

Supplementary Information

The preferred geometry of hydroperoxides is the result of an interplay between electrostatic and hyperconjugative effects

Marco A. S. Afonso,^a Rodrigo A. Cormanich^{*a}

University of Campinas, Institute of Chemistry, Department of Organic Chemistry, 13083-970 –
Campinas, SP, Brazil.

^{*}Corresponding author: cormanich@unicamp.br

Table of Contents

Computational Details	S4
Table S1. Mean absolute error (MAE) for single point calculations for the basis set benchmark.....	S5
Table S2. Mean absolute error (MAE) for single point calculations for the method benchmark.....	S5
Table S3. Mean absolute error (MAE) obtained from the potential energy curves	S6
Table S4. Global minimum dihedral angle obtained from the potential energy curves.....	S7
Figure S1. Atom numbering used for the molecules	S8
Figure S2. Total $\Delta E(T)$, non-Lewis $\Delta E(NL)$, Lewis $\Delta E(L)$, Natural Coulomb Electrostatic $\Delta E(NCE)$ and Natural steric analysis $\Delta E(STERIC)$ potential energy curves	S9
Figure S3. Individual natural coulombic potential energy curves and atomic charge	S10
Figure S4. Hyperconjugation interaction energies	S11
Table S5. Steric interactions energies for $\text{HOOC}(\text{CH}_3)_3$	S12
Table S6. Steric interactions energies for HOOCF_3	S12
Figure S5. Relative interatomic distances and bond angles	S13
Table S7. Bond lengths, bonds angles and dihedral angle at minimum	S15
Table S8. HOMO and LUMO energies at minimum	S16
Table S9. HOMO and LUMO surfaces at minimum.....	S16
Scheme S1. Free Energy and Enthalpy variation of the O-O bond homolytic clivage.....	S18
Scheme S2. Resonance structures and their weights obtained from the NRT analysis for $\text{HO}\bullet$, $\text{FO}\bullet$, $\text{ClO}\bullet$ and $\text{BrO}\bullet$ radicals	S19
Scheme S3. Main resonance structures and their weights obtained from the NRT analysis for analysis for $\text{H}_3\text{CO}\bullet$, $(\text{CH}_3)_3\text{CO}\bullet$ and $\text{F}_3\text{CO}\bullet$ radicals	S20
Scheme S10. Hyperconjugative interactions for the radicals	S21
Figure S6. NBO surfaces of the main hyperconjugative interactions for the radicals	S22
Figure S7. Total $\Delta E(T)$, non-Lewis $\Delta E(NL)$, Lewis $\Delta E(L)$, Natural Coulomb Electrostatic $\Delta E(NCE)$ and Natural steric analysis $\Delta E(STERIC)$ potential energy curves for HOOH at the HF/aug-cc-pVTZ//M06-2X/aug-cc-pVTZ level.....	S23
Figure S8. Number of remaining hyperconjugative interactions and the Amplitude of the “Others” curve in the function of the cutoff value	S24

Python Code for the individual natural coulombic potential energy curves calculation	S25
Cartesian Coordinates.....	S26

Computational Details

All potential energy curves (PEC) were optimized at the M06-2X/aug-cc-pVTZ level using ORCA 4.0.1^{1,2} by scanning the H-O-O-R dihedral angle from 0 to 180° in steps of 10°. For the peroxide HOOC(CH₃)₃, the fixed O1-O3-C4-C5 dihedral angle (Figure S1f) at 180° was also included throughout the scan.

The Natural Bond Orbitals analysis were carried out using Gaussian09 Rev D.01³ and NBO6.0⁴ at the M06-2X/aug-cc-pVTZ level by single point calculations over the geometries of the optimized potential energy curves. The curves of NBO Energetic Analysis were calculated by the deletion of all non-Lewis orbitals.

Individual natural coulombic interactions (Figure S3) for each point of the PEC were calculated from the interatomic distances (R_{AB}) and the total atomic charges (Q_A and Q_B) given by the Natural Coulomb Electrostatics (NCE) analysis using the Equation S1 and a Python code (page S25). The sum of all calculated interactions was compared with the NCE Potential Energy of each structure to verify the validity of the numerical values.

$$E_{AB} = \frac{Q_A Q_B}{R_{AB}} \quad (\text{S1})$$

The individual hyperconjugative interactions (Figure S4) were analyzed using the data from the NBO Energetic Analysis output (Second Order Perturbation Theory Analysis List) setting the threshold for printing equal to 0 kcal mol⁻¹ and excluding all interactions involving Rydberg (RY) and Core (CR) orbitals. To decrease the number of interactions shown in the graphs and make the qualitative analysis easily, some interactions were added into a new curve called “Others”. All interactions with the amplitude (the difference between the maximum and the minimum values of energy) lower than a cutoff value were chosen to “Others” curve and, a Python code was used to calculate the number of the remaining interactions and the amplitude of the “Others” curve in the function of the cutoff value (Figure S8) to set the best cutoff value for each molecule.

The individual steric interactions (Tables S5 and S6) were analyzed using the data from the NBO Natural Steric Analysis output setting the threshold for printing equal to 0 kcal mol⁻¹.

¹ F. Neese, *Wiley Interdiscip. Rev. Comput. Mol. Sci.*, 2012, **2**, 73–78.

² F. Neese, *Wiley Interdiscip. Rev. Comput. Mol. Sci.*, 2018, **8**, e1327.

³ Gaussian 09, Revision D.01, M. J. Frisch, G. W. Trucks, H. B. Schlegel, G. E. Scuseria, M. A. Robb, J. R. Cheeseman, G. Scalmani, V. Barone, B. Mennucci, G. A. Petersson, H. Nakatsuji, M. Caricato, X. Li, H. P. Hratchian, A. F. Izmaylov, J. Bloino, G. Zheng, J. L. Sonnenberg, M. Hada, M. Ehara, K. Toyota, R. Fukuda, J. Hasegawa, M. Ishida, T. Nakajima, Y. Honda, O. Kitao, H. Nakai, T. Vreven, J. A. Montgomery, Jr., J. E. Peralta, F. Ogliaro, M. Bearpark, J. J. Heyd, E. Brothers, K. N. Kudin, V. N. Staroverov, T. Keith, R. Kobayashi, J. Normand, K. Raghavachari, A. Rendell, J. C. Burant, S. S. Iyengar, J. Tomasi, M. Cossi, N. Rega, J. M. Millam, M. Klene, J. E. Knox, J. B. Cross, V. Bakken, C. Adamo, J. Jaramillo, R. Gomperts, R. E. Stratmann, O. Yazyev, A. J. Austin, R. Cammi, C. Pomelli, J. W. Ochterski, R. L. Martin, K. Morokuma, V. G. Zakrzewski, G. A. Voth, P. Salvador, J. J. Dannenberg, S. Dapprich, A. D. Daniels, O. Farkas, J. B. Foresman, J. V. Ortiz, J. Cioslowski, and D. J. Fox, Gaussian, Inc., Wallingford CT, 2013.

⁴ F. Weinhold, C. R. Landis and E. D. Glendening, *Int. Rev. Phys. Chem.*, 2016, **35**, 399–440.

Table S1. Basis set mean absolute error (MAE) (in kcal mol⁻¹) obtained from single point calculations for the M06-2X functional on MP2/cc-pVTZ calculated geometries and using the DLPNO-CCSD(T)/aug-cc-pVTZ//MP2/cc-pVTZ level as reference for compounds **1-7**.

Basis Set	HOOH	HOOF	HOOCI	HOObR	HOOCH ₃	HOOC(CH ₃) ₃	HOOCF ₃	Average
cc-pVDZ	0.303	0.407	0.527	0.767	0.112	0.115	0.113	0.335
cc-pVTZ	0.062	0.361	0.132	0.318	0.123	0.166	0.130	0.185
cc-pVQZ	0.027	0.327	0.080	0.255	0.193	0.226	0.151	0.180
cc-pV5Z	0.035	0.296	0.045	0.224	0.172	0.207	0.144	0.160
aug-cc-pVDZ	0.097	0.079	0.480	0.658	0.122	0.163	0.119	0.245
aug-cc-pVTZ	0.027	0.266	0.073	0.215	0.153	0.175	0.148	0.151

Table S2. Methods mean absolute error (MAE) (in kcal mol⁻¹) obtained with the aug-cc-pVTZ basis set on MP2/cc-pVTZ calculated geometries and using the DLPNO-CCSD(T)/aug-cc-pVTZ//MP2/cc-pVTZ level as reference for compounds **1-7**.

Method	HOOH	HOOF	HOOCI	HOObR	HOOCH ₃	HOOC(CH ₃) ₃	HOOCF ₃	Average
B3LYP	0.067	0.567	0.395	0.552	0.037	0.055	0.126	0.257
B3LYP-D3(BJ)	0.067	0.568	0.396	0.553	0.046	0.092	0.127	0.264
HF	0.107	1.035	0.151	0.263	0.232	0.310	0.309	0.344
M06	0.073	0.655	0.497	0.481	0.177	0.216	0.126	0.318
M062X	0.027	0.266	0.073	0.215	0.153	0.175	0.148	0.151
MP2	0.065	0.118	0.129	0.151	0.074	0.081	0.059	0.097
PBE0	0.040	0.672	0.476	0.633	0.154	0.189	0.110	0.325

Table S3. Mean absolute error (MAE) obtained for the potential energy curves optimized at the M06-2X/aug-cc-pVTZ level and using the DLPNO-CCSD(T)/aug-cc-pVTZ level single point calculations as reference for compounds **1-7**.

Peroxide	MAE (kcal mol ⁻¹)
HOOH (1)	0.0222
HOOF (2)	0.2188
HOOCI (3)	0.0501
HOObR (4)	0.1169
HOOCCH ₃ (5)	0.1175
HOOC(CH ₃) ₃ (6)	0.1175
HOOCF ₃ (7)	0.1398

Table S4. Global minimum dihedral angle (in degrees) obtained from a potential energy curve built from a 0-180° scan in steps of 10° of the H-O-O-R dihedral angle for peroxides **1-7**. The *cis* (H-O-O-R dihedral angle of 0°) and *trans* (H-O-O-R dihedral angle of 180°) barrier relative energies (ΔE) in comparison to the obtained minimum geometry were calculated at the DLPNO-CCSD(T)/aug-cc-pVTZ//M06-2X/aug-cc-pVTZ level and are given in kcal mol⁻¹.

