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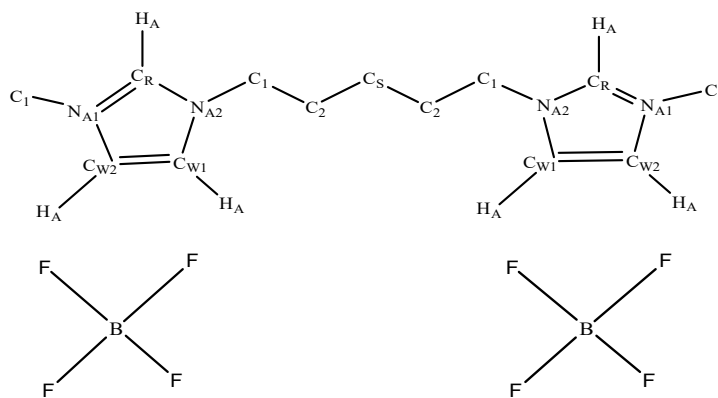
**Extension of Transferable Coarse-Grained Models to Dicationic
Ionic Liquids**

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Table S1 Force field parameters for $[\text{C}_5(\text{mim})_2][\text{BF}_4]_2$ DIL and also for $[\text{C}_n(\text{mim})_2][\text{BF}_4]_2$ (with $n=3,6,9,12$) DILs in the AA model



Lennard-Jones parameters

Atom	$\epsilon/\text{kcal.mol}^{-1}$	$\sigma/\text{\AA}$
C ₁	0.066	3.50
C ₂	0.066	3.50
C _C	0.066	3.50
C _R	0.070	3.55
C _W	0.070	3.55
H _A	0.030	2.42
H _C	0.030	2.50
H _I	0.030	2.50
N _A	0.170	3.25
B	0.095	3.58
F	0.061	3.12

Bond parameters

Atom	Atom	$\epsilon/\text{kcal.mol}^{-1}$	$\sigma/\text{\AA}$
C _W	H _A	183.500	1.08
C ₁	H _I	170.000	1.09
C ₂	H _C	170.000	1.09
C _S	H _C	170.000	1.09
C _R	N _A	238.528	1.32
C _{W1}	C _{W2}	260.038	1.34
C ₁	N _A	168.499	1.47
C ₁	C ₂	133.963	1.53
C ₂	C _S	133.963	1.53
C _W	N _A	213.552	1.38
B	F	290.000	1.39

Angle parameters

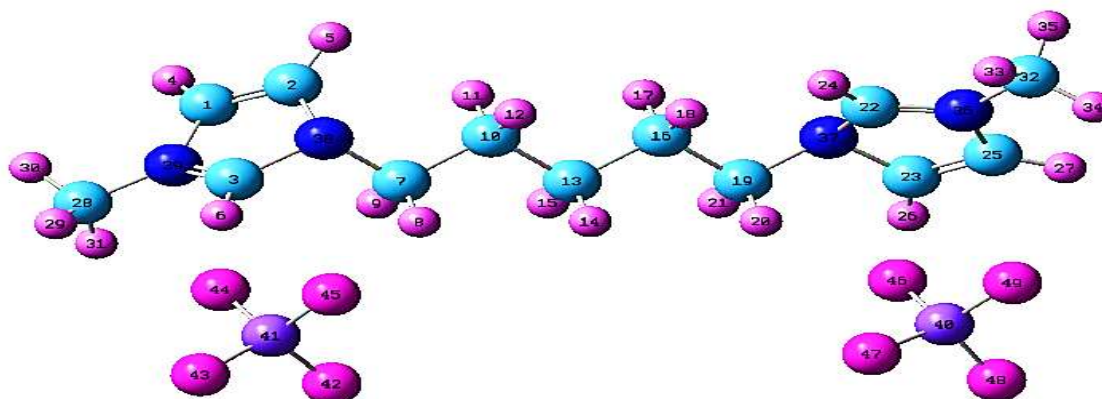
Atom	Atom	Atom	k_0 / kcal.mol ⁻¹ .rad ⁻²	θ_{eq} / deg
C _R	N _A	C _W	34.967	108.00
C _W	N _A	C ₁	34.967	125.60
C _R	N _A	C ₁	34.967	126.40
N _A	C _R	H _A	17.483	125.10
N _A	C _R	N _A	34.967	109.80
N _A	C _W	C _W	34.967	107.10
N _A	C _W	H _A	17.483	122.00
C _W	C _W	H _A	17.483	130.90
N _A	C ₁	H ₁	37.429	110.70
C ₁	C ₂	H _C	37.429	110.70
C ₂	C ₁	H ₁	37.429	110.70
C _S	C ₂	H _C	37.429	110.70
C ₂	C _S	H _C	37.429	110.70
C ₁	C ₂	C _S	50.000	112.70
C ₂	C _S	C ₂	50.000	112.70
N _A	C ₁	C ₂	50.000	112.70
H ₁	C ₁	H ₁	32.995	107.80
H _C	C ₂	H _C	32.995	107.80
H _C	C _S	H _C	32.995	107.80

Proper dihedral parameters

Atom	Atom	Atom	Atom	$k\phi,1$ /kcal.mol ⁻¹	$k\phi,2$ /kcal.mol ⁻¹	$k\phi,3$ /kcal.mol ⁻¹	$k\phi,4$ /kcal.mol ⁻¹
X	C _R	N _A	X	0.00	4.65	0.00	0.00
X	C _W	C _W	X	0.00	10.75	0.00	0.00
X	C _W	N _A	X	0.00	2.30	0.00	0.00
C _W	N _A	C ₁	H ₁	0.00	0.00	0.13	0.00
C _R	N _A	C ₁	H ₁	0.00	0.00	0.00	0.00
C _W	N _A	C ₁	C ₂	-1.38	1.06	0.21	0.00
C _R	N _A	C ₁	C ₂	-0.77	0.00	0.00	0.00
N _A	C ₁	C ₂	C _S	0.18	-0.16	0.24	0.00
N _A	C ₁	C ₂	H _C	0.00	0.00	0.00	0.00
C ₁	C ₂	C _S	H _C	0.00	0.00	0.37	0.00
C ₂	C _S	C ₂	H _C	0.00	0.00	0.37	0.00
H _C	C ₂	C ₁	H ₁	0.00	0.00	0.32	0.00
H _C	C ₂	C _S	H _C	0.00	0.00	0.32	0.00

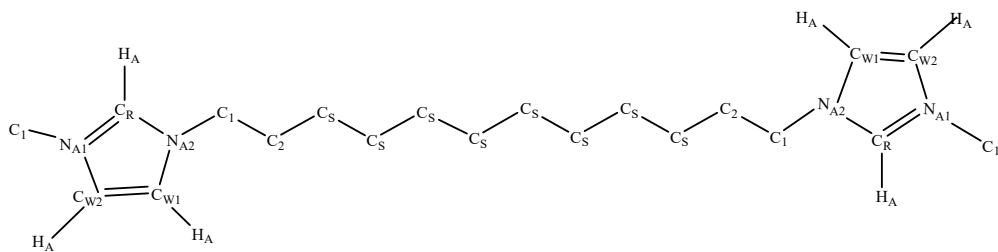
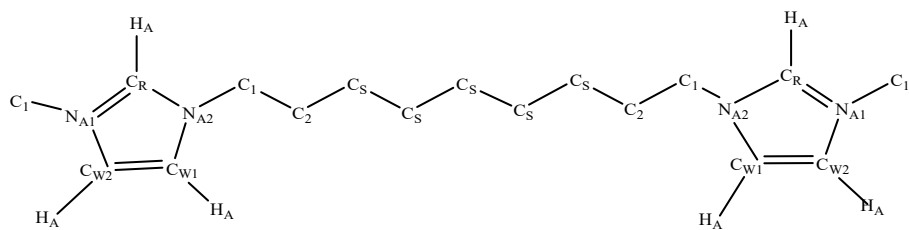
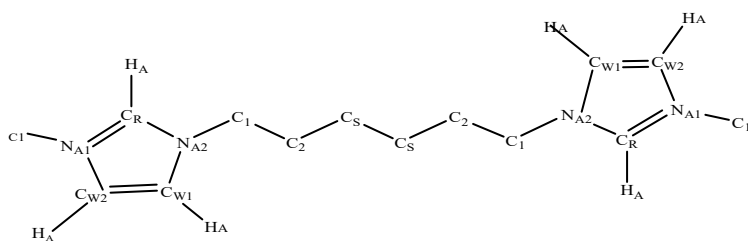
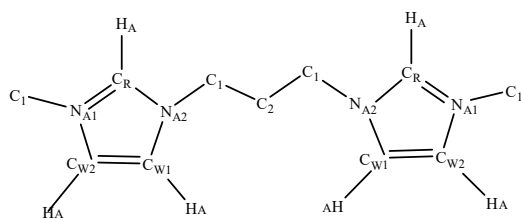
Improper dihedral parameters

Atom	Atom	Atom	Atom	K_ϕ / kcal.mol ⁻¹	d	n
X	N _A	X	X	1.00	-1.00	2.00
X	C _W	X	X	1.10	-1.00	2.00
X	C _R	X	X	1.10	-1.00	2.00



Partial charges in CHELPG method

$[\text{C}_5(\text{mim})_2]^{+2}$				$[\text{BF}_4]^-$	
Atom	q / e	Atom	q / e	Atom	q / e
C ₁	-0.1703	H ₁₇	0.1857	B ₁	1.1453
C ₂	-0.1307	C ₁₂	-0.1684	F ₁	-0.5250
C ₃	-0.0581	H ₁₈	0.1913	F ₂	-0.5236
H ₁	0.1765	H ₁₉	0.1808	F ₃	-0.4818
H ₂	0.1708	C ₁₃	-0.1841	F ₄	-0.4962
H ₃	0.2102	H ₂₀	0.1717	B ₁	1.0887
C ₄	0.0962	H ₂₁	0.0789	F ₅	-0.5154
H ₄	0.0783	H ₂₂	0.1146	F ₆	-0.5047
H ₅	0.0868	N ₁	0.1729	F ₇	-0.4680
C ₅	-0.1104	N ₂	0.0972	F ₈	-0.5045
H ₆	0.0259	N ₃	0.0843	Total charge	-1.785
H ₇	0.0279	N ₄	0.1773		
C ₆	-0.0628	C ₉	-0.1304		
H ₈	0.0451	H ₁₄	0.0990		
H ₉	-0.0091	H ₁₅	0.1457		
C ₇	0.3504	H ₁₆	0.0668		
H ₁₀	-0.0785	C ₁₀	-0.0105		
H ₁₁	-0.0065	C ₁₁	-0.1522		
C ₈	-0.1208	H ₁₃	0.0570		
H ₁₂	0.0872				
Total charge		1.785			



