

**Electronic Supplementary Information**

**Cocrystallization of racemic amino acids with ZnCl<sub>2</sub>: an investigation of chiral selectivity upon coordination to the metal centre.**

Oleksii Shemchuk, Fabrizia Grepioni, Dario Braga\*

*Molecular Crystal Engineering Laboratory, Dipartimento di Chimica “G. Ciamician”, Università di Bologna, Via F. Selmi 2, 40126 Bologna, Italy, E-mail: dario.braga@unibo.it*

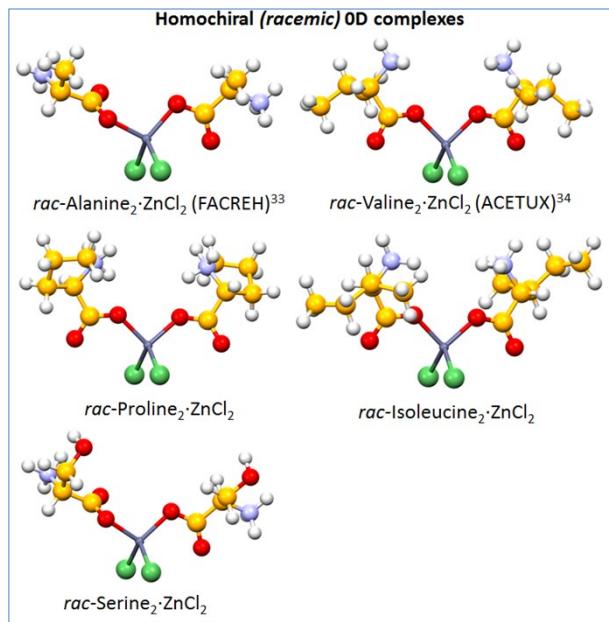
**Content**

<b>S1. Structural data</b>	p. 2
<b>S2. XRPD</b>	p. 4

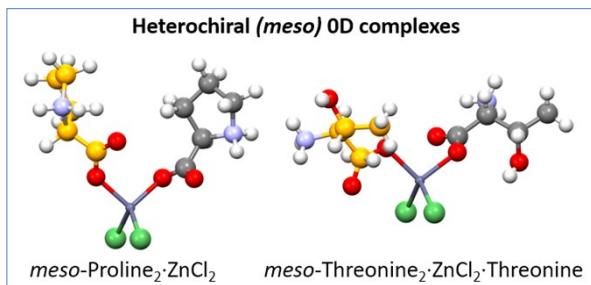
## S1. Structural data

**Table S-1.** Crystal data and details of measurements for all the complexes described in this work.

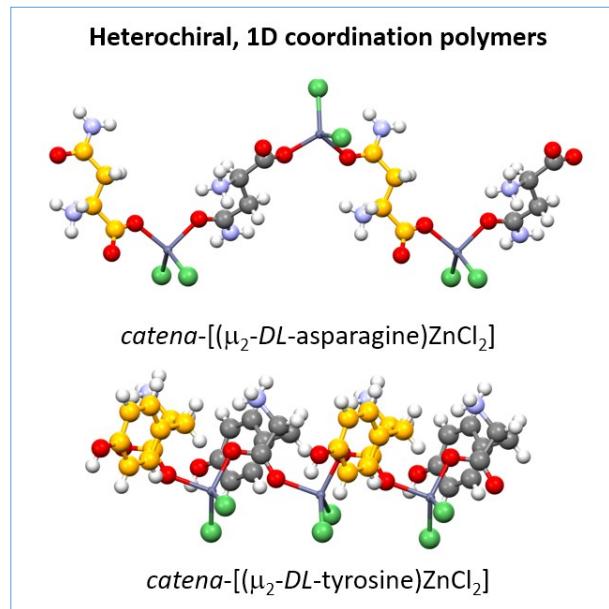
	<i>rac</i> -Isoleucine <sub>2</sub> ·ZnCl <sub>2</sub>	<i>meso</i> -Proline <sub>2</sub> ·ZnCl <sub>2</sub>	<i>rac</i> -Serine <sub>2</sub> ·ZnCl <sub>2</sub>	<i>meso</i> -Threonine <sub>2</sub> ·ZnCl <sub>2</sub> ·Threonine	<i>catena</i> -[(μ <sub>2</sub> -DL-Asparagine)ZnCl <sub>2</sub> ]	<i>catena</i> -[(μ <sub>2</sub> -DL-Tyrosine)ZnCl <sub>2</sub> ]	<i>rac</i> -Proline <sub>2</sub> ·ZnCl <sub>2</sub>	<i>rac</i> -Proline <sub>2</sub> ·ZnBr <sub>2</sub>
<b>Chemical formula</b>	C <sub>12</sub> H <sub>26</sub> Cl <sub>2</sub> N <sub>2</sub> O <sub>4</sub> Zn	C <sub>10</sub> H <sub>18</sub> Cl <sub>2</sub> N <sub>2</sub> O <sub>4</sub> Zn	C <sub>6</sub> H <sub>13</sub> Cl <sub>2</sub> N <sub>2</sub> O <sub>6</sub> Zn	C <sub>8</sub> H <sub>18</sub> Cl <sub>2</sub> N <sub>2</sub> O <sub>6</sub> Zn ·C <sub>4</sub> H <sub>9</sub> NO <sub>3</sub>	C <sub>4</sub> H <sub>8</sub> Cl <sub>2</sub> N <sub>2</sub> O <sub>3</sub> Zn	C <sub>9</sub> H <sub>11</sub> Cl <sub>2</sub> NO <sub>3</sub> Zn	C <sub>10</sub> H <sub>18</sub> Cl <sub>2</sub> N <sub>2</sub> O <sub>4</sub> Zn	C <sub>10</sub> H <sub>18</sub> Br <sub>2</sub> N <sub>2</sub> O <sub>4</sub> Zn
