

RADICAL PAIR FORMATION DUE TO COMPRESSION-INDUCED ELECTRON TRANSFER IN CRYSTALS OF ENERGETIC SALTS

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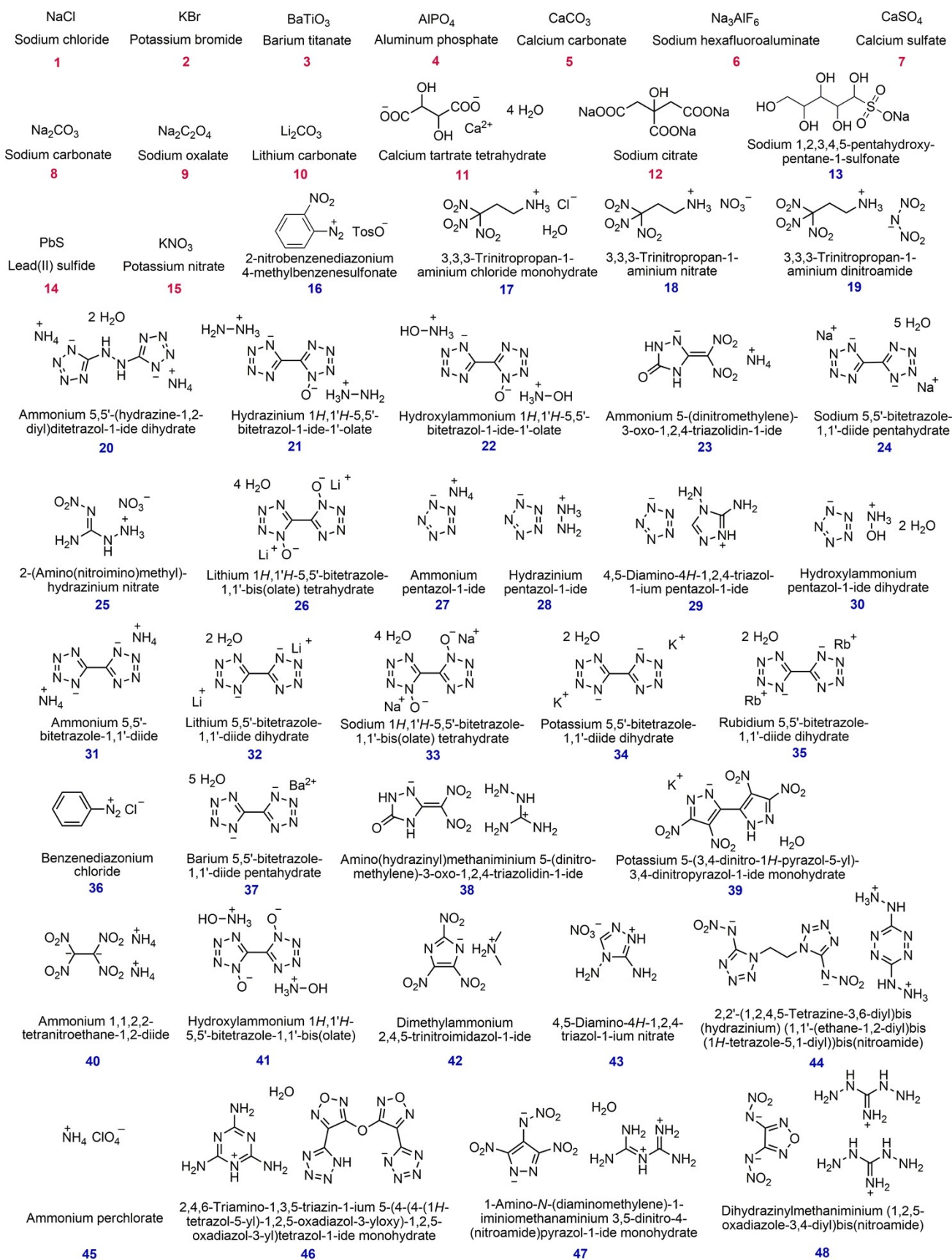


Fig. S1. Chemical formulas and full names of the salts studied.

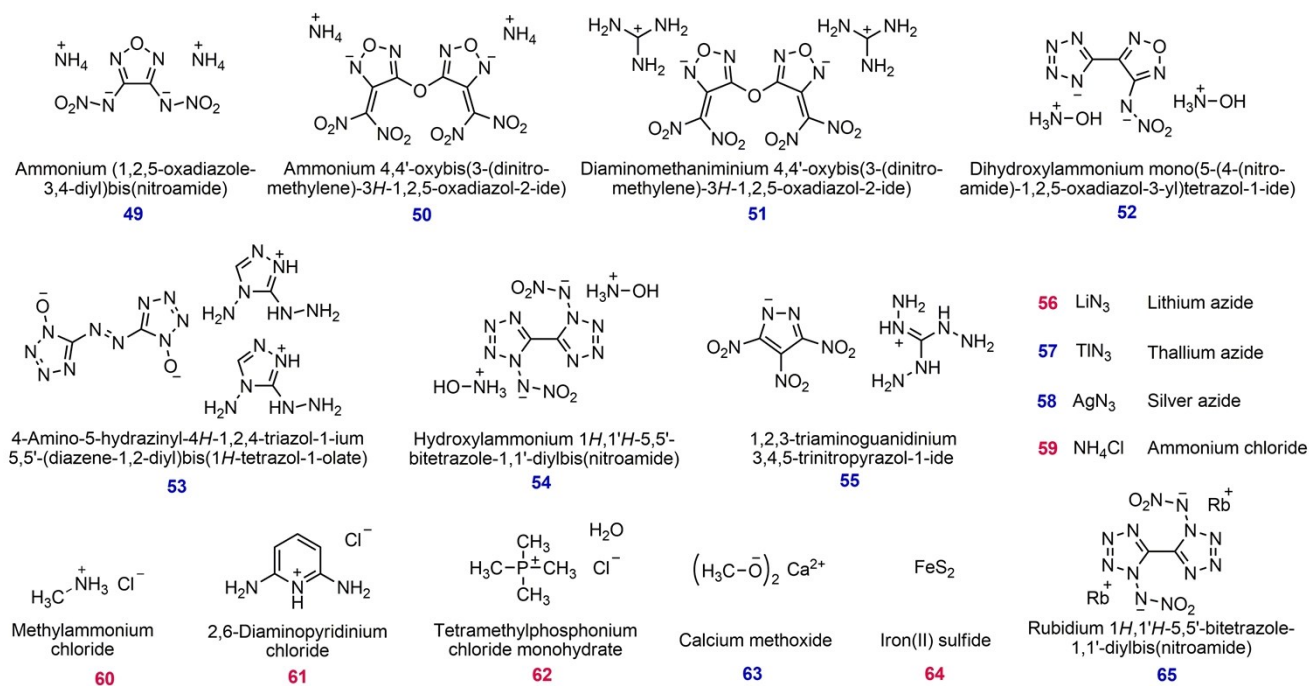


Fig. S1. Continue.

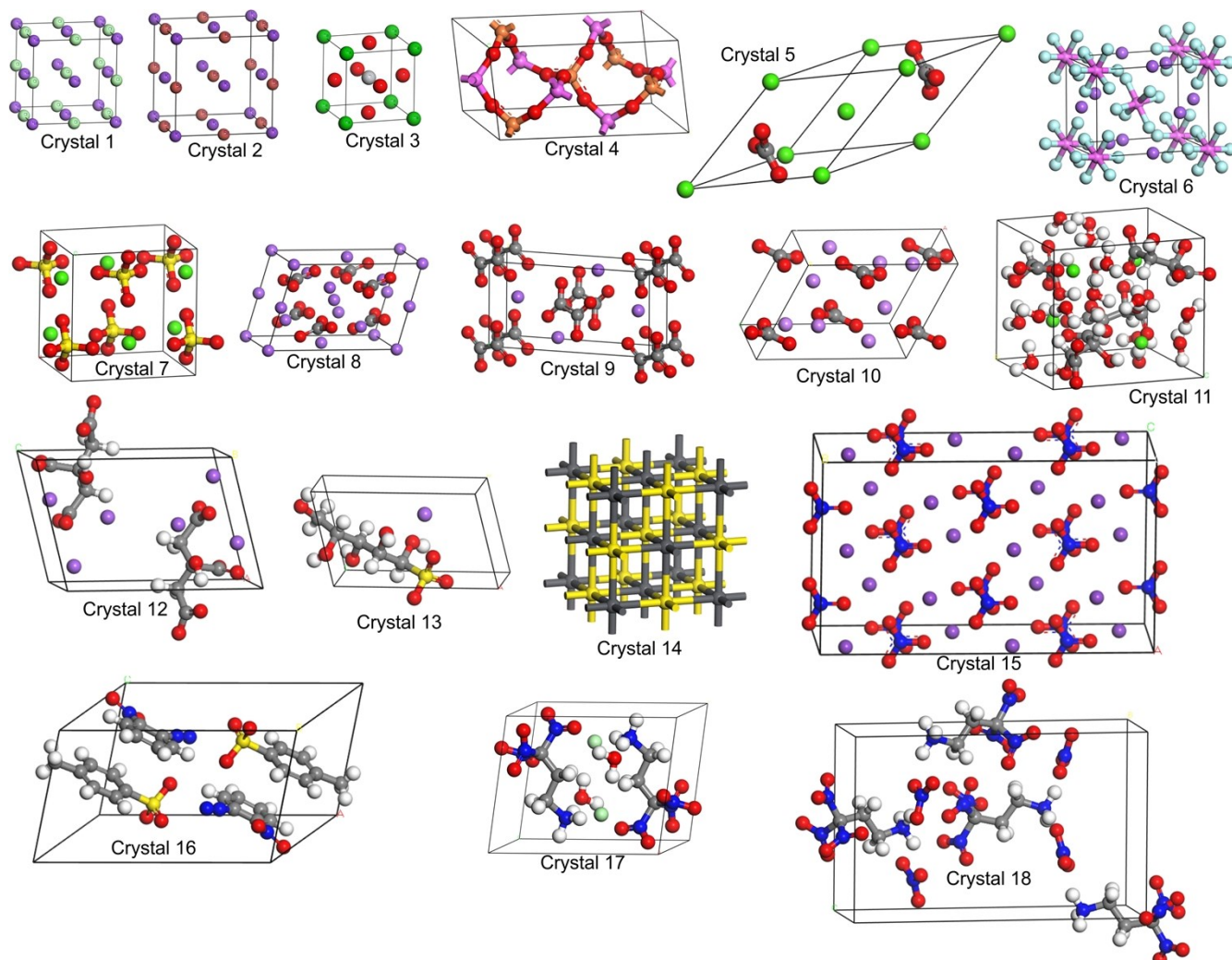


Fig. S2. Crystal structures of the salts studied.

