

Mining the Cambridge Database for theoretical chemistry. Mi-LJC: A new set of Lennard–Jones–Coulomb atom–atom potentials for the computer simulation of organic condensed matter

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ESI

ELECTRONIC SUPPLEMENTARY INFORMATION

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S1. Handling the SE–LE comparison.

The lattice energy LE is the difference in calculated intermolecular energy between one mole of molecules at infinite separation and one mole in the crystal (LE is stabilizing i.e. < 0). The sublimation enthalpy SE is a measure of purely intermolecular cohesion energy in the crystal when it does not include contributions from the enthalpy of any concomitant intramolecular process. A very approximate consideration of global equipartition energies for an ensemble of non-interacting particles leads to the following relationships: for vaporization enthalpy, $\Delta H(\text{gas-liq}) = -U(\text{liq}) + RT$, for sublimation enthalpy, $\Delta H(\text{gas-cry}) = -U(\text{cry}) - 2RT$ or $\text{SE} = -\text{LE} - 2RT$ (in molar quantities). In brief, the RT terms account for differences in kinetic energies between the various phases due to a change in the number of translational and rotational degrees of freedom, plus the Legendre transform molar $PV = RT$ term that converts internal energies into enthalpies. The main assumptions are revised and commented in detail in the following.

Thermodynamic internal energy also has translational and rotational kinetic terms and vibrational terms:

$$U(\text{tot}, \text{gas}) = U(\text{interm}, \text{gas}) + E(\text{kin}, \text{trasl}) + E(\text{kin}, \text{rot}) + E(\text{vib}, \text{intram}) \quad (\text{S1})$$

$$U(\text{tot}, \text{liq}) = U(\text{interm}, \text{liq}) + E(\text{kin}, \text{trasl}) + E(\text{kin}, \text{rot}) + E(\text{vib}, \text{intram}) \quad (\text{S2})$$

where $E(\text{vib}, \text{intram})$ denotes the intramolecular vibrational energy. In the crystal, translational and rotational energies are replaced by potential and kinetic terms for translational and rotational librations (t-lib and r-lib):

$$U(\text{tot}, \text{cry}) = U(\text{interm}, \text{cry}) + E(\text{kin}, \text{t-lib}) + E(\text{pot}, \text{t-lib}) + E(\text{kin}, \text{r-lib}) + E(\text{pot}, \text{r-lib}) + E(\text{vib}, \text{intram}) \quad (\text{S3})$$

If $E(\text{vib}, \text{intram})$ is the same in all aggregation states (a far-fetched assumption except for rigid molecules), it can be neglected in energy differences. $U(\text{interm}, \text{gas})$ is assumed as zero for vapors of a solid at room temperature (semi-ideal gas assumption). In the harmonic equipartition regime, each energy term is worth $3/2RT$; then:

$$U(\text{tot}, \text{gas}) = 3RT \quad (\text{S4})$$

$$U(\text{tot}, \text{liq}) = U(\text{interm}, \text{liq}) + 3RT \quad (\text{S5})$$

$$U(\text{tot}, \text{cry}) = U(\text{interm}, \text{cry}) + 6RT \quad (\text{S6})$$

Since $H = U + PV$, for the phase transition functions one gets at constant pressure:

Vaporization enthalpy:

$$\begin{aligned} \Delta H(\text{gas} - \text{liq}) &= \Delta U(\text{gas} - \text{liq}) + P\Delta V = \Delta U(\text{gas} - \text{liq}) + PV_{\text{gas}} = \\ \Delta U(\text{gas} - \text{liq}) + RT &= U(\text{tot}, \text{gas}) - U(\text{tot}, \text{liq}) + RT = \\ 3RT - U(\text{interm}, \text{liq}) - 3RT + RT &= -U(\text{interm}, \text{liq}) + RT \end{aligned} \quad (\text{S7})$$

where the V_{liq} is negligible against V_{gas} . For sublimation enthalpy (SE):

$$\begin{aligned}
\Delta H(gas - cry) &= \Delta U(gas - cry) + P\Delta V = \Delta U(gas - cry) + RT = \\
U(tot,gas) - U(tot,cry) + RT &= 3RT - U(interm,cry) - 6RT + RT = \quad (S8) \\
&\quad - U(cry) - 2RT \\
\Rightarrow SE &= -LE - 2RT \quad (S9)
\end{aligned}$$

Another way of seeing the matter is to assign a constant heat capacity of 3R to the gas and of 6R to the solid, and to perform the standard thermodynamic integration:

$$\Delta U_s(T) = \Delta U_s(0) + \int_0^T C_v dT = \Delta U_s(0) - 3R \int_0^T dT = \Delta U_s(0) - 3RT \quad (S10)$$

$$\Delta H_s = \Delta U_s + \Delta(PV) = \Delta U_s + PV = \Delta U_s + RT = \Delta U_s(0) - 2RT \quad (S11)$$

whereby the calculated LE is the sublimation internal energy at 0 K. However, the assumption of constant C_v breaks one of the basic laws of thermodynamics, namely the convergence to zero of solid heat capacities at low temperature. In general, these corrections depend on a number of heavy approximations (ideal or semi-ideal behavior, vibrational behaviour independent on the chemical environment, constant heat capacity), almost none of which apply to real organic molecules. At room temperature, the correction may introduce more inaccuracy than adhesion to physical principles; no correction is better than a bad correction.

Table S1.

Vaporization energies (VE) ($\text{kJ}\cdot\text{mol}^{-1}$) and their dispersive–repulsive (E_{dr}) and Coulomb (E_{C}) contributions, from NPT-MD simulations of liquids, 432 molecule box, $T = 298\text{ K}$ and $P = 1\text{ bar}$ (weak coupling barostat).

Substance ^a	VE	E_{dr}	E_{C}	Functional
(a) pyridine	−39.1	−32.0	−7.2	CLP
	−32.6	−26.3	−6.3	LJC
(b) aminobenzene	−48.4	−44.1	−4.4	CLP
	−56.8	−28.7	−28.0	LJC
(c) benzaldehyde	−50.4	−36.9	−13.4	CLP
	−41.2	−34.6	−6.7	LJC
(d) methoxybenzene	−48.1	−41.6	−6.5	CLP
	−41.7	−36.6	−5.1	LJC
(e) ethanol	−35.6	−20.3	−15.3	CLP
	−46.7	−3.5	−43.1	LJC
(f) dichloroethane	−26.1	−22.3	−3.8	CLP
	−32.3	−25.2	−7.1	LJC
(g) n-pentane	−27.3	−27.1	−0.1	CLP
	−23.1	−23.1	0.0	LJC

^a See Figure 1 in the main text

Table S2.

Cell dimensions and densities ($\text{\AA}$, deg and $\text{g}\cdot\text{cm}^{-3}$) from NPT-MD simulations of molecular crystals (1)–(6) (Figure 1 in the main text) at $P = 1\text{ bar}$ and variable T . Starting reference parameters, in bold, CLP with weak-coupling barostat, LJC with weak-coupling barostat and LJC with Parrinello–Rahman (PR) barostat. Computational boxes from duplication of unit cells to dimensions of at least $35\text{ \AA}$.

Crystal	Space group	T / K	<i>a</i>	<i>b</i>	<i>c</i>	α	β	γ	Density	Functional
(1) n-hexane	$P\bar{1}$	90	4.131	4.696	8.539	83.4	87.3	75.2	0.900	MC (starting)
			4.206	4.527	8.440	83.5	87.2	74.3	0.931	CLP
			4.070	4.429	8.474	83.7	87.3	74.5	0.978	LJC
			4.255	4.467	8.546	83.7	87.8	75.7	0.915	LJC (PR)
(2) Naphthalene	$P2_1/a$	298	8.266	5.968	8.669	90.0	122.9	90.0	1.186	MC (starting)
			7.804	5.808	8.693	90.0	120.8	90.0	1.257	CLP
			8.510	6.013	8.761	90.0	123.6	90.0	1.140	LJC
			8.368	5.982	8.502	90.0	115.7	90.6	1.110	LJC (PR)
(3) Maleic anhydride	$P2_12_12_1$	130	7.032	11.020	5.332	90.0	90.0	90.0	1.576	MC (starting)
			6.587	11.753	5.158	90.0	90.0	90.0	1.631	CLP
			7.030	11.503	5.249	90.0	90.0	90.0	1.535	LJC
			7.091	11.519	5.281	91.3	90.3	89.7	1.510	LJC (PR)
(4) Methylurea	$P2_12_12_1$	298	8.477	6.981	6.923	90.0	90.0	90.0	1.201	MC (starting)
			8.466	7.012	6.912	90.0	90.0	90.0	1.199	CLP
			7.890	7.331	6.511	90.0	90.0	90.0	1.307	LJC
			8.006	7.366	6.555	90.2	89.3	90.5	1.273	LJC (PR)
(5) p-nitrophenol	$P2_1/c$	110	6.166	8.837	11.543	90.0	103.4	90.0	1.510	MC (starting)
			6.914	8.466	10.432	90.0	106.7	90.0	1.580	CLP
			6.031	8.857	11.704	90.0	102.9	90.0	1.516	LJC
			6.122	8.724	11.932	89.5	105.1	89.2	1.502	LJC (PR)
(6) Trimethyltrioxane CHCl ₃ solvate	$P2_1/c$	190	11.579	13.627	7.941	90.0	107.0	90.0	1.394	MC (starting)
			13.070	11.866	8.004	90.0	109.1	90.0	1.424	CLP
			12.279	12.012	8.005	90.0	107.9	90.0	1.487	LJC
			12.362	12.214	8.111	90.0	111.3	90.0	1.464	LJC (PR))

Table S3.

Lattice energies (LE, $\text{kJ}\cdot\text{mol}^{-1}$) and their dispersive–repulsive (E_{dr}) and Coulomb (E_{C}) contributions, from NPT-MD simulations of molecular crystals. See Table S2 above for explanation of symbols. .

