Electronic Supplementary Information (ESI)

A bifunctional fluorescent sensor for CCCP-induced cancer cell

apoptosis imaging

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Experimental

Materials and Methods

All chemicals were gained from reagent companies (Sigma-Aldrich, Aladdin, Macklin). Annexin V/7-AAD Apoptosis Detection Kit was gained from Shanghai Xinyu Biological Technology CO., LTD. Deionized water was used all over the work. The pH values were detected by a Model PHS-3C meter (Shanghai, China). Absorption spectrums were detected by UV-2102 double-beam UV/VIS spectrometer. Fluorescence spectrums were recorded by F-4500 FL Spectrophotometer. The ¹H NMR and ¹³C NMR spectra were measured by Bruker DTX-400 spectrometer. ESI mass spectra were gained by an HPLC Q-Exactive HR-MS spectrometer (Thermo, USA). Flow cytometry analysis was taken on BD Canto plus (USA). Cofocal fluorescence images were gained by Zeiss LSM 880 confocal microscope.

Synthesis and characterization

Compounds 1 and 2 were gained by the literature methods.^{S1,S2}

Synthesis of probe NPCF. 240 mg of compound 1 (1.5 equiv.) and 269 mg of compound 2 (1 equiv.) were added to a flask (50 mL), followed by adding ethanol (20 mL) and piperidine (0.2 mL). After refluxing for 16 h under nitrogen, a yellow product was filtered, washed with ice-cold ethanol. 117.0 mg of **NPCF** was obtained in a yield of 28.4%. ¹H NMR (400 MHz, DMSO-*d*₆, ppm): 2.39 (s, 3H), 7.35 (t, 1H, J = 15.0 Hz), 7.42 (t, 1H, J = 15.0 Hz), 7.52 (t, 1H, J = 15.0 Hz), 7.60 (t, 2H, J = 14.0 Hz), 7.77 (s, 1H), 7.84 (s, 1 H), 7.89 (d, 1 H, J = 8.1 Hz), 8.15 (d, 1H, J = 8.1 Hz), 8.19–8.34 (m, 3H), 13.20 (s, 1H), 13.48 (s, 1H); ¹³C NMR (100 MHz, DMSO-*d*₆, ppm): $\delta = 20.27$, 113.33, 117.44, 121.60, 122.45, 122.77, 123.25, 123.58, 126.14, 126.46, 127.54, 129.57, 131.80, 132.83, 133.70, 135.28, 138.96, 143.57, 149.60, 151.28, 155.32, 168.84, 181.62. HR-MS: Calcd for $[C_{24}H_{17}N_3O_2S + H]^+$: 412.1114, found 412.1117.

Synthesis of compound NPM. 269 mg of compound **2** (1 equiv.) and 553 mg of K₂CO₃ (4 equiv.) were added to a flask (50 mL), and then acetone (7 mL) were added. After refluxing for 12 h, a red-purple product was filtered, washed with acetone and deionized water. Compound **NPM** was received as a red-purple solid (136.6 mg, yield 44.2%). ¹H NMR (400 MHz, DMSO-*d*₆, ppm): 2.20 (s, 3H), 2.25 (s, 3H), 7.05 (d, 2H, J = 16.0 Hz), 7.21 (s, 2H), 7.36 (s, 1H), 7.79–7.93 (m, 3H), 8.02 (s, 1H); ¹³C NMR (100 MHz, DMSO-*d*₆, ppm): $\delta = 20.73$, 27.31, 115.96, 116.34, 120.74, 121.68, 121.74, 122.61, 123.69, 125.31, 131.11, 133.64, 136.11, 144.35, 152.25, 166.29, 170.87, 198.56. HR-MS: Calcd for [C₁₈H₁₅NO₂S + H]⁺: 310.0896, found 310.0900.

