

## Supporting Information

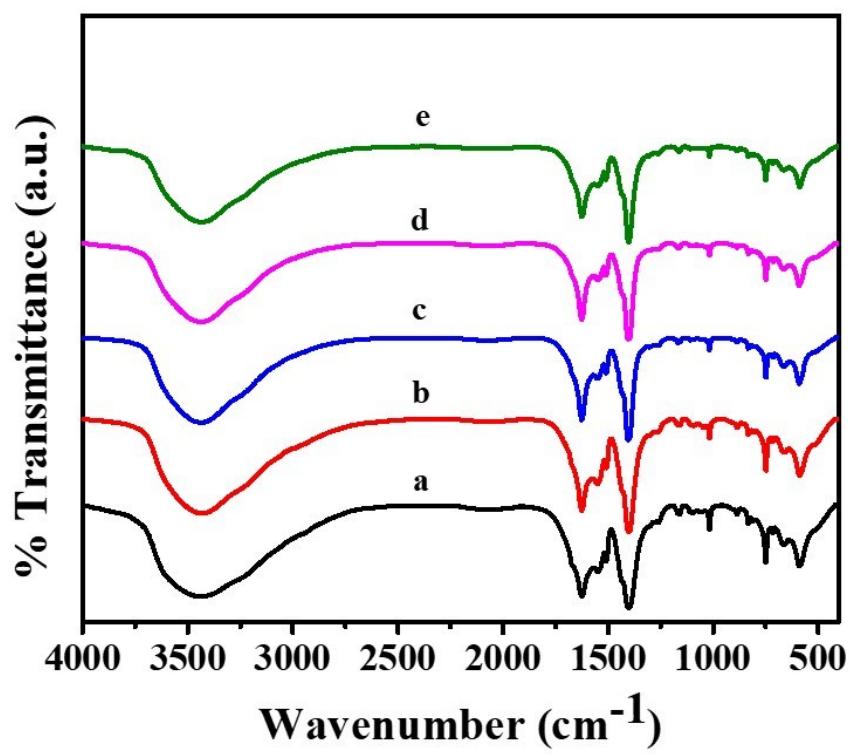
Platinum and cobalt intermetallic nanoparticles confined within MIL-101(Cr) for enhanced selective hydrogenation of the carbonyl bond in  $\alpha$ ,  $\beta$ -unsaturated aldehydes: synergistic effects of electronically-modified Pt sites and Lewis acid sites

