

Electronic Supplementary Information

Interfacial and bulk properties of concentrated solutions of ammonium nitrate

Sara Mosallanejad^a, Ibukun Oluwoye^a, Mohammednoor Altarawneh^b, Jeff Gore^c, Bogdan Z.

Dlugogorski^{d,*}

^a Discipline of Chemistry and Physics, College of Science, Health, Engineering and

Education, Murdoch University

90 South Street, Murdoch, WA 6150, Australia

^b Department of Chemical and Petroleum Engineering, United Arab Emirates University,

Sheikh Khalifa bin Zayed Street, Al-Ain, 15551, United Arab Emirates

^c Dyno Nobel Asia Pacific Pty Ltd, Mt Thorley, NSW 2330, Australia

^d Energy and Resources Institute

Charles Darwin University

Casuarina Campus, Orange 12.3.49, Ellengowan Drive

Northern Territory 0909, Australia

*Corresponding author's email:

bogdan.dlugogorski@cdu.edu.au

Section I: Non-polarisable water models

Table S1. The molecular parameters for the three water models employed in the current study.

	σ (Å)	ε (kJ mol ⁻¹)	q ($ e $)
SPC/E model¹			
O	3.166	0.650	-0.8476
H	0	0	0.4238
O-H distance (Å)	1.0		
H-O-H angle (°)	109.47		
k_b (kJ (mol·Å ²) ⁻¹)	2319.13		For SPCE/F model ²
k_θ (kJ (mol·rad ²) ⁻¹)	191.5		For SPCE/F model ²
TIP4P/2005 model³			
O	3.1589	0.774	0
H	0	0	0.5564
M	0	0	-1.1128
O-H distance (Å)	0.9572		
H-O-H angle (°)	104.52		
O-M distance (Å)	0.1546		

Section II. Various properties of water with non-polarisable force fields

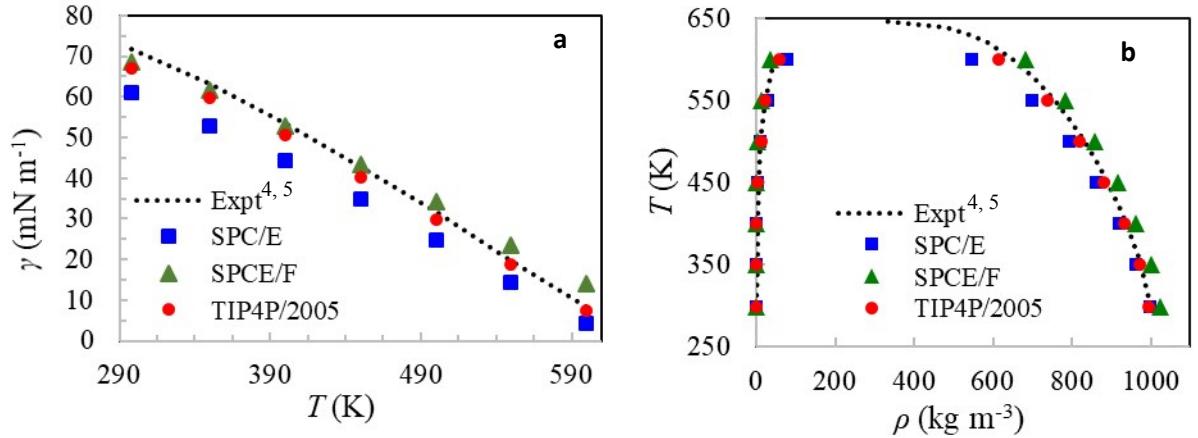


Figure S1. a) Values of the surface tension and b) the orthobaric densities of water. The dashed line represents the experimental values^{4, 5}, whereas the blue squares, green rectangles and red circles denote the present results with $N = 2133$ using SPC/E, SPCE/F and TIP4P/2005 models, respectively.

