

Supplementary Information:

**Semi-conductive helical homochiral metal–organic frameworks
based on enantiomeric proline derivatives**

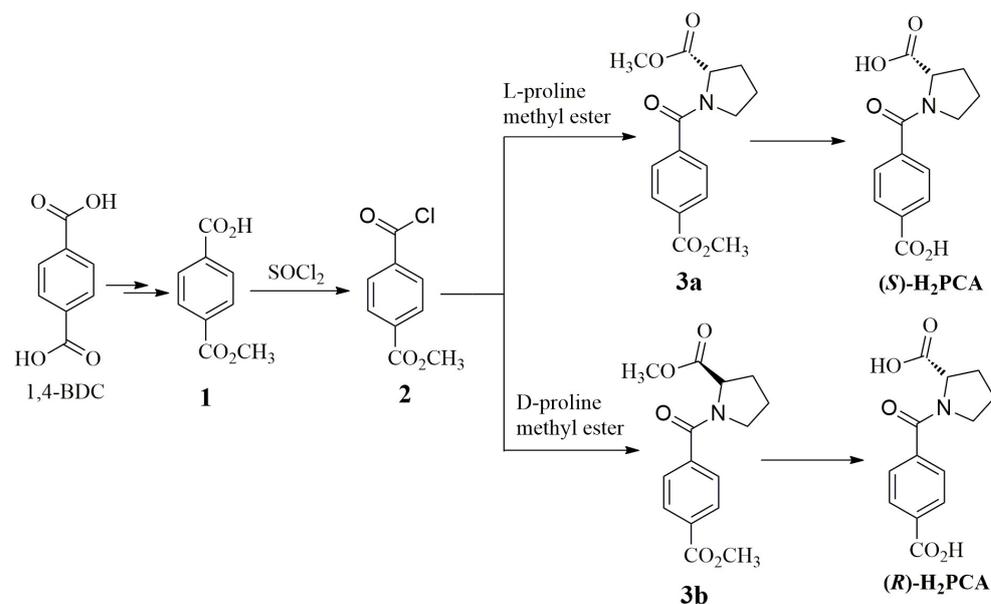
Yu-Lu Ma,^b Qin Meng^a and Zhong-Xuan Xu^{*a}

^aSchool of chemistry and chemical engineering, Zunyi Normal College, Zunyi,
Guizhou 563002, P.R China

^bSchool of Pharmaceutical Sciences and Yunnan Key Laboratory of Pharmacology for
Natural Products, Kunming Medical University, Kunming, 650500, P.R China.

E-mail: xuzhongxuan4201@163.com

1. Synthesis of (*S*)-H₂PCA and (*R*)-H₂PCA



Scheme S1. Synthetic routes to the compounds (*S*)-H₂PCA and (*R*)-H₂PCA

1.1 Synthesis of methyl 1-(4-(methoxycarbonyl)benzoyl)pyrrolidine-2-carboxylate (**3**)

Compound 1 (1.8g, 10mmol), freshly distilled SOCl₂ (20 mL) and dimethylformamide (0.1 mL) were added in a round-bottomed flask under a nitrogen atmosphere. After the reaction mixture was heated at 90 °C for three hours, the excess SOCl₂ was removed under in vacuo, giving compound 2 as a colorless liquid. Compound 2 in dry CH₂Cl₂ (15 mL) was added dropwise into a solution of methyl ester of L-proline or D-proline (1.6 g, 12 mmol) in the dry CH₂Cl₂ (50 mL) and triethylamine (2.2 g, 22 mmol) under a nitrogen atmosphere and ice-water bath. The reaction mixture was washed with 1.0 M HCl (2 × 150 mL) and saturated NaCl (2 × 15 mL), then dried over anhydrous sodium sulfate. The solvent was removed in vacuo and the residue was purified by flash column chromatography (EtOAc: petroleum ether = 1:2) to give compound 3 as a white solid (2g, 68%). ¹HNMR (400 MHz, CDCl₃), δ (ppm): 8.11–8.06 (2H, d), 7.65–7.63 (2H, d), 3.94 (3H, s), 3.79 (3H, s),

3.65–3.61 (1H, m), 3.52–3.46 (1H, m), 2.35–2.33(1H, m), 2.01–1.90(3H, m), ^{13}C NMR (100 MHz, CDCl_3), $\delta(\text{ppm})$: 172.46, 168.97, 165.49, 140.46, 131.75, 129.86, 127.45, 58.83, 52.40, 49.52, 29.36, 24.78; LRSM (ESI): Mass calcd for $\text{C}_{15}\text{H}_{17}\text{NO}_5$ $[\text{M}+\text{H}]^+$, 292.3; found 292.3.

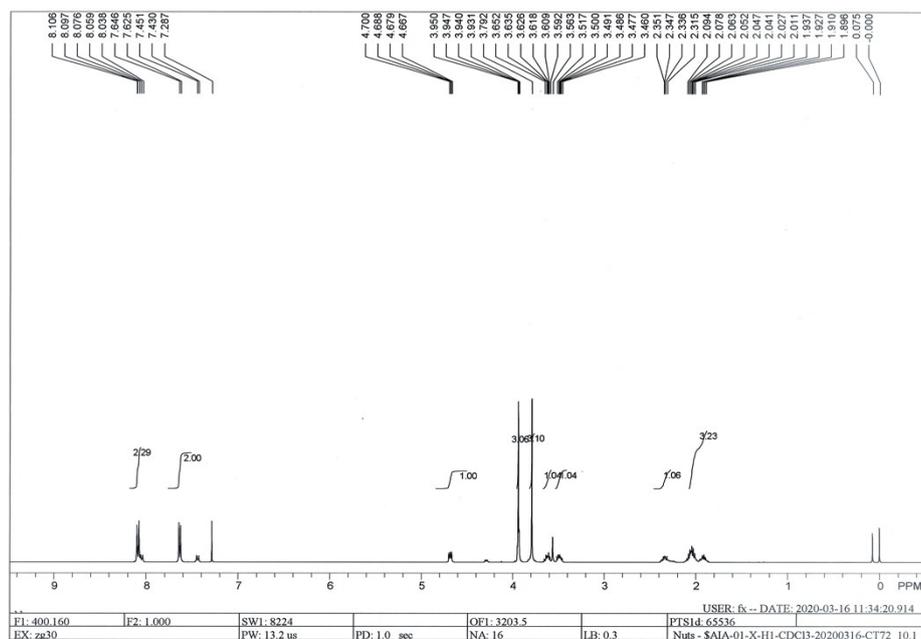


Figure S1. ^1H NMR of compound **3** in CDCl_3

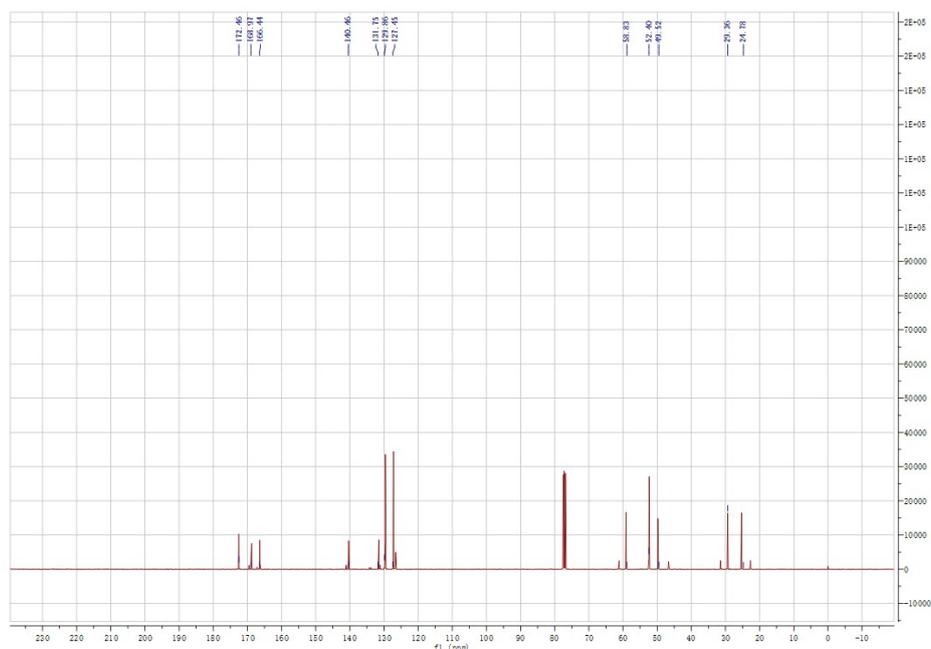
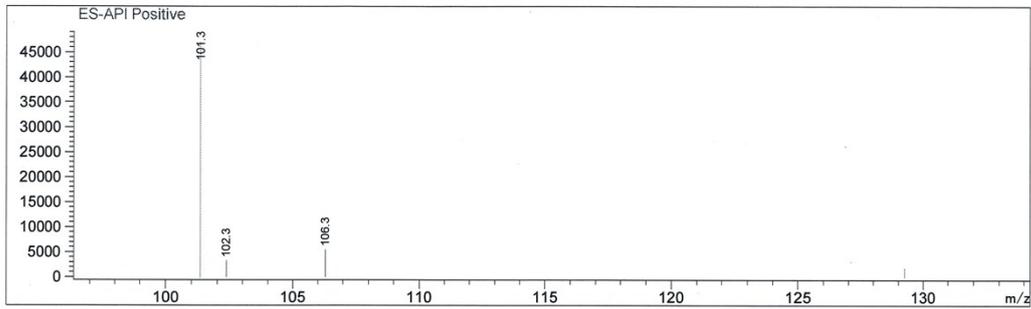


