

Supplementary information

**Ultra-sensitive Sb speciation analysis in water samples by magnetic ionic liquid
dispersive liquid-liquid microextraction and multivariate optimization**

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Figure S1

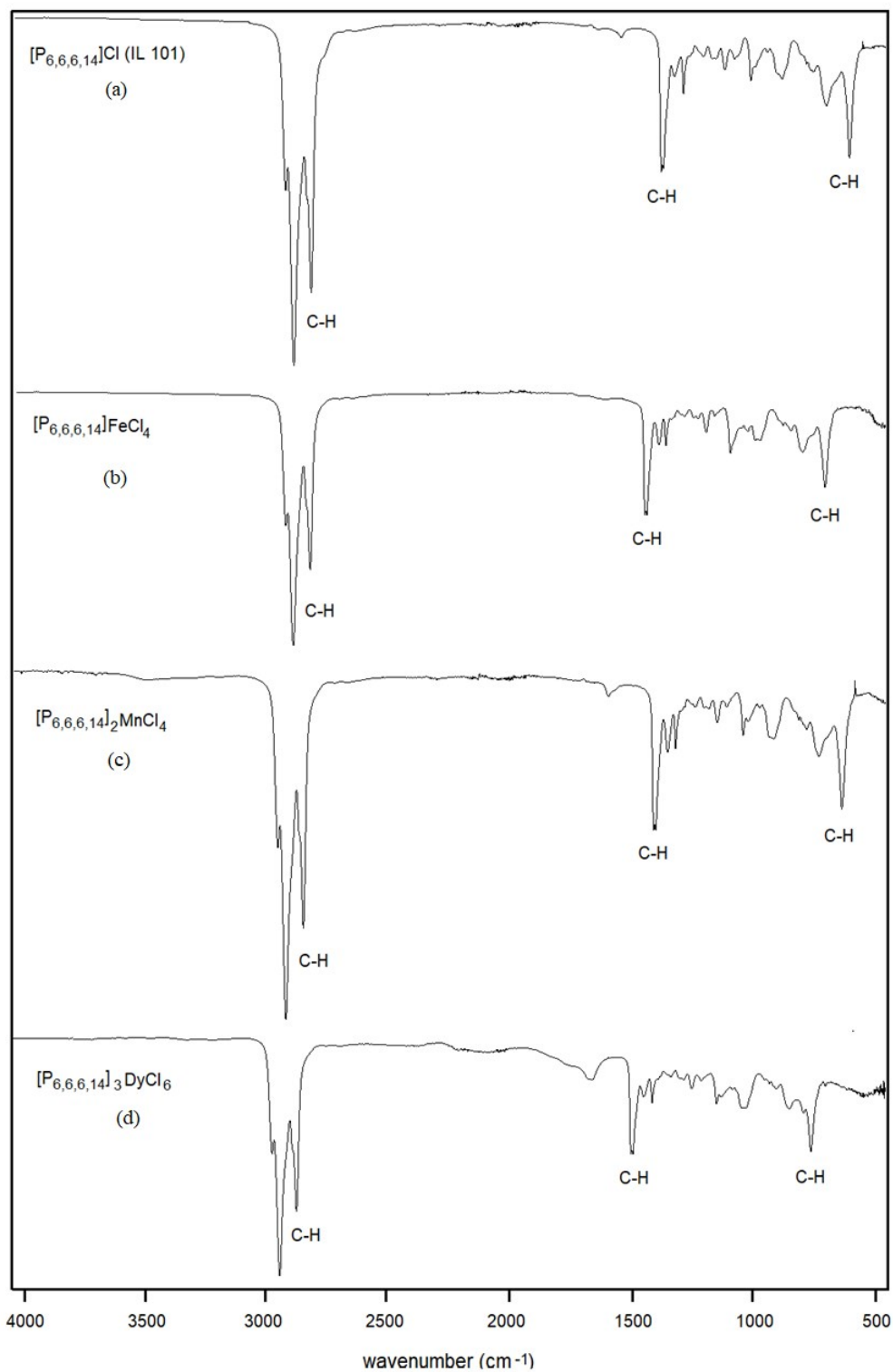


Fig. S1. FTIR spectra of (a) $[P_{6,6,6,14}]Cl$ and MILs studied in this work. (b) $[P_{6,6,6,14}]FeCl_4$, (c) $[P_{6,6,6,14}]_2MnCl_4$. (d) $[P_{6,6,6,14}]_3DyCl_6$.

