

Supplementary material

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Supplementary material for Nitroethane at high density: an experimental and computational vibrational study

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Summary

The supplementary material section of this publication contains figures showing the comparison between DFT calculations and experimental results, additional pressure induced vibrational mode progressions and vibrational band shapes.

This section also incorporates the tabulated summary of all vibron modes of the free nitroethane (NE) molecule. A video animation file for the free NE molecule (in Gaussian output format¹) has also been included.

Finally, video animations of the molecular motions within the monoclinic unit cell, as predicted by DFT calculations, for pressures ranging between 1.5 and 16 GPa are also included electronically in this section. The output format for these animations is compatible with the Molden² software.

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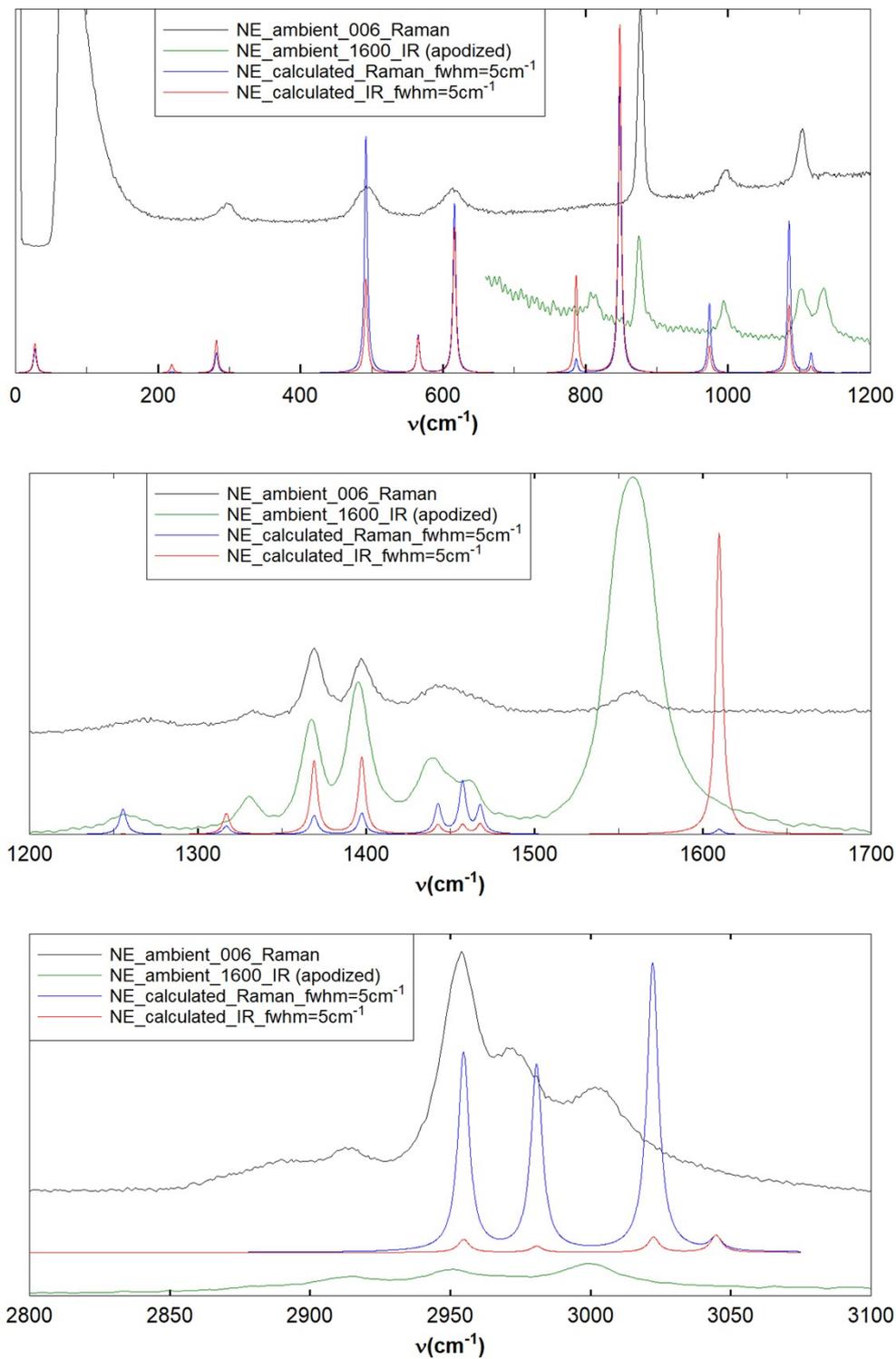


Figure S-1: Calculated vibrational spectra for NE free molecule compared with experimental Raman and IR vibrational spectra of ambient liquid NE. Calculated mode wavenumbers were scaled by 0.961.

