

Electronic Supplementary Information (ESI) for

Multiresponsive chemosensing and photocatalytic properties of three luminescent coordination polymers derived from bifunctional 1,1'-Di(4-carbonylphenyl)-2,2'-biimidazole ligand

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Table S1 Crystal data and structure refinement parameters of **1–3**

CP	1	2	3
Formula	C ₂₀ H ₂₄ CdN ₄ O ₁₀	C ₆₃ H ₄₅ Cd ₄ Cl ₂ N ₁₃ O ₁₄	C ₂₀ H ₁₄ N ₄ O ₅ Zn
Formula weight	592.84	1728.66	455.75
Crystal system	Monoclinic	Hexagonal	Trigonal
Space group	<i>P</i> 2 ₁ / <i>c</i>	<i>P</i> 6 ₁	<i>R</i> -3
<i>a</i> (Å)	12.4871(10)	16.2489(13)	32.2321(19)
<i>b</i> (Å)	14.3971(12)	16.2489(13)	32.2321(19)
<i>c</i> (Å)	16.4209(10)	45.5303(17)	14.2264(11)
α (°)	90	90	90
β (°)	114.354(5)	90	90
γ (°)	90	120	120
<i>V</i> (Å ³)	2689.4(4)	10411(2)	12800(2)
<i>Z</i>	4	6	18
<i>D</i> _{calcd} (Mg/m ³)	1.464	1.637	1.022
μ (mm ⁻¹)	0.867	1.355	0.887
Temperature (K)	296(2)	291(2)	291(2)
<i>F</i> (000)	1200	5052	3996
<i>R</i> _{int}	0.0208	0.0062	0.0252
<i>R</i> ₁ [<i>I</i> > 2σ(<i>I</i>)] ^a	0.0421	0.0317	0.0383
w <i>R</i> ₂ [<i>I</i> > 2σ(<i>I</i>)] ^b	0.1283	0.1034	0.0923
Gof	1.075	1.032	1.091

$$R_1 = \frac{\sum ||F_o| - |F_c||}{\sum |F_o|}, \quad \omega R_2 = \frac{\sum [w(F_o^2 - F_c^2)^2]}{\sum [w(F_o^2)^2]}^{1/2}$$

Table S2 Selected bond lengths (Å) and angles (°) for 1–3

CP 1							
Cd(1)-O(1)	2.450(3)	Cd(1)-O(2)	2.323(3)	Cd(1)-O(5)	2.410(4)	Cd(1)-O(4)#1	2.237(3)
Cd(1)-N(4)#2	2.292(3)	Cd(1)-N(2)#3	2.352(3)				
O(2)-Cd(1)-O(1)	54.49(11)	O(5)-Cd(1)-O(1)	109.88(16)	O(2)-Cd(1)-O(5)	86.46(14)	N(2)#3-Cd(1)-O(1)	86.92(11)
N(4)#2-Cd(1)-O(1)	146.72(11)	O(4)#1-Cd(1)-O(1)	80.06(14)	N(2)#3-Cd(1)-O(5)	161.04(14)	N(4)#2-Cd(1)-O(5)	81.50(14)
O(4)#1-Cd(1)-O(5)	83.25(15)	O(2)-Cd(1)-N(2)#3	111.12(10)	N(4)#2-Cd(1)-N(2)#3	89.12(9)	O(4)#1-Cd(1)-N(2)#3	91.39(13)
N(4)#2-Cd(1)-O(2)	96.90(10)	O(4)#1-Cd(1)-O(2)	126.08(13)	O(4)#1-Cd(1)-N(4)#2	133.09(13)		
Symmetry codes: #1 -x+2,-y,-z+1; #2 -x+1,y-1/2,-z+1/2; #3 x,-y+1/2,z+1/2.							
CP 2							
Cd(1)-Cl(1)	2.459(2)	Cd(1)-O(5)	2.422(7)	Cd(1)-O(13)	2.262(6)	Cd(1)-O(10) ^{#4}	2.263(6)
Cd(1)-N(2)	2.301(6)	Cd(2)-Cl(2)	2.513(3)	Cd(2)-O(6)	2.170(6)	Cd(2)-O(9)	2.265(6)
Cd(2)-O(13)	2.274(6)	Cd(2)-N(3) ^{#5}	2.232(7)	Cd(3)-O(7)	2.305(5)	Cd(3)-O(8)	2.402(5)
Cd(3)-O(1) ^{#7}	2.302(7)	Cd(3)-O(2) ^{#7}	2.375(5)	Cd(3)-N(6) ^{#6}	2.230(6)	Cd(3)-N(11) ^{#3}	2.264(6)
Cd(4)-N(10)	2.282(7)	Cd(4)-O(3) ^{#6}	2.392(5)	Cd(4)-O(4) ^{#6}	2.354(6)	Cd(4)-O(11) ^{#9}	2.283(5)
Cd(4)-O(12) ^{#9}	2.392(5)	Cd(4)-N(7) ^{#8}	2.284(6)				
O(10) ^{#4} -Cd(1)-O(13)	87.4(2)	O(10) ^{#4} -Cd(1)-N(2)	94.6(2)	O(13)-Cd(1)-N(2)	176.2(2)	O(10) ^{#4} -Cd(1)-O(5)	81.9(2)
O(13)-Cd(1)-O(5)	82.6(2)	N(2)-Cd(1)-O(5)	94.4(2)	O(10) ^{#4} -Cd(1)-Cl(1)	160.62(19)	O(13)-Cd(1)-Cl(1)	80.51(15)
N(2)-Cd(1)-Cl(1)	96.69(17)	O(5)-Cd(1)-Cl(1)	81.59(15)	O(6)-Cd(2)-N(3) ^{#5}	104.0(2)	O(6)-Cd(2)-O(9)	97.1(2)
N(3) ^{#5} -Cd(2)-O(9)	82.4(2)	O(6)-Cd(2)-O(13)	111.1(2)	N(3) ^{#5} -Cd(2)-O(13)	101.9(2)	O(9)-Cd(2)-O(13)	149.2(2)
O(6)-Cd(2)-Cl(2)	90.42(17)	N(3) ^{#5} -Cd(2)-Cl(2)	164.27(17)	O(9)-Cd(2)-Cl(2)	89.68(16)	O(13)-Cd(2)-Cl(2)	78.24(16)
N(6) ^{#6} -Cd(3)-N(11) ^{#3}	106.2(3)	N(6) ^{#6} -Cd(3)-O(7)	102.1(2)	N(11) ^{#3} -Cd(3)-O(7)	111.2(2)	N(6) ^{#6} -Cd(3)-O(1) ^{#7}	104.0(3)
N(11) ^{#3} -Cd(3)-O(1) ^{#7}	83.7(2)	O(7)-Cd(3)-O(1) ^{#7}	144.7(2)	N(6) ^{#6} -Cd(3)-O(2) ^{#7}	85.2(2)	N(11) ^{#3} -Cd(3)-O(2) ^{#7}	137.9(2)
O(7)-Cd(3)-O(2) ^{#7}	105.39(19)	O(1) ^{#7} -Cd(3)-O(2) ^{#7}	54.2(2)	N(6) ^{#6} -Cd(3)-O(8)	158.2(2)	N(11) ^{#3} -Cd(3)-O(8)	85.9(2)
O(7)-Cd(3)-O(8)	56.22(16)	O(1) ^{#7} -Cd(3)-O(8)	95.1(2)	O(2) ^{#7} -Cd(3)-O(8)	97.95(17)	N(7) ^{#8} -Cd(4)-N(10)	102.8(2)
N(7) ^{#8} -Cd(4)-O(11) ^{#9}	108.6(2)	N(10)-Cd(4)-O(11) ^{#9}	100.8(2)	N(7) ^{#8} -Cd(4)-O(4) ^{#6}	83.0(2)	N(10)-Cd(4)-O(4) ^{#6}	100.5(2)
O(11) ^{#9} -Cd(4)-O(4) ^{#6}	152.80(18)	N(7) ^{#8} -Cd(4)-O(12) ^{#9}	90.89(19)	N(10)-Cd(4)-O(12) ^{#9}	155.9(2)	O(11) ^{#9} -Cd(4)-O(12) ^{#9}	55.64(19)
O(4) ^{#6} -Cd(4)-O(12) ^{#9}	100.86(18)	N(7) ^{#8} -Cd(4)-O(3) ^{#6}	138.4(2)	N(10)-Cd(4)-O(3) ^{#6}	85.7(2)	O(11) ^{#9} -Cd(4)-O(3) ^{#6}	109.57(19)
O(4) ^{#6} -Cd(4)-O(3) ^{#6}	55.50(17)	O(12) ^{#9} -Cd(4)-O(3) ^{#6}	97.15(18)				
Symmetry codes: #3 y+1,-x+y+1,z-1/6; #4 y,-x+y,z-1/6; #5 x-y,x,z+1/6; #6 x-y+1,x,z+1/6; #7 -y+1,x-y+1,z+1/3; #8 -y+1,x-y,z+1/3; #9 x-y,x-1,z+1/6.							
CP 3							
O(1)-Zn(1)	1.9031(15)	N(2)-Zn(1)#1	2.0291(18)	N(4)-Zn(1)#2	1.9983(17)	O(3)-Zn(1)#3	1.8885(16)
O(3)#3-Zn(1)-O(1)	117.82(8)	O(3)#3-Zn(1)-N(4)#4	106.07(8)	O(1)-Zn(1)-N(4)#4	117.40(6)	O(3)#3-Zn(1)-N(2)#5	107.80(7)
O(1)-Zn(1)-N(2)#5	101.11(7)	N(4)#4-Zn(1)-N(2)#5	105.50(7)				
Symmetry codes: #1 -x+y-1/3,-x+1/3,z+1/3; #2 x-y+2/3,x+1/3,-z+1/3; #3 -x,-y+1,-z+1; #4 y-1/3,-x+y+1/3,-z+1/3; #5 -y+1/3,x-y+2/3,z-1/3.							

