

Supporting Information

Peroxidase and UV-Triggered Oxidase Mimetic Activities of UiO-66-NH₂/Chitosan Composite Membrane for Antibacterial Property

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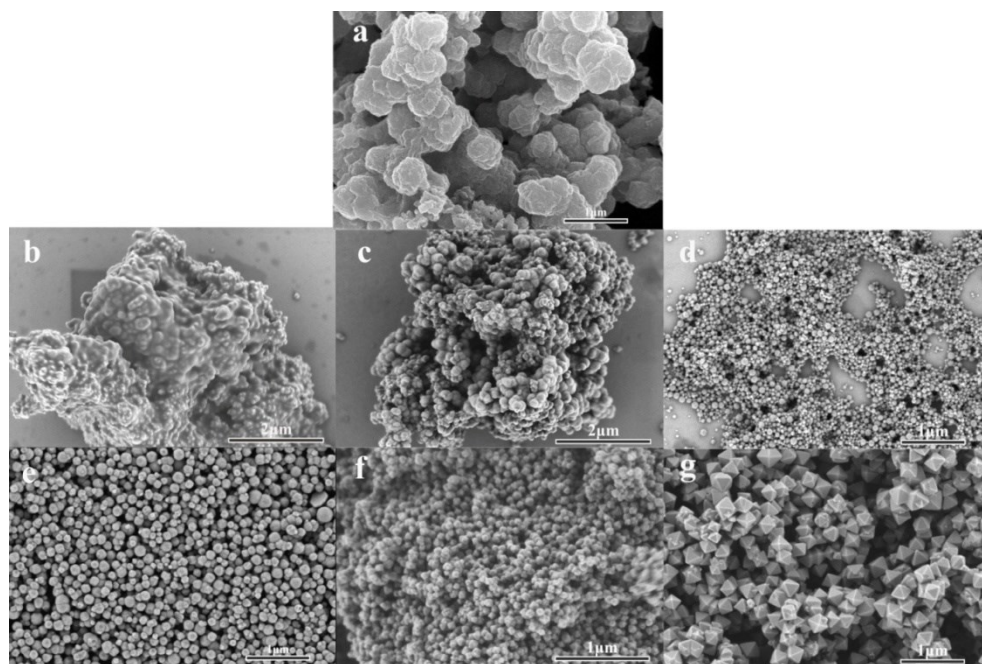


Figure S1. SEM image of UiO-66-NH₂ nanoparticles with the varied equivalent ratio of PABA to H₂ATA, a) 0, b) 10, c) 30, d) 50, e) 75, f) 100, and g) 150.

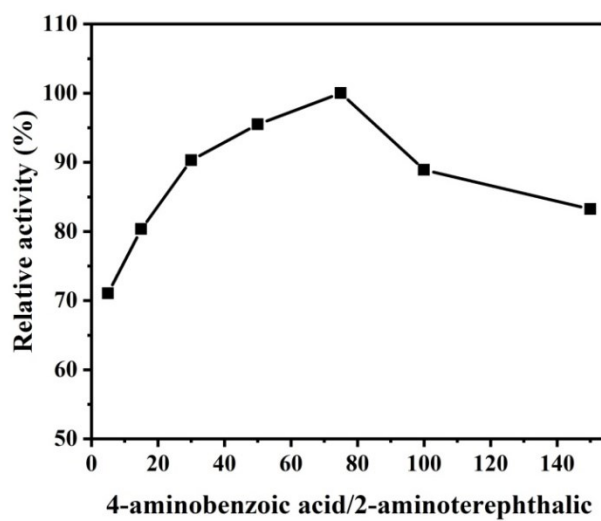


Figure S2. Catalytic oxidation of peroxidase substrate TMB·2HCl by H₂O₂ with different equivalent ratios of PABA to H₂ATA.

