Supporting Information

Solvatomorphism and Phase Transformation of CL-20: Probing

Properties and Investigating Mechanisms

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Preparation of the powdered solvate samples

Samples of CL-20·DMC and CL-20·DES were prepared by adding a requisite amount of ϵ -CL-20 into the pure solvents and heating the mixtures to certain temperatures below their respective boiling points. Then the solvates were crystallized from the solutions by spontaneous cooling to 20 °C.

CL-20·EMC and CL-20·DEO was prepared by producing the saturated solution of ϵ -CL-20 in EMC and DEO respectively at 50 °C and subsequently evaporating the solution at 20 °C for a few hours to obtain the solvates.

CL-20·ACN, CL-20·DMA, CL-20·DMF and CL-20· γ -BL were prepared by antisolvent crystallization method. Concentrated solutions of CL-20 in ACN, DMA, DMF and γ -BL were prepared at 20 °C. A certain amount of CTC was added into these solutions and kept stirring for 30 min, then we obtained corresponding solvates.

CL-20·PX was prepared by slurry method wherein excess ε-CL-20 was added to PX and allowed to equilibrated for more than 72 h at 60 °C. The mixture was cooled to 20 °C and CL-20·PX formed.

These precipitated solvates were stored in the mother liquor, filtered and dried before characterization.

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Solvent	Group ^a	Classification ^b	Crystal form
<i>n</i> -Heptane	1	AAA	N/A ^c
Cyclohexane	1	AAA	N/A ^c
Ethyl formate	2	AP	Form ε
Ethyl acetate	2	AP	Form ε
Butyl acretate	2	AP	Form ε
Dimethyl carbonate	2	AP	CL-20. DMC solvate
Diethyl carbonate	2	AP	Form ε
Methanol	3	HBD	Form ε
Ethanol	3	HBD	Form ε
1-Propanol	3	HBD	Form ε
2-Propanol	3	HBD	Form <i>e</i>
Toluene	4	AALP	N/A ^c
o-Xylene	4	AALP	Form <i>e</i>
<i>m</i> -Xylene	4	AALP	Form <i>e</i>
Acetone	5	AP	Form <i>e</i>
Cyclohexanone	5	AP	Form <i>e</i>
N,N-dimethylacetamide	6	AP	CL-20·DMA solvate
Dichloromethane	7	AP	Form <i>e</i>
Chloroform	7	EPD/HBD/AP	N/A ^c
Nitromethane	8	AP	Form <i>ε</i>

Table S1 Solvent classification and the crystal forms obtained during solvate screening.

^a Groups are based on the following solvent parameters: hydrogen bonding capability, dielectric constant, polarity/dipolarity and dipole moment.

^b Abbreviated instructions: AAA = aliphatic aprotic apolar, AP = aprotic polar, HBD = hydrogen bond donors, AALP = aromatic apolar or lightly polar, EPD = electron pair donors.
^c No crystals were obtained due to too low solubility.

FTIR spectra

The FTIR spectra of CL-20 solvates and polymorphs are presented in Fig. S1. Some new peaks appeared due to the inclusion of different solvent molecules, which were marked in the spectra using asterisks. The noticeable differences can be observed in the carbonyl group (C=O) stretching region, since most of the solvent molecules are esters and amides. Compared with the FTIR spectra of pure solvents, some red shifts can be observed in the C=O stretching vibration, appearing at 1742, 1737, and 1738 cm⁻¹ for CL-20·DMC, CL-20·EMC, and CL-20·DEO, respectively.



Fig. S1 The FTIR spectra of CL-20 solvates and polymorphs (a) CL-20·DMC, (b) CL-20·EMC, (c) CL-20·DES, (d) CL-20·DEO, (e) CL-20·ACN, (f) CL-20·DMA, (g) CL-20·DMF, (h) CL-20·PX, (i) CL-20·γ-BL, (j) form β, (k) form γ, and (l)form ε. The asterisks mark the bands due to the incorporation of solvent molecules.

Raman spectra

Raman measurements were also conducted to characterize CL-20 solvates owing to the convenient sample preparation method, which has no destructive effect on the crystal samples. The Raman spectra of the nine solvates are summarized and compared with the figures of three polymorphs in Fig. S2. The CL-20 solvates are readily distinguishable from the other three polymorphs, based on distinct differences in peak positions and relative intensities. The most easily noticeable difference in Raman spectra was in range from 200 cm⁻¹ to 400 cm⁻¹. The peak at 301 cm⁻¹ appears for CL-20·DMC and CL-20·EMC, whereas the peak for CL-20·DES moves slightly to 305 cm⁻¹ with a tiny hump following it. CL-20·DEO shows peak at 288 cm⁻¹ and CL-20·ACN shows a single peak at 273 cm⁻¹. CL-20·DMA and CL-20·DMF are characterized by the peak at 308 cm⁻¹ and 314 cm⁻¹, respectively. CL-20·PX and CL-20· γ -BL exhibit characteristic peaks at 321 cm⁻¹ and 310 cm⁻¹, respectively. Moreover, CL-20·DMC and CL-20·EMC show similar Raman spectra throughout the frequency range owing to the structural similarities.