Peroxide/Disulfide	Min. dihedral (degrees)	ΔE_{cis}	ΔE_{trans}
HOOH (1)	110	7.22	1.21
HOOF (2)	80	8.00	8.69
HOOCI (3)	90	5.48	3.94
HOObR (4)	90	5.60	3.34
HOOCCH ₃ (5)	120	5.59	0.29
HOOC(CH ₃) ₃ (6)	120	7.03	0.70
HOOCF ₃ (7)	100	5.85	1.97

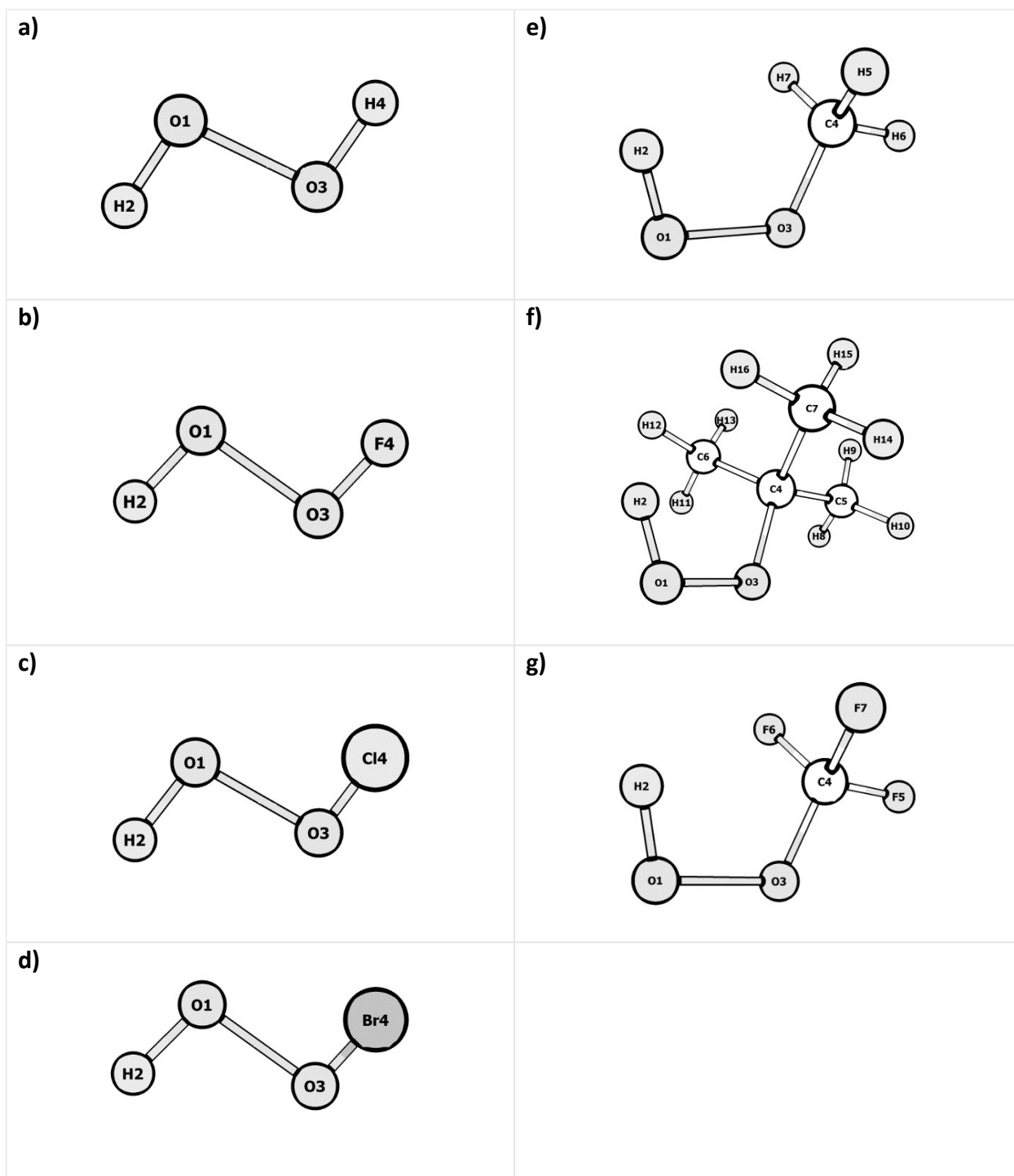


Figure S1. Atom numbering used for a) HOOH, b) HOOF c) HOOCI d) HOOBr e) HOOCH₃ f) HOOC(CH₃)₃ and g) HOO CF₃ molecules.

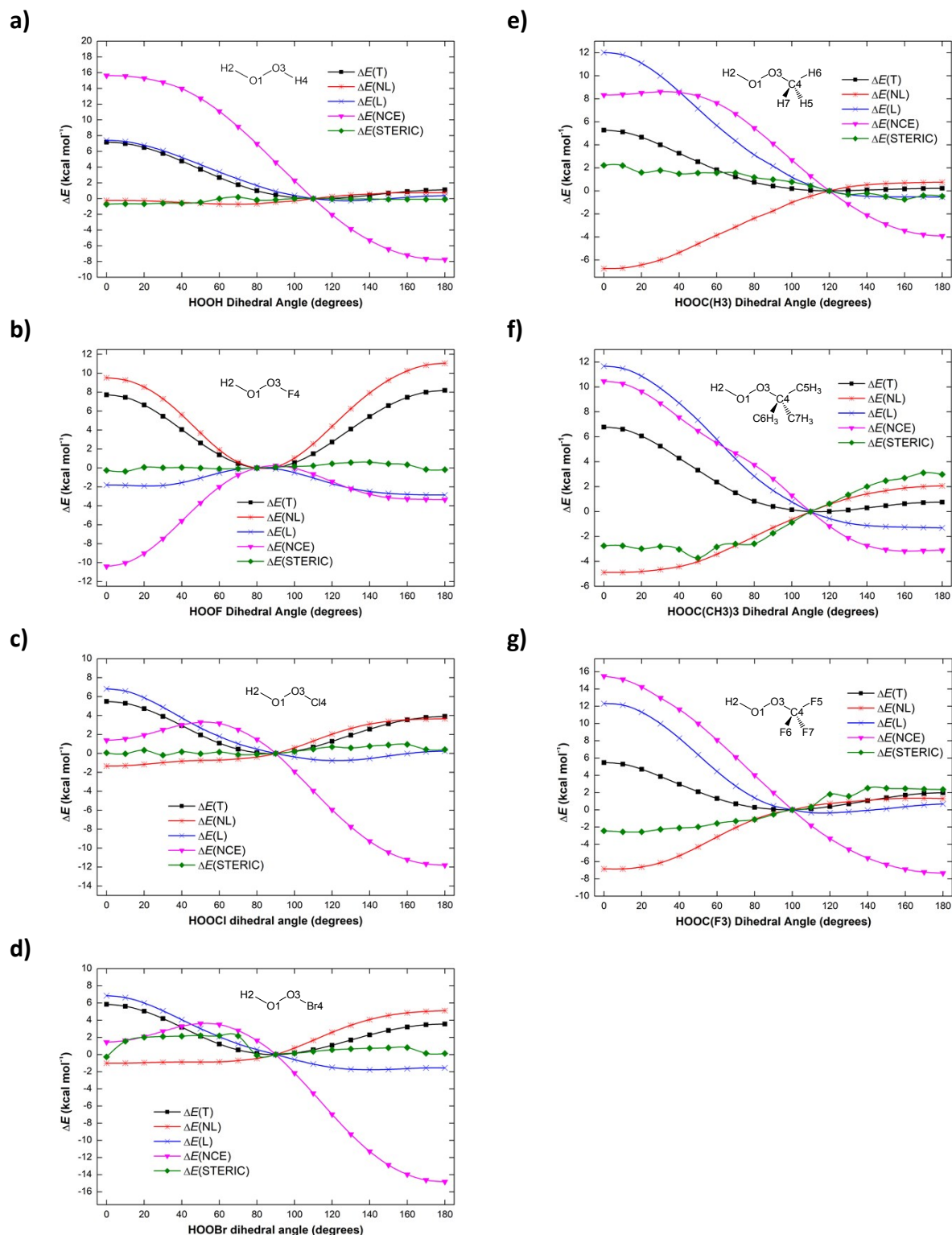


Figure S2. Total $\Delta E(T)$, non-Lewis $\Delta E(NL)$, Lewis $\Delta E(L)$, Natural Coulomb Electrostatic $\Delta E(NCE)$ and Natural steric analysis $\Delta E(STERIC)$ potential energy curves (kcal mol⁻¹) for **a)** HOOH, **b)** HOOF, **c)** HOOCI, **d)** HOOBr, **e)** HOOCH₃, **f)** HOOCH₃CH₃ and **g)** HOOCH₃F₃ obtained at the M06-2X/aug-cc-pVTZ level from the NBO analysis.

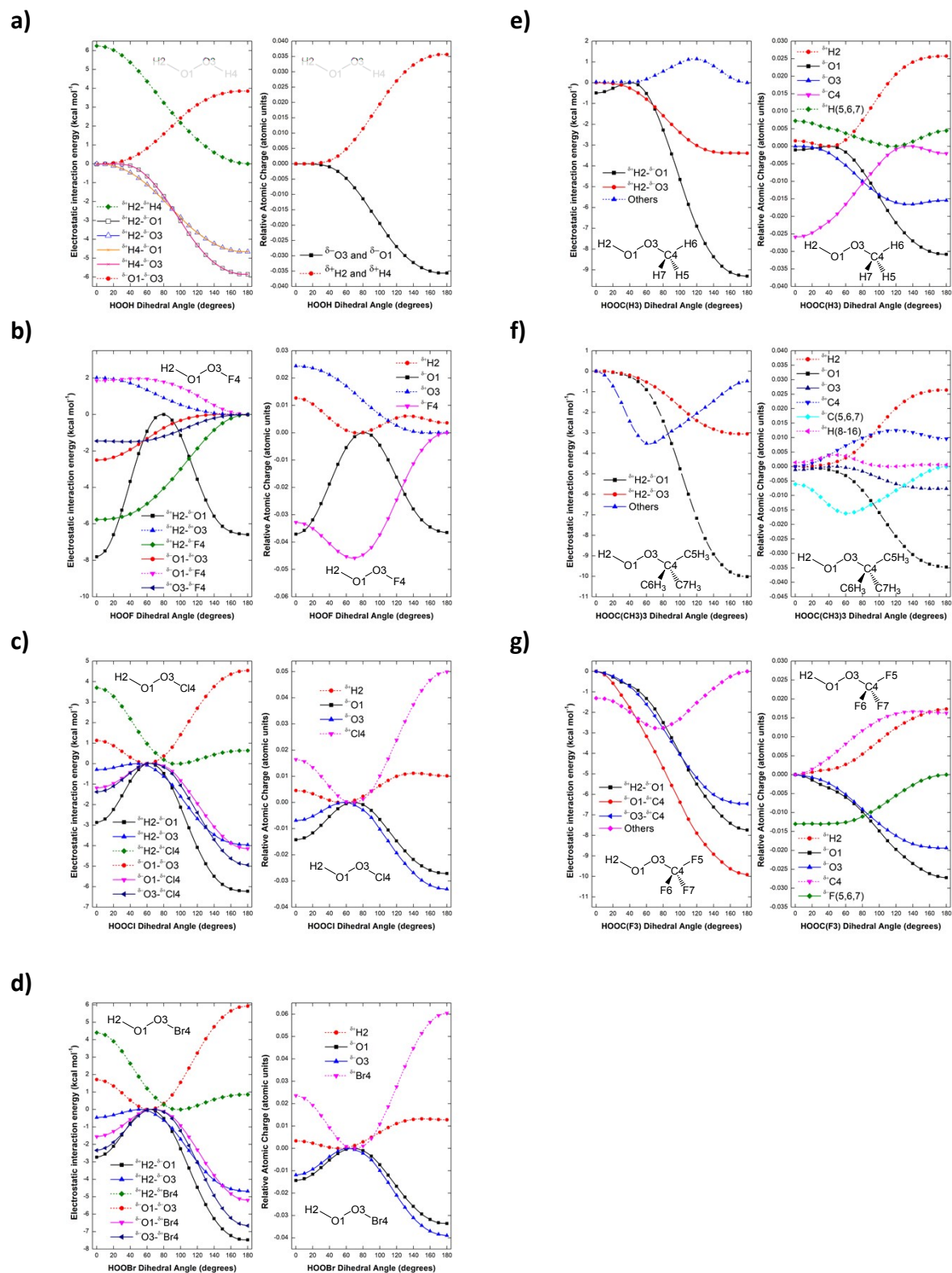


Figure S3. Individual natural coulombic potential energy curves (kcal mol⁻¹, left) and atomic charges (atomic units, right) for **a)** HOOH, **b)** HOOH, **c)** HOOCl, **d)** HOOBr, **e)** HOOCH₃, **f)** HOOCH₃ and **g)** HOOCF₃ obtained at the M06-2X/aug-cc-pVTZ level from the NBO analysis.