Table S3 Standard deviation and detection limit calculation for Fe³⁺, Cr₂O₇²⁻ and NZF in 1

	Fe ³⁺	Cr ₂ O ₇ ²⁻	NZF
1	597.223388	597.137779	609.978518
2	597.525673	597.251263	609.631232
3	597.021232	597.457365	609.593265
4	597.441231	597.034332	610.012316
5	597.282718	597.001232	609.891231
Standard deviation (σ)	0.17592	0.16547	0.17563
Ksv	1.41×10 ⁴	1.58×10 ⁴	3.78×10 ⁴
Detection limit (3σ/Ksv)	3.74×10 ⁻⁵	3.14×10 ⁻⁵	1.39×10 ⁻⁵

Table S4 Standard deviation and detection limit calculation for Fe³⁺, Cr₂O₇²⁻ and NZF in 2

	Fe ³⁺	Cr ₂ O ₇ ²⁻	NZF
1	502.762131	520.508999	521.254927
2	502.531144	520.421232	521.373216
3	502.827535	520.671313	521.094313
4	502.914577	520.831752	521.512132
5	502.521374	520.412266	521.136545
Standard deviation (σ)	0.15871	0.16094	0.15369
Ksv	1.14×10 ⁴	1.96×10 ⁴	2.87×10 ⁴
Detection limit (3σ/Ksv)	4.18×10 ⁻⁵	2.46×10 ⁻⁵	1.61×10 ⁻⁵

Table S5 Standard deviation and detection limit calculation for Fe³⁺, Cr₂O₇²⁻ and NZF in **3**

	Fe ³⁺	Cr ₂ O ₇ ²⁻	NZF
1	439.068312	450.361476	449.693751
2	439.214273	450.413143	449.603243
3	438.912313	450.291399	449.814321
4	439.212645	450.466366	449.513354
5	439.308066	450.781232	449.923171
Standard deviation (σ)	0.13854	0.16942	0.14605
Ksv	1.38×10^4	1.35×10^4	2.20×10^4
Detection limit ($3\sigma/Ksv$)	3.01×10^{-5}	3.76×10^{-5}	1.99×10^{-5}

Table S6 Comparison of various CPs sensors for the detection of Fe³⁺, Cr₂O₇²⁻ and NZF.

