

## Carbon-Doped Hexagonal Boron Nitride: an Effective Catalyst for Oxidative Desulfurization

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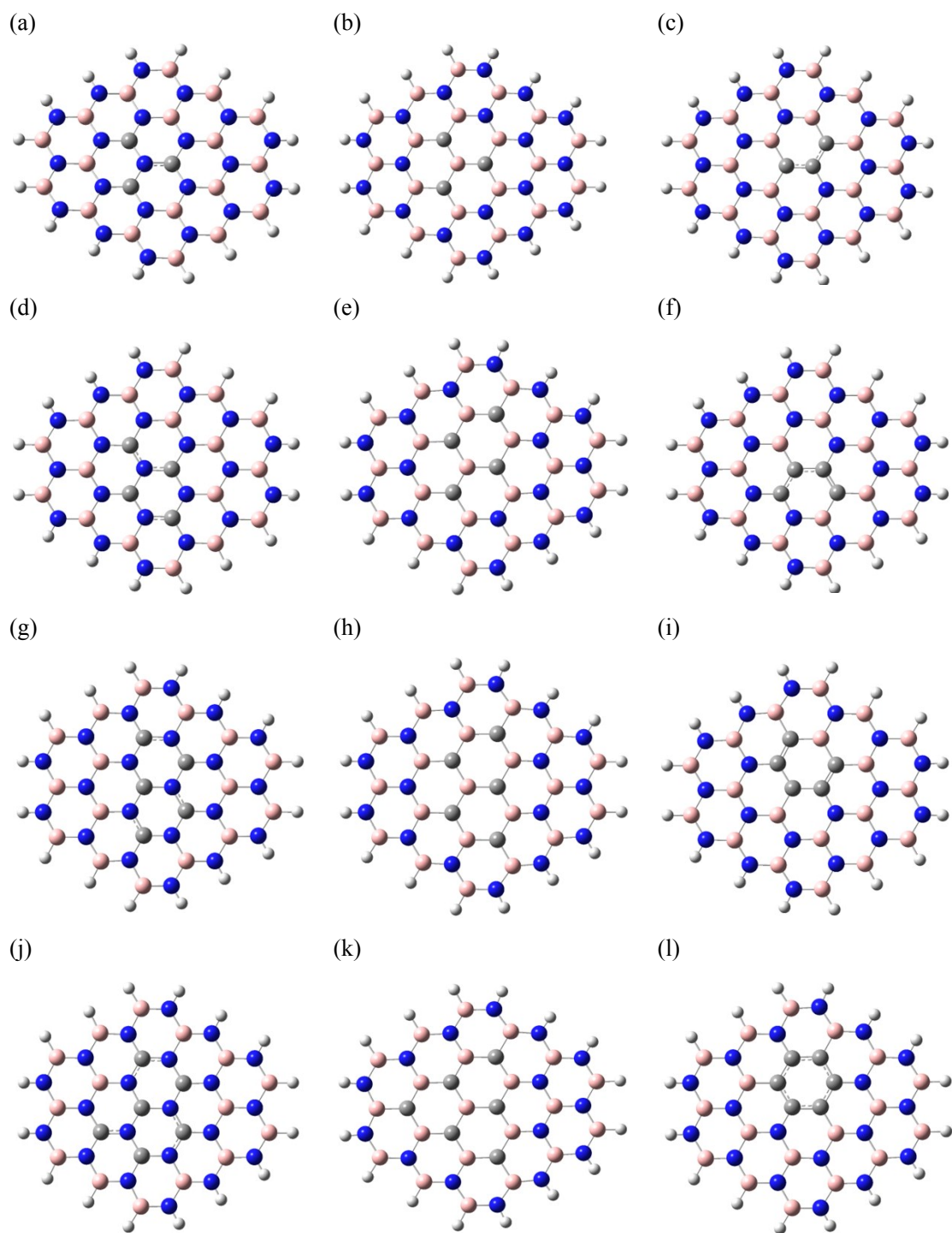


Figure S1. Cluster models for C-doped h-BN sheet (a)  $C_{3B}$ , (b)  $C_{3N}$ , (c)  $C_{B_2N}$ , (d)  $C_{4B}$ , (e)  $C_{4N}$ , (f)  $C_{2B_2N}$ , (g)  $C_{5B}$ , (h)  $C_{5N}$ , (i)  $C_{2B_3N}$ , (j)  $C_{6B}$ , (k)  $C_{6N}$ , (l)  $C_{3B_3N}$ .