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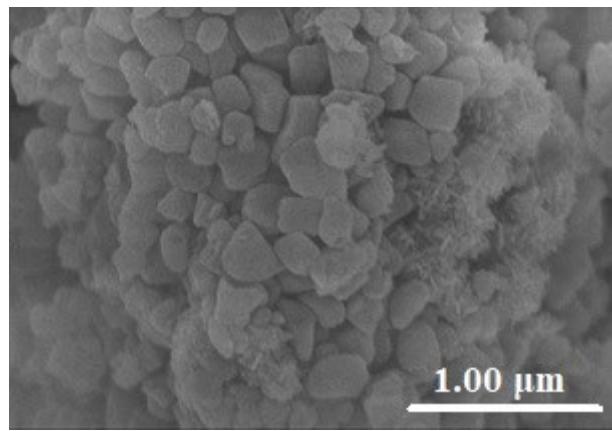
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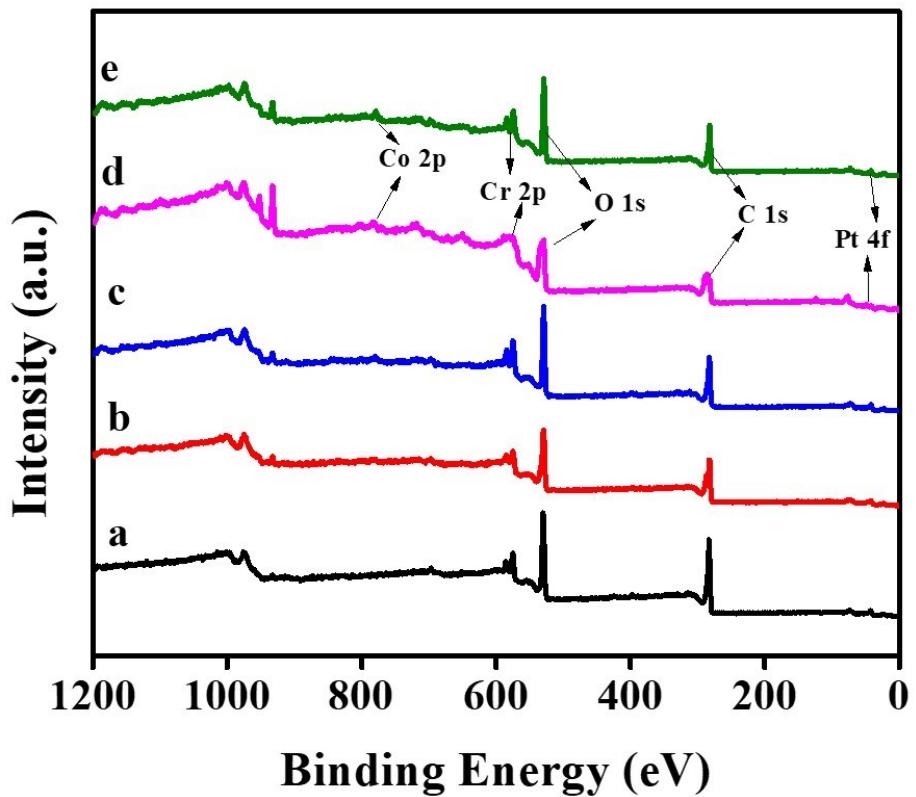
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**Fig. S1** FTIR spectra of (a) MIL-101(Cr), (b) 3%Pt/MIL-101(Cr), (c) 3%Pt1%Co/MIL-101(Cr), (d) 3%Pt3%Co/MIL-101(Cr), and (e) 3%Pt5%Co/MIL-101(Cr).



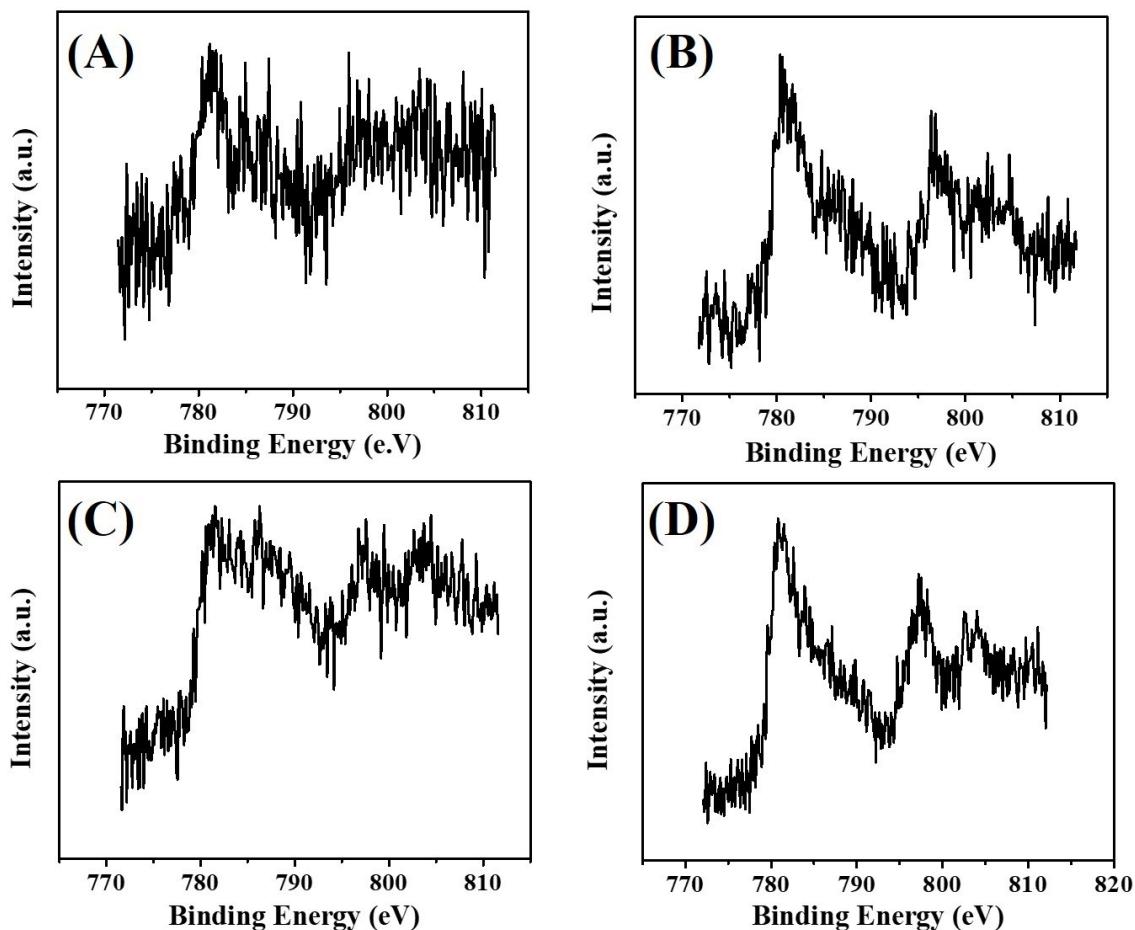
**Fig. S2** SEM image of MIL-101(Cr)



