

**Thermal conversion of a pyridine solvate to a de-solvate facilitated by rearrangement of chalcogen bonds. The solvate and non-solvate structures of N-(2-nitro-4-(3-oxobenzo[d][1,2]selenazol-2(3H)-yl)phenyl)picolinamide**

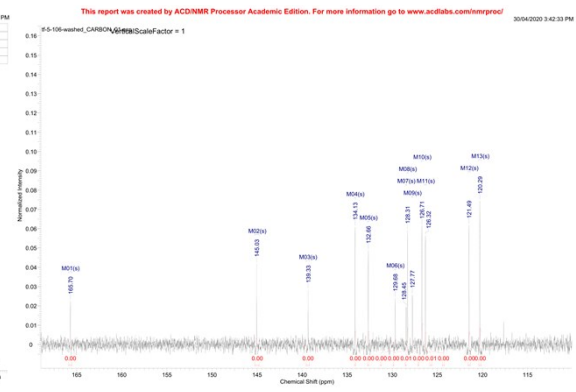
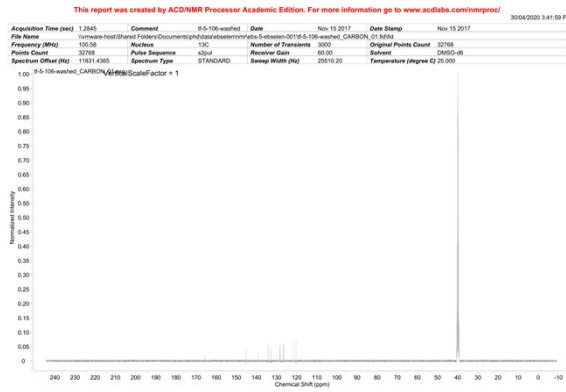
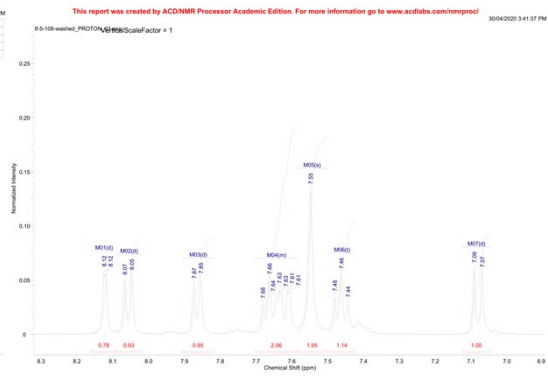
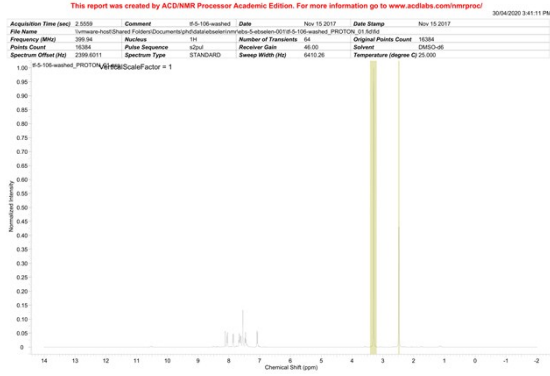
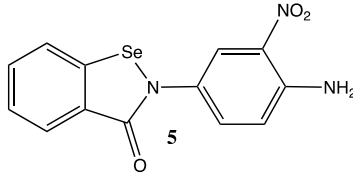
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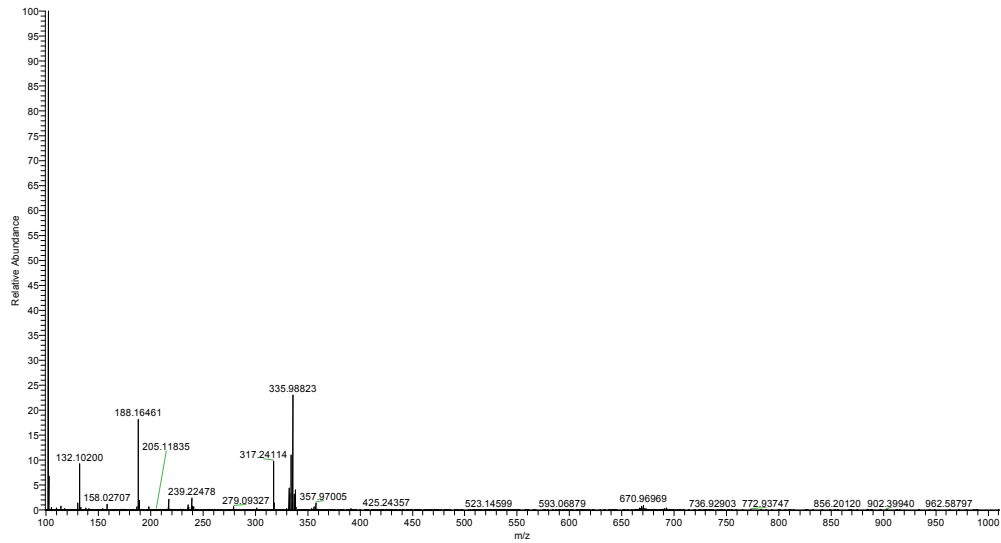
