

Achievement of enhanced solubility and improved optics in the molecular complexes based on sulfonate-pyridinium supramolecular synthon

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KEYWORDS: Co-crystal, molecular salt, solubility, optical properties

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Fig. ESI-2 FT-IR Spectrum of **2**.

Fig. ESI-3 FT-IR Spectrum of **3**.

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Fig. ESI-5 ¹H NMR Spectrum of **2** reported in DMSO-*d*₆ at 400 MHz.

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Table ESI-1 Comparative MPs of formers with molecular salts.

Table ESI-2 Solubility of compounds **1-3** in different solvents.

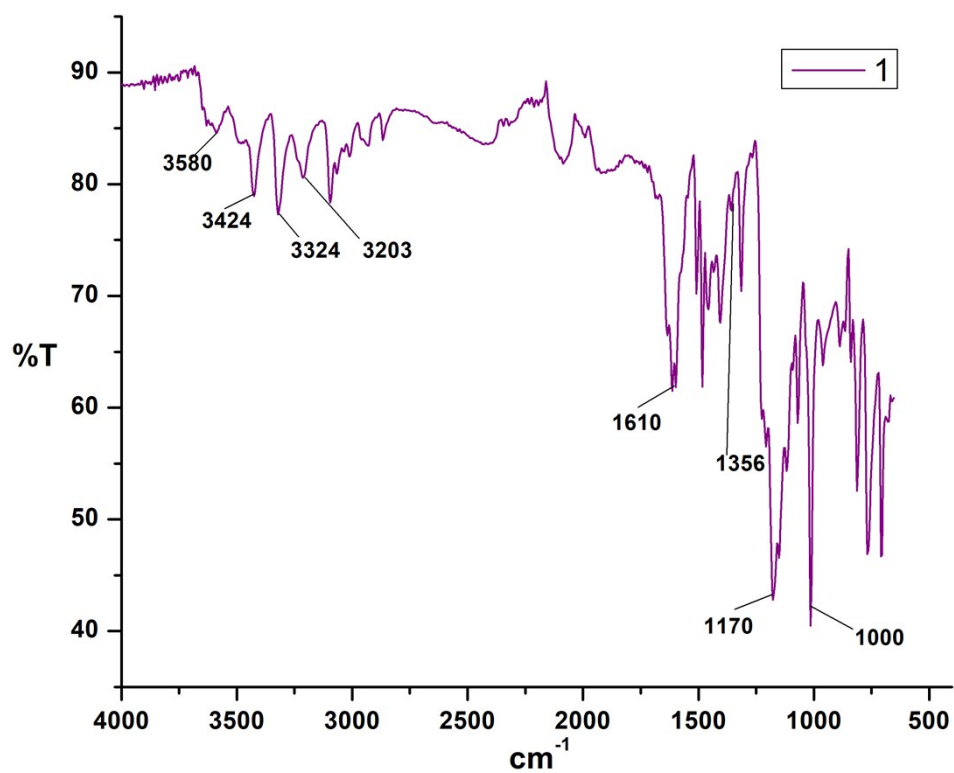


Fig. ESI-1 FT-IR Spectrum of 1.

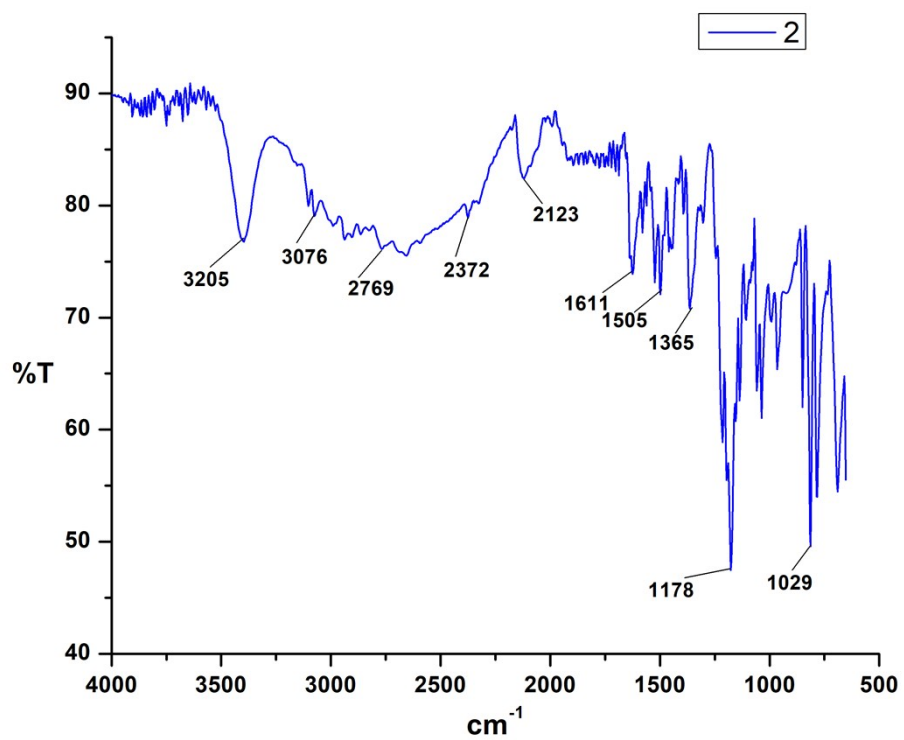


Fig. ESI-2 FT-IR Spectrum of 2.

