

Supplementary Information

Visible-Light Driven Photo-enhanced Zinc-Air Batteries using Synergistic Effect of Different Type of MnO₂ Nanostructures

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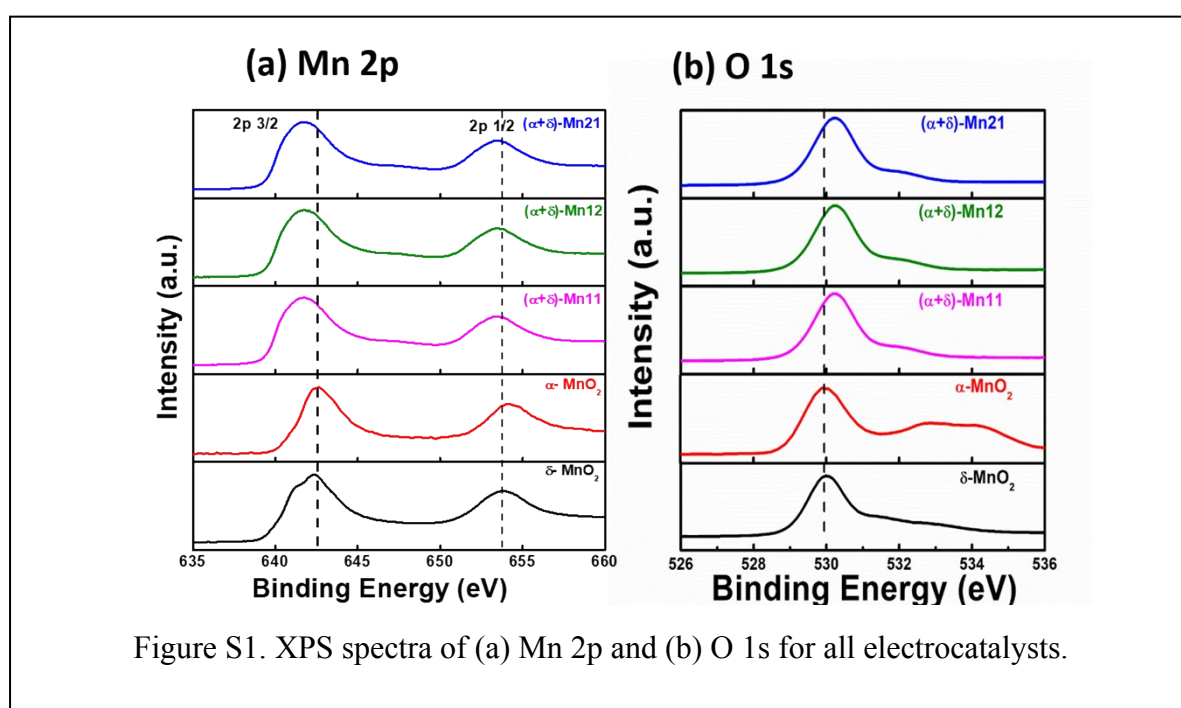


Figure S1. XPS spectra of (a) Mn 2p and (b) O 1s for all electrocatalysts.

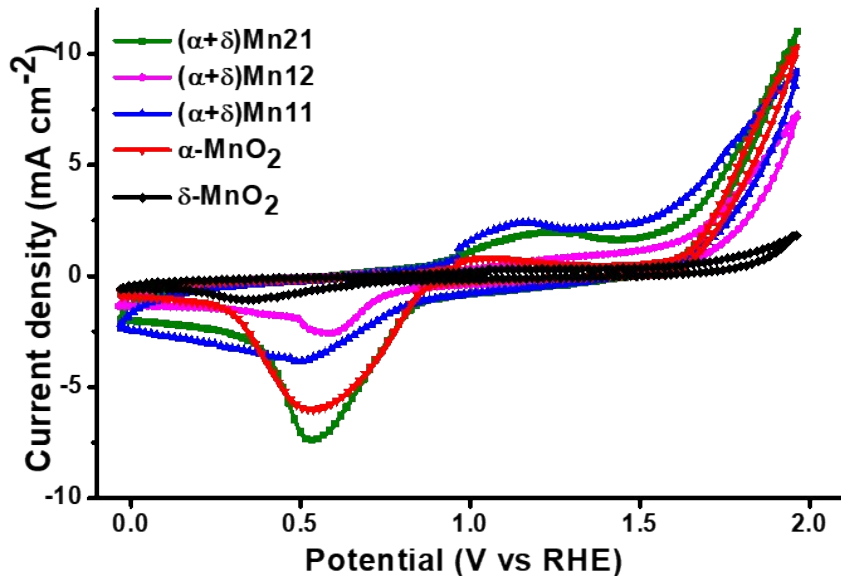


Fig. S2 Cyclic voltammety studies of composite $(\alpha+\delta)$ - Mn21, $(\alpha+\delta)$ - Mn12 and $(\alpha+\delta)$ - Mn11, along with α - MnO₂ and δ - MnO₂ taken in 0.1 M KOH electrolyte in O₂ atmosphere.