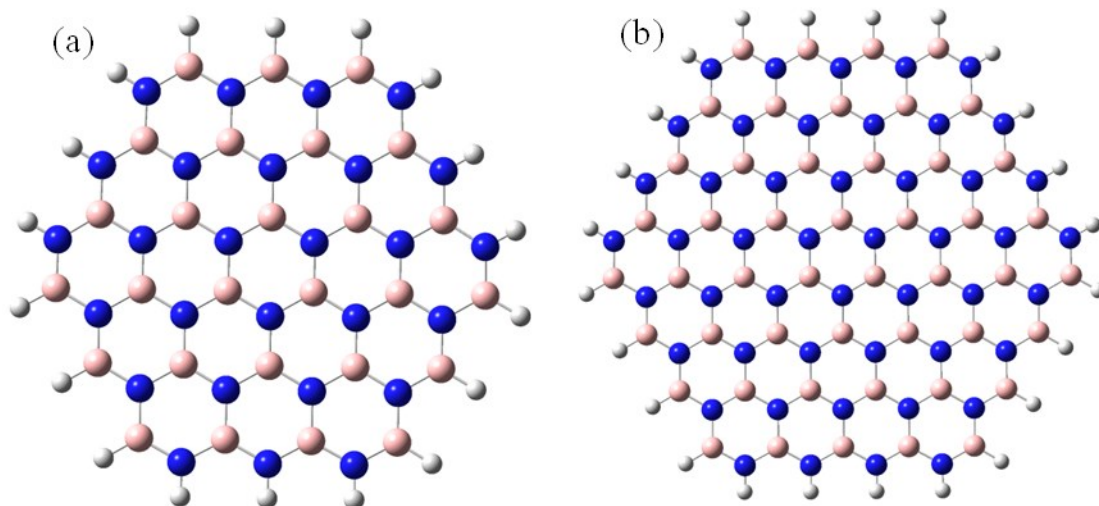


Figure S2. Cluster models for h-BN sheet (a)  $B_{27}N_{27}$ , (b)  $B_{48}N_{48}$

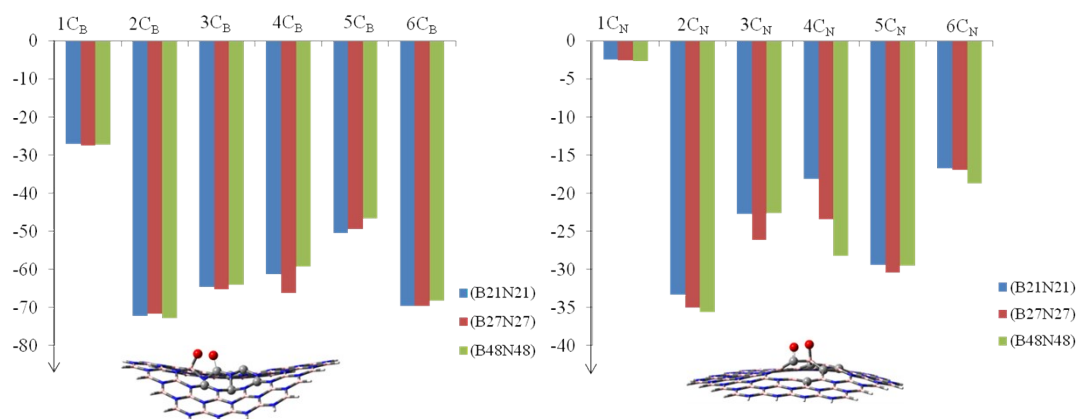


Figure S3. The adsorption energies for  $O_2$  on the different C-doped h-BN models. Left:  $O_2$  on the dispersed  $C_B$  surface. Right:  $O_2$  on the dispersed  $C_N$  surface.

