

Electronic Supporting Information for

New Type of Carborane-based Electron-accepting Material

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EXPERIMENTAL SECTION

General Information

All the experimental procedures were performed under a dry nitrogen (N_2) or argon (Ar) atmosphere using standard Schlenk techniques. All solvents were distilled freshly and used under dry nitrogen or argon-purging. Phenylacetylene, TMEDA, copper iodide, copper chloride, NBS, silver nitrate, 2-methyl-3-butyn-2-ol, *n*-butylamine, hydroxylamine hydrochloride, sodium hydroxide, decaborane, *N,N*-dimethylaniline were purchased from Aldrich or TCI and used without further purification. The starting materials, diphenylbutadiyne (1)¹, (bromoethynyl)benzene (2)², 2-methyl-6-phenylhexa-3,5-diyne-2-ol (3)³, phenylbutadiyne (4)⁴, and diphenylhexatriyne (5)⁵ were prepared according to previously reported procedures. Reference compound **I** was also prepared according to previously reported procedure.⁶ 1H and ^{13}C nuclear magnetic resonance (NMR) spectra were recorded using a Bruker Fourier 300 MHz spectrometer operated at 300.1 and 75.4 MHz, respectively. 1H and ^{13}C NMR chemical shifts were measured in $CDCl_3$. Referenced to the relative peaks of $CDCl_3$ are 7.26 ppm for 1H NMR and 77.16 ppm for ^{13}C NMR, respectively. ^{11}B NMR were acquired on 500MHz NMR spectrometer (AVANCE III HD, Bruker, Germany) at KBSI Western Seoul center. Elemental analyses were performed using a Carlo Erba Instruments CHNS-O EA 1108 analyser. A high-resolution mass spectrometry (HR-MS) analysis was performed using a highly sensitive liquid chromatography–multistage MS (LC/MS/MSⁿ) ($n = 10$) spectrometer (Thermo Fisher Scientific, LCQ Fleet Hyperbolic Ion Trap MS/MSⁿ Spectrometer).

Crystal Structures Determination

The structures of the (**II**) and (**III**) single crystal were determined at 223 K using an X-ray diffraction system (Bruker AXS GMBH) at the Korea Basic Science Institute, Seoul Western Centre that was equipped with a sealed-tube X-ray source (50 kV 30 mA); monochromatic Mo $K\alpha$ radiation ($\lambda = 0.71073 \text{ \AA}$) corresponding to graphite was obtained from the X-ray source. At this preliminary stage,

the unit cell constants were determined from a set of 45 narrow-frame (0.3° in ω) scans. A double-pass method of scanning was used to exclude any noise. The collected frames were integrated using an orientation matrix, which was developed using narrow frame scans. The *SMART* software package was used for data collection, while the *SAINT* software package⁷ was used for frame integration. Final cell constants were determined by conducting global refinement of the *xyz* centroids of the reflections, which were harvested from the entire data set. The structural solution and refinement were carried out using *SHELXTL PLUS* software package.⁸ The crystal structures of II and III have been deposited in the CCDC database (deposit numbers 1986098 and 1986099).

Absorption and Fluorescence Measurements

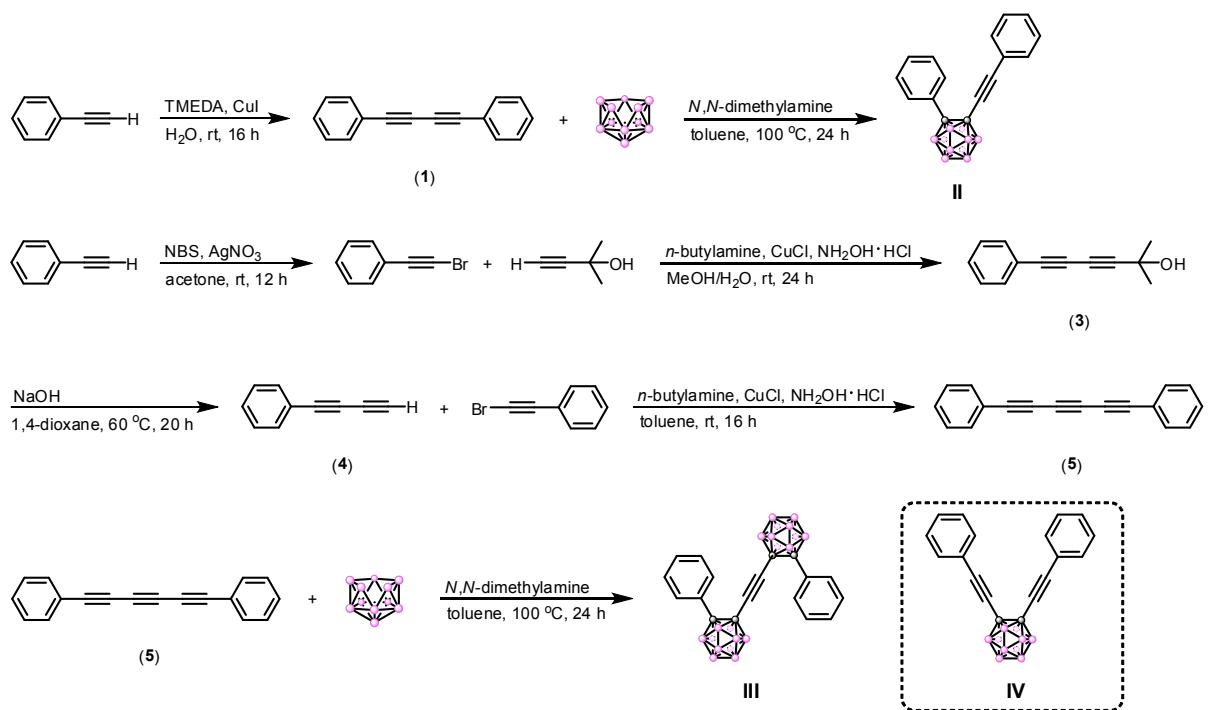
Absorption spectra were recorded using a Shimadzu (UV-3101PC) scanning spectrophotometer (Japan). Emission and excitation spectra were measured by using a Varian fluorescence spectrophotometer (Cary Eclipse) (Scinco, Korea).

DFT Calculations

Theoretical calculations for all complexes were conducted by the *Gaussian 16* Package.⁹ The ground-state geometries of **I–III** were optimised by using the B3LYP density functional theory (DFT) and 6-31G(d,p) basis set. The contours of the electron density were plotted by using *Chem3D version.10.0.*¹⁰

Cyclic Voltammetry Measurements

Cyclic voltammetry (CV) was performed in an electrolytic solution prepared using 0.1 M tetra-butyl ammonium hexafluorophosphate (NBu₄PF₆) at room temperature under an atmosphere of argon. For this purpose, CHI600E (CH Instrument, Inc., USA) was used. Glassy carbon, platinum wire, and Ag/AgNO₃ (0.1 M) were used as the working, counter, and reference electrodes, respectively. All the potentials were calibrated to the ferrocene/ferrocenium (Fc/Fc⁺) redox couple.



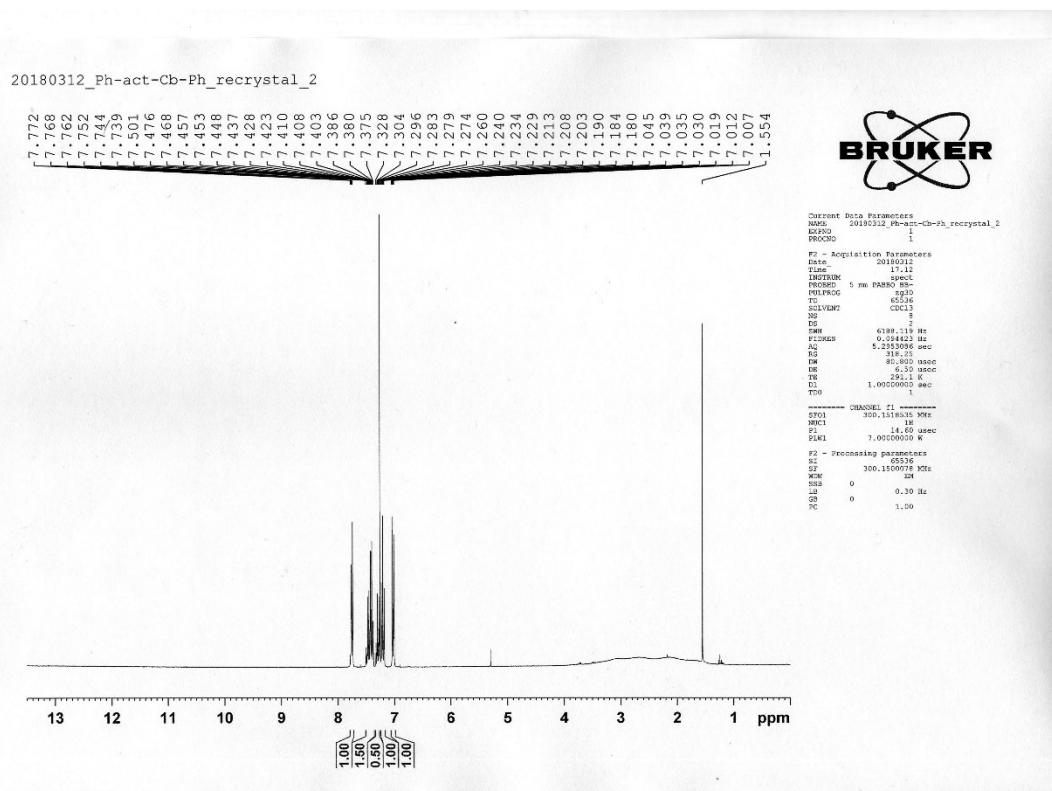
Scheme S1. Synthetic procedures for **II** and **III**.

Synthesis

1-(Phenylethynyl)-2-phenyl-*o*-carborane (II) A mixture of decaborane (0.39 g, 3.21 mmol), diphenylbutadiyne (1) (0.65 g, 1.63 mmol), and *N,N*-dimethylaniline (0.9 mL, 7.11 mmol) in toluene (15 mL) was stirred at 100 °C for 17 h under argon atmosphere. After cooling to room temperature, the mixture was filtered by vacuum filtration, washed with 1 M HCl aqueous solution and extracted with dichloromethane (DCM). The combined organic layers were dried over MgSO₄ and evaporated under reduced pressure. The crude product was purified via recrystallisation with DCM and methanol after silica-gel column chromatography using toluene:hexane (*v:v*=1:1) as an eluent. White crystal; Yield: 74%; ¹H NMR (300 MHz, CDCl₃, δ): 7.76 (dt, *J* = 7.2 Hz, 1.5 Hz, 2H), 7.50–7.38 (m, 3H), 7.33–7.28 (m, 1H), 7.23–7.18 (m, 2H), 7.02 (dt, *J* = 6.9 Hz, 1.5 Hz), 3.85–1.40 (br, 10H); ¹³C NMR (75 MHz, CDCl₃, δ): 131.96, 131.22, 130.70, 129.78, 128.58, 128.42, 120.07, 85.44, 82.45, 82.12, 69.65; ¹¹B NMR (500 MHz, CDCl₃, δ): -2.1, -3.9, -9.2, -9.8. HRMS (FAB) calcd for C₁₆H₂₀B₁₀: 322.2496, found 322.2494; Elem. anal. calcd for C₁₆H₂₀B₁₀: C, 59.97; H, 6.29; B, 33.74 Found: C, 59.96; H, 6.27.

