

## Supporting Information

### Hydrogen-bonding tuned hydroxo-bridged tetra-copper $\text{Cu}_4(\text{bipy})_4$ -cluster supramolecular network to layer coordination polymer

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## Materials and Instruments

Three benzenedicarboxylic acids (Adamas-beta®) were purchased from Adamas Reagent Co. Ltd (Shanghai, China). Other chemicals were commercially available and used as received without further purification.

The C, H and N elemental analyses were carried out with a Vario EL elemental analyzer. The ATR-FTIR spectra of the powders without KBr were collected in the range of 500-4000 cm<sup>-1</sup> by a Thermo Nicolet 6700 spectrometer. Powder X-ray diffraction (PXRD) data were collected on a Rigaku D/max 2200 diffractometer with Cu-K $\alpha$  radiation ( $\lambda = 1.5418 \text{ \AA}$ ).

**Table S1.** Crystallographic data for complexes **1-3**.

Compound	1	2	3
Formula	C <sub>56</sub> H <sub>60</sub> Cu <sub>4</sub> N <sub>8</sub> O <sub>20</sub>	C <sub>56</sub> H <sub>72</sub> Cu <sub>4</sub> N <sub>8</sub> O <sub>26</sub>	C <sub>28</sub> H <sub>42</sub> Cu <sub>2</sub> N <sub>4</sub> O <sub>16</sub>
Mr	1419.28	1527.37	817.73
Temp (K)	296	296	296
Cryst system	triclinic	monoclinic	monoclinic
Space group	$P\bar{1}$	$P2_1/c$	$P2_1/c$
$a/\text{\AA}$	9.0471(7)	14.6945(17)	16.609(5)
$b/\text{\AA}$	12.6873(9)	17.218(2)	16.984(5)
$c/\text{\AA}$	13.3662(9)	25.968(3)	13.707(4)
$\alpha/^\circ$	89.490(3)	90	90
$\beta/^\circ$	73.453(3)	94.777(4)	110.014(5)
$\gamma/^\circ$	75.303(3)	90	90
$V/\text{\AA}^3$	1419.24(18)	6547.4(13)	3633.0(18)
$Z$	1	4	4
$D_s/\text{g cm}^{-3}$	1.661	1.549	1.495
$\mu/\text{mm}^{-1}$	1.565	1.369	1.245
$F(000)$	728	3152	1696
$R(\text{int})$	0.0298	0.0932	0.0394
Total reflections	14891	179831	36818
Unique reflections	6522	15129	8317
$I > 2\sigma(I)$	4661	10407	7190
$R_1$	0.0321	0.0481	0.0753
$wR_2$	0.1016	0.1142	0.2668
$S$	1.004	1.042	1.156

**Table S2.** Selected bond distances (Å) and angles (°) for complexes **1-3**.

<b>1</b>		<b>2</b>				<b>3</b>	
Bond lengths							
Cu1-O1	1.9436(17)	Cu1-O1	1.922(2)	Cu2-O1W	2.246(2)	Cu1-O5	1.950(3)
Cu1-O2	1.9277(17)	Cu1-O2	1.954(2)	Cu3-O4	1.916(2)	Cu1-O1	2.408(3)
Cu1-N3	1.977(2)	Cu1-O3	2.312(2)	Cu3-O3	1.9547(19)	Cu1-N2	2.005(4)
Cu1-N4	2.027(2)	Cu1-N1	2.023(3)	Cu3-O2	2.270(2)	Cu1-N1	1.993(4)
Cu2-O2	1.9129(18)	Cu1-N2	1.990(3)	Cu3-N8	2.020(3)	Cu2-O4	1.943(3)
Cu2-O1	1.9586(17)	Cu2-O1	1.923(2)	Cu4-O4	1.929(2)	Cu2-O1	1.965(3)
Cu2-N1	1.993(2)	Cu2-O2	1.959(2)	Cu4-O3	1.980(2)	Cu2-O5	2.392(3)
Cu2-N2	2.012(2)	Cu2-N3	2.005(3)	Cu4-N6	2.021(3)	Cu2-N3	2.012(4)
Cu2-O1W	2.2437(19)	Cu2-N4	2.023(3)	Cu4-O6	2.203(2)	Cu2-N4	1.992(4)
<b>1</b>		<b>2</b>				<b>3</b>	
Bond angles							
O2-Cu1-O1	82.26(7)	O1-Cu1-O2	81.90(9)	O5-Cu1-N1	174.76(14)		
O2-Cu1-N3	95.87(9)	O1-Cu1-N2	98.04(11)	N1-Cu1-N2	81.22(16)		
O1-Cu1-N3	177.72(8)	O2-Cu1-N2	177.25(10)	O5-Cu1-O1	81.24(11)		
O2-Cu1-N4	162.17(8)	N2-Cu1-N1	80.77(11)	N2-Cu1-O1	93.21(13)		
O1-Cu1-N4	100.53(8)	N1-Cu1-O3	95.07(9)	N1-Cu1-O1	94.64(13)		
N3-Cu1-N4	80.85(9)	O1-Cu2-O2	81.76(9)	O5-Cu1-N2	102.17(14)		
O1-Cu1-O1	83.24(7)	O1-Cu2-N3	170.52(10)	O1-Cu2-N4	175.27(14)		
O2-Cu2-O1	82.25(7)	O2-Cu2-N3	97.83(10)	N4-Cu2-N3	80.92(16)		
O2-Cu2-N1	170.44(8)	N3-Cu2-N4	80.46(11)	O1-Cu2-O5	81.34(11)		
O1-Cu2-N1	97.62(8)	O1-Cu2-O1W	93.79(9)	N4-Cu2-O5	100.09(13)		
O2-Cu2-N2	95.37(8)	O4-Cu3-O3	83.38(8)	N3-Cu2-O5	96.50(12)		
O1-Cu2-N2	156.68(8)	O4-Cu3-N7	97.05(10)				
		N7-Cu3-N8	80.76(10)				
		O3-Cu4-O6	98.36(8)				
		N5-Cu4-O6	102.23(9)				

