

Electronic Supplementary Information

A PMMA-based heterogeneous photocatalyst for visible light-promoted (4+2) cycloaddition

Niklas Huber,^a Run Li,^a Calum T. J. Ferguson,^a Dominik W. Gehrig,^a Charusheela Ramanan,^a Paul W. M. Blom,^a Katharina Landfester,^a and Kai. A. I. Zhang^{*a,b}

^a Max Planck Institute for Polymer Research, Ackermannweg 10, 55128 Mainz, Germany

E-mail: kai.zhang@mpip-mainz.mpg.de

^b Department of Materials Science, Fudan University, Shanghai 200433, P. R. China

E-mail: kai_zhang@fudan.edu.cn

Characterization of Photocatalyst cPMMA-BTPh₂

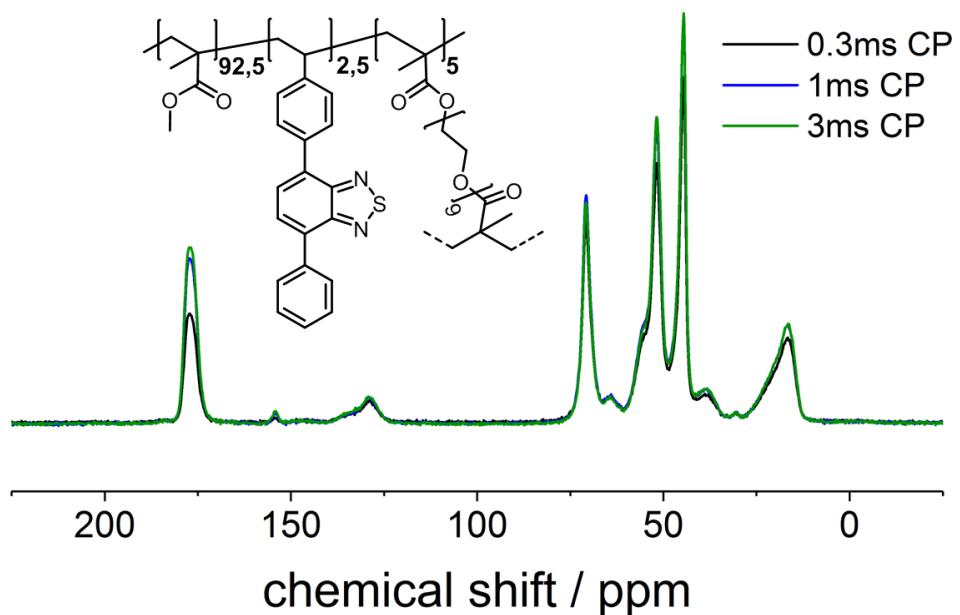


Fig. S1. ¹³C CP-MAS NMR spectrum for different contact times (0.3 ms, 1 ms, 3 ms) of cPMMA-BTPh₂.

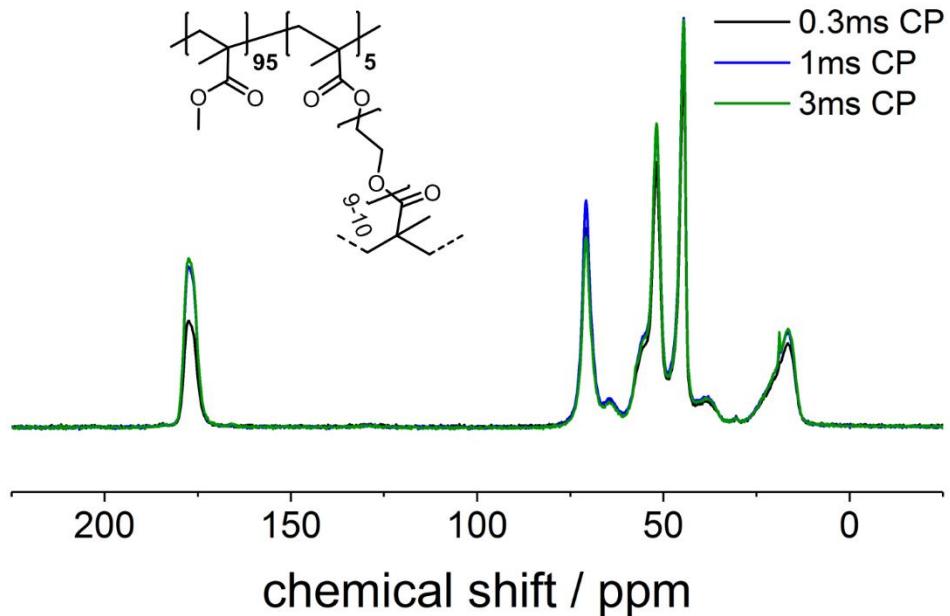


Fig. S2. ¹³C CP-MAS NMR spectrum for different contact times (0.3 ms, 1 ms, 3 ms) of cPMMA.

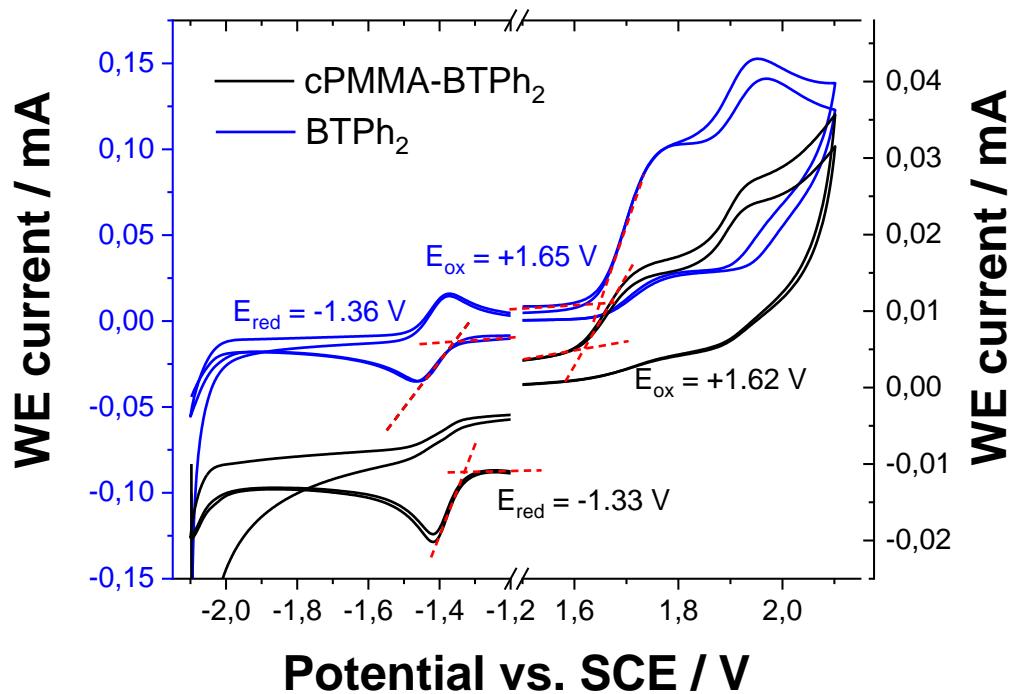
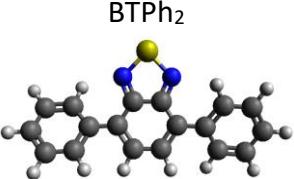
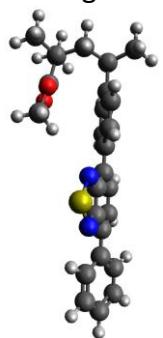
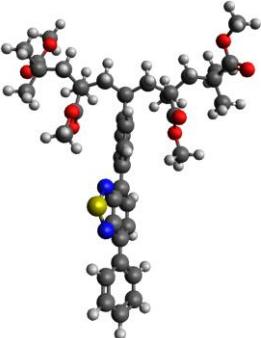


Fig. S3. Cyclic voltammetry of cPMMA-BTPh₂ and BTPh₂ with a scan rate of 100 mV/s in CH₃CN.

Table S1. DFT and TDDFT-calculated electrooptical parameters of the catalysts (rB3LYP, 6-31+g(d))

	HOMO vs. SCE	LUMO vs. SCE	Band Gap (electr.)	λ_{\max} (calc.)	Band Gap (optic. calc. from λ_{\max})
BTPh ₂ 	+1.64 V	-1.70 V	2.93 eV	480 nm	2.58 eV
cPMMA-BTPh ₂ analogue 1 	+1.36 V	-1.51 V	2.87 eV	496 nm	2.50 eV
cPMMA-BTPh ₂ analogue 2 	+1.40 V	-1.47 V	2.87 eV	496 nm	2.50 eV

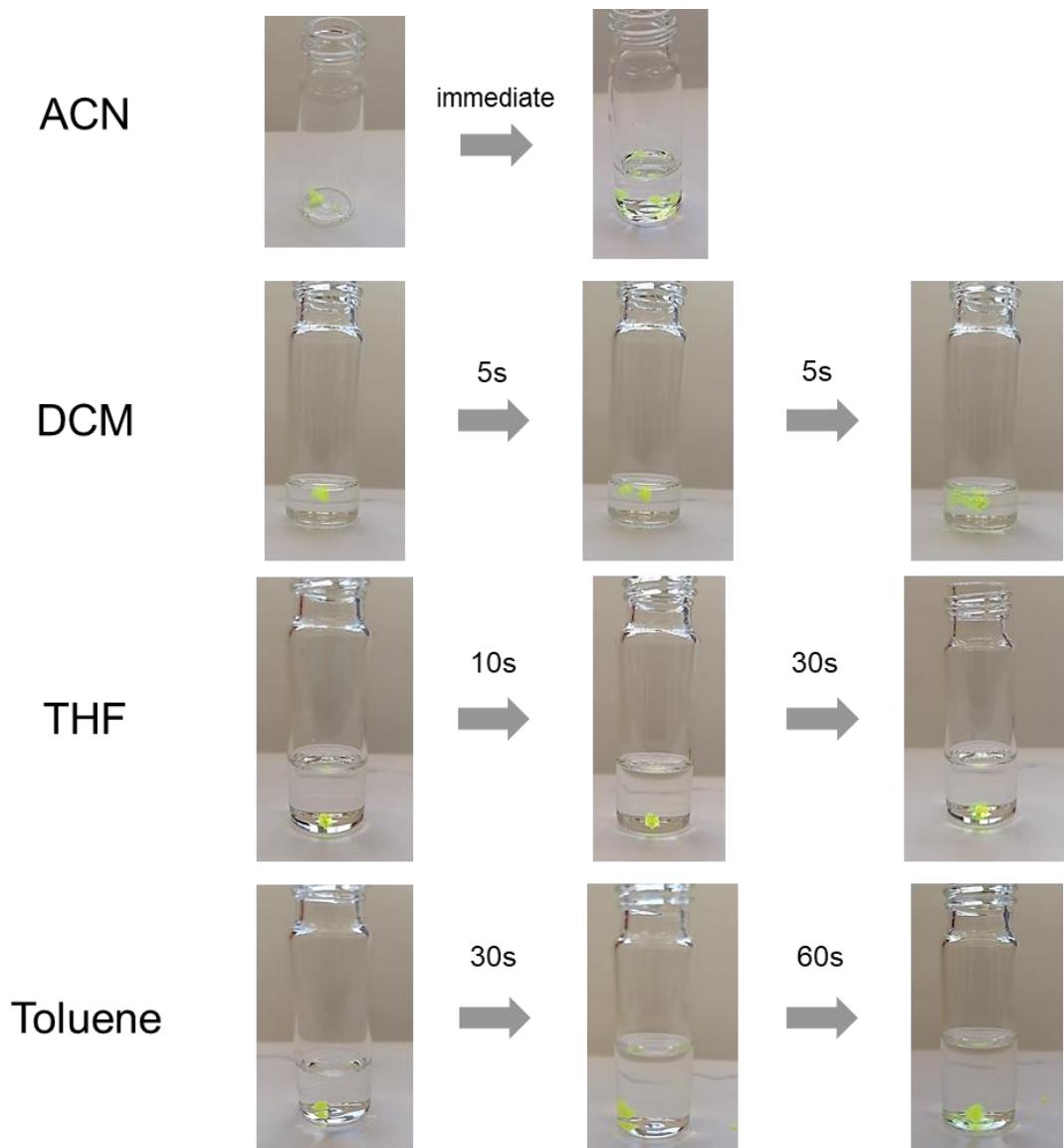


Fig. S4. Swelling experiments in acetonitrile, dichloromethane, THF and toluene. The pictures were taken after the times indicated.

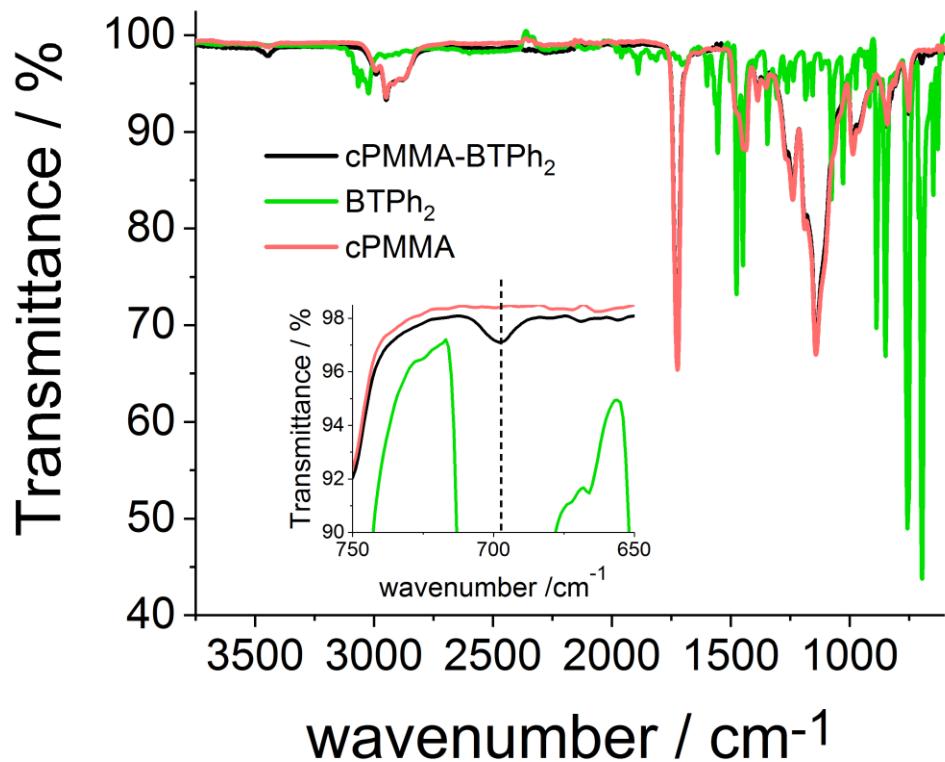


Fig. S5. Comparison of Fourier transform infrared (FTIR) spectra of cPMMA-BTPh₂, cPMMA and BTPh₂. cPMMA and cPMMA-BTPh₂ differ only slightly at 696 cm⁻¹, the major resonance of BTPh₂.

Current prices for used chemicals:

MMA – 18€ / 500g (Alfa)

Poly(ethylene glycol) dimethacrylate – 67,80€ / 250g (Sigma)

4,7-Dibromo-2,1,3-benzothiadiazole – 27,30 € / 1g (Alfa Aesar)

Phenylboronic acid – 37€ / 10g (Alfa Aesar)

4-Vinylphenylboronic acid – 95€ / 5g (TCI)

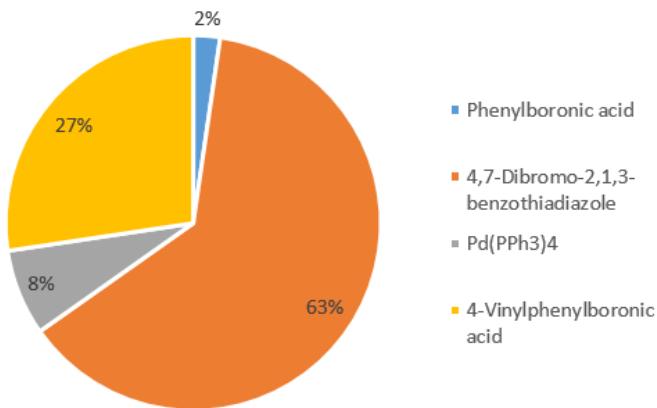
Pd(PPh₃)₄ – 105€ / 5g (TCI)

AIBN – 42€ / 25g (Sigma)

Cost of material for cPMMA-BTPh₂ (solvents not included):

Monomer Synthesis: Phenylboronic acid (509 mg), 4,7-Dibromo-2,1,3-benzothiadiazole (1840 mg), Pd(PPh₃)₄ (283 mg), 4-Vinylphenylboronic acid (1150 mg) = 79,91€ for 950 mg of Monomer = 84,11 €/g

Monomer Price per 1 g



Polymer Synthesis: MMA (294 mg), PEGDMA (93 mg), AIBN (4 mg), Photocatalytic unit (25,5 mg) = 2,07€ for 300 mg of Polymer = 6,92 € / g

Fig. S6. Material cost estimation for the synthesis of 1g of cPMMA-BTPh₂

Photocatalysis

Part A: Kinetic Study

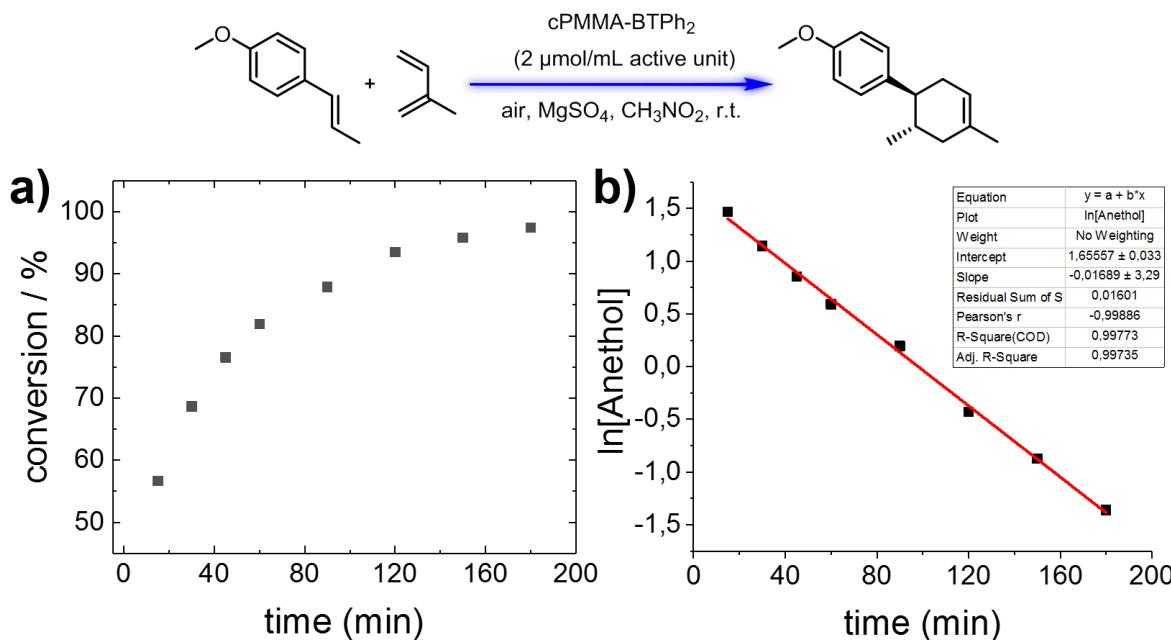


Fig. S7. Monitoring experiment of the photocatalytic Diels-Alder reaction. (a) Reaction conversion and (b) logarithmic plot of the concentration of *trans*-anethole during the reaction. The initial concentration of *trans*-anethole was 0.1 M.

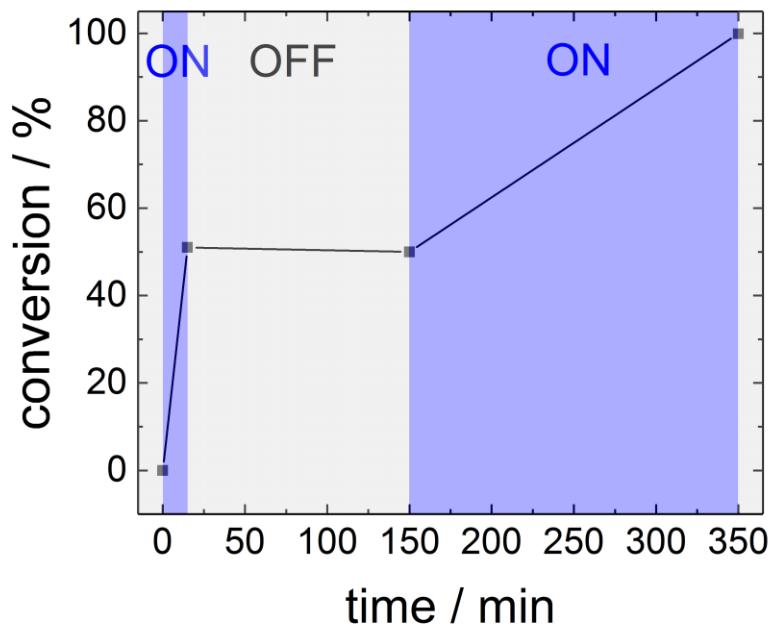


Fig. S8. Reaction conversion change during the light on and off experiment.

Part B: Apparent Quantum Yield Calculation

$$\Phi_{AQY} = \frac{\text{number of product molecules formed}}{\text{number of incident photons}}$$

$$= \frac{N_e}{N_p}$$

$$= \frac{n * N_A}{\left(\frac{P * t}{\frac{h * c}{\lambda}} \right)}$$

$$= \frac{n * N_A * h * c}{P * t * \lambda}$$

$$= \frac{0.125 * 10^{-3} \text{ mol} * 6.023 * 10^{23} \text{ mol}^{-1} * 6.63 * 10^{-34} \frac{\text{m}^2 \text{kg}}{\text{s}} \times 3 * 10^8 \frac{\text{m}}{\text{s}}}{\left(0.8 \text{ cm} * 2.3 * 0.065 \frac{\text{W}}{\text{cm}^2} \right) \times 240 * 60 \text{ s} \times 460 * 10^{-9} \text{ m}}$$

$$= 0.0227 = 2.27\%$$

Part C: Recycle Experiment

The recycle experiment was conducted similar to the general procedure. Specifically, a dried vial with a magnetic stir bar was charged with *trans*-anethole (35.6 mg, 0.24 mmol), isoprene (240 μ L, 2.4 mmol), cPMMA-BTPh₂ (2.4 mg) and nitromethane (2.4 mL). After 4 h of reaction, the conversion was analyzed by GC-MS. The reaction mixture was centrifuged at 1000 rpm for 2 min and the supernatant was replaced by CH₃NO₂. Then, another equivalent of *trans*-anethole (35.6 mg, 0.24 mmol) and isoprene (240 μ L, 2.4 mmol) were added into the mixture and the solution was further irradiated under blue light. The catalyst was recycled 4 times.