Synthesis of SO₂ donor. SO₂ donor were synthesized by reported methods.^{S3} 2, 4-dinitrobenzensulfonyl chloride (0.594 g, 2.25 mmol) in 18 mL dry DCM at 0 °C was added to a solution of benzylamine (0.477 g, 6.0 mmol) with trimethylamine (0.963 g, 9 mmol) and 10 mL DCM, the mixture was stirred and reacted for 3 h at room temperature, then 100 mL of water was added to the solution to stop the reaction. Extracted by DCM, and the organic layer was dried by anhydrous Na₂SO₄. The solvent was removed under vacuum to get the crude product and the yellow product was collected by chromatography using DCM as eluent. SO₂ donor was obtained as a yellow solid (580.0 mg, yield 77.3%). ¹H NMR (400 MHz, CDCl₃, ppm): 4.66 (d, 2H, J = 5.6 Hz), 6.92 (d, 1H, J = 9.5 Hz), 7.34–7.42 (m, 5H), 8.20–8.24 (dd, 1H, J = 2.5 Hz), 8.92 (s, 1H) 9.14 (d, 1H, J = 2.6 Hz); ¹³C NMR (100 MHz, CDCl₃, ppm): $\delta = 47.58$, 114.45, 124.22, 127.09, 128.37, 129.29, 130.39, 130.76, 135.59, 136.45, 148.23.

Detection Limit Calculation

For **NPCF**, the fluorescence intensity of 542 nm versus HSO_3^- concentration (0.6 to 1.8 equiv.) was investigated, and the detection limit of **NPCF** was counted as 22.7 nM (R² = 0.985), using the formula (1): Detection limit = 3 $\sigma/|k|$ (1)

Cell imaging

All cells were cultured in Dulbecco's modified Eagle's medium (DMEM) supplemented with 10% fetal bovine serum at 37°C

Cytotoxicity assay. Cytotoxicity assay was executed by using Cell Counting Kit-8 (CCK-8) according to our report article.^{S4} The concentrations of **NPCF** changed from 0 μ M to 30 μ M.

Bio-imaging of NPCF for sensing HSO₃⁻ **in MCF-7 cells.** In order to detect endogenous HSO₃⁻, MCF-7 cells were treated with NEM (1 mM) for 0.5 h, and treated with **NPCF** (10 μ M) for 0.5 h and SO₂ donor (100 μ M) for 1h. For detecting exogenous HSO₃⁻, MCF-7 cells were incubated with **NPCF** (10 μ M) for 0.5 h, and treated with 1equiv., 5equiv. and 10 equiv. of HSO₃⁻ for 1h, and then imaged. Conditions: $\lambda_{ex} = 405$ nm, $\lambda_{em} = 520$ nm-560 nm.

Bioimaging application of NPCF for detecting pH in Hela cells. NPCF (10 μ M) was incubated in different pH PBS buffers (pH 4.5, 5.5, 6.5, 7.5 and 8.5) for 0.5 h, and imaged by Zeiss LSM 880 confocal microscope. Conditions: $\lambda_{ex} = 552 \text{ nm}, \lambda_{em} = 630-670 \text{ nm}.$

SO₂ reduces LPS-induced inflammation and alleviates acidification environment. For the blank group, EC1 cells were incubated with NPCF (10 μ M) for 0.5 h; For the control groups, EC1 cells were treated with LPS (10 mg/L) for 12 h, followed by treated with PMA (1 μ g/mL) for 1 h, and imaged after incubation with NPCF (10 μ M) for 0.5 h; EC1 cells were treated with SO₂ donor (100 μ M) for 1 h, and then imaged after incubation with NPCF (10 μ M) for 0.5 h; For the experimental group, EC1 cells were incubated with LPS (10 mg/L) for 12 h, followed by treated with PMA (1 μ g/mL) for 1 h, and incubated with SO₂ donor (100 μ M) for 1 h, and finally imaged after incubation with **NPCF** (10 μ M) for 0.5 h.

Distinguishing between normal cells and cancer cells. NPCF were incubated with cancer cells (MCF-7 cells) and normal cells (MCF-10A cells) for 0.5 h, respectively, and then imaged. Conditions: for green channel, $\lambda_{ex} = 405 \text{ nm}$, $\lambda_{em} = 520 \text{ nm}-560 \text{ nm}$; for red channel, $\lambda_{ex} = 552 \text{ nm}$, $\lambda_{em} = 630-670 \text{ nm}$.

Changes of SO₂ and pH in cells with the increase of CCCP concentrations. Hela cells were incubated with CCCP (Ranged from 0 μ M to 30 μ M) for 1 h, 12 h or 24 h, respectively. Conditions: for green channel, $\lambda_{ex} = 405$ nm, $\lambda_{em} = 520$ nm–560 nm; for red channel, $\lambda_{ex} = 552$ nm, $\lambda_{em} = 630-670$ nm.

