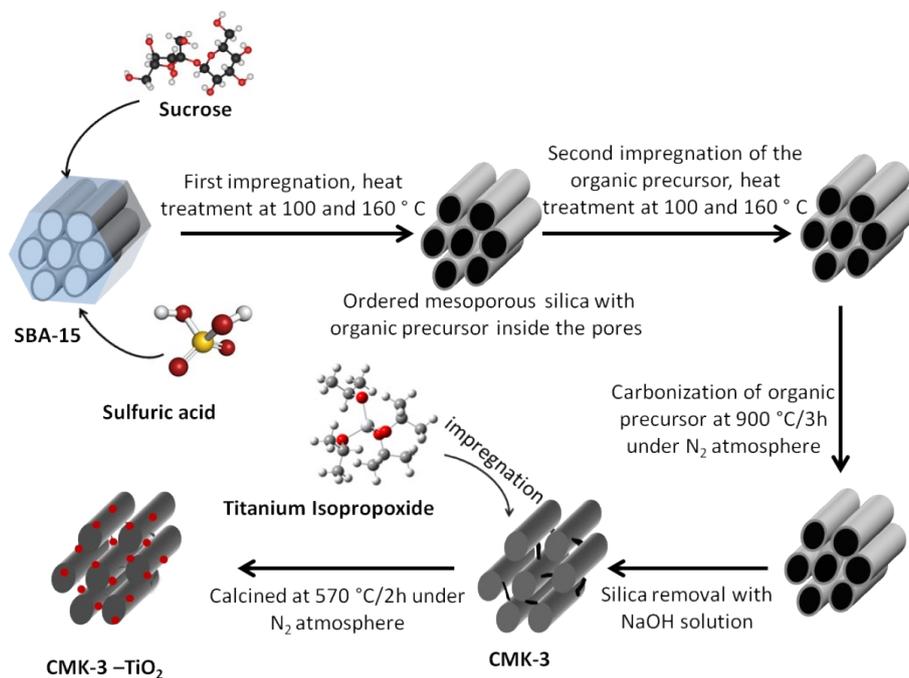
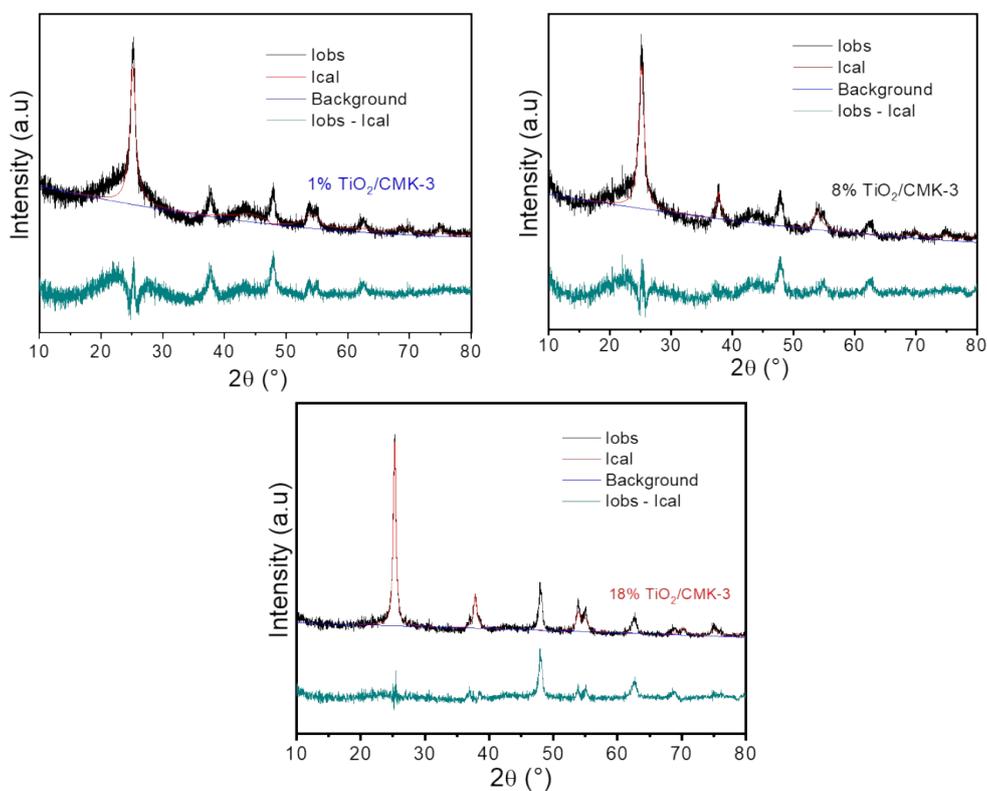