Fig. S2 The Raman spectra of CL-20 solvates and polymorphs (a) CL-20·DMC, (b) CL-20·EMC,
(c) CL-20·DES, (d) CL-20·DEO, (e) CL-20·ACN, (f) CL-20·DMA, (g) CL-20·DMF, (h) CL-20·PX, (i) CL-20·γ-BL, (j) form β, (k) form γ, and (l)form ε.

Table S2 The torsion angles of CL-20 molecules in CL-20 polymorphs and solvates.

Crystal form	C(1)-C(2)-N(1)-N(2) ^a	C(2)-C(1)-N(5)-N(4)	C(4)-C(3)-N(5)-N(6)
Form β	-67.24	138.95	150.45
Form γ	-141.07	155.33	164.92
Form ε	-142.26	67.14	147.88
CL-20·DMC	-149.27	154.56	169.22
CL-20·EMC	-151.80	152.14	167.72
CL-20. DES	-146.18	152.13	167.47
CL-20·DEO	-136.80	141.56	158.14
CL-20·ACN	-70.66	152.02	143.42
CL-20·DMA	-140.73	153.43	162.15
CL-20·DMF	-139.95	126.86	143.22
CL-20·PX	-66.98	146.27	143.67
CL-20·γ-BL	-145.95	155.96	159.79

^a The definition of atomic numberings is provided in Fig. 5a.



Fig. S3 Intermolecular interactions of (a) CL-20 molecular layer and (b) DMC molecular layer in

CL-20 · DMC (viewed along *c*-axis).



Fig. S4 Intermolecular interactions of CL-20 molecular layer in CL-20 EMC (viewed along c-



Fig. S5 Intermolecular interactions of (a) CL-20 molecular layer and (b) DES molecular layer in CL-20·DES (viewed along *c*-axis).



Fig. S6 Intermolecular interactions of (a)-(b) the two kinds of CL-20 molecular layers and (c)-(d) the two kinds of DMA molecular layers in CL-20 DMA (viewed along *c*-axis).



Fig. S7 (a) Intermolecular interactions between adjacent infinite CL-20 chains (viewed along *a*-axis); and (b) the host-guest intermolecular interactions (viewed along *b*-axis) in CL-20·DEO.

Structural information of CL-20·ACN (CCDC: 1569059)

CL-20·ACN belongs to the monoclinic system, space group C2/c, the molecular stoichiometry ratio of CL-20 to acetonitrile is 1:1. Compared with the published crystal structures of CL-20 polymorphs, the molecular conformation of CL-20 in acetonitrile solvate is close to form β . It can also be observed that the CL-20 acetonitrile solvate is a channel-type solvate, in which the acetonitrile molecules

occupy the channels along *b*-axis. CL-20 and ACN molecules interact depending on the C–H \cdots O, C–H \cdots N, and O \cdots C intermolecular interactions.

Structural information of CL-20·DMF (CCDC: 832144)

CL-20·DMF is crystallized in the triclinic crystal system, space group $P\bar{1}$, and molecules are arranged in alternate layers of CL-20 and DMF. In DMF layer, two independent molecules form dimers over inversion centers. In CL-20 layers, the molecules adopt the molecular conformation of γ -polymorph, and the CL-20 molecules are lined by weak C–H···O interactions along two dimensions. In the third direction (perpendicular to the layers), the most significant intermolecular interaction occurs between the O atoms of DMF and the H atoms of CL-20 (C–H···O, 2.241 to 2.732 Å).

Structural information of CL-20·PX (CCDC: 1022276)

CL-20·PX belongs to the orthorhombic system, space group *Pbca*, and the molecular number ratio of CL-20 to *p*-xylene is 2:1. The γ -conformation of the CL-20 molecules can be observed in the crystal structure of CL-20·PX. Alternation of two CL-20 layers and one *p*-xylene layer can be found. In CL-20 layers, a network of weakly linked CL-20 molecules is established through C–H···O intermolecular interactions. The most important host-guest interactions exist between the methyl H atoms of *p*-xylene and the nitro O atoms of CL-20. Besides, the O···C and O···N stacking interactions between CL-20 and *p*-xylene also stabilize the structure of CL-20·PX.

