# Supporting information 

# Computational Design of Janus Polymersomes with Controllable Fission from Double Emulsions 

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## S1. DPD fundamentals.

In DPD method, like molecular dynamics, the motion of all the DPD beads obeys Newton's equation of motion.

$$
\begin{equation*}
\frac{d r_{i}}{d t}=v_{i}, \quad \frac{d v_{i}}{d t}=f_{i} / m_{i} \tag{S1}
\end{equation*}
$$

where $\boldsymbol{r}_{i}, \boldsymbol{v}_{i}$ and $\mathrm{m}_{i}$ denote the position vector, velocity vector and mass of beads respectively, and $\boldsymbol{f}_{i}$ is the force acting on bead $i$. The force $\boldsymbol{F}_{i j}$ exerted on bead $i$ by bead $j$ is consisted of a conservative force ${ }^{F_{i j}}$, a dissipative force ${ }^{F_{i j}^{D}}$ and a random force $F_{i j}^{R}$. Thus, the total force $\boldsymbol{f}_{i}$ is given by

$$
\begin{equation*}
f_{i}=\sum_{j \neq i}\left(F_{i j}^{C}+F_{i j}^{D}+F_{i j}^{R}\right) \tag{S2}
\end{equation*}
$$

where the sum runs over all other beads within the cutoff radius $r_{c}$. And the conservative force ${ }_{i j}^{C}$, dissipative force $F_{i j}^{D}$ and random force $F_{i j}^{R}$ are given by

$$
\begin{align*}
& F_{i j}^{C}= \begin{cases}a_{i j}\left(r_{c}-r_{i j}\right) e_{i j} & r_{i j} \leq r_{c} \\
r_{i j}>r_{c}\end{cases}  \tag{S3}\\
& F_{i j}^{D}=-\gamma \omega^{D}\left(r_{i j}\right)\left(e_{i j} \cdot v_{i j}\right) e_{i j}  \tag{S4}\\
& F_{i j}^{R}=\sigma \omega^{R}\left(r_{i j}\right) \theta_{i j} e_{i j} \tag{S5}
\end{align*}
$$

where $a_{i j}$ is a constant that describes the maximum repulsion between two interacting beads. $r_{i j}$ is the distance between beads $i$ and $j$. $e_{i j}=r_{i j} / r_{i j}, v_{i j}=v_{i}-v_{j}, v_{i}$ and $v_{j}$ are the velocities of beads $i$ and $j$, respectively. $\gamma$ and $\sigma$ are the amplitudes of dissipative and random forces, respectively. $\theta_{i j}$ is a randomly fluctuating variable. $\omega^{D}$ and $\omega^{R}$ are r-dependent weight functions for dissipative and random forces, respectively. According to the fluctuation-dissipation theorem, $\omega_{i j} D(r)=\left[\omega_{i j}^{R}(r)\right]^{2}$ and $\sigma^{2}=2 \gamma k_{B} T$ ( $\sigma=3$ and $\gamma=4.5$ ). The following simple form of $\omega^{D}$ and $\omega^{R}$ was chosen by Groot and Warren ${ }^{1}$ :

$$
\omega^{D}(r)=\left[\omega^{R}(r)\right]^{2}= \begin{cases}\left(1-r / r_{c}\right)^{2} & r \leq r_{c}  \tag{S6}\\ 0 & r>r_{c}\end{cases}
$$

Also, the chain beads of polymers are connected by a harmonic bond:

$$
\begin{equation*}
F_{i j}^{B}=-\sum_{j} C^{B}\left(r_{i j}-r_{e q}\right) e_{i j} \tag{S7}
\end{equation*}
$$

where $C^{B}$ and $r_{e q}$ denotes the bond rigidity and the equilibrium bond length, respectively. Here, the most common and reliable values $\left(C^{B}=4.0 \text { and } r_{e q}=0\right)^{1-5}$ were adopted in our simulations.

The repulsive interaction parameters ( $a_{i j}$ in eq S 3 ) can be estimated from the $\chi$-parameter in Flory-Huggins theory according to the study of Groot and Warren ${ }^{1}$ as follows:

$$
\begin{equation*}
a_{i j}=a_{i i}+3.27 \chi_{i j} \tag{S8}
\end{equation*}
$$

## S2. Finite size effect.



Fig. S1. Snapshots of Janus polymersomes obtained in different boxes sized $\left(40 r_{\mathrm{c}}\right)^{3},\left(50 r_{\mathrm{c}}\right)^{3}$ and $\left(100 r_{\mathrm{c}}\right)^{3}$, respectively. Color code: block A, red; block B, cyan; block C, yellow.

## S3. The stability of double emulsions.

(a)

(b)


(c)


Fig. S2. Sequential sectional snapshots of the double emulsions with different $\mathrm{A}_{4} \mathrm{~B}_{8}$ diblock copolymer concentrations $\left(\varphi_{\text {polymer }}\right)$. (a) $\varphi_{\text {polymer }}=0.02$, and the simulation times are $0,2.0 \times 10^{5}$, $5.0 \times 10^{5}$, and $1.0 \times 10^{6}$ steps, respectively. (b) $\varphi_{\text {polymer }}=0.10$, and the simulation times are 0 and $2.0 \times 10^{6}$ steps. $d_{1}$ and $d_{2}$ are the thickness of the oil phase in different positions. (c) $\varphi_{\text {polymer }}=0.40$, and the simulation times are 0 and $2.0 \times 10^{6}$ steps. Color code: Water, pink; Oil, mauve; block A, red; block B, cyan.

