

<Electronic Supplementary Information>

**Supramolecular isomerism between cyclodimeric and sinusoidal 1D coordination polymers: competition of tunable argentophilic vs electrostatic interactions**

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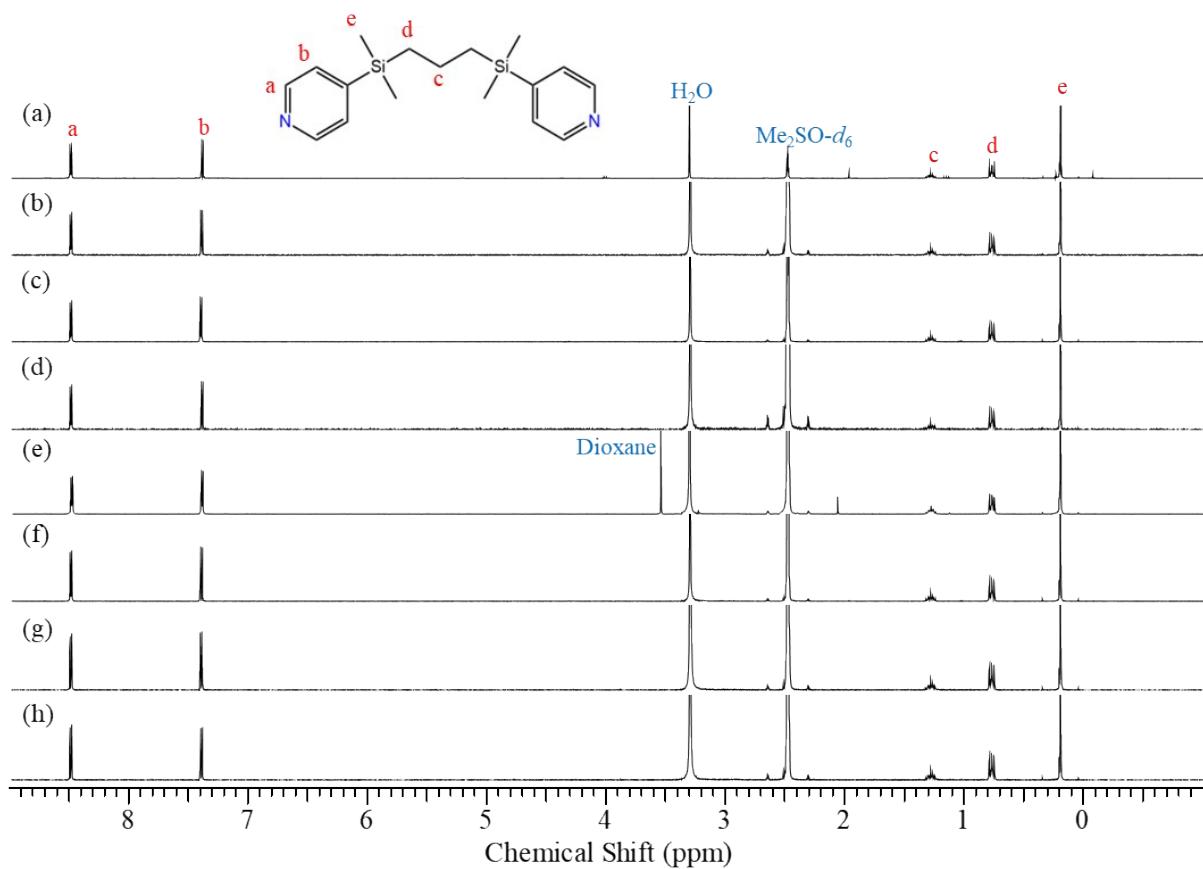
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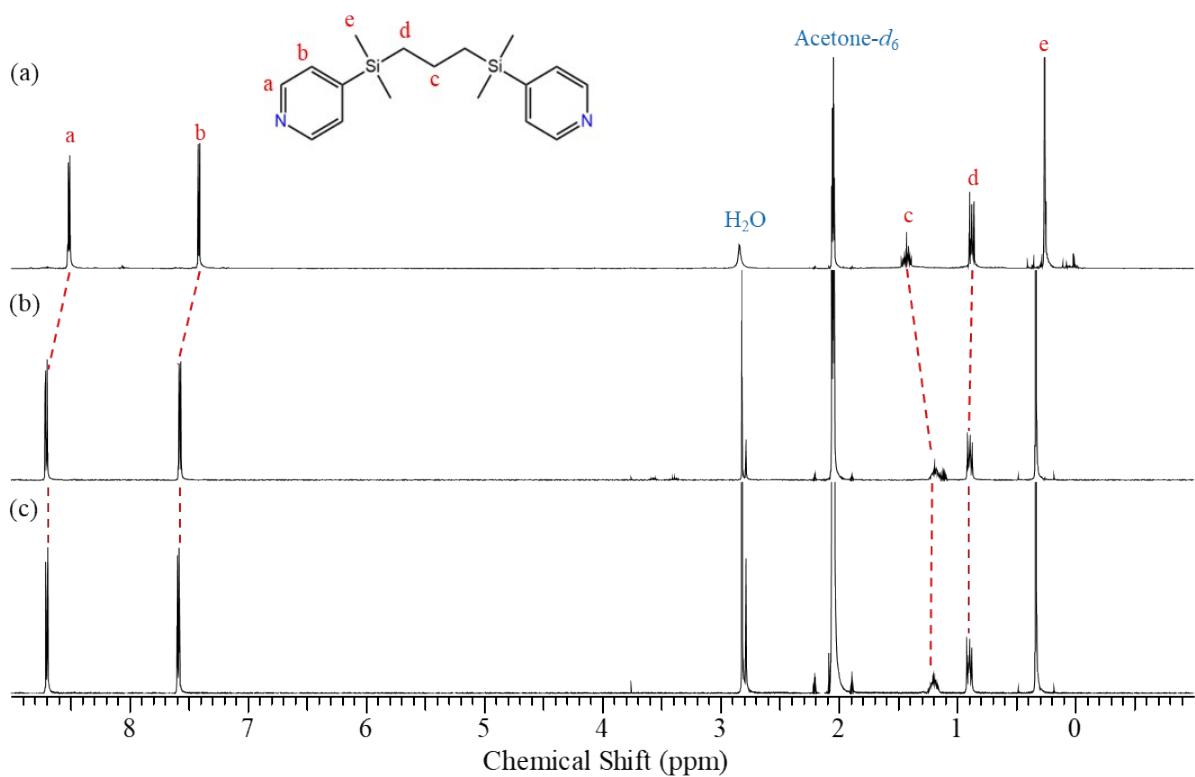
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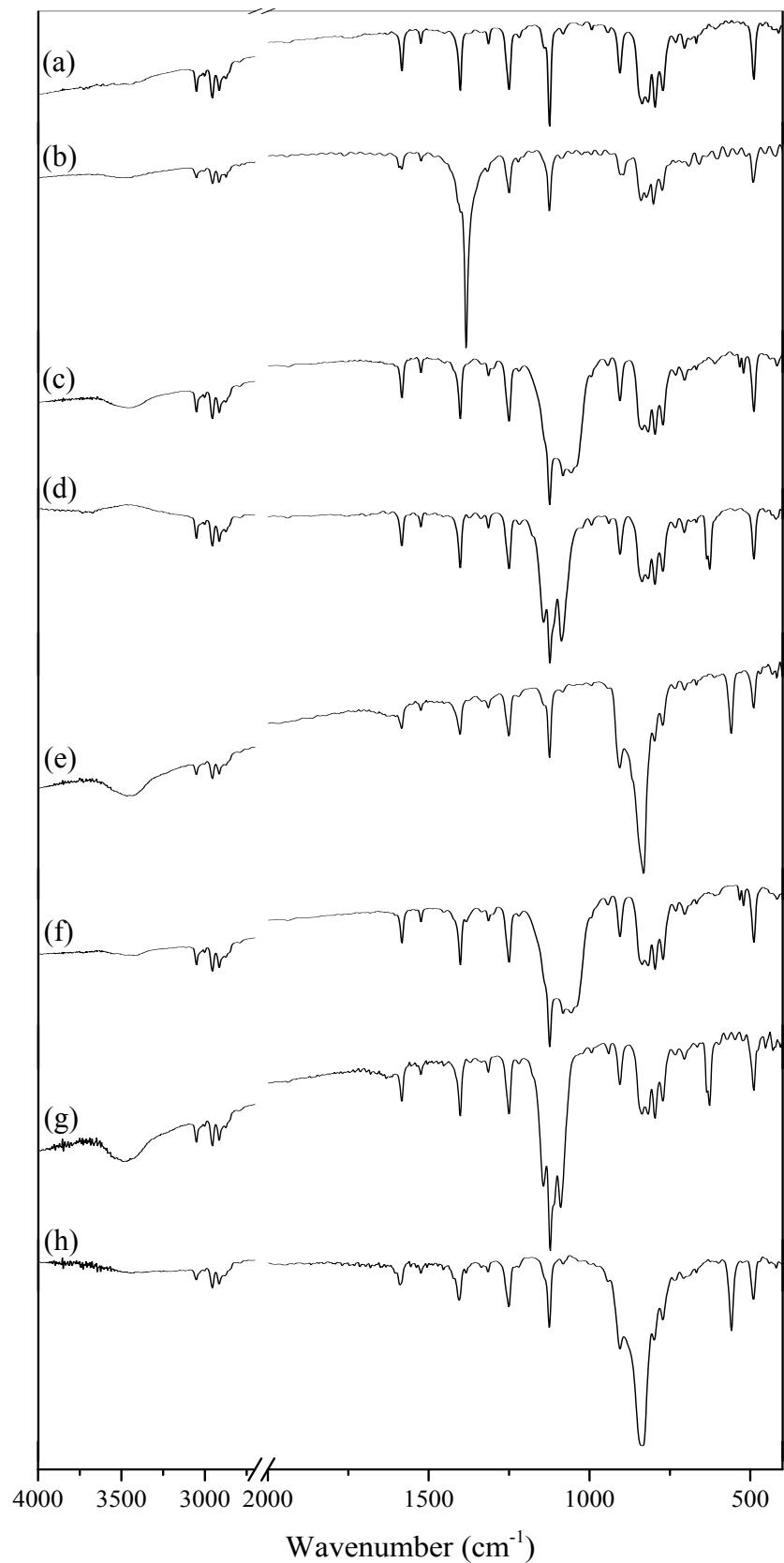
