

Hydrogen Bonds and Halogen Bonds in Complexes of Carbones $L \rightarrow C \leftarrow L$ as Electron Donors to HF and ClF, for $L = CO, N_2, HNC, PH_3,$ and SH_2

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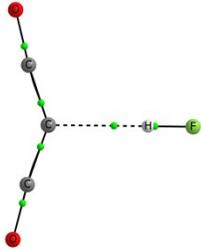
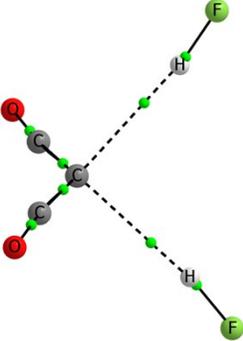
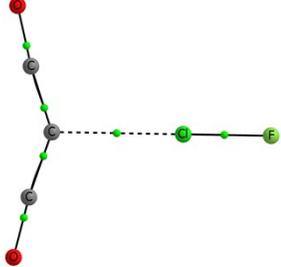
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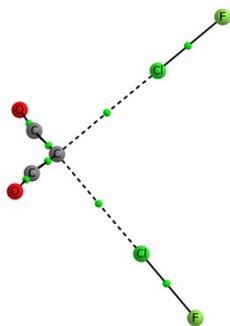
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Pgs. S2-S3.	Table S1. Structures (Å, deg), total energies (au), and molecular graphs of complexes of $OC \rightarrow C \leftarrow CO$ with one and two HF and ClF molecules.
Pgs. S4-S5.	Table S2. Structures (Å, deg), total energies (au), and molecular graphs of complexes of $NN \rightarrow C \leftarrow NN$ with one and two HF and ClF molecules.
Pgs. S6-S7.	Table S3. Structures (Å, deg), total energies (au), and molecular graphs of complexes of $HNC \rightarrow C \leftarrow CNH$ with one and two HF and ClF molecules.
Pgs. S8-S9.	Table S4. Structures (Å, deg), total energies (au), and molecular graphs of complexes of $H_3P \rightarrow C \leftarrow PH_3$ with one and two HF and ClF molecules.
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Pgs. S12-S13	Table S6. Paramagnetic spin orbit (PSO), diamagnetic spin orbit (DSO), Fermi contact (FC), and spin dipole (SD) components and total ${}^2hJ(F-C)$ (Hz) for coupling across hydrogen bonds in complexes with one and two HF molecules, and ${}^1xJ(Cl-C)$ (Hz) for coupling across halogen bonds in complexes with 1 and 2 ClF molecules.

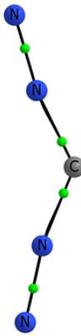
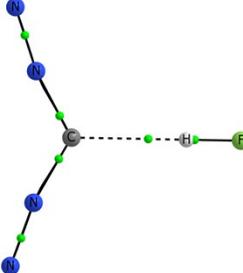
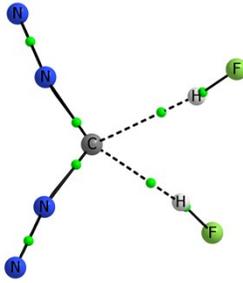
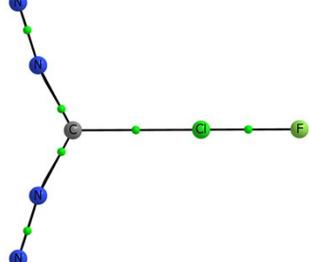
Table S1. Structures (Å, deg), total energies (au), and molecular graphs of complexes of OC→C←CO with one and two HF and ClF molecules.