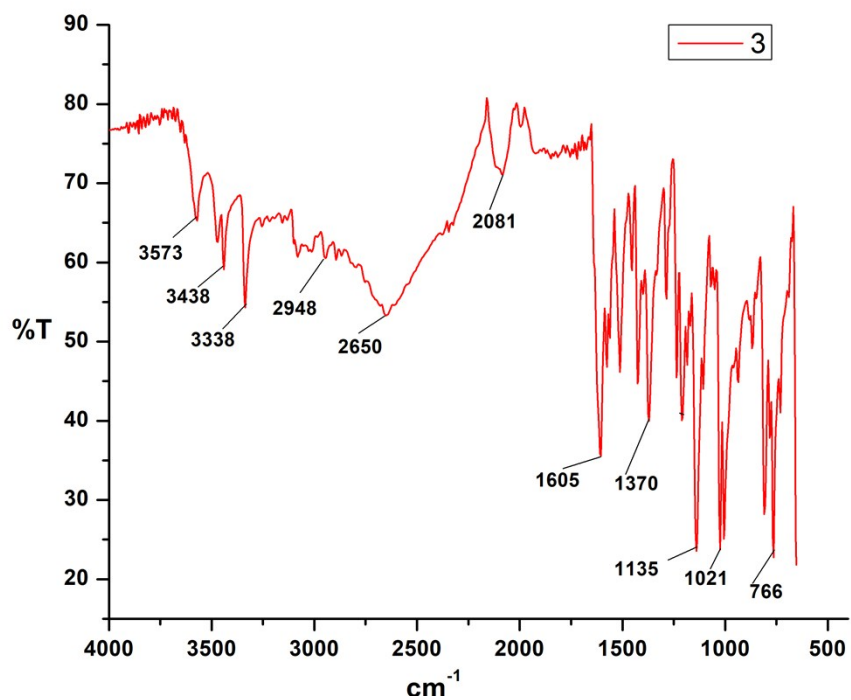


Fig. ESI-3 FT-IR Spectrum of **3**.

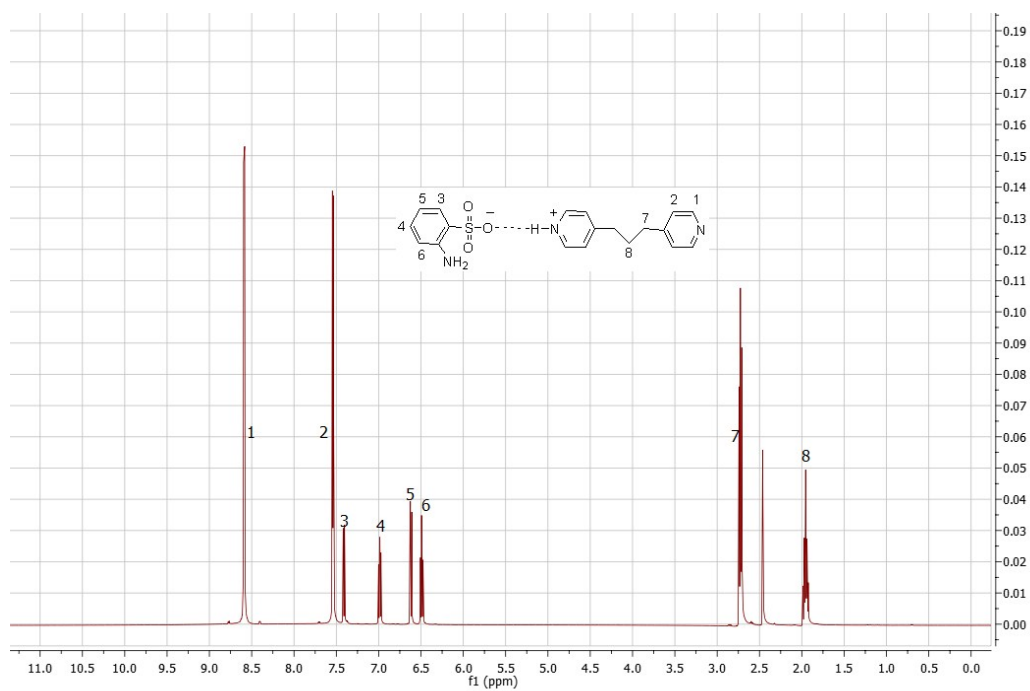


Fig. ESI-4 ^1H NMR Spectrum of **1** reported in $\text{DMSO-}d_6$ at 400 MHz.