Crystal ^a	Space group	T / K	LE	E_{dr}	E_{C}	Functional
(1) n-hexane	$P\bar{1}$	90	–55.8 –51.9 –46.4	–55.74 –52.00 –46.40	–0.03 0.08 0.00	CLP LJC LJC (PR)
(2) Naphthalene	$P2_1/a$	298	–74.1 –59.3 –56.1	–66.79 –54.13 –51.81	–7.35 –5.13 –4.27	CLP LJC LJC (PR)
(3) Maleic anhydride	$P2_12_12_1$	130	–84.6 –68.3 –65.9	–46.68 –40.84 –39.28	–37.95 –27.47 –26.65	CLP LJC LJC (PR)
(4) Methylurea	$P2_12_12_1$	298	–83.5 –95.0 –91.7	–46.94 –16.69 –17.28	–36.58 –78.31 –74.44	CLP LJC LJC (PR)
(5) p-nitrophenol	$P2_1/c$	110	–101.2 –95.0 –92.9	–71.50 –55.88 –54.81	–29.66 –39.10 –38.07	CLP LJC LJC (PR)
Crystal ^a	Space group	T / K	LE(solute)	LE(solvent)	LE(solvation)	Functional
(6) Trimethyltrioxane CHCl ₃ solvate ^b	$P2_1/c$	190	–50.6 –39.8 –39.8	–10.4 –8.9 –8.5	–56.5 –60.4 –58.7	CLP LJC LJC (PR)

^a See Figure 1 in the main text

^b Trimethyltrioxane part, chloroform part, and solvation part (left to right).

Table S4

Force field data for crystal structures and liquids in Molecular Dynamics. Atomic coordinates and charges, stretching functions for $E_s = 1/2K(R-R^0)^2$, bending functions for $E_b = 1/2K(\cos(t)-\cos(t^0))^2$, torsional functions, K , f and n for $E(t) = K(1-f(\cos nt))$.

#HEXANE01					
20					
1	-0.00257	0.22174	-3.19137	13	-0.8984
2	0.00280	-0.53915	-1.87570	13	-0.5518
3	0.00541	0.37514	-0.66189	13	-0.5632
4	-0.00541	-0.37514	0.66189	13	-0.5632
5	-0.00280	0.53916	1.87570	13	-0.5518
6	0.00257	-0.22174	3.19137	13	-0.8984
7	0.86830	0.85839	-3.24260	3	0.2957
8	0.01515	-0.47928	-4.01275	3	0.2957
9	-0.89498	0.82678	-3.25368	3	0.2957
10	0.87757	-1.17152	-1.84005	3	0.2816
11	-0.86989	-1.17406	-1.83437	3	0.2816
12	-0.86320	1.01541	-0.70640	3	0.2815
13	0.88423	1.00175	-0.69984	3	0.2815
14	0.86320	-1.01540	0.70640	3	0.2815
15	-0.88423	-1.00174	0.69984	3	0.2815
16	-0.87757	1.17152	1.84005	3	0.2817
17	0.86989	1.17407	1.83437	3	0.2816
18	0.87563	-0.85586	3.23697	3	0.2957
19	0.02357	0.47934	4.01263	3	0.2957
20	-0.88768	-0.82941	3.25943	3	0.2957
19 stretching					
1	2	1.520	2146.2	C- C	
1	7	1.080	3600.0	C- H	
1	8	1.080	3600.0	C- H	
1	9	1.080	3600.0	C- H	
2	3	1.520	2151.1	C- C	
2	10	1.080	3600.0	C- H	
2	11	1.080	3600.0	C- H	
3	4	1.522	2108.7	C- C	
3	12	1.080	3600.0	C- H	
3	13	1.080	3600.0	C- H	
4	5	1.520	2151.1	C- C	
4	14	1.080	3600.0	C- H	
4	15	1.080	3600.0	C- H	
5	6	1.520	2146.2	C- C	
5	16	1.080	3600.0	C- H	
5	17	1.080	3600.0	C- H	
6	18	1.080	3600.0	C- H	
6	19	1.080	3600.0	C- H	
6	20	1.080	3600.0	C- H	
36 bending					
1	2	3	113.00	534.6	C- C- C
1	2	10	109.50	642.5	C- C- H
1	2	11	109.50	642.5	C- C- H
2	1	7	109.50	642.5	C- C- H
2	1	8	109.50	642.5	C- C- H
2	1	9	109.50	642.5	C- C- H
2	3	4	113.00	534.6	C- C- C
2	3	12	109.50	642.5	C- C- H
2	3	13	109.50	642.5	C- C- H
3	2	10	109.50	642.5	C- C- H
3	2	11	109.50	642.5	C- C- H
3	4	5	113.00	534.6	C- C- C
3	4	14	109.50	642.5	C- C- H
3	4	15	109.50	642.5	C- C- H
4	3	12	109.50	642.5	C- C- H
4	3	13	109.50	642.5	C- C- H
4	5	6	113.00	534.6	C- C- C
4	5	16	109.50	642.5	C- C- H
4	5	17	109.50	642.5	C- C- H
5	4	14	109.50	642.5	C- C- H
5	4	15	109.50	642.5	C- C- H
5	6	18	109.50	642.5	C- C- H
5	6	19	109.50	642.5	C- C- H
5	6	20	109.50	642.5	C- C- H
6	5	16	109.50	642.5	C- C- H
6	5	17	109.50	642.5	C- C- H
7	1	8	109.50	470.0	H- C- H
7	1	9	109.50	470.0	H- C- H
8	1	9	109.50	470.0	H- C- H

10	2	11	108.00	470.0	H-	C-	H
12	3	13	108.00	470.0	H-	C-	H
14	4	15	108.00	470.0	H-	C-	H
16	5	17	108.00	470.0	H-	C-	H
18	6	19	109.50	470.0	H-	C-	H
18	6	20	109.50	470.0	H-	C-	H
19	6	20	109.50	470.0	H-	C-	H
5 torsion							
7	1	2	3	7.50	1.0	3.0	H- C- C- C
1	2	3	4	6.00	1.0	3.0	C- C- C- C
2	3	4	5	6.00	1.0	3.0	C- C- C- C
3	4	5	6	6.00	1.0	3.0	C- C- C- C
4	5	6	18	7.50	1.0	3.0	C- C- C- H

```
#HOZNOB methyltrioxane,chloroform solvate
21 trioxane
1 -0.12691 0.73456 -1.11331 23 -0.9197
2 -0.10989 -1.33712 -0.08012 23 -0.9284
3 -0.10486 0.59215 1.19698 23 -0.9194
4 0.34204 1.34148 0.07572 13 0.6932
5 0.34521 -0.59651 -1.21136 13 0.6942
6 0.33810 -0.74301 1.12703 13 0.6943
7 -0.22339 -1.50492 2.28066 13 -0.9286
8 -0.21827 2.72887 0.17243 13 -0.9296
9 -0.21817 -1.21549 -2.44823 13 -0.9281
10 1.42161 1.31869 0.05450 3 0.2388
11 1.42501 -0.57578 -1.20586 3 0.2404
12 1.41795 -0.72750 1.11479 3 0.2367
13 0.12787 -2.52554 2.24321 3 0.3061
14 0.09772 -1.04739 3.20477 3 0.3066
15 -1.30208 -1.49325 2.22896 3 0.3055
16 -1.29607 2.68644 0.11814 3 0.3057
17 0.07772 3.17192 1.11183 3 0.3070
18 0.15998 3.32760 -0.64301 3 0.3064
19 0.05224 -0.61694 -3.30560 3 0.3072
20 0.18003 -2.21259 -2.56524 3 0.3061
21 -1.29409 -1.26455 -2.36833 3 0.3056

5 chcl3
1 -0.07039 1.54371 -0.61151 42 -0.2877
2 -0.07161 -0.24633 1.64739 42 -0.2856
3 -0.06953 -1.29841 -1.03956 42 -0.2923
4 0.49245 0.00240 0.00856 13 0.6209
5 1.57230 0.00760 0.02729 3 0.2447

21 stretching
1 4 1.415 3755.2 O- C
1 5 1.416 3737.8 O- C
2 5 1.427 3492.9 O- C
2 6 1.418 3685.4 O- C
3 4 1.421 3625.7 O- C
3 6 1.408 3900.3 O- C
4 8 1.499 2577.6 C- C
4 10 1.080 3600.0 C- H
5 9 1.493 2702.5 C- C
5 11 1.080 3600.0 C- H
6 7 1.492 2728.8 C- C
6 12 1.080 3600.0 C- H
7 13 1.080 3600.0 C- H
7 14 1.080 3600.0 C- H
8 15 1.080 3600.0 C- H
8 16 1.080 3600.0 C- H
8 17 1.080 3600.0 C- H
8 18 1.080 3600.0 C- H
9 19 1.080 3600.0 C- H
9 20 1.080 3600.0 C- H
9 21 1.080 3600.0 C- H
4 nstr-v
1 4 1.754 3600.0 Cl- C
2 4 1.751 3600.0 Cl- C
3 4 1.763 3600.0 Cl- C
4 5 1.080 3600.0 C- H
39 bending
1 4 3 109.00 475.3 O- C- O
1 4 8 109.00 475.3 O- C- C
1 4 10 108.00 655.0 O- C- H
1 5 2 109.00 475.3 O- C- O
1 5 9 109.00 475.3 O- C- C
1 5 11 108.00 655.0 O- C- H
```

