

Electronic supplementary information

Growth of LaCoO₃ crystals in molten salt: Effects of synthesis

conditions

Sanzhao Song^a, Jian Sun^{a,b}, Jing Zhou^{a,c}, Chengzhi Guan^{a,c}, Zhiwei Hu^d, Ting-Shan, Chan^e, Xian-Long Du^{*a,b,c}, Xiao Lin^{a,c}, Jun Hu^{a,c}, Linjuan Zhang^{*a,b,c} and Jian-Qiang Wang^{*a,b,c}

^a Key Laboratory of Interfacial Physics and Technology, Shanghai Institute of Applied Physics, Chinese Academy of Sciences, Shanghai 201800, China

^b University of Chinese Academy of Sciences, Beijing 100049, China

^c Dalian National Laboratory for Clean Energy, Dalian 116023, China

^d Affiliation Max Planck Institute for Chemical Physics of Solids, Nöthnitzer Strasse 40, Dresden 01187, Germany.

^e National Synchrotron Radiation Research Center, Hsinchu 30076, Taiwan.

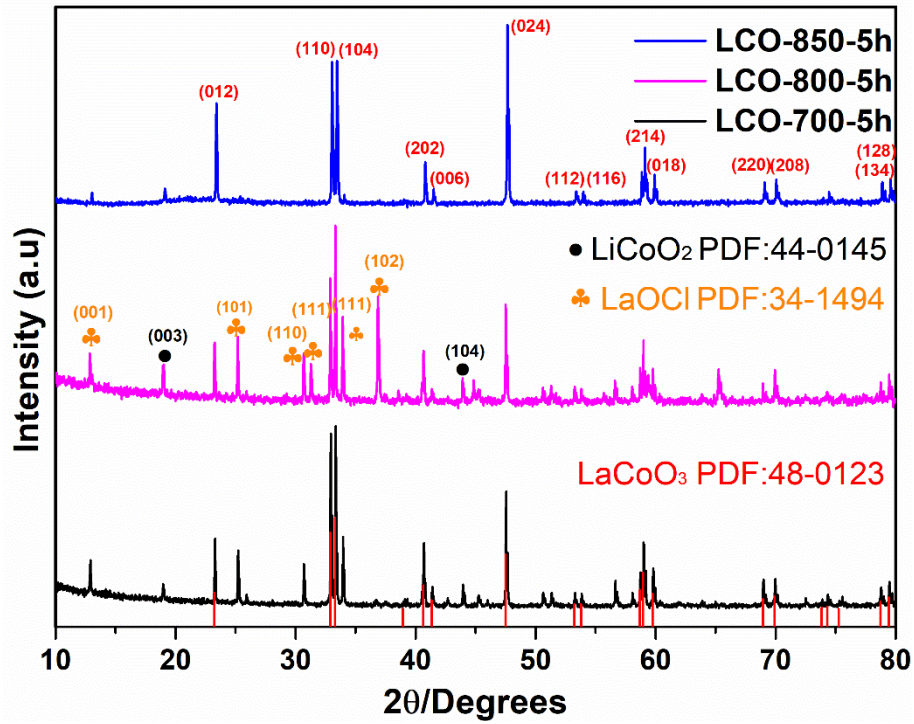


Fig. S1 XRD patterns of samples obtained at different temperature for 5 h with LiCl-KCl molten salts.