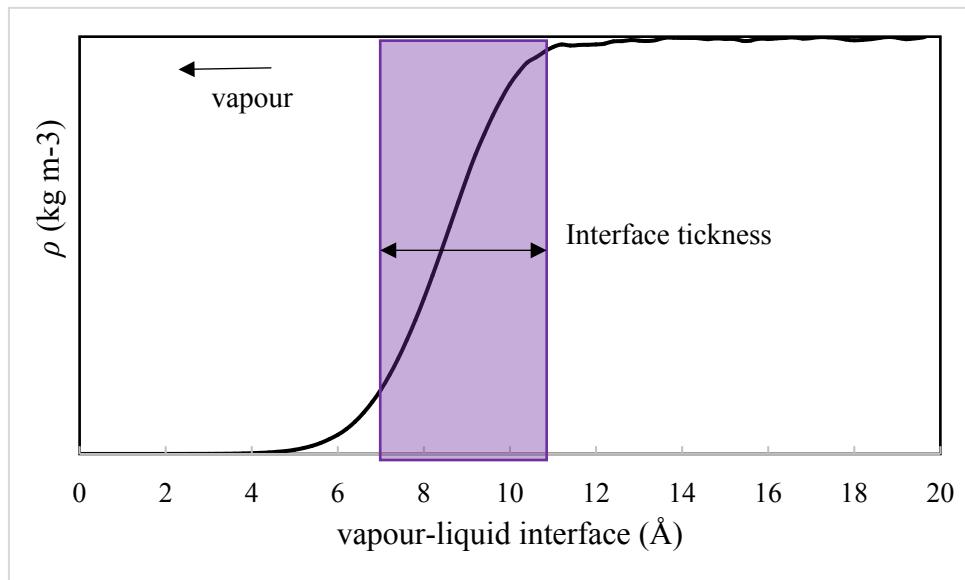


Figure S2. A schematic diagram of the density profile across the vapour-liquid interface and thickness of the interface for the SPC/E water model at 350 K.

Table S2. Liquid and vapour density, interfacial thickness and surface tension of SPC/E, SPCE/F and TIP4P/2005 water at the temperature range of 298.15 – 600 K. The interfacial thickness was calculated by fitting the hyperbolic tangent function to the density profiles. The numbers in parentheses represent one standard deviation.

<i>T</i> (K)	ρ_{liq} (kg m ⁻³)	ρ_{vap} (kg m ⁻³)	$\delta^{t_{10-90}}$ (Å)	γ (mN m ⁻¹)
SPC/E				
298.15	996.6 (0.1)	0.0	3.550 (0.008)	61.2 (0.1)
350	963.2 (0.1)	0.11 (0.03)	4.513 (0.007)	52.9 (0.1)
400	918.4 (0.1)	0.62 (0.02)	5.670 (0.008)	44.4 (0.1)
450	861.3 (0.2)	2.8 (0.1)	6.591 (0.022)	34.8 (0.2)
500	791.3 (0.1)	9.4 (0.1)	9.003 (0.012)	24.7 (0.1)
550	699.5 (0.2)	30.7 (1.1)	15.448 (0.018)	14.4 (0.2)
600	546.0 (1.1)	76.8 (0.9)	27.860 (0.044)	4.2 (0.2)
SPCE/F				
298.15	1025.2 (0.1)	0.0	3.131 (0.021)	68.7 (0.2)
350	999.6 (0.1)	0.0	4.001 (0.006)	61.8 (0.1)
400	961.8 (0.2)	0.21 (0.01)	4.816 (0.006)	52.8 (0.2)
450	915.0 (0.1)	1.3 (0.1)	6.771 (0.008)	43.4 (0.1)
500	857.4 (0.3)	4.4 (0.1)	7.329 (0.008)	34.1 (0.2)
550	783.4 (0.1)	12.2 (0.1)	12.150 (0.029)	23.6 (0.2)
600	683.1 (0.5)	35.0 (0.4)	17.797 (0.027)	14.2 (0.2)
TIP4P/2005				
298.15	995.0 (0.1)	0.0	3.405 (0.008)	67.2 (0.1)
350	970.4 (0.2)	0.0	4.293 (0.009)	59.8 (0.1)
400	932.0 (0.1)	0.61 (0.01)	5.150 (0.009)	50.5 (0.1)
450	881.4 (0.1)	2.4 (0.1)	7.193 (0.008)	40.4 (0.2)
500	818.2 (0.7)	13.2 (1.5)	10.590 (0.045)	29.9 (0.2)
550	737.5 (0.3)	21.4 (0.4)	15.318 (0.039)	19.0 (0.1)
600	614.0 (1.8)	58.9 (0.2)	26.691 (0.051)	7.5 (0.1)