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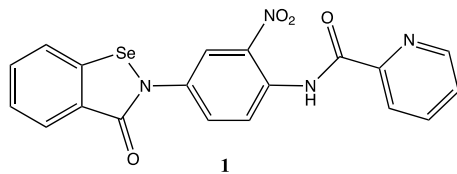
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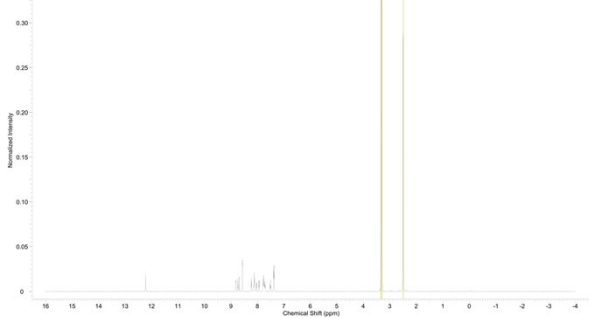




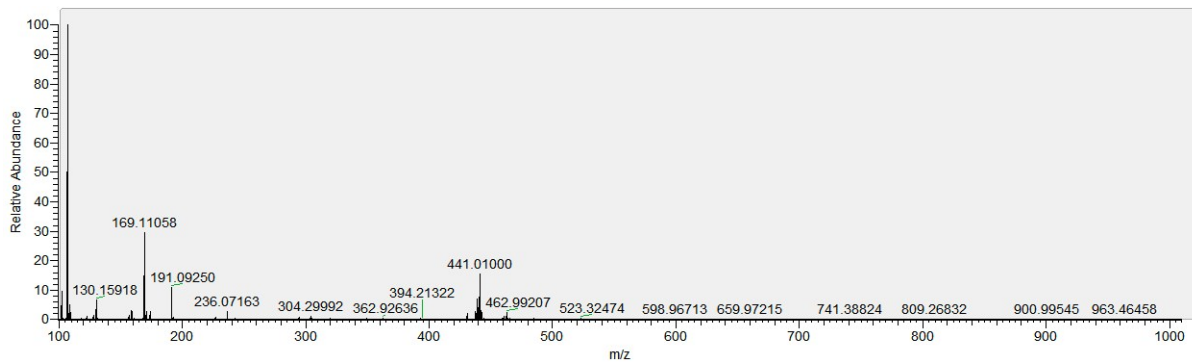
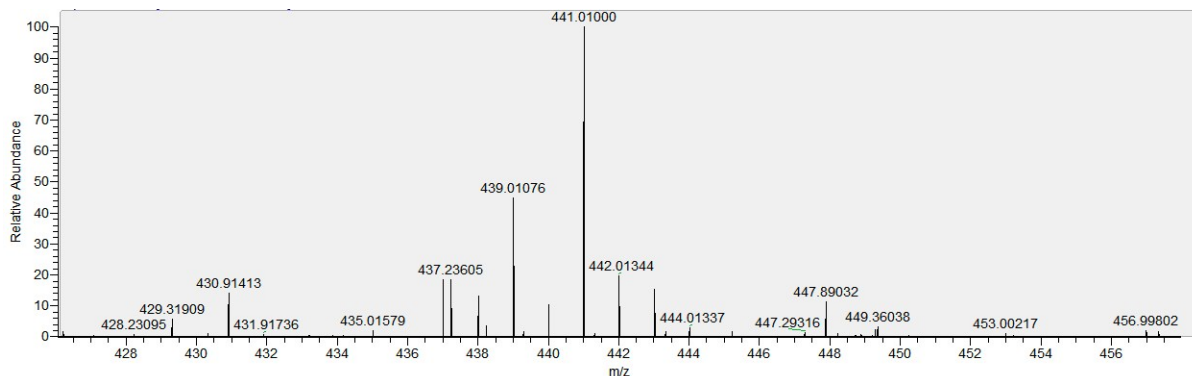
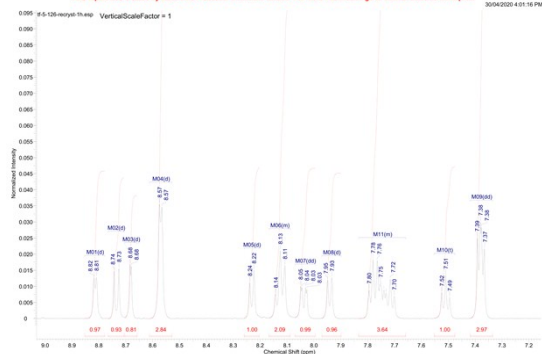
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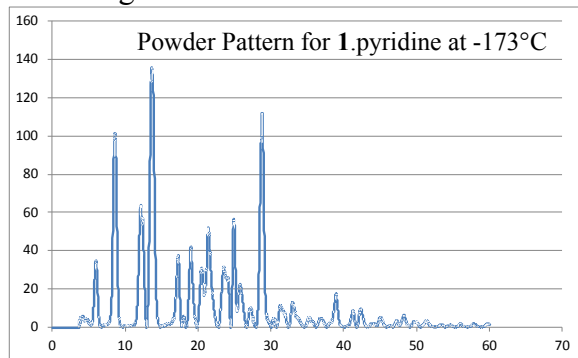