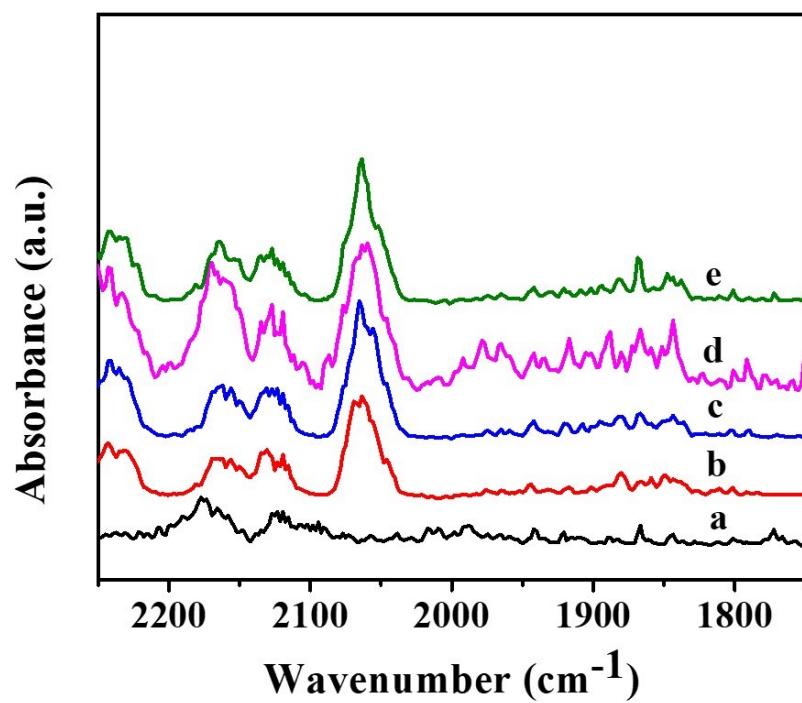
**Fig. S3** XPS survey scans for (a) 3%Pt/MIL-101(Cr), (b) 3%Pt1%Co/MIL-101(Cr), (c) 3%Pt3%Co/MIL-101(Cr), (d) 3%Pt5%Co/MIL-101(Cr), and (e) Used-3%Pt3%Co/MIL-101(Cr)

