# Plasmonic enhanced furfural hydrogenation catalyzed by stable carbon coated copper nanoparticles driven from metal-organic

## frameworks

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Fig. S1 TGA-DTA curves of HKUST-1

Fig. S2 N<sub>2</sub> adsorption–desorption isotherms of (a) HKUST-1, (b) Cu/AC (c) Cu<sub>2</sub>O-Cu@C.



Fig. S2  $N_2$  adsorption–desorption isotherms of (a) HKUST-1, (b) Cu/AC (c) Cu<sub>2</sub>O-Cu@C.

Fig. S3 N<sub>2</sub> adsorption/desorption isotherms and the corresponding BJH pore-size distribution curves of various Cu@C-T samples



Fig. S3 (a)  $N_2$  adsorption/desorption isotherms and (b) the corresponding BJH poresize distribution curves of various Cu@C-T samples.

Fig. S4 Gas chromatographic analysis of gaseous products during pyrolysis of HKUST-1



**Fig. S4** Gas chromatographic analysis of gaseous products during pyrolysis of HKUST-1.

## Table S1 Textural properties and compositions of various samples

Samples	$\begin{array}{c} S_{BET} \\ (m^2 \ g^{\text{-1}}) \end{array}$	V <sub>pore</sub> (cm <sup>3</sup> g <sup>-1</sup> )	D <sub>pore</sub> (nm)	Content of elements			
				Cu	С	0	Н
Cu@C-400	9.6	0.029	17.4	52.2	31.9	14.3	1.6
Cu@C-600	154.7	0.079	9.0	53.4	32.4	13.0	1.3
Cu@C-800	119.4	0.069	13.1	54.6	33.6	11.0	0.7

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## Fig. S5 Raman spectrum of various Cu@C-T samples



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Fig. S6 FT-IR spectrum of various Cu@C-T samples



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Fig. S7 The DRIFT spectra of FAL absorbed on Cu@C-T catalysts.



Fig. S7 The DRIFT spectra of FAL absorbed on Cu@C-T catalysts.



Fig. S8 Recyclability and stability of Cu@C-600 photocatalyst. Reaction conditions: 30 mg of catalyst, 0.2 mmol of substrate, and 5 mL of isopropanol as solvent, 1 atm H<sub>2</sub>. The reaction mixture was stirred under visible light irradiation (0.5 W/cm<sup>2</sup>) at 100 °C for 24 h.

## Fig. S9 The output spectra of LED light sources.



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## Fig. S10 UV-vis-DR spectra of various samples.



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## Fig. S11 XRD pattern of Cu/AC.



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