Section III. Coordination numbers and comparison of OPLS and OPLS/ECC models

Table S3. The calculated coordination numbers (CN) for the first minima in $g(r)$ of Figure 4.

Concentration (mol kg ⁻¹)	First minima							
	N _n -O _w		N _n -H _w		N _a -O _w		N _a -H _w	
	r (Å)	$CN(r)$						
0.65	5.10	16.17	3.25	6.20	3.75	6.67	4.45	21.53
2.5	5.10	14.67	3.25	5.55	3.75	6.04	4.45	19.85
4.1	5.10	13.57	3.25	5.12	3.75	5.60	4.45	18.55
5.9	5.10	12.52	3.25	4.72	3.75	5.18	4.45	17.23
13.6	5.10	9.57	3.25	3.59	3.75	3.97	4.45	13.40
18.7	5.10	8.50	3.25	3.20	3.75	3.55	4.45	11.95

RDF using OPLS and OPLS/ECC models

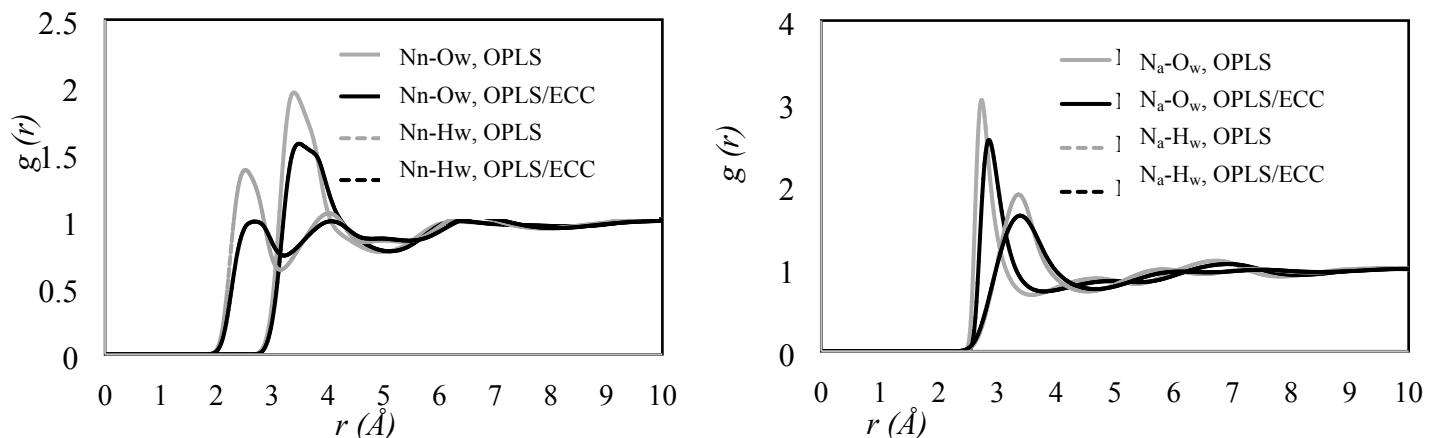


Figure S3. Nitrogen of ions-water radial distribution functions for 18.7 mol kg⁻¹ AN solution using the OPLS and OPLS/ECC models and the TIP4P/2005 force field for water at 298.15 K.