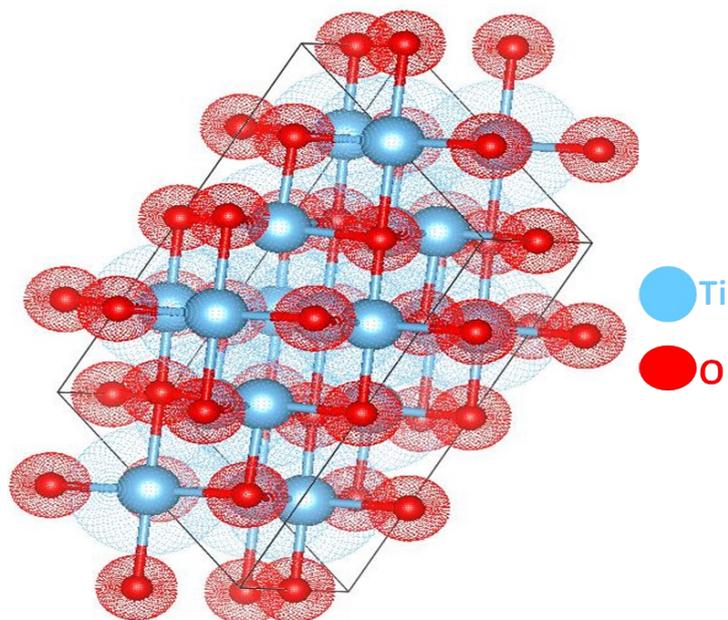
### Supplementary material



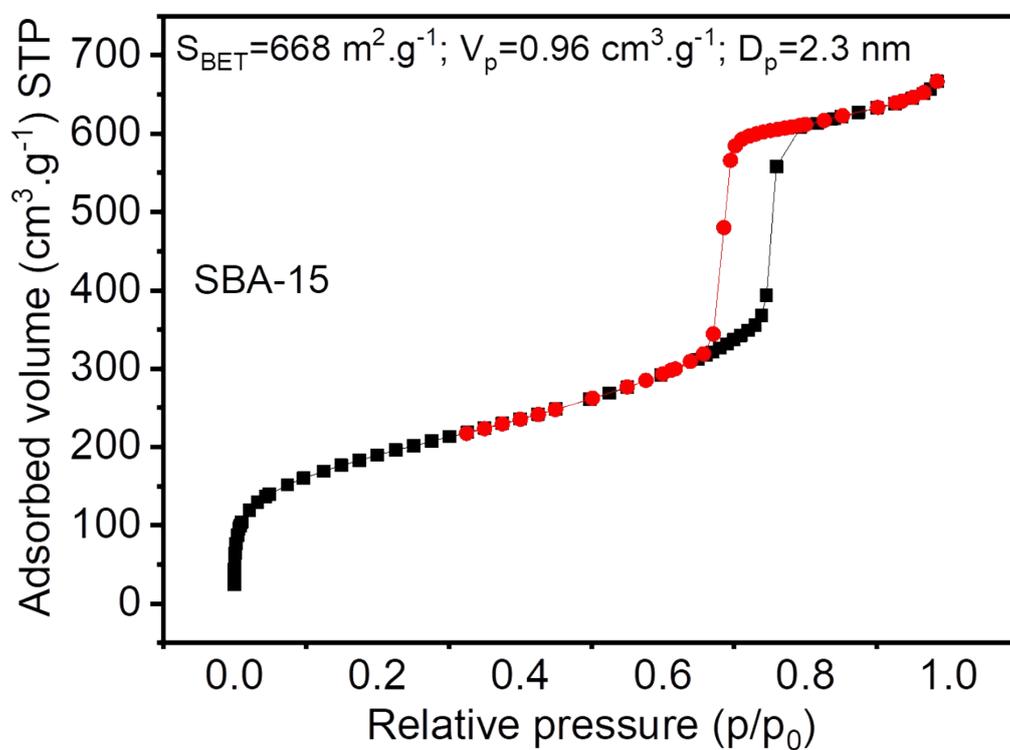
**Scheme S1.** Illustrative sequence of the methodology for preparing pure CMK-3 and TiO<sub>2</sub> dispersed on CMK-3 by impregnation.