M <sub>r</sub> , g mol <sup>-1</sup>	394.62	366.53	345.45	493.63	268.39	317.46	366.53	455.45
T / K	293 (2)	293 (2)	293 (2)	293 (2)	293 (2)	293 (2)	293 (2)	293 (2)
<b>Crystal system</b>	Triclinic	Monoclinic	Monoclinic	Triclinic	Monoclinic	Monoclinic	Monoclinic	Monoclinic
<b>Space group</b>	P-1	P2 <sub>1</sub> /c	P2 <sub>1</sub> /c	P-1	P2 <sub>1</sub> /c	P2 <sub>1</sub> /n	C2/c	C2/c
a / Å	6.1647 (4)	8.7133 (3)	10.1190 (4)	9.2147 (5)	10.0109 (19)	8.8022 (10)	18.7127 (13)	18.6238 (11)
b / Å	12.6246 (7)	6.6937 (3)	13.5305 (7)	9.3042 (5)	6.8396 (8)	17.1083 (15)	5.9481 (6)	5.9591 (4)
c / Å	13.4501(8)	25.1323 (9)	8.9330 (4)	12.7858 (8)	13.3557 (16)	8.8138 (8)	13.6392 (11)	14.4189 (9)
α / °	62.666 (6)	90	90	110.722 (6)	90	90	90	90
β / °	84.162 (5)	94.159 (3)°	91.634 (4)	103.789 (5)	107.375 (15)	116.611 (13)	104.335 (8)	104.296 (6)
γ / °	89.311 (5)	90	90	90.346 (5)	90	90	90	90
V / Å <sup>3</sup>	924.37 (11)	1461.96 (10)	1222.57 (10)	990.79 (11)	872.7 (2)	1186.7 (2)	1470.8 (2)	1550.67 (17)
Z, Z'	2, 1	4, 1	4, 1	2, 1	4, 1	4, 1	4, 0.5	4, 0.5
d / mg cm <sup>-3</sup>	1.418	1.665	1.877	1.655	2.043	1.777	1.655	1.951
μ / mm <sup>-1</sup>	1.63	2.06	2.46	1.56	3.39	2.51	2.04	6.75
<b>Measd reflns</b>	7348	6952	5665	8166	3851	5109	3307	3594
<b>Indep reflns</b>	4238	3369	2773	4576	1995	2729	1711	1799
<b>Reflns with I &gt; 2σ(I)</b>	3308	2819	2100	3479	1371	1292	1139	1346
R <sub>int</sub>	0.028	0.040	0.047	0.037	0.047	0.080	0.046	0.025
R[F <sup>2</sup> > 2σ(F <sup>2</sup> )]	0.050	0.045	0.058	0.048	0.067	0.072	0.064	0.044
wR(F <sup>2</sup> )	0.124	0.099	0.140	0.146	0.189	0.203	0.164	0.107



