High dual-state blue emission of functionalized pyrazoline

derivative for picric acid detection

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Fig. S2. ¹³C NMR spectrum of NF in DMSO.



Fig. S3. thermogravimetric (TG) analysis of NF

Table S1. Optical data of the compound NF in different solvents and solid state

Compound	solvents	$\lambda_{ab}^{a}(nm)$	$\lambda_{em}^{b}(nm)$	Δv^{c} (cm ⁻¹)	$arPsi_{ ext{PL}}{}^{ ext{d}}$
	Benzene	369	439	4321	0.40

DCM	369	439	4321	0.51
THF	363	434	4507	0.58
EA	362	433	4530	0.44
EtOH	362	453	5549	0.25
MeCN	362	438	4793	0.53
DMF	362	437	4741	0.53
Solid state	390	453	3566	0.43

^a Peak position of the maximum absorption band. ^b Peak position of fluorescence emission, excited at the absorption maximum. ^c Stokes' shift in cm⁻¹. ^d fluorescence quantum yields (Φ_{PL}).



Fig. S4 the UV-vis absorption in different solvents (a) and solid state (b).



Fig. S5 Time-resolved emission decay curves of NF in THF solution and solid state.



Fig. S6 (a) Corresponding Stern-Volmer plot for lower concentration of PA detection. Inset:Stern-Volmer plot obtained at concentration of PA. (b) The linear relationship of NF between the fluorescence intensity and the PA concentration



Fig. S7 the UV-vis absorption (a) and PL spectra (b) in THF/water (v/v = 8 : 2) containing PA and TFA.

Compounds	NF	Z, Calculated density	4, 1.238 (g/cm ⁻³)
Formula	$C_{23}H_{22}FN_3$	Absorption coefficient	0.081 mm ⁻¹
Mr	359.44	Data/restraints/pa rameters	3383/0/247
Temperature	293	Goodness-of-fit on F ²	1.120
Crystal system	orthorhombic	Largest diff.peak and hole	0.12 and -0.13 e.A ⁻³
Space group	P212121	Theta range for data collection	2.24 to 49.98 deg.
<i>a</i> (Å)	5.9014		$-7 \le h \le 6$
<i>b</i> (Å)	8.9662	Limiting indices	$-10 \le k \le 10$
<i>c</i> (Å)	36.435		$-43 \le 1 \le 37$
β (°)	90	F(000)	760
α (°)	90	Reflection	2668 /3383 [R

Table S2. Summary of crystallographic data and structure refinement details for NF

		collected / unique	(int) = 0.0453]
Γ (°)	90	Final R indices [I>2 σ (I)]	$R_1 = 0.0554, wR_2$ = 0.1052
Volume	1927.9	R indices (all)	$R_1 = 0.0769, wR_2 = 0.11128$