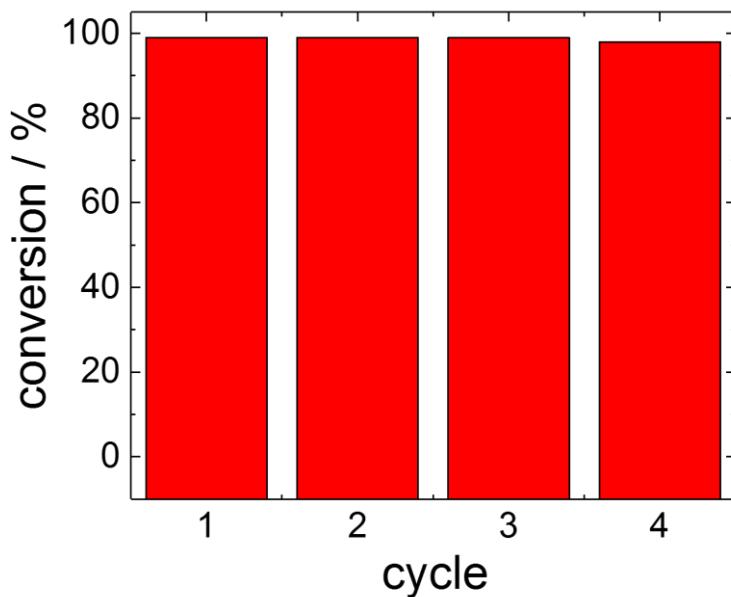


Fig. S9. Repeating experiment of the photocatalytic Diels-Alder reaction with cPMMA-BTPh₂.

In the same way, BTPh₂ was tested. Without purification process, another equivalent of *trans*-anethole (35.6 mg, 0.24 mmol), isoprene (240 μ L, 2.4 mmol), BTPh₂ (0.6 mg, 2.0 μ mol) and nitromethane (2.4 mL). After 2 h of reaction, the conversion was analyzed by GC-MS. Another equivalent of *trans*-anethole (35.6 mg, 0.24 mmol) and isoprene (240 μ L, 2.4 mmol) were added into the mixture and the solution was put under blue light irradiation. The procedure was repeated 3 times.

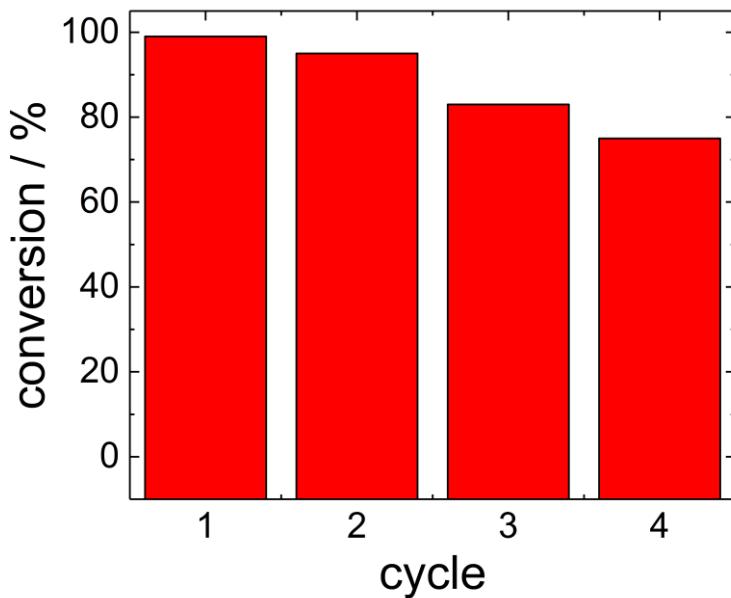


Fig. S10. Repeating experiment of the photocatalytic Diels-Alder reaction with BTPh_2 .

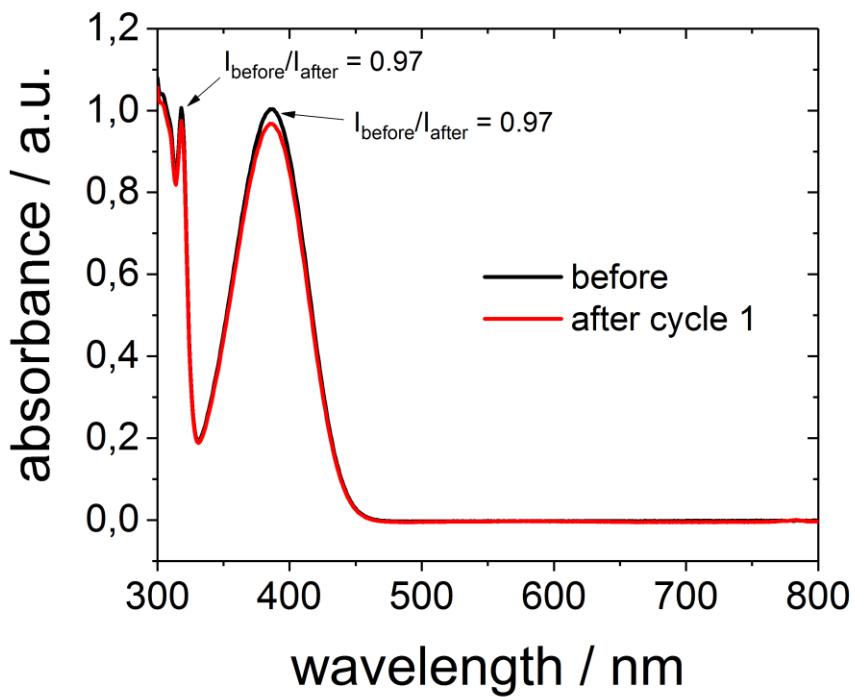


Fig. S11. UV-Vis spectra of cPMMA-BTPh₂ before and after the photocatalytic Diels-Alder reaction. The maximum intensities on both peaks show a similar decrease indicating this to be a concentration-related effect.

Part D: Solubility Experiments

In order to prove the heterogeneity of the cPMMA-BTPh₂ photocatalyst the catalyst was irradiated with blue light while being stirred in CH₃NO₂ (I), CDCl₃ (II) and *d*₈-THF (III) respectively. For sample I, after 4h the mixture was filtered through a syringe filter. The solution was centrifuged for 10min at 10000 rpm and the supernatant was evaporated to dryness on the rotary evaporator. CDCl₃ was added to the round-bottom flask and taken for ¹H-NMR analysis. For the setups II and III the mixture was filtered through a syringe filter, centrifuged as above and the supernatant directly taken for ¹H-NMR.

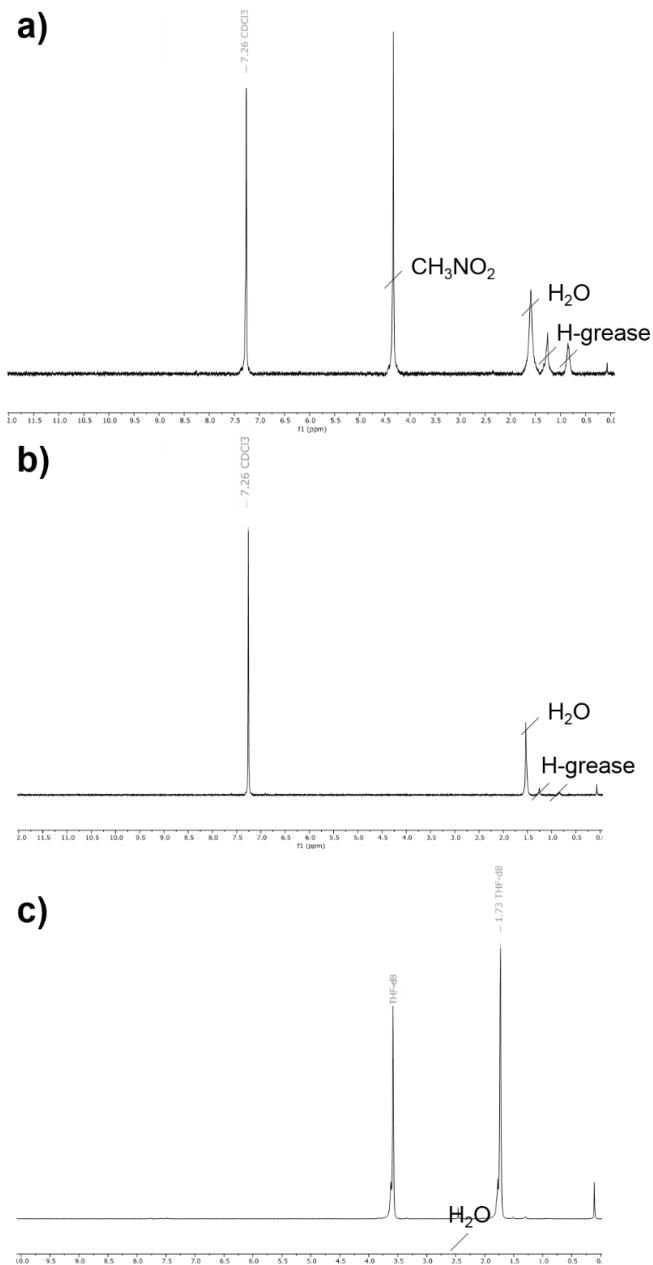


Fig. S12. ¹H-NMR spectra of a) experiment I b) experiment II and c) experiment III

Part E: Transient Absorption Spectra

Transient absorption spectra were recorded to further investigate the concrete electron transfer process between the photocatalyst BTPh_2 and substrates in the Diels-Alder reaction.

The signal peaking at 660 nm represents the photo-induced absorption (PIA) of BTPh_2 in its excited state (Fig. S11). The lifetime of the singlet excited state of BTPh_2 in a N_2 -purged solution was determined to be 11.1 ns, which is consistent to that obtained from the time-resolved photoluminescence spectroscopy (TRPL). However, the addition of O_2 did not affect the PIA of BTPh_2 (Fig. S12). The possible reason might be that the phosphorescence signal for active oxygen species such as ${}^1\text{O}_2$ normally localized on 1270 nm was beyond the detection limit of the transient absorption measurement. In comparison, the PIA curves of BTPh_2 under the addition of *trans*-anethole remained first the similar shape as that of pure BTPh_2 within the initial time from 1 ns to 3.1 ns, and then a new absorption peak at 610 nm appeared after 9.9 ns pulse irradiation (Fig. S13). This observation indicates that the formation of the radical cation of *trans*-anethole began to occur after 9.9 ns, which is similar as previous report,¹ thus supporting the proposed hole initiation mechanism. By increasing the concentration of *trans*-anethole, the absorption intensity of the mixture decayed faster, demonstrating the active electron transfer from the substrate to the photo-generated holes of the photocatalyst (Fig. S14). The decay profiles of the TA were consistent with the ones of the TRPL (Fig. S16 and Table S2).

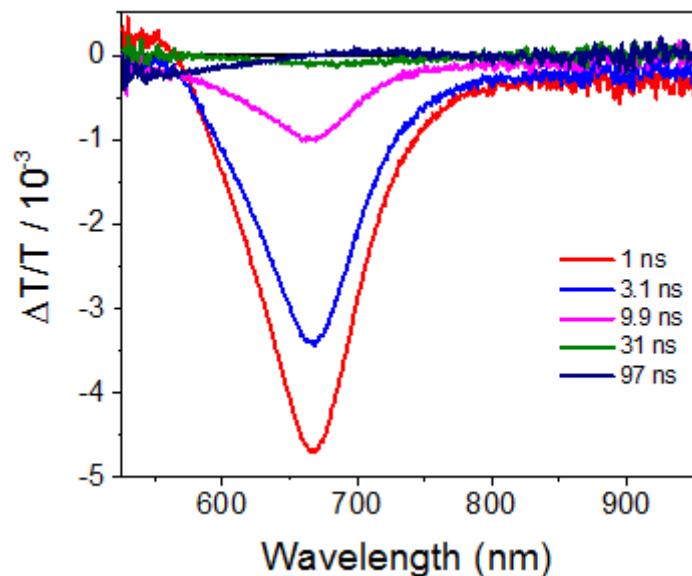


Fig. S13. Transient absorption spectra of BTPh_2 (20 mg/L in CH_3CN) under N_2 at different times.
Excitation wavelength 355 nm.

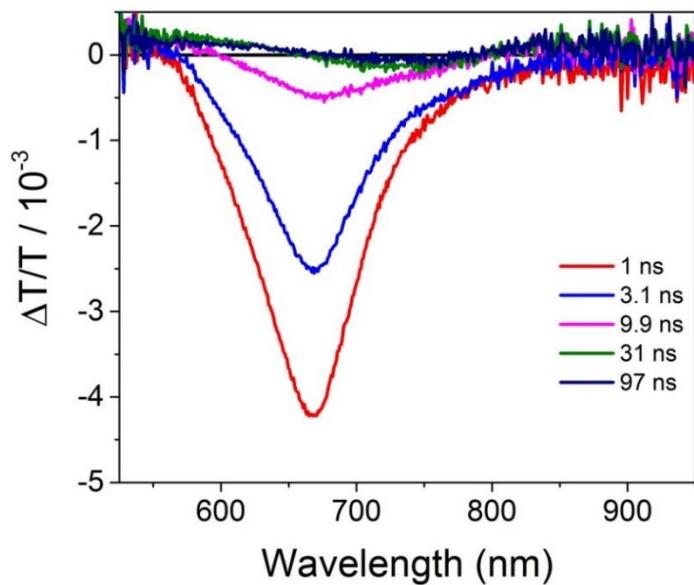


Fig. S14. Transient absorption spectra of BTPh₂ (20 mg/L in CH₃CN) under O₂ at different times.
Excitation wavelength 355 nm.

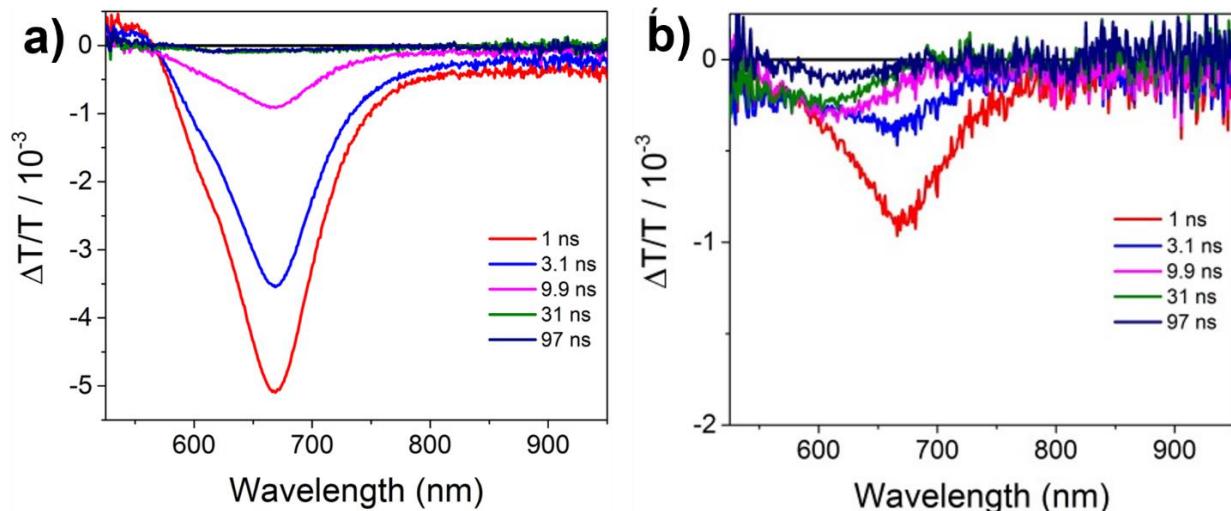


Fig. S15. Transient absorption spectra of BTPh₂ (20 mg/L) and *trans*-anethole (2 mM for a) and 20 mM for b) under N₂ atmosphere in CH₃CN at different times. Excitation wavelength 355 nm.

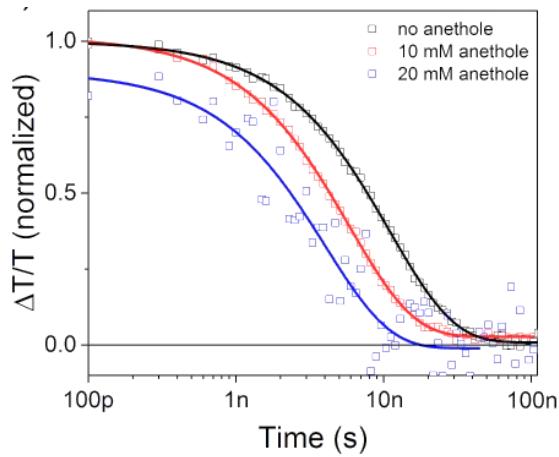


Fig. S16. Time profiles of the absorbance decay at 675 nm of pure BTPH₂ (black), with 10 mM *trans*-anethole (red) and 20 mM *trans*-anethole (blue) in CH₃CN.

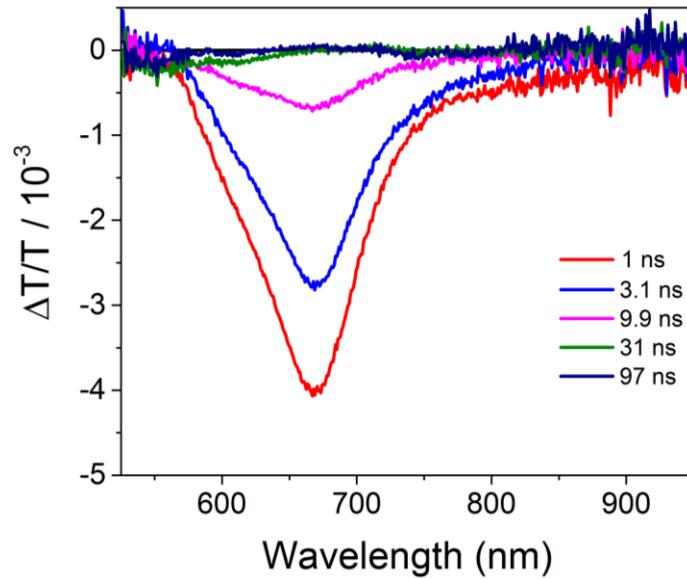


Fig. S17. Transient absorption spectra of BTPH₂ (20 mg/L) and *trans*-anethole (2 mM) under O₂ atmosphere in CH₃CN at different times. Excitation wavelength 355 nm.

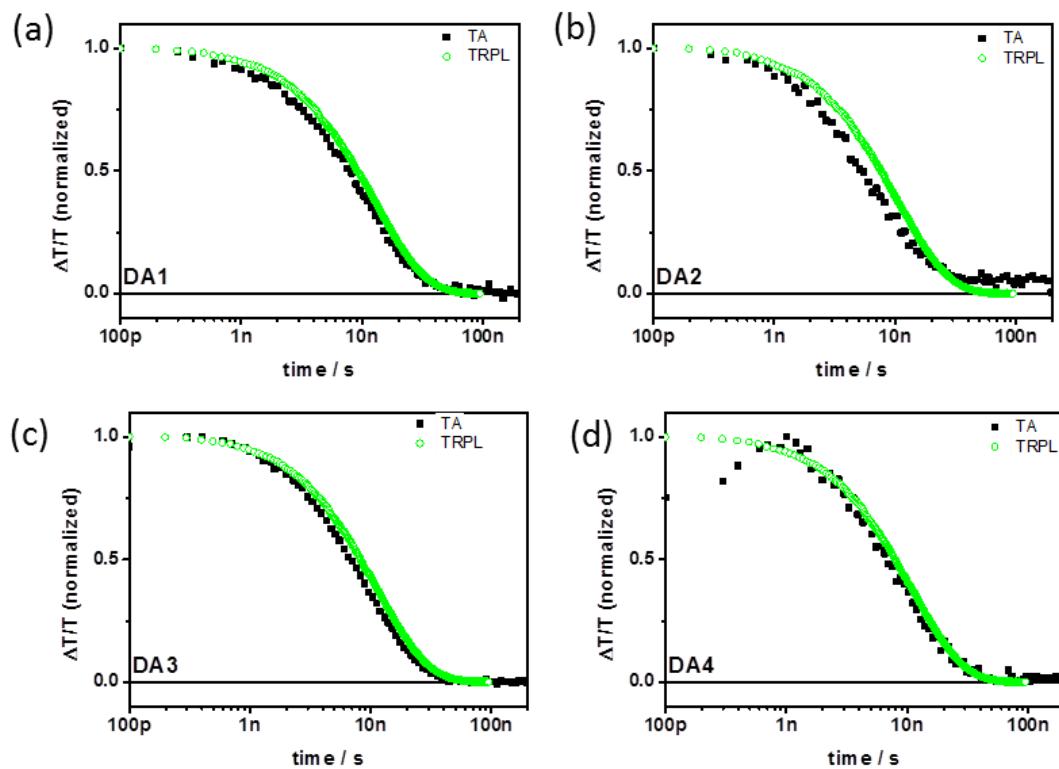
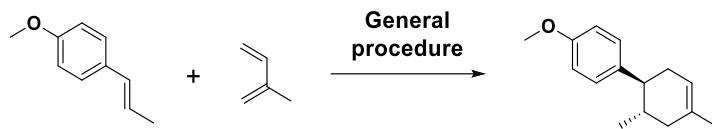


Fig. S18. Comparison of decay profiles of transient absorption (TA) at 660 – 690 nm and time-resolved photoluminescence (TRPL) at 480-520 nm for the Diels-Alder reaction. (a) pure BTPH_2 (20 mg/L in CH_3CN) under N_2 , (b) pure BTPH_2 (20 mg/L in CH_3CN) under O_2 , (c) BTPH_2 (20 mg/L) and *trans*-anethole (2 mM) under N_2 atmosphere in CH_3CN , (d) BTPH_2 (20 mg/L) and *trans*-anethole (2 mM) under O_2 atmosphere in CH_3CN .

Table S2. Excited state life time of the measured mixtures

Entry	Sample name	Fluorescence life time (ns)	Transient life time (ns)
1	BTPH_2 (CH_3CN)	12.9	11.1
2	$\text{BTPH}_2 + \text{O}_2$	10.7	7.5
3	$\text{BTPH}_2 + \text{anethole}$	11.5	9.1
4	$\text{BTPH}_2 + \text{anethole} + \text{O}_2$	11.2	9.4

Part F: Compound Characterization

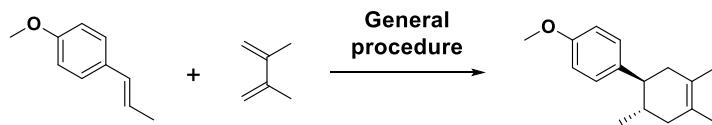


(1S,2S)-4'-methoxy-2,4-dimethyl-1,2,3,6-tetrahydro-1,1'-biphenyl: According to the general procedure, a dried vial with a magnetic stir bar was charged with *trans*-anethole (35.6 mg, 0.24 mmol), isoprene (240 μ L, 2.4 mmol), catalyst (2.4 mg), $MgSO_4$ (15mg) and nitromethane (2.4 mL). After the reaction was finished, the mixture was worked up according to the protocol and purified by chromatography on silica with hexane/ethyl acetate (12/1) as the elution to afford the pure cycloadduct as clear oil.

