Supporting Information

Defect engineering: an effective tool for enhancing the catalytic performance of copper-MOFs for the click reaction and the A³ coupling

Zhiying Fan,^[a] Zheng Wang,^[b] Mirza Cokoja,^[a] Roland A. Fischer^{*[a]}

- [a] Chair of Inorganic and Metal-Organic Chemistry, Catalysis Research Center and Department of Chemistry, Technical University of Munich, Ernst-Otto-Fischer-Straße 1, D-85748 Garching bei München, Germany. e-mail: roland.fischer@tum.de
- [b] Key Laboratory of Synthetic and Natural Functional Molecule Chemistry of Ministry of Education, College of Chemistry and Materials Science, Northwest University, 710127 Xi'an, Shaanxi, China.



Fig. S1 PXRD patterns of CuBTC and DE-CuBTC MOFs.



Fig. S2 ¹H NMR spectra of base-digested, activated CuBTC and DE-CuBTC MOFs.

The ¹H NMR results were achieved by following our previously report.^{S1} The activated samples were digested in NaOD/D₂O and the mixtures were heated. During this process, the Cu contents were transformed into a black solid which could be filtered off, and the organic linkers (BTC) in the frameworks were dissolved. Therefore, the shielding effect of the paramagnetic Cu ions on NMR is excluded. As shown in Figure S2, the peak at around 8.17 ppm (orange region) corresponds to the three protons on the benzene ring of BTC. The peak at around 8.35 ppm (green region) is assigned to one of the protons of PyDC. The calculation is shown below:

Doping ratio (PyDC) = I_{green} /[(I_{orange}/3)+I_{green}]×100%

the amount of 1 ypc to that of total linkers of the obtained be-cubic mores.				
Sample	Igreen	I _{orange}	Doping Ratio of PyDC to BTC (%)	
			Feeding	Obtained
CuBTC-PyDC_10	1	31.13	10	8.8
CuBTC-PyDC_20	1	14.38	20	17.2
CuBTC-PyDC_30	1	8.16	30	26.8

Table S1. The integration of peaks labelled by coloured regions and doping ratios of the amount of PyDC to that of total linkers of the obtained DE-CuBTC MOFs.



Fig. S3 N_2 adsorption isotherms of CuBTC and DE-CuBTC MOFs.



Fig. S4 The density functional theory (DFT) pore size distribution curves of CuBTC and DE-CuBTC MOFs.



Fig. S5 Deconvoluted XP spectra in the Cu 2p 3/2 region of CuBTC and DE-CuBTC MOFs.



Fig. S6 a) deconvoluted XP spectra in the Cu 2p 3/2 region, b) PXRD pattern, c) N_2 adsorption isotherm, d) pore size distribution curve of CuBTC-PyDC_40.



Fig. S7 TG curves of the prepared DE-CuBTC MOFs in comparison with the pristine CuBTC.



Fig. S8 FTIR spectra of CuBTC-PyDC_30 before and after phenylacetylene treatment.



Fig. S9 a) PXRD pattern, b) N_2 adsorption isotherm, c) pore size distribution curve, d) TG curve, and e) deconvoluted XP spectra in the Cu 2p 3/2 region of CuBTC-RVT.



Fig. S10 PXRD patterns of CuBTC-PyDC_30 before and after A³ coupling reaction.

References

S1 Z. Fan, J. Wang, W. Wang, S. Burger, Z. Wang, Y. Wang, C. Wöll, M. Cokoja and R. A. Fischer, *ACS Applied Materials & Interfaces*, 2020. DOI: 10.1021/acsami.0c07249