1,2-bis(2-Phenylcarboranyl)ethyne (III) A mixture of decaborane (0.107 g, 0.88 mmol), diphenylhexatriyne (5) (0.2 g, 0.88 mmol), and *N,N*-dimethylaniline (0.25 mL, 1.95 mmol) in toluene (5 mL) was stirred at 100 °C for 17 h under argon atmosphere. After cooling to room temperature, the mixture was filtered by vacuum filtration, washed with 1 M HCl aqueous solution and extracted with ethylacetate and DCM. The combined organic layers were dried over MgSO₄ and evaporated under reduced pressure. The crude product was purified via recrystallisation with DCM and methanol after silica-gel column chromatography using toluene:hexane (*v:v*=1:1) as an eluent. White transparent crystal; Yield: 53%; ¹H NMR (300 MHz, CDCl₃, δ): 7.56–7.51 (m, 2H), 7.44–1.37 (m, 8H), 3.90–1.20 (br, 10H); ¹³C NMR (75 MHz, CDCl₃, δ): 131.14, 130.89, 130.35, 128.83, 128.60, 128.44; ¹¹B NMR (500 MHz, CDCl₃, δ): -1.1, -3.9, -9.8. HRMS (FAB) calcd for C₁₈H₂₀B₁₀: 346.2496, found 346.2495; Elem. anal. calcd for C₁₈H₂₀B₁₀: C, 62.76; H, 5.85; B, 31.39; Found: C, 62.74; H, 5.84.

(a)



(b)

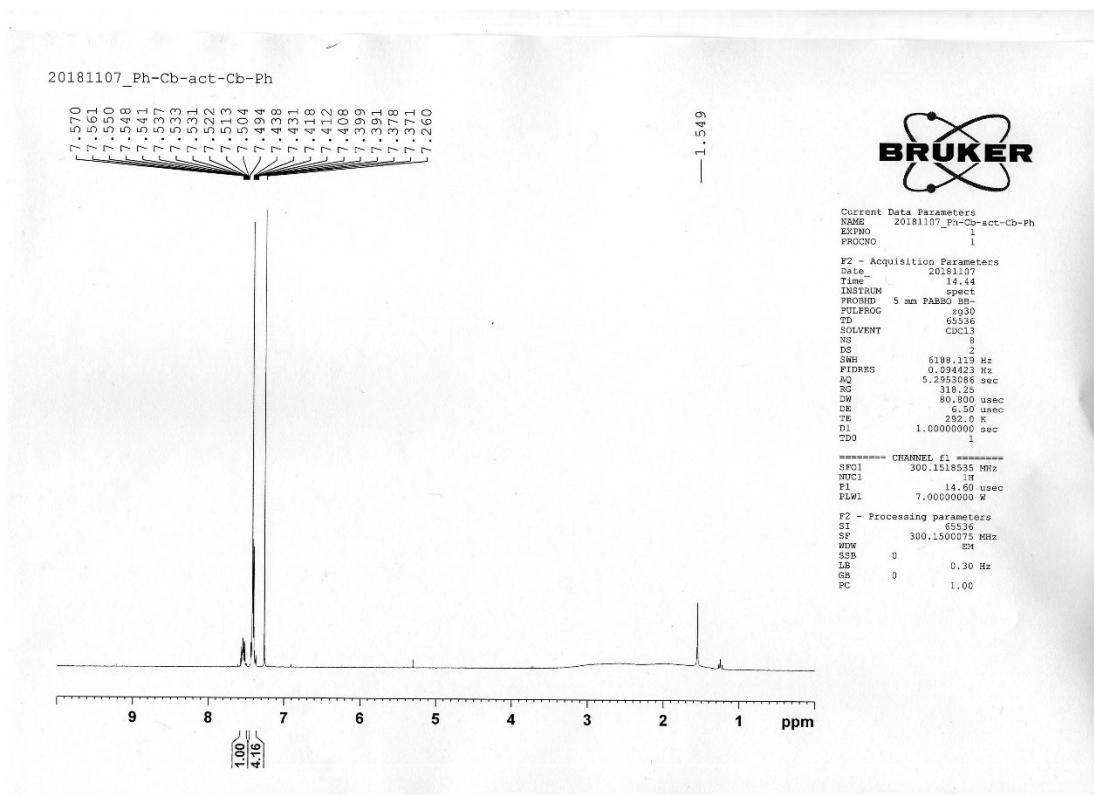
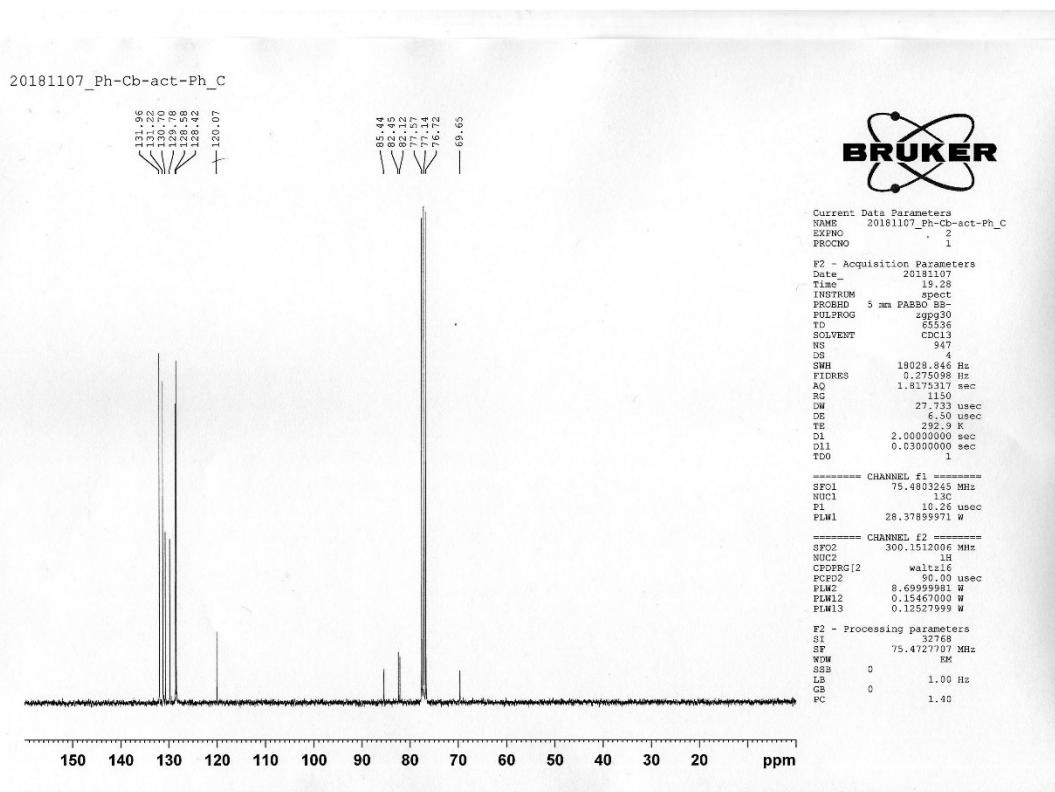


Figure S1. ¹H NMR spectra for (a) **II** and (b) **III** in CDCl_3 .

(a)



(b)

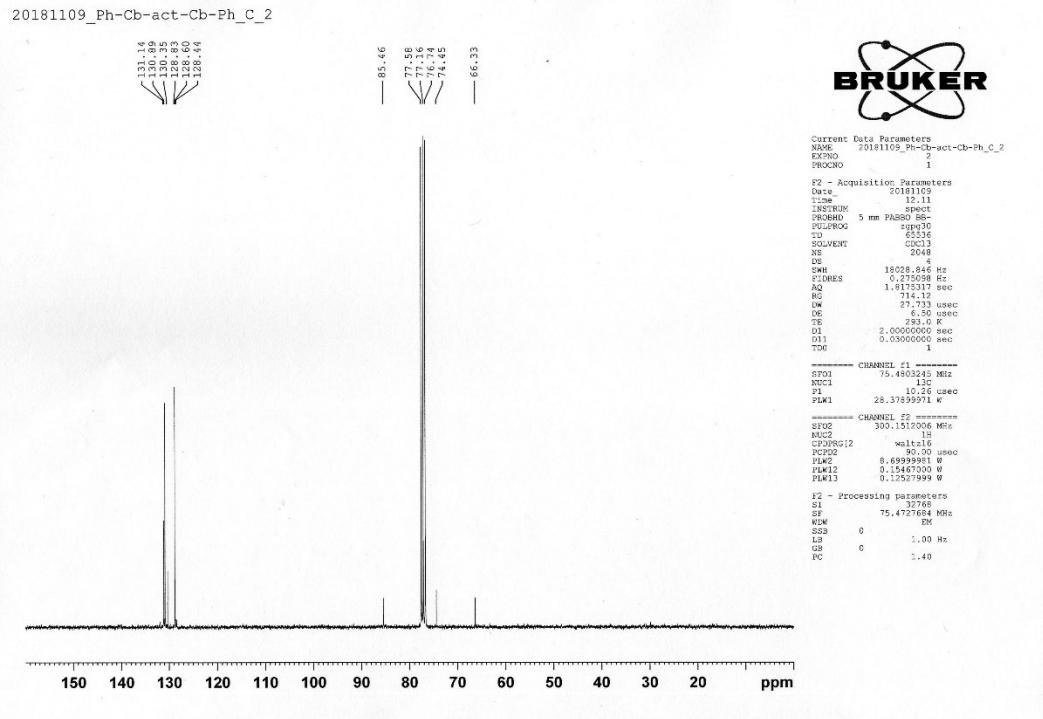
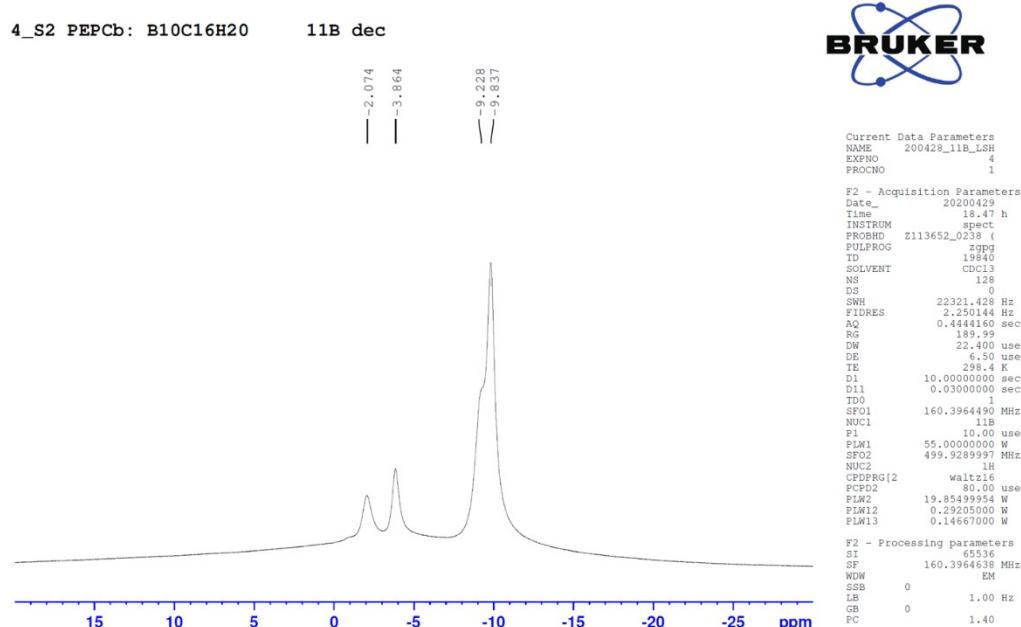


Figure S2. ^{13}C NMR spectra for (a) **II** and (b) **III** in CDCl_3 .

