

Electronic Supplementary Material (ESI) for Catalysis Science & Technology.
This journal is © The Royal Society of Chemistry 2020

Supporting Information

Data mining of amine dehydrogenases for the synthesis of enantiopure amino alcohols

Hongyue Wang,^{ab#} Ge Qu,^{b#} Jun-Kuan Li,^{bc#} Jun-An Ma,^c Jinggong Guo,^d Yuchen Miao^d and Zhoutong Sun^{b*}

^a University of Chinese Academy of Sciences, Beijing, 100049, China.

^b Tianjin Institute of Industrial Biotechnology, Chinese Academy of Sciences, 32 West 7th Avenue, Tianjin Airport Economic Area, Tianjin 300308, China.

^c Department of Chemistry, Tianjin Key Laboratory of Molecular Optoelectronic Sciences, and Tianjin Collaborative Innovation Center of Chemical Science and Engineering, Tianjin University, Tianjin 300072, China.

^d State Key Laboratory of Cotton Biology, Department of Biology, Institute of Plant Stress Biology, Henan University, 85 Minglun Street, Kaifeng, 475001, China.

[#] These three authors contributed equally to this work.

^{*} Corresponding Authors, E-mail: sunzht@tib.cas.cn, Tel.: (+86) 22-84861981.

Table of Contents

Figure S1. SDS-PAGE analysis of the protein expression and purification of AmdHs.	S3
Figure S2. The thermostability analysis of AmdHs.	S4
Figure S3. Docking analysis of substrate 1c in GsAmdH.	S5
Figure S4. HPLC profiles of the amino alcohols product catalyzed by <i>GsAmdH</i>	S6
Table S1. Analytical conditions of HPLC.	S7
Synthesis of α -hydroxy ketone substrates 1d , 1e , 1f , 1g	S9
¹ H NMR and ¹³ C NMR spectra of product (<i>S</i>)- 2a	S11
¹ H NMR and ¹³ C NMR spectra of substrate 1d	S12
¹ H NMR and ¹³ C NMR spectra of substrate 1e	S13
¹ H NMR and ¹³ C NMR spectra of substrate 1f	S14
¹ H NMR and ¹³ C NMR spectra of substrate 1g	S15
References	S16

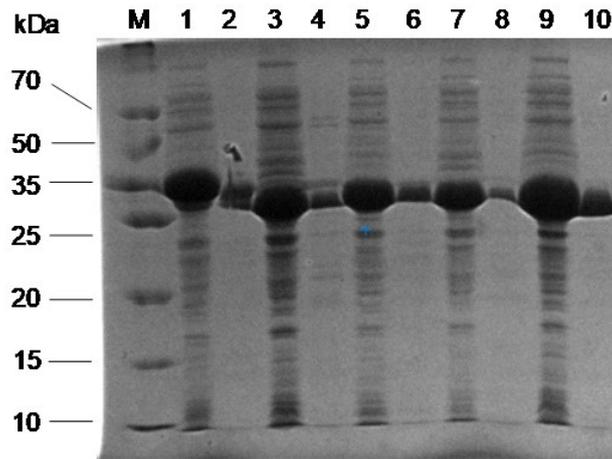


Figure S1. SDS-PAGE analysis of the protein expression and purification of AmdHs. M: protein marker. Lane 1-2: the crude extract and purified enzyme of *GsAmdH*; Lane 3-4: the crude extract and purified enzyme of *TiAmdH*; Lane 5-6: the crude extract and purified enzyme of *BsAmdH*; Lane 7-8: the crude extract and purified enzyme of *SpAmdH*; Lane 9-10: the crude extract and purified enzyme of *LsAmdH*.

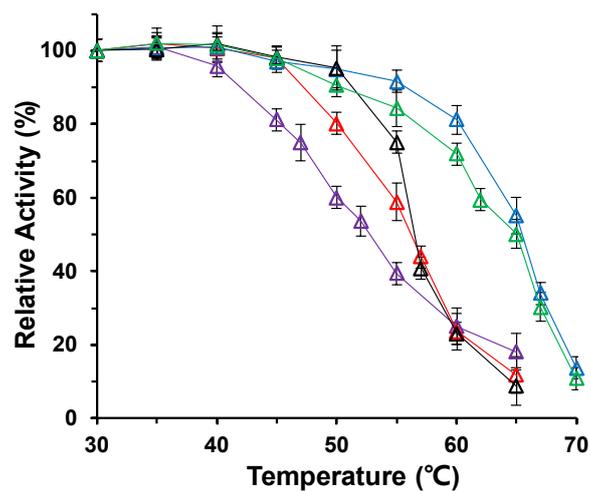


Figure S2. The thermostability analysis of AmDHs (\blacktriangle : *GsAmDH*, \triangle : *BsAmDH*, \triangle : *LsAmDH*, \triangle : *SpAmDH*, \triangle : *TiAmDH*), purified AmDHs (1 mg/mL) were incubated at different temperatures (30~70 °C) for 15 min, followed by measuring the residual activity towards substrate **1a** (10 mM). The initial activities of *GsAmDH*, *BsAmDH*, *LsAmDH*, *SpAmDH* and *TiAmDH* were determined to be 0.38 U/mg, 0.30 U/mg, 0.35 U/mg, 0.28 U/mg, 0.14 U/mg, respectively, towards substrate **1a**, after incubating at 30 °C for 15 min.

