

Supplementary information

Complete Capillary Electrophoresis Process on a Drone: Towards a Flying Micro-lab

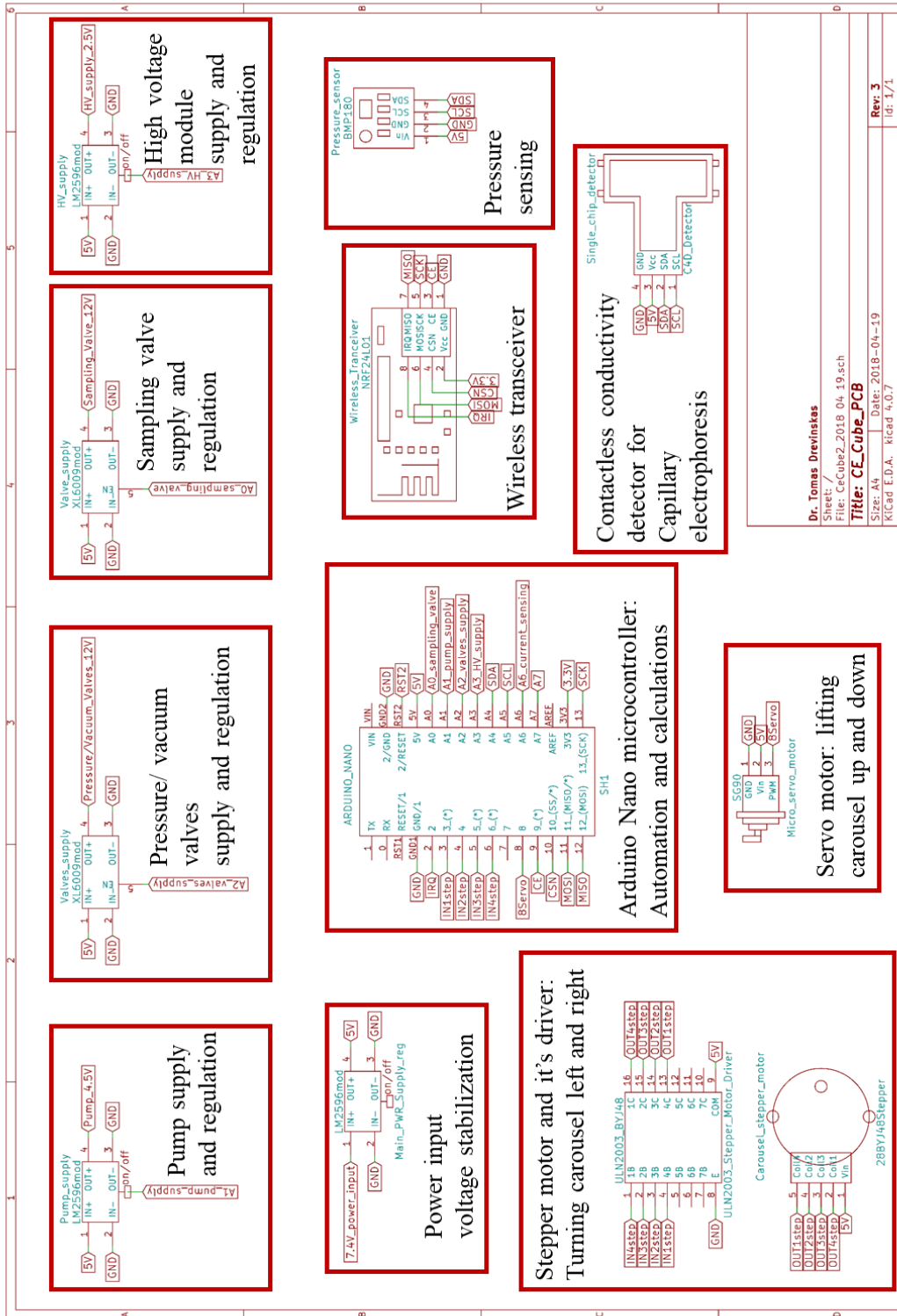