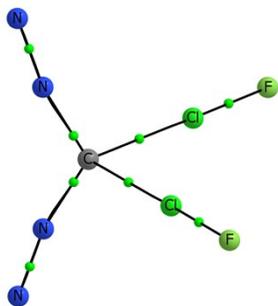
	<p>occo MP2= -264.29338266 NIMAG= 0 C X,1,1. C,1,r1,2,a1 C,1,r1,2,a1,3,180.,0 O,1,r2,2,a2,4,180.,0 O,1,r2,2,a2,3,180.,0</p> <p>r1=1.278099 r2=2.448793 a1=95.966 a2=95.133</p>
	<p>occo_hf_c2v MP2= -364.63865271 NIMAG= 0 C,0,0,0,0.28834188 C,0,1,2.2477998,0.70779174 C,0,-1,2.2477998,0.70779174 O,0,2.35481081,0.98938332 O,0,-2.35481081,0.98938332 H,0,0,-1.76735027 F,0,0,-2.69848166</p>
	<p>occo_2hf MP2= -464.98179312 NIMAG= 0 C,0,0,0,0.3504254809 C,0,1,2.032127612,0.8554973592 C,0,-1,2.032127612,0.8554973592 O,0,2.3129354785,1.1951814587 O,0,-2.3129354785,1.1951814587 H,-1.5312063788,0,-1.0935531994 H,1.5312063788,0,-1.0935531994 F,-2.2943101987,0,-1.6232500574 F,2.2943101987,0,-1.6232500574</p>
	<p>occo_clf MP2= -823.66354607 NIMAG= 0 C,0,0,0,0.4392692222 C,0,1,2.271109383,0.8555149771 C,0,-1,2.271109383,0.8555149771 O,0,2.3587629508,1.1293137208 O,0,-2.3587629508,1.1293137208 Cl,0,0,-2.0601043485 F,0,0,-3.7319622285</p>



occo_2clf MP2= -1383.03276020 NIMAG= 0
C,0.,0.,0.6226376912
C,0.,1.2176798287,1.0861864783
C,0.,-1.2176798287,1.0861864783
O,0.,2.3386460841,1.3893269997
O,0.,-2.3386460841,1.3893269997
Cl,-1.8632267091,0.,-1.167328749
Cl,1.8632267091,0.,-1.167328749
F,-3.0640898633,0.,-2.3154403102
F,3.0640898633,0.,-2.3154403102

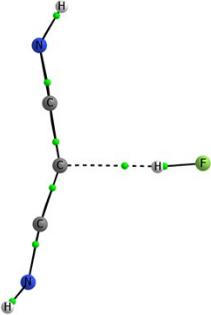
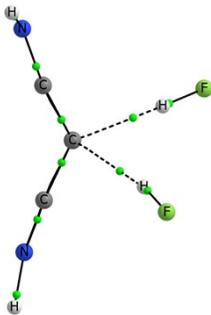
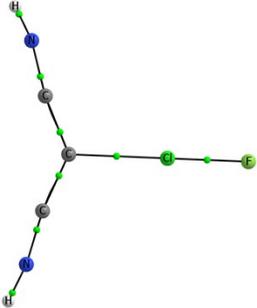
Table S2. Structures (Å, deg), total energies (au), and molecular graphs of complexes of NN→C←NN with one and two HF and ClF molecules.