Lennard-Jones parameters

Atom	$\epsilon/\text{kcal.mol}^{-1}$	$\sigma/\text{\AA}$
[C₃(mim)₂]²⁺		
C ₁	0.066	3.50
C ₂	0.066	3.50
C _R	0.070	3.55
C _W	0.070	3.55
H _A	0.030	2.42
H _C	0.030	2.50
H ₁	0.030	2.50
N _A	0.170	3.25
[C₆(mim)₂]²⁺		
C ₁	0.066	3.50
C ₂	0.066	3.50
C _S	0.066	3.50
C _R	0.070	3.55
C _W	0.070	3.55
H _A	0.030	2.42
H _C	0.030	2.50
H ₁	0.030	2.50
N _A	0.170	3.25
[C₉(mim)₂]²⁺		
C ₁	0.066	3.50
C ₂	0.066	3.50
C _S	0.066	3.50
C _R	0.070	3.55
C _W	0.070	3.55
H _A	0.030	2.42
H _C	0.030	2.50
H ₁	0.030	2.50
N _A	0.170	3.25
[C₁₂(mim)₂]²⁺		
C ₁	0.066	3.50
C ₂	0.066	3.50
C _S	0.066	3.50
C _R	0.070	3.55
C _W	0.070	3.55
H _A	0.030	2.42
H _C	0.030	2.50
H ₁	0.030	2.50
N _A	0.170	3.25

Bond parameters

Atom	Atom	$\epsilon/\text{kcal.mol}^{-1}$	$\sigma/\text{\AA}$
[C₃(mim)₂]²⁺			
C _W	H _A	183.500	1.08
C _R	H _A	183.500	1.08
C ₁	H ₁	170.000	1.09
C ₂	H _C	170.000	1.09
C _R	N _A	238.528	1.32
C _{W1}	C _{W2}	260.038	1.34
C ₁	N _A	168.499	1.47
C ₁	C ₂	133.963	1.53
C _W	N _A	213.552	1.38
[C₆(mim)₂]²⁺			
C _W	H _A	183.500	1.08
C _R	H _A	183.500	1.08
C ₁	H ₁	170.000	1.09
C ₂	H _C	170.000	1.09
C _S	H _C	170.000	1.09
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C ₁	N _A	168.499	1.47
C ₁	C ₂	133.963	1.53
C ₂	C _S	133.963	1.53
C _S	C _S	133.963	1.53
C _W	N _A	213.552	1.38
[C₉(mim)₂]²⁺			
C _W	H _A	183.500	1.08
C _R	H _A	183.500	1.08
C ₁	H ₁	170.000	1.09
C ₂	H _C	170.000	1.09
C _S	H _S	170.000	1.09
C _R	N _A	238.528	1.32
C _{W1}	C _{W2}	260.038	1.34
C ₁	N _A	168.499	1.47
C ₁	C ₂	133.963	1.53
C ₂	C _S	133.963	1.53
C _S	C _S	133.963	1.53
C _W	N _A	213.552	1.38

Atom	Atom	$\epsilon/\text{kcal.mol}^{-1}$	$\sigma/\text{\AA}$
[C₁₂(mim)₂]²⁺			
C _W	H _A	183.500	1.08
C _R	H _A	183.500	1.08
C ₁	H ₁	170.000	1.09
C ₂	H _C	170.000	1.09
C _S	H _C	170.000	1.09
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C ₁	N _A	168.499	1.47
C ₁	C ₂	133.963	1.53
C ₂	C _S	133.963	1.53
C _S	C _S	133.963	1.53
C _W	N _A	213.552	1.38

Angle parameters

Atom	Atom	Atom	$k_{\theta} / \text{kcal.mol}^{-1}.\text{rad}^{-2}$	$\theta_{\text{eq}} / \text{deg}$
[C₃(mim)₂]²⁺				
C _R	N _A	C _W	34.967	108.00
C _W	N _A	C ₁	34.967	125.60
C _R	N _A	C ₁	34.967	126.40
N _A	C _R	H _A	17.483	125.10
N _A	C _R	N _A	34.967	109.80
N _A	C _W	C _W	34.967	107.10
N _A	C _W	H _A	17.483	122.00
C _W	C _W	H _A	17.483	130.90
N _A	C ₁	H ₁	37.429	110.70
C ₁	C ₂	H _C	37.429	110.70
C ₂	C ₁	H ₁	37.429	110.70
C ₁	C ₂	C ₁	50.000	112.70
N _A	C ₁	C ₂	50.000	112.70
H ₁	C ₁	H ₁	32.995	107.80
H _C	C ₂	H _C	32.995	107.80
[C₆(mim)₂]²⁺				
C _R	N _A	C _W	34.967	108.00
C _W	N _A	C ₁	34.967	125.60
C _R	N _A	C ₁	34.967	126.40
N _A	C _R	H _A	17.483	125.10
N _A	C _R	N _A	34.967	109.80
N _A	C _W	C _W	34.967	107.10
N _A	C _W	H _A	17.483	122.00
C _W	C _W	H _A	17.483	130.90
N _A	C ₁	H ₁	37.429	110.70
C ₁	C ₂	H _C	37.429	110.70
C ₂	C ₁	H ₁	37.429	110.70
C _S	C ₂	H _C	37.429	110.70
C ₂	C _S	H _C	37.429	110.70
C _S	C _S	H _C	37.429	110.70
C ₁	C ₂	C _S	50.000	112.70
C ₂	C _S	C _S	50.000	112.70
N _A	C ₁	C ₂	50.000	112.70
H ₁	C ₁	H ₁	32.995	107.80
H _C	C ₂	H _C	32.995	107.80
H _C	C _S	H _C	32.995	107.80

Angle parameters

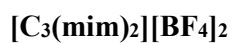
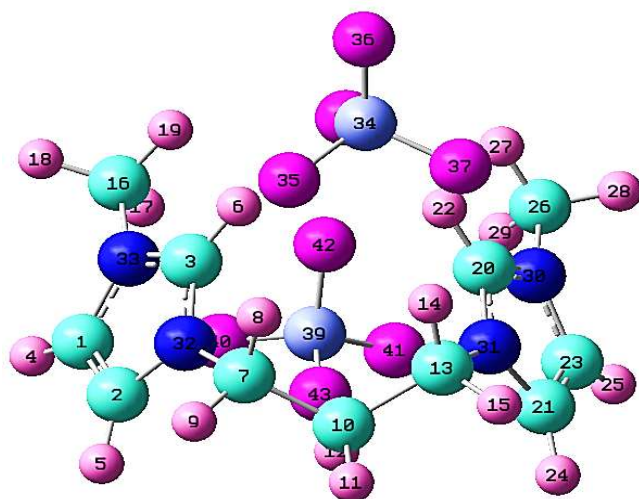
Atom	Atom	Atom	$k_{\theta} / \text{kcal.mol}^{-1}.\text{rad}^{-2}$	$\theta_{\text{eq}} / \text{deg}$
[C₉(mim)₂]²⁺				
C _R	N _A	C _W	34.967	108.00
C _W	N _A	C ₁	34.967	125.60
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N _A	C _W	H _A	17.483	122.00
C _W	C _W	H _A	17.483	130.90
N _A	C ₁	H ₁	37.429	110.70
C ₁	C ₂	H _C	37.429	110.70
C ₂	C ₁	H ₁	37.429	110.70
C _S	C ₂	H _C	37.429	110.70
C ₂	C _S	H _C	37.429	110.70
C _S	C _S	H _C	37.429	110.70
C ₁	C ₂	C _S	50.000	112.70
C ₂	C _S	C _S	50.000	112.70
C _S	C _S	C _S	50.000	112.70
N _A	C ₁	C ₂	50.000	112.70
H ₁	C ₁	H ₁	32.995	107.80
H _C	C ₂	H _C	32.995	107.80
H _C	C _S	H _C	32.995	107.80
[C₁₂(mim)₂]²⁺				
C _R	N _A	C _W	34.967	108.00
C _W	N _A	C ₁	34.967	125.60
C _R	N _A	C ₁	34.967	126.40
N _A	C _R	H _A	17.483	125.10
N _A	C _R	N _A	34.967	109.80
N _A	C _W	C _W	34.967	107.10
N _A	C _W	H _A	17.483	122.00
C _W	C _W	H _A	17.483	130.90
N _A	C ₁	H ₁	37.429	110.70
C ₁	C ₂	H _C	37.429	110.70
C ₂	C ₁	H ₁	37.429	110.70
C _S	C ₂	H _C	37.429	110.70
C ₂	C _S	H _C	37.429	110.70
C _S	C _S	H _C	37.429	110.70
C ₁	C ₂	C _S	50.000	112.70
C ₂	C _S	C _S	50.000	112.70
C _S	C _S	C _S	50.000	112.70
N _A	C ₁	C ₂	50.000	112.70
H ₁	C ₁	H ₁	32.995	107.80
H _C	C ₂	H _C	32.995	107.80
H _C	C _S	H _C	32.995	107.80

