

Supplementary Information

for

Carbon Supported Hybrid Catalysts for Controlled Product Selectivity in the Hydrosilylation of Alkynes

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1. Syntheses

General

Experiments were conducted under Ar atmosphere unless stated otherwise. Starting materials and solvents were purchased from Sigma-Aldrich, AK Scientific, Honeywell and Merck. Solvents were purified using an LC Technology Solution Inc. solvent purification system, and further dried by storage over 4 Å molecular sieves. NMR spectra were recorded on Bruker AV

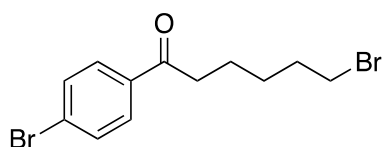
NEO 300, -AVIIIHD 300 MHz, -AVIIIHD 400 MHz, AV 400MHz, and AVIIIHD 500 MHz spectrometers. NMR spectra were referenced to residual solvent signals.¹ Mass spectra were acquired on a Q Exactive Plus hybrid quadrupole-Orbitrap mass spectrometer (ThermoFisher Scientific) equipped with a Vanquish UHPLC, a Bruker amaZon SL mass spectrometer, a Bruker Solarix 2xR 7T Fourier Transform Ion Cyclotron Resonance Mass Spectrometer and on an Agilent 7820A GC coupled to an Agilent 5977B MSD instrument. Microanalyses were obtained from the Macquarie University Chemical Analyses Facility. The reported value is the average of two individual measurements. IR spectra were recorded at a Thermo Scientific Nicolet iS5 spectrometer equipped with an iD5 ATR accessory and on a Shimadzu IRSpirit spectrometer equipped with QATR-S single-reflection ATR accessory.

Syntheses of Ligands

Propargylpyrazole: Pyrazole (3.0 g, 44.1 mmol) and propargylbromide (10.5 mL, 80 wt% in toluene) were dissolved in acetone (60 mL). An excess of K₂CO₃ (5 eq., 30.5 g, 220.3 mmol) was added and the suspension was refluxed for 16 h. The mixture was allowed to cool down to room temperature and was filtered through a glass frit. The acetone was removed using rotary evaporator and the crude product was dried under vacuum. Subsequently, it was filtered through a plug of celite using ethyl acetate as solvent. Solvent evaporation and drying under high vacuum afforded the product as a dark oil (4.10 g, 38.7 mmol, 88 %). ¹H and ¹³C NMR spectroscopic data was in agreement with the literature.²

(4-Bromophenyl)tri-*n*-butylstannane (2): 1,4-Dibromobenzene (3.5 g, 14.8 mmol) was dissolved in THF (140 mL) in a Schlenk flask and cooled to - 78 °C. *n*-Butyllithium (9.7 mL, 15.6 mmol, 1.05 eq.) was added slowly over 30 min under stirring using a syringe pump. Stirring was continued for another 30 min before tri-*n*-butyltin chloride (5.32 g, 16.3 mmol)

were added over 15 min using a syringe pump. The cooling bath was removed and the mixture was allowed to warm up to room temperature. Water (150 mL) was added and the mixture was transferred into a separatory funnel. The mixture was extracted with hexane (3×200 mL). The combined organic layers were dried over MgSO_4 and solvent evaporation afforded a colourless oil, which was either subject to column chromatography on silica using hexane eluent giving a major product fraction as first fraction, and another fraction containing minor impurities as second fraction, which was re-chromatographed. The obtained clean product fractions were combined and the solvent was evaporated using a rotary evaporator. This was followed by drying under vacuum, affording 5.25 g (11.8 mmol, 79 %) of a colorless oil. ^1H and ^{13}C NMR spectroscopic data was in agreement with the literature.³

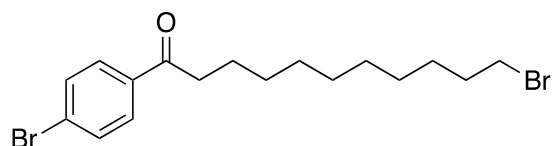


3a: A Schlenk flask was charged with 6-bromohexanoic acid (1.1 g, 5.6 mmol) and dichloromethane (30 mL).

Oxalylchloride (1.4 mL, 16.8 mmol) was added while

stirring at room temperature, followed two drops of DMF. The mixture was stirred for 45 min after which the gas evolution had ceased. The solvents and the excess of oxalylchloride were removed under vacuum and the colorless oily solids were dried and subsequently re-dissolved in 50 mL dichloromethane, followed by addition of aluminum(III) chloride (0.75 g, 5.6 mmol). The mixture was stirred for 15 min at room temperature. A second Schlenk flask was charged with **2** (2.5 g, 5.6 mmol) and dichloromethane (50 mL) and cooled to $-78\text{ }^\circ\text{C}$. The acid chloride / aluminum(III) chloride mixture was taken up with a syringe and added slowly over 1 h under stirring to the solution of **2** using a syringe pump. After completed addition, the cooling bath was removed and the mixture was allowed to warm up to room temperature, at which point water (100 mL) was added and the work-up was conducted on air. Phases were separated and the aqueous layer was extracted twice with dichloromethane (100 mL). The combined organic

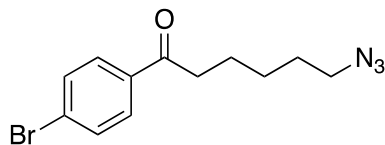
extracts were dried over MgSO_4 and concentrated to dryness using a rotary evaporator. The crude mixture was flash-filtered through a short column (12 cm $\times$ 3 cm), filled with silica/KF (1:1 /vol) in the upper half and silica in the lower half with pure silica, using dichloromethane as eluent, which removed efficiently the tin by-products.^{4, 5} Solvent evaporation gave a colorless oil, which crystallized upon standing. Re-crystallization from pentane at -20 °C afforded colorless needles, which was repeated twice to yield the title compound as white crystalline solid (859 mg, 2.6 mmol, 46 %). ^1H NMR (400 MHz, CDCl_3): δ = 7.82 (dm, $^3J_{\text{H-H}}$ = 8.7 Hz, 2H, Ar-*H*), 7.61 (dm, $^3J_{\text{H-H}}$ = 8.7 Hz, 2H, Ar-*H*), 3.43 (t, $^3J_{\text{H-H}}$ = 6.7 Hz, 2H, CH_2Br), 2.92 (t, $^3J_{\text{H-H}}$ = 7.2 Hz, 2H, COCH_2), 1.92 (m, 2H), 1.77 (m, 2H), 1.54 (m, overlaps with H_2O signal, 2H) ppm. ^{13}C NMR (101 MHz, CDCl_3): δ = 199.0, 135.8, 132.0, 129.7, 128.3, 38.4, 33.7, 32.7, 27.9, 23.3 ppm. MS (APCI) m/z : 385 $[\text{M} + \text{H}]^+$ (100%). HRMS (APCI): m/z $[\text{M} + \text{H}]$ calcd for $\text{C}_{12}\text{H}_{15}\text{Br}_2\text{O}$: 334.94637; found: 334.94636.



3b: The title compound was prepared analogously to compound **3a**. 11-bromoundecanoic acid (1.55 g, 5.8 mmol) and **2**

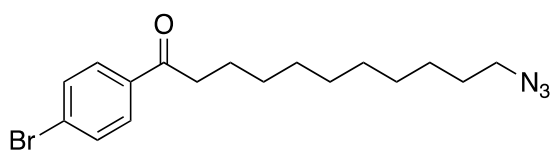
(2.6 g, 5.8 mmol) gave a colorless oil, which crystallized upon standing. Re-crystallization from pentane (250 mL) at 6 °C gave the product as white solid. The solids were collected using a Büchner funnel, washed with little pentane and dried under high vacuum, affording the product as white microcrystalline powder (1250 mg, 3.1 mmol, 53 %). ^1H NMR (400 MHz, CD_2Cl_2): δ = 7.82 (dm, $^3J_{\text{H-H}}$ = 8.7 Hz, 2H, Ar-*H*), 7.60 (dm, $^3J_{\text{H-H}}$ = 8.7 Hz, 2H, Ar-*H*), 3.41 (t, $^3J_{\text{H-H}}$ = 6.9 Hz, 2H, CH_2Br), 2.92 (t, $^3J_{\text{H-H}}$ = 7.4 Hz, 2H, COCH_2), 1.85 (m, 2H), 1.72 (m, 2H), 1.44-1.29 (brd, 12H) ppm. ^{13}C NMR (101 MHz, CD_2Cl_2): δ = 199.5, 135.9, 132.0, 131.71, 131.66, 130.0, 129.7, 128.7, 128.1, 38.7, 34.2, 32.9, 29.51, 29.47, 29.45, 29.4,

28.8, 28.3, 24.3 ppm. MS (APCI) m/z : 405 $[M + H]^+$ (100%), HRMS (APCI) m/z : calcd for $C_{17}H_{25}Br_2O$: 405.02462; found: 405.02474.



4a: Compound **3a** (600 mg, 1.8 mmol) was dissolved DMF (20 mL). Sodium azide (175 mg, 2.7 mmol) was added and the mixture was heated to 120 °C under stirring for 3h. The

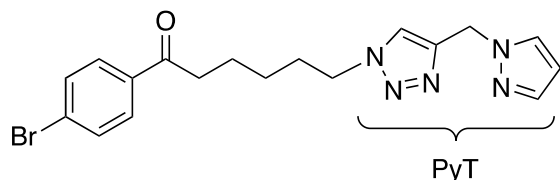
mixture was allowed to cool down to room temperature, diluted with ethyl acetate (120 mL) and extracted with water (3×150 mL). The organic layer was collected, dried over $MgSO_4$ and solvents were evaporated using a rotary evaporator, which was followed by drying under high vacuum. The product was obtained as pale yellow oil (0.526 mg, 1.8 mmol, 99 %). The reactive azide was only characterized by 1H and ^{13}C NMR spectroscopy and immediately used in the next step. 1H NMR (400 MHz, $CDCl_3$): δ = 7.80 (dm, $^3J_{H-H}$ = 8.6 Hz, 2H, Ar-*H*), 7.59 (dm, $^3J_{H-H}$ = 8.6 Hz, 2H, Ar-*H*), 3.28 (t, $^3J_{H-H}$ = 6.8 Hz, 2H, CH_2N_3), 2.94 (t, $^3J_{H-H}$ = 7.3 Hz, 2H, $COCH_2$), 1.75 (m, 2H, CH_2), 1.64 (m, 2H, CH_2), 1.45 (m, 2H, CH_2) ppm. ^{13}C NMR (101 MHz, $CDCl_3$): δ = 199.0, 135.7, 132.0, 129.6, 128.2, 51.3, 38.3, 28.9, 26.5, 23.7 ppm.