	Analyte	CPs-based fluorescent Materials	Quenching constant (K _{SV} , M ⁻¹)	Detection Limits (DL)	Media	Ref
1	Fe ³⁺	[H ₂ N(CH ₃) ₂] ₂ [Zn ₂ L(HPO ₃) ₂]	3.96 × 10 ⁵	1.16 × 10 ⁻⁴ mM	H ₂ O	21
2		{[Tb(Cmdep)(H ₂ O) ₃] ₂ (NO ₃) ₂ ·5H ₂ O} _n	5532	1.5 mM	H ₂ O	22
3		{[Cu ^I ₂ (tpta) ₂][Cu ^{II} (bptc)]·3H ₂ O·DMF} _n	3.817 × 10 ³	2.59 μM	H ₂ O	23
4		{[Co ₄ (timb) ₂ (Br-IPA) ₄]·5H ₂ O} _n	1.79 × 10 ⁴	3.01 × 10 ⁻⁵ M	H ₂ O	24
5		[Tb(HMDIA)(H ₂ O) ₃]·H ₂ O	1.73 × 10 ⁴		H ₂ O	25(a)
6		{[Zn(oba)(L) _{0.5}]-dma} _n	9.3 × 10 ³		DMF/ H ₂ O	25(b)
7		{[Cd ₂ (SA) ₂ (L) ₂]·H ₂ O} _n	2.1 × 10 ⁴	2.4 × 10 ⁻⁶ M	DMF/ H ₂ O	25(c)
	{[Cd(CDC)(L)]} _n	4.9 × 10 ³	7.4 × 10 ⁻⁵ M			
1	Cr ₂ O ₇ ²⁻	[Cd ₂ (L ₁)(1,4-NDC) ₂] _n	5.86 × 10 ⁴	0.031 ppm	H ₂ O	26
2		[Zr ₆ O ₄ (OH) ₈ (H ₂ O) ₄ (TCPP) ₄] □9DMF□3.5H ₂ O	5.91 × 10 ⁴		H ₂ O	27
3		{[Zn(H ₂ BCA)(m-bib)]·H ₂ O} _n	5.3 × 10 ⁴	0.07 μM	H ₂ O	28
4		[Zn ₅ (TDA) ₄ (TZ) ₄]·4DMF] _n	6.77 × 10 ³		H ₂ O	29
5		[Zn(NH ₂ -bdc)(4,4'-bpy)]	7.62 × 10 ³	1.30 μM	H ₂ O	30
6		[Cd _{1.5} (L) ₂ (bpy)(NO ₃) ₂]·2DMF·2H ₂ O	5.42 × 10 ⁴	320 ppb	H ₂ O	31
1		{[Cd ₃ (TDCPB)·2DMAc]·DMAc·4H ₂ O} _n	7.46 × 10 ⁴		DMAc	32
2		{[Tb(TATMA)(H ₂ O)·2H ₂ O] _n	3.00 × 10 ⁴		H ₂ O	33
3		[Zn(L) ₂]·CH ₂ Cl ₂ ·CH ₃ OH	1.62 × 10 ⁴		CH ₃ OH	34
4		[Cd(tptc) _{0.5} (o-bimb)] _n	4.4 × 10 ⁴		DMF	35
5		[Cd(H ₂ tptc) _{0.5} (mbimb)(Cl)] _n	2.1 × 10 ⁵			
6		[Zn ₂ (azdc) ₂ (dpta)]·(DMF) ₄	1.30 × 10 ⁵	0.63 ppm	DMF	36

H₃CmdepBr = N-carboxymethyl-(3,5-dicarboxyl)pyridinium bromide;

tpta = tris(4-(1,2,4-triazol-1-yl)phenyl)amine, H₄bptc = 3,3',4,4'-biphenyltetracarboxylic acid;

timb=1,3,5-tris(2-methylimidazol-1-yl)benzene;

H₄MDIA= 5,5'-methylenediisophthalic acid;

L = 3,3'-azodipyridine, H₂oba = 4,4'-oxydibenzoic acid;

H₂SA = succinic acid, H₂CDC = 1,4-Cyclohexanedicarboxylic acid, L = [3,3'-azobis(pyridine)];

L1 = 1,4-bis(benzimidazol-1-yl)-2-butylene, 1,4-H₂NDC = 1,4-naphthalenedicarboxylic acid;

H₄TCPP=2,3,5,6-tetrakis(4-carboxyphenyl)pyrazine;

H₂BCA= bis(4-carboxybenzyl)amine, m-bib = 1,3-bis(1-imidazolyl)benzene;

H₂TDA = thiophene-2,5-dicarboxylic acid, HTZ = 1H-1,2,4-Triazole;

NH₂-H₂bdc = 2-amino-1,4-benzenedicarboxylic acid, 4,4'-bpy = 4,4'-bipyridine;

HL = 4-(4-carboxyphenyl)-1,2,4-triazole, bpy = 4,4'-bipyridine;

H₆TDCPB = 1,3,5-tris[3,5-bis(3-carboxylphenyl-1-yl)phenyl-1-yl]benzene;

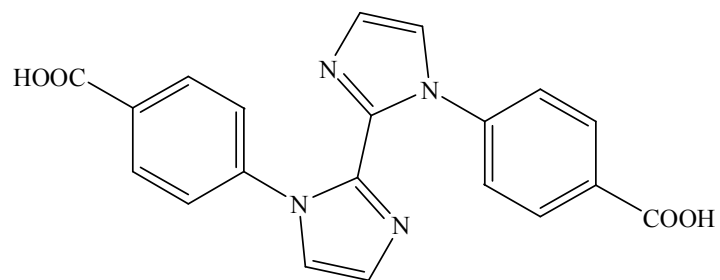
H₃TATMA = 4, 4',4''-s-triazine-1,3,5-triyltri-m-aminobenzoate