2	5	9	109.00	475.3	O- C- C
2	5	11	109.00	642.5	O- C- H
2	6	3	110.00	493.3	O- C- O
2	6	7	109.00	475.3	O- C- C
2	6	12	108.00	655.0	O- C- H
3	4	8	109.00	475.3	O- C- C
3	4	10	109.00	642.5	O- C- H
3	6	7	109.00	475.3	O- C- C
3	6	12	108.00	655.0	O- C- H
4	1	5	111.00	511.3	C- O- C
4	3	6	111.00	511.3	C- O- C
4	8	16	109.50	642.5	C- C- H
4	8	17	109.50	642.5	C- C- H
4	8	18	109.50	642.5	C- C- H
5	2	6	111.00	511.3	C- O- C
5	9	19	109.50	642.5	C- C- H
5	9	20	109.50	642.5	C- C- H
5	9	21	109.50	642.5	C- C- H
6	7	13	109.50	642.5	C- C- H
6	7	14	109.50	642.5	C- C- H
6	7	15	109.50	642.5	C- C- H
7	6	12	113.00	592.5	C- C- H
8	4	10	113.00	592.5	C- C- H
9	5	11	113.00	592.5	C- C- H
13	7	14	109.50	470.0	H- C- H
13	7	15	109.50	470.0	H- C- H
14	7	15	109.50	470.0	H- C- H
16	8	17	109.50	470.0	H- C- H
16	8	18	109.50	470.0	H- C- H
17	8	18	109.50	470.0	H- C- H
19	9	20	109.50	470.0	H- C- H
19	9	21	109.50	470.0	H- C- H
20	9	21	109.50	470.0	H- C- H
6 nbend-v					
1	4	2	111.00	500.0	C- C- C ?
1	4	3	110.00	500.0	C- C- C
1	4	5	109.00	642.5	C- C- H
2	4	3	110.00	500.0	C- C- C
2	4	5	108.00	655.0	C- C- H
3	4	5	109.00	642.5	C- C- H
9 torsion					
1	4	3	6	6.00	1.0 3.0 C- O- C- O
4	3	6	2	6.00	1.0 3.0 C- O- C- O
3	6	2	5	6.00	1.0 3.0 C- O- C- O
2	5	1	4	6.00	1.0 3.0 C- O- C- O
5	1	4	3	6.00	1.0 3.0 C- O- C- O
1	4	3	6	6.00	1.0 3.0 C- O- C- O
1	4	8	16	7.50	1.0 3.0 O- C- C- H
1	5	9	19	7.50	1.0 3.0 O- C- C- H
2	6	7	13	7.50	1.0 3.0 O- C- C- H

#MEUREA methylurea					
11					
1	0.00434	-0.05564	-0.47598	10	1.2133
2	-0.00657	0.02736	1.95453	13	-0.6380
3	-0.01806	0.64533	-1.61806	21	-0.9755
4	0.01035	0.64993	0.65834	21	-0.6511
5	0.00121	-1.30400	-0.47945	27	-1.2843
6	0.78152	-0.70899	2.01059	3	0.2871
7	0.14679	0.77753	2.71621	3	0.2881
8	-0.96075	-0.45405	2.11023	3	0.2857
9	0.09215	0.13648	-2.47186	7	0.4947
10	0.02762	1.64337	-1.57484	7	0.4930
11	0.02701	1.64259	0.53844	7	0.4870
10 stretching					
1	3	1.340	5845.9	C- N	
1	4	1.336	5949.1	C- N	
1	5	1.248	7484.0	C- O	
2	4	1.438	3532.4	C- N	
2	6	1.080	3600.0	C- H	
2	7	1.080	3600.0	C- H	
2	8	1.080	3600.0	C- H	
3	9	1.000	5300.0	N- H	
3	10	1.000	5300.0	N- H	
4	11	1.000	5300.0	N- H	
15 bending					
1	3	9	117.00	460.0	C- N- H
1	3	10	119.00	460.0	C- N- H
1	4	2	122.00	627.4	C- N- C
1	4	11	115.00	460.0	C- N- H

2	4	11	123.00	460.0	C- N- H
3	1	4	117.00	580.9	N- C- N
3	1	5	121.00	702.0	N- C- O
4	1	5	122.00	702.0	N- C- O
4	2	6	109.00	642.5	N- C- H
4	2	7	109.00	642.5	N- C- H
4	2	8	109.00	642.5	N- C- H
6	2	7	109.00	470.0	H- C- H
6	2	8	109.00	470.0	H- C- H
7	2	8	109.00	470.0	H- C- H
9	3	10	123.00	445.0	H- N- H
6 torsion					
5	1	3	9	45.00	-1.0 2.0 O- C- N- H amide
5	1	4	11	45.00	-1.0 2.0 O- C- N- H amide
1	3	4	5	100.00	-1.0 1.0 C- N- N- O
6	2	4	1	7.50	1.0 3.0 H- C- N- C methyl tors
3	1	9	10	100.00	-1.0 1.0 N- C- H- H
4	1	2	11	100.00	-1.0 1.0 N- C- C- H
#MLEICA01 maleic anhydride					
9					
1	0.02552	0.92681	0.00431	23	-0.7536
2	0.00300	0.10952	-1.11818	10	1.3867
3	0.00238	-1.29766	-0.67111	12	-0.2214
4	0.00257	-1.30367	0.66051	12	-0.2185
5	0.00028	0.10436	1.12150	10	1.3867
6	-0.01648	0.57192	-2.22305	27	-1.0952
7	-0.01560	0.56736	2.22536	27	-1.0936
8	0.00186	-2.16790	-1.31070	2	0.3046
9	0.00409	-2.17954	1.29236	2	0.3044
9 stretching					
1	2	1.389	4342.7	O- C	
1	5	1.388	4369.2	O- C	
2	3	1.476	3059.6	C- C	
2	6	1.198	8614.2	C- O	
3	4	1.332	6109.9	C- C	
3	8	1.080	3600.0	C- H	
4	5	1.482	2952.4	C- C	
4	9	1.080	3600.0	C- H	
5	7	1.197	8631.0	C- O	
13 bending					
1	2	3	108.00	457.2	O- C- C
1	2	6	121.00	691.6	O- C- O
1	5	4	108.00	457.2	O- C- C
1	5	7	121.00	691.6	O- C- O
2	1	5	108.00	457.2	C- O- C
2	3	4	108.00	499.5	C- C- C
2	3	8	126.00	430.0	C- C- H
3	2	6	130.00	853.9	C- C- O
3	4	5	108.00	499.5	C- C- C
3	4	9	126.00	430.0	C- C- H
4	3	8	126.00	430.0	C- C- H
4	5	7	131.00	871.9	C- C- O
5	4	9	126.00	430.0	C- C- H
9 torsion					
5	1	2	3	50.00	-1.0 1.0 C- O- C- C
2	1	5	4	50.00	-1.0 1.0 C- O- C- C
1	2	3	4	50.00	-1.0 1.0 O- C- C- C
2	1	3	6	100.00	-1.0 1.0 C- O- C- O
2	3	4	5	50.00	-1.0 1.0 C- C- C- C
3	2	4	8	100.00	-1.0 1.0 C- C- C- H
3	4	5	1	50.00	-1.0 1.0 C- C- C- O
4	3	5	9	100.00	-1.0 1.0 C- C- C- H
5	1	4	7	100.00	-1.0 1.0 C- O- C- O
#NAPHTA13 naphthalene					
18					
1	0.00004	0.70617	-0.00009	14	0.0511
2	-0.00023	1.39916	1.24243	12	-0.3093
3	0.00020	1.39946	-1.24251	12	-0.3095
4	0.00013	0.70047	2.42124	12	-0.2917
5	-0.00006	0.70021	-2.42146	12	-0.2917
6	-0.00004	-0.70617	0.00009	14	0.0511
7	0.00006	-0.70021	2.42146	12	-0.2917
8	-0.00013	-0.70047	-2.42124	12	-0.2917
9	0.00023	-1.39916	-1.24243	12	-0.3093
10	-0.00020	-1.39946	1.24251	12	-0.3095
11	-0.00075	2.47905	1.25665	2	0.2886
12	0.00049	2.47939	-1.25671	2	0.2886
13	0.00049	1.23520	3.35956	2	0.2868
14	-0.00022	1.23460	-3.35999	2	0.2869

15	0.00022	-1.23460	3.35999	2	0.2869
16	-0.00049	-1.23520	-3.35956	2	0.2868
17	0.00075	-2.47905	-1.25665	2	0.2886
18	-0.00049	-2.47939	1.25671	2	0.2886
19	stretching				
1	2	1.423	4192.1	C- C	
1	3	1.423	4191.0	C- C	
1	6	1.412	4410.2	C- C	
2	4	1.370	5295.3	C- C	
2	11	1.080	3600.0	C- H	
3	5	1.371	5286.6	C- C	
3	12	1.080	3600.0	C- H	
4	7	1.401	4656.0	C- C	
4	13	1.080	3600.0	C- H	
5	8	1.401	4656.0	C- C	
5	14	1.080	3600.0	C- H	
6	9	1.423	4192.1	C- C	
6	10	1.423	4191.0	C- C	
7	10	1.371	5286.6	C- C	
7	15	1.080	3600.0	C- H	
8	9	1.370	5295.3	C- C	
8	16	1.080	3600.0	C- H	
9	17	1.080	3600.0	C- H	
10	18	1.080	3600.0	C- H	
30	bending				
1	2	4	120.00	583.8	C- C- C
1	2	11	120.00	505.0	C- C- H
1	3	5	120.00	583.8	C- C- C
1	3	12	120.00	505.0	C- C- H
1	6	9	119.00	576.7	C- C- C
1	6	10	119.00	576.7	C- C- C
2	1	3	122.00	597.8	C- C- C
2	1	6	119.00	576.7	C- C- C
2	4	7	121.00	590.8	C- C- C
2	4	13	120.00	505.0	C- C- H
3	1	6	119.00	576.7	C- C- C
3	5	8	121.00	590.8	C- C- C
3	5	14	120.00	505.0	C- C- H
4	2	11	120.00	505.0	C- C- H
4	7	10	121.00	590.8	C- C- C
4	7	15	120.00	505.0	C- C- H
5	3	12	120.00	505.0	C- C- H
5	8	9	121.00	590.8	C- C- C
5	8	16	120.00	505.0	C- C- H
6	9	8	120.00	583.8	C- C- C
6	9	17	120.00	505.0	C- C- H
6	10	7	120.00	583.8	C- C- C
6	10	18	120.00	505.0	C- C- H
7	4	13	120.00	505.0	C- C- H
7	10	18	120.00	505.0	C- C- H
8	5	14	120.00	505.0	C- C- H
8	9	17	120.00	505.0	C- C- H
9	6	10	122.00	597.8	C- C- C
9	8	16	120.00	505.0	C- C- H
10	7	15	120.00	505.0	C- C- H
21	torsion				
6	1	2	4	50.00	-1.0 1.0 C- C- C- C
6	1	3	5	50.00	-1.0 1.0 C- C- C- C
2	1	6	10	50.00	-1.0 1.0 C- C- C- C
1	2	3	6	100.00	-1.0 1.0 C- C- C- C
1	2	4	7	50.00	-1.0 1.0 C- C- C- C
2	1	4	11	100.00	-1.0 1.0 C- C- C- H
1	3	5	8	50.00	-1.0 1.0 C- C- C- C
3	1	5	12	100.00	-1.0 1.0 C- C- C- H
2	4	7	10	50.00	-1.0 1.0 C- C- C- C
4	2	7	13	100.00	-1.0 1.0 C- C- C- H
3	5	8	9	50.00	-1.0 1.0 C- C- C- C
5	3	8	14	100.00	-1.0 1.0 C- C- C- H
1	6	9	8	50.00	-1.0 1.0 C- C- C- C
1	6	10	7	50.00	-1.0 1.0 C- C- C- C
6	1	9	10	100.00	-1.0 1.0 C- C- C- C
4	7	10	6	50.00	-1.0 1.0 C- C- C- C
7	4	10	15	100.00	-1.0 1.0 C- C- C- H
5	8	9	6	50.00	-1.0 1.0 C- C- C- C
8	5	9	16	100.00	-1.0 1.0 C- C- C- H
9	6	8	17	100.00	-1.0 1.0 C- C- C- H
10	6	7	18	100.00	-1.0 1.0 C- C- C- H

