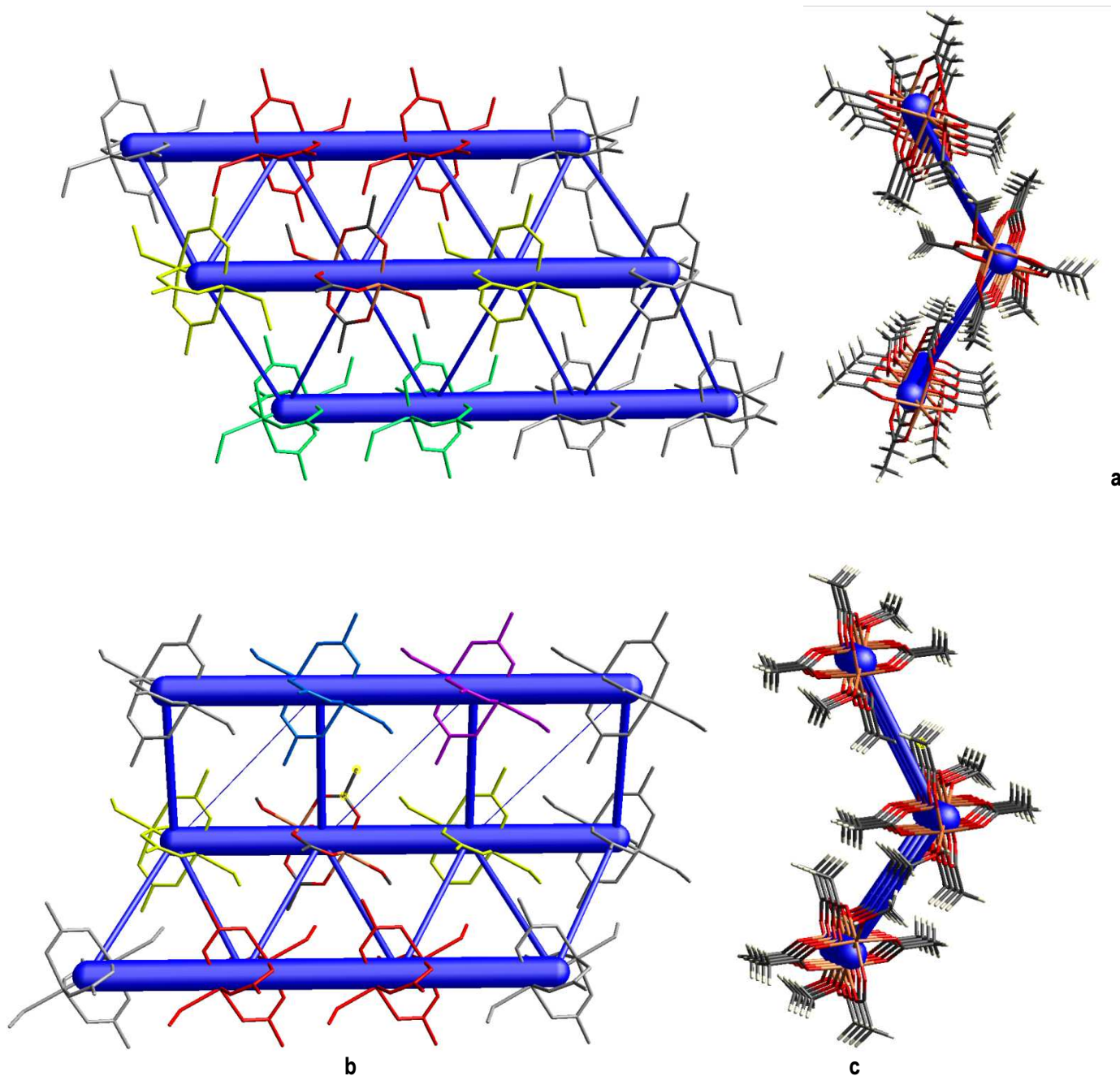


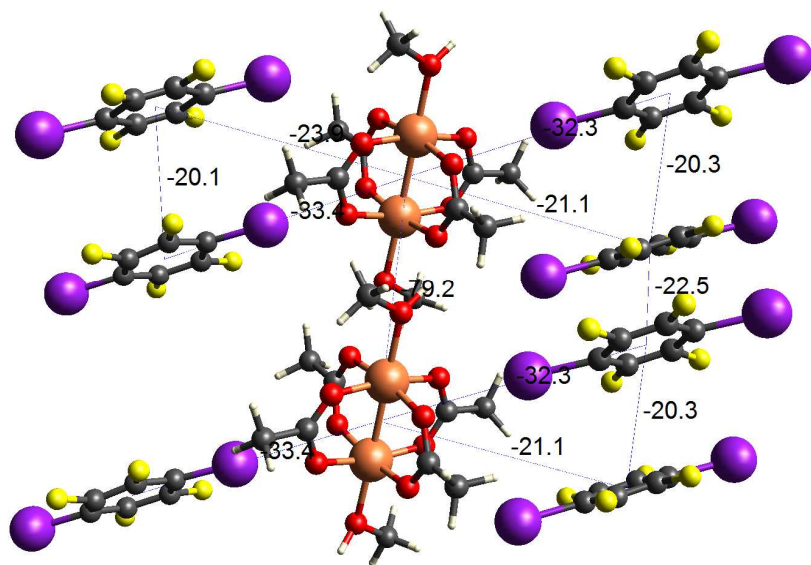
## New Supramolecular Heterosynthon [C-I...O=C(carboxylate)] at Work:

### Engineering of the Copper Acetates Cocrystals.

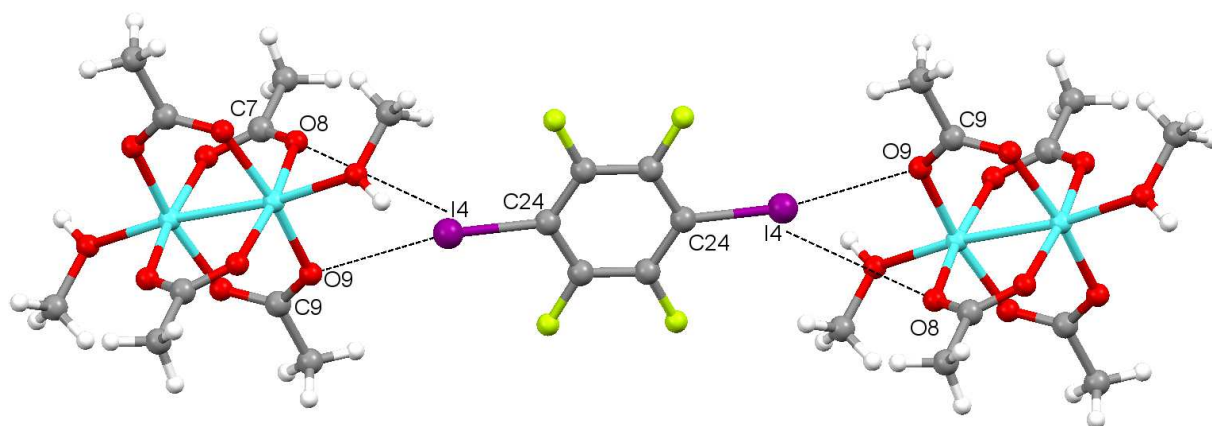
Yury V. Torubaev, Ivan V. Skabitsky



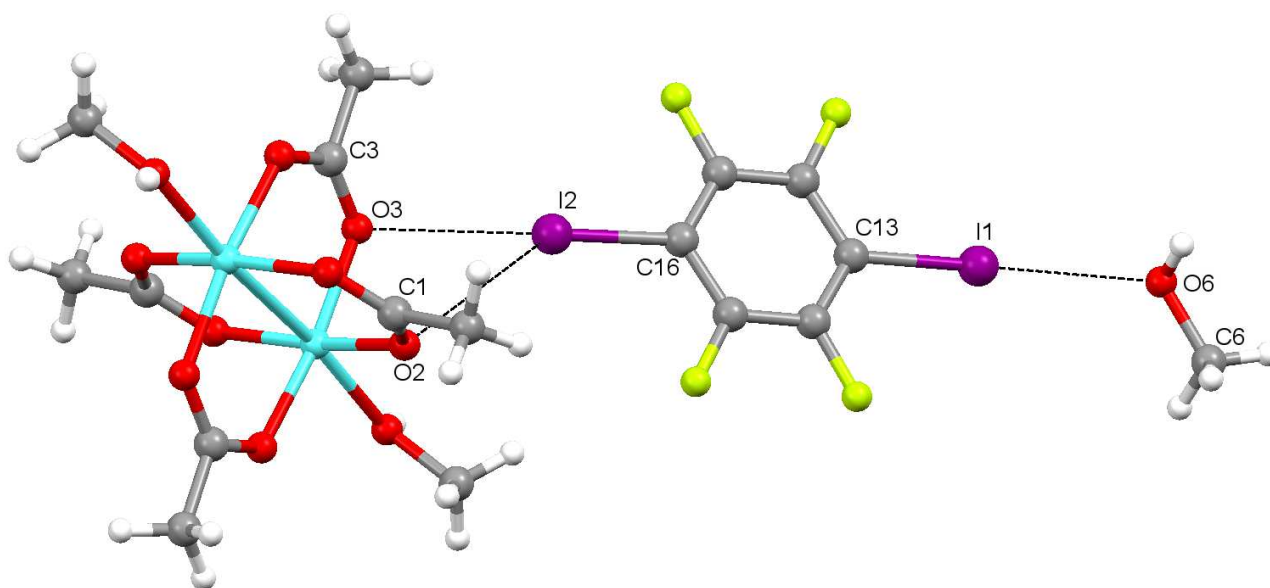
**Figure S1.** Showing the energy framework of the native crystal of  $[\text{Cu}(\text{OAc})_2\text{MeOH}]_2$  as the intersection of three