Section IV. Experimental values of surface tension of AN solutions

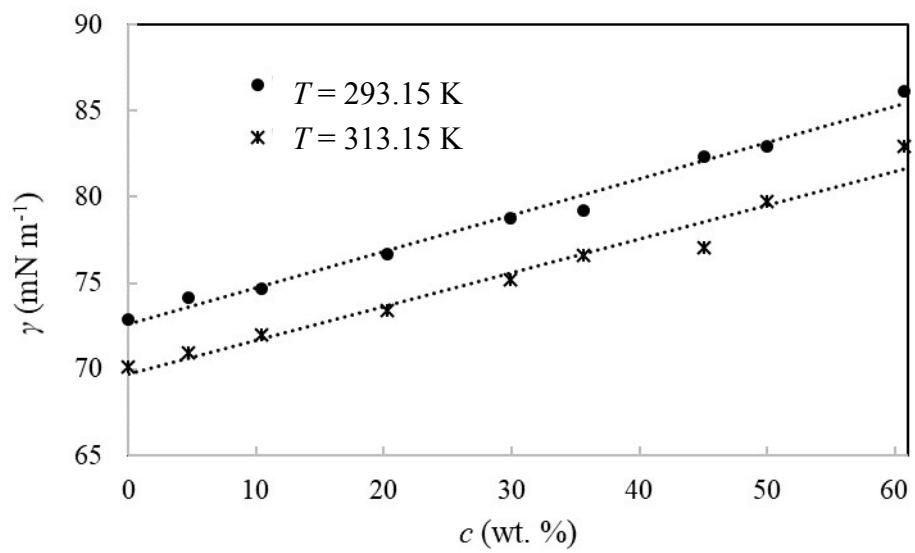


Figure S4. Experimental values of surface tension of AN solutions at temperatures of 293.15 K and 313.15 K.⁶

Section V. RDF and snapshots of coordination shell around one nitrogen of nitrate in the bulk and at the interface of the slab