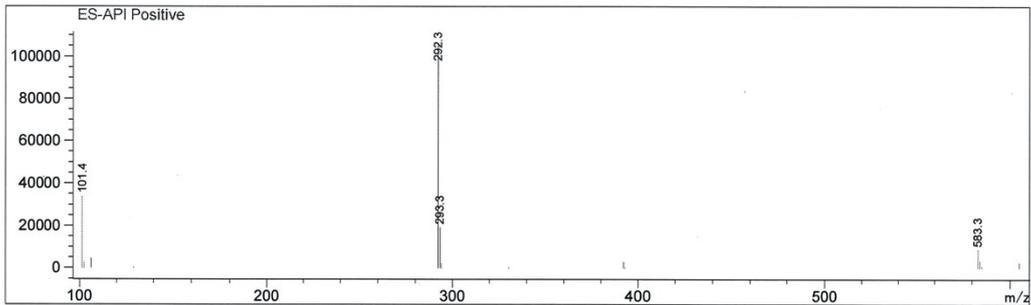
Figure S2. ^{13}C NMR of compound **3** in CDCl_3

LC/MS Report

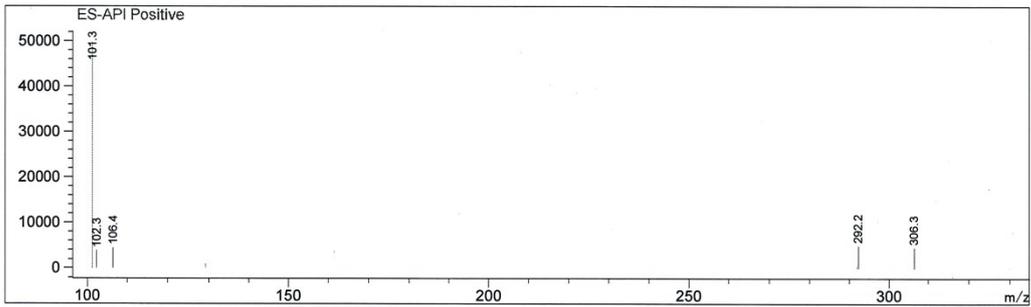
Ret. Time: 1.69 <<<< POSITIVE SPECTRA >>>>



Ret. Time: 1.77



Ret. Time: 1.88



Ret. Time: 1.98

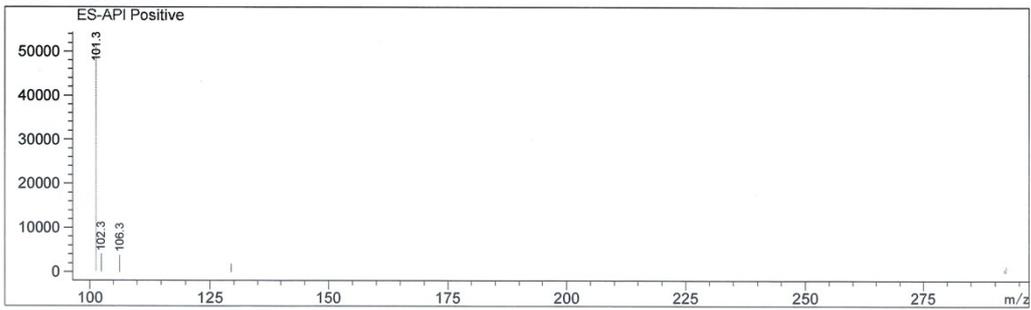


Figure S3. The LRSM of compound 3

1.2 Synthesis of (*S*)-H₂PCA and (*R*)-H₂PCA

Compound **3** (1.45g, 5mmol), CH₃OH (10 mL), H₂O (20 mL) and sodium hydroxide (0.8 g, 20 mmol) were added to a 50mL flask with a stirring bar. After the reaction mixture was stirred and heated at 50 °C for 6 h, the result solution was slowly acidified to pH 1–2 in an ice bath. The (*S*)-H₂PCA ((*R*)-H₂PCA) was obtained as a white solid: ¹H NMR (400 MHz, CD₃OD), δ (ppm): 8.12–8.10 (2H, d), 7.67–7.65 (2H, d), 4.61–4.58 (2H, q), 3.61–3.50 (2H, m), 2.41–2.38 (1H, m), 2.01–1.90 (3H, m), ¹³C NMR (100 MHz, CD₃OD), δ(ppm): 173.25, 167.72, 167.06, 140.72, 132.57, 129.79, 127.64, 59.12, 49.51, 29.42, 25.07; LRSM (ESI): Mass calcd for C₁₃H₁₃NO₅ [M+H]⁺, 264.3; found 264.3.