HL = 2-hydroxy-4-(pyridin-4-yl)benzaldehyde

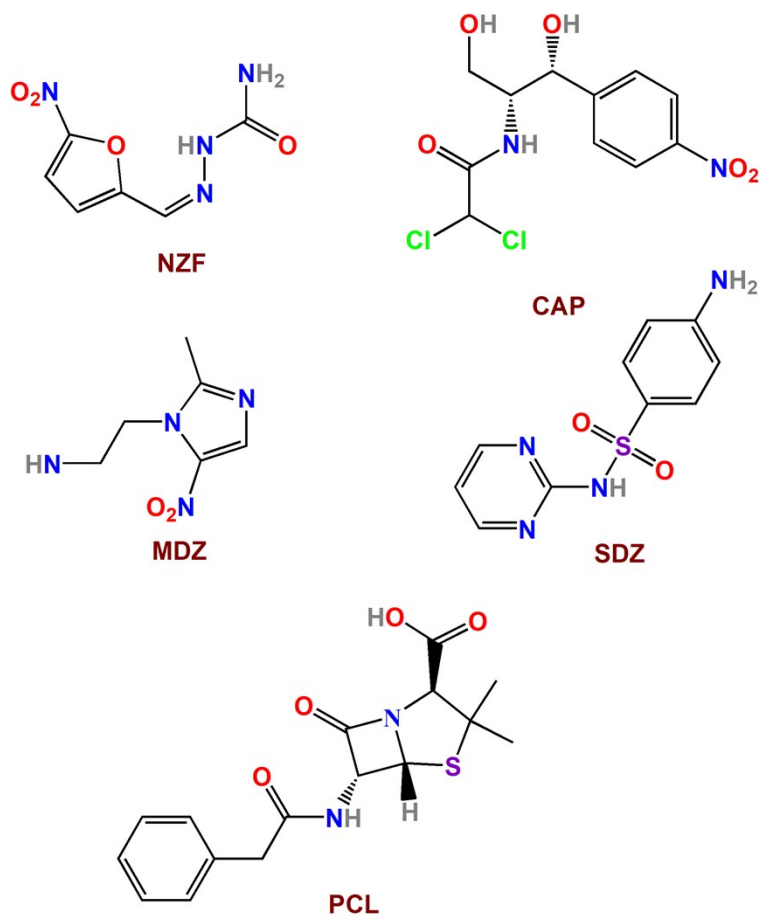
H₄tptc = p-terphenyl-2,2'',5'',5'''-tetracarboxylate acid

bimb = ortho/meta-bis(imidazol-1-ylmethyl)benzene

H₂azdc = Azobenzene-4,4'-dicarboxylic Acid



Scheme S1. The structure of 1,1'-di(4-carboxyphenyl)-2,2'-biimidazole (H_2L)



Scheme S2. The structures of selected antibiotics

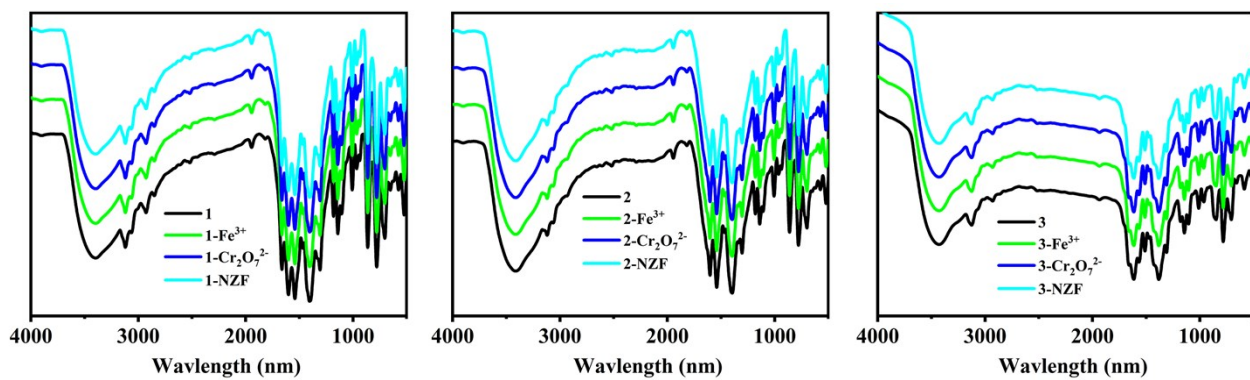


Fig. S1 FT-IR spectra of 1-3 and 1-3 recovered from Fe^{3+} , $\text{Cr}_2\text{O}_7^{2-}$ and NZF



Fig. S2 The binuclear $[\text{Cd}_2(\text{BIM})_2]$ SBU in **1**

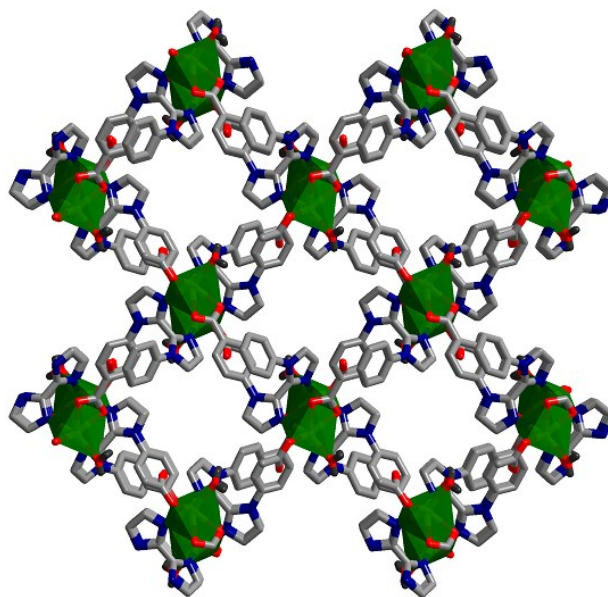


Fig. S3 The 3D porous network with the 1D circular channels of **1**

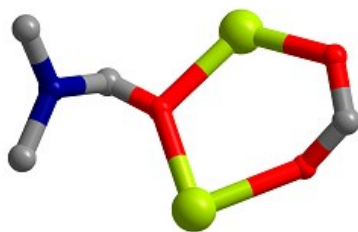


Fig. S4 The binuclear $[\text{Cd}_2(\text{COO})_2(\text{DMF})]$ cluster in **2**

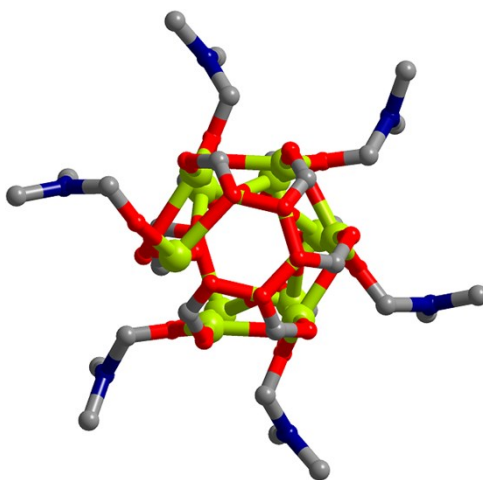


Fig. S5 The 1D $[\text{Cd}_2(\text{COO})_2(\text{DMF})]_n$ chain with 6-fold helix in **2**