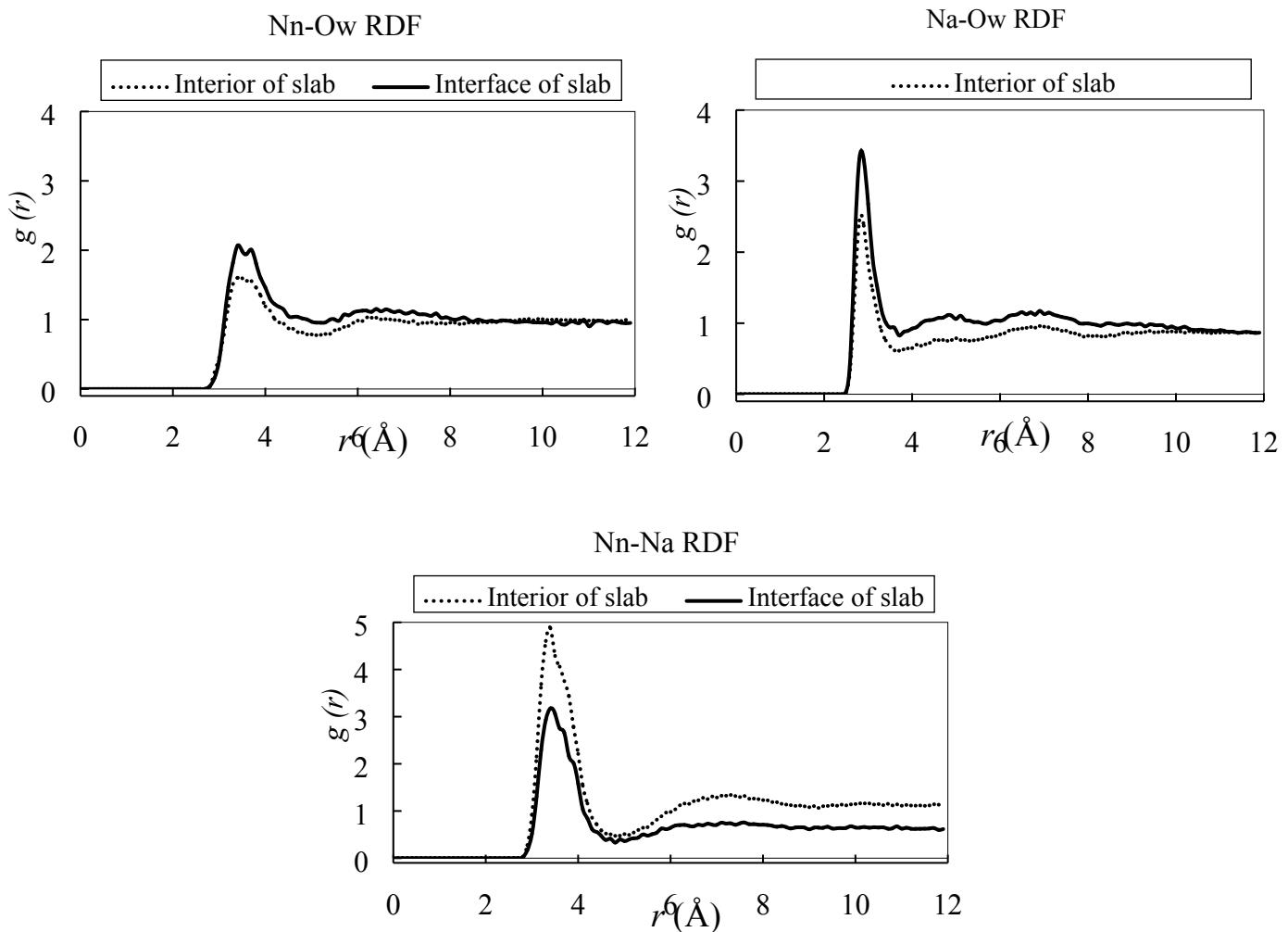


Figure S5. Radial distribution functions for the molality of AN solution of 18.7 mol kg^{-1} , for the OPLS/ECC model for the ions and the TIP4P/2005 model for water at 298.15 K ; N_n and N_a denote nitrogen atoms in nitrate and ammonium ions and O_w signifies the oxygen atoms in water.