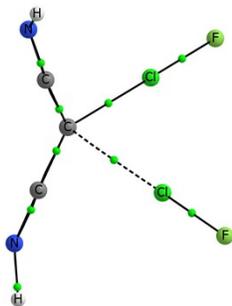
 <p>The diagram shows a central carbon atom (grey) bonded to two nitrogen atoms (blue) in a bent arrangement. The nitrogen atoms are further bonded to hydrogen atoms (white) and fluorine atoms (green), forming a chain-like structure.</p>	<p>n2cn2 MP2= -256.56479818 NIMAG= 0 C X,1,1. N,1,r1,2,a1 N,1,r1,2,a1,3,180.,0 N,1,r2,2,a2,4,180.,0 N,1,r2,2,a2,3,180.,0</p> <p>r1=1.26707563 r2=2.42717611 a1=114.27859507 a2=109.60147599</p>
 <p>The diagram shows a central carbon atom (grey) bonded to two nitrogen atoms (blue). One nitrogen is bonded to two hydrogens (white) and one fluorine (green). The other nitrogen is bonded to two hydrogens (white) and one fluorine (green). A dashed line indicates a hydrogen bond between a hydrogen atom and a fluorine atom.</p>	<p>n2cn2_hf_c2v MP2= -356.91190534 NIMAG= 0 C,0,0,0,0.12040519 N,0,.,1.12833163,0.74731132 N,0,.-1.12833163,0.74731132 N,0,.,2.21901314,1.12240416 N,0,.-2.21901314,1.12240416 H,0,0,.-1.84822557 F,0,0,.-2.78335803</p>
 <p>The diagram shows a central carbon atom (grey) bonded to two nitrogen atoms (blue). Each nitrogen is bonded to two hydrogens (white) and one fluorine (green). Two dashed lines indicate hydrogen bonds between the hydrogens of one nitrogen and the fluorines of the other nitrogen.</p>	<p>n2cn2_2hf MP2= -457.25669476 NIMAG= 0 C,0.0000000348,0.23432971,0. N,0.0000001377,0.91712429,1.11636807 N,0.0000001377,0.91712429,-1.11636807 N,0.0000001999,1.32970398,2.18522301 N,0.0000001999,1.32970398,-2.18522301 H,-1.42190871,-1.18048332,0. H,1.4219083531,-1.1804837486,0. F,-2.20083722,-1.69447812,0. F,2.2008367081,-1.6944787834,0.</p>
 <p>The diagram shows a central carbon atom (grey) bonded to two nitrogen atoms (blue). One nitrogen is bonded to two hydrogens (white) and one fluorine (green). The other nitrogen is bonded to two hydrogens (white) and one chlorine (green). A dashed line indicates a hydrogen bond between a hydrogen atom and a fluorine atom.</p>	<p>n2cn2_clf MP2= -815.94038615 NIMAG= 0 C,0,0,0,0.3274875081 N,0,.,1.1373100471,0.933216079 N,0,.-1.1373100471,0.933216079 N,0,.,2.2356897269,1.2816031898 N,0,.-2.2356897269,1.2816031898 Cl,0,0,.-1.9074965694 F,0,0,.-3.6213769262</p>



n2cn2_2clf MP2= -1375.31392417 NIMAG= 0
 C,0.0000000001,0.,0.3461521923
 N,0.0000000001,1.1251504367,1.0186234049
 N,0.0000000001,-1.1251504367,1.0186234049
 N,0.0000000002,2.2063803792,1.3949269225
 N,0.0000000002,-2.2063803792,1.3949269225
 Cl,-1.8509820125,0.,-1.0505355792
 Cl,1.8509820123,0.,-1.0505355796
 F,-3.2272028758,0.,-2.0422253554
 F,3.2272028753,0.,-2.0422253562

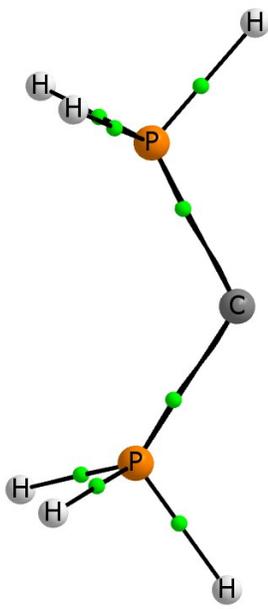
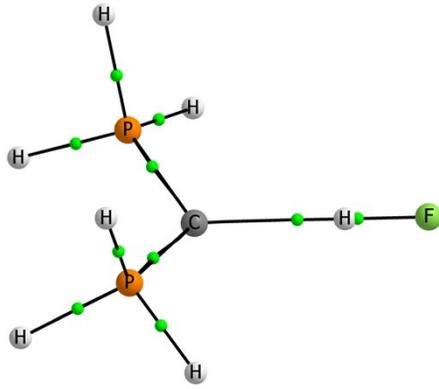
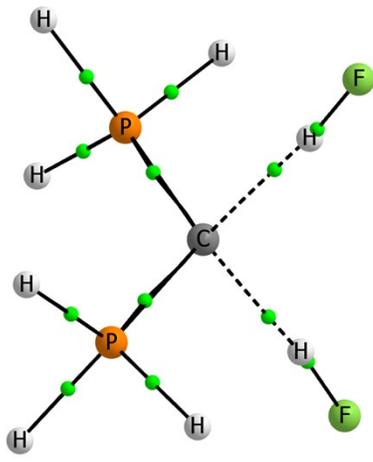
Table S3. Structures (Å, deg), total energies (au), and molecular graphs of complexes of HNC→C←CNH with one and two HF and ClF molecules.