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Table S-1: Vibrational modes of nitroethane. Vibrational wavenumbers (in cm⁻¹) of liquid nitroethane under ambient temperature and pressure observed in this study compared to that from previous studies. Predicted mode wavenumbers and assignments are based on DFT calculations as described in the text. Each mode assignment typically results from vibrations in more than one bond; strongest contributions (stretching and/or angular deformation) appear first in bold based on relative magnitude of displacements. Wavenumbers without assignment are due to harmonics or combination bands.

Smith ³			Geiseler ⁴			Courtecuisse ⁵			This work			
IR	IR	Raman	IR	Raman		Raman		IR	Raman	DFT calculation (B3LYP, 6-31G**)		
Vapor	Liq.	Liq.	Liq.	Liq.	Assignment	Liq.	Assignment	Liq.	Liq.	Single molecule	Assignment	Mode #
										27.1	τ skeletal	1
										219.3	τ+v_s CH₃ , τ skeletal	2
		295		298	-			297		281.9	δ C-C-N , ρ NO₂ , ω CH ₂ , ρ(i.pl.) CH ₃	3
	~495	491	493	494	ρ NO₂	494	ρ NO₂	492		491.8	ρ+δ NO₂ , δ C-C-N , ρ(i.pl.) CH ₃ , ω CH ₂ , ν C-N, ν C-C, ν N-O	4
						614	ω NO₂	592		565.1	ω+v_s+δ NO₂ , ρ+τ+v_s CH₂ , τ skeletal	5
~621	616	614	614	615	δ(sym)+ρ NO₂	618	δ(sym) NO₂	614		616.0	δ+v_s NO₂ , δ O-N-C , δ(i.pl.)+v_s CH₃ , δ C-C-N , ν C-N, ρ(i.pl.) CH ₃ , ω+δ CH ₂	6
807	810		810		?			811	811*	786.9	ρ+τ+v_s CH₂ , ρ(o.pl.) CH₃ , ω NO₂ , δ(o.pl.)+v _s CH ₃ , τ skeletal	7
866												
876	875	874.7	878	878	ν C-N	878	ν C-N	875	877	848.2	δ+v_s NO₂ , ν C-N , ρ(i.pl.) CH₃ , δ(i.pl.)+v _s CH ₃ , ω+δ CH ₂ , ν C-C, δ O-N-C	8
884												
985												
996	993	993	992	996	ν C-C			995	997	973.9	ρ(i.pl.) CH₃ , ω+δ CH₂ , ν C-C , ν C-N , v _s +δ NO ₂ , δ(i.pl.)+v _s CH ₃ , δ C-C-N, δ O-N-C	9
1005												
1103	1103	1101.6	1101	1106	ρ(o.pl.) CH₃	1095–995	ρ CH₃	1103	1104	1085.6	ρ(i.pl.) CH₃ , ν C-C , δ C-C-N , δ(i.pl.)+v _s CH ₃ , ρ+v _s NO ₂ , ω CH ₂ , ν C-N	10
1118												
~1124	1134		1131		ρ(i.pl.) CH₃			1134	1138	1116.6	ρ+τ+v_s CH₂ , ρ(o.pl.) CH₃ , δ(o.pl.)+v _s CH ₃ , ω NO ₂ , τ skeletal	11
~1150												
~1260	1255	1267	1253	1269	τ CH₂			1257	1265	1255.5	τ+v_s CH₂ , ρ(o.pl.) CH₃ , δ(o.pl.)+v _s CH ₃ , ω NO ₂ , τ skeletal	12
~1326	1330	1331	1330	1333	ω CH₂			1330	1332	1316.9	ρ(i.pl.) CH₃ , ν C-N , ν C-C , ν N-O, δ O-N-C, δ NO ₂	13
~1352												

Table continues on next page.