Electronic Supplementary Information (ESI) available:  $^1\text{H}$  NMR spectra and IR spectra of ligand,  $[\text{AgL}]_2(\text{NO}_3)_2$ ,  $[\text{AgL}]_2(\text{BF}_4)_2$ ,  $[\text{AgL}]_2(\text{ClO}_4)_2$ ,  $[\text{AgL}]_2(\text{PF}_6)_2 \cdot 5\text{C}_4\text{H}_8\text{O}_2$ ,  $[\text{AgL}](\text{BF}_4)$ ,  $[\text{AgL}](\text{ClO}_4)$  and  $[\text{AgL}](\text{PF}_6)$ . TGA-DSC curves of  $[\text{AgL}]_2(\text{NO}_3)_2$ ,  $[\text{AgL}]_2(\text{BF}_4)_2$ ,  $[\text{AgL}]_2(\text{ClO}_4)_2$ ,  $[\text{AgL}]_2(\text{PF}_6)_2 \cdot 5\text{C}_4\text{H}_8\text{O}_2$ ,  $[\text{AgL}](\text{BF}_4)$ ,  $[\text{AgL}](\text{ClO}_4)$  and  $[\text{AgL}](\text{PF}_6)$ . Crystal structures of  $[\text{AgL}]_2(\text{NO}_3)_2$ ,  $[\text{AgL}]_2(\text{BF}_4)_2$ ,  $[\text{AgL}]_2(\text{ClO}_4)_2$  and  $[\text{AgL}]_2(\text{PF}_6)_2 \cdot 5\text{C}_4\text{H}_8\text{O}_2$ . Solid-state PL spectra of  $[\text{AgL}](\text{ClO}_4)_2$  and  $[\text{AgL}](\text{ClO}_4)$ .