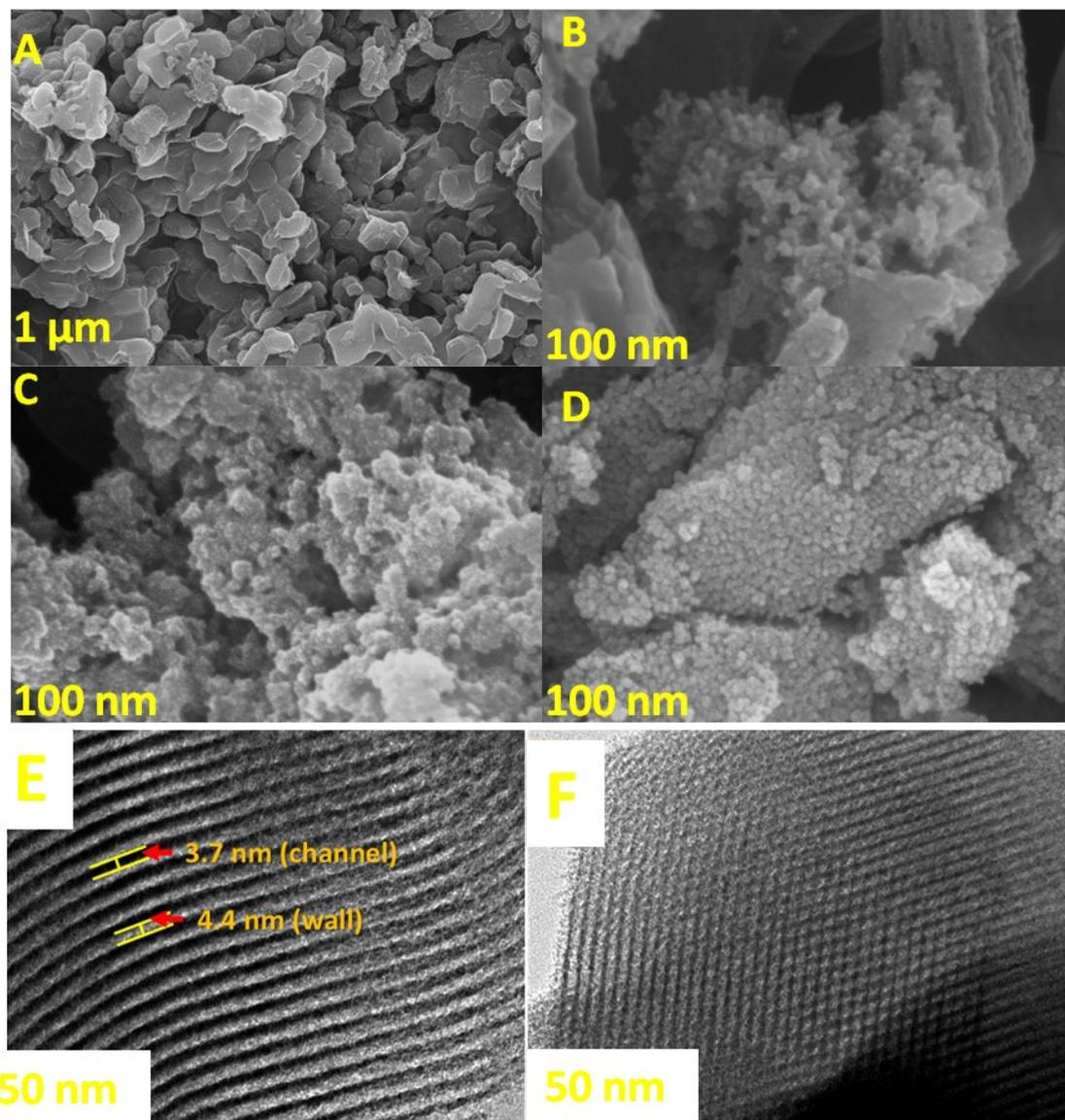
**Figure S1.** Rietveld refinement results for Ti-based photocatalysts.