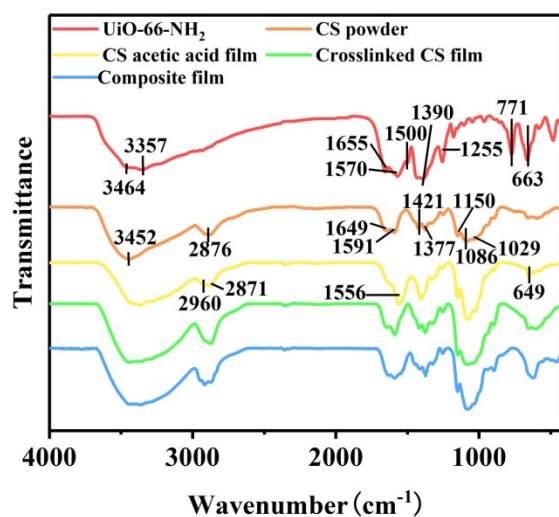


Figure S3. FTIR spectra of UiO-66-NH₂, CS powder, CS acetic acid film, Na₂SO₄ crosslinked CS film, and UiO-66-NH₂/CS composite film.

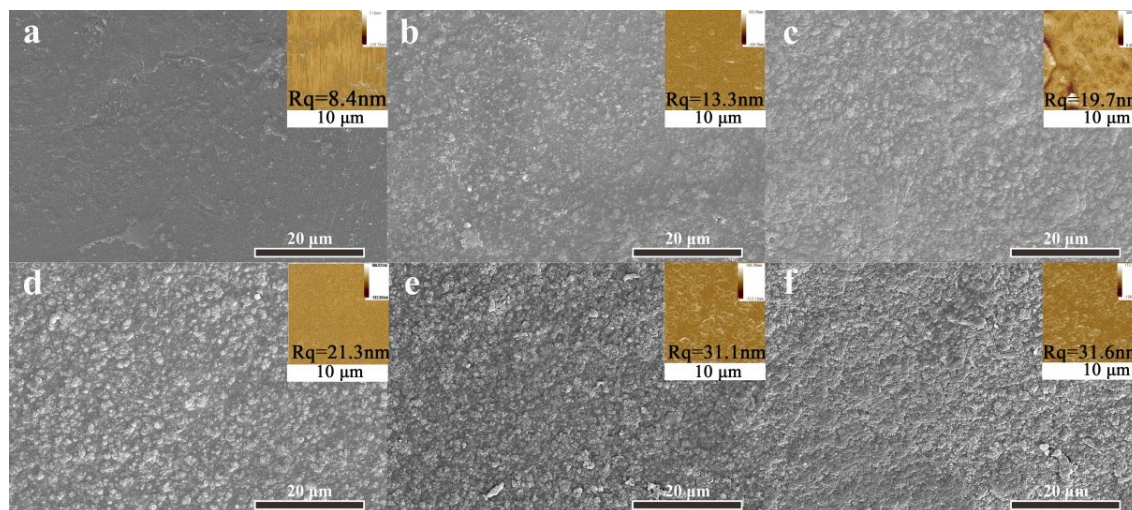


Figure S4. SEM ($\times 3000$) and AFM images of the surface morphology of the UiO-66-NH₂/CS composite membrane with the weight ratio of UiO-66-NH₂ to CS, a) 0:10, b) 1:10, c) 3:10, d) 5:10, e) 7:10, and f) 9:10.

