

Supporting Information

Low-spin 1,1'-diphoshametalloenates of Chromium and Iron

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S1. Experimental Details

General considerations

All syntheses and manipulations were conducted under UHP argon (AirGas) with rigorous exclusion of oxygen and water using Schlenk line and glove box techniques. Molecular sieves (4 Å, Sigma Aldrich, 8-12 mesh) were activated at 200 °C for 12 hours, at 5×10^{-4} mbar. All solvents were degassed (by bubbling Ar through them, or under dynamic vacuum) prior to drying procedures. Anhydrous THF (Sigma Aldrich) was degassed, refluxed over K/Ph₂CO until a persistent purple color was obtained, then distilled and stored over 4 Å molecular sieves. Anhydrous hexane and toluene (Sigma Aldrich) were degassed, refluxed over NaK/Ph₂CO for several days until dark coloration (purple solutions or purple/green solids) were observed, then distilled and stored over 4 Å molecular sieves. All solvents were tested with a dilute THF solution of Na₂Ph₂CO (150 mg Ph₂CO in 20 mL of THF with an excess of Na metal) such that ethereal solvents required 1 drop / mL to retain purple coloration and hydrocarbon solvents required 1 drop / 2 mL. Anhydrous C₆D₆ and C₄D₈O (both Sigma Aldrich) for NMR spectroscopy were degassed, stirred with NaK for 72 hours and filtered through a glass-fiber filter disc prior to use. KC₈ was prepared by heating graphite flakes (200 mesh, Sigma Aldrich) and 1/8th of a molar equivalent of freshly cleaned K metal together in a glass scintillation vial inside an argon glovebox with vigorous stirring using a metal spatula. 2-butyne was vacuum transferred and stored over 4 Å sieves for 4 days prior to use. 2.2.2-cryptand (Alfa Aesar) was dried under vacuum (5×10^{-4} mbar) for 48 hours and stored in a glovebox. PhPCl₂, anhydrous CrCl₂ (Alfa Aesar), anhydrous FeCl₂ (Alfa Aesar) and anhydrous CoCl₂ (Fisher Scientific) were used as received. [M(TMP)₂] (**1-M**, M = Cr or Fe) were prepared using a procedure slightly modified from those previously described;^{1,2} anhydrous FeCl₂ and CrCl₂ were combined with base free KTMP in THF at room temperature, and the complexes were crystallized from hexane. The glovebox atmosphere was periodically checked using a dilute toluene solution of [Ti(Cp)₂(μ-Cl)]₂ (200 mg of commercial [Ti(Cp)₂(Cl)]₂) reduced over an excess of Zn powder in 20 mL of toluene, and filtered. A drop of the green solution was allowed to dry on a vial lid (Urea, VWR) prior to any manipulations, such that the residue dried to a dark green color each time (a color change to yellow or orange indicates decomposition of the Ti test compound and atmospheric O₂/H₂O removal is required). All glassware, and glass-fiber filter discs, were stored in an oven (>150 °C) for 24 hours prior to being brought into the glovebox, and FEP (fluorinated ethylene propylene) NMR liners were brought into the box *via* overnight vacuum cycles. The NMR solutions were loaded into a fresh FEP NMR liner, stoppered with a PTFE plug, and then placed in J. Young tap appended NMR tubes. This was done as it facilitates the easy handling of material in facilities primarily concerned with Th/dU work and is a local requirement for the area this work had to be performed in. Crystals for single-crystal X-ray diffraction were mounted on nylon loops in Paratone-N oil. Solution phase electronic absorption spectra were collected at ambient temperature using a Varian Cary 6000i UV-vis-

residue. Multiple crops of colorless crystals were collected by repeated concentration and cooling of the orange solution. This material was then fully desolvated *in vacuo* to afford a white powder, which was confirmed to be base-free by elemental analysis. (TMP)₂ was prepared as previously described, and purified by crystallization from hexane.⁵

S1.2. Synthesis of 2-M (M = Cr, Fe)

Synthesis of [K(2.2.2-crypt)][Cr(TMP)₂] 2-Cr: A solution of **1-Cr** (55 mg, 0.17 mmol) and 2.2.2-crypt (62 mg, 0.17 mmol) in THF (1 mL) was prepared in a 20 mL glass scintillation vial equipped with a Teflon stir bar, then cooled to –35°C for 1 hour. Separately, a vial of KC₈ was also chilled to –35°C along with an aluminum vial block. Cold KC₈ (25 mg, 0.18 mmol, 1.1 eq.) was added rapidly to the vigorously stirred THF solution which was maintained close to –35°C by the vial block. The solution immediately turned from dark red to dark brown/black. The brown/black solution was filtered from black solids (presumably graphite) through a glass pipette loaded with a glass-fiber disc padded with a 2 mm layer of KC₈ into a 4 mL glass vial. Hexane (3 mL) was carefully layered on top of the THF solution, and the vial was then stored at –35°C for 18 hours. The pale brown/colorless supernatant was decanted and the brown/black crystals were washed with room temperature hexane (3 x 2 mL) followed by drying *in vacuo* to afford **2-Cr** (yield 102 mg, 82%). CrKP₂N₂O₆C₃₄H₆₀; calcd. C 54.75, H 8.11, N 3.76; found C 54.32, H 8.18, N 3.49. ¹H NMR (C₄D₈O, 400 MHz, 298 K): 2.63, 3.62, 3.65, only peaks attributable to free 2.2.2-cryptand, and to the solvent residual signals were observed. ³¹P{¹H} NMR (C₄D₈O, 162 MHz, 298 K): 74.55, diamagnetic impurity, corresponds to free {TMP}[–] anion. ATR-IR ($\bar{\nu}$, cm^{–1}): 2958 (m), 2935 (w), 2922 (w), 2866 (m), 2825 (s), 2727 (w), 2698 (w), 2360 (m), 2343 (m), 1477 (m), 1454 (w), 1442 (m), 1412 (w), 1398 (w), 1369 (m), 1358 (m), 1350 (s), 1300 (m), 1292 (m), 1257 (m), 1240 (w), 1169 (w), 1132 (s), 1101 (vs), 1080 (vs), 1057 (m), 1018 (m), 947 (s), 931 (s), 831 (m), 820 (m), 752 (m), 681 (w), 669 (w), 658 (vw), 571 (m). UV-vis-nIR (THF) λ_{max} (cm^{–1}, ϵ): 502 (19904, 4180, vbr).

Synthesis of [K(2.2.2-crypt)][Fe(TMP)₂] 2-Fe: A solution of **1-Fe** (150 mg, 0.45 mmol) and 2.2.2-crypt (170 mg, 0.45 mmol) in THF (1 mL) was prepared in a 20 mL glass scintillation vial equipped with a Teflon stir bar, then cooled to –35°C for 1 hour. Separately, a vial of KC₈ was also chilled to –35°C along with an aluminum vial block. Cold KC₈ (67 mg, 0.50 mmol, 1.1 eq.) was added rapidly to the vigorously stirred THF solution which was maintained close to –35°C by the vial block. The solution immediately turned from dark red to dark green/black. The green/black solution was filtered from black solids (presumably graphite) through a glass pipette loaded with a glass-fiber disc padded with a 2 mm layer of KC₈ into a 4 mL glass vial. Hexane (3 mL) was carefully layered on top of the THF solution, and the vial was then stored at –35°C for 18 hours. The pale brown/colorless supernatant was decanted and the green/black crystals were washed with room temperature hexane

(3 x 2 mL) followed by drying in vacuo to afford **2-Fe** (yield 243 mg, 72%). FeKP₂N₂O₆C₃₄H₆₀; calcd. C 54.47, H 8.07, N 3.74; found C 54.96, H 8.35, N 3.61. ¹H NMR (C₄D₈O, 400 MHz, 298 K): 1.94 (s, 12H, TMP-3,4-Me₂), 2.19 (d, ³J_{HP} = 10.29 Hz, 12H, TMP-2,5-Me₂), 2.28 (s, 12H, 2.2.2-crypt), 3.30 (s, 12H, 2.2.2-crypt), 3.37 (s, 24H, 2.2.2-crypt). ³¹P{¹H} NMR (C₄D₈O, 162 MHz, 298 K): 74.46, diamagnetic impurity, corresponds to free {TMP}⁻ anion. ATR-IR ($\bar{\nu}$, cm⁻¹): 2954 (w), 2873 (m), 2835 (m), 2729 (w), 2698 (w), 2360 (w), 2343 (w), 2324 (w), 1475 (m), 1442 (m), 1414 (w), 1390 (w), 1358 (m), 1350 (m), 1296 (m), 1257 (m), 1240 (w), 1173 (w), 1132 (s), 1099 (vs), 1078 (vs), 1020 (m), 947 (s), 931 (s), 829 (m), 820 (m), 796 (w), 752 (m), 679 (w), 667 (w), 638 (vw), 625 (vw), 588 (w), 569 (w), 553 (w), 536 (w). UV-vis-nIR (THF) λ_{max} (cm⁻¹, ϵ): 428 (23375, 2220, vbr), 533 (18754, 1830, vbr), 616 (16239, 1720, vbr), 885 (11297, 1760), 924 (10818, 1730).

S1.3. Attempted extension to Mn and Co

The synthesis of a redox series, such as that for **3-M** and **4-M** (M = Mn, Fe, Co) was desirable. Attempts to synthesize the required **1-Mn** from commercial MnCl₂, MnCl₂(THF)_n prepared *in-situ*, MnCl₂(DME)_n prepared *in-situ*, or MnCl₂(PMe₃) with two equivalents of KTMP in THF, DME or toluene resulted in intractable mixtures of dark powder that were insoluble in all common solvents that are compatible with this ligand set. The synthesis of **1-Co** was attempted using anhydrous CoCl₂ and two equivalents of K-TMP in THF (see below).

Synthesis of [Co(η^5 -TMP)(μ^2 : η^1 : η^1 -TMP)]₂ 7: CoCl₂ (65 mg, 0.5 mmol) was slurried in THF (4 mL) in a 20 mL glass scintillation vial at room temperature. K-TMP (178 mg, 1 mmol) was added to the blue suspension, which immediately turned dark brown. The brown mixture was stirred for 16 hours at room temperature. Volatiles were removed *in vacuo* and the black tacky residue was extracted with hot (60 °C) hexane (6 mL) which afforded a dark brown/black solution which was concentrated to 1.5 mL and stored at room temperature for 16 hours to afford black crystals of **7** along with sticky brown residues (mass 14 mg). We were unable to optimize the yield of this complex. A ³¹P{¹H} NMR taken of the crystalline material indicated several products (Figure S21).

Attempted reduction of crude 7: The reaction above was repeated on the same scale. We then attempted reduction of the crude mixture, assuming an arbitrary 80% conversion of the reactants to **7**, presuming any reduced product might form a salt-like complex as **2-M**, and be separable from **7** based on solubility. After stirring for 16 hours at room temperature, the crude mixture was reduced to dryness in vacuo, and extracted with warm (45 °C) toluene (5 mL), filtered through a glass pipette loaded with a glass-fiber disc into a 20 mL glass scintillation vial, and the brown/black solution was reduced to dryness in

vacuo. Solid 2.2.2-crypt (151 mg, 0.4 mmol, 0.8 eq. assuming 80% conversion) was added along with a Teflon stir bar, and then the tacky black residue was dissolved in THF (2 mL), and cooled to $-35\text{ }^{\circ}\text{C}$. Pre-cooled ($-35\text{ }^{\circ}\text{C}$) solid KC_8 (54 mg, 0.40 mmol, 0.8 eq. assuming 80% conversion) was added to the rapidly stirred solution which was maintained at $-35\text{ }^{\circ}\text{C}$ using a chilled vial block. No perceptible color change was observed. After stirring for 1 minute the black solution was filtered underneath pre-cooled hexane (10 mL) in a 20 mL glass scintillation vial, and stored overnight at $-35\text{ }^{\circ}\text{C}$ which afforded several low quality crystals of $[\text{K}(2.2.2\text{-crypt})][\text{TMP}]$ (**8**), 2.2.2-crypt, and no other crystalline products. The brown solution was decanted from the crystals and reduced to a thin oil in vacuo, which formed biphasic mixtures in hydrocarbon solvents (pentane, hexane, toluene), and pale brown solutions in ethereal solvents (THF, DME, Et_2O), but no further crystalline products could be isolated. When the reaction was repeated at room temperature, the observed colors and biphasic mixtures etc. were all identical, and likewise no crystalline products that contained Co could be isolated in our hands.

S2. Crystallography

General considerations

The crystal data for complexes **1-Cr**, **1-Fe**, **2-Cr**, **2-Fe**, **5** – **8** are compiled in Tables S1-3. Crystals of **1-Fe**, **2-Cr**, **2-Fe**, **7**, and **8** were examined with a Bruker D8 Quest diffractometer equipped with a CMOS detector and using mirror-monochromated Mo K α radiation ($\lambda = 0.71073 \text{ \AA}$) operating in shutterless mode; crystals of **1-Cr** was examined with a Bruker APEX II diffractometer equipped with a CCD detector and using mirror-monochromated Mo K α radiation ($\lambda = 0.71073 \text{ \AA}$); crystals of **5** were examined using an Oxford Diffraction Supernova diffractometer, equipped with CCD area detector and a mirror-monochromated Mo K α radiation ($\lambda = 0.71073 \text{ \AA}$); and crystals of **4** were examined with a Rigaku XtalLAB AFC11 diffractometer, equipped with CCD detector and mirror-monochromated Cu K α radiation ($\lambda = 1.54184 \text{ \AA}$). APEX II (APEX II), or APEX III (D8 Quest), or CrysAlisPro (Rigaku XtalLAB AFC11) software was used for control and solving the unit cells prior to data collection. Intensities were integrated from data recorded on 0.5° frames by ω rotation with 10s frame exposure (**1-Cr**), 13s frame exposure (**1-Fe**), 25s frame exposure (**2-Fe**), 30s frame exposure (**2-Cr**), or 90s frame exposure; or by 0.7° frames by ω rotation with 1s or 2s frame exposure (**6**); or finally by 0.8° frames by ω rotation with 30s frame exposure (**5**). CrysAlisPro was used for final unit cell determination and parameters were refined from the observed positions of all strong reflections in each data set and an analytical absorption correction was applied.⁷ The Olex2⁸ GUI was used for structure solution and refinement utilizing the ShelX software packages.^{9,10} The structures were solved using ShelXT;¹⁰ the datasets were refined by ShelXL⁹ using full-matrix least-squares on all unique F^2 values, with anisotropic displacement parameters for all non-hydrogen atoms, and with constrained riding hydrogen geometries; $U_{\text{iso}}(\text{H})$ was set at 1.2 (1.5 for methyl groups) times U_{eq} of the parent atom. The largest features in final difference syntheses were close to heavy atoms and were of no chemical significance. Olex2 combined with Inkscape was employed for molecular graphics.^{8,11} CCDC 2003385 (**1-Cr**), 2003386 (**1-Fe**), 2003387 (**2-Cr**), 2003388 (**2-Fe**), 2003389 (**5**), 2003390 (**6**), 2017380 (**7**), and 2017381 (**8**) contain the supplementary crystal data for this article.[§] These data can be obtained free of charge from the Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request/cif.

Table S1. Crystallographic data for 1-Cr, 1-Fe, 2-Cr, and 2-Fe.