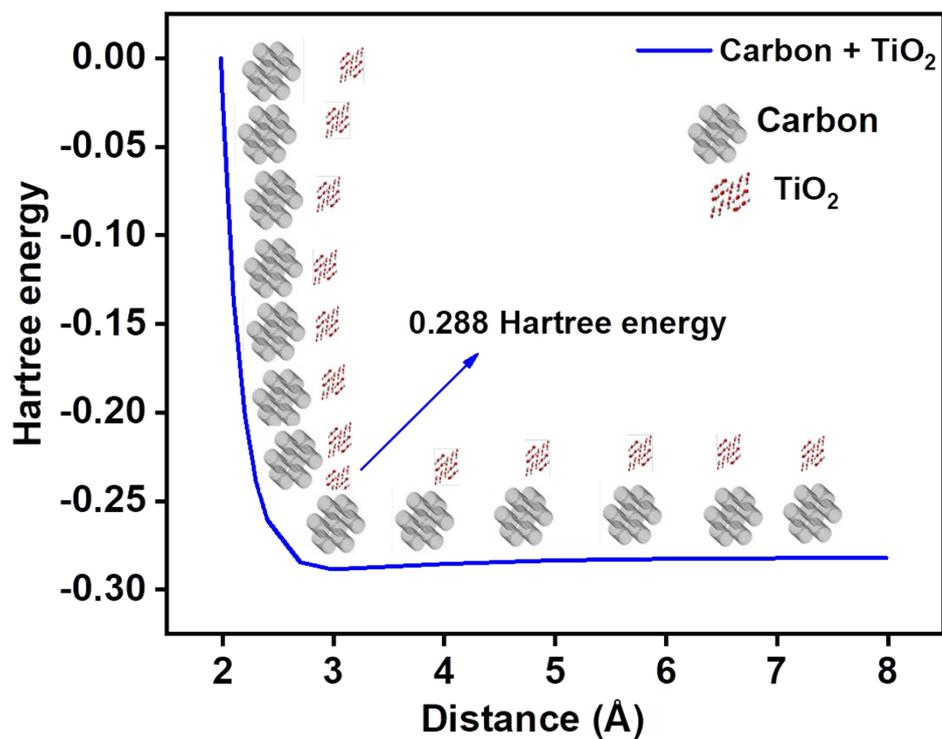
**Figure S2.** Unit cell for the TiO<sub>2</sub> anatase phase with the direction of parameters a and c. Image extracted from the Vesta software.



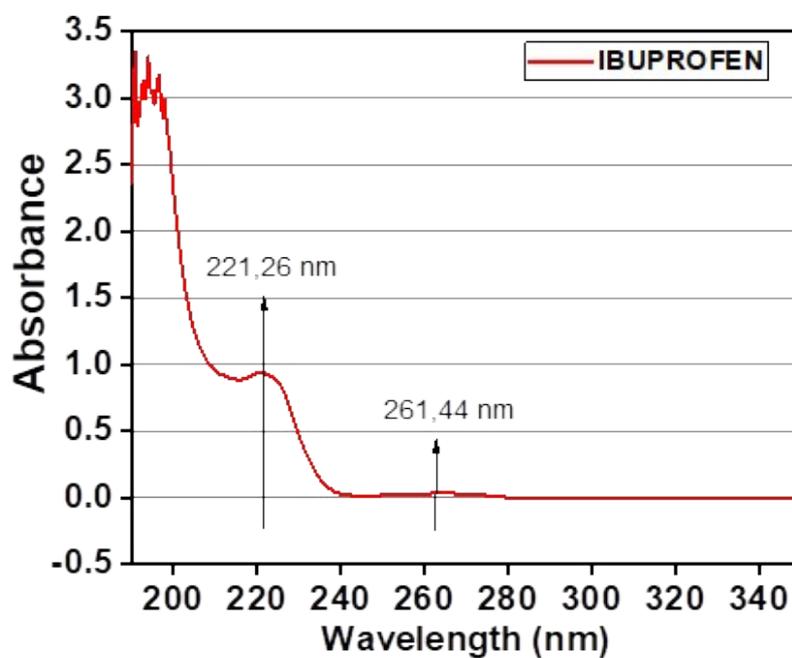
**Figure S3.** N<sub>2</sub> adsorption-desorption isotherms for SBA-15 template.



