#### SUPPLEMENTARY INFORMATION

#### Photophysics and photochemistry of carminic acid and related natural pigments

Margherita Zucchelli, <sup>a,b,†</sup> Fernando D. Villarruel, <sup>a,c,†</sup> Pedro David-Gara, <sup>d</sup> Mariana R. Costante, <sup>c</sup> Marcos Tascón, <sup>b, e</sup> Fernando Marte, <sup>b,\*</sup> Fernando S. García Einschlag, <sup>c,\*</sup> Franco M. Cabrerizo <sup>a,\*</sup>

<sup>a</sup> Instituto Tecnológico de Chascomús (INTECH), Universidad Nacional de San Martín (UNSAM) Consejo Nacional de Investigaciones Científicas y Técnicas (CONICET), Av. Intendente Marino Km 8.2,
CC 164 (B7130IWA), Chascomús, Argentina. E-mail: <u>fcabrerizo@intech.gov.ar</u>

<sup>b</sup> IIPC-Tarea, Universidad Nacional de San Martin, Quinquela Martin 1784, CABA, Argentina. E-mail: <u>fmarte.iipc@gmail.com</u>

<sup>c</sup> INIFTA - CONICET, Universidad Nacional de La Plata, Diag. 113 y 64 (1900), La Plata, Argentina. Email: <u>fgarciae@quimica.unlp.edu.ar</u>

<sup>d</sup> Centro de Investigaciones Ópticas (CIOP – CONICET – CIC), Universidad Nacional de La Plata, C.C. 3,
(1897) La Plata, Argentina.

<sup>*e*</sup> Instituto de Investigación e Ingeniería Ambiental (IIIA), CONICET-UNSAM, San Martin, Buenos Aires, Argentina.

\* To whom correspondence should be addressed (<u>fcabrerizo@intech.gov.ar</u>, <u>fgeins@inifta.unlp.edu.ar and</u> <u>fmarte.iipc@gmail.com</u>)

<sup>†</sup>Both authors contributed equally

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1. UV-visible absorption spectra of air-equilibrated CA aqueous solution irradiated at 350 nm and 366 nm



**Figure SI.1.** Evolution of the UV-visible absorption spectra of air-equilibrated CA aqueous solution ([CA]<sub>0</sub> = 340  $\mu$ M) when subject to irradiation: (a) hv = 350 nm, and (b) hv = 366 nm.

## 2. EEMs of CA irradiated aqueous solution (420 nm)



**Figure SI.2.** Evolution of EEMs of air-equilibrated CA aqueous solution ( $[CA]_0 = 110 \mu M$ ) when subject to irradiation (*hv* = 420 nm). Irradiation times are depicted as inset inside each EEM.

# 3. EEMs of CA irradiated aqueous solution (300 nm)



**Figure SI.3.** Evolution of EEMs of air-equilibrated CA aqueous solution ( $[CA]_0 = 110 \mu M$ ) when subject to irradiation (hv = 300 nm). Irradiation times are depicted as inset inside each EEM.

4. Difference UV-visible absorption and emission spectra of CA aqueous solution irradiated under air-equilibrated and N<sub>2</sub>-saturated conditions



**Figure SI.4**. (a) Experimental difference (ED) absorption spectra of CA irradiated (hv = 420 nm; Irradiation time = 24 h) under air-equilibrated (**black**) and N<sub>2</sub>-saturated (**red**) atmosphere. (b) Fluorescence emission spectra of CA recorded under air-equilibrated and N<sub>2</sub>-saturated solutions.

## 5. Hydrogen peroxide production upon CA photoexcitation



**Figure SI.5.** Generation of  $H_2O_2$  upon photoexcitation (hv = 420 nm, irradiation times = 0, 6 and 12 h) of CA in air-equilibrated aqueous solution ([CA]<sub>0</sub> = 110  $\mu$ M).



Absorbance / a.u.



**Figure SI.6. (a)** UV-vis absorption and **(b)** Normalized Difference (ND) spectra of CA solution ( $[CA]_0 = 110 \ \mu\text{M}$ ) irradiated ( $hv = 420 \ \text{nm}$ ) in the absence (**red**) and in the presence (**black**) of SOD (40 U/ml). **(c)** Evolution of EEMs of CA irradiated solutions. Irradiation times are depicted as inset inside each EEM.

#### 7. Effect of sodium azide (NaN<sub>3</sub>) on the photodegradation of CA



**Figure SI.7. (a)** Normalized Difference (ND) UV-vis absorption spectra of CA solution ( $[CA]_0 = 110 \mu M$ ) irradiated (hv = 420 nm, irradiation time = 7 h) in the absence (**black**) and in the presence (**red**) of sodium azide ( $[NaN_3]_0 = 11 \text{ mM}$ ). (b) Evolution of absorbance at 495 nm as a function of the irradiation time.

8. Normalized Difference spectra of CA Lake air-equilibrated aqueous solution irradiated at 300 nm and 420 nm



**Figure SI.8.** Normalized Difference (ND) absorption spectra of Lake aqueous solutions irradiated at hv = 420 nm (red) and 300 nm (black). Total irradiation time = 24 h.

## 9. EEMs of CA Lake irradiated aqueous solution (420 nm)



**Figure SI.9.** Evolution of EEMs of air-equilibrated Lake aqueous solution when subject to irradiation (hv = 420 nm. Irradiation times are depicted as inset inside each EEM.

## 10. EEMs of CA Lake irradiated aqueous solution (300 nm)



**Figure SI.10.** Evolution of EEMs of air-equilibrated Lake aqueous solution when subject to irradiation (hv = 300 nm). Irradiation times are depicted as inset inside each EEM.



#### 11. Comparative analysis of CA, Lake and the photoproducts fluorescence emission spectra

**Figure SI.11. (a)** Normalized fluorescence emission spectra of CA (**red**) and CA-like chromophore (**black**) produced upon irradiation (hv = 300 nm) of Lake. (**b**) Evolution of the total emission intensity of CA-like chromophore (calculated as the integral below the entire emission spectra) as a function of the irradiation time (hv = 300 and 420 nm). (**c**) and (**d**) Normalized excitation (dashed lines,  $\lambda_{em} = 495$  nm) and emission (solid lines,  $\lambda_{exc} = 340$  nm) spectra of photoproducts F#3 and F#4, respectively, obtained during the irradiation (24 h) of CA (**red**) and Lake (**black**).