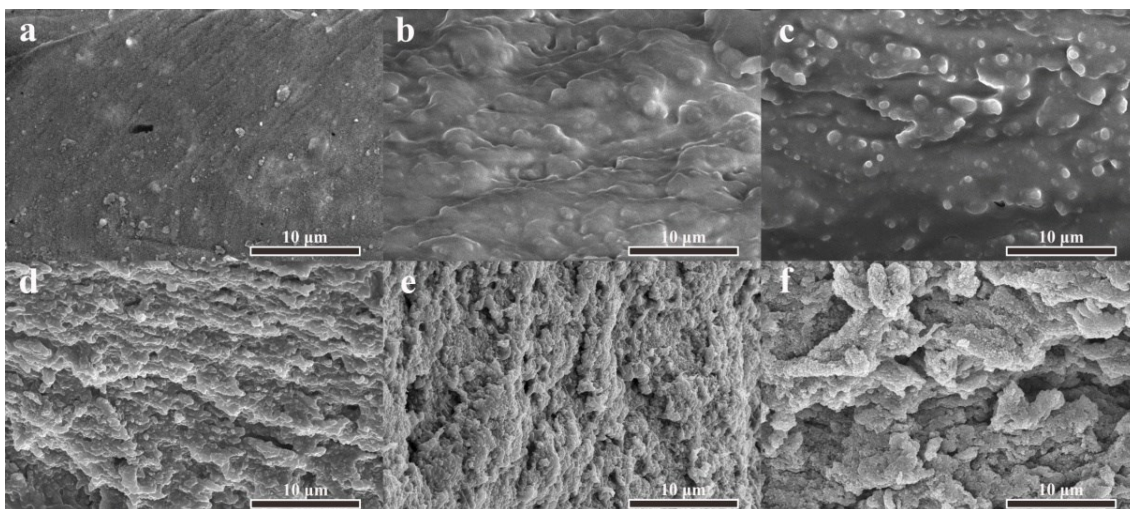


Figure S5. The cross-section SEM ($\times 6000$) images of the UiO-66-NH₂/CS composite membrane with the weight ratio of UiO-66-NH₂ to CS, a) 0:10; b) 1:10, c) 3:10, d) 5:10, e) 7:10 and f) 9:10.

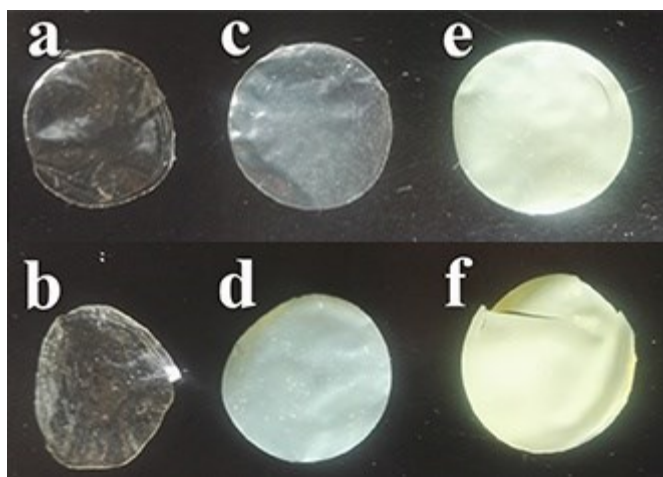


Figure S6. Photographs of the UiO-66-NH₂/CS composite membranes with the varied weight ratios of UiO-66-NH₂ NPs to CS, a-b) 0:10, c) 1:10, d) 3:10, e) 6:10, and f) 9:10.

Peroxidase-like activity measurements

Acetate buffer solution (0.1 M, pH=4.0, 4 mL), TMB·2HCl (1 mM), H₂O₂ (3 wt%, 45 μ L) and 0.25 cm² CS or UiO-66-NH₂/CS composite membrane were added and the reaction mixture was incubated at room temperature for 30 min followed by monitoring the formation of oxidation product of TMB·2HCl at 652 nm using UV-vis spectrophotometry.