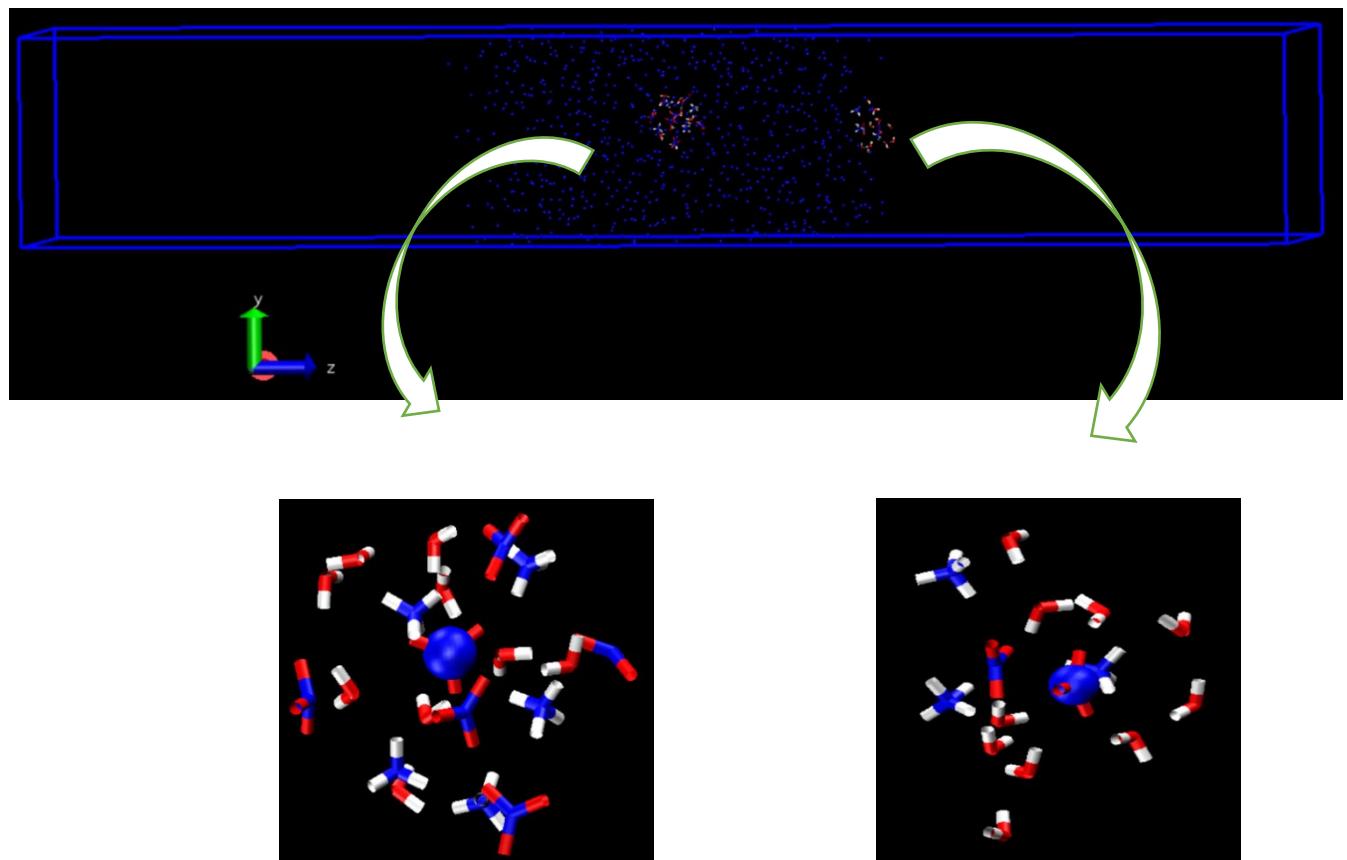


Figure S6. Solvation shells around nitrogen atoms in a nitrate anion (all water molecules and ions are within 5.1 Å of the target nitrogen atom) in the bulk and at the interface for the AN molality of 18.7 mol kg⁻¹. The water molecules, nitrate anions and ammonium cations are illustrated in red – white, blue – red and blue – white, respectively.