**Table S1** Pt 4f<sub>7/2</sub> XPS peak positions for the different catalysts reported here (binding energies, BE, in eV)

Catalyst	Pt(0)	Pt(II)	Pt(IV)
3%Pt/MIL-101(Cr)	70.80	73.65	75.55
3%Pt1%Co/MIL-101(Cr)	71.35	73.65	75.55
3%Pt3%Co/MIL-101(Cr)	71.15	73.45	75.35
3%Pt5%Co/MIL-101(Cr)	71.05	73.50	75.50
Used-3%Pt3%Co/MIL-101(Cr)	71.35	74.45	75.70



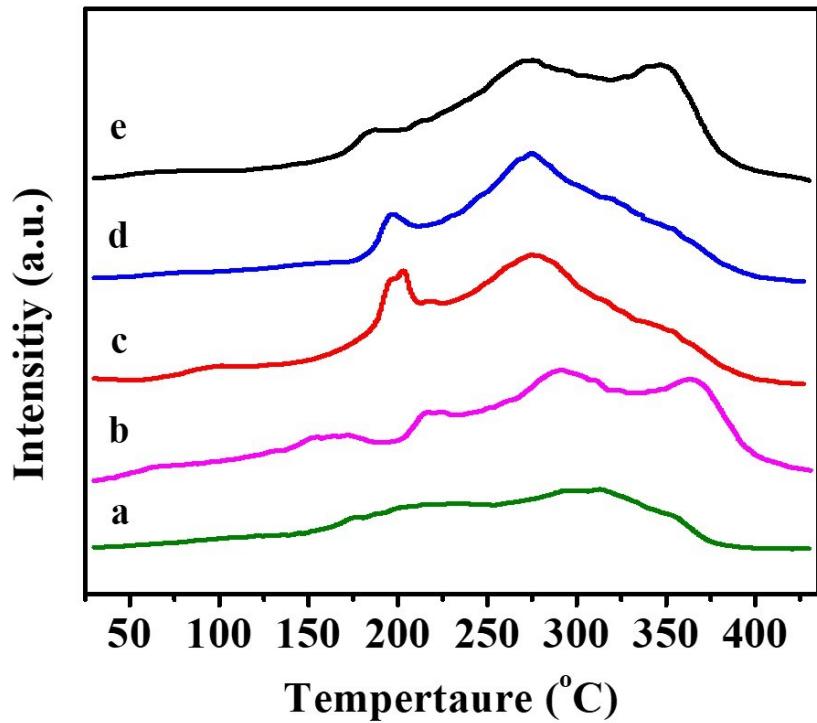
**Fig. S4** Co 2p XPS spectra for (A) 3%Pt1%Co/MIL-101(Cr), (B) 3%Pt3%Co/MIL-101(Cr), (C) 3%Pt5%Co/MIL-101(Cr), and (D) Used-3%Pt3%Co/MIL-101(Cr)



**Fig. S5** FTIR spectra of CO adsorbed on (a) MIL-101(Cr) (b) 3%Pt/MIL-101(Cr), (c) 3%Pt1%Co/MIL-101(Cr), (d) 3%Pt3%Co/MIL-101(Cr), and (e) 3%Pt5%Co/MIL-101(Cr)

**Table S2** Peak positions for the C–O stretching mode of carbon monoxide linearly adsorbed on Pt<sup>0</sup> sites in the different catalysts reported here

Catalyst	Wavenumber (cm <sup>-1</sup> )
MIL-101	-
3%Pt/MIL-101(Cr)	2068
3%Pt1%Co/MIL-101(Cr)	2065
3%Pt3%Co/MIL-101(Cr)	2059
3%Pt5%Co/MIL-101(Cr)	2063

