Supporting Information

## Catalyst-Electrolyte Interface Chemistry for Electrochemical CO<sub>2</sub> Reduction

Young Jin Sa,<sup>ab⊥</sup> Chan Woo Lee,<sup>c⊥</sup> Si Young Lee,<sup>ad⊥</sup> Jonggeol Na,<sup>e</sup> Ung Lee\* <sup>adf</sup> and Yun

## Jeong Hwang\*adg

<sup>a</sup> Clean Energy Research Center, Korea Institute of Science and Technology (KIST), Seoul 02792, Republic of Korea.

<sup>b</sup> Department of Chemistry, Kwangwoon University, Seoul 01897, Republic of Korea.

<sup>c</sup> Department of Chemistry, Kookmin University, Seoul 02707, Republic of Korea.

<sup>d</sup> Division of Energy and Environmental Technology, KIST School, Korea University of Science and Technology (UST), Seoul 02792, Republic of Korea.

<sup>e</sup> Division of Chemical Engineering and Materials Science, Ewha Womans University, Seoul 03760, Republic of Korea.

<sup>f</sup> Green School, Korea University, Seoul 02841, Republic of Korea.

<sup>g</sup> Department of Chemical and Biomolecular Engineering and Yonsei-KIST Convergence Research Institute, Yonsei University, Seoul 03722, Republic of Korea.

 $^{\perp}$ These authors contributed to equally to this work

\*Corresponding author Email: ulee@kist.re.kr (U.L.) and yjhwang@kist.re.kr (Y.J.H.)

**Table S1** Summary on the title, contents, and theme of previous review articles for electrochemical  $CO_2$  reduction. The review articles can be categorized based on main focus: catalyst, electrolyte, cell design & process, mechanism, stability and comprehensive. The number of articles for each group corresponds to 63, 2, 13, 11, 1 and 10 in order.

Ref.	Title	Contents	Theme
This work	Catalyst-Electrolyte Interface Chemistry for Electrochemical CO <sub>2</sub> Reduction	<ol> <li>Introduction</li> <li>Technoeconomic perspectives of electrochemical CO<sub>2</sub> reduction</li> <li>Fundamentals of electrochemical CO<sub>2</sub> reduction</li> <li>Design of interfaces between catalyst and surface modulator</li> <li>Understanding catalyst-electrolyte interfaces</li> </ol>	Control of organic modulators, electrolyte ions, electrode structures and the three- phase boundary at the catalyst-electrolyte interface
1	Advances and challenges in electrochemical CO <sub>2</sub> reduction processes: an engineering and design perspective looking beyond new catalyst materials	<ul> <li>6. Conclusions and perspectives <ol> <li>Introduction</li> </ol> </li> <li>2. Working principles of electrochemical CO<sub>2</sub> reduction <ol> <li>Electrolyzer configurations</li> <li>Electrolyzer selection</li> <li>Electrolyte selection</li> <li>pH effects</li> </ol> </li> <li>7. Pressure and temperature effects <ol> <li>Summary and outlook</li> </ol> </li> </ul>	Electrolyzer configuration, electrode structure, electrolyte selection, pH control, and the electrolyzer's operating pressure and temperature
2	A disquisition on the active sites of heterogeneous catalysts for electrochemical reduction of CO <sub>2</sub> to value-added chemicals and fuel	<ol> <li>Introduction</li> <li>Active Sites in Metal-Based Catalysts</li> <li>Active Sites in Metal-Carbon Catalysts</li> <li>Active Sites in Carbon-Based Catalysts</li> <li>Advanced Tools for Active Site Determination         <ul> <li>Machine Learning in CO<sub>2</sub>RR</li> <li>Summary and Outlook</li> </ul> </li> </ol>	Design strategies of active sites
3	Electrolytic cell design for electrochemical CO <sub>2</sub> reduction	<ol> <li>Introduction</li> <li>Electrochemical cell for CO<sub>2</sub>RR</li> <li>Summary and outlook</li> </ol>	Electrolytic cell design
4	An overview on the recent developments of Ag-based electrodes in the electrochemical reduction of CO <sub>2</sub> to CO	<ol> <li>I. Introduction</li> <li>Ag-based nanostructured electrode materials</li> <li>Outlook on surface engineering</li> </ol>	Material factors in Ag- based electrodes
5	Metal-organic frameworks for electrochemical reduction of carbon dioxide: The role of metal centers	<ol> <li>Introduction</li> <li>Mechanism of electrocatalytic reduction of CO<sub>2</sub></li> <li>MOFs-based materials for electrochemical reduction of CO<sub>2</sub></li> <li>Conclusions and outlooks</li> </ol>	MOF-based electrocatalysts
6	Pushing the activity of CO <sub>2</sub> electroreduction by system engineering	1. Introduction     2. Catalytic reactor design     3. Renewable energy-driven systems     4. System optimization     5. Summary and outlook	Reactor architectures and system engineering
7	Strategies for designing nanoparticles for electro- and photocatalytic CO <sub>2</sub> reduction	<ol> <li>Introduction</li> <li>Designer Metal Nanoparticles for Electrocatalytic CO<sub>2</sub>RR</li> <li>Designer Semiconductor Nanoparticles for Photocatalytic CO<sub>2</sub>RR</li> <li>Conclusion and Perspectives</li> </ol>	Strategy of designing nanoparticles
8	Rational design of Ag-based catalysts for the electrochemical CO <sub>2</sub> reduction to CO: A review	<ol> <li>Introduction</li> <li>Fundamentals of ECR</li> <li>Advances in silver-based CO<sub>2</sub>-to-CO electrocatalysts</li> </ol>	Material factors in Ag- based electrodes

		4. Summary and Outlook	
9	Electrochemical conversion of carbon dioxide to high value chemicals using gas-diffusion electrodes	<ol> <li>Introduction         <ol> <li>Electrochemical CO<sub>2</sub> reduction</li> <li>Figures of merit for electrochemical CO<sub>2</sub> reduction</li> </ol> </li> <li>Scaling up of electrochemical CO<sub>2</sub> reduction processes using gas diffusion electrodes</li> <li>GDE systems for various CO<sub>2</sub> reduction products         <ol> <li>Outlook for future research</li> </ol> </li> </ol>	gas diffusion electrode and electrolyzers
10	Modelling chemical reactions on surfaces: The roles of chemical bonding and van der Waals interactions	1. Introduction     2. Methods     3. Applications: Modelling chemical reactions on     metal surfaces     4. More realistic conditions in chemical reactions     5. Remaining challenges     6. Conclusions	Modelling electrocatalytic reduction of CO <sub>2</sub>
11	Current progress of metallic and carbon-based nanostructure catalysts towards the electrochemical reduction of $CO_2$	<ol> <li>Introduction</li> <li>Size and morphology effect in ECR</li> <li>Crystal facet control</li> <li>Defect engineering</li> <li>Interface and surface modification</li> <li>Oxide derivation</li> <li>Conclusion and outlook</li> </ol>	Design of nanostructured inorganic catalysts
12	Emerging nanostructured carbon-based non-precious metal electrocatalysts for selective electrochemical CO <sub>2</sub> reduction to CO	1. Introduction     2. Metal-free heteroatom-doped carbon materials     3. Transition metal heteroatom codoped carbon     materials     4. Carbon-based hybrid materials     5. Summary and outlook	Carbon-based non- precious metal electrocatalysts
13	Interfacial effects in supported catalysts for electrocatalysis	<ol> <li>Introduction</li> <li>Strategies to improve the formation of interfacial effects</li> <li>Application of interfacial effects in electrocatalysis</li> <li>Conclusion and Perspective</li> </ol>	Interfacial effects in supported catalysts
14	Electrolyte effects on the electrochemical reduction of CO <sub>2</sub>	<ol> <li>Introduction</li> <li>CO<sub>2</sub> reduction in aqueous electrolytes</li> <li>CO<sub>2</sub> reduction in non-aqueous electrolytes</li> <li>CO<sub>2</sub> reduction in electrolyte mixtures</li> <li>Summary and outlook</li> </ol>	Electrolyte effects
15	Surface strategies for catalytic CO <sub>2</sub> reduction: from two- dimensional materials to nanoclusters to single atoms	<ol> <li>Introduction</li> <li>Material synthesis and surface strategies for CO<sub>2</sub> reduction</li> <li>Challenges and opportunities</li> </ol>	Nano-to-atomic surfac strategies for catalysts
16	Carbon-based catalysts for electrochemical CO <sub>2</sub> reduction	<ol> <li>Introduction</li> <li>Heteroatoms modification</li> <li>Transition metal- nitrogen- carbon Structure</li> <li>Combination with functional groups</li> <li>Conclusions and outlook</li> </ol>	Carbon-based catalyst and synthetic strategy
17	Review of two-dimensional materials for electrochemical $CO_2$ reduction from a theoretical perspective	<ol> <li>Introduction</li> <li>Possible reaction pathways of electrochemical of CO<sub>2</sub> reduction</li> <li>Discovery of 2D catalysts for electrochemical CO<sub>2</sub> reduction</li> <li>2D transition metal dichalcogenides</li> <li>2D structures of group-VA elements</li> <li>2D metal-organic materials</li> <li>Transition-metal oxides</li> <li>Transition-metal carbides (MXENEs)</li> <li>Summary and outlook</li> </ol>	2D materials for CO <sub>2</sub> reduction