	1-Cr	1-Fe	2-Cr	2-Fe
Internal identifier	Apx2920	Dq0894	Dq1027	Dq0907
CCDC ref code	2003385	2003386	2003387	2003389
Formula	CrP ₂ C ₁₆ H ₂₄	FeP ₂ C ₁₆ H ₂₄	KCrP ₂ N ₂ O ₆ C ₃₄ H ₆₀	KFeP ₂ N ₂ O ₆ C ₃₄ H ₆₀
Fw	330.29	334.14	745.88	749.73
Crystal syst	Monoclinic	Monoclinic	Triclinic	Triclinic
Space group	<i>P</i> 2 ₁ / <i>n</i>	<i>C</i> 2/ <i>c</i>	<i>P</i> -1	<i>P</i> -1
<i>a</i> , Å	7.8420(3)	14.2948(7)	10.1994(4)	10.1947(2)
<i>b</i> , Å	12.5620(4)	12.8549(6)	14.2193(7)	14.2373(4)
<i>c</i> , Å	8.8611(3)	8.8448(4)	15.0882(6)	15.0677(3)
α , °	90	90	62.328(4)	62.275(2)
β , °	109.444(4)	104.364(5)	86.597(3)	85.840(2)
γ , °	90	90	87.156(3)	87.045(2)
<i>V</i> , Å ³	823.13(5)	1574.50(13)	1933.91(16)	1930.42(8)
<i>Z</i>	2	4	2	2
ρ_{calcd} , g cm ⁻³	1.333	1.410	1.281	1.290
μ , mm ⁻¹	0.874	1.146	0.529	0.624
<i>F</i> (000)	348	704	798	802
Cryst size, mm	0.30 x 0.35 x 0.42	0.16 x 0.19 x 0.29	0.30 x 0.30 x 0.30	0.05 x 0.22 x 0.25
Temperature, K	100(2)	100(2)	100(2)	100(2)
no. reflections (unique)	6774 (1508)	6215 (1605)	23292 (7060)	25573 (7807)
<i>R</i> _{int}	0.017	0.032	0.033	0.036
<i>R</i> ₁ (<i>wR</i> ₂) (<i>F</i> ² > 2 σ (<i>F</i> ²))	0.0252 (0.0667)	0.0312 (0.0789)	0.0338 (0.0872)	0.0400 (0.1035)
<i>S</i> ^a	1.09	1.12	1.10	1.09
min./max. diff map, Å ⁻³	-0.26, 0.33	-0.24, 0.56	-0.25, 0.49	-0.26, 0.62

^a $R = \sum ||F_o| - |F_c|| / \sum |F_o|$; $R_w = [\sum w(F_o^2 - F_c^2)^2 / \sum w(F_o^2)^2]^{0.5}$; $S = [\sum w(F_o^2 - F_c^2)^2 / (\text{no. data} - \text{no. params})]^{0.5}$ for all data.

Table S2. Crystallographic data for 5 – 8.

	5	6	7	8
Internal identifier	Adpm457	Ldpm35	Dq1025	Dq0919
Formula	ZrC ₁₈ H ₂₂	PC ₁₄ H ₁₇	Co ₂ P ₄ C ₃₂ H ₄₈	K ₂ P ₂ O ₁₂ N ₄ C ₅₂ H ₉₆ ·C ₄ H ₈ O
Fw	329.57	216.24	674.44	1181.57
Crystal syst	Monoclinic	Triclinic	Monoclinic	Triclinic
Space group	<i>I</i> 2/a	<i>P</i> -1	<i>C</i> 2/c	<i>P</i> -1
<i>a</i> , Å	14.1554(8)	8.0727(2)	20.161(3)	12.7189(8)
<i>b</i> , Å	9.6976(4)	8.74422(19)	9.0620(4)	15.9437(10)
<i>c</i> , Å	12.0182(6)	18.4065(4)	20.214(3)	17.8937(12)
α , °	90	103.1739(17)	90	64.066(6)
β , °	112.336(6)	92.6248(19)	119.71(2)	85.655(5)
γ , °	90	100.705(2)	90	86.321(5)
<i>V</i> , Å ³	1526.00(15)	1237.81(5)	3207.6(9)	3251.9(4)
<i>Z</i>	4	4	4	2
ρ_{calcd} , g cm ⁻³	1.434	1.160	1.397	1.207
μ , mm ⁻¹	0.705	1.664	1.254	0.254
<i>F</i> (000)	680	464	1416	1280
Cryst size, mm	0.04 x 0.12 x 0.17	0.09 x 0.18 x 0.19	0.10 x 0.12 x 0.40	0.05 x 0.14 x 0.37
Temperature, K	150(2)	150(2)	100(2)	100(2)
no. reflections (unique)	2872 (1558)	13067 (4482)	17531 (2935)	31744 (13040)
<i>R</i> _{int}	0.028	0.019	0.065	0.065
<i>R</i> ₁ (w <i>R</i> ₂) (<i>F</i> ² > 2σ(<i>F</i> ²))	0.0330 (0.0805)	0.0329 (0.0899)	0.0333 (0.0834)	0.0673 (0.1979)
<i>S</i> ^a	1.07	1.04	1.01	1.06
min./max. diff map, Å ⁻³	-0.54, 0.46	-0.26, 0.32	-0.34, 0.50	-0.54, 0.73

^a $R = \sum ||F_o| - |F_c|| / \sum |F_o|$; $R_w = [\sum w(F_o^2 - F_c^2)^2 / \sum w(F_o^2)^2]^{0.5}$; $S = [\sum w(F_o^2 - F_c^2)^2 / (\text{no. data} - \text{no. params})]^{0.5}$ for all data.

S3. Molecular structures

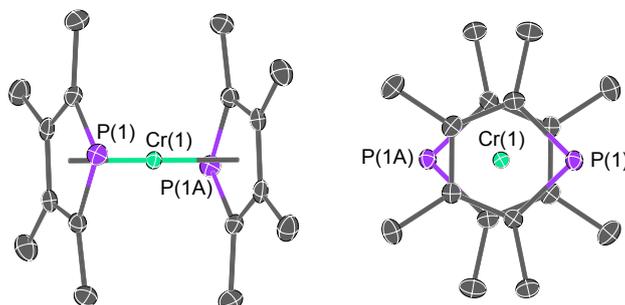


Figure S1. Side view (left) and top view (right) of the molecular structure of **1-Cr** with ellipsoids set at 50%, and hydrogen atoms removed for clarity. Cr(1) sits on a site of symmetry, and thus the inter-ring angle is necessarily 180°. (operations: $x, y, z; 1-x, 1-y, 1-z$).

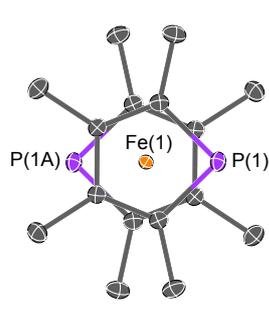


Figure S2. Side view (left) and top view (right) of the molecular structure of **1-Fe** with ellipsoids set at 50%, and hydrogen atoms removed for clarity. Fe(1) sits on a site of symmetry, and thus the inter-ring angle is necessarily 180°. (operations: $x, y, z; \frac{1}{2}-x, \frac{1}{2}-y, 1-z$).

Table S3. Structural parameters for 1-Cr and 1-Fe.

M = Cr or Fe^a	1-Cr	1-Fe
P–M / Å	2.3812(4)	2.2932(4)
TMP _{cent} ···M / Å	1.795(1)	1.660(1)
PC _{2plane} ···C _{4plane} ^b / °	4.45(12)	1.27(13)

^a Prior-reported CCDC codes for **1-Cr**: NORRAN; **1-Fe**: ABIGAU.

^b A plane is defined by the three atoms, P and the two adjoining C atoms; a second plane is defined by all four C atoms of the TMP ring. The “hinge angle”, HA, between these two planes is reported.

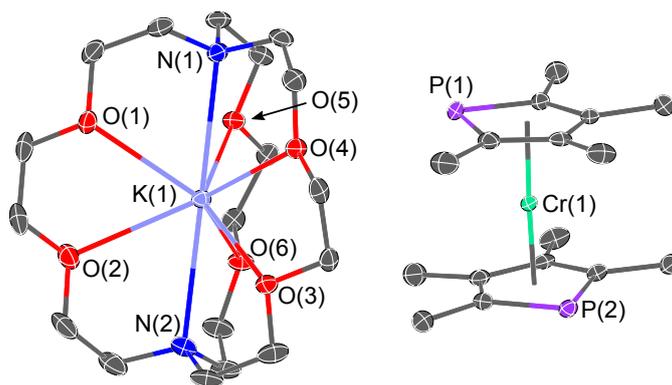


Figure S3. Molecular structure of **2-Cr** with ellipsoids set at 50%, and hydrogen atoms removed for clarity. The structure has been inverted for this image to show the similarity to **2-Fe** (below).

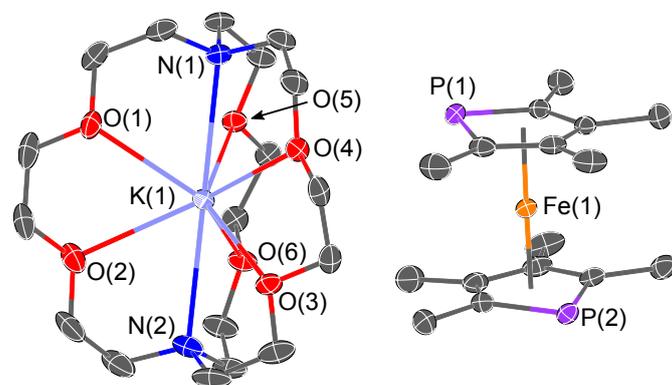


Figure S4. Molecular structure of **2-Fe** with ellipsoids set at 50%, and hydrogen atoms removed for clarity.

Table S4. Structural parameters for 2-Cr and 2-Fe.

M = Cr or Fe	2-Cr	2-Fe
P–M / Å	2.3596(8) / 2.3595(7)	2.3842(8) / 2.4078(7)
TMP _{cent} ···M / Å	1.739(1) / 1.737(1)	1.714(1) / 1.711(1)
TMP _{cent} ···M···TMP _{cent} / °	177.60(4)	178.78(2)
PC _{2plane} ···C _{4plane} ^a / °	3.57(2) / 4.64(2)	6.58(2) / 8.73(2)
P–M–P ring twist / °	139.82(5)	146.35(5)

^a A plane is defined by the three atoms, P and the two adjoining C atoms; a second plane is defined by all four C atoms of the TMP ring. The “hinge angle”, HA, between these two planes is reported.

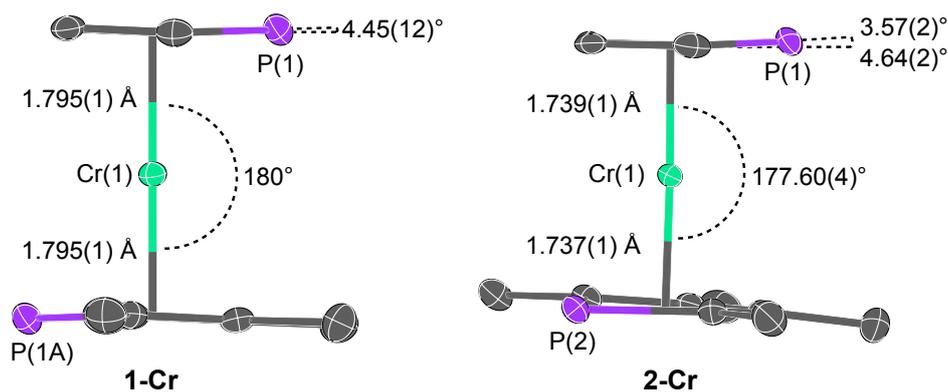


Figure S5. Molecular structure of **1-Cr** (left), and the anion of **2-Cr** (right) with ellipsoids set at 50%. Hydrogen atoms, methyl groups on P(1) ring, and the cation of **2-Cr** removed for clarity. Intended to show structural comparisons of the 1,1'-diphosphametalocene fragment. The pair of numbers for the hinge angle corresponds to the P(1) ring / P(2) ring respectively.

Table S5. Structural comparisons between the 1-Cr and 2-Cr pair.

M = Cr	1-Cr	2-Cr	Δ
P-M / Å	2.3812(4)	2.3596(8) / 2.3595(7)	-0.022(1)
TMP _{cent} ...M / Å	1.795(1)	1.739(1) / 1.737(1)	Avg. -0.057(1)
TMP _{cent} ...M...TMP _{cent} ^b / °	180	177.60(4)	-2.40°
PC _{2plane} ...C _{4plane} ^a / °	4.45(12)	3.57(2) / 4.64(2)	<i>c</i>

^a A plane is defined by the three atoms, P and the two adjoining C atoms; a second plane is defined by all four C atoms of the TMP ring. The “hinge angle”, HA, between these two planes is reported.

^b The propagated error is too large and precludes meaningful analysis beyond that **2-Cr** has approximately the same deviation of the P atoms from the TMP plane as **1-Cr**.

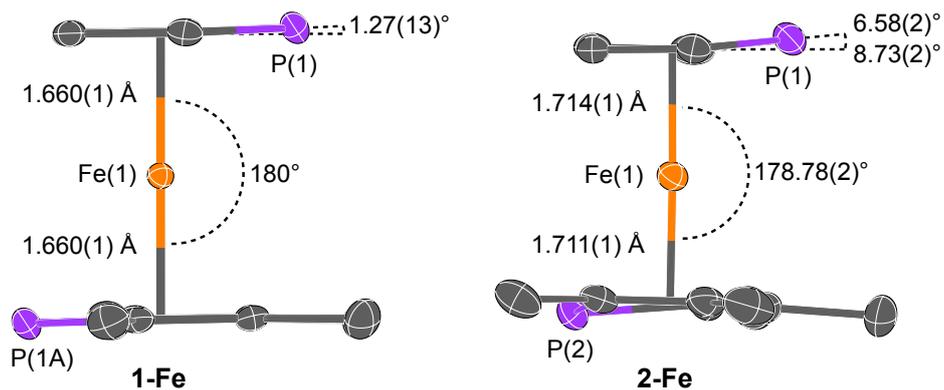


Figure S6. Molecular structure of **1-Fe** (left) and the anion of **2-Fe** (right) with ellipsoids set at 50%. Hydrogen atoms, methyl groups on P(1) ring, and the cation of **2-Fe** removed for clarity. Intended to show structural comparisons of the 1,1'-diphosmetalocene fragment. The pair of numbers for the hinge angle corresponds to the P(1) ring / P(2) ring respectively.

Table S6. Structural comparisons between the 1-Fe and 2-Fe pair.

M = Fe	1-Fe	2-Fe	Δ
P–M / Å	2.2932(4)	2.3842(8) / 2.4078(7)	+0.103(1)
TMP _{cent} ···M / Å	1.660(1)	1.714(1) / 1.711(1)	Avg. +0.053(1)
TMP _{cent} ···M···TMP _{cent} ^b / °	180	178.78(2)	–1.12
PC _{2plane} ···C _{4plane} ^a / °	1.27(13)	6.58(2) / 8.73(2)	<i>c</i>

^a A plane is defined by the three atoms, P and the two adjoining C atoms; a second plane is defined by all four C atoms of the TMP ring. The “hinge angle”, HA, between these two planes is reported.

^b The propagated error is too large and precludes meaningful analysis beyond that **2-Fe** has a much larger deviation of the P atoms from the TMP plane than **1-Fe**.

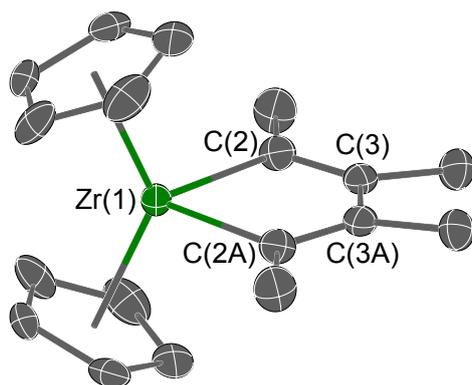


Figure S7. Molecular structure of **5** with ellipsoids set at 50%. A disordered component for both Cp rings, and hydrogen atoms removed for clarity. Zr(1) sits on a site of symmetry, thus half the molecule is symmetry generated (operations: x, y, z ; $\frac{1}{2}-x, y, 1-z$).