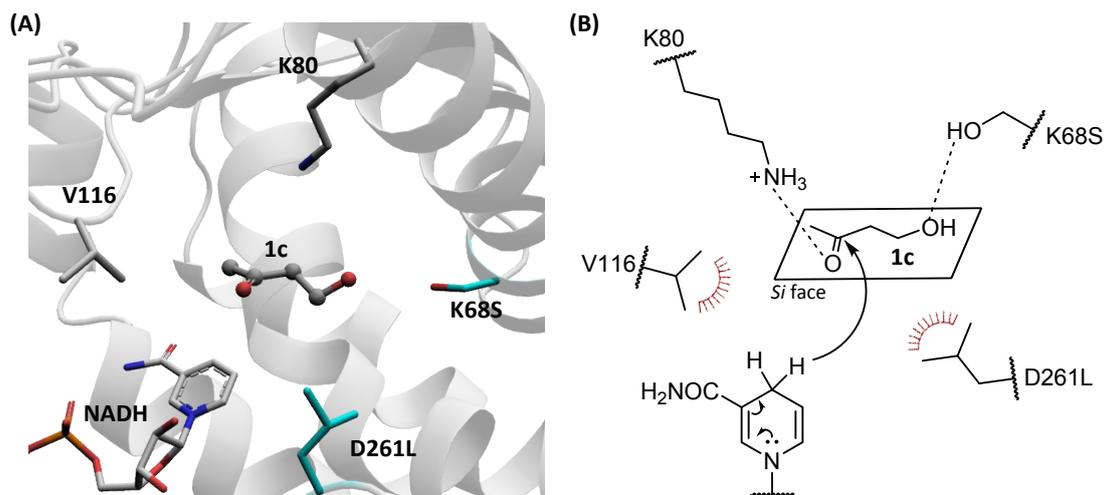


Figure S3. Docking analysis of substrate **1c** in GsAmDH. **(A)** Highest ranked docking pose for substrate **1c** in the catalytic pocket of GsAmDH; **(B)** Schematic representation of the interactions between substrate and surrounding residues. Nucleophilic attack (black arrow) at the *Si* face of the carbonyl group of **1c**, resulting in formation of (*S*)-**2c**. Hydrogen bonds are indicated by black lines, while penitential hydrophobic interactions are shown in red spikes.

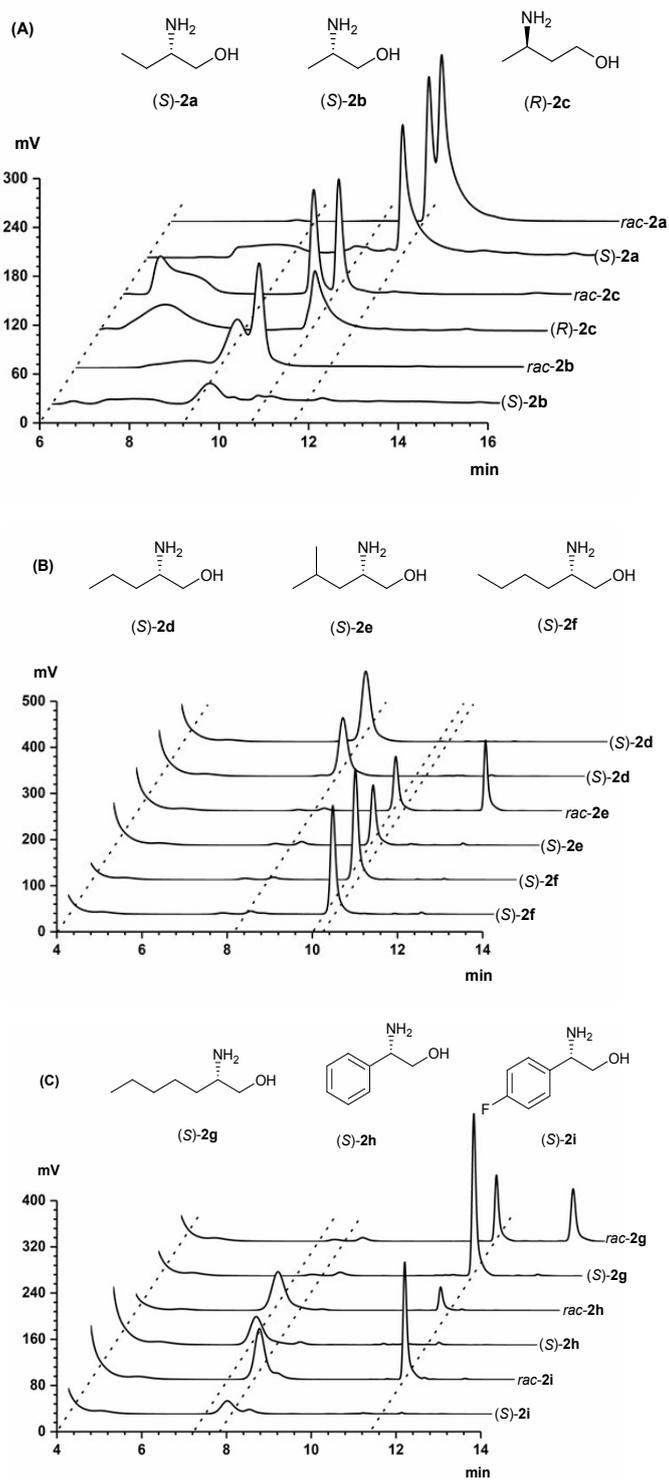
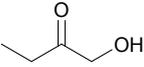
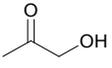
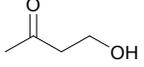
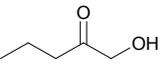
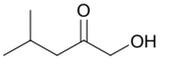
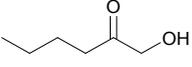
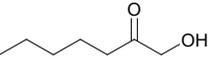
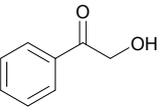
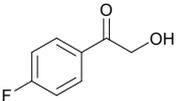


Figure S4. HPLC profiles of amino alcohols standards and products catalyzed by *GsAmdH*. (A) HPLC profiles of standards *rac*-2a, *rac*-2c, *rac*-2b and products (S)-2a, (R)-2c, (S)-2b; (B) HPLC profiles of standards (S)-2d, *rac*-2e, (S)-2f and products (S)-2d, (S)-2e, (S)-2f; (C) HPLC profiles of standards *rac*-2g, *rac*-2h, *rac*-2i and products (S)-2g, (S)-2h, (S)-2i.

Table S1. Analytical conditions of HPLC.