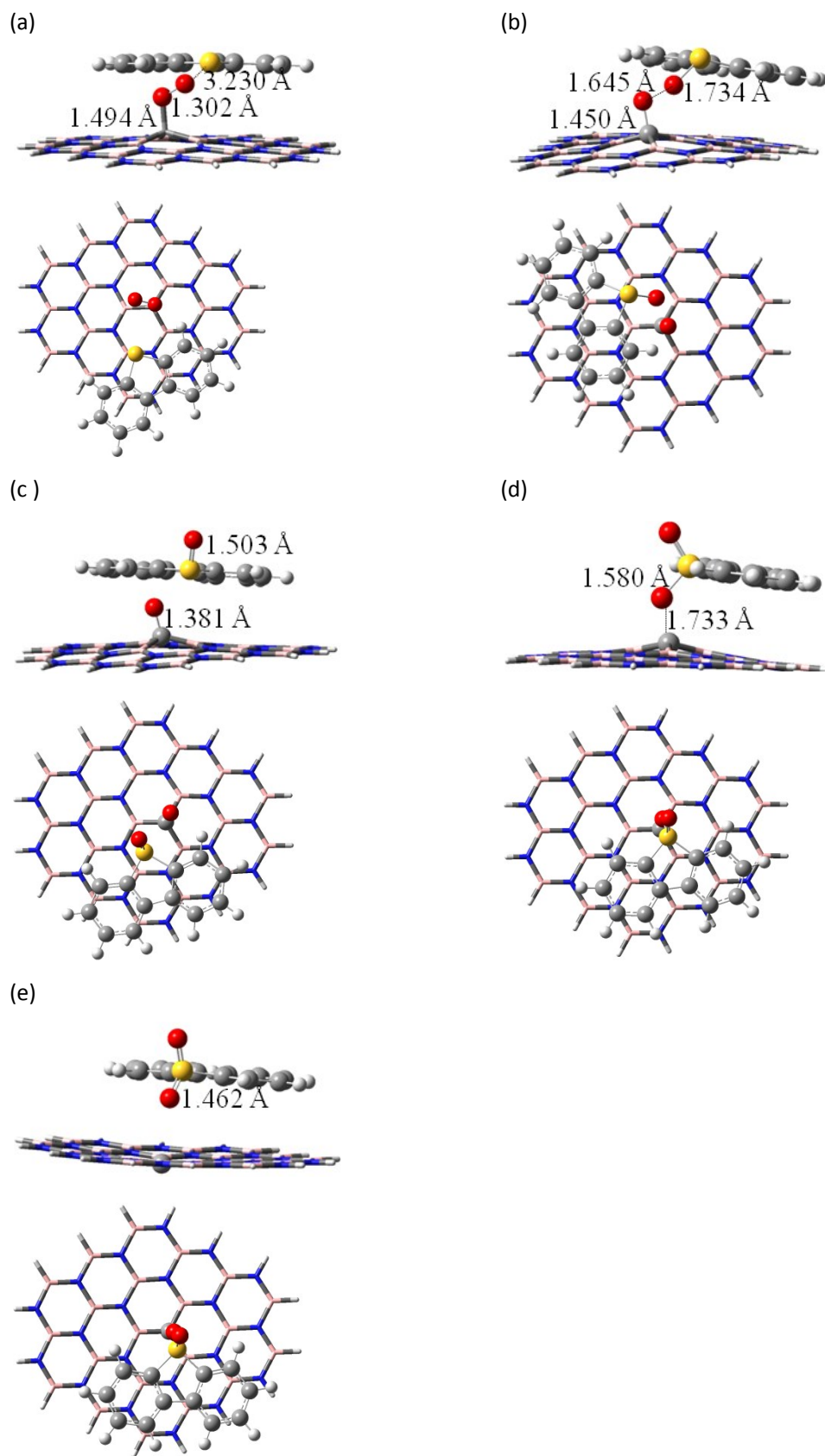


Figure S4 DBT oxidation on  $C_N$  model. (a)  $O_2$  and DBT co-adsorption (b)TS1 (c) IM1, DBTO (d) TS2 (e)FS, DBTOO