Figure S2

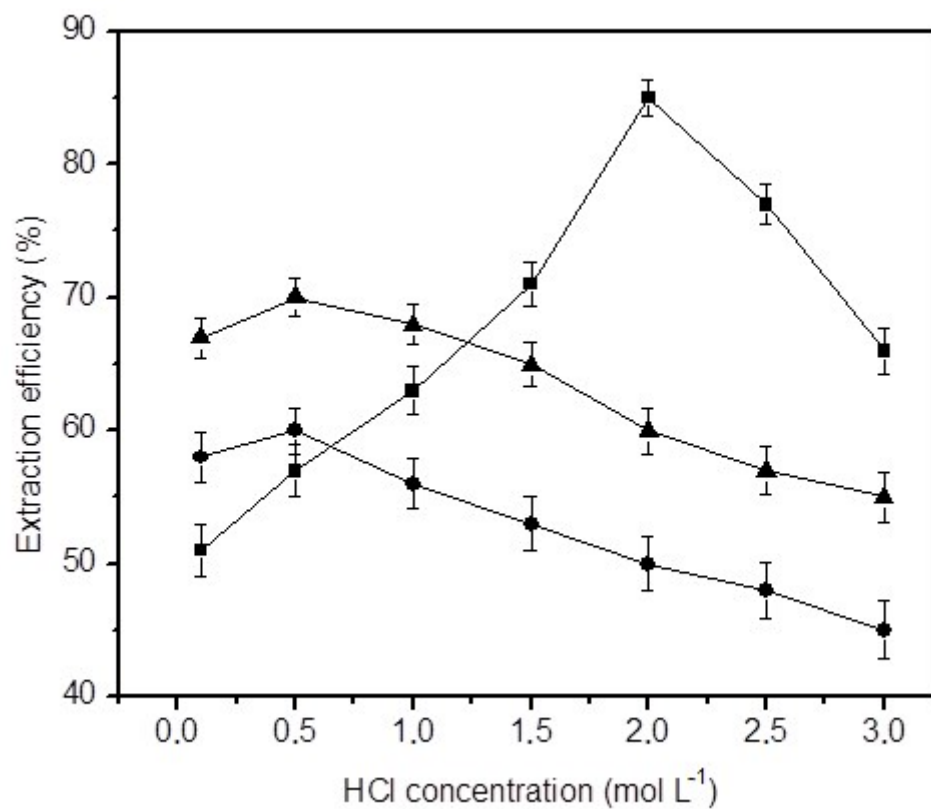


Fig. S2. Effect of hydrochloric acid concentration on Sb(III) extraction efficiency using [P_{6,6,6,14}]FeCl₄ (■), [P_{6,6,6,14}]₃DyCl₆ (▲) and [P_{6,6,6,14}]₂MnCl₄ (●) as extraction solvents. Other conditions were as mentioned in Table 1 (95% confidence interval, n = 6).