Substrates	Structures	Products	Retention time (min)		
			Substrate	(<i>S</i>)-Product	(<i>R</i>)-Product
1a	 1-hydroxybutan-2-one	2a ^a	n.a. ^c	11.7	12.0
1b	 1-hydroxypropan-2-one	2b ^a	n.a. ^c	9.6	10.1
1c	 4-hydroxybutan-2-one	2c ^a	n.a. ^c	10.3	10.8
1d	 1-hydroxypentan-2-one	2d ^b	n.a. ^c	8.3	n.a. ^c
1e	 1-hydroxy-4-methylpentan-2-one	2e ^b	n.a. ^c	10.1	12.2
1f	 1-hydroxyhexan-2-one	2f ^b	n.a. ^c	10.2	n.a. ^c
1g	 1-hydroxyheptan-2-one	2g ^b	n.a. ^c	11.4	13.2
1h	 2-hydroxy-1-phenylethan-1-one	2h ^b	n.a. ^c	7.4	11.2
1i	 1-(4-fluorophenyl)-2-hydroxyethan-1-one	2i ^b	n.a. ^c	8.0	11.4

^a HPLC conditions: Agilent SB-Aq C18 column (4.6*250 mm, 5 μm), injection volume 10 μL, column temperature 35 °C, flow rate 1 mL/min, detection wavelength 334 nm; buffer A: 0.05M sodium acetate, buffer B: methanol, gradient program: 70% A/30% B; 30% B, hold for 6 min, increase B to 45% in 1 min, hold for 8 min; decrease B to 30% in 0.5 min, hold for 4.5 min.

^b HPLC conditions: Zorbax SB-C18 column (4.6*150 mm, 5 µm), injection volume 10 µL, column temperature 25 °C, flow rate 1 mL/min, detection wavelength 340 nm; buffer A: ddH₂O (0.1% trifluoroacetic acid), buffer B: methanol (0.1% trifluoroacetic acid), gradient program: 40% A/60% B; decrease B to 40% in 3 min, hold for 0 min; increase B to 60% in 4 min, hold for 0 min; increase B to 80% in 3 min, hold for 3 min, decrease B to 60% in 2 min, hold for 0 min.

^c n.a. = not available.

Synthesis of α -hydroxy ketone substrates

General information:

^1H , ^{13}C spectra were recorded on Bruker AV 400 MHz instrument at 400 MHz (^1H NMR), 100 MHz (^{13}C NMR). or Bruker AV 600 MHz instrument at 600 MHz (^1H NMR), 150 MHz (^{13}C NMR). Chemical shifts were reported in ppm down field from internal Me_4Si and external CCl_3F , respectively. Multiplicity was indicated as follows: s (singlet), d (doublet), t (triplet), q (quartet), m (multiplet), dd (doublet of doublet), br (broad). Coupling constants are reported in Hertz (Hz).

General procedure A: synthesis of **1d** and **1f**^{1,2}

To a solution of the corresponding diol (37 mmol) in acetic acid (30 mL) was slowly added aqueous 2.8 M NaClO solution (13 mL, 1.1 equiv) at room temperature. After completion of the reaction (monitored by TLC), the mixture was then extracted with dichloromethane (3 x 100 mL). The combined organic phases were washed with aqueous NaHCO_3 solution and brine, dried over anhydrous MgSO_4 and concentrated in vacuo. Flash chromatography on silica gel with petroleum ether/ethyl acetate (6:1-2:1) afforded the α -hydroxy ketone product.

General procedure B: synthesis of **1e** and **1g**^{1,2}

To a mixture of the corresponding olefin (21 mmol), acetone (170 mL), water (38 mL) and acetic acid (8 mL) at room temperature was added dropwise a KMnO_4 solution (24 mmol, 5 g in 64 mL acetone and 21 mL water) and the mixture was stirred at this temperature until the olefin was completely converted (by TLC). Then the reaction mixture was filtered through a celite pad. The combined filtrate was concentrated under reduced pressure to remove the acetone and then extracted with dichloromethane (3 x 100 mL). The combined organic layers were neutralized by repeated washings with aqueous NaHCO_3 solution (1 M, 3 x 100 mL), dried over anhydrous MgSO_4 , and concentrated in vacuo. Purification by flash chromatography on silica gel eluting with petroleum ether/ethyl acetate (6:1-2:1) gave the α -hydroxy ketone product.

1-hydroxypentan-2-one (**1d**)

Colorless liquid; 1.80 g; 50% yield; ^1H NMR (400 MHz, CDCl_3) δ 4.19 (d, $J = 3.6$ Hz, 1H), 3.69 – 3.08 (m, 1H), 2.35 (t, $J = 7.4$ Hz, 1H), 1.76 – 1.52 (m, 1H), 0.90 (t, $J = 7.4$ Hz, 2H). ^{13}C NMR (101 MHz, CDCl_3) δ 209.9, 68.1, 40.3, 17.2, 13.7.

1-hydroxy-4-methylpentan-2-one (**1e**)

Colorless liquid; 0.39 g; 18% yield; **¹H NMR** (400 MHz, CDCl₃) δ 5.33 (s, 1H), 4.24 (s, 4H), 2.32 (d, *J* = 7.0 Hz, 5H), 2.28 – 2.14 (m, 3H), 0.97 (d, *J* = 6.6 Hz, 14H). **¹³C NMR** (101 MHz, CDCl₃) δ 209.5, 68.7, 47.3, 24.9, 22.5.