	electrocatalytic reactions provided by SERS	<ol> <li>Oxygen electrochemistry</li> <li>Electrochemistry of carbon monoxide and carbon dioxide</li> <li>Formic acid oxidation</li> <li>Summary</li> </ol>	Raman spectroscopy fo CO <sub>2</sub> reduction
19	Electrode materials engineering in electrocatalytic CO <sub>2</sub> reduction: Energy input and conversion efficiency	<ol> <li>S. Summary</li> <li>1. Introduction</li> <li>2. Fundamentals</li> <li>3. Electrode materials for ECR system</li> <li>4. Conclusions and perspectives</li> </ol>	Anodic and cathodic materials in the photo- anode-assisted electrochemical CO <sub>2</sub> reduction
20	Advanced engineering of core/shell nanostructures for electrochemical carbon dioxide reduction	<ol> <li>Introduction</li> <li>Synthesis of core/shell materials</li> <li>A brief introduction of CO<sub>2</sub>RR</li> <li>Advanced regulations of core/shell</li> <li>nanomaterials for enhanced CO<sub>2</sub> electrocatalytic activity</li> <li>Conclusions and perspectives</li> </ol>	Electrocatalysts with a core/shell structure
21	Nitrogen-doped metal-free carbon catalysts for (electro)chemical CO <sub>2</sub> conversion and valorisation	1. Introduction         2. N-doped carbon materials in CO2 valorisation         3. Concluding remarks	Nitrogen doped metal free carbon catalysts
22	Solvents and supporting electrolytes in the electrocatalytic reduction of CO <sub>2</sub>	<ol> <li>Introduction</li> <li>Aqueous electrocatalytic CO<sub>2</sub> reduction</li> <li>Non-aqueous electrocatalytic CO<sub>2</sub> reduction</li> </ol>	The effects of supporting electrolyte and solvents
23	In-situ infrared spectroscopy applied to the study of the electrocatalytic reduction of CO <sub>2</sub> : Theory, practice and challenges	<ol> <li>Introduction</li> <li>Fundamental aspects of infrared spectroscopy at electrode-electrolyte interfaces</li> <li>Experimental details of infrared spectroscopy at electrode-electrolyte interfaces</li> <li>Applications in the electrocatalytic reduction of CO<sub>2</sub></li> </ol>	In-situ IR spectroscop
24	Recent advances in metal- organic frameworks for photo- /electrocatalytic CO <sub>2</sub> reduction	5. Challenges and opportunities 1. Introduction 2. Photocatalytic CO <sub>2</sub> reduction in MOFs 3. Electrocatalytic CO <sub>2</sub> reduction in MOFs 4. Summary and Prospects	MOF-based CO <sub>2</sub> reduction
25	Rational design of carbon-based metal-free catalysts for electrochemical carbon dioxide reduction: A review	<ol> <li>Introduction</li> <li>Engineering physical structure of CMs</li> <li>Engineering electronic structure of CMs</li> <li>Engineering CMs for CO<sub>2</sub>RR</li> <li>Conclusions and perspectives</li> </ol>	Design strategies of carbon-based metal-fro materials
26	Supported single atoms as new class of catalysts for electrochemical reduction of carbon dioxide	<ol> <li>1. Introduction</li> <li>2. Electrocatalysts for CO<sub>2</sub>RR: from nanoparticles (NPs) to single atoms</li> <li>3. Electrochemical CO<sub>2</sub> conversion on single- atom catalysts</li> <li>4. Understanding the origin of CO<sub>2</sub>RR activity and reaction mechanisms</li> <li>5. Challenges and opportunities</li> </ol>	Single-atom catalysts for CO <sub>2</sub> reduction
27	Recent advances in intensified ethylene production—A review	<ol> <li>Introduction</li> <li>Alternative approaches for ethylene production</li> <li>Challenges and opportunities for intensified ethylene production</li> <li>Summary</li> </ol>	Processes for intensific ethylene production
28	Advances and challenges in understanding the electrocatalytic conversion of carbon dioxide to fuels	<ol> <li>Introduction         <ol> <li>Initial activation of CO2</li> <li>Carbon-carbon bond formation</li> <li>Reaction and process conditions</li> </ol> </li> <li>Electrode morphology and (sub)surface atoms         <ol> <li>In situ spectroscopic investigation of CO2 reduction</li> </ol> </li> </ol>	Mechanistic understandings of CO reduction reaction

		7. Computational approaches for CO <sub>2</sub> reduction 8. Future directions	
29	Electrochemical CO <sub>2</sub> reduction into chemical feedstocks: From mechanistic electrocatalysis models to system design	<ol> <li>Introduction</li> <li>Techno-economic analysis of CO<sub>2</sub>RR</li> <li>Computational insight into the reaction mechanisms</li> <li>Catalysts for CO<sub>2</sub> electroreduction</li> <li>System design for CO<sub>2</sub> electroreduction</li> <li>Conclusions</li> </ol>	Comprehensive review including techno- economic analysis, computational understandings, catalysts materials and system design
30	Engineering metal-organic frameworks for the electrochemical reduction of CO <sub>2</sub> : A minireview	<ol> <li>Introduction</li> <li>Main challenges for CO<sub>2</sub>ER</li> <li>MOF-related catalysts for CO<sub>2</sub>ER</li> <li>Concluding remarks</li> </ol>	Advantages and limitation of MOF- based catalysts in CO <sub>2</sub> reduction
31	Structure-sensitivity and electrolyte effects in CO <sub>2</sub> electroreduction: From model studies to applications	<ol> <li>Introduction</li> <li>Structure sensitivity in CO<sub>2</sub> reduction</li> <li>Tuning the selectivity by engineering the electrolyte interface</li> <li>Challenges and future directions</li> </ol>	The effects of surface structure and electrolyte
32	Current achievements and the future direction of electrochemical CO <sub>2</sub> reduction: A short review	<ol> <li>Introduction</li> <li>Commercial plants and projects</li> <li>An economic feasibility of produced chemicals</li> <li>Liquid-phase electrochemical CO<sub>2</sub> reduction</li> <li>Evolution into the gas-phase CO<sub>2</sub> electrolysis</li> <li>Conclusion</li> </ol>	CO <sub>2</sub> electrolysis systems and their industrial feasibility
33	Progress and perspectives of electrochemical CO <sub>2</sub> reduction on copper in aqueous electrolyte	<ol> <li>Introduction</li> <li>Considerations for conducting and comparing electrochemical CO<sub>2</sub> reduction experiments</li> <li>Experimental probes of CO<sub>2</sub> reduction mechanisms</li> <li>Theoretical studies on copper</li> <li>Electrochemical CO<sub>2</sub> reduction pathways on Cu</li> <li>Nanostructured Cu</li> <li>Copper bimetallics</li> <li>Conclusions and future outlook</li> </ol>	Various factors and reaction mechanisms affecting Cu-based CO reduction
34	CO <sub>2</sub> conversion by membrane reactors	<ol> <li>Introduction</li> <li>Low temperature (≤100 °C) methods for CO<sub>2</sub> conversion</li> <li>High temperature (&gt;100 °C) membrane reactors for CO<sub>2</sub> conversion         <ul> <li>4. Conclusions</li> </ul> </li> </ol>	Electrochemical, photochemical, thermochemical membrane reactors fo $CO_2$ reduction
35	CO <sub>2</sub> reduction on gas-diffusion electrodes and why catalytic performance must be assessed at commercially-relevant conditions	<ol> <li>Introduction</li> <li>Effect of cell configuration and reaction rate on CO<sub>2</sub> reduction environments</li> <li>Impact of high current densities on CO<sub>2</sub> reduction catalyst testing</li> <li>Impact of high current densities on system design</li></ol>	The effects of cell configuration and reaction rate on electrocatalytic system
36	Rational design of transition metal-based materials for highly efficient electrocatalysis	<ol> <li>Introduction</li> <li>Creating more active sites</li> <li>Improving the utilization of active sites</li> <li>Modulation of electronic configuration</li> <li>Control lattice facets</li> <li>Conclusion and perspective</li> </ol>	Strategies to design metal-based electrocatalysts
37	Carbon-supported single atom catalysts for electrochemical energy conversion and storage	1. Introduction     2. Sample preparation     3. Electrocatalytic performance     4. Summaries and perspectives	Synthetic strategies and electrocatalytic performances of single atom catalysts
	Surface and interface engineering in copper-based	<ol> <li>Introduction</li> <li>Fundamental understanding of the CO<sub>2</sub>RR</li> <li>Improving selectivity by interfacial</li> </ol>	The fundamental role o the secondary metal in