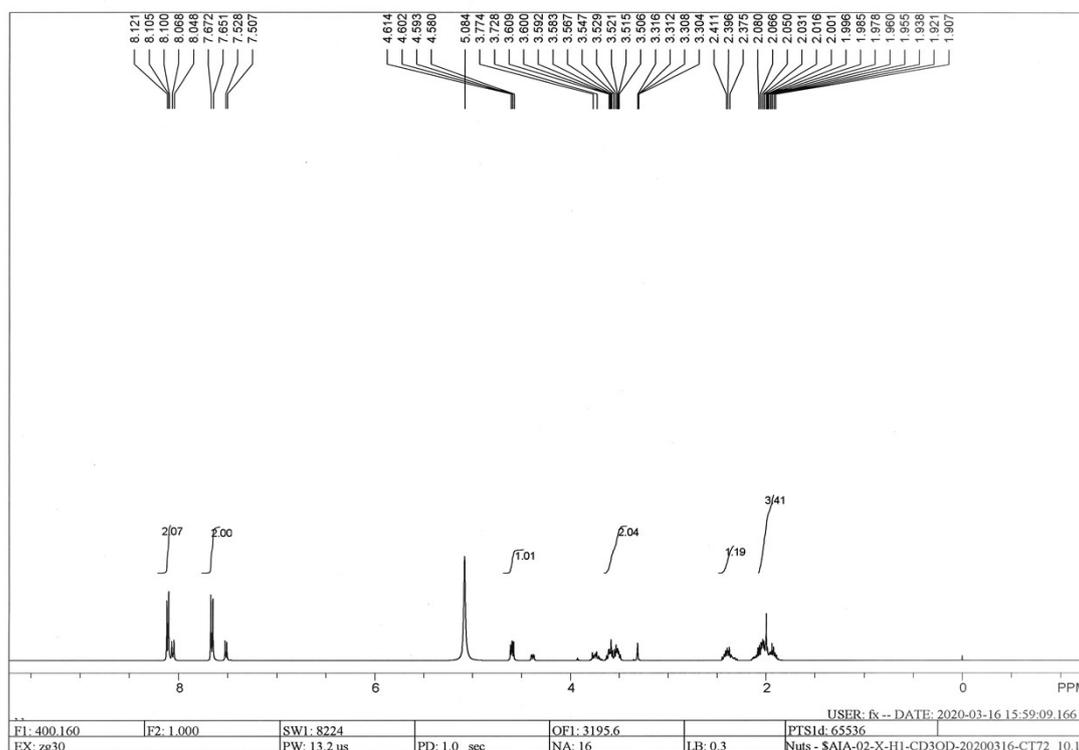


Figure S4. ¹H NMR of H₂PCA in CD₃OD

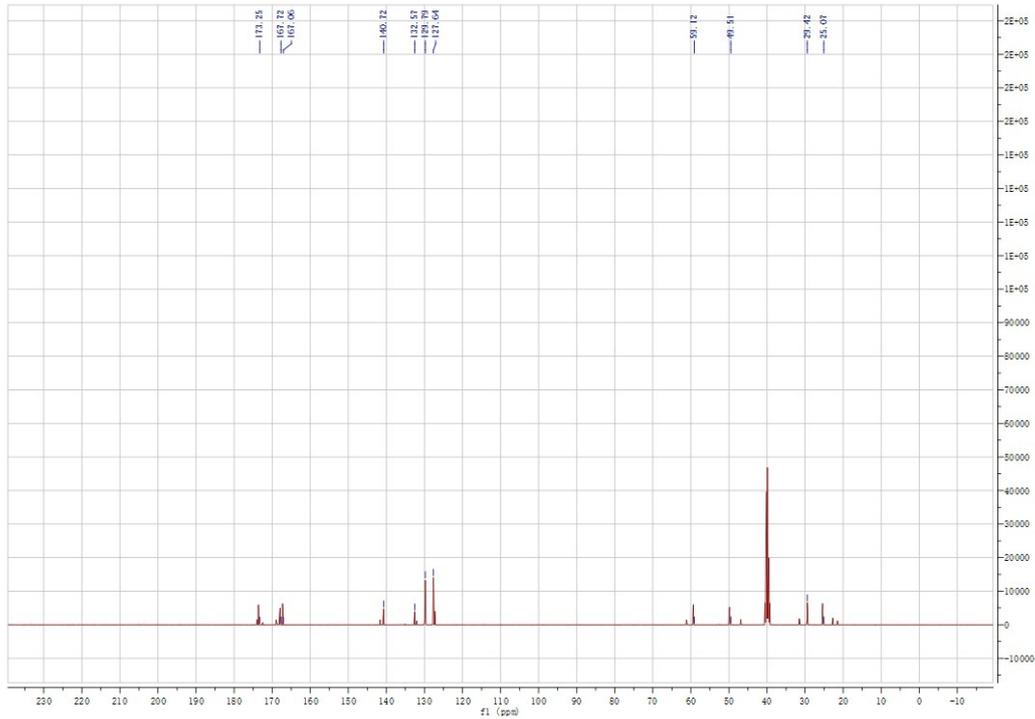


Figure S5. ^{13}C NMR of H_2PCA in CD_3OD

LC/MS Report

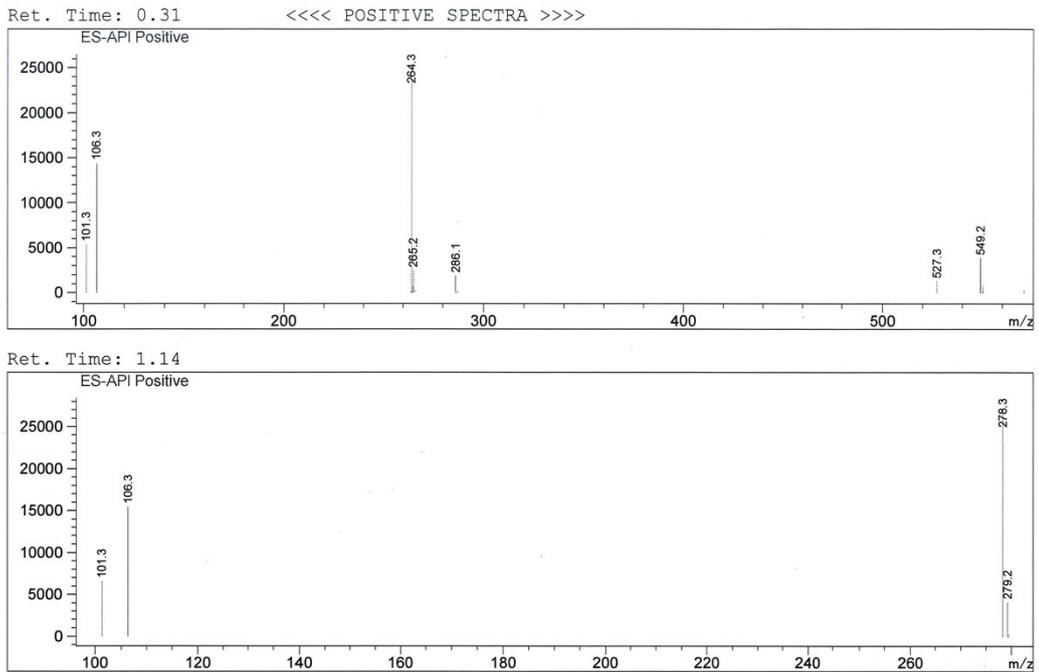


Figure S6. The LRMS of H_2PCA

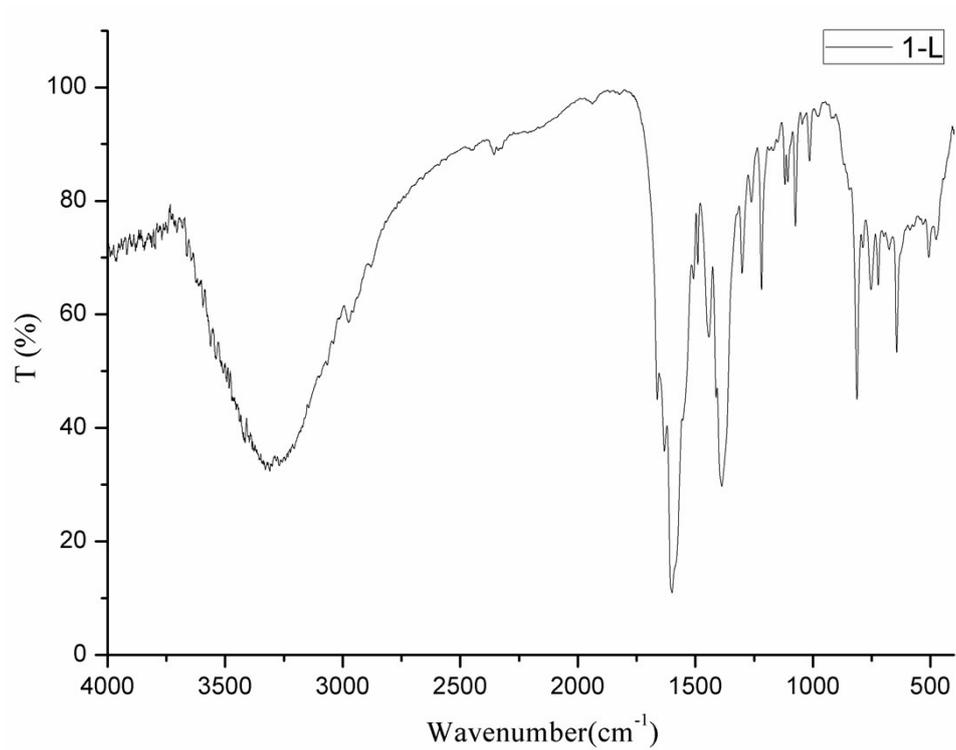