1-hydroxyhexan-2-one (**1f**)

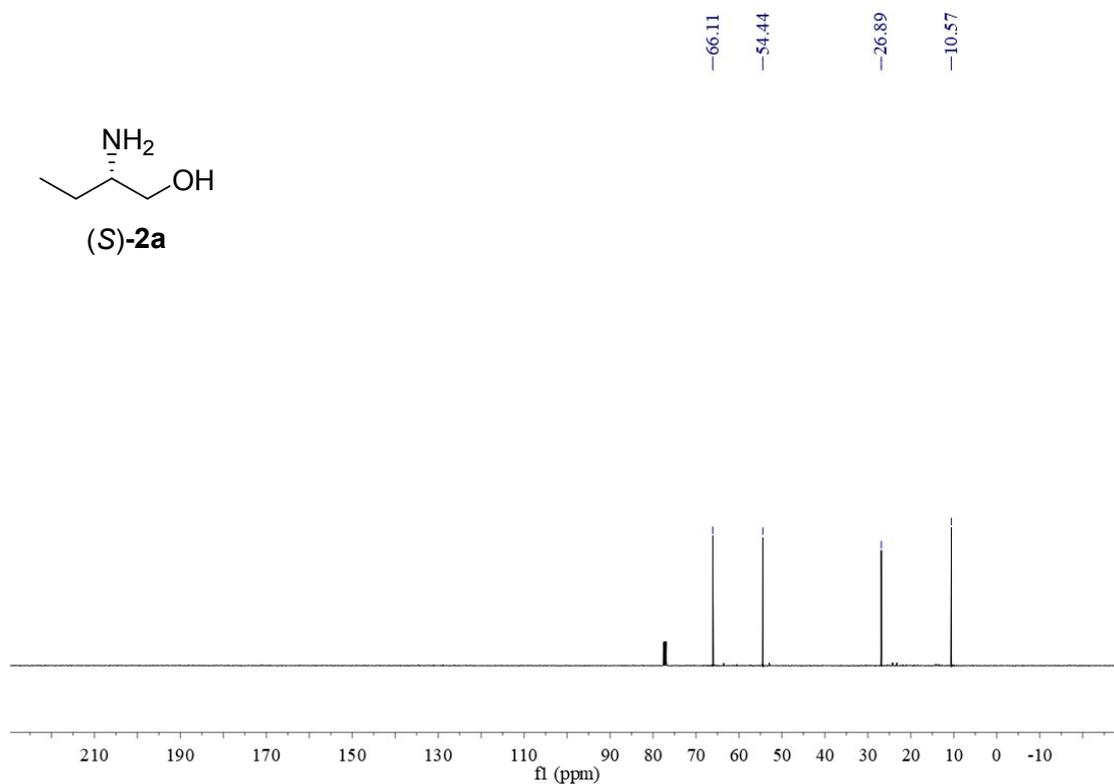
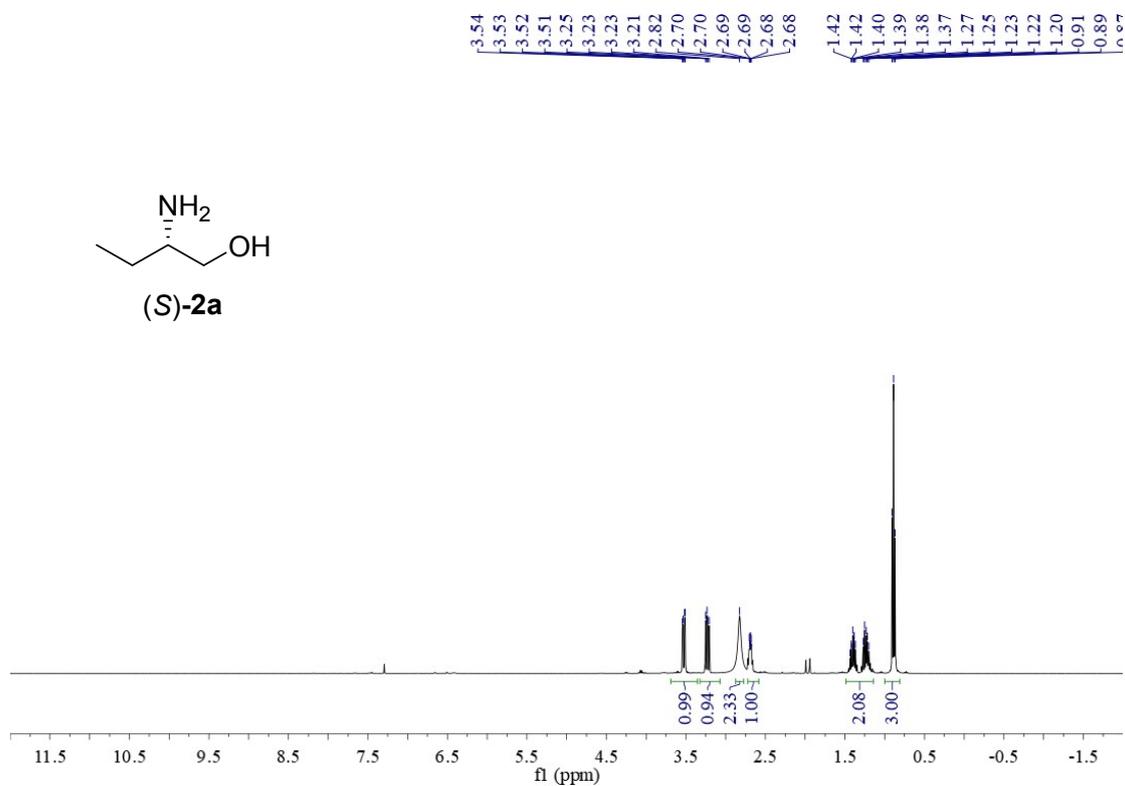
Colorless liquid; 2.56 g; 60% yield; **¹H NMR** (400 MHz, CDCl₃) δ 4.27 (d, *J* = 3.4 Hz, 1H), 3.19 (s, 1H), 2.44 (t, *J* = 7.5 Hz, 1H), 1.65 (dt, *J* = 15.2, 7.5 Hz, 1H), 1.49 – 1.21 (m, 1H), 0.95 (t, *J* = 7.3 Hz, 2H). **¹³C NMR** (101 MHz, CDCl₃) δ 210.0, 68.0, 38.1, 25.7, 22.3, 13.7.

1-hydroxyheptan-2-one (**1g**)

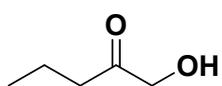
Colorless liquid; 0.93 g; 34% yield; **¹H NMR** (400 MHz, CDCl₃) δ 4.24 (s, 1H), 3.16 (s, 1H), 2.41 (t, *J* = 7.5 Hz, 1H), 1.78 – 1.54 (m, 1H), 1.31 (dt, *J* = 7.4, 4.8 Hz, 2H), 0.89 (t, *J* = 6.9 Hz, 2H). **¹³C NMR** (101 MHz, CDCl₃) δ 210.0, 68.1, 38.4, 31.3, 23.4, 22.3, 13.8.