4b: The title compound was prepared analogously to compound **4a**. Compound **3b** (1.05 g,

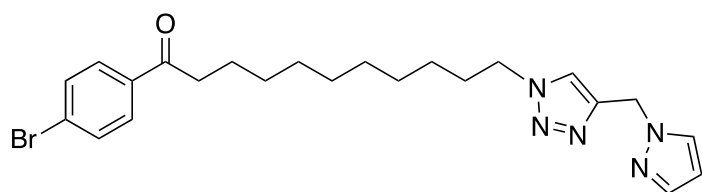
2.6 mmol) and sodium azide (255 mg, 3.9 mmol) in DMF (40 mL) gave the product as pale yellow oil (0.949 mg, 2.6 mmol, 99%), which crystallized upon standing. The reactive azide was only characterized by 1H and ^{13}C NMR spectroscopy and immediately used in the next step. 1H NMR (400 MHz, CD_2Cl_2): δ = 7.81 (dm, $^3J_{H-H}$ = 8.7 Hz, 2H, Ar-*H*), 7.59 (dm, $^3J_{H-H}$ = 8.7 Hz, 2H, Ar-*H*), 3.25 (t, $^3J_{H-H}$ = 7.0 Hz, 2H, CH_2N_3), 2.91 (t, $^3J_{H-H}$ = 7.3 Hz, 2H, 2H), 1.71 (m, 2H), 1.58 (m, 2H), 1.34 - 1.29 (brd, 14 H) ppm. ^{13}C NMR (101 MHz, CD_2Cl_2): δ = 199.6,

135.9, 132.0, 129.7, 128.1, 77.5, 77.2, 76.8, 51.6, 38.7, 29.51, 29.45, 29.4, 29.2, 28.9, 26.8, 24.3 ppm.



5a: A round bottom flask was charged with compound **4a** (500 mg, 1.7 mmol), *tert.*-butanol (20 mL) and water (40 mL). Propargylpyrazole (223 mg, 2.1 mmol), copper sulfate · 5 H₂O (34 mg, 0.1 mmol) and sodium ascorbate (27 mg, 0.1 mmol) were added while stirring. The mixture was stirred for 16 h at 50 °C. After the reaction was completed, water (100 mL) was added, which led to the formation of a precipitate. Work-up was carried out by extraction with ethyl acetate (3 × 80 mL). The combined organic layers were dried over MgSO₄, concentrated to dryness and dried under high vacuum to afford a light brown oil which crystallized upon standing. Yield: 656 mg (1.6 mmol, 97%). ¹H NMR (400 MHz, CD₂Cl₂): δ = 7.80 (dm, ³J_{H-H} = 8.6 Hz, 2H, Ar-*H*), 7.61 (dm, ³J_{H-H} = 8.6 Hz, 2H, Ar-*H*), 7.57 (brm, 2H, PyT), 7.52 (brm, 1H, PyT), 6.31 (brm, 1H, PyT), 5.39 (s, 2H, N₃CHC-CH₂), 4.33 (t, ³J_{H-H} = 7.1 Hz, 2H, CH₂PyT), 2.92 (t, ³J_{H-H} = 7.2 Hz, 2H, COCH₂), 1.92 (m, 2H, CH₂), 1.73 (m, 2H, CH₂), 1.37 (m, 2H, CH₂) ppm. ¹³C NMR (101 MHz, CD₂Cl₂): δ = 198.9, 144.5 (brd¹), 140.3 (brd¹), 136.1, 132.2, 129.8, 128.2, 123.5 (brd¹), 106.9 (brd¹), 50.5, 47.6, 38.4, 30.4, 26.4, 23.5 ppm. MS (ESI) *m/z*: 402 [M + H]⁺ (100%). HRMS (ESI) *m/z*: calcd for C₁₈H₂₀BrN₅O: 402.0930; found: 402.0914.

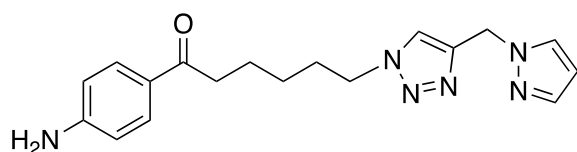
¹ These signals consistently appeared broadened, thus resulting in loss of intensity.



5b: A round bottom flask was charged with compound **4b** (850 mg, 2.6 mmol), *tert.*-butanol (70 mL) and water (80 mL). While stirring, propargylpyrazole (365 mg, 3.4 mmol), copper sulfate · 5 H₂O

(57 mg, 0.2 mmol) and sodium ascorbate (45 mg, 0.2 mmol) of were added. The mixture was stirred for 16 h, after which a thick off-white precipitation formed. The mixture was diluted with water (100 mL) and the solids were collected using a Büchner funnel and washed with several portions of water. Drying under high vacuum afforded 1.103 g of an off-white solid. Extraction of the remaining solution of the reaction mixture and washings 3 × with 100 mL ethyl acetate, drying the combined organic phases over MgSO₄, solvent evaporation and drying under high vacuum gave another 110 mg of the title compound. Combined yield: 1.213 g (0.25 mmol, 90 %). ¹H NMR (400 MHz, CD₂Cl₂): δ = 7.82 (dm, ³J_{H-H} = 8.7 Hz, 2H, Ar-*H*), 7.61 (dm, ³J_{H-H} = 8.7 Hz, 2H, Ar-*H*), 7.52 (m, 1H, PyT), 7.49 (m, 1H, PyT), 7.47 (m, 1H, PyT), 6.25 (t, ³J_{H-H} = 2.0 Hz, 1H, PyT), 5.40 (s, 2H, N₃CHC-CH₂), 4.29 (t, ³J_{H-H} = 7.2 Hz, 2H, CH₂PyT), 2.92 (t, ³J_{H-H} = 7.2 Hz, 2H, COCH₂), 1.86 (m, 2H, CH₂), 1.68 (m, 2H, CH₂), 1.36-1.28 (brd, 12H, CH₂) ppm. ¹³C NMR (101 MHz, CD₂Cl₂): δ = 199.6, 143.9 (brd¹), 139.9 (brd¹), 136.4, 132.2, 129.9, 128.1, 122.8 (brd¹), 106.3 (brd¹), 50.8, 47.8, 38.9, 30.6, 29.8, 29.76, 29.69, 29.6, 29.3, 26.8, 24.5 ppm. MS (ESI) m/z: 472 [M + H]⁺ (100%). HRMS (ESI) m/z: calcd for C₂₃H₃₁BrN₅O: 472.17120; found: 472.1701.

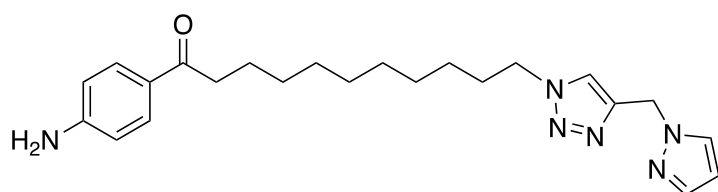
¹ These signals consistently appeared broadened, thus resulting in loss of intensity.



6a: Aryl bromide **5a** (180 mg, 0.45 mmol) was dissolved in ethanol/water (7:3; 9 mL). Copper iodide (85 mg, 0.45 mmol), sodium azide (117

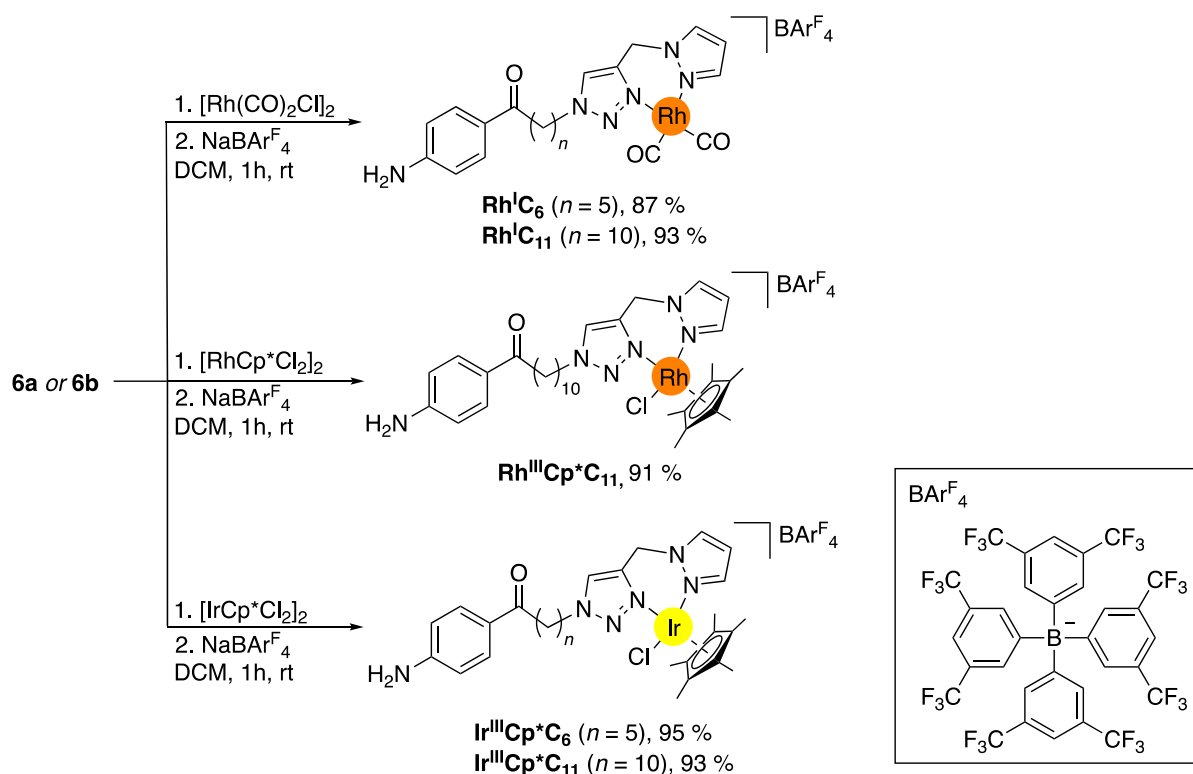
mg, 1.80 mmol) and sodium ascorbate (89 mg, 0.45 mmol) were added and the mixture was refluxed for 24 h. After cooling down to room temperature, water was added (80 mL) and the mixture was extracted with ethyl acetate (3 × 80 mL). The combined organic extracts were dried over magnesium sulfate, the solvents were removed using a rotary evaporator and the crude solids were dried under vacuum. The crude mixture was re-dissolved in little