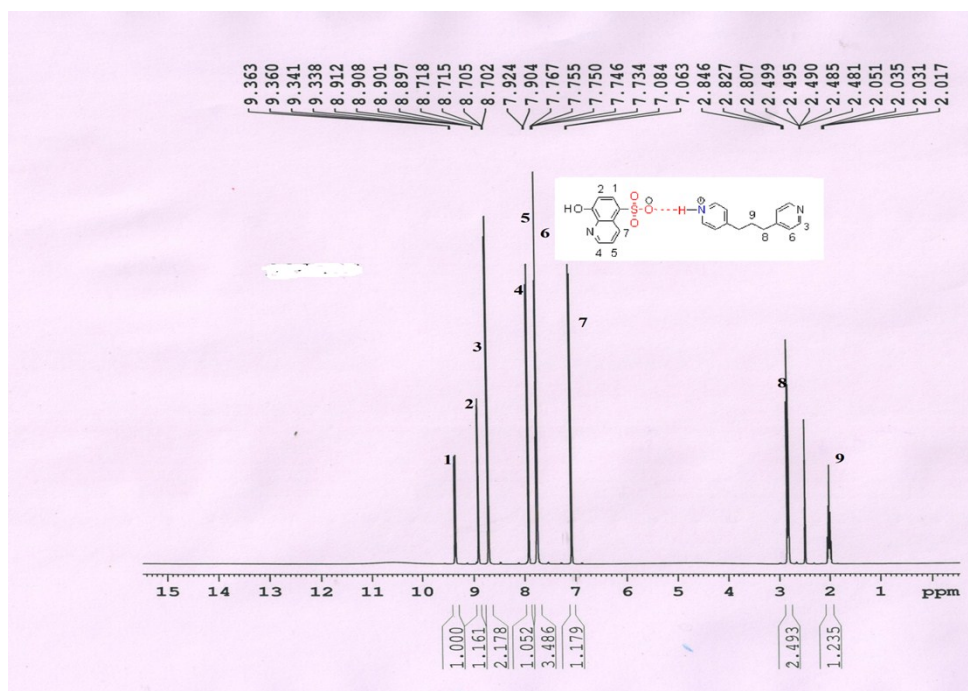


Fig. ESI-5 ¹H NMR Spectrum of **2** reported in DMSO-*d*₆ at 400 MHz.

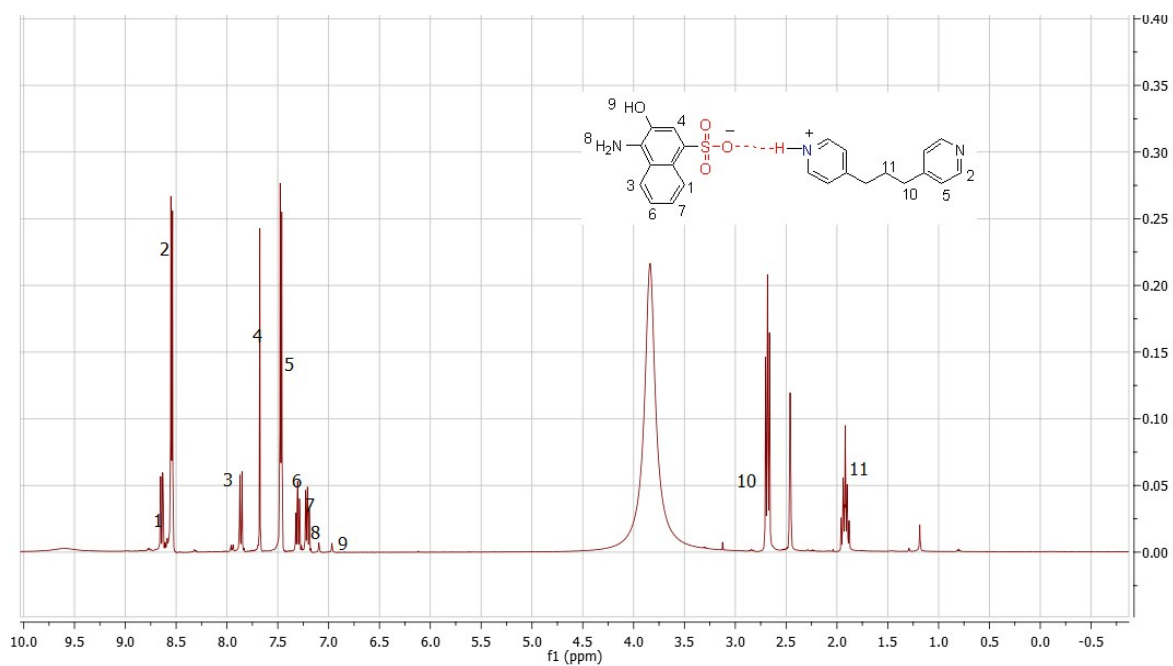


Fig. ESI-6 ¹H NMR Spectrum of **3** reported in DMSO-*d*₆ at 400 MHz.

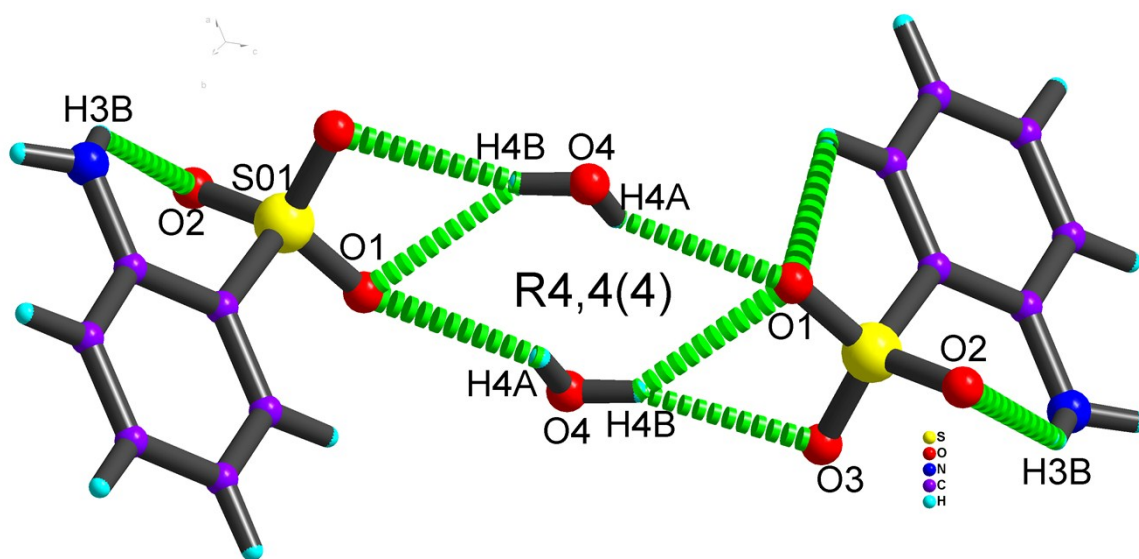


Fig ESI-7 Centro-symmetric hydrated amino-aryl sulfonate dimer in **1**.