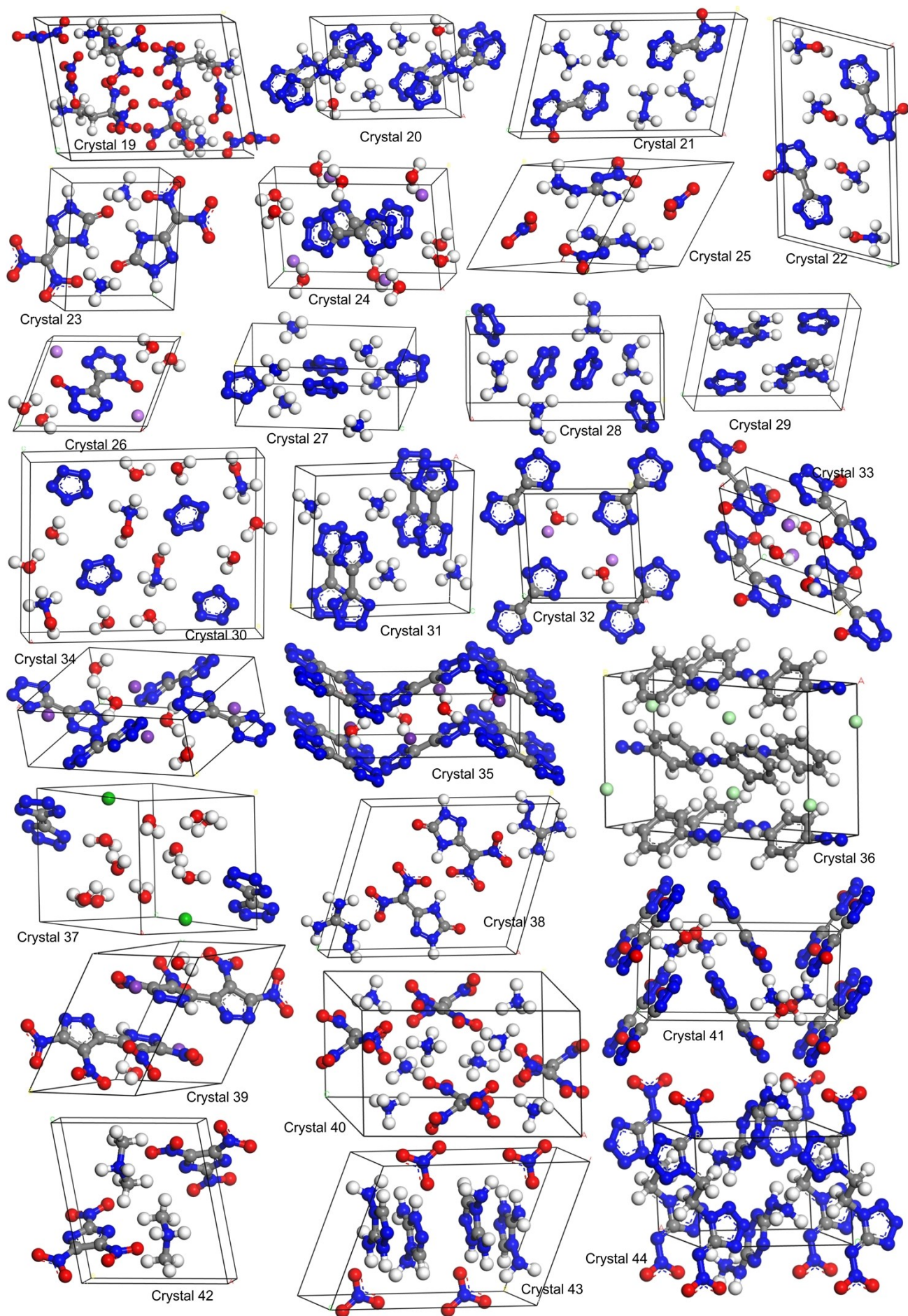


Fig. S2. Continue.
S5

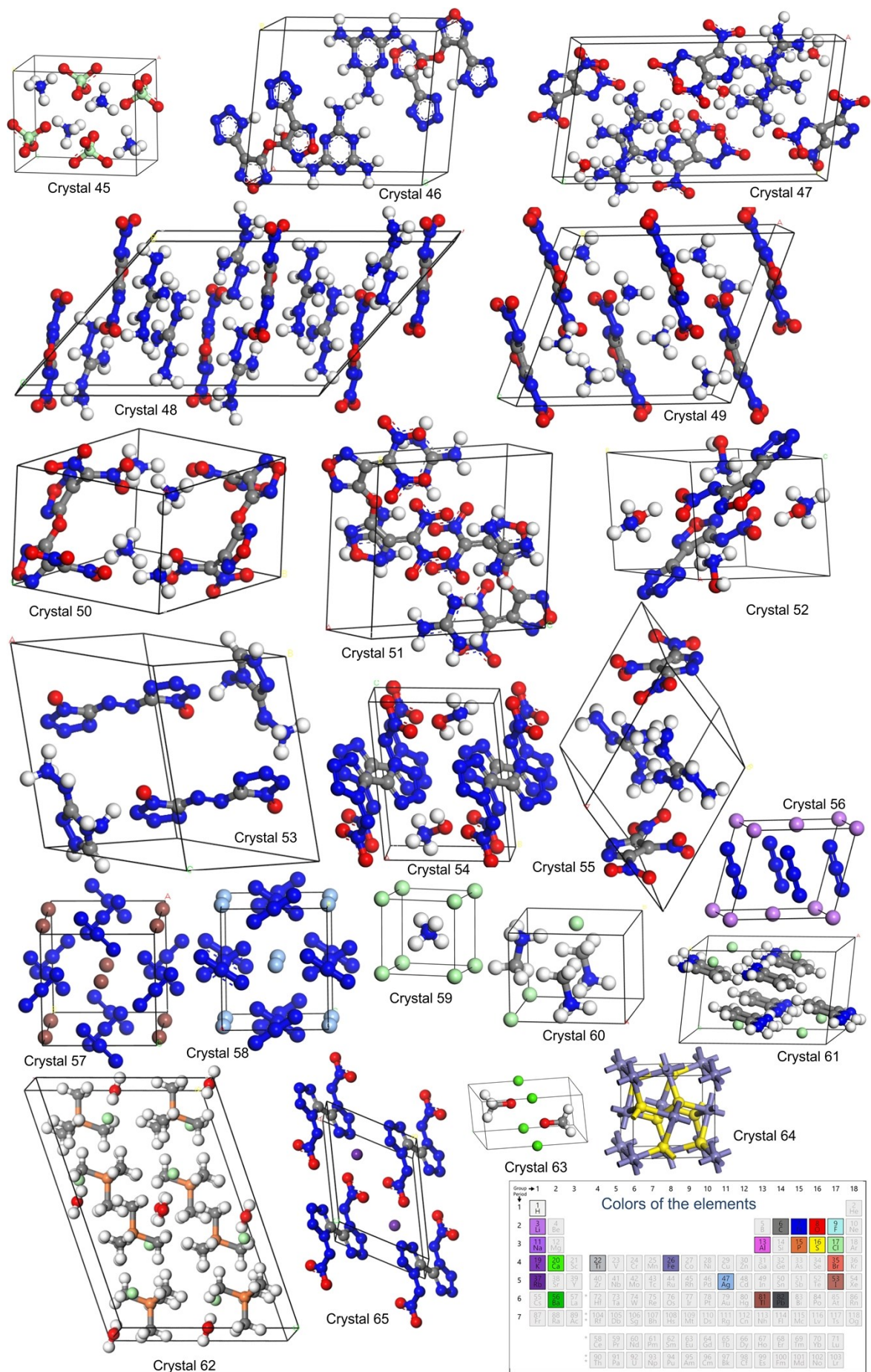


Fig. S2. Continue.
S6

Table S1. The calculated and experimental (in parentheses) asymmetric cell parameters of the salts studied.