Figure S3

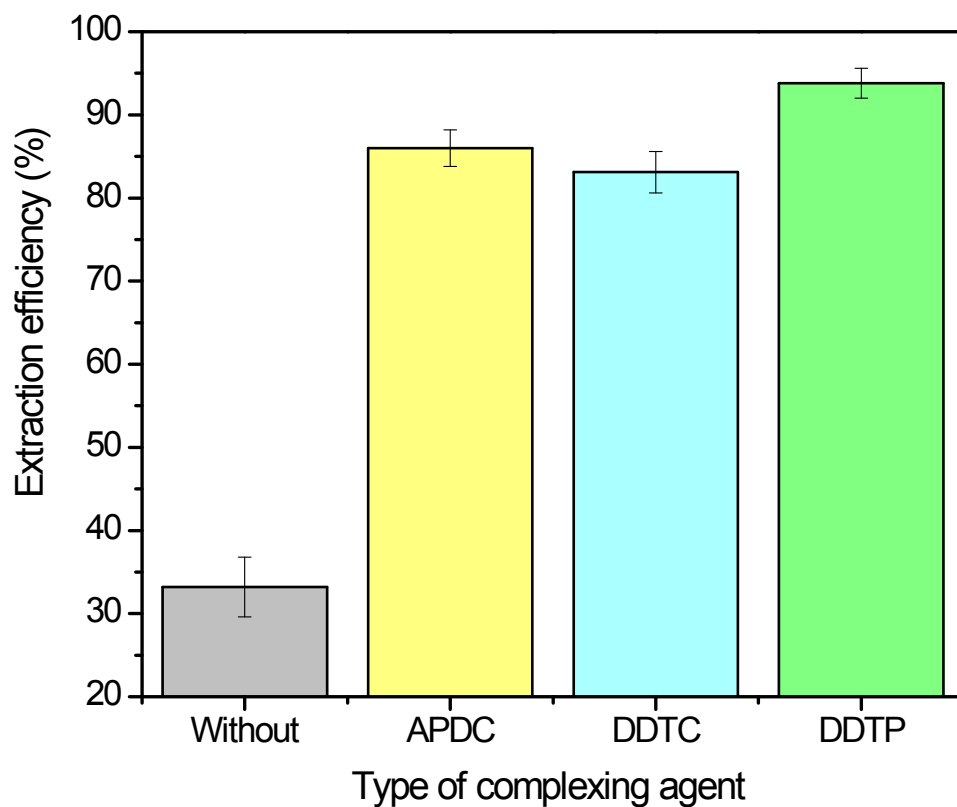


Fig. S3. Effect of different complexing agents on Sb(III) extraction efficiency (6 mL of sample at 2 mol L⁻¹ HCl). ■ Without, ■ APDC, ■ DDTC and ■ DDTP. Other conditions were as indicated in Table 1 (95% confidence interval, n = 6).