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Smith ³			Geiseler ⁴			Courtecuisse ⁵			This work					
IR	IR	Raman	IR	Raman		Raman		IR	Raman	DFT calculation (B3LYP, 6-31G**)				
Vapor	Liq.	Liq.	Liq.	Liq.	Assignment	Liq.	Assignment	Liq.	Liq.	Single molecule	Assignment	Mode #		
1366	1365	1366.5	1368	1370	$\nu_s \text{NO}_2$	1395–1368	$\nu_s \text{NO}_2$	1367	1369	1369.0	$\delta(\text{umbrella, i.pl.})+\nu_s \text{CH}_3$, $\nu_s+\delta \text{NO}_2$, $\nu \text{C-N}$, $\delta+\omega \text{CH}_2$, $\delta \text{O-N-C}$, $\nu \text{C-C}$	14		
1379														
1397	1395	1395	1395	1399	$\delta(\text{sym}) \text{CH}_3$			1395	1398	1397.4	$\delta(\text{umbrella, i.pl.})+\nu_s \text{CH}_3$, $\nu \text{C-C}$, $\nu \text{C-N}$, $\omega+\delta \text{CH}_2$, $\nu_s+\delta \text{NO}_2$, $\delta \text{O-N-C}$	15		
1406														
~1440	1441	1440	1440	1440	$\delta(\text{sym}) \text{CH}_2$	1440	$\delta(\text{sym}) \text{CH}_3$	1439	1444	1442.6	$\delta+\nu_s \text{CH}_2$, $\nu \text{C-C}$, $\delta(\text{i.pl.}) \text{CH}_3$, $\nu_s+\delta \text{NO}_2$, $\delta \text{C-C-N}$, $\delta \text{O-N-C}$	16		
1458	~1452	1456	1450		?			1452	1453	1457.3	$\delta(\text{o.pl.})+\nu_s \text{CH}_3$, τ skeletal	17		
~1480	~1465		1463	1463	$\delta \text{CH}_3+\text{CH}_2$			1463	1465	1467.7	$\delta(\text{i.pl.})+\nu_s \text{CH}_3$, $\delta+\omega \text{CH}_2$, $\nu \text{C-N}$, $\nu \text{C-C}$	18		
1573	1560	1554	1559	1559	$\nu_a \text{NO}_2$	1555	$\nu_a \text{NO}_2$	1558	1558	1609.7	$\nu_a+\delta \text{NO}_2$, $\delta \text{O-N-C}$, $\delta \text{C-C-N}$, $\omega+\delta \text{CH}_2$, $\delta(\text{umbrella, i.pl.}) \text{CH}_3$, $\nu \text{C-N}$, $\nu \text{C-C}$	19		
1588														
1862														
1946														
1976														
2075														
2128														
2208														
2247														
2315														
2358														
~2435	2420													
~2470														
2560														
2650														
2652														
~2750	2750	2749												
2750														
2890 2880 2889 -														
2882 2889														
2912 2913 2913														
2950	2950	2951	2903	2944	$\nu_s \text{CH}_3+\text{CH}_2$	2950	$\nu_s \text{CH}_3$	2950	2953	2954.7	$\nu_s+\delta(\text{i.pl.}) \text{CH}_3$, $\nu \text{C-C}$, $\delta \text{C-C-N}$, $\omega+\nu_s \text{CH}_2$	20		
2970 2973 2980.7														
$\nu_s+\delta \text{CH}_2$, $\nu_a(\text{i.pl.}) \text{CH}_3$, $\delta \text{C-C-N}$, $\nu \text{C-C}$														
2971 2938 2971 $\nu_a \text{CH}_2$														
3001 3022.0 $\nu_a \text{CH}_2$, $\nu_a(\text{o.pl.}) \text{CH}_3$, τ skeletal														
3010	3000	3001	2990	3013	$\nu_a+\nu_s \text{CH}_3$					3000	3004	3022.4	$\nu_a(\text{i.pl.}) \text{CH}_3$, $\nu_s \text{CH}_2$, $\delta \text{C-C-N}$, $\delta \text{O-N-C}$	23
3030 3044.7 $\nu_a(\text{o.pl.}) \text{CH}_3$, $\nu_a \text{CH}_2$, τ skeletal														
24														
~3225														
3600														
3680														
4100														
4270														
4430														

τ , torsion; δ , bending/scissoring; ρ , rocking; ω , wagging; ν , stretching (ν_s , symmetric stretching; ν_a , asymmetric stretching)