Section VI. Lammps code for calculating surface tension-Part 1

```
units      real
atom_style full
dimension 3
boundary p p f
```

```
pair_style lj/long/tip4p/long long long 6 2 2 2 0.1546 12
kspace_style pppm/disp/tip4p 1e-4
kspace_modify force/disp/real 0.0001
kspace_modify force/disp/kspace 0.002
kspace_modify slab 3.0
```

```
bond_style harmonic
angle_style harmonic
improper_style cvff
```

```
special_bonds      lj/coul 0.0 0.0 0.5
```

```
read_data 0-65.data
```

```
pair_coeff 1 1 0 0
pair_coeff 1 2 0 0
pair_coeff 1 3 0 0
pair_coeff 1 4 0 0
pair_coeff 1 5 0 0
pair_coeff 1 6 0 0
pair_coeff 2 2 0 0
pair_coeff 2 3 0 0
pair_coeff 2 4 0 0
pair_coeff 2 5 0 0
pair_coeff 2 6 0 0
pair_coeff 3 3 0.17 3.25
pair_coeff 3 4 0.17 3.19961
pair_coeff 3 5 0.188944 3.04877
pair_coeff 3 6 0.177437 3.20413
pair_coeff 4 4 0.17 3.15
pair_coeff 4 5 0.188944 3.0015
pair_coeff 4 6 0.177437 3.15445
pair_coeff 5 5 0.21 2.86
pair_coeff 5 6 0.197211 3.00574
pair_coeff 6 6 0.1852 3.1589
```

```
set type 1 charge 0.2625
```

```

set type 2 charge 0.5564
set type 3 charge -0.3
set type 4 charge 0.5955
set type 5 charge -0.4485
set type 6 charge -1.1128

group nitrate type 4 5 # Nn On
group ammonium type 1 3 # Ha Na
group water type 2 6 # Hw Ow

# neighbor skin
neighbor 2.0 bin
neigh_modify every 1 delay 0 check yes

compute pe all pe
compute ke all ke
compute 1 all chunk/atom bin/1d z center 0.2 units box

fix 23 water ave/chunk 1 500000 1000000 13 density/mass density/number file
density_water.txt
fix 2 all ave/chunk 1 500000 1000000 1 density/mass density/number file
density_all.txt

thermo 1000
thermo_style multi
#-----Energy minimization-----
#min_style cg
minimize 1.0e-8 1.0e-8 5000 10000
fix 3 all shake 0.0001 20 0 b 2 a 2
fix walls all wall/reflect zlo 40.5 zhi 82.5 units box

thermo 1000
thermo_style custom step spcpu temp density press ke pe etotal lx ly lz
dump 1 all custom 10000 traj.lammpstrj id mol type mass q xu yu zu ix iy iz vx vy vz
dump_modify 1 sort id
restart 1000000 ANwater.restart
#
#-----Initial velocity-----
velocity all create 298.15 123456 dist gaussian
#
#-----Equilibrium-----
fix 5 all npt temp 298.15 298.15 100.0 x 1.0 1.0 1000.0 y 1.0 1.0 1000.0 couple xy
timestep 1
run 500000

```

```
write_data npt_wall_1run.data
write_coeff pair.coeff
```

Lammps code for calculating surface tension-Part 2

```
units          real
atom_style    full
dimension     3
boundary      p p p

pair_style    lj/long/tip4p/long long long 6 2 2 2 0.1546 12
kspace_style  pppm/disp/tip4p 1e-6
kspace_modify force/disp/real 0.0001
kspace_modify force/disp/kspace 0.002
#kspace_modify slab 3.0

bond_style    harmonic
angle_style   harmonic
improper_style cvff

special_bonds    lj/coul 0.0 0.0 0.5

read_data      npt_wall_1run.data

pair_coeff 1 1 0 0
pair_coeff 1 2 0 0
pair_coeff 1 3 0 0
pair_coeff 1 4 0 0
pair_coeff 1 5 0 0
pair_coeff 1 6 0 0
pair_coeff 2 2 0 0
pair_coeff 2 3 0 0
pair_coeff 2 4 0 0
pair_coeff 2 5 0 0
pair_coeff 2 6 0 0
pair_coeff 3 3 0.17 3.25
pair_coeff 3 4 0.17 3.19961
pair_coeff 3 5 0.188944 3.04877
pair_coeff 3 6 0.177437 3.20413
pair_coeff 4 4 0.17 3.15
pair_coeff 4 5 0.188944 3.0015
pair_coeff 4 6 0.177437 3.15445
pair_coeff 5 5 0.21 2.86
pair_coeff 5 6 0.197211 3.00574
pair_coeff 6 6 0.1852 3.1589
```

```

set type 1 charge 0.2625
set type 2 charge 0.5564
set type 3 charge -0.3
set type 4 charge 0.5955
set type 5 charge -0.4485
set type 6 charge -1.1128

group nitrate type 4 5 # Nn On
group ammonium type 1 3 # Ha Na
group water type 2 6 # Hw Ow
group oxygen type 6 #Owater

# neighbor skin
neighbor 2.0 bin
neigh_modify every 1 delay 0 check yes

compute pe all pe
compute ke all ke
compute 1 all chunk/atom bin/1d z center 0.2 units box

fix 23 water ave/chunk 1 1000000 1000000 13 density/mass density/number file
density_water.txt
fix 21 nitrate ave/chunk 1 1000000 1000000 11 density/mass density/number file
density_nitrate.txt
fix 22 ammonium ave/chunk 1 1000000 1000000 12 density/mass density/number
file density_ammonium.txt
fix 2 all ave/chunk 1 1000000 1000000 1 density/mass density/number file
density_all.txt
fix 24 oxygen ave/chunk 1 1000000 1000000 14 density/mass density/number file
density_oxygen.txt

# Set up pressure tensor components
# c_thermo_press[1] is the XX component of the pressure tensor
variable xPress equal c_thermo_press[1]
variable yPress equal c_thermo_press[2]
variable zPress equal c_thermo_press[3]

# Computes the symmetric per-atom stress tensor
compute CC2 all chunk/atom bin/1d z center 0.5 units box
compute stpa all stress/atom NULL
fix stress all ave/chunk 1 1000000 1000000 CC2 c_stpa[1] c_stpa[2] c_stpa[3] &
c_stpa[4] c_stpa[5] c_stpa[6] file "out.stress.zProfile.txt"

# Compute surface tension