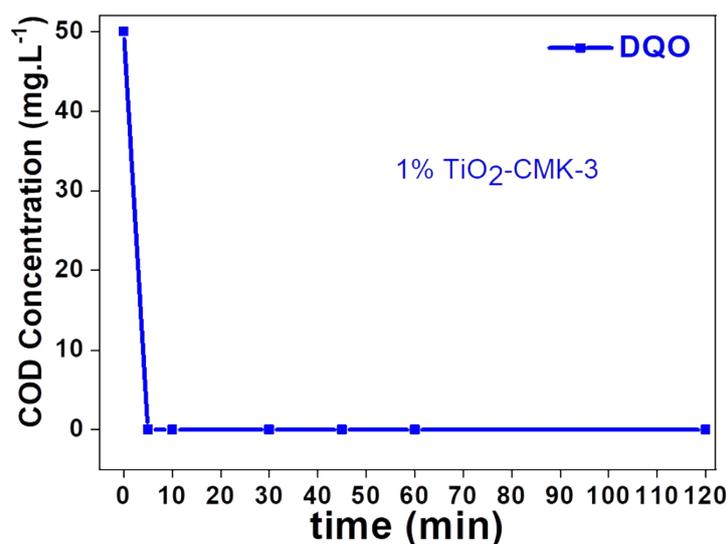
**Figure S4.** Images obtained by SEM analysis: A) CMK-3; B) 8%-TiO<sub>2</sub>-CMK-3 C) 18%-TiO<sub>2</sub>-CMK-3 D) 1%-TiO<sub>2</sub>-CMK-3. E) and F) Images obtained by TEM analysis for pure SBA-15 template.



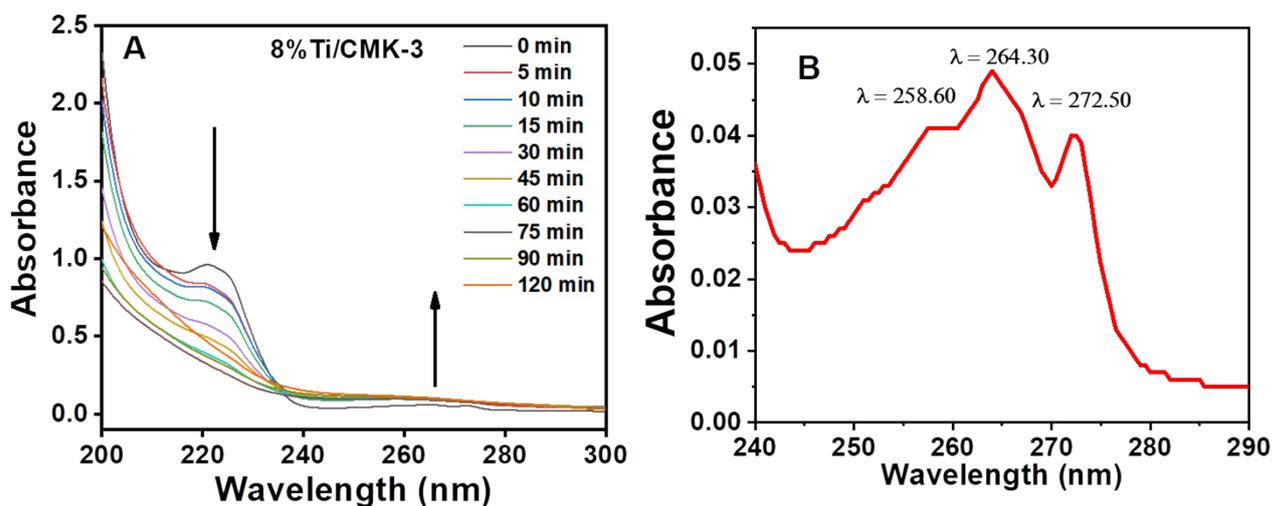
**Figure S5.** Dependence of potential energy and the molecular interaction distance between carbon and TiO<sub>2</sub>.