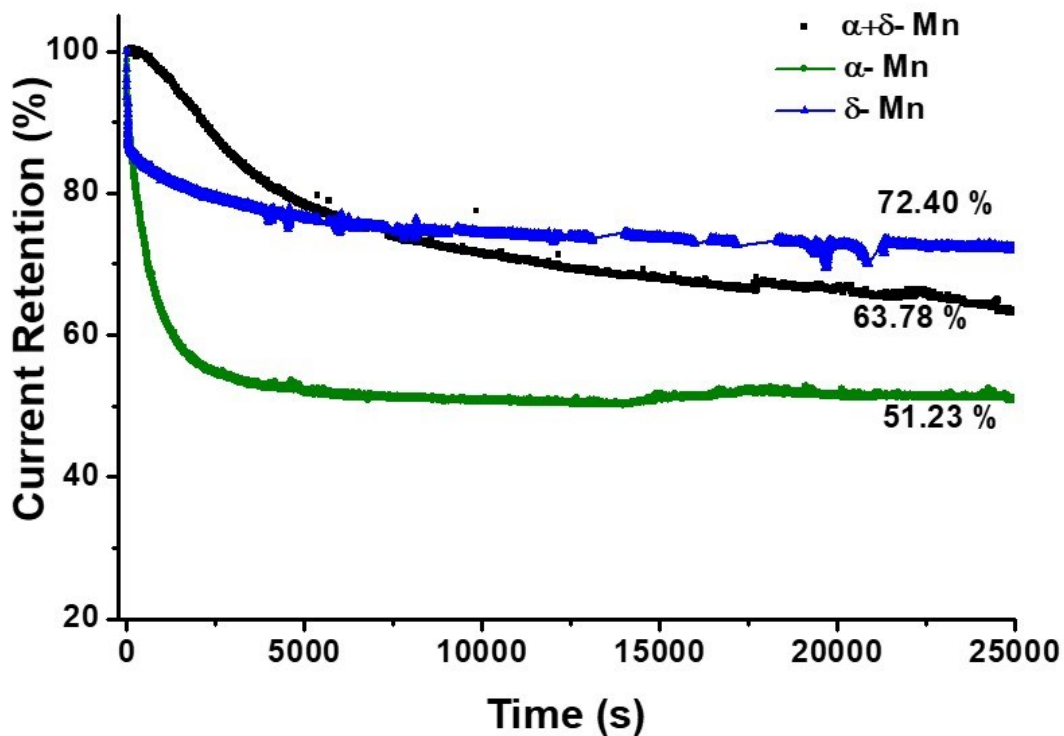


Fig. S3 Chronoamperometric response of $(\alpha+\delta)$ -Mn11, α - MnO₂ and δ - MnO₂ composite taken at -0.6 V (vs Ag/AgCl) and 600 rpm in O₂ saturated 0.1 M KOH electrolyte solution.

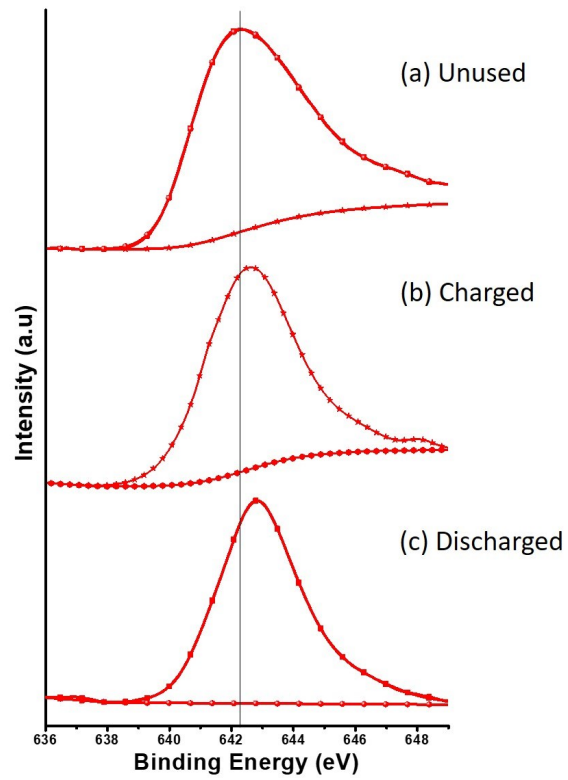


Fig. S4 XPS of Mn 2p for (a) Unused; (b) After charging; and (c) After discharging ($\alpha+\delta$)-Mn11 sample, showing a shift in the peak binding energy.