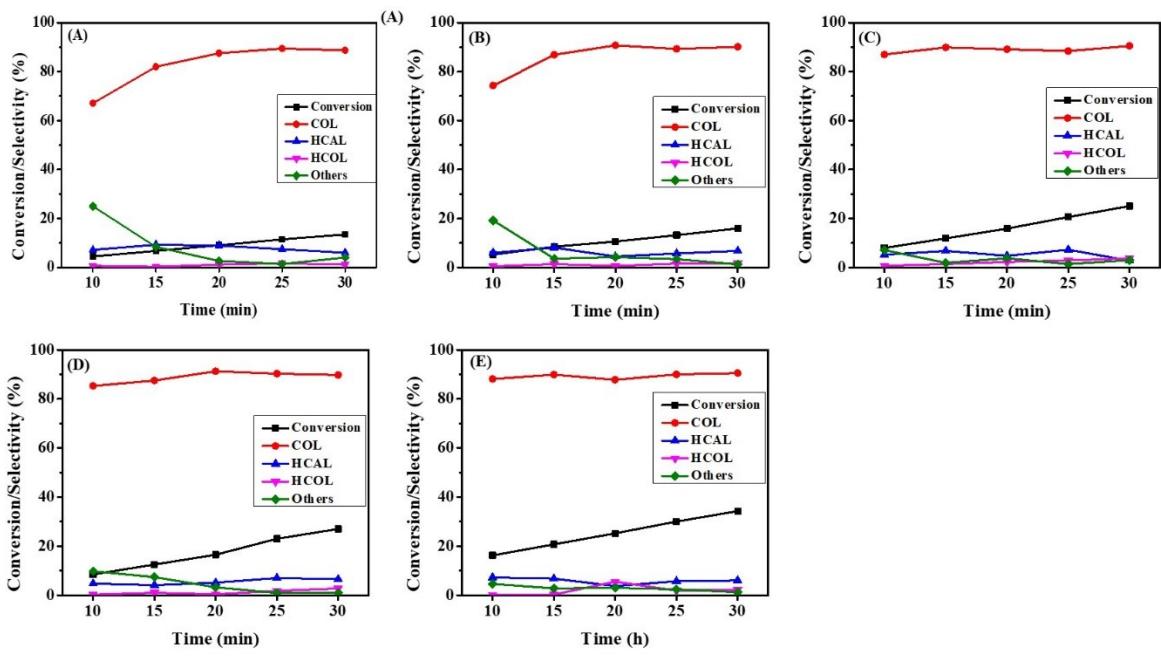


**Fig. S6** NH<sub>3</sub>-TPD profiles from (a) MIL-101(Cr), (b) 3%Pt/MIL-101(Cr), (c) 3%Pt1%Co/MIL-101(Cr), (d) 3%Pt3%Co/MIL-101(Cr), and (e) 3%Pt5%Co/MIL-101(Cr)

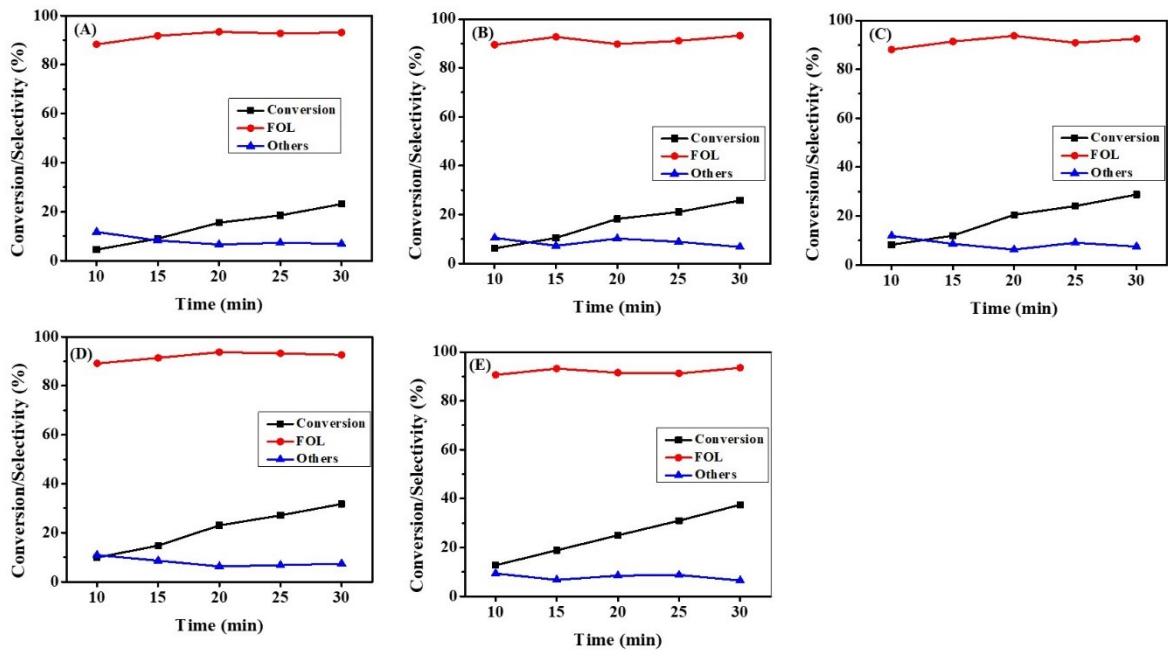
**Table S3** EXAFS fitted parameters for 3%Pt/MIL-101(Cr) and 3%Pt3%Co/MIL-101(Cr).