Flow cytometry analysis. Hela cells were cultured in 6-well plates at a density of 2.0×10^5 cells/well, and the cells were washed with PBS buffers and centrifuged. The cells were resuspended in PBS, stained with **NPCF** (10 µM) for 30 min, and checked by flow cytometry.

Zebrafish Confocal Fluorescence Imaging

Wild zebrafish were gained from Shanghai FishBio Co., Ltd. Zebrafish were fed in E3 media at 28°C. For the control groups, the two days old zebrafish were incubated with **NPCF** for 0.5 h and imaged. For the experimental group, zebrafish were incubated with 10 μ M of **NPCF** for 0.5 h, treated with 100 μ M of HSO₃⁻ for 1 h, and then imaged. Zebrafish were imaged on Zeiss LSM 880 confocal microscope. Conditions: $\lambda_{ex} = 405$ nm, $\lambda_{em} = 520$ nm–560 nm.

References

S1 Y. Q. Ge, J. Jia, H. Yang, X. T. Tao and J. W. Wang, *Dyes Pigments*, 2011, 88, 344-349
S2 X. Yang, Y. Liu, Y. Wu, X. Ren, D. Zhang and Y. Ye, *Sensor. Actuat. B-Chem*, 2017, 253, 488–494

S3 S. R. Malwal, D. Sriram, P. Yogeeswari, V. B. Konkimalla and H. Chakrapani, *J. Med. Chem.*, 2012, **55**, 553–557.

S4 X. Yang, W. Liu, J. Tang, P. Li, H. Weng, Y. Ye, M. Xian, B. Tang and Y. Zhao, *Chem. Commun.*, 2018, **54**, 11387–11390.



Scheme S1. The proposed mechanism of NPCF (10 μ M) for detecting both HSO₃⁻ and pH. (b) Changes in pH during SO₂ relieving inflammation and dual indicators to detect changes in SO₂ and pH during the apoptotic process in living cells.

Fig. S1 HR-MS spectrum of probe NPCF with HSO₃⁻







¹³C-NMR spectrum of probe NPCF in DMSO- d_6



HR-MS spectrum of probe NPCF

Fig. S3 Structure characterization of compound NPM



¹H-NMR spectrum of compound **NPM** in DMSO- d_6



¹³C-NMR spectrum of compound NPM in DMSO- d_6



HR-MS spectrum of probe NPM



Fig. S4 UV–vis absorption spectra changes. The black line means only NPCF was added, the red line means NPCF and 10 equiv. of HSO_3^- were added.



Fig. S5 (a) $\lambda_{ex} = 350$ nm, $\lambda_{em} = 542$ nm, slit: 10 nm/10 nm. (b) $\lambda_{ex} = 543$ nm, $\lambda_{em} = 653$ nm, slit: 10 nm/10 nm. All data were acquired in HEPES buffer solution (pH = 7.4, 10 mM, containing 1 mM CTAB). The black line means only **NPCF** was added, the red line means **NPCF** and 10 equiv of HSO₃⁻ were added.



Fig. S6 Calibration curve of probe **NPCF** to HSO_3^- (0–10 equiv.). Conditions: $\lambda_{\text{ex}} = 350 \text{ nm}$, slit: 10 nm/10 nm.



Fig. S7 The linear responses at low HSO_3^- concentrations (0.6–1.8 equiv.).



Fig. S8 UV–vis absorption changes of **NPCF** in the presence of different analytes (10 equiv. of Na⁺, K⁺, HS⁻, Cys, Hcy, GSH, NO₃⁻, CO₃²⁻, HCO₃⁻, Ac⁻, SO₄²⁻, PO₄³⁻, F⁻, Cl⁻, Br⁻, I⁻, NO₂⁻, •OH, ONOO⁻, •NO, O₂•⁻, and H₂O₂) in HEPES buffer solution (pH = 7.4, 10 mM, containing 1 mM CTAB).



Fig. S9 Emission changes of **NPCF** in the presence of different analytes (10 equiv. of Na⁺, K⁺, HS⁻, Cys, Hcy, GSH, NO₃⁻, CO₃²⁻, HCO₃⁻, Ac⁻, SO₄²⁻, PO₄³⁻, F⁻, Cl⁻, Br⁻, I⁻, NO₂⁻, •OH, ONOO⁻, •NO, O₂•⁻, and H₂O₂) in HEPES buffer solution (pH = 7.4, 10 mM, containing 1 mM CTAB), slits: 10/10 nm. (a) $\lambda_{ex} = 350$ nm, $\lambda_{em} = 542$ nm; (b) $\lambda_{ex} = 553$ nm, $\lambda_{em} = 653$ nm.