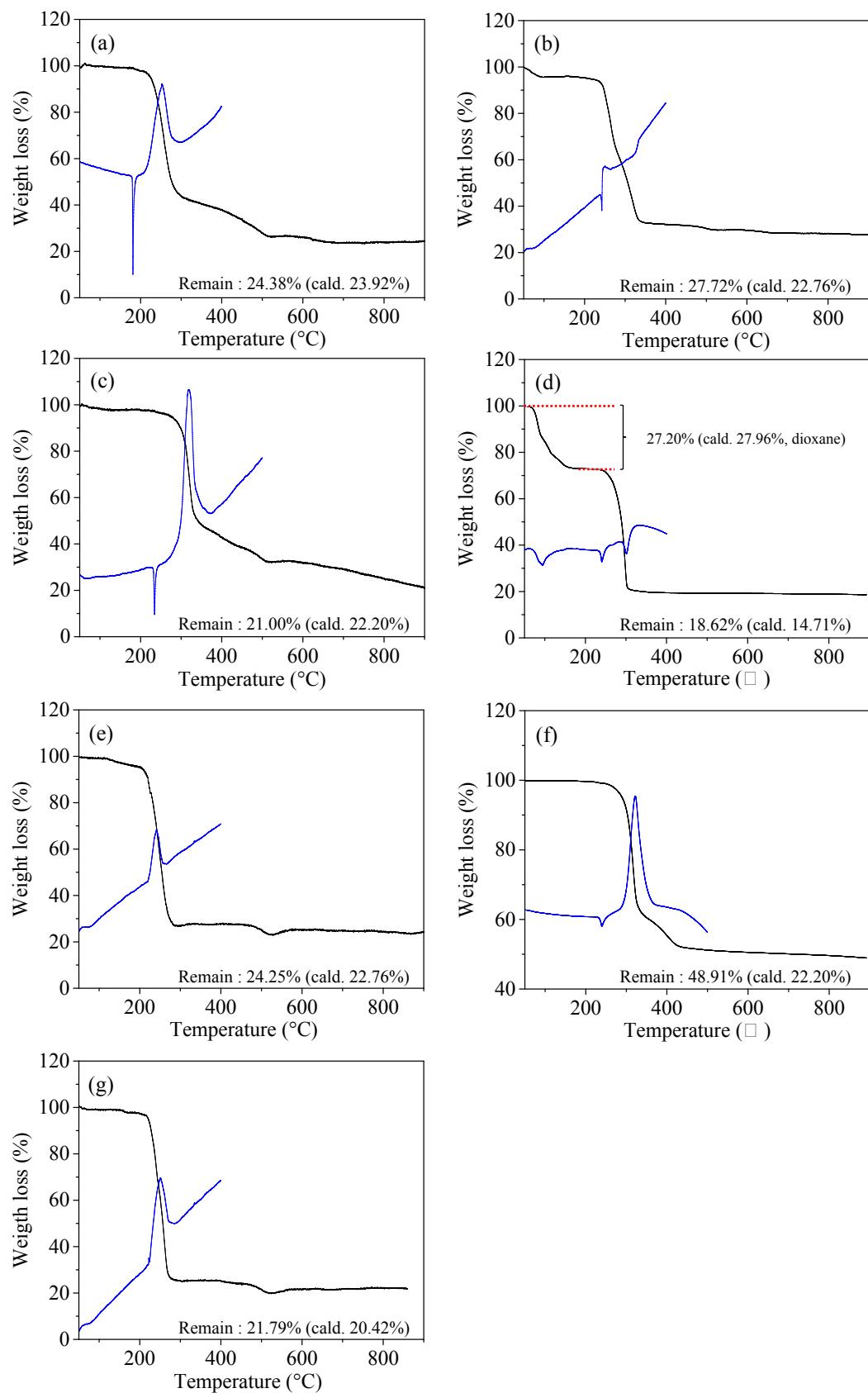
**Fig. S1**  $^1\text{H}$  NMR spectra for L (a),  $[\text{AgL}]_2(\text{NO}_3)_2$  (b),  $[\text{AgL}]_2(\text{BF}_4)_2$  (c),  $[\text{AgL}]_2(\text{ClO}_4)_2$  (d),  $[\text{AgL}]_2(\text{PF}_6)_2 \cdot 5\text{C}_4\text{H}_8\text{O}_2$  (e),  $[\text{AgL}](\text{BF}_4)$  (f),  $[\text{AgL}](\text{ClO}_4)$  (g), and  $[\text{AgL}](\text{PF}_6)$  (h) in  $\text{Me}_2\text{SO}-d_6$ .