Figure S4

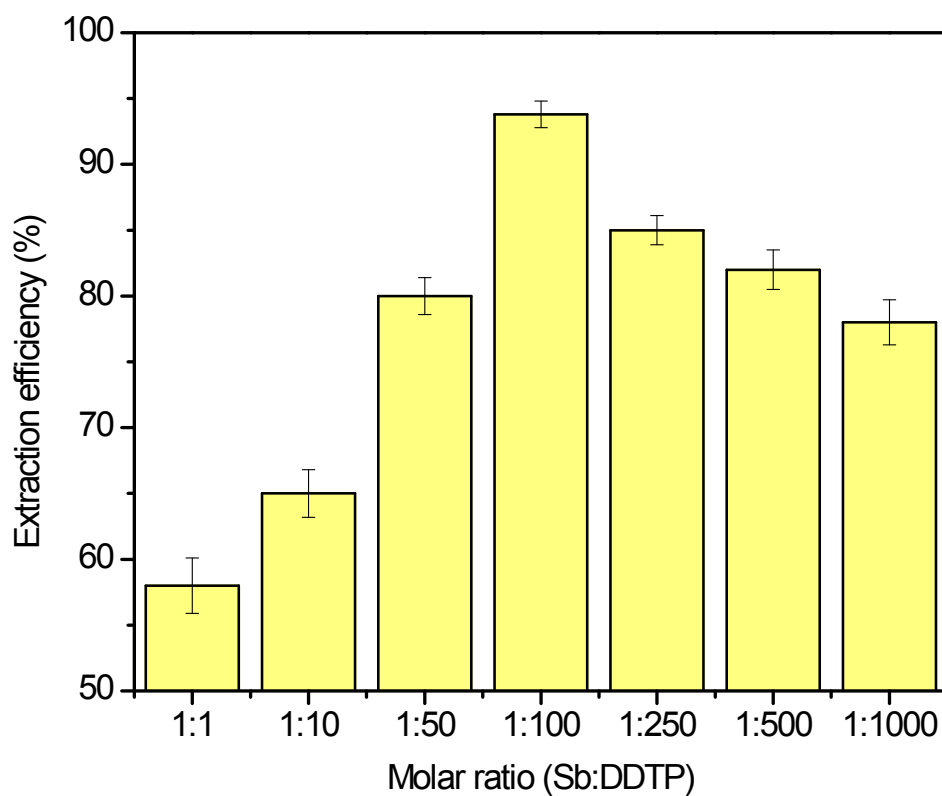
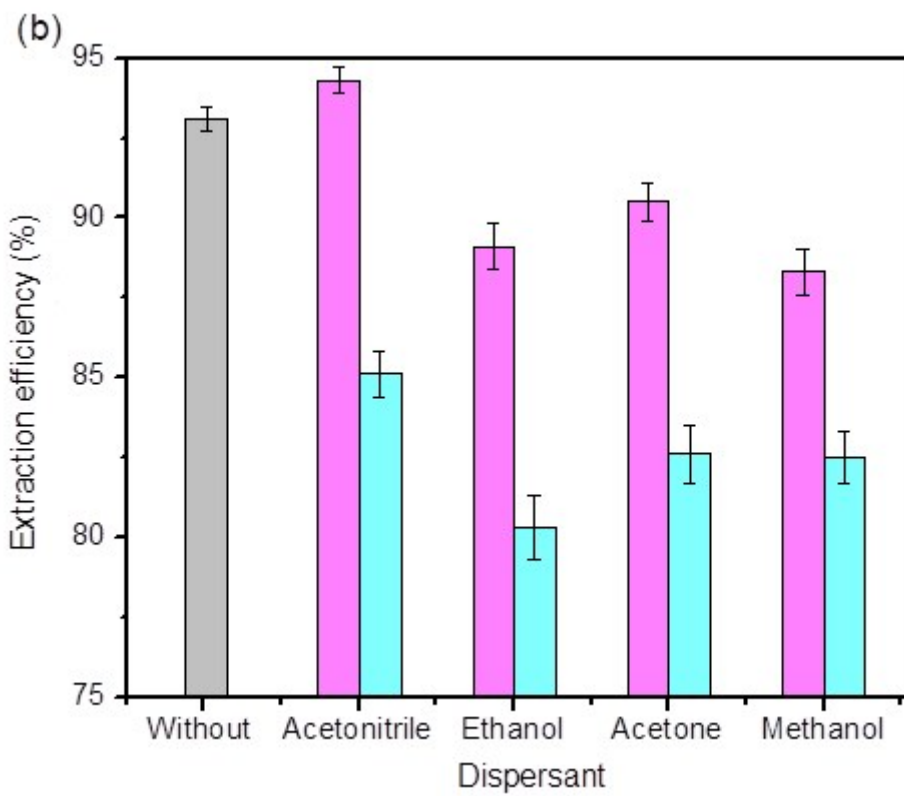
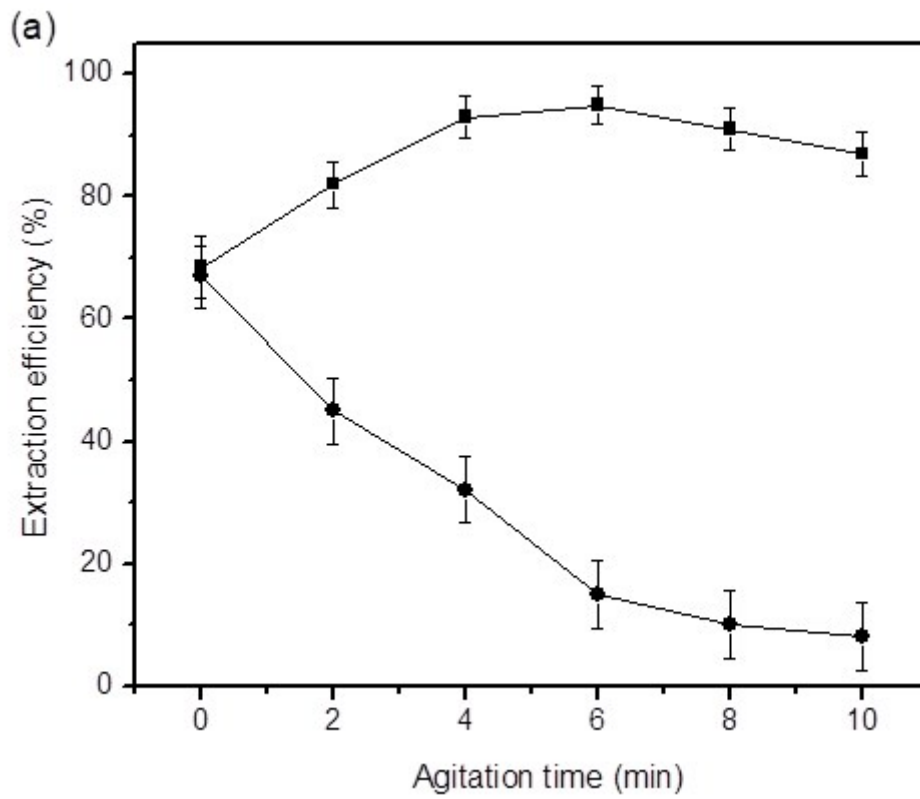