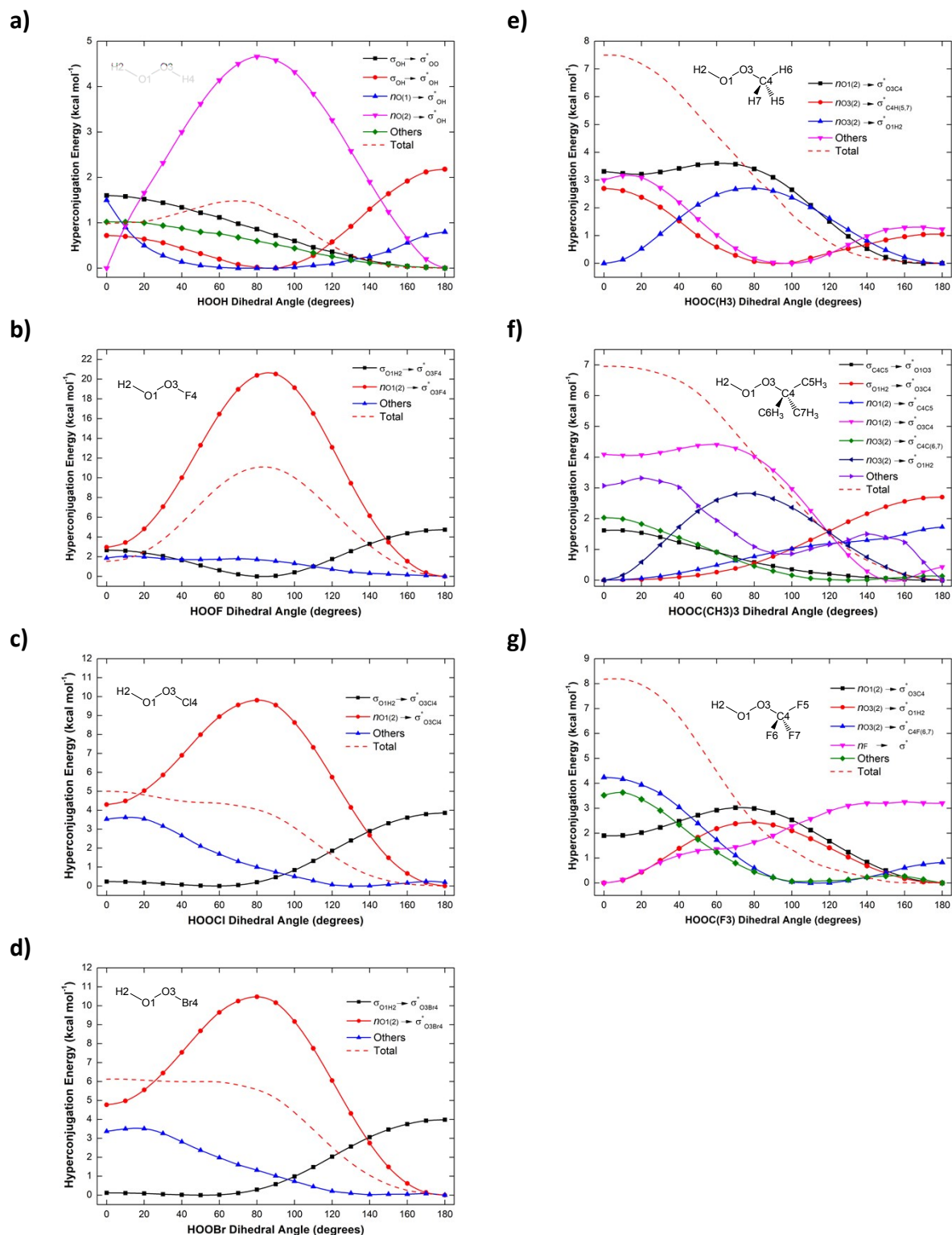


Figure S4. Hyperconjugation interaction energies (kcal mol⁻¹) for **a)** HOOH, **b)** HOOF, **c)** HOOCI, **d)** HOObR, **e)** HOOCH₃, **f)** HOOCH₃CH₃ and **g)** HOOCHF₃ obtained at the M06-2X/aug-cc-pVTZ level from the NBO analysis.

Table S5. Steric interactions energies for $\text{HOOC}(\text{CH}_3)_3$ at *cis* and *trans* geometries obtained at the M06-2X/aug-cc-pVTZ level from the NBO analysis.

Interactions	E_{cis} (kcal mol ⁻¹)	E_{trans} (kcal mol ⁻¹)	ΔE (kcal mol ⁻¹)
$n_{O1} - (\sigma_{OC}, \sigma_{CC}, \sigma_{CH})$	1.82	16.39	14.57
$\sigma_{O1H2} - (\sigma_{OC}, \sigma_{CC}, \sigma_{CH})$	10.53	3.09	-7.44
Others	140.74	139.44	-1.31
Total	153.10	158.92	5.82

Table S6. Steric interactions energies for HOOCF_3 at *cis* and *trans* geometries obtained at the M06-2X/aug-cc-pVTZ level from the NBO analysis.

Interactions	E_{cis} (kcal mol ⁻¹)	E_{trans} (kcal mol ⁻¹)	ΔE (kcal mol ⁻¹)
$n_{O1} - (\sigma_{CO}, \sigma_{CF}, n_F)$	2.68	9.70	7.02
$\sigma_{O1H2} - (\sigma_{CO}, \sigma_{CF}, n_F)$	9.06	5.01	-4.05
Others	136.90	138.72	1.82
Total	148.64	153.43	4.79

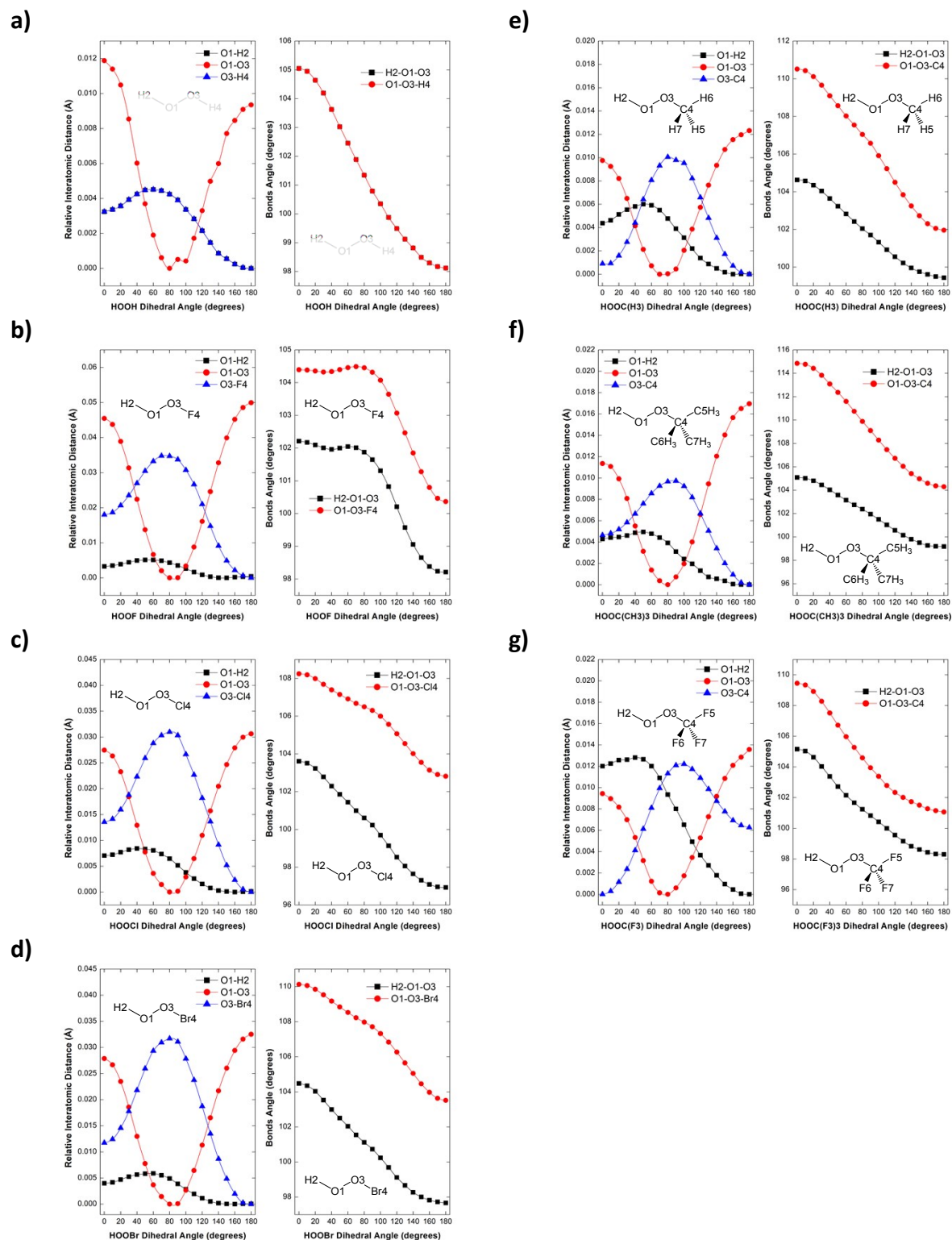
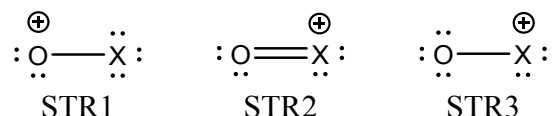


Figure S5. Relative Interatomic Distances and Bonds Angles for **a)** HOOH, **b)** HOOF, **c)** HOOCI, **d)** HOObR, **e)** HOOCH₃, **f)** HOOCH(CH₃)₃ and **g)** HOOCHF₃ obtained at the M06-2X/aug-cc-pVTZ level.