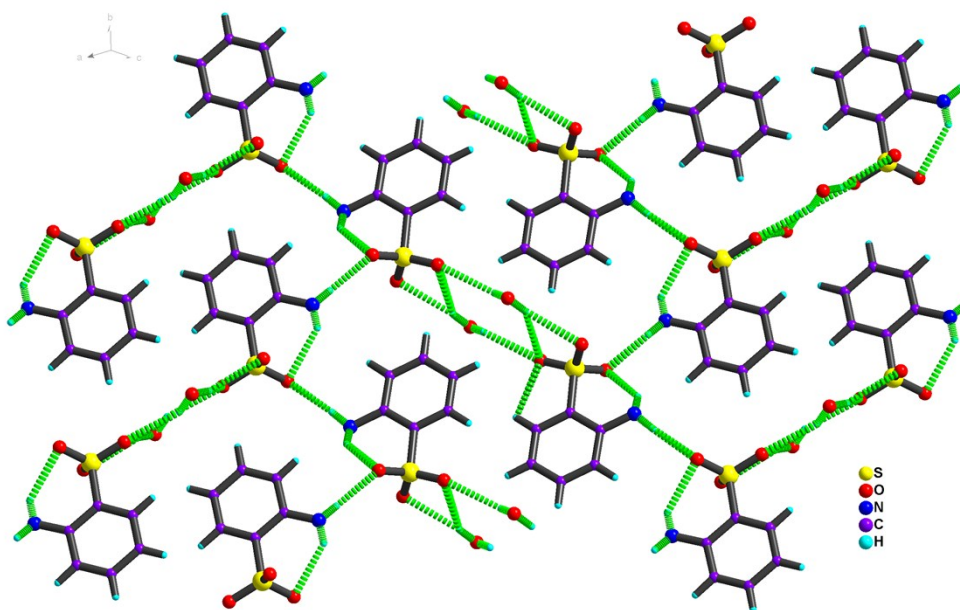


Fig. ESI-8 Anionic hydrated layers of amino-aryl sulfonate dimer in **1**.

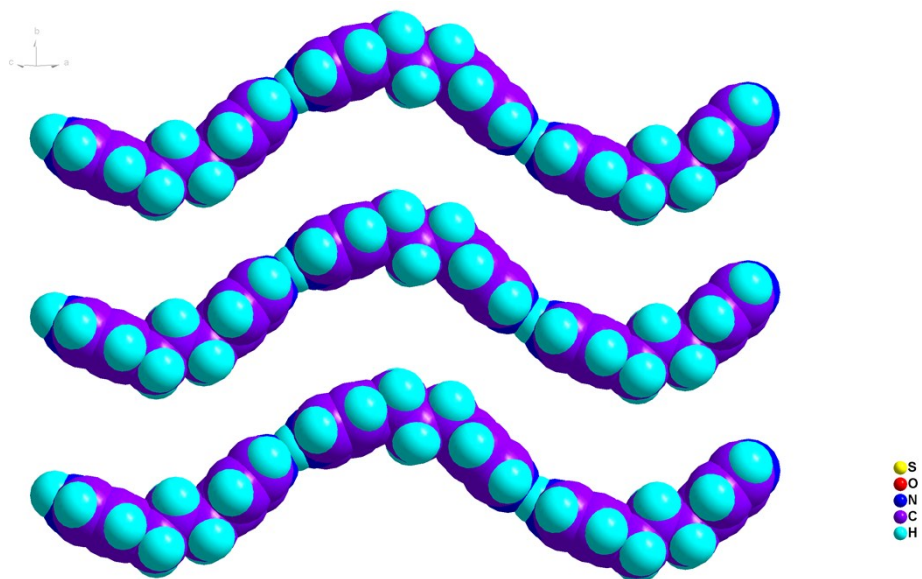


Fig. ESI-9 Isolated bis-pyridinium chains in **1**.

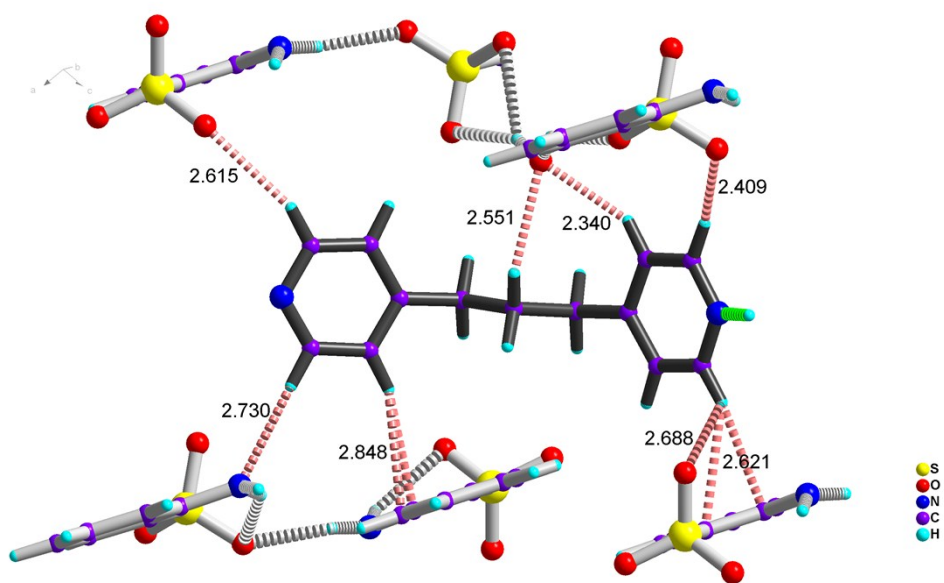


Fig. ESI-10 Interactions between bis-pyridinium chains and anionic layers of amino-aryl sulfonate layers in **1**.