Figure S7. The IR spectra of **1-L**

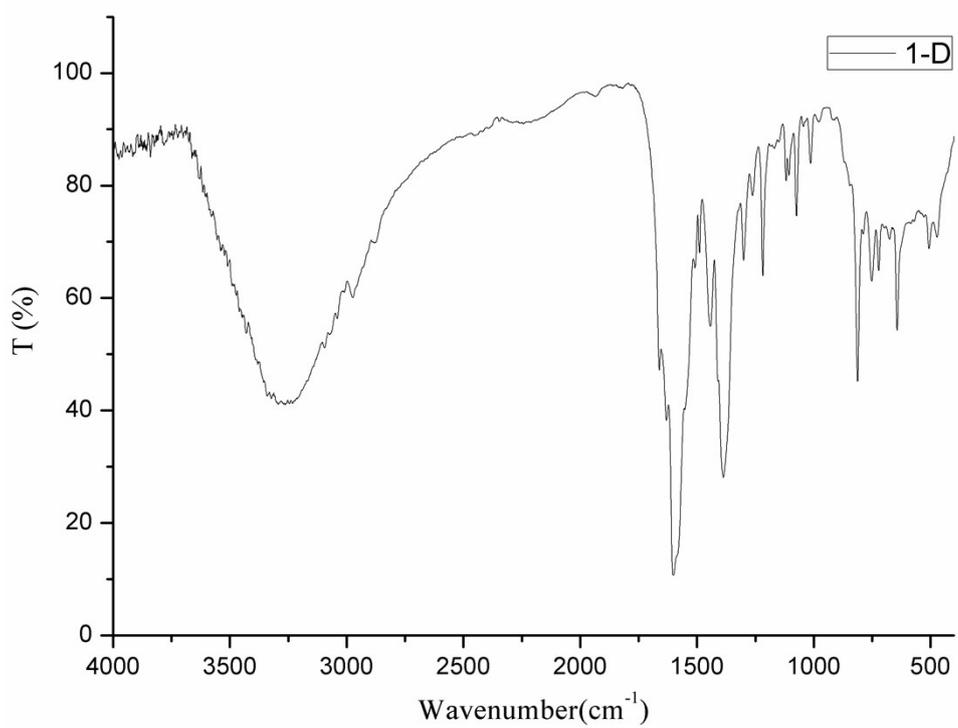


Figure S8. The IR spectra of **1-D**

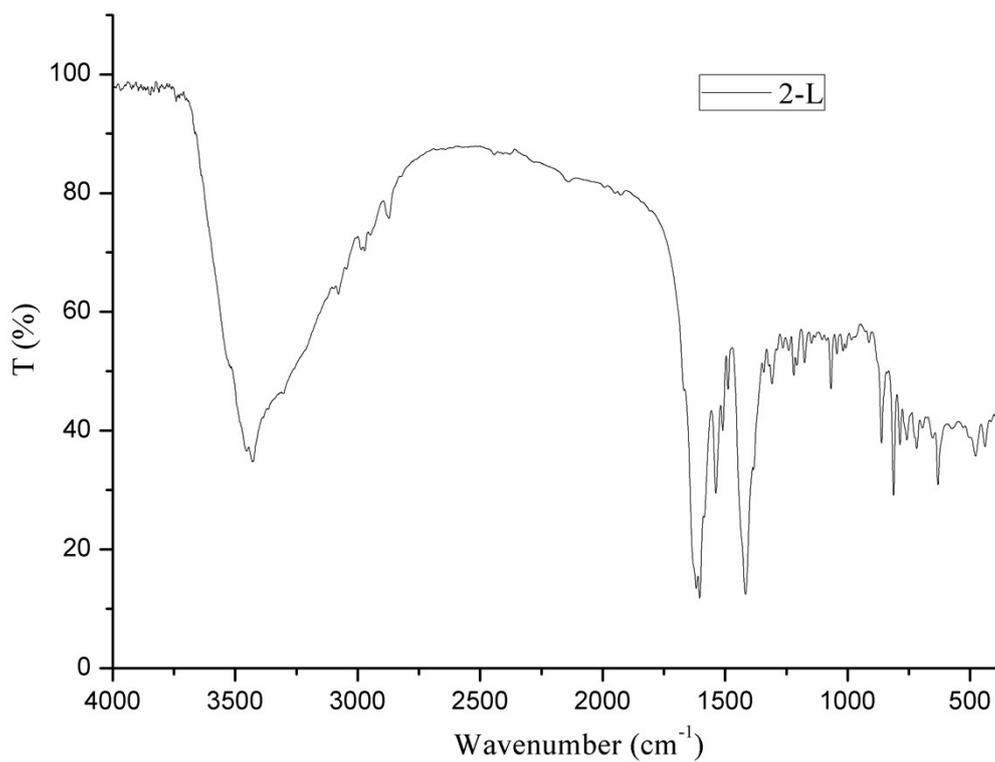


Figure S9. The IR spectra of **2-L**

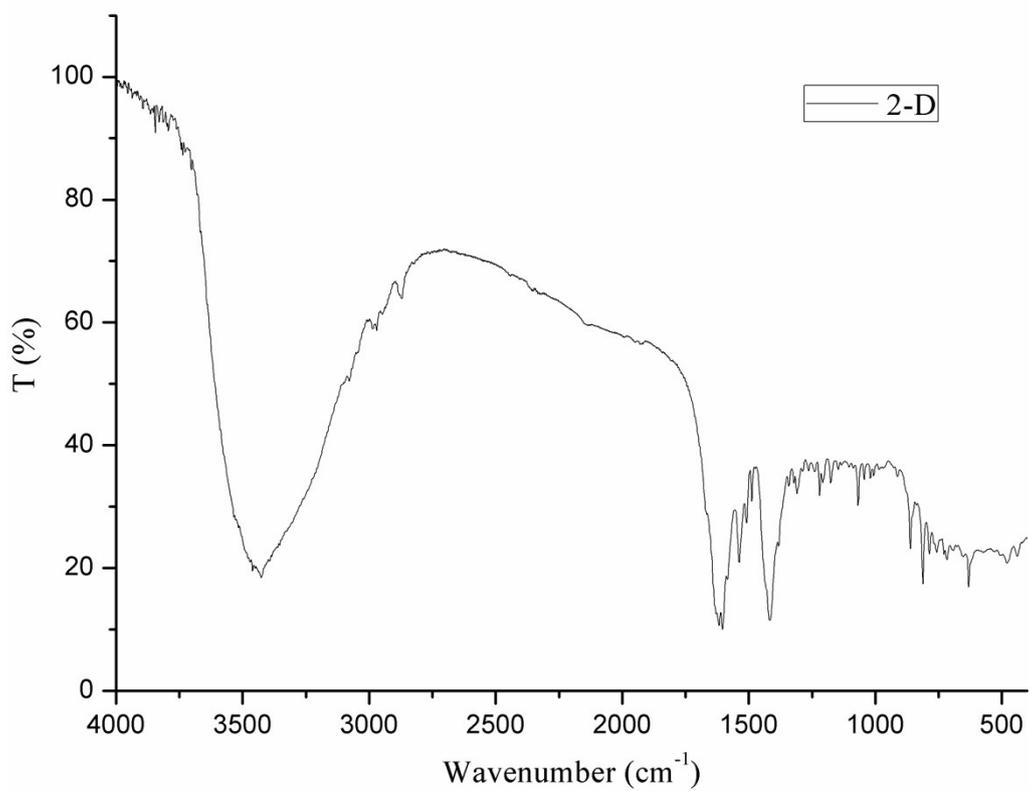


Figure S10. The IR spectra of **2-D**

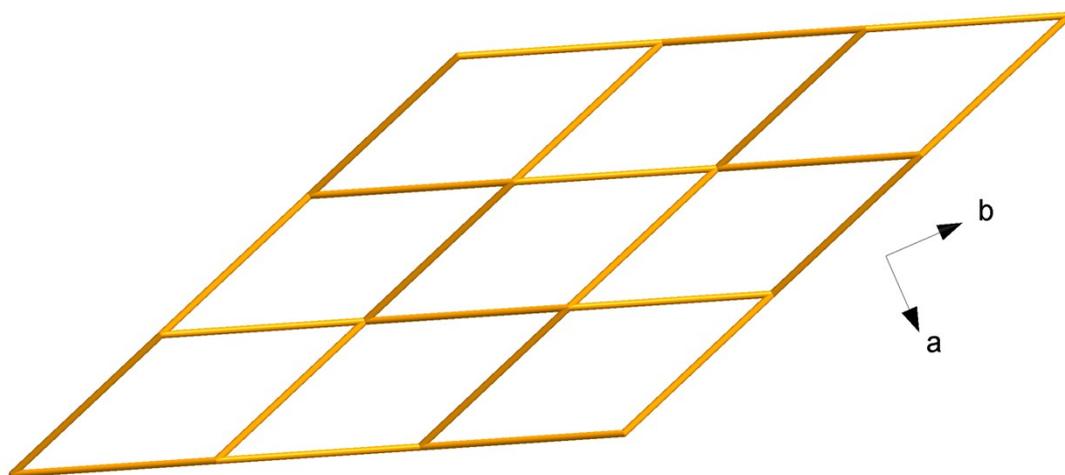