Crystal	Space group	<i>a</i>	<i>b</i>	<i>c</i>	α	β	γ
1	<i>Fm</i> $\bar{3}m$	5.672 (5.640)					
2	<i>Fm</i> $\bar{3}m$	6.597 (6.597)					
3	<i>Pm</i> $\bar{3}m$	3.954 (4.010)					
4	<i>P</i> 3 ₁ 2 ₁	5.053 (4.942)		11.092 (10.945)			
5	<i>R</i> $\bar{3}c$	6.351 (6.375)			46.5 (46.1)		
6	<i>P</i> 2 ₁ / <i>n</i>	5.276 (5.402)	5.546 (5.596)	7.534 (7.756)		90.9 (90.3)	
7	<i>Amma</i>	7.016 (6.991)	7.012 (6.996)	6.326 (6.238)			
8	<i>C</i> 2/ <i>m</i>	8.804 (8.825)	5.288 (5.194)	5.921 (5.953)		104.8 (101.8)	
9	<i>P</i> 2 ₁ / <i>a</i>	10.299 (10.350)	5.334 (5.260)	3.442 (3.460)		91.4 (92.9)	
10	<i>C</i> /2 <i>c</i>	8.369 (8.359)	4.972 (4.973)	6.111 (6.198)		114.5 (114.8)	
11	<i>P</i> 2 ₁ 2 ₁ 2 ₁	9.488 (9.631)	10.455 (10.573)	9.109 (9.215)			
12	<i>P</i> 2 ₁	7.337 (7.347)	5.366 (5.435)	10.747 (11.034)		103.9 (103.9)	
13	<i>P</i> 1	4.840 (4.856)	5.911 (5.850)	8.880 (8.795)	76.0 (76.5)	81.6 (81.5)	71.5 (71.4)
14	<i>Fm</i> $\bar{3}m$	5.863 (5.930)					
15	<i>Cmc</i> 2 ₁	10.892 (10.904)	18.429 (18.455)	6.362 (6.421)			
16	<i>P</i> $\bar{1}$	7.489 (7.509)	7.572 (7.546)	13.611 (13.672)	78.6 (78.6)	76.7 (75.7)	75.3 (75.6)
17	<i>P</i> $\bar{1}$	6.770 (6.743)	7.826 (7.805)	10.154 (10.066)	89.8 (90.4)	99.2 (98.8)	113.9 (114.1)
18	<i>P</i> 2 ₁ 2 ₁ 2 ₁	5.698 (5.662)	10.293 (10.283)	16.382 (16.258)			
19	<i>P</i> $\bar{1}$	6.766 (6.709)	11.381 (11.255)	15.394 (15.214)	75.1 (75.5)	79.4 (79.3)	75.5 (75.7)
20	<i>P</i> $\bar{1}$	4.018 (3.824)	7.093 (6.967)	9.670 (9.920)	82.4 (80.5)	90.6 (88.7)	86.1 (79.5)
21	<i>P</i> $\bar{1}$	3.614 (3.655)	8.441 (8.436)	14.289 (14.338)	102.8 (103.6)	96.4 (96.6)	93.2 (93.3)
22	<i>P</i> $\bar{1}$	3.628 (3.606)	8.020 (8.019)	14.833 (14.798)	105.5 (105.1)	91.8 (91.7)	96.5 (96.0)
23	<i>P</i> $\bar{1}$	6.764 (6.785)	7.092 (7.137)	8.070 (7.965)	84.3 (85.1)	75.0 (74.0)	84.6 (84.9)
24	<i>P</i> $\bar{1}$	7.199 (7.209)	7.429 (7.462)	10.679 (10.775)	82.2 (81.3)	76.6 (76.5)	68.5 (67.6)
25	<i>P</i> $\bar{1}$	7.288 (7.175)	7.181 (7.202)	7.909 (7.856)	65.8 (66.1)	63.1 (65.3)	66.1 (68.5)
26	<i>P</i> $\bar{1}$	5.280 (5.345)	6.343 (6.363)	8.228 (8.300)	105.4 (105.9)	97.6 (98.2)	112.3 (111.5)
27	<i>Pcca</i>	9.488 (9.399)	3.956 (3.954)	10.353 (10.362)			

Table S1. Continue.

Crystal	Space group	<i>a</i>	<i>b</i>	<i>c</i>	α	β	γ
28	$P2_1/c$	6.554 (6.453)	12.076 (12.247)	5.369 (5.407)		97.9 (97.8)	
29	$P\bar{1}$	5.315 (5.298)	7.128 (7.156)	10.174 (10.106)	100.6 (100.0)	103.9 (104.1)	103.7 (104.1)
30	$P2_1/c$	3.630 (3.685)	14.796 (14.638)	11.443 (11.575)		90.8 (90.3)	
31	$Pban$	9.123 (9.130)	10.691 (10.763)	3.669 (3.666)			
32	$P\bar{1}$	3.353 (3.368)	7.368 (7.362)	7.490 (7.501)	83.2 (82.9)	86.5 (87.1)	80.4 (80.8)
33	$P\bar{1}$	5.587 (5.681)	6.471 (6.451)	8.354 (8.355)	99.5 (100.3)	97.0 (97.3)	111.7 (111.8)
34	$P2_1/c$	3.836 (3.851)	11.347 (11.316)	9.750 (9.735)		93.6 (93.0)	
35	$P2_1/c$	4.051 (4.088)	11.355 (11.328)	9.990 (9.948)		94.6 (93.5)	
36	$C222_1$	15.050 (15.152)	4.860 (4.928)	9.011 (9.044)			
37	$P\bar{1}$	7.417 (7.440)	7.876 (8.162)	9.272 (9.259)	89.3 (88.6)	74.0 (73.7)	88.7 (88.9)
38	$P\bar{1}$	3.630 (3.648)	10.997 (10.984)	13.276 (13.201)	107.4 (107.2)	94.6 (95.4)	98.6 (99.1)
39	$P\bar{1}$	7.336 (7.354)	8.853 (9.047)	10.809 (10.724)	105.1 (105.6)	94.9 (94.5)	108.7 (108.9)
40	$P2_1/n$	9.322 (9.255)	7.865 (7.800)	13.152 (13.135)		107.8 (107.9)	
41	$P2_1/c$	5.527 (5.426)	11.523 (11.660)	6.480 (6.501)		96.10 (95.26)	
42	$P2_1$	8.516 (8.484)	6.163 (6.144)	9.507 (9.439)		100.9 (100.6)	
43	$P2_1/c$	8.764 (8.527)	6.132 (6.265)	12.401 (12.760)		110.0 (112.0)	
44	$P\bar{1}$	6.796 (6.771)	7.639 (7.562)	8.737 (8.712)	96.6 (96.5)	101.7 (101.9)	115.5 (115.0)
45	$Pnma$	8.859 (9.130)	6.043 (5.790)	7.241 (7.470)			
46	$P\bar{1}$	6.535 (6.691)	10.776 (10.822)	12.326 (12.259)	84.8 (85.2)	76.4 (76.7)	84.2 (84.7)
47	$P2_1/n$	9.758 (9.688)	7.062 (7.154)	18.405 (18.317)		97.9 (98.0)	
48	$C2/c$	18.570 (18.761)	8.268 (8.292)	12.572 (12.498)		131.4 (131.3)	
49	$C2/c$	12.136 (12.031)	7.489 (7.401)	10.175 (9.981)		111.2 (109.4)	
50	$P\bar{1}$	7.724 (7.691)	9.367 (9.290)	10.495 (10.564)	93.8 (94.5)	104.4 (104.2)	99.8 (100.4)
51	$P\bar{1}$	8.747 (8.882)	10.374 (10.251)	10.661 (10.718)	79.0 (78.1)	86.0 (83.7)	72.2 (71.3)
52	$P\bar{1}$	7.118 (7.130)	8.205 (8.131)	9.176 (9.057)	83.7 (83.4)	71.2 (71.7)	73.0 (72.5)
53	$P\bar{1}$	8.262 (7.972)	8.374 (8.192)	10.743 (10.269)	89.3 (90.4)	109.5 (110.2)	117.5 (115.5)
54	$P\bar{1}$	5.176 (5.242)	6.649 (6.769)	8.476 (8.400)	96.9 (96.4)	90.2 (89.2)	97.9 (97.5)
55	$P\bar{1}$	5.065 (5.113)	10.742 (10.676)	11.565 (11.579)	65.3 (66.3)	77.7 (78.5)	85.8 (84.9)

Table S1. Continue.

Crystal	Space group	<i>a</i>	<i>b</i>	<i>c</i>	α	β	γ
56	<i>C2/m</i>	5.602 (5.627)	3.282 (3.319)	4.854 (4.979)		104.1 (107.4)	
57	<i>I4/mcm</i>	6.131 (6.208)		7.591 (7.355)			
58	<i>Ibam</i>	6.186 (5.617)	5.366 (5.915)	6.364 (6.006)			
59	<i>P43m</i>	3.806 (3.860)					
60	<i>Pmn2₁</i>	5.971 (6.040)	4.813 (5.050)	6.202 (6.040)			
61	<i>C2/m</i>	10.805 (10.763)	9.046 (9.076)	6.811 (6.635)		96.7 (95.4)	
62	<i>C2/c</i>	11.818 (11.762)	8.667 (8.628)	17.591 (17.604)		108.6 (107.5)	
63	<i>P3m1</i>	3.688 (3.642)		8.019 (8.311)			
64	<i>Pa3</i>	5.317 (5.428)					
65	<i>P1</i>	5.356 (5.303)	7.077 (7.063)	8.621 (8.568)	66.5 (66.8)	86.3 (86.7)	71.1 (70.0)

Table S2. The relative errors $\delta = (V_{theor} - V_{exper})/V_{exper}$ of the asymmetric cell volumes estimation.

Salt	V_{theor} (Å ³)	V_{exper} (Å ³)	δ (%)	Salt	V_{theor} (Å ³)	V_{exper} (Å ³)	δ (%)
1	182.48	179.41	1.71	34	423.55	423.65	-0.02
2	287.10	287.10	0.00	35	458.05	459.82	-0.38
3	61.82	64.48	-4.13	36	659.09	675.31	-2.40
4	283.21	267.31	5.95	37	520.51	539.45	-3.51
5	123.07	122.72	0.29	38	495.64	493.36	0.46
6	220.42	234.46	-5.99	39	630.95	639.35	-1.31
7	311.22	305.09	2.01	40	918.12	902.31	1.75
8	266.51	267.10	-0.22	41	410.36	409.57	0.19
9	189.03	188.12	0.48	42	489.97	483.62	1.31
10	231.39	233.89	-1.07	43	626.26	632.03	-0.91
11	903.59	938.35	-3.70	44	390.27	385.38	1.27
12	410.73	427.70	-3.97	45	387.65	394.88	-1.83
13	233.10	229.52	1.56	46	837.34	858.38	-2.45
14	201.54	208.53	-3.35	47	1256.27	1257.16	-0.07
15	1277.04	1292.12	-1.17	48	1447.96	1460.71	-0.87
16	718.54	719.40	-0.12	49	862.20	838.27	2.85
17	484.41	476.49	1.66	50	720.04	713.73	0.88
18	960.80	946.58	1.50	51	904.08	903.47	0.07
19	1099.94	1068.31	2.96	52	485.07	475.31	2.05
20	272.48	256.29	6.32	53	612.48	558.43	9.68
21	420.96	425.21	-1.00	54	286.79	293.67	-2.34
22	412.37	410.15	0.54	55	558.44	567.10	-1.53
23	371.13	368.54	0.70	56	86.56	88.73	-2.45
24	516.10	519.82	-0.72	57	285.34	283.46	0.66
25	324.17	327.54	-1.03	58	211.25	199.55	5.86
26	237.27	243.11	-2.40	59	55.13	57.51	-4.14
27	388.59	385.09	0.91	60	178.24	184.23	-3.25
28	420.90	423.36	-0.58	61	661.18	645.27	2.47
29	351.47	349.18	0.66	62	1707.69	1703.83	0.23
30	614.54	624.37	-1.57	63	109.07	110.24	-1.06
31	357.85	360.24	-0.66	64	150.31	159.93	-6.02
32	181.00	182.10	-0.60	65	282.72	276.06	2.41
33	271.21	273.38	-0.79				

Table S3. The Mulliken (M) and Hirshfeld (H) partial charges (e^-) at different pressures along with the frontier molecular orbitals energies (eV).