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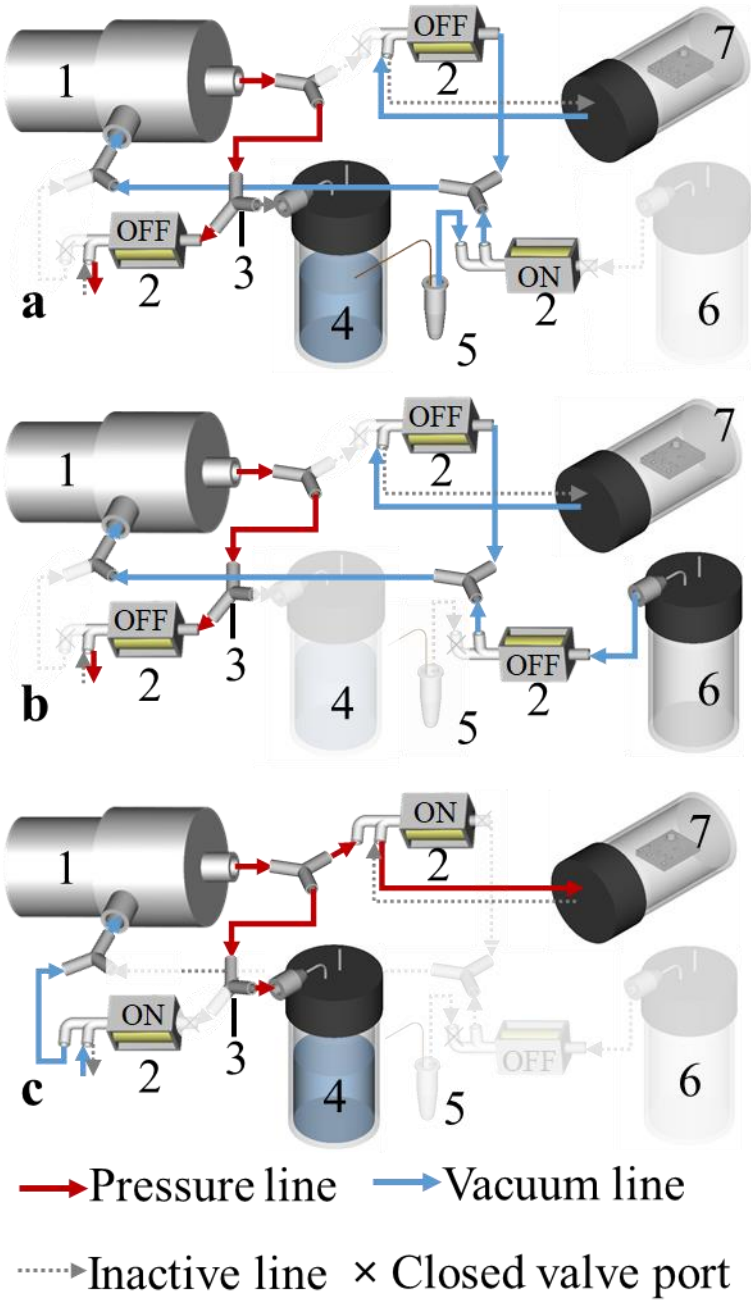
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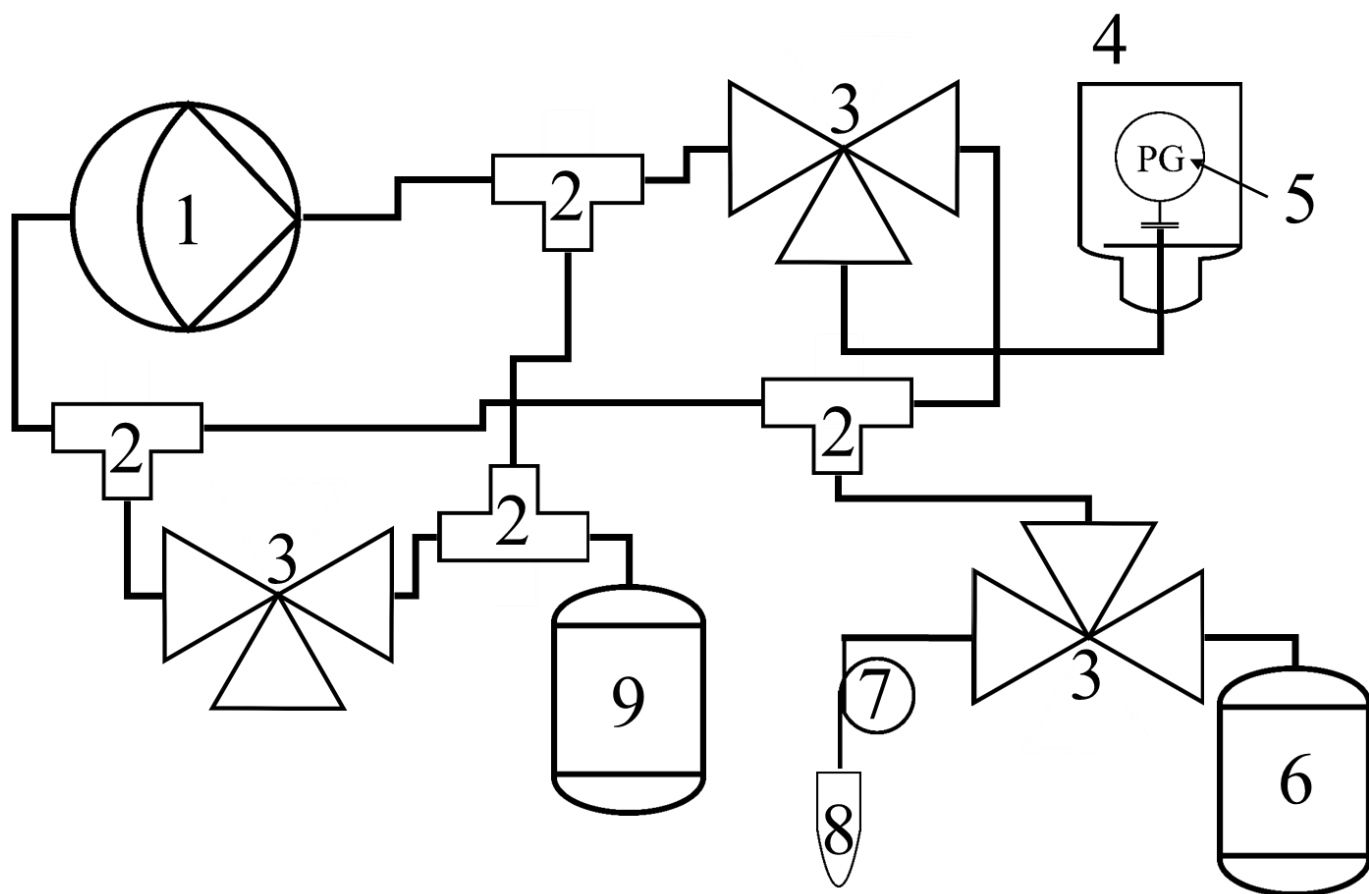