**Figure S-1.** Homochiral complexation: the crystals contain complexes of DD and LL in racemic mixture. No conglomerate formation is observed.

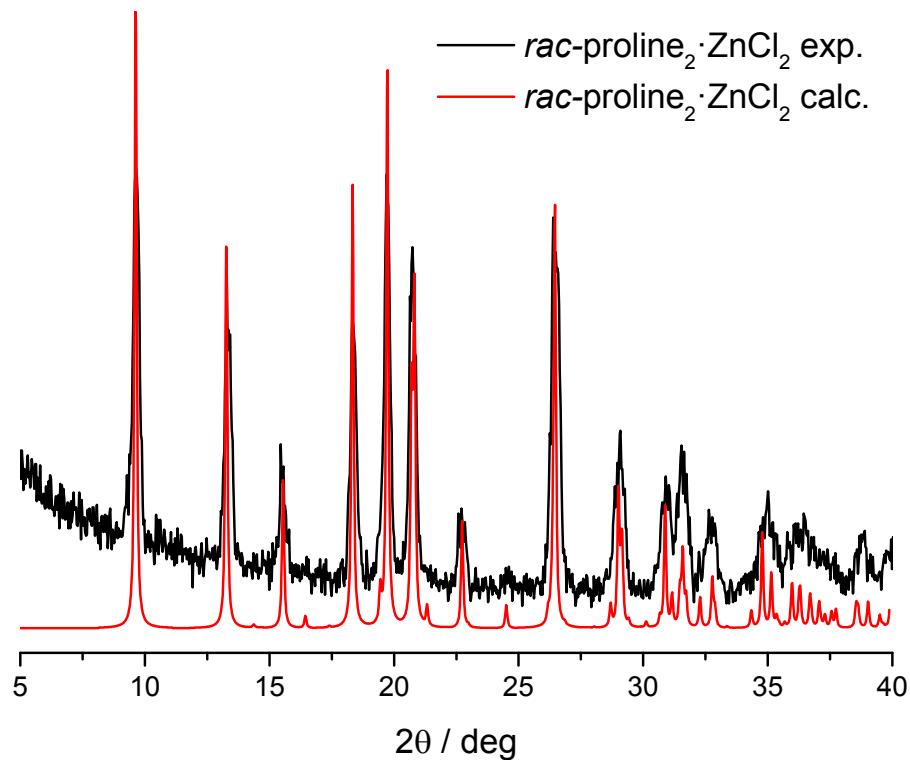


**Figure S-2.** Heterochiral complexes, which crystallize in *meso*-form.

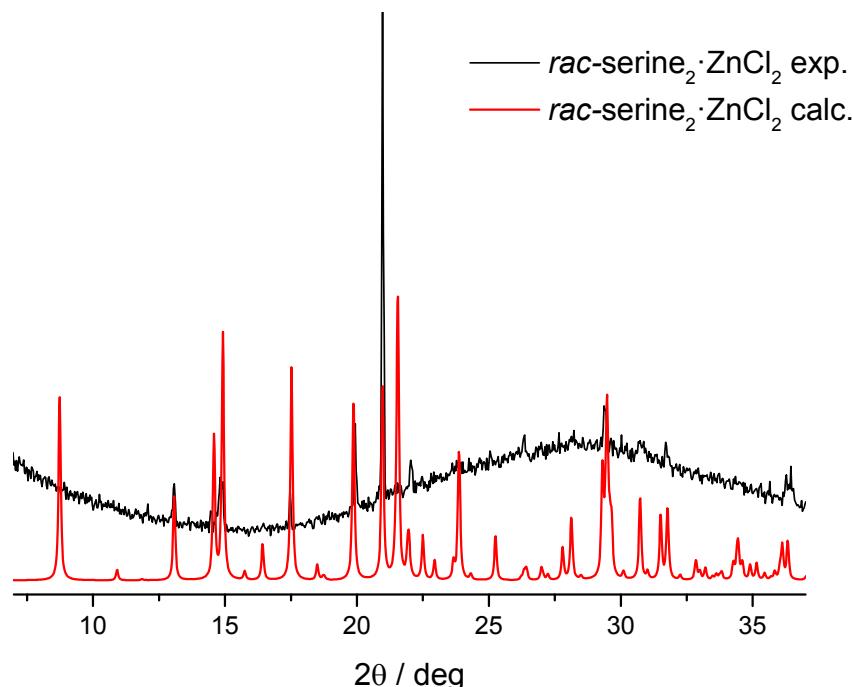


**Figure S-3.** The 1D polymers in *catena*-[(<math>\mu\_2</math>-*DL*-asparagine)ZnCl<sub>2</sub>] and *catena*-[(<math>\mu\_2</math>-*DL*-tyrosine)ZnCl<sub>2</sub>]: the chains contain alternating D and L amino acids.

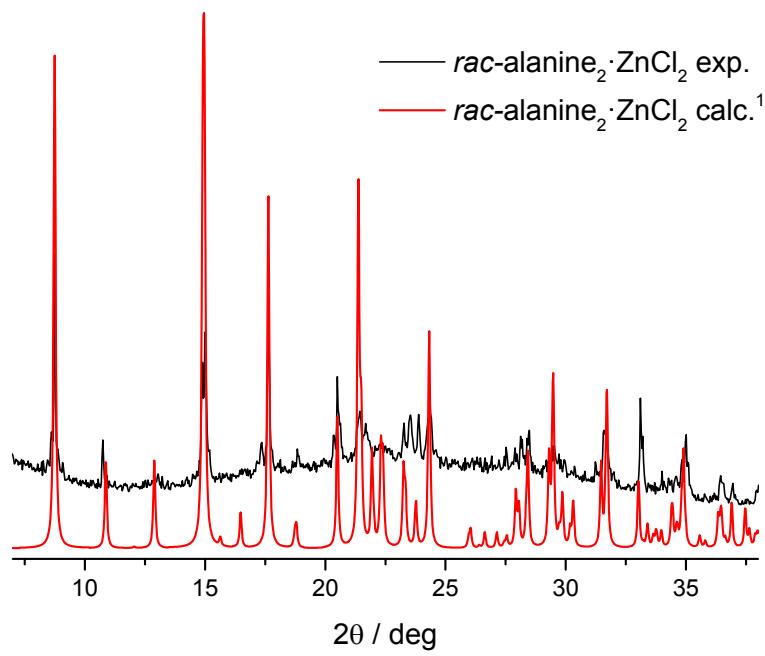
## S2. XRPD



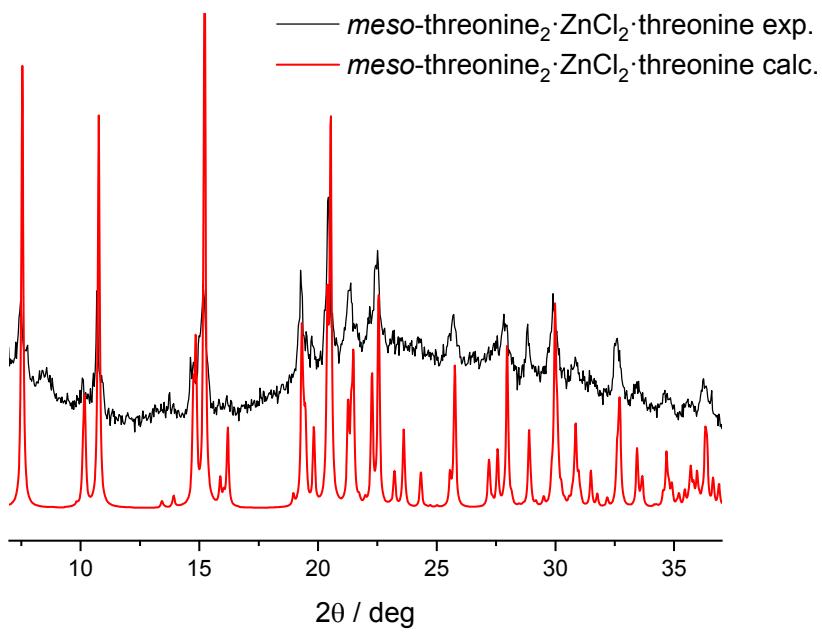
**Fig. S-4.** Comparison of (top, black line) the room temperature experimental XRPD patterns for *rac*-proline<sub>2</sub>·ZnCl<sub>2</sub>, obtained by slurry experiment in EtOH, and (bottom, red line) the pattern calculated on the basis of single crystal data.