Proper dihedral parameters

Atom	Atom	Atom	Atom	$k_{\phi,1}/\text{kcal.mol}^{-1}$	$k_{\phi,2}/\text{kcal.mol}^{-1}$	$k_{\phi,3}/\text{kcal.mol}^{-1}$	$k_{\phi,4}/\text{kcal.mol}^{-1}$
[C₃(mim)₂]²⁺							
X	C _R	N _A	X	0.00	4.65	0.00	0.00
X	C _W	C _W	X	0.00	10.75	0.00	0.00
X	C _W	N _A	X	0.00	2.30	0.00	0.00
C _W	N _A	C ₁	H ₁	0.00	0.00	0.13	0.00
C _R	N _A	C ₁	H ₁	0.00	0.00	0.00	0.00
C _W	N _A	C ₁	C ₂	-1.38	1.06	0.21	0.00
C _R	N _A	C ₁	C ₂	-0.77	0.00	0.00	0.00
N _A	C ₁	C ₂	C ₁	0.18	-0.16	0.24	0.00
N _A	C ₁	C ₂	H _C	0.00	0.00	0.00	0.00
C ₁	C ₂	C ₁	H ₁	0.00	0.00	0.37	0.00
H _C	C ₂	C ₁	H _C	0.00	0.00	0.32	0.00
[C₆(mim)₂]²⁺							
X	C _R	N _A	X	0.00	4.65	0.00	0.00
X	C _W	C _W	X	0.00	10.75	0.00	0.00
X	C _W	N _A	X	0.00	2.30	0.00	0.00
C _W	N _A	C ₁	H ₁	0.00	0.00	0.13	0.00
C _R	N _A	C ₁	H ₁	0.00	0.00	0.00	0.00
C _W	N _A	C ₁	C ₂	-1.38	1.06	0.21	0.00
C _R	N _A	C ₁	C ₂	-0.77	0.00	0.00	0.00
N _A	C ₁	C ₂	C _S	0.18	-0.16	0.24	0.00
N _A	C ₁	C ₂	H _C	0.00	0.00	0.00	0.00
C ₁	C ₂	C _S	H _C	0.00	0.00	0.37	0.00
C ₂	C _S	C _S	H _C	0.00	0.00	0.37	0.00
C _S	C _S	C ₂	H _C	0.00	0.00	0.37	0.00
H _C	C ₂	C ₁	H ₁	0.00	0.00	0.32	0.00
H _C	C ₂	C _S	H _C	0.00	0.00	0.32	0.00
H _C	C _S	C _S	H _C	0.00	0.00	0.32	0.00

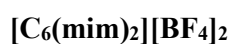
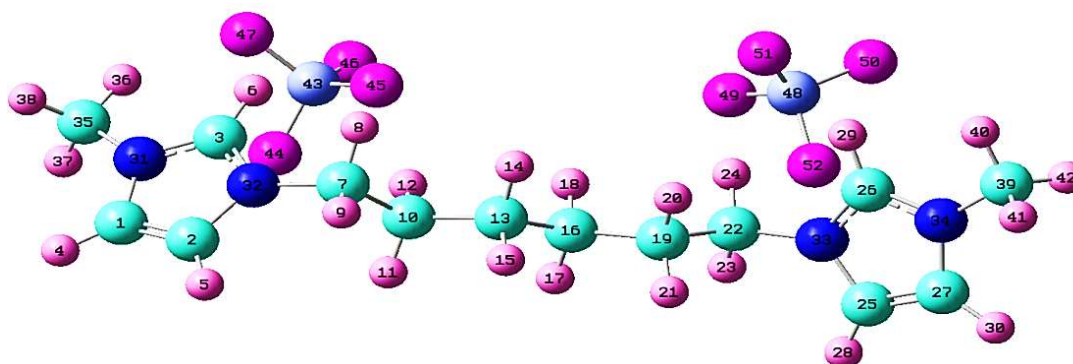
Proper dihedral parameters

Atom	Atom	Atom	Atom	$k_{\phi,1}/\text{kcal.mol}^{-1}$	$k_{\phi,2}/\text{kcal.mol}^{-1}$	$k_{\phi,3}/\text{kcal.mol}^{-1}$	$k_{\phi,4}/\text{kcal.mol}^{-1}$
[C₉(mim)₂]²⁺							
X	C _R	N _A	X	0.00	4.65	0.00	0.00
X	C _W	C _W	X	0.00	10.75	0.00	0.00
X	C _W	N _A	X	0.00	2.30	0.00	0.00
C _W	N _A	C ₁	H ₁	0.00	0.00	0.13	0.00
C _R	N _A	C ₁	H ₁	0.00	0.00	0.00	0.00
C _W	N _A	C ₁	C ₂	-1.38	1.06	0.21	0.00
C _R	N _A	C ₁	C ₂	-0.77	0.00	0.00	0.00
N _A	C ₁	C ₂	C _S	0.18	-0.16	0.24	0.00
N _A	C ₁	C ₂	H _C	0.00	0.00	0.00	0.00
C ₁	C ₂	C _S	H _C	0.00	0.00	0.37	0.00
C ₂	C _S	C _S	H _C	0.00	0.00	0.37	0.00
C _S	C _S	C ₂	H _C	0.00	0.00	0.37	0.00
C _S	C _S	C _S	H _C	0.00	0.00	0.37	0.00
H _C	C ₂	C ₁	H ₁	0.00	0.00	0.32	0.00
H _C	C ₂	C _S	H _C	0.00	0.00	0.32	0.00
H _C	C _S	C _S	H _C	0.00	0.00	0.32	0.00
[C₁₂(mim)₂]²⁺							
X	C _R	N _A	X	0.00	4.65	0.00	0.00
X	C _W	C _W	X	0.00	10.75	0.00	0.00
X	C _W	N _A	X	0.00	2.30	0.00	0.00
C _W	N _A	C ₁	H ₁	0.00	0.00	0.13	0.00
C _R	N _A	C ₁	H ₁	0.00	0.00	0.00	0.00
C _W	N _A	C ₁	C ₂	-1.38	1.06	0.21	0.00
C _R	N _A	C ₁	C ₂	-0.77	0.00	0.00	0.00
N _A	C ₁	C ₂	C _S	0.18	-0.16	0.24	0.00
N _A	C ₁	C ₂	H _C	0.00	0.00	0.00	0.00
C ₁	C ₂	C _S	H _C	0.00	0.00	0.37	0.00
C ₂	C _S	C _S	H _C	0.00	0.00	0.37	0.00
C _S	C _S	C ₂	H _C	0.00	0.00	0.37	0.00
C _S	C _S	C _S	H _C	0.00	0.00	0.37	0.00
H _C	C ₂	C ₁	H ₁	0.00	0.00	0.32	0.00
H _C	C ₂	C _S	H _C	0.00	0.00	0.32	0.00
H _C	C _S	C _S	H _{CC}	0.00	0.00	0.32	0.00



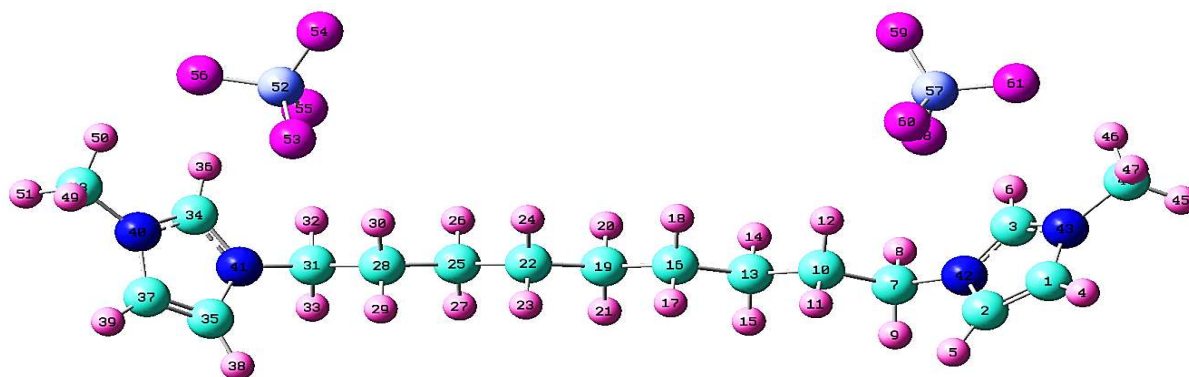
Partial charges in CHELPG method

[C ₃ (mim) ₂] ⁺²				[BF ₄] ⁻	
Atom	q / e	Atom	q / e	Atom	q / e
C ₁	-0.1226	H ₁₈	0.0790	B ₃₄	0.7217
C ₂	-0.1629	H ₁₉	0.1231	F ₃₅	-0.3809
C ₃	0.0114	C ₂₀	-0.0337	F ₃₆	-0.3535
H ₄	0.1985	C ₂₁	-0.1602	F ₃₇	-0.3913
H ₅	0.1804	H ₂₂	0.2140	F ₃₈	-0.3803
H ₆	0.1750	C ₂₃	-0.1213	F ₃₉	0.6565
C ₇	0.0617	H ₂₄	0.1734	B ₄₀	-0.3653
H ₈	0.0411	H ₂₅	0.1919	F ₄₁	-0.3676
H ₉	0.0307	C ₂₆	-0.1508	F ₄₂	-0.3709
C ₁₀	0.1547	H ₂₇	0.1140	F ₄₃	-0.3474
H ₁₁	-0.0310	H ₂₈	0.0761	Total charge -1.579	
H ₁₂	-0.0081	H ₂₉	0.1318		
C ₁₃	0.0480	N ₃₀	0.1041		
H ₁₄	0.0562	N ₃₁	0.0691		
H ₁₅	0.0322	N ₃₂	0.0406		
C ₁₆	-0.1564	N ₃₃	0.0891		
H ₁₇	0.1301				
Total charge		1.579			