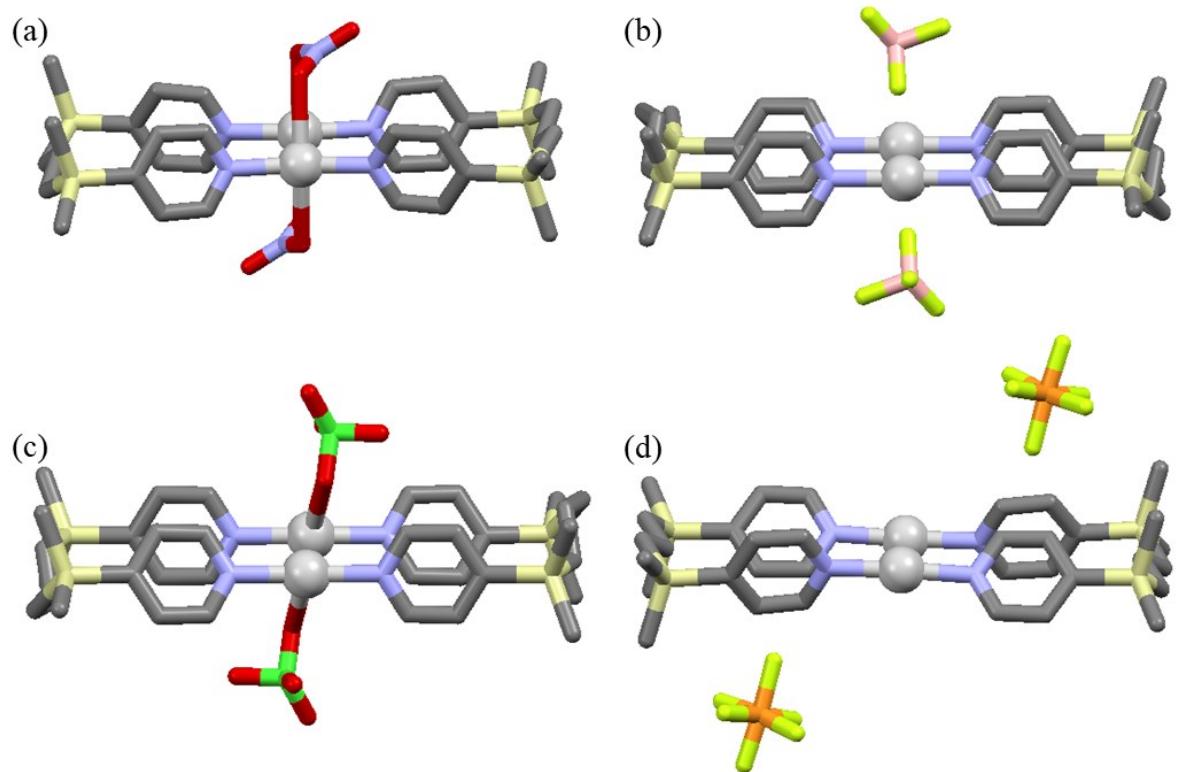
**Fig. S2**  $^1H$  NMR spectra for L (a),  $[AgL]_2(ClO_4)_2$  (b),  $[AgL](ClO_4)$  (c) in acetone- $d_6$ .



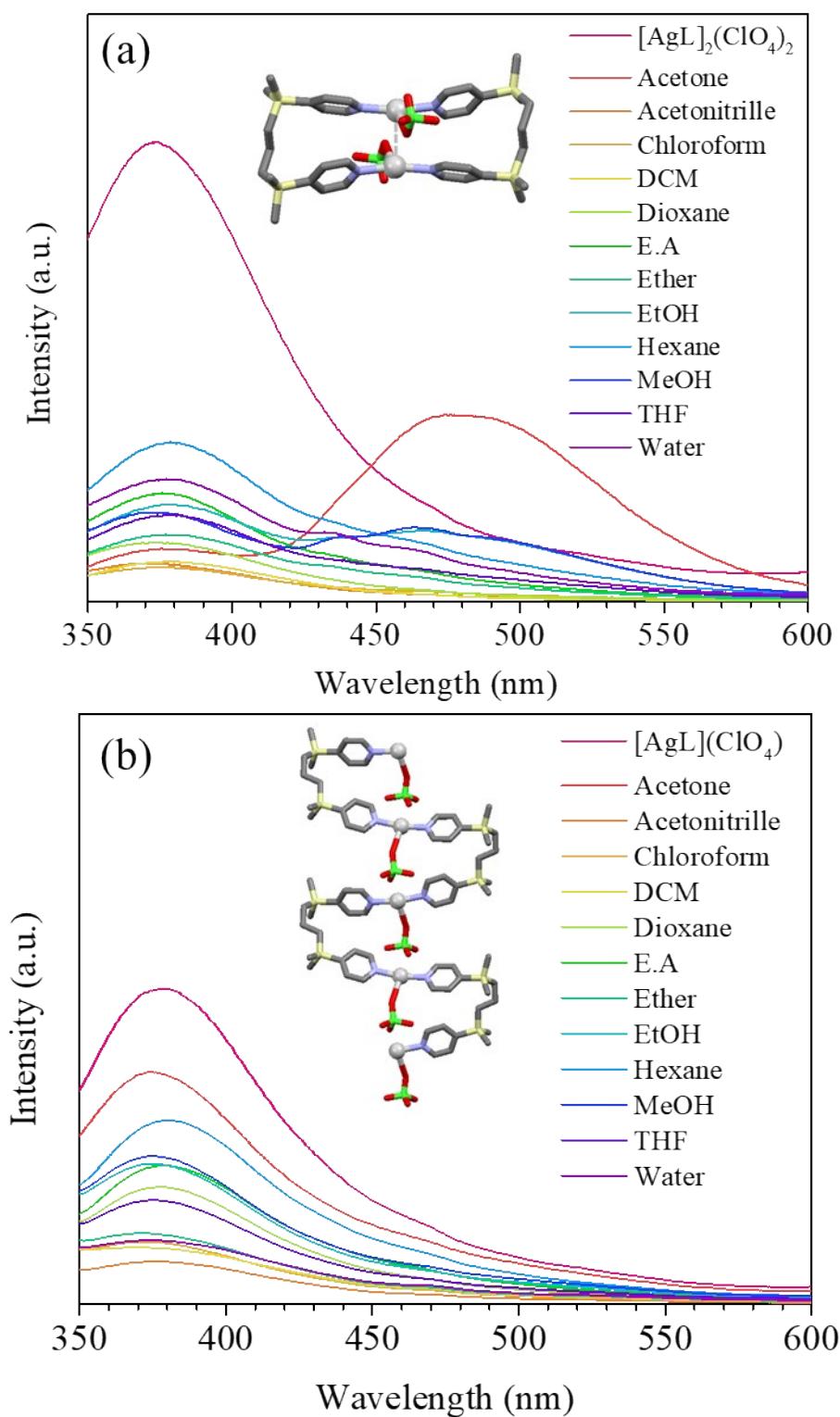
**Fig. S3** IR spectra for L (a), [AgL]<sub>2</sub>(NO<sub>3</sub>)<sub>2</sub> (b), [AgL]<sub>2</sub>(BF<sub>4</sub>)<sub>2</sub> (c), [AgL]<sub>2</sub>(ClO<sub>4</sub>)<sub>2</sub> (d), [AgL]<sub>2</sub>(PF<sub>6</sub>)<sub>2</sub>·5C<sub>4</sub>H<sub>8</sub>O<sub>2</sub> (e), [AgL](BF<sub>4</sub>) (f), [AgL](ClO<sub>4</sub>) (g), and [AgL](PF<sub>6</sub>) (h).



**Fig. S4** TGA and DSC curves for [AgL]<sub>2</sub>(NO<sub>3</sub>)<sub>2</sub> (a), [AgL]<sub>2</sub>(BF<sub>4</sub>)<sub>2</sub> (b), [AgL]<sub>2</sub>(ClO<sub>4</sub>)<sub>2</sub> (c), [AgL]<sub>2</sub>(PF<sub>6</sub>)<sub>2</sub>·5C<sub>4</sub>H<sub>8</sub>O<sub>2</sub> (d), [AgL](BF<sub>4</sub>) (e), [AgL](ClO<sub>4</sub>) (f) and [AgL](PF<sub>6</sub>) (g).



**Fig. S5** Crystal structures (side views) of  $[\text{AgL}]_2(\text{NO}_3)_2$  (a),  $[\text{AgL}]_2(\text{BF}_4)_2$  (b),  $[\text{AgL}]_2(\text{ClO}_4)_2$  (c),  $[\text{AgL}]_2(\text{PF}_6)_2 \cdot 5\text{C}_4\text{H}_8\text{O}_2$  (d).



**Fig. S6** Solid-state PL spectra for  $[\text{AgL}]_2(\text{ClO}_4)_2$  (top, a) and  $[\text{AgL}](\text{ClO}_4)$  (bottom, b) upon additions of a drop of solvent at  $\lambda_{\text{ex}} = 321 \text{ nm}$ .