$\text{Zr}(1) \cdots \text{Cp}_{\text{cent}} = 2.210(5) \text{ \AA}$; $\text{Zr}(1)-\text{C}(2) = 2.232(3) \text{ \AA}$; $\text{C}(2)-\text{C}(3) = 1.355(4) \text{ \AA}$; $\text{C}(3)-\text{C}(3\text{A}) = 1.505(6) \text{ \AA}$.

$\text{Cp}_{\text{cent}} \cdots \text{Zr}(1) \cdots \text{Cp}_{\text{cent}} = 133.6(2)^\circ$; $\text{C}(2)-\text{Zr}(1)-\text{C}(2\text{A}) = 81.22(15)^\circ$; $\text{Cp}_{\text{cent}} \cdots \text{Zr}(1) \cdots \text{Cp}_{\text{cent}}$ dihedral $\text{C}(2)-\text{Zr}(1)-\text{C}(2\text{A}) = 91.66^\circ$.

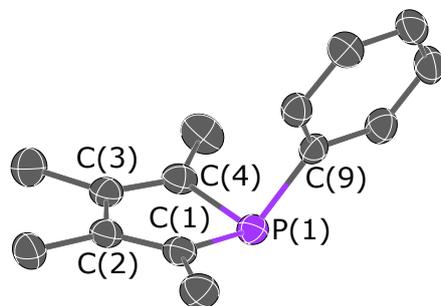


Figure S8. Molecular structure of **6** with ellipsoids set at 50%. A second molecule in the asymmetric unit, and hydrogen atoms removed for clarity.

$\text{P}(1)-\text{C}(1) = 1.7987(13) \text{ \AA}$; $\text{P}(1)-\text{C}(4) = 1.8057(17) \text{ \AA}$; $\text{P}(1)-\text{C}(9) = 1.8313(16) \text{ \AA}$; $\text{C}(1)-\text{C}(2) = 1.353(2) \text{ \AA}$; $\text{C}(2)-\text{C}(3) = 1.477(2) \text{ \AA}$; $\text{C}(3)-\text{C}(4) = 1.347(2) \text{ \AA}$.

$\text{C}(1)-\text{P}(1)-\text{C}(4) = 1.7979(13)^\circ$; $\text{C}(1)-\text{P}(1)-\text{C}(9) = 106.06(6)^\circ$; $\text{C}(4)-\text{P}(1)-\text{C}(9) = 103.36(7)^\circ$.

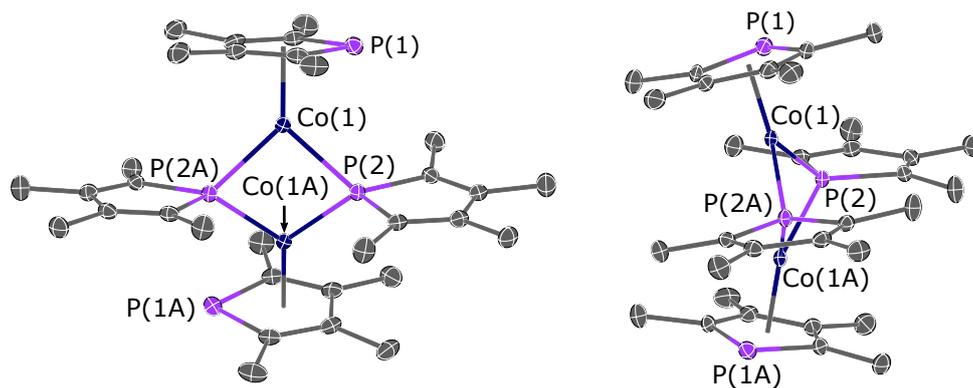


Figure S9. Front view (left) and side view (right) of the molecular structure of **7** with ellipsoids set at 50%, and hydrogen atoms removed for clarity.

$\text{Co}(1)\cdots\text{Co}(1\text{A}) = 2.685(1) \text{ \AA}$; $\text{Co}(1)\text{--P}(1) = 2.2963(10) \text{ \AA}$; $\text{Co}(1)\text{--P}(2) = 2.1697(10) \text{ \AA}$; $\text{P}(1)\text{TMP}_{\text{cent}}\cdots\text{Co}(1) = 1.705(1) \text{ \AA}$.

$\text{TPM}_{\text{cent}}\cdots\text{Co}(1)\cdots\text{Co}(1\text{A}) = 166.57(4)^\circ$; $\text{Co}(1)\text{--P}(2)\text{--Co}(1\text{A}) = 76.38(4)^\circ$; $\text{P}(2)\text{--Co}(1)\text{--P}(2\text{A}) = 94.06(4)^\circ$; $\sum_{\text{Co}(1)\text{--P}(2\text{A})\text{--Co}(1\text{A})\text{--P}(2)} = 340.88^\circ$.

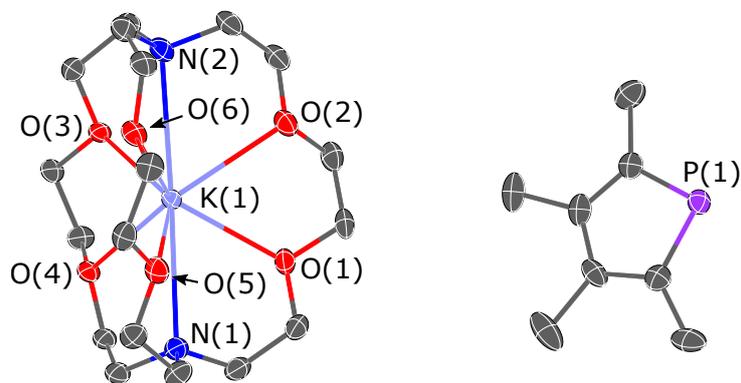


Figure S10. Molecular structure of **8** with ellipsoids set at 50%. Hydrogen atoms, a second set of [K(2.2.2-crypt)] and TMP, and a THF molecule removed for clarity. The extensive disorder components were also removed, and preclude meaningful discussion of the metrical parameters for this compound.

S4. NMR Spectroscopy

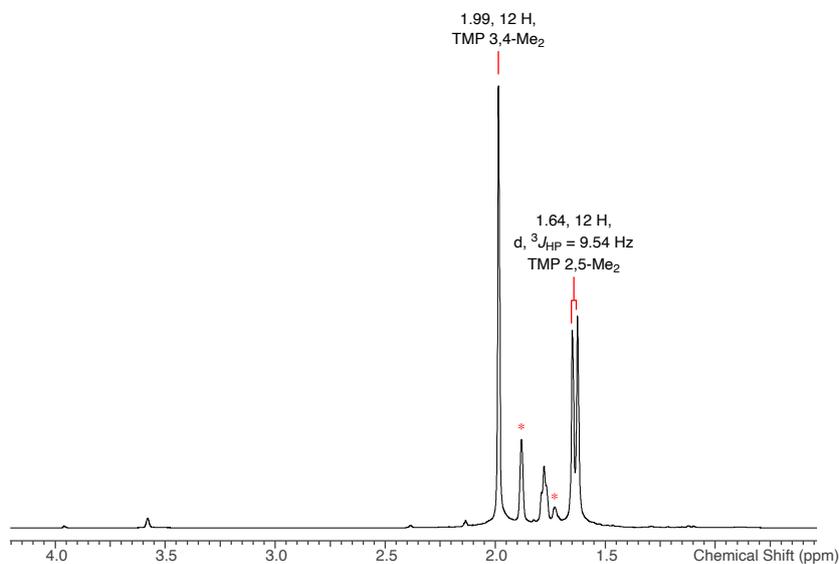


Figure S11. ^1H NMR spectrum of **1-Fe** in C_6D_6 at room temperature, spectrum cropped to show all observed peaks. * denotes a small amount of $(\text{TMP})_2$, presumably formed *via* oxidative coupling of K-TMP from traces of FeCl_3 in commercial samples of FeCl_2 . An authentic sample of $(\text{TMP})_2$ was prepared and the ^1H NMR spectra were compared (*vide infra*, Figure S17) which agreed well with literature values.²

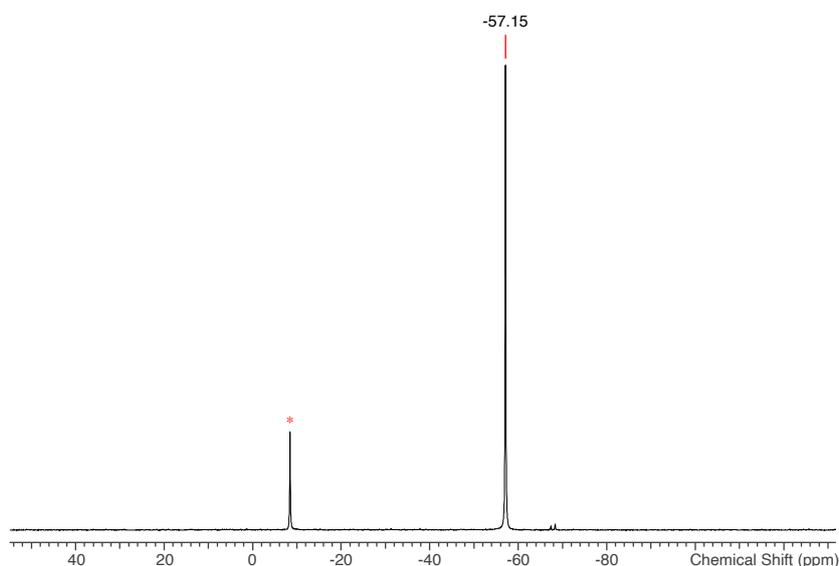


Figure S12. $^{31}\text{P}\{^1\text{H}\}$ NMR spectrum of **1-Fe** in C_6D_6 at room temperature, spectrum cropped to show all observed peaks. *denotes a small amount of $(\text{TMP})_2$.

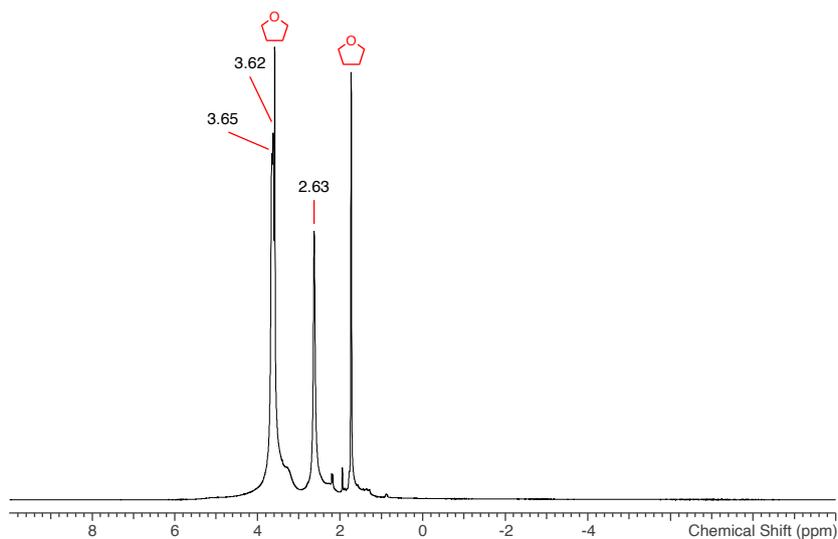


Figure S13. ^1H NMR spectrum of **2-Cr** in $\text{C}_4\text{D}_8\text{O}$ at room temperature, spectrum cropped to show all observed peaks.

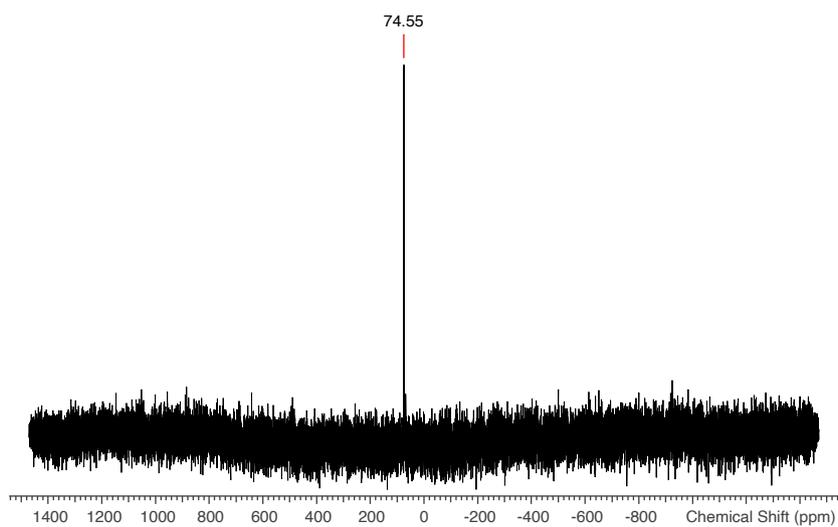


Figure S14. $^{31}\text{P}\{^1\text{H}\}$ NMR spectrum of **2-Cr** in $\text{C}_4\text{D}_8\text{O}$ at room temperature. The sole observed peak at 74.55 ppm corresponds well to the reported ^{31}P resonance for $[\text{K}(18\text{-crown-6})(\text{TMP})]$, and is likely a diamagnetic impurity due to decomposition.

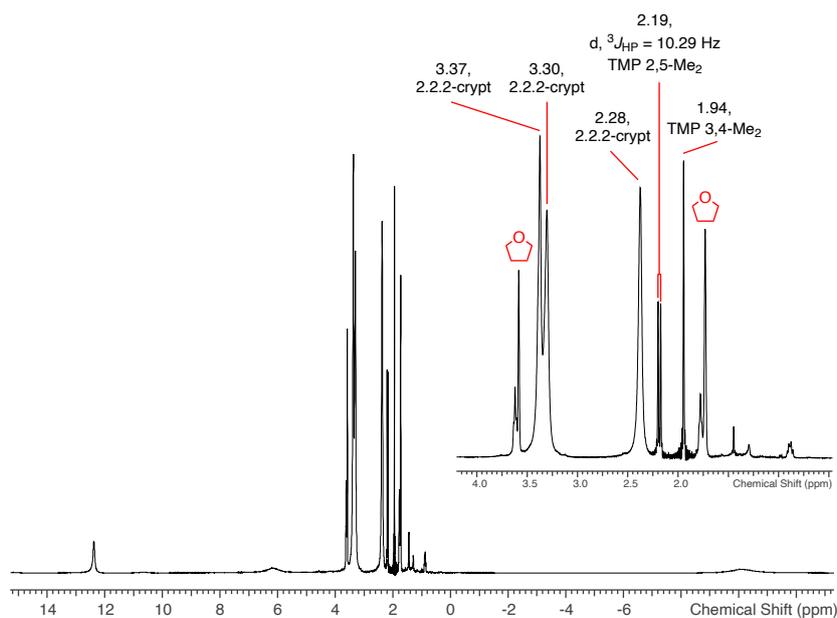


Figure S15. ^1H NMR spectrum of **2-Fe** in $\text{C}_4\text{D}_8\text{O}$ at room temperature, spectrum cropped to show all observed peaks. Broad features around -10 ppm, 6 ppm and 12.5 ppm could not be identified by integration or by comparison with known complexes. The spectrum can be compared with that of **1-Fe** (Figure S11).