(a)



(b)

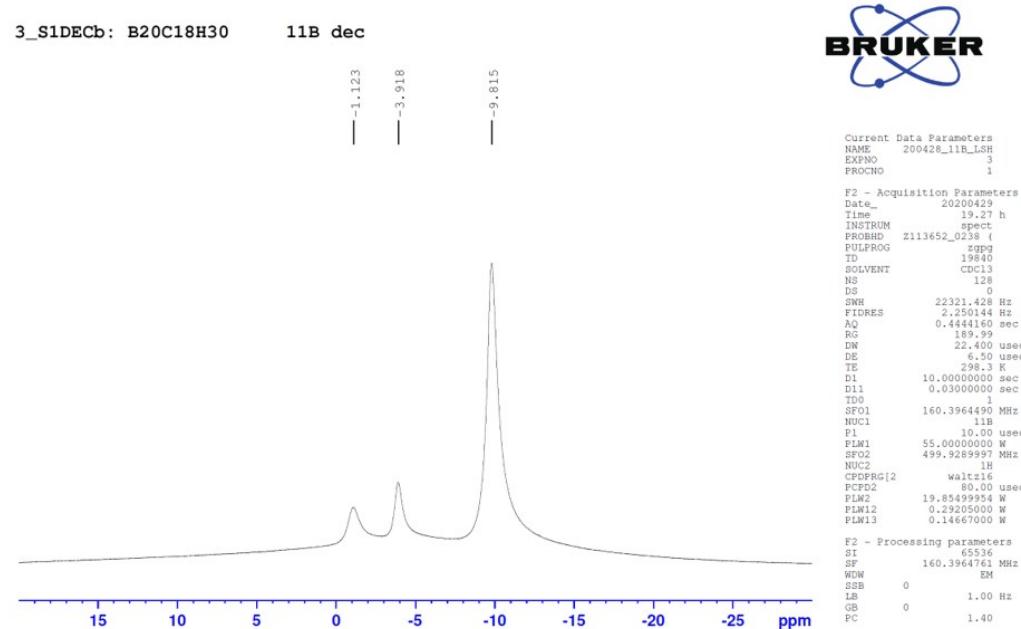
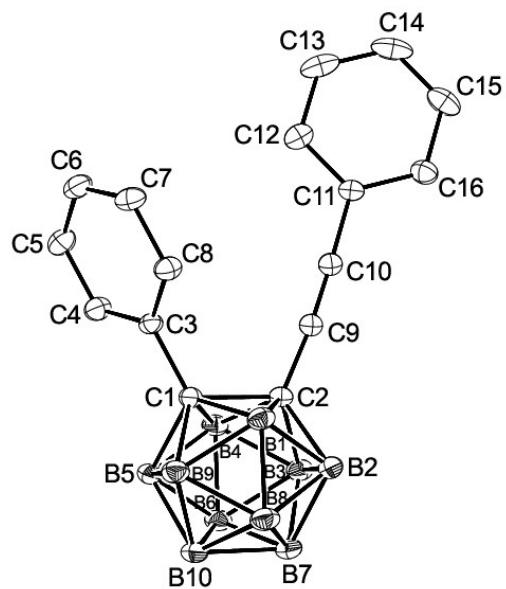


Figure S3. ¹¹B NMR spectra for (a) **II** and (b) **III** in CDCl₃.

(a)



(b)

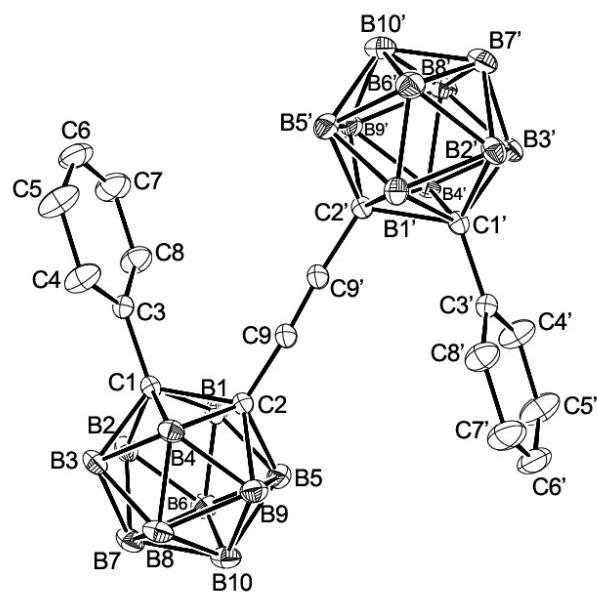


Figure S4. ORTEP(III)¹¹ drawings of (a) **II** and (b) **III** showing the thermal anisotropic displacement parameters at 30% probability. Hydrogen atoms are omitted for clarity.

Table S1. Crystal data and structure refinement for compounds **II** and **III**.

Identification code	II	III
Empirical formula	C ₁₆ H ₁₀ B ₁₀	C ₁₈ H ₁₀ B ₁₀
Formula weight	310.34	334.36
Temperature	223(2) K	223(2) K
Wavelength	0.71073 Å	0.71073 Å
Crystal system, space group	Triclinic, <i>P</i> −1	Monoclinic, <i>P</i> 2 ₁ /n
Unit cell dimensions	<i>a</i> = 7.2684(6) Å, α = 74.227(3) $^\circ$ <i>b</i> = 10.407(1) Å, β = 81.330(3) $^\circ$ <i>c</i> = 12.774 (1) Å, γ = 86.111(3) $^\circ$	<i>a</i> = 10.796(1) Å <i>b</i> = 9.7951(9) Å, β = 112.281(3) $^\circ$ <i>c</i> = 13.782(1) Å
Volume	918.89(15) Å ³	1348.5(2) Å ³
Z, Calculated density	2, 1.122 Mg/m ³	4, 1.647 Mg/m ³
Absorption coefficient, μ	0.056 mm ^{−1}	0.083 mm ^{−1}
<i>F</i> (000)	312	672
Crystal size	0.22 × 0.16 × 0.09 mm	0.21 × 0.14 × 0.10 mm
θ range for data collection	2.265 to 28.426 $^\circ$	2.622 to 28.515 $^\circ$
Limiting indices	-9 ≤ <i>h</i> ≤ 9, -13 ≤ <i>k</i> ≤ 13, -17 ≤ <i>l</i> ≤ 17	-14 ≤ <i>h</i> ≤ 14, -13 ≤ <i>k</i> ≤ 13, -18 ≤ <i>l</i> ≤ 18
Reflections collected /unique	29616 / 4611 [<i>R</i> _{int} = 0.0798]	43152 / 3392 [<i>R</i> _{int} = 0.1113]
Completeness to θ = 28.40	99.9 %	100.0 %
Refinement method	Full-matrix least-squares on <i>F</i> ²	Full-matrix least-squares on <i>F</i> ²
Data / restraints / parameters	4611 / 0 / 235	3392 / 0 / 173
Goodness-of-fit on <i>F</i> ²	1.035	1.025
Final <i>R</i> indices [<i>I</i> >2 σ (<i>I</i>)]	<i>R</i> ₁ ^a = 0.0597, <i>wR</i> ₂ ^b = 0.1319	<i>R</i> ₁ ^a = 0.0624, <i>wR</i> ₂ ^b = 0.1498
<i>R</i> indices (all data)	<i>R</i> ₁ = 0.1110, <i>wR</i> ₂ = 0.1559	<i>R</i> ₁ = 0.1065, <i>wR</i> ₂ = 0.1764
Largest diff. peak and hole	0.258 and -0.221 e. Å ^{−3}	0.254 and -0.276 e. Å ^{−3}

^a*R*₁ = $\sum ||F_o| - |F_c||$ (based on reflections with $F_o^2 > 2\sigma F^2$), ^b*wR*₂ = [$\sum [w(F_o^2 - F_c^2)^2] / \sum [w(F_o^2)^2]$]^{1/2}; $w = 1 / [\sigma^2(F_o^2) + (0.095P)^2]$; $P = [\max(F_o^2, 0) + 2F_c^2]/3$ (also with $F_o^2 > 2\sigma F^2$)

Table S2. Bond lengths [\AA] for **II**.

Compound II			
Bond	Length [\AA]	Bond	Length [\AA]
C(1)-C(9)	1.440(2)	C(14)-C(15)	1.372(3)
C(1)-B(2)	1.702(3)	C(15)-C(16)	1.374(3)
C(1)-B(3)	1.703(3)	B(1)-B(8)	1.760(3)
C(1)-C(2)	1.708(3)	B(1)-B(9)	1.771(3)
C(1)-B(4)	1.723(3)	B(1)-B(2)	1.781(3)
C(1)-B(1)	1.727(3)	B(2)-B(8)	1.771(3)
C(2)-C(3)	1.506(2)	B(2)-B(7)	1.772(3)
C(2)-B(5)	1.703(3)	B(2)-B(3)	1.778(3)
C(2)-B(9)	1.708(3)	B(3)-B(7)	1.770(3)
C(2)-B(1)	1.738(3)	B(3)-B(6)	1.777(3)
C(2)-B(4)	1.742(3)	B(3)-B(4)	1.779(3)
C(3)-C(4)	1.388(3)	B(4)-B(6)	1.761(3)
C(3)-C(8)	1.390(3)	B(4)-B(5)	1.769(3)
C(4)-C(5)	1.384(3)	B(5)-B(10)	1.773(3)
C(5)-C(6)	1.378(3)	B(5)-B(6)	1.777(3)
C(6)-C(7)	1.368(3)	B(5)-B(9)	1.783(3)
C(7)-C(8)	1.390(3)	B(6)-B(7)	1.777(3)
C(9)-C(10)	1.187(2)	B(6)-B(10)	1.783(3)
C(10)-C(11)	1.435(2)	B(7)-B(10)	1.772(3)
C(11)-C(16)	1.392(3)	B(7)-B(8)	1.783(3)
C(11)-C(12)	1.392(3)	B(8)-B(9)	1.777(3)
C(12)-C(13)	1.386(3)	B(8)-B(10)	1.781(3)
C(13)-C(14)	1.378(4)	B(9)-B(10)	1.781(3)

Symmetry transformations used to generate equivalent atoms:

Table S3. Bond lengths [\AA] for **III**.

Compound III			
Bond	Length [\AA]	Bond	Length [\AA]
C(1)-C(3)	1.502(2)	B(1)-B(5)	1.780(3)
C(1)-C(2)	1.697(2)	B(2)-B(3)	1.777(3)
C(1)-B(3)	1.698(2)	B(2)-B(7)	1.780(3)
C(1)-B(2)	1.706(3)	B(2)-B(6)	1.781(3)
C(1)-B(4)	1.745(3)	B(3)-B(7)	1.777(3)
C(1)-B(1)	1.746(2)	B(3)-B(4)	1.780(3)
C(2)-C(9)	1.436(2)	B(3)-B(8)	1.780(3)
C(2)-B(5)	1.708(3)	B(4)-B(8)	1.756(3)
C(2)-B(9)	1.710(3)	B(4)-B(9)	1.778(3)
C(2)-B(4)	1.722(3)	B(5)-B(10)	1.769(3)
C(2)-B(1)	1.722(3)	B(5)-B(6)	1.775(3)
C(3)-C(4)	1.372(3)	B(5)-B(9)	1.778(3)
C(3)-C(8)	1.379(3)	B(6)-B(10)	1.777(3)
C(4)-C(5)	1.381(3)	B(6)-B(7)	1.782(3)
C(5)-C(6)	1.368(3)	B(7)-B(10)	1.777(4)
C(6)-C(7)	1.361(3)	B(7)-B(8)	1.785(3)
C(7)-C(8)	1.380(3)	B(8)-B(9)	1.776(3)
C(9)-C(9) ^{#1}	1.182(3)	B(8)-B(10)	1.780(3)
B(1)-B(6)	1.757(3)	B(9)-B(10)	1.770(3)
B(1)-B(2)	1.765(3)		

Symmetry transformations used to generate equivalent atoms: #1 -x+1,-y+1,-z

Table S4. Bond angles [°] for **II**.