	selective CO <sub>2</sub> electroreduction	engineering 4. Theoretical prediction 5. General trends	electrocatalysts
39	Cu-based nanocatalysts for electrochemical reduction of CO <sub>2</sub>	1. Introduction     2. Electrochemical reduction of CO <sub>2</sub> 3. Cu-based nanocatalysts for electrochemical reduction of CO <sub>2</sub> 4. Conclusions	Design strategies of Cu based electrocatalysts
40	Recent advances in the nanoengineering of electrocatalysts for CO <sub>2</sub> reduction	<ol> <li>Introduction</li> <li>Bulk metallic catalysts for the ECR</li> <li>Nanoengineering of catalysts for the ECR</li> <li>Summary and outlook</li> </ol>	Nanoengineering strategies of electrocatalysts
41	Electrolytic CO <sub>2</sub> reduction in a flow cell	<ol> <li>Introduction</li> <li>Flow-cell architectures</li> <li>Gas phase CO<sub>2</sub> electrolysis flow cells</li> <li>CO<sub>2</sub> flow cell optimization</li> <li>Conclusions and perspectives</li> </ol>	System-level strategies of membrane-based flow cells and microfluidic reactors
42	Electrocatalytic alloys for CO <sub>2</sub> reduction	<ol> <li>Introduction</li> <li>CO<sub>2</sub> → CO with electrocatalytic alloys</li> <li>CO<sub>2</sub> → formate with electrocatalytic alloys</li> <li>CO<sub>2</sub> → C<sub>2</sub> with electrocatalytic alloys</li> <li>Summary and perspective</li> </ol>	Summary on various reported alloys
43	Progress and perspective of electrocatalytic CO <sub>2</sub> reduction for renewable carbonaceous fuels and chemicals	<ol> <li>Introduction</li> <li>Electrocatalysts for electrocatalytic CO<sub>2</sub> reduction</li> <li>Product selectivity in electrocatalytic CO<sub>2</sub> reduction</li> <li>Challenges and perspectives</li> </ol>	Metal–organic complexes, metals, metal alloys, inorganic metal compounds and carbon-based metal-free nanomaterials for CO <sub>2</sub> reduction
44	Understanding the heterogeneous electrocatalytic reduction of carbon dioxide on oxide-derived catalysts	<ol> <li>Introduction</li> <li>Effects of pH and electrolyte on the performance and selectivity of CO<sub>2</sub> reduction on oxide catalysts</li> <li>Analysis of the performance and mechanisms of oxide-derived catalysts</li> <li>Conclusions</li> </ol>	The effect of oxygen, surface morphologies, and local pH gradients on the catalysis of CO <sub>2</sub> reduction
45	Tuning of CO <sub>2</sub> reduction selectivity on metal electrocatalysts	<ol> <li>Introduction</li> <li>Option of electrolytes</li> <li>Design of electrocatalysts</li> <li>Conclusion and perspective</li> </ol>	The rational selection of electrolytes and design of electrocatalysts
46	Metal-free carbon materials for CO <sub>2</sub> electrochemical reduction	<ol> <li>Introduction</li> <li>Mechanics of the CO<sub>2</sub>RR on metal-free carbon materials</li> <li>Metal-free carbon electrocatalysts for the CO<sub>2</sub>RR</li> <li>Conclusions and outlook</li> </ol>	Carbon materials with heteroatom doping as metal-free catalysts
47	CO <sub>2</sub> reduction: From the electrochemical to photochemical approach	<ol> <li>Introduction</li> <li>Fundamentals of electrocatalytic and photocatalytic CO<sub>2</sub> reduction</li> <li>Electrocatalytic materials for CO<sub>2</sub> reduction</li> <li>Photocatalytic materials for CO<sub>2</sub> reduction</li> <li>Conclusion and perspectives</li> </ol>	Electrocatalysts and photocatalysts for CO <sub>2</sub> reduction
48	Fundamentals and challenges of electrochemical CO <sub>2</sub> reduction using two-dimensional materials	<ol> <li>Introduction         <ol> <li>A perspective of electrochemical CO<sub>2</sub> reduction</li> <li>2D nanosheet catalysis of CO<sub>2</sub> electroreduction</li> <li>Strategies for improving CO<sub>2</sub> electrocatalytic activity of 2D nanosheets</li> <li>Summary and outlook</li> </ol> </li> </ol>	Strategies for tuning catalytic activities of 21 materials
_49	Nanostructured materials for heterogeneous electrocatalytic	1. Introduction2. Crucial parameters for CO2 electroreduction tests	Material factors determining the

