

## New insight into the interaction between divacancy and H/He impurity in $\text{Ti}_3\text{AlC}_2$ by first-principles studies

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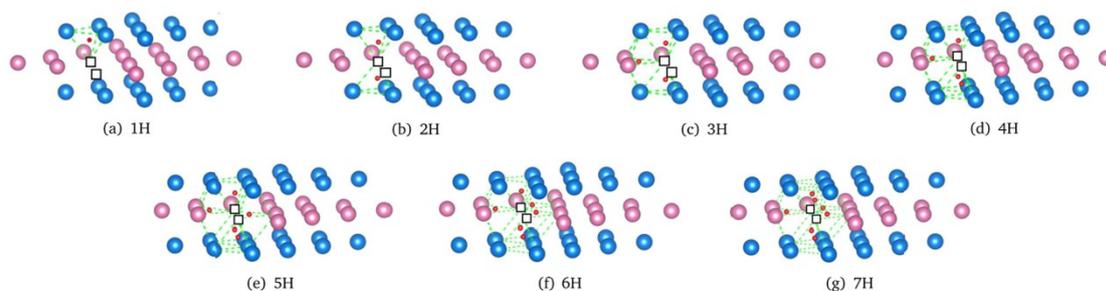
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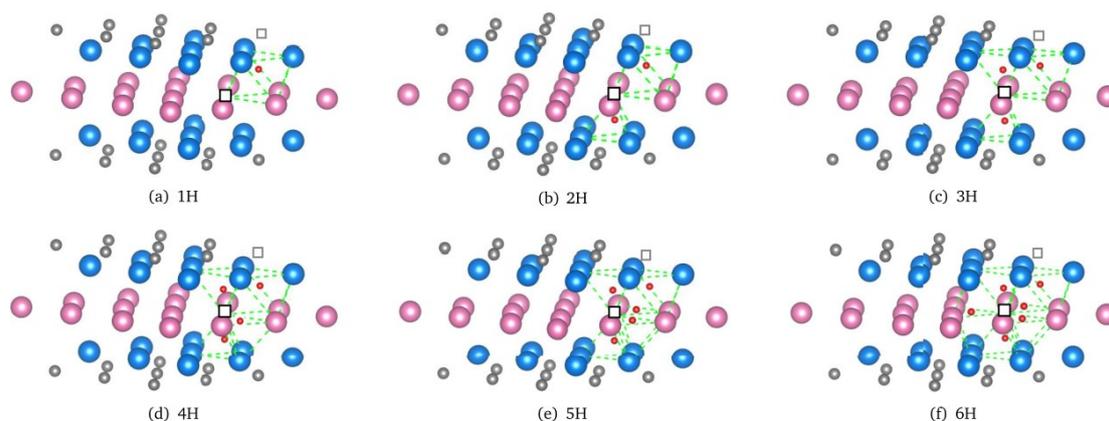
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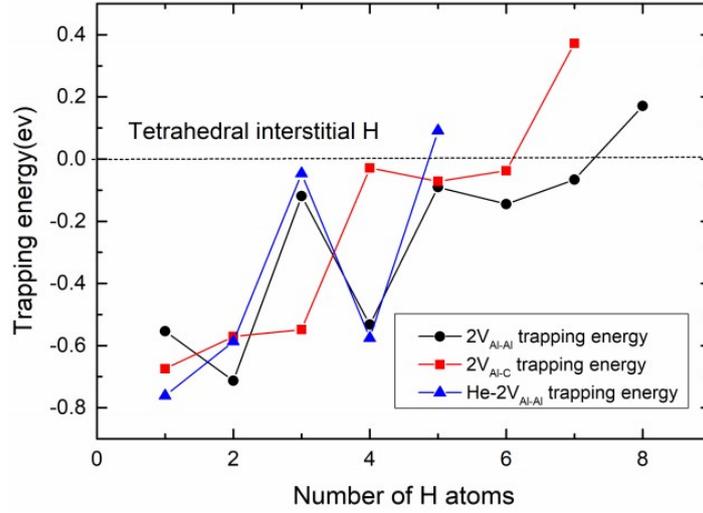
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**Fig. S1** Atomic configurations for different numbers of H atoms trapped in the  $2V_{\text{Al-Al}}$  divacancy. Blue spheres, pink spheres, and red spheres denote Ti atoms, Al atoms and H atoms, respectively. The black squares denote Al vacancies.