#NITPOL03 p-nitropheno1

15									
1	0.02946	-0.06511	-3.48537	29	-1.2281				
2	0.00996	-1.09037	2.68977	30	-0.8337				
3	0.03677	1.08737	2.70110	30	-0.8620				
4	0.00634	-0.00689	2.09485	20	1.1626				
5	0.00138	0.00264	-2.12840	12	0.7539				
6	-0.02829	1.22814	-1.42918	12	-0.3430				
7	-0.04283	1.23164	-0.03756	12	-0.2172				
8	-0.01954	0.00631	0.64288	12	0.2097				
9	-0.01716	-1.22231	-0.03470	12	-0.2201				
10	-0.01655	-1.21766	-1.42576	12	-0.3475				
11	-0.03968	2.16123	-1.97287	2	0.2975				
12	-0.07158	2.16273	0.50893	2	0.2963				
13	-0.01581	-2.15321	0.51286	2	0.2964				
14	-0.02986	-2.15165	-1.96790	2	0.2980				
15	0.32503	0.81508	-3.85675	5	0.7370				
15	stretching								
1	5	1.359	5008.5	O- C					
1	15	1.000	4250.0	O- H					
2	4	1.236	6000.0	O- N					
3	4	1.251	6000.0	O- N					
4	8	1.452	3196.5	N- C					
5	6	1.411	4433.3	C- C					
5	10	1.408	4496.7	C- C					
6	7	1.392	4844.9	C- C					
6	11	1.080	3600.0	C- H					
7	8	1.402	4632.9	C- C					
7	12	1.080	3600.0	C- H					
8	9	1.403	4605.5	C- C					
9	10	1.391	4858.2	C- C					
9	13	1.080	3600.0	C- H					
10	14	1.080	3600.0	C- H					
22	bending								
1	5	6	123.00	727.7	O- C- C				
1	5	10	117.00	619.5	O- C- C				
2	4	3	122.00	700.0	O- N- O				
2	4	8	119.00	701.0	O- N- C				
3	4	8	118.00	701.0	O- N- C				
4	8	7	120.00	608.8	N- C- C				
4	8	9	118.00	590.2	N- C- C				
5	1	15	109.00	450.0	C- O- H				
5	6	7	120.00	583.8	C- C- C				
5	6	11	120.00	505.0	C- C- H				
5	10	9	120.00	583.8	C- C- C				
5	10	14	120.00	505.0	C- C- H				
6	5	10	120.00	583.8	C- C- C				
6	7	8	119.00	576.7	C- C- C				
6	7	12	121.00	492.5	C- C- H				
7	6	11	120.00	505.0	C- C- H				
7	8	9	122.00	597.8	C- C- C				
8	7	12	121.00	492.5	C- C- H				
8	9	10	119.00	576.7	C- C- C				
8	9	13	121.00	492.5	C- C- H				
9	10	14	120.00	505.0	C- C- H				
10	9	13	121.00	492.5	C- C- H				
15	torsion								
15	1	5	6	7.00	-1.0	2.0	H- O- C- C	phenol OH	
2	4	8	9	10.00	-1.0	2.0	O- N- C- C	nitro group	
4	2	3	8	100.00	-1.0	1.0	N- O- O- C		
10	5	6	7	50.00	-1.0	1.0	O- C- C- C		
6	5	10	9	50.00	-1.0	1.0	O- C- C- C		
5	1	6	10	100.00	-1.0	1.0	C- O- C- C		
5	6	7	8	50.00	-1.0	1.0	C- C- C- C		
6	5	7	11	100.00	-1.0	1.0	C- C- C- H		
6	7	8	9	50.00	-1.0	1.0	C- C- C- N		
7	6	8	12	100.00	-1.0	1.0	C- C- C- H		
7	8	9	10	50.00	-1.0	1.0	N- C- C- C		
8	4	7	9	100.00	-1.0	1.0	C- N- C- C		
8	9	10	5	50.00	-1.0	1.0	C- C- C- C		
9	8	10	13	100.00	-1.0	1.0	C- C- C- H		
10	5	9	14	100.00	-1.0	1.0	C- C- C- H		

benzaldehyde liquid									
14	1	0.00000	0.23168	0.48549	12	0.0197			
	2	0.00000	1.29040	-0.41518	12	-0.2284			
	3	0.00000	1.03976	-1.78240	12	-0.2915			
	4	0.00000	-0.26961	-2.24894	12	-0.2261			
	5	-0.00000	-1.32833	-1.34827	12	-0.2912			
	6	-0.00000	-1.07768	0.01894	12	-0.2285			
	7	0.00000	0.49855	1.94123	10	0.6560			
	8	0.00000	2.30775	-0.05269	2	0.2875			
	9	0.00000	1.86237	-2.48220	2	0.2878			
	10	0.00000	-0.46435	-3.31124	2	0.2874			
	11	-0.00000	-2.34568	-1.71077	2	0.2878			
	12	0.00000	-1.90029	0.71874	2	0.2876			
	13	0.00000	-0.35124	2.80261	27	-1.1086			
	14	0.00000	1.53074	2.25901	3	0.2606			
14	stretching								
	2	1.390	4880.7	C- C					
	6	1.390	4880.9	C- C					
	7	1.480	2985.6	C- C					
	3	1.390	4880.8	C- C					
	8	1.080	3600.0	C- H					
	4	1.390	4880.7	C- C					
	9	1.080	3600.0	C- H					
	5	1.390	4880.9	C- C					
	10	1.080	3600.0	C- H					
	6	1.390	4880.8	C- C					
	11	1.080	3600.0	C- H					
	12	1.080	3600.0	C- H					
	13	1.210	8342.8	C- O					
	14	1.080	3600.0	C- H					
21	bending								
	2	3	120.00	583.8	C- C- C				
	2	8	120.00	505.0	C- C- H				
	6	5	120.00	583.8	C- C- C				
	6	12	120.00	505.0	C- C- H				
	7	13	125.00	763.8	C- C- O				
	7	14	118.00	530.0	C- C- H				
	1	6	120.00	583.8	C- C- C				
	1	7	120.00	583.8	C- C- C				
	3	9	120.00	505.0	C- C- H				
	2	8	120.00	505.0	C- C- H				
	4	5	120.00	583.8	C- C- C				
	4	10	120.00	505.0	C- C- H				
	3	9	120.00	505.0	C- C- H				
	4	6	120.00	583.8	C- C- C				
	5	11	120.00	505.0	C- C- H				
	4	10	120.00	505.0	C- C- H				
	5	12	120.00	505.0	C- C- H				
	6	7	120.00	583.8	C- C- C				
	5	11	120.00	505.0	C- C- H				
13	7	14	117.00	542.5	O- C- H				
14	torsion								
	1	2	3	50.00	-1.0	1.0	C- C- C- C		
	1	6	5	50.00	-1.0	1.0	C- C- C- C		
	1	7	13	16.00	-1.0	2.0	C- C- C- O ald.		
	2	6	7	100.00	-1.0	1.0	C- C- C- C		
	1	3	4	50.00	-1.0	1.0	C- C- C- C		
	2	3	8	100.00	-1.0	1.0	C- C- C- H		
	2	4	5	50.00	-1.0	1.0	C- C- C- C		
	3	4	9	100.00	-1.0	1.0	C- C- C- H		
	3	5	6	50.00	-1.0	1.0	C- C- C- C		
	4	3	10	100.00	-1.0	1.0	C- C- C- H		
	4	5	1	50.00	-1.0	1.0	C- C- C- C		
	5	6	11	100.00	-1.0	1.0	C- C- C- H		
	6	1	12	100.00	-1.0	1.0	C- C- C- H		
	7	1	13	100.00	-1.0	1.0	C- C- O- H		
aniline liquid									
14	1	0.00000	-0.00000	0.92675	12	0.3359			
	2	0.00000	1.20378	0.23175	12	-0.3800			
	3	0.00000	1.20378	-1.15825	12	-0.2824			
	4	0.00000	-0.00001	-1.85325	12	-0.3584			
	5	-0.00000	-1.20377	-1.15825	12	-0.2824			
	6	0.00000	-1.20377	0.23175	12	-0.3800			
	7	0.00000	0.00000	2.32175	17	-1.0719			
	8	0.00000	2.13909	0.77175	2	0.2926			
	9	0.00000	2.13909	-1.69825	2	0.2876			
	10	0.00000	-0.00001	-2.93325	2	0.2877			