	reaction mechanisms	<ol> <li>Reaction setup for CO<sub>2</sub> electroreduction</li> <li>Reaction mechanism and pathways</li> <li>Summary and perspectives</li> </ol>	electroreduction
50	Continuous-flow electroreduction of carbon dioxide	<ol> <li>Summary and perspectives         <ol> <li>Introduction</li> <li>Reactor designs</li> <li>Materials</li> <li>Operation</li> <li>How to benchmark a CO<sub>2</sub> electrolyzer correctly?</li> <li>Photoelectrochemical reduction of CO<sub>2</sub> in continuous-flow</li> <li>Summary and outlook</li> </ol> </li> </ol>	The effects of cell design, employed materials and operational conditions
51	Recent progress on bismuth- based nanomaterials for electrocatalytic carbon dioxide reduction	<ol> <li>Introduction</li> <li>Fundamentals for electrocatalytic CO<sub>2</sub>RR</li> <li>Various Bi-based nanomaterials for electrocatalytic CO<sub>2</sub>RR</li> <li>Summary and outlook</li> </ol>	Bi-based electrocatalytic materials
52	Recent advances in atomic-level engineering of nanostructured catalysts for electrochemical CO <sub>2</sub> reduction	<ol> <li>Introduction</li> <li>Measurement system and evaluation parameters for ECR</li> <li>Reaction mechanism</li> <li>Advanced characterization techniques for verifying structural information at atomic scale</li> <li>Modifying the nanostructured electrocatalysts at atomic scale toward ECR</li> <li>Summary and outlook</li> </ol>	Design strategies of nanostructured electrocatalysts at the atomic level
53	Mechanistic understanding of the electrocatalytic CO <sub>2</sub> reduction reaction - New developments based on advanced instrumental techniques	<ol> <li>Introduction         <ol> <li>Reaction pathways</li> </ol> </li> <li>Commonly used theoretical methods for mechanistic investigations         <ol> <li>Advanced instrumental techniques for mechanistic investigations</li> <li>Conclusions and outlook</li> </ol> </li> </ol>	Instrumental techniques for the mechanistic study of the electrochemical CO <sub>2</sub> reduction
54	In-situ spectroscopic techniques as critical evaluation tools for electrochemical carbon dioxide reduction: A mini review	<ol> <li>Introduction</li> <li>Reaction mechanism of CO<sub>2</sub>RR</li> <li>In-situ techniques for electrochemical CO<sub>2</sub> reduction</li> <li>Summary and outlook</li> </ol>	Electrochemical and non-electrochemical techniques as critical evaluation tools for electrocatalysts
55	Heterostructured catalysts for electrocatalytic and photocatalytic carbon dioxide reduction	<ol> <li>Introduction</li> <li>CO<sub>2</sub> reduction pathways</li> <li>Heterostructures in electrocatalytic CO<sub>2</sub> reduction</li> <li>Heterostructures in photocatalytic CO<sub>2</sub> reduction</li> <li>Mechanisms of photocatalysis: Type II, Z- Scheme, p-n heterojunctions</li> <li>Conclusions and outlook</li> </ol>	Heterostructured catalysts pertaining to electrocatalytic and photocatalytic carbon dioxide reduction
56	Current progress in electrocatalytic carbon dioxide reduction to fuels on heterogeneous catalysts	<ol> <li>Introduction</li> <li>Research progress</li> <li>Conclusions and perspectives</li> </ol>	The design of effective catalysts with lower overpotential, high FE and product selectivity
57	Transition metal-nitrogen sites for electrochemical carbon dioxide reduction reaction	<ol> <li>Introduction</li> <li>Metal-nitrogen containing macrocyclic complexes</li> <li>Metal organic frameworks (MOFs)</li> <li>Carbon-based metal-nitrogen materials</li> <li>Reaction parameters and proposed mechanisms</li> <li>Conclusion and outlooks</li> </ol>	M–N <sub>x</sub> sites-containing transition metal- centered macrocyclic complexes, metal organic frameworks, and M–N <sub>x</sub> -doped carbon materials
<u>-50</u>	Metal-nitrogen-carbon electrocatalysts for CO <sub>2</sub>	<ol> <li>CO<sub>2</sub> valorization and syngas generation</li> <li>Metal-nitrogen-carbon electrocatalysts for syngas generation, a selectivity and economical perspective</li> </ol>	M-N-C electrocatalysts

		<ol> <li>The active sites of metal-nitrogen-carbon electrocatalysts for CO<sub>2</sub>RR</li> <li>Toward syngas generation at an industrial scale 5. Summary and outlook</li> </ol>	
59	An overview of Cu-based heterogeneous electrocatalysts for CO <sub>2</sub> reduction	<ol> <li>Introduction</li> <li>A note for conducting EC CO<sub>2</sub> reduction</li> <li>Significant progress in the study of Cu-based heterogeneous for EC CO<sub>2</sub> reduction</li> <li>A brief review for Cu-based heterogeneous catalysts for EC CO<sub>2</sub> reduction</li> <li>Electrolyte effect on CO<sub>2</sub> reduction with Cu- based heterogeneous electrocatalysts</li> <li>EC/PEC CO<sub>2</sub> reduction and H<sub>2</sub>O oxidation as an overall reaction system for Cu-based electrocatalysts</li> <li>Summary and outlook</li> </ol>	Cu-based heterogeneous electrocatalysts and electrolyte effects
60	Electrochemical CO <sub>2</sub> reduction: from nanoclusters to single atom catalysts	<ol> <li>Introduction</li> <li>Fundamentals of electrochemical CO<sub>2</sub> reduction</li> <li>ECR applications of NCs catalysts</li> <li>ECR applications of SACs</li> <li>Summary and perspectives</li> </ol>	Electrocatalytic properties of nanoclusters and single atom catalysts
61	Transforming the carbon economy: challenges and opportunities in the convergence of low-cost electricity and reductive CO <sub>2</sub> utilization	1. Introduction     2. Direct electrochemical     3. Non-thermal plasma     4. Direct bioelectrochemical     5. Indirect bioelectrochemical     6. Indirect thermochemical     7. Summary and cross-comparison of CO <sub>2</sub> R     pathways     8. General considerations for CO <sub>2</sub> R     9. Evaluation of CO <sub>2</sub> R products     10. Conclusions	Technical barriers and economic viability of electrochemical, thermocatalytic and biological CO <sub>2</sub> utilization
62	Electrocatalytic reduction of carbon dioxide: opportunities with heterogeneous molecular catalysts	<ol> <li>Introduction</li> <li>Fundamentals of heterogeneous molecular catalysts for CO<sub>2</sub> reduction</li> <li>Non-covalent immobilization technique</li> <li>Covalent immobilization technique</li> <li>Periodic immobilization technique</li> <li>The stability of heterogeneous molecular catalysts</li> <li>The influence of catalysts' utilization on the CO<sub>2</sub> conversion</li> <li>The influence of the supports on the CO<sub>2</sub> conversion</li> <li>Designing suitable catalysts for CO<sub>2</sub> conversion</li> <li>Conclusions and perspectives</li> </ol>	Methods for heterogeneous immobilization of homogeneous molecular catalysts for $CO_2$ reduction
63	Electrochemical CO <sub>2</sub> reduction to CO catalyzed by 2D nanostructures	1. Introduction         2. Fundamentals of electrochemical CO2 reduction         3. 2D electrocatalysts         4. Conclusions	Two-dimensional graphene and transition metal chalcogenides for CO <sub>2</sub> reduction
64	Electrochemical CO <sub>2</sub> reduction on nanostructured metal electrodes: fact or defect?	<ol> <li>Introduction</li> <li>Onset potential</li> <li>Selectivity and stability</li> <li>Activity and current density</li> <li>Mass transport effects</li> <li>Conclusions</li> </ol>	The effects of active surface area and thickness of the catalytic layer on activity, selectivity, stability and mass transfer
65	Strategies in catalysts and electrolyzer design for	<ol> <li>Introduction</li> <li>Multicarbon hydrocarbons</li> <li>Multicarbon oxygenates</li> </ol>	Electrocatalysts, electrode/reactor design and corresponding