Fig. S4. Influence of Sb:DDTP molar ratio on Sb(III) extraction efficiency using $[P_{6,6,6,14}]FeCl_4$ as extracting phase. Other conditions were as mentioned in Table 1 (95% confidence interval, $n = 6$).

Figure S5



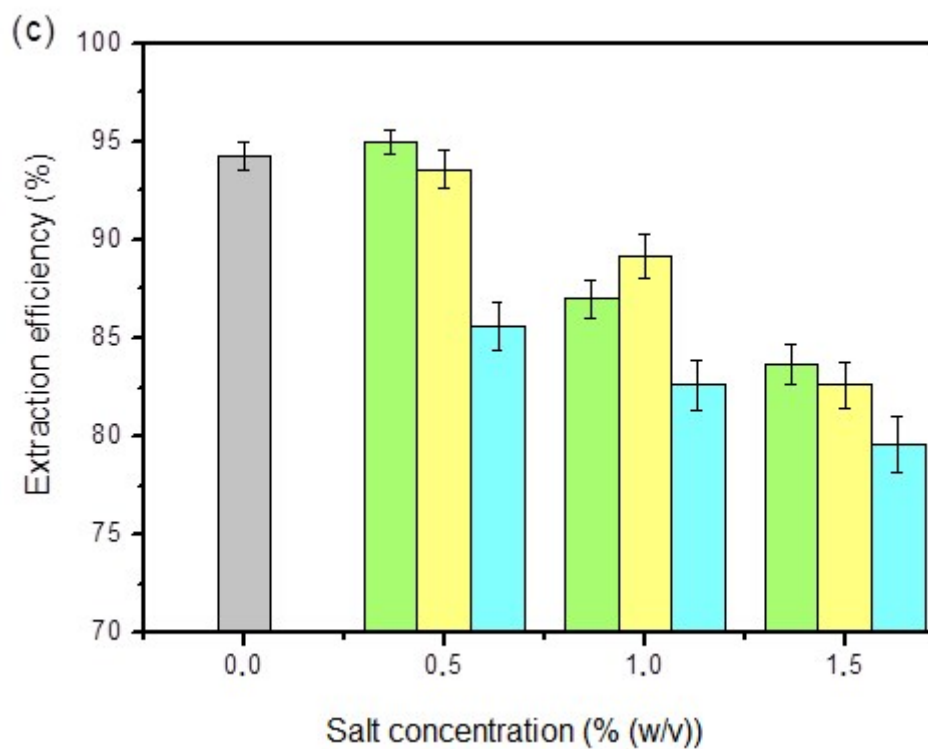


Fig. S5. Study of the categorical factors involved in the proposed method. (a) Type of agitation: (■) vortex and (●) ultrasound. (b) Dispersants volume: ■ 25 μ L and ■ 50 μ L. (c) Ionic strength: ■ NaCl, ■ NaNO₃ and ■ NaClO₄. Other conditions are shown in Table 1 (95% confidence interval, n = 6).