NMR spectra

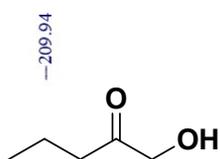
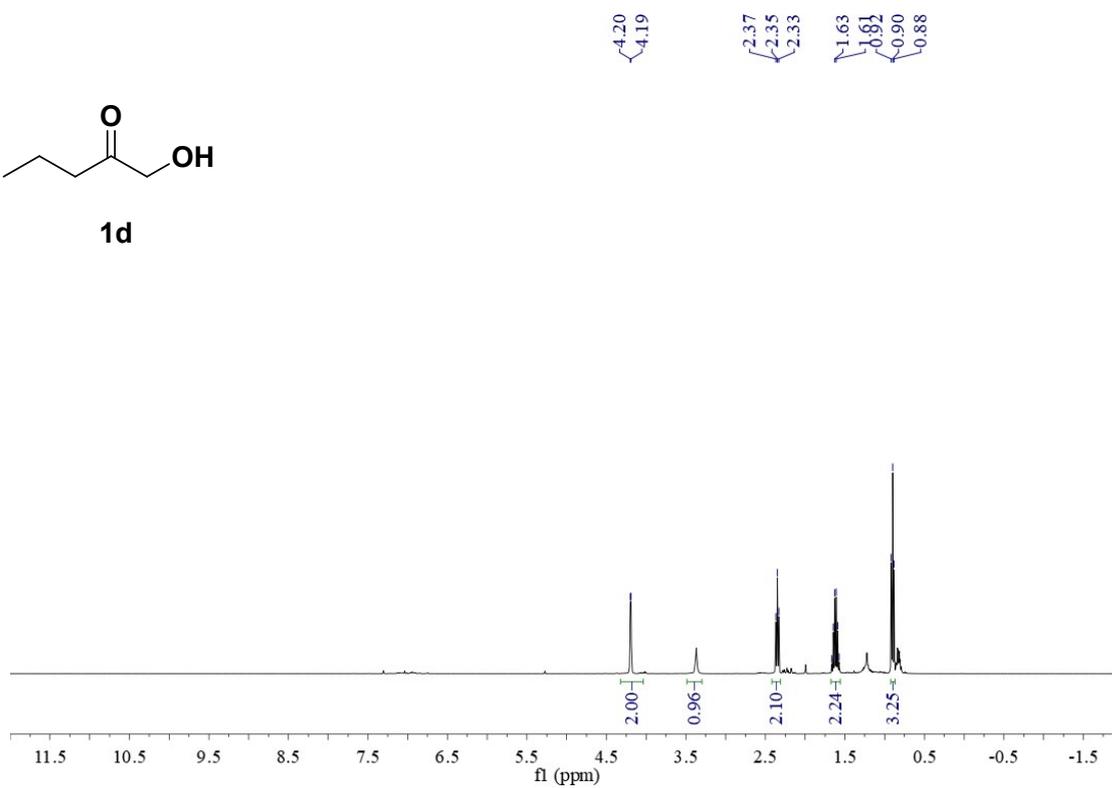
(S)-2-aminobutan-1-ol (2a)



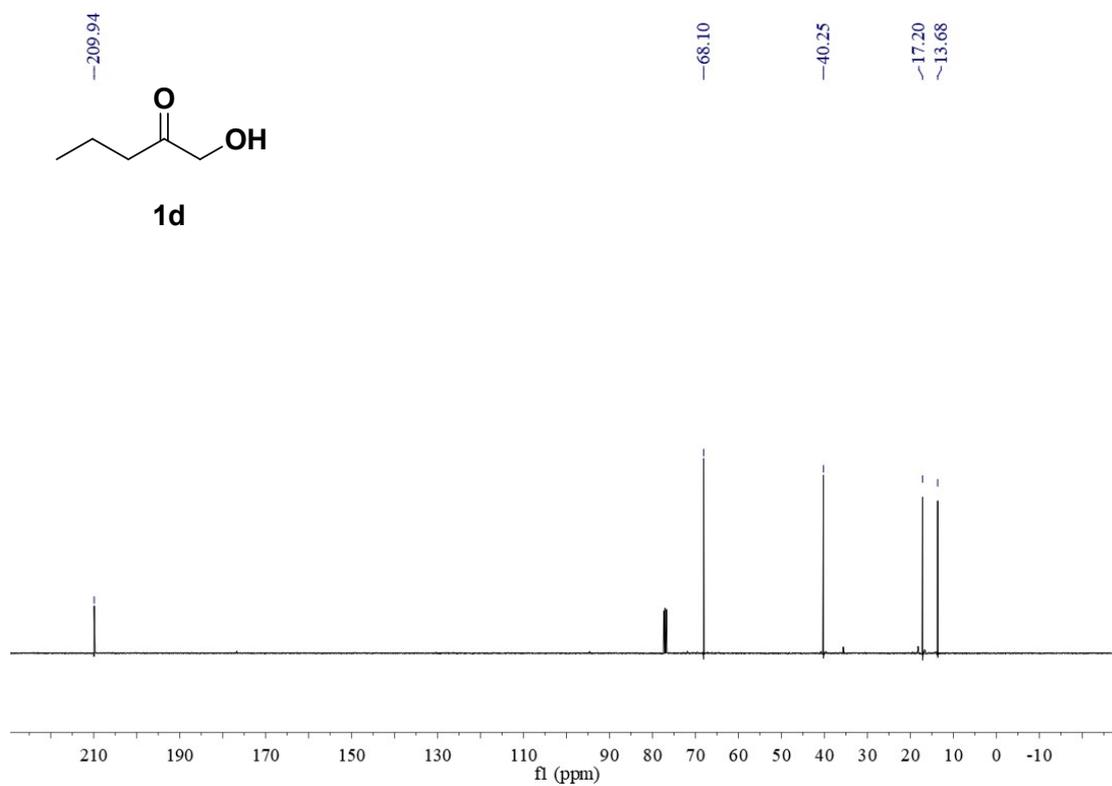
1-hydroxypentan-2-one (1d)