Partial charges in CHELPG method

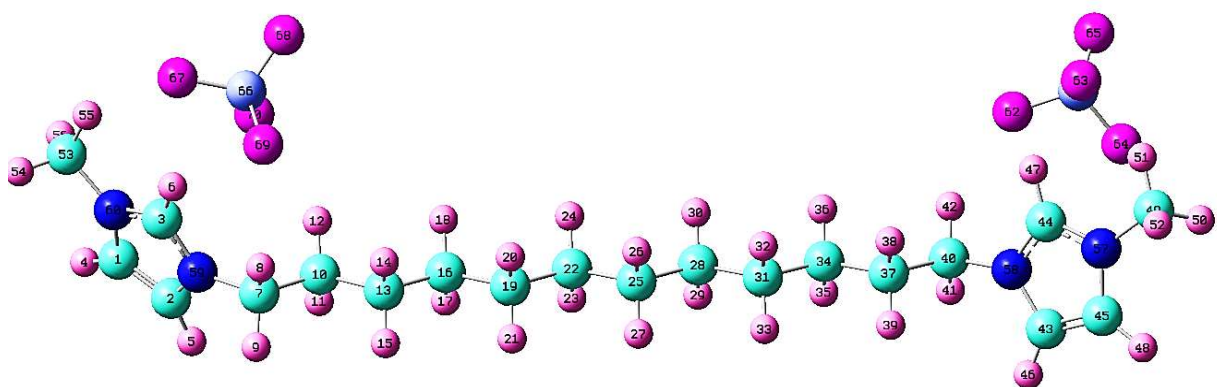
$[\text{C}_6(\text{mim})_2]^{+2}$				$[\text{BF}_4]^-$	
Atom	q / e	Atom	q / e	Atom	q / e
C ₁	-0.1288	C ₂₂	-0.0848	B ₄₃	1.0951
C ₂	-0.1623	H ₂₃	0.0462	F ₄₄	-0.5151
C ₃	-0.0037	H ₂₄	0.0734	F ₄₅	-0.4817
H ₄	0.1662	C ₂₅	-0.1662	F ₄₆	-0.4730
H ₅	0.1863	C ₂₆	-0.0028	F ₄₇	-0.5066
H ₆	0.1934	C ₂₇	-0.1255	B ₄₈	1.0946
C ₇	-0.0829	H ₂₈	0.1874	F ₄₉	-0.4816
H ₈	0.0729	H ₂₉	0.1930	F ₅₀	-0.5064
H ₉	0.0456	H ₃₀	0.1655	F ₅₁	-0.4729
C ₁₀	0.3808	N ₃₁	0.1468	F ₅₂	-0.5148
H ₁₁	-0.0770	N ₃₂	0.0683	Total charge -1.763	
H ₁₂	-0.0413	N ₃₃	0.0700		
C ₁₃	-0.1018	N ₃₄	0.1448		
H ₁₄	0.0420	C ₃₅	-0.1415		
H ₁₅	-0.0119	H ₃₆	0.1532		
C ₁₆	-0.1008	H ₃₇	0.1155		
H ₁₇	-0.0120	H ₃₈	0.0614		
H ₁₈	0.0418	C ₃₉	-0.1407		
C ₁₉	0.3804	H ₄₀	0.1532		
H ₂₀	-0.0412	H ₄₁	0.1153		
H ₂₁	-0.0768	H ₄₂	0.0613		
Total charge		1.763			



[C₉(mim)₂][BF₄]₂

Partial charges in CHELPG method

[C ₉ (mim) ₂] ⁺²				[BF ₄] ⁻	
Atom	q / e	Atom	q / e	Atom	q / e
C ₁	-0.1534	H ₂₇	-0.0066	B ₅₂	1.0953
C ₂	-0.1558	C ₂₈	0.3231	F ₅₃	-0.4991
C ₃	0.0230	H ₂₉	-0.0769	F ₅₄	-0.4675
H ₄	0.1738	H ₃₀	-0.0220	F ₅₅	-0.4832
H ₅	0.1879	C ₃₁	-0.0341	F ₅₆	-0.5163
H ₆	0.1718	H ₃₂	0.0767	B ₅₇	1.0591
C ₇	0.0038	H ₃₃	0.0259	F ₅₈	-0.4755
H ₈	0.0661	C ₃₄	0.0002	F ₅₉	-0.4602
H ₉	0.0161	C ₃₅	-0.1431	F ₆₀	-0.4794
C ₁₀	0.2907	H ₃₆	0.1851	F ₆₁	-0.5049
H ₁₁	-0.0804	C ₃₇	-0.1564	Total charge -1.732	
H ₁₂	-0.0198	H ₃₈	0.1844		
C ₁₃	-0.0612	H ₃₉	0.1722		
H ₁₄	0.0118	N ₄₀	0.1514		
H ₁₅	-0.0097	N ₄₁	0.0448		
C ₁₆	-0.0372	N ₄₂	0.0482		
H ₁₇	-0.0140	N ₄₃	0.1348		
H ₁₈	0.0109	C ₄₄	-0.1197		
C ₁₉	0.203	H ₄₅	0.0625		
H ₂₀	-0.0354	H ₄₆	0.1465		
H ₂₁	-0.0592	H ₄₇	0.0992		
C ₂₂	0.0084	C ₄₈	-0.1333		
H ₂₃	-0.0246	H ₄₉	0.1024		
H ₂₄	0.0020	H ₅₀	0.1552		
C ₂₅	-0.1077	H ₅₁	0.0623		
H ₂₆	0.0277				
Total charge		1.732			



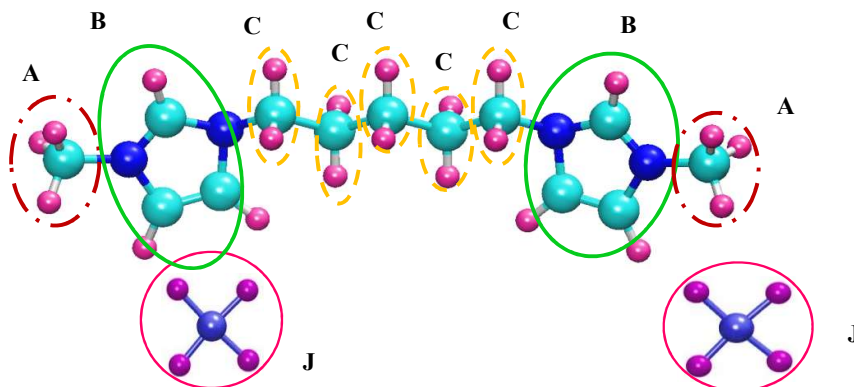
[C₁₂(mim)₂][BF₄]₂

Partial charges in CHELPG method

[C ₁₂ (mim) ₂] ⁺²				[BF ₄] ⁻			
Atom	q / e	Atom	q / e	Atom	q / e	Atom	q / e
C ₁	-0.1422	H ₂₁	-0.0401	H ₄₁	0.0680	B ₆₁	1.0711
C ₂	-0.1729	C ₂₂	0.0124	H ₄₂	0.1117	F ₆₂	-0.4966
C ₃	-0.0111	H ₂₃	-0.0233	C ₄₃	-0.2051	F ₆₃	-0.4949
H ₄	0.1696	H ₂₄	-0.0233	C ₄₄	-0.0475	F ₆₄	-0.5016
H ₅	0.1927	C ₂₅	0.1008	C ₄₅	-0.1255	F ₆₅	-0.4643
H ₆	0.1818	H ₂₆	-0.0348	H ₄₆	0.1982	B ₆₆	1.0777
C ₇	-0.0655	H ₂₇	-0.0432	H ₄₇	0.2107	F ₆₇	-0.5099
H ₈	0.0806	C ₂₈	0.0442	H ₄₈	0.1664	F ₆₈	-0.4651
H ₉	0.0298	H ₂₉	-0.0249	C ₄₉	-0.1024	F ₆₉	-0.4793
C ₁₀	0.3291	H ₃₀	-0.0187	H ₅₀	0.1126	F ₇₀	-0.4902
H ₁₁	-0.0825	C ₃₁	-0.0182	H ₅₁	0.1312	Total charge	-1.753
H ₁₂	-0.0290	H ₃₂	-0.0012	H ₅₂	0.0502		
C ₁₃	-0.0828	H ₃₃	-0.0097	C ₅₃	-0.1342		
H ₁₄	0.0231	C ₃₄	0.1793	H ₅₄	0.0629		
H ₁₅	-0.0069	H ₃₅	-0.0506	H ₅₅	0.1533		
C ₁₆	0.0224	H ₃₆	-0.0281	H ₅₆	0.1029		
H ₁₇	-0.0251	C ₃₇	0.0573	N ₅₇	0.1331		
H ₁₈	-0.0023	H ₃₈	-0.0269	N ₅₈	0.1780		
C ₁₉	0.1164	H ₃₉	-0.0407	N ₅₉	0.0884		
H ₂₀	-0.0282	C ₄₀	-0.0737	N ₆₀	0.1535		
Total charge		1.753					

Table S2 Force field parameters for $[\text{C}_5(\text{mim})_2][\text{BF}_4]_2$ DIL in the three mapping schemes of CG model. The similar parameters have been used for $[\text{C}_n(\text{mim})_2][\text{BF}_4]_2$ (with $n=3,6,9,12$) in the CG-1 model.

Force-field parameters for CG (scheme-1)



Lennard-Jones parameters

Bead	Bead	Potential Type	$\epsilon/\text{kcal.mol}^{-1}$	$\sigma/\text{\AA}$
A	A	lj12-6	0.086	3.41
A	B	lj12-6	0.229	3.90
A	C	lj12-6	0.089	3.68
A	J	lj12-6	0.258	4.11
B	B	lj12-6	0.612	4.38
B	C	lj12-6	0.236	4.17
B	J	lj12-6	0.688	4.59
C	C	lj12-6	0.091	3.95
C	J	lj12-6	0.266	4.38
J	J	lj12-6	0.774	4.80

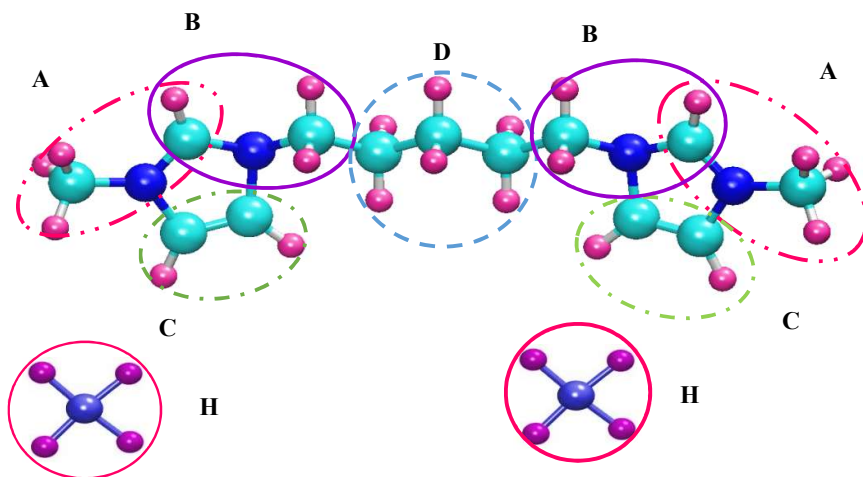
Bond parameters

Bead	Bead	$k_b/\text{kcal.mol}^{-1}.\text{\AA}^{-2}$	$r_{eq}/\text{\AA}$
A	B	210.792	2.71
B	C	219.647	2.69
C	C	297.635	1.66