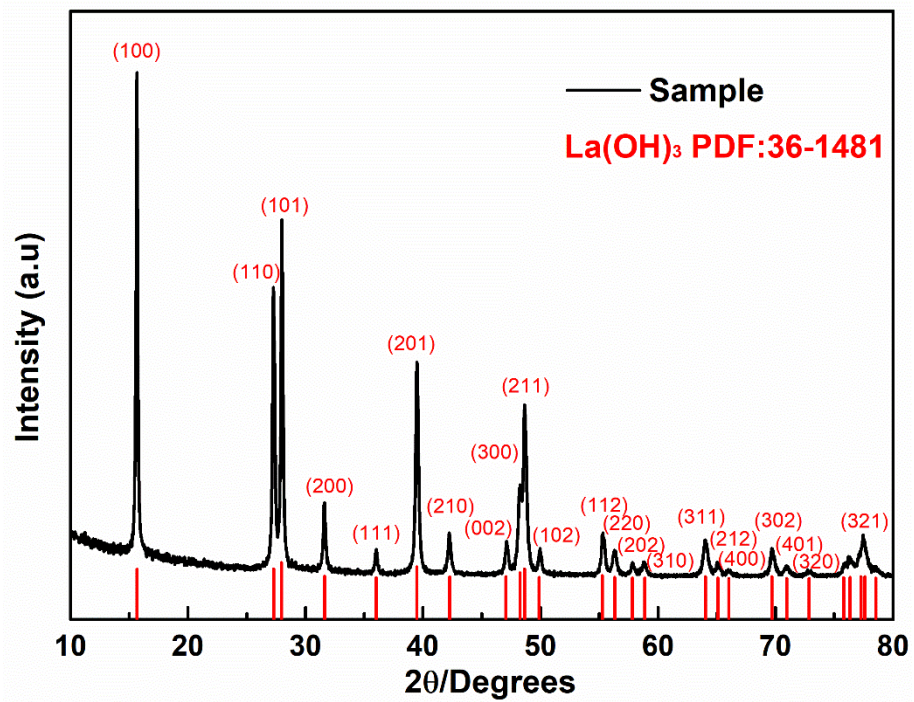


Fig. S2 XRD patterns of the sample obtained after the La_2O_3 soaked into the Na_2CO_3 - K_2CO_3 eutectic salt at 850 °C for 5 h.

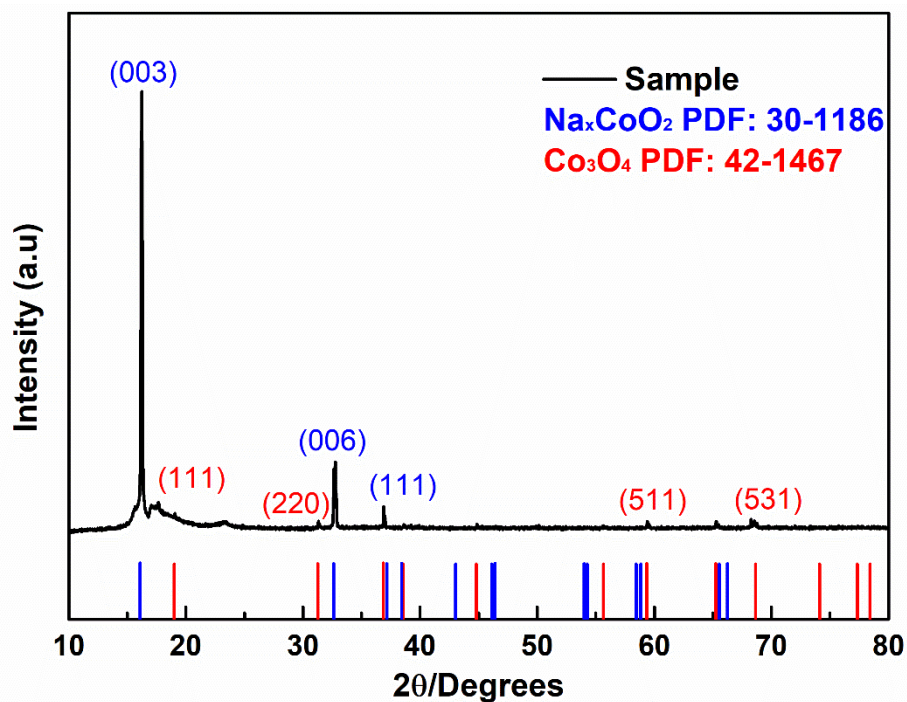


Fig. S3 XRD patterns of the sample obtained after Co_3O_4 soaked into the Na_2CO_3 - K_2CO_3 eutectic salts at 850°C for 5 h.