11	-0.00000	-2.13908	-1.69825	2	0.2876
12	0.00000	-2.13908	0.77175	2	0.2926
13	0.00000	0.86602	2.82175	8	0.4856
14	0.00000	-0.86603	2.82175	8	0.4856
14	stretching				
1	2	1.390	4880.7	C- C	
1	6	1.390	4880.9	C- C	
1	7	1.395	4550.6	C- N	
2	3	1.390	4880.8	C- C	
2	8	1.080	3600.0	C- H	
3	4	1.390	4880.7	C- C	
3	9	1.080	3600.0	C- H	
4	5	1.390	4880.9	C- C	
4	10	1.080	3600.0	C- H	
5	6	1.390	4880.8	C- C	
5	11	1.080	3600.0	C- H	
6	12	1.080	3600.0	C- H	
7	13	1.000	5300.0	N- H	
7	14	1.000	5300.0	N- H	
21	bending				
1	2	3	120.00	583.8	C- C- C
1	2	8	120.00	505.0	C- C- H
1	6	5	120.00	583.8	C- C- C
1	6	12	120.00	505.0	C- C- H
1	7	13	120.00	460.0	C- N- H
1	7	14	120.00	460.0	C- N- H
2	1	6	120.00	583.8	C- C- C
2	1	7	120.00	608.8	C- C- N
2	3	4	120.00	583.8	C- C- C
2	3	9	120.00	505.0	C- C- H
3	2	8	120.00	505.0	C- C- H
3	4	5	120.00	583.8	C- C- C
3	4	10	120.00	505.0	C- C- H
4	3	9	120.00	505.0	C- C- H
4	5	6	120.00	583.8	C- C- C
4	5	11	120.00	505.0	C- C- H
5	4	10	120.00	505.0	C- C- H
5	6	12	120.00	505.0	C- C- H
6	1	7	120.00	608.8	C- C- N
6	5	11	120.00	505.0	C- C- H
13	7	14	120.00	445.0	H- N- H
14	torsion				
6	1	2	3	50.00	-1.0 1.0 C- C- C- C
2	1	6	5	50.00	-1.0 1.0 C- C- C- C
2	1	7	13	20.00	-1.0 2.0 C- C- N- H amine
1	2	6	7	100.00	-1.0 1.0 C- C- C- N
1	2	3	4	50.00	-1.0 1.0 C- C- C- C
2	1	3	8	100.00	-1.0 1.0 C- C- C- H
2	3	4	5	50.00	-1.0 1.0 C- C- C- C
3	2	4	9	100.00	-1.0 1.0 C- C- C- H
3	4	5	6	50.00	-1.0 1.0 C- C- C- C
4	3	5	10	100.00	-1.0 1.0 C- C- C- H
4	5	6	1	50.00	-1.0 1.0 C- C- C- C
5	4	6	11	100.00	-1.0 1.0 C- C- C- H
6	1	5	12	100.00	-1.0 1.0 C- C- C- H
7	1	13	14	100.00	-1.0 1.0 N- C- H- H
methoxybenzene liquid					
16					
1	-0.00000	0.31045	0.46528	12	0.5237
2	0.00000	1.30648	-0.50428	12	-0.3534
3	0.00000	0.96483	-1.85163	12	-0.2818
4	0.00000	-0.37285	-2.22943	12	-0.3315
5	-0.00000	-1.36887	-1.25988	12	-0.2812
6	-0.00000	-1.02722	0.08748	12	-0.3615
7	-0.00000	0.64596	1.78841	23	-0.8067
8	0.00000	2.34582	-0.21073	2	0.2970
9	0.00000	1.73871	-2.60496	2	0.2875
10	0.00000	-0.63831	-3.27630	2	0.2879
11	-0.00000	-2.40821	-1.55342	2	0.2873
12	-0.00000	-1.80111	0.84081	2	0.3023
13	-0.00000	-0.42953	2.73087	13	-0.4209
14	-0.00000	-0.04727	3.74096	3	0.2832
15	0.87608	-1.04734	2.59974	3	0.2841
16	-0.87607	-1.04734	2.59974	3	0.2841
16	stretching				
1	2	1.390	4880.7	C- C	
1	6	1.390	4880.9	C- C	

1	7	1.365	4873.1	C- O
2	3	1.390	4880.8	C- C
2	8	1.080	3600.0	C- H
3	4	1.390	4880.7	C- C
3	9	1.080	3600.0	C- H
4	5	1.390	4880.9	C- C
4	10	1.080	3600.0	C- H
5	6	1.390	4880.8	C- C
5	11	1.080	3600.0	C- H
6	12	1.080	3600.0	C- H
7	13	1.430	3418.1	O- C
13	14	1.080	3600.0	C- H
13	15	1.080	3600.0	C- H
13	16	1.080	3600.0	C- H
25 bending				
1	2	3	120.00	583.8 C- C- C
1	2	8	120.00	505.0 C- C- H
1	6	5	120.00	583.8 C- C- C
1	6	12	120.00	505.0 C- C- H
1	7	13	117.00	619.5 C- O- C
2	1	6	120.00	583.8 C- C- C
2	1	7	120.00	673.6 C- C- O
2	3	4	120.00	583.8 C- C- C
2	3	9	120.00	505.0 C- C- H
3	2	8	120.00	505.0 C- C- H
3	4	5	120.00	583.8 C- C- C
3	4	10	120.00	505.0 C- C- H
4	3	9	120.00	505.0 C- C- H
4	5	6	120.00	583.8 C- C- C
4	5	11	120.00	505.0 C- C- H
5	4	10	120.00	505.0 C- C- H
5	6	12	120.00	505.0 C- C- H
6	1	7	120.00	673.6 C- C- O
6	5	11	120.00	505.0 C- C- H
7	13	14	110.00	630.0 O- C- H
7	13	15	111.00	617.5 O- C- H
7	13	16	111.00	617.5 O- C- H
14	13	15	108.00	470.0 H- C- H
14	13	16	108.00	470.0 H- C- H
15	13	16	108.00	470.0 H- C- H
14 torsion				
6	1	2	3	50.00 -1.0 1.0 C- C- C- C
2	1	6	5	50.00 -1.0 1.0 C- C- C- C
1	2	6	7	100.00 -1.0 1.0 C- C- C- O
1	2	3	4	50.00 -1.0 1.0 C- C- C- C
2	1	3	8	100.00 -1.0 1.0 C- C- C- H
2	3	4	5	50.00 -1.0 1.0 C- C- C- C
3	2	4	9	100.00 -1.0 1.0 C- C- C- H
3	4	5	6	50.00 -1.0 1.0 C- C- C- C
4	3	5	10	100.00 -1.0 1.0 C- C- C- H
4	5	6	1	50.00 -1.0 1.0 C- C- C- C
5	4	6	11	100.00 -1.0 1.0 C- C- C- H
6	1	5	12	100.00 -1.0 1.0 C- C- C- H
6	1	7	13	2.00 -1.0 2.0 C- C- O- C
1	7	13	14	2.50 1.0 3.0 C- O- C- H

1,2-dichloroethane liquid					
8					
1	0.00000	-0.53608	-0.52454	13	-0.2698
2	-0.00000	0.53604	0.52454	13	-0.2698
3	0.00000	-0.05451	2.20371	42	-0.2840
4	-0.88468	1.13932	0.38390	3	0.2769
5	0.88468	1.13932	0.38390	3	0.2769
6	0.00000	0.05452	-2.20371	42	-0.2839
7	-0.88469	-1.13937	-0.38391	3	0.2769
8	0.88468	-1.13936	-0.38391	3	0.2769
7 stretching					
1	2	1.500	2564.5	C- C	
1	6	1.780	2500.0	C-Cl	
1	7	1.080	3600.0	C- H	
1	8	1.080	3600.0	C- H	
2	3	1.780	2500.0	C-Cl	
2	4	1.080	3600.0	C- H	
2	5	1.080	3600.0	C- H	
12 bending					
1	2	3	115.00	601.0	C- C-Cl
1	2	4	108.00	655.0	C- C- H
1	2	5	108.00	655.0	C- C- H
2	1	6	115.00	601.0	C- C-Cl
2	1	7	108.00	655.0	C- C- H
2	1	8	108.00	655.0	C- C- H

3	2	4	108.00	655.0	C1-	C-	H
3	2	5	108.00	655.0	C1-	C-	H
4	2	5	110.00	470.0	H-	C-	H
6	1	7	108.00	655.0	C1-	C-	H
6	1	8	108.00	655.0	C1-	C-	H
7	1	8	110.00	470.0	H-	C-	H
1 torsion							
6	1	2	3	7.50	1.0	3.0	C1- C- C-C1
1 intram.nonbonded							
3 6							

ethanol liquid							
9							
1	-0.00000	-0.23739	-1.29910	13	-0.9094		
2	0.00000	0.54635	0.00909	13	0.0717		
3	-0.00000	-0.31271	1.14603	29	-1.3356		
4	0.00000	0.21725	1.96443	5	0.7314		
5	-0.87374	1.18024	0.04314	3	0.2680		
6	0.87374	1.18024	0.04314	3	0.2680		
7	-0.87608	-0.86566	-1.36361	3	0.2995		
8	0.00000	0.43602	-2.14345	3	0.3068		
9	0.87608	-0.86567	-1.36360	3	0.2995		
8	stretching						
1	2	1.525	2038.1	C-	C		
1	7	1.080	3600.0	C-	H		
1	8	1.080	3600.0	C-	H		
1	9	1.080	3600.0	C-	H		
2	3	1.425	3529.9	C-	O		
2	5	1.080	3600.0	C-	H		
2	6	1.080	3600.0	C-	H		
3	4	1.000	4250.0	O-	H		
13	bending						
1	2	3	112.00	529.4	C-	C-	O
1	2	5	109.00	642.5	C-	C-	H
1	2	6	109.00	642.5	C-	C-	H
2	1	7	110.00	630.0	C-	C-	H
2	1	8	110.00	630.0	C-	C-	H
2	1	9	110.00	630.0	C-	C-	H
2	3	4	110.00	450.0	C-	O-	H
3	2	5	109.00	642.5	O-	C-	H
3	2	6	109.00	642.5	O-	C-	H
5	2	6	108.00	470.0	H-	C-	H
7	1	8	108.00	470.0	H-	C-	H
7	1	9	108.00	470.0	H-	C-	H
8	1	9	108.00	470.0	H-	C-	H
2	torsion						
7	1	2	3	7.50	1.0	3.0	H- C- C- O
1	2	3	4	2.00	1.0	3.0	C- C- O- H
3	intram.nonbonded						
4	7	4	8	4	9		