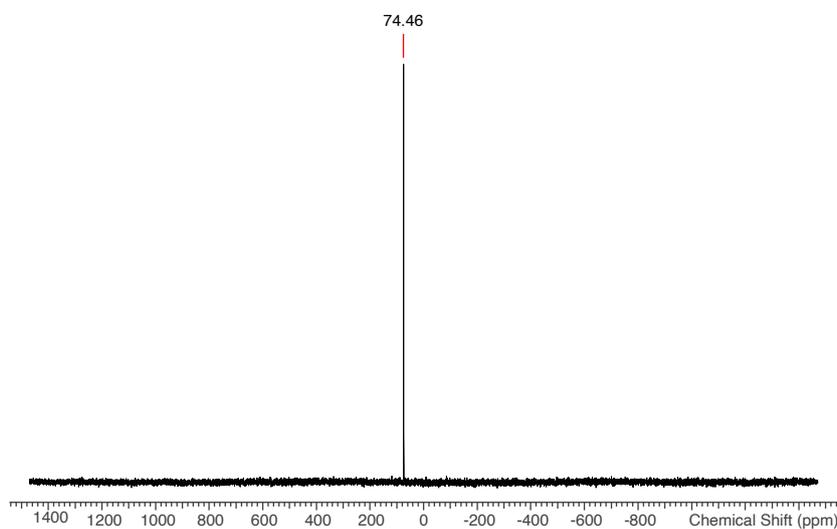


Figure S16. $^{31}\text{P}\{^1\text{H}\}$ NMR spectrum of **2-Fe** in $\text{C}_4\text{D}_8\text{O}$ at room temperature. The sole observed peak at 74.46 ppm corresponds well to the reported ^{31}P resonance for $[\text{K}(18\text{-crown-6})(\text{TMP})]$, and is likely a diamagnetic impurity due to decomposition.

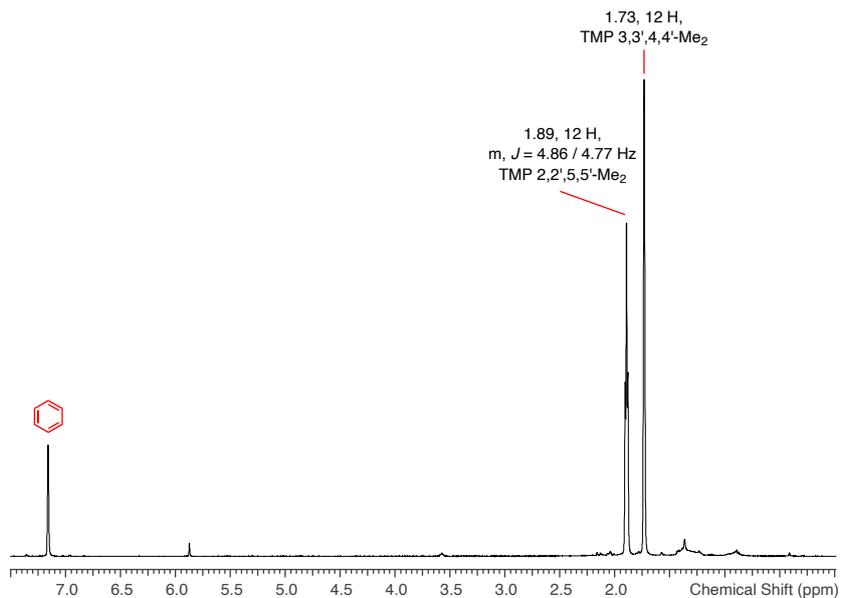


Figure S17. ^1H NMR spectrum of $(\text{TMP})_2$, in C_6D_6 at room temperature, spectrum cropped to show all observed peaks. The peak at 1.89 ppm presents as a triplet but could plausibly be a doublet of doublets due to ABB' coupling of the CH_3 group with the 1,3-P and the 1,4-P.

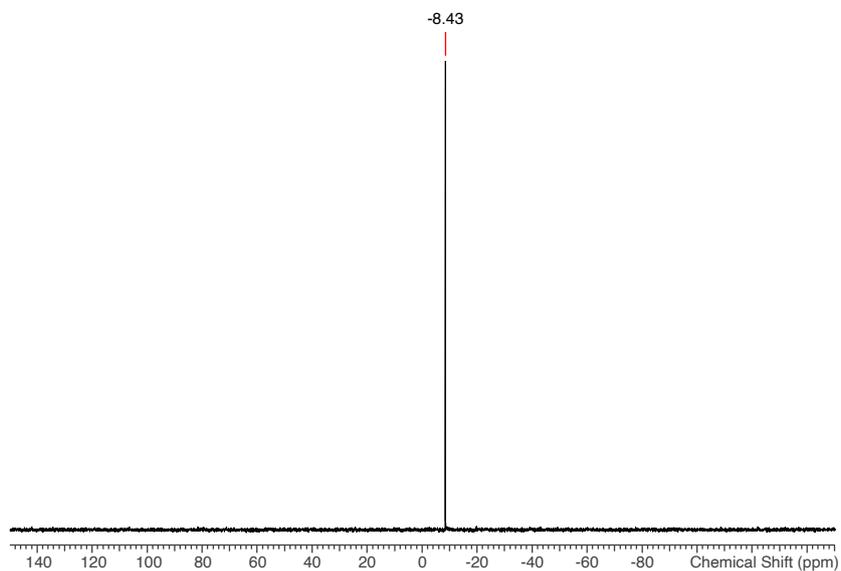


Figure S18. $^{31}\text{P}\{^1\text{H}\}$ NMR spectrum of $(\text{TMP})_2$, in C_6D_6 at room temperature.

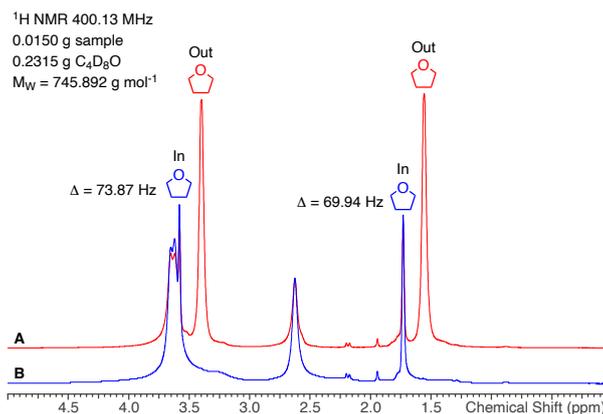


Figure S19. ^1H NMR spectrum of **2-Cr** in $\text{C}_4\text{D}_8\text{O}$, with an external drop of $\text{C}_4\text{H}_8\text{O}$ outside the FEP liner to act as a reference. Collected at room temperature.

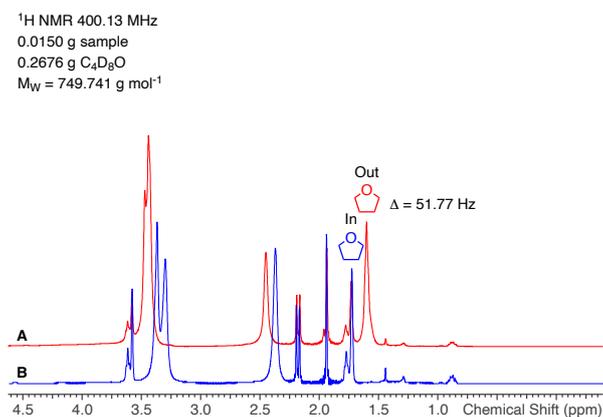


Figure S20. ^1H NMR spectrum of **2-Fe** in $\text{C}_4\text{D}_8\text{O}$, with an external drop of $\text{C}_4\text{H}_8\text{O}$ outside the FEP liner to act as a reference. Collected at room temperature. The second external THF resonance could not be distinguished from overlapping sample peaks.

Table S7. Solution magnetic susceptibility data for 2-Cr and 2-Fe via Evans NMR method.^a

^b Sample / peak	$\mu_{\text{eff}} / \text{B.M}$	$\chi'_m T \text{ (S.I.)} / \text{mol}^{-1}$	$\chi'_m T \text{ (c.g.s. e.m.u.)} / \text{cm}^3 \text{ mol}^{-1} \text{ K}$	$\chi'_m T \text{ (S.I.)} / \text{m}^3 \text{ mol}^{-1} \text{ K}$	mass of sample / g	mass of solvent + sample / g	$M_r / \text{g mol}^{-1}$	$\Delta \text{ peak} / \text{Hz}$
[K(2.2.2-crypt)][Cr(TMP) ₂] peak 1	1.455	1.12E-08	0.265	3.33E-06	0.0150	0.2465	745.892	73.87
[K(2.2.2-crypt)][Cr(TMP) ₂] peak 2	1.433	1.12E-08	0.257	3.33E-06	0.0150	0.2465	745.892	69.94
[K(2.2.2-crypt)][Fe(TMP) ₂]	1.376	1.12E-08	0.237	3.33E-06	0.0150	0.2826	749.741	51.77

^a The small masses engender large errors in this methodology, the results should be cautiously interpreted along with other data. ^b Spectrometer frequency 400.130 MHz. Simple diamagnetic correction of $M_r / -2,000,000$ applied. $\rho_{\text{D8THF}} = 0.985 \text{ g mL}^{-1}$.

1.

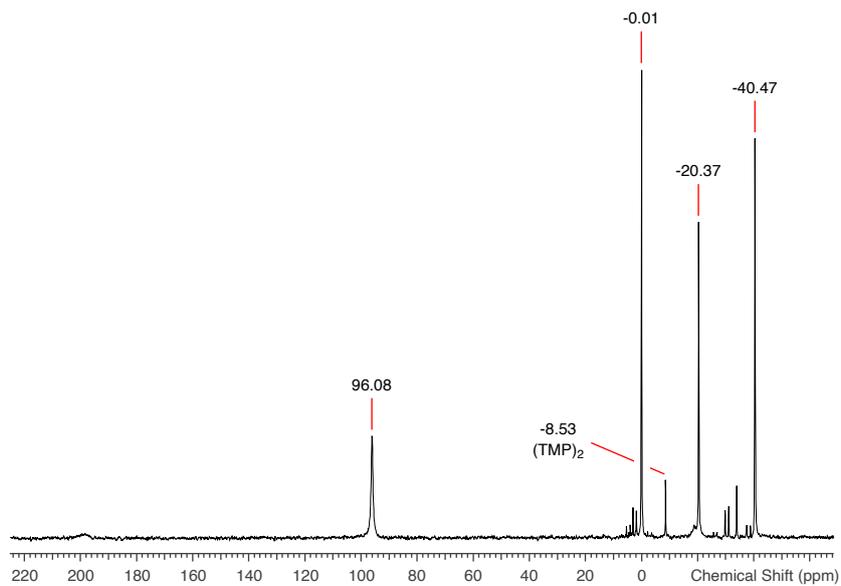


Figure S21. $^{31}\text{P}\{^1\text{H}\}$ NMR spectrum of crystals of **7** in C_6D_6 at room temperature.

S5. UV-vis-nIR Spectroscopy

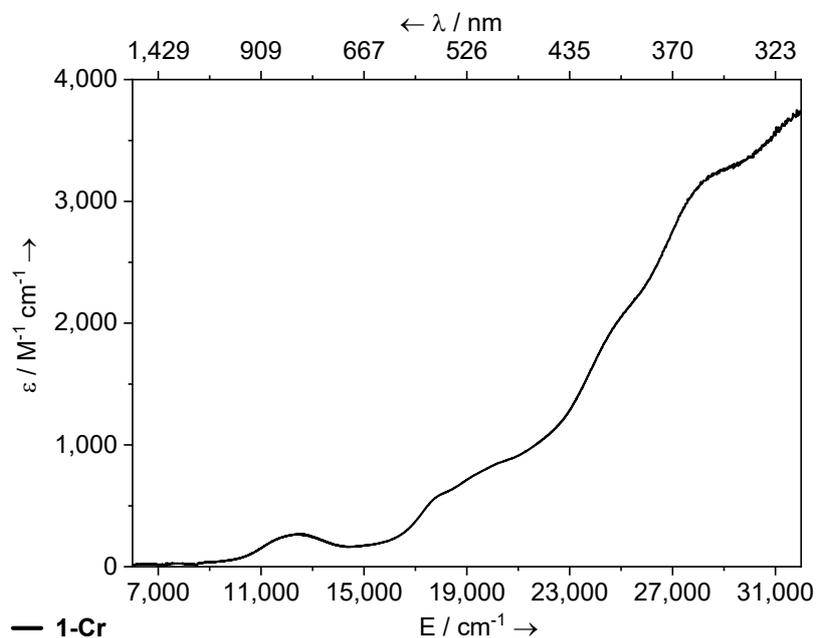


Figure S22. UV-vis-nIR spectrum of **1-Cr** in THF (0.63 mM) at room temperature recorded between 5,556–36,364 cm^{-1} (1,800–275 nm), window shown between 6,000–32,000 cm^{-1} (1,667–313 nm).

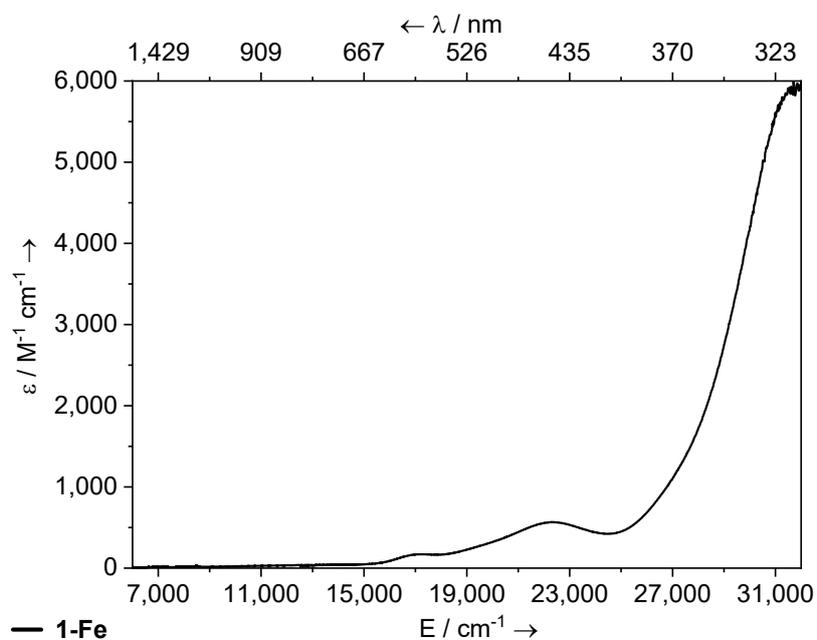


Figure S23. UV-vis-nIR spectrum of **1-Fe** in THF (0.50 mM) at room temperature recorded between 5,556–36,364 cm^{-1} (1,800–275 nm), window shown between 6,000–32,000 cm^{-1} (1,667–313 nm).

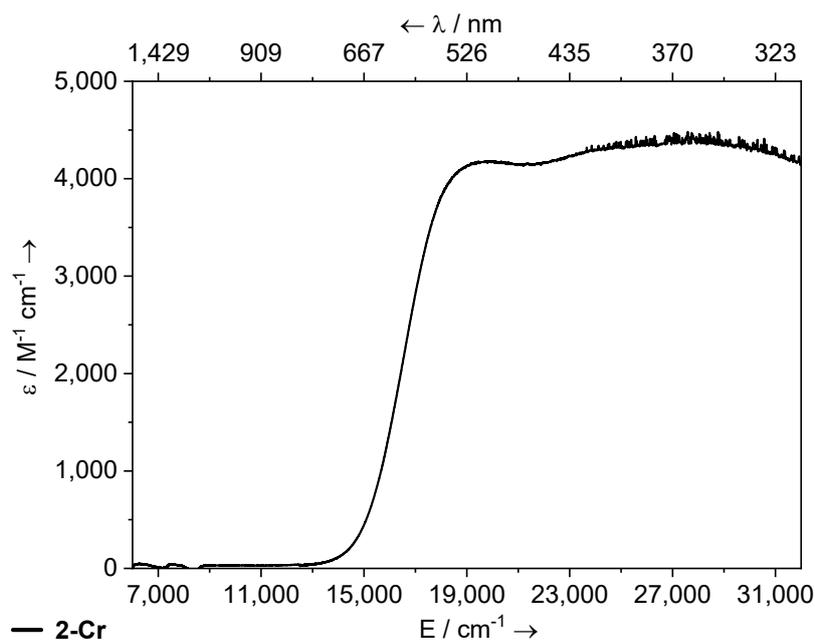


Figure S24. UV-vis-nIR spectrum of **2-Cr** in THF (0.40 mM) at room temperature recorded between 5,556–36,364 cm^{-1} (1,800–275 nm), window shown between 6,000–32,000 cm^{-1} (1,667–313 nm). Features either side of 7,000 cm^{-1} are due to imperfect background subtraction.