Table 1 showing atomic percentage of Mn in various oxidation states calculated from XPS

	Atomic percentage				
	α - MnO ₂	δ - MnO ₂	(α + δ)- Mn11	(α + δ)- Mn12	(α + δ)- Mn21
Mn (VI)	28.01	11.15	21.50	14.26	9.96
Mn(IV)	71.99	81.91	53.71	63.68	69.33
Mn(III)	---	2.87	6.86	12.23	10.08
Mn(II)	---	1.63	5.33	4.88	5.05

Table 2 showing surface area and pore radius measurement

	α - MnO ₂	δ - MnO ₂	(α + δ)- MnO ₂
Specific surface area (m ² /g)	57.63	171.40	52.03
Pore diameter (Å)	38.14	34.02	31.28

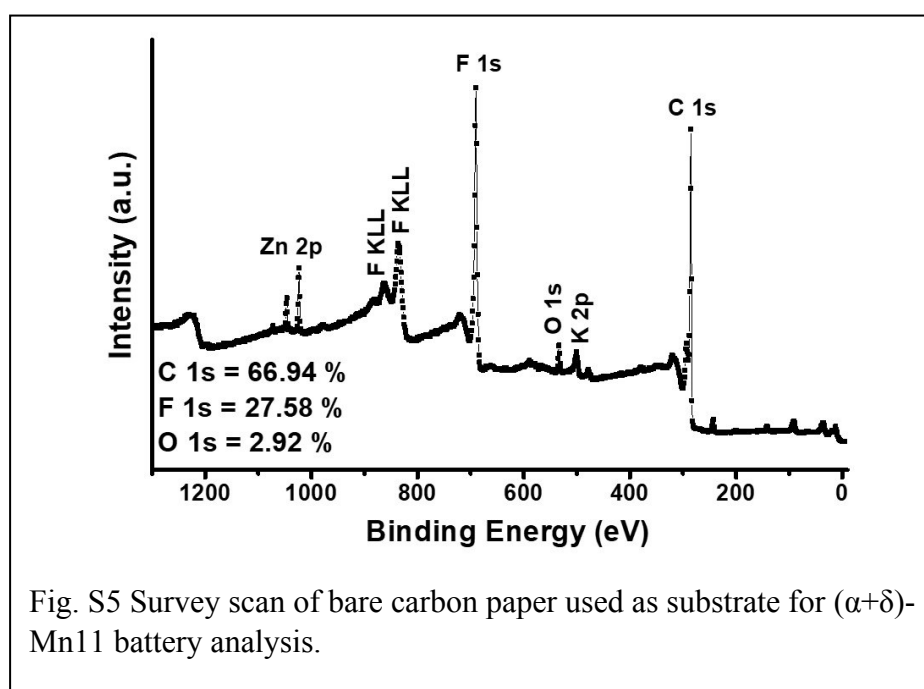


Fig. S5 Survey scan of bare carbon paper used as substrate for (α + δ)-Mn11 battery analysis.

Table 3 showing survey of various cathode materials used for Zn-air battery

	Specific Capacity at 5 mA cm ⁻²	Power density (mW cm ⁻²)	Reference
δ- MnO ₂	169.5 mAh	---	1
δ- MnO ₂	160.3 mAh	---	1
α- MnO ₂	175.2 mAh	---	1
α- MnO ₂	171.5 mAh	---	1
Fe doped α- MnO ₂	669.0 mAh g ⁻¹	30.65	2
α- MnO ₂ /CNT10	---	66.30	3
α- MnO ₂ /CNT20	----	65.40	3
α- MnO ₂ /DWCNT	---	61.60	3
α- MnO ₂ /XC72	---	61.50	3
α- MnO ₂ /SWCNT	---	59.50	3
α- MnO ₂ /AC	---	57.60	3
α- MnO ₂ /SMGP	---	40.50	3
MnO ₂ -IL _{0.5}	762.0 mAh g ⁻¹	166.00	4
CoMn ₂ O ₄ /rGO	460.0 mAh g ⁻¹ (20 mA cm ⁻²)	---	5
CoMn ₂ O ₄ /N- rGO	610.0 mAh g ⁻¹ (20 mA cm ⁻²)	---	5
Co ₃ O ₄ / MnO ₂	---	43.00	6
Co ₃ O ₄	---	15.00	6
MnO ₂	---	32.00	6
Pt/C	625 mAh g ⁻¹	140.00	7
α- MnO ₂	710.0 mAh g ⁻¹	6.91	This work
δ- MnO ₂	671.0 mAh g ⁻¹	15.77	This work
(α+δ)- MnO ₂	715.0 mAh g ⁻¹	28.53	This work

Reference:

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