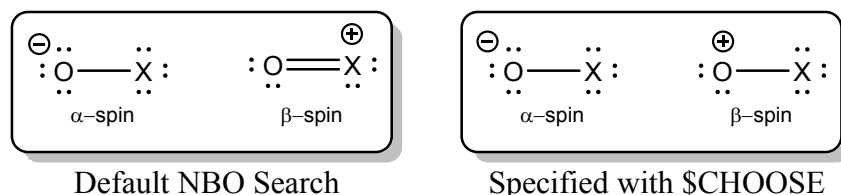
For the stability analysis of the peroxides, the molecules and the respective radical fragments were optimized at the M06-2X/aug-cc-pVTZ and UM06-2X/aug-cc-pVTZ levels without any constraints using the software Gaussian 09. Frequencies calculations were carried out to evaluate the Gibbs and Enthalpy energies. The Natural Bond Orbitals calculations were carried out using the NBO6.0 program and the NBO open-shell analysis.⁵ All the Lewis structures of the α and β spin sets are represented as analogues closed-shell structures.⁶

First, we carried out a Natural Resonance Theory (NRT)⁷ analysis to evaluate the resonance structures of the radical fragments and their respective weights. For the FO•, ClO• and BrO• radicals we specified the following resonance structures for the β -spin set using the \$NTRSTRB option:



The NRT results are shown in the Schemes S2 and S3 and, for the FO•, ClO• and BrO• radicals, the STR2 structure (with an O-X double bond) is the main resonance structure and with weights greater than 99.6%. That is a result of a radical stabilization via highly favourable hyperconjugative interactions between the lone pairs of the halogens atoms and the partially occupied O p orbital.

Finally, to allow the analysis and comparison of the $(\sigma, n) \rightarrow n_{\text{O}}$ hyperconjugative interactions in all radical fragments studied in this work, we specified the following structures for the radicals FO•, ClO• and BrO• using the \$CHOOSE option and we carried out a DEL analysis.



⁵ J. E. Carpenter and F. Weinhold, *J. Mol. Struct. THEOCHEM*, 1988, **169**, 41–62.

⁶ F. Weinhold, C. R. Landis and E. D. Glendening, *Int. Rev. Phys. Chem.*, 2016, **35**, 399–440.

⁷ a) E. D. Glendening and F. Weinhold, *J. Comput. Chem.*, 1998, **19**, 593–609 b) E. D. Glendening and F. Weinhold, *J. Comput. Chem.*, 1998, **19**, 610–627. c) E. D. Glendening, J. K. Badenhoop and F. Weinhold, *J. Comput. Chem.*, 1998, **19**, 628–646.

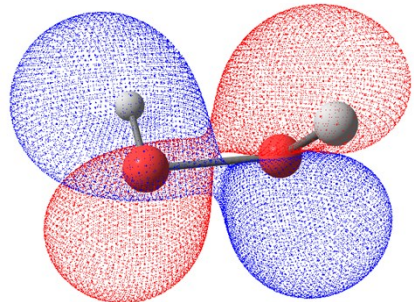
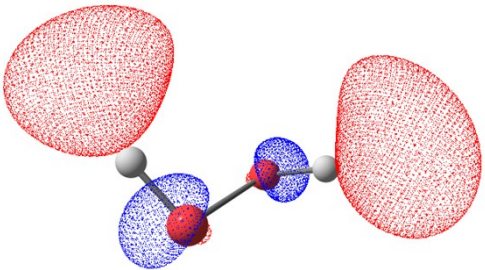
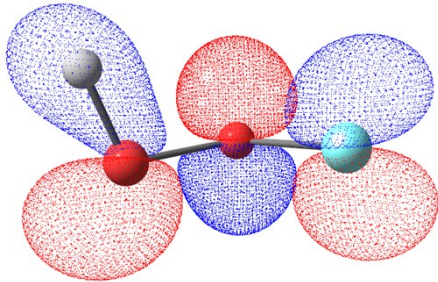
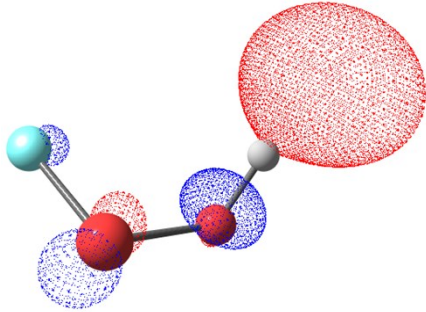
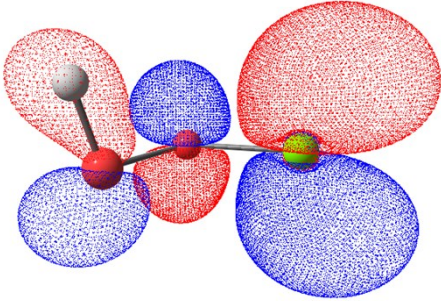
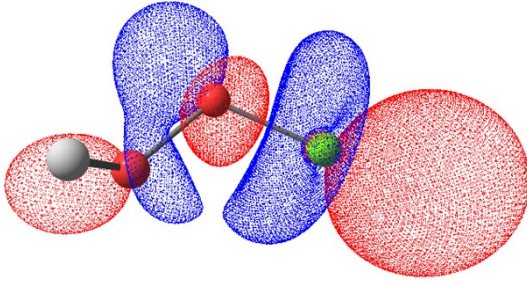
Table S7. Bond lengths, bonds angles and dihedral angle for molecules 1-7 obtained from the minimization at the M06-2X/aug-cc-pVTZ level using the software Gaussian 09.

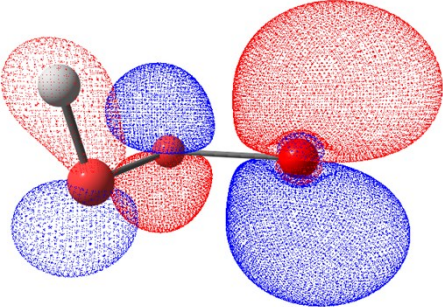
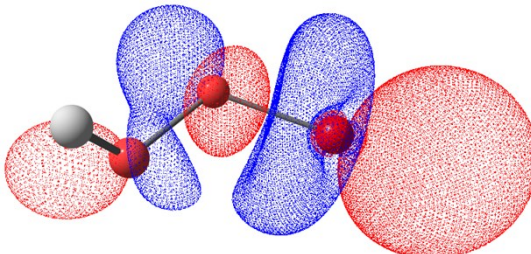
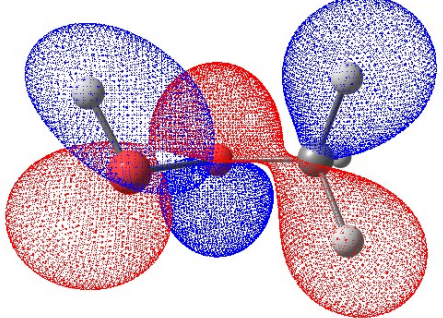
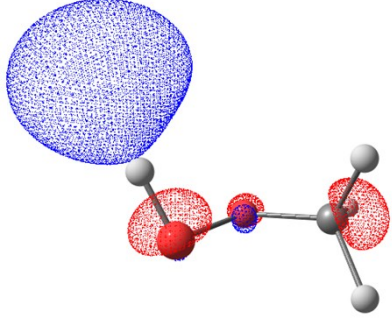
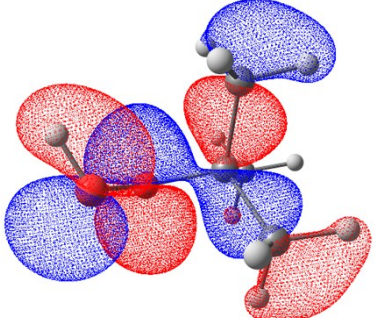
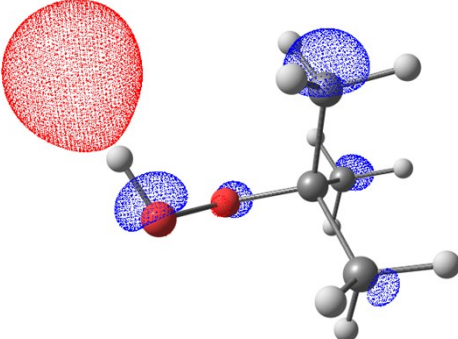
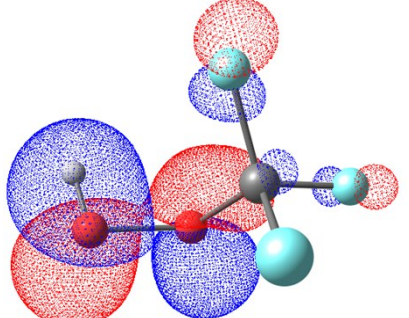
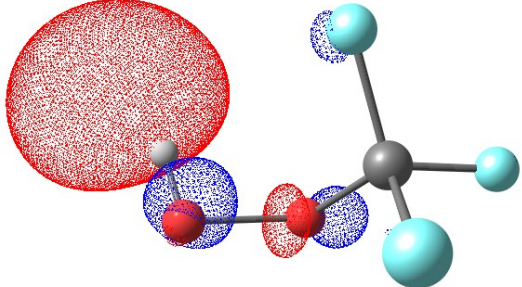
Molecule	Bond Length (Å)			Bonds Angle (degrees)		Dihedral Angle (degrees)
	O-O	H-O	O-R	H-O-O	O-O-R	
HOOH (1)	1.423	0.964	0.964	101.479	101.479	112.780
HOOF (2)	1.361	0.970	1.415	103.589	105.392	84.654
HOOCI (3)	1.399	0.968	1.693	102.626	109.309	88.211
HOObR (4)	1.400	0.968	1.837	102.629	110.110	90.433
HOOCCH ₃ (5)	1.423	0.965	1.410	101.567	106.564	112.919
HOOC(CH ₃) ₃ (6)	1.422	0.964	1.438	101.545	109.286	108.829
HOOCF ₃ (7)	1.414	0.967	1.368	101.785	107.257	98.353

Table S8. HOMO and LUMO energies for molecules **1-7** obtained from the minimization at the M06-2X/aug-cc-pVTZ level.