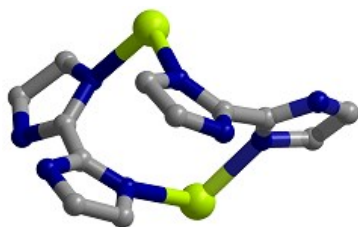


Fig. S6 The binuclear $[\text{Cd}_2(\text{BIM})_2]$ SBU in **2**

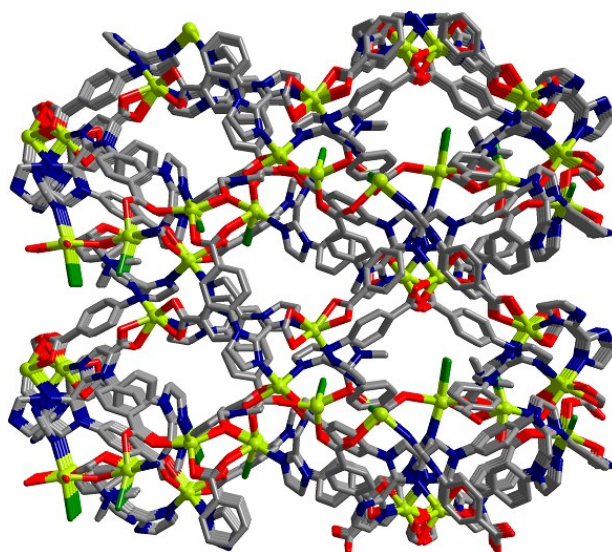


Fig. S7 The 3D porous network of **2**

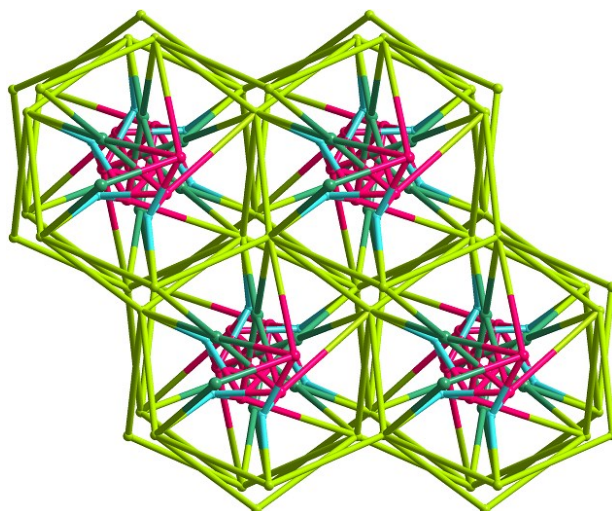


Fig. S8 The 3D 4-nodal (3,4,5,8)-c $\{4.5^2\} \{4^3.5^3\} \{4^3.5^7\} \{4^5.5^{11}.6^7.7^4.8\}$ net of **2**