	toward C <sub>2+</sub> products	4. Design of the electroreduction cell 5. Summary and outlook	mechanisms for C-C coupling
66	Recent progress in self- supported catalysts for CO <sub>2</sub> electrochemical reduction	<ol> <li>Introduction</li> <li>Self-supported catalysts for electrochemical CO<sub>2</sub> reduction</li> <li>Organically doped metal electrocatalysts</li> <li>Nanostructured materials on metal foils</li> <li>Metal nanoarrays on conductive substrate</li> <li>Heteroatom-doped carbon materials</li> <li>Conclusions</li> </ol>	Synthesis methods, chemical composition nanostructures, and catalytic efficiencies of self-supported catalys
67	Mechanistic understanding of CO <sub>2</sub> reduction reaction (CO <sub>2</sub> RR) toward multicarbon products by heterogeneous copper-based catalysts	<ol> <li>Introduction</li> <li>CO<sub>2</sub>RR and CORR: DFT insights</li> <li>Summary and perspectives</li> </ol>	Mechanistic reaction pathways for multicarbon products
68	Durable cathodes and electrolyzers for the efficient aqueous electrochemical reduction of CO <sub>2</sub>	<ol> <li>Introduction</li> <li>Recent durability and stability studies of cathodes for CO<sub>2</sub>RR</li> <li>Failure modes of CO<sub>2</sub> electrolyzer components</li> <li>CO<sub>2</sub>RR durability and degradation characterization protocols</li> <li>Summary and outlook</li> </ol>	Reported durability studies and degradatio mechanisms
69	CO <sub>2</sub> reduction: From homogeneous to heterogeneous electrocatalysis	<ol> <li>Introduction</li> <li>Homogeneous catalysis</li> <li>Surface catalysis</li> <li>Surface nanocatalysis</li> <li>Conclusions and outlook</li> </ol>	Homogeneous and heterogeneous catalys of transition-metal complex catalysts for CO <sub>2</sub> reduction
70	Promises of main group metal- based nanostructured materials for electrochemical CO <sub>2</sub> reduction to formate	<ol> <li>Introduction         <ol> <li>Fundamentals of electrochemical CO<sub>2</sub> reduction</li> <li>Main group metal-based electrocatalysts for selective CO<sub>2</sub>RR to formate</li> <li>Flow cells or membrane electrode assembly cells</li> <li>Mechanistic studies using in situ characterization techniques                 <ul> <li>Summary and outlook</li> </ul> </li> </ol></li> </ol>	Main group metal-bass (Sn, Bi, In, Pb, Sb) electrocatalysts, cell design and mechanisti studies for selective CO <sub>2</sub> reduction to form acid
71	Carbon-rich nonprecious metal single atom electrocatalysts for CO <sub>2</sub> reduction and hydrogen evolution	<ol> <li>Introduction</li> <li>Characterization and evaluation of carbon-rich NPMSACs for CRR and HER</li> <li>Carbon-rich NPMSACs for CRR and HER</li> <li>Conclusion</li> </ol>	Structure-activity relationship of nonprecious metal single atom catalysts
72	Two-dimensional electrocatalysts for efficient reduction of carbon dioxide	<ol> <li>Introduction</li> <li>Overview of the structure and properties of 2D materials</li> <li>Synthesis of 2D materials</li> <li>Applications of 2D materials in the eCO<sub>2</sub>RR</li> <li>Challenges and outlook</li> </ol>	Structures and catalyt properties of 2D catalysts
73	A review on photochemical, biochemical and electrochemical transformation of CO <sub>2</sub> into value-added products	<ol> <li>Introduction</li> <li>Carbon cycle and CO<sub>2</sub> emission</li> <li>CO<sub>2</sub> capture and methodologies</li> <li>Products obtained from CO<sub>2</sub> transformation</li> <li>Photochemical transformation of CO<sub>2</sub></li> <li>Biochemical transformation of CO<sub>2</sub></li> <li>Electrochemical transformation of CO<sub>2</sub></li> <li>Future research</li> <li>Conclusion</li> </ol>	Overview of CO <sub>2</sub> reduction using photochemical, biochemical and electrochemical methods
74	Strategies for bioelectrochemical CO <sub>2</sub> reduction	1. Introduction     2. Electrochemical CO <sub>2</sub> reduction by enzymes     3. Electrochemical CO <sub>2</sub> reduction by cells     4. Conclusion and outlooks     1. Introduction	CO <sub>2</sub> reduction catalyze by electroactive enzymes and whole cells

	catalysts with atomic layer deposition for the reduction of carbon dioxide	<ol> <li>Atomic layer deposition designed catalysts</li> <li>The applications of the ALD-designed novel catalyst materials for CO<sub>2</sub> reduction</li> <li>The significance of ALD-prepared materials in CO<sub>2</sub> reduction</li> </ol>	for the designs of the efficient catalyst nanomaterials in CO <sub>2</sub> reduction
		5. Summary and outlook	
76	Electrocatalytic water splitting and CO <sub>2</sub> reduction: Sustainable solutions via single-atom catalysts supported on 2D materials	<ol> <li>Introduction</li> <li>SACs supported on 2D materials</li> <li>Energy harvesting applications with 2D materials</li> <li>Conclusion and perspectives</li> </ol>	Single-atom catalysts supported on 2D materials
77	Heterogeneous molecular catalysts for electrocatalytic CO <sub>2</sub> reduction	<ol> <li>Introduction         <ol> <li>Metal center</li> <li>Extrinsic and intrinsic activity</li> <li>Mechanism of CO<sub>2</sub> reduction                 <ul> <li>Ligand effects</li> <li>Electrode support effects                              <li>Catalyst stability</li></li></ul></li></ol></li></ol>	Heterogeneous molecular catalysts for electrochemical CO <sub>2</sub> reduction including polymers, metal-organic frameworks, and covalent-organic frameworks
78	Alloy nanocatalysts for the electrochemical oxygen reduction (ORR) and the direct electrochemical carbon dioxide reduction reaction (CO <sub>2</sub> RR)	<ol> <li>Introduction</li> <li>Pt-based nanoalloys for the electrochemical oxygen reduction reaction</li> <li>Cu-based nanoalloys for the carbon dioxide reduction reaction</li> <li>Concluding future perspectives on nanoalloys</li> </ol>	Pt-based and Cu-based nanoalloy electrocatalysts for ORR and CO <sub>2</sub> RR
79	Supported molecular catalysts for the heterogeneous $CO_2$ electroreduction	<ol> <li>Introduction</li> <li>Catalyst structure</li> <li>Immobilization strategy and support material</li> <li>Conclusions and perspectives</li> </ol>	Catalyst structure, immobilization strategy and support material of molecular catalysts for heterogeneous CO <sub>2</sub> reduction
80	Advances in the electrochemical catalytic reduction of CO <sub>2</sub> with metal complexes	<ol> <li>Introduction</li> <li>Selected examples of molecular catalyst development</li> <li>Conclusions and perspectives</li> </ol>	Electrochemical catalytic reduction of CO <sub>2</sub> with metal complexes
81	Pathways to industrial-scale fuel out of thin air from CO <sub>2</sub> electrolysis	<ol> <li>Introduction</li> <li>Solar-driven synthesis routes for MeOH using CO<sub>2</sub> electrolyzers</li> <li>Energy efficiency and distribution of a solar- driven MeOH synthesis process</li> <li>Relative scale of each sub-process</li> <li>Future outlook and summary</li> </ol>	Integration of direct CO <sub>2</sub> air capture with CO <sub>2</sub> and H <sub>2</sub> O electrolyzers and a traditional MeOH synthesis step
82	Probing CO <sub>2</sub> conversion chemistry on nanostructured surfaces with operando vibrational spectroscopy	<ol> <li>Introduction</li> <li>Raman spectroelectrochemistry</li> <li>Surface-enhanced Raman spectroelectrochemistry</li> <li>Surface-selective infrared spectroelectrochemistry</li> <li>Surface-enhanced infrared absorption spectroelectrochemistry</li> <li>Surface-enhanced infrared absorption spectroelectrochemistry</li> <li>Emerging vibrational spectroscopy-based techniques</li> <li>Concluding remarks</li> </ol>	Operando probing of electrochemical CO <sub>2</sub> reduction using Raman and infrared spectroscopy
83	One-dimensional nanomaterial electrocatalysts for CO <sub>2</sub> fixation	<ol> <li>I. Introduction</li> <li>Metallic 1D nanostructured materials</li> <li>Metal-based 1D nanomaterials</li> <li>Carbon-based 1D nanomaterials</li> <li>Conclusion and Outlook</li> </ol>	1D nanomaterials for CO <sub>2</sub> eletroreduction, including including metals, transition-metal oxides/nitrides, transition-metal chalcogenides, and carbon-based materials