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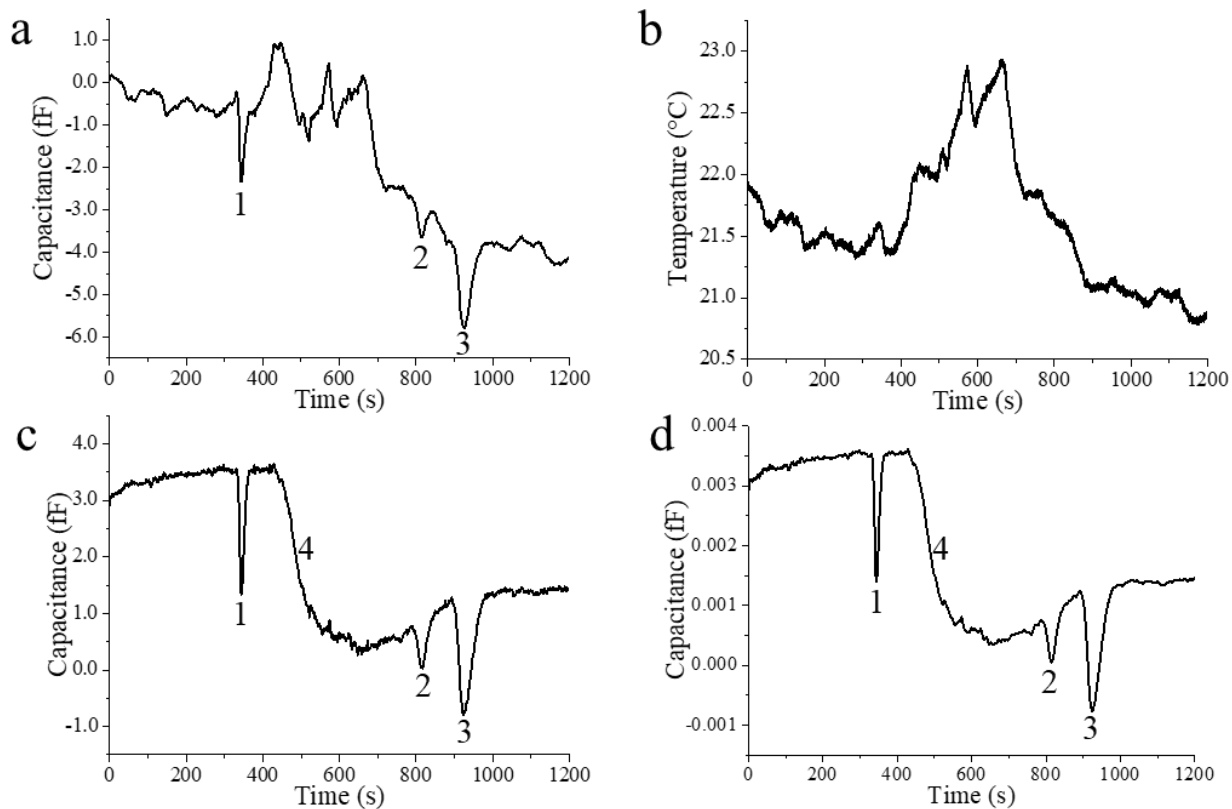
Supplementary Fig. S1. Schematic electrical wiring diagram. Designed using KiCAD.