**Fig. S2** Atomic configurations for different numbers of H atoms trapped in the  $2V_{\text{Al-C}}$  divacancy. Blue spheres, pink spheres, gray spheres, and red spheres denote Ti atoms, Al atoms, C atoms and H atoms, respectively. The black square and gray square denote Al vacancy and C vacancy, respectively.



**Fig. S3** Trapping energy for H atom as a function of the number of H atoms trapped by divacancy and He-2V complex in  $\text{Ti}_3\text{AlC}_2$ .

**Table S1**

Summary of solution energies, lattice parameters and volume changes with increasing number of trapped H atoms in divacancies and He-2V complex.

Configuration	$E_s$ (ev)	$a$ (Å)	$c$ (Å)	$V$ (Å <sup>3</sup> )	$\Delta a/a$ (%)	$\Delta c/c$ (%)	$\Delta V/V$ (%)
$\text{Ti}_3\text{AlC}_2$	–	3.0820	18.6374	1380.20	–	–	–
$2V_{Al-Al}$	–	3.0870	18.5271	1378.33	0.1622	-0.5918	-0.1355
1H- $2V_{Al-Al}$	-0.81	3.0888	18.5599	1379.28	0.2206	-0.4158	-0.0667
2H- $2V_{Al-Al}$	-1.77	3.0871	18.5818	1379.41	0.1655	-0.2983	-0.0572
3H- $2V_{Al-Al}$	-2.04	3.0879	18.5523	1378.09	0.1914	-0.4566	-0.1529
4H- $2V_{Al-Al}$	-2.82	3.0857	18.5829	1378.95	0.1201	-0.2924	-0.0906
5H- $2V_{Al-Al}$	-3.16	3.0862	18.5486	1376.90	0.1363	-0.4765	-0.2391
6H- $2V_{Al-Al}$	-3.55	3.0868	18.5269	1376.01	0.1557	-0.5929	-0.3036
7H- $2V_{Al-Al}$	-3.87	3.0867	18.5051	1374.72	0.1525	-0.7097	-0.3970
$2V_{Al-C}$	–	3.0830	18.6025	1378.43	0.0324	-0.1873	-0.1282
1H- $2V_{Al-C}$	-0.93	3.0825	18.6060	1378.67	0.0162	-0.0017	-0.1109
2H- $2V_{Al-C}$	-1.75	3.0809	18.6346	1379.33	-0.0357	-0.0150	-0.0630
3H- $2V_{Al-C}$	-2.54	3.0802	18.6506	1379.17	-0.0584	0.0708	-0.0746
4H- $2V_{Al-C}$	-2.82	3.0814	18.6228	1377.66	-0.0195	-0.0783	-0.1840
5H- $2V_{Al-C}$	-3.14	3.0819	18.6014	1377.01	-0.0032	-0.1932	-0.2311
6H- $2V_{Al-C}$	-3.43	3.0827	18.5743	1375.94	0.0227	-0.3386	-0.3087
He- $2V_{Al-Al}$	–	3.0883	18.5537	1381.20	0.2044	-0.4491	0.0725
1H-He- $2V_{Al-Al}$	-0.30	3.0878	18.5693	1381.07	0.1882	-0.3654	0.0630
2H-He- $2V_{Al-Al}$	-1.14	3.0871	18.6031	1382.20	0.1655	-0.1840	0.1449
3H-He- $2V_{Al-Al}$	-1.41	3.0880	18.5641	1380.07	0.1947	-0.3933	-0.0094
4H-He- $2V_{Al-Al}$	-2.24	3.0866	18.5808	1379.65	0.1493	-0.3037	-0.0398