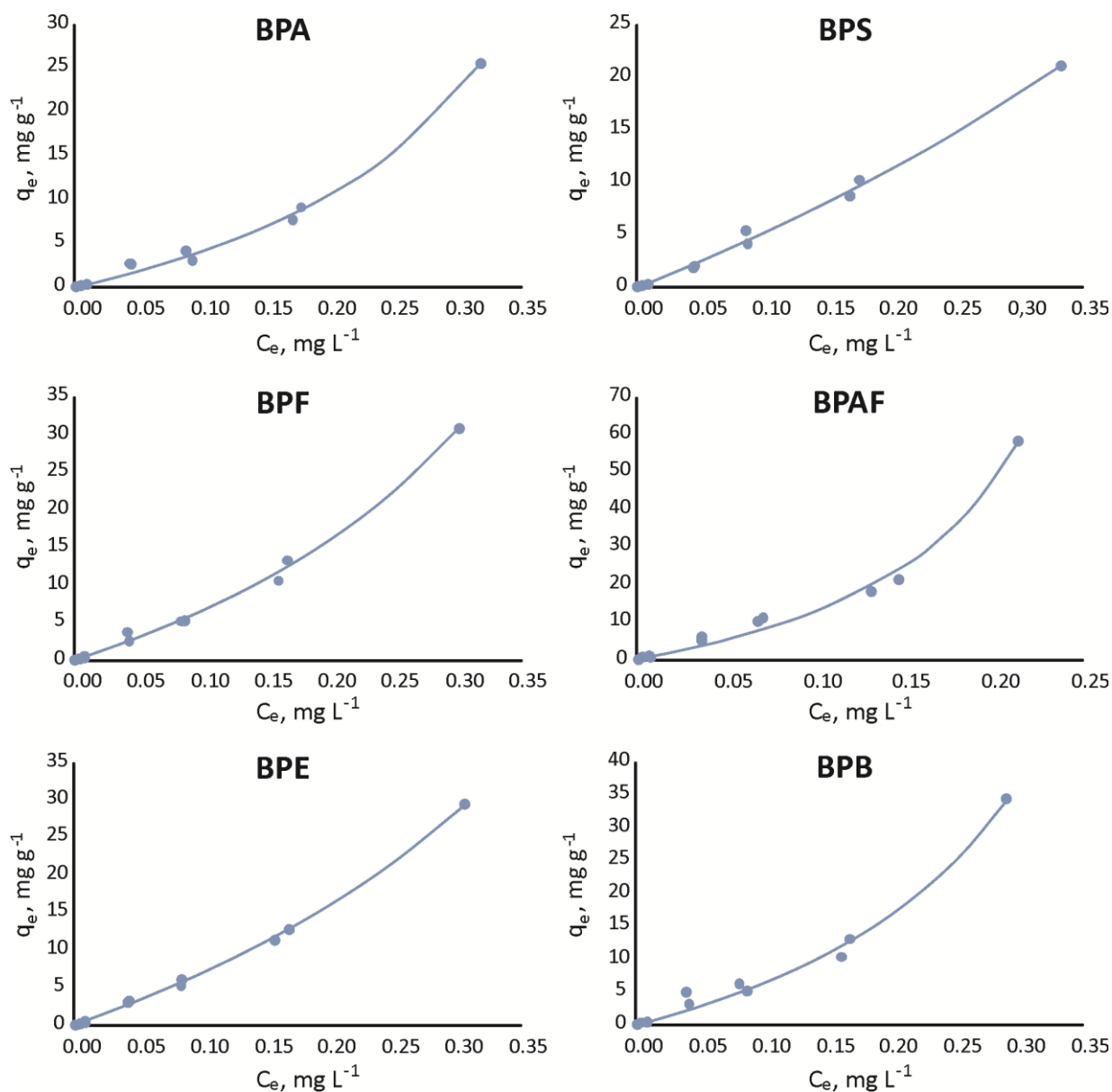


Figure S3. The adsorption isotherms obtained for the tested bisphenols (BPA – bisphenol A, BPS – bisphenol S, BPF – bisphenol F, BPAF – bisphenol AF, BPE – bisphenol E, BPB – bisphenol B).  $C_e$  – the equilibrium concentration in the water phase,  $q_e$  – the amount of adsorbed bisphenol per one gram of sorbent.