i.pl., deformation is symmetric with, or in the C-C-N-O₂ plane of symmetry

o.pl., deformation is asymmetric with, or out of the C-C-N-O₂ plane of symmetry

* Weak, peak requires fitting before baseline removal

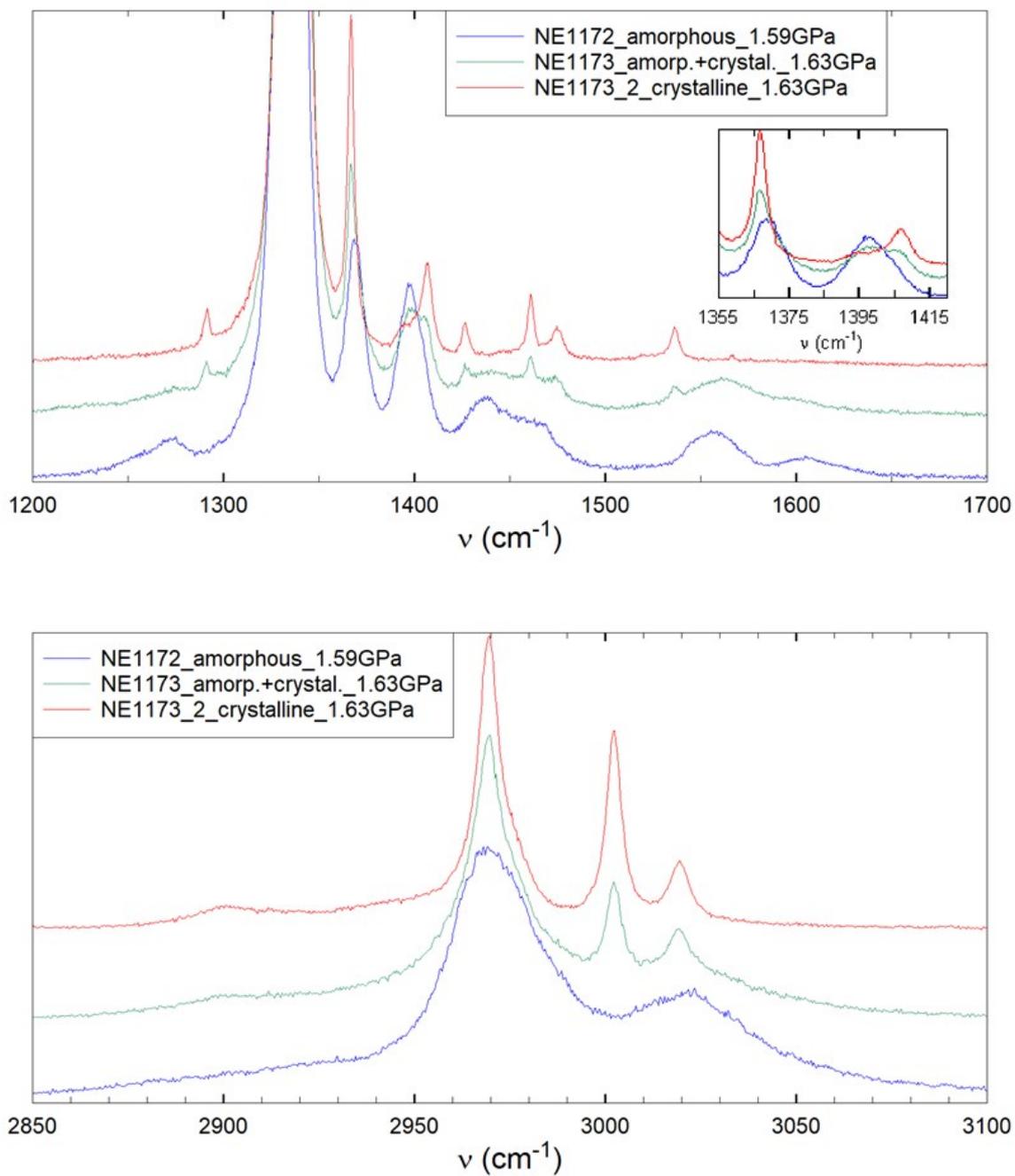


Figure S-2: Raman vibrational spectra of NE at ~1.6 GPa in the high fingerprint (top for $\nu \lesssim 1500\text{cm}^{-1}$) and functional group (top for $\nu \gtrsim 1500\text{cm}^{-1}$ and bottom) regions.

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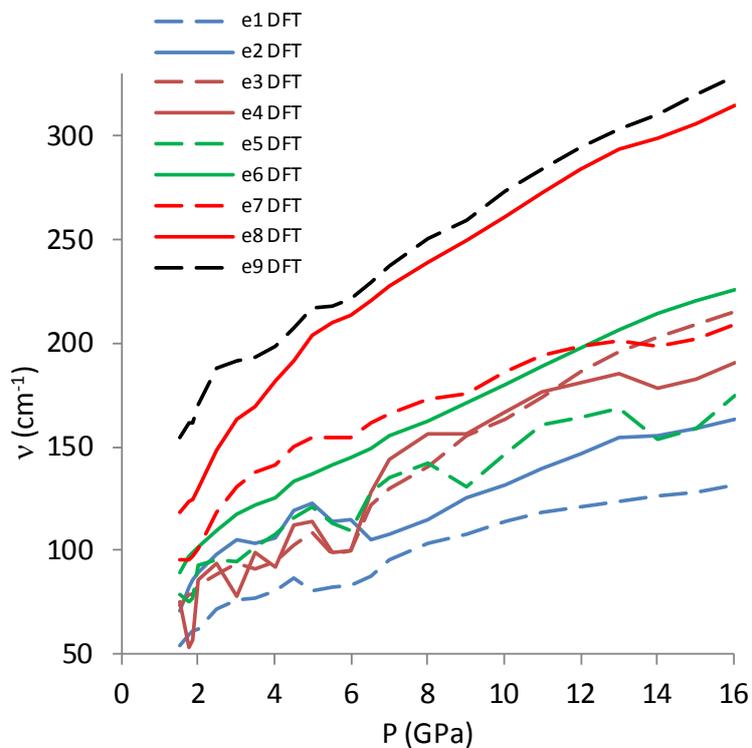


Figure S-3: Calculated pressure-induced vibrational mode shifts for external modes of nitroethane in the NE crystalline phase.