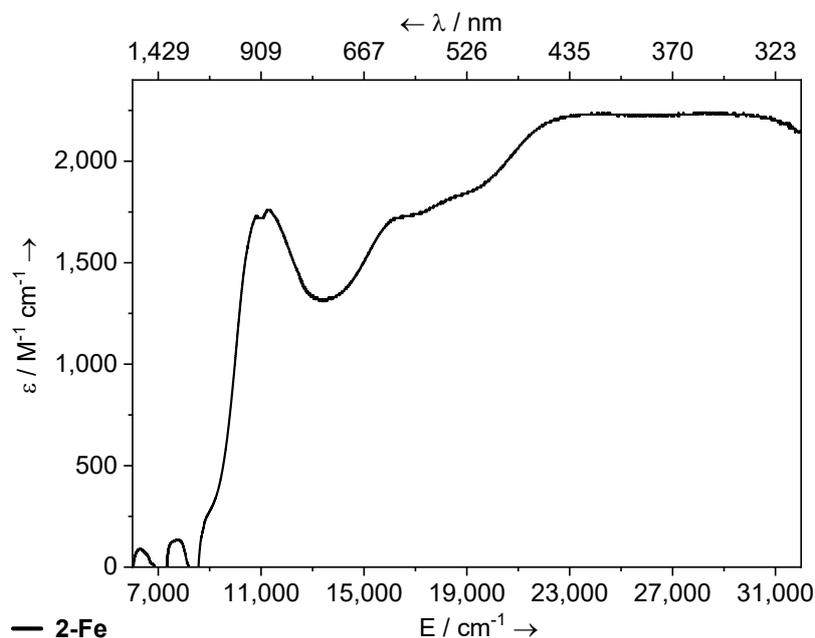


Figure S25. UV-vis-nIR spectrum of **2-Fe** in THF (0.40 mM) at room temperature recorded between 5,556–36,364 cm^{-1} (1,800–275 nm), window shown between 6,000–32,000 cm^{-1} (1,667–313 nm). Features either side of 7,000 cm^{-1} are due to imperfect background subtraction.

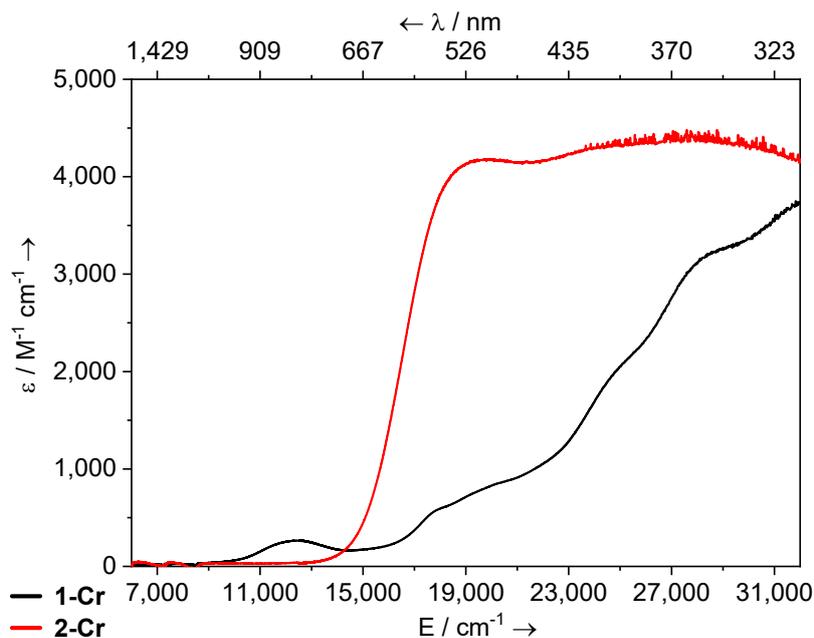


Figure S26. UV-vis-nIR spectrum of both **1-Cr** (black, bottom line) and **2-Cr** (red, top line) in THF at room temperature recorded between 5,556–36,364 cm^{-1} (1,800–275 nm), window shown between 6,000–32,000 cm^{-1} (1,667–313 nm). Intended to highlight the spectral differences upon reduction.

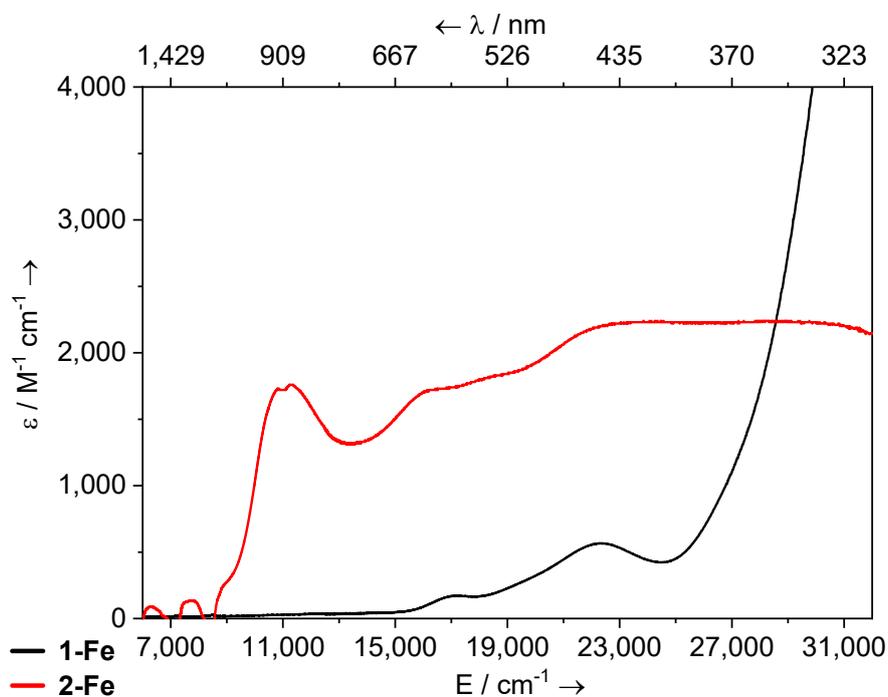


Figure S27. UV-vis-nIR spectrum of both **1-Fe** (black, bottom line) and **2-Fe** (red, top line) in THF at room temperature recorded between 5,556–36,364 cm^{-1} (1,800–275 nm), window shown between 6,000–32,000 cm^{-1} (1,667–313 nm). Intended to highlight the spectral differences upon reduction.

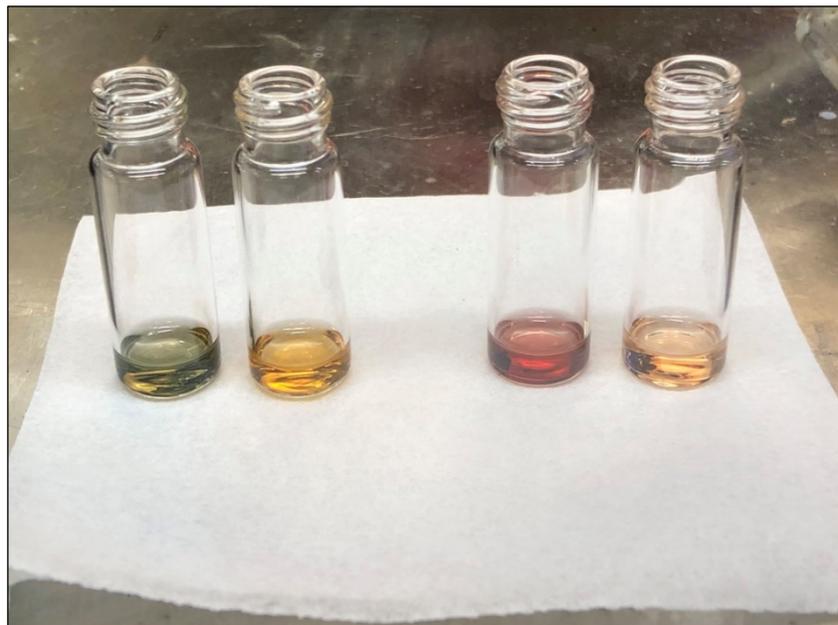


Figure S28. A photograph of the solutions used for the UV-vis-nIR measurements. From left to right, **2-Fe**, **1-Fe**, **2-Cr**, and **1-Cr**.

S6. ATR-FTIR Spectroscopy

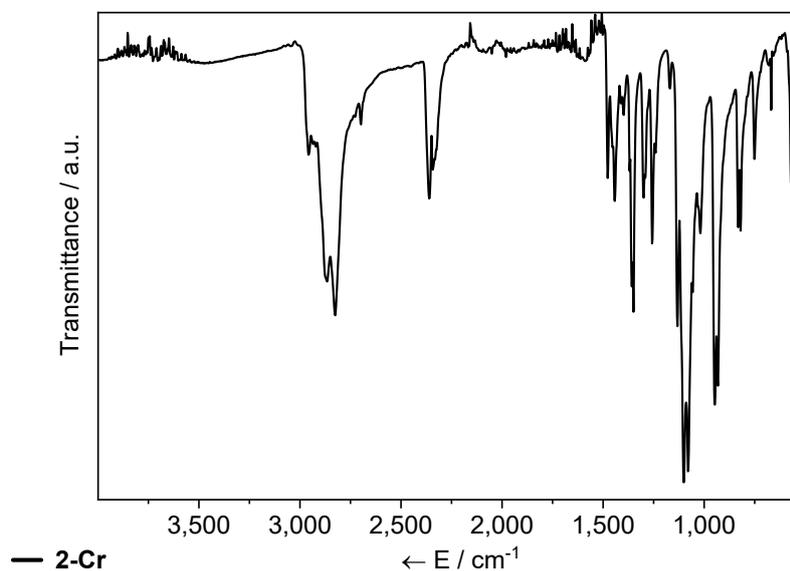


Figure S29. FT-IR spectrum of both 2-Cr a microcrystalline powder, between 4,000-525 cm⁻¹.

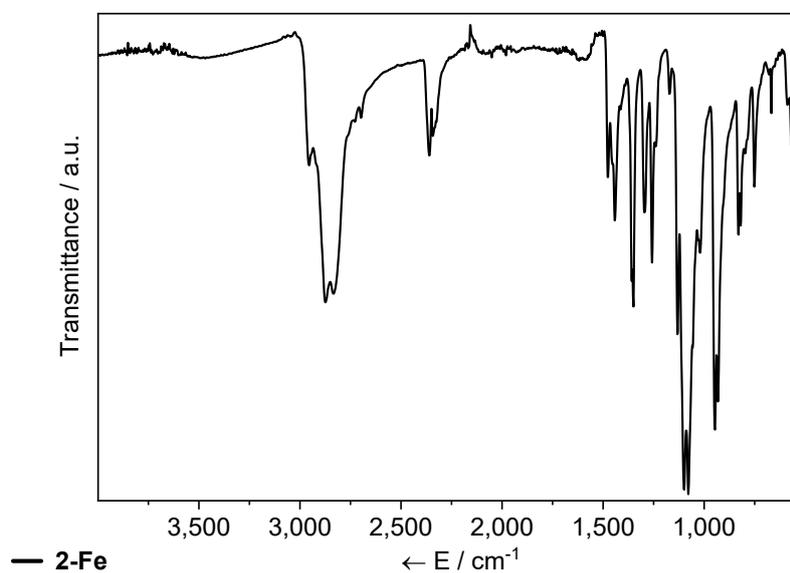


Figure S30. FT-IR spectrum of both 2-Fe a microcrystalline powder, recorded between 4,000-525 cm⁻¹.

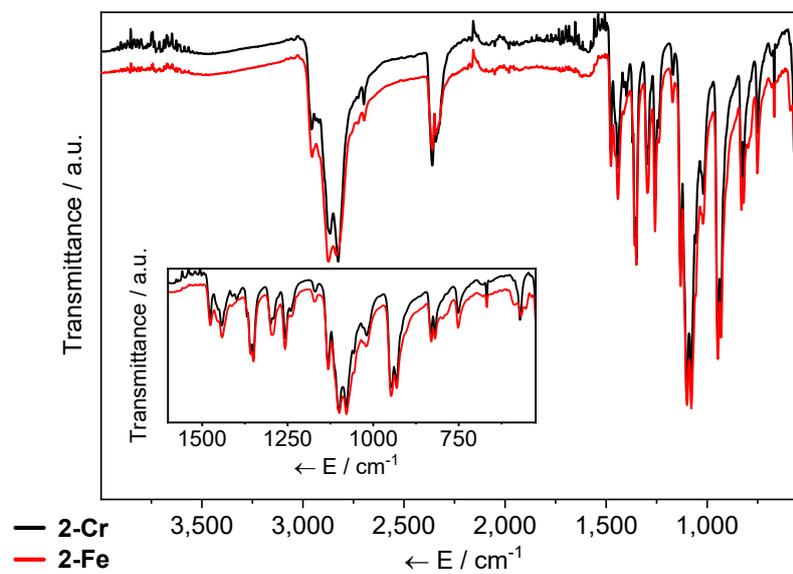


Figure S31. FT-IR spectrum of both **2-Cr** (black, top line) and **2-Fe** (red, bottom line) as a microcrystalline powder, recorded between 4,000-525 cm⁻¹. Inset shown between 1,600-500 cm⁻¹, to show that the two complexes are essentially identical in the fingerprint region.

S7. EPR Spectroscopy

General considerations: High-Field/Frequency Electron Paramagnetic Resonance spectra were recorded on powder samples of **2-Cr** and **2-Fe**. Samples were loaded into polyethylene cups and immobilized with a PTFE stopper. All sample manipulations were performed in an inert atmosphere glovebox. The transmission-type spectrometer used in this study combined a 17 T superconducting magnet with a phase-locked source (Virginia Diodes Inc., Charlottesville, VA, USA) with a series of frequency multipliers. The field modulated signal was detected by an InSb hot-electron bolometer (QMC Ltd., Cardiff, U.K.). Temperature control was achieved using an Oxford Instruments (Oxford, U.K.) continuous-flow cryostat. Spectral simulations were generated using EasySpin.¹²

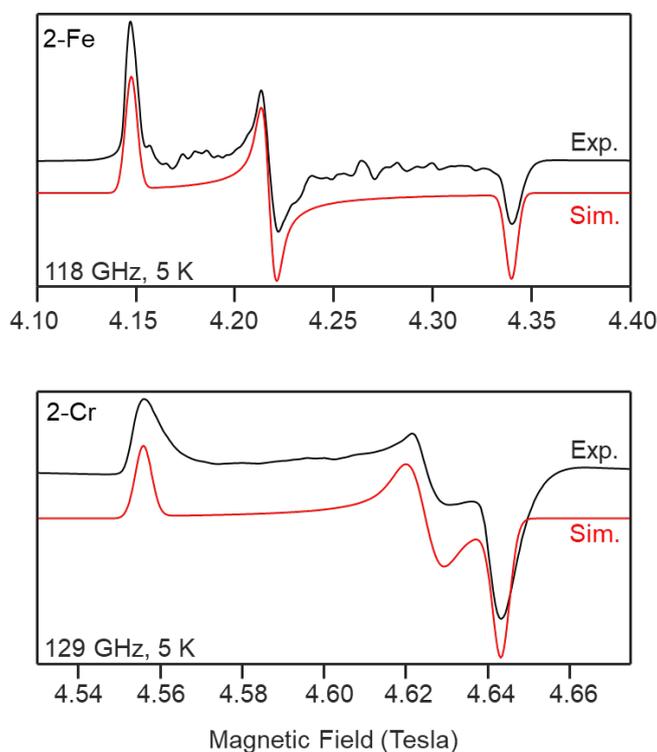


Figure S32. Experimental (black) and simulated (red) EPR spectra of **2-Fe** (top), **2-Cr** (bottom), note the x-axis scale for **2-Cr**. Spectra were recorded at the temperatures and frequencies denoted on the respective figures. Note: the baseline features in the **2-Fe** spectrum results from micro-crystallites in the imperfectly ground powder. Multi-frequency studies of **2-Cr** and **2-Fe** reveal no additional features arising from higher spin states. However, **2-Fe** exhibits a minor impurity at lower field which is omitted here for clarity and is shown below in Figure S33. The g -values used in the simulations are reported the main text as well as in Table S8.