Salt	0 GPa		20 GPa		50 GPa		100 GPa		Anion		Cation	
	M	H	M	H	M	H	M	H	E_{HOMO}	E_{LUMO}	E_{HOMO}	E_{LUMO}
1	0.621	0.210	0.595	0.170	0.565	0.170	0.525	0.180	-0.769	5.89	-39.53	-6.98
2	0.713	0.250	0.707	0.190	0.687	0.170	0.673	0.170	-0.860	4.81	-26.73	-6.00
3	0.714	0.200	0.681	0.194	0.663	0.200	0.651	0.205	3.820	4.79	-32.66	-11.98
4	0.610	0.193	0.600	0.173	0.640	0.160	0.720	0.149	9.031	11.42	-108.48	-32.58
5	0.731	0.195	0.684	0.166	0.651	0.146	0.609	0.130	5.791	9.00	-45.15	-14.24
6	0.793	0.288	0.822	0.262	0.852	0.259	0.877	0.260	5.912	10.34	-39.53	-6.98
7	0.890	0.181	0.920	0.139	0.960	0.120	1.000	0.106	3.606	8.33	-45.15	-14.24
8	0.950	0.270	0.970	0.262	0.987	0.257	1.010	0.263	5.791	9.00	-39.53	-6.98
9	0.750	0.283	0.705	0.240	0.710	0.238	0.740	0.243	3.685	8.11	-39.53	-6.98
10	1.062	0.190	1.218	0.150	1.373	0.128	1.510	0.113	5.791	9.00	-64.01	-6.93
11	0.750	0.220	0.705	0.120	0.710	0.107	0.740	0.100	1.523	6.38	-45.15	-14.24
12	1.062	0.290	1.217	0.260	1.373	0.257	1.397	0.270	4.891	8.89	-39.53	-6.98
13	1.230	0.260	1.721	0.250	1.979	0.249	2.310	0.241	-3.467	2.43	-39.53	-6.98
14	0.600	0.270	0.530	0.220	0.440	0.190	0.310	0.170	5.643	9.25	-28.28	-16.70
15	0.912	0.312	0.960	0.232	1.008	0.210	1.058	0.192	-1.204	4.42	-26.73	-6.00
16	0.861	0.600	0.686	0.510	0.510	0.419	0.286	0.296	-2.326	2.28	-13.18	-8.75
17	0.700	0.280	0.570	0.130	0.460	0.070	0.350	0.010	-0.769	5.89	-13.08	-7.86
18	0.792	0.383	0.733	0.188	0.690	0.100	0.600	0.000	-1.204	4.42	-13.08	-7.86
19	0.830	0.364	0.784	0.136	0.744	0.027	0.719	-0.085	-2.905	1.93	-13.08	-7.86
20	0.665	0.275	0.641	0.151	0.615	0.095	0.565	0.045	1.521	6.59	-22.46	-6.36
21	0.742	0.299	0.672	0.145	0.643	0.057	0.678	0.022	1.879	7.17	-14.16	-6.01
22	0.650	0.232	0.590	0.112	0.567	0.042	0.545	-0.005	1.879	7.17	-15.97	-6.38
23	0.773	0.343	0.725	0.163	0.690	0.080	0.688	0.010	-2.483	1.12	-22.46	-6.36
24	0.830	0.185	0.890	0.045	0.945	-0.005	1.005	-0.035	1.362	7.31	-39.53	-6.98
25	0.750	0.308	0.646	0.107	0.614	0.032	0.652	0.012	-1.204	4.42	-12.25	-6.46
26	0.820	0.125	0.830	0.021	0.835	-0.040	0.880	-0.085	1.645	7.08	-64.01	-6.93
27	0.730	0.310	0.660	0.115	0.635	0.040	0.725	0.005	-3.095	5.07	-22.46	-6.36
28	0.794	0.334	0.750	0.140	0.782	0.080	0.780	0.040	-3.095	5.07	-14.16	-6.01
29	0.800	0.343	0.737	0.136	0.743	0.043	0.763	0.003	-3.095	5.07	-12.02	-4.99
30	1.852	0.187	1.579	0.020	1.367	-0.023	1.100	-0.034	-3.095	5.07	-15.97	-6.38
31	0.675	0.270	0.620	0.115	0.655	0.070	0.645	0.015	1.362	7.31	-22.46	-6.36
32	0.836	0.134	0.820	0.024	0.816	-0.030	0.850	-0.070	1.362	7.31	-64.01	-6.93
33	0.794	0.165	0.817	0.045	0.862	-0.005	0.932	-0.035	1.645	7.08	-39.53	-6.98
34	0.910	0.216	0.996	0.116	1.056	0.056	1.120	0.024	1.362	7.31	-26.73	-6.00
35	0.940	0.234	1.010	0.124	1.090	0.066	1.160	0.016	1.362	7.31	-22.91	-5.50
36	0.819	0.420	0.689	0.221	0.560	0.130	0.411	0.041	-0.769	5.89	-12.59	-8.18
37	0.665	0.162	0.610	0.043	0.612	-0.006	0.591	-0.050	1.362	7.31	-32.66	-11.98
38	0.757	0.304	0.734	0.141	0.731	0.051	0.824	-0.001	-2.483	1.12	-12.77	-4.66
39	0.730	0.270	0.660	0.066	0.666	0.030	0.676	0.014	-4.350	-0.75	-26.73	-6.00
40	0.804	0.337	0.745	0.134	0.730	0.047	0.725	-0.032	0.970	4.98	-22.46	-6.36
41	0.730	0.265	0.695	0.130	0.650	0.075	0.635	0.025	1.645	7.08	-15.97	-6.38
42	0.860	0.310	0.791	0.014	0.744	-0.114	0.740	-0.210	-4.044	0.05	-16.45	-5.41
43	0.795	0.342	0.770	0.170	0.765	0.075	0.740	-0.008	-1.204	4.42	-12.02	-4.99
44	0.686	0.210	0.684	0.107	0.660	0.065	0.696	0.037	-0.716	4.20	-15.89	-13.76
45	0.810	0.355	0.703	0.120	0.657	-0.005	0.613	-0.085	-3.002	4.29	-22.46	-6.36
46	0.714	0.210	0.638	-0.042	0.604	-0.124	0.578	-0.204	-4.370	-0.01	-11.60	-5.23
47	0.743	0.283	0.681	0.091	0.614	0.003	0.768	0.000	1.052	4.29	-17.46	-10.76
48	0.780	0.294	0.747	0.070	0.727	-0.020	0.706	-0.114	1.168	5.52	-12.40	-4.33
49	0.790	0.334	0.720	0.166	0.720	0.108	0.682	0.018	1.168	5.52	-22.46	-6.36
50	0.803	0.361	0.759	0.134	0.755	0.041	0.755	-0.035	-1.244	3.24	-22.46	-6.36
51	0.807	0.319	0.737	0.062	0.784	-0.018	0.826	-0.080	-1.244	3.24	-13.39	-5.21
52	0.707	0.280	0.645	0.125	0.638	0.047	0.632	0.033	0.388	5.10	-15.97	-6.38
53	0.695	0.285	0.628	0.120	0.615	0.095	0.710	0.078	1.166	4.38	-11.67	-4.79
54	0.730	0.270	0.660	0.066	0.666	0.030	0.676	0.014	-0.502	4.35	-15.97	-6.38
55	0.845	0.325	1.414	0.304	1.395	0.180	1.340	0.035	-4.168	0.12	-12.09	-3.97

Table S3. Continue.

Salt	0 GPa		20 GPa		50 GPa		100 GPa		Anion		Cation	
	M	H	M	H	M	H	M	H	E_{HOMO}	E_{LUMO}	E_{HOMO}	E_{LUMO}
56	1.030	0.130	1.080	0.098	1.130	0.090	1.183	0.080	-0.04	6.12	-64.01	-6.93
57	0.700	0.320	0.730	0.230	0.820	0.220	1.120	0.303	-0.04	6.12	-17.08	-7.83
58	0.623	0.268	0.600	0.250	0.630	0.248	0.683	0.273	-0.04	6.12	-17.15	-10.66
59	0.700	0.257	0.568	0.078	0.468	-0.010	0.372	-0.080	-0.77	5.89	-22.46	-6.36
60	0.710	0.269	0.441	-0.079	0.511	-0.020	0.439	-0.079	-0.77	5.89	-17.12	-5.90
61	0.710	0.258	0.559	0.061	0.449	-0.041	0.331	-0.120	-0.77	5.89	-10.31	-5.66
62	0.711	0.269	0.570	0.020	0.469	-0.080	0.351	-0.160	-0.77	5.89	-14.73	-4.25
63	1.390	0.205	1.320	0.175	1.270	0.155	1.190	0.145	1.08	4.73	-45.15	-14.24
64	-0.020	0.060	-0.133	0.040	-0.287	0.020	-0.420	0.020	5.25	7.49	-28.09	-21.24
65	1.060	0.279	1.250	0.170	1.440	0.141	1.588	0.129	-0.50	4.35	-22.91	-5.50

Table S4. Decomposition equations applied for the calculations of the stored energy content (E_c).