Catalysts	shell	CN <sup>[a]</sup>	R <sup>[b]</sup>	$\sigma^2 (10^{-3} \text{ \AA}^2)$ <sup>[c]</sup>	$\Delta E_0 (\text{eV})$ <sup>[d]</sup>	R <sup>[e]</sup>
3%Pt/MIL-101(Cr)	Pt–O	$1.00 \pm 0.01$	$2.00 \pm 0.02$	$2.65 \pm 0.2$	9.87	0.003
	Pt–Pt	$4.27 \pm 0.02$	$2.74 \pm 0.01$	$8.98 \pm 0.2$	4.91	
3%Pt3%Co/MIL-101(Cr)	Pt–O	$0.81 \pm 0.03$	$1.98 \pm 0.02$	$0.97 \pm 0.1$	6.31	0.001
	Pt–Pt	$5.28 \pm 0.07$	$2.73 \pm 0.01$	$10.2 \pm 0.2$	3.71	
Pt-foil	Pt–Pt	12	$2.76 \pm 0.05$	$5.63 \pm 0.3$	5.90	0.0018

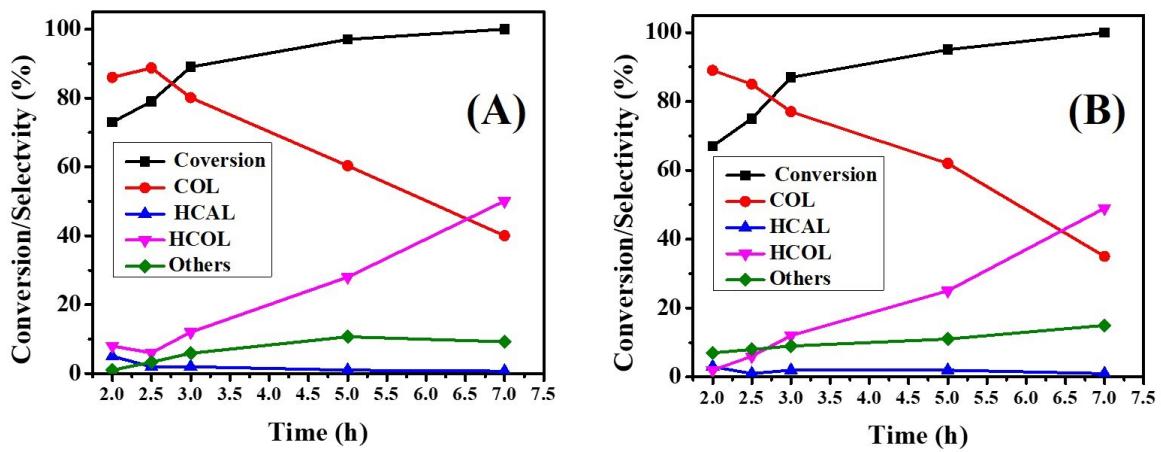
[a] Coordination number; [b] Distance between absorber and backscatterer atoms; [c] Debye–Waller factor; [d] Inner potential correction; [e] Correlation coefficient.