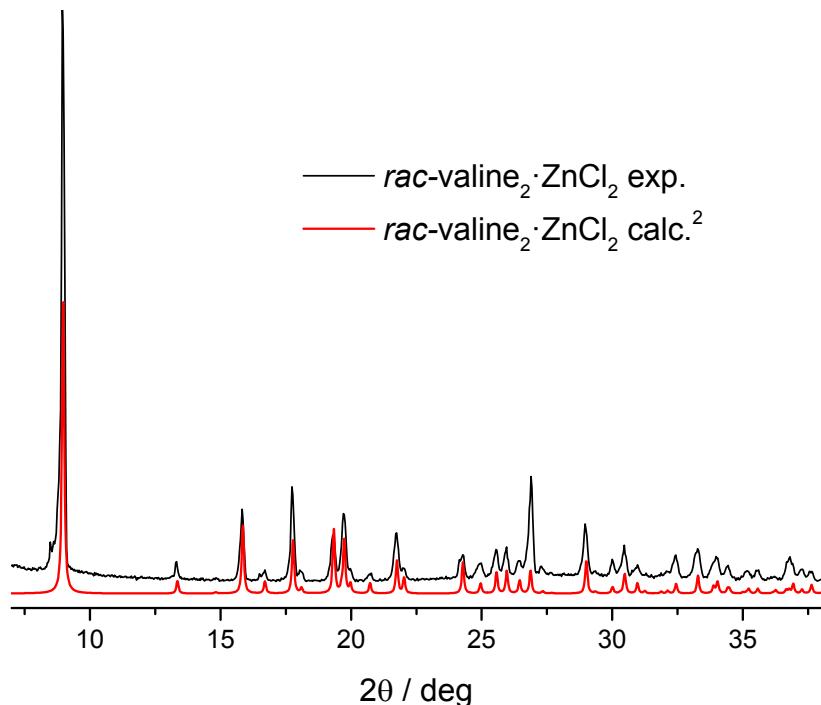
**Fig. S-5.** Comparison of the experimental (black line) and calculated (red line) XRPD patterns for *rac*-serine<sub>2</sub>·ZnCl<sub>2</sub>.



**Fig. S-6.** Comparison of the experimental (black line) and calculated (red line) XRPD patterns for *rac*-alanine<sub>2</sub>·ZnCl<sub>2</sub> [refcode FACREH<sup>1</sup>].



**Fig. S-7.** Comparison of the experimental (black line) and calculated (red line) XRPD patterns for meso-threonine<sub>2</sub>·ZnCl<sub>2</sub>·threonine.



**Fig. S-8.** Comparison of the experimental (black line) and calculated (red line) XRPD patterns for *rac*-valine<sub>2</sub>·ZnCl<sub>2</sub> [refcode ACETUX<sup>2</sup>].

#### References:

1. M. Subha Nandhini, R. V. Krishnakumar and S. Natarajan, *Acta Crystallogr. Sect. E: Struct. Rep. Online*, 2002, **58**, m127-m129.
2. M. S. Nandhini, R. V. Krishnakumar and S. Natarajan, *Acta Crystallogr. Sect. E: Struct. Rep. Online*, 2001, **57**, M498-M500.