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## Efficient Cu catalyst for 5hydroxymethylfurfural hydrogenolysis by forming Cu-O-Si bond

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Catalysts	HMF amount (g)	Catalyst amount (g)	Т (°С)	P (MPa)	t (h)	DMF yield (%)	Ref.
Cu20-PMO	0.5	0.1	220	5	6	66.0	1
$CuCo \mathbb{R}/NGr/\alpha - Al_2O_3$	0.126	0.1	180	2	16	99.0	2
CuZn-2	1.5	0.5	220	1.5	5	91.8	3
NC-Cu/MgAlO	0.5	0.1	220	<i>in situ</i> cyclohexanol	0.5	96.1	4
Cu-Co@C	0.25	0.02	180	5	8	99.4	5
Cu/Al <sub>2</sub> O <sub>3</sub>	0.1	0.03	240	<i>in situ</i> methanol	6	73.9	6
Cu/ZnO	0.1	0.03	240	<i>in situ</i> methanol	10	76.5	6
Cu-Co/Al <sub>2</sub> O <sub>3</sub>	0.315	0.5	220	3	8	78.0	7
Cu-Ni/TiO <sub>2</sub>	0.5	0.3	200	2.5	8	84.3	8
CuZn alloy	0.5	0.1	220	3	18	97.0	9
CuSi-PS	0.75	0.1	180	1.5	2	93.4	This work

 Table S1. Catalytic performance of different Cu based catalysts for HMF conversion

 to DMF/DMTHF in batch reactor.



Figure S1. The XRD patterns of dried CuSi-PS-x catalysts.



Figure S2. The IR spectra of dried CuSi-PS-x catalysts.



Figure S3. The XRD patterns of calcined CuSi-IM-30 catalyst.



Figure S4. Cu2p spectra of the calcined catalysts.



Figure S5. O1s spectra of the calcined catalysts.



Figure S6. Si2p spectra of the calcined catalysts.



Figure S7. XRD spectra of the calcined CuSi-PS-x catalysts.



Figure S8 TPR results of calcined catalysts.



Figure S9. XRD patterns of the reduced catalysts.



Figure S10. Cu2p spectra of the reduced catalysts.



Figure S11. O1s spectra of the reduced catalysts.



Figure S12. Si2p spectra of the reduced catalysts.



Figure S13. Correlations of Cu dispersion measured by N<sub>2</sub>O titration and Cu dispersion measured by XPS results.



Figure S14. NH<sub>3</sub>-TPD results of the calcined and reduced CuSi-PS-30 catalysts.



Figure S15 TOF values over different catalysts.

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