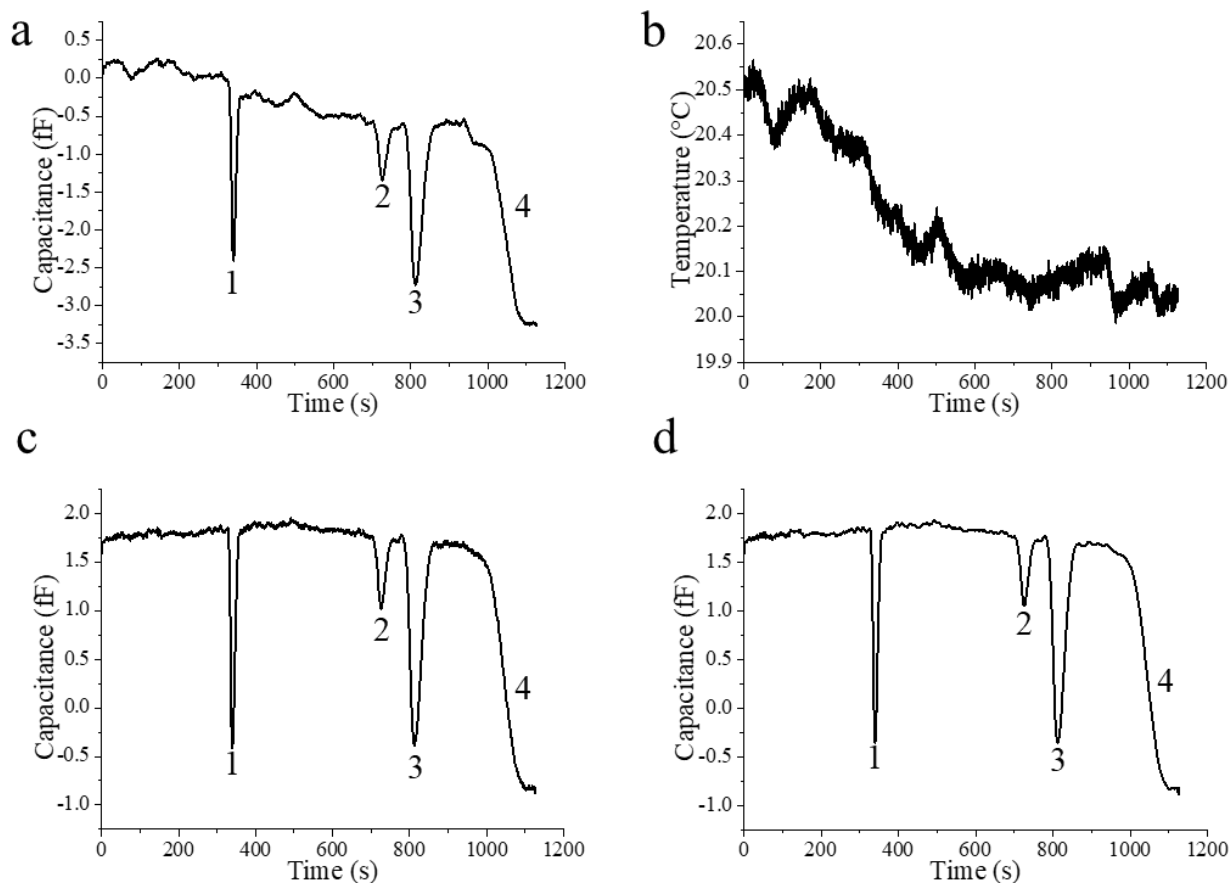
Supplementary Fig. S2. Schematic diagram of pressure, vacuum and liquid distribution system: **(a)** sampling mode, sample collection procedure from the air to the vial, **(b)** vacuum mode, capillary flushing procedure from the vial, **(c)** pressure mode, separation capillary flushing procedure from the BGE bottle. Numbers: 1 – pressure/ vacuum pump, 2 – 3-port, 2-way air valve, 3 – air split, 4 – BGE bottle, 5 – sample vial, 6 – waste bottle, 7 – bottle with pressure sensor