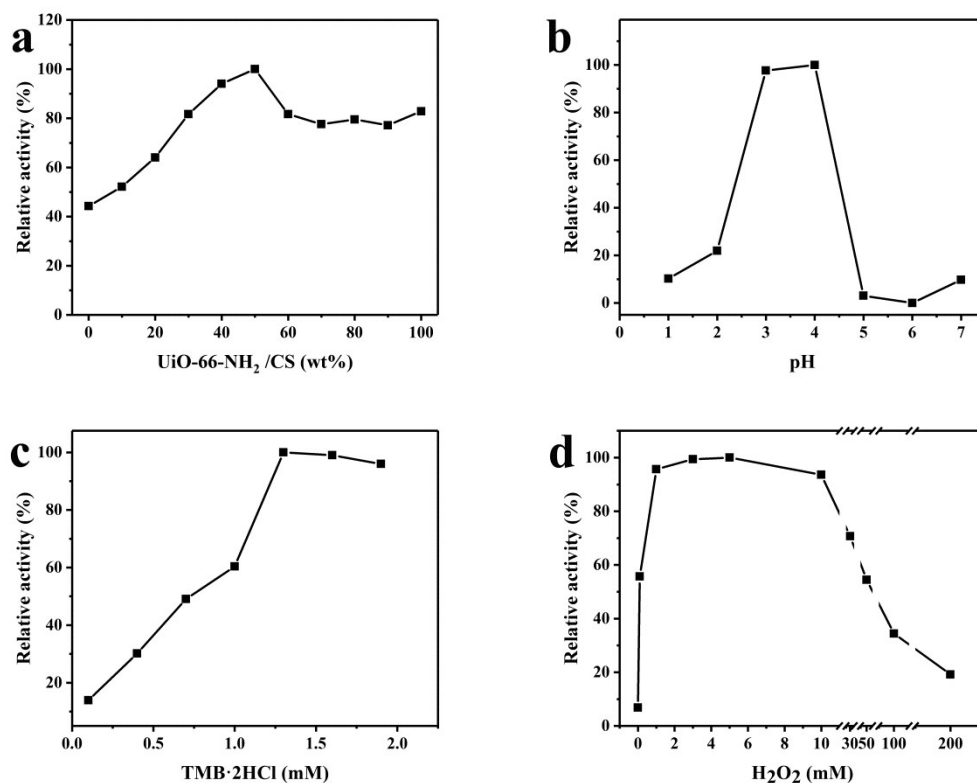


Figure S7. The catalytic oxidation of peroxidase substrate TMB·2HCl under different conditions: a) with different weight ratios of UiO-66-NH₂ to CS, b) under different pH conditions, c) with various TMB·2HCl amounts, and d) with various H₂O₂ amounts.

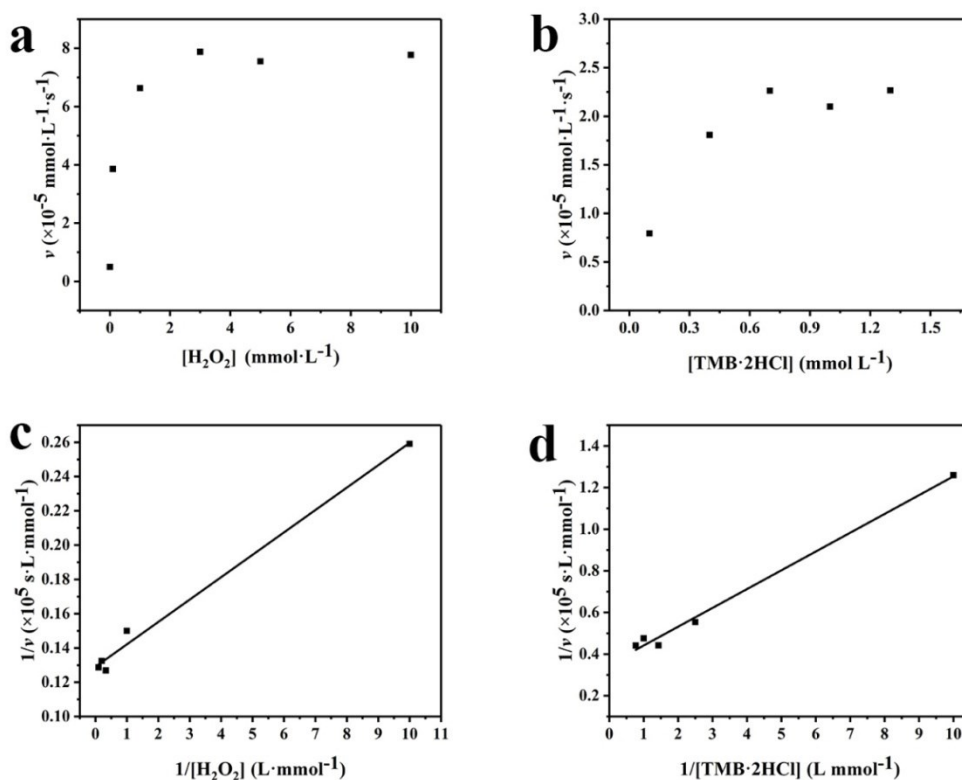


Figure S8. The kinetics of the reaction was measured using the UiO-66-NH₂/CS composite membrane ($r=0.25$ cm) and TMB·2HCl (1.3 mM) under different (a) H₂O₂, and (b) TMB·2HCl concentrations. Corresponding double-reciprocal plots of peroxidase-like activity of the UiO-66-NH₂/CS composite membrane at a fixed concentration of one substrate versus varying concentration of another substrate for (c) H₂O₂ or (d) TMB·2HCl.

Table S1. The Michaelis-Menten (K_m) constant and maximum reaction rate (V_{max}) of the UiO-66-NH₂/CS composite membrane.