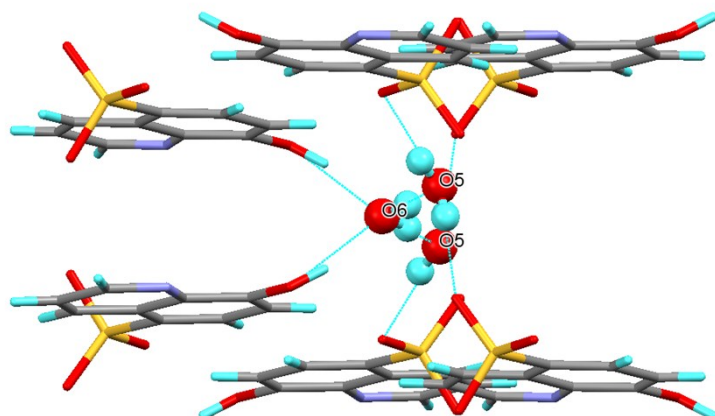


Fig. ESI-11 Hexa-sulfonate:tri-aqua clusters in **2**.

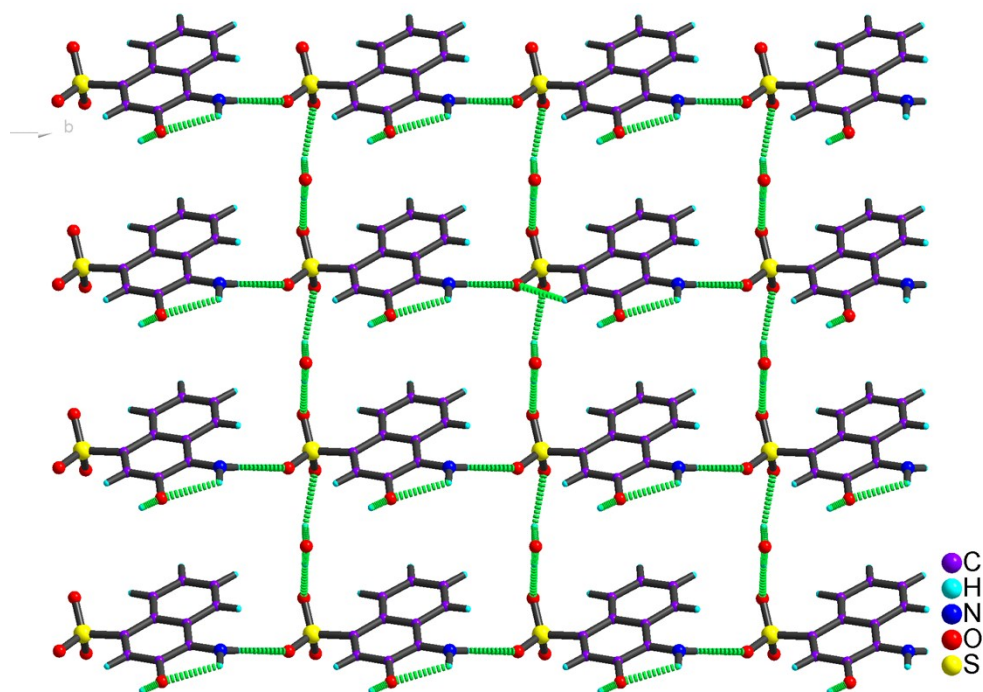


Fig. ESI-12 2-D hydrated anionic nets of organo-sulfonates in **3**.

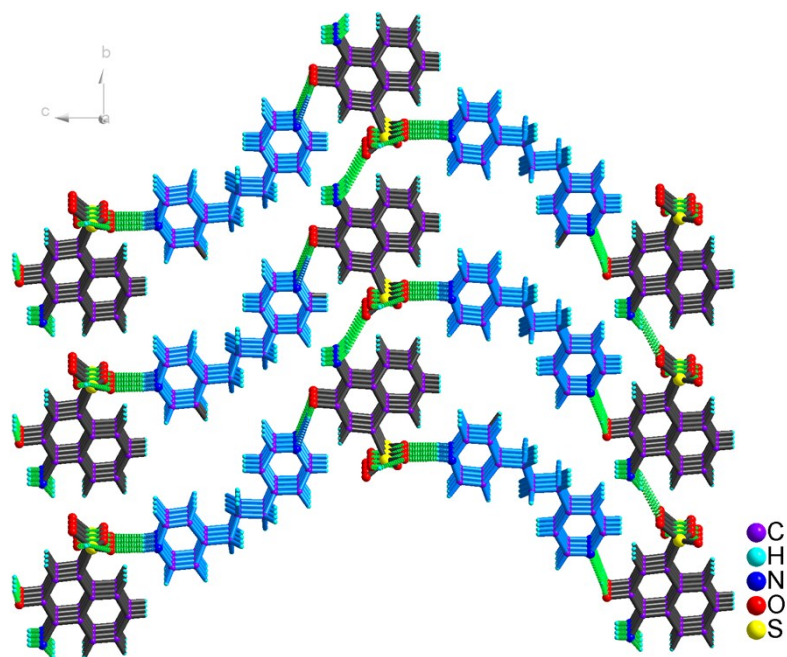


Fig. ESI-13 3-D reticular nets formed in 3 by bridging of anionic layers of hydrated organo-sulfonate layers by 4,4'-BPP in 3.