Molecule	HOMO (Eh)	LUMO (Eh)	HOMO-LUMO Gap (eV)
HOOH (1)	-0.3742	-0.0090	9.94
HOOF (2)	-0.4112	-0.0128	10.84
HOOCI (3)	-0.3656	-0.0413	8.83
HOObR (4)	-0.3483	-0.0636	7.75
HOCH ₃ (5)	-0.3502	-0.0069	9.34
HOOC(CH ₃) ₃ (6)	-0.3424	-0.0080	9.10
HOOCF ₃ (7)	-0.4319	-0.0144	11.36

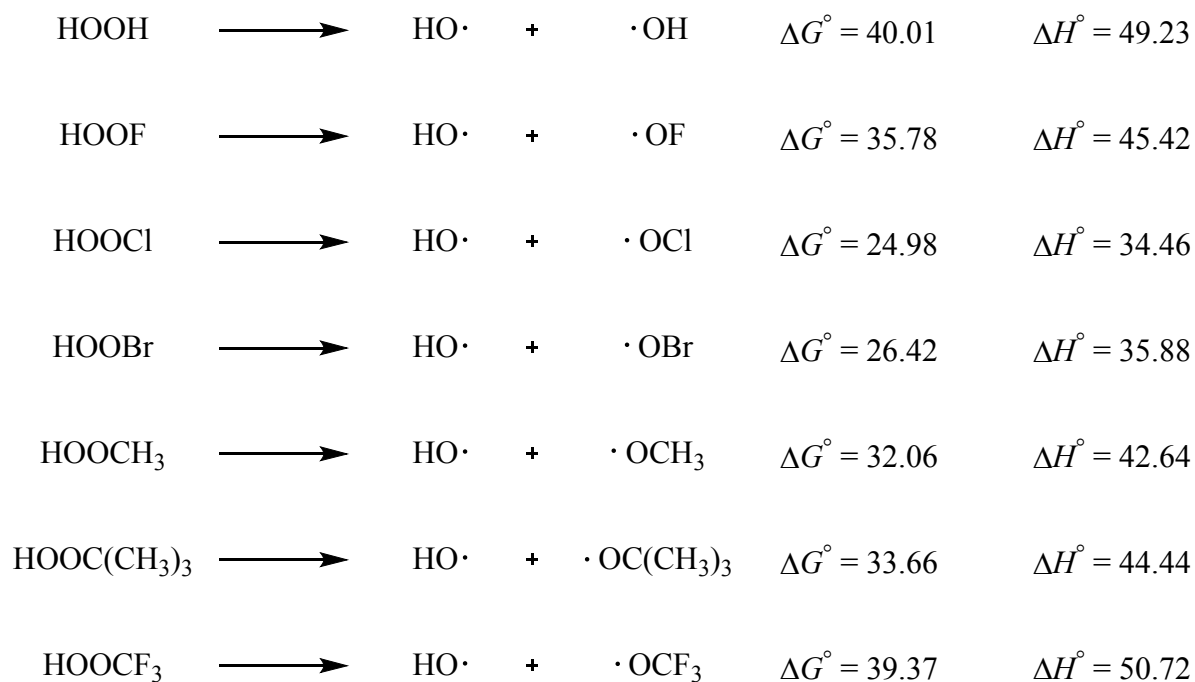
Table S9. HOMO and LUMO surfaces^a for molecules **1-7** obtained from the minimization at the M06-2X/aug-cc-pVTZ level.

Molecule	HOMO	LUMO
HOOH		
HOOF		
HOOCI		

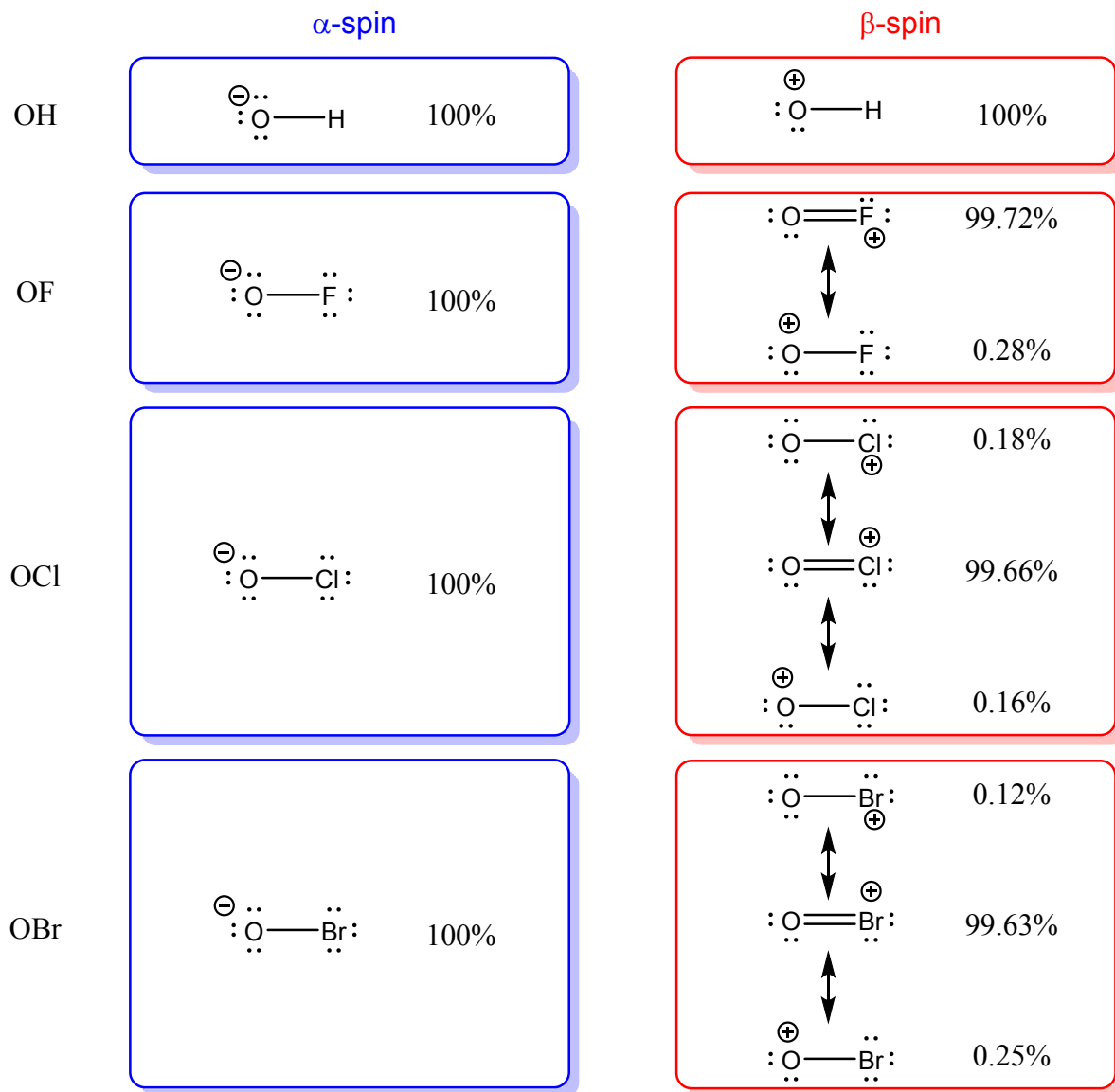
HOObBr		
HOOCCH ₃		
HOOC(CH ₃) ₃		
HOOCF ₃		

^a The molecular orbitals surfaces were rendered using the software Chemcraft 1.8 (Build 572b)⁸ with a cube density of 20 map points per angstrom and the following contour values: 0.03 (HOOC(CH₃)₃), 0.036 (HOOH, HOOCI, HOObBr and HOOCCH₃), 0.0432 (HOOCF₃) and 0.051840 (HOOF).

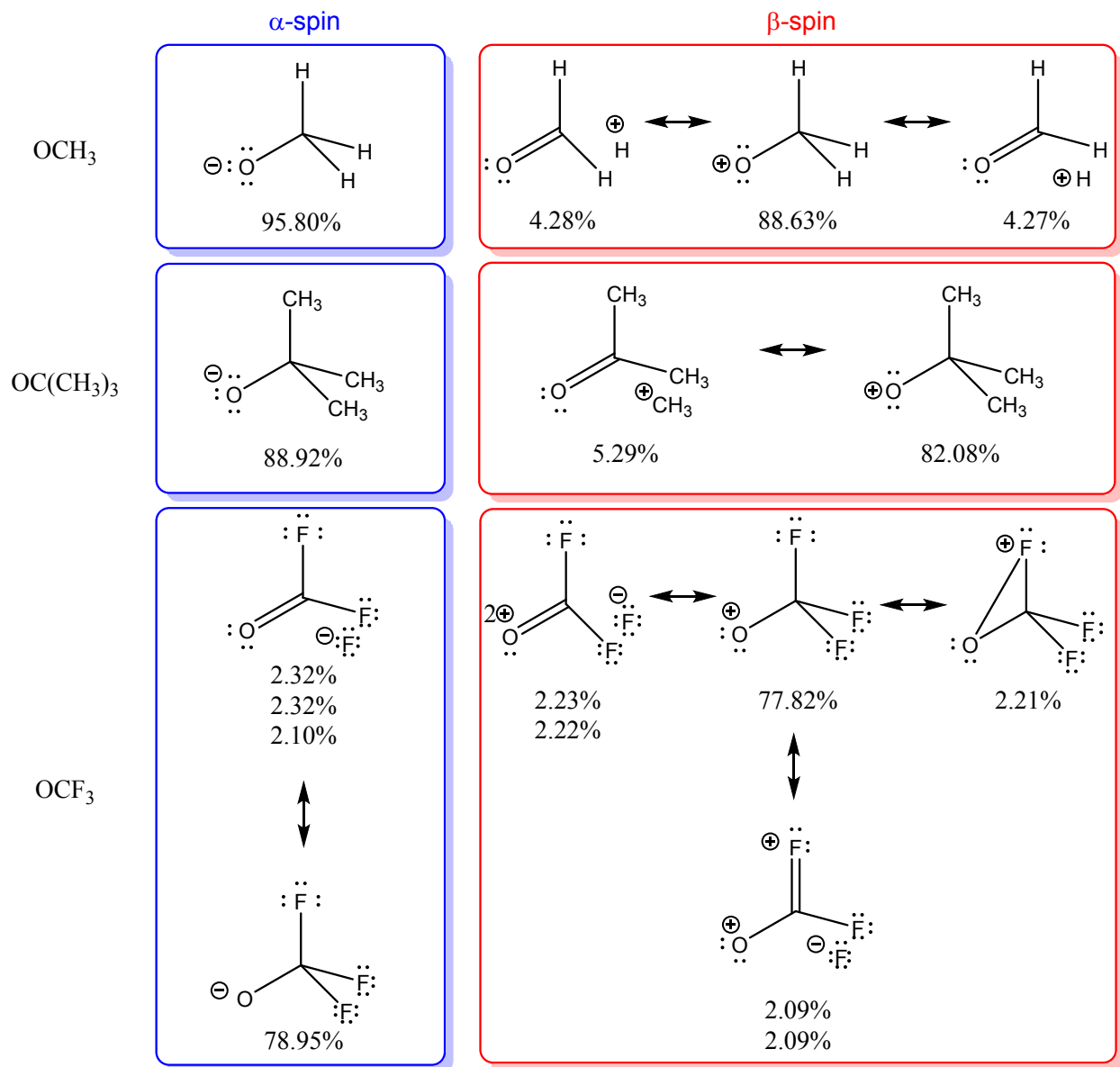
⁸ G. A. Zhurko, Chemcraft - graphical program for visualization of quantum chemistry computations. Ivanovo, Russia, 2005. <https://chemcraftprog.com>



Scheme S1. Free Energy and Enthalpy variation of the O-O bond homolytic cleavage for molecules 1-7 at the M06-2X/aug-cc-pVTZ level using the software Gaussian 09.