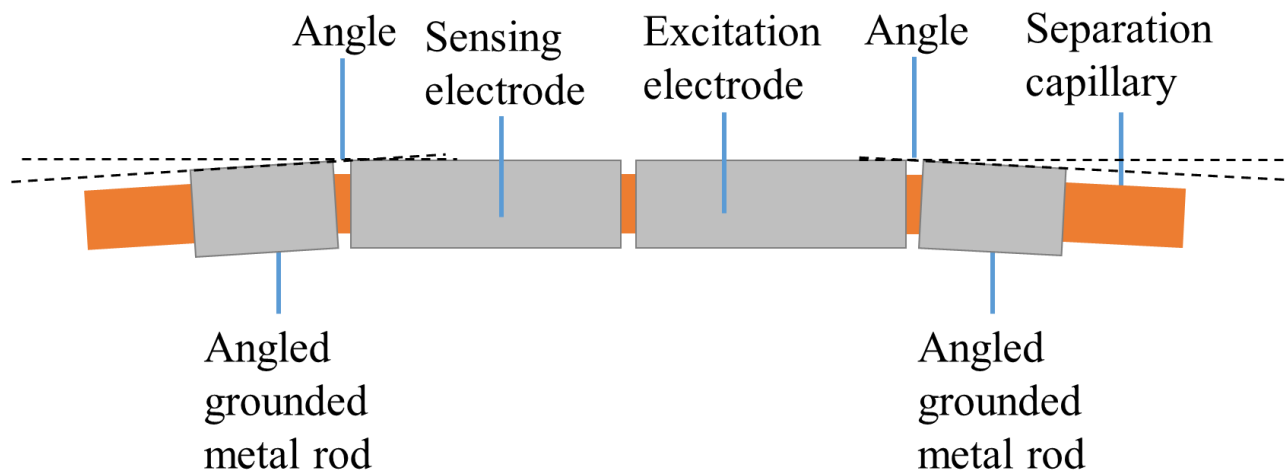
Supplementary Fig. S3 P&ID diagram of a pressure/ vacuum distribution system. Markings: (1) – mini diaphragm 6 V pressure/ vacuum pump, (2) 3-port air split, (3) 3-port 2-way solenoid valve, (4) Pressure/ vacuum gas bottle, (5) pressure sensor, (6) waste bottle, (7) fused silica capillary, (8) sample vial, (9) background electrolyte bottle.