Structural information of CL-20·γ-BL (CCDC: 832146)

CL-20· γ -BL is crystallized in the monoclinic system, space group *Cc*, and the molecular number ratio of CL-20 to γ -BL is 1:1. There are four CL-20 molecules and four γ -BL molecules in one asymmetric unit. The CL-20 molecules adapt the γ -

conformation and are linked by C–H···O (2.326(6) to 2.663(6) Å) to form chains running parallel to the *a*-axis. Moreover, the chains can be arranged in two layers, resulting in a zig-zag network of CL-20 molecules in the *ab*-plane. The γ -BL molecules form two chains running parallel to the *a*-axis, and no significant interactions are observed between γ -BL molecules themselves. The host and guest molecules are contacted by C–H···O interactions between the H atoms of CL-20 and the ring/carbonyl oxygen of γ -BL.



Fig. S8 The packing diagrams for the already published CL-20 solvates. Packing diagrams for (b),(c) and (d) are viewed along *a*-axis; packing diagrams for (a) is viewed along *b*-axis.

D—H···A ^b	H…A/Å	D…A/Å	D—H····A/°	D—H…A	H…A/Å	D…A/Å	D—H···A/°
	Form	β			CL-20·DM	ſA	
С7-Н7⋯О4А	2.55	3.273(5)	135	С4-Н4…О25	2.11	2.857(5)	130
С9-Н9…О6А	2.58	3.233(4)	134	С7-Н7⋯О26	2.13	2.795(5)	122

Table S3 Geometric parameters of intermolecular interactions in CL-20 polymorphs and solvates.^a

C11-H11O10B	2.54	3.161(4)	127	С8-Н8…О28	2.42	2.905(5)	109
	Form	γ		С9-Н9…О28	2.49	2.933(5)	106
С1-Н1…О6А	2.55	3.324(4)	137	С10-Н10…О26	2.53	3.018(6)	110
С9-Н9…О2А	2.47	3.326(3)	156	С15-Н15С…О24	2.55	3.300(8)	133
Form <i>ε</i>			С16-Н16В…О24	2.58	3.441(8)	147	
С1-Н1…О6В	2.44	3.177(3)	136	С20-Н20В…О10	2.46	3.174(8)	129
С9-Н9…О2А	2.52	3.360(3)	150	С23-Н23С…О23	2.54	3.499(9)	165
C11-H11O8B	2.54	3.052(3)	116		CL-20·D	MF	
	CL-20·I	DMC		C2-H21…O40	2.54	2.913(2)	102
C4-H4…O31	2.34	2.933(3)	117	C5-H51…O31	2.40	2.891(2)	110
С7-Н7⋯О29	2.38	3.067(3)	126	С6-Н61…О31	2.24	2.797(3)	115
С9-Н9…О29	2.26	3.008(3)	131	С34-Н342…О26	2.51	3.245(2)	134
С10-Н10…О25	2.34	3.117(3)	134		CL-20·γ·	-BL	
С1-Н1…О32	2.28	3.235(3)	159	C107-H1071O138	2.40	2.900(10)	111
C15-H15A…O13	2.53	3.488(4)	165	C114-H1141O138	2.45	2.916(10)	109
С20-Н20С…О20	2.59	3.167(4)	118	С122-Н1221…Об	2.47	3.417(10)	163
С20-Н20С…О8	2.60	3.500(4)	153	С22-Н221…О137	2.39	2.898(11)	112
CL-20.DES			С23-Н231…О137	2.50	2.973(11)	109	
С1-Н1…О14	2.58	3.042(3)	108	С30-Н301…О7	2.55	3.506(10)	166
С2-Н2…О14	2.54	3.013(3)	109	С59-Н591…О7	2.32	2.845(10)	114
C4-H4…O13	2.34	3.022(3)	125	С82-Н821…Об	2.28	2.851(10)	116
С1-Н1…О16	2.46	3.419(3)	161	С112-Н1121…О1	2.29	3.122(10)	142
С7-Н7А…О8	2.55	3.512(4)	163	С17-Н171…О9	2.32	3.128(10)	140
С7-Н7В…О5	2.58	3.424(4)	143	С57-Н571…О133	2.35	3.149(10)	139
	CL-20·1	DEO		C87-H871…O140	2.41	3.183(10)	135
С1-Н1…О15	2.50	3.028(2)	112	C134-H1342…O117	2.46	3.067(10)	121
С2-Н2…О15	2.57	3.072(2)	111	C134-H1341O99	2.50	3.139(11)	124
С3-Н3…О13	2.49	2.903(2)	104	C141-H1412O26	2.44	3.257(11)	142
C4-H4…O13	2.59	2.961(2)	102	C141-H1411O35	2.47	3.037(12)	116
С1-Н1…О14	2.55	3.414(2)	145	C1-H11O131	2.48	3.348(13)	146
C8-H8B…O1	2.59	3.371(2)	136	C1-H12…O27	2.54	3.356(13)	142
C12-H12A…O5	2.54	3.206(2)	125	C135-H1351…O42	2.39	3.132(13)	135
	CL-20.4	ACN		C2-H21…O49	2.45	3.130(11)	126
C1-H1A…N13	2.41	3.205(3)	138	C2-H22…O67	2.42	3.054(12)	122
							120
C4-H4A…N13	2.36	3.180(3)	140	C10-H101O20	2.59	3.189(10)	
С7-Н7В…О9	2.57	3.348(5)	145	C10-H101O80	2.40	3.056(10)	125
	CL-20·I	DMA		C10-H102O105	2.45	3.245(11)	138
С2-Н2…О27	2.52	2.945(5)	105	С13-Н132…О129	2.45	3.052(12)	120
С3-Н3…О27	2.44	2.900(5)	108				