Table S8. Experimental and CASSCF/NEVPT2 Calculated Spin Hamiltonian Parameters.

		S	g^1, g^2, g^3
2-Fe	Exp.	1/2	2.033(5), 1.999(5), 1.943(5)
	NEVPT2	1/2	2.116, 2.029, 1.919

2-Cr	Exp.	1/2	2.023(5), 1.993(5), 1.985(5)
	NEVPT2	1/2	2.007, 1.984, 1.978

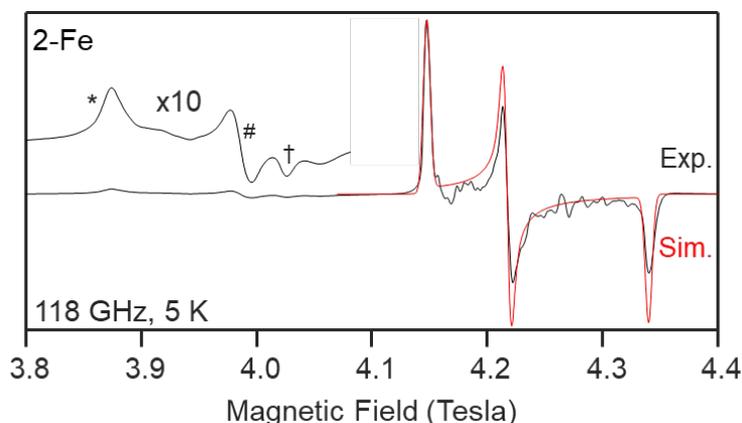


Figure S33. Spectrum of **2-Fe** (same as Figure S32) with an extended axis to show the unknown minor species. The g -values of these features are: ~ 2.09 (*), 2.12 (#), and ~ 2.18 (†). No evidence of this species was present in **2-Cr**.

The sign of Δg_{μ} ($g_{\mu} - g_e$), i.e., whether the g - value is less than or greater than the free electron value, g_e , is determined by the competition of the low-energy α - and β - electron transitions. If the excited states contributing to Δg are predominately accessed by transitions involving α -electrons, then $\Delta g < 0$ while those involving low-lying β -electrons will result in $\Delta g > 0$. The g -values are reasonably well reproduced by the CASSCF + NEVPT2 calculations (see Table S8) and some qualitative insight may be gleaned from examining the AILFT orbital diagrams in Figure 3. In the case of **2-Cr** the lowest energy state (transitions: $\{d_{xy}/d_{y^2-y^2}\} \rightarrow \{d_{z^2}\}$) do not mix by spin-orbit coupling and the sign and magnitude of Δg are determined by the mixing of numerous higher energy excited states. This explains the relatively small Δg values (**2-Cr**, $|\Delta g_{\text{avg}}| = 0.0158$). For **2-Fe** the first excited state can be described as an α - electron transition from $\{d_{xz}\} \rightarrow \{d_{yz}\}$ which will mix with the ground state and negative Δg along z . As in the case of **2-Cr**, several higher energy excited states make up the remaining contributions. However, the comparatively large mixing induced by the first excited state is likely a key contributor to the larger observed $|\Delta g_{\text{avg}}| = 0.0311$ in **2-Fe**.

S8. ^{57}Fe Mössbauer Spectroscopy

General considerations: Spectra were recorded at 120 K in zero applied field using a constant acceleration spectrometer and a $^{57}\text{Co}/\text{Rh}$ source. The samples used for these measurements consisted of ground powders of **1-Fe** and **2-Fe** that were contained in polyethylene sample cups with tightly fitted lids. The isomer shift is reported relative to that of $\alpha\text{-Fe}$ at room temperature. Spectral simulations were generated using the WMOSS software package (SEE Co. Minneapolis, MN).¹³

Table S9. Experimental and DFT Calculated ^{57}Fe Mössbauer Parameters.

		δ (mm/sec)	ΔE_Q (mm/sec)	Γ (mm/sec) ^a
2-Fe	Exp.	0.65	1.28	-0.602
	DFT	0.66	1.46	-

1-Fe	Exp.	0.48	2.02	0.318
	DFT	0.50	1.91	-

^a Positive and negative linewidths indicate Lorentzian and Gaussian lineshapes, respectively.

S9. Quantum Chemical Calculations

DFT Calculations: Calculations for **1-Cr**, **2-Cr**, **1-Fe**, and **2-Fe** were performed using the atomic coordinates determined by X-ray crystallography (with hydrogen positions optimized). These calculations were performed using ORCA V4.1 with the BP86 functional along with the def2-tzvp basis sets.¹⁴⁻¹⁸ The functional was chosen from our previous success in reproducing ⁵⁷Fe Mössbauer parameters using the BP86 functional.¹⁹ The resolution of the identity approximation was used along with auxiliary basis sets generated using the ‘autoaux’ command.²⁰ The quadrupole splitting and isomer shifts were then calculated using the BP86 functional with CP(PPP) (Fe)/def2-tzvp(C)/def2-svp(H). The RI approximation was not used for calculation of ⁵⁷Fe Mössbauer Parameters. The computed density at the Fe nucleus was converted to experimental values of the isomer shift using the calibration curve described by Römelt, *et al.*²¹ Example input files are shown below.

Complete Active Space Self-Consistent Field/AILFT Calculations: State-averaged complete active space self-consistent field (SA-CASSCF) followed by second-order N-electron valence perturbation theory (NEVPT2) calculations were performed on the same structures used in the DFT calculations (*vide supra*). Scalar relativistic effects were accounted for by the Douglas–Kroll–Hess (DKH) procedure. SOC was accounted for by quasi-degenerate perturbation theory. This methodology of accounting for SOC has proven successful in numerous other studies.²²⁻³⁰ The dkh-def2-tzvp (Fe,C)/dkh-def2-svp (H) basis set combination was used. The active space consisted of the five 3d – orbitals and all of the 3d – electrons (six for Fe²⁺, seven for Fe⁺, four for Cr²⁺, five for Cr⁺). All roots for all multiplicities were included. This means that for Cr²⁺/Fe²⁺ 5 quintet, 45 triplet, and 50 singlet states were included. For Cr⁺ 1 sextet, 24 quartet, and 75 singlet states were included. For Fe⁺ 10 quartet states and 40 doublet states were included. The *ab-initio* ligand field analysis was performed as implemented in ORCA V4.1 and the energies presented in the text are those associated with the mapping of the CASSCF+NEVPT2 energies. Example input files are shown below.²⁷

S9.1. Example ORCA input files

Optimize Hydrogen Positions:

```
! BP86 def2-tzvp autoaux opt
%geom
optimizehydrogens true
end
*xyz charge multiplicity
xyz coordinates of structure
*
```

⁵⁷Fe Mössbauer Parameters:

```
! BP86 def2-tzvp NoFinalGrid
%method
SpecialGridAtoms 26
SpecialGridIntAcc 7
end
%basis
NewGTO H "def2-svp"
NewGTO Fe "CP(PPP)"
end
*xyz charge multiplicity
xyz coordinates of structure
*
%eprnmr
nuclei = all Fe {rho, fgrad}
end
```

CASSCF + NEVPT2:

```

!DKH dkh-def2-tzvp autoaux normalprint moread
%moinp "guess.gbw" # Here the guess orbitals were from a single point DFT calculation
%casscf # Here the Fe+/d7 settings are used as an example
nel 7
norb 5
mult 4,2
nroots 10,40
actorbs dorbs
trafostep RI
nevpt2 true
rel
dosoc true
gtensor true
end
end
%basis
NewGTO H "def2-svp"
NewGTO Fe "CP(PPP)"
end
*xyz -1 2
xyz coordinates of structure
*
```

Table S10. Summary of CASSCF+NEVPT2 and AILFT results for 1-Fe, 2-Fe, 1-Cr, and 2-Cr.

	Method	Spin Ground State	AILFT Energies (cm ⁻¹)
1-Fe	CASSCF	2	0.0, 1018.7, 3671.5, 18514.2, 21138.6
	CASSCF+NEVPT 2	0	0.0, 899.0, 2984.5, 23425.8, 25707.1
2-Fe	CASSCF	3/2	0.0, 389.9, 4677.5, 14412.4, 19698.6
	CASSCF+NEVPT 2	1/2	0.0, 354.7, 4364.2, 19153.3, 25100.0
1-Cr	CASSCF	2	0.0, 2226.9, 5170.3, 18790.0, 22484.6
	CASSCF+NEVPT 2	1	0.0, 2518.0, 4620.0, 21382.7, 24956.9
2-Cr	CASSCF	1/2	0.0, 461.3, 7483.5, 21077.6, 25943.8
	CASSCF+NEVPT 2	1/2	0.0, 343.7, 6144.3, 24766.6, 29492.1

S10. SQUID Magnetometry

General considerations: Samples of **1-Cr**, **2-Cr**, and **2-Fe** used for measurements consisted of crushed microcrystalline powders weighed (28.0 mg, **1-Cr**; 23.6 mg, **2-Cr**; 29.0 mg, **2-Fe**) into thin-walled precision borosilicate NMR tubes, which were flame sealed under vacuum. A diamagnetic correction was performed using two further NMR tubes from the same batch also sealed under vacuum.

Magnetic measurements were performed with a Quantum Design SQUID magnetometer MPMS-XL. The temperature dependence of magnetic susceptibility was measured in a direct current (DC) applied magnetic field of 0.1 T in the range from 1.8 to 300 K.

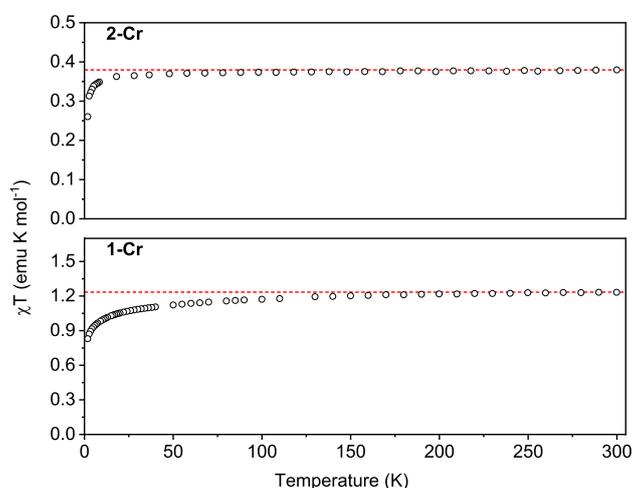


Figure S34. Temperature dependence of χT for **1-Cr** and **2-Cr** between 1.8 and 300 K. The red lines are to guide the eye for values at 298 K (**1-Cr**, 1.24 emu·K/mol; **2-Cr**, 0.38 emu·K/mol).

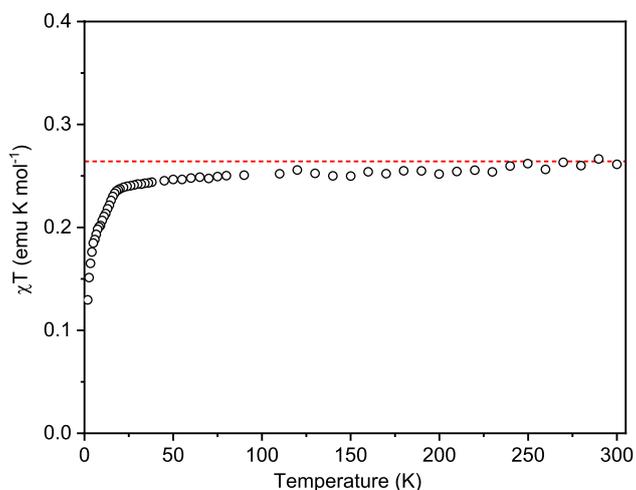


Figure S35. Temperature dependence of χT for **2-Fe** between 1.8 and 300 K. The red line is to guide the eye for the value at 298 K (0.26 emu·K/mol).

S11. Qualitative ligand HOMOs in Cp^- vs $\{\text{PC}_4\text{H}_4\}^-$

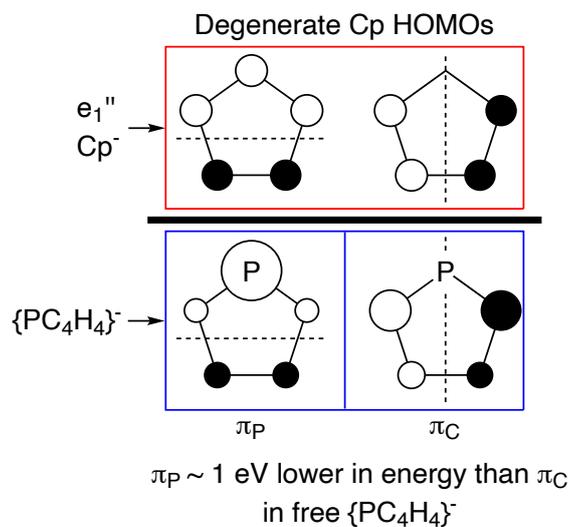


Figure S36. Illustrative pictures of the ligand HOMOs relevant to the LUMO in FcH , and $[\text{Fe}(\text{PC}_4\text{H}_4)_2]$.

The ligand HOMO in Cp^- in D_{5h} symmetry is comprised of two degenerate orbitals. However, for $\{\text{PC}_4\text{H}_4\}^-$, the P atom breaks the 5-fold symmetry, and thus the analogous orbitals are not degenerate. While both the π_P and π_C orbitals shown above are of appropriate symmetry to interact with the metal d_{xz} , d_{yz} orbitals, they differ in energy by $\sim 1 \text{ eV}$, and thus the $[\text{Fe}(\text{PC}_4\text{H}_4)_2]$ complex LUMO is mostly comprised of the π_P orbital and a metal d_{yz} orbital. Please see Fenske *et al* for a much more in depth analysis,³¹ as the above discussion does not fully account for the lower symmetry of **2-Fe**.

Author Contributions

CAPG performed all synthesis, UV-vis-nIR and NMR characterization and sample preparation, and X-ray diffraction data collection and interpretation with supervision by BLS. SMG performed EPR and Mössbauer spectroscopies and interpreted the data, and also performed all theoretical work, with supervision by BWS. OU performed SQUID magnetometry measurements and interpretation. RJB sealed samples for analysis and was supervised by JLK. CAPG lead the project with project scope and design helped by SMG. The manuscript was prepared by CAPG with input from all authors.