planes (a, only half of each plane is shown for clarity), and its extended side projections (b) and (c). Intermolecular interactions are given in kJ/mol



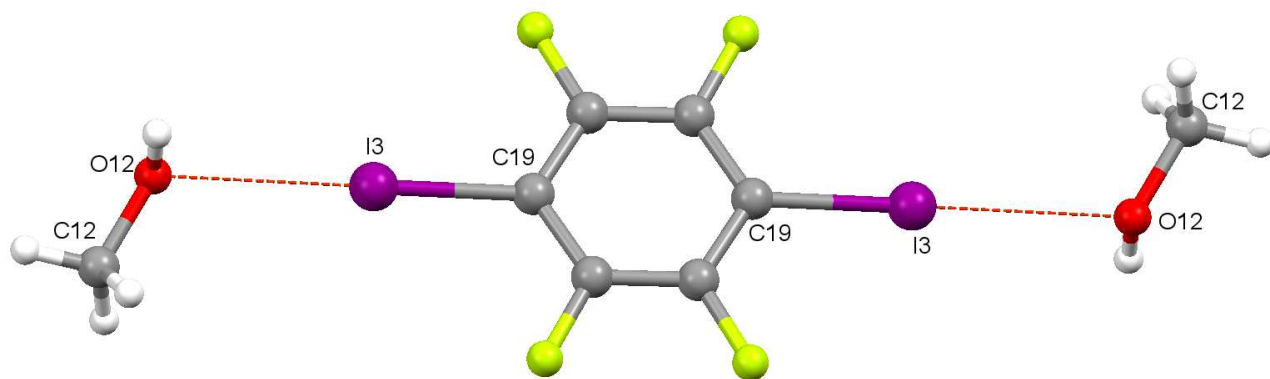
**Figure S2.** Showing the intermolecular interaction energies (in kJ/mol) in the fragment of cocrystal 2 packing,