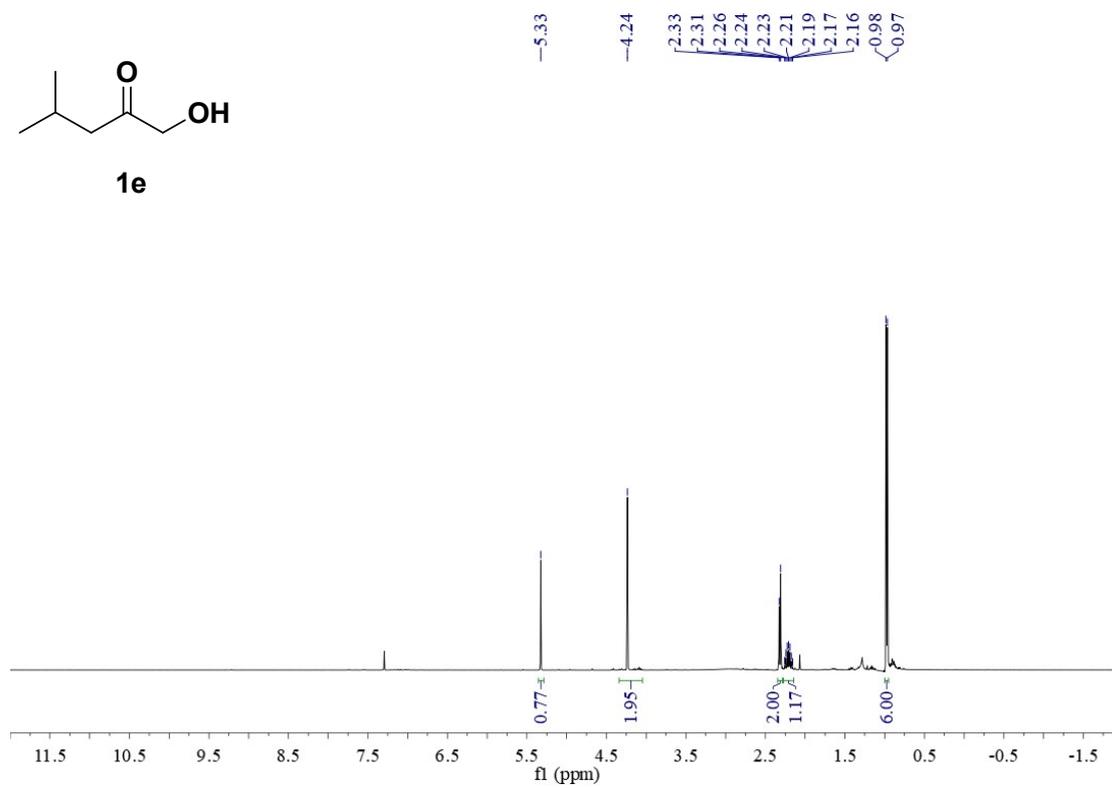
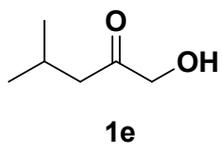
1d