^a Definition of the atomic numberings can be obtained in CIF files.

^b D = Donor, A = Acceptor.



Fig. S9. Overlay of the packing motifs of CL-20 molecules in (a) CL-20 DMC + CL-20 EMC, (b)

 $CL-20 \cdot DES + CL-20 \cdot EMC.$



Fig. S10 The packing coefficients of CL-20 polymorphs and solvates.

Hirshfeld surface analysis

For CL-20 solvates, the middle part between the spikes refer to $H \cdots H$ and $O \cdots O$ contacts. CL-20·DES, CL-20·ACN, CL-20·DMA, CL-20·DMF and CL-20· γ -BL exhibit obvious $H \cdots H$ contacts. The wings regions are more complicated and represent the $H \cdots H$, $O \cdots H$, $N \cdots H$ and $C \cdots H$ contacts. The wings regions of CL-20·DES and CL-20·DMC correspond to the $C \cdots H$ contact. The wings regions of CL-20·DES and CL-10

20·DEO represent the H···H, O···H and N···H contacts (see Fig. S11a-b). For CL-20·ACN, the C···H contact exists on the two sides of the plot, and the two small wings represent the H···H, O···H, N···H, and C···H contacts (see Fig. S11c). The wings regions of CL-20·DMA depict the H···H, O···H, N···H and C···H contacts (see Fig. S11d). The C···H contact also exists on the two sides of the plot of CL-20·DMF. The wings regions of CL-20· γ -BL represent the O···H and N···H contacts (see Fig. S11e). In addition, the C···O and C···N contacts are also observed in the plots of the solvates.

For CL-20 polymorphs, the middle part between the two sharp spikes is the $O \cdots O$ contact between the O atoms of nitro groups. Besides, there also exist $O \cdots N$, $H \cdots H$, $N \cdots H$ and $N \cdots N$ contacts.



Fig. S11 The 2D fingerprint plots delineated into H····H, O···H, N···H and C···H contacts for (a) CL-20·DES, (b) CL-20·DEO, (c) CL-20·ACN, (d) CL-20·DMA, and (e) CL-20·γ-BL.



Fig. S12 The TG-DTA curves of (a)-(i) CL-20 solvates and (j)-(l) CL-20 polymorphs.



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Fig. S13 The VT-PXRD patterns of CL-20 solvates.



Fig. S14 The PXRD patterns of desolvation products of (a) CL-20 DES, (b) CL-20 DEO, (c) CL-

20 · DMF, (d) CL-20 · PX, and (e) CL-20 · γ-BL.

Table S4 Comparison of the desolvation and phase transformation temperatures observed in

Solvates or polymorphs	Desolvation onset temperature in DSC T _{desol} (°C)	Observed start temperature of desolvation in VT- PXRD (°C)	Peak temperature of phase transformation $T_{\rm P}$ (°C)	Observed start temperature of phase transformation in VT- PXRD (°C)
CL-20·DMC	58.1	60	128.5	125
CL-20·EMC	26.2	30	127.6	125
CL-20. DES	98.0	100		
CL-20 DEO	70.6	65	128.1	125
CL-20·ACN	45.7	45	137.7	135
CL-20·DMA	63.7	65		
CL-20·DMF	66.2	65	131.2	135
CL-20·PX	87.0	85		
CL-20·γ-BL	95.4	90		
Form β			141.83	140
From ε			159.47	155

DSC and VT-PXRD measurements, respectively.



Fig. S15 Packing similarity tree diagram of CL-20 polymorphs and solvates.