**(a)** Selected intermolecular distances (Å): O(8)-I(4) 3.46(1), I(4)-O(9) 3.090(4)



**(b)** Selected intermolecular distances (Å): O(3)-I(2) 3.032(3), I(2)-O(2) 3.748(3), I(1)-O(6) 2.917(3)



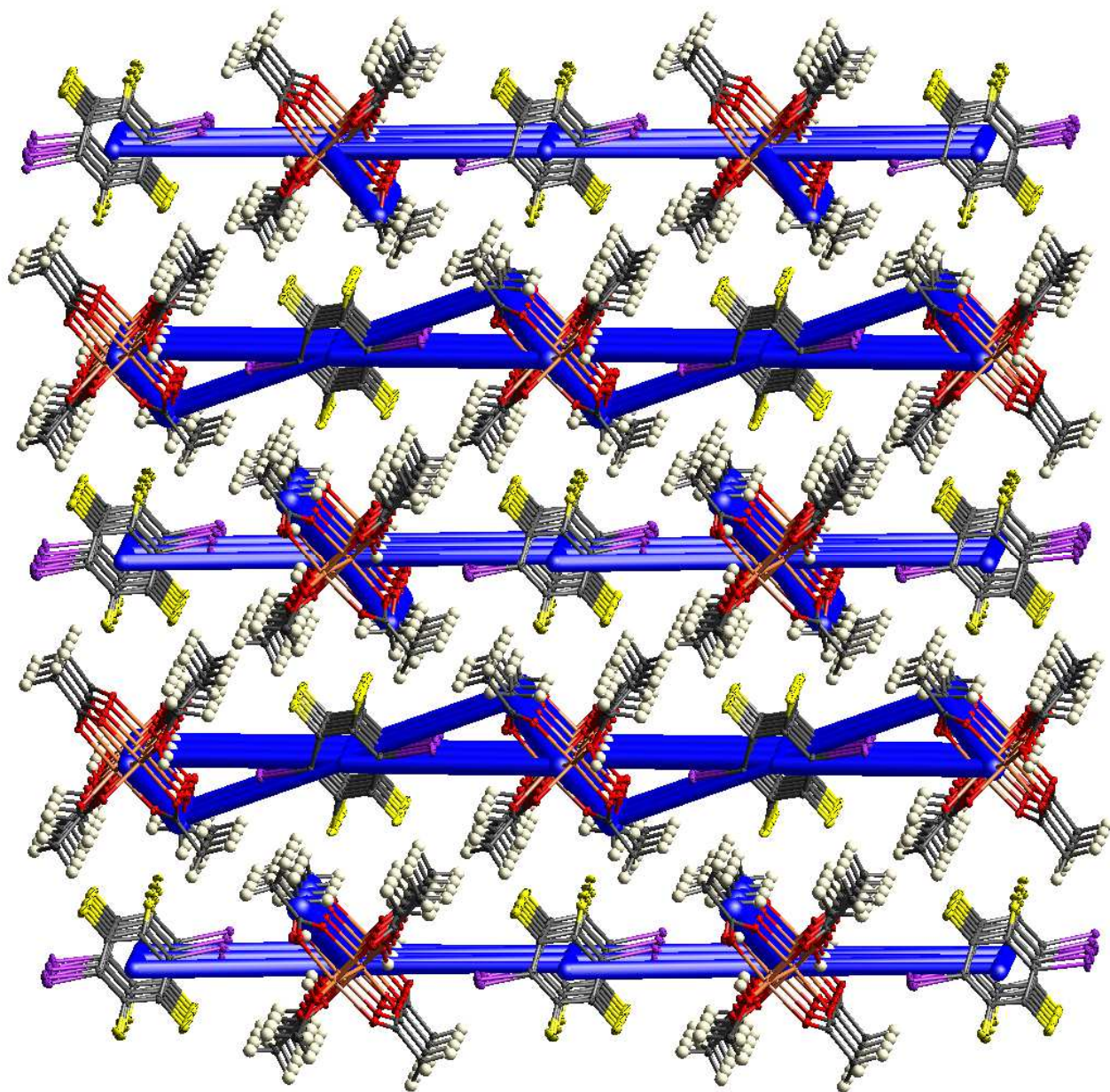
(c) Selected intermolecular distances (Å): I3---O12 2.911(3), C19-I3---O12 176.2(1), C12-O12—I3 120.0(3)