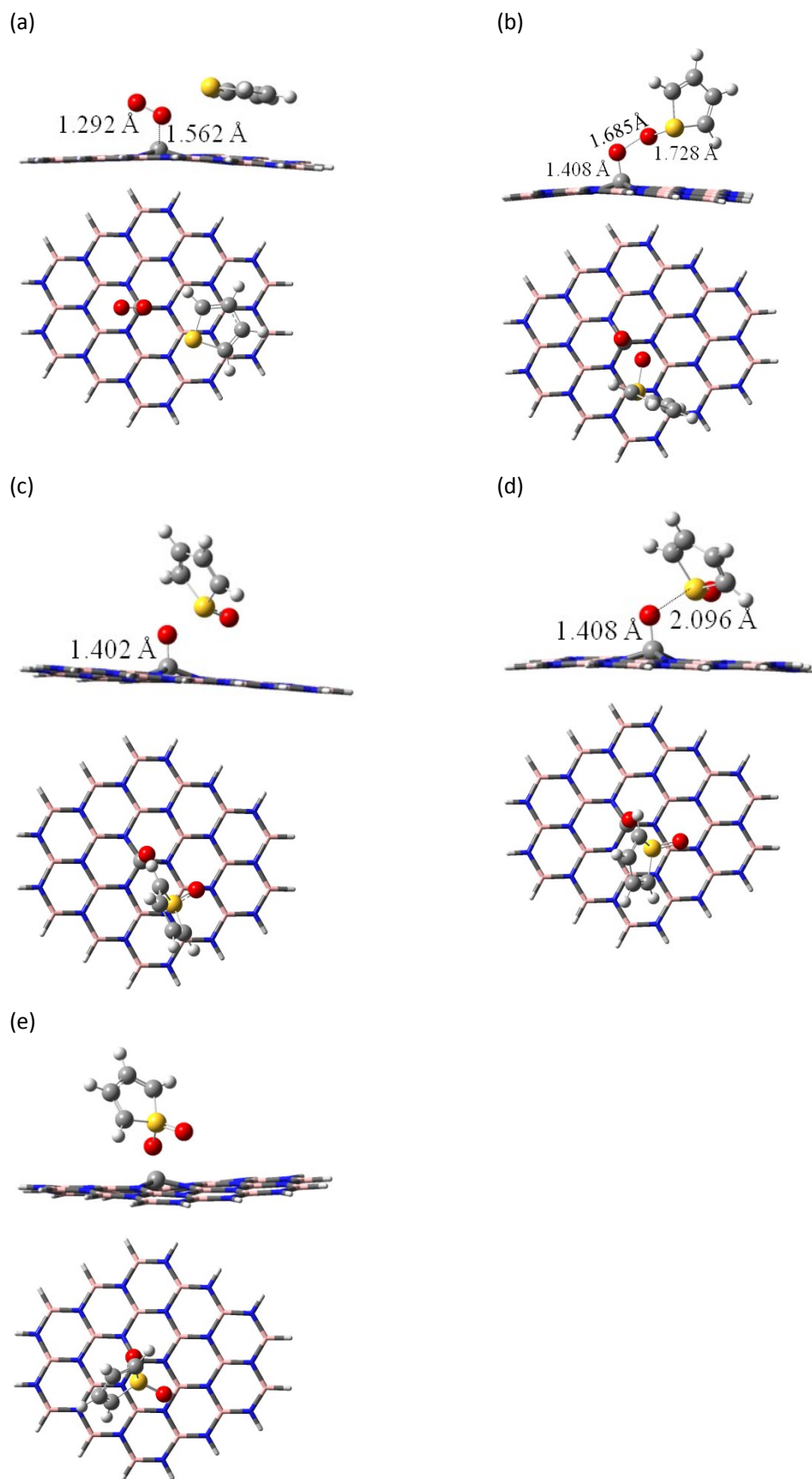


Figure S5 TH oxidation on  $C_B$  model. (a)  $O_2$  and TH co-adsorption (b)TS1 (c) IM1 (d) TS2 (e) FS