R_f (hexane/ethyl acetate 12/1): 0.76

1H NMR ($CDCl_3$, 300 MHz): δ 6.99 (d, $J=9$ Hz, 2H), 6.75 (d, $J=9$ Hz, 2H), 5.36 (bs, 1H), 3.71 (s, 3H), 2.21 (m, 1H), 2.11 (m, 1H), 2.04 (m, 1H), 1.98 (m, 1H), 1.81 (m, 1H), 1.71 (m, 1H), 1.61 (s, 3H), 0.62 (d, $J=6$ Hz, 3H);

^{13}C NMR ($CDCl_3$, 75 MHz): δ 157.8, 138.2, 133.8, 128.5, 120.9, 113.7, 55.2, 47.0, 39.9, 35.3, 34.0, 23.4, 20.2.



(1S,2S)-4'-methoxy-2,4,5-trimethyl-1,2,3,6-tetrahydro-1,1'-biphenyl: According to the general procedure, a dried vial with a magnetic stir bar was charged with *trans*-anethole (35.6 mg, 0.24 mmol), 2,3-dimethylbuta-1,3-diene (81.5 μ L, 0.72 mmol), catalyst (2.4 mg), $MgSO_4$ (15mg) and nitromethane (2.4 mL). After the reaction was finished, the mixture was worked up according to the protocol and purified by chromatography on silica with hexane/dichloromethane (1/1) as the elution to afford the pure cycloadduct as clear oil.

R_f (hexane/dichloromethane 1/1): 0.81

1H NMR ($CDCl_3$, 300 MHz): δ 7.02 (d, $J=9$ Hz, 2H), 6.78 (d, $J=12$ Hz, 2H), 3.71 (s, 3H), 2.33 (td, $J=6$, 12 Hz, 1H), 2.07 (m, 1H), 1.98 (m, 2H), 1.77 (m, 2H), 1.56 (s, 3H), 1.54 (s, 3H), 0.63 (d, $J=9$ Hz, 3H);

^{13}C NMR ($CDCl_3$, 75 MHz): δ 157.6, 138.2, 128.5, 125.5, 125.4, 113.7, 55.2, 47.8, 41.9, 41.7, 34.3, 20.1, 18.8, 18.7.

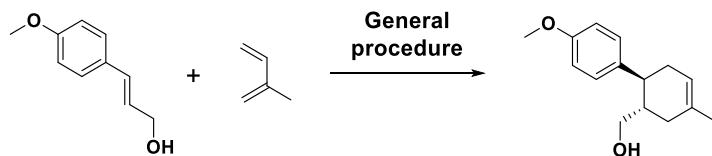


(1R,2S)-4'-methoxy-2,4,6,6-tetramethyl-1,2,3,6-tetrahydro-1,1'-biphenyl: According to the general procedure, a dried vial with a magnetic stir bar was charged with *trans*-anethole (35.6 mg, 0.24 mmol), 2,4-dimethylpenta-1,3-diene (93.1 μ L, 0.72 mmol), catalyst (2.4 mg), $MgSO_4$ (15mg) and nitromethane (2.4 mL). After the reaction was finished, the mixture was worked up according to the protocol and purified by chromatography on silica with hexane/ethyl acetate (25/1) as the elution to afford the pure cycloadduct as clear oil.

R_f (hexane/ethyl acetate 25/1): 0.61

1H NMR ($CDCl_3$, 300 MHz): δ 6.97 (dd, $J=6, 18$ Hz, 2H), 6.76 (dd, $J=9, 18$ Hz, 2H), 5.12 (s, 1H), 3.72 (s, 3H), 2.11 (d, 9 Hz, 1H), 2.05 (m, 1H), 1.98 (m, 1H), 1.70 (m, 1H), 1.59 (s, 3H), 0.75 (s, 3H), 0.70 (s, 3H), 0.64 (d, $J=6$ Hz, 3H);

^{13}C NMR ($CDCl_3$, 75 MHz): δ 157.7, 133.3, 130.3, 129.3, 113.2, 112.2, 57.4, 56.1, 40.5, 36.3, 30.0, 28.8, 24.6, 23.3, 20.8.

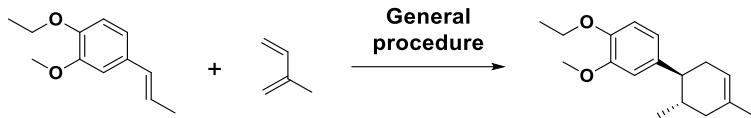


((1S,2S)-4'-methoxy-4-methyl-1,2,3,6-tetrahydro-[1,1'-biphenyl]-2-yl)methanol: According to the general procedure, a dried vial with a magnetic stir bar was charged with (E)-3-(4-methoxyphenyl)prop-2-en-1-ol (39.4 mg, 0.24 mmol), isoprene (240 μ L, 2.4 mmol), catalyst (2.4 mg), $MgSO_4$ (15mg) and nitromethane (2.4 mL). After the reaction was finished, the mixture was worked up according to the protocol and purified by chromatography on silica with hexane/ethyl acetate (4/1) as the elution to afford the pure cycloadduct as clear oil.

R_f (hexane/ethyl acetate 4/1): 0.56

1H NMR ($CDCl_3$, 300 MHz): δ 7.06 (d, $J=9$ Hz, 2H), 6.78 (d, $J=9$ Hz, 2H), 5.37 (bs, 1H), 3.70 (s, 3H), 3.33 (dd, $J=3, 9$ Hz, 1H), 3.22 (dd, 6, 12 Hz, 1H), 2.45 (m, 1H), 2.14 (m, 1H), 2.12 (m, 1H), 2.03 (t, $J=12$ Hz, 1H), 1.93 (m, 1H), 1.90 (m, 1H), 1.64 (s, 3H);

^{13}C NMR ($CDCl_3$, 75 MHz): δ 158.1, 137.0, 133.4, 128.4, 120.6, 114.1, 65.9, 55.2, 42.2, 41.5, 34.7, 33.9, 23.5.

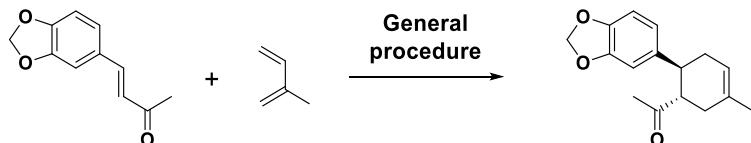


(1*S*,2*S*)-4'-ethoxy-3'-methoxy-2,4-dimethyl-1,2,3,6-tetrahydro-1,1'-biphenyl: According to the general procedure, a dried vial with a magnetic stir bar was charged with (E)-1-ethoxy-2-methoxy-4-(prop-1-en-1-yl)benzene (48.1 mg, 0.25 mmol), isoprene (250 μ L, 2.5 mmol), catalyst (2.4 mg), $MgSO_4$ (15mg) and nitromethane (2.4 mL). After the reaction was finished, the mixture was worked up according to the protocol and purified by chromatography on silica with hexane/ethyl acetate (12/1) as the elution to afford the pure cycloadduct as clear oil.

R_f (hexane/ethyl acetate 12/1): 0.46

1H NMR ($CDCl_3$, 300 MHz): δ 6.71 (m, 1H), 6.63 (m, 2H), 5.37 (bs, 1H), 4.02 (m, 2H), 3.78 (s, 3H), 2.21 (m, 1H), 2.14 (m, 1H), 2.09 (m, 1H), 1.99 (m, 1H), 1.83 (m, 1H), 1.72 (m, 1H), 1.62 (s, 3H), 1.38 (t, $J=6$ Hz, 3H), 0.66 (d, $J=6$ Hz, 3H);

^{13}C NMR ($CDCl_3$, 75 MHz): δ 149.1, 146.4, 138.8, 133.8, 120.8, 119.6, 112.6, 111.0, 64.3, 55.9, 47.4, 39.8, 35.2, 34.0, 23.4, 20.2, 14.9.

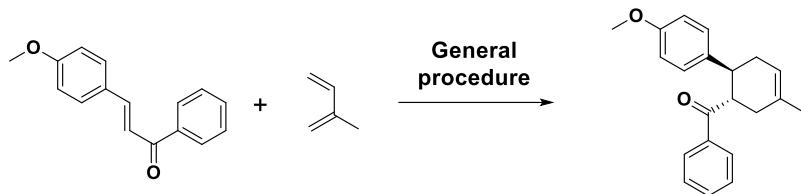


1-((1*S*,6*S*)-6-(benzo[d][1,3]dioxol-5-yl)-3-methylcyclohex-3-en-1-yl)ethan-1-one: According to the general procedure, a dried vial with a magnetic stir bar was charged with (E)-4-(benzo[d][1,3]dioxol-5-yl)but-3-en-2-one (45.6 mg, 0.24 mmol), isoprene (240 μ L, 2.4 mmol), catalyst (2.4 mg), $MgSO_4$ (15mg) and nitromethane (2.4 mL). After the reaction was finished, the mixture was worked up according to the protocol and purified by chromatography on silica with hexane/ethyl acetate (10/1) as the elution to afford the pure cycloadduct as clear oil.

R_f (hexane/ethyl acetate 10/1): 0.36

1H NMR ($CDCl_3$, 300 MHz): δ 6.72 (d, $J=9$ Hz, 1H), 6.70 (d, $J=1.6$ Hz, 1H), 6.67 (dd, $J=3, 9$ Hz, 1H), 5.93 (s, 2H), 5.48 (m, 1H), 2.99 (dt, $J=5.3, 10.3$ Hz, 1H), 2.87 (dt, $J=5.4, 11$ Hz, 1H), 2.31 (m, 1H), 2.26 (m, 1H), 2.17 (m, 1H), 2.07 (m, 1H), 1.90 (s, 3H), 1.73 (s, 3H);

¹³C NMR (CDCl₃, 75 MHz): δ 212.0, 147.7, 146.1, 138.0, 132.1, 120.6, 120.5, 108.3, 107.7, 100.9, 53.7, 42.5, 34.2, 33.3, 29.8, 23.2.



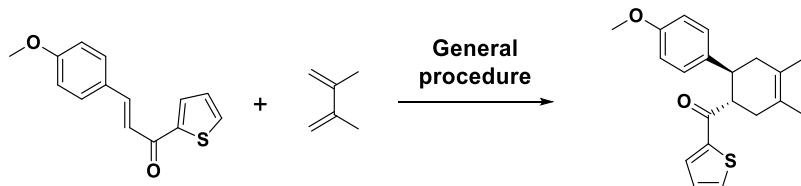
((1S,2S)-4'-methoxy-4-methyl-1,2,3,6-tetrahydro-[1,1'-biphenyl]-2-yl)(phenyl)methanone:

According to the general procedure, a dried vial with a magnetic stir bar was charged with (E)-3-(4-methoxyphenyl)-1-phenylprop-2-en-1-one (57.2 mg, 0.24 mmol), isoprene (240 μL, 2.4 mmol), catalyst (2.4 mg), MgSO₄ (15mg) and nitromethane (2.4 mL). After the reaction was finished, the mixture was worked up according to the protocol and purified by chromatography on silica with hexane/dichloromethane (1/1) as the elution to afford the pure cycloadduct as clear oil.

R_f (hexane/dichloromethane 1/1): 0.28

¹H NMR (CDCl₃, 300 MHz): δ 7.84 (d, J=7.4 Hz, 2H), 7.50 (t, J=6.8 Hz, 1H), 7.39 (t, J=7.5 Hz, 2H), 7.13 (d, J=8.5 Hz, 2H), 6.72 (d, J=8.4 Hz, 2H), 5.56 (bs, 1H), 3.98 (m, 1H), 3.71(s, 3H), 3.22 (m, 1H), 2.34 (m, 4H), 1.74 (s, 3H);

¹³C NMR (CDCl₃, 75 MHz): δ 203.8, 157.8, 137.4, 136.5, 132.7, 132.5, 128.41, 128.38, 128.0, 121.0, 113.7, 55.1, 47.4, 41.6, 35.2, 34.2, 23.2.



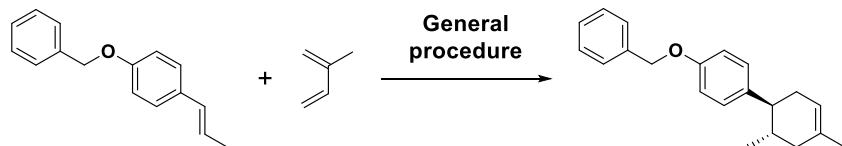
((1S,2S)-4'-methoxy-4,5-dimethyl-1,2,3,6-tetrahydro-[1,1'-biphenyl]-2-yl)(thiophen-2-yl)methanone:

According to the general procedure, a dried vial with a magnetic stir bar was charged with (E)-3-(4-methoxyphenyl)-1-(thiophen-2-yl)prop-2-en-1-one (58.6 mg, 0.24 mmol), 2,3-dimethylbuta-1,3-diene (271.6 μL, 2.4 mmol), catalyst (2.4 mg), MgSO₄ (15mg) and nitromethane (2.4 mL). After the reaction was finished, the mixture was worked up according to the protocol and purified by chromatography on silica with hexane/ethyl acetate (4/1) as the elution to afford the pure cycloadduct as clear oil.

R_f (hexane/ethyl acetate 4/1): 0.61

¹H NMR (CDCl₃, 300 MHz): δ 7.54 (d, J=4.1 Hz, 2H), 7.10 (d, J=8.2 Hz, 2H), 7.00 (t, J=4.1 Hz, 1H), 6.79 (d, J=8.5 Hz, 2H), 4.77 (d, J=15 Hz, 2H), 3.92 (m, 2H), 3.72 (s, 3H), 2.41 (t, J=9.7 Hz 1H), 2.02 (t, J=10 Hz, 1H), 1.65 (s, 3H), 1.07 (s, 3H);

¹³C NMR (CDCl₃, 75 MHz): δ 193.4, 158.2, 152.5, 143.8, 133.6, 131.9, 131.7, 129.3, 128.1, 113.5, 108.7, 55.2, 48.7, 45.0, 42.0, 35.0, 21.9, 18.8.

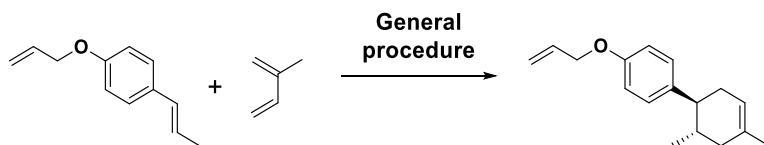


(1S,2S)-4'-(benzyloxy)-2,4-dimethyl-1,2,3,6-tetrahydro-1,1'-biphenyl: According to the general procedure, a dried vial with a magnetic stir bar was charged with (E)-1-(benzyloxy)-4-(prop-1-en-1-yl)benzene (53.8 mg, 0.24 mmol), isoprene (240 μL, 2.4 mmol), catalyst (2.4 mg), MgSO₄ (15mg) and nitromethane (2.4 mL). After the reaction was finished, the mixture was worked up according to the protocol and purified by chromatography on silica with hexane/ethyl acetate (8/1) as the elution to afford the pure cycloadduct as clear oil.

R_f (hexane/ethyl acetate 8/1): 0.37

¹H NMR (CDCl₃, 300 MHz) δ 7.42 (m, 5H), 7.13 (d, 2H), 6.96 (d, 2H), 5.49 (s, 1H), 5.08 (s, 2H), 2.34 (m, 1H), 2.23 (m, 2H), 2.13 (m, 1H), 1.94 (m, 1H), 1.86 (m, 1H), 1.74 (s, 3H), 0.76 (d, 3H).

¹³C NMR (CDCl₃, 75 MHz) δ 157.18, 138.56, 137.39, 133.87, 128.63, 127.97, 127.62, 121.01, 114.73, 70.12, 47.07, 39.94, 35.36, 34.04, 30.99, 23.50, 20.37.

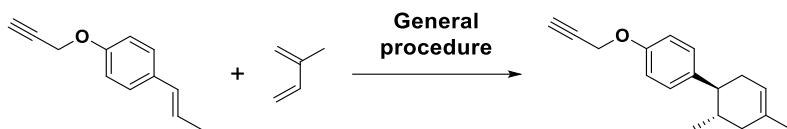


(1S,2S)-4'-(allyloxy)-2,4-dimethyl-1,2,3,6-tetrahydro-1,1'-biphenyl: According to the general procedure, a dried vial with a magnetic stir bar was charged with (E)-1-(allyloxy)-4-(prop-1-en-1-yl)benzene (41.8 mg, 0.24 mmol), isoprene (240 μL, 2.4 mmol), catalyst (2.4 mg), MgSO₄ (15mg) and nitromethane (2.4 mL). After the reaction was finished, the mixture was worked up according to the protocol and purified by chromatography on silica with hexane/ethyl acetate (8/1) as the elution to afford the pure cycloadduct as clear oil.

R_f (hexane/ethyl acetate 8/1): 0.32

¹H NMR (CDCl₃, 300 MHz) δ 7.11 (d, 2H), 6.89 (d, 2H), 6.10 (m, 1H), 5.44 (d, 2H), 5.30 (d, 2H), 4.55 (d, 2H), 2.33 (m, 1H), 2.22 (m, 2H), 2.12 (m, 1H), 1.91 (m, 1H), 1.85 (m, 1H), 1.73 (s, 3H), 0.74 (d, 3H).

¹³C NMR (CDCl₃, 75 MHz) δ 156.88, 138.37, 133.64, 128.51, 120.96, 117.45, 114.56, 68.85, 47.00, 39.89, 35.29, 34.00, 30.91, 23.43, 20.28.



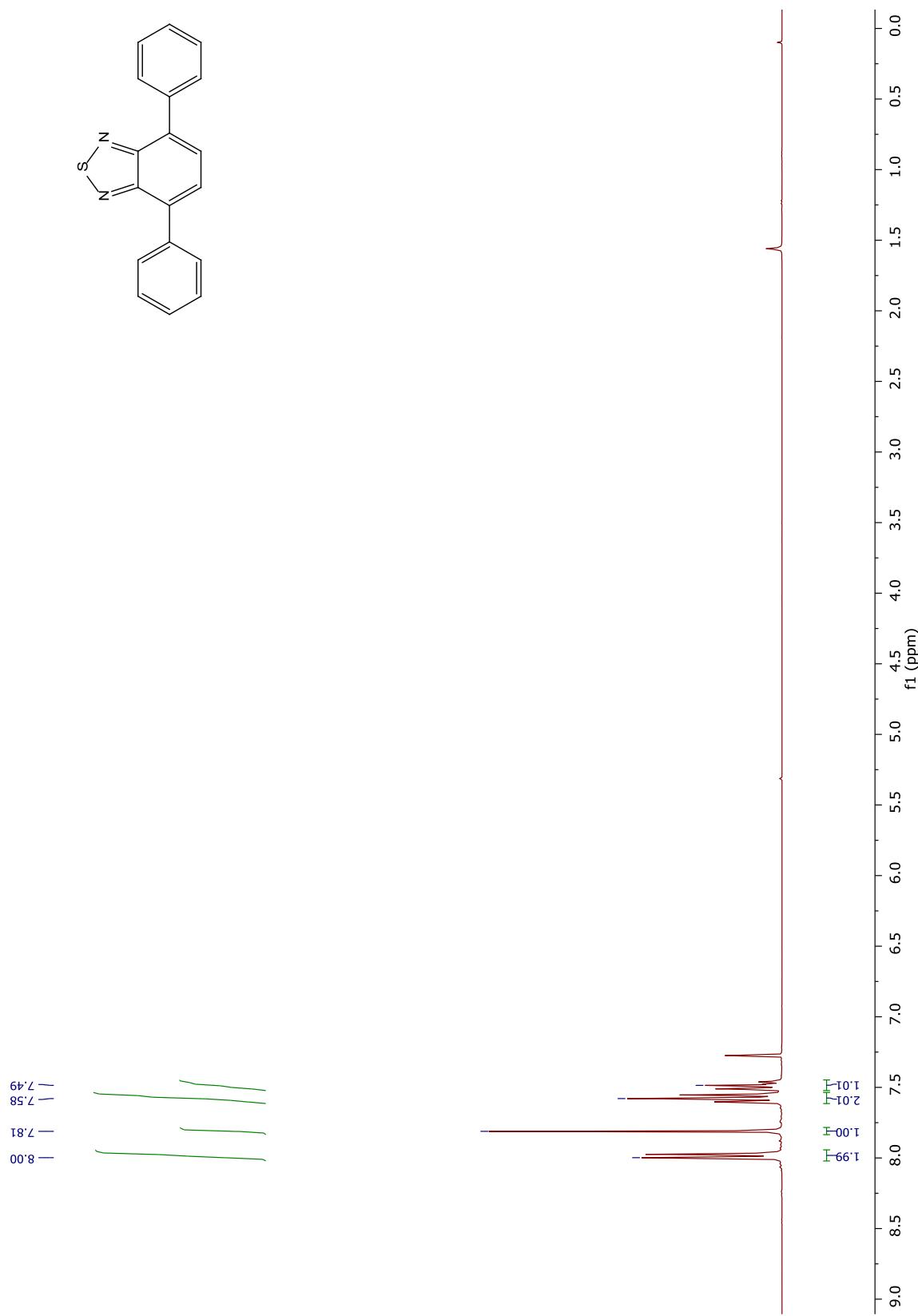
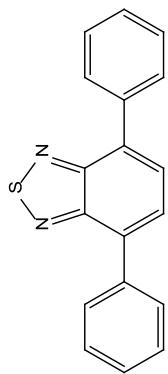
(1S,2S)-2,4-dimethyl-4'-(prop-2-yn-1-yloxy)-1,2,3,6-tetrahydro-1,1'-biphenyl: According to the general procedure, a dried vial with a magnetic stir bar was charged with (E)-1-(prop-1-en-1-yl)-4-(prop-2-yn-1-yloxy)benzene (41.3 mg, 0.24 mmol), isoprene (240 μL, 2.4 mmol), catalyst (2.4 mg), MgSO₄ (15mg) and nitromethane (2.4 mL). After the reaction was finished, the mixture was worked up according to the protocol and purified by chromatography on silica with hexane/ethyl acetate (8/1) as the elution to afford the pure cycloadduct as clear oil.