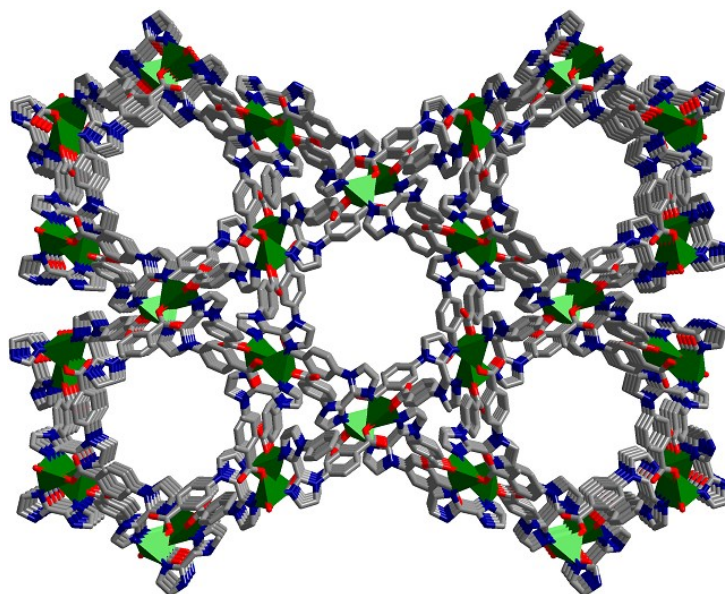


Fig. S9 The 3D porous framework containing the 1D circle channels of 3

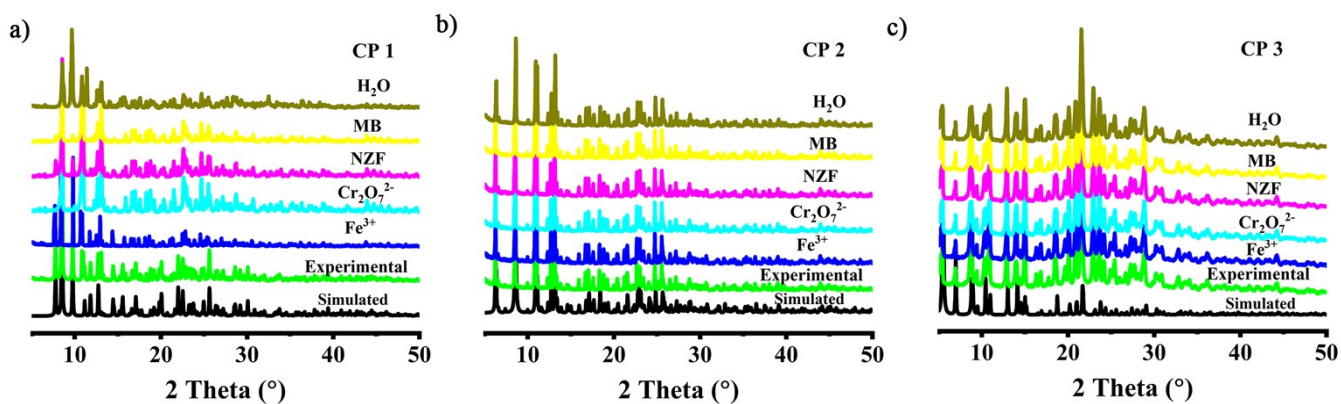


Fig. S10 PXR D patterns of 1-3

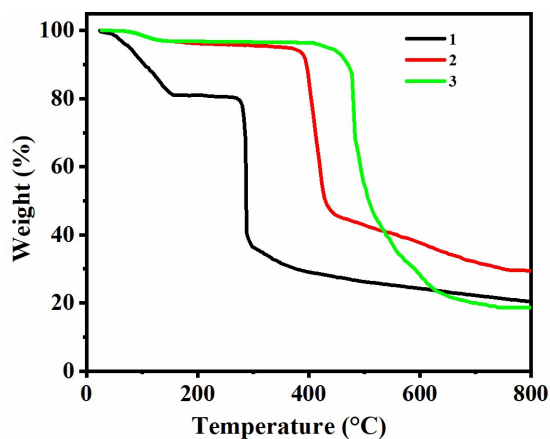


Fig. S11 TGA curves for 1-3

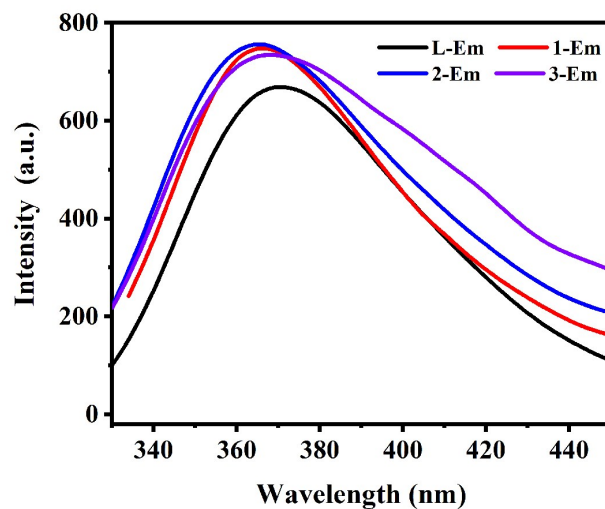


Fig. S12 The solid state diffuse-reflectance spectra for 1, 2, 3 and H₂L ligand

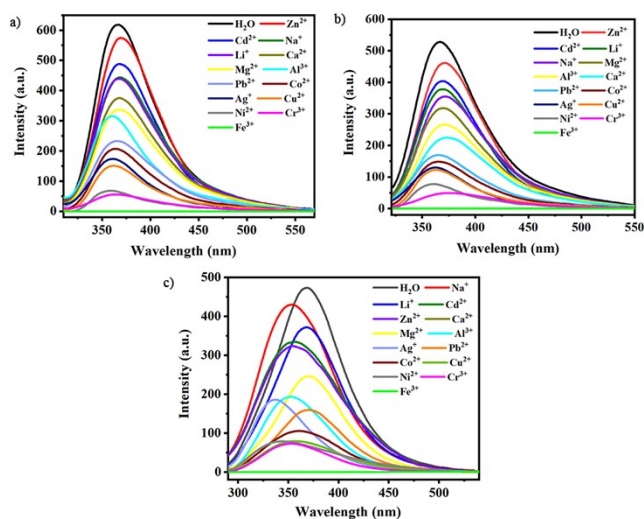


Fig. S13 Fluorescence response of 1(a), 2(b) and 3(c) toward different metal cations in H₂O solution

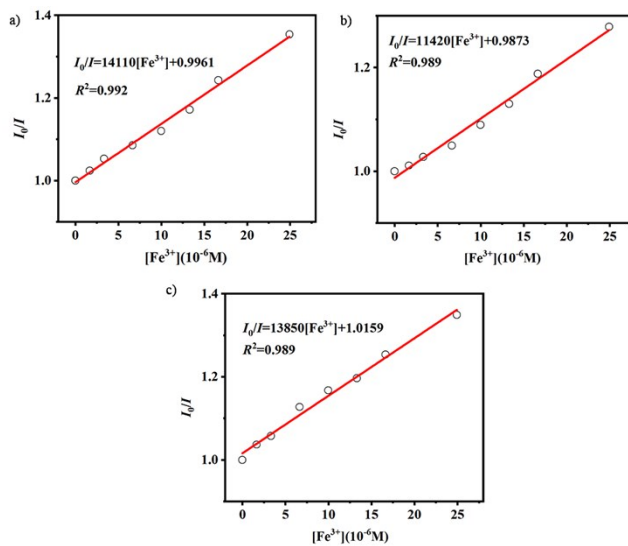


Fig. S14 The stern–Volmer plots of Fe³⁺ in 1(a), 2(b) and 3(c).

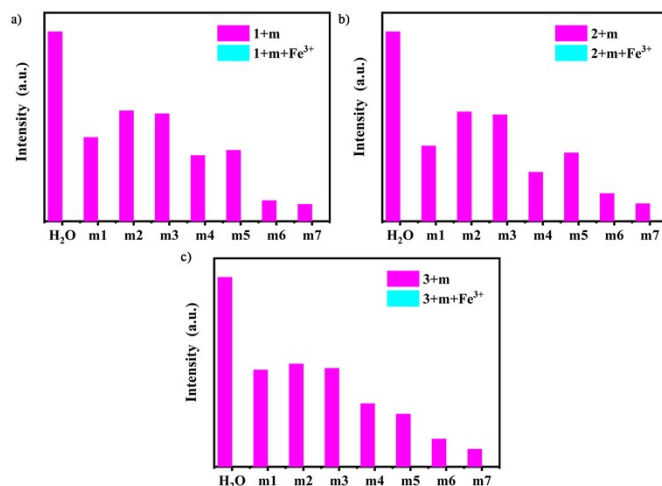


Fig. S15 Luminescence intensity of **1(a)**, **2(b)** and **3(c)** with different cations solution added Fe³⁺ ions (10⁻² M) (m1: Ag⁺/Na⁺/Co²⁺; m2: Li⁺/Ni²⁺/Zn²⁺; m3: Mg²⁺/Pb²⁺/Cd²⁺; m4: Cr³⁺/Ca²⁺; m5: Al³⁺/Cu²⁺; m6: Ni²⁺; m7: Cr³⁺).