Compound II	Bond	Angles [°]	Bond	Angles [°]
C(9)-C(1)-B(2)	123.29(16)	C(1)-B(3)-B(4)	59.27(11)	
C(9)-C(1)-B(3)	123.12(16)	B(7)-B(3)-B(4)	107.37(16)	
B(2)-C(1)-B(3)	62.96(13)	B(6)-B(3)-B(4)	59.36(12)	
C(9)-C(1)-C(2)	114.96(14)	B(2)-B(3)-B(4)	107.86(16)	
B(2)-C(1)-C(2)	111.07(14)	C(1)-B(4)-C(2)	59.07(11)	
B(3)-C(1)-C(2)	111.33(14)	C(1)-B(4)-B(6)	104.93(15)	
C(9)-C(1)-B(4)	116.19(15)	C(2)-B(4)-B(6)	105.60(15)	
B(2)-C(1)-B(4)	114.14(14)	C(1)-B(4)-B(5)	105.26(15)	
B(3)-C(1)-B(4)	62.58(12)	C(2)-B(4)-B(5)	58.05(11)	
C(2)-C(1)-B(4)	61.02(11)	B(6)-B(4)-B(5)	60.48(12)	
C(9)-C(1)-B(1)	116.15(16)	C(1)-B(4)-B(3)	58.15(12)	
B(2)-C(1)-B(1)	62.57(13)	C(2)-B(4)-B(3)	106.23(15)	
B(3)-C(1)-B(1)	114.34(14)	B(6)-B(4)-B(3)	60.26(13)	
C(2)-C(1)-B(1)	60.78(11)	B(5)-B(4)-B(3)	108.45(15)	
B(4)-C(1)-B(1)	113.06(14)	C(2)-B(5)-B(4)	60.18(11)	
C(3)-C(2)-B(5)	122.00(15)	C(2)-B(5)-B(10)	105.72(15)	
C(3)-C(2)-C(1)	117.66(14)	B(4)-B(5)-B(10)	107.78(16)	
B(5)-C(2)-C(1)	108.87(14)	C(2)-B(5)-B(6)	106.51(15)	
C(3)-C(2)-B(9)	124.21(15)	B(4)-B(5)-B(6)	59.53(12)	
B(5)-C(2)-B(9)	63.00(12)	B(10)-B(5)-B(6)	60.27(13)	
C(1)-C(2)-B(9)	109.17(14)	C(2)-B(5)-B(9)	58.63(11)	
C(3)-C(2)-B(1)	119.20(15)	B(4)-B(5)-B(9)	108.34(15)	
B(5)-C(2)-B(1)	112.98(14)	B(10)-B(5)-B(9)	60.11(13)	
C(1)-C(2)-B(1)	60.17(11)	B(6)-B(5)-B(9)	108.50(16)	
B(9)-C(2)-B(1)	61.82(12)	B(4)-B(6)-B(3)	60.39(12)	
C(3)-C(2)-B(4)	115.62(14)	B(4)-B(6)-B(7)	107.87(15)	
B(5)-C(2)-B(4)	61.77(12)	B(3)-B(6)-B(7)	59.74(13)	
C(1)-C(2)-B(4)	59.92(11)	B(4)-B(6)-B(5)	59.99(12)	
B(9)-C(2)-B(4)	113.14(14)	B(3)-B(6)-B(5)	108.17(15)	
B(1)-C(2)-B(4)	111.61(14)	B(7)-B(6)-B(5)	107.54(16)	
C(4)-C(3)-C(8)	118.93(17)	B(4)-B(6)-B(10)	107.73(16)	
C(4)-C(3)-C(2)	119.27(16)	B(3)-B(6)-B(10)	107.66(16)	
C(8)-C(3)-C(2)	121.80(16)	B(7)-B(6)-B(10)	59.70(13)	
C(5)-C(4)-C(3)	120.54(19)	B(5)-B(6)-B(10)	59.75(13)	
C(6)-C(5)-C(4)	120.16(19)	B(3)-B(7)-B(10)	108.45(16)	
C(7)-C(6)-C(5)	119.77(18)	B(3)-B(7)-B(2)	60.27(13)	
C(6)-C(7)-C(8)	120.77(19)	B(10)-B(7)-B(2)	108.15(16)	
C(7)-C(8)-C(3)	119.82(18)	B(3)-B(7)-B(6)	60.12(13)	
C(10)-C(9)-C(1)	174.9(2)	B(10)-B(7)-B(6)	60.31(13)	
C(9)-C(10)-C(11)	177.8(2)	B(2)-B(7)-B(6)	108.35(15)	
C(16)-C(11)-C(12)	119.83(17)	B(3)-B(7)-B(8)	108.14(15)	
C(16)-C(11)-C(10)	120.37(17)	B(10)-B(7)-B(8)	60.14(14)	
C(12)-C(11)-C(10)	119.79(18)	B(2)-B(7)-B(8)	59.75(13)	

C(13)-C(12)-C(11)	119.1(2)	B(6)-B(7)-B(8)	108.33(16)
C(14)-C(13)-C(12)	120.4(2)	B(1)-B(8)-B(2)	60.61(12)
C(15)-C(14)-C(13)	120.3(2)	B(1)-B(8)-B(9)	60.09(12)
C(14)-C(15)-C(16)	120.2(2)	B(2)-B(8)-B(9)	108.63(15)
C(15)-C(16)-C(11)	120.1(2)	B(1)-B(8)-B(10)	108.02(15)
C(1)-B(1)-C(2)	59.05(11)	B(2)-B(8)-B(10)	107.79(17)
C(1)-B(1)-B(8)	104.57(15)	B(9)-B(8)-B(10)	60.09(13)
C(2)-B(1)-B(8)	105.39(15)	B(1)-B(8)-B(7)	108.17(16)
C(1)-B(1)-B(9)	105.47(15)	B(2)-B(8)-B(7)	59.82(13)
C(2)-B(1)-B(9)	58.26(11)	B(9)-B(8)-B(7)	107.96(16)
B(8)-B(1)-B(9)	60.43(13)	B(10)-B(8)-B(7)	59.63(13)
C(1)-B(1)-B(2)	58.03(12)	C(2)-B(9)-B(1)	59.92(11)
C(2)-B(1)-B(2)	106.05(15)	C(2)-B(9)-B(8)	105.94(16)
B(8)-B(1)-B(2)	60.00(13)	B(1)-B(9)-B(8)	59.48(12)
B(9)-B(1)-B(2)	108.41(16)	C(2)-B(9)-B(10)	105.18(16)
C(1)-B(2)-B(8)	105.16(15)	B(1)-B(9)-B(10)	107.52(16)
C(1)-B(2)-B(7)	104.65(15)	B(8)-B(9)-B(10)	60.08(13)
B(8)-B(2)-B(7)	60.43(13)	C(2)-B(9)-B(5)	58.37(11)
C(1)-B(2)-B(3)	58.52(12)	B(1)-B(9)-B(5)	107.72(15)
B(8)-B(2)-B(3)	108.31(16)	B(8)-B(9)-B(5)	107.67(17)
B(7)-B(2)-B(3)	59.81(13)	B(10)-B(9)-B(5)	59.69(12)
C(1)-B(2)-B(1)	59.40(11)	B(7)-B(10)-B(5)	107.95(15)
B(8)-B(2)-B(1)	59.39(13)	B(7)-B(10)-B(8)	60.23(13)
B(7)-B(2)-B(1)	107.68(16)	B(5)-B(10)-B(8)	107.88(15)
B(3)-B(2)-B(1)	108.14(15)	B(7)-B(10)-B(9)	108.24(16)
C(1)-B(3)-B(7)	104.72(16)	B(5)-B(10)-B(9)	60.20(12)
C(1)-B(3)-B(6)	105.10(16)	B(8)-B(10)-B(9)	59.83(13)
B(7)-B(3)-B(6)	60.14(13)	B(7)-B(10)-B(6)	59.99(13)
C(1)-B(3)-B(2)	58.52(12)	B(5)-B(10)-B(6)	59.98(13)
B(7)-B(3)-B(2)	59.92(13)	B(8)-B(10)-B(6)	108.15(16)
B(6)-B(3)-B(2)	108.08(17)	B(9)-B(10)-B(6)	108.34(14)

Symmetry transformations used to generate equivalent atoms:

Table S5. Bond angles [°] for **III**.

Compound III	Bond	Angles [°]	Bond	Angles [°]
C(3)-C(1)-C(2)	117.43(13)	C(2)-B(4)-B(8)	104.84(14)	
C(3)-C(1)-B(3)	122.40(14)	C(1)-B(4)-B(8)	105.13(14)	
C(2)-C(1)-B(3)	109.35(13)	C(2)-B(4)-B(9)	58.47(11)	
C(3)-C(1)-B(2)	123.95(15)	C(1)-B(4)-B(9)	106.13(14)	
C(2)-C(1)-B(2)	108.99(13)	B(8)-B(4)-B(9)	60.34(12)	
B(3)-C(1)-B(2)	62.96(11)	C(2)-B(4)-B(3)	104.52(14)	
C(3)-C(1)-B(4)	116.14(14)	C(1)-B(4)-B(3)	57.57(11)	
C(2)-C(1)-B(4)	60.01(11)	B(8)-B(4)-B(3)	60.45(13)	
B(3)-C(1)-B(4)	62.24(11)	B(9)-B(4)-B(3)	108.44(15)	
B(2)-C(1)-B(4)	113.32(14)	C(2)-B(5)-B(10)	104.49(14)	
C(3)-C(1)-B(1)	118.67(14)	C(2)-B(5)-B(6)	104.73(15)	
C(2)-C(1)-B(1)	60.01(10)	B(10)-B(5)-B(6)	60.19(13)	
B(3)-C(1)-B(1)	112.80(14)	C(2)-B(5)-B(9)	58.71(11)	
B(2)-C(1)-B(1)	61.51(11)	B(10)-B(5)-B(9)	59.85(12)	
B(4)-C(1)-B(1)	111.47(14)	B(6)-B(5)-B(9)	108.18(15)	
C(9)-C(2)-C(1)	114.39(14)	C(2)-B(5)-B(1)	59.13(11)	
C(9)-C(2)-B(5)	123.01(15)	B(10)-B(5)-B(1)	107.35(15)	
C(1)-C(2)-B(5)	111.67(14)	B(6)-B(5)-B(1)	59.25(13)	
C(9)-C(2)-B(9)	123.69(15)	B(9)-B(5)-B(1)	108.13(14)	
C(1)-C(2)-B(9)	111.51(13)	B(1)-B(6)-B(5)	60.53(12)	
B(5)-C(2)-B(9)	62.71(12)	B(1)-B(6)-B(10)	108.01(16)	
C(9)-C(2)-B(4)	116.35(15)	B(5)-B(6)-B(10)	59.75(13)	
C(1)-C(2)-B(4)	61.39(10)	B(1)-B(6)-B(2)	59.86(12)	
B(5)-C(2)-B(4)	114.19(13)	B(5)-B(6)-B(2)	108.32(15)	
B(9)-C(2)-B(4)	62.40(11)	B(10)-B(6)-B(2)	107.87(15)	
C(9)-C(2)-B(1)	115.47(15)	B(1)-B(6)-B(7)	107.90(15)	
C(1)-C(2)-B(1)	61.42(10)	B(5)-B(6)-B(7)	107.96(16)	
B(5)-C(2)-B(1)	62.53(12)	B(10)-B(6)-B(7)	59.93(14)	
B(9)-C(2)-B(1)	114.17(14)	B(2)-B(6)-B(7)	59.97(13)	
B(4)-C(2)-B(1)	113.82(13)	B(3)-B(7)-B(10)	107.91(16)	
C(4)-C(3)-C(8)	118.19(18)	B(3)-B(7)-B(2)	59.94(12)	
C(4)-C(3)-C(1)	119.96(16)	B(10)-B(7)-B(2)	107.87(15)	
C(8)-C(3)-C(1)	121.85(17)	B(3)-B(7)-B(6)	107.93(14)	
C(3)-C(4)-C(5)	120.8(2)	B(10)-B(7)-B(6)	59.91(13)	
C(6)-C(5)-C(4)	120.6(2)	B(2)-B(7)-B(6)	59.99(12)	
C(7)-C(6)-C(5)	118.9(2)	B(3)-B(7)-B(8)	59.98(13)	
C(6)-C(7)-C(8)	120.9(2)	B(10)-B(7)-B(8)	59.96(13)	
C(3)-C(8)-C(7)	120.5(2)	B(2)-B(7)-B(8)	107.89(15)	
C(9) ^{#1} -C(9)-C(2)	175.7(3)	B(6)-B(7)-B(8)	107.88(16)	
C(2)-B(1)-C(1)	58.58(10)	B(4)-B(8)-B(9)	60.43(12)	
C(2)-B(1)-B(6)	104.87(15)	B(4)-B(8)-B(10)	108.03(15)	
C(1)-B(1)-B(6)	105.53(14)	B(9)-B(8)-B(10)	59.69(13)	