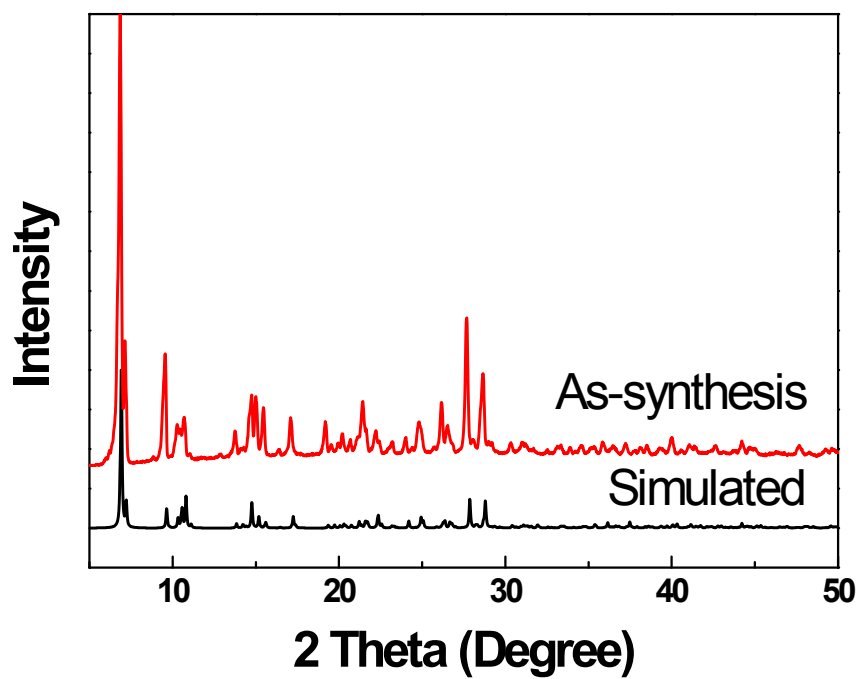


Figure S1. The PXRD curves of 1.

