Supplementary information for "Theoretical study of the mechanism of the solvent dependency of ESIPT in HBT"

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Figure S1. Calculated S₀ energy profiles along C=C twisting at the ω B97X-D/6-31G(d,p) level.



Figure S2. Calculated S1 energy profiles along ESIPT in the gas phase and in solutions at the M11/6-31G(d,p) level.



Figure S3. Calculated S₁ energy profiles along ESIPT in the gas phase and in solutions at the EOM-CCSD/6-31G(d,p) level.

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Calculation level	gas	DMSO	Acetone	CH_2CI_2
ω B97X-D/6-31G(d)	0.53	1.16	1.18	1.13
M11/6-31G(d,p)	0.24	0.51	0.50	0.47
EOM-CCSD/6-31G(d,p)	0.08	0.17	0.17	0.16

Table S1. Calculated activation energies of ESIPT ΔE_{pt} (in kcal/mol).



Figure S4. Calculated S_1 energy profiles along C=C twisting at the M11/6-31G(d,p) level.



Figure S5. Calculated S_1 energy profiles along C=C twisting at the EOM-CCSD/6-31G(d,p) level.



Figure S6. Calculated S_0 energy profiles along C=C twisting at the M11/6-31G(d,p) level.



Figure S7. Calculated S_0 energy profiles along C=C twisting at the EOM-CCSD/6-31G(d,p) level.

Table S2. Calculated activation energies of C=C twisting ΔE_{tw} (in kcal/mol).

Calculation level	gas	DMSO	Acetone	CH_2CI_2
ωB97X-D/6-31G(d)	No barrier	4.56	4.21	3.45
M11/6-31G(d,p)	0.50	8.59	8.20	7.31
EOM-CCSD/6-31G(d,p)	No barrier	6.04	5.70	4.90



Figure S8. Optimized geometry of complex between the keto HBT and two CH_2CI_2 molecules at the PCM- ω B97X-D/6-31G(d,p) level.

transition energies $\Delta E_{1 \rightarrow 0}$ of the complexes (in kcal/mol).									
	one CH ₂ Cl ₂		two CH ₂ Cl ₂						
Calculation level	E _{int}	$\Delta E_{1 \rightarrow 0}$	E _{int}	$\Delta E_{1 ightarrow 0}$					
ωB97X-D/6-31G(d,p)	-18.35	22.32	-27.43	20.66					
ωB97X-D/6-311G(d,p)	-17.56	24.71	-29.03	23.58					
PCM-ωB97X-D/6-31G(d,p)	-7.04	63.75	-7.46	62.74					
PCM-ωB97X-D/6-311G(d,p)	-7.46	63.73	-12.34	63.03					

Table S3. Calculated interaction energies E_{int} between the keto HBT and one or two CH_2Cl_2 molecules in the S_1 state and $S_1 \rightarrow S_0$ transition energies AE_{int} of the complexes (in kcal/mol)