Figure S6

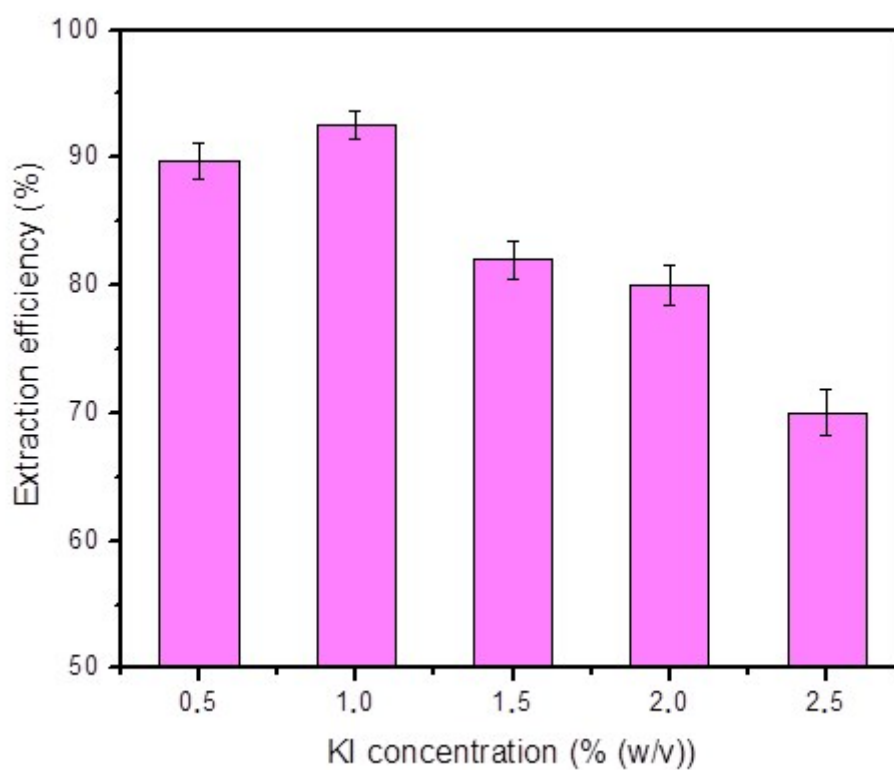


Fig. S6. Effect of KI concentration on Sb extraction efficiency of the MIL-DLLME method. Other conditions were as mentioned in Table 1 (95% confidence interval, n = 6).

Figure S7

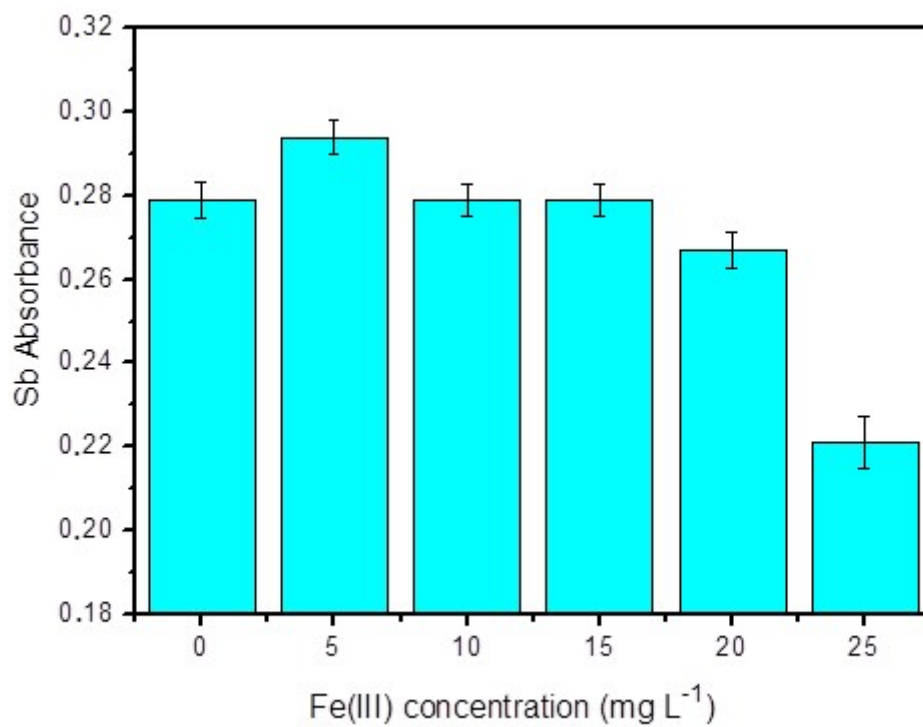


Fig. S7. Effect of Fe(III) concentration on Sb signal. A volume of 20 μL of 200 $\mu\text{g L}^{-1}$ Sb aqueous standard solution was injected. ETAAS instrumental conditions were as indicated in Table 1 (95% confidence interval, $n = 6$).