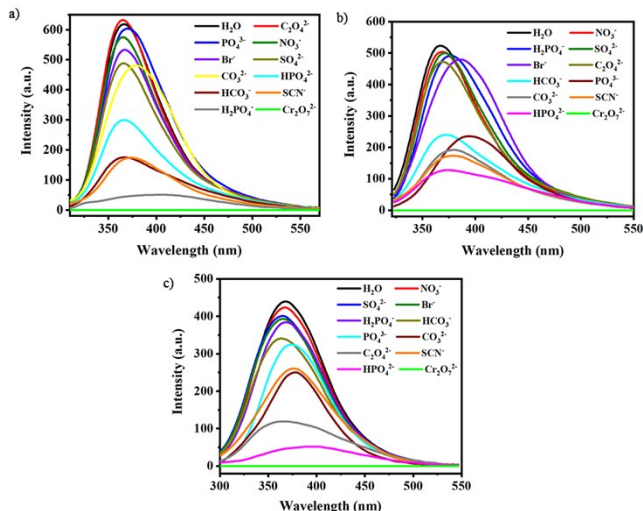


Fig. S16 Fluorescence response of **1(a)**, **2(b)** and **3(c)** toward different anions in H₂O solution.

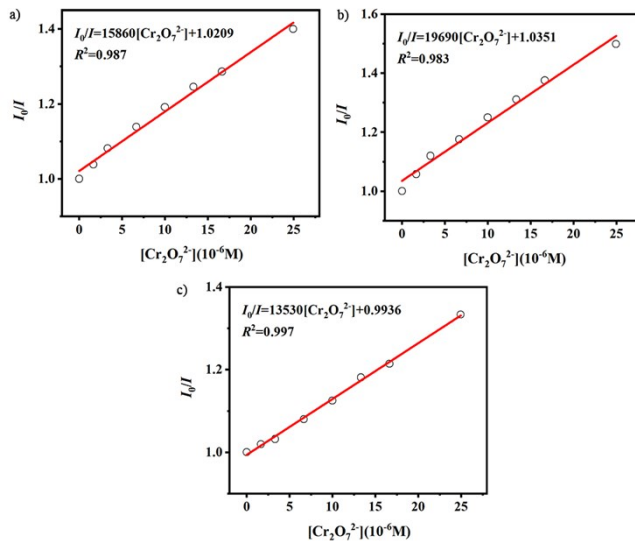


Fig. S17 The stern–Volmer plots of Cr₂O₇²⁻ in **1(a)**, **2(b)** and **3(c)**.

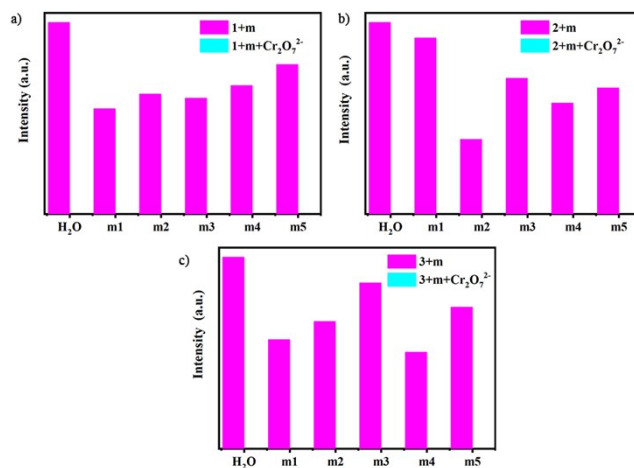


Fig. S18 Luminescence intensity of **1**(a), **2**(b) and **3**(c) with different mixed cations solution added $\text{Cr}_2\text{O}_7^{2-}$ ions (10^{-2} M) (m1: $\text{C}_2\text{O}_4^{2-}/\text{H}_2\text{PO}_4^-$; m2: $\text{PO}_4^{3-}/\text{SCN}^-$; m3: $\text{NO}_3^-/\text{HCO}_3^-$; m4: $\text{Br}^-/\text{HPO}_4^{2-}$; m5: $\text{SO}_4^{2-}/\text{CO}_3^{2-}$).

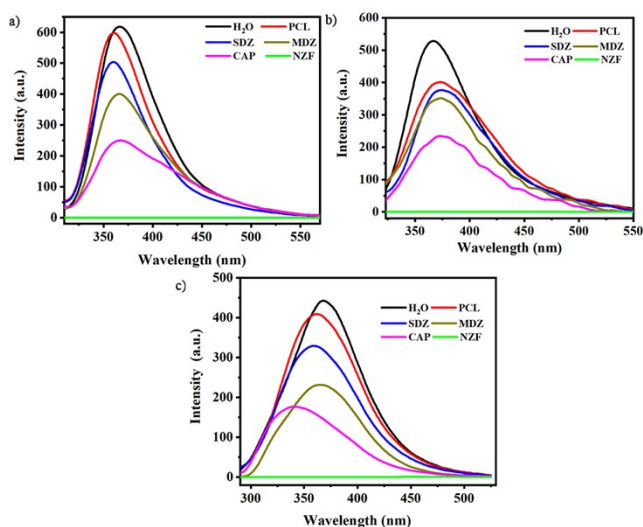


Fig. S19 Fluorescence response of **1**(a), **2**(b) and **3**(c) toward different antibiotics in H₂O solution.

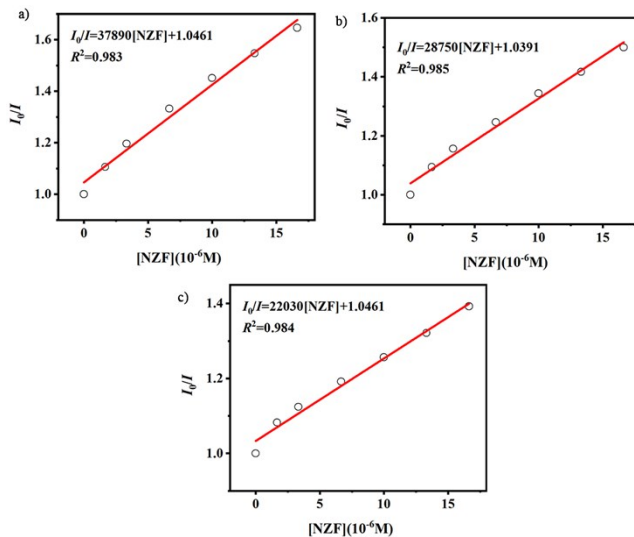


Fig. S20 The stern-Volmer plots of NZF in **1**(a), **2**(b) and **3**(c).