Angle parameters

Bead	Bead	Bead	$k_\theta/\text{kcal.mol}^{-1}.\text{rad}^{-2}$	Θ_{eq}/deg
A	B	C	138.983	142.38
B	C	C	68.269	109.45
C	C	C	62.049	114.00

Force-field parameters for CG (scheme-2)



Lennard-Jones parameters

Bead	Bead	Potential Type	$\epsilon/\text{kcal.mol}^{-1}$	$\sigma/\text{\AA}$
A	A	lj9-6	0.376	4.10
A	B	lj9-6	0.360	4.05
A	C	lj9-6	0.274	4.05
A	D	lj9-6	0.348	4.30
A	H	lj12-6	0.539	4.45
B	B	lj9-6	0.346	4.10
B	C	lj9-6	0.263	4.05
B	D	lj9-6	0.334	4.30
B	H	lj12-6	0.517	4.45
C	C	lj9-6	0.200	3.60
C	D	lj9-6	0.254	3.76
C	H	lj12-6	0.393	4.20
D	D	lj9-6	0.420	4.51
D	H	lj12-6	0.570	4.65

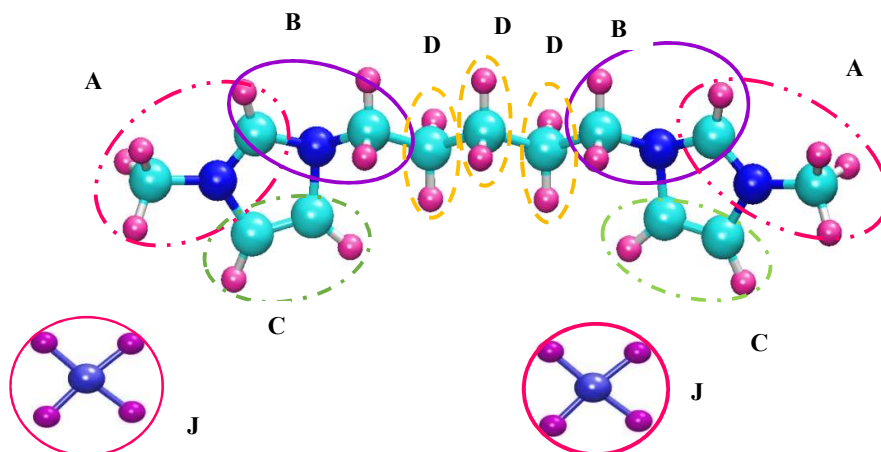
Bond parameters

Bead	Bead	$k_b/\text{kcal.mol}^{-1}.\text{\AA}^{-2}$	$r_{eq}/\text{\AA}$
A	B	179.540	2.97
A	C	223.750	2.30
B	C	230.650	2.27
B	D	7.980	3.23

Angle parameters

Bead	Bead	Bead	$k_\theta/\text{kcal.mol}^{-1}.\text{rad}^{-2}$	θ_{eq}/deg
A	B	C	932.500	51.22
A	C	B	433.260	81.88
C	A	B	748.700	50.39
A	B	D	4.910	143.82
C	B	D	1.400	147.83

Force-field parameters for CG (scheme-3)



Lennard-Jones parameters

Bead	Bead	Potential Type	$\epsilon/\text{kcal.mol}^{-1}$	$\sigma/\text{\AA}$
A	A	lj9-6	0.376	4.10
A	B	lj9-6	0.360	4.05
A	C	lj9-6	0.274	4.05
A	D	lj12_6	0.185	4.03
A	J	lj12_6	0.539	4.45
B	B	lj9_6	0.346	4.10
B	C	lj9_6	0.263	4.05
B	D	lj12_6	0.178	4.03
B	J	lj12_6	0.517	4.45
C	C	lj9_6	0.200	3.60
C	D	lj12_6	0.135	3.78
C	J	lj12_6	0.393	4.20
D	D	lj12_6	0.091	3.95
D	J	lj12_6	0.266	4.38
J	J	lj12_6	0.774	4.80

Bond parameters

Bead	Bead	$k_b / \text{kcal.mol}^{-1}.\text{\AA}^{-2}$	$r_{eq} / \text{\AA}$
A	B	179.540	2.97
A	C	223.750	2.30
B	C	230.650	2.27
B	D	7.980	3.23
D	D	297.635	1.66

Angle parameters

Bead	Bead	Bead	$k_\theta / \text{kcal.mol}^{-1}.\text{rad}^{-2}$	θ_{eq} / deg
A	B	C	932.500	51.22
A	C	B	433.260	81.88
C	A	B	748.700	50.39
A	B	D	4.910	143.82
C	B	D	1.400	147.83
D	D	D	62.049	114.00
B	D	D	68.269	109.45

Table S3 Diffusion coefficients for the studied DILs in both CG and AA models obtained from the simulations and corrected for system-size dependences (Yeh-Hummer correlation) at 500 K and at 1 atm with confidence intervals expressed as $\pm\sigma$.

	D_{cation} ($10^{-11} \text{ m}^2 \cdot \text{s}^{-1}$)		D_{anion} ($10^{-11} \text{ m}^2 \cdot \text{s}^{-1}$)	
	D_{PBC}	D_0	D_{PBC}	D_0
$[\text{C}_3(\text{mim})_2][\text{BF}_4]_2$				
CG (1)	201.37 $\pm$ 0.7492	247.19 $\pm$ 0.7492	253.39 $\pm$ 0.6577	284.55 $\pm$ 0.6577
AA	46.36 $\pm$ 0.4824	55.22 $\pm$ 0.4824	69.09 $\pm$ 0.2826	76.33 $\pm$ 0.2826
$[\text{C}_6(\text{mim})_2][\text{BF}_4]_2$				
CG (1)	155.98 $\pm$ 0.5886	192.25 $\pm$ 0.5886	234.05 $\pm$ 0.4385	262.35 $\pm$ 0.4385
AA	33.18 $\pm$ 0.2236	39.33 $\pm$ 0.2236	46.60 $\pm$ 0.2147	51.18 $\pm$ 0.2147
$[\text{C}_9(\text{mim})_2][\text{BF}_4]_2$				
CG (1)	184.75 $\pm$ 1.4868	227.52 $\pm$ 1.4868	327.25 $\pm$ 0.5111	363.04 $\pm$ 0.5111
AA	30.35 $\pm$ 0.3543	36.21 $\pm$ 0.3543	45.23 $\pm$ 0.2716	49.42 $\pm$ 0.7492
$[\text{C}_{12}(\text{mim})_2][\text{BF}_4]_2$				
CG (1)	143.08 $\pm$ 2.5839	174.40 $\pm$ 2.5839	305.41 $\pm$ 1.8920	336.72 $\pm$ 1.8920
AA	19.20 $\pm$ 0.1293	22.79 $\pm$ 0.1293	45.68 $\pm$ 0.1823	49.74 $\pm$ 0.1823

Table S4 The fitting parameters of VFT equation for $[\text{C}_{12}(\text{mim})_2][\text{BF}_4]_2$ in both AA and CG models.

	All-Atom		Coarse-Grained	
	Cation	Anion	Cation	Anion
D_{PBC}				
$D' / 10^{-11} \text{ m}^2 \cdot \text{s}^{-1}$	0.0002	0.0003	0.0001	0.0004
B' / K	-7154.8	-5846.9	-10793.6	-10256.1
T_0 / K	1113	984	1265	1252
R^2	0.9723	0.9921	0.9776	0.9819
D_0				
$D' / 10^{-11} \text{ m}^2 \cdot \text{s}^{-1}$	0.0002	0.0002	0.0005	0.0003
B' / K	-6981.0	-6166.7	-8912.2	-10727.4
T_0 / K	1105	998	1192	1270
R^2	0.9720	0.9922	0.9768	0.9820

Table S5 Diffusion coefficients of $[\text{C}_{12}(\text{mim})_2][\text{BF}_4]_2$ in both CG and AA models in the temperature range of 350-500 K and at 1 atm obtained from the simulations and corrected for system-size dependences (Yeh-Hummer correlation) with confidence intervals expressed as $\pm\sigma$.

	Diffusion Coefficient of Cation ($10^{-11} \text{ m}^2.\text{s}^{-1}$)		Diffusion Coefficient of Anion ($10^{-11} \text{ m}^2.\text{s}^{-1}$)	
	D_{PBC}	D_0	D_{PBC}	D_0
T (350K)				
CG (1)	2.25 $\pm$ 0.0277	2.74 $\pm$ 0.0277	6.69 $\pm$ 0.0465	7.38 $\pm$ 0.0465
AA	0.29 $\pm$ 0.0076	0.35 $\pm$ 0.0076	0.43 $\pm$ 0.0155	0.47 $\pm$ 0.0155
T (400K)				
CG (1)	25.64 $\pm$ 0.1485	31.25 $\pm$ 0.1485	61.65 $\pm$ 0.1333	67.97 $\pm$ 0.1333
AA	3.13 $\pm$ 0.0219	3.73 $\pm$ 0.0219	4.56 $\pm$ 0.0113	4.98 $\pm$ 0.0113
T (450K)				
CG (1)	71.03 $\pm$ 0.8100	86.58 $\pm$ 0.8100	148.58 $\pm$ 0.5479	163.81 $\pm$ 0.5479
AA	9.67 $\pm$ 0.1235	11.51 $\pm$ 0.1235	16.67 $\pm$ 0.1774	18.17 $\pm$ 0.1774
T (500K)				
CG (1)	143.08 $\pm$ 2.5839	174.40 $\pm$ 2.5839	305.41 $\pm$ 1.8920	336.72 $\pm$ 1.8920
AA	19.20 $\pm$ 0.1293	22.79 $\pm$ 0.1293	45.68 $\pm$ 0.1823	49.74 $\pm$ 0.1823

Table S6 The hydrodynamic radii of the cations and anions of the studied DILs.

