

Oxygen-substituted borophene as a potential anode material for Li/Na-ion batteries: A first principles study

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Supplementary information:

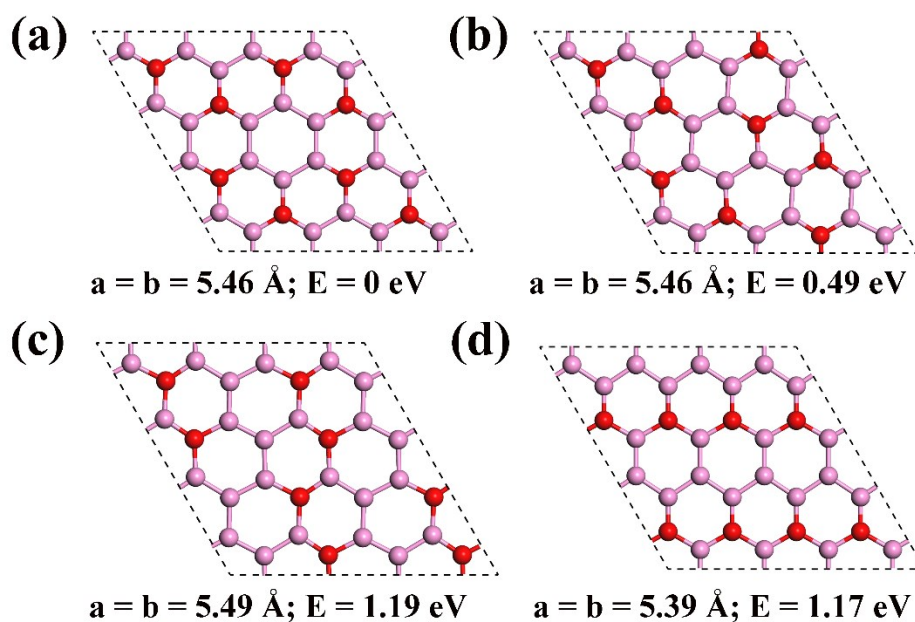


Figure S1 Optimized low-energy geometries of HBS oxides. The corresponding lattice constants and energies are labeled down below.

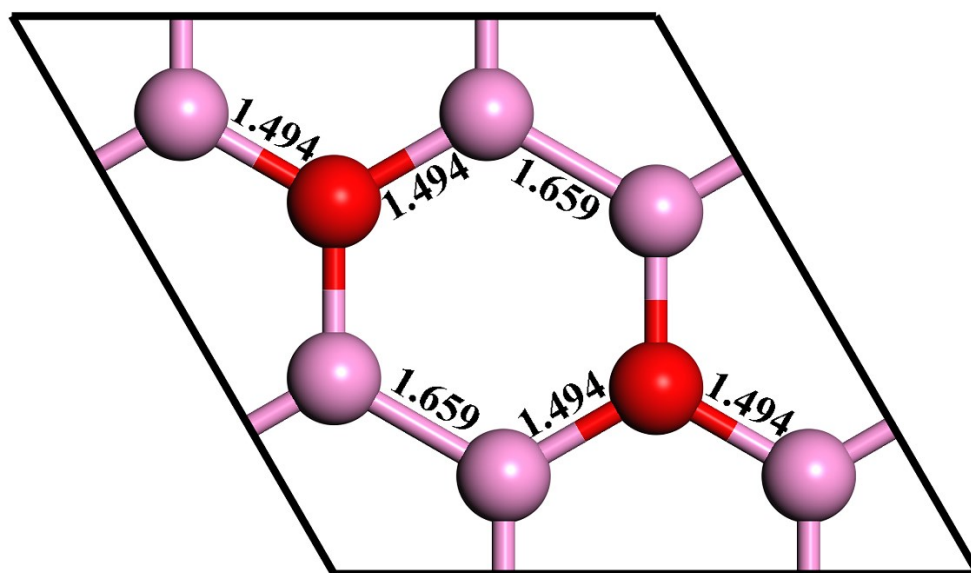


Figure S2 Optimized unit structure of *h*-B₃O monolayer, bond lengths are in angstroms (Å).