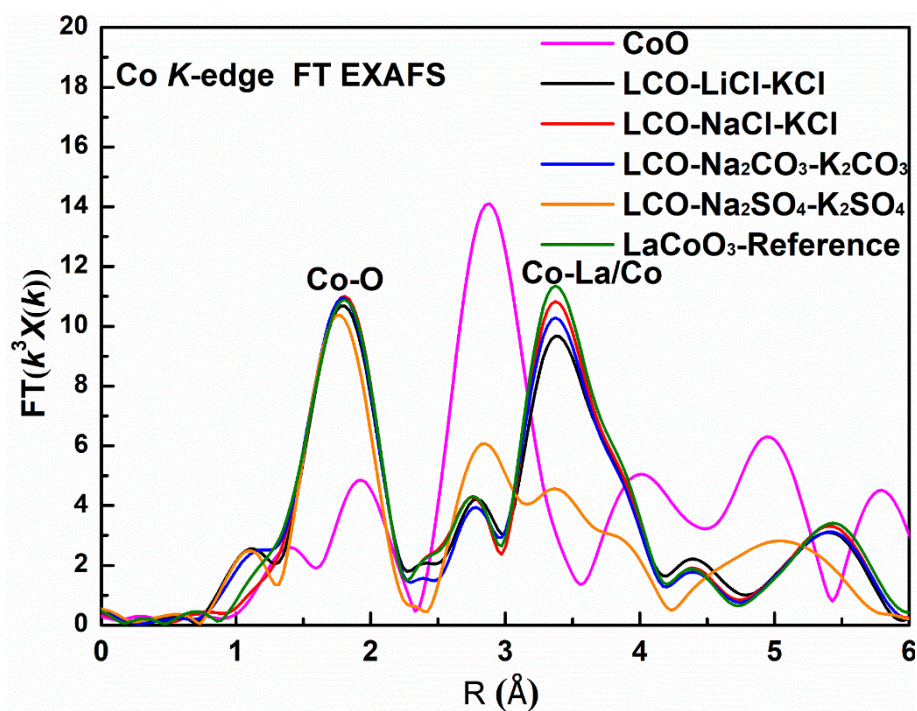


Fig. S4 Fourier transform (FT) of the Co K-edge EXAFS of synthesized samples and references.

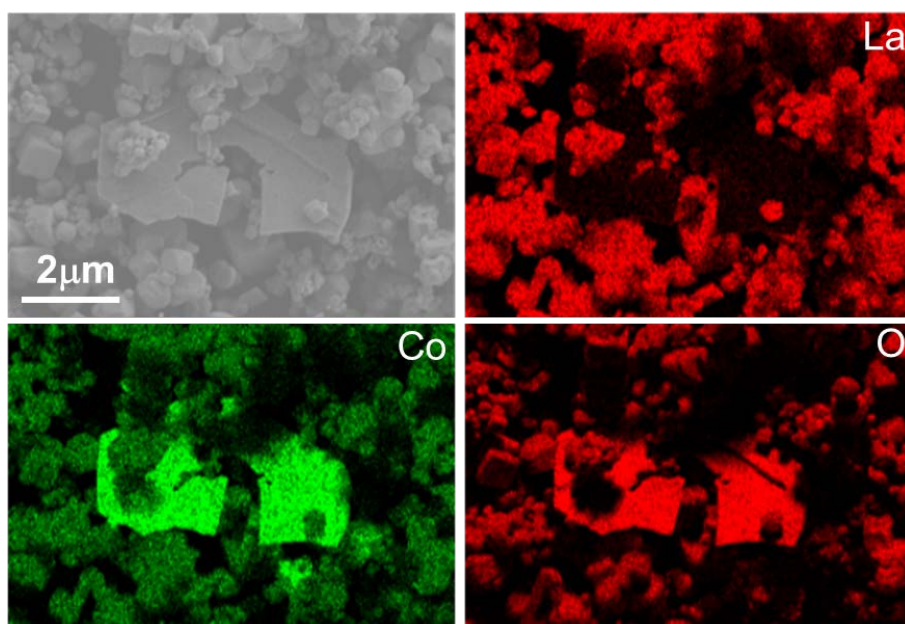


Fig. S5 SEM image and corresponding elemental mapping images of the synthesized sample in the LiCl-KCl eutectic salt at 850 °C for 5 h.