Supplementary Fig. S4. Demonstration of signal compensation and conditioning, when analysis performed on a hovering drone. (a) Original electropherogram, (b) temperature change during analysis, (c) temperature-compensated electropherogram, (d) sensitivity-enhanced electropherogram. Peaks: 1 – NH_4^+ , 2 – DEA, 3 – TEA, 4 – system valley. Average wind speed 7 m/ s, gusts up to 10 m/ s. Sampling – 32 min. Added volatile compounds: no more than 1.0 ppm NH_3 , 1.3 ppm DEA, 1.0 ppm TEA. Separation conditions: BGE – 500 mM CH_3COOH , injection at $10\text{s}\times 20$ kPa, L_{tot} 30 cm, L_{eff} 23.5 cm, separation voltage potential + 4.0 kV, detection – C4D at 3.3 V 32 kHz square wave.



Supplementary Fig. S5. Demonstration of signal compensation and conditioning, when analysis performed on a landed drone. (a) Original electropherogram, (b) temperature change during analysis, (c) temperature-compensated electropherogram, (d) sensitivity-enhanced electropherogram. Peaks: 1 – NH_4^+ , 2 – DEA, 3 – TEA, 4 – system valley. Average wind speed 7 m/s, gusts up to 10 m/s. Sampling – 32 min. Added volatile compounds: no more than 1.0 ppm NH_3 , 1.3 ppm DEA, 1.0 ppm TEA. Separation conditions: BGE – 500 mM CH_3COOH , injection at $10\text{s} \times 20$ kPa, L_{tot} 30 cm, L_{eff} 23.5 cm, separation voltage potential + 4.0 kV, detection – C4D at 3.3 V 32 kHz square wave.



Supplementary Fig. S6. Schematic diagram representing how the capillary was fixed in order to prevent vibration caused effects on the detection

Supplementary Table S1. Calibration conditions and results

No	Analyte	Calibration equation y=	R ²	Average Precision (%)	Calibration Range (mM)	Background electrolyte	Injection	Ltot (cm)*	Leff (cm)*
1	K ⁺	0.3335x	0.9964	<10.3	0.0058-0.3700	500 mM CH ₃ COOH	10s×20kPa	30	23.5
2	Ca ²⁺	0.8642x	0.9913	<9.6	0.0059-0.3818				
3	Na ⁺	0.6226x	0.9953	<6.5	0.0039-0.2500				
4	NH ₄ ⁺	0.3363x	0.9913	<4.5	0.0133-0.3127				
5	Diethylamine	0.7199x	0.9984	<2.2	0.0151-0.9680				
6	Triethylamine	0.9175x	0.9967	<3.3	0.0112-0.7165				
7	BSA*	18.202x	0.9925	<11.1	0.0005-0.0341				
8	Formic acid	0.1212x + 0.0021	0.9922	<11.6	0.0078-0.2500	150 mM AMP**	10s×20kPa	35	28.5
9	Acetic acid	0.2909x + 0.0023	0.9857	<10.3	0.0078-0.2500				
10	Methylphosphonic acid	0.2465x	0.9935	<3.8	0.0400-0.2000	10 mM MES*** / L- His****	5s×20kPa	35	28.5
11	Ethylphosphonic acid	0.0867x + 0.0022	0.9925	<8.0	0.0400-0.1000				
12	Propylphosphonic acid	0.0857x + 0.0028	0.9934	<8.8	0.0400-0.1000				
13	Ethyl methylphosphonate	0.0725x + 0.0011	0.9956	<12.6	0.0400-0.1000				
14	Pinacoyl methylphosphonate	0.0587x + 0.0007	0.9809	<6.2	0.0400-0.1000				

* BSA – Bovine serum albumin

- ** AMP - 2-Amino-2-methyl-1-propanol
- *** MES - 2-(N-morpholino)ethanesulfonic acid
- **** L-His - L-histidine