	<p>hncccnh MP2= -224.49828538 NIMAG= 0 C,0,0,-0.0727410902 C,-1.2777541366,0.015305167,0.0077287239 C,1.2777541366,-0.015305167,0.0077287239 N,-2.4959363321,-0.088106447,-0.0479416778 N,2.4959363321,0.088106447,-0.0479416778 H,-3.1391210905,0.4521763263,0.5140337039 H,3.1391210905,-0.4521763263,0.5140337039</p>
	<p>hncccnh_hf_2 MP2= -324.84631720 NIMAG= 0 C,0.025855665,-0.3979920643,-0.0317893391 C,0.1841603418,-0.6401983839,1.2290566534 C,0.21511312,-0.7137646619,-1.2725233241 N,0.2282660238,-0.9093448869,2.4105826751 N,0.1795367962,-0.9170026582,-2.4689658497 H,0.6329667368,-0.3835777431,3.1697390467 H,0.9381958626,-1.1491725868,-3.0906529625 H,-0.4561089312,1.5467894875,0.0446170894 F,-0.6209856151,2.4611864976,0.1549360107</p>
	<p>hncccnh_clf_cs MP2= -783.87597986 NIMAG= 0 C,0.02688357,0.85464306,0. C,-0.02246988,1.3796153,1.20599586 C,-0.02246988,1.3796153,-1.20599586 N,0.08903734,1.73637556,2.34731754 N,0.08903734,1.73637556,-2.34731754 H,-0.56066104,1.93162951,3.09076707 H,-0.56066104,1.93162951,-3.09076707 Cl,0.00602361,-1.30361233,0. F,-0.01330197,-3.07334689,0.</p>
	<p>hncccnh_2hf MP2= -425.19351407 NIMAG= 0 C,0.1694741939,0.4133533394,0. C,0.0681299915,0.9258423686,1.205488668 C,0.0681299915,0.9258423686,-1.205488668 N,0.1320682366,1.2910133393,2.3477813064 N,0.1320682366,1.2910133393,-2.3477813064 H,-0.5227183113,1.4396504955,3.0953767211 H,-0.5227183113,1.4396504955,-3.0953767211 H,1.3780930569,-1.1356032572,0. H,-1.3313288755,-0.9408341046,0. F,1.9905693891,-1.8448503324,0. F,-2.1548686351,-1.3836076298,0.</p>



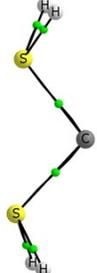
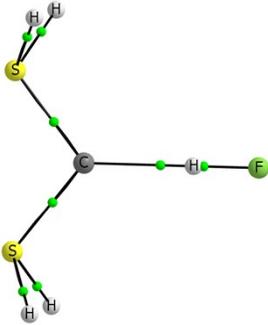
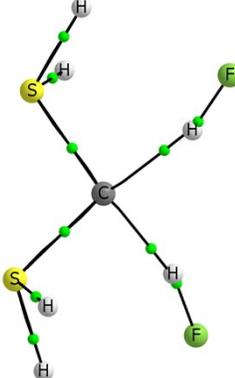
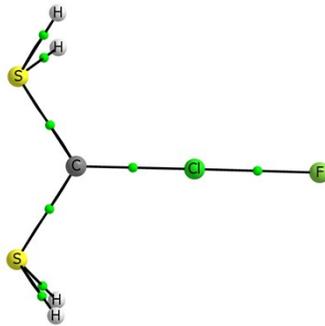
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 C,0.3357542363,0.5939811706,0.
 C,0.1217887327,1.0867625783,1.2084329851
 C,0.1217887327,1.0867625783,-1.2084329851
 N,0.0996139036,1.4617184883,2.3427688888
 N,0.0996139036,1.4617184883,-2.3427688888
 H,-0.5275600911,1.4355898664,3.1281612559
 H,-0.5275600911,1.4355898664,-3.1281612559
 Cl,1.7702191452,-1.0960962607,0.
 Cl,-1.7797177138,-0.8296738095,0.
 F,2.9074618733,-2.4161139948,0.
 F,-3.1690571101,-1.7666616793,0.