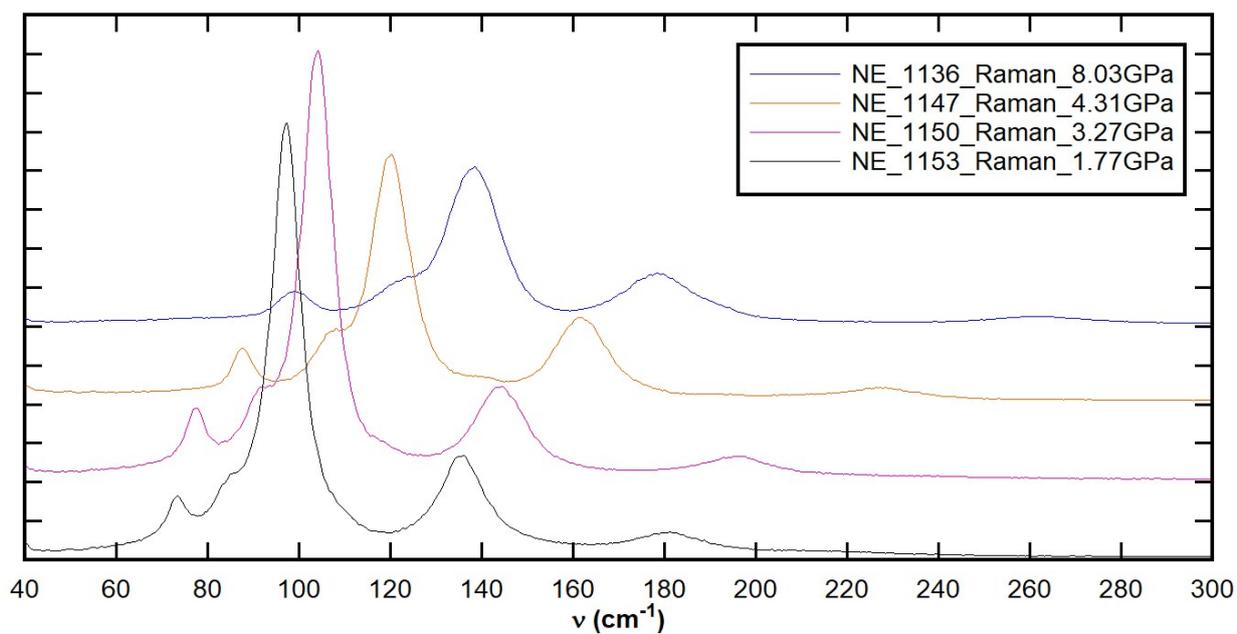


Figure S-4: Observed Raman vibrational band shape in the low wavenumber range at various pressures.