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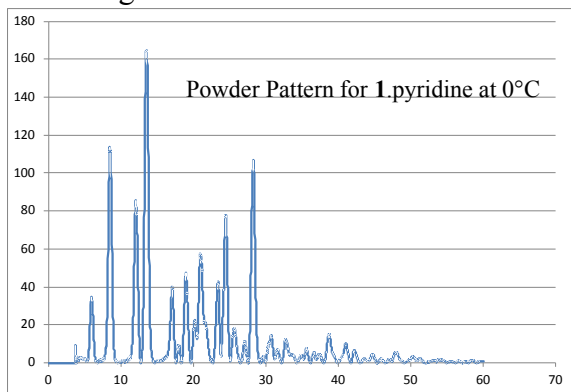
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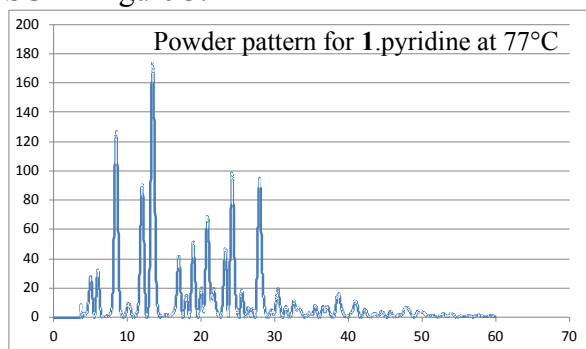
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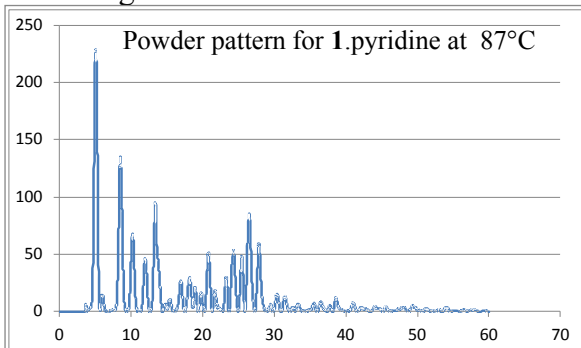
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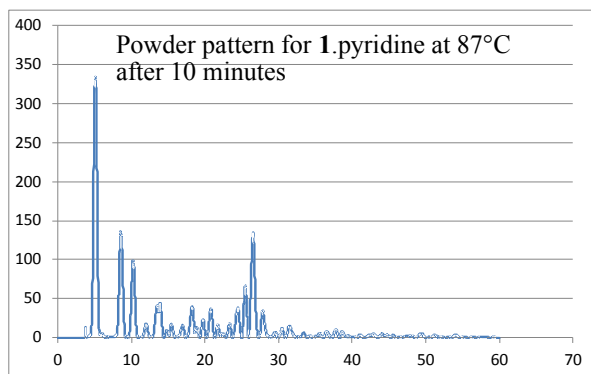
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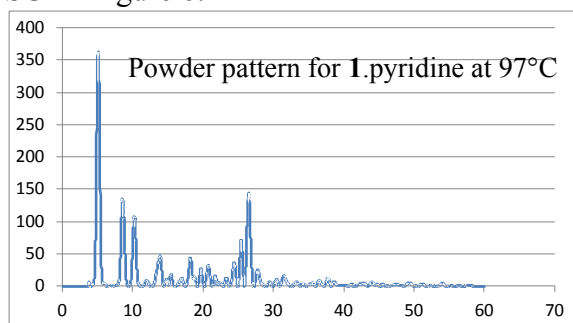
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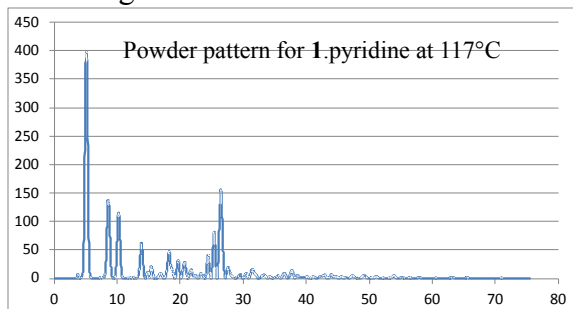