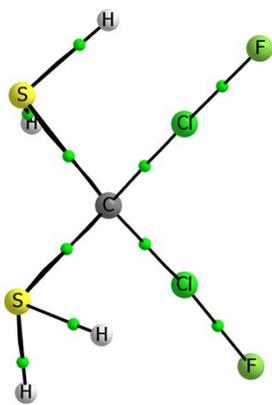
Table S4. Structures (Å, deg), total energies (au), and molecular graphs of complexes of $\text{H}_3\text{P}\rightarrow\text{C}\leftarrow\text{PH}_3$ with one and two HF and ClF molecules.

 <p>The image shows the molecular structure of the ph3cph3 complex. It consists of two phosphorus atoms (P, orange) and one carbon atom (C, grey). Each phosphorus atom is bonded to three hydrogen atoms (H, white). The carbon atom is bonded to both phosphorus atoms, forming a central C-P-P bridge. The phosphorus atoms are also bonded to each other, forming a P-P bond. The structure is shown in a perspective view.</p>	<p>ph3cph3 MP2= -723.24780957 NIMAG= 0 C X,1,1. P,1,r1,2,a1 P,1,r1,2,a1,3,180.,0 H,3,r2,1,a2,2,0.,0 H,4,r2,1,a2,2,0.,0 H,3,r3,1,a3,2,d3,0 H,3,r3,1,a3,2,-d3,0 H,4,r3,1,a3,2,d3,0 H,4,r3,1,a3,2,-d3,0</p> <p>r1=1.65748192 r2=1.40235366 r3=1.42132697 a1=119.87711784 a2=112.22000049 a3=121.80824346 d3=117.38873328</p>
 <p>The image shows the molecular structure of the ph3cph3_hf complex. It consists of two phosphorus atoms (P, orange), one carbon atom (C, grey), and one hydrogen fluoride molecule (HF, white and green). Each phosphorus atom is bonded to three hydrogen atoms (H, white). The carbon atom is bonded to both phosphorus atoms and the hydrogen atom of the HF molecule. The phosphorus atoms are also bonded to each other. The structure is shown in a perspective view.</p>	<p>ph3cph3_hf MP2= -823.61200529 NIMAG= 0 C,0.,-0.3319616825,-0.1966273382 P,1.458436208,-0.0443277036,-0.9383488443 P,-1.458436208,-0.0443277036,-0.9383488443 H,2.5269454885,-0.138816188,-0.0390045554 H,-2.5269454885,-0.138816188,-0.0390045554 H,1.9400601182,-0.8814524035,-1.9754472824 H,1.7174967,1.211075746,-1.5397666355 H,-1.7174967,1.211075746,-1.5397666355 H,-1.9400601182,-0.8814524035,-1.9754472824 H,0.,-1.4738086288,1.1148845846 F,0.,-2.1038252147,1.847770359</p>
 <p>The image shows the molecular structure of the ph3cph3_hf2 complex. It consists of two phosphorus atoms (P, orange), one carbon atom (C, grey), and two hydrogen fluoride molecules (HF, white and green). Each phosphorus atom is bonded to three hydrogen atoms (H, white). The carbon atom is bonded to both phosphorus atoms and the hydrogen atoms of both HF molecules. The phosphorus atoms are also bonded to each other. The structure is shown in a perspective view.</p>	<p>ph3cph3_hf2 MP2= -923.97263358 NIMAG= 0 C,0.,0.,-0.0618881618 P,1.4586416725,0.,-0.8875404704 P,-1.4586416725,0.,-0.8875404704 H,2.5366944211,0.,0.0018151859 H,-2.5366944211,0.,0.0018151859 H,1.7969573545,-1.080127172,-1.7323299583 H,1.7969573545,1.080127172,-1.7323299583 H,-1.7969573545,1.080127172,-1.7323299583 H,-1.7969573545,-1.080127172,-1.7323299583 H,0.,-1.4779508483,0.9342302452 F,0.,-2.3277938952,1.3860682585 H,0.,1.4779508483,0.9342302452 F,0.,2.3277938952,1.3860682585</p>