Figure S11. The *sql* net of Cu-(*S*)-PCA layer in compound **1-L**

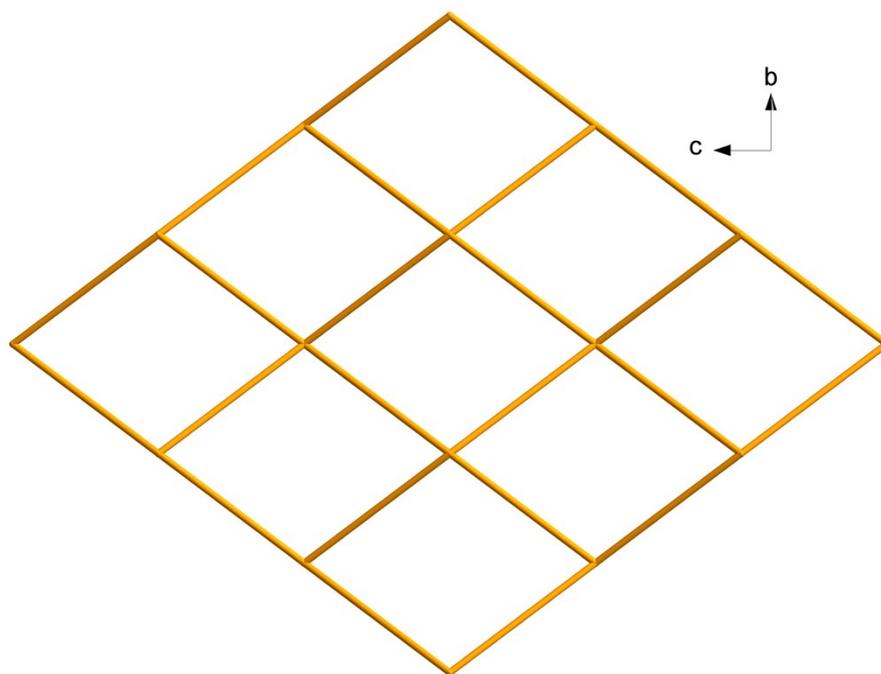


Figure S12. The *sql* net of Co-(*S*)-PCA layer in compound **2-L**

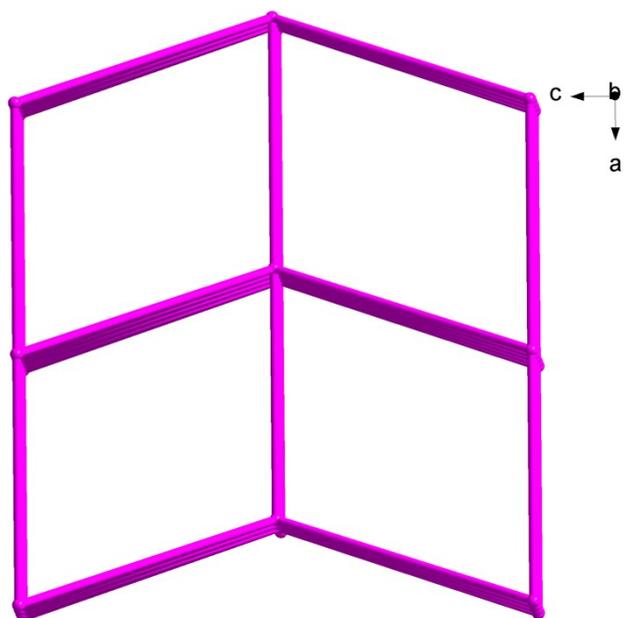


Figure S13. The 6-Connected *pcu* net of **2-L**