C(2)-B(1)-B(2)	105.15(13)	B(4)-B(8)-B(3)	60.43(12)
C(1)-B(1)-B(2)	58.11(10)	B(9)-B(8)-B(3)	108.50(14)
B(6)-B(1)-B(2)	60.73(13)	B(10)-B(8)-B(3)	107.67(16)
C(2)-B(1)-B(5)	58.34(11)	B(4)-B(8)-B(7)	108.30(15)
C(1)-B(1)-B(5)	106.03(14)	B(9)-B(8)-B(7)	107.91(15)
B(6)-B(1)-B(5)	60.22(13)	B(10)-B(8)-B(7)	59.82(13)
B(2)-B(1)-B(5)	108.77(15)	B(3)-B(8)-B(7)	59.81(13)
C(1)-B(2)-B(1)	60.38(11)	C(2)-B(9)-B(10)	104.37(15)
C(1)-B(2)-B(3)	58.31(11)	C(2)-B(9)-B(8)	104.48(15)
B(1)-B(2)-B(3)	108.15(15)	B(10)-B(9)-B(8)	60.26(13)
C(1)-B(2)-B(7)	105.36(15)	C(2)-B(9)-B(4)	59.12(11)
B(1)-B(2)-B(7)	107.60(16)	B(10)-B(9)-B(4)	107.53(16)
B(3)-B(2)-B(7)	59.95(12)	B(8)-B(9)-B(4)	59.23(12)
C(1)-B(2)-B(6)	106.25(15)	C(2)-B(9)-B(5)	58.58(11)
B(1)-B(2)-B(6)	59.41(12)	B(10)-B(9)-B(5)	59.81(13)
B(3)-B(2)-B(6)	107.97(16)	B(8)-B(9)-B(5)	108.05(16)
B(7)-B(2)-B(6)	60.04(13)	B(4)-B(9)-B(5)	108.13(15)
C(1)-B(3)-B(2)	58.73(11)	B(5)-B(10)-B(9)	60.34(13)
C(1)-B(3)-B(7)	105.82(14)	B(5)-B(10)-B(6)	60.07(13)
B(2)-B(3)-B(7)	60.11(12)	B(9)-B(10)-B(6)	108.48(15)
C(1)-B(3)-B(4)	60.19(10)	B(5)-B(10)-B(7)	108.41(16)
B(2)-B(3)-B(4)	108.28(14)	B(9)-B(10)-B(7)	108.53(16)
B(7)-B(3)-B(4)	107.58(16)	B(6)-B(10)-B(7)	60.16(13)
C(1)-B(3)-B(8)	106.13(14)	B(5)-B(10)-B(8)	108.31(15)
B(2)-B(3)-B(8)	108.21(15)	B(9)-B(10)-B(8)	60.06(13)
B(7)-B(3)-B(8)	60.21(13)	B(6)-B(10)-B(8)	108.29(16)
B(4)-B(3)-B(8)	59.12(12)	B(7)-B(10)-B(8)	60.22(13)
C(2)-B(4)-C(1)	58.60(10)		

Symmetry transformations used to generate equivalent atoms: #1 -x+1, -y+1, -z

Computational method. The calculations were performed with the *Gaussian16* package, using the density functional theory (DFT) and time-dependent density functional theory (TD-DFT) method with the B3LYP hybrid functional. All structures were optimized using DFT or TD-DFT method with a 6-31G(d,p) basis set. The starting coordinates were obtained from the X-ray structures. The calculations of the anion species were performed in the spin unrestricted regime. All calculations were performed in the gas phase.

Table S6. Optimized geometry coordinates for **I**.

Atom	x	y	z	Atom	x	y	z
C	2.372990	2.753790	-1.193885	B	-1.176926	-0.009037	1.411491
C	1.182791	2.028967	-1.196289	B	-2.941994	-0.009428	1.443365
C	2.988240	3.093017	0.010814	H	3.915482	-3.657923	-0.011601
C	2.397339	-2.709640	-1.214272	H	-4.499255	1.536850	0.009834
C	1.206961	-1.984955	-1.214415	H	-4.499368	-1.536647	-0.009950
C	0.587488	1.624821	0.008264	H	2.858286	2.978434	2.159887
C	-0.723015	0.879370	0.005960	H	-1.918626	2.434048	1.515338
C	2.988194	-3.093048	-0.010657	H	0.753743	1.711023	2.158849
C	2.397035	2.710070	1.214400	H	2.814606	-3.057348	2.138184
C	0.587421	-1.624864	-0.008254	H	-1.919369	-2.453295	1.482715
C	-0.723072	-0.879389	-0.006042	H	-0.452677	-0.016521	2.338830
C	1.206670	1.985368	1.214469	H	0.710105	-1.789583	2.140695
C	2.372585	-2.754291	1.193993	H	-3.515458	-0.016614	2.480754
C	1.182375	-2.029483	1.196316	H	2.815294	3.056483	-2.138060
B	-1.176898	0.008999	-1.411577	H	0.710788	1.788702	-2.140708
B	-2.941987	0.009560	-1.443474	H	-0.452650	0.016494	-2.338903
B	-2.068979	1.448514	-0.877465	H	2.858874	-2.977636	-2.159725
B	-2.069381	-1.437226	-0.897125	H	-3.515431	0.016836	-2.480885
B	-3.506714	0.888566	0.005855	H	0.754304	-1.710253	-2.158823
B	-3.506767	-0.888454	-0.005965	H	3.915521	3.657906	0.011816
B	-2.069279	1.437254	0.897033	H	-1.919245	2.453319	-1.482780
B	-2.069110	-1.448474	0.877366	H	-1.918743	-2.434044	-1.515416

Table S7. Optimized geometry coordinates for **II**.

Atom	x	y	z	Atom	x	y	z
C	-0.992789	-1.108877	0.014698	H	5.822366	-2.898748	-0.487707
C	-1.822206	0.413703	0.023664	C	3.844080	-2.090722	-0.277712
C	-1.019726	1.690408	-0.006459	H	3.344895	-3.031035	-0.486559
C	-0.837240	2.377647	-1.216335	B	-1.786544	-0.574448	1.453904
H	-1.258457	1.981554	-2.132043	H	-1.105885	-0.251524	2.358281
C	-0.125733	3.574960	-1.254422	B	-1.871382	-2.269720	0.913356
H	0.001794	4.090064	-2.201504	H	-1.239284	-3.066982	1.514777
C	0.415807	4.109085	-0.085214	B	-1.887620	-2.270599	-0.867787
H	0.970035	5.042343	-0.115357	H	-1.266523	-3.068990	-1.479296
C	0.230314	3.440297	1.124688	B	-1.811563	-0.575010	-1.409797
H	0.634648	3.851434	2.044655	H	-1.141495	-0.234123	-2.316440
C	-0.483977	2.243544	1.165893	B	-3.278059	0.258252	-0.851713
H	-0.628775	1.746539	2.116367	H	-3.645229	1.209386	-1.450560
C	0.435510	-1.067321	0.003929	B	-3.354677	-1.426639	-1.409290
C	1.646707	-1.016109	-0.001912	H	-3.872327	-1.705921	-2.438210
C	3.072068	-0.945719	-0.005006	B	-3.401164	-2.477549	0.037391
C	3.718907	0.275382	0.265792	H	-3.958414	-3.523685	0.043062
H	3.121454	1.156774	0.474562	B	-3.329156	-1.426706	1.482644
C	5.109144	0.343374	0.262140	H	-3.825236	-1.707732	2.521604
H	5.601051	1.288532	0.471189	B	-3.261902	0.260439	0.927965
C	5.869508	-0.796763	-0.008167	H	-3.611113	1.213748	1.534338
H	6.953782	-0.739192	-0.009329	B	-4.261671	-0.920036	0.045777
C	5.233826	-2.010990	-0.277158	H	-5.443797	-0.833743	0.055406

Table S8. Optimized geometry coordinates for **III**.

Atom	x	y	z	Atom	x	y	z
C	-2.976471	3.788453	-0.888616	H	-5.938849	-2.592139	-0.087309
C	-2.783084	4.378561	0.360179	H	-2.828867	1.585947	2.287331
C	-3.103793	2.405483	-1.000607	H	-5.096983	-0.110081	1.601601
C	-2.730135	3.575201	1.498960	H	3.027906	-4.403245	-1.781915
C	-3.035120	1.586765	0.137085	H	3.992435	2.916187	2.272078
C	-2.858421	2.191723	1.390607	H	5.097149	0.109780	1.601237
C	-3.210688	0.094218	0.015850	H	-3.447871	-4.365027	-0.386619
C	1.816000	0.918975	-0.156092	H	-2.164772	-0.416676	2.261443
C	0.538763	0.273182	-0.161643	H	2.165047	0.416441	2.261168
C	-0.538792	-0.272854	-0.161607	H	-1.118384	-2.922558	1.124418
C	-1.816013	-0.918708	-0.155986	H	-3.991925	-2.916577	2.272139
C	3.210875	-0.094334	0.015524	H	2.680472	-5.455966	0.446011
C	3.102176	-2.405535	-1.001003	H	2.831382	-1.585969	2.287271
C	2.974603	-3.788472	-0.888879	H	2.594194	-4.022858	2.478365
C	3.035054	-1.586820	0.136784	B	3.868025	2.082826	-1.607124
C	2.782537	-4.378549	0.360135	B	2.158769	2.262830	-1.158464
C	2.859697	-2.191756	1.390511	B	3.394065	3.185965	-0.280374
C	2.731178	-3.575198	1.498999	B	2.839831	0.653597	-1.516359
H	-3.031004	4.403223	-1.781580	B	-2.839903	-0.653447	-1.516297
H	-2.681213	5.456005	0.445957	B	-4.494728	-0.589549	-0.873593
H	4.261804	2.456838	-2.660144	B	4.836249	2.158886	-0.106615
H	-3.265664	1.965884	-1.976840	B	4.494777	0.589483	-0.873959
H	1.281371	2.647983	-1.850178	B	-3.867933	-2.082757	-1.606994
H	3.448365	4.365042	-0.386476	B	2.061953	2.426327	0.613730
H	-2.417028	0.021538	-2.383207	B	-2.158622	-2.262590	-1.158496
H	-5.258257	0.164933	-1.368676	B	3.712032	2.350227	1.269587
H	2.416873	-0.021118	-2.383436	B	-4.836007	-2.159046	-0.106399
H	-4.261790	-2.456712	-2.660007	B	-4.399970	-0.754428	0.897017
H	5.939144	2.591847	-0.087559	B	4.400167	0.754191	0.896677
H	-2.592090	4.022878	2.478170	B	2.687104	0.917924	1.332979
H	5.258138	-0.164996	-1.369311	B	-3.393711	-3.185954	-0.280391
H	3.263035	-1.965971	-1.977416	B	-2.686898	-0.918103	1.333270
H	1.118867	2.922689	1.124707	B	-2.061631	-2.426267	0.613671
H	-1.281162	-2.647592	-1.850215	B	-3.711672	-2.350467	1.269691

Table S9. Optimized geometry coordinates for \mathbf{I}^- .