n-pentane liquid					
17					
1	0.00000	0.33158	2.52857	13	-0.8988
2	0.00000	-0.52118	1.26428	13	-0.5522
3	-0.00000	0.33159	-0.00000	13	-0.5622
4	-0.00000	-0.52119	-1.26428	13	-0.5522
5	-0.00000	0.33159	-2.52856	13	-0.8988
6	-0.87608	0.96242	2.55929	3	0.2952
7	0.00000	-0.29557	3.40781	3	0.2950
8	0.87608	0.96242	2.55929	3	0.2952
9	0.87374	-1.15599	1.26428	3	0.2824
10	-0.87374	-1.15599	1.26428	3	0.2824
11	0.87374	0.96640	0.00000	3	0.2818
12	-0.87374	0.96640	-0.00000	3	0.2818
13	-0.87374	-1.15599	-1.26428	3	0.2824
14	0.87374	-1.15599	-1.26428	3	0.2824
15	0.87608	0.96242	-2.55928	3	0.2952
16	-0.00000	-0.29556	-3.40781	3	0.2950
17	-0.87607	0.96242	-2.55929	3	0.2952
16	stretching				
1	2	1.525	2037.9	C-	C
1	6	1.080	3600.0	C-	H
1	7	1.080	3600.0	C-	H
1	8	1.080	3600.0	C-	H
2	3	1.525	2038.1	C-	C
2	9	1.080	3600.0	C-	H
2	10	1.080	3600.0	C-	H
3	4	1.525	2038.0	C-	C
3	11	1.080	3600.0	C-	H

3	12	1.080	3600.0	C-	H
4	5	1.525	2038.1	C-	C
4	13	1.080	3600.0	C-	H
4	14	1.080	3600.0	C-	H
5	15	1.080	3600.0	C-	H
5	16	1.080	3600.0	C-	H
5	17	1.080	3600.0	C-	H
30 bending					
1	2	3	112.00	527.6	C- C- C
1	2	9	109.00	642.5	C- C- H
1	2	10	109.00	642.5	C- C- H
2	1	6	110.00	630.0	C- C- H
2	1	7	111.00	617.5	C- C- H
2	1	8	110.00	630.0	C- C- H
2	3	4	112.00	527.6	C- C- C
2	3	11	109.00	642.5	C- C- H
2	3	12	109.00	642.5	C- C- H
3	2	9	109.00	642.5	C- C- H
3	2	10	109.00	642.5	C- C- H
3	4	5	112.00	527.6	C- C- C
3	4	13	109.00	642.5	C- C- H
3	4	14	109.00	642.5	C- C- H
4	3	11	109.00	642.5	C- C- H
4	3	12	109.00	642.5	C- C- H
4	5	15	111.00	617.5	C- C- H
4	5	16	111.00	617.5	C- C- H
4	5	17	111.00	617.5	C- C- H
5	4	13	109.00	642.5	C- C- H
5	4	14	109.00	642.5	C- C- H
6	1	7	108.00	470.0	H- C- H
6	1	8	108.00	470.0	H- C- H
7	1	8	108.00	470.0	H- C- H
9	2	10	108.00	470.0	H- C- H
11	3	12	108.00	470.0	H- C- H
13	4	14	108.00	470.0	H- C- H
15	5	16	108.00	470.0	H- C- H
15	5	17	108.00	470.0	H- C- H
16	5	17	108.00	470.0	H- C- H
4 torsion					
6	1	2	3	6.00	1.0 3.0 H- C- C- C
1	2	3	4	6.00	1.0 3.0 C- C- C- C
2	3	4	5	6.00	1.0 3.0 C- C- C- C
3	4	5	15	6.00	1.0 3.0 C- C- C- H
21 intram.nonbonded					
6	13	6	14	6	15 6 16 6 17
7	13	7	14	7	15 7 16 7 17
8	13	8	14	8	15 8 16 8 17
9	15	9	16	9	17 10 15 10 16 10 17

pyridine liquid					
11					
1	0.00002	0.00155	-1.39998	18	-0.8204
2	0.00017	-1.14382	-0.69520	12	0.1038
3	-0.00023	-1.19767	0.69858	12	-0.3235
4	0.00011	-0.00166	1.41442	12	-0.1626
5	0.00006	1.19593	0.70196	12	-0.3235
6	-0.00014	1.14567	-0.69277	12	0.1039
7	0.00063	-2.07623	-1.24010	2	0.2726
8	-0.00077	-2.14735	1.21297	2	0.2956
9	0.00041	-0.00314	2.49442	2	0.2857
10	0.00020	2.14425	1.21887	2	0.2956
11	-0.00040	2.07937	-1.23557	2	0.2727
11 stretching					
1	2	1.345	5736.7	N- C	
1	6	1.345	5731.9	N- C	
2	3	1.395	4779.5	C- C	
2	7	1.080	3600.0	C- H	
3	4	1.394	4799.2	C- C	
3	8	1.080	3600.0	C- H	
4	5	1.393	4807.4	C- C	
4	9	1.080	3600.0	C- H	
5	6	1.396	4762.2	C- C	
5	10	1.080	3600.0	C- H	
6	11	1.080	3600.0	C- H	
16 bending					
1	2	3	124.00	646.0	N- C- C
1	2	7	118.00	530.0	N- C- H
1	6	5	124.00	646.0	N- C- C

1	6	11	118.00	530.0	N-	C-	H
2	1	6	117.00	580.9	C-	N-	C
2	3	4	119.00	576.7	C-	C-	C
2	3	8	121.00	492.5	C-	C-	H
3	2	7	118.00	530.0	C-	C-	H
3	4	5	118.00	569.7	C-	C-	C
3	4	9	121.00	492.5	C-	C-	H
4	3	8	121.00	492.5	C-	C-	H
4	5	6	119.00	576.7	C-	C-	C
4	5	10	121.00	492.5	C-	C-	H
5	4	9	121.00	492.5	C-	C-	H
5	6	11	118.00	530.0	C-	C-	H
6	5	10	121.00	492.5	C-	C-	H
11	torsion						
6	1	2	3	50.00	-1.0	1.0	C- N- C- C
2	1	6	5	50.00	-1.0	1.0	C- N- C- C
1	2	3	4	50.00	-1.0	1.0	N- C- C- C
2	1	3	7	100.00	-1.0	1.0	C- N- C- H
2	3	4	5	50.00	-1.0	1.0	C- C- C- C
3	2	4	8	100.00	-1.0	1.0	C- C- C- H
3	4	5	6	50.00	-1.0	1.0	C- C- C- C
4	3	5	9	100.00	-1.0	1.0	C- C- C- H
4	5	6	1	50.00	-1.0	1.0	C- C- C- N
5	4	6	10	100.00	-1.0	1.0	C- C- C- H
6	1	5	11	100.00	-1.0	1.0	C- N- C- H

Table S5

Data for outliers in the sublimation enthalpy-lattice energy match with LJC force field. CSD refcode, sublimation enthalpy, lattice energy and deviations, number of experimental determinations of SE, and compound name. Data from Table S1 arranged by chemical class.