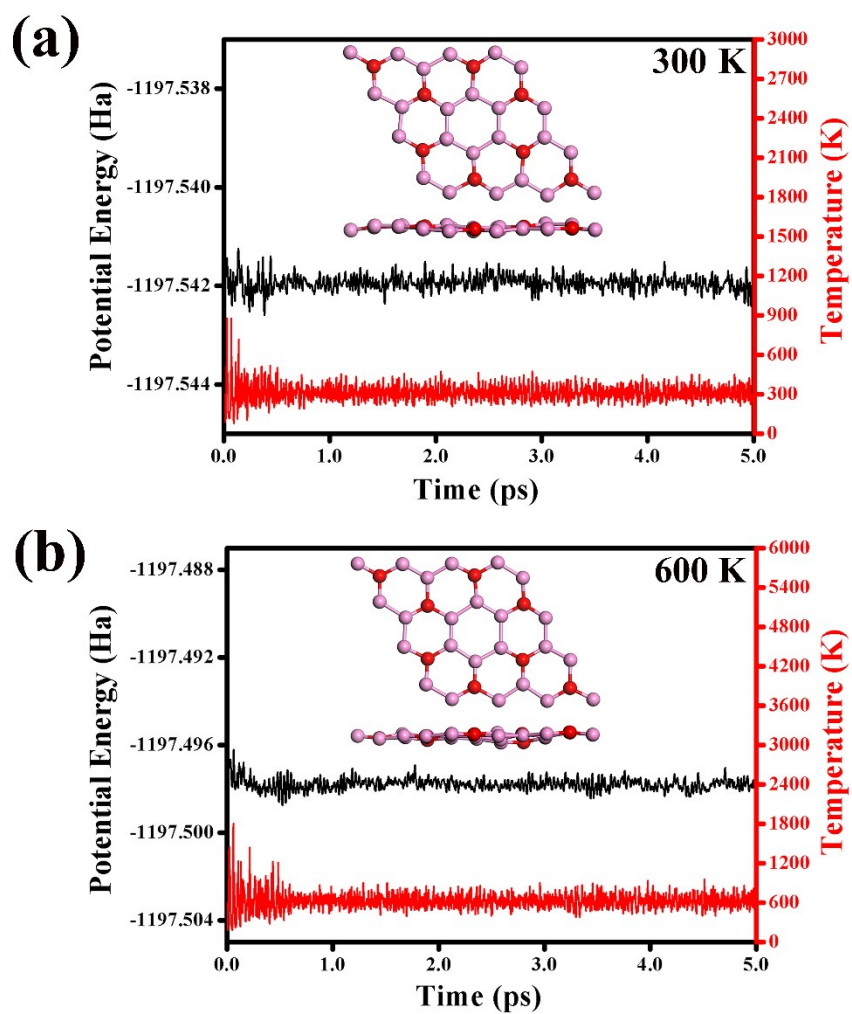


Figure S3 Variation of the potential energies and the temperatures of h -B₃O monolayer in molecular dynamics (MD) simulations at (a) 300 K and (b) 600 K, the corresponding snapshots of the final frame of h -B₃O monolayer after lasting for 5.0 ps.