**Figure S3.** Fragments of crystal packing of **2**, showing three independent molecules of 1,4-DITFB bridging (a) two [Cu(OAc)<sub>2</sub>MeOH]<sub>2</sub> molecules, (b) [Cu(OAc)<sub>2</sub>MeOH]<sub>2</sub> molecule and MeOH solvate, and (c) two MeOH solvates.

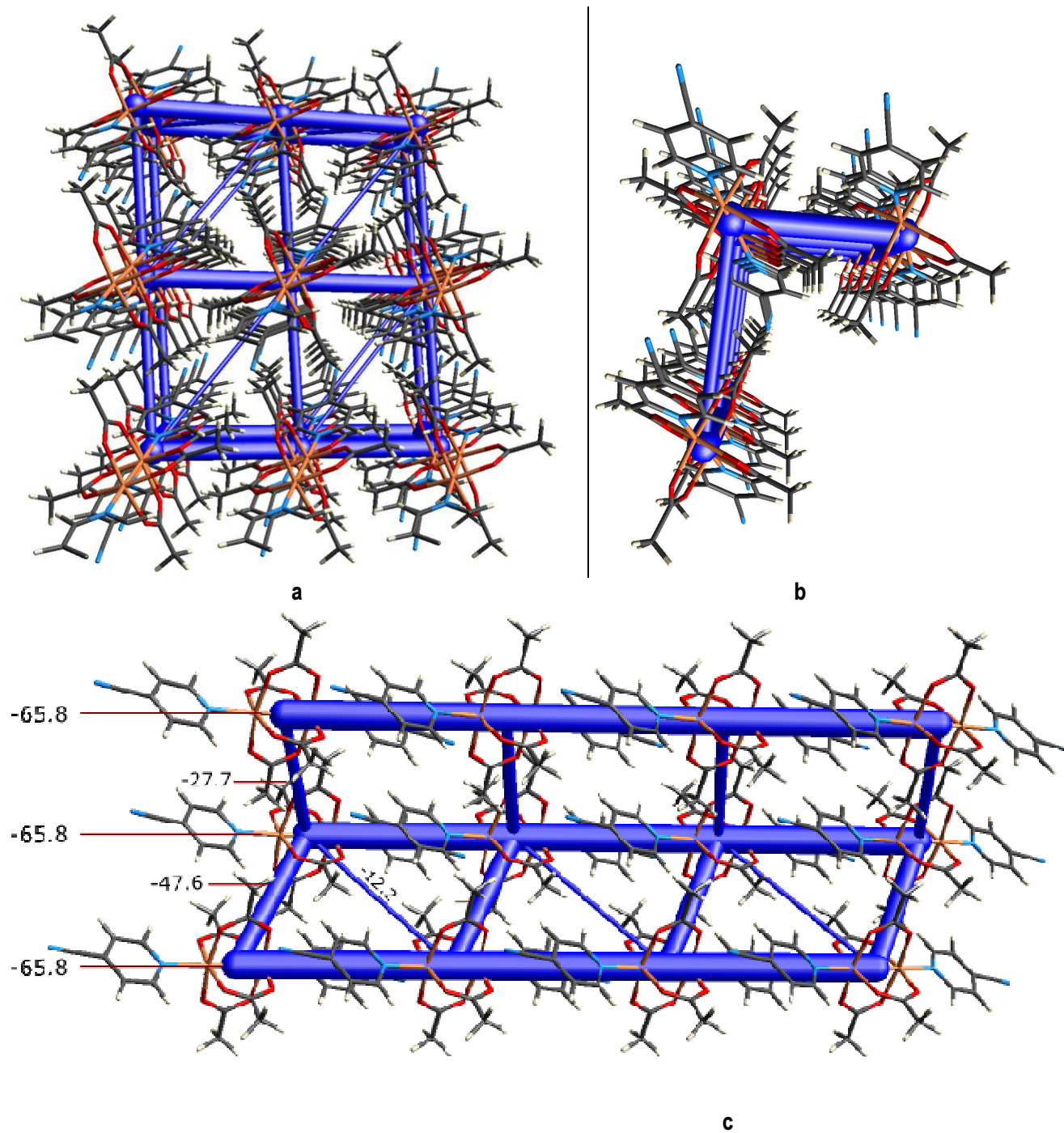
	E_ele	E_pol	E_dis	E_rep	E_tot
	-22.0	-3.5	-19.6	11.7	-35.6
	-37.1	-9.0	-21.6	44.3	-37.3
	-58.0	-13.4	-16.0	67.4	-43.5