Catalyst	K_m (mM)		V_{max} (10^{-8} M s ⁻¹)	
	TMB·2HCl	H ₂ O ₂	TMB·2HCl	H ₂ O ₂
UiO-66-NH ₂ /CS composite membrane	0.10	0.26	7.75	2.85
HRP ^[1]	0.17	10.9	41.8	58.5
GO-Fe ₃ O ₄ ^[2]	0.43	0.71	13.08	5.31
Cu-Ag/rGO ^[3]	0.63	8.63	4.26	7.02
Hemin@MOF ^[4]	10.9	0.07	8.98	6.07
MOF-808 ^[5]	0.08	1.06	3.12	1.39

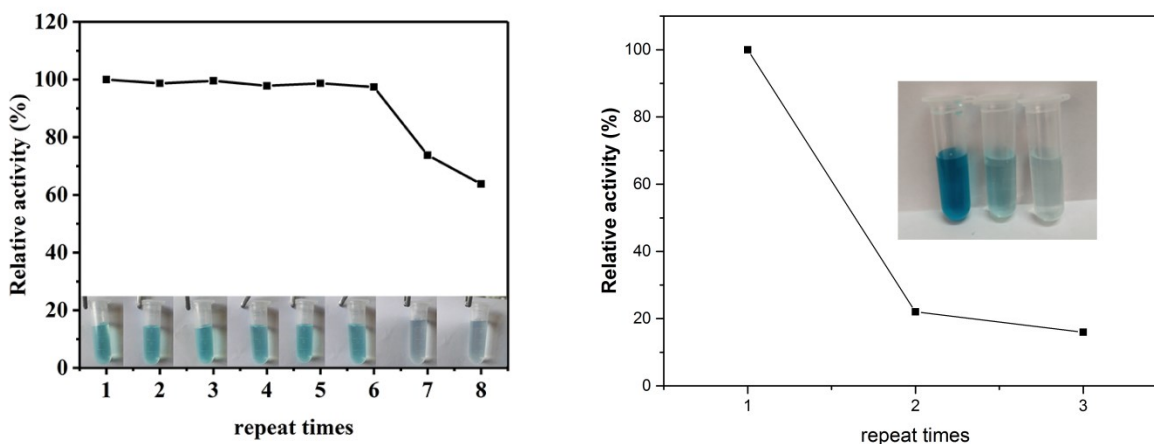


Figure S9. a) Photocatalytic stability of UiO-66-NH₂/CS (50%) composite membrane by 8 cycles of tests; b) Catalytic cycles of UiO-66-NH₂/CS (50%) composite membrane in the presence of H₂O₂ at neutral pH. The result showed that the peroxidase-like catalytic activity of UiO-66-NH₂/chitosan composite membrane obviously decreased.

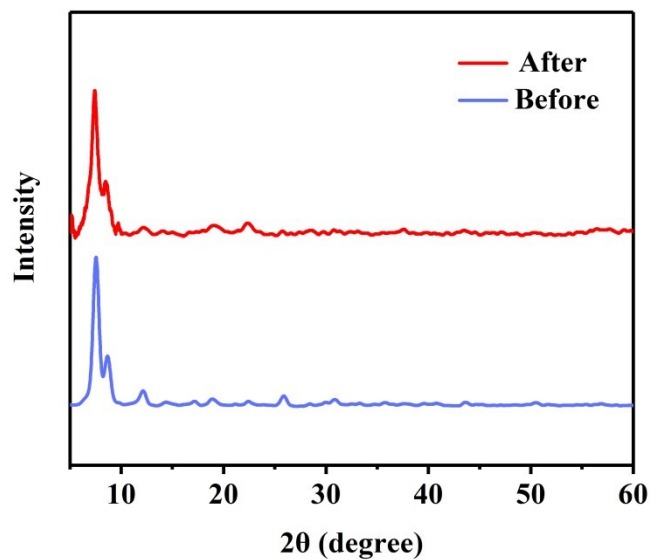


Figure S10. Powder XRD profiles for the UiO-66-NH₂/CS composite membrane before and after photo catalysis.