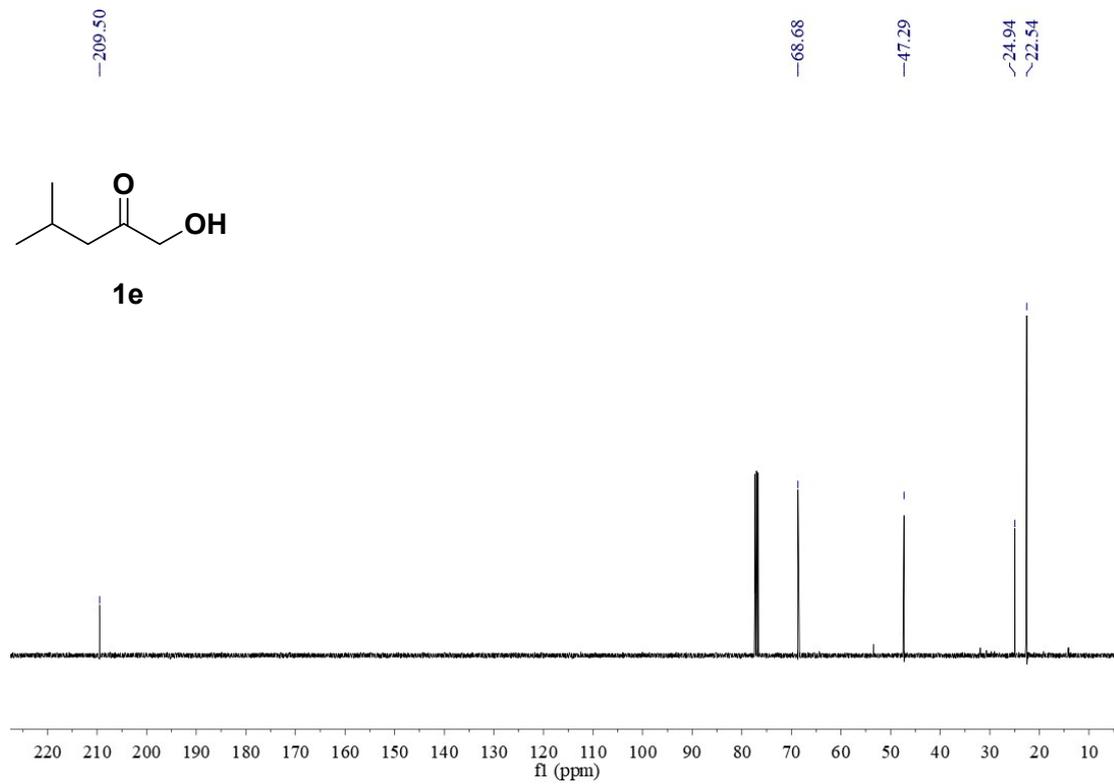


1d

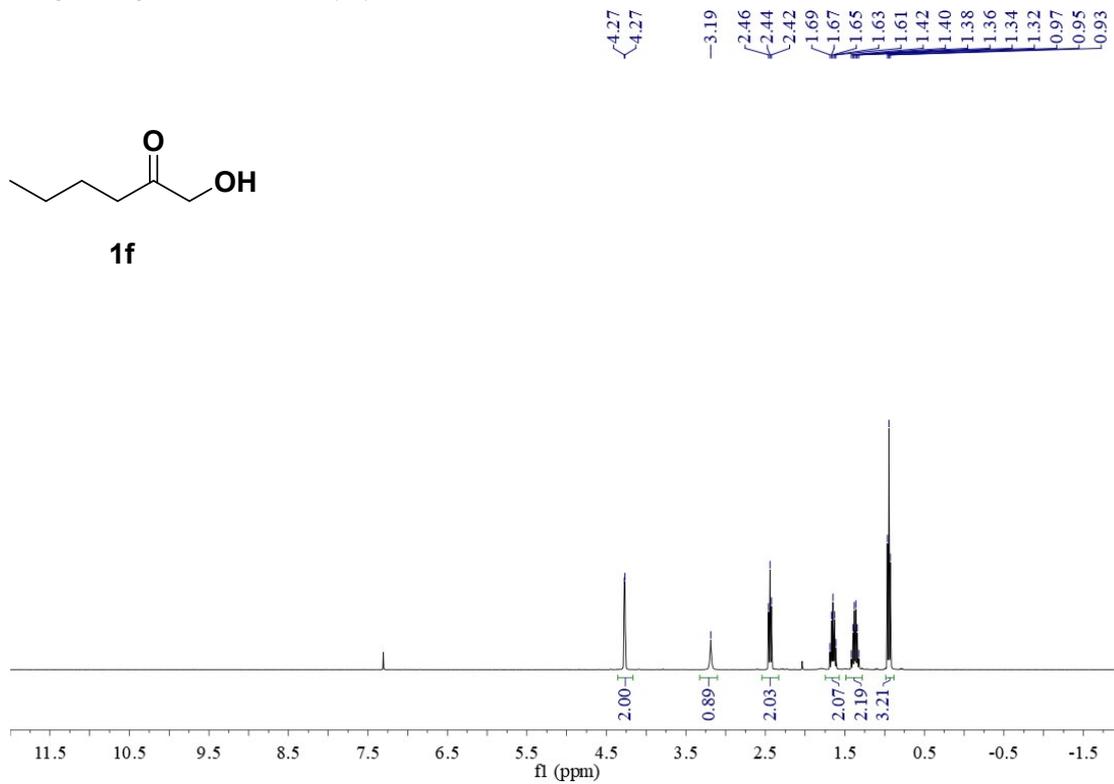


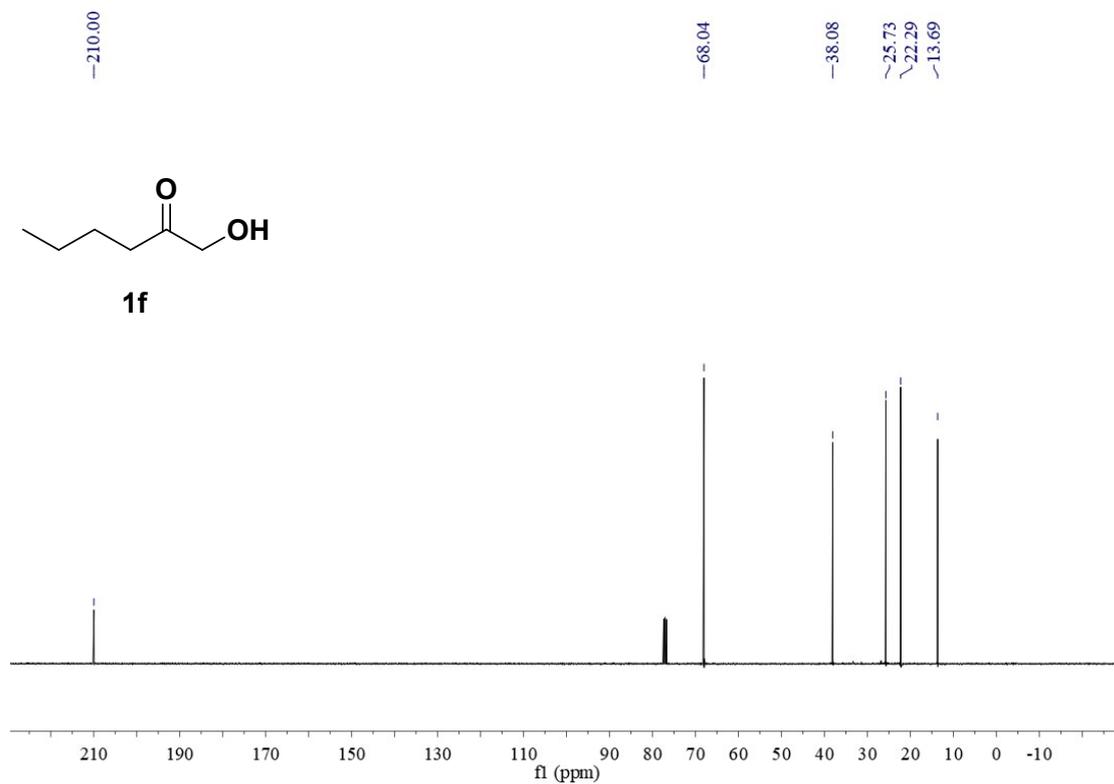
1-hydroxy-4-methylpentan-2-one (1e)