	<p>ph3cph3_clf_cs MP2= -1282.66020425 NIMAG= 0 C,-0.5702208707,0.7820187856,0. P,0.0877149421,1.2013921191,1.4940238996 P,0.0877149421,1.2013921191,-1.4940238996 H,-0.6418889533,0.5951188924,2.5189372346 H,-0.6418889533,0.5951188924,-2.5189372346 H,1.4091888811,0.823366933,1.8154764299 H,0.0736271937,2.5629495925,1.8454539394 H,0.0736271937,2.5629495925,-1.8454539394 H,1.4091888811,0.823366933,-1.8154764299 Cl,-0.1644036178,-1.22931212,0. F,0.2116043722,-3.0850092358,0.</p>
	<p>ph3cph3_clf2 MP2= -1842.06423199 NIMAG= 0 C,0.,0.,-0.197510476 P,1.5473265889,0.,-0.9366014009 P,-1.5473265889,0.,-0.9366014009 H,2.4886749959,0.,0.0920023721 H,-2.4886749959,0.,0.0920023721 H,1.8634832179,-1.112175768,-1.7293150167 H,1.8634832179,1.112175768,-1.7293150167 H,-1.8634832179,1.112175768,-1.7293150167 H,-1.8634832179,-1.112175768,-1.7293150167 Cl,0.,-1.8044991034,0.7579002304 F,0.,-3.4682002611,1.5930108333 Cl,0.,1.8044991034,0.7579002304 F,0.,3.4682002611,1.5930108333</p>

Table S5. Structures (Å, deg), total energies (au), and molecular graphs of complexes of $\text{H}_2\text{S} \rightarrow \text{C} \leftarrow \text{SH}_2$ with one and two HF and ClF molecules.

 <p>The structure shows a central carbon atom (grey) bonded to two sulfur atoms (yellow). Each sulfur atom is bonded to two hydrogen atoms (white). The sulfur atoms are positioned above and below the carbon atom, with their respective hydrogens extending outwards.</p>	<p>sh2csh2 MP2= -835.64105675 NIMAG= 0 C,0.,0.,0.9156757047 S,0.,1.321528952,-0.1985578964 S,0.,-1.321528952,-0.1985578964 H,-0.9564879678,2.1958664427,0.2168109512 H,0.9564879678,-2.1958664427,0.2168109512 H,0.9564879678,2.1958664427,0.2168109512 H,-0.9564879678,-2.1958664427,0.2168109512</p>
 <p>The structure shows a central carbon atom (grey) bonded to two sulfur atoms (yellow) and one hydrogen atom (white). The sulfur atoms are bonded to two hydrogen atoms each. The hydrogen atom is bonded to a fluorine atom (green). The fluorine atom is positioned to the right of the carbon atom.</p>	<p>sh2csh2_hf MP2= -936.01013497 NIMAG= 0 C,0.,0.,0.9640528482 S,0.,1.3418492879,-0.0945105931 S,0.,-1.3418492879,-0.0945105931 H,-0.9617198222,2.1946574259,0.3453265972 H,0.9617198222,-2.1946574259,0.3453265972 H,0.9617198222,2.1946574259,0.3453265972 H,-0.9617198222,-2.1946574259,0.3453265972 H,0.,0.,2.6228632357 F,0.,0.,3.6066024262</p>
 <p>The structure shows a central carbon atom (grey) bonded to two sulfur atoms (yellow) and two hydrogen atoms (white). Each sulfur atom is bonded to two hydrogen atoms. The two hydrogen atoms are bonded to two fluorine atoms (green). The fluorine atoms are positioned to the right of the carbon atom.</p>	<p>sh2csh2_2hf_2 MP2= -1036.37304874 NIMAG= 0 C,0.,0.,0.1223139218 S,1.342779589,0.2488535012,-0.9319605674 S,-1.342779589,-0.2488535012,-0.9319605674 H,1.9729297489,1.3313596823,-0.4158558868 H,-1.9729297489,-1.3313596823,-0.4158558868 H,2.3271173107,-0.5758722217,-0.4997453782 H,-2.3271173107,0.5758722217,-0.4997453782 H,-0.1990084851,-1.3265015897,1.1693142726 H,0.1990084851,1.3265015897,1.1693142726 F,-0.547177965,-2.1459717069,1.5655800979 F,0.547177965,2.1459717069,1.5655800979</p>
 <p>The structure shows a central carbon atom (grey) bonded to two sulfur atoms (yellow) and one hydrogen atom (white). The sulfur atoms are bonded to two hydrogen atoms each. The hydrogen atom is bonded to a chlorine atom (green), which is in turn bonded to a fluorine atom (green). The chlorine and fluorine atoms are positioned to the right of the carbon atom.</p>	<p>sh2csh2_clf_2_cs MP2= -1395.07297178 NIMAG= 0 C,0.5410515897,-0.4385920457,0. S,1.385213613,0.0866048172,1.3952385412 S,1.385213613,0.0866048172,-1.3952385412 H,0.687808939,1.1415365109,1.9255657793 H,0.8604759567,-0.7551203492,-2.3085802082 H,0.8604759567,-0.7551203492,2.3085802082 H,0.687808939,1.1415365109,-1.9255657793 Cl,-1.2794012454,-0.1709949123,0. F,-3.2122183692,0.2215653473,0.</p>