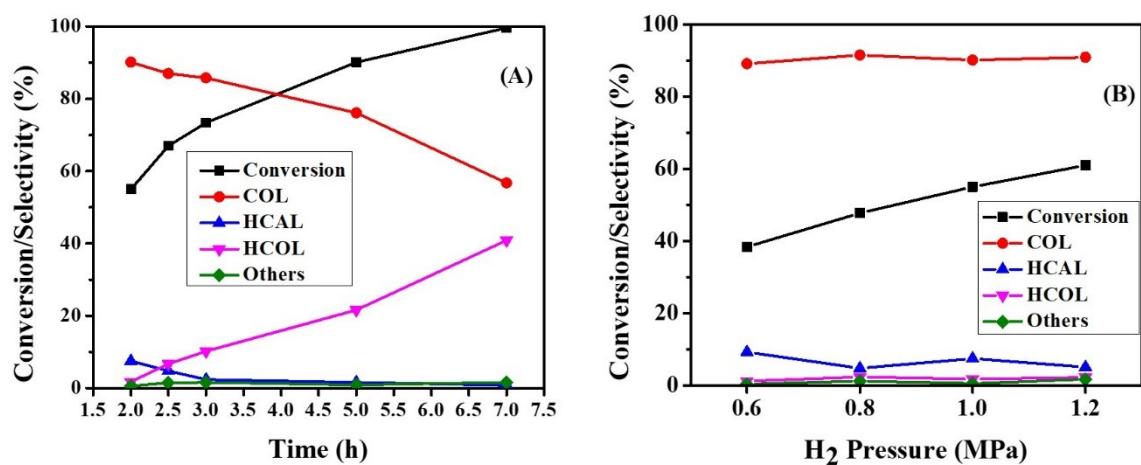
**Fig. S7** Kinetic data for the conversion of CAL using a 3%Pt3%Co/MIL-101(Cr) catalyst at (A) 30 °C (B) 45 °C (C) 60 °C (D) 70 °C and (E) 80 °C. Reaction conditions: 1 MPa H<sub>2</sub> pressure, 0.2525 g CAL, 20 mg catalysts, 15 mL isopropanol, 800 rpm.



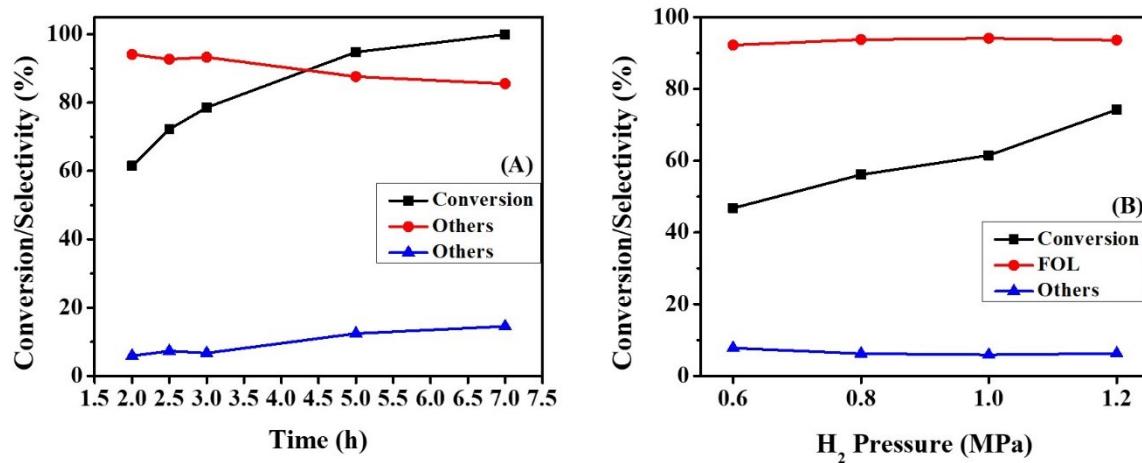
**Fig. S8** Kinetic data for the conversion of FFL using a 3%Pt3%Co/MIL-101(Cr) catalyst at (A) 30 °C (B) 45 °C (C) 60 °C (D) 70 °C and (E) 80 °C. Reaction conditions: 1 MPa H<sub>2</sub> pressure, 0.2525 g CAL, 20 mg catalysts, 15 mL isopropanol, 800 rpm.