ILs	Hydrodynamic Radii R / Å	
	Cation	Anion
$[\text{C}_3(\text{mim})_2][\text{BF}_4]_2$	4.18	2.47
$[\text{C}_5(\text{mim})_2][\text{BF}_4]_2$	4.37	2.47
$[\text{C}_6(\text{mim})_2][\text{BF}_4]_2$	4.32	2.47
$[\text{C}_9(\text{mim})_2][\text{BF}_4]_2$	4.71	2.47
$[\text{C}_{12}(\text{mim})_2][\text{BF}_4]_2$	4.77	2.47

Table S7 Electrical conductivity of $[\text{C}_{12}(\text{mim})_2][\text{BF}_4]_2$ in both CG and AA models in the temperature range of 350-500 K and at 1 atm with confidence intervals expressed as $\pm\sigma$.

T (K)	CG (1)	AA
σ (mS.cm ⁻¹)		
350	$2.45 \times 10^{-6} \pm 2.4967 \times 10^{-7}$	$2.04 \times 10^{-7} \pm 3.9232 \times 10^{-8}$
400	$2.05 \times 10^{-5} \pm 1.6532 \times 10^{-6}$	$1.82 \times 10^{-6} \pm 1.1954 \times 10^{-7}$
450	$4.40 \times 10^{-5} \pm 4.6099 \times 10^{-6}$	$5.35 \times 10^{-6} \pm 5.1905 \times 10^{-7}$
500	$1.58 \times 10^{-4} \pm 9.9291 \times 10^{-6}$	$1.15 \times 10^{-5} \pm 6.3096 \times 10^{-7}$

Table S8 Transference numbers for the studied DILs in Both CG and AA models at 500 K and at 1 atm with confidence intervals expressed as $\pm\sigma$.

	t_+	t_-
$[\text{C}_3(\text{mim})_2][\text{BF}_4]_2$		
CG (1)	0.61 ± 0.0043	0.39 ± 0.0023
AA	0.57 ± 0.0104	0.43 ± 0.0050
$[\text{C}_6(\text{mim})_2][\text{BF}_4]_2$		
CG (1)	0.57 ± 0.0039	0.42 ± 0.0021
AA	0.59 ± 0.0074	0.41 ± 0.0043
$[\text{C}_9(\text{mim})_2][\text{BF}_4]_2$		
CG (1)	0.53 ± 0.0069	0.45 ± 0.0031
AA	0.57 ± 0.0120	0.43 ± 0.0065
$[\text{C}_{12}(\text{mim})_2][\text{BF}_4]_2$		
CG (1)	0.48 ± 0.0145	0.52 ± 0.0088
AA	0.46 ± 0.0055	0.54 ± 0.0050

Figure Captions

Fig. S1 Temperature-dependent density for $[\text{C}_3(\text{mim})_2][\text{BF}_4]_2$ and $[\text{C}_{12}(\text{mim})_2][\text{BF}_4]_2$ in both CG and AA models at different temperatures and at 1 atm with confidence intervals expressed as $\pm\sigma$.

Fig. S2 Isobaric expansion coefficients for the studied DILs with increasing linkage alkyl chain for both CG and AA models at 1 atm with confidence intervals expressed as $\pm\sigma$.

Fig. S3 Temperature-dependent isobaric expansion coefficients for $[\text{C}_3(\text{mim})_2][\text{BF}_4]_2$ and $[\text{C}_{12}(\text{mim})_2][\text{BF}_4]_2$ in both CG and AA models at different temperatures and at 1 atm. The error bars are smaller than the symbols.

Fig. S4 Snapshots of the simulation boxes of the studied DILs in both CG and AA models at 400 K and at 1 atm. The polar (depicted in pink) and nonpolar (depicted in blue) domains have been also shown in the figure.

Fig. S5 Comparison of anion-anion RDFs of $[\text{C}_6(\text{mim})_2][\text{BF}_4]_2$ in systems containing 252, 512 and 1000 DIL molecules in both CG and AA models.

Fig. S6 Mean square displacement (MSD) plots of cation and anion of the studied DILs at 500 K and at 1 atm in coarse-grained model. The evolution of β parameter with time has been also shown in each plot.

Fig. S7 Mean square displacement (MSD) plots of cation and anion of the studied DILs at 500 K and at 1 atm in all-atom model. The evolution of β parameter with time has been also shown in each plot.

Fig. S8 Mean square displacement (MSD) plots of cation and anion of $[\text{C}_{12}(\text{mim})_2][\text{BF}_4]_2$ in coarse-grained model in the temperature range of 350-500 K and at 1 atm. The evolution of β parameter with time has been also shown in each plot.

Fig. S9 Mean square displacement (MSD) plots of cation and anion of $[\text{C}_{12}(\text{mim})_2][\text{BF}_4]_2$ in all-atom model in the temperature range of 350-500 K and at 1 atm. The evolution of β parameter with time has been also shown in each plot.

Fig. S1

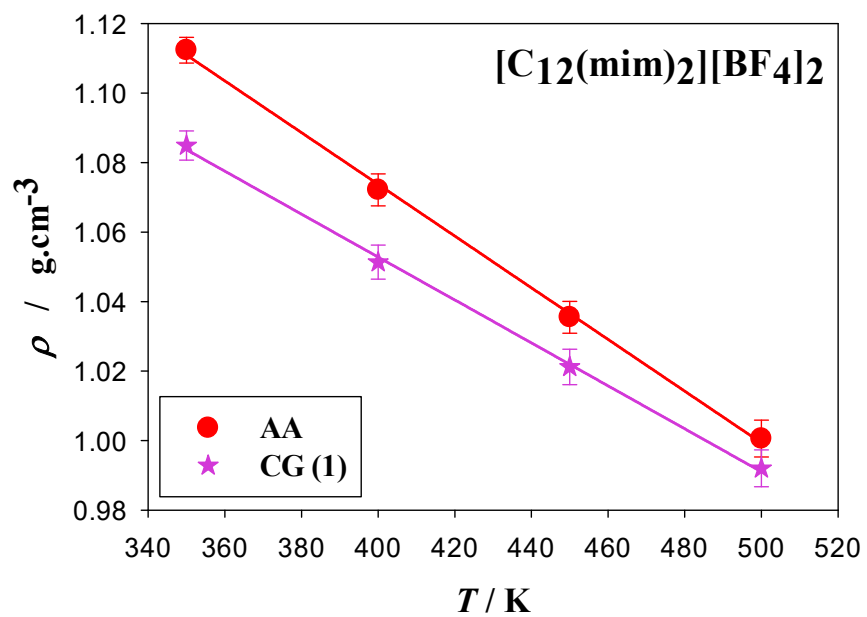
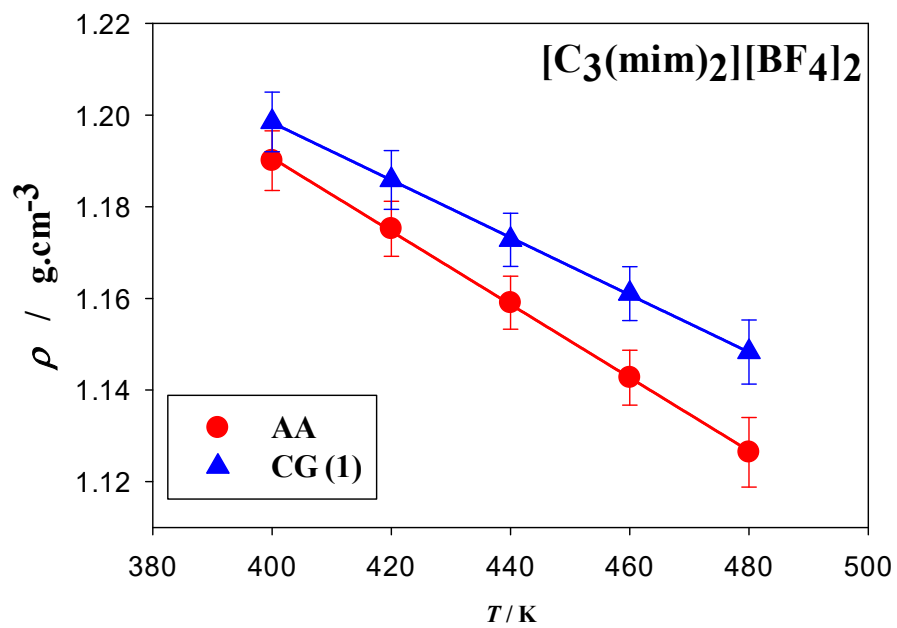