Atom	x	y	z	Atom	x	y	z
C	3.254688	2.274956	1.203184	B	0.000052	-0.956246	-1.221065
C	2.454801	1.135422	1.204531	B	-0.000213	-2.775812	-1.434435
C	3.659916	2.858041	-0.000341	H	-4.282370	3.749101	0.000253
C	-3.252057	2.277010	1.203971	H	1.490137	-4.438778	-0.000072
C	-2.452136	1.137497	1.205507	H	-1.490133	-4.438779	0.000072
C	2.032261	0.536444	-0.000545	H	3.559687	2.714820	-2.150894
C	1.195049	-0.675110	-0.000637	H	2.466004	-1.988255	-1.587412
C	-3.659918	2.858038	0.000339	H	2.144107	0.695677	-2.146858
C	3.252059	2.277008	-1.203972	H	-3.564396	2.711152	-2.150175
C	-2.032261	0.536443	0.000545	H	-2.466036	-1.987671	-1.586528
C	-1.195049	-0.675111	0.000639	H	-0.000702	-0.236651	-2.168242
C	2.452138	1.137495	-1.205507	H	-2.148860	0.691962	-2.145793
C	-3.254693	2.274949	-1.203186	H	-0.000730	-3.343151	-2.482869
C	-2.454806	1.135416	-1.204531	H	3.564387	2.711163	2.150173
B	-0.000052	-0.956245	1.221065	H	2.148853	0.691971	2.145793
B	0.000216	-2.775813	1.434435	H	0.000702	-0.236653	2.168243
B	1.495554	-2.009817	0.897673	H	-3.559682	2.714826	2.150892
B	-1.495586	-2.010315	0.898463	H	0.000733	-3.343151	2.482870
B	0.879707	-3.414554	-0.000119	H	-2.144101	0.695683	2.146859
B	-0.879704	-3.414555	0.000120	H	4.282368	3.749105	-0.000257
B	1.495588	-2.010313	-0.898462	H	2.466038	-1.987670	1.586529
B	-1.495551	-2.009817	-0.897672	H	-2.466003	-1.988258	1.587413

Table S10. Optimized geometry coordinates for $\text{H}^{\cdot-}$.

Atom	x	y	z	Atom	x	y	z
C	-0.431780	-1.462745	0.000095	H	6.087619	0.077960	-2.151003
C	-2.085260	0.254424	-0.000206	C	4.210520	-0.368415	-1.212187
C	-1.790453	1.699859	-0.000105	H	3.677777	-0.494254	-2.149267
C	-1.650267	2.417678	-1.205124	B	-1.471368	-0.790712	1.232537
H	-1.754713	1.889416	-2.146405	H	-0.932465	-0.297679	2.169554
C	-1.384311	3.784483	-1.203782	B	-1.181387	-2.605269	0.903275
H	-1.282567	4.309959	-2.150431	H	-0.492480	-3.286159	1.592000
C	-1.247421	4.480205	-0.000032	B	-1.181677	-2.604706	-0.903572
H	-1.036759	5.546480	0.000003	H	-0.492821	-3.285422	-1.592539
C	-1.384391	3.784427	1.203680	B	-1.471439	-0.790297	-1.232376
H	-1.282690	4.309872	2.150352	H	-0.932617	-0.297649	-2.169643
C	-1.650367	2.417630	1.204962	B	-3.262621	-0.449360	-0.896770
H	-1.754893	1.889315	2.146200	H	-3.920262	0.266468	-1.582874
C	0.938410	-1.172096	0.000103	B	-2.772524	-2.057370	-1.434522
C	2.123173	-0.867420	0.000050	H	-3.177666	-2.454076	-2.482501
C	3.500832	-0.537157	0.000018	B	-2.617593	-3.132468	-0.000004
C	4.210341	-0.367574	1.212207	H	-2.929386	-4.282864	-0.000069
H	3.677478	-0.492764	2.149305	B	-2.772196	-2.057618	1.434660
C	5.563609	-0.046074	1.206471	H	-3.177423	-2.454279	2.482616
H	6.087315	0.079448	2.150974	B	-3.261913	-0.449459	0.897025
C	6.251510	0.116573	-0.000014	H	-3.919066	0.266342	1.583639
H	7.308548	0.367952	-0.000023	B	-3.842882	-1.869622	0.000146
C	5.563778	-0.046911	-1.206490	H	-5.002470	-2.144983	0.000237

Table S11. Optimized geometry coordinates for **III⁻**.

Atom	x	y	z	Atom	x	y	z
C	4.109775	-3.283769	-1.203361	H	5.568208	3.534931	-0.000242
C	4.207453	-3.981614	0.001676	H	3.840749	-1.372454	2.148425
C	3.913482	-1.905168	-1.205191	H	5.385270	0.867494	1.572342
C	4.111084	-3.282613	1.206157	H	-4.182827	3.814995	-2.148465
C	3.812485	-1.187279	0.000550	H	-3.754107	-3.203899	2.477589
C	3.914790	-1.904020	1.206874	H	-5.385130	-0.868118	1.572031
C	3.632030	0.284958	0.000005	H	2.830963	4.785840	-0.001090
C	-1.576930	-1.231383	-0.000737	H	2.442564	0.392211	2.221161
C	-0.461317	-0.405568	-0.000629	H	-2.442366	-0.392049	2.220694
C	0.462070	0.405625	-0.000551	H	0.940320	2.931656	1.589514
C	1.577725	1.231421	-0.000531	H	3.754760	3.203768	2.478068
C	-3.632101	-0.284892	-0.000265	H	-4.354578	5.058049	0.002156
C	-3.914232	1.905214	-1.205330	H	-3.841503	1.372288	2.148246
C	-4.110980	3.283750	-1.203421	H	-4.185412	3.812646	2.151632
C	-3.813068	1.187266	0.000366	B	-3.521540	-2.681516	-1.436814
C	-4.208959	3.981488	0.001657	B	-1.839299	-2.572970	-0.907607
C	-3.915674	1.903903	1.206730	B	-2.977810	-3.605675	-0.001256
C	-4.112422	3.282433	1.206095	B	-2.770599	-1.042914	-1.289711
H	4.181492	-3.814975	-2.148436	B	2.771104	1.043251	-1.289705
H	4.352704	-5.058224	0.002112	B	4.528928	1.332668	-0.895348
H	-3.754678	-3.202438	-2.479683	B	-4.578419	-2.874728	-0.000834
H	3.838423	-1.374512	-2.147171	B	-4.528531	-1.332809	-0.895766
H	-0.940272	-2.930621	-1.592064	B	3.522322	2.681683	-1.436403
H	-2.830020	-4.785754	-0.001621	B	-1.839054	-2.573527	0.905429
H	2.443406	0.391423	-2.221626	B	1.840044	2.573219	-0.907251
H	5.385932	0.866821	-1.571875	B	-3.521147	-2.682351	1.434993
H	-2.442929	-0.390952	-2.221566	B	4.579120	2.874520	-0.000324
H	3.755645	3.202653	-2.479200	B	4.528492	1.333022	0.895304
H	-5.567367	-3.535356	-0.000862	B	-4.528205	-1.333287	0.894923
H	4.183839	-3.812914	2.151660	B	-2.770216	-1.043736	1.288714
H	-5.385597	-0.867220	-1.572403	B	2.978613	3.605750	-0.000797
H	-3.838940	1.374634	-2.147338	B	2.770562	1.043894	1.289247
H	-0.939808	-2.931607	1.589372	B	1.839687	2.573646	0.905707
H	0.940978	2.930894	-1.591631	B	3.521749	2.682316	1.435442

Table S12. TD-DFT calculated transitions for **I**.

Excited state	Energ y (eV)	Wavelengt h (nm)	<i>f</i>	Transitions			
1	5.0816	243.99	0.0077	57→80 68→81 69→81 71→80 74→79 74→84 75→81 75→82	-0.01040 -0.01272 0.01505 -0.02194 -0.12328 0.01252 -0.18339 -0.01145	76→78 76→80 76→83 77→78 77→80 77→85 75→81 77→80	0.60281 -0.02467 0.02037 0.06966 0.28214 0.02129 -0.01324 0.01227
2	5.2103	237.96	0.1286	57→78 58→78 65→78 67→82 69→84 71→78 73→82 74→81 75→79	-0.01147 -0.01011 -0.01719 -0.02410 -0.01134 0.01756 0.01612 0.02943 0.09473	75→84 76→78 76→80 77→78 77→80 77→83 77→87 75→79 77→78	-0.01462 -0.07390 -0.08184 0.68796 -0.01736 -0.03032 0.01545 0.01453 -0.01892
3	5.3273	232.73	0.0901	60→78 65→79 68→78 69→78 74→80 75→78 75→83	0.01362 -0.01173 -0.01453 0.02317 -0.06896 0.65914 -0.03255	75→87 76→79 76→81 77→79 75→78 77→79	0.01473 -0.02084 0.04402 0.23404 -0.01439 0.01583
4	5.6116	220.94	0.0018	71→80 72→78 74→79 74→84 75→81 76→78	0.01067 -0.01365 0.52991 -0.02241 0.11217 0.30014	76→80 76→83 76→87 77→78 77→80	0.02050 -0.05879 0.01624 0.02482 -0.33197
5	5.8437	212.17	0.0098	65→78 68→79 71→78 74→81 75→79 75→84	0.01228 -0.01028 -0.02602 -0.01272 0.59086 0.01635	76→80 77→78 77→80 77→83 75→79 77→78	0.38105 -0.03153 0.01881 0.02688 -0.01636 0.01165

Table S13. TD-DFT calculated transitions for **II**.