refco	DHs	Eljc,	%Dev	D(abs)	Nexpt	name	
FUMAAC01	136.4	119.6	12.3	16.8	2	trans-butenedioic	acids
FURDCA	122.8	109.4	11.0	13.5	1	furan-2,5-dicarboxylic	
MXCINN02	134	117.3	12.5	16.7	1	trans-4-methoxycinnamic	
AMBAC008	110	96.5	12.3	13.5	2	2-aminobenzoic	
NICOAC02	115	102.2	11.1	12.8	2	3-pyridinecarboxylic	
QQQFDJ01	122.1	135.5	-11.0	-13.4	1	9-anthracenecarboxylic	
PCPHOL01	93.2	107.9	-15.8	-14.7	1	pentachlorophenol	alcohols
UCAYED01	67.2	79.7	-18.6	-12.5	1	pentafluorophenol	
CATCOL16	85	96.1	-13.0	-11.1	2	1,2-dihydroxybenzene	
FECCOF01	110	125.6	-14.2	-15.6	2	1,6-hexanediol	
HYQUIN08	100.4	111.2	-10.7	-10.8	6	1,4-dihydroxybenzene	
RESORA03	93	105.7	-13.6	-12.7	5	1,3-dihydroxybenzene	
PHBALD11	101	90.8	10.1	10.2	3	4-hydroxybenzaldehyde	H..O=C,no alc-alc Hbond
XAYCIJ	100	88.3	11.7	11.7	2	3-hydroxybenzaldehyde	id.
FAMDIG	112	93.6	16.4	18.4	2	5-chloro-8-hydroxyquinoline	OH...N
CLNOBE02	103	119.3	-15.9	-16.3	2	1-chloro-2,4,6-trinitrobenzene	nitro
BECEJY	90	100.1	-11.2	-10.1	4	2,3-dinitro-2,3-dimethylbutane	
DNBENZ15	85	95.1	-11.9	-10.1	3	1,3-dinitrobenzene	
MLEICA01	86.4	69.9	19.1	16.5	1	maleic anhydride	diones N-oxides
NIKVUZ	69	57.3	17.0	11.8	1	cyclobutane-1,2-dione	
INDDON	97	84.9	12.5	12.1	1	1,3-indandione	
ZZZPRC03	106	88.4	16.6	17.6	1	1-methyl-1H-indole-2,3-dione	
RAJGUG	89	70.6	20.7	18.4	2	2-methylpyridine Noxide	
VAKJUM	124	105	15.3	19	1	2,3-dimethylquinoxaline 1,4dioxide	
CAACTY	41.2	30	27.3	11.2	1	cyanoacetylene	cyano
CYNGEN	31.2	35.8	-14.6	-4.6	4	cyanogen	
MALONT	79	61.4	22.3	17.6	2	malononitrile	
BETHAZ01	71	82	-15.5	-11	1	2,1,3-benzothiadiazole	sulfa
DMSULO04	77	87.5	-13.6	-10.5	1	dimethyl sulfone	
THIOUR16	108.6	84.4	22.3	24.2	7	thiourea	
PRYLTS	96	111.6	-16.2	-15.6	1	p-tolyl propenylsulfone	
MOYGUC01	62.2	77.6	-24.8	-15.4	1	thiocamphor	
TUXFIC02	125	142.7	-14.1	-17.7	1	dibenzyl sulfone	
PHUREA02	138.2	121	12.4	17.2	1	monophenylurea	NH..O=
DIKPIPO3	106.6	123.9	-16.3	-17.3	1	2,5-piperazinedione	
MODYAG01	138.2	122.1	11.6	16.1	1	4-aminophthalimide	
ACANIL06	87	98.3	-13	-11.3	2	acetanilide	
HQOXAL01	120.9	104.5	13.6	16.4	2	2-hydroxyquinoxaline	keto form NH..O=
NONMEI	93	83.4	10.4	9.6	1	8-aminoquinoline	NH..N
NICOAM06	120	104.2	13.2	15.8	2	3-pyridinecarboxamide	
FULPIM	88.4	77.3	12.5	11.1	1	2-methylimidazole	
IMAZOL13	83	71.8	13.5	11.2	4	imidazole	
ZBCNON	58	65.8	-13.4	-7.8	1	3-azabicyclo 3.2.2nonane	aza
HXMTAM23	80	89.5	-11.9	-9.5	5	1,3,5,7-tetraazatricyclo 3.3.1.13,7decane	
HEYJOK01	68	76.7	-12.8	-8.7	2	quinoxaline 1,4	
MPYRAZ02	95	78.1	17.8	16.9	1	tetramethylpyrazine	wrong DHs pyrazine 56
TROXAN	56	63.3	-12.9	-7.3	3	1,3,5-trioxane	ether Oxygen
GOCWEA	82	91.7	-11.8	-9.7	1	trans-1,3,5,7-tetraoxadecalin	
XOGJAG	53.6	63.4	-18.3	-9.8	1	trifluoromethylbenzene	
NMACRO11	105	118.5	-12.8	-13.5	1	10-methylacridin-9	10H -one
TXOXBZ10	100	89.2	10.8	10.8	1	1,3,5-trimethoxybenzene	
TEKQIJ	100.6	113.2	-12.5	-12.6	2	2,2',5,5'-tetrachlorobiphenyl	
FELSEU	112.9	124.7	-10.5	-11.9	1	1,2,3,4-tetrachlorodibenzo-1,4	dioxin
BEQXAZ	65	75.2	-15.6	-10.2	2	benzofurazan	unreliable Xray 1951 22%R
CLANIC06	86	73.1	15.0	12.9	2	4-chloroaniline	short CH..Cl

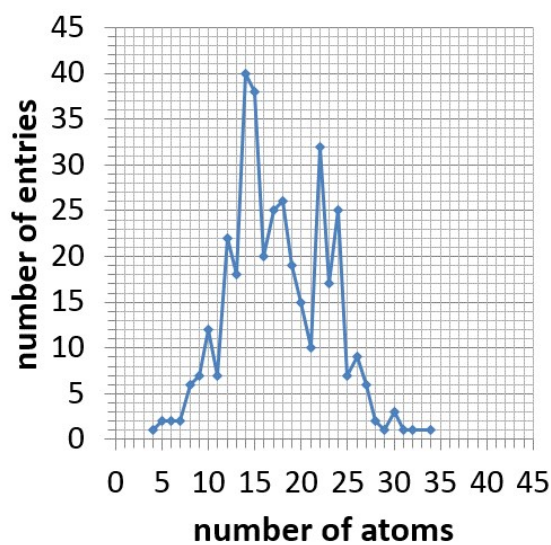


Figure S1.

Distribution of number of atoms per molecule in the 377-molecule data set.

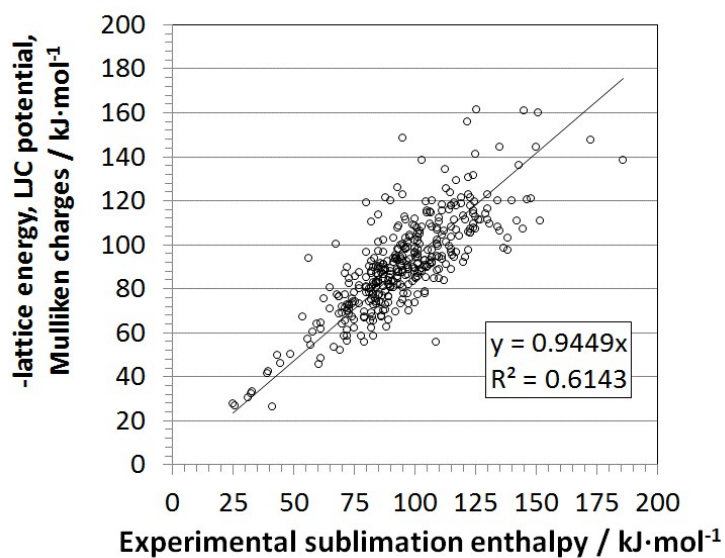


Figure S2.

Performance of the LJC potential with Mulliken charges to reproduce experimental sublimation enthalpies for the 377 crystal structures in the current dataset (see text).

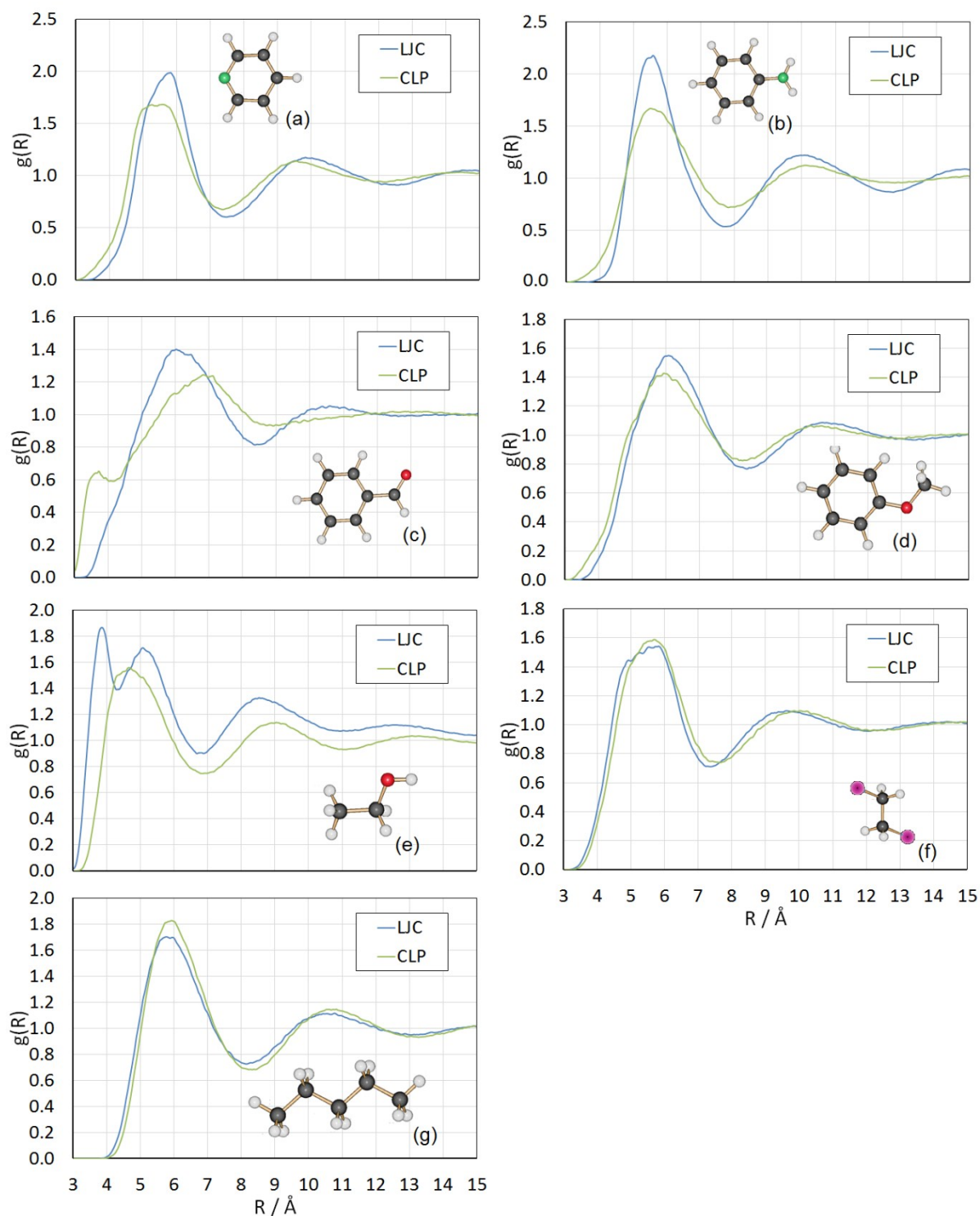


Figure S3.

Centre-of-mass radial distribution functions, $g(R)$, averaged through the last 200 ps of the simulation for liquids (a)–(g) (see main text, Figure 1). Atom code: C: black; H: white; O: red; N: green; Cl: purple. CLP (LJC) results are shown as green (blue) lines.