dichloromethane (~ 2 mL) and applied on a short column prefilled with silica and wetted with ethyl acetate. Flash chromatography using ethyl acetate as eluent afforded a pure fraction, which was concentrated to dryness using a rotary evaporator, followed by drying under high vacuum. The product crystallized upon standing, giving an off-white crystalline solid (113 mg, 0.33 mmol, 74 %). ¹H NMR (400 MHz, CD₂Cl₂): δ = 7.75 (dm, ³J_{H-H} = 8.7 Hz, 2H, Ar-*H*), 7.52 (m, 2H, PyT), 7.47 (m, 1H, PyT), 6.64 (dm, ³J_{H-H} = 8.7 Hz, 2H, Ar-*H*), 6.25 (t, ³J_{H-H} = 2.1 Hz, 1H, PyT), 5.40 (s, 2H, N₃CHC-CH₂), 4.31 (t, ³J_{H-H} = 7.2 Hz, 2H, PyTCH₂), 3.41 (brd, 2H, NH₂), 2.84 (t, ³J_{H-H} = 7.2 Hz, 2H, COCH₂), 1.90 (m, 2H, CH₂), 1.70 (m, 2H, CH₂), 1.36 (m, 2H, CH₂) ppm. ¹³C NMR (101 MHz, CD₂Cl₂): δ = 198.1, 151.8, 143.7, 139.8, 130.6, 129.6, 127.7, 122.9, 113.9, 106.1, 50.6, 47.7, 37.8, 30.5, 26.5, 24.1 ppm. MS (ESI) m/z: 339 [M + H]⁺ (100%). HRMS (ESI) m/z: calcd for C₁₈H₂₃N₆O: 339.1933; 339.1920.



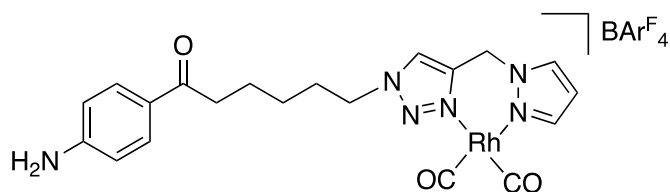
6b: The compound was prepared analogously to compound **6a**. Aryl bromide **5b** (200 mg, 0.42 mmol),

copper iodide (81 mg, 0.2 mmol), sodium azide (110 mg, 1.7 mmol) and sodium ascorbate (84 mg, 0.42 mmol) in EtOH/H₂O (7:3; 9 mL) gave the title compound as off-white crystalline solid (120 mg, 0.28 mmol, 67 %). ¹H NMR (400 MHz, CD₂Cl₂): δ = 7.77 (dm, ³J_{H-H} = 8.7 Hz, 2H, Ar-*H*), 7.52 (m, 1H, PyT), 7.51 (m, 1H, PyT), 7.47 (m, 1H, PyT), 6.64 (dm, ³J_{H-H} = 8.7 Hz, 2H, Ar-*H*), 6.25 (t, ³J_{H-H} = 2.0 Hz, 1H, PyT), 5.40 (s, 2H, N₃CHC-CH₂), 4.30 (concealed s, 2H, NH₂), 4.27 (t, ³J_{H-H} = 7.4 Hz, 2H, CH₂PyT), 2.83 (t, ³J_{H-H} = 7.2 Hz, 2H, COCH₂), 1.86 (m, 2H, CH₂), 1.66 (m, 2H, CH₂), 1.35-1.27 (brd, 12H, CH₂) ppm. ¹³C NMR (101 MHz, CD₂Cl₂): δ = 198.8, 151.7, 143.7, 139.8, 130.7, 129.6, 127.8, 122.8, 113.9, 106.1, 50.8, 47.7, 38.2, 30.5, 29.8, 29.73, 29.68, 29.65, 29.3, 26.8, 25.1 ppm. MS (ESI) m/z: 409 [M + H]⁺ (100%), HRMS (ESI) m/z: calcd for C₂₃H₃₃N₆O: 409.27160; found: 409.2701.



Syntheses of Metal Complexes

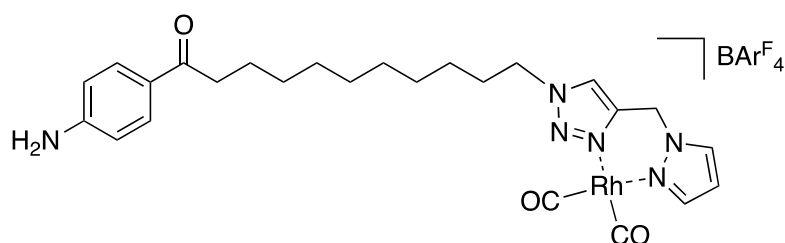
Scheme S1. Syntheses of complexes $\text{Rh}^{\text{I}}\text{C}_6$, $\text{Rh}^{\text{I}}\text{C}_{11}$, $\text{Rh}^{\text{III}}\text{Cp}^*\text{C}_{11}$, $\text{Ir}^{\text{III}}\text{Cp}^*\text{C}_6$, $\text{Ir}^{\text{III}}\text{Cp}^*\text{C}_{11}$ from the ligands **6a/b** and metal precursors.



Rh^IC₆: A Schlenk flask was charged with ligand **6a** (50 mg, 0.15 mmol) of and dichloromethane (6 mL). Di-μ-

chloro-tetracarbonyldirrhodium(I) (39 mg, 0.07 mmol) was added and the solution was stirred for 30 min at room temperature after which sodium tetrakis[(3,5-trifluoromethyl)phenyl]borate (NaBARF_4) (131 mg, 0.15 mmol) was added and the mixture was stirred for another 30 min. After the addition of NaBARF_4 , the solution turned cloudy. The crude mixture was filtered through a plug of celite (4×3 cm) using dichloromethane as eluent and concentrated to a volume of about 2 mL using rotary evaporator. Pentane (4 mL) was added to precipitate the product as red-brown oil. The slightly turbid solution was allowed to stand for 20 min, and the

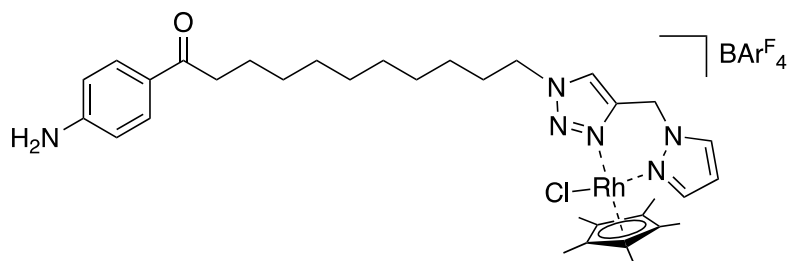
now colorless liquids were removed and discarded. The oil was re-dissolved in dichloromethane (1 mL) and the procedure was repeated, followed by drying under high vacuum, which afforded 174 mg (0.12 mmol, 87 %) of a red-brown solid. ^1H NMR (500 MHz, CD_2Cl_2): δ = 8.02 (m, 1H, PyT), 7.82 (m, 1H, PyT), 7.77-7.72 (brd, 11H, PyT/Ar-*H*/BAr $^{\text{F}_4}$), , 7.57 (m, 4H, BAr $^{\text{F}_4}$), 6.65 (dm, $^3J_{\text{H-H}}$ = 8.7 Hz, 2H, Ar-*H*), 6.53 (t, $^3J_{\text{H-H}}$ = 2.6 Hz, 1H, PyT), 5.43 (s, 2H, N $_3$ CHC- CH_2), 4.53 (t, $^3J_{\text{H-H}}$ = 7.1 Hz, 2H, CH_2 PyT), 4.22 (brd, 2H, NH $_2$), 2.90 (t, $^3J_{\text{H-H}}$ = 6.7 Hz, 2H, COCH $_2$), 2.00 (m, 2H, CH $_2$), 1.76 (m, 2H, CH $_2$), 1.39 (m, 2H, CH $_2$) ppm. ^{13}C NMR (126 MHz, CD_2Cl_2): δ = 198.1, 182.9 (d, $^1J_{\text{C-Rh}}$ = 68.8 Hz), 182.0 (d, $^1J_{\text{C-Rh}}$ = 70.2 Hz), 162.2 (q, $^1J_{\text{C-B}}$ = 49.8 Hz, ipso-C, BAr $^{\text{F}_4}$), 151.9, 147.1 (d, $^2J_{\text{C-Rh}}$ $\approx$ 2.1 Hz), 139.7, 135.2 (brd, m-CH, BAr $^{\text{F}_4}$), 135.1, 130.7, 129.3 (qq, $^2J_{\text{C-F}}$ = 31.5 Hz, $^3J_{\text{C-B}}$ = 2.9 Hz, C-CF $_3$, BAr $^{\text{F}_4}$), 128.3, 125.0 (q, $^1J_{\text{C-F}}$ = 272.5 Hz, CF $_3$, BAr $^{\text{F}_4}$), 123.9, 117.9 (m, p-CH, BAr $^{\text{F}_4}$), 114.0, 108.90 (d, $^3J_{\text{C-Rh}}$ $\approx$ 1.3 Hz), 52.9, 45.9, 37.3, 29.9, 26.0, 23.3 ppm. MS (ESI) m/z : 497 [M] $^+$ (100%). HRMS (ESI) m/z : calcd for $\text{C}_{20}\text{H}_{22}\text{N}_6\text{O}_3\text{Rh}$: 497.08084; found: 497.08047. Elemental analysis: calcd (%) for $\text{C}_{52}\text{H}_{34}\text{BF}_{24}\text{N}_6\text{O}_3\text{Rh}$: C (45.91); H (2.52); N (6.18); found: C (45.66); H (2.15); N (6.16). IR (ATR): 2107 (m), 2048 (m), 1661 (w), 1662 (w), 1598 (w), 1572 (w), 1522 (w), 1465 (w), 1439 (w), 1421 (w), 1353 (m), 1273 (s), 1114 (s), 1001 (w), 973 (w), 933 (w), 886 (m), 893 (m), 790 (w), 762 (w), 745 (w), 712 (m), 682 (m), 669 (m), 638 (w), 622 (w), 609 (w), 602 (w), 577 (w), 566 (w), 556 (w).