	Emerging carbon-based heterogeneous catalysts for	CO <sub>2</sub> 3. Linear scaling relations for metal surfaces	Metal-free doped carbon and aromatic N-
92	electrochemical CO <sub>2</sub> conversion: From liquid-phase to gas-phase systems	<ol> <li>CO<sub>2</sub> reduction reaction with liquid-phase reactor</li> <li>CO<sub>2</sub> reduction reaction with gas-phase reactor         <ol> <li>Summary and outlook</li> <li>Introduction</li> </ol> </li> <li>Challenges of the electrochemical reduction of</li> </ol>	Gas-phase reactor systems for CO <sub>2</sub> reduction
91	Understanding the roadmap for electrochemical reduction of CO <sub>2</sub> to multi-carbon oxygenates and hydrocarbons on copper- based catalysts Towards higher rate	<ol> <li>Introduction</li> <li>Atomistic mechanism of various C<sub>2</sub> and C<sub>3</sub> products</li> <li>Factors influencing the thermodynamics and kinetics of CRR to C<sub>2</sub> products</li> <li>Design principles of C<sub>2</sub> electrocatalysts</li> <li>Technical issues</li> <li>Challenges and outlook</li> <li>Introduction</li> </ol>	An in-depth discussion of the mechanistic aspects of various C <sub>2</sub> reaction pathways on copper-based catalysts
90	Defect engineering in earth- abundant electrocatalysts for CO <sub>2</sub> and N <sub>2</sub> reduction	<ol> <li>Introduction</li> <li>Concept, characterization and engineering strategy of defect</li> <li>Defect engineering for electrocatalytic CRR and NRR</li> <li>Conclusions</li> </ol>	The type, regulation strategy, fine characterization methods of defect and the application in electrocatalytic CRR and NRR
89	Theoretical insights into heterogeneous (photo)electrochemical CO <sub>2</sub> reduction	<ol> <li>Introduction</li> <li>Heterogeneous CO<sub>2</sub> photoelectrochemical reduction</li> <li>Heterogeneous CO<sub>2</sub> electrochemical reduction</li> <li>Discussion and outlook</li> <li>Summary</li> </ol>	Theoretical studies of heterogeneous (photo)electrochemical reduction
88	Secondary-sphere effects in molecular electrocatalytic CO <sub>2</sub> reduction	1. Introduction2. Enzymes for the interconversion CO2 and COCO or formic acid3. Emerging systems4. Conclusions and outlook	Secondary-sphere strategies of molecular catalysts to facilitate rapid and selective CO <sub>2</sub> reduction
87	A look at periodic trends in d- block molecular electrocatalysts for CO <sub>2</sub> reduction	<ol> <li>Introduction</li> <li>Electrochemical CO<sub>2</sub> reduction with transition metal complexes</li> <li>Horizontal trends in transition metal complexes for CO<sub>2</sub> reduction</li> <li>Vertical trends in transition metal complexes for CO<sub>2</sub> reduction</li> <li>Ligand design and evolution for CO<sub>2</sub> reduction</li> <li>Formic acid producing catalysts</li> <li>Conclusions and prospects</li> </ol>	Trends of activity, electronic structure of catalytic intermediates and product selectivity in mononuclear complexes
86	Carbon dioxide photo/electroreduction with cobalt	<ol> <li>Introduction         <ol> <li>Catalytic reduction mechanism</li> <li>Types of Co-containing catalysts for CO<sub>2</sub> reduction</li> </ol> </li> <li>Performance of CO<sub>2</sub> reduction based on Co-containing catalysts         <ol> <li>Conclusion and perspective</li> </ol> </li> </ol>	CO <sub>2</sub> photo/electroreduction with Co catalysts
85	Recent trends, benchmarking, and challenges of electrochemical reduction of CO <sub>2</sub> by molecular catalysts	<ol> <li>Introduction</li> <li>Metal molecular catalysts</li> <li>Metal-free organic catalysts</li> <li>Conclusions and outlook</li> </ol>	Molecular catalysts and design principles for CO <sub>2</sub> electroreduction
84	Advances in Sn-based catalysts for electrochemical CO <sub>2</sub> reduction	<ol> <li>Introduction</li> <li>Reaction mechanism and pathways of the Sn-based electrocatalysts</li> <li>Advanced Sn-based catalysts for electrochemical CO<sub>2</sub> reduction</li> <li>Conclusions and outlook</li> </ol>	Various Sn-based electrocatalysts and reaction mechanisms

	chemicals	4. Nonmetal heteroatom-doped carbon nanostructures	reduction of CO <sub>2</sub>
		<ul> <li>5. Aromatic nitrogen substituted heterocycles</li> <li>6. Metal–nitrogen–carbon structure</li> <li>7. Summary and outlook</li> </ul>	
94	Electrochemical carbon dioxide splitting	<ol> <li>Introduction</li> <li>Fundamental principles</li> <li>Recent progress towards practical electrochemical CO<sub>2</sub> splitting</li> <li>Summary and perspectives</li> </ol>	The roles of main components of electrochemical CO <sub>2</sub> splitting device, such as catalyst, electrode, electrolyzer, membrane, and electrolyte
95	Efficient and selective electrochemically driven enzyme-catalyzed reduction of carbon dioxide to formate using formate dehydrogenase and an artificial cofactor	<ol> <li>Introduction</li> <li>Carbon dioxide reduction using metal- independent FDH and its natural cofactor, NADH</li> <li>Replacing NADH with an artificial redox cofactor, methyl viologen</li> <li>Methyl viologen is a unidirectional cofactor for FDH</li> <li>Continuous reduction of carbon dioxide to formate in two-chamber and three-chamber electrolyzer configurations</li> <li>Conclusions</li> </ol>	Enzyme-catalyzed reduction of carbon dioxide to formate using formate dehydrogenase
96	Single-atom catalysis toward efficient CO <sub>2</sub> conversion to CO and formate products	<ol> <li>Introduction</li> <li>Thermally stable SACS for selective CO<sub>2</sub>-to- CO conversion</li> <li>Atomically dispersed Ni(I) SACs for highly efficient electrochemical CO<sub>2</sub> reduction reaction</li> <li>A quasi-homogeneous SAC for aqueous-phase CO<sub>2</sub> hydrogenation under mild conditions</li> <li>Cconclusions and perspectives</li> </ol>	Single-atom catalysts in thermal catalysis and electrocatalysis for CO <sub>2</sub> reduction
97	Electrochemical reduction of $CO_2$ over heterogeneous catalysts in aqueous solution: recent progress and perspectives	<ol> <li>Introduction</li> <li>Mechanistic pathways of CO<sub>2</sub> electroreduction in aqueous media</li> <li>Recent advances of electrocatalysts for improved performance and mechanistic understanding</li> <li>Summary and perspectives</li> </ol>	Underlying mechanistic studies and novel heterogeneous catalysts for electrochemical reduction of CO <sub>2</sub>
98	Rational catalyst and electrolyte design for CO <sub>2</sub> electroreduction towards multicarbon products	<ol> <li>Introduction</li> <li>Catalyst design for C<sub>2+</sub> products</li> <li>Electrolyte design for C<sub>2+</sub> products</li> <li>Summary and outlook</li> </ol>	Rational design of catalyst and electrolyte for multicarbon products
99	Theoretical insights into selective electrochemical conversion of carbon dioxide	<ol> <li>Introduction</li> <li>Challenges of electrochemical CO<sub>2</sub> reduction reactions</li> <li>Strategies to tune intermediate binding</li> <li>4. Conclusions</li> </ol>	Design principles of catalyst and electrolyte based on calculational studies
100	Nickel complexes as molecular catalysts for water splitting and CO <sub>2</sub> reduction	<ol> <li>Introduction</li> <li>Water splitting</li> <li>Carbon dioxide reduction</li> <li>Conclusions</li> </ol>	Nickel complexes as molecular catalysts for water splitting and CO <sub>2</sub> reduction

## Supplementary references

- 1. S. Garg, M. Li, A. Z. Weber, L. Ge, L. Li, V. Rudolph, G. Wang and T. E. Rufford, *J. Mater. Chem. A*, 2020, **8**, 1511–1544.
- 2. R. Daiyan, W. H. Saputera, H. Masood, J. Leverett, X. Lu and R. Amal, *Adv. Energy Mater.*, 2020, **10**, 1902106.