Fig. S2

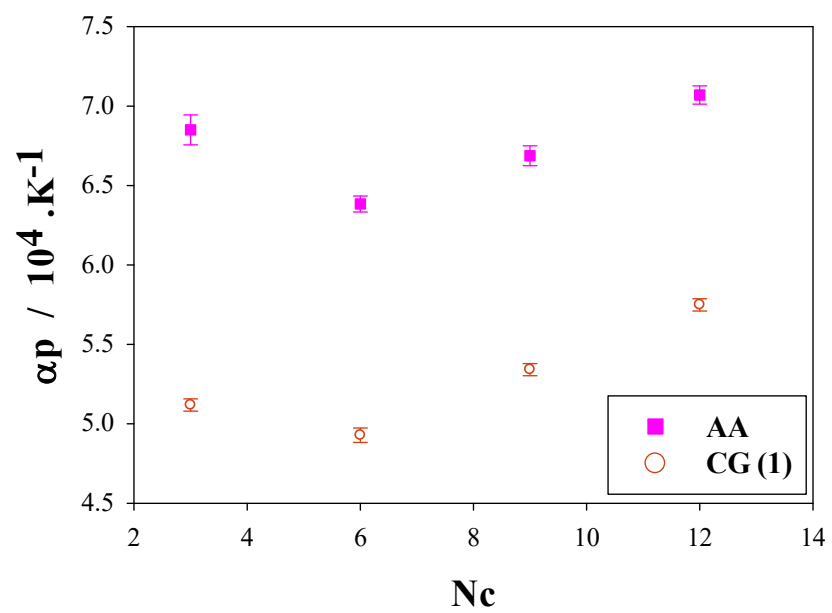


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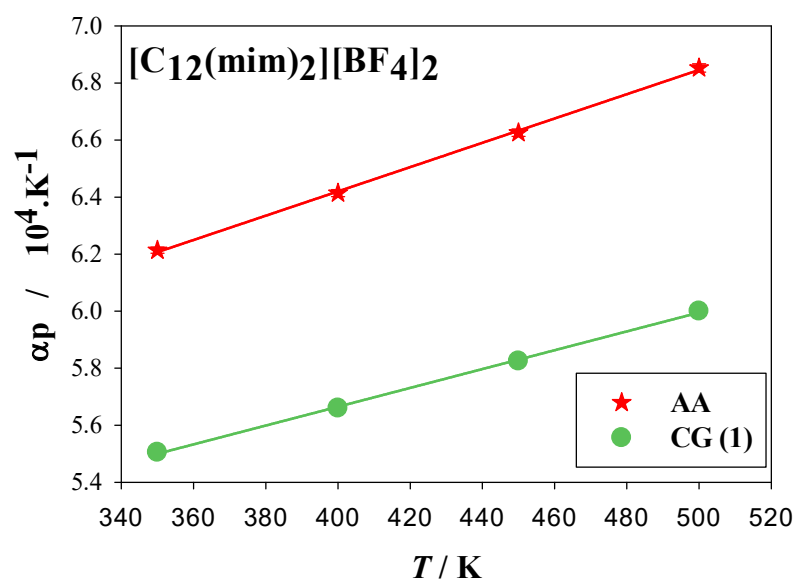
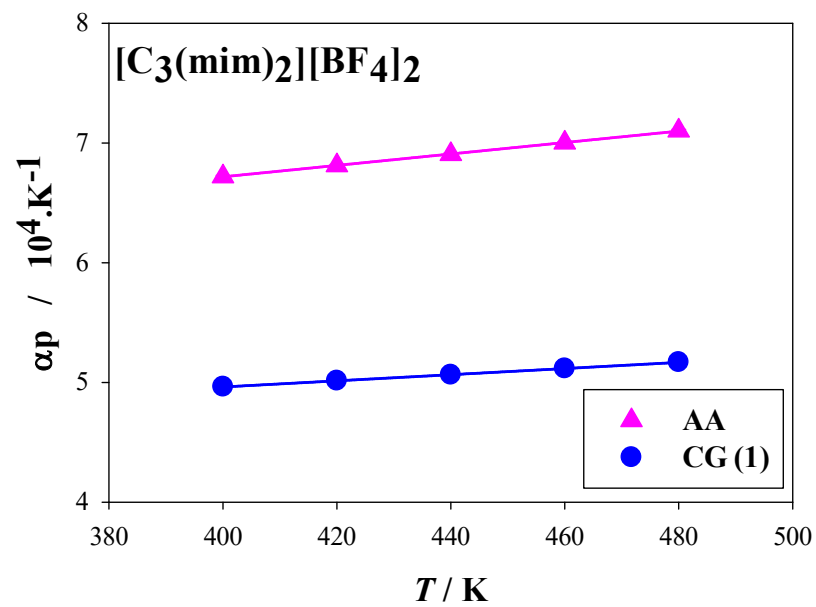
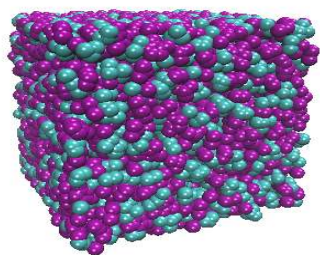
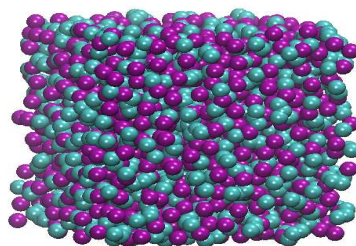


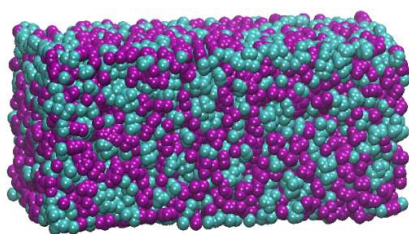
Fig. S4



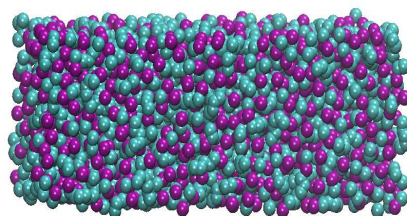
[C₃(mim)₂][BF₄]₂ (AA)



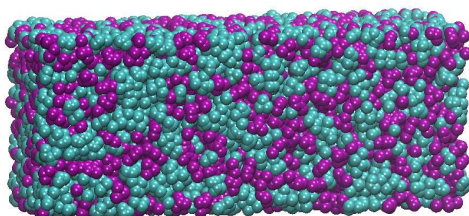
[C₃(mim)₂][BF₄]₂ (CG (1))



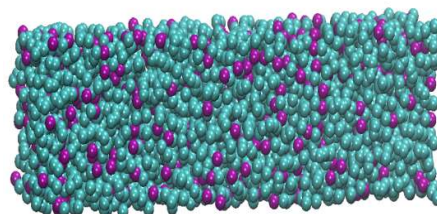
[C₆(mim)₂][BF₄]₂ (AA)



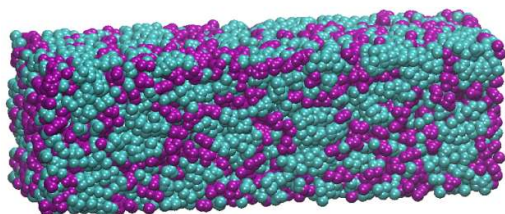
[C₆(mim)₂][BF₄]₂ (CG (1))



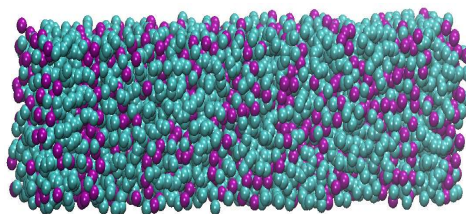
[C₉(mim)₂][BF₄]₂ (AA)



[C₉(mim)₂][BF₄]₂ (CG (1))



[C₁₂(mim)₂][BF₄]₂ (AA)



[C₁₂(mim)₂][BF₄]₂ (CG (1))