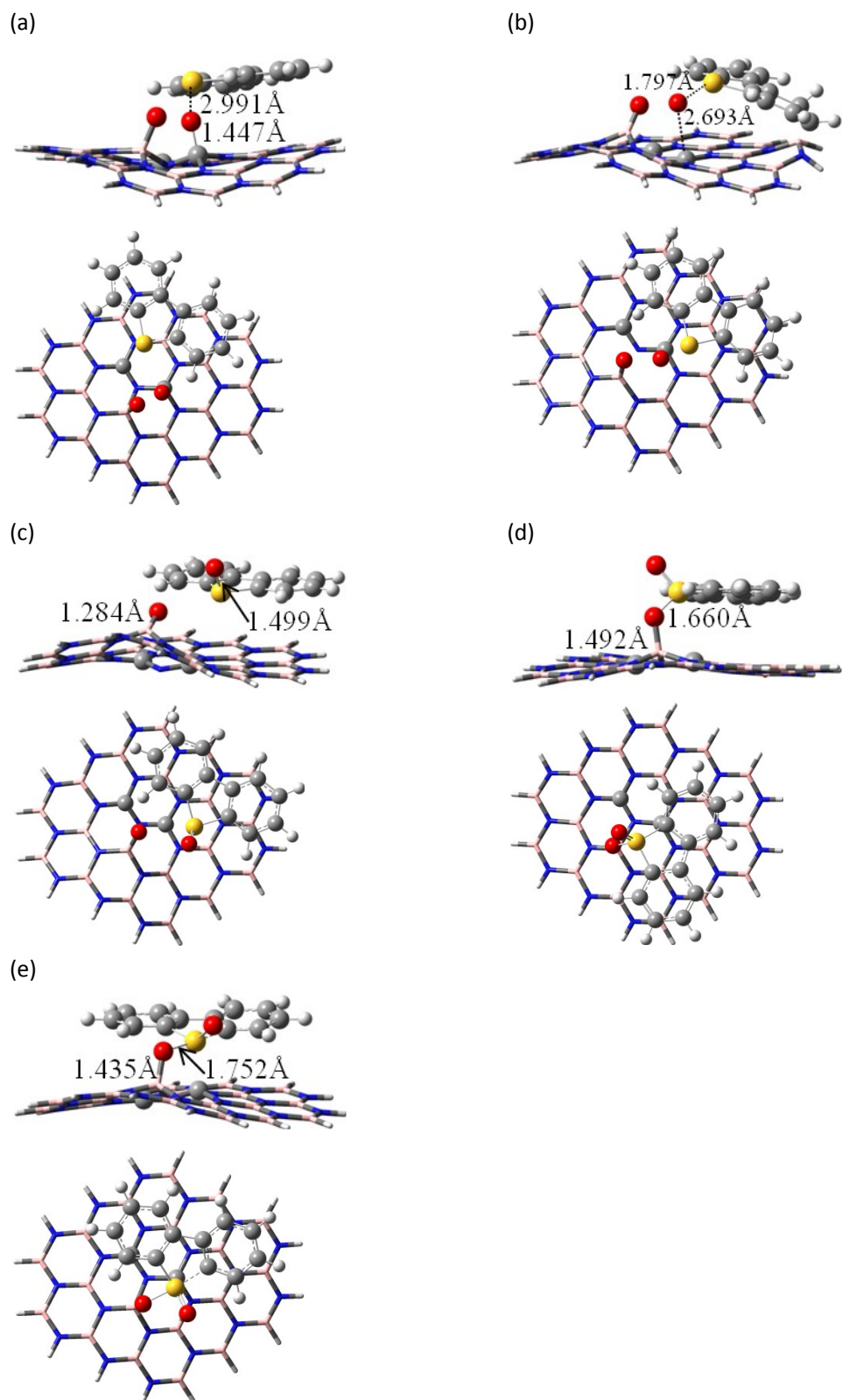


Figure S6 DBT oxidation on  $C_{2B}$  model. (a)  $O_2$  and DBT co-adsorption (b) TS1 (c) IM1 (d) TS2 (e) FS