- 3. S. Liang, N. Altaf, L. Huang, Y. Gao and Q. Wang, J. CO2 Util., 2020, 35, 90–105.
- 4. S. A. Mahyoub, F. A. Qaraah, C. Chen, F. Zhang, S. Yan and Z. Cheng, *Sustain*. *Energy Fuels*, 2019, **4**, 50–67.
- 5. P. Shao, L. Yi, S. Chen, T. Zhou and J. Zhang, J. Energy Chem., 2020, 40, 156–170.
- 6. H. Shen, Z. Gu and G. Zheng, *Sci. Bull.*, 2019, **64**, 1805–1816.
- 7. J. Y. Choi, W. Choi, J. W. Park, C. K. Lim and H. Song, *Chem. Asian J.*, 2020, **15**, 253–265.
- 8. D. Sun, X. Xu, Y. Qin, S. P. Jiang and Z. Shao, *ChemSusChem*, 2020, 13, 39–58.
- 9. S. Malkhandi and B. S. Yeo, Curr. Opin. Chem. Eng., 2019, 26, 112–121.
- 10. G. Su, S. Yang, Y. Jiang, J. Li, S. Li, J.-C. Ren and W. Liu, *Prog. Surf. Sci.*, 2019, **94**, 100561.
- 11. L. Hou, J. Yan, L. Takele, Y. Wang, X. Yan and Y. Gao, *Inorg. Chem. Front.*, 2019, **6**, 3363–3380.
- 12. X. Wang, Q. Zhao, B. Yang, Z. Li, Z. Bo, K. H. Lam, N. M. Adli, L. Lei, Z. Wen, G. Wu and Y. Hou, *J. Mater. Chem. A*, 2019, **7**, 25191–25202.
- 13. H. Li, C. Chen, D. Yan, Y. Wang, R. Chen, Y. Zou and S. Wang, *J. Mater. Chem. A*, 2019, 7, 23432–23450.
- 14. M. M. de Salles Pupo and R. Kortlever, *ChemPhysChem*, 2019, 20, 2926–2935.
- L. Wang, W. Chen, D. Zhang, Y. Du, R. Amal, S. Qiao, J. Wu and Z. Yin, *Chem. Soc. Rev.*, 2019, 48, 5310–5349.
- 16. C. Jia, K. Dastafkan, W. Ren, W. Yang and C. Zhao, *Sustainable Energy Fuels*, 2019, **3**, 2890–2906.
- 17. X. Zhu and Y. Li, Wiley Interdiscip. Rev. Comput. Mol. Sci., 2019, 9, e1416.
- A. J. Keeler, G. R. Salazar-Banda and A. E. Russell, *Curr. Opin. Electrochem.*, 2019, 17, 90–96.
- 19. R.-B. Song, W. Zhu, J. Fu, Y. Chen, L. Liu, J.-R. Zhang, Y. Lin and J.-J. Zhu, *Adv. Mater.*, 2019. DOI: 10.1002/adma.201903796.
- 20. Q. Shao, P. Wang, S. Liu and X. Huang, J. Mater. Chem. A, 2019, 7, 20478–20493.
- 21. D. M. Fernandes, A. F. Peixoto and C. Freire, *Dalton Trans.*, 2019, 48, 13508–13528.
- 22. M. König, J. Vaes, E. Klemm and D. Pant, *iScience*, 2019, 19, 135–160.
- 23. R. Kas, O. Ayemoba, N. J. Firet, J. Middelkoop, W. A. Smith and A. Cuesta, *ChemPhysChem*, 2019, **20**, 2904–2925.
- 24. Q. Wang, Y. Zhang, H. Lin and J. Zhu, Chem. Eur. J., 2019, 25, 14026–14035.
- 25. S. Liu, H. Yang, X. Su, J. Ding, Q. Mao, Y. Huang, T. Zhang and B. Liu, *J. Energy Chem.*, 2019, **36**, 95–105.

- 26. Y. Cheng, S. Yang, S. P. Jiang and S. Wang, *Small Methods*, 2019, **3**, 1800440.
- Y. Gao, L. Neal, D. Ding, W. Wu, C. Baroi, A. M. Gaffney and F. Li, *ACS Catal.*, 2019, 9, 8592–8621.
- Y. Y. Birdja, E. Pérez-Gallent, M. C. Figueiredo, A. J. Göttle, F. Calle-Vallejo and M. T. M. Koper, *Nat. Energy*, 2019, 4, 732–745.
- 29. M. G. Kibria, J. P. Edwards, C. M. Gabardo, C.-T. Dinh, A. Seifitokaldani, D. Sinton and E. H. Sargent, *Adv. Mater.*, 2019, **31**, 1807166.
- 30. R. Wang, F. Kapteijn and J. Gascon, Chem. Asian J., 2019, 14, 3452–3461.
- 31. P. Sebastián-Pascual, S. Mezzavilla, I. E. L. Stephens and M. Escudero-Escribano, *ChemCatChem*, 2019, **11**, 3626–3645.
- 32. M.-Y. Lee, K. T. Park, W. Lee, H. Lim, Y. Kwon and S. Kang, *Crit. Rev. Environ. Sci. Technol.*, 2020, **50**, 769–815.
- S. Nitopi, E. Bertheussen, S. B. Scott, X. Liu, A. K. Engstfeld, S. Horch, B. Seger, I. E. L. Stephens, K. Chan, C. Hahn, J. K. Nørskov, T. F. Jaramillo and I. Chorkendorff, *Chem. Rev.*, 2019, **119**, 7610–7672.
- 34. A. Brunetti and E. Fontananova, J. Nanosci. Nanotechnol., 2019, 19, 3124–3134.
- 35. T. Burdyny and W. A. Smith, *Energy Environ. Sci.*, 2019, **12**, 1442–1453.
- 36. S. Dou, X. Wang and S. Wang, *Small Methods*, 2019, **3**, 1800211.
- 37. Y. Peng, B. Lu and S. Chen, Adv. Mater., 2018, 30, 1801995.
- 38. A. Vasileff, C. Xu, Y. Jiao, Y. Zheng and S. Z. Qiao, Chem, 2018, 4, 1809–1831.
- 39. H. Xie, T. Wang, J. Liang, Q. Li and S. Sun, Nano Today, 2018, 21, 41–54.
- 40. F. Li, D. R. MacFarlane and J. Zhang, *Nanoscale*, 2018, 10, 6235–6260.
- 41. D. M. Weekes, D. A. Salvatore, A. Reyes, A. Huang and C. P. Berlinguette, *Acc. Chem. Res.*, 2018, **51**, 910–918.
- 42. J. He, N. J. J. Johnson, A. Huang and C. P. Berlinguette, *ChemSusChem*, 2018, **11**, 48–57.
- 43. W. Zhang, Y. Hu, L. Ma, G. Zhu, Y. Wang, X. Xue, R. Chen, S. Yang and Z. Jin, *Adv. Sci.*, 2018, **5**, 1700275.
- 44. J. E. Pander, D. Ren, Y. Huang, N. W. X. Loo, S. H. L. Hong and B. S. Yeo, *ChemElectroChem*, 2018, **5**, 219–237.
- 45. Y. Wang, J. Liu, Y. Wang, A. M. Al-Enizi and G. Zheng, *Small*, 2017, 13, 1701809.
- X. Duan, J. Xu, Z. Wei, J. Ma, S. Guo, S. Wang, H. Liu and S. Dou, *Adv. Mater.*, 2017, 29, 1701784.
- 47. J. Wu, Y. Huang, W. Ye and Y. Li, Adv. Sci., 2017, 4, 1700194.
- 48. Z. Sun, T. Ma, H. Tao, Q. Fan and B. Han, *Chem*, 2017, **3**, 560–587.