Scheme S2. Resonance structures and their weights obtained from the NTR analysis for HO•, FO•, ClO• and BrO• radical fragments. The structures of the spin sets are represented as analogues closed-shell structures.



Scheme S3. Main resonance structures (>2%) and their weights obtained from the NTR analysis for $\text{H}_3\text{CO}\bullet$, $(\text{CH}_3)_3\text{CO}\bullet$ and $\text{F}_3\text{CO}\bullet$ radicals. The structures of the spin sets are represented as analogues closed-shell structures.

Table S10. Hyperconjugative interactions energies between bonds (σ) and lone pairs (n) and the O partially empty p orbital (n_o) of the β -spin set for HO•, FO•, ClO•, BrO•, H₃CO•, (CH₃)₃CO• and F₃CO• radicals obtained from the minimization at the M06-2X/aug-cc-pVTZ level using the software Gaussian 09.

Radical	Hyperconjugative Interactions (kcal mol ⁻¹)		
	$\sigma \rightarrow n_o$	$n \rightarrow n_o$	Total
OH	0.00	0.00	0.00
OF	0.00	48.23	48.23
OCI	0.00	119.09	119.09
OBBr	0.00	104.93	104.93
OCH ₃	20.73	0.00	20.73
OC(CH ₃) ₃	18.34	0.00	18.34
OCF ₃	2.80	3.84	6.64

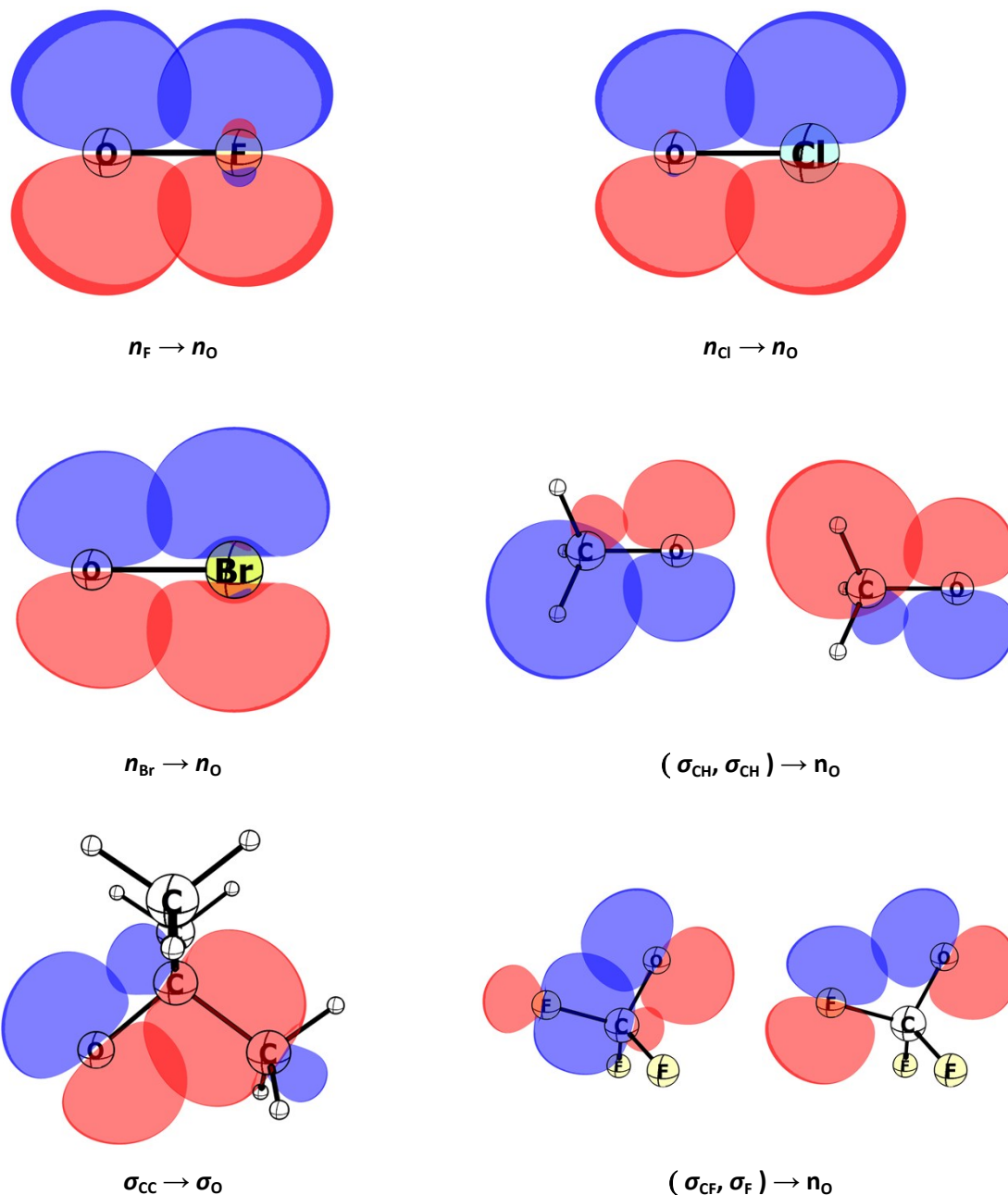


Figure S6. NBO surfaces of the main hyperconjugative interactions between bonds (σ) and lone pairs (n) and the O partially empty p orbital (n_{O}) of the β -spin set for FO•, ClO•, BrO•, H₃CO•, (CH₃)₃CO• and F₃CO• radicals obtained from the minimization at the M06-2X/aug-cc-pVTZ level using the software Gaussian 09. The molecular orbitals surfaces were rendered using the software Chemcraft 1.8 (Build 572b) with a cube density of 40 map points per angstrom and the following contour values: 0.06 (FO•, ClO•, BrO• and H₃CO•,) and 0.05 ((CH₃)₃CO• and F₃CO•).

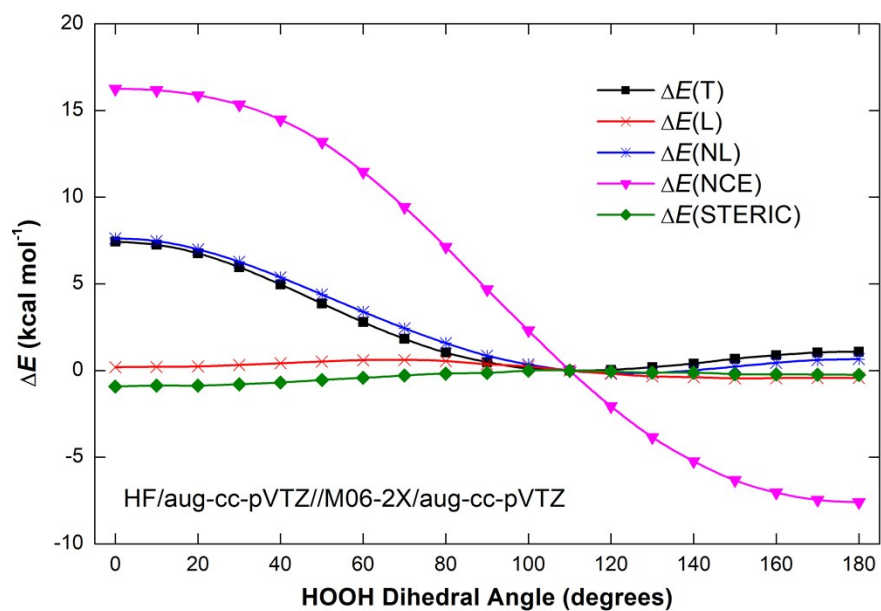


Figure S7. Total $\Delta E(T)$, non-Lewis $\Delta E(NL)$, Lewis $\Delta E(L)$, Natural Coulomb Electrostatic $\Delta E(NCE)$ and Natural steric analysis $\Delta E(STERIC)$ potential energy curves (kcal mol⁻¹) for HOOH obtained at the HF/aug-cc-pVTZ//M06-2X/aug-cc-pVTZ level from the NBO analysis