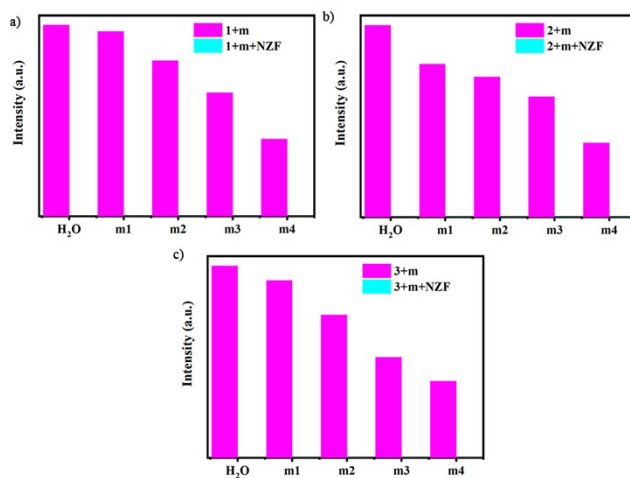


Fig. S21 Luminescence intensity of **1**(a), **2**(b) and **3**(c) with different mixed cations solution added NZF (10⁻² M) (m1: PCL; m2: SDZ; m3: MDZ; m4: CAP).

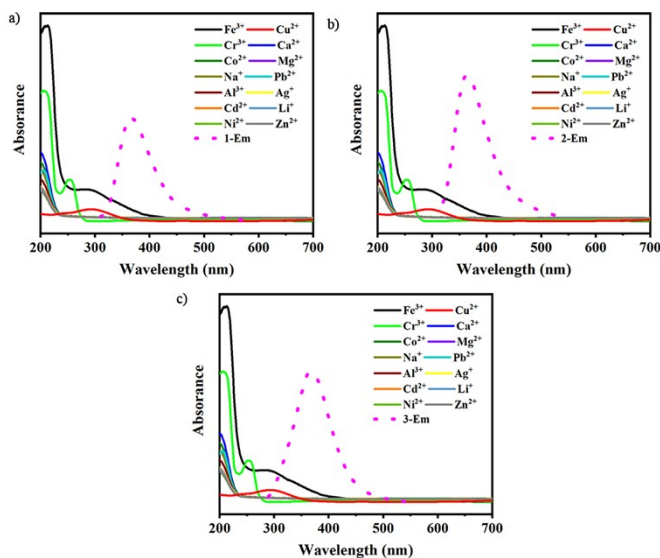


Fig. S22 UV-vis spectra of different cations in H₂O solutions, and the emission spectra of **1**(a), **2**(b) and **3**(c).

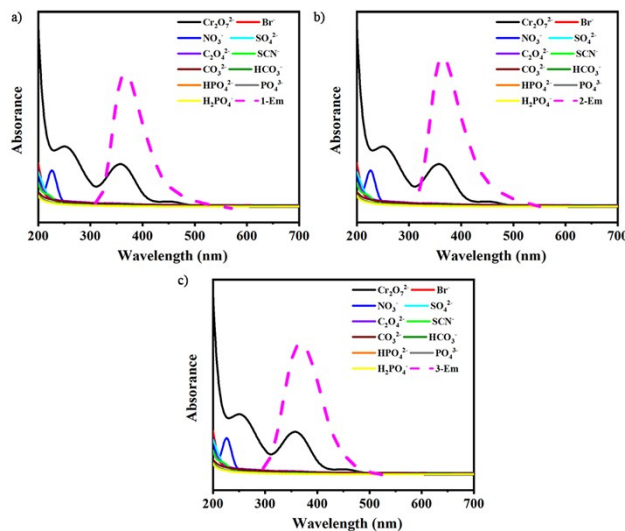


Fig. S23 UV-vis spectra of different anions in H₂O solutions, and the emission spectra of **1**(a), **2**(b) and **3**(c).

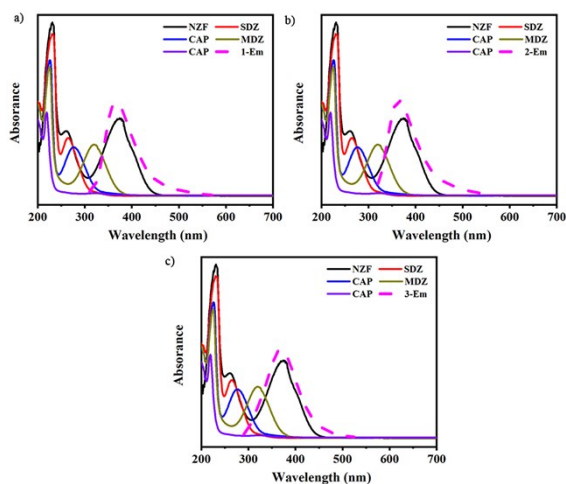


Fig. S24 UV-vis spectra of different antibiotics in H₂O solutions, and the emission spectra of **1(a)**, **2(b)** and **3(c)**.

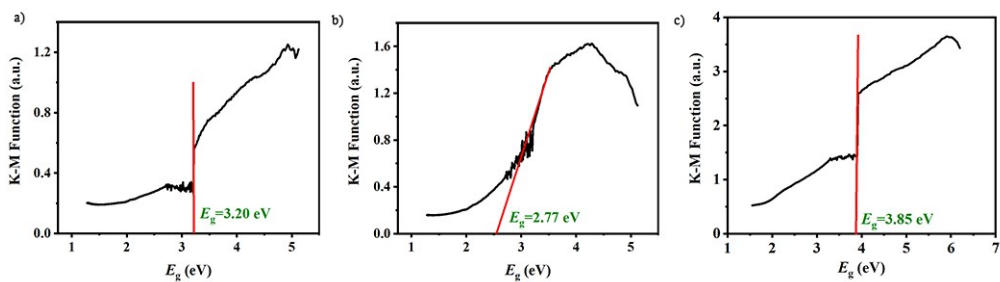


Fig. S25 Kubelka–Munk-transformed diffuse reflectance spectra of **1(a)**, **2(b)** and **3(c)**.