Salt	Decomposition equation
1	$2\text{NaCl} = \text{Na}_2 + \text{Cl}_2$
2	$2\text{KBr} = \text{K}_2 + \text{Br}_2$
3	$\text{BaTiO}_3 = \text{BaO} + \text{TiO}_2$
4	$2\text{AlPO}_4 = \text{Al}_2\text{O}_3 + \text{P}_2\text{O}_5$
5	$\text{CaCO}_3 = \text{CaO} + \text{CO}_2$
6	$\text{Na}_3\text{AlF}_6 = 3\text{NaF} + \text{AlF}_3$
7	$\text{CaSO}_4 = \text{CaO} + \text{SO}_3$
8	$\text{Na}_2\text{CO}_3 = \text{Na}_2\text{O} + \text{CO}_2$
9	$\text{Na}_2\text{C}_2\text{O}_4 = \text{Na}_2\text{O} + \text{CO}_2 + \text{CO}$
10	$\text{Li}_2\text{CO}_3 = \text{Li}_2\text{O} + \text{CO}_2$
11	$\text{C}_4\text{H}_2\text{O}_6\text{Ca} + 4\text{H}_2\text{O} = \text{CaO} + 4\text{CO} + 5\text{H}_2\text{O}$
12	$2\text{C}_6\text{H}_5\text{O}_7\text{Na}_3 = 3\text{Na}_2\text{O} + 5\text{H}_2\text{O} + 6\text{CO} + 6\text{C}$
13	$8\text{C}_5\text{H}_{11}\text{O}_8\text{SNa} + 8\text{H}_2\text{O} = 8\text{NaOH} \cdot \text{H}_2\text{O} + 40\text{H}_2\text{O} + 8\text{CO}_2 + 32\text{C} + \text{S}_8$
14	$8\text{PbS} = 8\text{Pb} + \text{S}_8$
15	$4\text{KNO}_3 = 2\text{N}_2 + 2\text{K}_2\text{O} + 5\text{O}_2$
16	$16\text{C}_{13}\text{H}_{11}\text{N}_3\text{O}_5\text{S} = 24\text{N}_2 + 80\text{H}_2\text{O} + 8\text{H}_2\text{S} + \text{S}_8 + 208\text{C}$
17	$\text{C}_3\text{H}_7\text{ClN}_4\text{O}_6 + \text{H}_2\text{O} = 2\text{N}_2 + \text{HCl} + 4\text{H}_2\text{O} + 3\text{CO}$
18	$2\text{C}_3\text{H}_7\text{N}_5\text{O}_9 = 5\text{N}_2 + 7\text{H}_2\text{O} + \text{CO} + 5\text{CO}_2$
19	$4\text{C}_3\text{H}_7\text{N}_7\text{O}_{10} = 14\text{N}_2 + 14\text{H}_2\text{O} + 12\text{CO}_2 + \text{O}_2$
20	$\text{C}_2\text{H}_{14}\text{N}_{12}\text{O}_2 = \text{N}_2 + \text{CH}_4 + \text{H}_2 + \text{H}_2\text{O}$
21	$\text{C}_2\text{H}_{10}\text{N}_{12}\text{O} = 6\text{N}_2 + \text{H}_2\text{O} + 2\text{CH}_4$
22	$2\text{C}_2\text{H}_8\text{N}_{10}\text{O}_3 = 10\text{N}_2 + 6\text{H}_2\text{O} + \text{CH}_4 + 3\text{C}$
23	$\text{C}_3\text{H}_6\text{N}_6\text{O}_5 = 3\text{N}_2 + 3\text{H}_2\text{O} + 2\text{CO} + \text{C}$
24	$2\text{C}_2\text{H}_{10}\text{N}_8\text{O}_5\text{Na}_2 = 8\text{N}_2 + 4\text{NaOH} \cdot \text{H}_2\text{O} + 2\text{H}_2\text{O} + \text{CH}_4 + 3\text{C}$
25	$\text{CH}_6\text{N}_6\text{O}_5 = 3\text{N}_2 + 3\text{H}_2\text{O} + \text{CO}_2$
26	$\text{C}_2\text{Li}_2\text{H}_8\text{N}_8\text{O}_6 = 4\text{N}_2 + 2\text{LiOH} \cdot \text{H}_2\text{O} + \text{CO} + \text{C} + \text{H}_2\text{O}$
27	$\text{H}_4\text{N}_6 = 2\text{H}_2 + 3\text{N}_2$
28	$2\text{H}_5\text{N}_7 = 5\text{H}_2 + 7\text{N}_2$
29	$2\text{C}_2\text{H}_6\text{N}_{10} = 3\text{CH}_4 + \text{C} + 10\text{N}_2$
30	$\text{H}_4\text{N}_6\text{O} + 2\text{H}_2\text{O} = 3\text{H}_2\text{O} + \text{H}_2 + 3\text{N}_2$
31	$\text{C}_2\text{H}_8\text{N}_{10} = 2\text{CH}_4 + 5\text{N}_2$
32	$2\text{C}_2\text{Li}_2\text{H}_8\text{N}_8\text{O}_4 = 4\text{H}_3\text{LiO}_2 + \text{CH}_4 + 3\text{C} + 8\text{N}_2$
33	$\text{C}_2\text{H}_8\text{N}_8\text{Na}_2\text{O}_6 = 2\text{H}_3\text{NaO}_2 + \text{H}_2\text{O} + 4\text{N}_2 + \text{CO} + \text{C}$
34	$2\text{C}_2\text{H}_4\text{K}_2\text{N}_8\text{O}_2 = 4\text{KOH} + 8\text{N}_2 + \text{CH}_4 + 3\text{C}$
35	$2\text{C}_2\text{H}_4\text{N}_8\text{O}_2\text{Rb}_2 = 4\text{RbOH} + 8\text{N}_2 + \text{CH}_4 + 3\text{C}$
36	$\text{C}_6\text{H}_5\text{ClN}_2 = \text{N}_2 + \text{HCl} + \text{CH}_4 + 5\text{C}$
37	$2\text{BaC}_2\text{H}_{10}\text{N}_8\text{O}_5 + 10\text{H}_2\text{O} = 2\text{BaO}_{10}\text{H}_{18} + 8\text{N}_2 + \text{CH}_4 + 3\text{C}$
38	$2\text{C}_4\text{H}_9\text{N}_9\text{O}_5 = 9\text{N}_2 + 9\text{H}_2\text{O} + \text{CO} + 7\text{C}$
39	$\text{C}_6\text{H}_3\text{KN}_8\text{O}_9 = \text{KOH} + 4\text{N}_2 + \text{H}_2\text{O} + \text{CO}_2 + 5\text{CO}$
40	$\text{C}_2\text{H}_8\text{N}_6\text{O}_8 = 3\text{N}_2 + 4\text{H}_2\text{O} + 2\text{CO}_2$
41	$\text{C}_2\text{H}_8\text{N}_{10}\text{O}_4 = 5\text{N}_2 + 4\text{H}_2\text{O} + 2\text{C}$
42	$\text{C}_5\text{H}_8\text{N}_6\text{O}_6 = 3\text{N}_2 + 4\text{H}_2\text{O} + 2\text{CO} + 3\text{C}$

Table S4. Continue.

Salt	Decomposition equation
43	$C_2H_6N_6O_3 = 3N_2 + 3H_2O + 2C$
44	$C_2H_{14}N_{12}O_2 = 6N_2 + 2H_2O + 2CH_4 + H_2$
45	$2ClH_4NO_4 = N_2 + 4H_2O + Cl_2 + O_2$
46	$2C_9H_{10}N_{18}O_4 = 18N_2 + 8H_2O + CH_4 + 17C$
47	$2C_5H_{11}N_{11}O_7 = 11N_2 + 11H_2O + 3CO + 7C$
48	$2C_4H_{16}N_{16}O_5 = 16N_2 + 10H_2O + 3CH_4 + 5C$
49	$C_2H_8N_8O_5 = 4N_2 + 4H_2O + CO + C$
50	$C_6H_8N_{10}O_{11} = 5N_2 + 4H_2O + 5CO + CO_2$
51	$C_8H_{12}N_{14}O_{11} = 7N_2 + 6H_2O + 5CO + 3C$
52	$C_3H_8N_{10}O_5 = 5N_2 + 4H_2O + CO + 2C$
53	$C_4H_8N_{16}O_2 = 8N_2 + 2H_2O + CH_4 + 3C$
54	$C_2H_8N_{14}O_6 = 7N_2 + 4H_2O + 2CO$
55	$2C_4H_9N_{11}O_6 = 11N_2 + 9H_2O + 3CO + 5C$
56	$2LiN_3 = 2Li + 3N_2$
57	$2TlN_3 = 2Tl + 3N_2$
58	$2AgN_3 = 2Ag + 3N_2$
59	$2NH_4Cl = N_2 + 2HCl + 3H_2$
60	$2CH_6NCl = N_2 + 2HCl + 2CH_4 + H_2$
61	$4C_5H_8N_3Cl = 6N_2 + 4HCl + 7CH_4 + 13C$
62	$4C_4H_{14}OPCl = 4H_2O + 4HCl + 11CH_4 + 5C + 4P$
63	$C_2H_6O_2Ca = CaO + H_2O + CH_4 + C$
64	$4FeS_2 = 4Fe + S_8$
65	$C_2N_{12}O_4Rb_2 = 6N_2 + Rb_2O + CO_2 + CO$

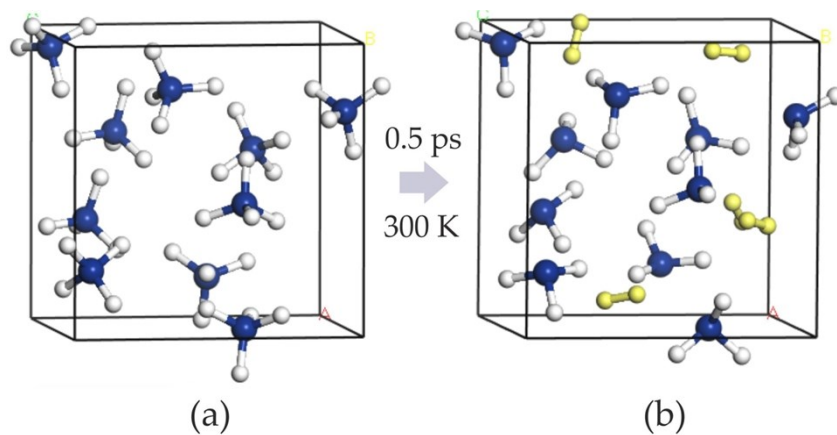


Fig. S3. $NH_4\bullet$ radicals before (a) and after (b) MD simulation at ambient conditions.