References

1. F. Nief, F. Mathey, L. Ricard, F. Robert, *Organometallics*, 1988, **7**, 921.
2. R. Feher, F. H. Köhler, F. Nief, L. Ricard, S. Rossmayer, *Organometallics*, 1997, **16**, 4606.
3. L. E. Manxzer, J. Deaton, P. Sharp, R. R. Schrock, in *Inorganic Syntheses*, 2007, pp. 135.
4. T. K. Panda, M. T. Gamer, P. W. Roesky, *Organometallics*, 2003, **22**, 877.
5. P. J. Fagan, W. A. Nugent, *J. Am. Chem. Soc.*, 1988, **110**, 2310.
6. P. J. Fagan, W. A. Nugent, *Org. Syn.*, 1992, **70**.
7. *CrysAlisPro 39.27b*, Oxford Diffraction / Agilent Technologies UK Ltd, Yarnton, U.K., **2017**.
8. O. V. Dolomanov, L. J. Bourhis, R. J. Gildea, J. A. K. Howard, H. Puschmann, *J. Appl. Crystallogr.*, 2009, **42**, 339.
9. G. M. Sheldrick, *Acta Crystallogr. A*, 2008, **64**, 112.
10. G. M. Sheldrick, *Acta Crystallogr. C*, 2015, **71**, 3.
11. Inkscape: Open Source Scalable Vector Graphics Editor. <https://inkscape.org/>.
12. S. Stoll, A. Schweiger, *J. Magn. Reson.*, 2006, **178**, 42.
13. I. Prisecaru, WMOSS4 Mössbauer Spectral Analysis Software. <http://wmoos.org/>.
14. A. D. Becke, *Phys. Rev. A*, 1988, **38**, 3098.
15. A. D. Becke, *J. Chem. Phys.*, 1993, **98**, 5648.
16. F. Neese, *WIREs Comput. Mol. Sci.*, 2018, **8**, e1327.
17. J. P. Perdew, *Phys. Rev. B*, 1986, **33**, 8822.

18. F. Weigend, R. Ahlrichs, *Phys. Chem. Chem. Phys.*, 2005, **7**, 3297.
19. C. A. P. Goodwin, M. Giansiracusa, S. M. Greer, H. M. Nicholas, P. Evans, M. Vonci, S. Hill, N. F. Chilton, D. P. Mills, *Nat. Chem.*, 2020.
20. G. L. Stoychev, A. A. Auer, F. Neese, *J. Chem. Theory. Comput.*, 2017, **13**, 554.
21. M. Römelt, S. Ye, F. Neese, *Inorg. Chem.*, 2009, **48**, 784.
22. S. K. Singh, J. Eng, M. Atanasov, F. Neese, *Coord. Chem. Rev.*, 2017, **344**, 2.
23. D. Maganas, J. Krzystek, E. Ferentinos, A. M. Whyte, N. Robertson, V. Psycharis, A. Terzis, F. Neese, P. Kyritsis, *Inorg. Chem.*, 2012, **51**, 7218.
24. S. Ye, F. Neese, *J. Chem. Theory. Comput.*, 2012, **8**, 2344.
25. J. Jung, M. Atanasov, F. Neese, *Inorg. Chem.*, 2017, **56**, 8802.
26. E. A. Suturina, J. Nehr Korn, J. M. Zadrozny, J. Liu, M. Atanasov, T. Weyhermuller, D. Maganas, S. Hill, A. Schnegg, E. Bill, J. R. Long, F. Neese, *Inorg. Chem.*, 2017, **56**, 3102.
27. M. Atanasov, D. Aravena, E. Suturina, E. Bill, D. Maganas, F. Neese, *Coord. Chem. Rev.*, 2015, **289-290**, 177.
28. E. A. Suturina, D. Maganas, E. Bill, M. Atanasov, F. Neese, *Inorg. Chem.*, 2015, **54**, 9948.
29. M. Autillo, M. A. Islam, J. Jung, J. Pilme, N. Galland, L. Guerin, P. Moisy, C. Berthon, C. Tamain, H. Bolvin, *Phys. Chem. Chem. Phys.*, 2020, **22**, 14293.
30. J. Jung, M. A. Islam, V. L. Pecoraro, T. Mallah, C. Berthon, H. Bolvin, *Chemistry*, 2019, **25**, 15112.
31. N. M. Kostic, R. F. Fenske, *Organometallics*, 1983, **2**, 1008.

S13. CIF Reports

checkCIF/PLATON report

Structure factors have been supplied for datablock(s) 1-Cr

THIS REPORT IS FOR GUIDANCE ONLY. IF USED AS PART OF A REVIEW PROCEDURE FOR PUBLICATION, IT SHOULD NOT REPLACE THE EXPERTISE OF AN EXPERIENCED CRYSTALLOGRAPHIC REFEREE.

No syntax errors found. CIF dictionary Interpreting this report

Datablock: 1-Cr

Bond precision: C-C = 0.0021 A Wavelength=0.71073

Cell: a=7.8420(3) b=12.5620(4) c=8.8611(3)
 alpha=90 beta=109.444(4) gamma=90

Temperature: 100 K

	Calculated	Reported
Volume	823.13(5)	823.13(5)
Space group	P 21/n	P 1 21/n 1
Hall group	-P 2yn	-P 2yn
Moiety formula	C16 H24 Cr P2	C16 H24 Cr P2
Sum formula	C16 H24 Cr P2	C16 H24 Cr P2
Mr	330.29	330.29
Dx,g cm-3	1.333	1.333
Z	2	2
Mu (mm-1)	0.874	0.874
F000	348.0	348.0
F000'	349.12	
h,k,lmax	9,15,10	9,15,10
Nref	1507	1508
Tmin,Tmax	0.700,0.769	0.804,1.000
Tmin'	0.686	

Correction method= # Reported T Limits: Tmin=0.804 Tmax=1.000
AbsCorr = MULTI-SCAN

Data completeness= 1.001 Theta(max)= 25.349

R(reflections)= 0.0252(1441) wR2(reflections)= 0.0667(1508)

S = 1.091 Npar= 92

The following ALERTS were generated. Each ALERT has the format
test-name_ALERT_alert-type_alert-level.
Click on the hyperlinks for more details of the test.

Alert level G

PLAT328_ALERT_4_G	Possible Missing H on sp3? Phosphorus	P1 Check
PLAT380_ALERT_4_G	Incorrectly? Oriented X(sp2)-Methyl Moiety	C6 Check
PLAT883_ALERT_1_G	No Info/Value for _atom_sites_solution_primary .	Please Do !
PLAT941_ALERT_3_G	Average HKL Measurement Multiplicity	4.5 Low
PLAT978_ALERT_2_G	Number C-C Bonds with Positive Residual Density.	6 Info

0 **ALERT level A** = Most likely a serious problem - resolve or explain
0 **ALERT level B** = A potentially serious problem, consider carefully
0 **ALERT level C** = Check. Ensure it is not caused by an omission or oversight
5 **ALERT level G** = General information/check it is not something unexpected

1 ALERT type 1 CIF construction/syntax error, inconsistent or missing data
1 ALERT type 2 Indicator that the structure model may be wrong or deficient
1 ALERT type 3 Indicator that the structure quality may be low
2 ALERT type 4 Improvement, methodology, query or suggestion
0 ALERT type 5 Informative message, check

It is advisable to attempt to resolve as many as possible of the alerts in all categories. Often the minor alerts point to easily fixed oversights, errors and omissions in your CIF or refinement strategy, so attention to these fine details can be worthwhile. In order to resolve some of the more serious problems it may be necessary to carry out additional measurements or structure refinements. However, the purpose of your study may justify the reported deviations and the more serious of these should normally be commented upon in the discussion or experimental section of a paper or in the "special_details" fields of the CIF. checkCIF was carefully designed to identify outliers and unusual parameters, but every test has its limitations and alerts that are not important in a particular case may appear. Conversely, the absence of alerts does not guarantee there are no aspects of the results needing attention. It is up to the individual to critically assess their own results and, if necessary, seek expert advice.

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Structure factors have been supplied for datablock(s) 1-Fe

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No syntax errors found. CIF dictionary Interpreting this report

Datablock: 1-Fe

Bond precision: C-C = 0.0030 A Wavelength=0.71073

Cell: a=14.2948(7) b=12.8549(6) c=8.8448(4)
 alpha=90 beta=104.364(5) gamma=90

Temperature: 100 K

	Calculated	Reported
Volume	1574.50(13)	1574.50(13)
Space group	C 2/c	C 1 2/c 1
Hall group	-C 2yc	-C 2yc
Moiety formula	C16 H24 Fe P2	C16 H24 Fe P2
Sum formula	C16 H24 Fe P2	C16 H24 Fe P2
Mr	334.14	334.14
Dx, g cm-3	1.410	1.410
Z	4	4
Mu (mm-1)	1.146	1.146
F000	704.0	704.0
F000'	706.33	
h,k,lmax	17,16,11	17,16,11
Nref	1608	1605
Tmin,Tmax	0.770,0.832	0.850,1.000
Tmin'	0.717	

Correction method= # Reported T Limits: Tmin=0.850 Tmax=1.000
AbsCorr = MULTI-SCAN

Data completeness= 0.998 Theta(max)= 26.370

R(reflections)= 0.0312(1407) wR2(reflections)= 0.0798(1605)

S = 1.117 Npar= 92

The following ALERTS were generated. Each ALERT has the format
test-name_ALERT_alert-type_alert-level.
Click on the hyperlinks for more details of the test.

● Alert level C		
PLAT094_ALERT_2_C	Ratio of Maximum / Minimum Residual Density	2.29 Report
PLAT911_ALERT_3_C	Missing FCF Refl Between Thmin & STh/L= 0.600	4 Report

● Alert level G		
PLAT328_ALERT_4_G	Possible Missing H on sp3? Phosphorus	P1 Check
PLAT941_ALERT_3_G	Average HKL Measurement Multiplicity	3.9 Low
PLAT978_ALERT_2_G	Number C-C Bonds with Positive Residual Density.	6 Info

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2 ALERT type 3 Indicator that the structure quality may be low
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0 ALERT type 5 Informative message, check

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Structure factors have been supplied for datablock(s) 2-Cr

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No syntax errors found. CIF dictionary Interpreting this report

Datablock: 2-Cr

Bond precision: C-C = 0.0030 A Wavelength=0.71073

Cell: a=10.1994(4) b=14.2193(7) c=15.0882(6)
 alpha=62.328(4) beta=86.597(3) gamma=87.156(3)
Temperature: 100 K

	Calculated	Reported
Volume	1933.91(16)	1933.91(16)
Space group	P -1	P -1
Hall group	-P 1	-P 1
Moiety formula	C18 H36 K N2 O6, C16 H24 Cr P2	C16 H24 Cr P2, C18 H36 K N2 O6
Sum formula	C34 H60 Cr K N2 O6 P2	C34 H60 Cr K N2 O6 P2
Mr	745.88	745.88
Dx, g cm-3	1.281	1.281
Z	2	2
Mu (mm-1)	0.529	0.529
F000	798.0	798.0
F000'	799.73	
h,k,lmax	12,17,18	12,17,18
Nref	7083	7060
Tmin,Tmax	0.853,0.853	0.902,1.000
Tmin'	0.853	

Correction method= # Reported T Limits: Tmin=0.902 Tmax=1.000
AbsCorr = MULTI-SCAN

Data completeness= 0.997 Theta(max)= 25.350

R(reflections)= 0.0338(5848) wR2(reflections)= 0.0872(7060)

S = 1.100 Npar= 423

The following ALERTS were generated. Each ALERT has the format

test-name_ALERT_alert-type_alert-level.

Click on the hyperlinks for more details of the test.

Alert level C

PLAT911_ALERT_3_C Missing FCF Refl Between Thmin & STh/L= 0.600 13 Report

Alert level G

PLAT042_ALERT_1_G Calc. and Reported MoietyFormula Strings Differ	Please Check
PLAT328_ALERT_4_G Possible Missing H on sp3? Phosphorus	P1 Check
PLAT328_ALERT_4_G Possible Missing H on sp3? Phosphorus	P2 Check
PLAT380_ALERT_4_G Incorrectly? Oriented X(sp2)-Methyl Moiety	C7 Check
PLAT380_ALERT_4_G Incorrectly? Oriented X(sp2)-Methyl Moiety	C16 Check
PLAT910_ALERT_3_G Missing # of FCF Reflection(s) Below Theta(Min).	1 Note
PLAT912_ALERT_4_G Missing # of FCF Reflections Above STh/L= 0.600	10 Note
PLAT933_ALERT_2_G Number of OMIT Records in Embedded .res File ...	1 Note
PLAT941_ALERT_3_G Average HKL Measurement Multiplicity	3.3 Low
PLAT978_ALERT_2_G Number C-C Bonds with Positive Residual Density.	8 Info

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Structure factors have been supplied for datablock(s) 2-Fe

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No syntax errors found. CIF dictionary Interpreting this report

Datablock: 2-Fe

Bond precision: C-C = 0.0034 A Wavelength=0.71073

Cell: a=10.1947(2) b=14.2373(4) c=15.0677(3)
 alpha=62.275(2) beta=85.840(2) gamma=87.045(2)
Temperature: 100 K

	Calculated	Reported
Volume	1930.42(8)	1930.42(8)
Space group	P -1	P -1
Hall group	-P 1	-P 1
Moiety formula	C18 H36 K N2 O6, C16 H24 Fe P2	C16 H24 Fe P2, C18 H36 K N2 O6
Sum formula	C34 H60 Fe K N2 O6 P2	C34 H60 Fe K N2 O6 P2
Mr	749.73	749.73
Dx, g cm-3	1.290	1.290
Z	2	2
Mu (mm-1)	0.624	0.624
F000	802.0	802.0
F000'	803.77	
h,k,lmax	12,17,18	12,17,18
Nref	7890	7807
Tmin,Tmax	0.856,0.969	0.872,1.000
Tmin'	0.856	

Correction method= # Reported T Limits: Tmin=0.872 Tmax=1.000
AbsCorr = MULTI-SCAN

Data completeness= 0.989 Theta(max)= 26.370

R(reflections)= 0.0400(6351) wR2(reflections)= 0.1035(7807)

S = 1.090 Npar= 423

The following ALERTS were generated. Each ALERT has the format

test-name_ALERT_alert-type_alert-level.

Click on the hyperlinks for more details of the test.

Alert level C

PLAT094_ALERT_2_C	Ratio of Maximum / Minimum Residual Density	2.41	Report
PLAT906_ALERT_3_C	Large K Value in the Analysis of Variance	2.110	Check
PLAT911_ALERT_3_C	Missing FCF Refl Between Thmin & STh/L= 0.600	60	Report

Alert level G

PLAT042_ALERT_1_G	Calc. and Reported MoietyFormula Strings Differ	Please	Check
PLAT154_ALERT_1_G	The s.u.'s on the Cell Angles are Equal ..(Note)	0.002	Degree
PLAT328_ALERT_4_G	Possible Missing H on sp3? Phosphorus	P1	Check
PLAT328_ALERT_4_G	Possible Missing H on sp3? Phosphorus	P2	Check
PLAT883_ALERT_1_G	No Info/Value for _atom_sites_solution_primary .	Please	Do !
PLAT910_ALERT_3_G	Missing # of FCF Reflection(s) Below Theta(Min).	1	Note
PLAT912_ALERT_4_G	Missing # of FCF Reflections Above STh/L= 0.600	22	Note
PLAT933_ALERT_2_G	Number of OMIT Records in Embedded .res File ...	1	Note
PLAT941_ALERT_3_G	Average HKL Measurement Multiplicity	3.3	Low
PLAT978_ALERT_2_G	Number C-C Bonds with Positive Residual Density.	6	Info

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checkCIF/PLATON report

Structure factors have been supplied for datablock(s) 5

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No syntax errors found. CIF dictionary Interpreting this report

Datablock: 5

Bond precision: C-C = 0.0040 A Wavelength=0.71073

Cell: a=14.1554(8) b=9.6976(4) c=12.0182(6)
 alpha=90 beta=112.336(6) gamma=90

Temperature: 150 K

	Calculated	Reported
Volume	1526.00(15)	1525.99(14)
Space group	I 2/a	I 1 2/a 1
Hall group	-I 2ya	-I 2ya
Moiety formula	C18 H22 Zr	C18 H22 Zr
Sum formula	C18 H22 Zr	C18 H22 Zr
Mr	329.58	329.57
Dx, g cm-3	1.434	1.435
Z	4	4
Mu (mm-1)	0.705	0.705
F000	680.0	680.0
F000'	668.18	
h,k,lmax	17,12,15	17,12,15
Nref	1559	1558
Tmin,Tmax	0.905,0.974	0.775,1.000
Tmin'	0.889	

Correction method= # Reported T Limits: Tmin=0.775 Tmax=1.000
AbsCorr = GAUSSIAN

Data completeness= 0.999 Theta(max)= 26.386

R(reflections)= 0.0330(1401) wR2(reflections)= 0.0805(1558)

S = 1.068 Npar= 117

The following ALERTS were generated. Each ALERT has the format
test-name_ALERT_alert-type_alert-level.
Click on the hyperlinks for more details of the test.