Rh $^{\text{I}}$ C $_{11}$: A Schlenk flask was charged with ligand **6b** (135 mg, 0.32 mmol) and dichloromethane (15 mL).

Reaction with di- μ -chloro-tetracarbonyldirhodium(I) (62 mg, 0.16 mmol) and NaBAr $^{\text{F}_4}$ (283 mg, 0.32 mmol), followed by work-up analogously to the above described preparation of **RhC $_6$**

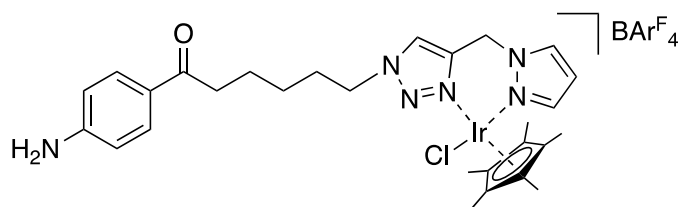
gave 425 mg (0.29 mmol, 93 %) of a red-brown solid. ^1H NMR (500 MHz, CD_2Cl_2): δ = 7.86 (m, 1H, PyT), 7.84 (m, 1H, PyT), 7.77 (dm, $^3J_{\text{H-H}}$ = 8.7 Hz, 2H, Ar-*H*), 7.75 (m, 1H, PyT), 7.72 (m, 8H, BAr^{F_4}), 7.55 (m, 4H, BAr^{F_4}), 6.65 (dm, $^3J_{\text{H-H}}$ = 8.7 Hz, 2H, Ar-*H*), 6.53 (t, $^3J_{\text{H-H}}$ = 2.6 Hz, 1H, PyT), 5.45 (s, 2H, $\text{N}_3\text{CHC-CH}_2$), 4.47 (t, $^3J_{\text{H-H}}$ = 7.2 Hz, 2H, CH_2PyT), 4.19 (brd, 2H, NH_2), 2.84 (t, $^3J_{\text{H-H}}$ = 7.3 Hz, 2H, COCH_2), 1.95 (m, 2H, CH_2), 1.65 (m, 2H, CH_2), 1.33-1.28 (brd, 12H, CH_2) ppm. ^{13}C NMR (101 MHz, CD_2Cl_2): δ = 199.3, 182.7 (d, $^1J_{\text{C-Rh}}$ $\approx$ 70.6 Hz), 182.0 (d, $^1J_{\text{C-Rh}}$ $\approx$ 70.9 Hz), 162.2 (q, $^1J_{\text{C-B}}$ = 49.8 Hz, ipso-C, BAr^{F_4}), 151.8, 147.1 (d, $^2J_{\text{C-Rh}}$ $\approx$ 2.1 Hz), 139.9, 135.3 (brd, m-CH, BAr^{F_4}), 135.2, 130.8, 129.3 (qq, $^2J_{\text{C-F}}$ = 31.5 Hz, $^3J_{\text{C-B}}$ = 2.9 Hz, C- CF_3 , BAr^{F_4}), 127.8, 125.0 (q, $^1J_{\text{C-F}}$ = 272.5 Hz, CF_3 , BAr^{F_4}), 124.0, 118.0 (m, p-CH, BAr^{F_4}), 114.0, 108.9 (d, $^3J_{\text{C-Rh}}$ $\approx$ 1.3 Hz), 53.4, 45.8, 38.3, 30.0, 29.71, 29.68, 29.5, 29.4, 29.0, 26.4, 25.2 ppm. MS (ESI) m/z : 567 $[\text{M}]^+$ (100%). HRMS (ESI) m/z : calcd for $\text{C}_{25}\text{H}_{32}\text{N}_6\text{O}_3\text{Rh}$: 567.15909; found: 567.15857. Elemental analysis: calcd (%) for $\text{C}_{57}\text{H}_{44}\text{BF}_{24}\text{N}_6\text{O}_3\text{Rh}$: C (47.85); H (3.10); N (5.87); found: C (48.19); H (2.72); N (6.12). IR (ATR): 2934 (w), 2860 (w), 2107 (m), 2048 (m), 1784 (w), 1660 (w), 1622 (w), 1598 (w), 1522 (w), 1465 (w), 1439 (w), 1421 (w), 1553 (s), 1272 (s), 1158 (m), 1113 (s), 1094 (s), 1001 (w), 945 (w), 933 (w), 886 (m), 838 (m), 796 (w), 761 (w), 745 (w), 712 (m), 681 (m), 669 (m), 638 (w), 621 (w), 609 (w), 602 (w), 586 (w), 577 (w), 566 (w), 556 (w).



Rh^{III}Cp*Cl₁: A Schlenk flask was charged with ligand **6b** (50 mg, 0.12 mmol) and dichloromethane (6 mL).

Pentamethylcyclopentadienylrhodium(III)chloride dimer (38 mg, 0.06 mmol) was added and the mixture was stirred for 30 min at room temperature. This was followed by addition of $\text{NaBAr}^{\text{F}_4}$ (108 mg, 0.12 mmol). Stirring was continued for another 30 min before work-up was

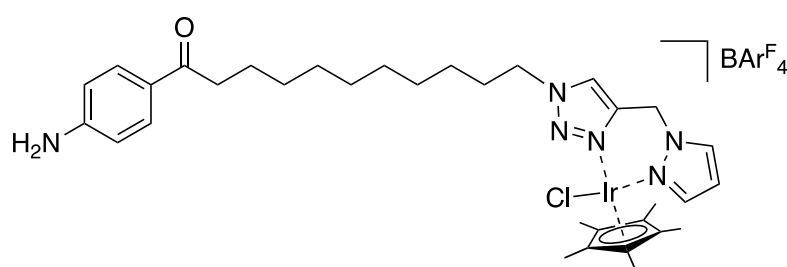
carried out analogously to the above described procedure, giving the title compound (172 mg, 0.11 mmol, 91 %) as orange solid. ^1H NMR (400 MHz, CD_2Cl_2): δ = 7.80 (m, 2H, PyT), 7.76 (dm, $^3J_{\text{H-H}}$ = 8.7 Hz, 2H, Ar-*H*), 7.72 (brm, 9H, $\text{BAr}^{\text{F}}_4/\text{PyT}$), 7.56 (m, 4H, BAr^{F}), 6.65 (t, $^3J_{\text{H-H}}$ = 2.6 Hz, 1H, Ar-*H*), 6.56 (t, $^3J_{\text{H-H}}$ = 2.5 Hz, 1H, PyT), 5.57 (d, $^2J_{\text{H-H}}$ = 15.7 Hz, $\text{N}_3\text{CHC-CH}_2$), 5.07 (d, $^2J_{\text{H-H}}$ = 15.7 Hz, $\text{N}_3\text{CHC-CH}_2$), 4.49 (m, 2H, CH_2PyT), 4.17 (brd, 2H, NH_2), 2.84 (t, $^3J_{\text{H-H}}$ = 7.3 Hz, 2H, COCH_2), 1.96 (m, 2H, CH_2), 1.68 (s, 15H, Cp^* , CH_3), 1.66 (m, 2H, CH_2) 1.33-1.26 (brm, 12H) ppm. ^{13}C NMR (101 MHz, CD_2Cl_2): δ = 198.9, 162.2 (q, $^1J_{\text{C-B}}$ = 49.8 Hz, ipso-C, BAr^{F}_4), 151.7, 145.4, 139.6, 135.2 (brd, m-CH, BAr^{F}_4), 134.7, 130.7, 129.3 (qq, $^2J_{\text{C-F}}$ = 31.5 Hz, $^3J_{\text{C-B}}$ = 2.9 Hz, C- CF_3 , BAr^{F}_4), 125.0 (q, $^1J_{\text{C-F}}$ = 272.4 Hz, CF_3 , BAr^{F}_4), 124.1, 117.9 (m, p-CH, BAr^{F}_4), 114.0, 109.1, 97.5 (d, $^1J_{\text{C-Rh}}$ = 8.3 Hz), 45.4, 38.2, 30.2, 29.70, 29.66, 29.5, 29.0, 26.5, 25.0, 9.4 ppm. MS (ESI) m/z : 681 $[\text{M}]^+$ (100%), 323 (94 %). HRMS (ESI) m/z : calcd for $\text{C}_{33}\text{H}_{47}\text{ClN}_6\text{ORh}$: 681.25494, found: 681.25440. Elemental analysis: calcd (%) for $\text{C}_{65}\text{H}_{59}\text{BCl F}_{24}\text{N}_6\text{ORh}$: C (50.52), H (3.85), N (5.44); found: C (50.79), H (3.79), N (5.33). IR (ATR): 3341 (w), 2932 (w), 2859 (w), 2361 (w), 2340 (w), 1659 (w), 1623 (w), 1597 (w), 1569 (w), 1519 (w), 1487 (w), 1463 (w), 1438 (w), 1417 (w), 1353 (m), 1323 (w), 1273 (s), 1159 (m), 1116 (s), 1072 (m), 1023 (w), 993 (w), 945 (w), 932 (w), 866 (m), 839 (m), 792 (w), 758 (w), 745 (w), 712 (m), 682 (m), 669 (m), 640 (w), 621 (w), 603 (w), 580 (w), 569 (w).