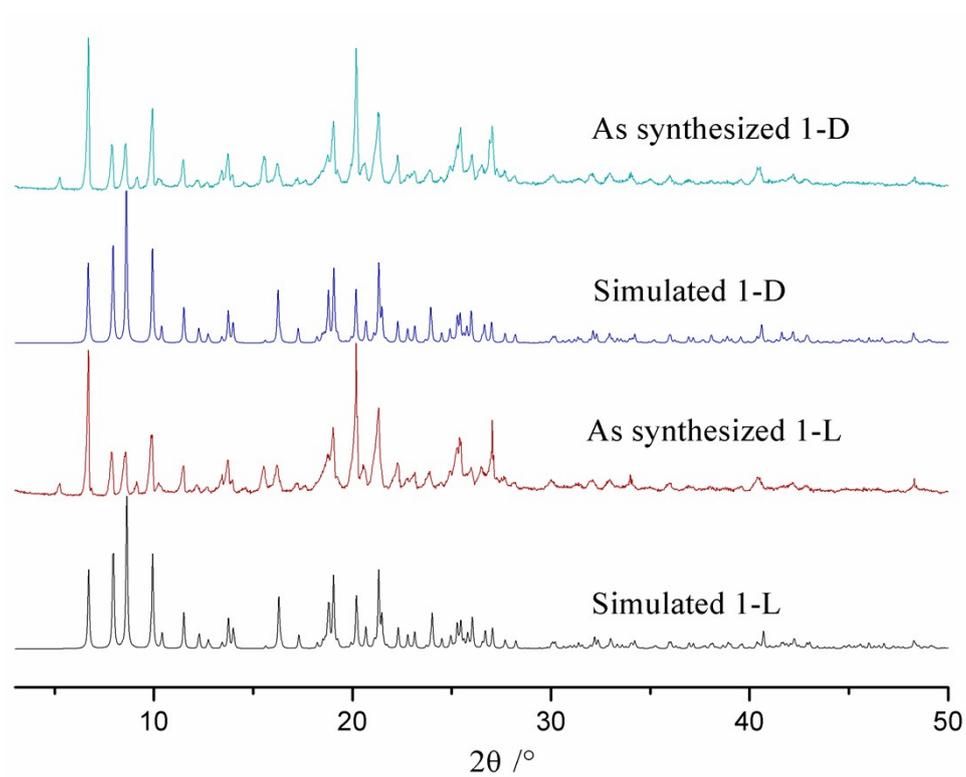


Figure S14. PXRD patterns of **1-D** and **1-L**

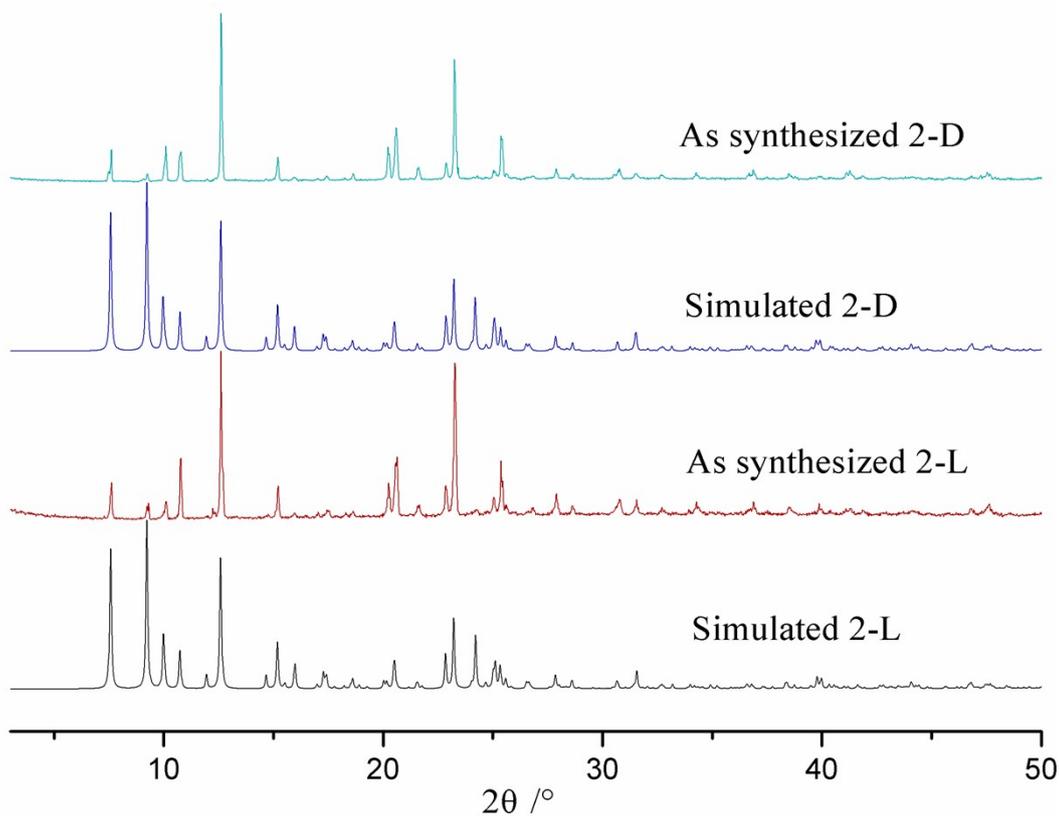


Figure S15. PXRD patterns of **2-D** and **2-L**

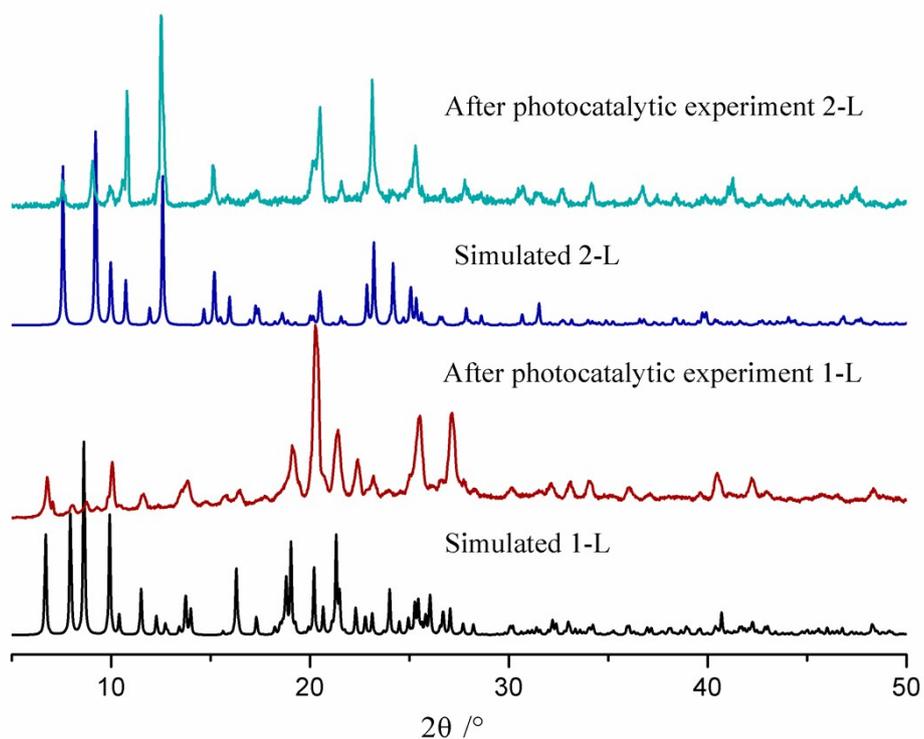


Figure S16. PXRD patterns of **1-L** and **2-L** after photocatalytic experiments

