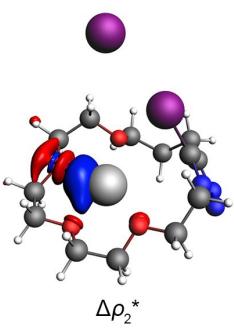


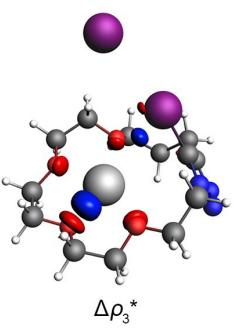
$\Delta\rho_1^*$

$$\Delta q_1 = 0.174 \text{ a.u.}$$



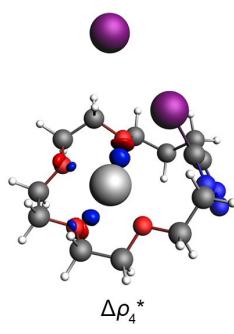
$\Delta\rho_2^*$

$$\Delta q_2 = 0.168 \text{ a.u.}$$



$\Delta\rho_3^*$

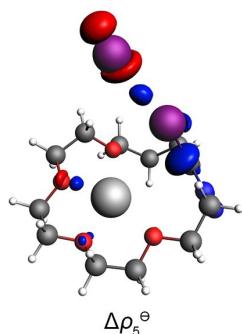
$$\Delta q_3 = 0.145 \text{ a.u.}$$



$\Delta\rho_4^*$

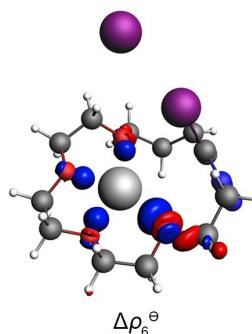
$$\Delta q_4 = 0.118 \text{ a.u.}$$

$$\Delta E_{\text{oi},1} = -6.69 \text{ kcal mol}^{-1} \quad \Delta E_{\text{oi},2} = -6.59 \text{ kcal mol}^{-1} \quad \Delta E_{\text{oi},3} = -4.10 \text{ kcal mol}^{-1} \quad \Delta E_{\text{oi},4} = -2.82 \text{ kcal mol}^{-1}$$



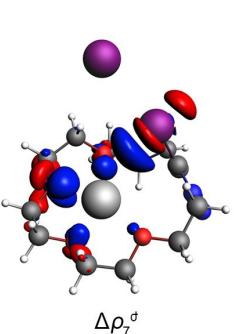
$\Delta\rho_5^\ominus$

$$\Delta q_5 = 0.097 \text{ a.u.}$$



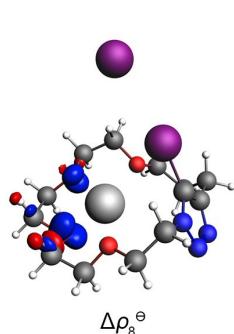
$\Delta\rho_6^\ominus$

$$\Delta q_6 = 0.083 \text{ a.u.}$$



$\Delta\rho_7^\ddagger$

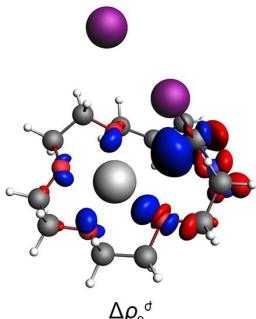
$$\Delta q_7 = 0.074 \text{ a.u.}$$



$\Delta\rho_8^\ddagger$

$$\Delta q_8 = 0.072 \text{ a.u.}$$

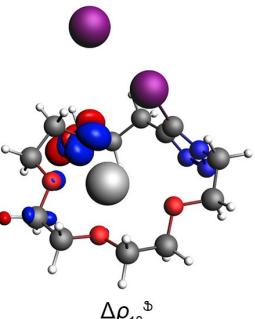
$$\Delta E_{\text{oi},5} = -1.47 \text{ kcal mol}^{-1} \quad \Delta E_{\text{oi},6} = -1.99 \text{ kcal mol}^{-1} \quad \Delta E_{\text{oi},7} = -1.58 \text{ kcal mol}^{-1} \quad \Delta E_{\text{oi},8} = -1.58 \text{ kcal mol}^{-1}$$



$\Delta\rho_9^\ddagger$

$$\Delta q_9 = 0.063 \text{ a.u.}$$

$$\Delta E_{\text{oi},9} = -1.12 \text{ kcal mol}^{-1}$$



$\Delta\rho_{10}^\ddagger$

$$\Delta q_{10} = 0.051 \text{ a.u.}$$

$$\Delta E_{\text{oi},10} = -0.88 \text{ kcal mol}^{-1}$$