

Supporting Information

Tetraphenylethene probe based fluorescent silica nanoparticles for the selective detection of nitroaromatic explosives

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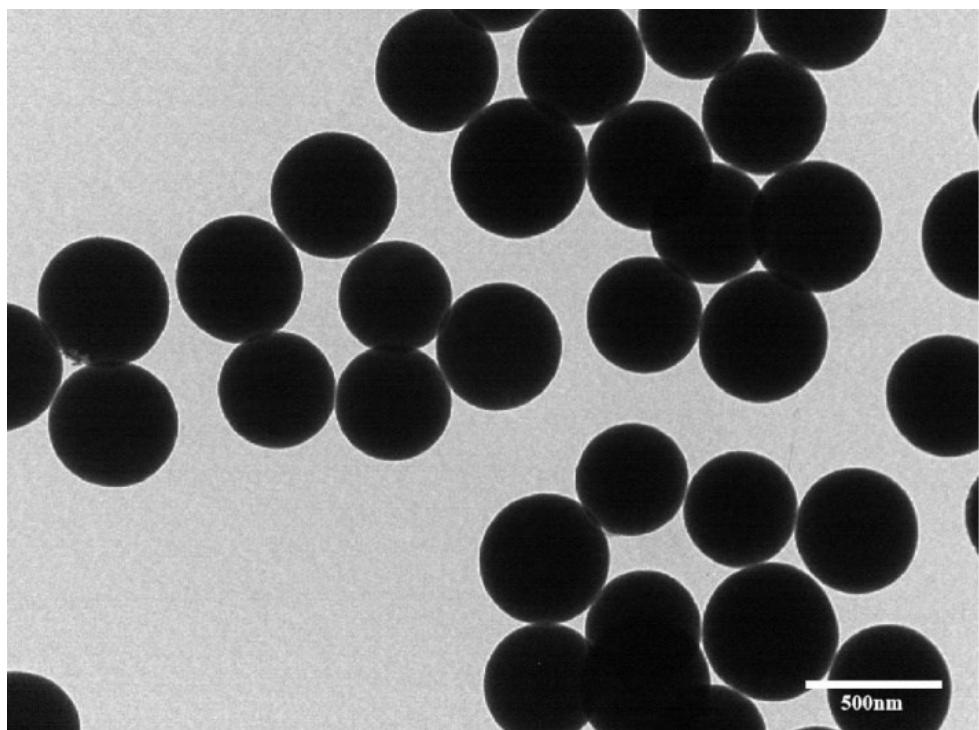


Fig. S1. TEM image of the prepared TPE-SiO₂ nanoparticles at 800 μM concentration of TPE-C2-2+ probe.

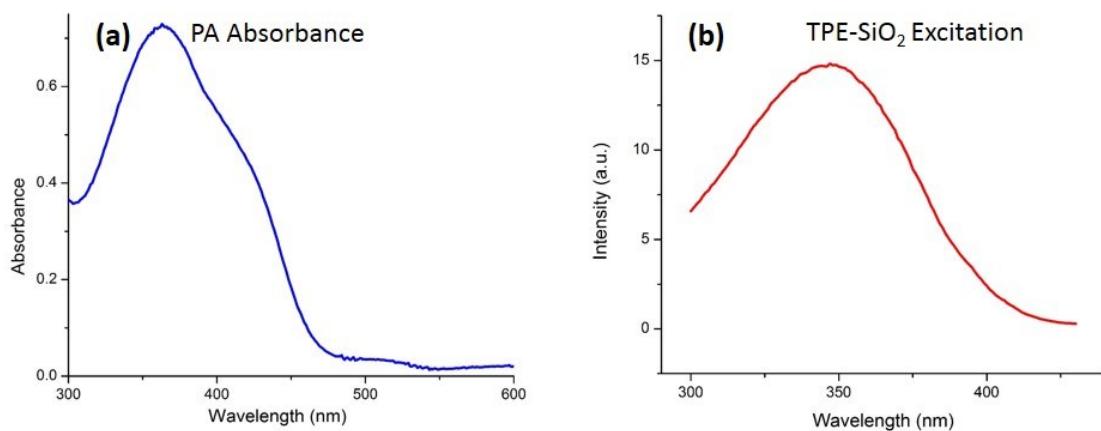


Fig. S2. (a) UV–vis light absorption spectrum of picric acid (PA). (b) Excitation spectrum of TPE-SiO₂ nanoparticles.