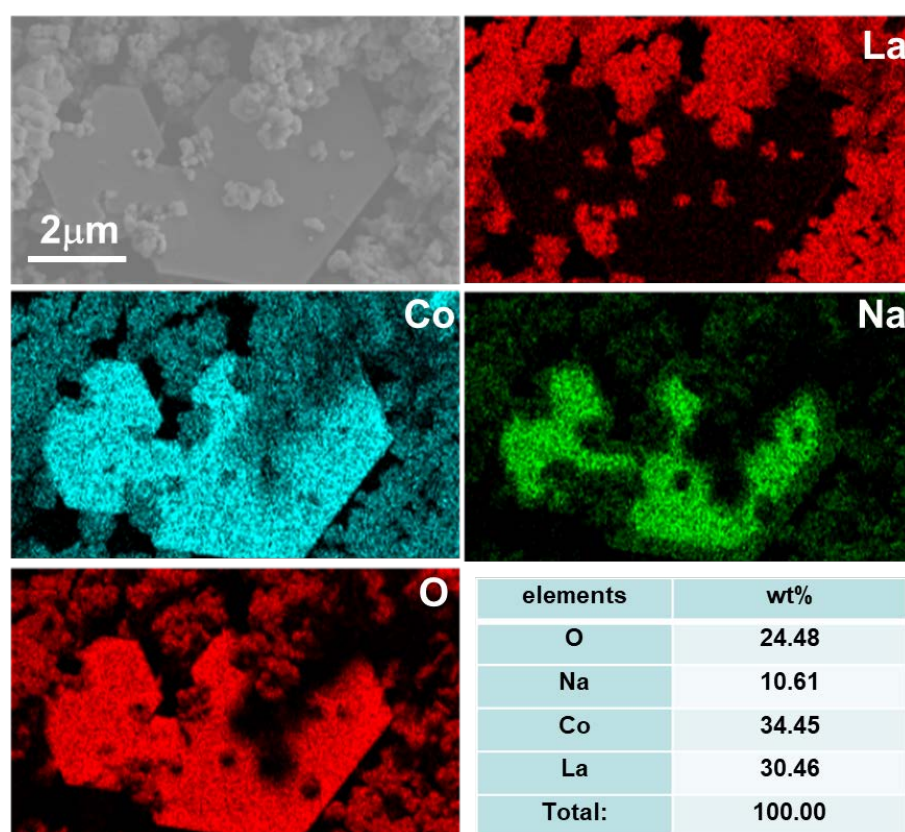


Fig. S6 SEM image and corresponding elemental mapping images of the synthesized sample in the $\text{Na}_2\text{CO}_3\text{-K}_2\text{CO}_3$ eutectic salt at 850 °C for 5 h.

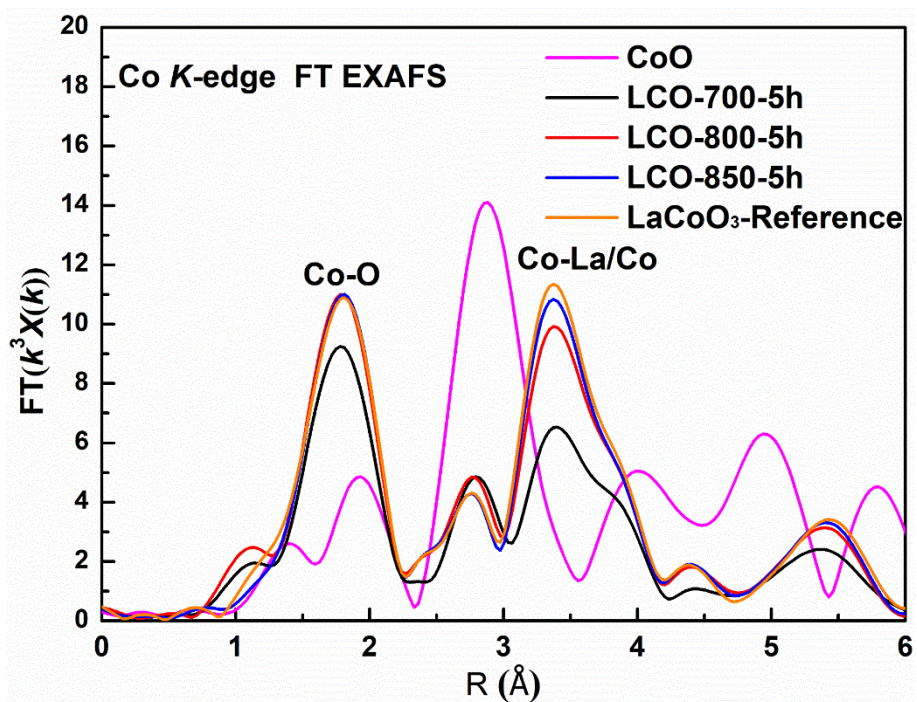


Fig. S7 Fourier transform (FT) of the Co K-edge EXAFS of synthesized samples under different temperatures and references.