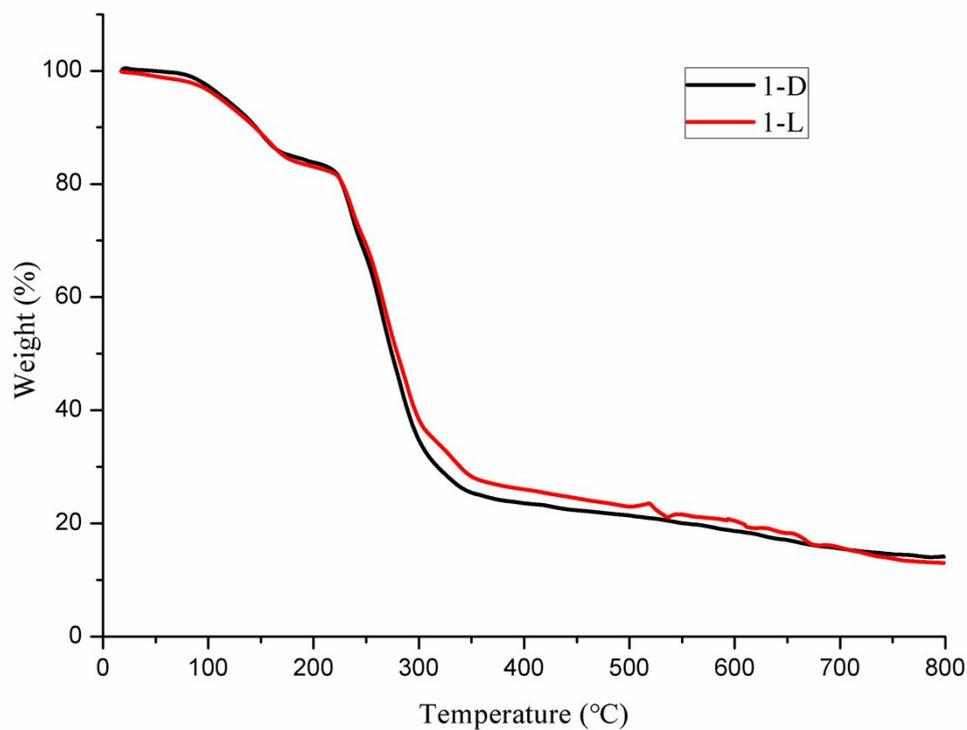


Figure S17. TGA curves of **1-L** and **1-D**

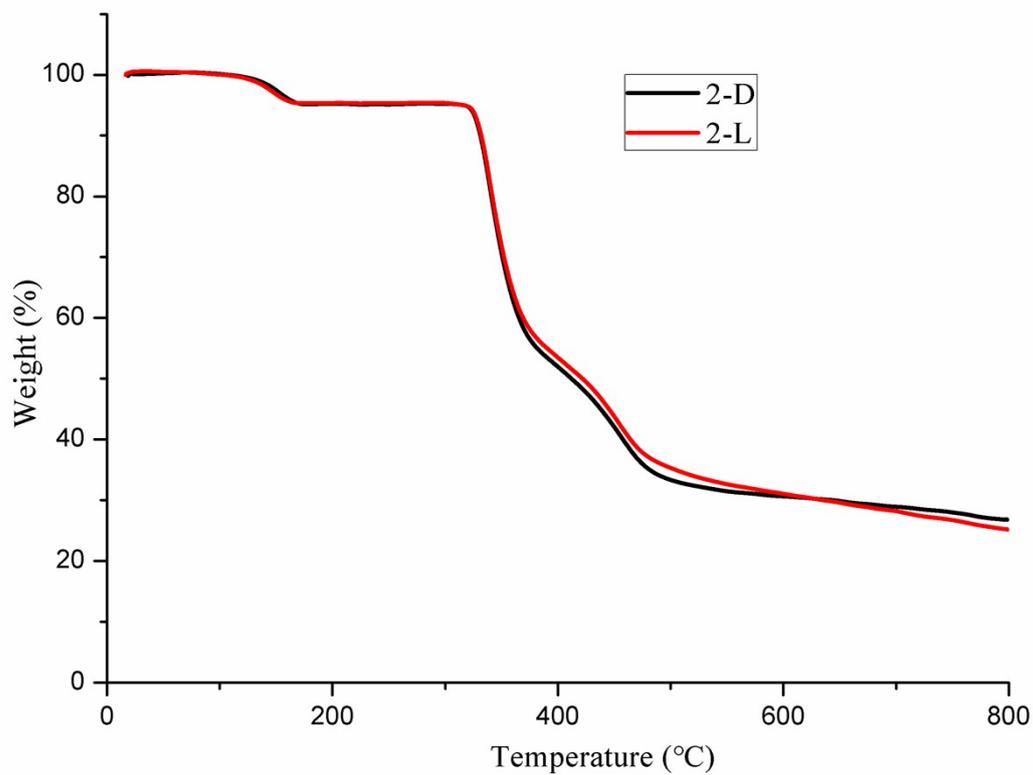


Figure S18. TGA curves of **2-L** and **2-D**

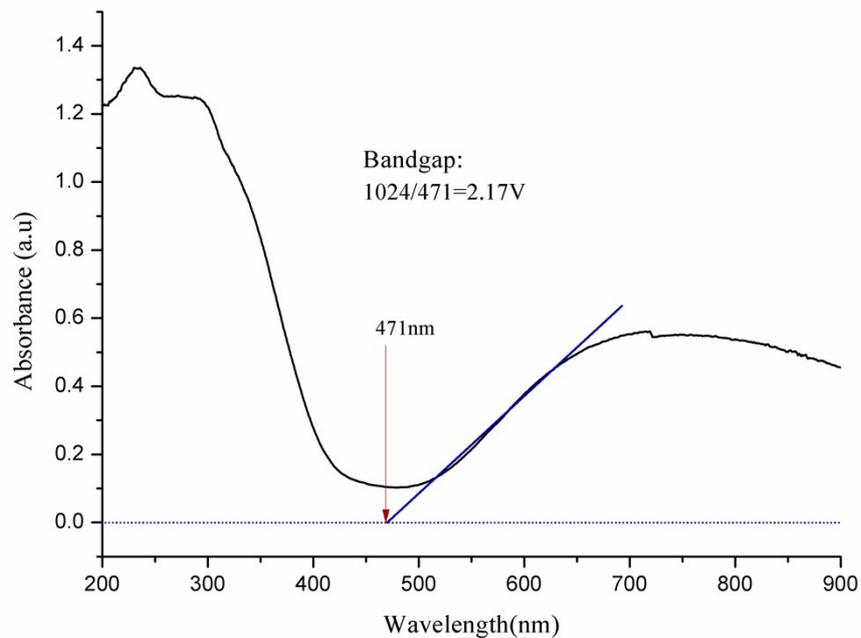


Figure S19. The bandgap of **1-L** from solid UV-vis absorption spectrum according to KM method

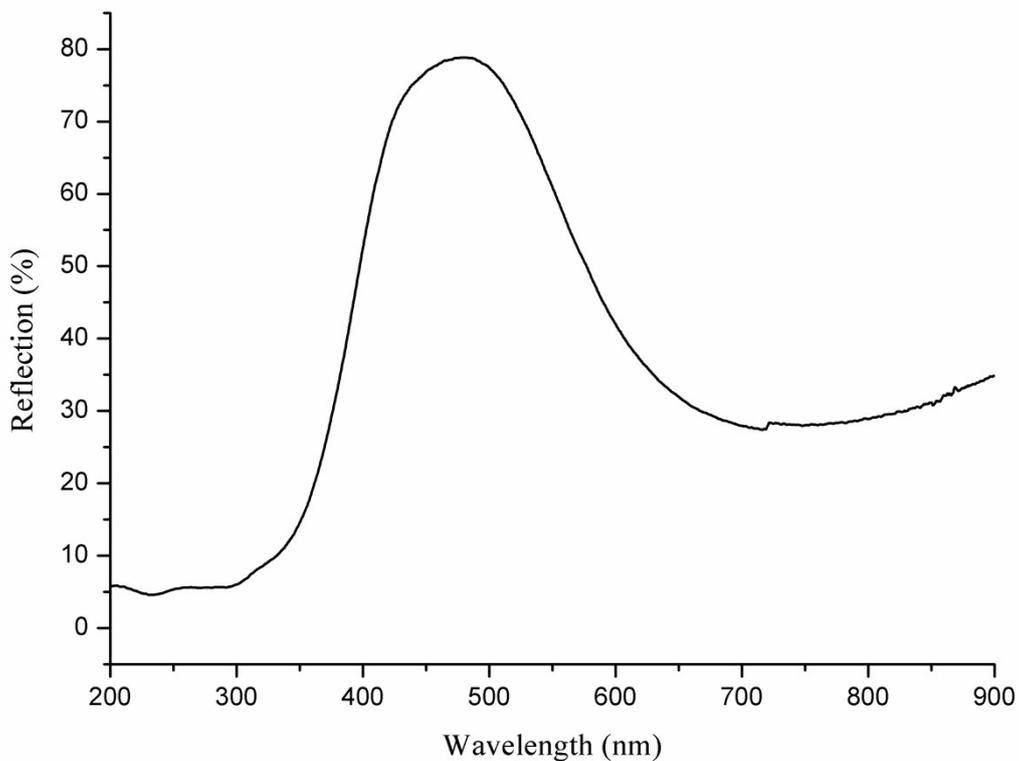


Figure S20. Solid UV-vis reflection spectrum of **1-L**

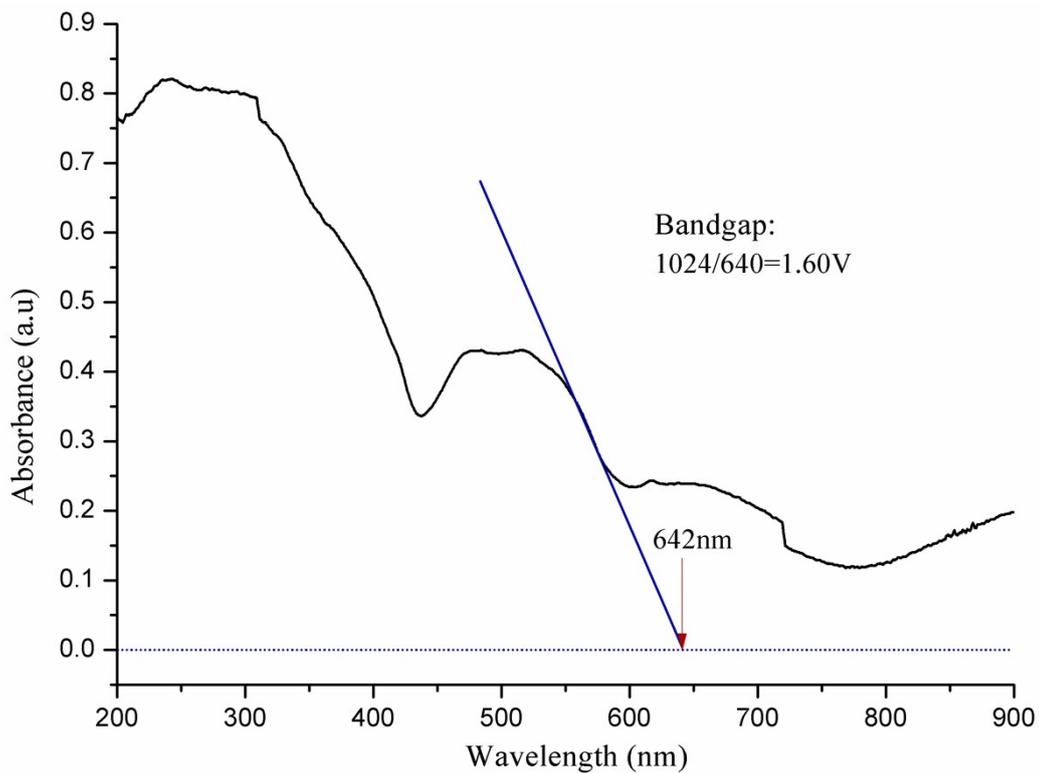


Figure S21. The bandgap of **2-L** from solid UV-vis absorption spectrum according to KM method

