Non-covalent interactions involving remote substituents influence the topologies of supramolecular chains featuring hydroxyl-O–H···O(hydroxyl) hydrogen bonding in crystals of $(4-YC_6H_4)N(H)C(=S)N(CH_2CH_2OH)_2$ for Y = H, Me, Cl and NO₂

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Electronic Supplementary Information

Table S1. A listing of selected electrostatic potentials and charge deviations (ΔV_{ESP}) for molecules 1-4.

Table S2. List of natural charges for 1-4.

Table S3 The d_{norm} contact distances for all interactions with $d_i + d_e < \Sigma$ (van der Waals radii) in **1-4** as determined through Hirshfeld surface analysis, with all X–H bond lengths adjusted to their neutron values.

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Figure S3 Crystallographic diagrams for **3**: Side-on and end-on views of the supramolecular chain sustained by hydroxyl-O–H \cdots O(hydroxyl) hydrogen bonding, shown as orange dashed lines.

Figure S4 The overall two-dimensional fingerprint plots and selected decomposed fingerprint plots delineated into $H \cdots O/O \cdots H$, $H \cdots C/C \cdots H$ and $H \cdots S/S \cdots H$ surface contacts for (a) 1a, (b) 1b, (c) 2, (d) 3, (e) 4a and (f) 4b, with the internal/external percentage contributions specified for each contact.

Figure S5 Molecular topology obtained through QTAIM analysis, showing the bond critical point (orange dot) and ring critical point (yellow dot) for C–H··· π (C=S) interactions present in (a) **2**, (b) **3** and (c) **4**. The embedded diagram shows the corresponding NCI plot of the interaction for the corresponding pairwise molecules. The analogous C–H··· π (C=S) interaction is absence in **1** and hence, the topology is not shown.

Figure S6 NCI plots for the molecular dimers of (a) **1**, (b) **2**, (c) **3** and (d) **4** featuring complementary interactions involving C–H···O, C–H··· π or C–H···C contacts.

Contact	$V_{\rm ESP}$ (kcal/mol)			
	H-donor	H-acceptor	$\Delta V_{ m ESP}$	
1				
O1−H1o…O2	+55.73	-42.01	97.74	
C8–H8a…S1	+29.03	-36.09	65.12	
C4–H4…S1	+5.92	-35.13	41.05	
2				
O1−H1o…O2	+55.99	-42.45	98.44	
С6–Н6…О2	+12.07	-45.78	57.85	
3				
O1–H1o…O2	+59.12	-39.60	98.72	
С6–Н6…О2	+13.38	-40.42	53.80	
C10–H10b…Cl1	+33.16	+0.76	32.40	
Cl1…Cl1	-0.94	-0.94	-1.88	
4				
O1–H1o…O2	+59.49	-39.30	98.79	
C10–H10a…S1	+36.24	-30.37	66.61	
С11-Н11b…О3	+12.52	-36.91	49.43	
С7–Н7…О3	+10.23	-34.49	44.72	

Table S1. A listing of selected electrostatic potentials and charge deviations (ΔV_{ESP}) for molecules 1-4.

	Natural Charge, e				
Atom	1	2	3	4	
S1	-0.393	-0.397	-0.387	-0.361	
O1	-0.737	-0.737	-0.737	-0.736	
H1O	0.496	0.496	0.496	0.498	
O2	-0.748	-0.749	-0.747	-0.743	
H2O	0.480	0.480	0.480	0.480	
N1	-0.613	-0.611	-0.613	-0.614	
H1N	0.444	0.444	0.445	0.448	
N2	-0.396	-0.397	-0.393	-0.384	
C1	0.345	0.346	0.344	0.338	
C2	0.141	0.126	0.141	0.195	
C3	-0.215	-0.205	-0.197	-0.222	
C4	-0.208	-0.212	-0.224	-0.169	
C5	-0.226	-0.024	-0.053	-0.010	
C6	-0.205	-0.209	-0.222	-0.164	
C7	-0.234	-0.224	-0.216	-0.235	
C8	-0.249	-0.249	-0.249	-0.25	
C9	-0.08	-0.08	-0.08	-0.08	
C10	-0.25	-0.249	-0.25	-0.251	
C11	-0.075	-0.075	-0.075	-0.075	
C5'	-	-0.624	-	-	
Cl1	-	-	-0.011	-	
O3	-	-	-	-0.398	
O4	-	-	-	-0.399	
N3	-	-	-	0.532	
Н3	0.23	0.229	0.237	0.245	
H4	0.222	0.218	0.237	0.252	
Н5	0.222	-	-	-	
Н6	0.223	0.218	0.238	0.252	
H7	0.225	0.224	0.232	0.235	
H8a	0.224	0.223	0.224	0.226	
H8b	0.227	0.227	0.228	0.229	
H9a	0.182	0.18	0.181	0.182	
H9b	0.181	0.181	0.182	0.184	
H10a	0.233	0.233	0.234	0.234	
H10b	0.216	0.216	0.217	0.219	
H11a	0.169	0.169	0.17	0.171	
H11b	0.17	0.17	0.17	0.171	
H5a'	-	0.222	-	-	
H5b'	-	0.22	-	-	
H5c'	-	0.221	-	-	

 Table S2. List of natural charges for 1-4.