- 49. L. Zhang, Z.-J. Zhao and J. Gong, Angew. Chem., Int. Ed., 2017, 56, 11326–11353.
- 50. B. Endrődi, G. Bencsik, F. Darvas, R. Jones, K. Rajeshwar and C. Janáky, *Prog. Energy Combust. Sci.*, 2017, **62**, 133–154.
- 51. C. Yang, J. Chai, Z. Wang, Y. Xing, J. Peng and Q. Yan, *Chem. Res. Chinese Univ.*, 2020, **36**, 410–419.
- 52. H. Liu, Y. Zhu, J. Ma, Z. Zhang and W. Hu, Adv. Funct. Mater., 2020, 30, 1910534.
- 53. Y. Zhang, S.-X. Guo, X. Zhang, A. M. Bond and J. Zhang, *Nano Today*, 2020, **31**, 100835.
- 54. K. S. Adarsh, N. Chandrasekaran and V. Chakrapani, Front. Chem., 2020, 8, 137.
- 55. P. Prabhu, V. Jose and J.-M. Lee, Adv. Funct. Mater., 2020, 1910768.
- 56. A. Liu, M. Gao, X. Ren, F. Meng, Y. Yang, L. Gao, Q. Yang and T. Ma, *J. Mater. Chem. A*, 2020, 8, 3541–3562.
- C. Yan, L. Lin, G. Wang and X. Bao, *Cuihua Xuebao/Chinese J. Catal.*, 2019, 40, 23– 37.
- 58. L. Delafontaine, T. Asset and P. Atanassov, ChemSusChem, 2020, 13, 1688–1698.
- 59. J. Zhao, S. Xue, J. Barber, Y. Zhou, J. Meng and X. Ke, *J. Mater. Chem. A*, 2020, **8**, 4700–4734.
- 60. F. Lü, H. Bao, Y. Mi, Y. Liu, J. Sun, X. Peng, Y. Qiu, L. Zhuo, X. Liu and J. Luo, *Sustainable Energy Fuels*, 2020, 4, 1012–1028.
- 61. R. G. Grim, Z. Huang, M. T. Guarnieri, J. R. Ferrell, L. Tao and J. A. Schaidle, *Energy Environ. Sci.*, 2020, **13**, 472–494.
- 62. L. Sun, V. Reddu, A. C. Fisher and X. Wang, *Energy Environ. Sci.*, 2020, 13, 374–403.
- 63. C. B. Hiragond, H. Kim, J. Lee, S. Sorcar, C. Erkey and S.-I. In, *Catalysts*, 2020, 10, 98.
- 64. R. Kas, K. Yang, D. Bohra, R. Kortlever, T. Burdyny and W. A. Smith, *Chem. Sci.*, 2020, **11**, 1738–1749.
- 65. L. Fan, C. Xia, F. Yang, J. Wang, H. Wang and Y. Lu, Sci. Adv., 2020, 6, eaay3111.
- 66. H. Yang, X. Wang, Q. Hu, X. Chai, X. Ren, Q. Zhang, J. Liu and C. He, *Small Methods*, 2020. DOI: 10.1002/smtd.201900826.
- 67. T. K. Todorova, M. W. Schreiber and M. Fontecave, ACS Catal., 2020, 10, 1754–1768.
- 68. U. O. Nwabara, E. R. Cofell, S. Verma, E. Negro and P. J. A. Kenis, *ChemSusChem*, 2020, **13**, 855–875.
- 69. S. Zhang, Q. Fan, R. Xia and T. J. Meyer, Acc. Chem. Res., 2020, 53, 255–264.
- 70. N. Han, P. Ding, L. He, Y. Li and Y. Li, Adv. Energy Mater., 2020, 10, 1902338.
- 71. T. Wang, Q. Zhao, Y. Fu, C. Lei, B. Yang, Z. Li, L. Lei, G. Wu and Y. Hou, *Small Methods*, 2019, **3**, 1900210.

- 72. Y. Zhang, L. Li, S. Guo, X. Zhang, F. Li, A. M. Bond and J. Zhang, *ChemSusChem*, 2020, **13**, 59–77.
- P. R. Yaashikaa, P. Senthil Kumar, S. J. Varjani and A. Saravanan, *J. CO2 Util.*, 2019, 33, 131–147.
- 74. M. Yuan, M. J. Kummer and S. D. Minteer, Chem. Eur. J., 2019, 25, 14258–14266.
- 75. Z. Chen, G. Zhang, J. Prakash, Y. Zheng and S. Sun, *Adv. Energy Mater.*, 2019, **9**, 1900889.
- 76. K. C. Kwon, J. M. Suh, R. S. Varma, M. Shokouhimehr and H. W. Jang, *Small Methods*, 2019, **3**, 1800492.
- 77. N. Corbin, J. Zeng, K. Williams and K. Manthiram, Nano Res., 2019, 12, 2093–2125.
- 78. C. Kim, F. Dionigi, V. Beermann, X. Wang, T. Möller and P. Strasser, *Adv. Mater.*, 2019, **31**, 1805617.
- 79. X. M. Hu, S. U. Pedersen and K. Daasbjerg, *Curr. Opin. Electrochem.*, 2019, **15**, 148–154.
- F. Franco, S. Fernández and J. Lloret-Fillol, *Curr. Opin. Electrochem.*, 2019, 15, 109– 117.
- 81. W. A. Smith, T. Burdyny, D. A. Vermaas and H. Geerlings, *Joule*, 2019, **3**, 1822–1834.
- 82. N. Heidary, K. H. Ly and N. Kornienko, Nano Lett., 2019, 19, 4817–4826.
- A. Guan, C. Yang, Y. Quan, H. Shen, N. Cao, T. Li, Y. Ji and G. Zheng, *Chem. Asian J.*, 2019, 14, 3969–3980.
- S. Zhao, S. Li, T. Guo, S. Zhang, J. Wang, Y. Wu and Y. Chen, *Nano-Micro Lett.*, 2019, 11, 62.
- 85. K. Elouarzaki, V. Kannan, V. Jose, H. S. Sabharwal and J. Lee, *Adv. Energy Mater.*, 2019, **9**, 1900090.
- C. Li, X. Tong, P. Yu, W. Du, J. Wu, H. Rao and Z. M. Wang, *J. Mater. Chem. A*, 2019, 7, 16622–16642.
- 87. C. Jiang, A. W. Nichols and C. W. Machan, Dalton Trans., 2019, 48, 9454–9468.
- 88. A. W. Nichols and C. W. Machan, Front. Chem., 2019, 7, 397.
- 89. S. Xu and E. A. Carter, Chem. Rev., 2019, 119, 6631–6669.
- 90. Q. Wang, Y. Lei, D. Wang and Y. Li, Energy Environ. Sci., 2019, 12, 1730-1750.
- 91. Y. Zheng, A. Vasileff, X. Zhou, Y. Jiao, M. Jaroniec and S. Z. Qiao, *J. Am. Chem. Soc.*, 2019, **141**, 7646–7659.
- 92. J. T. Song, H. Song, B. Kim and J. Oh, Catalysts, 2019, 9, 224.
- 93. J. Wu, T. Sharifi, Y. Gao, T. Zhang and P. M. Ajayan, Adv. Mater., 2019, 31, 1804257.
- 94. J. Xie, Y. Huang, M. Wu and Y. Wang, ChemElectroChem, 2019, 6, 1587–1604.

- 95. B. S. Jayathilake, S. Bhattacharya, N. Vaidehi and S. R. Narayanan, *Acc. Chem. Res.*, 2019, **52**, 676–685.
- 96. X. Su, X. F. Yang, Y. Huang, B. Liu and T. Zhang, Acc. Chem. Res., 2019, **52**, 656–664.
- 97. C. Long, X. Li, J. Guo, Y. Shi, S. Liu and Z. Tang, Small Methods, 2018, 3, 1800369.
- 98. D. Gao, R. M. Arán-Ais, H. S. Jeon and B. Roldan Cuenya, *Nat. Catal.*, 2019, **2**, 198–210.
- 99. C. W. Lee, C. Kim and B. K. Min, Nano Converg., 2019, 6, 8.
- 100. J. W. Wang, W. J. Liu, D. C. Zhong and T. B. Lu, *Coord. Chem. Rev.*, 2019, **378**, 237–261.