**Fig. S9** Selective hydrogenation of CAL as a function of reaction time over the (A) 3%Pt1%Co/MIL-101(Cr) and (B) 3%Pt5%Co/MIL-101(Cr) catalysts (Reaction conditions: 60 °C, 1 MPa H<sub>2</sub> pressure, 0.2525 g CAL, 20 mg catalysts, 15 mL isopropanol, 800 rpm).



**Fig. S10** (A) Kinetic data for the conversion of CAL using a 3%Pt3%Co/MIL-101(Cr) catalyst at 30 °C, 1 MPa H<sub>2</sub> pressure, 0.2525 g CAL, 20 mg catalysts, 15 mL isopropanol, 800 rpm, (B) CAL total conversion and selectivity after 2 h of reaction versus H<sub>2</sub> pressure.



**Fig. S11** (A) Kinetic data for the conversion of FFL using a 3%Pt3%Co/MIL-101(Cr) catalyst at 30 °C, 1 MPa H<sub>2</sub> pressure, 0.2525 g FFL, 20 mg catalysts, 15 mL isopropanol and 800 rpm. (B) FFL total conversion and selectivity after 2 h of reaction versus H<sub>2</sub> pressure.

**Table S4** Comparison of catalytic performance for the hydrogenation of cinnamaldehyde (CAL) of 3%Pt3%Co/MIL-101(Cr) versus other reported Pt-based catalysts.

Catalysts	Conversion (%)	TOF	Time (h)	T (°C) & P(H <sub>2</sub> ) (MPa)	COL	Reference
3%Pt3%Co/MIL-101(Cr)	95	546 h <sup>-1</sup>	2	60 & 1	91	Current work
3%Pt3%Co/MIL-101(Cr)	38.4	216 h <sup>-1</sup>	2	30 & 0.6	89.1	Current work
PtNi <sub>2.20</sub> NWs@Ni/Fe <sub>4</sub> -MOF	99.5	77.6 h <sup>-1</sup>	2	40 & 3	83.3	1
10c-Co/2c-Pt/MWCNTs	93.3	-	12	80 & 1	93.4	2
3Pt0.05Sn/HPZSM-5	99.5	-	1	90 & 2	81.9	3
PtFe <sub>0.25</sub> /Al <sub>2</sub> O <sub>3</sub> @SBA-15	77.4	1.54 s <sup>-1</sup>	0.5	90 & 2	76.9	4
19 %wt Pt <sub>3</sub> Co@Co(OH) <sub>2</sub>	99.6	-	2	70 & 0.5	91.3	5
3wt% Pt-Mo <sub>2</sub> N/SBA-15	85.8	423 h <sup>-1</sup>	2	80 & 1	78	6
2 wt% Pt-Mo <sub>2</sub> N/SBA-15	70.8	521 h <sup>-1</sup>	2	80 & 1	76.9	6
Pt-Co/SBA-15-0.6	71	-	2	80 & 1	91	7
Pt <sub>3</sub> Fe/CNT	62.1	-	0.5	60 & 2	97.2	8
Pt-FeOx/15TS	99.0	1.36 s <sup>-1</sup>	1	90 & 4	84.5	9

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