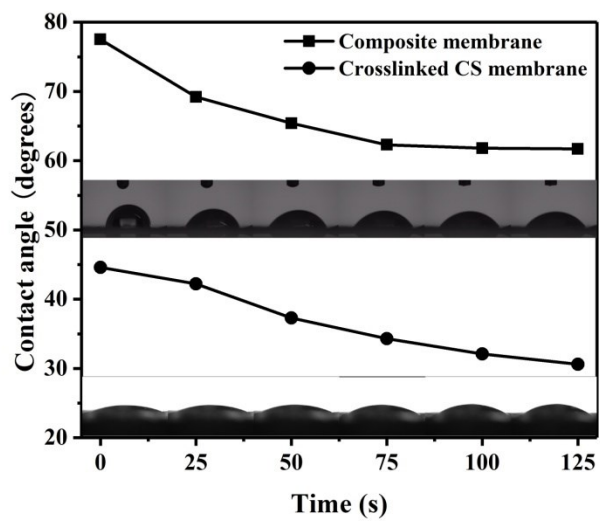


Figure S11. Dynamic contact angles of the crosslinked CS membrane and theUiO-66-NH₂/CS composite membrane.

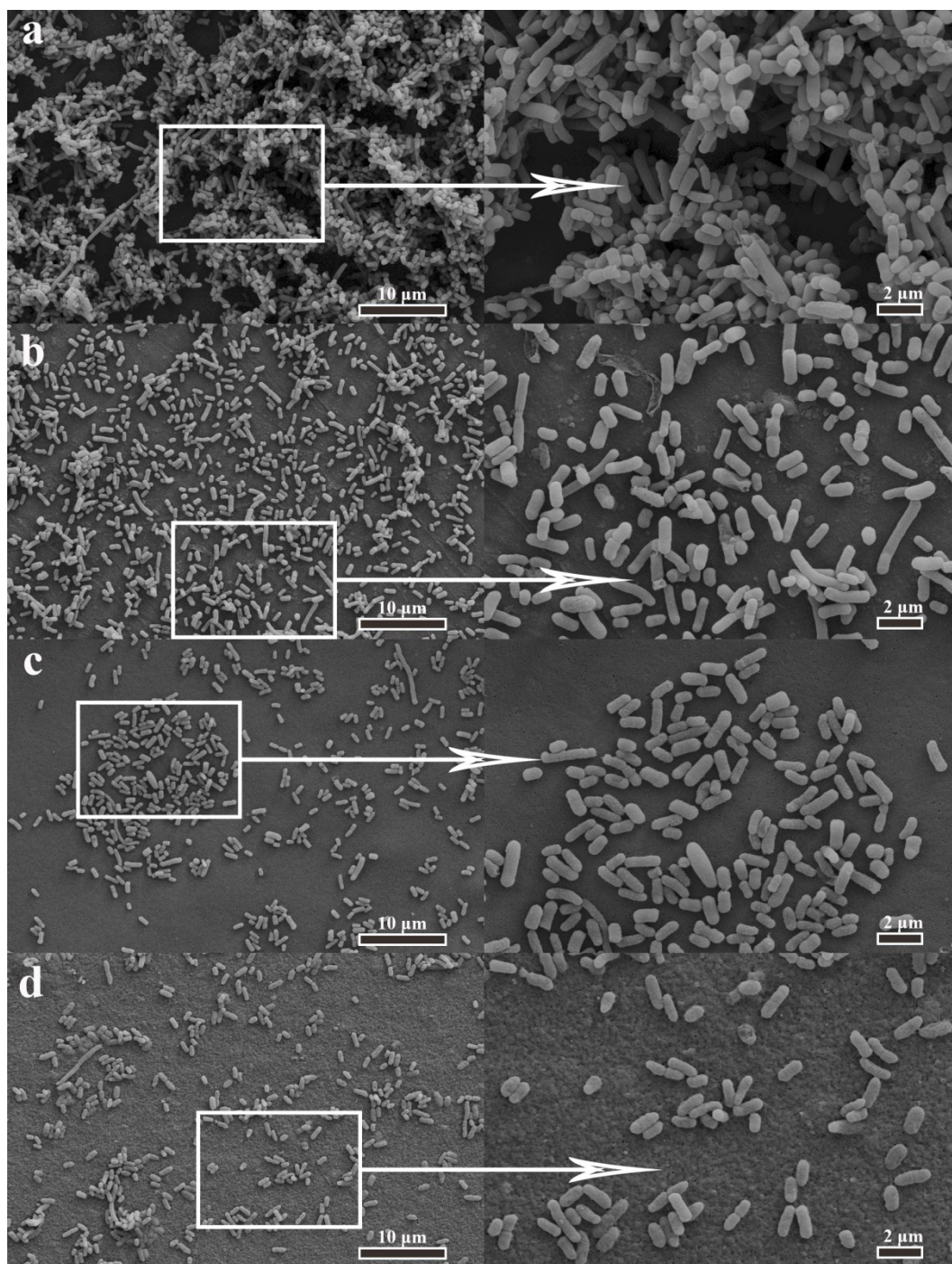


Figure S12. The morphology of *E. coli* in solid circumstance: a) CS membrane+PBS, b) CS membrane+H₂O₂, c) UiO-66-NH₂/CS composite membrane+PBS, and d) UiO-66-NH₂/CS composite membrane+H₂O₂, with UV irradiation of the UiO-66-NH₂/CS composite membrane for 30 min.

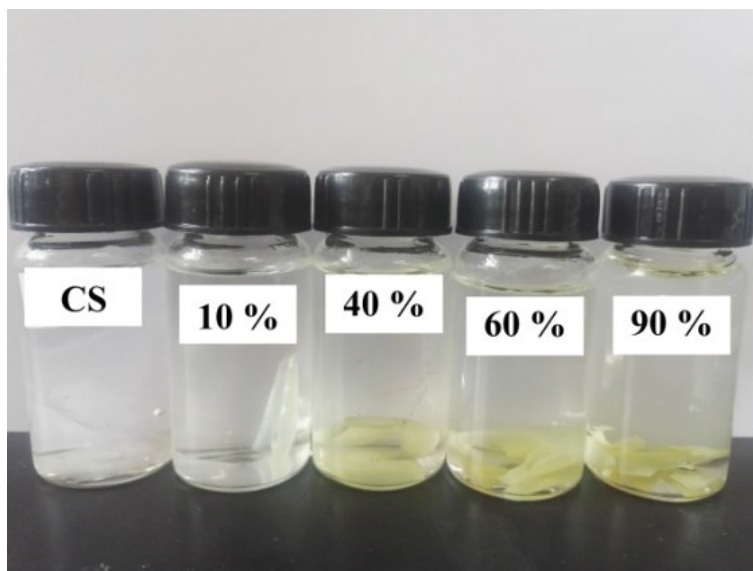


Figure S13. Photographs of the UiO-66-NH₂/CS composite membrane with different weight ratios of UiO-66-NH₂ to CS after immersion into water for 30 days followed by ultrasonication for 2 h.