sh2csh2_2clf_2_cs MP2= -1954.47793911 NIMAG= 0
 C,-0.2777685223,0.0499668431,0.
 S,-1.3187088323,-0.0673553052,-1.4708006374
 S,-1.3187088323,-0.0673553052,1.4708006374
 H,-1.2037371574,1.2074080245,-1.9201681324
 H,-0.3239526424,-0.5464710293,2.246524511
 H,-0.3239526424,-0.5464710293,-2.246524511
 H,-1.2037371574,1.2074080245,1.9201681324
 Cl,0.6029390113,-1.7281147203,0.
 Cl,0.6265188278,1.7230393204,0.
 F,1.4745926442,-3.3788087609,0.
 F,1.4187797445,3.4468264846,0.

Table S6. Paramagnetic spin orbit (PSO), diamagnetic spin orbit (DSO), Fermi contact (FC), and spin-dipole (SD) components and total ${}^2J(\text{F-C})$ (Hz) for coupling across hydrogen bonds in complexes with one and two HF molecules, and ${}^1J(\text{Cl-C})$ (Hz) for coupling across halogen bonds in complexes with 1 and 2 ClF molecules.

L = CO	PSO	DSO	FC	SD	${}^2J(\text{F-C})$
HF	-1.7	0.2	18.2	0.2	16.8
2HF	-1.8	0.2	14.5	0.2	13
					${}^1J(\text{Cl-C})$
ClF	1.8	0	27.6	0.3	29.7
2ClF	1.6	0.1	22.4	0.3	24.3

L = NN	PSO	DSO	FC	SD	${}^2J(\text{F-C})$
HF	-0.9	0.2	62.1	0.4	61.8
2HF	-1.3	0.2	45.4	0.1	44.4
					${}^1J(\text{Cl-C})$
ClF	1.3	0.1	64.6	0.1	66.1
2ClF	1.6	0.1	52.9	0.2	54.7

L = HNC	PSO	DSO	FC	SD	${}^2J(\text{F-C})$
HF	-2.1	0.1	19.1	0.2	17.3
2HF	-2.0	0.2	37.2	0.1 ^a	35.5
	-2.7	0.2	23.9	0.2 ^a	21.5 ^b
					${}^1J(\text{Cl-C})$
ClF	3.3	0.1	52.2	0.9	56.5
2ClF	2.9	0.1	53.2	0.9 ^a	57.1
	2.4	0.1	20.3	0.9 ^a	23.7 ^b

L = PH ₃	PSO	DSO	FC	SD	² hJ(F-C)
HF	-2.8	0.3	139	0	136.5
2HF	-3.7	0.3	98.5	0.8	95.2
					¹ xJ(C-Cl)
ClF	7.7	0.1	33.7	3.3	44.8
2ClF	4.8	0.1	46.8	2	53.7

L = SH ₂	PSO	DSO	FC	SD	² hJ(F-C)
HF	0.0	0.3	289.2	0.8	290.3
2HF	-1.5	0.3	166.4	-0.9	164.3
					¹ xJ(C-Cl)
ClF	-0.8	0.1	3.6	0.6	3.5

- a) Estimated.
b) For the interaction at the longer intermolecular distance.