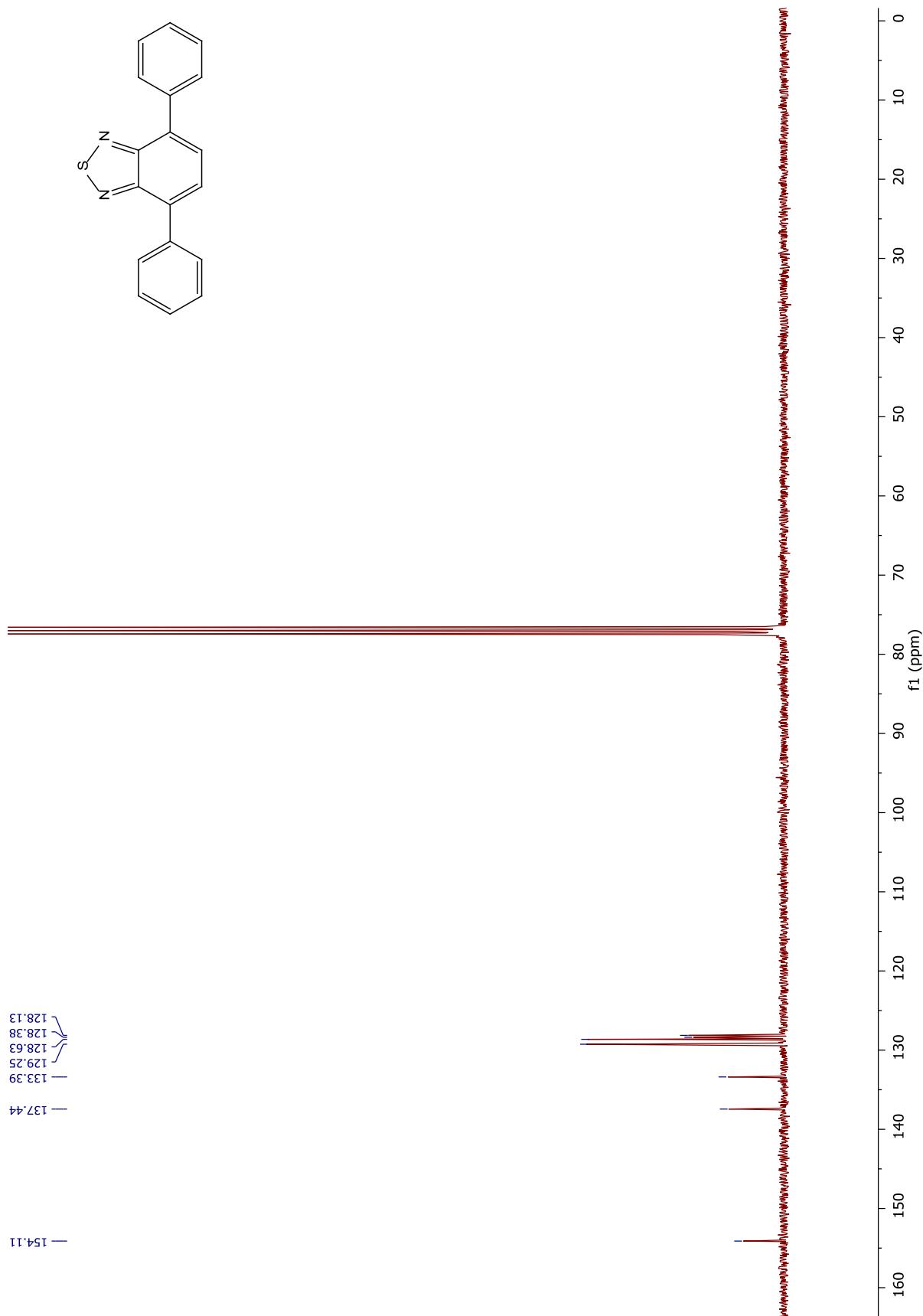
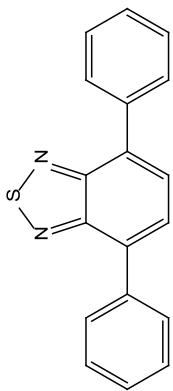
R_f (hexane/ethyl acetate 8/1): 0.30

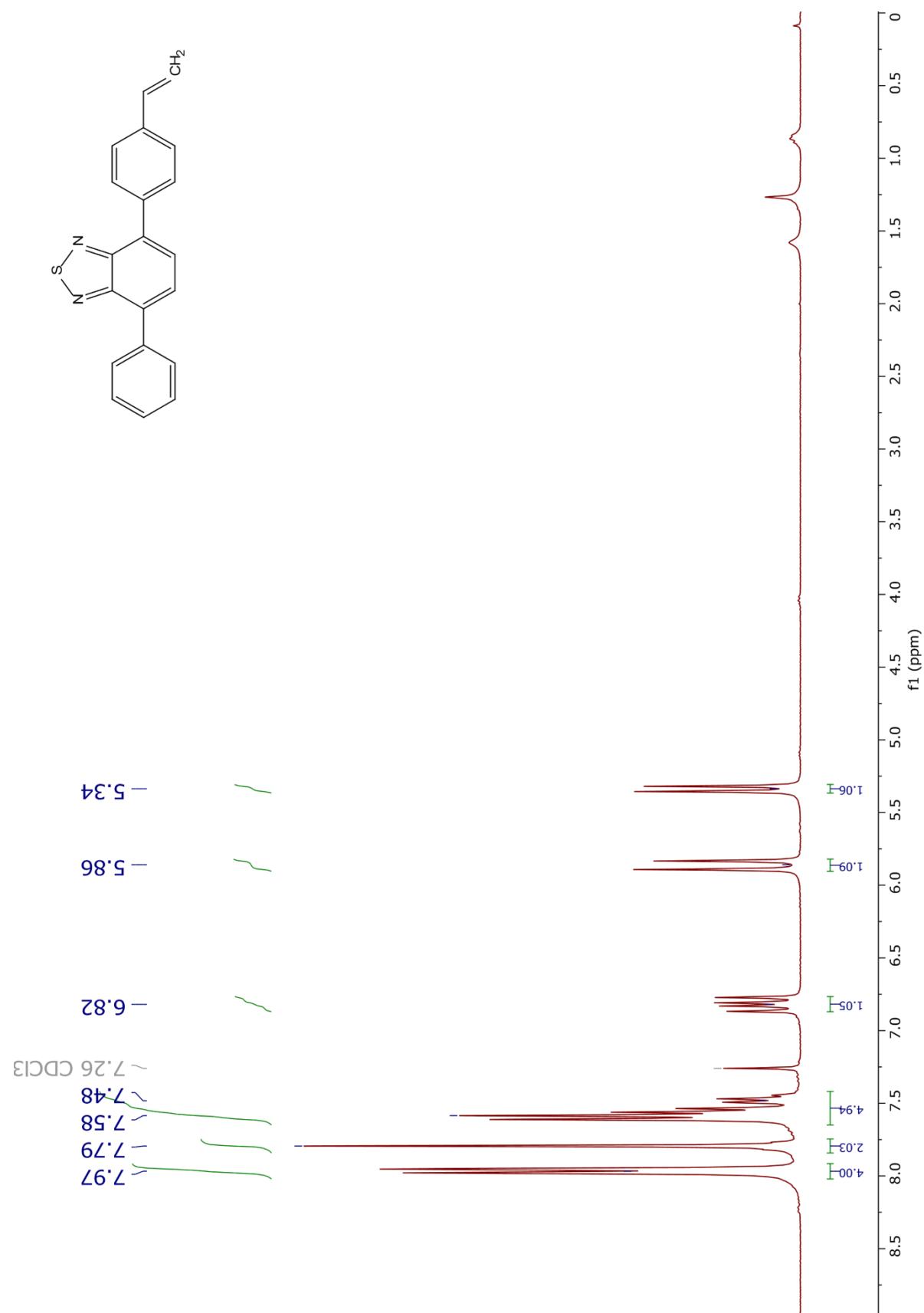
¹H NMR (CDCl₃, 300 MHz) δ 7.10 (d, 2H), 6.92 (d, 2H), 5.45 (s, 1H), 4.68 (d, 2H), 2.52 (s, 1H), 2.31 (m, 1H), 2.20 (m, 2H), 2.10 (m, 1H), 1.91 (m, 1H), 1.82 (m, 1H), 1.70 (s, 3H), 0.72 (d, 3H).

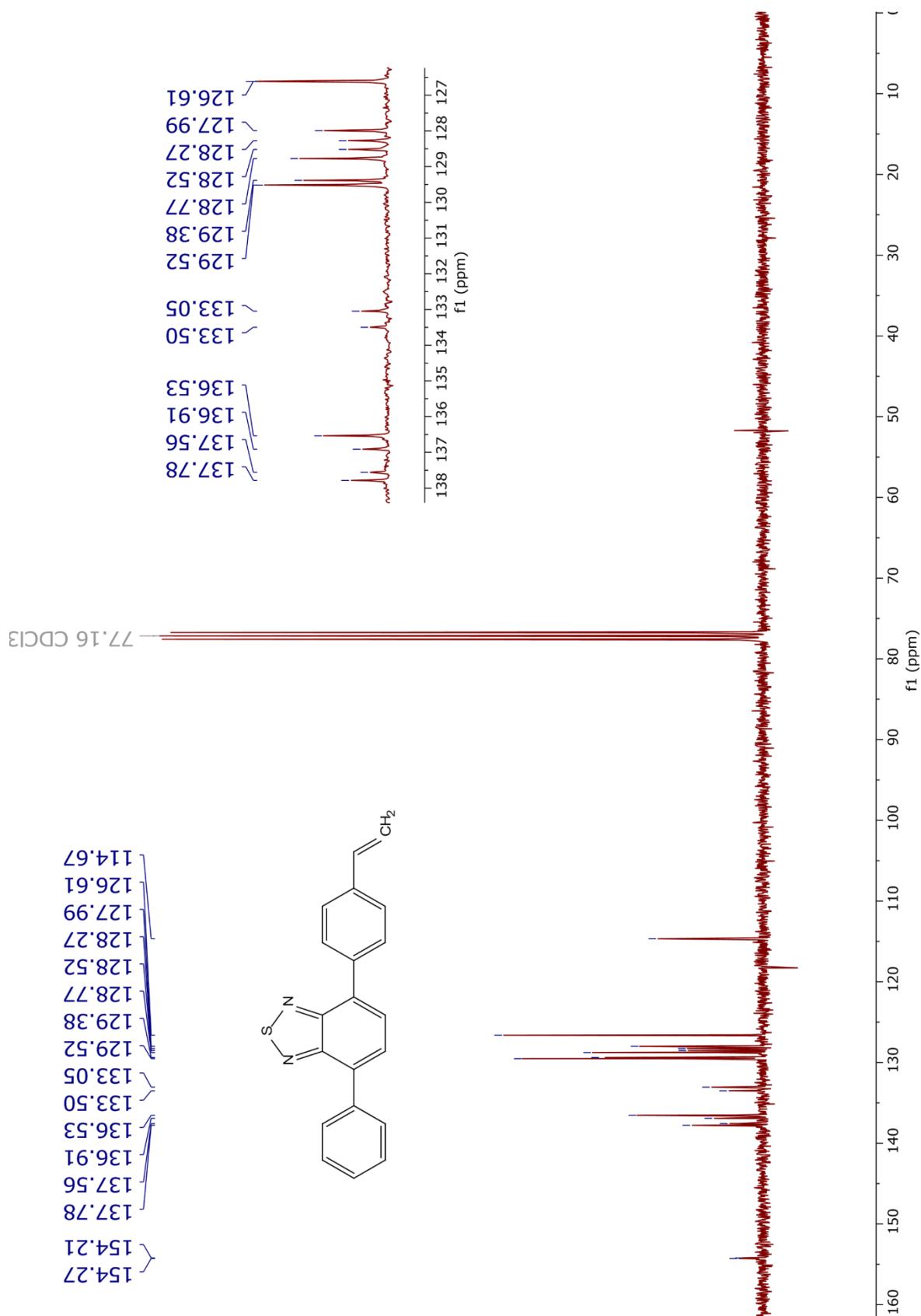
¹³C NMR (CDCl₃, 75 MHz) δ 155.95, 139.28, 133.93, 128.65, 120.98, 114.81, 79.01, 75.44, 55.99, 47.10, 39.93, 35.33, 34.05, 23.51, 20.37.