$\text{Ir}^{\text{III}}\text{Cp}^*\text{C}_6$: A Schlenk flask was charged with ligand **6a** (79 mg, 0.23 mmol) and dichloromethane (10 mL).

Pentamethylcyclopentadienyliridium(III)chloride dimer (93 mg, 0.12 mmol) was added and the mixture was stirred for 30 min at room temperature. This was followed by addition of $\text{NaBAr}^{\text{F}}_4$ (207 mg, 0.23 mmol). Stirring was continued for another 30 min before work-up was carried out analogously to the above described procedure for the rhodium complexes, giving

the title compound as light yellow solid (350 mg, 0.22 mmol, 95 %). ^1H NMR (400 MHz, CD_2Cl_2): δ = 7.93 (m, 1H, PyT), 7.78-7.73 (brm, 11H, BAr^{F_4} /PyT/Ar-*H*) 7.69 (m, 1H, PyT), 7.59 (m, 4H, BAr^{F_4}), 6.65 (dm, $^3J_{\text{H-H}}$ = 8.8 Hz, 2H, Ar-*H*), 6.55 (t, $^3J_{\text{H-H}}$ = 2.6 Hz, 1H, PyT-*H*), 5.54 (d, $^2J_{\text{H-H}}$ = 15.7 Hz, $\text{N}_3\text{CHC-CH}_2$), 4.93 (d, $^2J_{\text{H-H}}$ = 15.7 Hz, $\text{N}_3\text{CHC-CH}_2$), 4.46 (m, 2H, CH_2PyT), 4.22 (brd, 2H, NH_2), 2.89 (apparent t, $^3J_{\text{H-H}}$ $\approx$ 7 Hz, 2H, COCH), 2.00 (m, 2H, CH_2), 1.75 (m, 2H, CH_2), 1.62 (s, 15H, Cp^* , CH_3), 1.39 (m, 2H, CH_2) ppm. ^{13}C NMR (101 MHz, CD_2Cl_2): δ = 198.0, 162.2 (q, $^1J_{\text{C-B}}$ = 49.8 Hz, ipso-C, BAr^{F_4}), 152.0, 145.4, 138.7, 135.2 (brd, BAr^{F_4}), 134.3, 130.7, 129.3 (qq, $^2J_{\text{C-F}}$ = 31.5 Hz, $^3J_{\text{C-B}}$ = 2.8 Hz, C- CF_3 , BAr^{F_4}), 127.6, 125.0 (q, $^1J_{\text{C-F}}$ = 272.4 Hz, CF_3 , BAr^{F_4}), 124.4, 117.9 (m, p-CH, BAr^{F_4}), 114.0, 109.4, 89.4, 52.8, 45.7, 37.4, 30.1, 26.1, 23.4, 9.2 ppm. MS (ESI) m/z : 701 $[\text{M}]^+$ (100%), 333 (65 %). HRMS (ESI) m/z : calcd for $\text{C}_{28}\text{H}_{37}\text{ClIrN}_6\text{O}$: 701.23467, found: 701.23244. Elemental analysis: calcd (%) for $\text{C}_{60}\text{H}_{49}\text{Cl BF}_{24}\text{IrN}_6\text{O}$: C (46.06), H (3.16), N (5.37); found: C (45.80), H (2.99), N (5.37). IR (ATR): 3351 (w), 2341 (w), 2113 (w), 1829 (w), 1661 (w), 1622 (w), 1598 (w), 1570 (w), 1456 (w), 1436 (w), 1419 (w), 1354 (m), 1323 (w), 1274 (s), 1158 (m), 1116 (s), 1030 (w), 997 (w), 946 (w), 930 (w), 886 (m), 839 (m), 974 (w), 757 (w), 745 (w), 712 (m), 682 (m), 669 (m), 639 (w), 622 (w), 602 (w), 587 (w), 577 (w), 566 (w), 556 (w).



$\text{Ir}^{\text{III}}\text{Cp}^*\text{C}_{11}$: A Schlenk flask was charged with ligand **6b** (84 mg, 0.21 mmol) and dichloromethane (10 mL).

Pentamethylcyclopentadienyliridium(III)chloride dimer (82 mg, 0.10 mmol) was added and the mixture was stirred for 30 min at room temperature. This was followed by addition of $\text{NaBAr}^{\text{F}_4}$ (182 mg, 0.21 mmol). Stirring was continued for another 30 min before work-up was carried out analogously to the above described procedure, giving the title compound (311 mg,

0.19 mmol, 93 %) of as light yellow solid. ^1H NMR (400 MHz, CD_2Cl_2): δ = 7.82 (m, 1H, PyT), 7.79 - 7.74 (brm, 11H, $\text{BAr}^{\text{F}_4}/\text{PyT}/\text{Ar-H}$), 7.68 (m, 2H, PyT), 7.59 (m, 2H, BAr^{F_4}), 6.65 (dm, $^3J_{\text{H-H}} = 8.8$ Hz, 2H, Ar-H), 6.54 (t, $^3J_{\text{H-H}} = 2.6$ Hz, 1H, PyT), 5.56 (d, $^2J_{\text{H-H}} = 15.7$ Hz, $\text{N}_3\text{CHC-CH}_2$), 4.93 (d, $^2J_{\text{H-H}} = 15.7$ Hz, $\text{N}_3\text{CHC-CH}_2$), 4.51 (m, 2H, CH_2PyT), 4.21 (brd, 2H, NH_2), 2.86 (t, $^3J_{\text{H-H}} = 7.3$ Hz, 2H, COCH_2), 1.95 (m, 2H, CH_2), 1.67 (m, 2H, CH_2), 1.63 (s, 15H, Cp^* , CH_3), 1.34-1.30 (m, 12H, CH_2) ppm. ^{13}C NMR (101 MHz, CD_2Cl_2): δ = 199.1 (C^4), 162.2 (q, $^1J_{\text{C-B}} = 49.8$ Hz, ipso-C, BAr^{F_4}), 151.8, 145.4, 138.8, 135.2 (brd, m-CH, BAr^{F_4}), 134.3, 130.7, 129.3 (qq, $^2J_{\text{C-F}} = 31.5$ Hz, $^3J_{\text{C-B}} = 2.9$ Hz, C- CF_3 , BAr^{F_4}), 125.0 (q, $^1J_{\text{C-F}} = 272.5$ Hz, CF_3 , BAr^{F_4}), 127.8, 124.0, 118.0, 117.9, 117.9 (m, p-CH, BAr^{F_4}), 114.0, 109.4, 89.5, 53.2, 45.7, 38.3, 30.2, 29.70, 29.66, 29.5, 29.0, 26.5, 25.1, 9.1 ppm. MS (ESI) m/z: 771 $[\text{M}]^+$ (100%), 386 (90 %). HRMS (ESI) m/z: calcd for $\text{C}_{33}\text{H}_{47}\text{ClIrN}_6\text{O}$: 771.31293, found: 771.31174. Elemental analysis: calcd (%) for $\text{C}_{65}\text{H}_{59}\text{BClF}_{24}\text{IrN}_6\text{O}$: C (47.76), H (3.64), N (5.14); found: C (47.55), H (3.74), N (5.03). IR (ATR): 3351 (w), 2933 (w), 2860 (w), 2340 (w), 2340 (w), 2113 (w), 1660 (w), 1622 (w), 1598 (w), 1570 (w), 1520 (w), 1457 (w), 1436 (w), 1419 (w), 1354 (m), 1273 (s), 1157 (m), 1117 (s), 1030 (w), 997 (w), 944 (w), 933 (w), 866 (m), 839 (m), 794 (w), 757 (w), 745 (w), 712 (m), 682 (m), 669 (m), 639 (w), 622 (w), 602 (w), 587 (w), 577 (w), 566 (w), 555 (w).

2. X-ray Single Crystal Diffraction

Crystals suitable for X-ray single crystal diffraction were obtained by crystallization from pentane (**3a**), and slow diffusion of pentane into a solution in dichloromethane (**6a** and **6b**).

Crystals were selected using a microscope and mounted on the diffractometer. Details from the data collection, structure solution and refinement are reported in Table S1. Molecular graphics: OLEX2.⁶

Table S1. Crystal data and structure refinement for compound **3a**, **6a** and **6b**.