Fig. S10 Fluorescence response of NPCF (10 μ M) to HSO₃⁻ (10 equiv.) in the presence of the different analytes. in HEPES buffer solution (pH = 7.4, 10 mM, containing 1 mM CTAB). $\lambda_{ex} = 350$ nm, slit: 10 nm/10 nm.



Fig. S11 Fluorescence response of NPCF (10 μ M) to HSO₃⁻ (10 equiv.) in the presence of the different analytes in HEPES buffer solution (pH = 7.4, 10 mM, containing 1 mM CTAB). $\lambda_{ex} = 553$ nm, slit: 10 nm/10 nm.



Fig. S12 Time-dependent fluorescence spectral changes of **NPCF** with HSO_3^- (10 equiv.) in HEPES buffer solution (pH = 7.4, 10 mM, containing 1 mM CTAB), slits: 10/10 nm. (a) $\lambda_{ex} = 350$ nm, $\lambda_{em} = 542$ nm. (b) $\lambda_{ex} = 553$ nm, $\lambda_{em} = 653$ nm.



Fig. S13 The pH effect of the test solution on **NPCF** (10 μ M) was tested in the absence (black line) or presence (red line) of HSO₃⁻ (10 equiv.). (a) $\lambda_{ex} = 350$ nm, $\lambda_{em} = 542$ nm, slits: 10/10 nm. (b) $\lambda_{ex} = 553$ nm, $\lambda_{em} = 653$ nm, slits: 10/10 nm.



Fig. S14 UV-vis absorption spectra changes at pH 4.26 and 9.21.



Fig. S15 Fluorescence spectra changes at pH 4.26 and 9.21. NPCF: $\lambda_{ex} = 410$ nm, $\lambda_{em} = 610$ nm, slits: 5/5 nm. NPM: $\lambda_{ex} = 380$ nm, $\lambda_{em} = 573$ nm, slits: 5/5 nm.



Fig. S16 Absorption spectra changes of NPCF at different pH values.



Fig. S17 The pH titration curve of NPCF plotted by fluorescence as a function of pH. Conditions: $\lambda_{ex} = 410$ nm, slits: 5/5 nm.



Fig. S18 (a) Absorption and (b) fluorescence spectra changes of NPM at different pH values. (b) $\lambda_{ex} = 380$ nm, $\lambda_{em} = 573$ nm. Slits: 5/5 nm.



Fig. S19 The pH titration curve of **NPM** plotted by fluorescence as a function of pH. Conditions: $\lambda_{ex} = 380$ nm, $\lambda_{em} = 573$ nm, slit: 10 nm/10 nm.



Fig. S20 The fluorescence of NPCF in pH 4.26 and 9.21 under different potential interference of interference agents. Conditions: $\lambda_{ex} = 410$ nm, $\lambda_{em} = 610$ nm.



Fig. S21 The time courses of fluorescence intensity of NPCF in HEPES buffer solution (10 mM, containing 1 mM CTAB) at different pH values (4.26 and 9.21, respectively). Conditions: $\lambda_{ex} = 410$ nm, $\lambda_{em} = 610$ nm, slit: 5 nm/5 nm.



Fig. S22 Reversible fluorescence intensity ($\lambda_{em} = 610 \text{ nm}$) changes of **NPCF** between pH = 4.26 and 9.21 in HEPES buffer solution (10 mM, containing 1 mM CTAB). Conditions: $\lambda_{ex} = 410 \text{ nm}$, $\lambda_{em} = 610 \text{ nm}$.



Fig. S23 Different NPCF concentrations (0, 2.5 μ M, 5 μ M, 10 μ M, 15 μ M, 20 μ M and 30 μ M) were tested in Hela cells for toxicity.



Fig. S24 Fluorescence images of MCF-7 cells. (a, b and c) cells were incubated with NPCF (10 μ M) for 0.5 h; (d, e and f) images of cells after treatment with probe NPCF (10 μ M) for 0.5 h and then treatment of the cells with 100 μ M of HSO₃⁻ for 1

h. (a and d) Bright field images; (b and e) green emission of NPCF; (c and f) red emission of NPCF.