Excited state	Energ y (eV)	Wavelengt h (nm)	<i>f</i>	Transitions			
1	4.5593	271.93	0.0519	73→85	-0.01021	79→86	0.01167
				74→84	-0.02900	79→87	0.02587
				74→85	0.03196	79→88	0.02336
				74→87	0.01798	79→92	-0.01562
				74→88	0.01598	81→88	0.01275
				74→89	0.01012	82→85	-0.01115
				77→84	-0.02568	83→84	-0.44503
				77→85	0.02527	83→85	0.50745
				77→87	0.01336	83→86	0.05868
				77→88	0.01175	83→87	0.11546
				78→84	0.02212	83→88	0.10473
				78→85	-0.02444	83→89	0.05732
				78→87	-0.01083	83→92	-0.04401
				79→84	-0.03548	83→97	0.01312
				79→85	0.02134	83→84	0.01209
2	4.8151	257.49	0.4665	57→101	-0.01005	79→97	-0.02681
				72→87	0.01282	79→98	-0.01027
				72→88	0.01169	80→84	-0.02263
				73→85	-0.01683	80→85	0.02079
				73→87	-0.01315	80→87	0.01497
				73→88	-0.01197	81→87	0.04321
				74→84	0.03579	81→88	-0.04757
				74→85	0.02517	82→84	-0.02950
				76→85	-0.01128	82→85	0.03619
				77→84	0.03966	83→84	0.50279
				77→85	0.02340	83→85	0.43018
				77→89	-0.01163	83→86	0.04949
				78→84	-0.03082	83→87	0.07802
				78→85	-0.02569	83→88	0.07259
				78→87	-0.01147	83→89	-0.05613
				78→88	-0.01051	83→91	-0.01180
				79→84	0.07194	83→92	-0.01674
				79→85	-0.07317	83→94	0.02416
				79→86	-0.03594	79→92	0.01029
				79→87	-0.07672	83→84	-0.03639
				79→88	-0.06939	83→85	-0.01697
				79→89	-0.03069	83→89	-0.01314
				79→92	0.05048		
3	4.9722	249.36	0.0015	74→87	0.02090	81→85	0.16858
				74→88	-0.02374	81→86	-0.01547
				77→87	0.02562	81→87	-0.02049
				77→88	-0.02858	81→88	-0.01913
				78→87	-0.02542	81→89	0.07093
				78→88	0.02817	81→94	-0.01636
				80→84	0.04461	82→84	-0.01304
				80→85	0.01557	83→87	-0.31296
				81→84	0.50549	83→88	0.32349
4	5.1594	240.31	0.0101	72→84	0.01691	80→88	0.02192
				72→86	0.01172	80→90	-0.02303
				73→84	-0.01656	81→84	-0.04414
				75→84	-0.01078	81→85	0.04531
				75→86	-0.01066	82→84	-0.15719

				76→84	-0.01522	82→85	0.12650
				76→86	0.02254	82→86	0.29684
				79→84	-0.22522	82→87	-0.06271
				79→85	-0.08542	82→88	-0.06429
				79→94	-0.02638	82→91	0.02440
				80→84	0.40446	83→84	0.03168
				80→85	-0.34450	83→87	0.02544
				80→86	0.09814	83→88	0.03039
				80→87	0.02097	82→86	0.01553
5	5.2932	234.23	0.0239	67→84	-0.01508	80→86	-0.04424
				70→84	-0.01031	81→85	-0.01527
				72→84	0.01689	81→87	0.01074
				72→85	0.01218	81→88	-0.01175
				73→84	-0.02409	82→84	0.53221
				73→85	-0.01127	82→85	0.07690
				74→87	-0.01037	82→86	-0.04468
				76→84	-0.02786	82→89	-0.04301
				77→87	-0.01114	83→84	0.02124
				78→84	-0.01615	83→85	0.08624
				79→84	-0.29991	83→86	-0.13508
				79→85	-0.10737	83→87	-0.13804
				79→87	-0.01147	83→88	-0.13100
				79→88	-0.01061	83→89	-0.04597
				79→94	-0.03127	83→92	0.02067
				80→84	0.14428	83→97	-0.01188
				80→85	0.10304		
6	5.5219	224.53	0.1392	71→85	0.01239	80→85	0.19103
				71→89	-0.01304	80→86	0.13281
				72→85	-0.01773	81→84	-0.01099
				74→84	-0.01532	81→86	-0.01521
				75→85	0.01599	82→84	-0.08504
				76→84	0.01308	82→85	0.62016
				76→85	-0.01974	82→86	-0.02990
				78→84	0.02232	82→87	0.11310
				78→85	0.01383	82→88	0.10674
				78→89	-0.01360	82→90	-0.03762
				79→84	-0.02757	82→97	-0.01328
				79→85	0.12236	83→85	-0.01063
				79→86	0.01738	83→88	0.02307
				79→87	0.02451	83→89	0.04823
				79→88	0.02192	82→84	0.01151
				80→84	-0.03172	82→85	-0.02543
7	5.7733	214.76	0.1868	62→84	-0.01119	79→97	-0.02262
				72→84	-0.01588	80→84	0.03409
				73→84	0.03344	80→85	-0.06874
				74→84	-0.04049	80→86	-0.07819
				74→85	-0.01409	80→87	-0.01946
				74→94	-0.01278	80→88	-0.02144
				76→84	0.01257	81→87	0.03235
				76→85	0.01927	81→88	-0.03278
				77→84	-0.03716	82→84	0.01871
				77→85	-0.01524	82→85	-0.08242
				77→94	-0.01111	82→86	-0.04821
				78→84	0.06843	82→87	-0.05602
				78→85	0.01519	82→88	-0.04896
				78→89	-0.01680	83→84	0.10890
				79→84	-0.19091	83→85	0.04402
				79→85	0.62561	83→86	-0.01502
				79→86	-0.01263	83→89	0.11005
				79→87	-0.01423	83→94	-0.03106
				79→88	-0.01262	83→108	0.01236
				79→92	0.01118	83→84	-0.01747

Table S14. TD-DFT calculated transitions for III.

Excited state	Energ y (eV)	Wavelengt h (nm)	<i>f</i>	Transitions			
1	4.9420	250.88	0.0043	92→120	0.03582	115→139	0.01584
				92→123	-0.01791	115→142	-0.01259
				92→132	0.01323	116→121	-0.08933
				99→127	-0.01077	116→122	-0.05449
				107→120	0.02015	116→124	0.01459
				108→120	0.04062	117→120	0.43875
				108→123	-0.01531	117→125	-0.02984
				112→120	0.03242	118→120	0.20229
				112→123	-0.01168	118→123	-0.01262
				114→121	-0.01164	118→125	0.08993
				114→122	0.01318	118→131	-0.01085
				115→120	0.46339	119→121	0.02146
				115→123	-0.11235	119→124	0.09432
				115→132	0.03359	119→130	-0.01212
2	4.9640	249.77	0.0101	109→125	-0.01007	118→124	-0.13373
				114→120	-0.01577	118→130	0.01556
				116→120	0.61100	119→120	-0.23566
				116→123	0.02782	119→123	0.02487
				116→125	-0.02370	119→125	-0.13967
				117→121	-0.13051	119→131	0.01453
				117→122	-0.08562	118→124	-0.01012
				117→124	0.04524	119→125	-0.01053
				118→121	-0.05899		
3	4.9807	248.93	0.0019	92→120	0.03471	115→139	0.01497
				92→123	-0.01716	115→142	-0.01204
				92→132	0.01366	116→121	0.10984
				99→127	-0.01045	116→122	0.06476
				107→120	0.01715	116→124	-0.01359
				108→120	0.03804	117→120	-0.42553
				108→123	-0.01248	117→123	-0.02959
				112→120	0.02921	117→125	0.03021
				112→123	-0.01033	118→120	-0.12525
				113→127	-0.01006	118→125	-0.10468
				114→121	-0.01049	119→120	0.02286
				114→122	0.01477	119→121	-0.02693
				115→120	0.48946	119→122	0.01483
				115→123	-0.11277	119→124	-0.10818
				115→132	0.03206	119→130	0.01123
4	4.9847	248.73	0.1071	93→120	0.01259	117→124	-0.01785
				102→120	-0.01118	118→120	0.01048
				114→120	0.05830	118→122	0.01392
				114→123	-0.01687	118→124	-0.03908
				115→120	-0.01402	118→127	-0.02049
				115→121	-0.01052	118→129	0.01387
				115→122	0.01664	119→120	0.65215
				116→120	0.23196	119→123	-0.08518
				116→125	0.02325	119→125	-0.04072
				117→120	0.01567	119→128	-0.01392
				117→121	-0.04149	119→132	0.01111
				117→122	-0.01425		

5	5.0034	247.80	0.0017	96→120 115→120 115→126 116→121 116→124 117→120 117→123 117→125 118→120	-0.01204 -0.04682 0.01333 0.01514 0.02948 -0.24299 0.02815 -0.02580 0.65153	118→123 118→125 118→128 118→132 119→121 119→122 119→124 119→127 119→129	-0.08676 -0.02213 -0.01489 0.01014 0.01316 0.01032 -0.01317 -0.02257 0.01570
6	5.3500	231.75	0.0005	109→125 111→124 114→120 114→123 115→121 115→122 116→120 116→121 116→122 116→123 116→125 116→128 117→120 117→121 117→122 117→123 117→124 117→125	0.01317 0.01080 -0.07976 0.01331 0.04205 -0.04884 0.23760 0.07145 0.03563 -0.25461 0.02134 -0.03000 0.04424 0.39694 0.20593 -0.04871 -0.06485 0.01153	117→127 117→129 118→120 118→121 118→122 118→123 118→124 118→125 118→130 119→120 119→121 119→122 119→123 119→124 119→125 119→126 119→131	0.02440 -0.02806 0.01328 0.20479 -0.06537 -0.01159 0.21051 -0.04088 -0.01936 -0.02194 -0.03143 0.02006 0.01544 -0.04019 0.22580 -0.01684 -0.01824
7	5.3518	231.67	0.0021	109→124 111→125 114→120 114→122 114→124 115→120 116→120 116→121 116→122 116→123 116→124 116→127 116→129 117→120 117→121 117→122 117→123	-0.01329 -0.01093 0.01483 0.01005 0.01024 -0.03403 -0.04603 0.41537 0.20333 0.04965 -0.03048 0.02604 -0.02914 0.22712 -0.06892 -0.03514 -0.25152	117→124 117→125 117→128 118→120 118→121 118→122 118→123 118→124 118→125 118→126 118→131 119→121 119→122 119→124 119→125 119→126 119→130	0.01178 0.05620 -0.02920 0.06908 -0.03756 0.01342 -0.06096 -0.04182 -0.21977 0.01870 0.01822 -0.16142 0.11098 -0.22290 -0.03821 0.02035
8	5.3844	230.27	0.0112	86→120 88→120 92→121 92→122 92→133 93→120 99→126 102→120 102→123 103→121 104→120 105→121 106→120 107→121 108→121 108→122 108→133 109→120	0.01144 -0.01201 -0.01748 0.02685 0.02064 -0.01652 0.01338 -0.03997 0.01543 -0.01117 -0.02951 -0.01209 -0.01041 -0.01571 -0.01434 0.02727 0.01065 0.04162	112→121 112→122 112→133 114→120 114→123 114→132 115→121 115→122 115→124 115→130 115→133 116→120 116→123 117→122 118→121 118→122 118→124 119→120	-0.01692 0.02603 0.01028 0.53123 -0.09097 0.02460 -0.24696 0.32531 0.01523 -0.02665 0.05721 0.03575 -0.03886 0.08651 0.07657 0.02861 0.03602 -0.06741

				111→121 111→127	-0.01268 0.01071	119→123 119→125	0.02879 0.03810
9	5.4349	228.13	0.0001	93→121	-0.01006	117→120	-0.03914
				93→122	0.01407	117→123	0.05717
				96→126	-0.01584	117→126	-0.02256
				102→121	0.01075	118→125	0.07053
				102→122	-0.01334	118→126	0.08051
				111→126	-0.01524	119→121	-0.41521
				114→121	-0.04727	119→122	0.51708
				114→122	0.06466	119→124	0.09029
				116→121	-0.16291		
10	5.4504	227.48	0.0039	93→126	0.01288	117→121	0.21487
				96→121	0.01299	117→122	-0.09635
				96→122	-0.01863	117→124	-0.01160
				109→126	-0.01144	118→121	-0.39255
				111→122	-0.01275	118→122	0.51589
				115→122	0.02016	118→124	0.08132
				116→120	0.03252	119→125	0.06423
				116→123	-0.04733	119→126	0.08619

