

Supporting Information for “Water-Gas Shift Reaction Cocatalyzed by the Polyoxometalates (POMs)-gold Composites: The “Magic” Role of the POMs”

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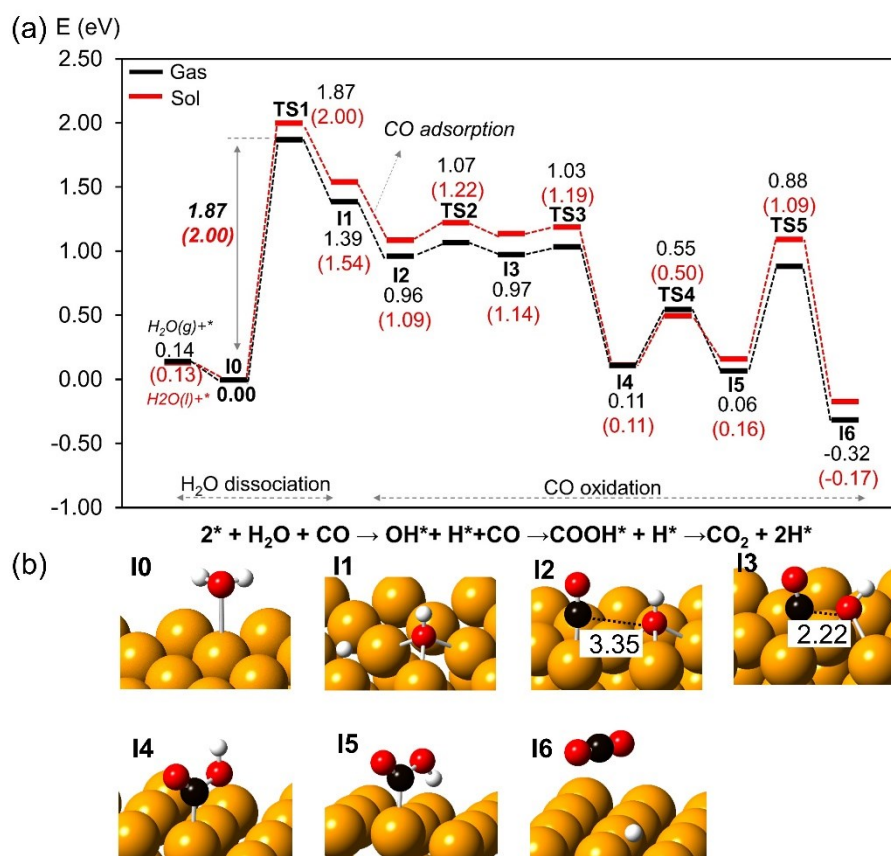


Fig. S1 (a) Calculated energy (eV) profile for WGS on bare Au(111) with (red) and without (black) solvent considered. All energies are referred to the water adsorbed state **I0** (defined as energy zero); (b) the main intermediates involved in the mechanism.

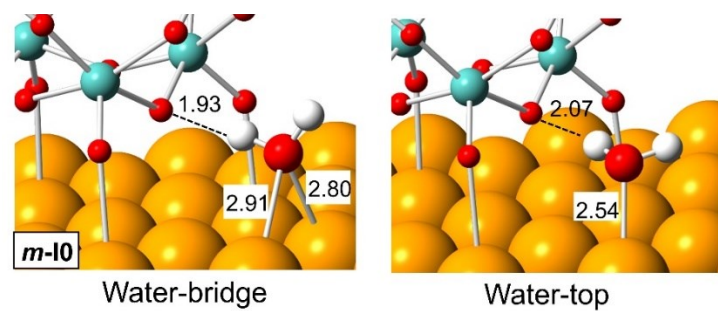


Fig. S2 The adsorption modes of water on K_3PMo_{12} -Au(111) through the bridge (left) and top (right) sites.

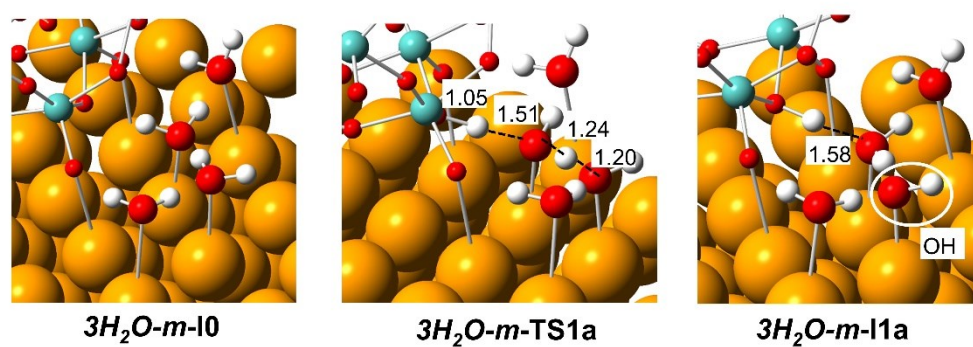


Fig. S3. The optimized geometries for water dissociation process in the presence of three explicit water molecules on PMo_{12} -Au(111).

Table S1. The *Bader AIM charge comparisons for the main intermediates and transition states involved in the proposed mechanism catalyzed by $\text{PMo}_{12}\text{-Au(111)}$ and $\text{PW}_{12}\text{-Au(111)}$.*

	ΣPM_{12}	ΣK	ΣAu	H_{w1}	H_{w2}	O_{w}	C	O_{Co}
<i>m-I1*</i>	-3.07	2.70	0.23	0.66	0.63	-1.22	1.10	-1.03
<i>m-TS1*</i>	-3.22	2.71	0.24	0.65	0.63	-1.07	1.13	-1.06
<i>m-I2*</i>	-3.49	2.71	0.25	0.67	0.64	-1.17	1.47	-1.08
<i>m-TS2*</i>	-3.54	2.71	0.20	0.65	0.63	-1.16	1.57	-1.06
<i>m-I3*</i>	-3.97	2.70	0.07	0.62	0.60	-1.07	2.09	-1.04
<i>w-I1*</i>	-2.84	2.72	0.12	0.62	0.63	-1.19	1.10	-1.04
<i>w-TS1*</i>	-2.85	2.72	-0.18	0.64	0.67	-1.09	1.19	-1.10
<i>w-I2*</i>	-3.04	2.72	-0.18	0.67	0.62	-1.12	1.42	-1.10
<i>w-TS2*</i>	-3.23	2.72	-0.21	0.69	0.63	-1.14	1.59	-1.05
<i>w-I3*</i>	-3.45	2.72	-0.44	0.58	0.63	-1.05	2.09	-1.06