Fig. S5

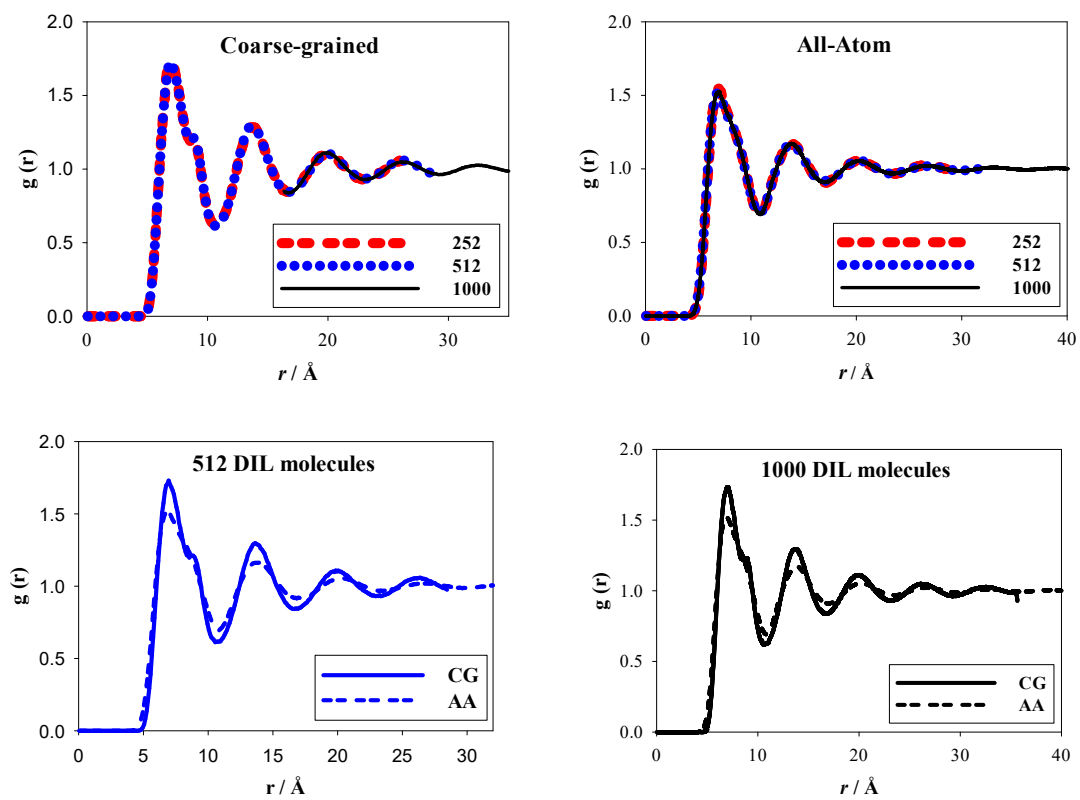


Fig. S6

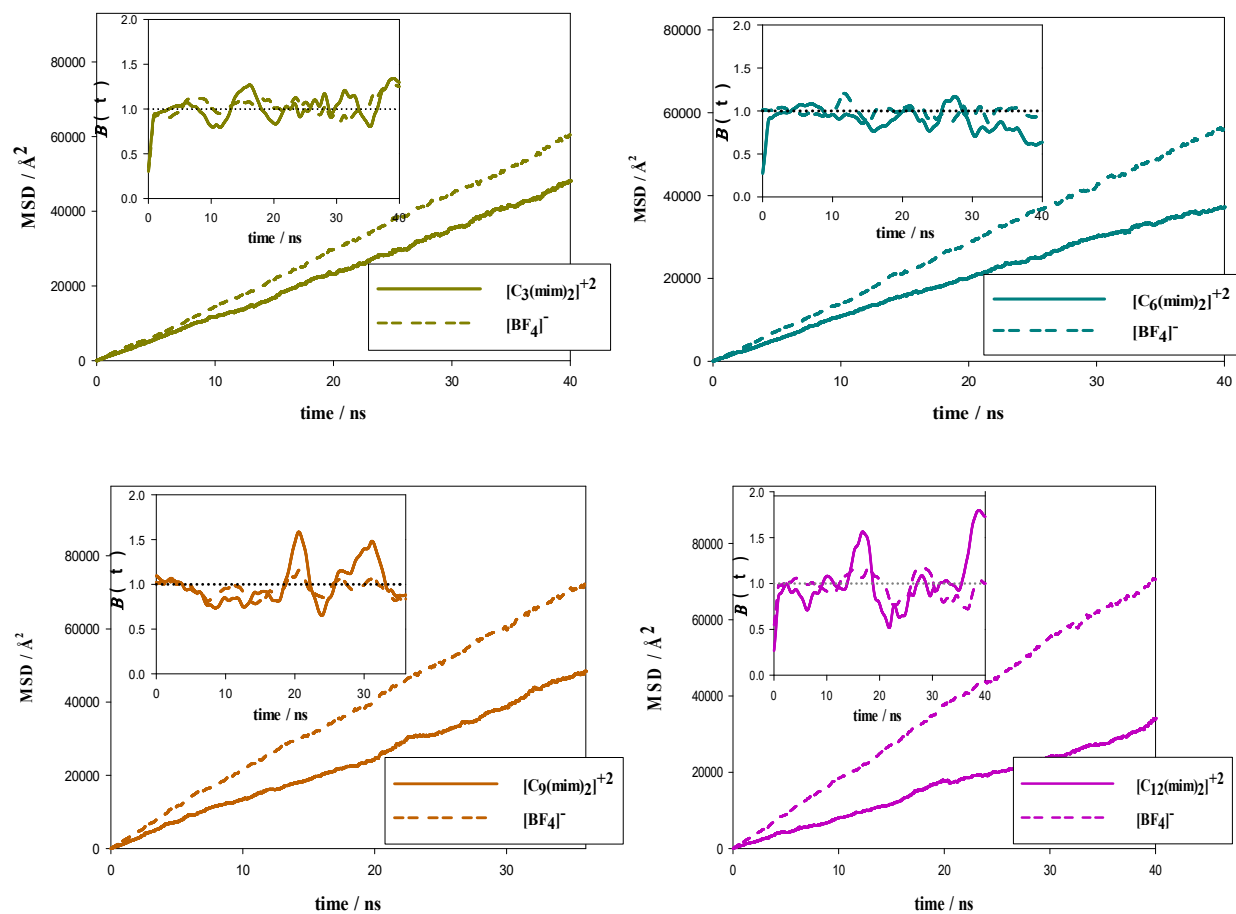


Fig. S7

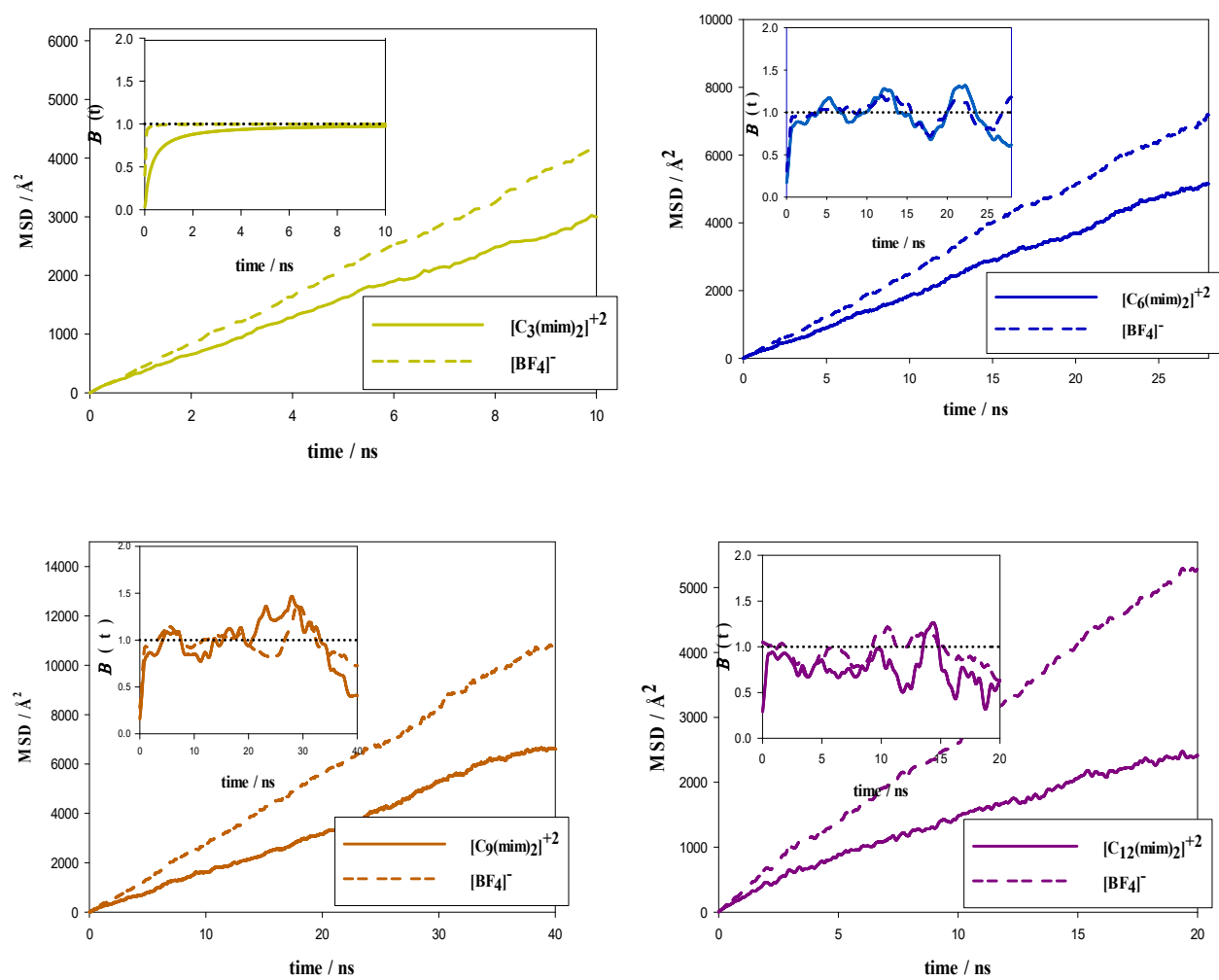


Fig. S8

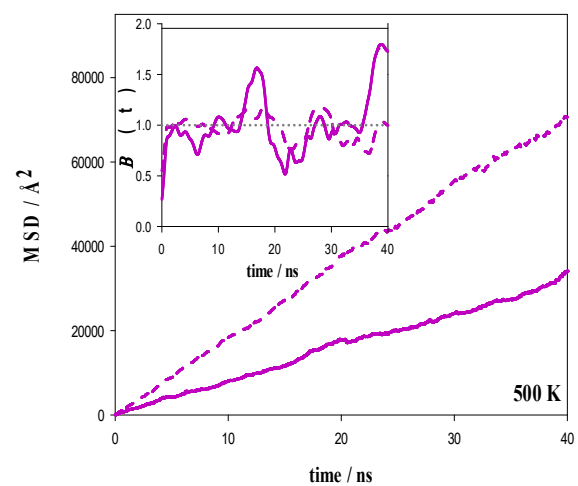
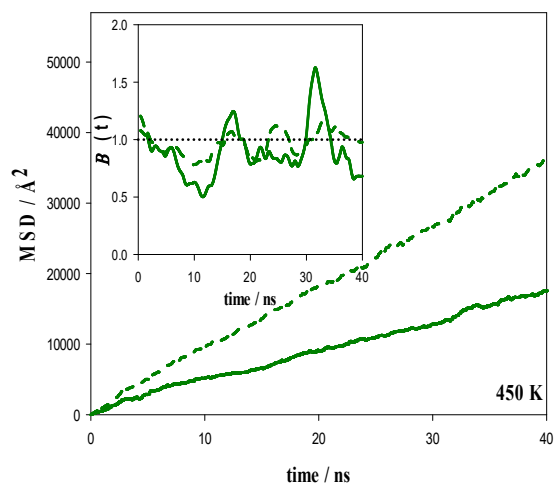
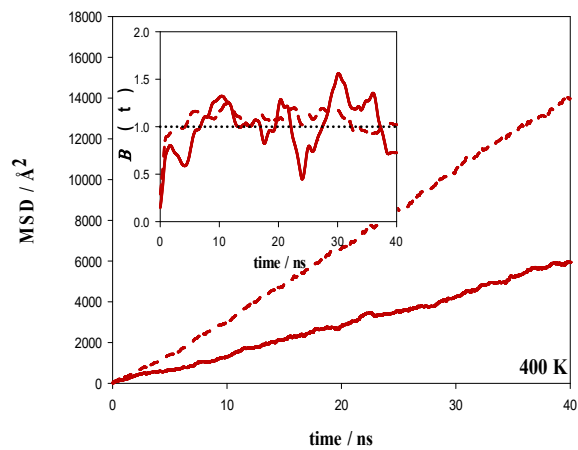
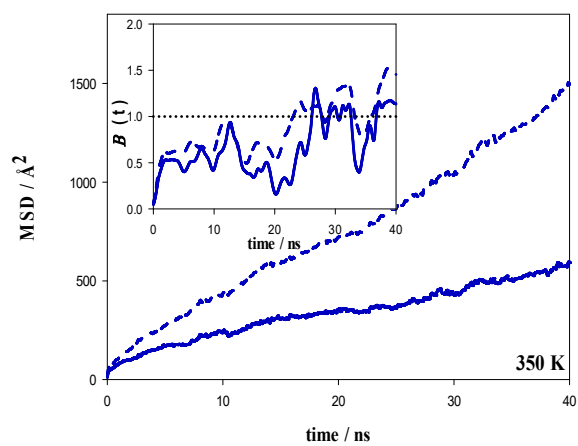


Fig. S9