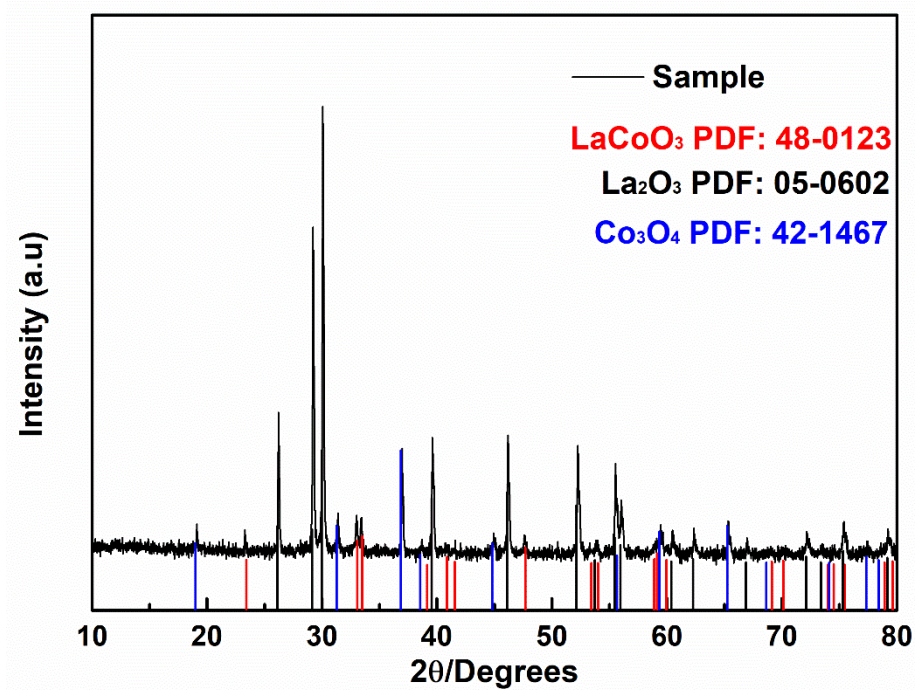


Fig. S8 XRD patterns of samples obtained by solid state method at 700 °C for 10 h.

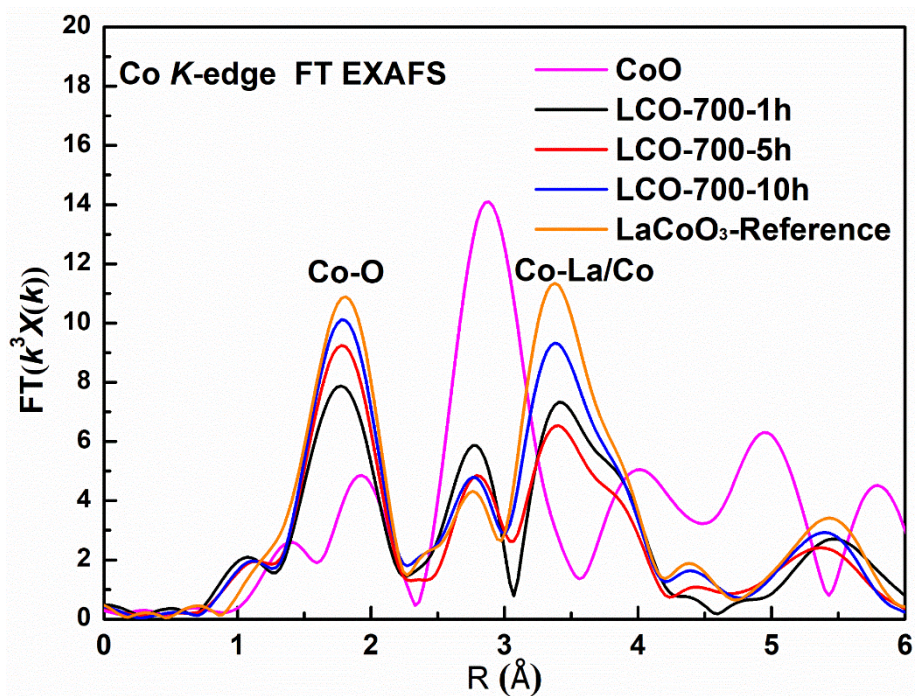


Fig. S9 Fourier transform (FT) of the Co K-edge EXAFS of the samples synthesized at 700 °C with different times and references.