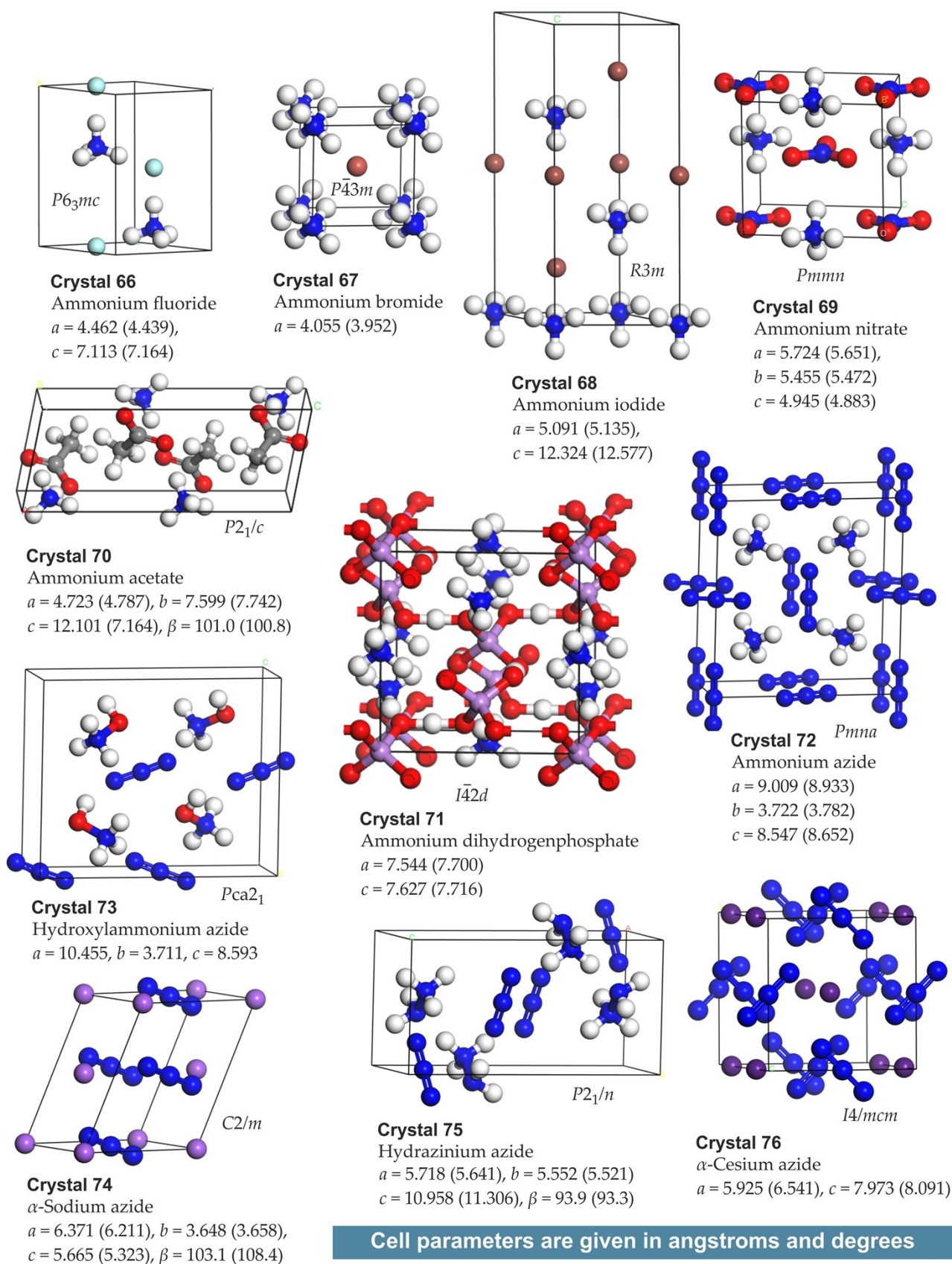


Fig. S4. Crystal structures along with the optimized and experimental (in parentheses) asymmetric cell parameters of salts 66-76.

Table S5. Hirshfeld charges (e^-) of the cation in the NH_4X and XN_3 series at various pressures along with conceptual DFT parameters (eV) of the corresponding radicals ($\text{X}\cdot$) obtained in terms of the adiabatic approximation.

Salt	X	0 GPa	20 GPa	50 GPa	100 GPa	I	A	η	χ	ω
66	F	0.260	0.172	0.110	0.080	17.78	3.51	14.27	10.65	3.97
67	Br	0.248	0.070	-0.033	-0.112	11.96	3.58	8.38	7.77	3.60
68	I	0.278	0.047	-0.057	-0.142	12.14	3.25	8.89	7.70	3.33
69	NO_3	0.370	0.176	0.070	-0.006	13.75	4.03	9.72	8.89	4.07
70	CH_3COO	0.306	0.220	0.174	0.116	11.41	3.15	8.26	7.28	3.21
71	H_2PO_4	0.330	0.140	0.014	-0.020	10.99	4.14	6.85	7.57	4.18
72	NH_4	0.280	0.133	0.053	-0.020	4.41	-1.27	5.68	1.57	0.22
73	NH_3OH	0.230	0.133	0.077	0.047	7.20	-3.76	10.96	1.72	0.13
74	Na	0.268	0.223	0.220	0.233	5.42	0.60	4.82	3.01	0.94
75	NH_3NH_2	0.310	0.173	0.090	0.047	7.20	-3.76	10.96	1.72	0.13
76	Cs	0.300	0.280	0.280	0.290	4.01	0.49	3.53	2.25	0.72

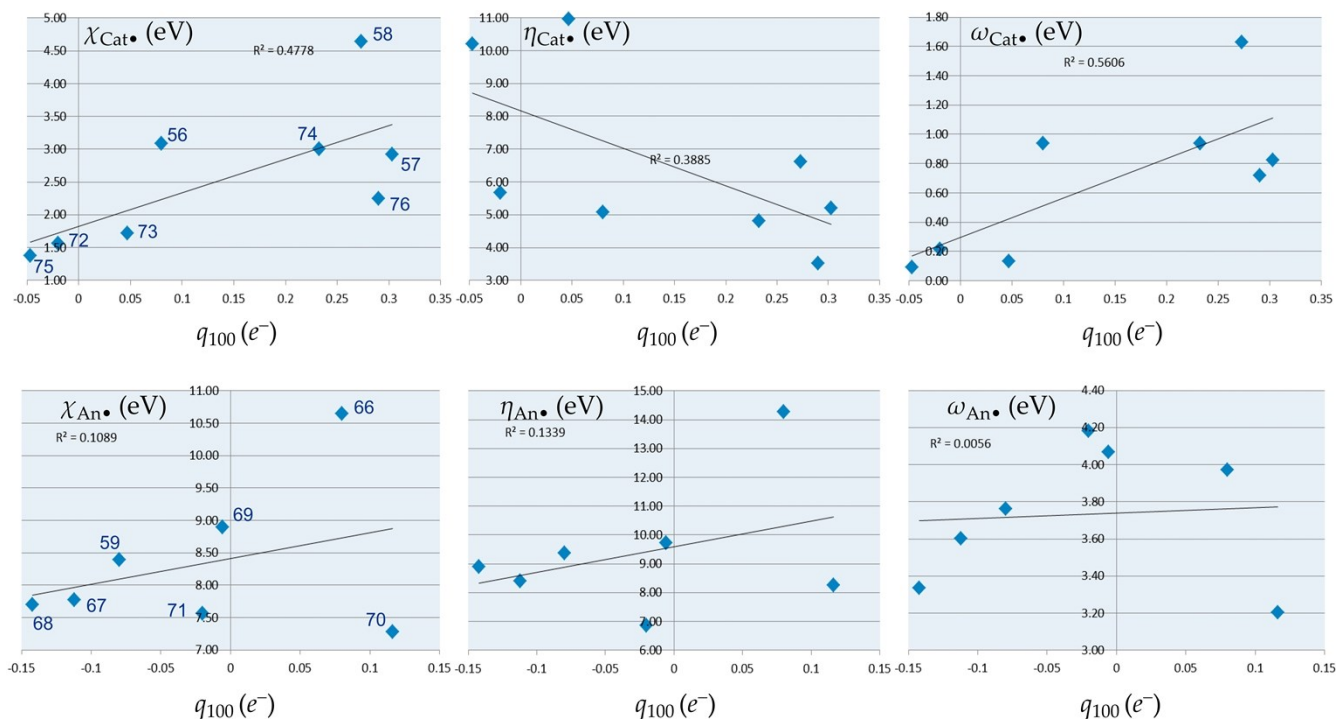


Fig. S5. Dependence of conceptual DFT parameters (χ , η and ω) on cationic charges at 100 GPa (q_{100}) for the NH_4X and XN_3 series salts.