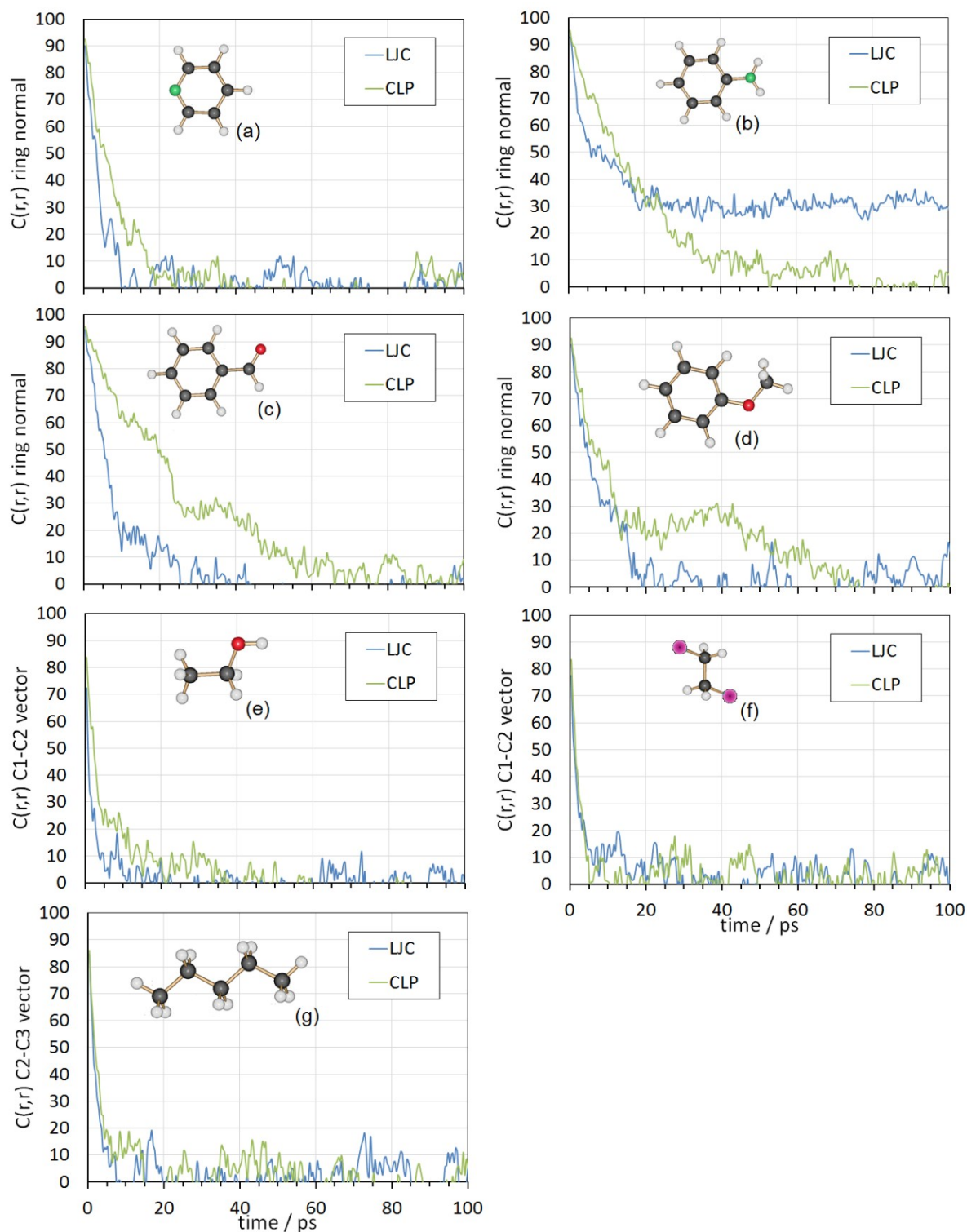


Figure S4.

Raw rotational correlation function, $C(\mathbf{u})$ (uncorrected for smoothing) as a function of time in the first 100 ps of the trajectory for liquids. See Fig. S3 above for explanation of symbols.

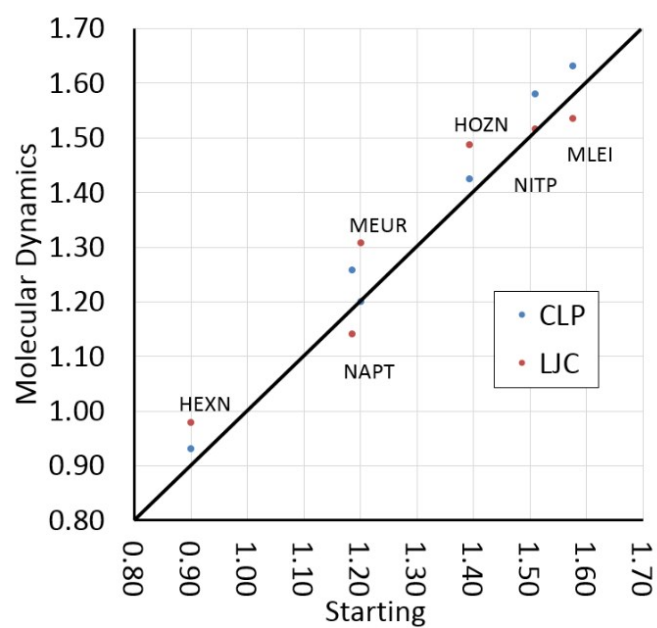


Figure S5.

Comparison between crystal densities from Molecular Dynamics simulations and experiment: HEXN: n-hexane at 90 K; NAPT: naphthalene at 298 K; MEUR: methylurea at 298 K; HOZN: trimethyltrioxane CHCl_3 solvate at 190 K; NITP: p-nitrophenol at 110 K; maleic anhydride at 130 K.

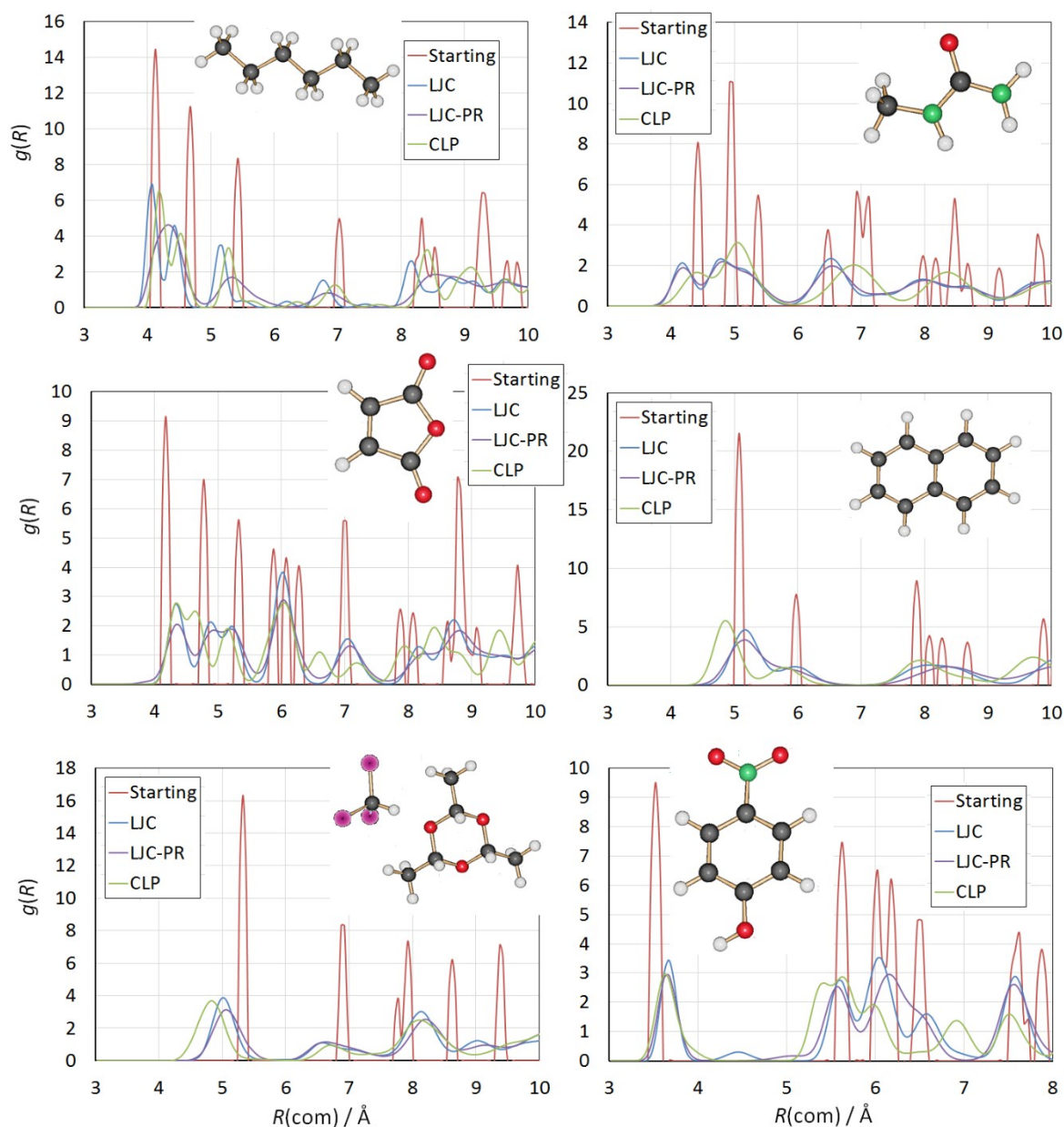


Figure S6.

Centre-of-mass radial distribution functions, $g(R)$, from MD simulation of crystals (1)–(6) (see main text, Figure 1). Red lines: starting crystal structures after short MC thermalization; CLP results, and 6-12-q with weak-coupling (LJC) or Parrinello–Rahman (LJC–PR) barostat. Atom code: C: black; H: white; O: red; N: green; Cl: purple.