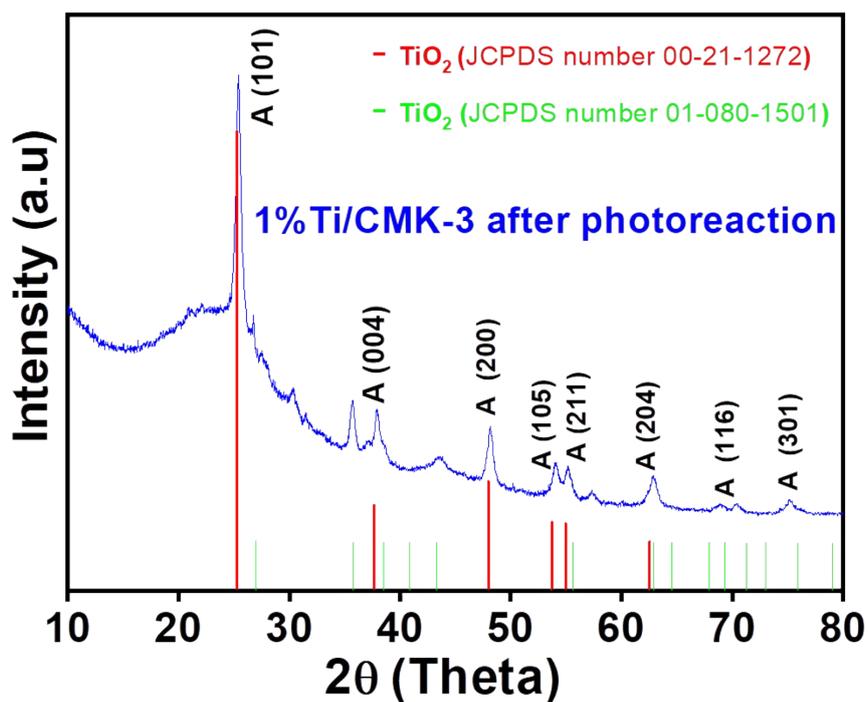
**Figure S6.** Absorptive properties of the ibuprofen molecule.



**Figure S7.** COD concentration over time in the degradation of ibuprofen for the sample 1% Ti/CMK-3.



**Figure S8.** (A) Absorption spectrum of ibuprofen at different times during the photocatalytic tests for the solid 8% TiO<sub>2</sub>-CMK-3; (B) UV spectra in the wavelength range between 250 and 280 nm, showing the bands referring to the by-products formation.



**Figure S9.** XRD results for the sample 1% Ti/CMK-3 after the photocatalytic test.

**Table S1.** Structural parameters such as lattice constants ( $c$  and  $a$ ),  $c/a$  ratio, interplanar spacing, cell volume ( $V_{\text{cell}}$ ), crystallite size ( $D$ ).

%Ti	$2\theta$ (q)	$d(\text{hkl})$ (Å)	$D$ (nm)	Anatase unit cell parameters (Å)			$V_{\text{cell}}$ (Å <sup>3</sup> )
				$c$	$a=b$	$c/a$ ratio	
1	25.24	3.523	8	9.489	3.795	2.5	136.66
8	25.24	3.507	9	9.511	3.801	2.502	137.411
18	25.34	3.519	15	9.509	3.788	2.51	136.444