Туре	Contact	Distance (Å)	Symmetry operation
1			
Ι	H10…O22	1.70	<i>x</i> , <i>y</i> , <i>z</i>
Ι	H210O2	1.70	x, y, 1+z
Ι	H28b…O1	2.54	1- <i>x</i> , 1- <i>y</i> , 1- <i>z</i>
Ι	H5…O1	2.54	$1^{1/2}-x$, $^{1/2}+y$, $^{1/2}-z$
II	H9a…C22	2.68	1- <i>x</i> , 1- <i>y</i> , 1- <i>z</i>
II	H9a…C23	2.70	1- <i>x</i> , 1- <i>y</i> , 1- <i>z</i>
II	H29b…C7	2.71	$-\frac{1}{2}+x$, $\frac{1}{2}-y$, $\frac{1}{2}+z$
II	H29a…C7	2.73	1- <i>x</i> , 1- <i>y</i> , 1- <i>z</i>
III	H4···S21	2.85	$1^{1/2}-x$, $^{1/2}+y$, $^{1/2}-z$
III	H8a…S21	2.83	$-\frac{1}{2}+x$, $\frac{1}{2}-y$, $-\frac{1}{2}+z$
IV	C21…O22	3.19	1- <i>x</i> , 1- <i>y</i> , 1- <i>z</i>
2			
Ι	H10…O2	1.74	$-\frac{1}{2}-x$, $-\frac{1}{2}+y$, $\frac{1}{2}-z$
Ι	Н6⋯О2	2.55	$\frac{1}{2}-x$, $-\frac{1}{2}+y$, $\frac{1}{2}-z$
II	H10b···C6	2.65	$-\frac{1}{2}+x$, $\frac{1}{2}-y$, $-\frac{1}{2}+z$
II	Н9а…С7	2.71	-1+ <i>x</i> , <i>y</i> , <i>z</i>
II	H4…C10	2.74	$\frac{1}{2}-x$, $\frac{1}{2}+y$, $\frac{1}{2}-z$
3			
Ι	H10…O2	1.73	$-\frac{1}{2}-x$, $\frac{1}{2}+y$, $\frac{1}{2}-z$
Ι	Н6⋯О2	2.51	$\frac{1}{2}-x, \frac{1}{2}+y, \frac{1}{2}-z$
II	H10a···C6	2.67	$-\frac{1}{2}+x$, $\frac{1}{2}-y$, $-\frac{1}{2}+z$
II	H4…C10	2.73	$\frac{1}{2}-x, -\frac{1}{2}+y, \frac{1}{2}-z$
II	Н9b…С7	2.75	-1+ <i>x</i> , <i>y</i> , <i>z</i>

Table S3 The d_{norm} contact distances for all interactions with $d_i + d_e < \Sigma$ (van der Waals radii) in **1-4** as determined through Hirshfeld surface analysis, with all X–H bond lengths adjusted to their neutron values.

II	H8a····C5	2.78	-1+x, y, z
V	H10b…Cl1	2.78	$-1\frac{1}{2}+x$, $1\frac{1}{2}-y$, $-\frac{1}{2}+z$
V	Cl1…Cl1	3.44	2- <i>x</i> , 1- <i>y</i> , 1- <i>z</i>
4			
Ι	H1o…O22	1.75	<i>x</i> , 1+ <i>y</i> , <i>z</i>
Ι	H210…O2	1.76	<i>x</i> , <i>y</i> , <i>z</i>
Ι	H29b…O1	2.29	- <i>x</i> , 1- <i>y</i> , 1- <i>z</i>
Ι	H11b…O3	2.30	<i>x</i> , <i>y</i> , 1+ <i>z</i>
Ι	H7…O23	2.53	<i>x</i> , 1+ <i>y</i> , -1+ <i>z</i>
Ι	H27…O3	2.55	<i>x</i> , <i>y</i> , 1+ <i>z</i>
Ι	H31a…O23	2.56	x, y, -1+z
II	H30a…C5	2.73	1- <i>x</i> , 1- <i>y</i> , 1- <i>z</i>
II	H30a…C4	2.75	1- <i>x</i> , 1- <i>y</i> , 1- <i>z</i>
II	H10b…C24	2.76	1- <i>x</i> , 1- <i>y</i> , 2- <i>z</i>
III	H30b…S1	2.83	1- <i>x</i> , 1- <i>y</i> , 1- <i>z</i>
III	H10a…S21	2.86	1- <i>x</i> , 1- <i>y</i> , 2- <i>z</i>



LUMO(+2)

(a)



LUMO(+1)

(b)



LUMO(+2)

(c)

LUMO(+1)



(d)

Figure S1 HOMO-LUMO mapping for (a) 1, (b) 2, (c) 3 and (d) 4.



(a)



(b)



(d)

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