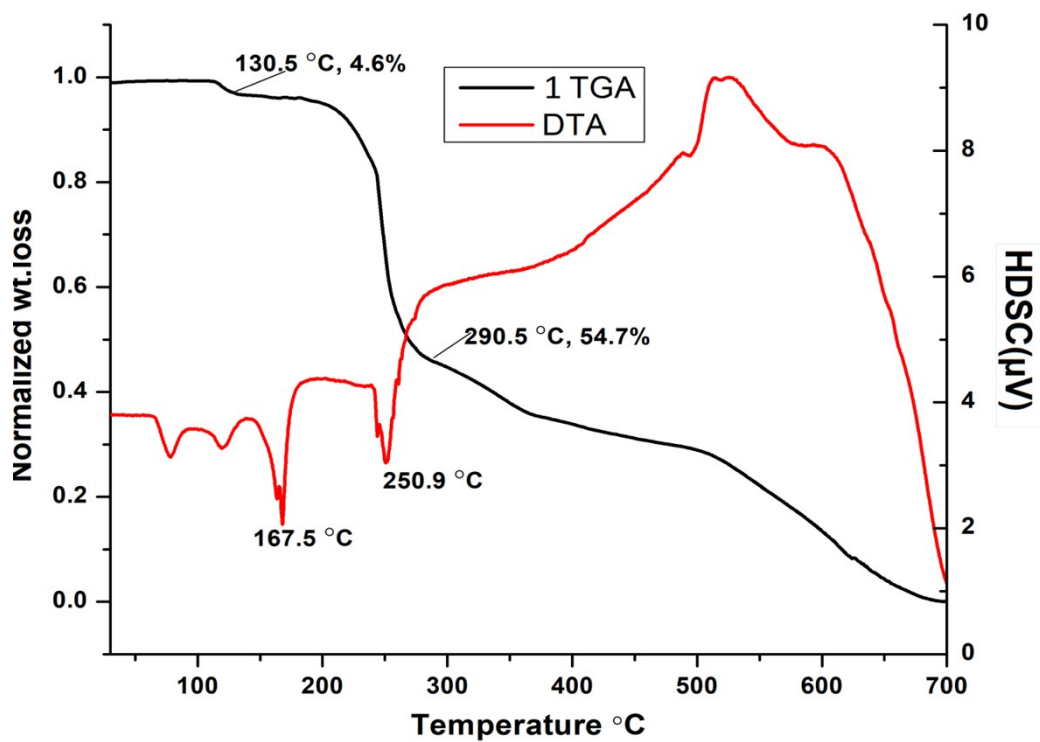


Fig. ESI-14 TGA-DSC curves of 1.

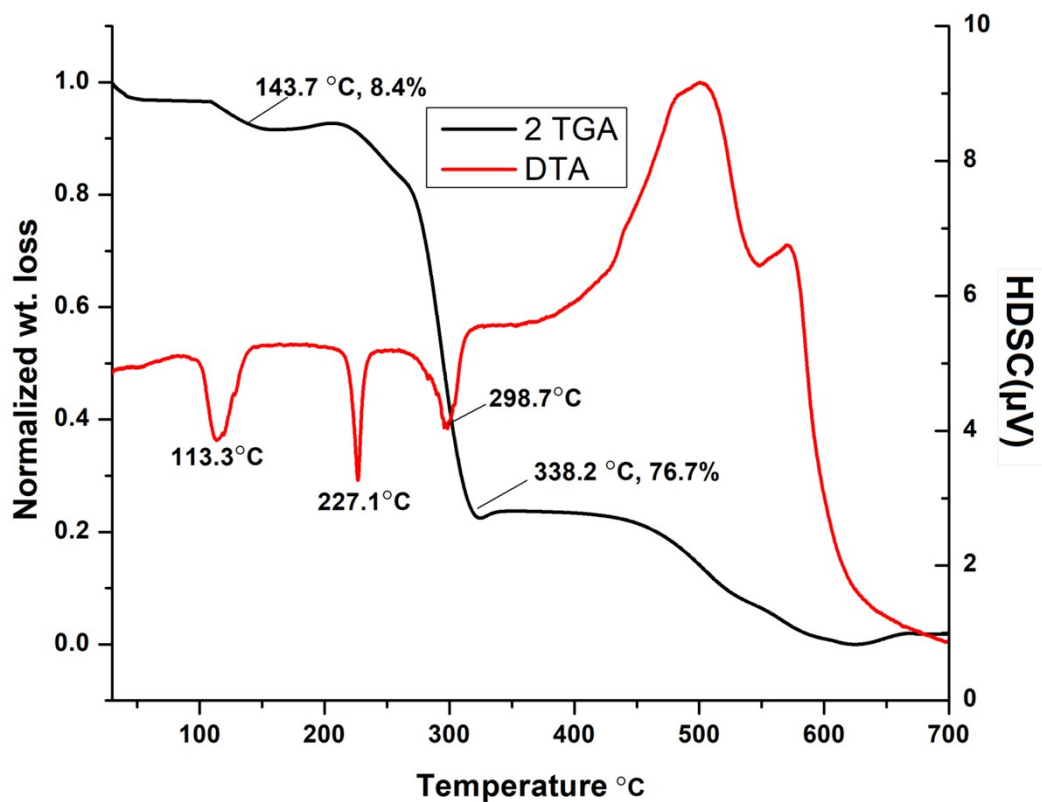


Fig. ESI-15 TGA-DSC curves of 2.