Supplementary material

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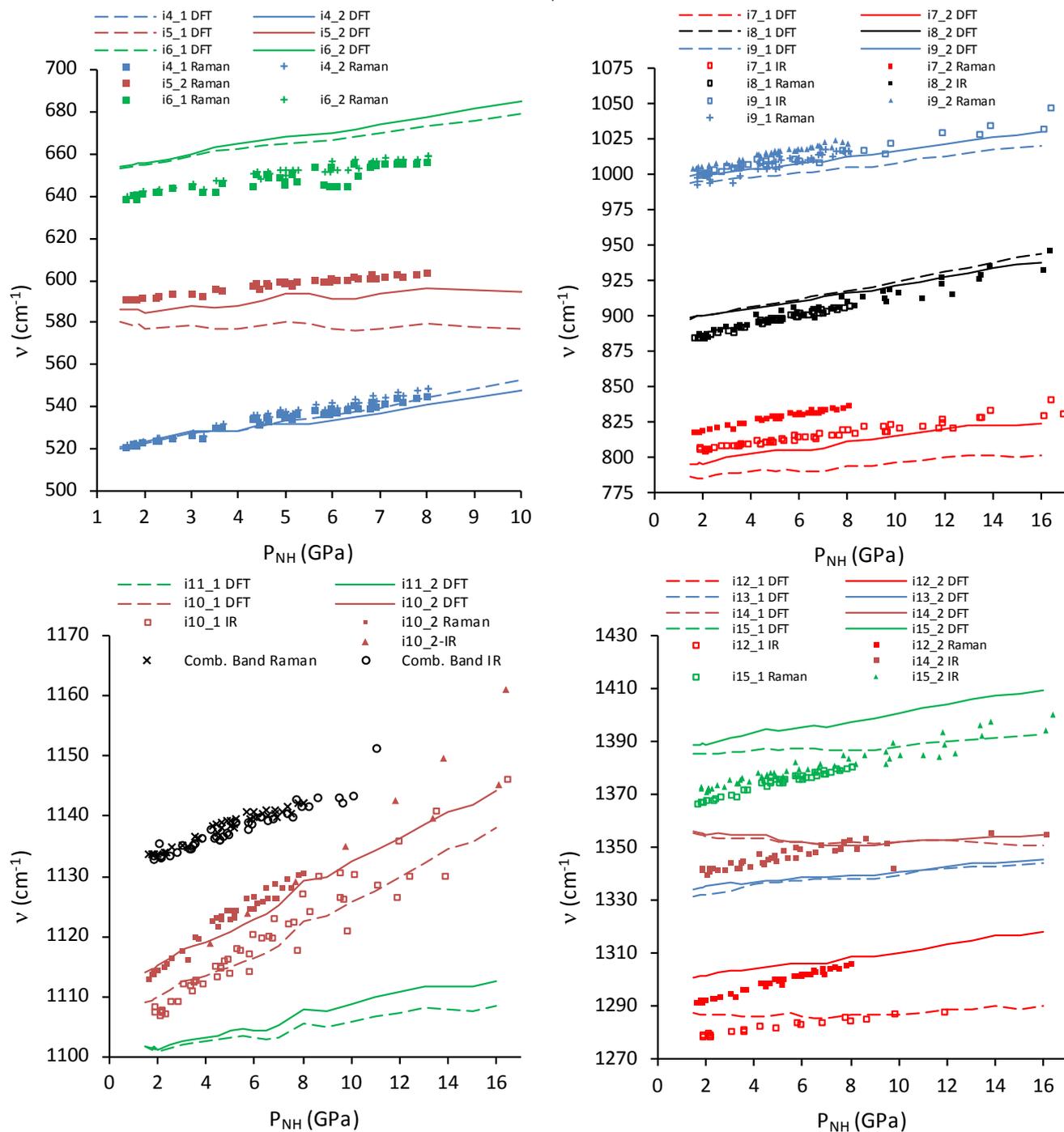


Figure S-5: Vibrational mode progressions as a function of pressure for assigned modes in the fingerprint region. Calculated (DFT) mode progressions for both Davydov components have been scaled as described in the text. Prefixes “e” and “i” denote external and internal modes, respectively. Suffixes “_1” and “_2” denote each Davydov components.

Supplementary material

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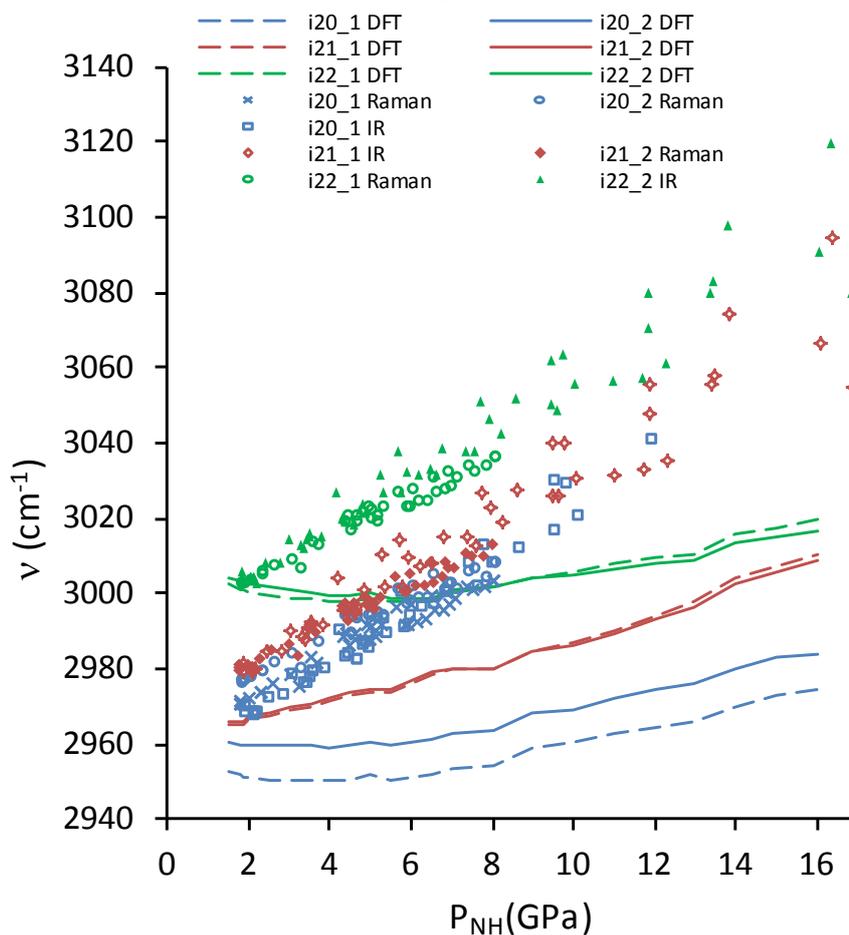


Figure S-6: Vibrational mode progression as a function of pressure for modes i23 and i24 in the functional group region. Calculated (DFT) mode progressions for both Davydov components have been scaled as described in the text. Prefixes “e” and “i” denote external and internal modes, respectively. Suffixes “_1” and “_2” denote each Davydov components.