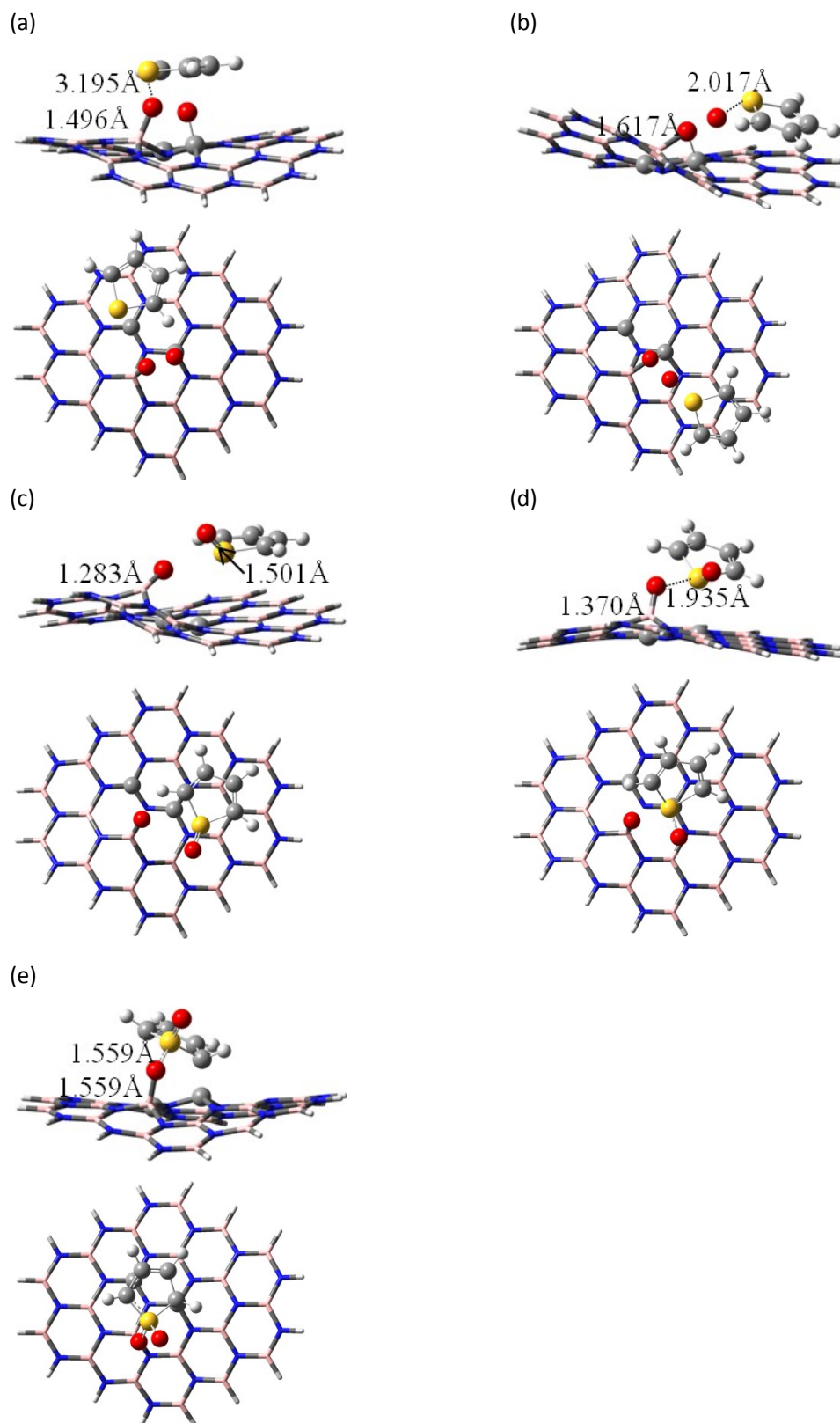


Figure S7 TH oxidation on C<sub>2B</sub> model. (a) O<sub>2</sub> and TH co-adsorption (b)TS1 (c) IM1 (d) TS2 (e) FS