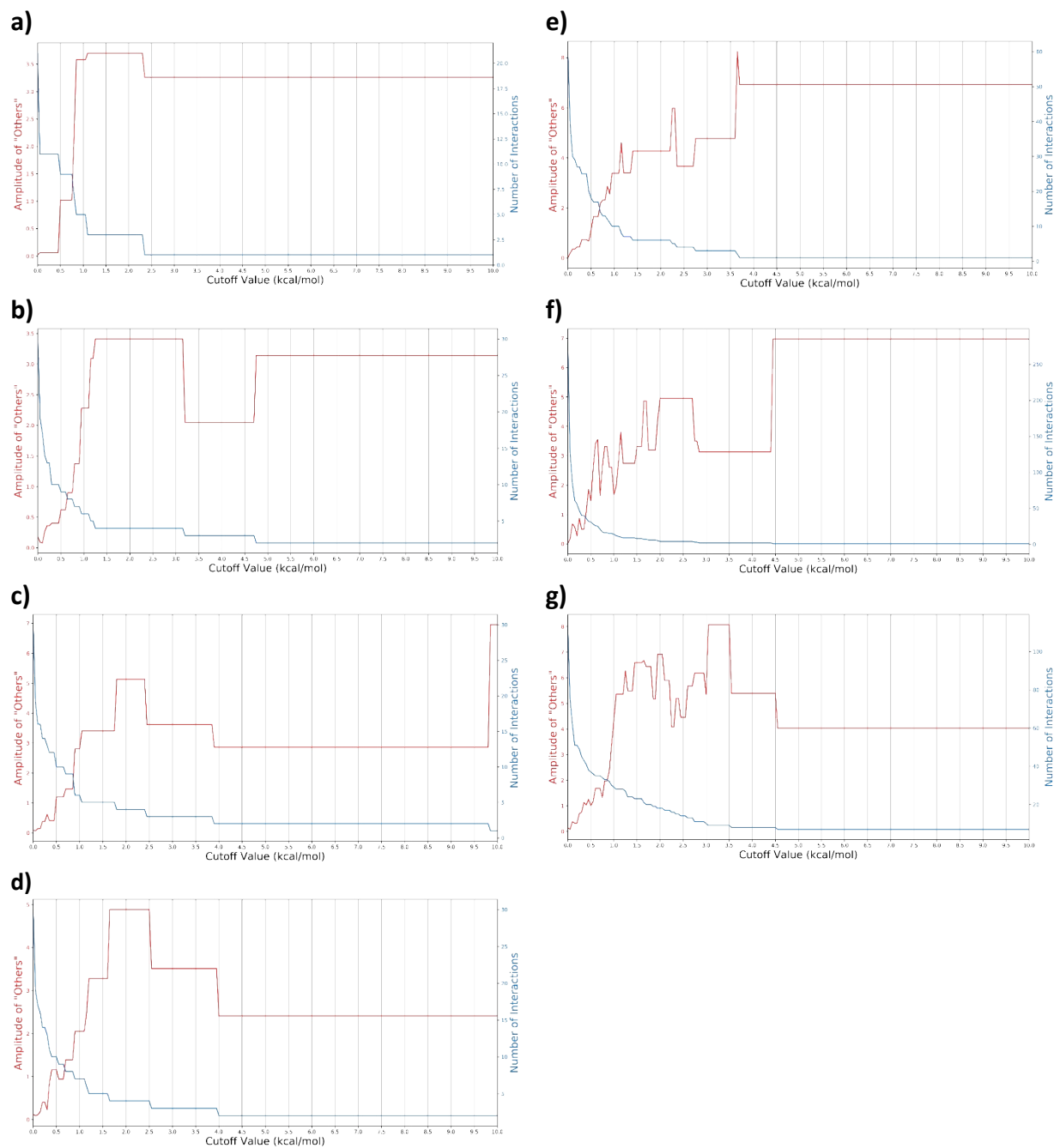


Figure S8. Number of remaining hyperconjugative interactions and the amplitude of the "Others" curve in the function of the cutoff value for **a)** HOOH, **b)** HOOF, **c)** HOOCI, **d)** HOObR, **e)** HOOCH₃, **f)** HOOC(CH₃)₃ and **g)** HOOCF₃ molecules.


```

import pandas as pd
import numpy as np
from itertools import combinations

input_data = pd.read_csv('HOOH_input.csv')

molecule_name = 'HOOH'
atoms_list = [ 'O1', 'H2', 'O3', 'H4' ]

interactions_list = list(combinations(atoms_list, 2))

interactions = np.zeros((19,len(interactions_list)))
distances = np.zeros((19,len(interactions_list)))

for n in range(0,len(interactions_list)):

    for i in range(0,19):
        a1 = interactions_list[n][0]
        a2 = interactions_list[n][1]
        dihedral = i * 10

        q1 = float(input_data.loc[input_data.Atom == a1].loc[input_data.Dihedral == dihedral]['Total'])
        q2 = float(input_data.loc[input_data.Atom == a2].loc[input_data.Dihedral == dihedral]['Total'])
        x1 = float(input_data.loc[input_data.Atom == a1].loc[input_data.Dihedral == dihedral]['X'])
        y1 = float(input_data.loc[input_data.Atom == a1].loc[input_data.Dihedral == dihedral]['Y'])
        z1 = float(input_data.loc[input_data.Atom == a1].loc[input_data.Dihedral == dihedral]['Z'])
        x2 = float(input_data.loc[input_data.Atom == a2].loc[input_data.Dihedral == dihedral]['X'])
        y2 = float(input_data.loc[input_data.Atom == a2].loc[input_data.Dihedral == dihedral]['Y'])
        z2 = float(input_data.loc[input_data.Atom == a2].loc[input_data.Dihedral == dihedral]['Z'])

        d = (((x1 - x2) ** 2) + ((y1 - y2) ** 2) + ((z1 - z2) ** 2)) ** (0.5)
        #interactions[i][n] = ((q1 * q2) / d) * 0.529177 #in a.u.
        interactions[i][n] = ((q1 * q2) / d) * 0.529177 * 627.509474 #in kcal mol-1
        distances[i][n] = d

charges = np.zeros((19,len(atoms_list)))

for n in range(0,len(atoms_list)):
    a1 = atoms_list[n]

    for i in range(0,19):
        dihedral = i * 10
        q = float(input_data.loc[input_data.Atom == a1].loc[input_data.Dihedral == dihedral]['Total'])
        charges[i][n] = q

np.savetxt(str(molecule_name + '_NCE_charges.csv'), charges, delimiter=",")
np.savetxt(str(molecule_name + '_NCE_interactions.csv'), interactions, delimiter=",")
np.savetxt(str(molecule_name + '_NCE_distances.csv'), distances, delimiter=",")
np.savetxt(str(molecule_name + '_NCE_interactions_list.txt'), interactions_list, fmt='%s', delimiter='-')

```

In the input file (csv format), the atoms are added in the lines with their total charge, cartesian coordinates and the dihedral angle of the corresponding geometry. As an example, the seven first lines for the HOOH input file:

Atom	Total	X	Y	Z	Dihedral
O1	-0.44844	0.000000	-0.718817	-0.103953	0
H2	0.44844	0.000000	-0.970354	0.831627	0
O3	-0.44844	0.000000	0.718817	-0.103953	0
H4	0.44844	0.000000	0.970354	0.831627	0
O1	-0.44841	-0.718486	-0.103618	0.011213	10
H2	0.44841	-0.969663	0.828946	-0.066466	10

Cartesians coordinates for molecules 1-7 in their *cis*, *trans* and global minimum geometries. Total energies (in atomic units) were obtained at the M06-2X/aug-cc-pVTZ level.

Molecule	<i>cis</i>	Minimum	<i>trans</i>
HOOH	Total Energy (M06-2X/aug-cc-pVTZ) = -151.541973614	Total Energy (M06-2X/aug-cc-pVTZ) = -151.553414001	Total Energy (M06-2X/aug-cc-pVTZ) = -151.551627385
	O 0.000000000 -0.718817000 -0.103953000	O -0.704974000 -0.111497000 -0.060801000	O 0.000000000 -0.717548000 0.000000000
	H 0.000000000 -0.970354000 0.831627000	H -0.991085000 0.634471000 0.486414000	H 0.955890000 -0.853887000 0.000000000
	O 0.000000000 0.718817000 -0.103953000	O 0.704973000 0.111497000 -0.060802000	O 0.000000000 0.717548000 0.000000000
	H 0.000000000 0.970354000 0.831627000	H 0.991093000 -0.634470000 0.486414000	H -0.955890000 0.853887000 0.000000000
HOOF	Total Energy (M06-2X/aug-cc-pVTZ) = -250.678124723	Total Energy (M06-2X/aug-cc-pVTZ) = -250.690386023	Total Energy (M06-2X/aug-cc-pVTZ) = -250.677373944
	O 1.135576000 -0.188247000 0.000000000	O -1.086507000 -0.223563000 -0.122953000	O -1.024800000 -0.385689000 0.000000000
	H 0.753756000 -1.084980000 -0.000001000	H -1.192471000 -0.599610000 0.771093000	H -1.789931000 0.213446000 0.000000000
	O -0.018439000 0.628092000 0.000000000	O -0.004741000 0.597335000 0.043297000	O 0.000000000 0.594455000 0.000000000
	F -1.076761000 -0.270421000 0.000000000	F 1.102495000 -0.265618000 -0.014872000	F 1.109815000 -0.209285000 0.000000000
HOOCI	Total Energy (M06-2X/aug-cc-pVTZ) = -611.09127427	Total Energy (M06-2X/aug-cc-pVTZ) = -611.100034706	Total Energy (M06-2X/aug-cc-pVTZ) = -611.093795228
	O -1.512691000 -0.280983000 0.000000000	O -1.455431000 -0.325767000 -0.121974000	O 0.584183000 -1.33534000 0.000000000
	H -1.042452000 -1.134494000 -0.000001000	H -1.615046000 -0.548712000 0.811162000	H 0.139185000 -2.192682000 0.000000000
	O -0.456008000 0.686046000 0.000000000	O -0.470475000 0.665708000 0.023354000	O -0.601581000 -0.524397000 0.000000000
	Cl 0.987767000 -0.123883000 0.000000000	Cl 1.001312000 -0.127695000 -0.001306000	Cl 0.000000000 1.003302000 0.000000000
HOObR	Total Energy (M06-2X/aug-cc-pVTZ) = -2725.136689391	Total Energy (M06-2X/aug-cc-pVTZ) = -2725.146016601	Total Energy (M06-2X/aug-cc-pVTZ) = -2725.140348337
	O -0.467355000 -1.920353000 0.000000000	O -1.892028000 -0.359410000 -0.122301000	O 0.589884000 -1.774850000 0.000000000
	H -1.262731000 -1.360005000 0.000000000	H -2.044345000 -0.597973000 0.808429000	H 0.169045000 -2.647686000 0.000000000
	O 0.625197000 -0.991782000 0.000000000	O -0.960041000 0.683214000 0.023888000	O -0.611015000 -0.982938000 0.000000000
	Br 0.000000000 0.704488000 0.000000000	Br 0.710311000 -0.056927000 -0.000604000	Br 0.000000000 0.706000000 0.000000000
HOOCH ₃	Total Energy (M06-2X/aug-cc-pVTZ) = -190.843035043	Total Energy (M06-2X/aug-cc-pVTZ) = -190.851449485	Total Energy (M06-2X/aug-cc-pVTZ) = -190.851087032
	O 1.227598000 0.167610000 0.000000000	O -1.134185000 -0.287522000 -0.094695000	O 1.097425000 -0.342731000 -0.000086000
	H 0.916895000 1.087253000 -0.000001000	H -1.615607000 -0.043177000 0.708435000	H 1.854244000 0.258130000 0.000178000
	O 0.029209000 -0.619343000 0.000001000	O -0.029900000 0.617693000 -0.023554000	O 0.032121000 0.620553000 0.000316000
	C -1.115102000 0.218067000 0.000000000	C 1.117820000 -0.223262000 0.025274000	C -1.115421000 -0.210884000 -0.000051000
	H -1.160751000 0.843800000 -0.898629000	H 1.032683000 -0.991222000 -0.741902000	H -1.121346000 -0.827440000 0.899992000
	H -1.959240000 -0.469393000 -0.000001000	H 1.967933000 0.423730000 -0.179621000	H -1.972042000 0.457610000 -0.015105000
	H -1.160752000 0.843799000 0.898630000	H 1.220748000 -0.691122000 1.007432000	H -1.104691000 -0.845575000 -0.886600000
	Total Energy (M06-2X/aug-cc-pVTZ) = -308.77792021	Total Energy (M06-2X/aug-cc-pVTZ) = -308.788729081	Total Energy (M06-2X/aug-cc-pVTZ) = -308.787524337
HOOC(CH ₃) ₃	O 2.054510000 -0.183464000 -0.000711000	O 1.921604000 0.020396000 -0.137301000	O -1.888718000 -0.000337000 -0.087125000
	H 1.838292000 0.760746000 0.005607000	H 2.332480000 -0.820371000 -0.375771000	H -2.561091000 0.000096000 -0.778476000
	O 0.788027000 -0.856335000 -0.005620000	O 0.721115000 -0.046131000 -0.910266000	O -0.709686000 0.000139000 -0.913403000
	C -0.369521000 0.021492000 -0.000021000	C -0.386646000 0.020428000 0.030420000	C 0.388466000 -0.000456000 0.030597000
	C -1.476757000 -1.022596000 -0.007394000	C -1.608310000 -0.056230000 -0.871972000	C 1.627577000 0.000051000 -0.849372000
	H -1.381245000 -1.659160000 0.871085000	H -1.671709000 -1.030766000 -1.353084000	H 1.653677000 0.886367000 -1.482523000
	H -2.452439000 -0.540920000 0.001033000	H -2.506292000 0.096357000 -0.274237000	H 2.511195000 0.003866000 -0.211956000
	H -1.387560000 -1.640664000 -0.899630000	H -1.567298000 0.715829000 -1.640660000	H 1.658344000 -0.888566000 -1.479601000
	C -0.445704000 0.900575000 -1.249273000	C -0.267917000 1.334404000 0.782892000	C 0.308070000 -1.269014000 0.869659000
	H 0.235370000 1.753750000 -1.212020000	H 0.620222000 1.315442000 1.412601000	H -0.615947000 -1.290831000 1.442827000