**Figure S4.** Showing the intermolecular interaction energies in the ([Cu(OAc)<sub>2</sub>MeOH]<sub>2</sub> 2MeOH)<sub>n</sub> chain of cocrystal solvate **2**. Intermolecular interactions are given in kJ/mol. (CE-B3LYP / DGDZVP).

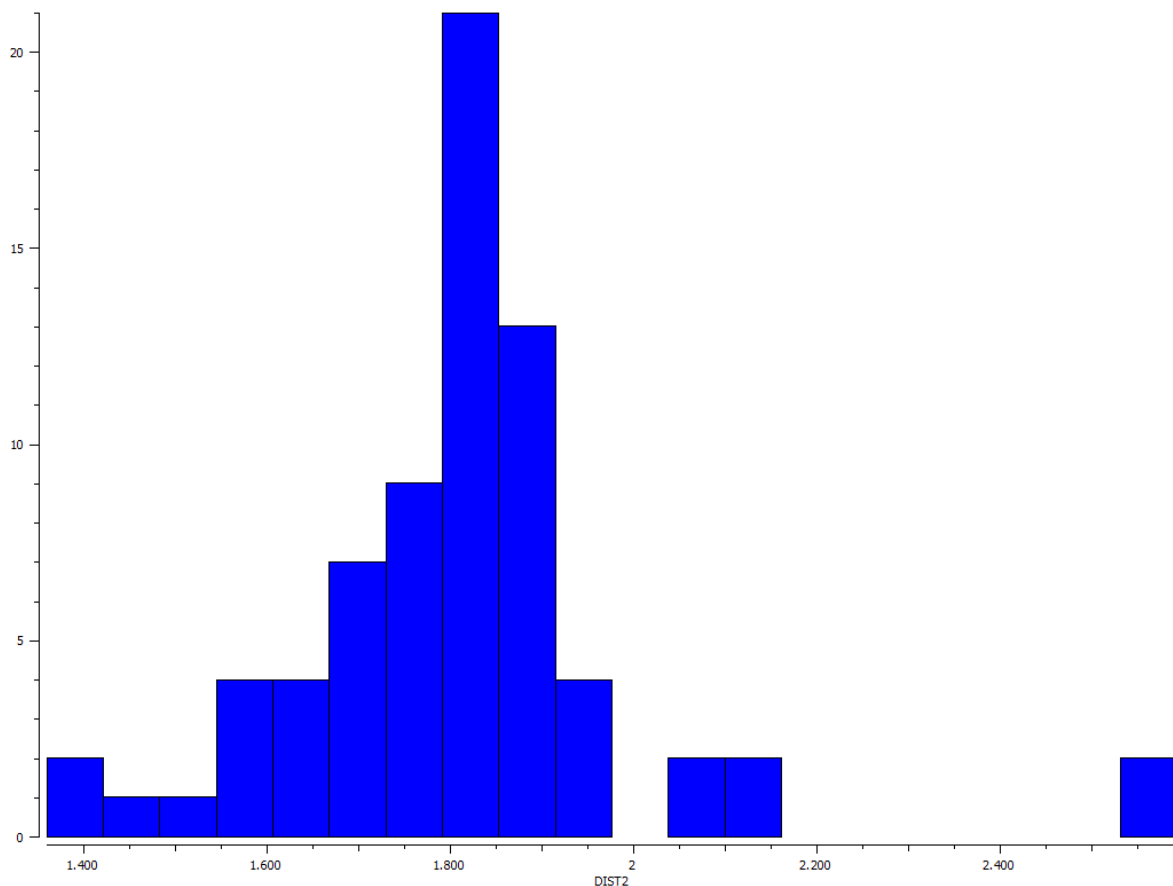




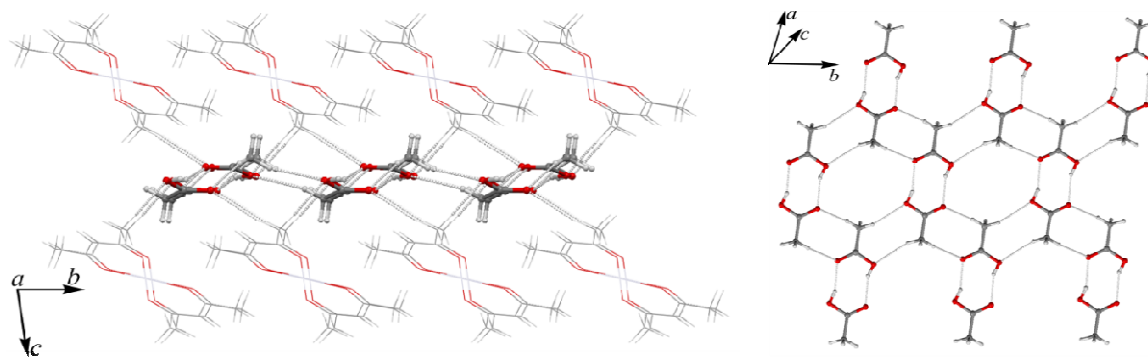
**Figure S5.** Showing the energy framework of **2** along the *c*-axis (cutoff 20 kJ/mol). Notice the absence of planar homomolecular energy frameworks.



**Figure S6.** Showing (a) the energy framework of the native  $[\text{Cu}(\text{OAc})_2(4\text{CNpy})]_2$  crystal (notice the intersection of three planes) and its side projections (b) and (c). Cutoff 5 kJ/mol. Intermolecular interactions are given in kJ/mol.



**Figure S7.** Distribution of O-H...O=C distance in AcOH dimers in the structures, deposited in CCDC CSD (Release 2020.0) [1].



**a**

**Figure S8.** AA dimers shown in the fragments of the crystal packing of bis(acetylacetonato)-platinum(II) acetic acid solvate [2]. Packing pattern along the *a* axis (a) and packing pattern of the AA sublattice (b). Dotted lines shows H...O HBs at a distances shorter than sum of vdW radii. Bis(acetylacetonato)-platinum(II) are displayed in wireframe style for clarity.

<sup>1</sup>. Macrae, C. F., Sovago, I., Cottrell, S. J., Galek, P. T. A., McCabe, P., Pidcock, E., ... Wood, P. A. Mercury 4.0: from visualization to analysis, design and prediction. *J. Appl. Cryst.* **2020**, 53, 226–235

<sup>2</sup>. Kozitsyna, N. Y., Gekhman, A. E., Nefedov, S. E., Vargaftik, M. N., & Moiseev, I. I. A trinuclear platinum–cobalt acetate-bridged complex  $\{\text{Pt}^{\text{II}}(\text{acac})\}_2\text{Co}^{\text{II}}(\mu\text{-OOCMe})_4$ : Synthesis and structure., I.I., *Inorg. Chem. Commun.* **2007**, 10, 956

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