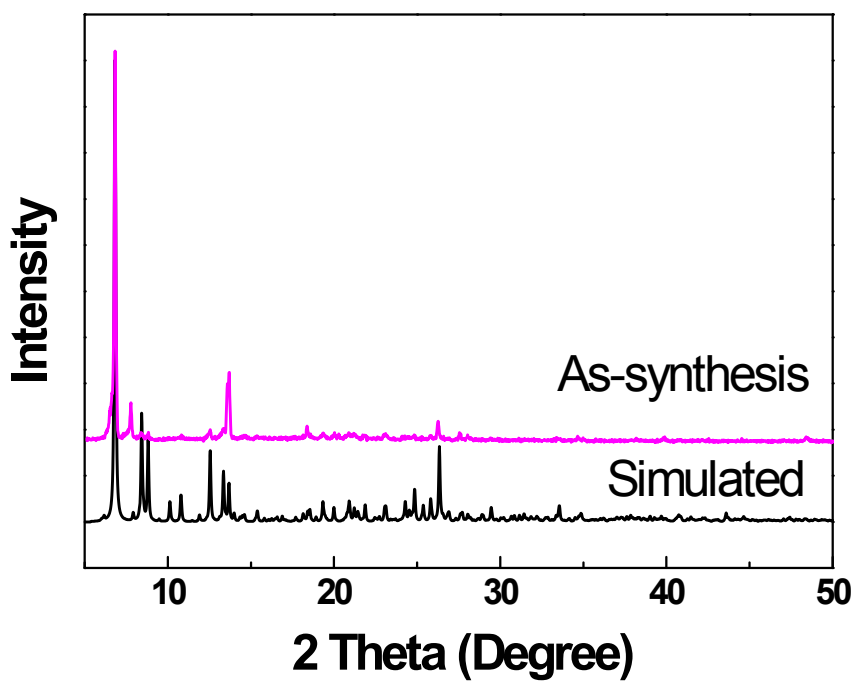


Figure S2. The PXRD curves of 2.

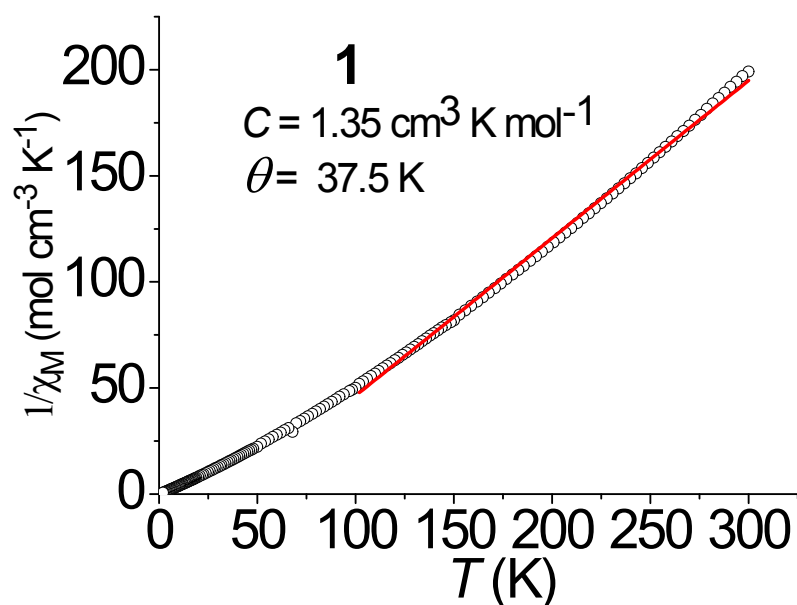


Figure S3. Curie-Weiss fit (red solid line) of the inverse magnetic susceptibility  $1/\chi_M$  for **1**

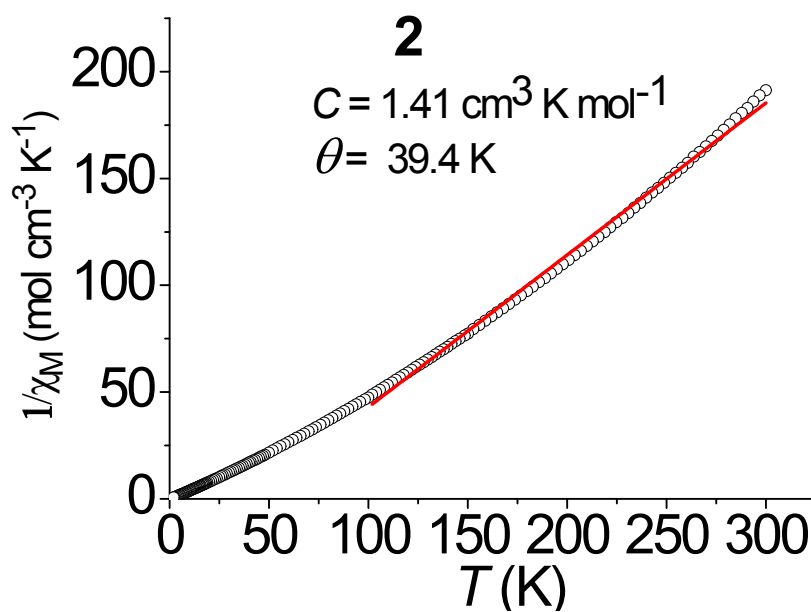


Figure S4. Curie-Weiss fit (red solid line) of the inverse magnetic susceptibility  $1/\chi_M$  for **2**