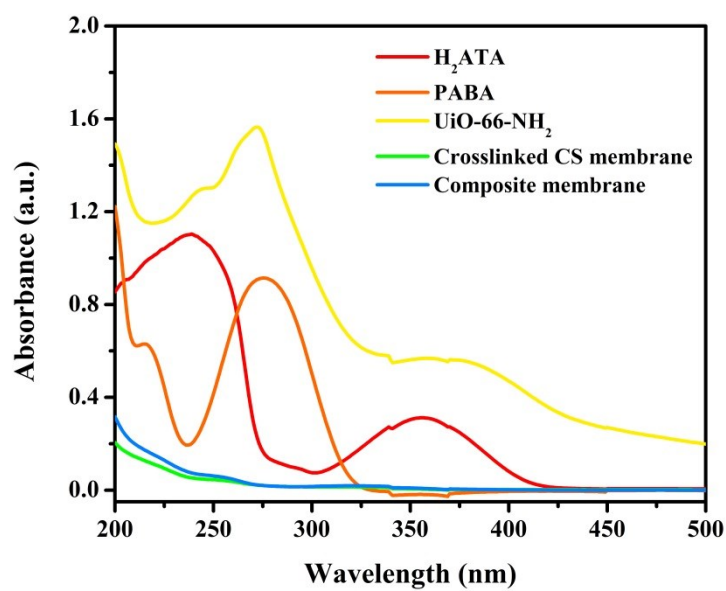


Figure S14. The UV-vis absorbance of the liquid taken from the vials with two pieces of membranes (1.5 cm × 2.5 cm) in 20 mL of water.



Figure S15 Peroxidase-like enzymatic activity of UiO-66-NH₂/chitosan composite membrane stored for two years. The long term stability for peroxidase mimicking was detected in the presence of TMB•2HCl and H₂O₂. The reaction solution quickly turned blue, which indicated that its peroxidase-like enzymatic activity was maintained well.

References

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