Supplementary material

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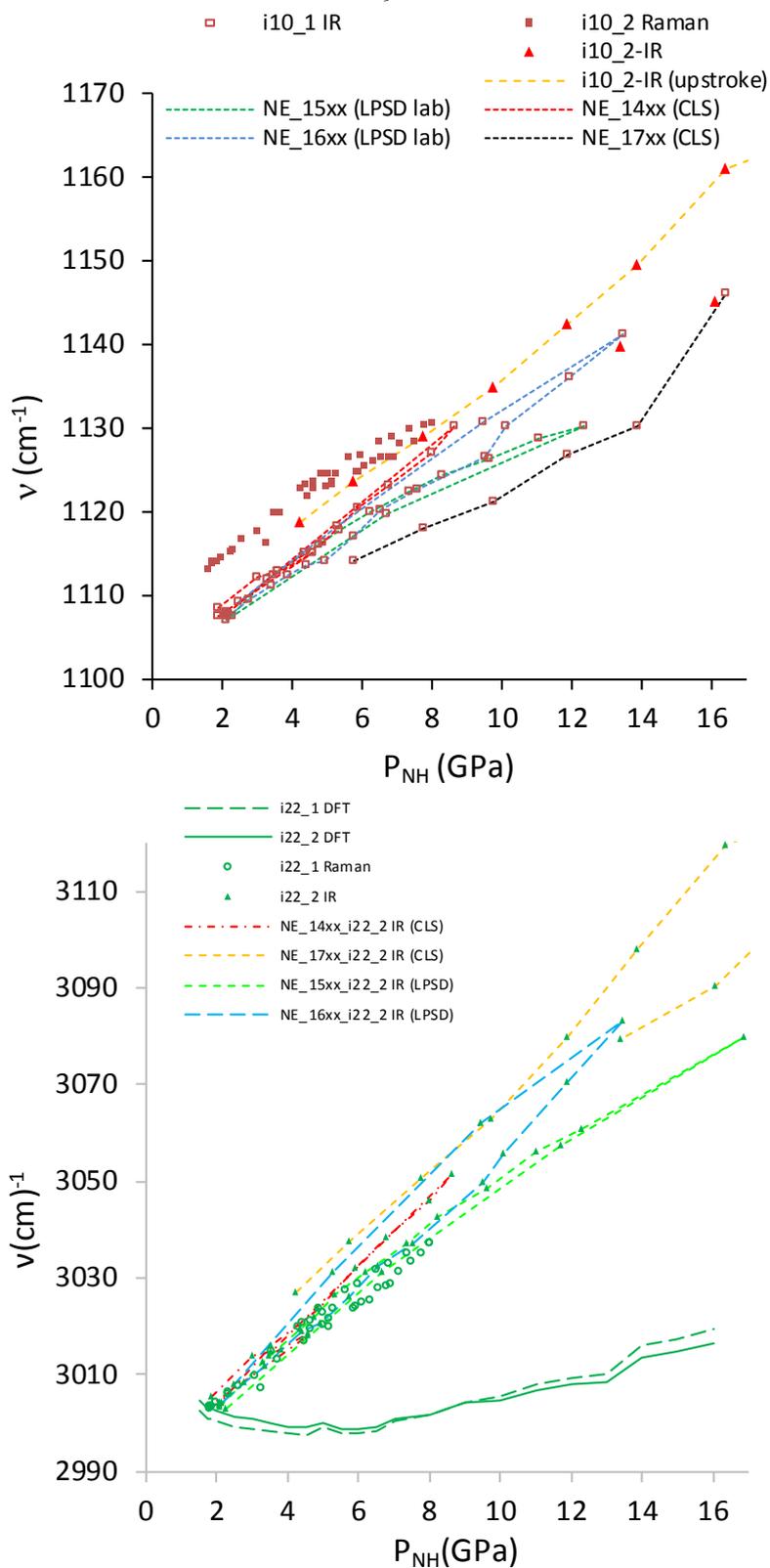


Figure S-7: Pressure induced evolutions of modes i10 (top) and i22 (bottom). Dotted lines highlight four different IR sample loading. Subscript “NH” denotes the use of a non-hydrostatic ruby pressure standard, as described in the text.