Identification code	3a	6a	6b
CCDC-number	1961819	1961820	1961821
Empirical formula	C ₁₂ H ₁₄ Br ₂ O	C ₁₈ H ₂₂ N ₆ O	C ₂₃ H ₃₂ N ₆ O
Formula weight	334.05	338.41	408.54
Temperature/K	149.97	149.95	99.57
Crystal system	monoclinic	monoclinic	monoclinic
Space group	P2 ₁ /c	P2 ₁ /c	P2 ₁ /n
a/Å	15.7970(11)	14.6540(14)	5.4266(5)
b/Å	10.6695(7)	7.0476(7)	52.494(5)
c/Å	7.5330(5)	17.0806(19)	8.0320(8)
α /°	90	90	90
β /°	91.679(3)	91.653(4)	107.835(3)
γ /°	90	90	90
Volume/Å ³	1269.11(15)	1763.3(3)	2178.1(4)
Z	4	4	4
ρ_{calc} /g/cm ³	1.748	1.275	1.246
μ /mm ⁻¹	6.360	0.084	0.080
F(000)	656.0	720.0	880.0
Crystal size/mm ³	0.334 × 0.218 × 0.02	0.19 × 0.154 × 0.03	0.303 × 0.274 × 0.088
Radiation	MoK α (λ = 0.71073)	MoK α (λ = 0.71073)	MoK α (λ = 0.71073)
2 θ range for data collection/°	4.608 to 55.448	4.772 to 55.236	4.656 to 55.02
Index ranges	-20 ≤ h ≤ 20 -13 ≤ k ≤ 13 -9 ≤ l ≤ 9	-19 ≤ h ≤ 19 -9 ≤ k ≤ 9 -22 ≤ l ≤ 22	-7 ≤ h ≤ 7 -68 ≤ k ≤ 68 -10 ≤ l ≤ 10
Reflections collected	40852	52028	110638
Independent reflections	2947 [R _{int} = 0.0769, R _{sigma} = 0.0344]	4072 [R _{int} = 0.1027, R _{sigma} = 0.0425]	4987 [R _{int} = 0.2120, R _{sigma} = 0.0524]
Data/restraints/ parameters	2947/0/136	4072/0/234	4987/0/279
Goodness-of-fit on F ²	1.021	1.053	1.052
Final R indexes [I ≥ 2 σ (I)]	R ₁ = 0.0338 wR ₂ = 0.0772	R ₁ = 0.0462 wR ₂ = 0.0985	R ₁ = 0.0425 wR ₂ = 0.1063
Final R indexes [all data]	R ₁ = 0.0495 wR ₂ = 0.0855	R ₁ = 0.0903 wR ₂ = 0.1202	R ₁ = 0.0511 wR ₂ = 0.1091
Largest diff. peak/hole / e Å ⁻³	1.03/-1.19	0.16/-0.23	0.27/-0.32

The molecular structure of dibromide **3a** is shown in Figure S1.

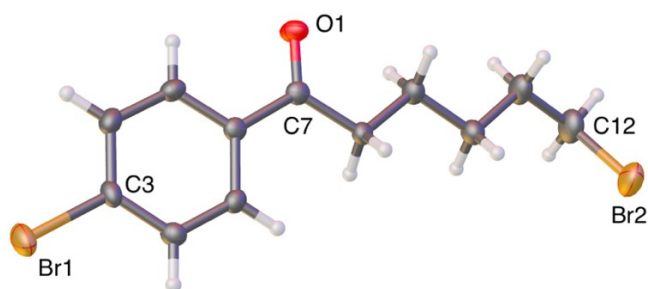


Figure S1. Molecular structure as determined by X-ray single crystal diffraction for 1-bromo-4-(6-bromoundecanoyl)benzene (**3a**).

The packing of the molecules in the crystal of **6a** shows hydrogen bonds between amine H and O (2.0 Å) and in-between H and N of the pyrazole-triazole (PyT) groups (Figure S2). The packing of compound **6b** is featuring a network of weak H-bonds involving PyT groups and amines with molecules aligned pair wise head to tail (Figure S3). Overall, the packing regarding **6b** is alkyl dominated with the long alkyl chains aligned straight, and the PyT head-group pointing along the imaginary axis through the alkyl chain. Regarding **6a**, PyT head-groups are also arranged pair-wise, however head-to-head.

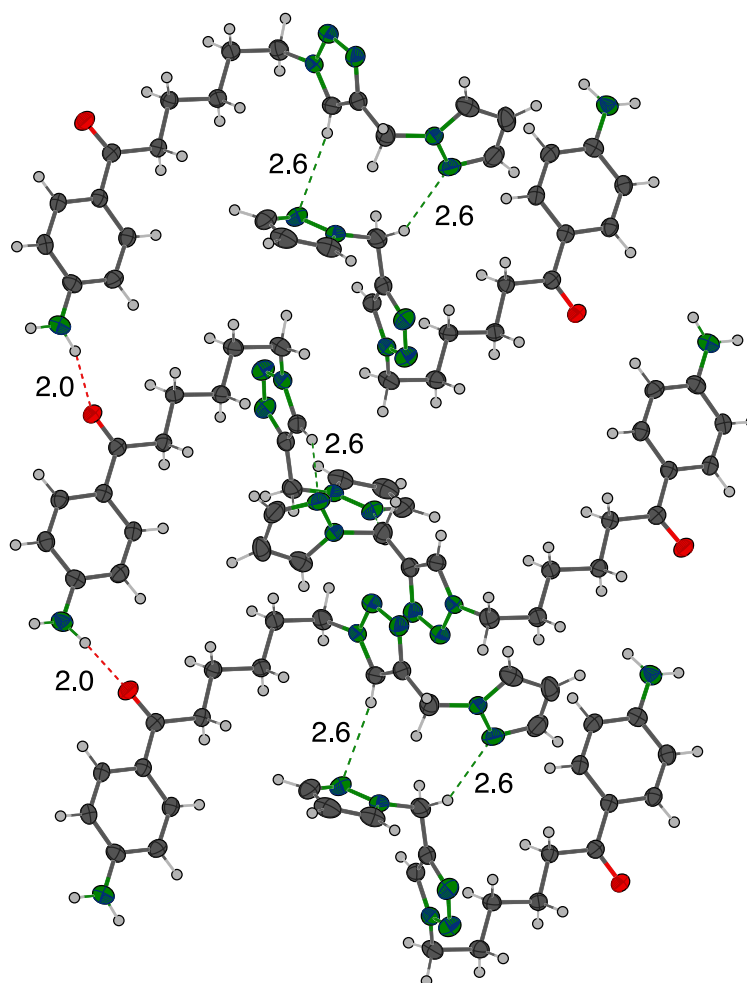


Figure S2. Crystal packing for compound **6a**. View along crystallographic b-axis. The lengths of selected hydrogen bonds are reported in Å.

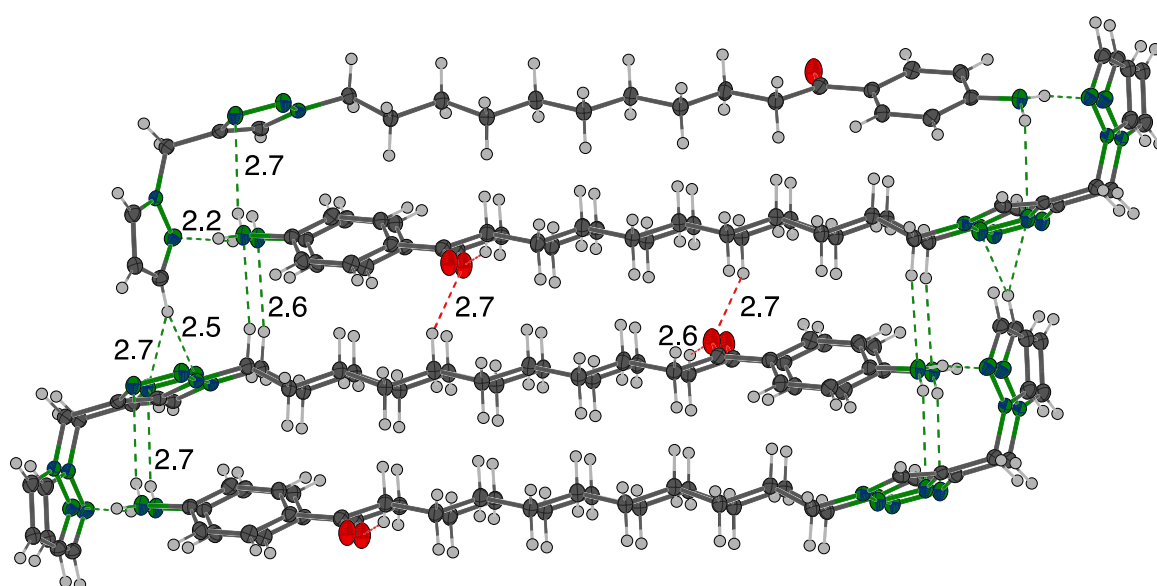


Figure S3. Crystal packing for compound **6b**. View along crystallographic a-axis. The lengths of selected hydrogen bonds are reported in Å.

3. Surface Immobilization and Characterization

Optimized Procedure for Catalyst Immobilization on Carbon Black

A Schlenk flask was charged with the Rh- or Ir-based metal complex (0.035 mmol) and degassed nitromethane (10 mL). The solution was cooled to 0 °C before degassed hydrochloric acid (0.1 M, 0.5 mL) was added slowly under stirring, followed by sodium nitrite (4.8 mg, 0.07 mmol). Stirring was continued for another 10 min. Carbon black (50 mg) was added and the mixture was stirred for 16 h at room temperature. Subsequently, the black mixture was transferred into glass centrifuge tubes and centrifuged for 20 min at 4000 rpm. The liquids were removed and discarded, and the remaining black solids were treated with washing and centrifugation steps: 8 mL methanol, 8 mL deionized water (MilliQ, 18 Ω), and finally 8 mL methanol. After each washing step, the samples were centrifuged for 20 min at 4000 rpm and the liquids were removed and discarded. The obtained black solids were allowed to dry overnight at room temperature on air and finally dried under vacuum ($\sim 2.0 \times 10^{-1}$ mbar), affording the hybrid catalysts as black powders. The isolated yields of the hybrid catalysts based on the Rh^I precursors were slightly higher compared to those of the Rh^{III}/Ir^{III} precursors (Table S2).

Table S2. Isolated yields of hybrid catalysts.