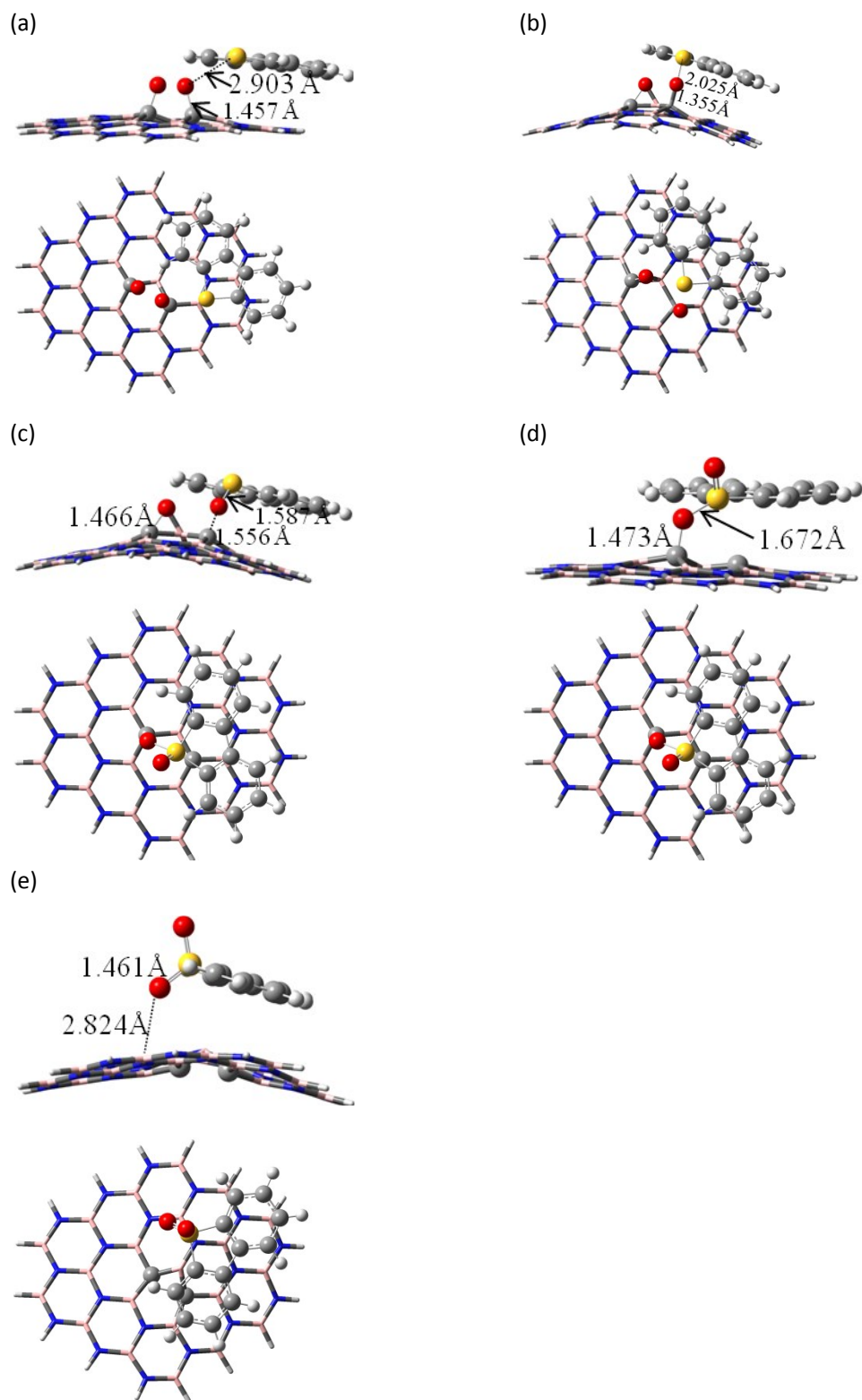


Figure S8 DBT oxidation on  $C_{2N}$  model. (a)  $O_2$  and DBT co-adsorption (b)TS1 (c) IM1 (d) TS2 (e) FS