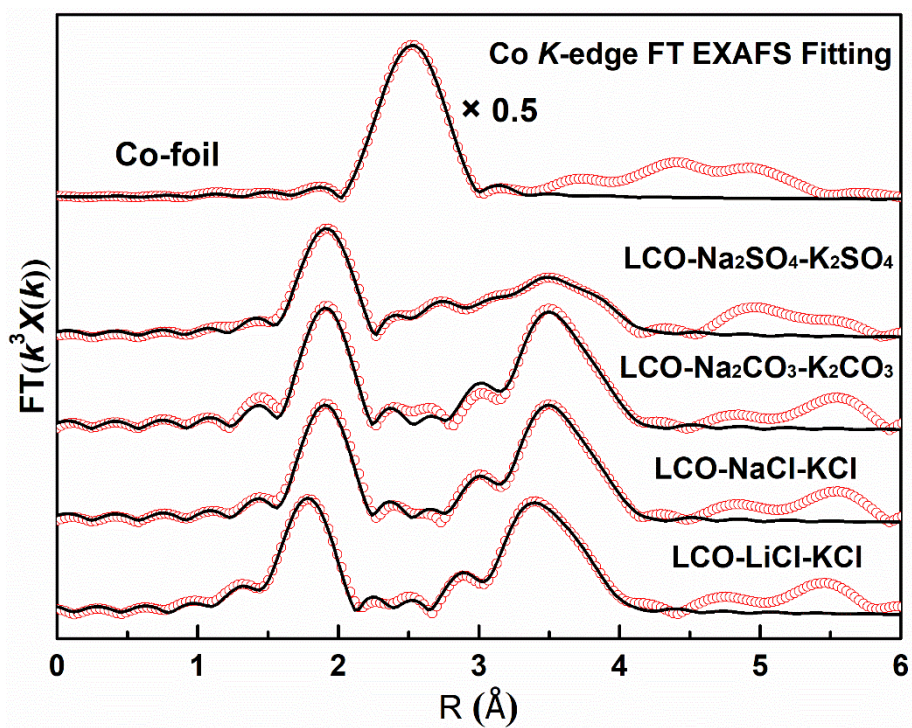


Fig. S10 EXAFS fitting curves for LCO with different molten salts. (FT range: 2.5 - 12.0 Å⁻¹).

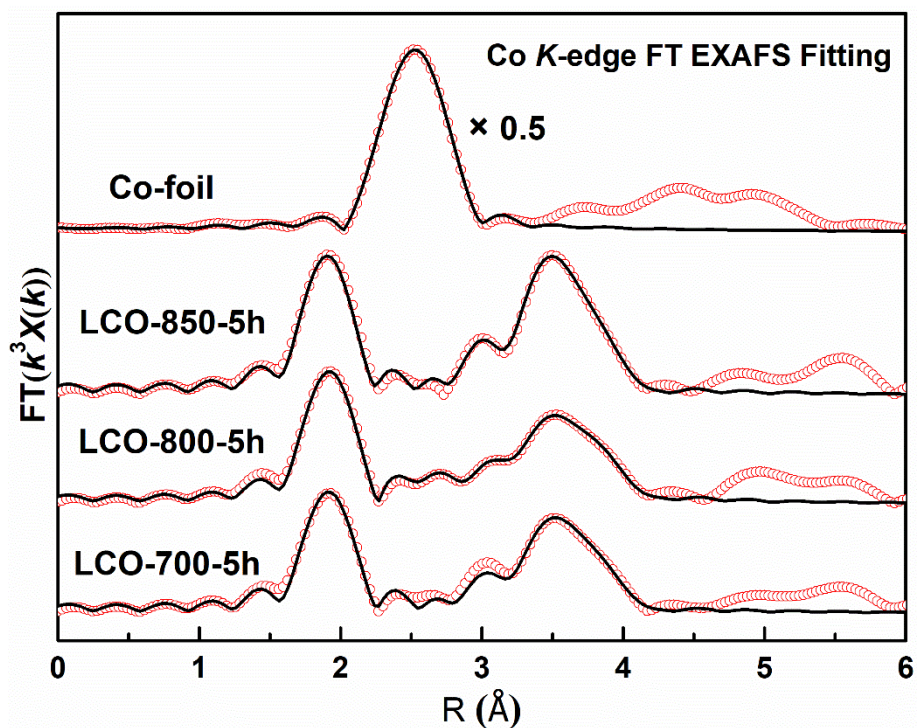


Fig. S11 EXAFS fitting curves for LCO with different temperature for 5 h. (FT range: 2.5 - 12.0 \AA^{-1}).