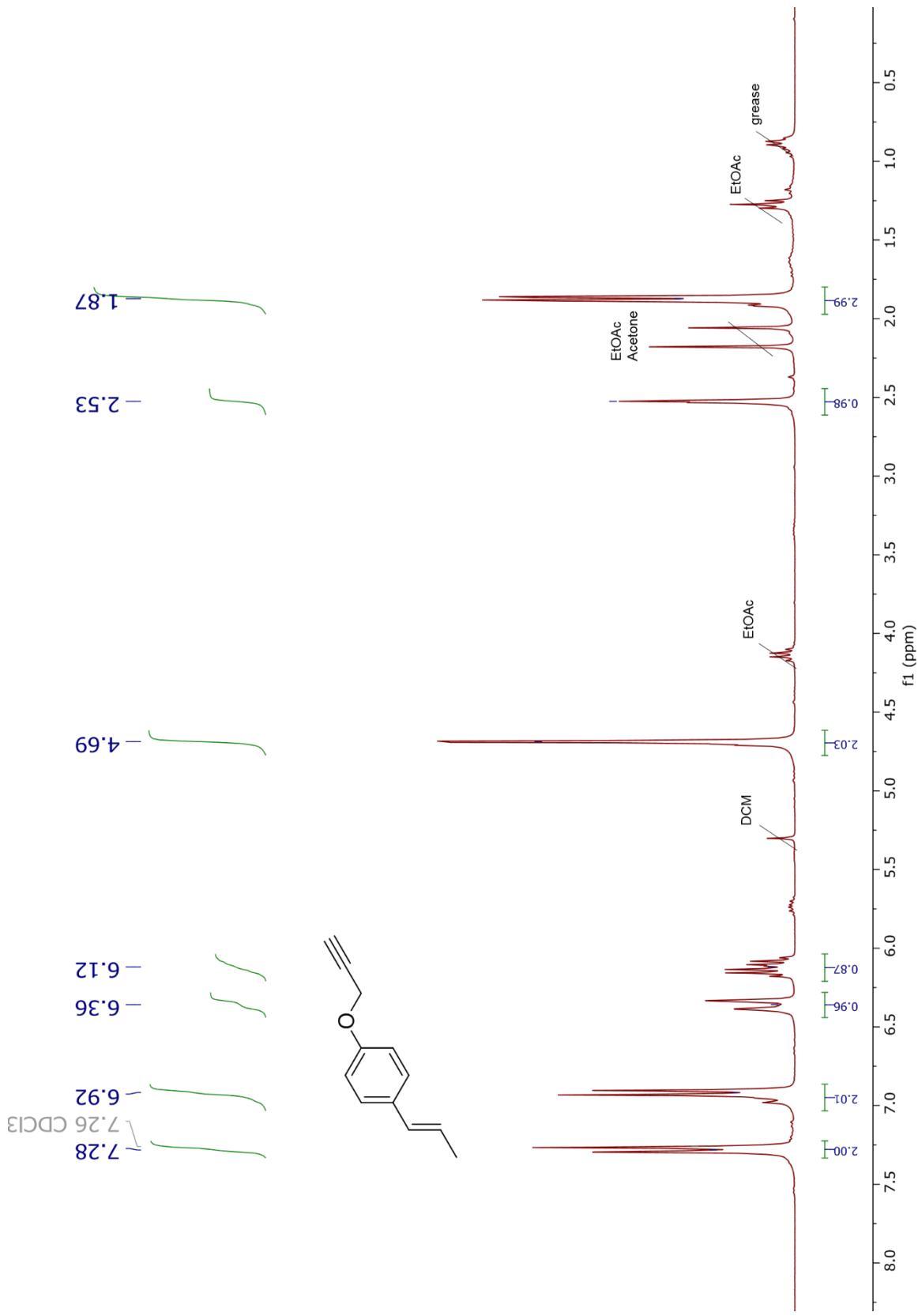
¹H and ¹³C NMR spectra

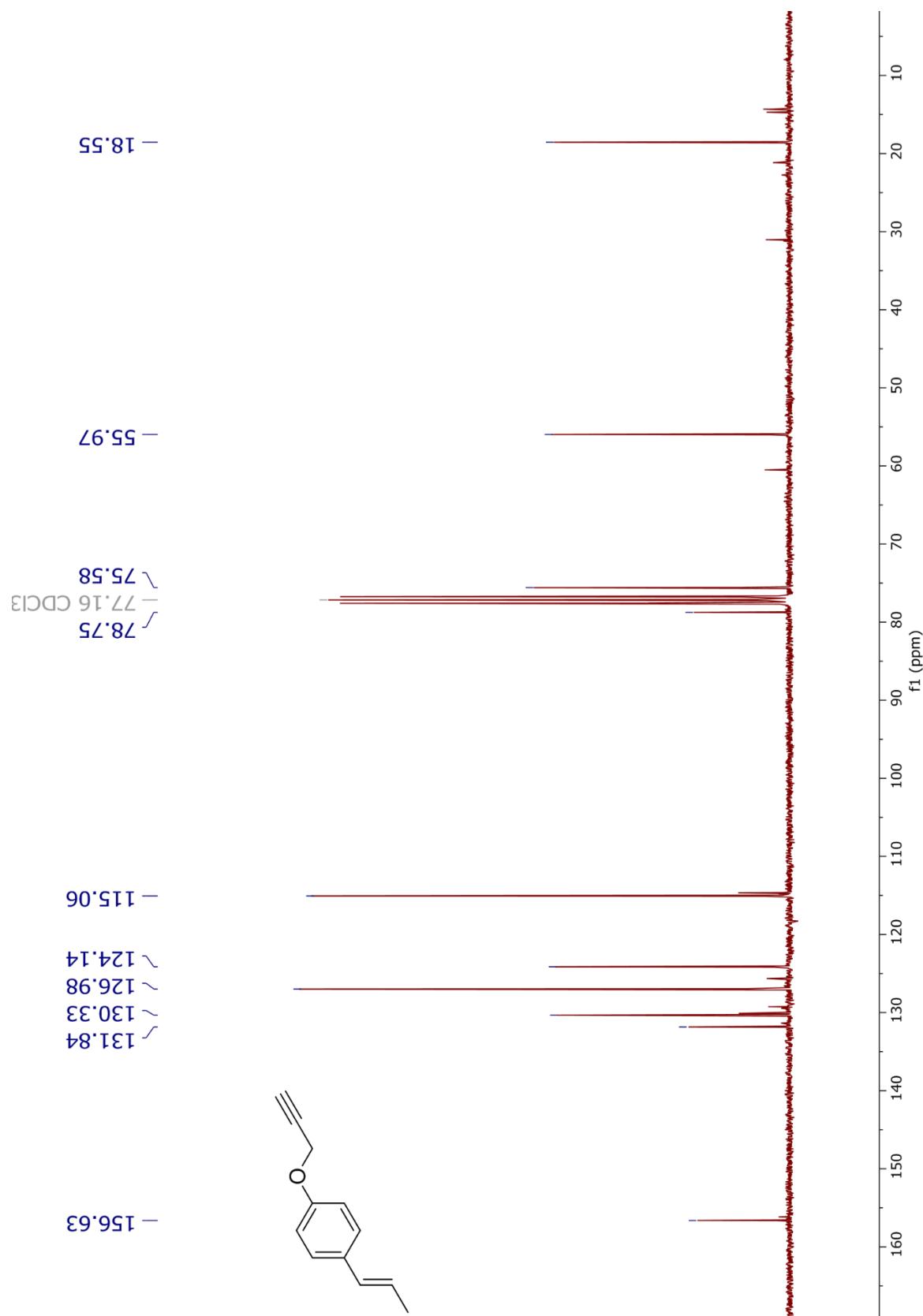


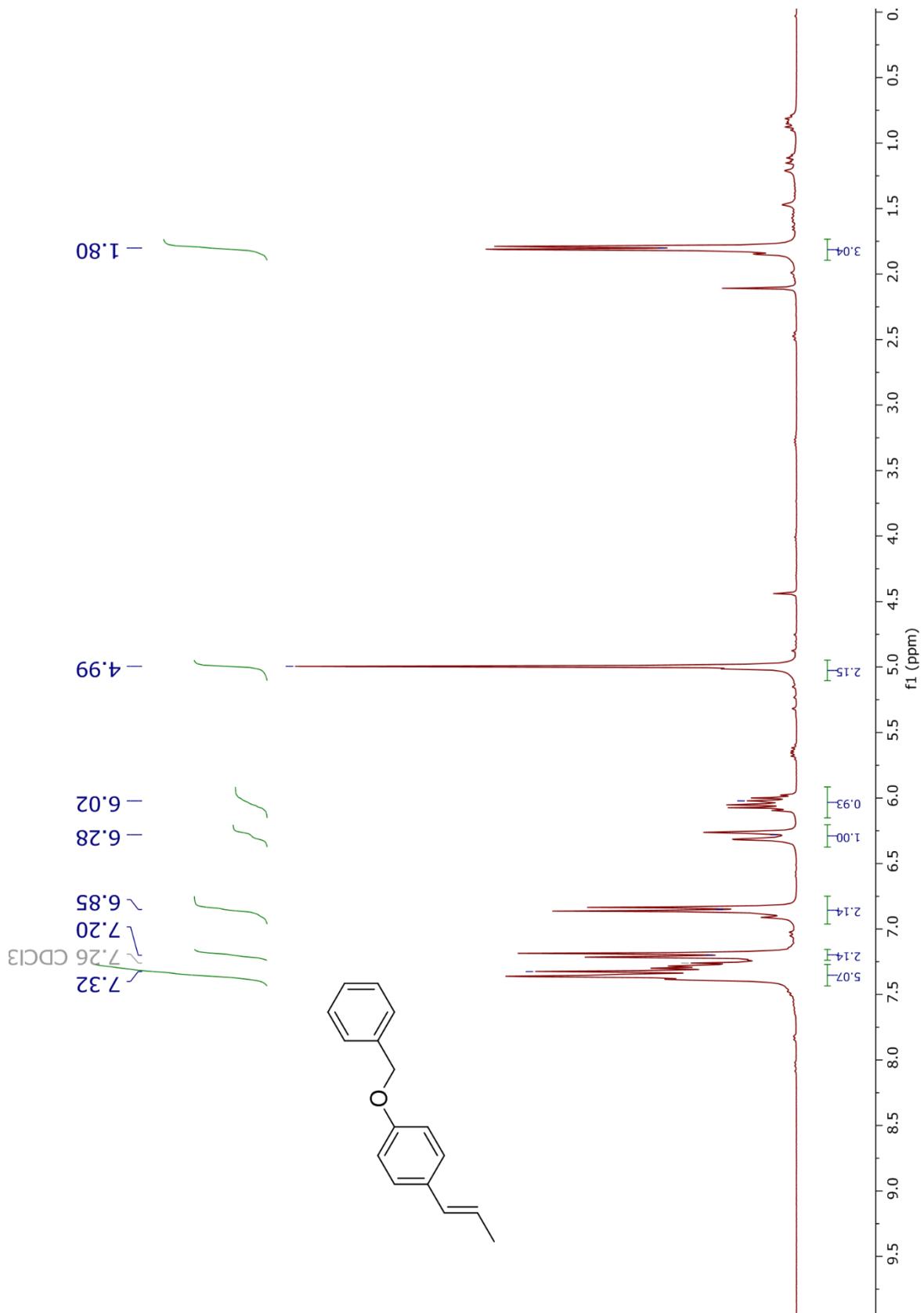


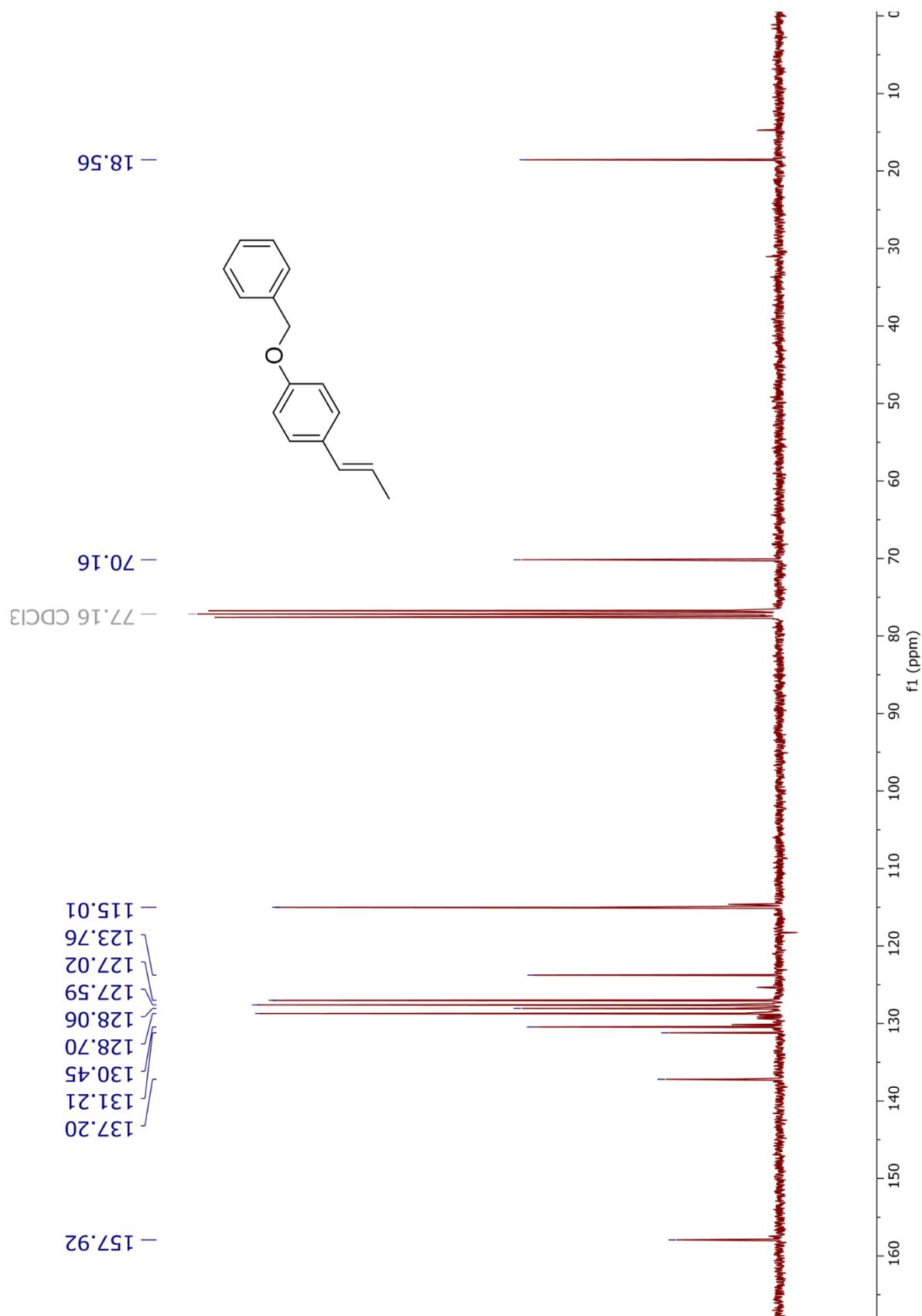


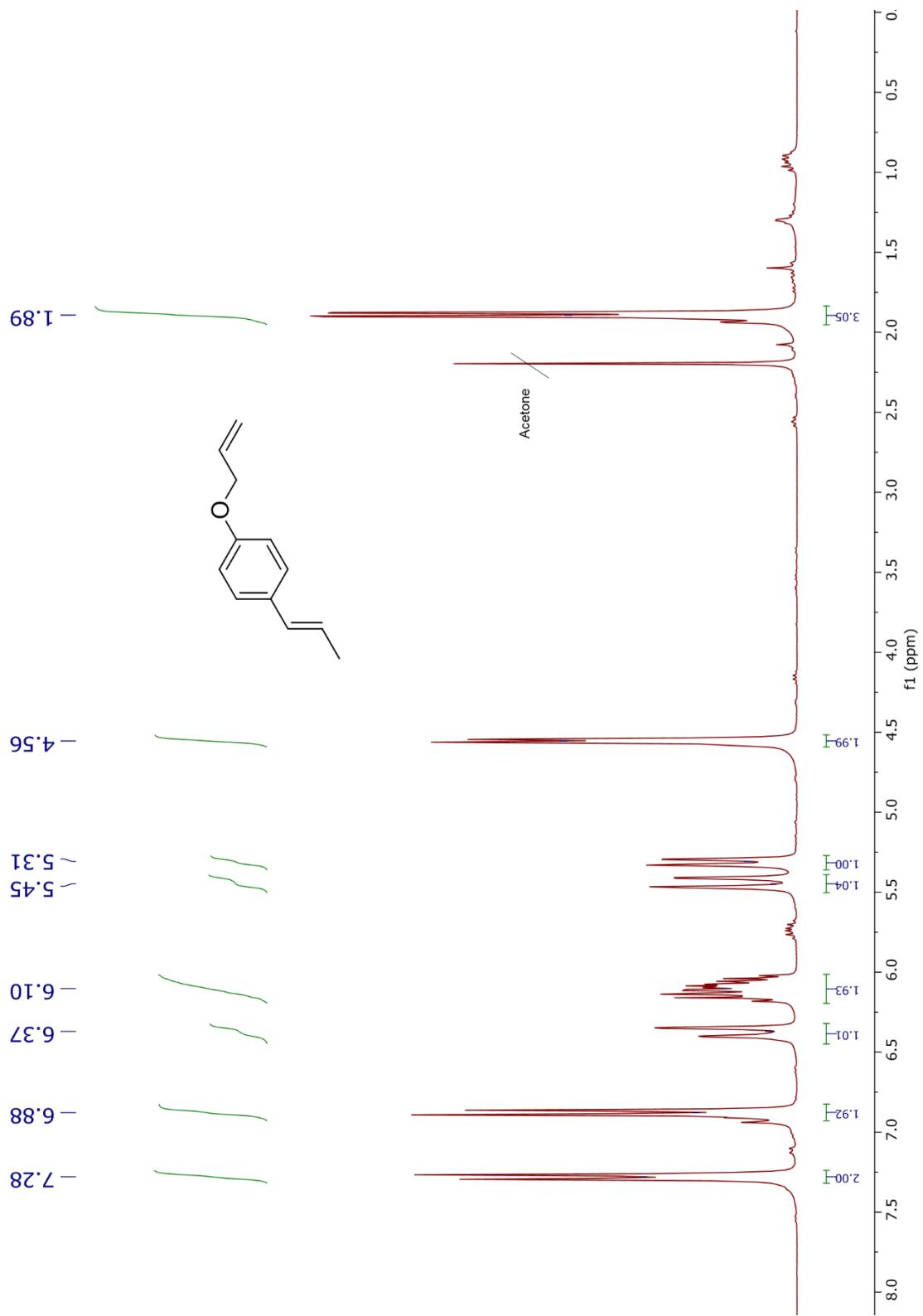




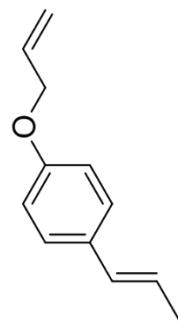








- 18.54

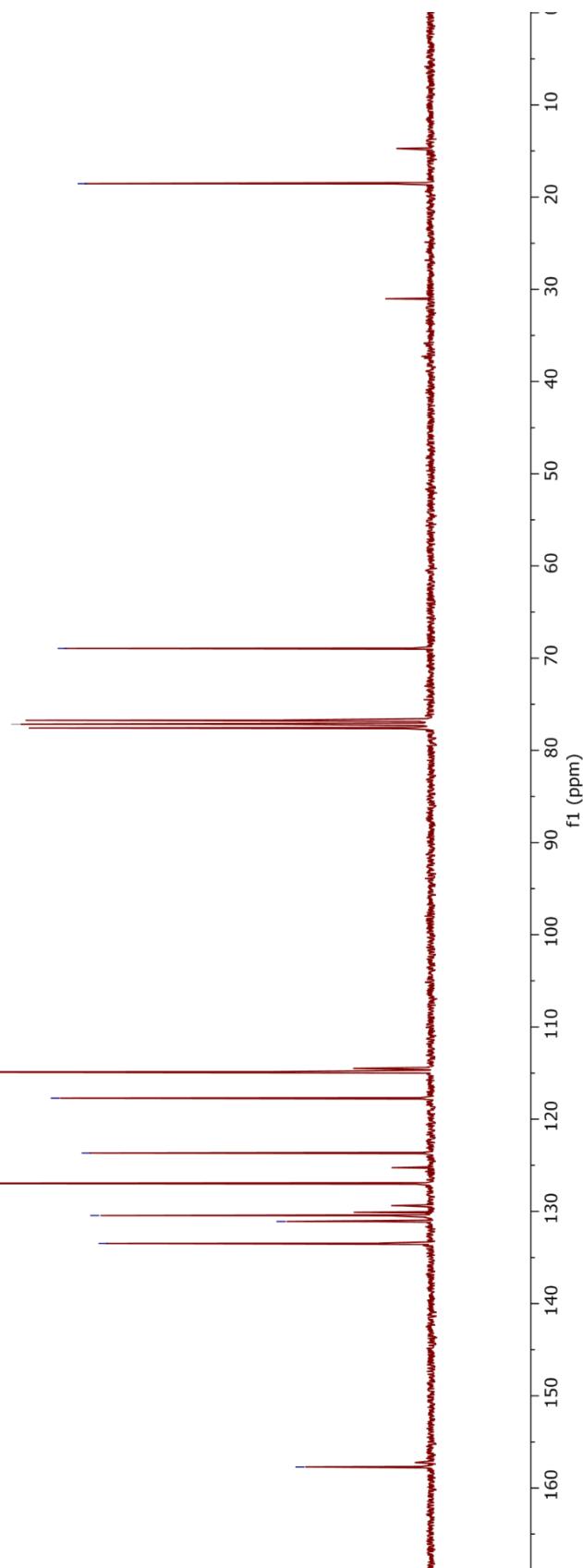


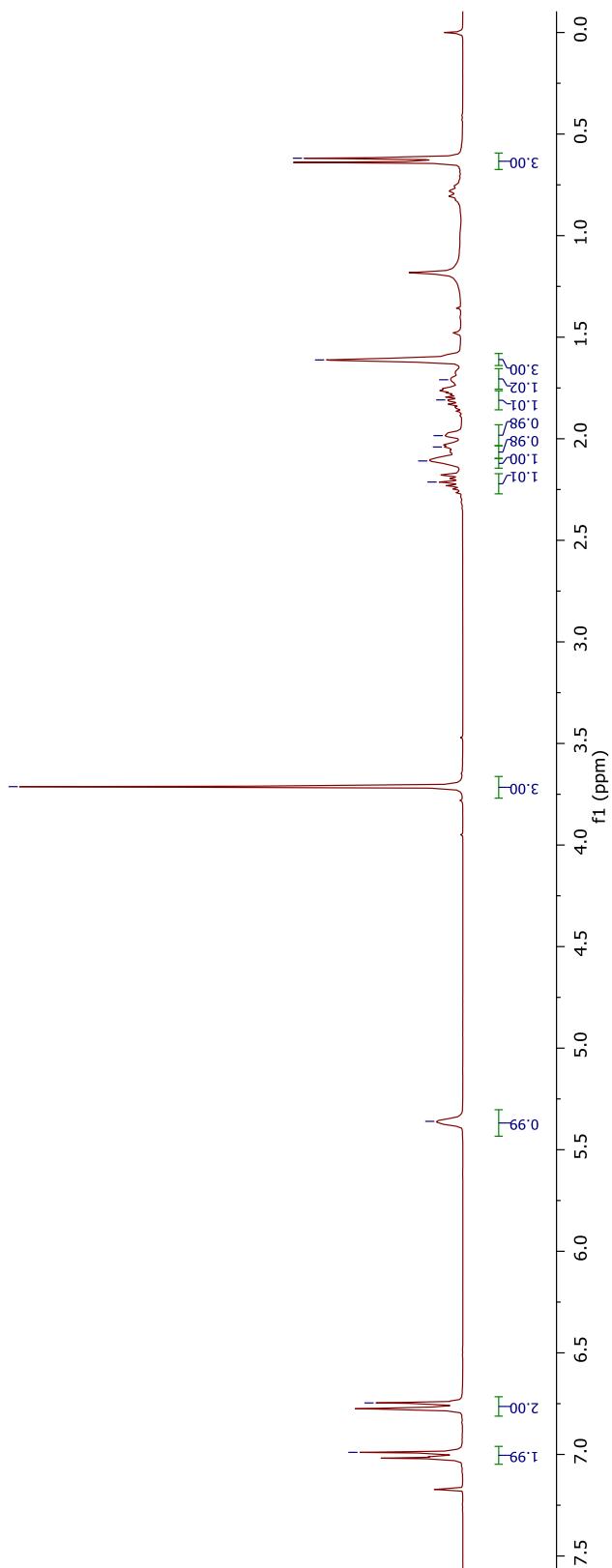
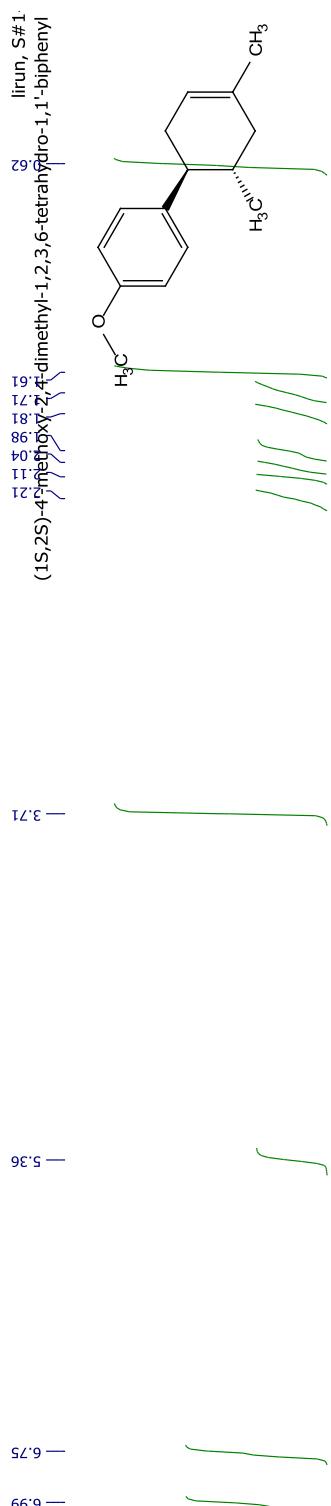
- 68.95

- 77.16 CDCl₃

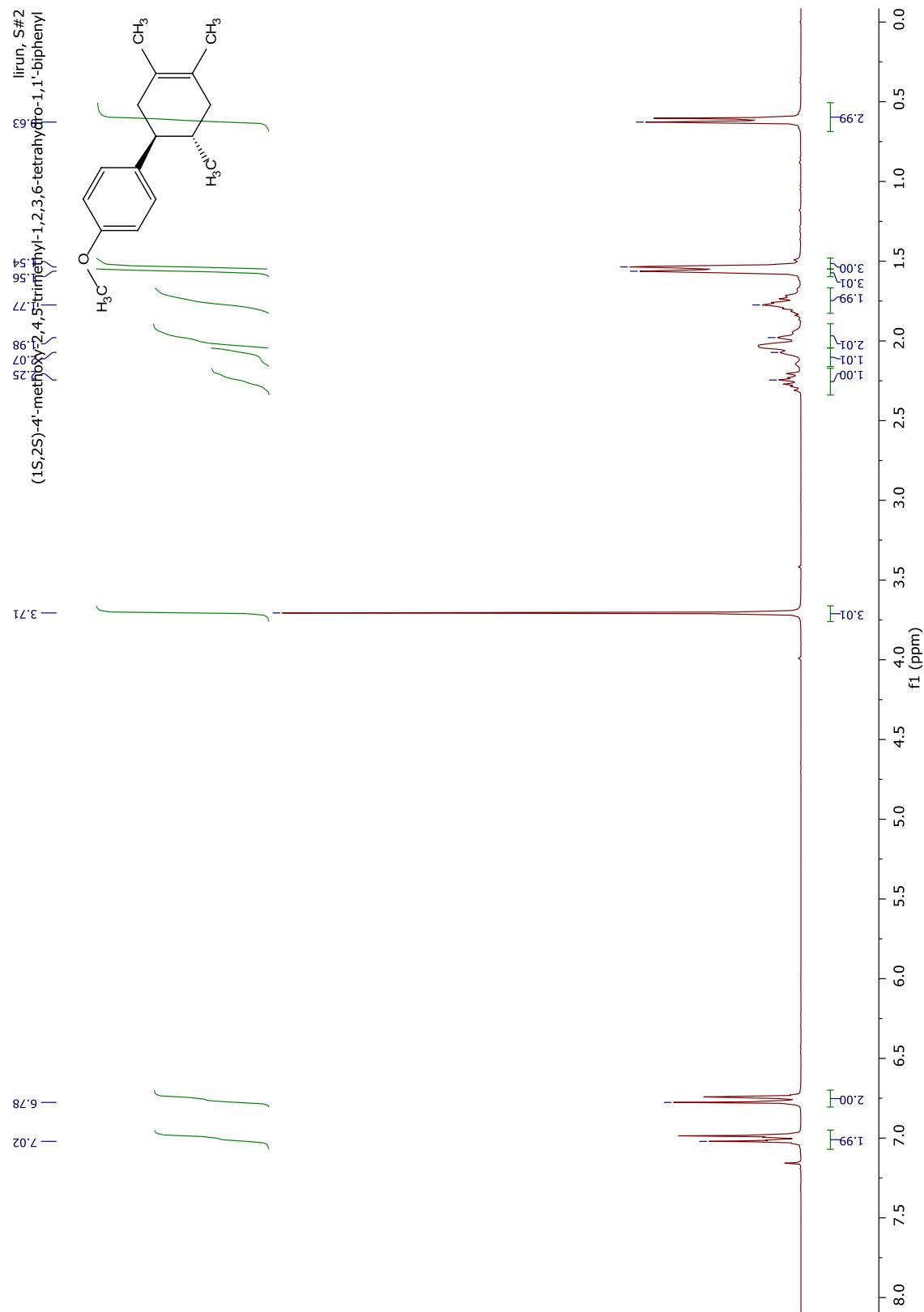
✓ 133.47
✓ 131.09
✓ 130.45
✓ 126.97
✓ 123.67
✓ 117.72
✓ 114.87