Supplementary Table S2. Sampling conditions

No	Analyte	Reagent concentration	Method	Concentration in the air (ppm)	Added	Sampling Time (min)	Determined in collected sample (mM)
1	CH ₃ COOH	99.9 %	Evaporation	79.3	10 µL	16	0.0647 ± 0.0026
2	CH ₃ COOH	99.9 %	Evaporation	15.9	2 µL	16	0.0229 ± 0.0013
3	HCOOH	99.9 %	Evaporation	120.3	10 µL	16	0.1074 ± 0.0114
4	HCOOH	99.9 %	Evaporation	24.0	2 µL	16	0.0490 ± 0.0036
5	NH ₄ ⁺	25 %	Evaporation	51.2	10 µL	4	0.0381 ± 0.0108
6	NH ₄ ⁺	25 %	Evaporation	51.2	10 µL	8	0.0985 ± 0.0070
7	NH ₄ ⁺	25 %	Evaporation	51.2	10 µL	16	0.1297 ± 0.0072
8	NH ₄ ⁺	0.5 %	Evaporation	1.0	10 µL	32	0.0184 ± 0.0009
9	DEA*	99.9 %	Evaporation	43.9	10 µL	4	0.0298 ± 0.0034
10	DEA	99.9 %	Evaporation	43.9	10 µL	8	0.0750 ± 0.0022
11	DEA	99.9 %	Evaporation	43.9	10 µL	16	0.0706 ± 0.0029
12	DEA	2.0 %	Evaporation	1.3	15 µL	32	0.0090 ± 0.0007
13	TEA**	99.9 %	Evaporation	32.5	10 µL	4	0.0209 ± 0.0035
14	TEA	99.9 %	Evaporation	32.5	10 µL	8	0.0577 ± 0.0037

15	TEA	99.9 %	Evaporation	32.5	10 μ L	16	0.1102 \pm 0.0038
16	TEA	2.0 %	Evaporation	1.0	15 μ L	32	0.0384 \pm 0.0042
17	K ⁺	32 mg/ L	Ultrasonic atomization	N/A***	N/A	4	0.0303 \pm 0.0020
18	K ⁺	32 mg/ L	Ultrasonic atomization	N/A	N/A	8	0.0305 \pm 0.0023
19	K ⁺	32 mg/ L	Ultrasonic atomization	N/A	N/A	16	0.0391 \pm 0.0044
20	Na ⁺	1727 mg/ L	Ultrasonic atomization	N/A	N/A	4	0.0750 \pm 0.0080
21	Na ⁺	1727 mg/ L	Ultrasonic atomization	N/A	N/A	8	0.2006 \pm 0.0104
22	Ca ²⁺	552 mg/ L	Ultrasonic atomization	N/A	N/A	4	0.0123 \pm 0.0020
23	Ca ²⁺	552 mg/ L	Ultrasonic atomization	N/A	N/A	8	0.0331 \pm 0.0024
24	BSA****	1 mg/ mL	Ultrasonic atomization	N/A	N/A	8	0.0034 \pm 0.0006
25	BSA	1 mg/ mL	Ultrasonic atomization	N/A	N/A	16	0.0076 \pm 0.0003

DEA* - Diethylamine

TEA** - Triethylamine

N/A*** - Not applicable

BSA**** - Bovine serum albumin

$$LOQ = \frac{10\sigma}{S} \text{ (Supplementary equation S1)}$$

¹Where LOQ is the limit of quantification, S is the slope of calibration curve and σ is the standard deviation of the response.

$$C_{mg/L} = \frac{V_s \times C_{\%} \times \rho_s}{V_{box} \times 100\%} \text{ (Supplementary equation S2)}$$

Where $C_{mg/L}$ – is the concentration of a substance in the air (mg/ L), V_s – added solution (μ L), $C_{\%}$ - percentage of substance solution (%), ρ_s – solution density (mg/ μ L), V_{box} – volume of the box (L)

$$C_{ppm} = \frac{m \times 10^3}{V} \times \frac{24.46}{MW} \text{ (Supplementary equation S3)}$$

²Where C_{ppm} - air concentration (ppm) by volume, at 25 °C and 760 mm Hg, m – actual mass of substance (mg), V – air volume (L), 24.46 – the volume (L) of 1 mole of gas (or evaporated substance) at 25 °C and pressure of 760 mm Hg, MW – molecular weight, grams/ mole.

References

- (1) Ich. ICH Topic Q2 (R1) Validation of Analytical Procedures : Text and Methodology. *International Conference on Harmonization*. 2005, p 17.
- (2) Air Concentration Calculations for Comparison to OSHA Standards. In *Niosh Manual of Analytical Methods (NMAM)*; Eller, P. M., Cassinelli, M. E., Eds.; U.S. Department of Health and Human Services: Cincinnati, Ohio, 1994; p A-2.