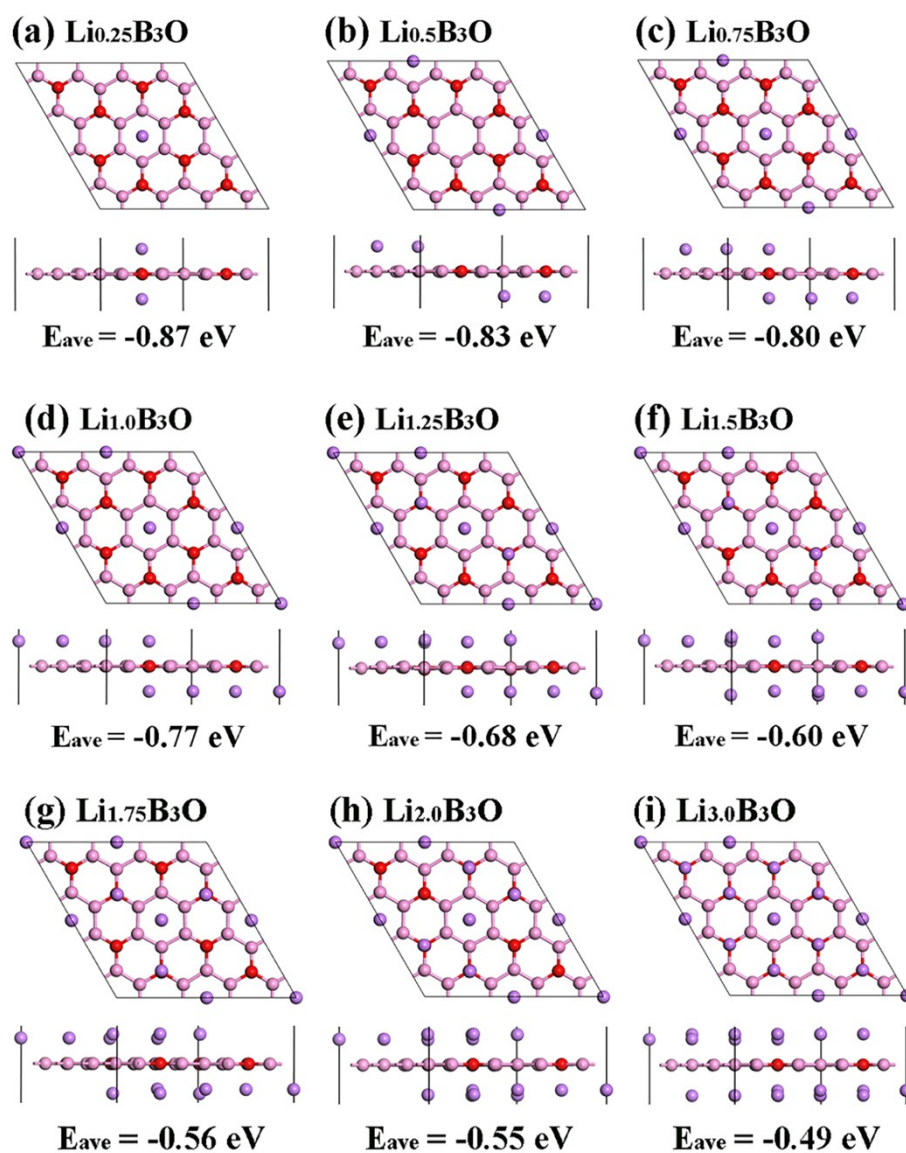


Figure S4 Top and side views of the optimized structures of (a) $\text{Li}_{0.25}\text{B}_3\text{O}$; (b) $\text{Li}_{0.5}\text{B}_3\text{O}$; (c) $\text{Li}_{0.75}\text{B}_3\text{O}$; (d) $\text{Li}_{1.0}\text{B}_3\text{O}$; (e) $\text{Li}_{1.25}\text{B}_3\text{O}$; (f) $\text{Li}_{1.5}\text{B}_3\text{O}$; (g) $\text{Li}_{1.75}\text{B}_3\text{O}$; (h) $\text{Li}_{2.0}\text{B}_3\text{O}$ and (i) $\text{Li}_{3.0}\text{B}_3\text{O}$ with Li atoms adsorbed on the surface of $h\text{-B}_3\text{O}$ monolayer. The B, O and Li atoms are denoted by pink, red and purple balls, respectively.