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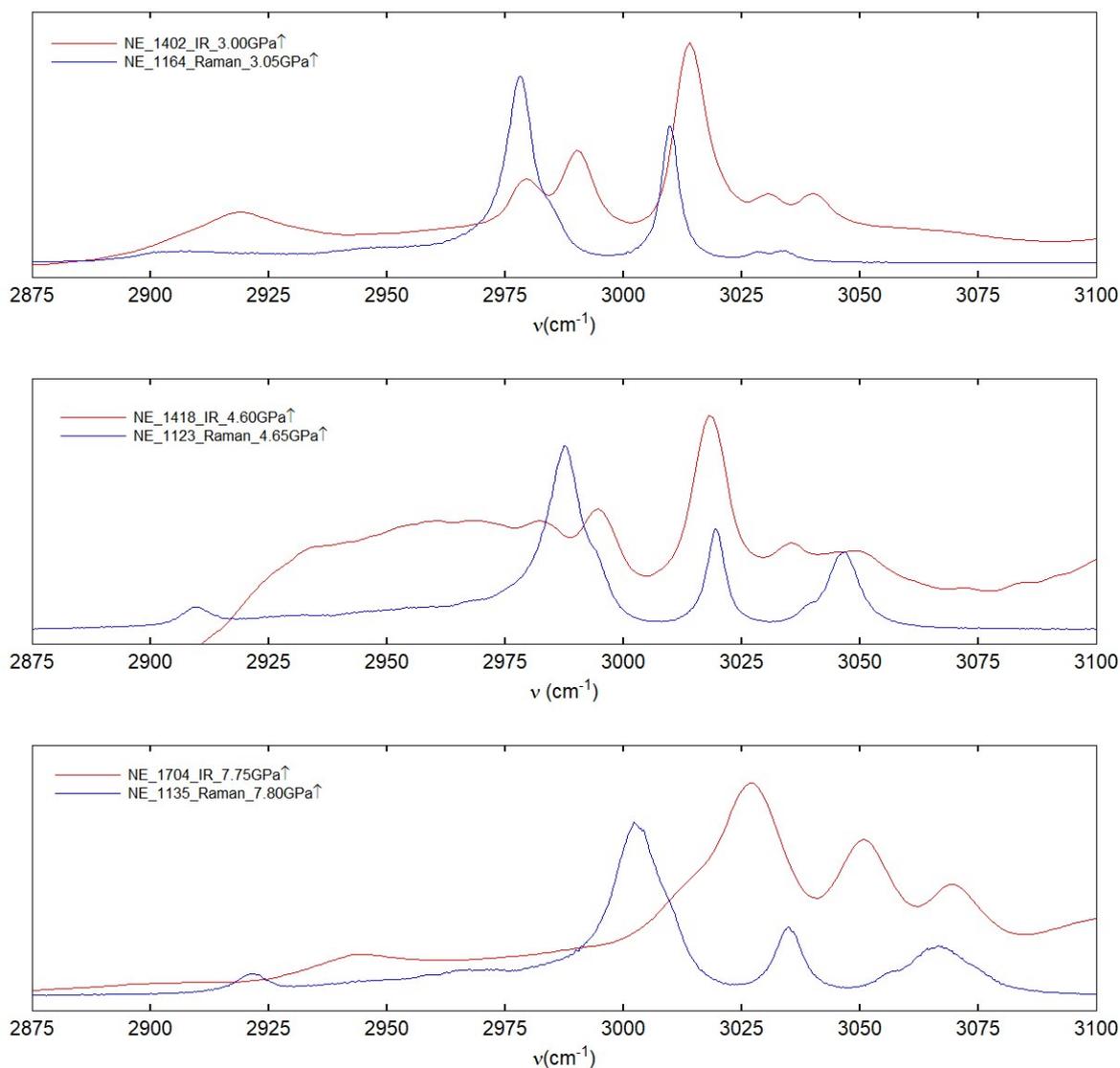


Figure S-8: Comparison between Raman and IR vibrational spectra in the functional group region between pressures ~3.0-7.8 GPa. Legend symbol “ \uparrow ” denotes a spectrum obtained during the pressure upstroke.

Supplementary References

- ¹ R. Dennington, T. Keith, and J. Millam, in *GaussView Version 5.0.8* (Gaussian, Inc. Wallingford CT 06492 USA), 2009).
- ² G. Schaftenaar, and J. H. Noordik, *Journal of Computer-Aided Molecular Design* **14** (2000) 123.
- ³ D. C. Smith, C. Y. Pan, and J. R. Nielsen, *The Journal of Chemical Physics* **18** (1950) 706.
- ⁴ G. Geiseler, and H. Kessler, *Berichte der Bunsengesellschaft für physikalische Chemie* **68** (1964) 571.
- ⁵ S. Courtecuisse *et al.*, *The Journal of Chemical Physics* **108** (1998) 7350.