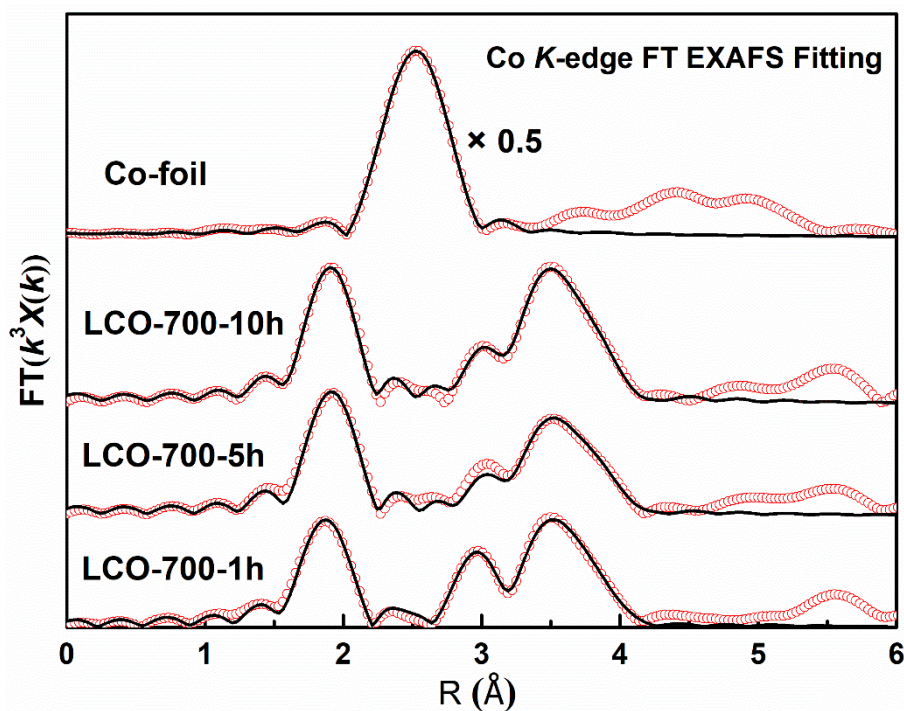


Fig. S12 EXAFS fitting curves for LCO at 700 $^{\circ}\text{C}$ with different time. (FT range: 2.5 - 12.0 \AA^{-1}).

Table S1 melting points and composition of some used metal salt systems.

Salt system	Composition (mol %)	Melting point ($^{\circ}\text{C}$)
NaCl	100	801
KCl	100	771
LiCl-KCl	59 / 41	353
NaCl- KCl	50 / 50	657

NaCl-Na ₂ CO ₃	55 / 45	632
NaCl-Na ₂ SO ₄	53 / 47	617
KCl-K ₂ CO ₃	62 / 38	631
KCl- K ₂ SO ₄	74 / 26	690
Na ₂ CO ₃ -K ₂ CO ₃	58 / 42	709
Na ₂ SO ₄ -K ₂ SO ₄	74 / 26	834

Table S2 Firing conditions and batch compositions of samples.

Samples No.	Firing conditions	Salt assembly (molar ratios)	Reactant: salt ratio (wt.) %
S0	850 °C / 5 h	LiCl-KCl (0.55:0.45)	1:2
S1		Na ₂ SO ₄ -K ₂ SO ₄ (0.74:0.26)	
S2		Na ₂ CO ₃ -K ₂ CO ₃ (0.58:0.42)	
S3		NaCl-Na ₂ CO ₃ (0.55:0.45)	
S4		NaCl-Na ₂ SO ₄ (0.50:0.50)	
S5		KCl-K ₂ CO ₃ (0.60:0.40)	
S6		KCl- K ₂ SO ₄ (0.75:0.25)	
S7		KCl	
S8		NaCl	
T1	800 °C / 5 h	NaCl-KCl (0.50:0.50)	
T2	700 °C / 10 h		
T3	700 °C / 5 h		
T4	700 °C / 1 h		
K1	700 °C / 10 h	No salt	
K2	700 °C / 10 h	NaCl-KCl (0.50:0.50)	1:5
K3	700 °C / 10 h	NaCl-KCl (0.50:0.50)	1:10

Table S3 Structural parameters of Co foil and samples synthesized via different molten salts at 850 °C, for 5 h.

