Supplementary Information for

Improvement of NOx uptake/release over Pd/Beta by propylene: shielding effect

of intermediates on adsorbed NOx species

Chao Liu,^a Jun Wang,^a Zexiang Chen,^a Jianqiang Wang^a and Meiqing Shen*abc

- a. Key Laboratory for Green Chemical Technology of State Education Ministry, School of Chemical Engineering & Technology, Tianjin University, Tianjin 300072, P. R. China.
- b. State Key Laboratory of Engines, Tianjin University, Tianjin 300072, P. R. China.
- c. Collaborative Innovation Centre of Chemical Science and Engineering (Tianjin), Tianjin 300072, P. R. China.

*To Whom Correspondence should be addressed E-mail: mqshen@tju.edu.cn Phone: (+86) 22-27407002

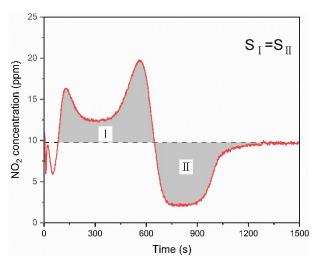


Fig. S1 NO₂ concentration curves of Pd/Beta during adsorption and the initial stage of desorption for a feed of 200 ppm NOx, 5%

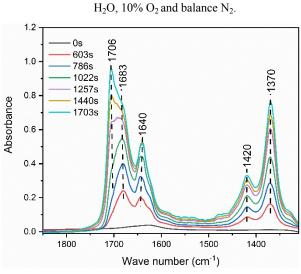


Fig. S2 In-situ FTIR spectra of Pd/Beta in a feed 400 ppm acetone and balance $N_{\rm 2}.$

To verify the peak assignments, we conducted an in-situ FTIR experiment under a flow of acetone and N₂, and the result was shown in Fig. S2. Two overlapping bands at 1683 cm⁻¹ and 1706 cm⁻¹ are assigned to C=O stretching mode of acetone molecules adsorbed on Pd species and silanol groups, respectively. There is no peak at 1706 cm⁻¹ in Fig. 5b due to the shielding effect of water on silanol. In addition, the peak at 1623 cm⁻¹ in Fig. 5b overlapped with the band at 1640 cm⁻¹.

Table S1 The comparison of the experimental and calculated frequency of C=O in Pd-NC₃H₆O and acetone

	Experimental frequency of C=O	Calculated frequency of C=O
C=O in Pd-NC ₃ H ₆ O	1746 cm ⁻¹	1593 cm ⁻¹
C=O in acetone	1692 cm ⁻¹	1554 cm ⁻¹
Difference	54 cm ⁻¹	39 cm ⁻¹