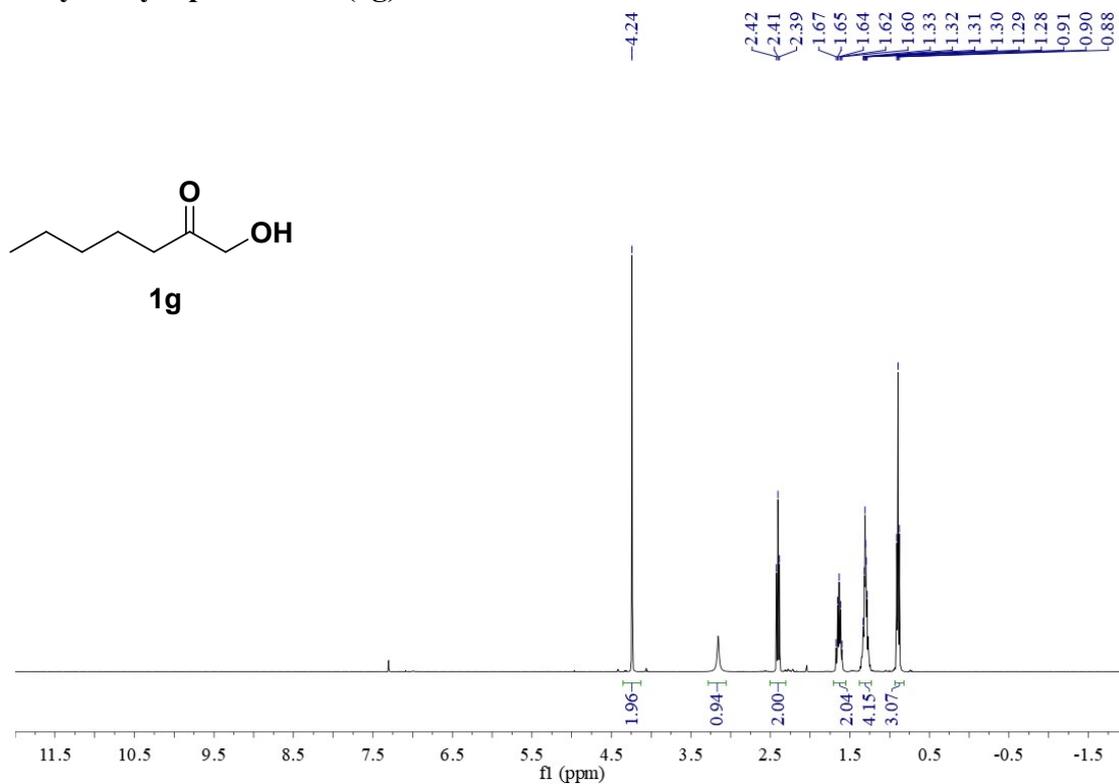


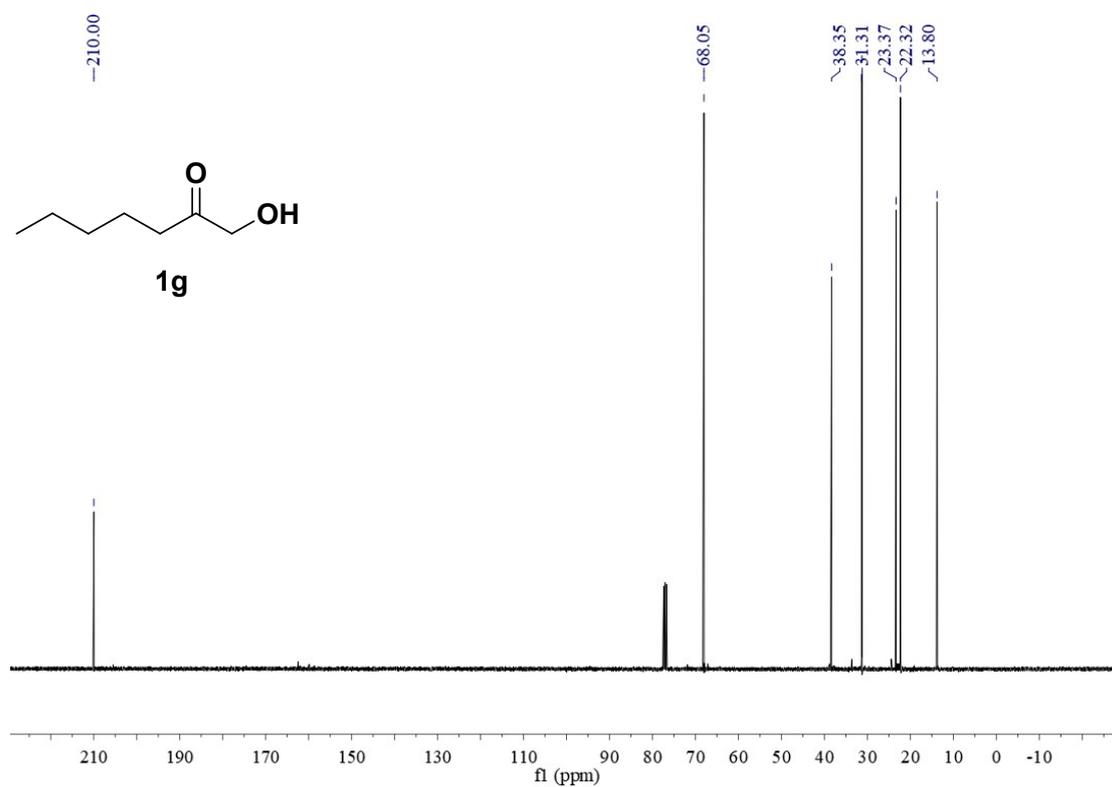
1-hydroxyhexan-2-one (1f)





1-hydroxyheptan-2-one (1g)





References

- 1 D. Guclu, M. Rale, and W. D. Fessner, *Eur. J. Org. Chem.*, 2015, **13**, 2960-2964.
- 2 F. F. Chen, S. C. Cosgrove, W. R. Birmingham, J. Mangas-Sanchez, J. Citoler, M. P. Thompson, G. W. Zheng, J. H. Xu, and N. J. Turner, *ACS. Catal.*, 2019, **9**, 11813-11818.