Entry	Precursor	m [mg]	n [mol]	CB [mg]	Hybrid catalyst	Yield [mg]
1 ^a	Rh ^I C ₁₁	50	0.035	50	Rh ^{III} C ₁₁ CB	49
2	Rh ^I C ₁₁	100	0.070	100	Rh ^{III} C ₁₁ CB	106
3 ^a	Rh ^I C ₆	48	0.035	50	Rh ^{III} C ₆ CB	43
4	Rh ^I C ₆	96	0.070	100	Rh ^{III} C ₆ CB	111
5 ^a	Rh ^I C ₀	44	0.035	50	Rh ^{III} C ₀ CB	51
6	Rh ^I C ₀	88	0.070	100	Rh ^{III} C ₀ CB	116
7 ^a	Rh ^{III} Cp [*] C ₁₁	27	0.017	25	Rh ^{III} Cp [*] C ₁₁ CB	20
8	Rh ^{III} Cp [*] C ₁₁	108	0.070	100	Rh ^{III} Cp [*] C ₁₁ CB	81
9 ^a	Ir ^{III} Cp [*] C ₁₁	57	0.035	50	Ir ^{III} Cp [*] C ₁₁ CB	36
10 ^a	Ir ^{III} Cp [*] C ₆	55	0.035	50	Ir ^{III} Cp [*] C ₆ CB	36
11 ^a	Ir ^{III} Cp [*] C ₀	51	0.035	50	Ir ^{III} Cp [*] C ₀ CB	34

^a These batches were used for characterization

3.1 Characterization by Energy Dispersive X-ray Spectroscopy (EDX)

Scanning electron microscope (SEM) images and EDX spectra were acquired using a Zeiss EVO MA15 Scanning electron microscope equipped with Oxford Instruments Aztec Synergy EDS/EBSD). The hybrid catalyst (~ 1-2 mg), in form of a black powder, was placed onto a round and flat sample holder equipped with a double sided adhesive tape. The sample was pressed onto the tape using a spatula and non-attached powder was carefully removed using a gentle stream of air. Sample holders were mounted on a revolving holder disc, which was placed in the chamber and evacuated. Calibration was carried out using a copper standard and the sample to detector distance was 12 mm using an energy of 12.5 keV.

Initially, the sample was screened by SEM and flat areas were identified. For determination of elemental compositions, typically five to six areas were selected and scanned. This was followed by EDX area mapping. Analyses were performed with the Oxford Instruments AZtec software package. Data deconvolution and background removal was performed with AZtec TruMap. Determined metal concentrations by EDX and XPS are reported in Table S3.

Obtained EDX spectra are shown for **Rh^{III}C₁₁CB** and **Ir^{III}Cp^{*}C₁₁CB** in Figure S4 and S9, respectively. EDX maps of all samples are shown in Figures S5-S11, and determined elemental compositions are reported in Tables S4-S10.

Table S3. Determined metal concentrations by EDX and XPS.

Hybrid catalyst	at% Rh EDX	wt% Rh EDX	at% Ir EDX	wt% Ir EDX	at% Rh XPS	at% Ir XPS
Rh^{III}C₁₁CB	0.49	3.93	-	-	0.96	-
Rh^{III}C₆CB	0.52	4.14	-	-	0.97	-
Rh^{III}C₀CB	0.82	6.37	-	-	1.22	-
Rh^{III}Cp[*]C₁₁CB	0.25	2.00	-	-	0.63	-
Ir^{III}Cp[*]C₁₁CB	-	-	0.13	2.08	-	0.24
Ir^{III}Cp[*]C₆CB	-	-	0.11	1.68	-	0.28
Ir^{III}Cp[*]C₀CB	-	-	0.08	1.20	-	0.22

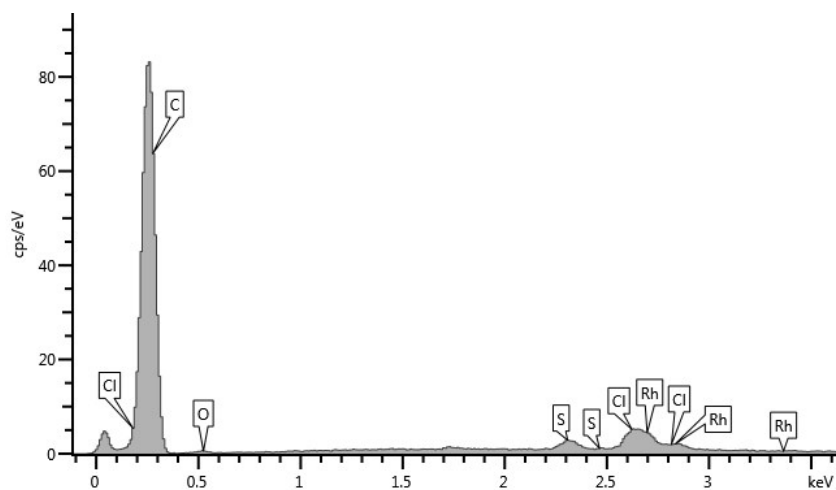


Figure S4. EDX spectrum of $\text{Rh}^{\text{III}}\text{C}_{11}\text{CB}$.

Table S4. Determined elemental composition of $\text{Rh}^{\text{III}}\text{C}_{11}\text{CB}$ in wt% by EDX.

Element	Area 1	Area 2	Area 3	Area 4	Area 5	Area 6	Average
C	88.54	89.07	87.99	88.84	89.71	89.03	88.86
O	3.96	3.55	4.42	4.03	3.55	3.99	3.92
S	0.95	1.06	1.06	0.92	0.98	1.02	1
Cl	2.35	2.35	2.42	2.27	2.22	2.17	2.3
Rh	4.21	3.96	4.11	3.94	3.54	3.79	3.93
Total	100	100	100	100	100	100	100

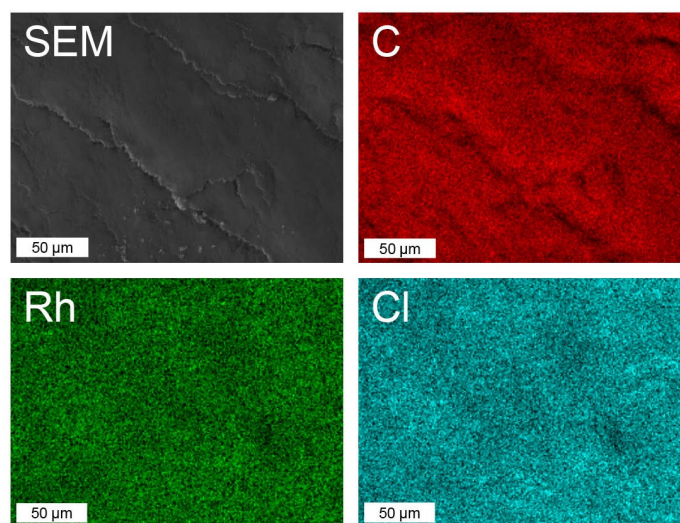
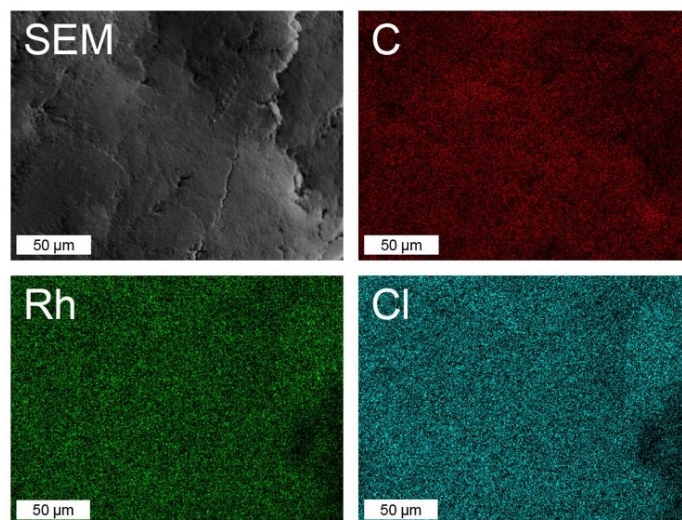


Figure S5. (a) SEM image of $\text{Rh}^{\text{III}}\text{C}_6\text{CB}$ and EDX maps for carbon (C), rhodium (Rh) and chlorine (Cl).

Table S5. Determined elemental composition of $\text{Rh}^{\text{III}}\text{C}_6\text{CB}$ in wt% by EDX.

Element	Area 1	Area 2	Area 3	Area 4	Area 5	Average
C	86.38	86.12	85.91	85.97	86.21	86.12
O	6.46	6.54	6.54	6.78	6.12	6.49
S	1	1.03	1.05	0.99	1.05	1.02
Cl	2.15	2.24	2.29	2.15	2.26	2.22
Rh	4.01	4.07	4.22	4.12	4.27	4.14
Total	100	100	100	100	100	100

**Figure S6.** (a) SEM image of $\text{Rh}^{\text{III}}\text{C}_0\text{CB}$ and EDX maps for carbon (C), rhodium (Rh) and chlorine (Cl).**Table S6.** Determined elemental composition of $\text{Rh}^{\text{III}}\text{C}_0\text{CB}$ in wt% by EDX.