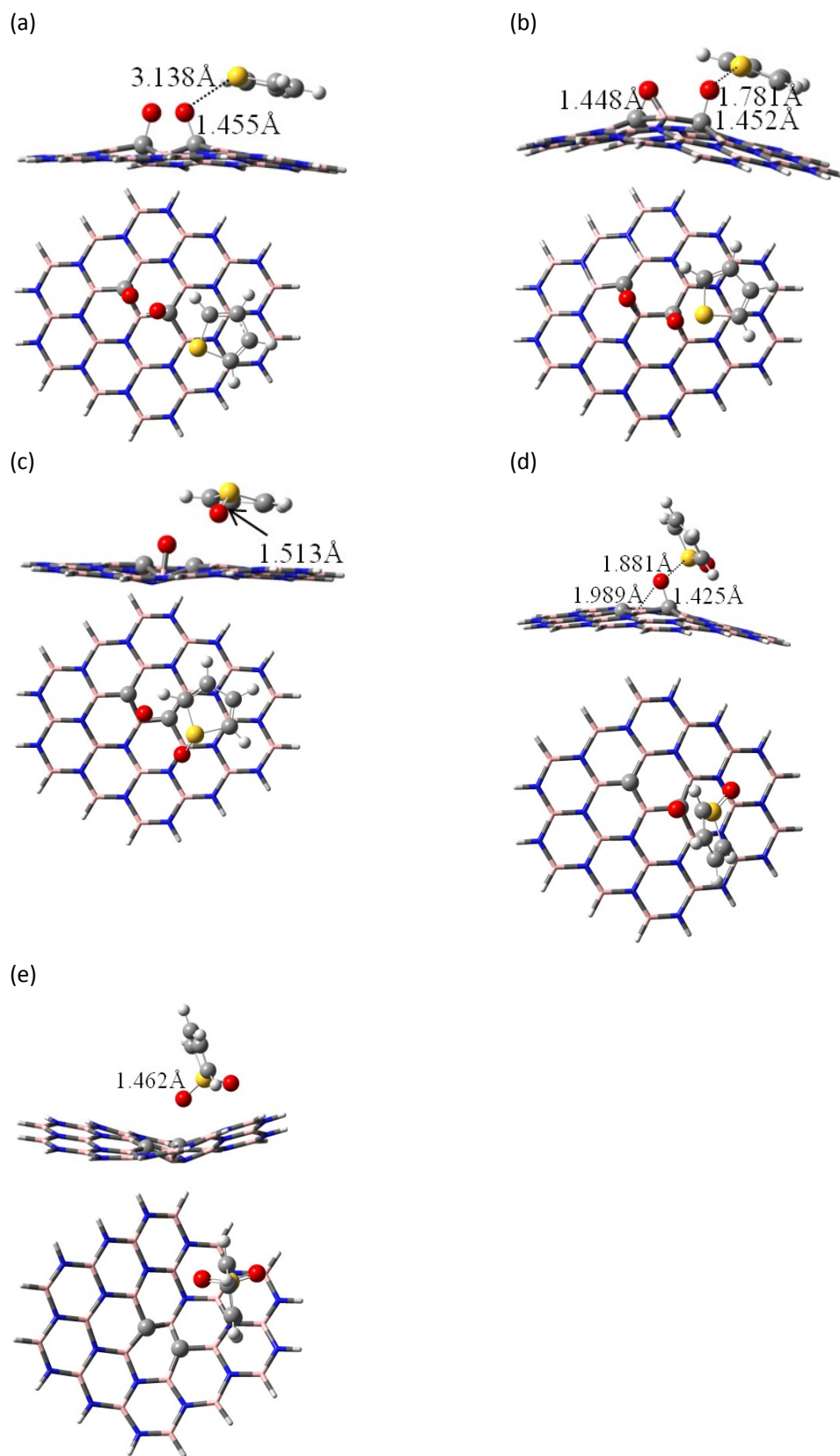


Figure S9 TH oxidation on  $C_{2N}$  model. (a)  $O_2$  and TH co-adsorption (b)TS1 (c) IM1 (d) TS2 (e) FS

Table S1 The adsorption Gibbs free energies ( $G_{ad}$ , kcal/mol) of thiophene (TH) on  $C_N$  monolayer.

	sulfcompound	sulfcompound +O <sub>2</sub>	TS1	IM1	TS2	FS
TH	3.5	5.2	57.3	14.9	30.5	23.8
DBT	-2.6	1.5	45.6	4.2	29.3	-11.7

Table S2 The adsorption Gibbs free energies ( $G_{ad}$ , kcal/mol) of TH and DBT on  $C_B$  monolayer.

	Sulfite	sulfite+O <sub>2</sub>	TS1	IM1	TS2	FS
TH	3.9	-13.6	43.7	8.5	15.5	6.3
DBT	-0.4	-26.2	30.0	-6.1	4.0	-13.1