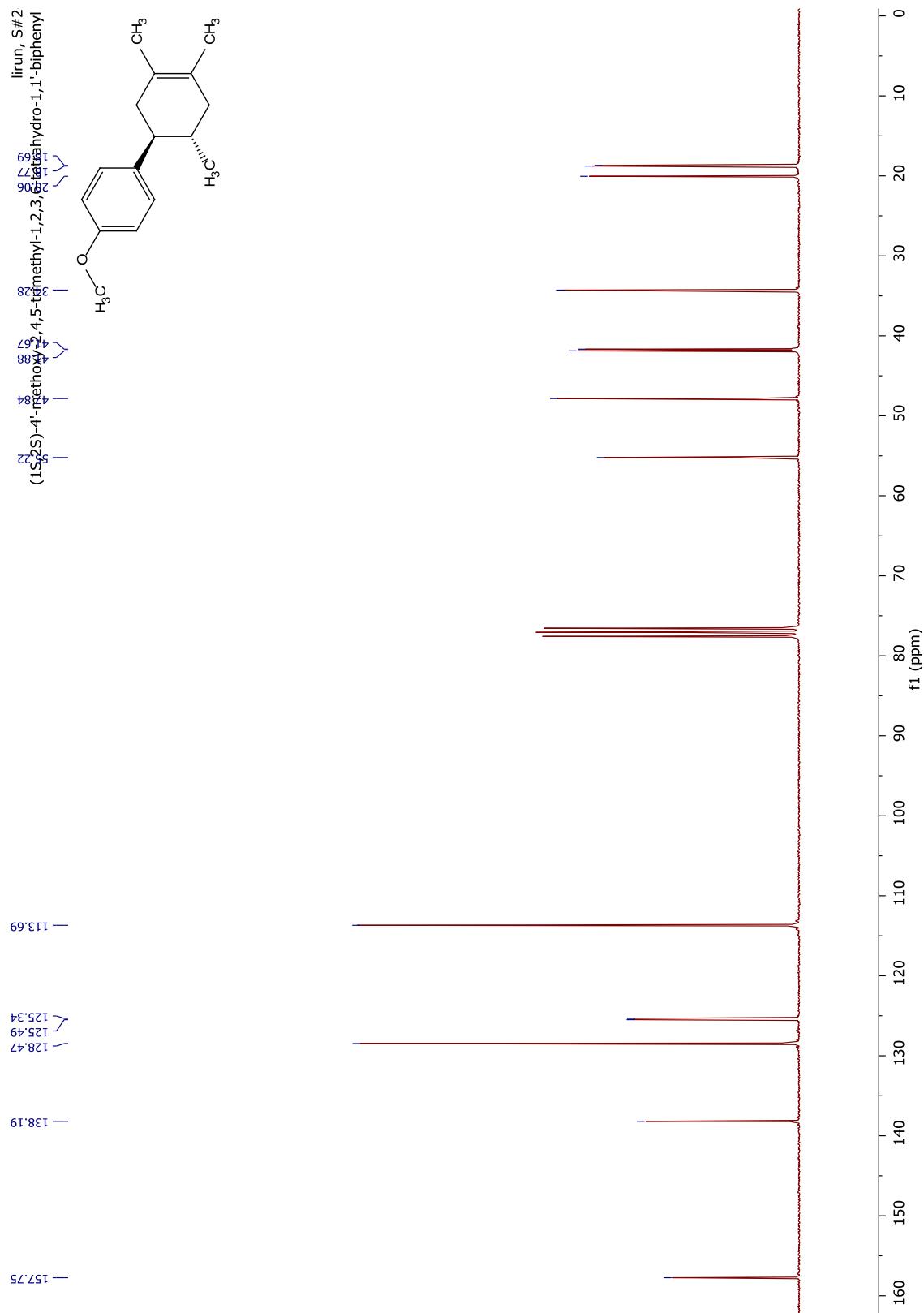
- 157.71

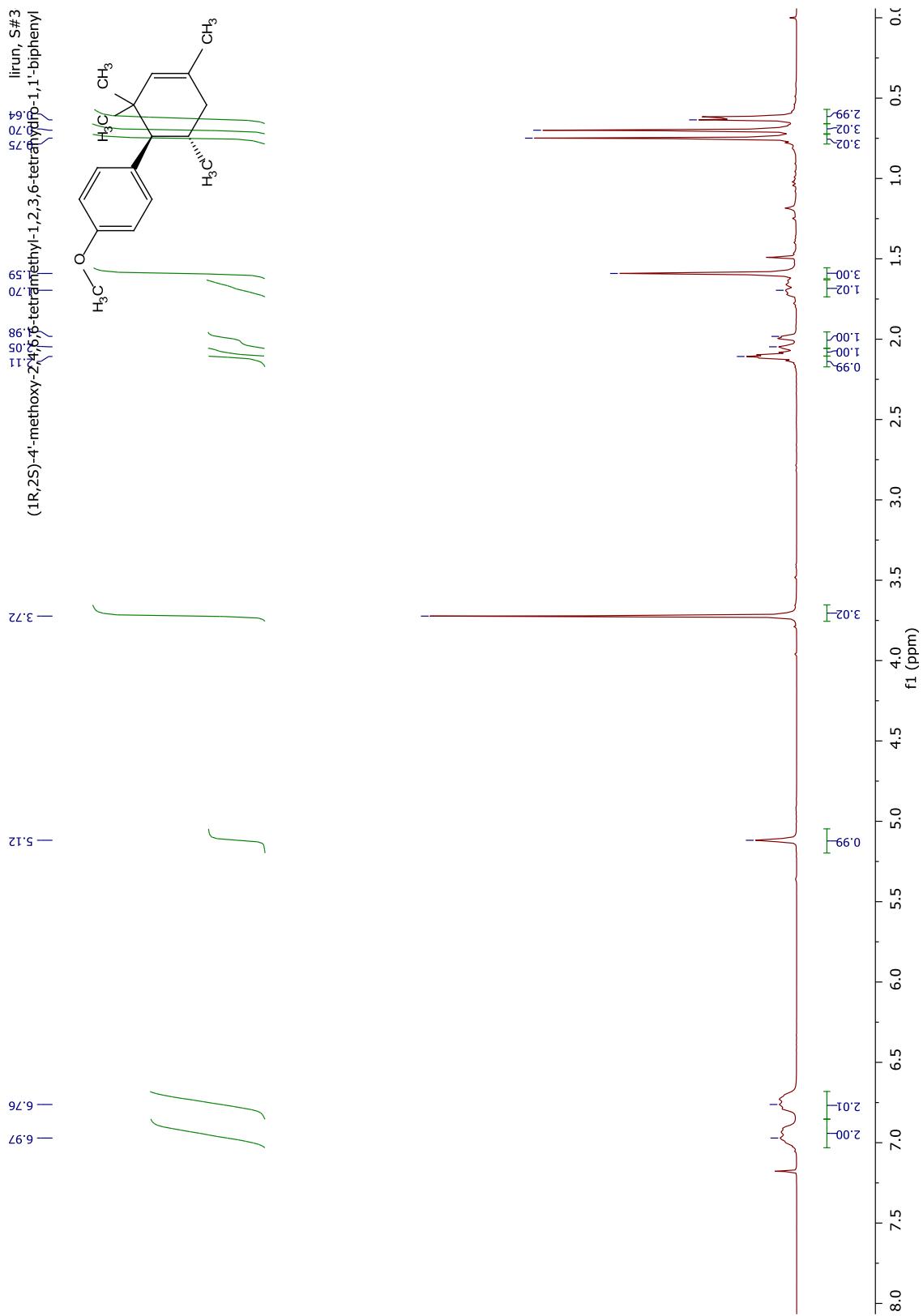


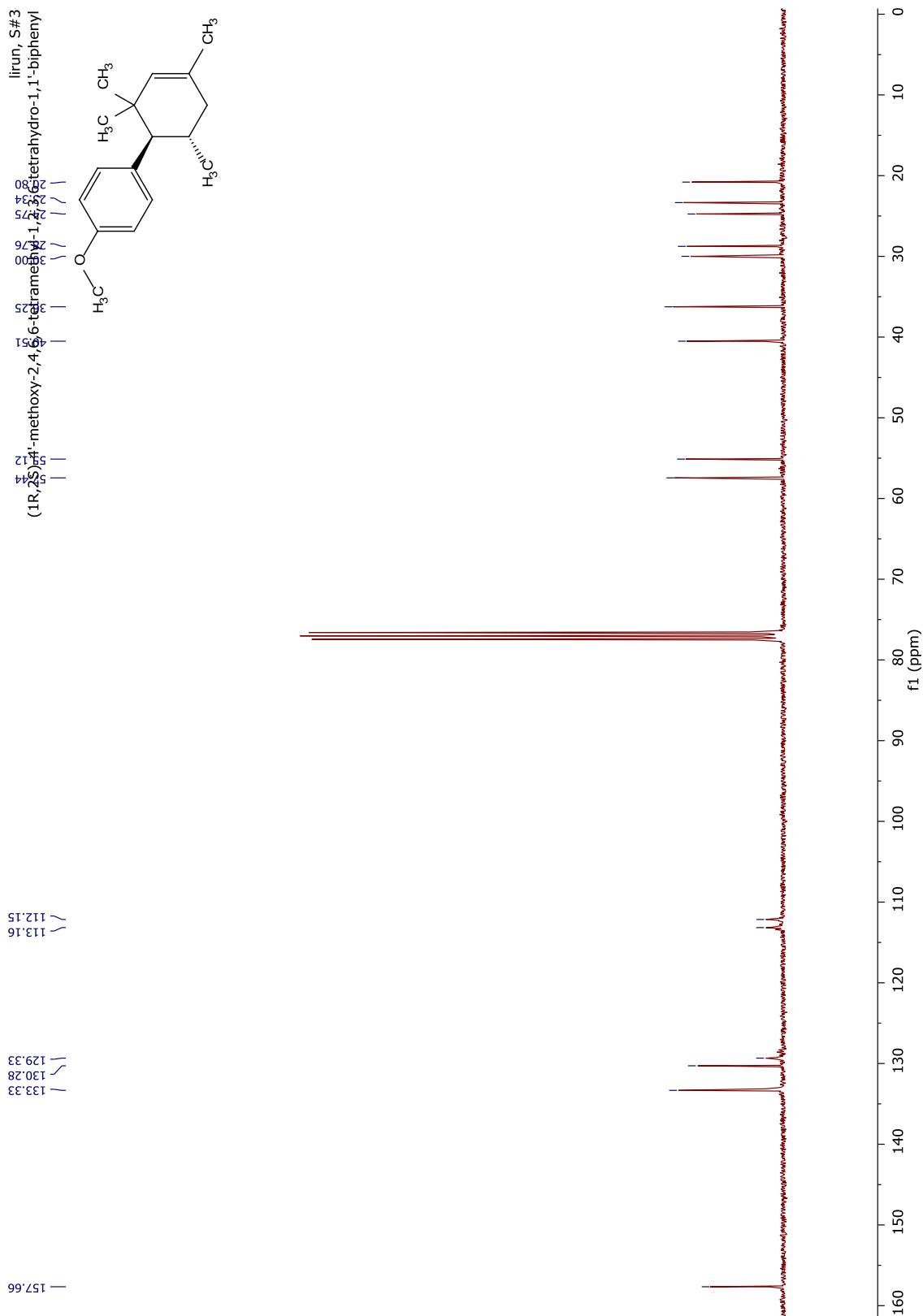


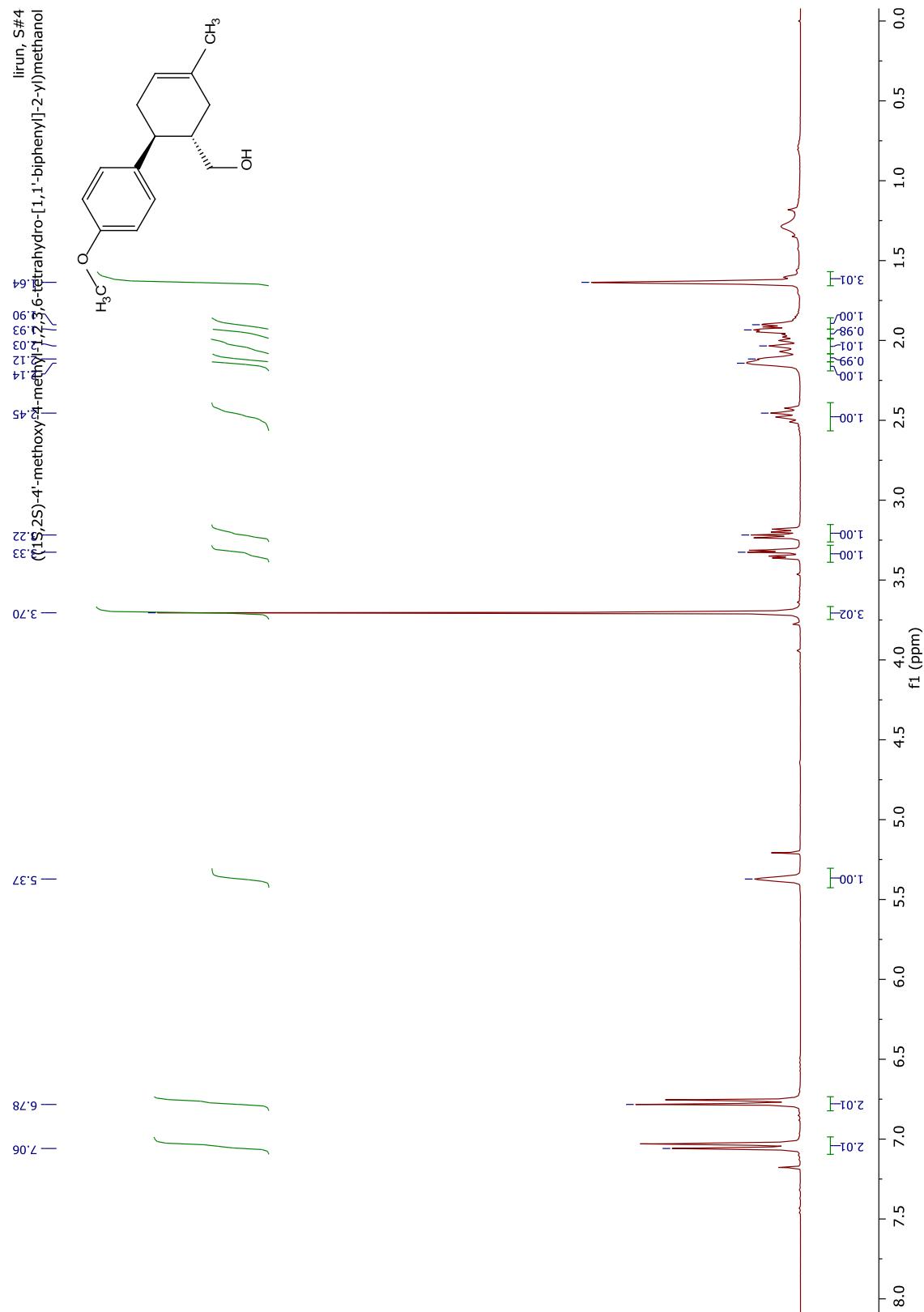


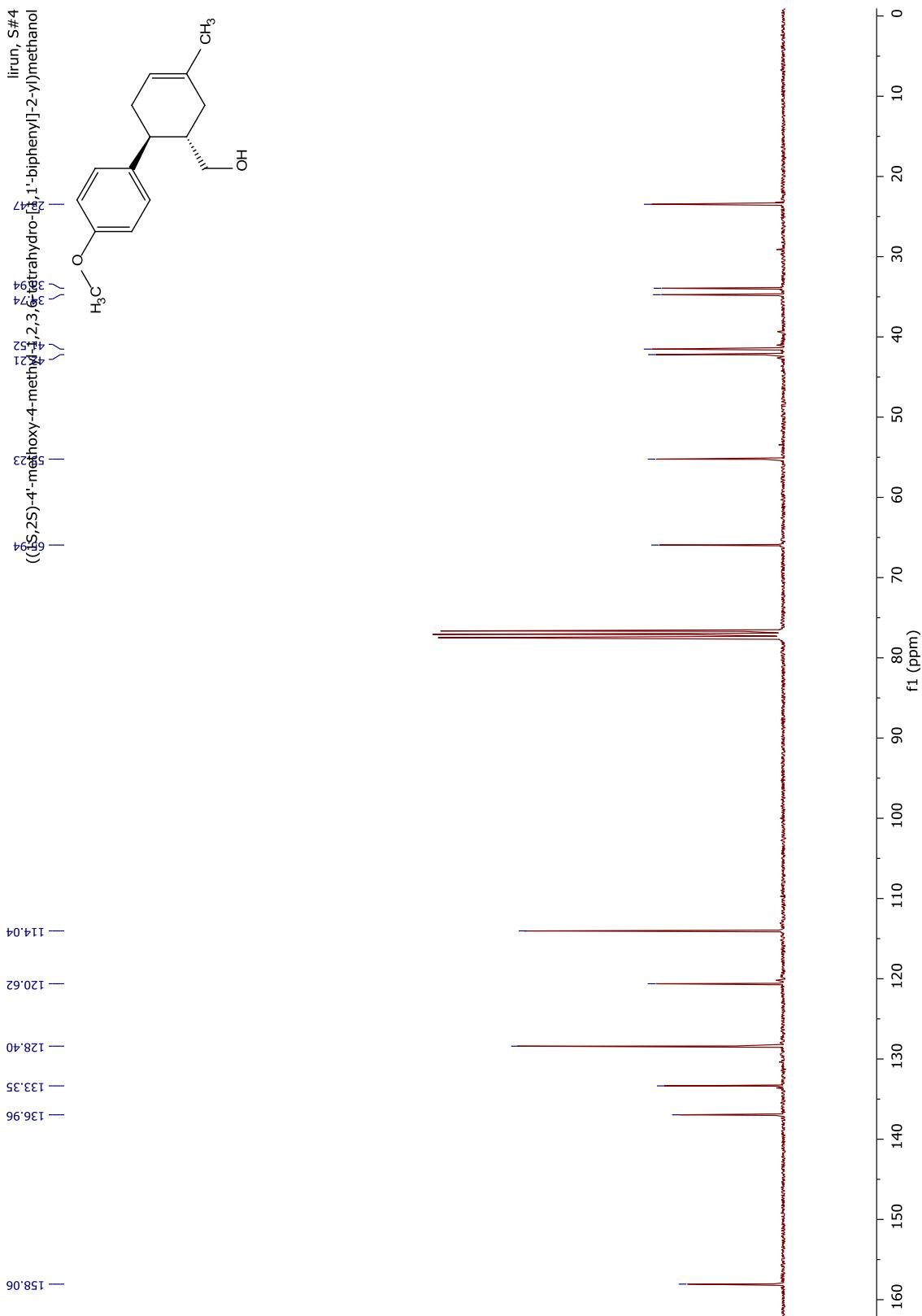


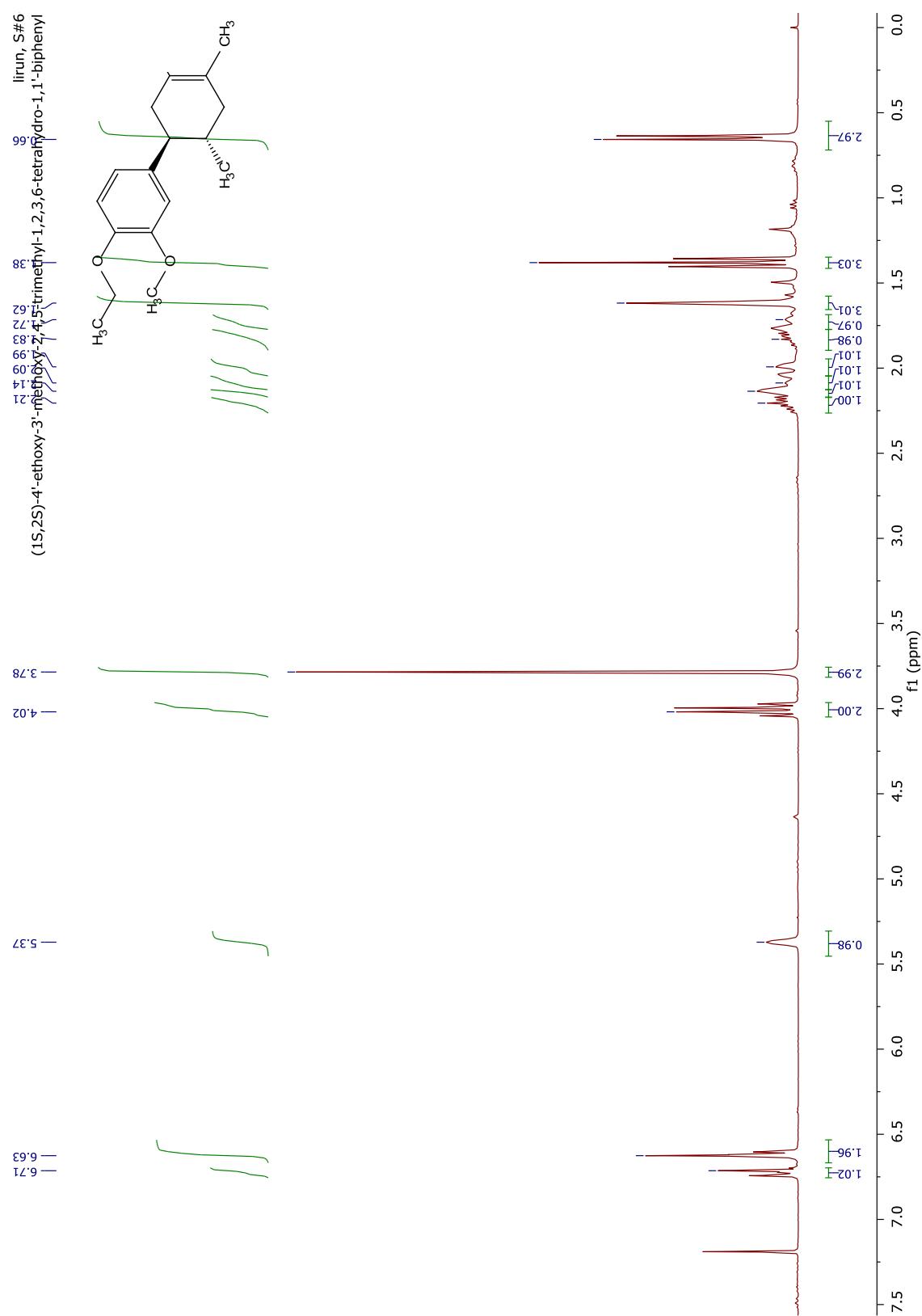


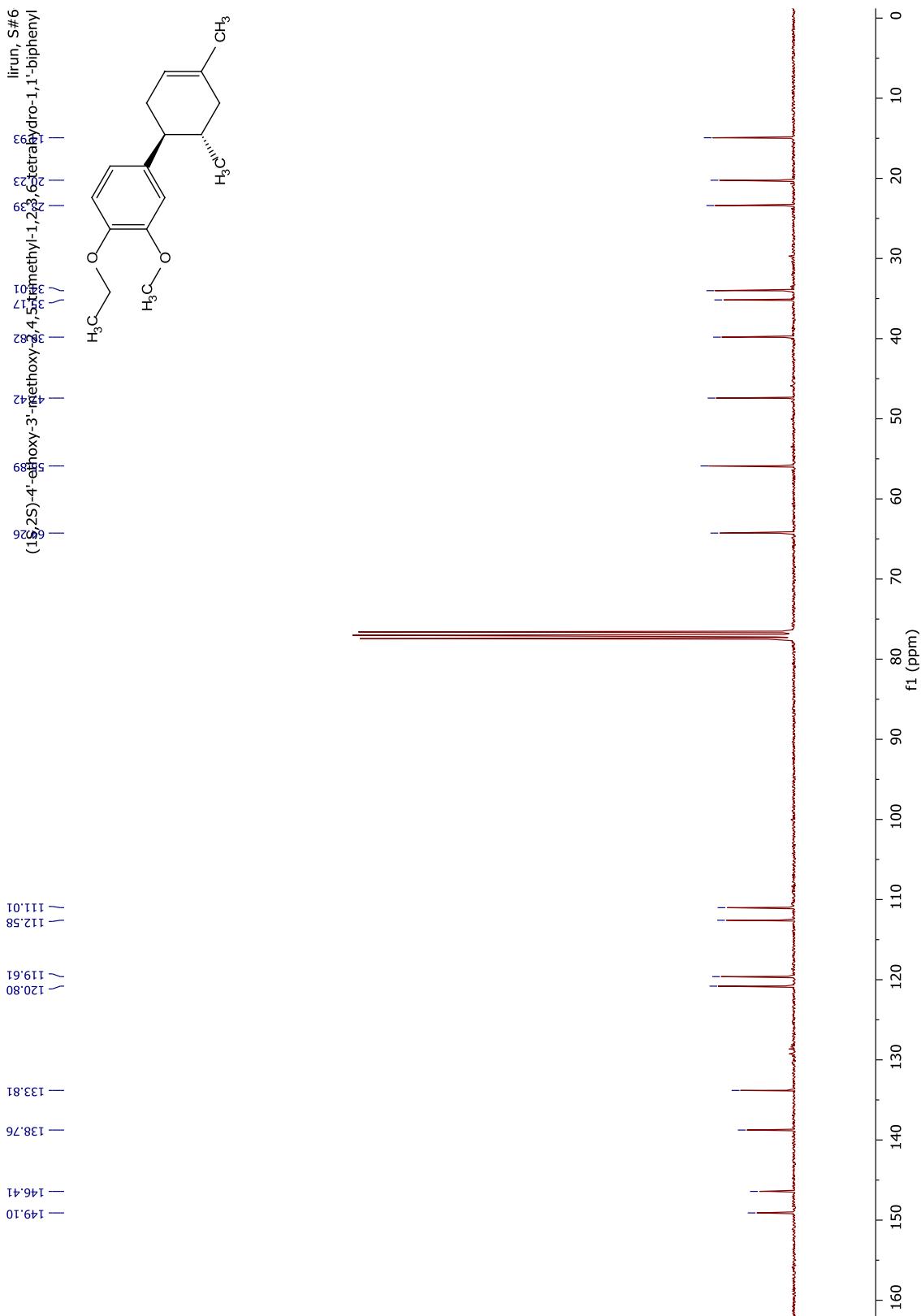