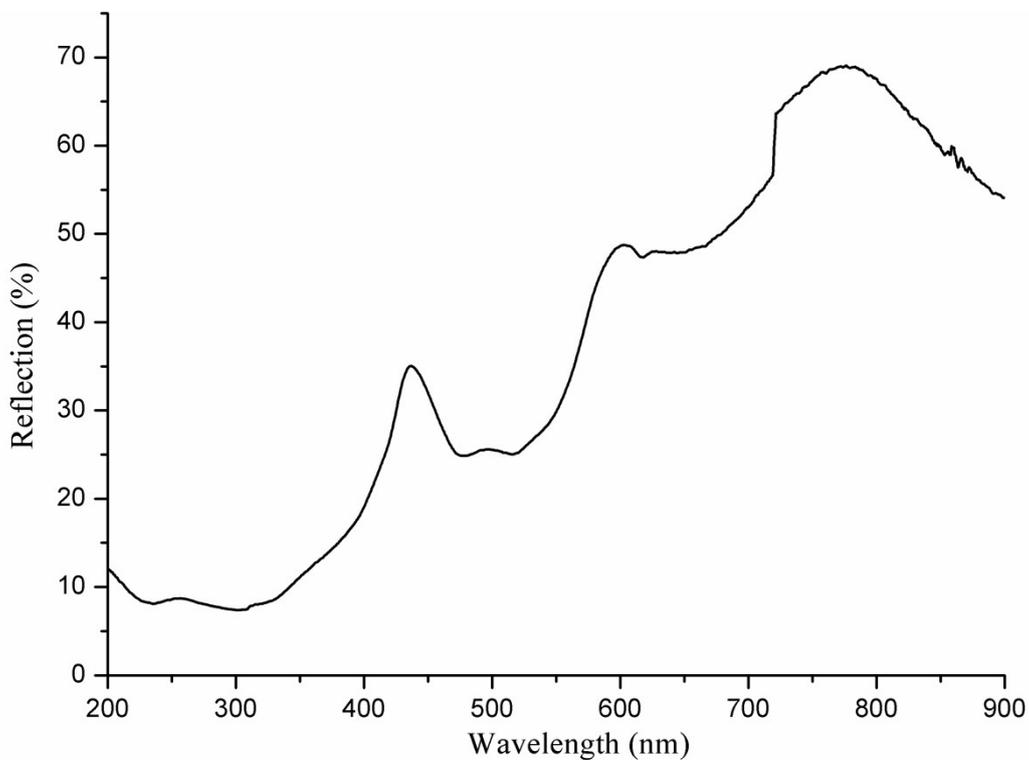


Figure S22. Solid UV-vis reflection spectrum of **2-L**

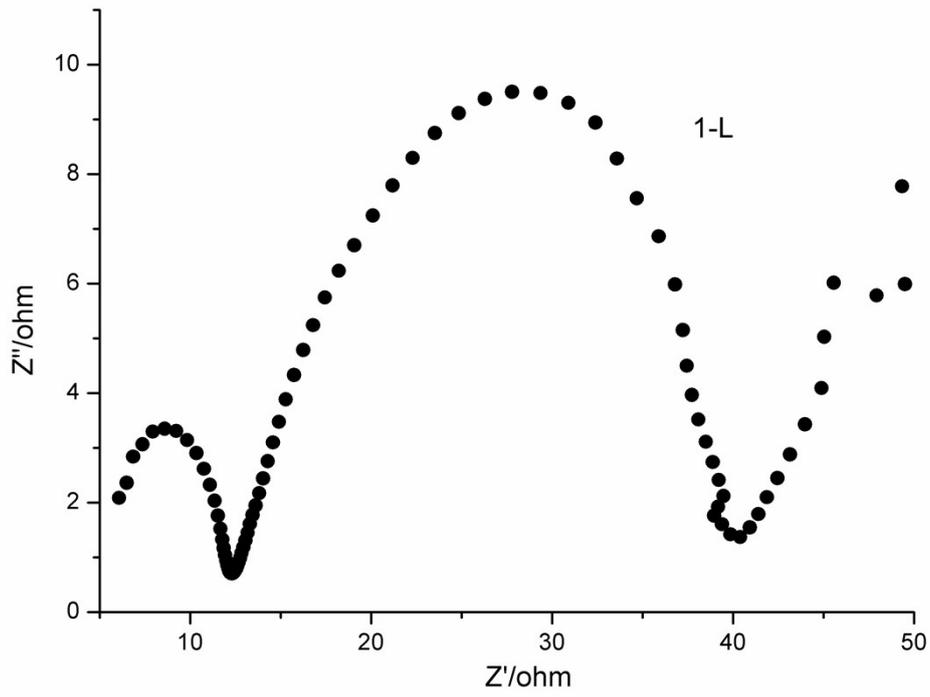


Figure S23. EIS plot of **1-L**

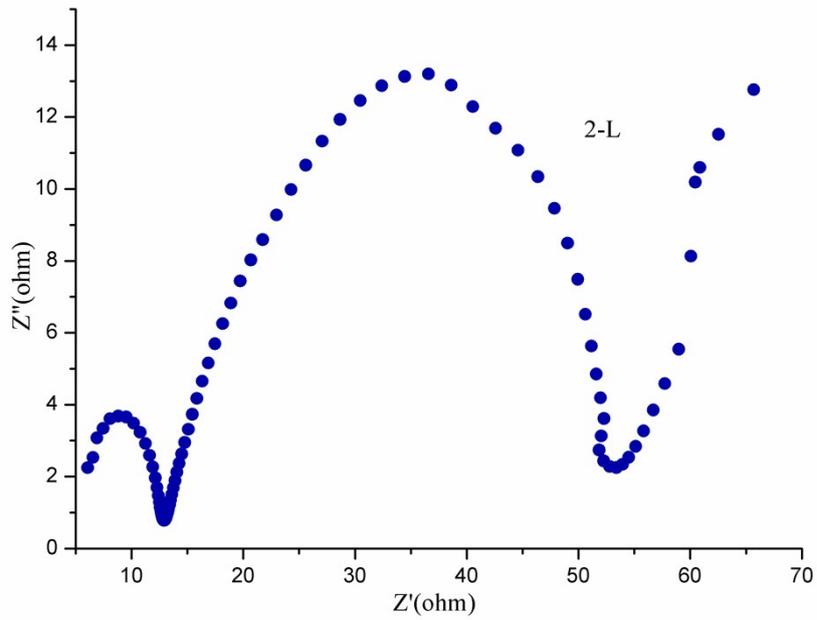


Figure S24. EIS plot of **2-L**

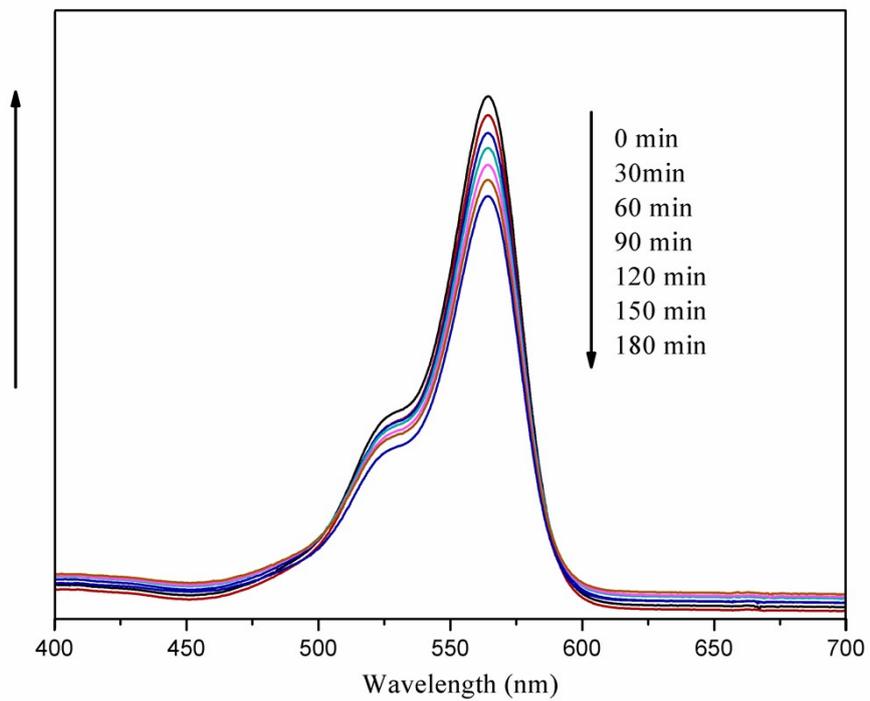


Figure S25. Photocatalytic degradation of RhB solution under UV irradiation without catalyst