Alert level G		
PLAT003_ALERT_2_G	Number of Uiso or Uij Restrained non-H Atoms ...	15 Report
PLAT187_ALERT_4_G	The CIF-Embedded .res File Contains RIGU Records	1 Report
PLAT301_ALERT_3_G	Main Residue Disorder(Resd 1)	53% Note
PLAT811_ALERT_5_G	No ADDSYM Analysis: Too Many Excluded Atoms	! Info
PLAT860_ALERT_3_G	Number of Least-Squares Restraints	144 Note
PLAT883_ALERT_1_G	No Info/Value for _atom_sites_solution_primary .	Please Do !
PLAT910_ALERT_3_G	Missing # of FCF Reflection(s) Below Theta(Min).	1 Note
PLAT912_ALERT_4_G	Missing # of FCF Reflections Above STh/L= 0.600	1 Note
PLAT941_ALERT_3_G	Average HKL Measurement Multiplicity	1.8 Low
PLAT955_ALERT_1_G	Reported (CIF) and Actual (FCF) Lmax Differ by .	1 Units
PLAT978_ALERT_2_G	Number C-C Bonds with Positive Residual Density.	8 Info

- 0 **ALERT level A** = Most likely a serious problem - resolve or explain
0 **ALERT level B** = A potentially serious problem, consider carefully
0 **ALERT level C** = Check. Ensure it is not caused by an omission or oversight
11 **ALERT level G** = General information/check it is not something unexpected
- 2 ALERT type 1 CIF construction/syntax error, inconsistent or missing data
2 ALERT type 2 Indicator that the structure model may be wrong or deficient
4 ALERT type 3 Indicator that the structure quality may be low
2 ALERT type 4 Improvement, methodology, query or suggestion
1 ALERT type 5 Informative message, check
-

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PLATON version of 16/07/2020; check.def file version of 12/07/2020

checkCIF/PLATON report

Structure factors have been supplied for datablock(s) 6

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No syntax errors found. CIF dictionary Interpreting this report

Datablock: 6

Bond precision: C-C = 0.0020 A Wavelength=1.54184
Cell: a=8.0727(2) b=8.74422(19) c=18.4065(4)
 alpha=103.1739(17) beta=92.6248(19) gamma=100.705(2)
Temperature: 150 K

	Calculated	Reported
Volume	1237.81(5)	1237.81(5)
Space group	P -1	P -1
Hall group	-P 1	-P 1
Moiety formula	C14 H17 P	C14 H17 P
Sum formula	C14 H17 P	C14 H17 P
Mr	216.25	216.24
Dx, g cm-3	1.160	1.160
Z	4	4
Mu (mm-1)	1.664	1.664
F000	464.0	464.0
F000'	466.15	
h,k,lmax	9,10,22	9,10,22
Nref	4541	4482
Tmin,Tmax	0.760,0.862	0.703,1.000
Tmin'	0.690	

Correction method= # Reported T Limits: Tmin=0.703 Tmax=1.000
AbsCorr = GAUSSIAN

Data completeness= 0.987 Theta(max)= 68.247

R(reflections)= 0.0329(4211) wR2(reflections)= 0.0899(4482)

S = 1.036 Npar= 279

The following ALERTS were generated. Each ALERT has the format
test-name_ALERT_alert-type_alert-level.
Click on the hyperlinks for more details of the test.

Alert level C
PLAT911_ALERT_3_C Missing FCF Refl Between Thmin & STh/L= 0.600 55 Report

Alert level G
PLAT412_ALERT_2_G Short Intra XH3 .. XHn H6A ..H5D . 2.02 Ang.
x,y,z = 1_555 Check
PLAT912_ALERT_4_G Missing # of FCF Reflections Above STh/L= 0.600 5 Note
PLAT941_ALERT_3_G Average HKL Measurement Multiplicity 2.9 Low
PLAT978_ALERT_2_G Number C-C Bonds with Positive Residual Density. 8 Info
PLAT992_ALERT_5_G Repd & Actual _reflns_number_gt Values Differ by 2 Check

0 **ALERT level A** = Most likely a serious problem - resolve or explain
0 **ALERT level B** = A potentially serious problem, consider carefully
1 **ALERT level C** = Check. Ensure it is not caused by an omission or oversight
5 **ALERT level G** = General information/check it is not something unexpected

0 ALERT type 1 CIF construction/syntax error, inconsistent or missing data
2 ALERT type 2 Indicator that the structure model may be wrong or deficient
2 ALERT type 3 Indicator that the structure quality may be low
1 ALERT type 4 Improvement, methodology, query or suggestion
1 ALERT type 5 Informative message, check

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checkCIF/PLATON report

Structure factors have been supplied for datablock(s) 7

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No syntax errors found. CIF dictionary Interpreting this report

Datablock: 7

Bond precision: C-C = 0.0040 A Wavelength=0.71073
Cell: a=20.161(3) b=9.0620(4) c=20.214(3)
 alpha=90 beta=119.71(2) gamma=90
Temperature: 100 K

	Calculated	Reported
Volume	3207.6(9)	3207.7(9)
Space group	C 2/c	C 1 2/c 1
Hall group	-C 2yc	-C 2yc
Moiety formula	C32 H48 Co2 P4	C32 H48 Co2 P4
Sum formula	C32 H48 Co2 P4	C32 H48 Co2 P4
Mr	674.44	674.44
Dx, g cm-3	1.397	1.397
Z	4	4
Mu (mm-1)	1.254	1.254
F000	1416.0	1416.0
F000'	1420.68	
h,k,lmax	24,10,24	24,10,24
Nref	2938	2935
Tmin,Tmax	0.835,0.882	0.861,1.000
Tmin'	0.606	

Correction method= # Reported T Limits: Tmin=0.861 Tmax=1.000
AbsCorr = MULTI-SCAN

Data completeness= 0.999 Theta(max)= 25.342

R(reflections)= 0.0333(2325) wR2(reflections)= 0.0834(2935)

S = 1.009 Npar= 180

The following ALERTS were generated. Each ALERT has the format
test-name_ALERT_alert-type_alert-level.
Click on the hyperlinks for more details of the test.

Alert level G

PLAT232_ALERT_2_G	Hirshfeld Test Diff (M-X) Col	--P1	.	5.5 s.u.
PLAT328_ALERT_4_G	Possible Missing H on sp3? Phosphorus			P1 Check
PLAT883_ALERT_1_G	No Info/Value for _atom_sites_solution_primary .			Please Do !
PLAT910_ALERT_3_G	Missing # of FCF Reflection(s) Below Theta(Min).			1 Note
PLAT912_ALERT_4_G	Missing # of FCF Reflections Above STh/L= 0.600			1 Note
PLAT933_ALERT_2_G	Number of OMIT Records in Embedded .res File ...			1 Note
PLAT978_ALERT_2_G	Number C-C Bonds with Positive Residual Density.			4 Info

0 **ALERT level A** = Most likely a serious problem - resolve or explain
0 **ALERT level B** = A potentially serious problem, consider carefully
0 **ALERT level C** = Check. Ensure it is not caused by an omission or oversight
7 **ALERT level G** = General information/check it is not something unexpected

1 ALERT type 1 CIF construction/syntax error, inconsistent or missing data
3 ALERT type 2 Indicator that the structure model may be wrong or deficient
1 ALERT type 3 Indicator that the structure quality may be low
2 ALERT type 4 Improvement, methodology, query or suggestion
0 ALERT type 5 Informative message, check

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PLATON version of 16/07/2020; check.def file version of 12/07/2020

checkCIF/PLATON report

Structure factors have been supplied for datablock(s) 8

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No syntax errors found. CIF dictionary Interpreting this report

Datablock: 8

Bond precision: C-C = 0.0050 A Wavelength=0.71073

Cell: a=12.7189(8) b=15.9437(10) c=17.8937(12)
 alpha=64.066(6) beta=85.655(5) gamma=86.321(5)

Temperature: 100 K

	Calculated	Reported
Volume	3251.9(4)	3251.9(4)
Space group	P -1	P -1
Hall group	-P 1	-P 1
Moiety formula	2(C18 H36 K N2 O6), 2(C8 H12 P), C4 H8 O	2(C8 H12 P), C4 H8 O, 2(C18 H36 K N2 O6)
Sum formula	C56 H104 K2 N4 O13 P2	C56 H104 K2 N4 O13 P2
Mr	1181.57	1181.57
Dx,g cm-3	1.207	1.207
Z	2	2
Mu (mm-1)	0.254	0.254
F000	1280.0	1280.0
F000'	1281.74	
h,k,lmax	15,19,22	15,19,22
Nref	13295	13040
Tmin,Tmax	0.958,0.987	0.826,1.000
Tmin'	0.910	

Correction method= # Reported T Limits: Tmin=0.826 Tmax=1.000
AbsCorr = MULTI-SCAN

Data completeness= 0.981 Theta(max)= 26.372

R(reflections)= 0.0673(8417) wR2(reflections)= 0.1979(13040)

S = 1.063 Npar= 1310

The following ALERTS were generated. Each ALERT has the format

test-name ALERT_alert-type_alert-level.

Click on the hyperlinks for more details of the test.

Alert level C

PLAT088_ALERT_3_C	Poor Data / Parameter Ratio	9.95	Note
PLAT329_ALERT_4_C	Carbon Atom Hybridisation Unclear for		C35B Check
PLAT340_ALERT_3_C	Low Bond Precision on C-C Bonds	0.005	Ang.
PLAT906_ALERT_3_C	Large K Value in the Analysis of Variance	3.712	Check
PLAT911_ALERT_3_C	Missing FCF Refl Between Thmin & STh/L= 0.600	168	Report
PLAT977_ALERT_2_C	Check Negative Difference Density on H6B	-0.35	eA-3
PLAT977_ALERT_2_C	Check Negative Difference Density on H6C	-0.40	eA-3

Alert level G

PLAT002_ALERT_2_G	Number of Distance or Angle Restraints on AtSite	10	Note
PLAT003_ALERT_2_G	Number of Uiso or Uij Restrained non-H Atoms ...	136	Report
PLAT042_ALERT_1_G	Calc. and Reported MoietyFormula Strings Differ		Please Check
PLAT176_ALERT_4_G	The CIF-Embedded .res File Contains SADI Records	3	Report
PLAT178_ALERT_4_G	The CIF-Embedded .res File Contains SIMU Records	5	Report
PLAT301_ALERT_3_G	Main Residue Disorder(Resd 1)	100%	Note
PLAT301_ALERT_3_G	Main Residue Disorder(Resd 3)	100%	Note
PLAT301_ALERT_3_G	Main Residue Disorder(Resd 4)	100%	Note
PLAT302_ALERT_4_G	Anion/Solvent/Minor-Residue Disorder (Resd 2)	100%	Note
PLAT302_ALERT_4_G	Anion/Solvent/Minor-Residue Disorder (Resd 6)	100%	Note
PLAT302_ALERT_4_G	Anion/Solvent/Minor-Residue Disorder (Resd 7)	100%	Note
PLAT302_ALERT_4_G	Anion/Solvent/Minor-Residue Disorder (Resd 8)	100%	Note
PLAT302_ALERT_4_G	Anion/Solvent/Minor-Residue Disorder (Resd 9)	100%	Note
PLAT304_ALERT_4_G	Non-Integer Number of Atoms in (Resd 1)	57.33	Check
PLAT304_ALERT_4_G	Non-Integer Number of Atoms in (Resd 2)	55.50	Check
PLAT304_ALERT_4_G	Non-Integer Number of Atoms in (Resd 3)	5.67	Check
PLAT304_ALERT_4_G	Non-Integer Number of Atoms in (Resd 4)	7.50	Check
PLAT304_ALERT_4_G	Non-Integer Number of Atoms in (Resd 6)	17.14	Check
PLAT304_ALERT_4_G	Non-Integer Number of Atoms in (Resd 7)	3.86	Check
PLAT304_ALERT_4_G	Non-Integer Number of Atoms in (Resd 8)	10.85	Check
PLAT304_ALERT_4_G	Non-Integer Number of Atoms in (Resd 9)	2.14	Check
PLAT380_ALERT_4_G	Incorrectly? Oriented X(sp2)-Methyl Moiety	C5	Check
PLAT380_ALERT_4_G	Incorrectly? Oriented X(sp2)-Methyl Moiety	C6	Check
PLAT380_ALERT_4_G	Incorrectly? Oriented X(sp2)-Methyl Moiety	C7	Check
PLAT380_ALERT_4_G	Incorrectly? Oriented X(sp2)-Methyl Moiety	C8	Check
PLAT380_ALERT_4_G	Incorrectly? Oriented X(sp2)-Methyl Moiety	C31A	Check
PLAT380_ALERT_4_G	Incorrectly? Oriented X(sp2)-Methyl Moiety	C32A	Check
PLAT380_ALERT_4_G	Incorrectly? Oriented X(sp2)-Methyl Moiety	C33A	Check
PLAT380_ALERT_4_G	Incorrectly? Oriented X(sp2)-Methyl Moiety	C34A	Check
PLAT398_ALERT_2_G	Deviating C-O-C Angle From 120 for O13A	100.9	Degree
PLAT398_ALERT_2_G	Deviating C-O-C Angle From 120 for O13B	87.6	Degree
PLAT413_ALERT_2_G	Short Inter XH3 .. XHn H5A ..H53C .	2.09	Ang.
	1-x,1-y,1-z =	2_666	Check
PLAT413_ALERT_2_G	Short Inter XH3 .. XHn H6C ..H42A .	2.09	Ang.
	1-x,-y,1-z =	2_656	Check
PLAT413_ALERT_2_G	Short Inter XH3 .. XHn H6C ..H42C .	2.09	Ang.
	1-x,-y,1-z =	2_656	Check
PLAT720_ALERT_4_G	Number of Unusual/Non-Standard Labels	4	Note
PLAT773_ALERT_2_G	Check long C-C Bond in CIF: C35B --C36B	1.78	Ang.
PLAT773_ALERT_2_G	Check long C-C Bond in CIF: C45B --C46B	1.86	Ang.
PLAT790_ALERT_4_G	Centre of Gravity not Within Unit Cell: Resd. # C18 H36 K N2 O6	2	Note
PLAT790_ALERT_4_G	Centre of Gravity not Within Unit Cell: Resd. # C18 H36 K N2 O6	4	Note
PLAT811_ALERT_5_G	No ADDSYM Analysis: Too Many Excluded Atoms		! Info

PLAT860_ALERT_3_G	Number of Least-Squares Restraints	2121	Note
PLAT910_ALERT_3_G	Missing # of FCF Reflection(s) Below Theta(Min).	4	Note
PLAT912_ALERT_4_G	Missing # of FCF Reflections Above STh/L= 0.600	84	Note
PLAT933_ALERT_2_G	Number of OMIT Records in Embedded .res File ...	4	Note
PLAT941_ALERT_3_G	Average HKL Measurement Multiplicity	2.4	Low
PLAT978_ALERT_2_G	Number C-C Bonds with Positive Residual Density.	5	Info
PLAT992_ALERT_5_G	Repd & Actual _reflns_number_gt Values Differ by	1	Check

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47 **ALERT level G** = General information/check it is not something unexpected

1 ALERT type 1 CIF construction/syntax error, inconsistent or missing data
13 ALERT type 2 Indicator that the structure model may be wrong or deficient
10 ALERT type 3 Indicator that the structure quality may be low
28 ALERT type 4 Improvement, methodology, query or suggestion
2 ALERT type 5 Informative message, check

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