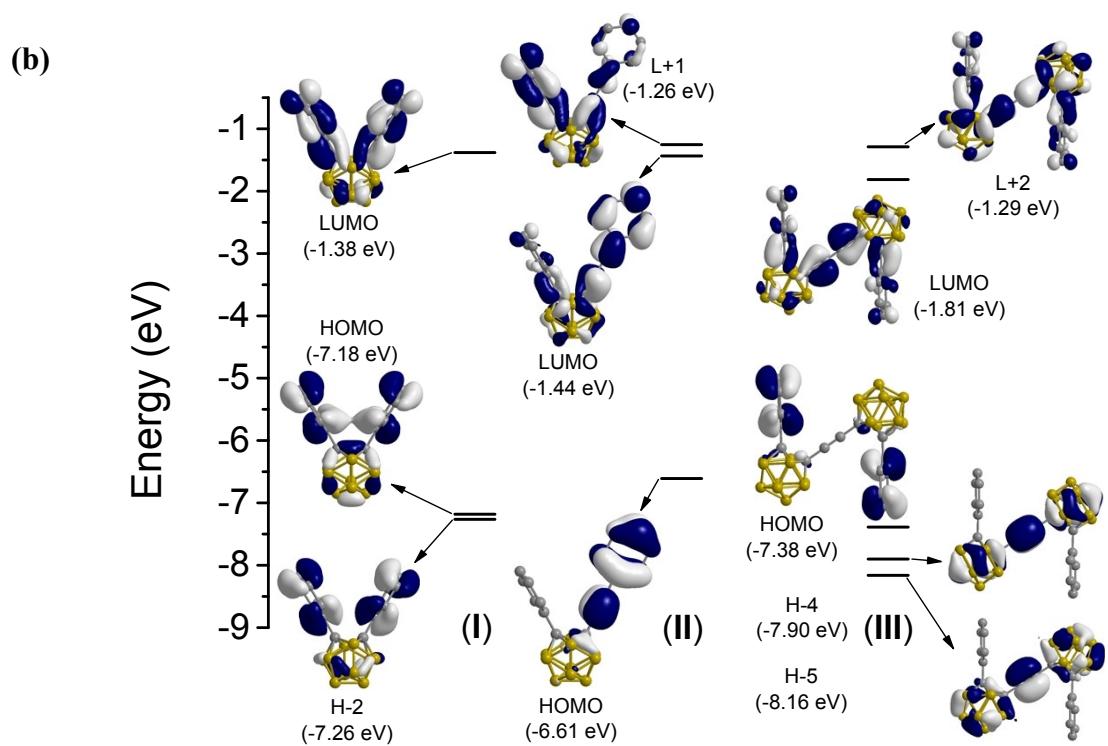
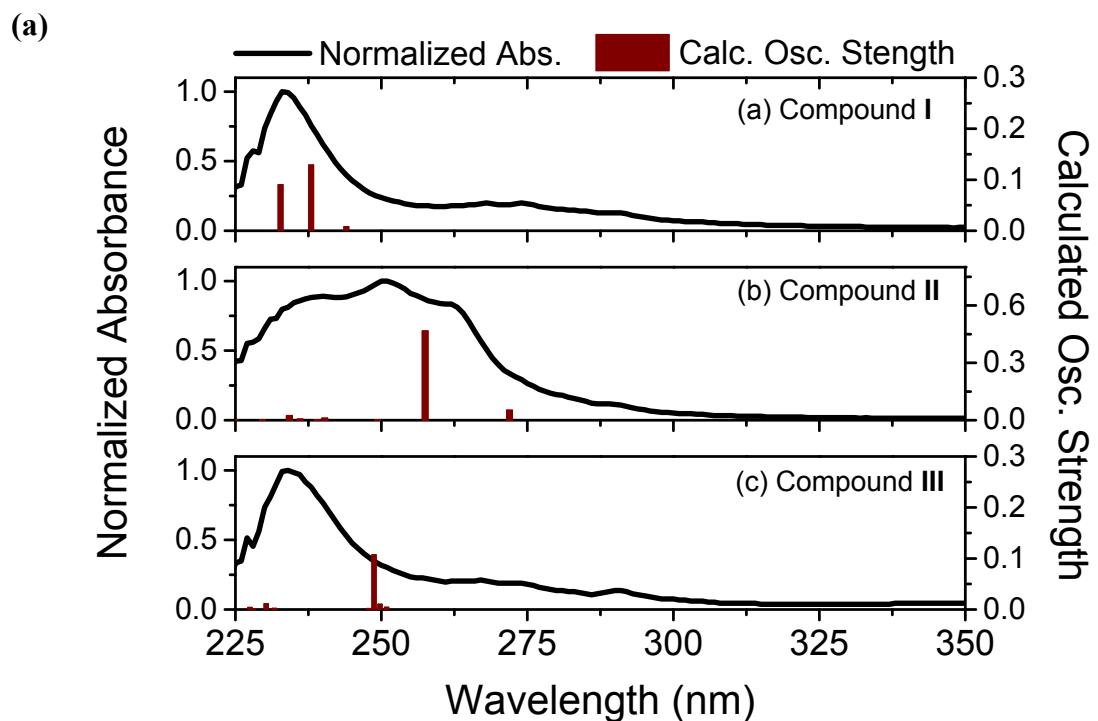


Figure S5. (a) TD-DFT calculated oscillation strengths with measured absorption spectra and (b) isodensity plots for the orbitals included in the major transitions of **I–III**, and their energy levels.

Table S15. Selected TD-DFT calculated transitions for **I**–**III** and their major contributions.

Compound	Wavelength (nm)	Osc. Strength	Major Contributions
I	237.96	0.1286	HOMO → LUMO (95%)
	232.73	0.0901	H-2 → LUMO (87%)
II	271.94	0.0519	HOMO → LUMO (40%), HOMO → L+1 (52%)
	257.49	0.4665	HOMO → LUMO (51%), HOMO → L+1 (37%)
III	248.73	0.1071	HOMO → LUMO (85%)
	230.27	0.0112	H-5 → LUMO (56%), H-4 → L+2 (21%)

Geometrical changes between neutral and anionic species. The optimised structures of the anionic species, **I**[–]–**III**[–], are depicted in Figure S6 along with the neutral species for comparison. Note that the atom labels are differed from Figure 2 for comparison. The most notable structural differences between the neutral and the anionic geometries of **I**–**III** are the lengthening of the carboranyl C–C bonds (C1–C2) upon reduction. Along with this structural change, **II**[–] and **III**[–] exhibit important geometrical changes. The bond lengths between the carboranyl atom and the phenyl substituent (C1–C3) and between the carboranyl carbon and ethynyl group (C2–C4) in **II**[–] and **III**[–] are slightly shorter than those of their corresponding neutral species, while the ethynyl C–C bonds (C4–C5) are longer than that of the neutral species (Table S6). Moreover, the bond distance between the phenyl-substituted carbon and neighbouring boron in the carborane cage of **II**[–] contracted upon reduction, while the distance between the ethynyl-substituted carbon and the neighbouring boron was extended. This trend is even more pronounced in **III**[–]. Therefore, we confirmed that the added electron is stabilized through the σ^* orbital of the carborane cage and the π^* orbital of the ethynyl group, rather than the π^* orbital of the phenyl group.

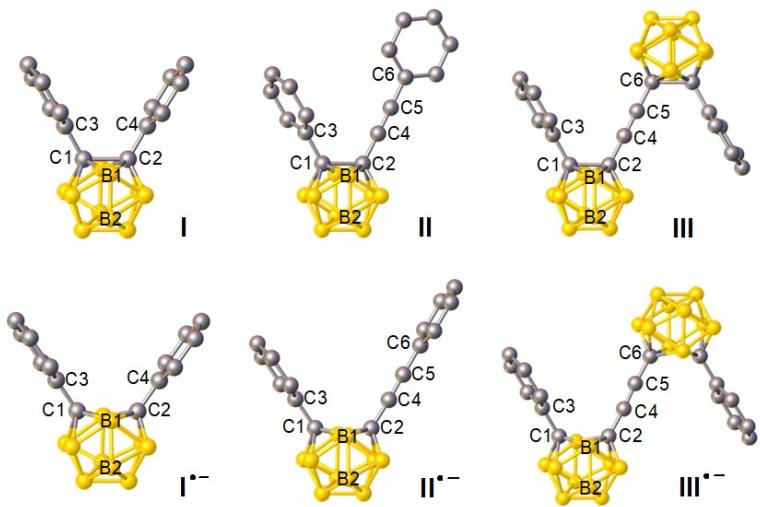


Figure S6. Optimized structures of **I–III** and **I^{•-}–III^{•-}**, with selected atoms labelled (note: the atom labels are differed from Figure 2 for comparison).

Table S16. Selected bond lengths (\AA) in the DFT-optimised geometries of the neutral and radical-anionic states of **I–III**. Atom labels are provided in Figure 4a.

	I	I^{•-}	II	II^{•-}	III	III^{•-}
C1–C2	1.759	2.390	1.734	2.384	1.733	2.262
C1–C3	1.508	1.473	1.508	1.475	1.508	1.483
C2–C4	1.508	1.473	1.429	1.401	1.431	1.388
C4–C5	–	–	1.212	1.223	1.208	1.229
C5–C6	–	–	1.427	1.417	1.431	1.388
C1–B1	1.724	1.731	1.739	1.729	1.742	1.726
C2–B1	1.725	1.732	1.728	1.747	1.725	1.767
B1–B2	1.765	1.832	1.763	1.827	1.762	1.808

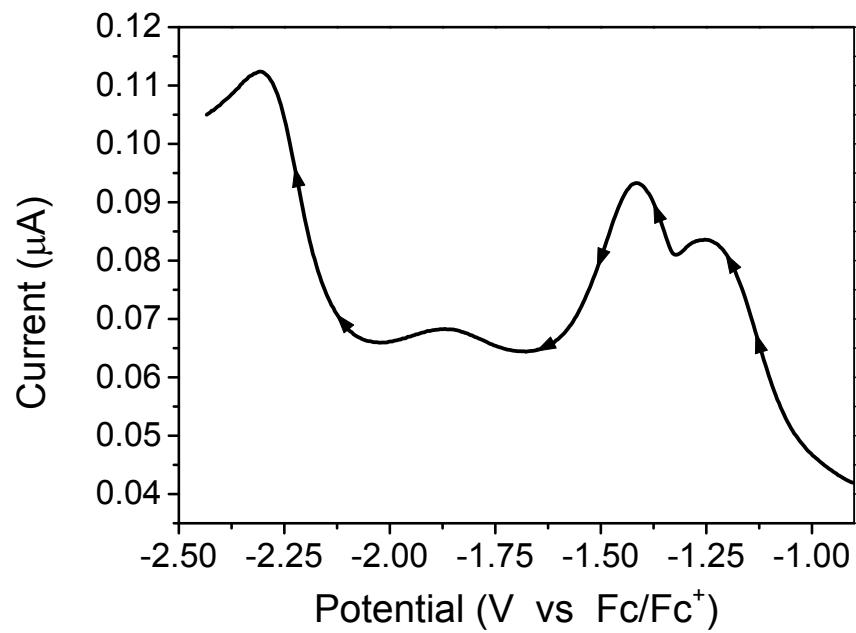


Figure S7. Square wave voltammetry scan for **III** from positive to negative potentials.

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