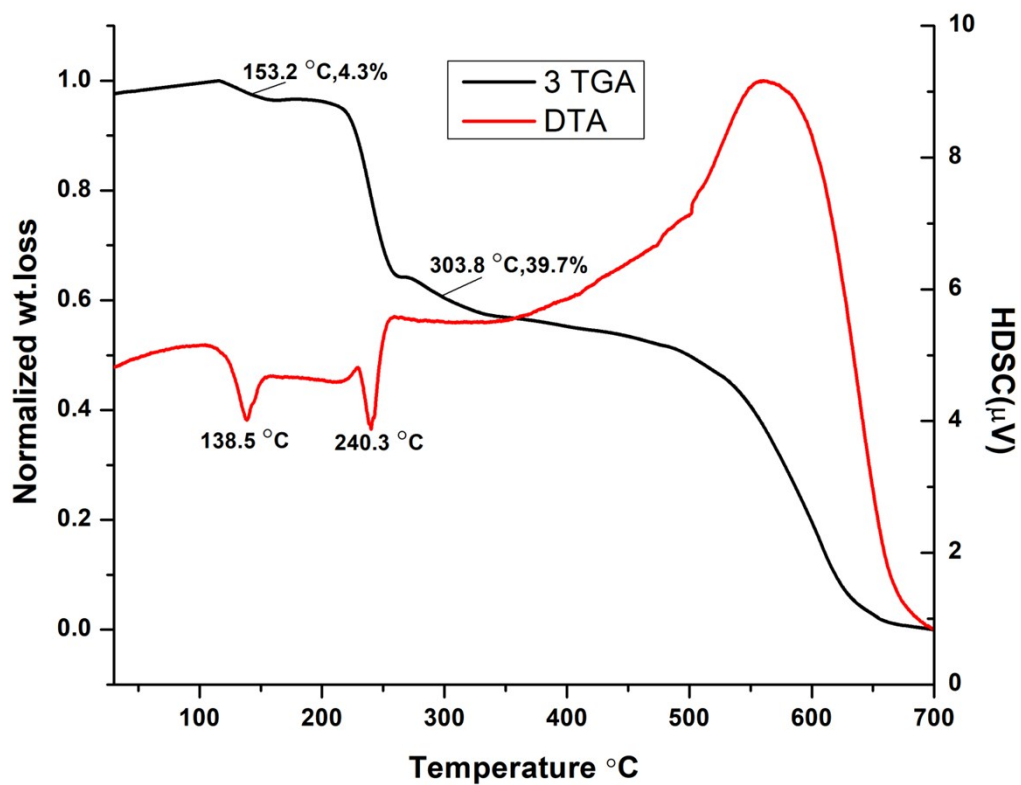


Fig. ESI-16 TGA-DSC curves of 3.

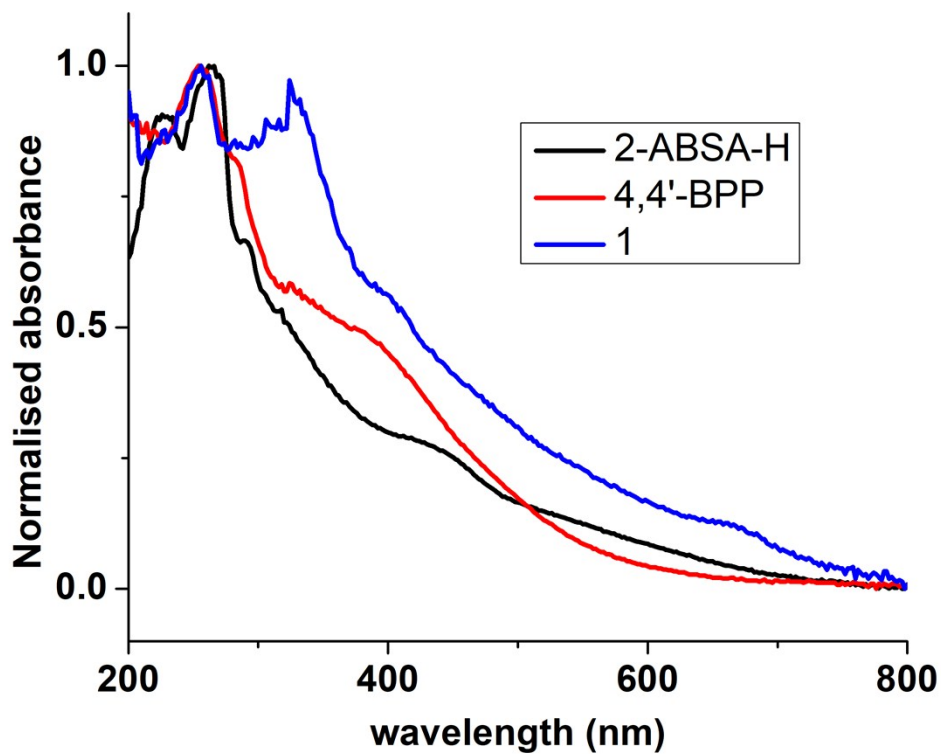


Fig. ESI-17 DR-UV-Vis spectra of starting materials and Compound 1.