Sample	Atomic Scatter	Coordination Number	Bond Length (Å)	Debye-Waller factor (10 ⁻³ ×Å ²)	R factor
Co -foil	Co-Co	12	2.49 ± 0.01	6.2 ± 0.2	0.001
LCO-NaCl-KCl	Co-O	5.5 ± 0.3	1.92 ± 0.01	3.3 ± 0.2	0.007
	Co-La	6.1 ± 0.2	3.33 ± 0.01	4.6 ± 0.3	
	Co-Co	3.0 ± 0.4	3.89 ± 0.01	1.7 ± 0.4	
LCO-LiCl-KCl	Co-O	5.6 ± 0.2	1.92 ± 0.01	3.3 ± 0.3	0.004
	Co-La	5.9 ± 0.2	3.33 ± 0.01	4.5 ± 0.2	
	Co-Co	3.3 ± 0.4	3.89 ± 0.01	2.0 ± 0.5	

LCO-Na ₂ CO ₃ -K ₂ CO ₃	Co-O	5.4 ± 0.3	1.92 ± 0.01	2.6 ± 0.6	0.014
	Co-La	6.3 ± 0.5	3.33 ± 0.01	4.7 ± 0.5	
	Co-Co	3.2 ± 0.6	3.89 ± 0.01	2.1 ± 1.0	
LCO-Na ₂ SO ₄ -K ₂ SO ₄	Co-O	4.9 ± 0.2	1.92 ± 0.01	2.5 ± 0.4	0.005
	Co-La	2.9 ± 0.4	3.36 ± 0.01	3.2 ± 0.8	
	Co-Co	2.6 ± 0.5	3.87 ± 0.01	3.4 ± 1.2	
	Co-Co''	3.1 ± 0.9	2.89 ± 0.01	6.7 ± 1.5	

Table S4 Structural parameters of Co foil and abtained products under different temperatures for 5 h extracted from the Co K-edge EXAFS fitting.

Sample	Atomic Scatter	Coordination Number	Bond Length (Å)	Debye-Waller factor (10 ⁻³ ×Å ²)	R factor
Co -foil	Co-Co	12	2.49 ± 0.01	6.2 ± 0.2	0.001
LCO-700-5h	Co-O	4.5 ± 0.4	1.93 ± 0.01	2.5 ± 0.7	0.028
	Co-La	3.7 ± 0.5	3.34 ± 0.01	3.8 ± 0.7	
	Co-Co	2.3 ± 0.7	3.90 ± 0.01	1.8 ± 1.2	
LCO-800-5h	Co-O	4.9 ± 0.2	1.92 ± 0.01	2.4 ± 0.4	0.007
	Co-La	3.5 ± 0.4	3.33 ± 0.01	3.7 ± 0.4	
	Co-Co	3.0 ± 0.5	3.89 ± 0.01	3.3 ± 0.9	
LCO-850-5h	Co-O	5.3 ± 0.2	1.93 ± 0.01	2.2 ± 0.4	0.004
	Co-La	3.7 ± 0.3	3.35 ± 0.01	3.4 ± 0.4	
	Co-Co	3.0 ± 0.5	3.89 ± 0.01	1.7 ± 0.7	

Table S5 Structural parameters of Co foil and synthesized samples under 700 °C with different times extracted from the Co K-edge EXAFS fitting.

Sample	Atomic Scatter	Coordination Number	Bond Length (Å)	Debye-Waller factor (10 ⁻³ ×Å ²)	R factor
Co -foil	Co-Co	12	2.49 ± 0.01	6.2 ± 0.2	0.001
LCO-700-1h	Co-O	4.0 ± 0.2	1.92 ± 0.01	2.6 ± 0.5	0.004
	Co-La	3.8 ± 0.3	3.32 ± 0.01	3.3 ± 0.4	
	Co-Co	2.8 ± 0.5	3.91 ± 0.01	1.9 ± 0.9	
	Co-Co'	6.0 ± 0.4	3.02 ± 0.01	14.5 ± 0.9	
	Co-O	4.5 ± 0.4	1.93 ± 0.01	2.5 ± 0.6	0.028

LCO-700-5h	Co-La	3.7 ± 0.5	3.34 ± 0.01	3.8 ± 0.2	
	Co-Co	2.3 ± 0.7	3.90 ± 0.01	1.8 ± 0.9	
LCO-700-10h	Co-O	5.6 ± 0.3	1.92 ± 0.01	3.5 ± 0.5	0.009
	Co-La	5.5 ± 0.3	3.33 ± 0.01	4.1 ± 0.3	
	Co-Co	3.6 ± 0.6	3.88 ± 0.01	3.0 ± 0.9	