Fig. S25 Fluorescence images of MCF-7 cells. The cells were incubated with **NPCF** (10 μ M) for 0.5 h (a, b, and c) and 100 μ M of SO₂ donor for 1h (d, e and f). The cells were treated with 1 mM of NEM for 0.5 h, incubated with **NPCF** (10 μ M) for 0.5 h, and then treated with 100 μ M of SO₂ donor for 1h (g, h and i). MCF-7 cells were incubated with **NPCF** (10 μ M) for 0.5 h and treated with several concentrations of HSO₃⁻ for 1 h (j–r).



Fig.S26 Confocal fluorescence images of zebrafish using a 10×objective. (a–d) **NPCF** (10 μ M) and zebrafish were incubated for 0.5 h. (e–h) **NPCF** (10 μ M) and zebrafish were incubated for 0.5 h, and then treated with HSO₃⁻ (100 μ M) for 1 h. (i) Relative pixel intensity of I_{green channel}/I_{red channel} for detecting HSO₃⁻. (a and e) $\lambda_{ex} = 405$ nm, $\lambda_{em} = 520$ nm–560 nm; (b and f) $\lambda_{ex} = 552$ nm, $\lambda_{em} = 630-670$ nm. Scale bar is 200 μ m.



Fig. S27 (a–d) EC1 cells were incubated with NPCF (10 μ M) for 0.5 h. (e–h) EC1 cells were incubated with SO₂ donor (100 μ M) for 1 h, and then incubated with NPCF (10 μ M) for 0.5 h. (i–l) EC1 cells were incubated with LPS (10 mg/L) for 12 h and PMA (1 μ g/mL) for 1 h, and then incubated with NPCF (10 μ M) for 0.5 h. (m–p) EC1 cells were incubated with LPS (10 mg/L) for 1 h, followed by adding SO₂ donor (100 μ M) for 1 h, and finally incubated with NPCF (10 μ M) for 0.5 h. (q) Relative pixel intensity of green channel (b, f, j and n, $\lambda_{ex} = 405$ nm, $\lambda_{em} = 520$ nm–560 nm) and red channel (c, g, k and o, $\lambda_{ex} = 552$ nm, $\lambda_{em} = 630-670$ nm). Scale bar is 25 μ m.



Fig. S28 Confocal fluorescence images of cancer cells vs normal cells. **NPCF** were incubated with MCF-10A cells (a–d) and MCF-7 cells (e–h) for 0.5 h, respectively. Scale bar is 25 μ m. (i–l) Flow cytometry analysis of (a–h). (m) The mean intensity of green channel (i and k, $\lambda_{ex} = 405$ nm, $\lambda_{em} = 520$ nm–560 nm) and (n) red channel (j and l, $\lambda_{ex} = 552$ nm, $\lambda_{em} = 630-670$ nm) on the flow cytometry analysis.



Fig.S29 Experiments of SO₂ and pH changes in CCCP-induced apoptosis. Flow cytometry analysis of (a–n), different concentrations of CCCP (a–d, 0 μ M; e–h, 10 μ M; i–l, 30 μ M) were co-incubated with the cells for 12h (a, b, e, f, i and j) or 24 h (c, d, g, h, k and l), respectively, followed by separately adding **NPCF** (10 μ M) and incubating for 0.5 h, and then analyzed by flow cytometry. (m and n) The green channel (a, e, i, c, g and k, $\lambda_{ex} = 405$ nm, $\lambda_{em} = 520$ nm–560 nm) and red channel (b, f, j, d, h and 1, $\lambda_{ex} = 552$ nm, $\lambda_{em} = 630-670$ nm). (o–v) Annexin V/7-AAD analysis of apoptosis of CCCP at different concentrations (p and t, 0 μ M; q and u, 10 μ M; r and v, 30 μ M) and different culture time (o–r, 12 h; s–v, 24 h). (x–y) The population of necrosis (Q1), early apoptosis (Q4), late apoptosis (Q2), and viable cells (Q3) for 12h (x) or 24h (y) in the Annexin V/7-AAD analysis of apoptosis (o–v).



Fig.S30 (a–j) Confocal fluorescence images of HeLa cells and **NPCF** (10 μ M) were treated with different pH PBS buffers (pH 4.5, 5.5, 6.5, 7.5 and 8.5) for 0.5 h, respectively. (k) Relative pixel intensity of red channel (λ_{ex} = 552 nm, λ_{em} = 630–670 nm) in different pH PBS buffers. (I) The linear responses in different pH PBS buffers. Scale bar is 25 μ m