	H -0.239600000 0.321416000 -2.146619000	H -0.169829000 2.152562000 0.070867000	H 0.333479000 -2.141546000 0.216314000
	H -1.454334000 1.311681000 -1.325648000	H -1.141081000 1.502119000 1.411235000	H 1.150900000 -1.318153000 1.558257000
	C -0.447202000 0.882341000 1.261349000	C -0.379783000 -1.180385000 0.973739000	C 0.307669000 1.269167000 0.869434000
	H -1.447791000 1.314785000 1.327199000	H -1.229624000 -1.104949000 1.654854000	H 1.141926000 1.312422000 1.568904000
	H -0.269075000 0.285401000 2.153003000	H -0.481774000 -2.110286000 0.413000000	H 0.347057000 2.141707000 0.217428000
	H 0.253187000 1.720485000 1.248673000	H 0.529095000 -1.219358000 1.571257000	H -0.623000000 1.297739000 1.431141000
HOOCF ₃	Total Energy (M06-2X/aug-cc-pVTZ) = -488.631258695	Total Energy (M06-2X/aug-cc-pVTZ) = -488.639985487	Total Energy (M06-2X/aug-cc-pVTZ) = -488.636820458
	O 1.622791000 -0.994487000 0.000000000	O 1.816256000 -0.004579000 0.117279000	O -1.781019000 -0.105126000 -0.000596000
	H 0.854989000 -1.597377000 0.000000000	H 2.102915000 -0.142984000 -0.799771000	H -2.519923000 0.514313000 0.002152000
	O 1.065457000 0.319809000 0.000000000	O 0.709416000 0.860883000 -0.087567000	O -0.712534000 0.847838000 0.004122000
	C -0.292773000 0.236045000 0.000000000	C -0.372334000 0.020231000 0.003454000	C 0.376223000 0.021110000 0.000477000
	F -0.763124000 1.444477000 0.000000000	F -1.445624000 0.757957000 -0.001691000	F 1.439324000 0.776072000 -0.007055000
	F -0.763124000 -0.412320000 -1.052224000	F -0.428135000 -0.807077000 -1.025276000	F 0.408346000 -0.751509000 1.059475000
	F -0.763124000 -0.412320000 1.052224000	F -0.356717000 -0.709639000 1.087117000	F 0.397998000 -0.755970000 -1.056111000

Cartesians coordinates for molecules 1-7 and total energy (in atomic units) from the minimization (Tables S7-S9) obtained at M06-2X/aug-cc-pVTZ level and using the software Gaussian 09:

HOOH	HOOF	HOOCI
Free Energy (M06-2X/aug-cc-pVTZ) = -151.548119 Enthalpy (M06-2X/aug-cc-pVTZ) = -151.522354	Free Energy (M06-2X/aug-cc-pVTZ) = -250.696462 Enthalpy (M06-2X/aug-cc-pVTZ) = -250.667645	Free Energy (M06-2X/aug-cc-pVTZ) = -611.108914 Enthalpy (M06-2X/aug-cc-pVTZ) = -611.078742
O 0.000000000000 0.711682000000 -0.058094000000	O -1.087168000000 -0.227688000000 -0.123721000000	O -1.491788000000 -0.308593000000 -0.122994000000
H 0.786655000000 0.903492000000 0.464752000000	H -1.263534000000 -0.551932000000 0.772771000000	H -1.661115000000 -0.585143000000 0.789205000000
O 0.000000000000 -0.711682000000 -0.058094000000	O -0.011817000000 0.590017000000 0.037869000000	O -0.480834000000 0.646066000000 0.029590000000
H -0.786655000000 -0.903492000000 0.464752000000	F 1.117269000000 -0.260744000000 -0.009551000000	Cl 1.026005000000 -0.124391000000 -0.002469000000
HOObR	HOOCCH₃	HOOC(CH₃)₃
Free Energy (M06-2X/aug-cc-pVTZ) = -2725.155996 Enthalpy (M06-2X/aug-cc-pVTZ) = -2725.124517	Free Energy (M06-2X/aug-cc-pVTZ) = -190.822027 Enthalpy (M06-2X/aug-cc-pVTZ) = -190.791404	Free Energy (M06-2X/aug-cc-pVTZ) = -308.681108 Enthalpy (M06-2X/aug-cc-pVTZ) = -308.641446
O -1.924889000000 -0.344578000000 -0.122415000000	O -1.148516000000 -0.267025000000 -0.102788000000	O -1.943512000000 -0.107227000000 -0.153233000000
H -2.106246000000 -0.600217000000 0.793150000000	H -1.573073000000 -0.118731000000 0.750475000000	H -2.314691000000 0.779625000000 -0.224091000000
O -0.967487000000 0.665545000000 0.025818000000	O -0.025364000000 0.601945000000 -0.017419000000	O -0.730462000000 0.032717000000 -0.882889000000
Br 0.721293000000 -0.056215000000 -0.000582000000	C 1.118140000000 -0.221810000000 0.020838000000	C 0.379807000000 0.000465000000 0.030703000000
	H 1.962964000000 0.465089000000 0.019969000000	C 1.571961000000 0.124229000000 -0.905128000000
	H 1.133687000000 -0.831783000000 0.926250000000	H 1.530252000000 1.063210000000 -1.456107000000
	H 1.158626000000 -0.863070000000 -0.860062000000	H 2.495009000000 0.097808000000 -0.327096000000
		H 1.580251000000 -0.700257000000 -1.616813000000
		C 0.310132000000 1.181251000000 0.988103000000
		H 1.205362000000 1.208798000000 1.609156000000
		H 0.248728000000 2.116835000000 0.430685000000
		H -0.554481000000 1.098278000000 1.645050000000
		C 0.400562000000 -1.324615000000 0.776961000000
		H 0.430167000000 -2.152707000000 0.069415000000
		H 1.282631000000 -1.376461000000 1.415603000000
		H -0.486203000000 -1.427020000000 1.399332000000

HOOCF ₃	Radical OH	Radical OF
Free Energy (M06-2X/aug-cc-pVTZ) = -488.637697 Enthalpy (M06-2X/aug-cc-pVTZ) = -488.602268	Free Energy (M06-2X/aug-cc-pVTZ) = -75.74218 Enthalpy (M06-2X/aug-cc-pVTZ) = -75.721954	Free Energy (M06-2X/aug-cc-pVTZ) = -174.897259 Enthalpy (M06-2X/aug-cc-pVTZ) = -174.873309
O -1.856092000000 0.002734000000 0.120459000000 H -2.145384000000 0.049238000000 -0.801526000000 O -0.715287000000 -0.827076000000 0.023679000000 C 0.380960000000 -0.009080000000 0.002872000000 F 1.428154000000 -0.807642000000 -0.068133000000 F 0.370577000000 0.795909000000 -1.056979000000 F 0.471343000000 0.745065000000 1.084133000000	O 0.000000000000 0.000000000000 0.107999000000 H 0.000000000000 0.000000000000 -0.863995000000	O 0.000000000000 0.000000000000 -0.703304000000 F 0.000000000000 0.000000000000 0.625159000000
Radical OCl	Radical OBr	Radical OCH ₃
Free Energy (M06-2X/aug-cc-pVTZ) = -535.326929 Enthalpy (M06-2X/aug-cc-pVTZ) = -535.301869	Free Energy (M06-2X/aug-cc-pVTZ) = -2649.371708 Enthalpy (M06-2X/aug-cc-pVTZ) = -2649.345384	Free Energy (M06-2X/aug-cc-pVTZ) = -115.02876 Enthalpy (M06-2X/aug-cc-pVTZ) = -115.001493
O 0.000000000000 0.000000000000 -1.066316000000 Cl 0.000000000000 0.000000000000 0.501796000000	O 0.000000000000 0.000000000000 -1.392606000000 Br 0.000000000000 0.000000000000 0.318310000000	O 0.791578000000 0.005110000000 -0.000013000000 C -0.577225000000 0.011165000000 -0.000084000000 H -1.018050000000 1.011836000000 -0.002338000000 H -0.925687000000 -0.562153000000 -0.871114000000 H -0.925538000000 -0.557549000000 0.874053000000
Radical OC(CH ₃) ₃	Radical OCF ₃	
Free Energy (M06-2X/aug-cc-pVTZ) = -232.885292 Enthalpy (M06-2X/aug-cc-pVTZ) = -232.848677	Free Energy (M06-2X/aug-cc-pVTZ) = -412.832775 Enthalpy (M06-2X/aug-cc-pVTZ) = -412.79949	
O 0.001692000000 -0.263560000000 1.427181000000 C -0.000190000000 0.023663000000 0.076131000000 C -1.270581000000 0.778629000000 -0.311781000000 H -2.149633000000 0.202044000000 -0.027932000000 H -1.304701000000 1.739706000000 0.200914000000 H -1.294695000000 0.960582000000 -1.386496000000 C 0.009399000000 -1.374270000000 -0.578993000000 H -0.874378000000 -1.937312000000 -0.285713000000 H 0.007809000000 -1.240597000000 -1.660612000000 H 0.901534000000 -1.924545000000 -0.286799000000 C 1.259906000000 0.795716000000 -0.311646000000 H 2.146635000000 0.230897000000 -0.028012000000 H 1.281501000000 0.978227000000 -1.386318000000 H 1.281190000000 1.757051000000 0.201252000000	O -0.785606000000 -0.011642000000 1.147947000000 C -0.006648000000 -0.000170000000 0.033180000000 F 0.773708000000 -1.068474000000 -0.032428000000 F 0.770148000000 1.071251000000 -0.014105000000 F -0.841108000000 0.007685000000 -0.995985000000	