## S4. Dewetting of the double emulsion.

Table S1. Inner size $\left(r_{\text {in }}\right)$, surface area $(A)$ and the critical number of polymer chains $\left(N_{\mathrm{c}}\right)$ for different double emulsions.

| $\boldsymbol{r}_{\text {in }}$ | $\mathbf{6}$ | $\mathbf{8}$ | $\mathbf{1 0}$ | $\mathbf{1 2}$ | $\mathbf{1 4}$ | $\mathbf{1 6}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\boldsymbol{A}$ | 452.4 | 804.2 | 1256.6 | 1809.6 | 2463.0 | 3217.0 |
| $\boldsymbol{N}_{\mathbf{c}}$ | 1000 | 1500 | 2500 | 3500 | 5000 | 6500 |


(b)

(c)


Fig. S3. Sequential sectional snapshots of the dewetting process with different polymer concentrations. (a) $N_{\text {polymer }}=2000$, and the simulation times are $0,1.0 \times 10^{6}, 2.0 \times 10^{6}$, and $2.5 \times$ $10^{6}$ steps, respectively. (b) $N_{\text {polymer }}=3500$, and the simulation times are $0,1.0 \times 10^{6}, 2.0 \times 10^{6}$, and $3.0 \times 10^{6}$ steps, respectively. (c) $N_{\text {polymer }}=4500$, and the simulation times are $0,1.0 \times 10^{6}, 1.6 \times$ $10^{6}$, and $3.0 \times 10^{6}$ steps, respectively. Water phase are omitted for clarity. Color code: Oil, mauve; block A, red; block B, cyan.

## S5. Calculation of the degree of phase separation.

As shown in Fig. S5, the degree of phase separation ( $d p s$ ) of blocks B and C is defined as ${ }^{6}$
$d p s=1-\frac{\left\langle\rho_{B} \rho_{C}\right\rangle}{\sqrt{\left\langle\rho_{B}^{2}\right\rangle\left\langle\rho_{C}^{2}\right\rangle}}$
where $\rho_{\mathrm{B}}$ and $\rho_{\mathrm{C}}$ are the number densities of beads B and C in the spherical space with each bead B or C as the origin and $r_{\text {cut }}$ as the cutoff radius. $\langle\cdots\rangle$ is the ensemble average. $d p s=0$ means beads B and C mix perfectly, while $d p s=1$ is for complete phase separation.


Fig. S4. Schematic diagram for the calculation of the degree of phase separation. Beads B and C are in cyan and yellow, respectively.

## S6. Co-assembly of $\mathrm{AB} / \mathbf{A C}$ blends through double emulsion.



Fig. S5. Sequential sectional snapshots of the formation of two-component polymersomes with different values of $a_{\mathrm{BC}}$. (a) $a_{\mathrm{BC}}=25$, and the simulation times are $0,0.8 \times 10^{6}, 1.2 \times 10^{6}$, and 2.0 $\times 10^{6}$ steps, respectively. (b) $a_{\mathrm{BC}}=30$, and the simulation times are $0,0.8 \times 10^{6}, 1.2 \times 10^{6}$, and 2.0 $\times 10^{6}$ steps, respectively. (c) $a_{\mathrm{BC}}=40$, and the simulation times are $0,0.8 \times 10^{6}, 1.2 \times 10^{6}$, and 2.0 $\times 10^{6}$ steps, respectively. Water phase are omitted for clarity. Color code: Oil, mauve; block A, red; block B, cyan; block C, yellow.

## S7. Co-assembly of $\mathbf{A B} / \mathbf{A C} / \mathrm{BC}$ blends through double emulsion.



Fig. S6. Sectional snapshots of the vesicles obtained from $A B / A C$ blends and $A B / A C / B C$ blends with different values of $a_{\mathrm{BC}}$. Color code: block A, red; block B, cyan; block C, yellow.


Fig. S7. Sequential sectional snapshots of the formation of Janus polymersome. The simulation times are $0,0.6 \times 10^{6}, 1.0 \times 10^{6}$, and $2.0 \times 10^{6}$ steps, respectively. Water phase are omitted for clarity. Color code: Oil, mauve; block A, red; block B, cyan; block C, yellow.


Fig. S8. Sectional snapshots of Janus polymersome formed through the co-assembly of $\mathrm{AB} / \mathrm{CD} / \mathrm{BC}$ blends with $a_{\mathrm{AD}}=25$ (a) and $a_{\mathrm{AD}}=40(\mathrm{~b})$. Water phase are omitted for clarity. Color code: block A, red; block B, cyan; block C, yellow; block D, magenta.

## S8. Effect of the number of BC.



Fig. S9. (a) Sectional snapshot of the Janus polymersome obtained with $N_{\mathrm{BC}}=285$. (b) Enlarged detail the extra BC domain in the membrane. (c) Snapshot of the extra BC domain attached to the Janus ring. Color code: block A, red; block B, cyan; block C, yellow.

## S9. Mean-square displacement curves of polymers.



Fig. S10. Mean-square displacement (MSD) curves of polymers with different polymer concentrations.

## S10. References.

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