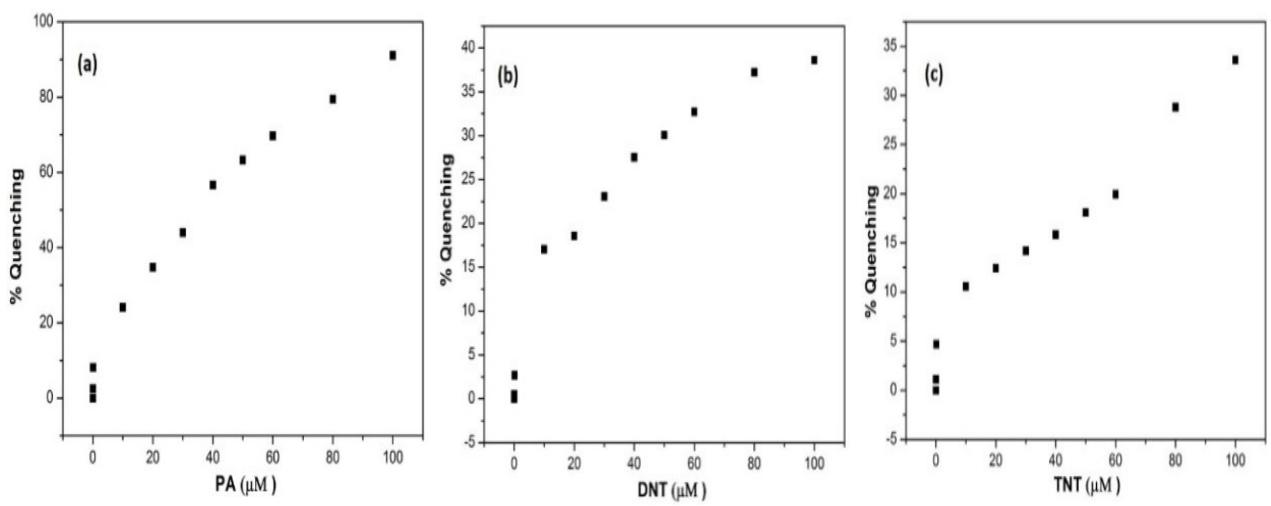


Fig. S3. Plots of % quenching of the TPE-SiO₂ fluorescent nanoparticles in the presence of different concentrations of the nitroaromatic explosives (a) PA, (b) DNT and (c) TNT.

Table S1. Comparison of the TPE-SiO₂ nanoparticles based fluorescence sensor and other reported fluorescent sensors for PA.

Materials	Detection Limit	Linear range (μM)	Reference
Fluorescent silica nanoparticles	1.01 ppm	—	1
Fluorescent AIE dots	32.1 nM	—	2
All inorganic Perovskite QDs	0.8nM	—	3
MoS ₂ QDs	0.095 μM	0.099-36.5	4
Pyrene-based probe	99 nM	—	5
carbon dots	1 μM	—	6
graphene QDs	0.3 μM	1-60	7
COF	0.289 μM	0-5	8
COF	—	5-60	9

COF	-	0-64	10
Gold nanoparticles	79 nM	-	11
Copper (I) metallogel	50 μ M	-	12
MOF	9.5 μ M	9.3-467.7	13
MOF	-	0-50	14
MOFs	0.0682 μ M 0.0694 μ M	0-60	15
covalent-organic polymers	4.37 μ M	-	16
conjugated polymer	-	-	17
conjugated polymers	0.5 μ M 1 μ M	0-220	18
conjugated polymer	0.11 μ M	0-20	19
nonconjugated polymer	0.026 μ M	0.05-70	20
small molecule	0.285 μ M	0-2.75 0-1	21
TPE-SiO ₂	0.01 μ M	0.1-50 μ M	This work

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