## Supplementary Information for

## Surface decoration of phosphorene nanoribbons via 4d transition metal atoms for spintronics

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Fig. S1 (a-b) Schemetic views and (c-f) relaxed structures of X-adsorbed hydrogenated armchair phosphorene nanoribbons (X = Y, Zr, Nb and Mo) with hollow, top, bridge and valley sites.

Atoms	Adsorption Site	$E_{\rm f}({\rm eV})$	М
	Hollow site	-3.85	0.70
V	Top site	-3.86	0.72
Ŷ	Bridge site	-3.86	0.72
	Valley site	-3.86 -3.86 -5.29 -4.62 -4.62 -4.42 -6.24 -4.29 -4.29	0.00
	Hollow site	-4.62	1.58
7	Valley site-5.29Hollow site-4.62Top site-4.62Bridge site-4.42Valley site-6.24Hollow site-4.29Top site-4.29	1.59	
Zr	Bridge site	-4.42	1.52
	Valley site	-6.24	0.00
	Hollow site	-4.29	0.95
NTL	Top site		0.94
IND	Bridge site		0.94
	Valley site	-6.04	0.00
Мо	Hollow site	-2.55	3.45
	Top site	-2.55	3.45
	Bridge site	-1.52	5.19
	Valley site	-4.37	0.00

**Table S1** The formation energy ( $E_f$ ) and magnetic moment (M) of the X-adsorbed (X = Y, Zr, Nb and Mo) phosphorene nanoribbons with different initial adsorption sites.

**Table S2.** Parameters for the adsorption with different hollow sites (H1-H4) and the substitutional doping (S) of 4d transition metal atoms (X = Y, Zr, Nb and Mo) on hydrogenated armchair phosphorene nanoribbons, including the distortion of nanoribbons ( $\Delta Y_{\text{max}}$ , the maximal displacement in the Y direction of P atoms from their initial position), the minimum X-P distance ( $d_{X-P}$ ), the formation energy ( $E_f$ ) and the magnetic moment (M).

Atoms	Site	$\Delta Y_{\rm max}({\rm \AA})$	$d_{\mathrm{X-P}}(\mathrm{\AA})$	$E_{\rm f}({\rm eV})$	M
Y	H1	0.338	2.68	-3.94	0.72
	H2	0.106	2.70	-3.84	0.75
	H3	0.143	2.72	-3.84	0.71
	H4	0.150	2.72	-3.85	0.70
	S	0.468	2.69	-5.93	0.00
Zr	H1	0.400	2.53	-4.68	1.48
	H2	0.122	2.55	-4.58	1.60
	H3	0.130	2.55	-4.61	1.59
	H4	0.137	2.55	-4.62	1.58
	S	0.573	2.58	-6.84	0.63
Nb	H1	0.436	2.39	-4.40	0.94
	H2	0.138	2.36	-4.25	0.95
	H3	0.149	2.36	-4.28	0.94
	H4	0.149	2.36	-4.29	0.95
	S	0.567	2.44	-6.40	0.00
Мо	H1	0.0904	2.44	-2.40	3.59
	H2	0.0782	2.45	-2.51	3.50
	H3	0.0866	2.45	-2.54	3.47
	H4	0.0899	2.45	-2.55	3.45
	S	0.2783	2.48	-4.51	2.82



Fig. S2 Charge density of (a) VBM and (b) CBM of the pure phosphorene nanoribbon.



**Fig. S3** Transmission spectra of the Mo-adsorbed phosphorene nanoribbons with central region containing 3-, 4- and 5-units for parallel spin configurations, respectively. The consistency of these transmission spectra indicates the convergence of the length of nanoribbons taken account in the transport calculation.



**Fig. S4** (a) The relaxed geometric structure and (b) the band structure of the pristine phosphorene nanoribbon with the single-atom vacancy.



**Fig. S5** (a)-(c) The schematic structure and the band structure of the X-adsorbed phosphorene nanoribbons (X=Y, Zr, Nb and Mo) with different concentrations.

Atoms	Concentration (%)	Band gap (eV)	М
	1.1	0.25	0.66
Y	1.5	0.19	0.70
	2.2	0.05	0.68
	1.1	0.48	1.60
Zr	1.5	0.43	1.58
	2.2	0.29	1.57
	1.1	0.53	0.95
Nb	1.5	0.52	0.95
	2.2	0.44	0.95
	1.1	0.12	3.46
Мо	1.5	0.13	3.45
	2.2	0.08	3.37

**Table S3** The band gap and magnetic moment (*M*) for the X-adsorbed phosphorene nanoribbons (X=Y, Zr, Nb and Mo) with the concentration of 1.1%, 1.5% and 2.2%.



Fig. S6 (a) and (b) The variation of the band gap and magnetic moment of the X-adsorbed phosphorene nanoribbons (X=Y, Zr, Nb and Mo) with adsorption concentration (1.1%, 1.5% and 2.2%).