Table S6. The frontier molecular orbital energies and conceptual DFT parameters (in eV) obtained in terms of the vertical approximation.

Salt	E_{HOMO}		E_{LUMO}		η		χ		ω	
	An•	Cat•	An•	Cat•	An•	Cat•	An•	Cat•	An•	Cat•
1	-9.30	-3.51	-6.85	-1.77	-2.44	-1.74	8.08	2.64	-13.34	-2.01
2	-8.89	-2.90	-6.42	-1.60	-2.47	-1.30	7.66	2.25	-11.88	-1.95
3	-1.96	-8.28	0.04	-6.90	-2.00	-1.38	0.96	7.59	-0.23	-20.88
4	3.65	-25.37	5.78	-22.19	-2.13	-3.19	-4.71	23.78	-5.22	-88.76
5	-2.28	-9.87	1.11	-8.07	-3.39	-1.80	0.58	8.97	-0.05	-22.41
6	0.76	-3.51	3.61	-1.77	-2.85	-1.74	-2.19	2.64	-0.84	-2.01
7	-2.83	-9.87	-0.37	-8.07	-2.45	-1.80	1.60	8.97	-0.52	-22.41
8	-2.28	-3.51	1.11	-1.77	-3.39	-1.74	0.58	2.64	-0.05	-2.01
9	-2.46	-3.51	-0.37	-1.77	-2.09	-1.74	1.42	2.64	-0.48	-2.01
10	-2.28	-3.65	1.11	-1.51	-3.39	-2.14	0.58	2.58	-0.05	-1.55
11	-1.58	-9.87	0.96	-8.07	-2.55	-1.80	0.31	8.97	-0.02	-22.41
12	0.46	-3.51	1.38	-1.77	-0.92	-1.74	-0.92	2.64	-0.46	-2.01
13	-7.57	-3.51	-6.77	-1.77	-0.80	-1.74	7.17	2.64	-32.20	-2.01
14	0.33	-10.97	2.26	-9.54	-1.93	-1.43	-1.29	10.25	-0.43	-36.82
15	-10.50	-2.90	-6.97	-1.60	-3.53	-1.30	8.74	2.25	-10.80	-1.95
16	-7.67	-5.28	-6.26	-3.40	-1.41	-1.88	6.96	4.34	-17.15	-5.02
17	-9.30	-6.72	-6.85	-3.52	-2.44	-3.20	8.08	5.12	-13.34	-4.09
18	-10.50	-6.72	-6.97	-3.52	-3.53	-3.20	8.74	5.12	-10.80	-4.09
19	-9.71	-6.72	-6.04	-3.52	-3.67	-3.20	7.87	5.12	-8.45	-4.09
20	-3.03	-6.35	-0.93	-1.11	-2.10	-5.24	1.98	3.73	-0.93	-1.33
21	-3.49	-7.39	-1.37	-2.64	-2.12	-4.75	2.43	5.01	-1.39	-2.65
22	-3.49	-7.78	-1.37	-4.03	-2.12	-3.75	2.43	5.90	-1.39	-4.64
23	-7.57	-6.35	-5.35	-1.11	-2.22	-5.24	6.46	3.73	-9.38	-1.33
24	-3.21	-3.51	-1.44	-1.77	-1.77	-1.74	2.33	2.64	-1.53	-2.01
25	-10.50	-7.40	-6.97	-4.11	-3.53	-3.28	8.74	5.76	-10.80	-5.04
26	-2.81	-3.65	-1.31	-1.51	-1.50	-2.14	2.06	2.58	-1.41	-1.55
27	-7.99	-6.35	-5.43	-1.11	-2.56	-5.24	6.71	3.73	-8.80	-1.33
28	-7.99	-7.39	-5.43	-2.64	-2.56	-4.75	6.71	5.01	-8.80	-2.65
29	-7.99	-3.99	-5.43	-1.38	-2.56	-2.62	6.71	2.68	-8.80	-1.38
30	-7.99	-7.78	-5.43	-4.03	-2.56	-3.75	6.71	5.90	-8.80	-4.64
31	-3.21	-6.35	-1.44	-1.11	-1.77	-5.24	2.33	3.73	-1.53	-1.33
32	-3.21	-3.65	-1.44	-1.51	-1.77	-2.14	2.33	2.58	-1.53	-1.55
33	-2.81	-3.51	-1.31	-1.77	-1.50	-1.74	2.06	2.64	-1.41	-2.01
34	-3.21	-2.90	-1.44	-1.60	-1.77	-1.30	2.33	2.25	-1.53	-1.95
35	-3.21	-2.80	-1.44	-1.60	-1.77	-1.20	2.33	2.20	-1.53	-2.01
36	-9.30	-5.66	-6.85	-2.18	-2.44	-3.48	8.08	3.92	-13.34	-2.21
37	-3.21	-8.28	-1.44	-6.90	-1.77	-1.38	2.33	7.59	-1.53	-20.88
38	-7.57	-3.96	-5.35	-0.91	-2.22	-3.05	6.46	2.44	-9.38	-0.97
39	-8.68	-2.90	-6.89	-1.60	-1.79	-1.30	7.78	2.25	-16.97	-1.95
40	-3.00	-6.35	-1.21	-1.11	-1.79	-5.24	2.11	3.73	-1.24	-1.33
41	-2.81	-7.78	-1.31	-4.03	-1.50	-3.75	2.06	5.90	-1.41	-4.64
42	-9.23	-2.07	-7.07	-0.69	-2.15	-1.38	8.15	1.38	-15.43	-0.69
43	-10.50	-3.99	-6.97	-1.38	-3.53	-2.62	8.74	2.68	-10.80	-1.38
44	-3.16	-9.87	-2.50	-7.20	-0.66	-2.67	2.83	8.54	-6.08	-13.65
45	-10.18	-6.35	-7.25	-1.11	-2.93	-5.24	8.71	3.73	-12.95	-1.33
46	-7.21	-2.81	-4.24	-0.56	-2.97	-2.25	5.73	1.68	-5.52	-0.63
47	-3.40	-8.37	-1.37	-5.37	-2.03	-2.99	2.39	6.87	-1.40	-7.89
48	-3.06	-3.20	-0.89	-0.70	-2.17	-2.50	1.97	1.95	-0.90	-0.76
49	-3.06	-6.35	-0.89	-1.11	-2.17	-5.24	1.97	3.73	-0.90	-1.33
50	-3.87	-6.35	-3.34	-1.11	-0.52	-5.24	3.61	3.73	-12.40	-1.33
51	-3.87	-3.59	-3.34	-0.64	-0.52	-2.95	3.61	2.12	-12.40	-0.76
52	-3.64	-7.78	-1.84	-4.03	-1.81	-3.75	2.74	5.90	-2.08	-4.64
53	-3.53	-4.25	-1.59	-1.60	-1.94	-2.65	2.56	2.92	-1.69	-1.62
54	-3.96	-7.78	-3.00	-4.03	-0.96	-3.75	3.48	5.90	-6.34	-4.64
55	-9.88	-3.43	-7.45	-0.87	-2.42	-2.55	8.66	2.15	-15.48	-0.90
56	-8.49	-3.65	-5.50	-1.51	-2.99	-2.14	6.99	2.58	-8.17	-1.55
57	-8.49	-3.25	-5.50	-2.03	-2.99	-1.22	6.99	2.64	-8.17	-2.86

Table S6. Continue.

Salt	E_{HOMO}		E_{LUMO}		η		χ		ω	
	An•	Cat•	An•	Cat•	An•	Cat•	An•	Cat•	An•	Cat•
58	-8.49	-5.33	-5.50	-3.46	-2.99	-1.87	6.99	4.39	-8.17	-5.16
59	-9.30	-6.35	-6.85	-1.11	-2.44	-5.24	8.08	3.73	-13.34	-1.33
60	-9.30	-2.37	-6.85	-0.98	-2.44	-1.39	8.08	1.68	-13.34	-1.01
61	-9.30	-2.97	-6.85	-0.73	-2.44	-2.24	8.08	1.85	-13.34	-0.77
62	-9.30	-1.85	-6.85	-1.02	-2.44	-0.83	8.08	1.44	-13.34	-1.24
63	-7.52	-9.87	-4.10	-8.07	-3.42	-1.80	5.81	8.97	-4.93	-22.41
64	0.40	-13.48	1.95	-11.64	-1.55	-1.83	-1.17	12.56	-0.44	-43.03
65	-3.96	-2.80	-3.00	-1.60	-0.96	-1.20	3.48	2.20	-6.34	-2.01

Table S7. The frontier molecular orbital energies and conceptual DFT parameters (in eV) obtained in terms of the adiabatic approximation.