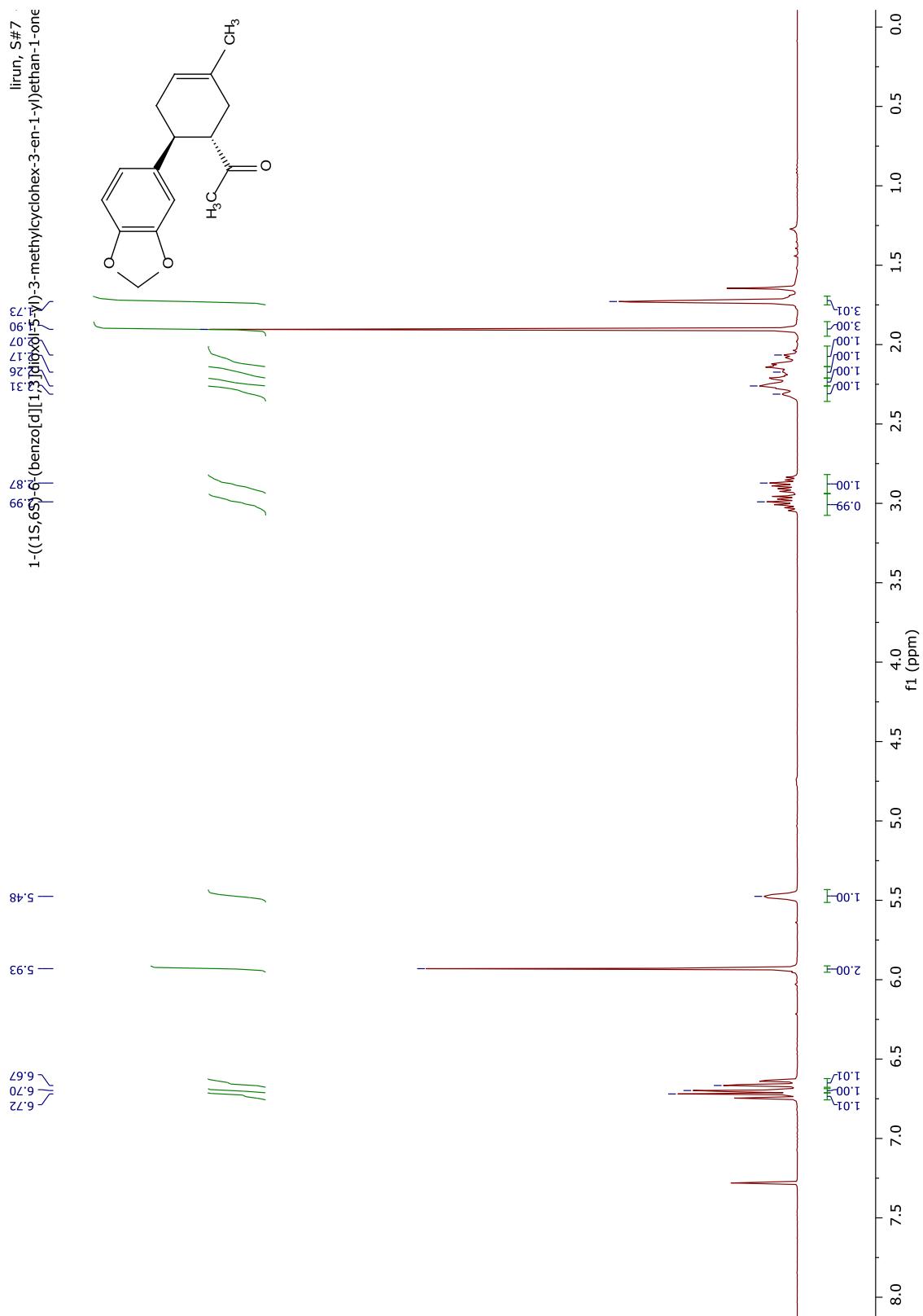


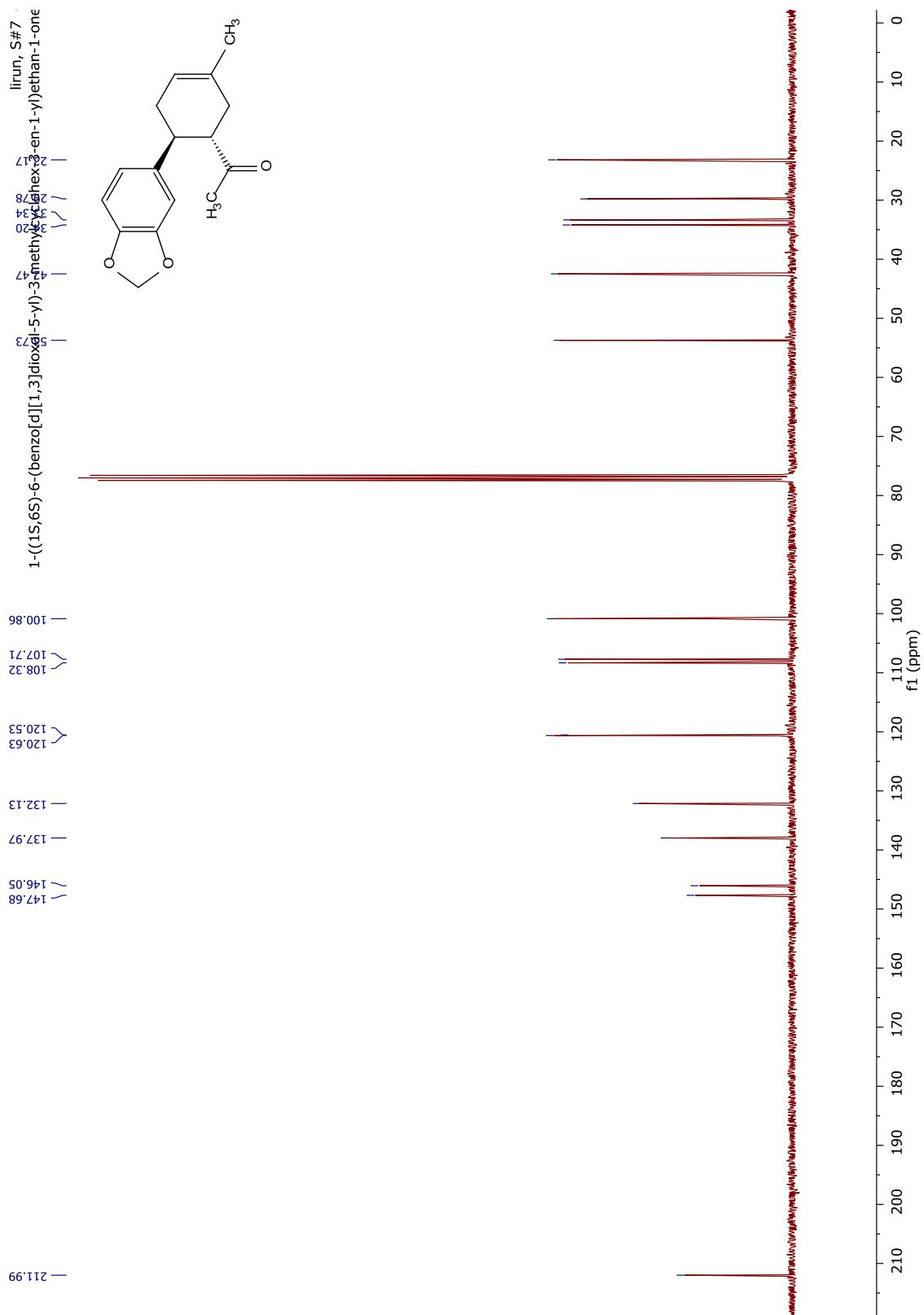


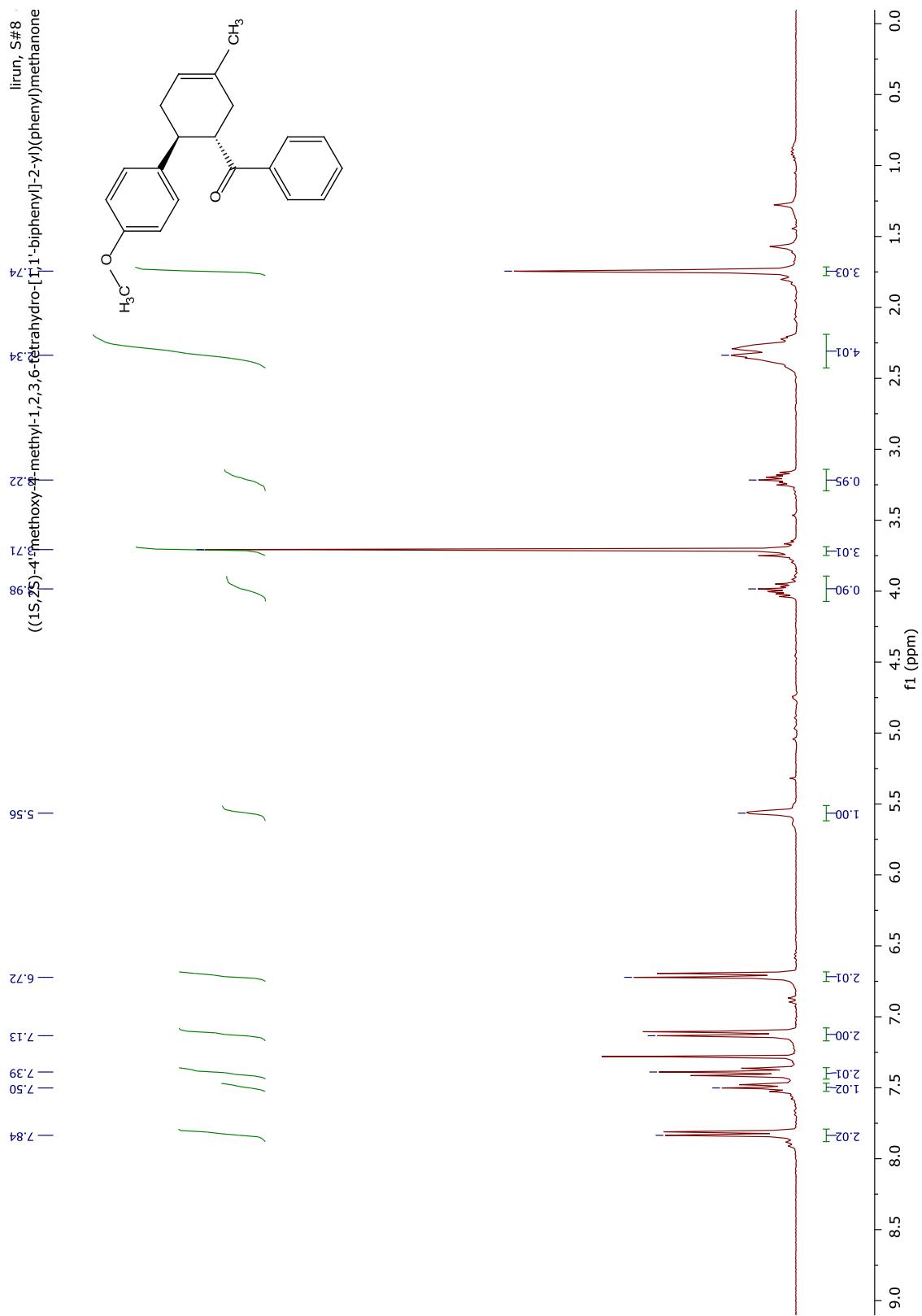


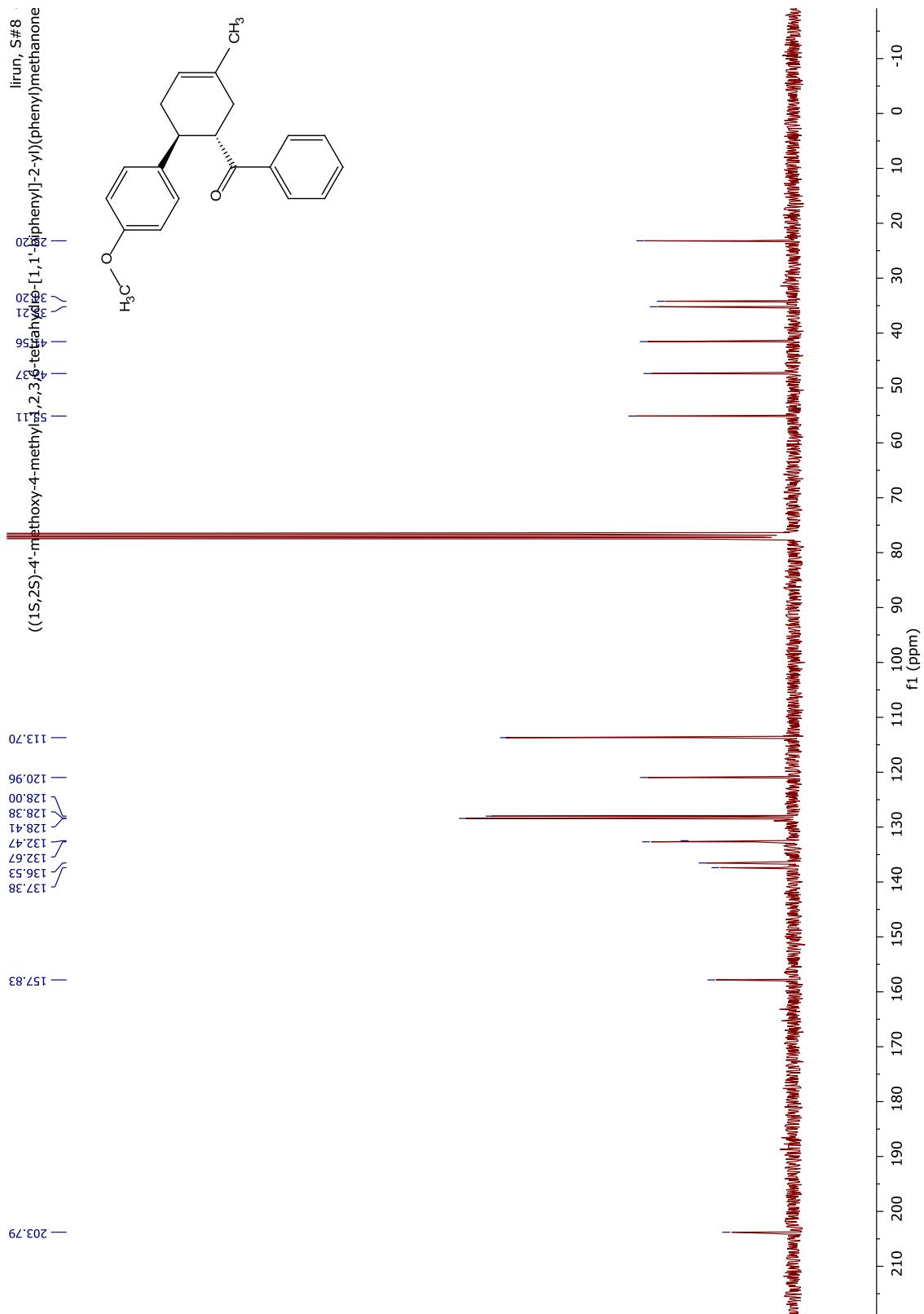


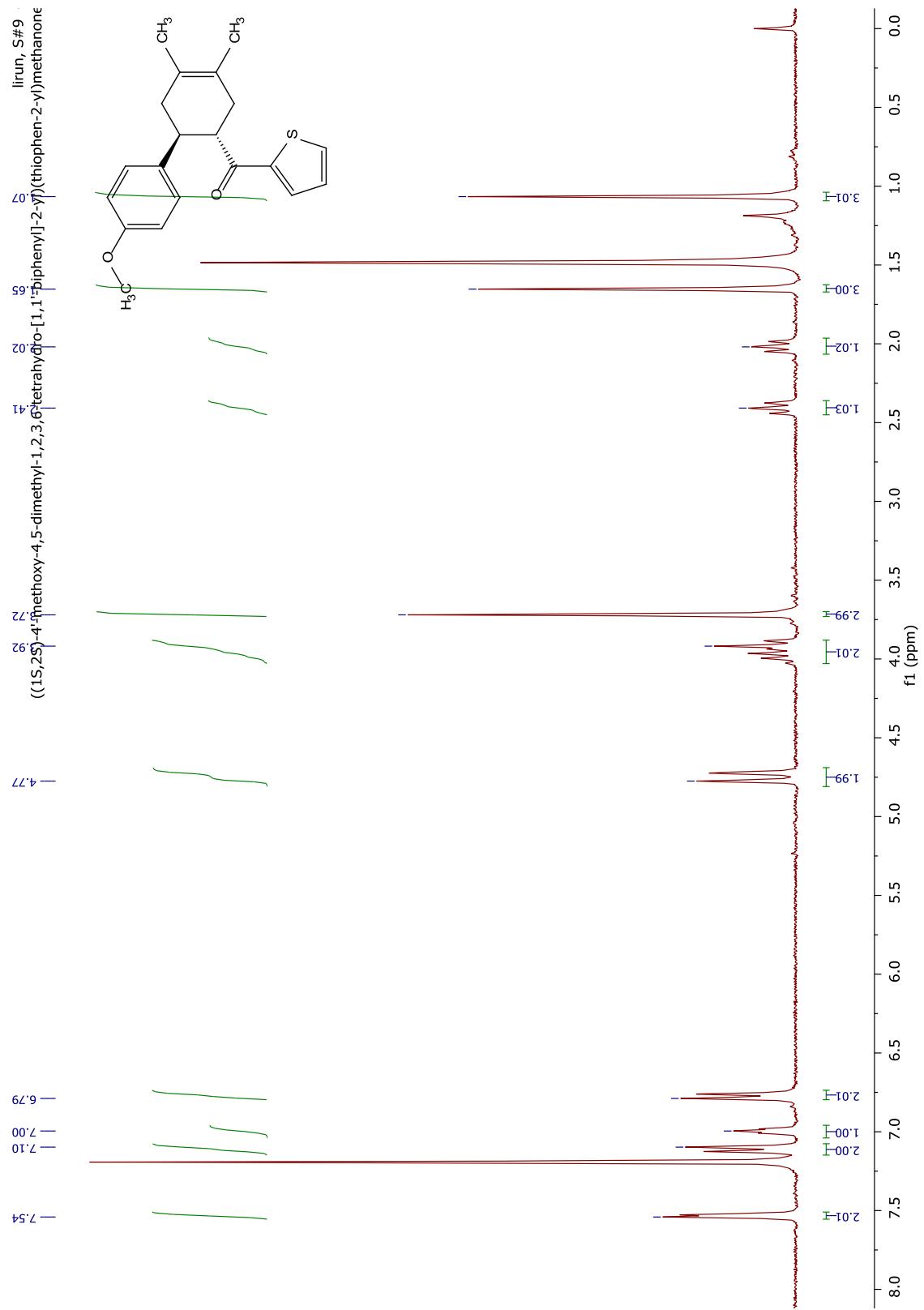


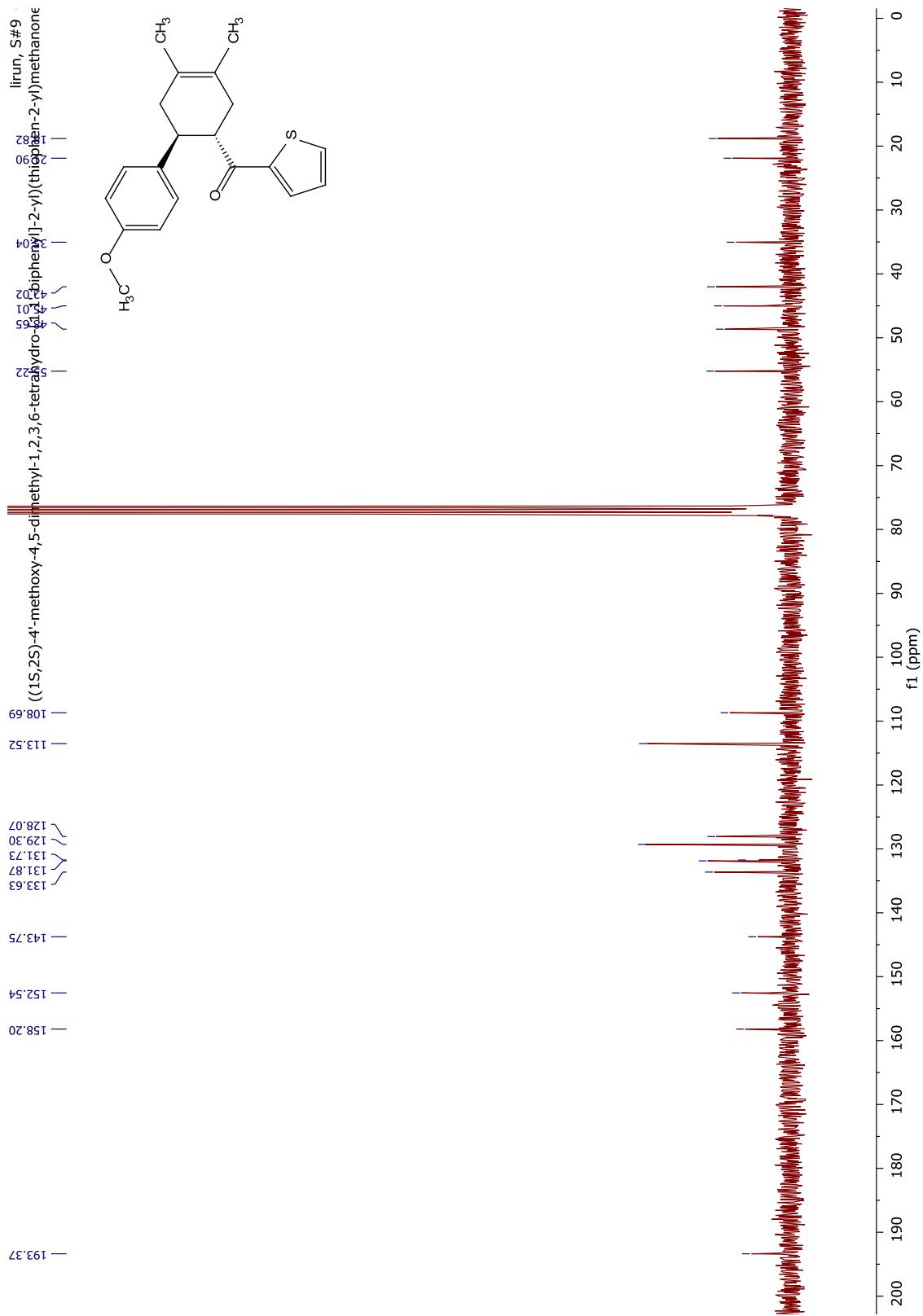


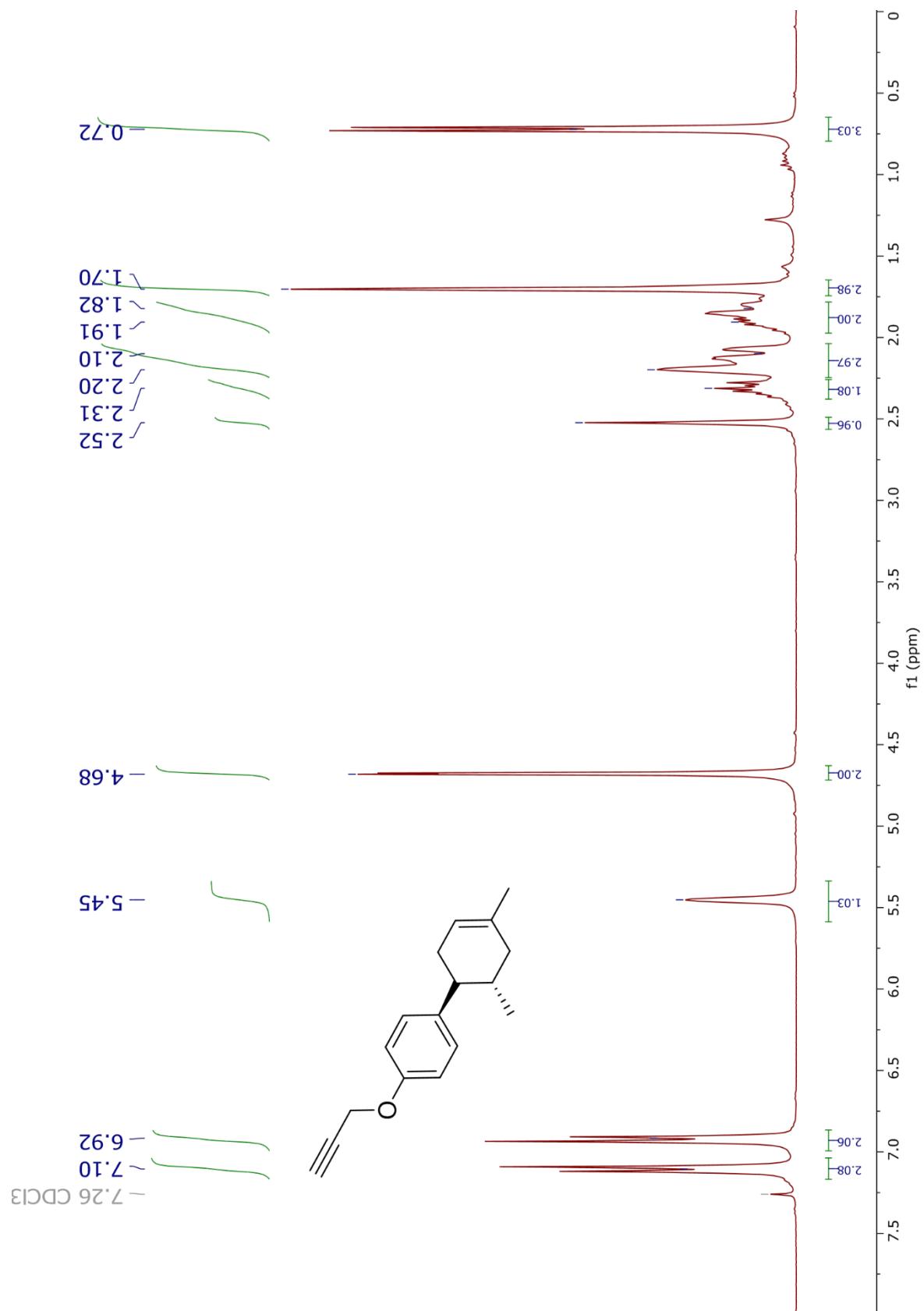