```

```

variable A_in_m equal 1e-10 # Angstrom in meter
variable atm_in_Pa equal 101325 # note: 1 Pa = 1 N/m^2
variable N_in_mN equal 1e3 # Newton in milliNewton
variable convFac equal ${A_in_m}*${atm_in_Pa}*${N_in_mN}
# lz is the box length in the z-dimention
variable st equal 0.5*lz*(v_zPress-0.5*(v_xPress+v_yPress))*${convFac} # in mN/m
fix st all ave/time 1 500 500 v_st file "out.surfTension.txt"
fix st average all ave/time 1 500 500 v_st file "out.AVesurfTension.txt" ave running

thermo 1000
thermo_style multi
#-----Energy minimization-----
#min_style cg
minimize 1.0e-8 1.0e-8 5000 10000
fix 3 all shake 0.0001 20 0 b 2 a 2
#fix walls all wall/reflect zlo 69.5 zhi 140.5 units box

thermo 1000
thermo_style custom step spcpu temp density press ke pe etotal lx ly lz pxx pyy pzz
dump 1 all custom 10000 traj.lammpstrj id mol type mass q xu yu zu ix iy iz vx vy vz
dump_modify 1 sort id
restart 1000000 ANwater.restart
#
#-----Initial velocity-----
#velocity all create 298.15 123456 dist gaussian
#
#-----Equilibrium-----
fix 5 all nvt temp 298.15 298.15 100.0
timestep 1
run 20000000
write_data nvt_final_1.data
write_coeff pair.coeff

```

References

1. Berendsen, H. J. C.; Grigera, J. R.; Straatsma, T. P., The Missing Term in Effective Pair Potentials. *J. Phys. Chem.* **1987**, *91* (24), 6269-6271.
2. Lopez-Lemus, J.; Chapela, G. A.; Alejandre, J., Effect of Flexibility on Surface Tension and Coexisting Densities of Water. *J. Chem. Phys.* **2008**, *128* (17), 174703.
3. Abascal, J. L.; Vega, C., A General Purpose Model for The Condensed Phases of Water: TIP4P/2005. *J. Chem. Phys.* **2005**, *123* (23), 234505.
4. Vargaftik, N. B.; Volkov, B. N.; Voljak, L. D., International Tables of the Surface Tension of Water. *J. Phys. Chem. Ref. Data* **1983**, *12* (3), 817-820.
5. Green, D. W.; Perry, R. H., *Perry's Chemical Engineers' Handbook* 8th ed.; McGraw-Hill: 2008.
6. Abramzon, A. A.; Gauberk, R. D., Surface Tension of Salt Solutions. *Russ. J. Appl. Chem.* **1993**, *66* (8), 1474.