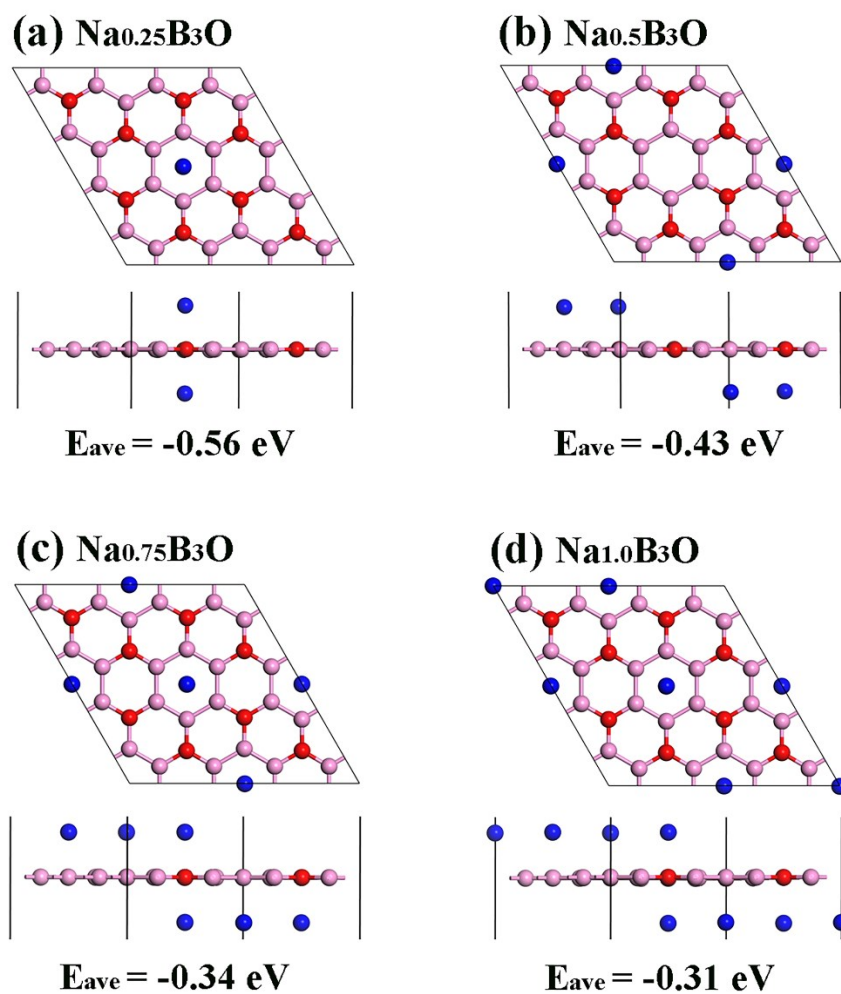


Figure S5 Top and side views of the optimized structures of (a) $\text{Na}_{0.25}\text{B}_3\text{O}$; (b) $\text{Na}_{0.5}\text{B}_3\text{O}$; (c) $\text{Na}_{0.75}\text{B}_3\text{O}$ and (d) $\text{Na}_{1.0}\text{B}_3\text{O}$ with Na atoms adsorbed on the surface of $h\text{-B}_3\text{O}$ monolayer. The B, O and Na atoms are denoted by pink, red and blue balls, respectively.

Table S1 The adsorption energies, initial sites, final sites and adsorption heights for Li/Na atom adsorbed on *h*-B₃O monolayer surface.

Metal atoms	Initial sites	Final sites	Adsorption energies (eV)	Adsorption heights (Å)
Li	H1	H1	-2.33	1.818
	H2	H2	-1.86	1.712
	T1	H1	-2.33	1.809
	T2	H2	-1.86	1.701
	B1	H1	-2.33	1.792
	B2	H2	-1.86	1.717
Na	H1	H1	-1.70	2.311
	H2	H2	-1.33	2.243
	T1	H1	-1.70	2.293
	T2	T2	-1.27	2.077
	B1	H2	-1.33	2.250
	B2	T2	-1.27	2.100