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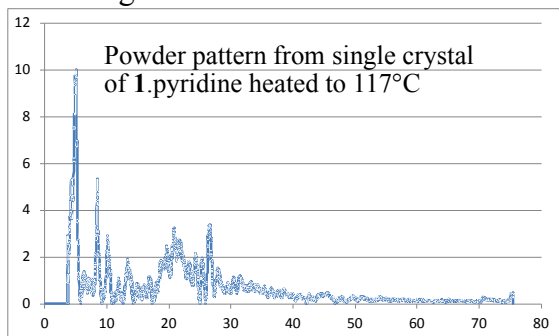
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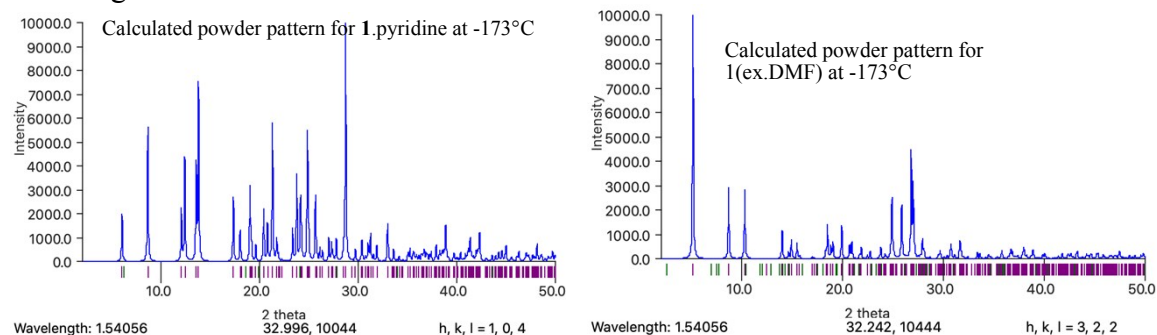
SUPP-Figure 7.



SUPP-Figure 8.



SUPP Figure 9



SUPP-Figure 10 Single crystal of **1**.Pyridine heated to 90°C. The faint powder rings were shown to be dues to formation of **1**(ex.DMF). See SUPP-Figure 8 for complete transformation at 113°C