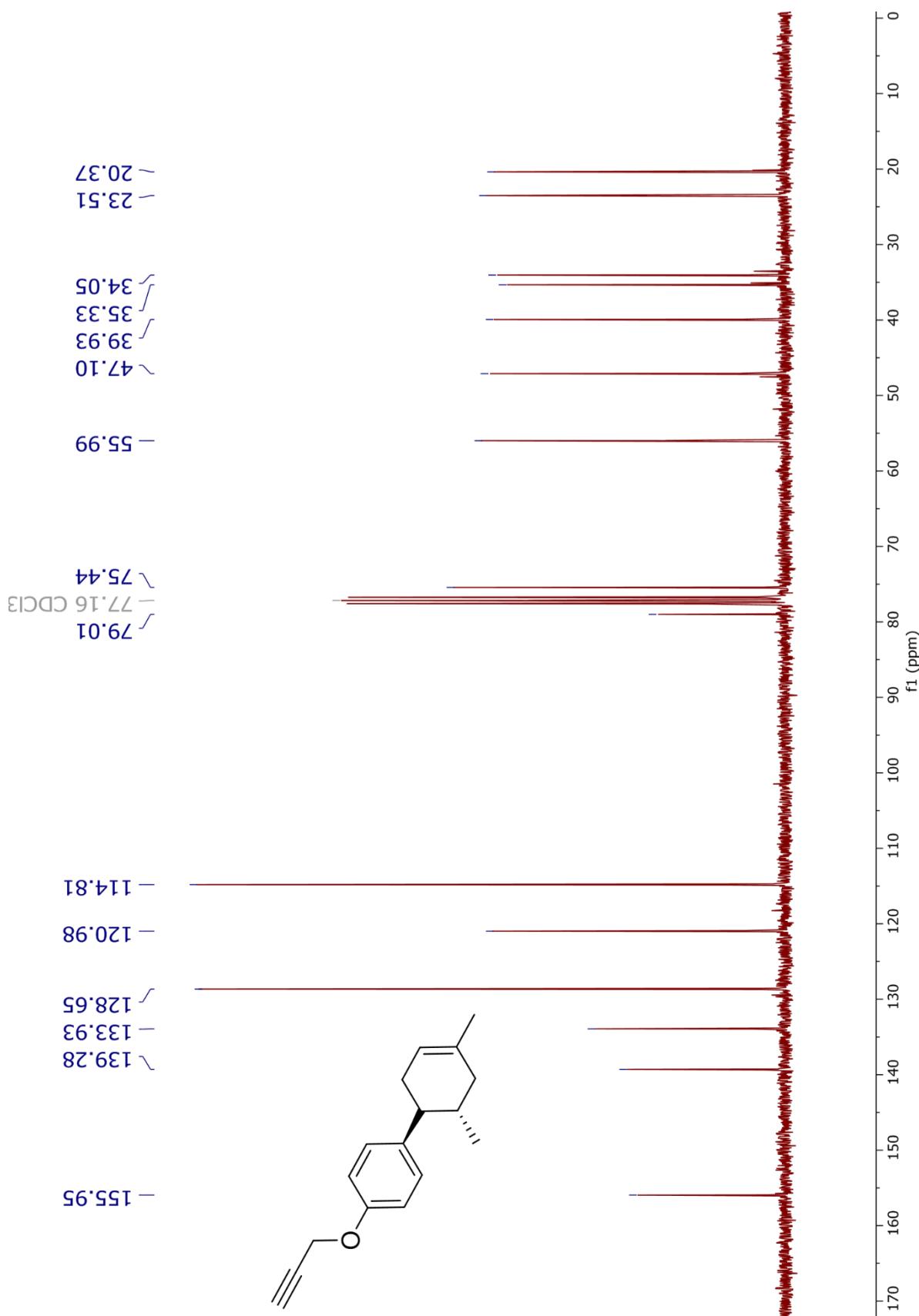


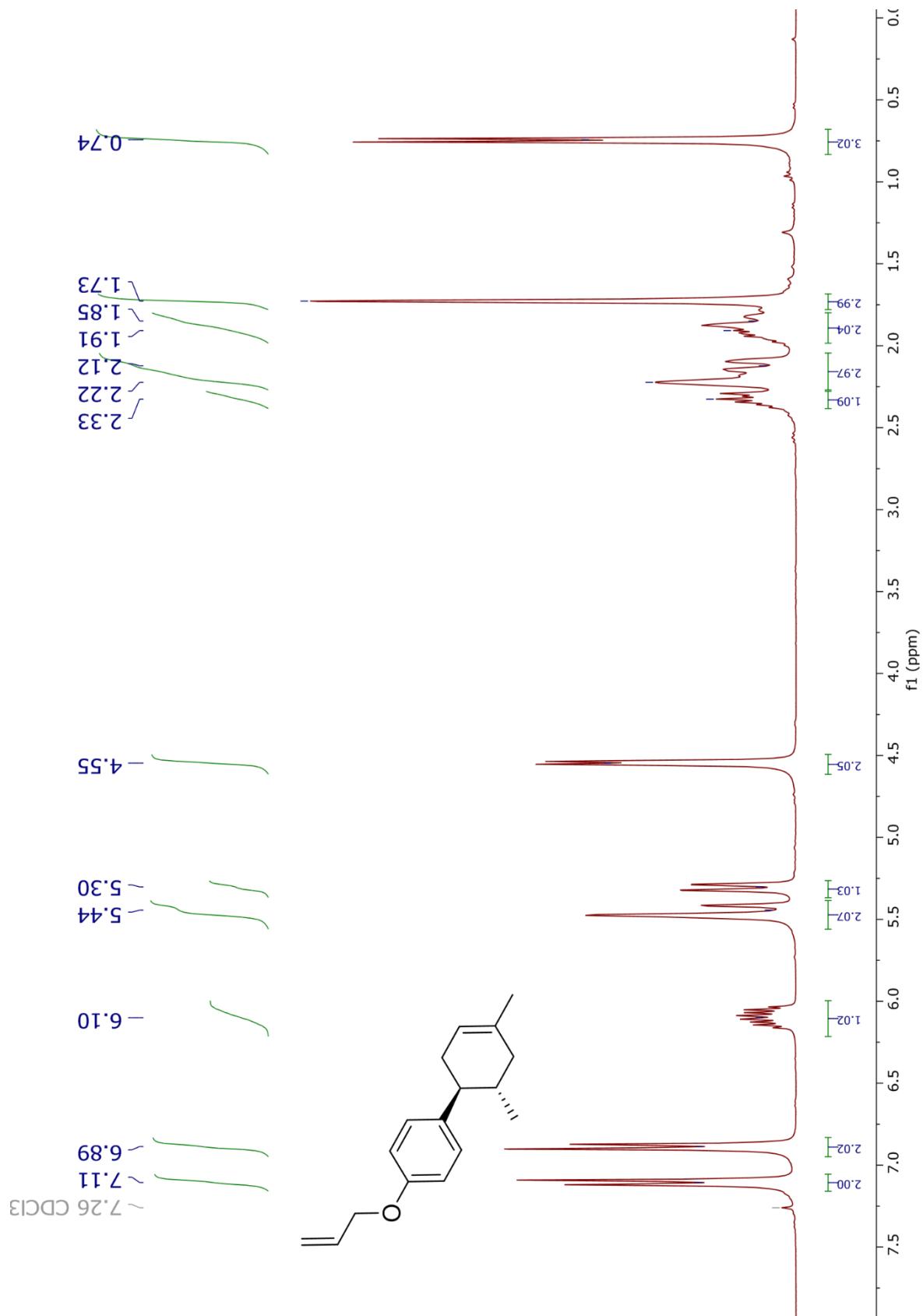


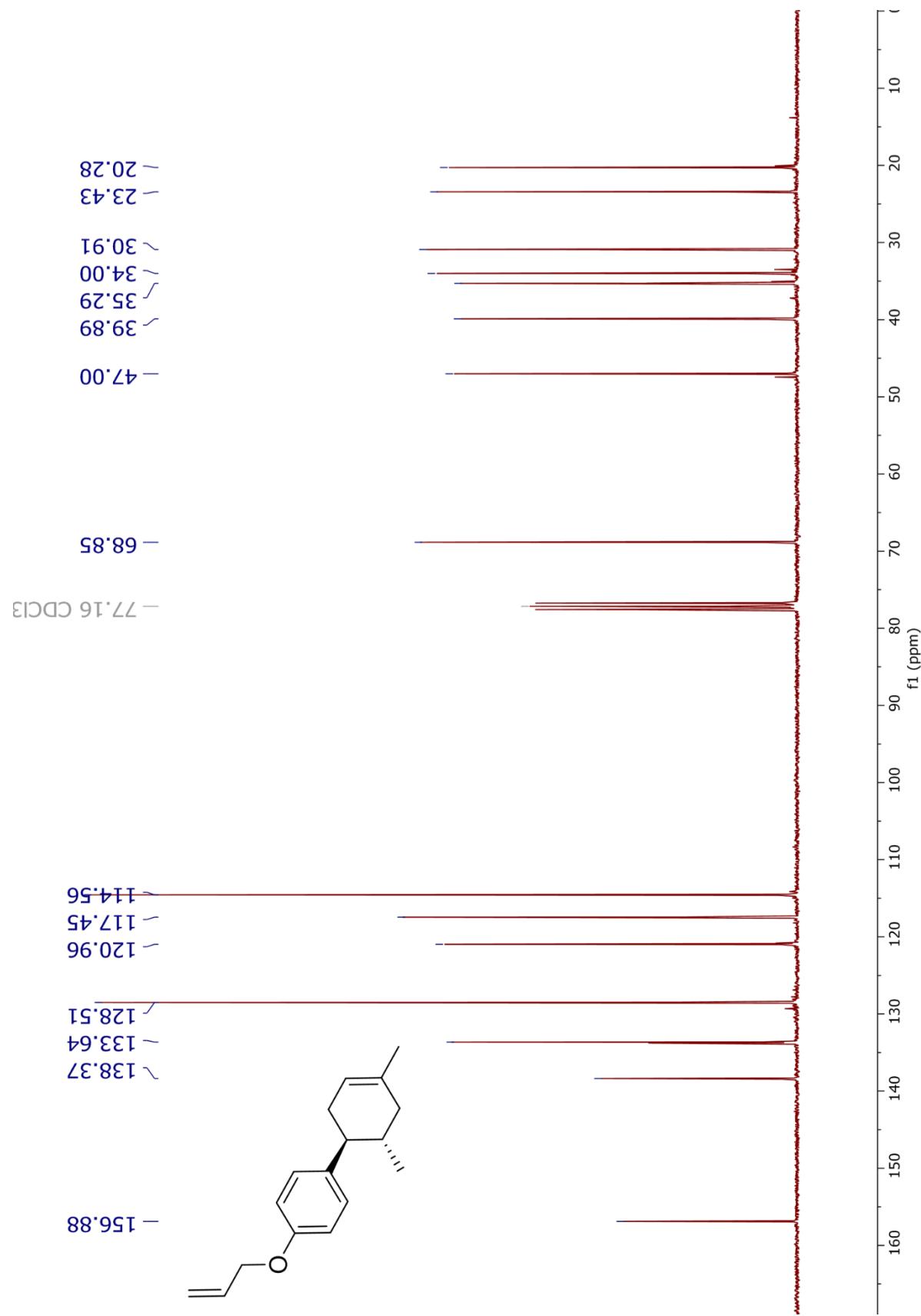


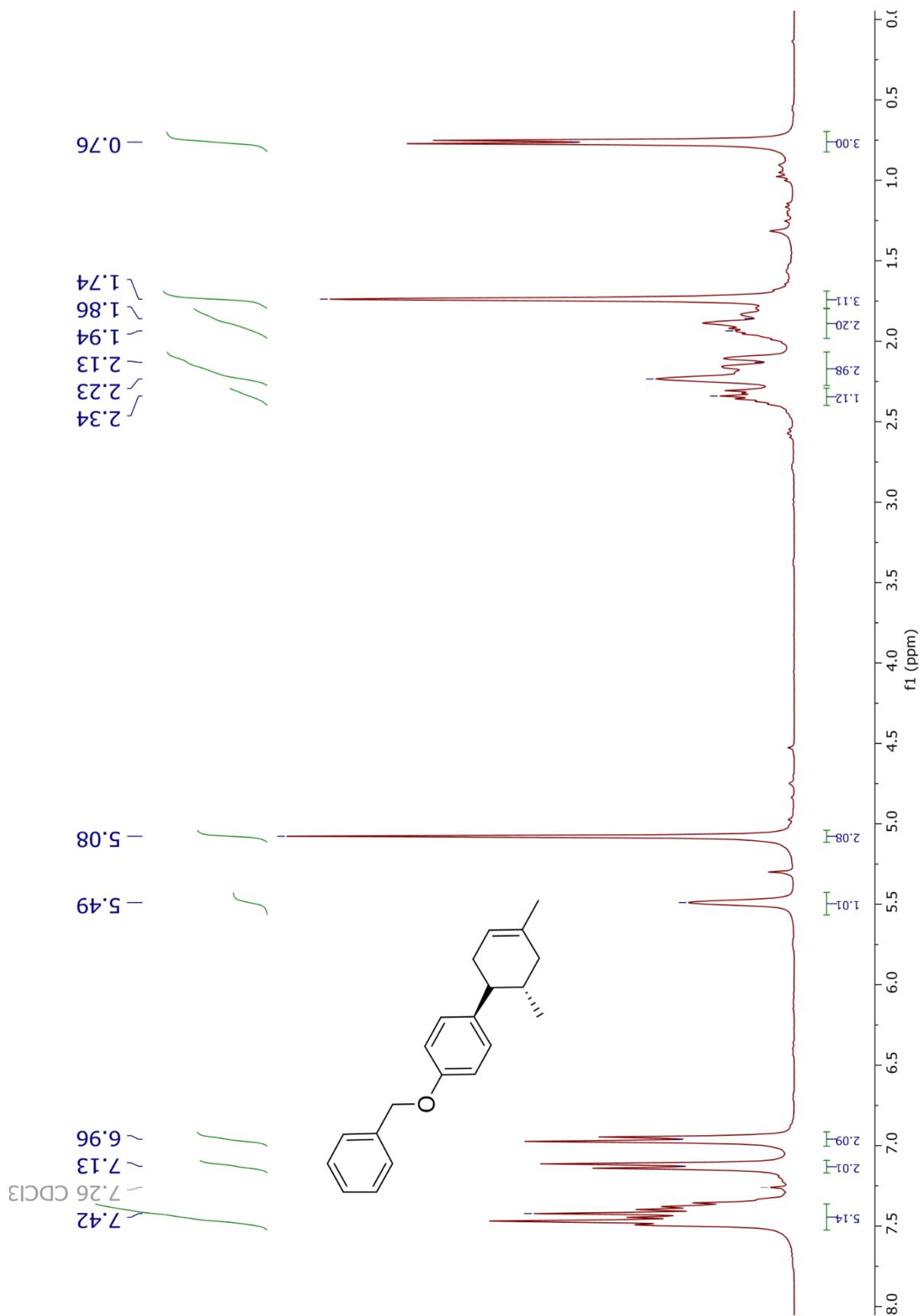


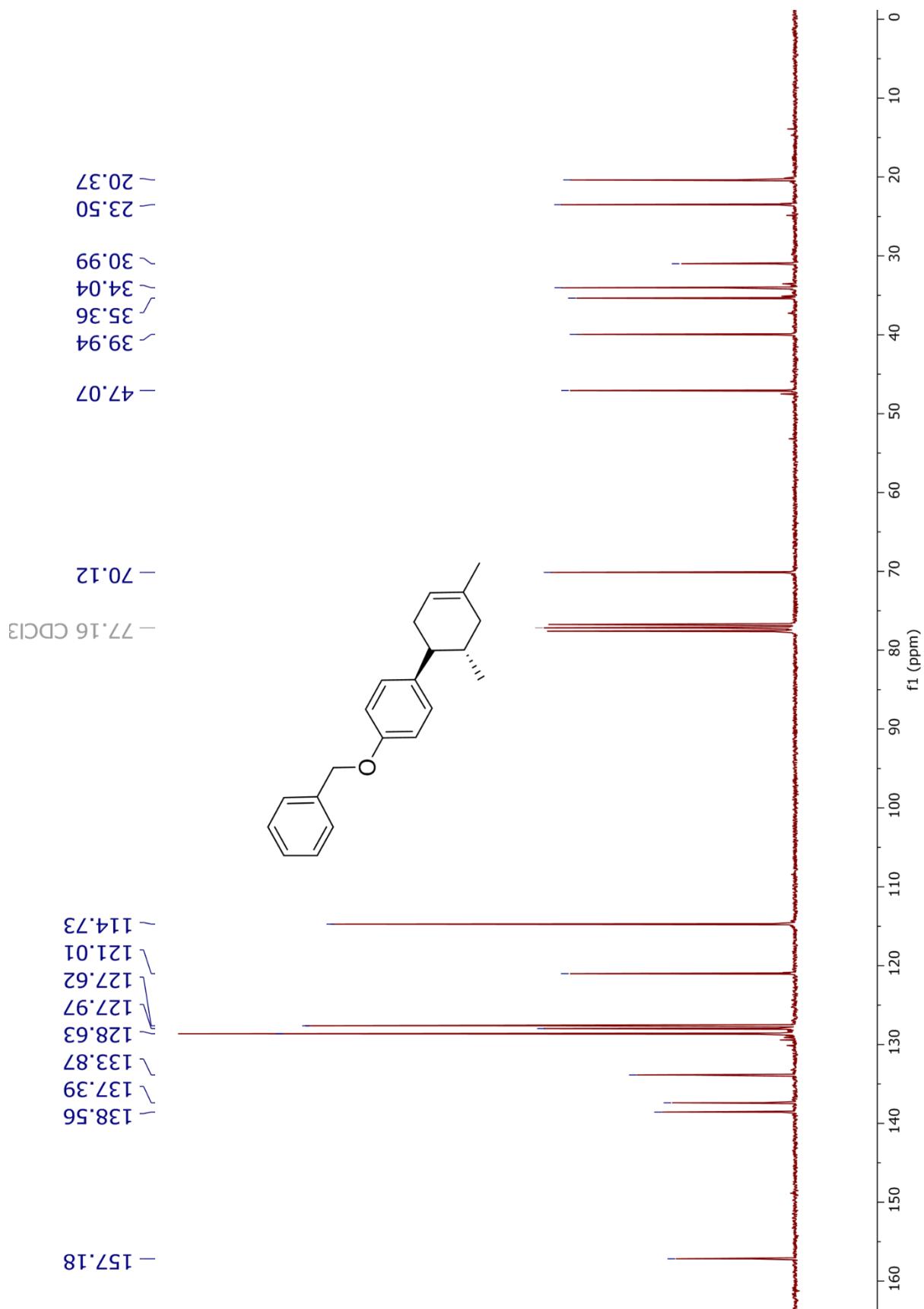












References

1. S. Tojo, S. Toki and S. Takamuku, *J. Org. Chem.*, 1991, **56**, 6240-6243.