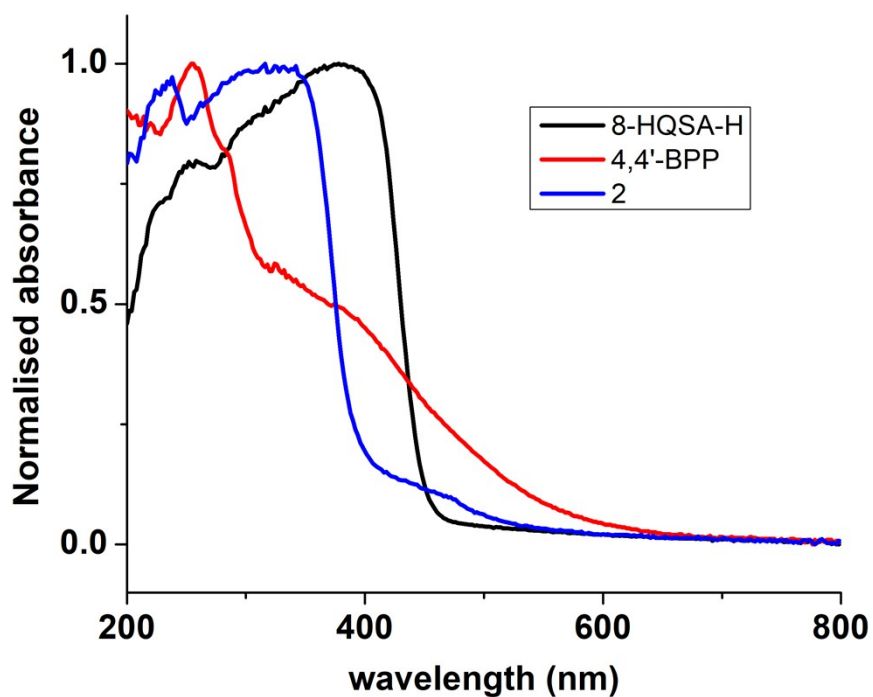


Fig. ESI-18 DR-UV-Vis spectra of starting materials and Compound 2.

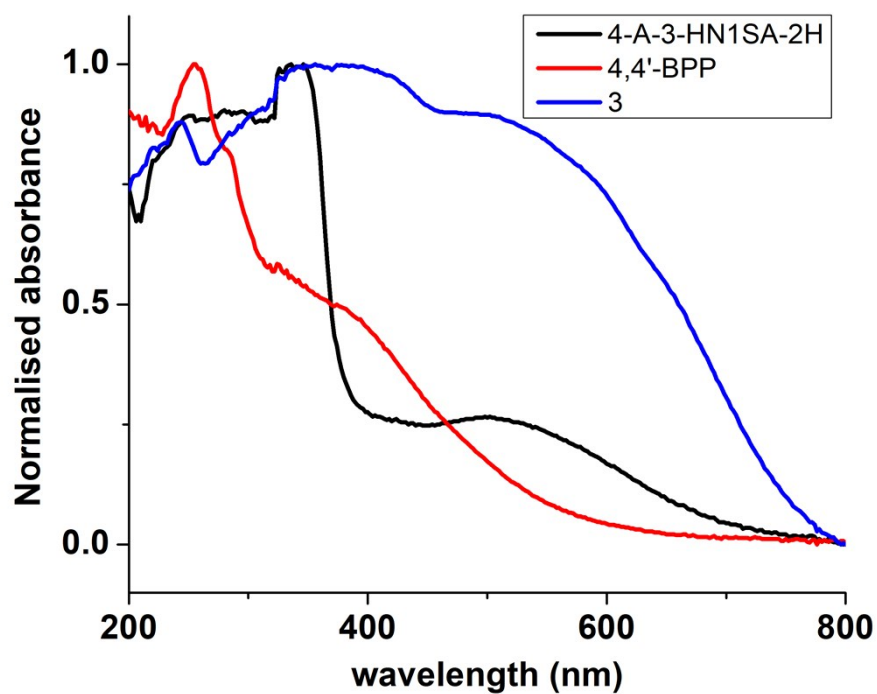


Fig. ESI-19 DR-UV-Vis spectra of starting materials and Compound 3.

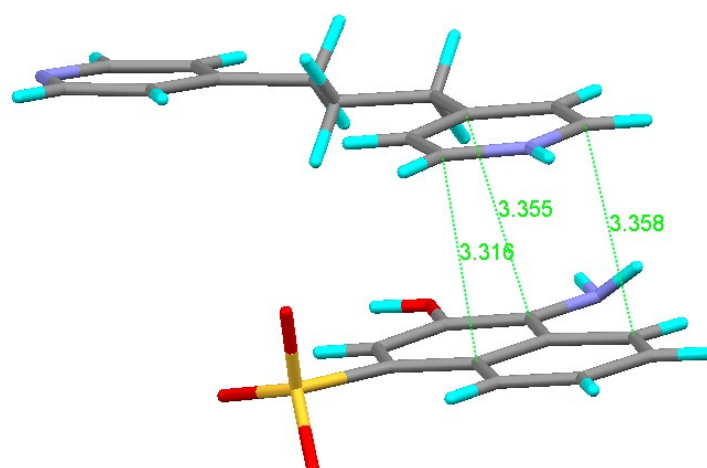


Fig. ESI-20 Strong π - π interactions between crystal formers of 3.

Table ESI-1. Comparative MPs of formers with molecular salts.

Acid former	MP. °C	Base former	MP. °C	Molecular salt	MP. °C
C₆H₇NO₃S	>300	C₁₃H₁₄N₂	60	1	150
C₉H₇NO₃S	>300	C₁₃H₁₄N₂	60	2	205
C₁₀H₇NO₄S	285	C₁₃H₁₄N₂	60	3	140

Table ESI-2. Solubility of compounds **1-3** in different solvents

Compound	CHCl₃	Acetone	MeOH	EtOH	CH₃CN	DCM	H₂O	DMSO
1	insoluble	insoluble	soluble	soluble	insoluble	partial	soluble	soluble
2	insoluble	insoluble	soluble	soluble	insoluble	partial	soluble	soluble
3	insoluble	insoluble	insoluble	insoluble	insoluble	insoluble	soluble	soluble