Salt	E_{HOMO}		E_{LUMO}		η		χ		ω	
	An•	Cat•	An•	Cat•	An•	Cat•	An•	Cat•	An•	Cat•
1	13.07	5.41	3.71	0.60	9.36	4.81	8.39	3.00	3.76	0.94
2	11.96	4.50	3.58	0.52	8.38	3.98	7.77	2.51	3.60	0.79
3	3.58	10.14	-1.91	5.22	5.49	4.91	0.84	7.68	0.06	6.00
4	-1.39	28.96	-7.37	18.93	5.99	10.03	-4.38	23.95	1.60	28.59
5	5.12	12.09	-3.54	6.15	8.65	5.94	0.79	9.12	0.04	7.00
6	0.57	5.41	-4.64	0.60	5.21	4.81	-2.04	3.00	0.40	0.94
7	5.27	12.09	-1.52	6.15	6.80	5.94	1.88	9.12	0.26	7.00
8	5.12	5.41	-3.54	0.60	8.65	4.81	0.79	3.00	0.04	0.94
9	0.32	5.41	-2.41	0.60	2.73	4.81	-1.05	3.00	0.20	0.94
10	5.12	5.62	-3.54	0.55	8.65	5.07	0.79	3.08	0.04	0.94
11	1.25	12.09	-0.57	6.15	1.82	5.94	0.34	9.12	0.03	7.00
12	-2.12	5.41	-4.06	0.60	1.94	4.81	-3.09	3.00	2.46	0.94
13	6.93	5.41	5.28	0.60	1.66	4.81	6.11	3.00	11.26	0.94
14	2.20	13.83	-4.04	7.01	6.23	6.82	-0.92	10.42	0.07	7.96
15	13.75	4.50	4.03	0.52	9.72	3.98	8.89	2.51	4.07	0.79
16	8.31	7.01	4.33	1.79	3.98	5.22	6.32	4.40	5.02	1.86
17	13.07	7.05	3.71	4.26	9.36	2.80	8.39	5.66	3.76	5.72
18	13.75	7.05	4.03	4.26	9.72	2.80	8.89	5.66	4.07	5.72
19	11.68	7.05	4.66	4.26	7.03	2.80	8.17	5.66	4.75	5.72
20	4.58	4.40	-0.33	-1.27	4.91	5.68	2.12	1.57	0.46	0.22
21	5.09	6.45	-0.25	-3.71	5.34	10.16	2.42	1.37	0.55	0.09
22	5.09	7.20	-0.25	-3.76	5.34	10.96	2.42	1.72	0.55	0.13
23	9.22	4.40	3.89	-1.27	5.33	5.68	6.55	1.57	4.03	0.22
24	4.62	5.41	-0.08	0.60	4.70	4.81	2.27	3.00	0.55	0.94
25	13.75	6.54	4.03	3.75	9.72	2.79	8.89	5.14	4.07	4.74
26	5.25	5.62	-0.15	0.55	5.40	5.07	2.55	3.08	0.60	0.94
27	10.55	4.40	2.39	-1.27	8.16	5.68	6.47	1.57	2.56	0.22
28	10.55	6.45	2.39	-3.71	8.16	10.16	6.47	1.37	2.56	0.09
29	10.55	4.41	2.39	0.28	8.16	4.14	6.47	2.35	2.56	0.67
30	10.55	7.20	2.39	-3.76	8.16	10.96	6.47	1.72	2.56	0.13
31	4.62	4.40	-0.08	-1.27	4.70	5.68	2.27	1.57	0.55	0.22
32	4.62	5.62	-0.08	0.55	4.70	5.07	2.27	3.08	0.55	0.94
33	5.25	5.41	-0.15	0.60	5.40	4.81	2.55	3.00	0.60	0.94
34	4.62	4.50	-0.08	0.52	4.70	3.98	2.27	2.51	0.55	0.79
35	4.62	4.34	-0.08	0.49	4.70	3.85	2.27	2.41	0.55	0.76
36	13.07	6.78	3.71	1.10	9.36	5.68	8.39	3.94	3.76	1.37
37	4.62	10.14	-0.08	5.22	4.70	4.91	2.27	7.68	0.55	6.00
38	9.22	3.94	3.89	0.17	5.33	3.77	6.55	2.06	4.03	0.56
39	10.04	4.50	5.60	0.52	4.45	3.98	7.82	2.51	6.88	0.79
40	4.20	4.40	0.09	-1.27	4.11	5.68	2.14	1.57	0.56	0.22
41	5.25	7.20	-0.15	-3.76	5.40	10.96	2.55	1.72	0.60	0.13
42	10.80	3.75	5.59	-0.75	5.20	4.51	8.20	1.50	6.46	0.25
43	13.75	4.41	4.03	0.28	9.72	4.14	8.89	2.35	4.07	0.67
44	5.50	-2.24	1.84	7.80	3.66	-10.03	3.67	2.78	1.84	-0.38

Table S7. Continue.

Salt	E_{HOMO}		E_{LUMO}		η		χ		ω	
	An•	Cat•	An•	Cat•	An•	Cat•	An•	Cat•	An•	Cat•
45	12.34	4.40	5.29	-1.27	7.05	5.68	8.82	1.57	5.52	0.22
46	8.18	4.00	5.54	0.31	2.64	3.69	6.86	2.16	8.92	0.63
47	4.70	9.22	0.17	4.54	4.53	4.68	2.44	6.88	0.66	5.06
48	4.76	3.72	0.26	-0.41	4.50	4.12	2.51	1.65	0.70	0.33
49	4.76	4.40	0.26	-1.27	4.50	5.68	2.51	1.57	0.70	0.22
50	5.92	4.40	2.31	-1.27	3.61	5.68	4.12	1.57	2.35	0.22
51	5.92	4.25	2.31	-0.44	3.61	4.70	4.12	1.90	2.35	0.39
52	5.38	7.20	0.71	-3.76	4.66	10.96	3.04	1.72	0.99	0.13
53	4.81	4.25	0.21	0.54	4.60	3.71	2.51	2.40	0.69	0.77
54	6.10	7.20	1.80	-3.76	4.30	10.96	3.95	1.72	1.81	0.13
55	11.55	3.71	5.89	-0.10	5.65	3.81	8.72	1.80	6.72	0.43
56	12.78	5.62	2.71	0.55	10.06	5.07	7.75	3.08	2.98	0.94
57	12.78	5.51	2.71	0.35	10.06	5.16	7.75	2.93	2.98	0.83
58	12.78	7.95	2.71	1.32	10.06	6.63	7.75	4.64	2.98	1.62
59	13.07	4.40	3.71	-1.27	9.36	5.68	8.39	1.57	3.76	0.22
60	13.07	4.14	3.71	-0.55	9.36	4.69	8.39	1.79	3.76	0.34
61	13.07	4.37	3.71	-0.67	9.36	5.04	8.39	1.85	3.76	0.34
62	13.07	3.05	3.71	-0.10	9.36	3.16	8.39	1.47	3.76	0.34
63	7.36	12.09	1.42	6.15	5.94	5.94	4.39	9.12	1.62	7.00
64	1.78	16.49	-3.67	7.44	5.45	9.05	-0.95	11.96	0.08	7.91
65	6.10	4.34	1.80	0.49	4.30	3.85	3.95	2.41	1.81	0.76

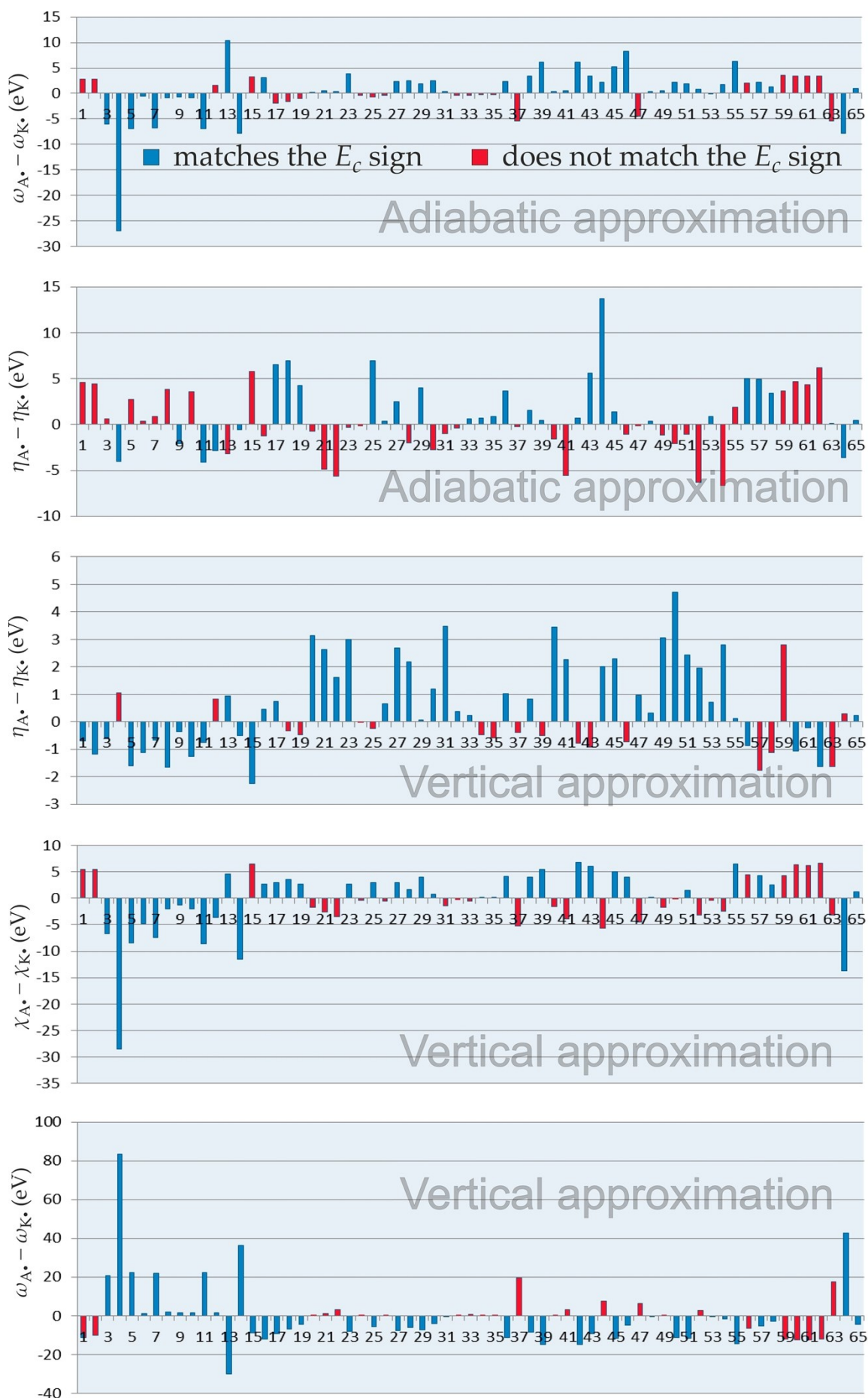


Fig. S6. Efficiency of various conceptual DFT parameters in distinguishing of compounds with the positive and negative energy content.