Element	Area 1	Area 2	Area 3	Area 4	Area 5	Area 6	Average
C	85.87	85.67	85.34	86.58	85.73	85.75	85.82
O	3.99	2.97	3.85	2.14	3.3	3.14	3.23
S	0.89	1.05	1.04	1.09	0.97	0.98	1.00
Cl	3.41	3.69	3.39	3.79	3.44	3.71	3.57
Rh	5.84	6.63	6.38	6.39	6.56	6.41	6.37
Total	100	100	100	100	100	100	100

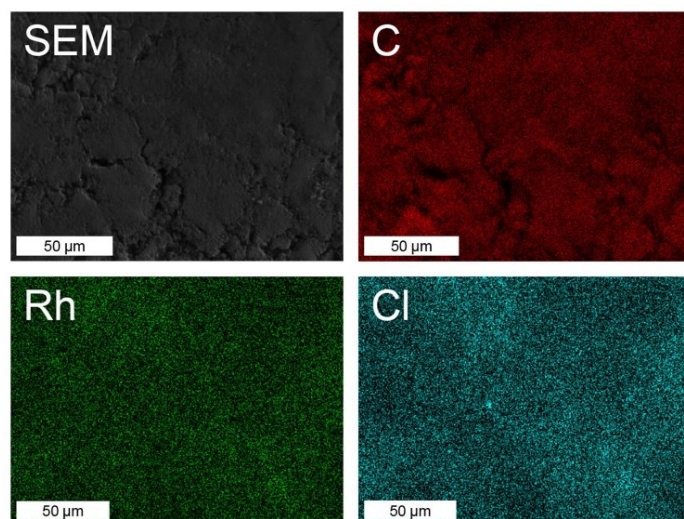


Figure S7. (a) SEM image of $\text{Rh}^{\text{III}}\text{Cp}^*\text{C}_{11}\text{CB}$ and EDX maps for carbon (C), rhodium (Rh) and chlorine (Cl).

Table S7. Determined elemental composition of $\text{Rh}^{\text{III}}\text{Cp}^*\text{C}_{11}\text{CB}$ in wt% by EDX.

Element	Area 1	Area 2	Area 3	Area 4	Area 5	Area 6
C	88.41	87.68	87.93	87.75	87.86	87.93
O	7.57	7.74	7.84	7.44	8.29	7.78
S	1.20	1.21	1.18	1.18	1.03	1.16
Cl	1.05	1.26	1.02	1.39	0.99	1.14
Rh	1.77	2.11	2.02	2.24	1.84	2.00
Total	100	100	100	100	100	100

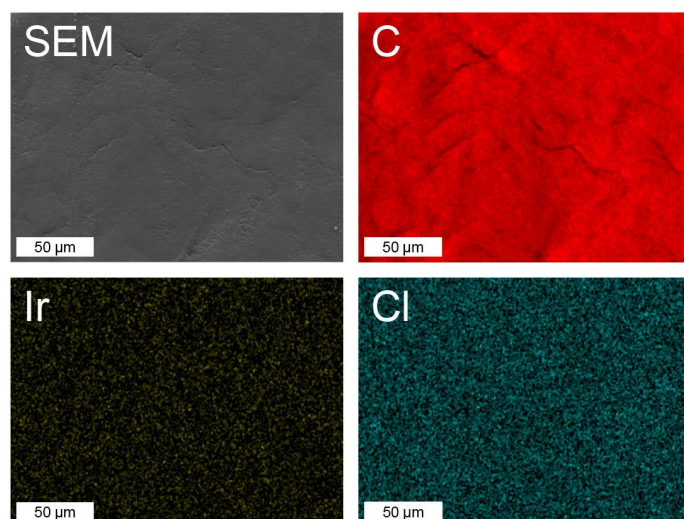


Figure S8. SEM image of $\text{Ir}^{\text{III}}\text{Cp}^*\text{C}_{11}\text{CB}$ and EDX maps for carbon (C), iridium (Ir) and chlorine (Cl).

Table S8. Determined elemental composition of $\text{Ir}^{\text{III}}\text{Cp}^*\text{C}_{11}\text{CB}$ in wt% by EDX.

Element	Area 1	Area 2	Area 3	Area 4	Area 5	Average
C	95.22	94.9	95.42	94.83	94.82	95.04
O	1.59	1.39	1.19	1.33	1.6	1.42
S	0.76	0.72	0.83	0.66	0.68	0.73
Cl	0.74	0.76	0.71	0.73	0.71	0.73
Ir	1.70	2.24	1.85	2.44	2.18	2.08
Total	100	100	100	100	100	100

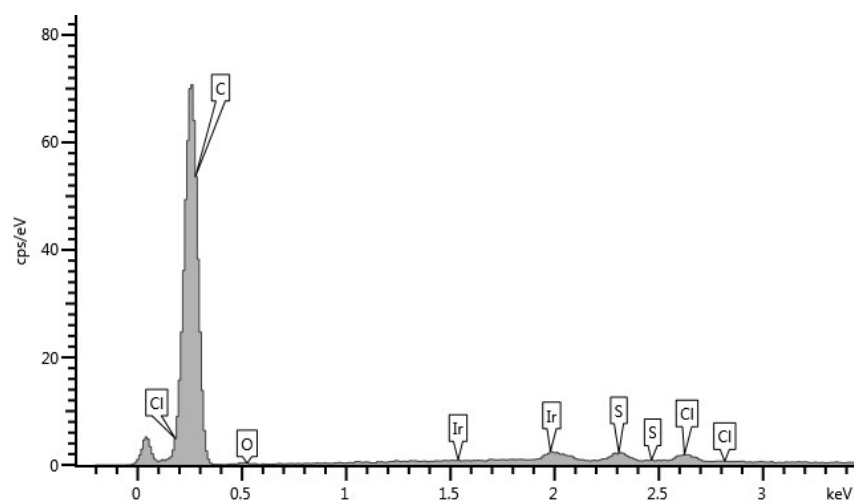


Figure S9. EDX spectrum of $\text{Ir}^{\text{III}}\text{Cp}^*\text{C}_{11}\text{CB}$.

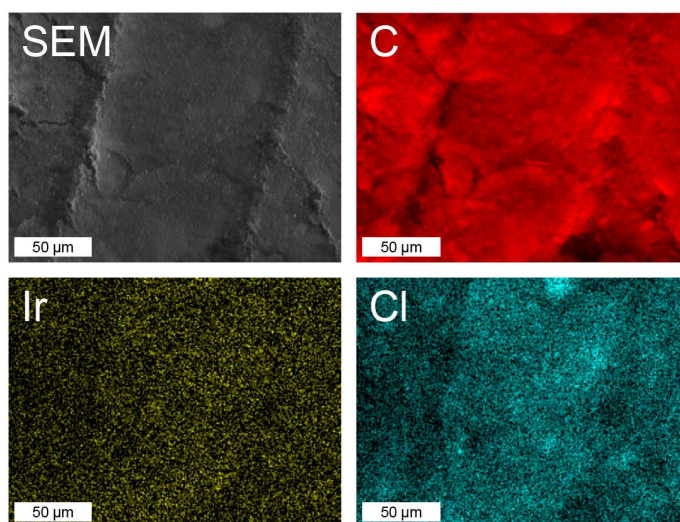


Figure S10. SEM image of $\text{Ir}^{\text{III}}\text{Cp}^*\text{C}_6\text{CB}$ and EDX maps for carbon (C), iridium (Ir) and chlorine (Cl).

Table S9. Determined elemental composition of $\text{Ir}^{\text{III}}\text{Cp}^*\text{C}_6\text{CB}$ in wt% by EDX.

Element	Area 1	Area 2	Area 3	Area 4	Area 5	Average
C	93.81	93.81	93.64	93.87	93.55	93.74
O	3.20	3.20	3.19	3.02	3.67	3.26
S	0.65	0.65	0.68	0.68	0.64	0.66
Cl	0.67	0.67	0.68	0.71	0.59	0.66
Ir	1.67	1.67	1.8	1.71	1.55	1.68
Total	100	100	100	100	100	100

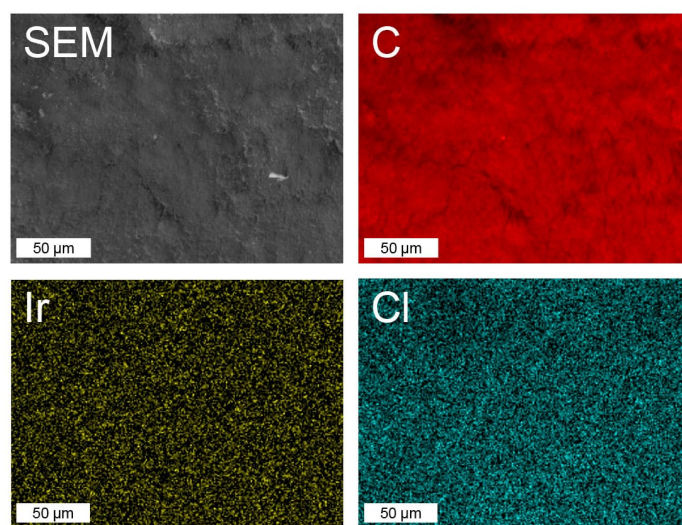


Figure S11. SEM image of $\text{Ir}^{\text{III}}\text{Cp}^*\text{C}_0\text{CB}$ and EDX maps for carbon (C), iridium (Ir) and chlorine (Cl).

Table S10. Determined elemental composition of $\text{Ir}^{\text{III}}\text{Cp}^*\text{C}_0\text{CB}$ in wt% by EDX.

Element	Area 1	Area 2	Area 3	Area 4	Area 5	Average
C	94.75	94.21	94.45	94.71	94.52	94.53
O	2.91	3.50	3.38	2.99	3.23	3.20
S	0.64	0.64	0.63	0.7	0.63	0.65
Cl	0.44	0.44	0.39	0.44	0.42	0.43
Ir	1.27	1.21	1.16	1.16	1.2	1.20
Total	100	100	100	100	100	100

Control Sample

We prepared a blank CB control sample by applying the same procedure as described above for the catalyst immobilization on CB, but under absence of a metal complex. EDX analysis revealed presence of only trace amounts of Cl (Table S11), which suggests that the bulk of the Cl found in the Rh^I hybrid catalysts is either bound to the ligand or attached to the metal center as counterion.

Table S11. Determined elemental composition of a control sample of CB in wt% by EDX.

Element	Area 1	Area 2	Area 3	Area 4	Area 5	Average
C	97.2	97.16	97.34	97.14	97.24	97.22
O	2.51	2.51	2.34	2.55	2.45	2.47
S	0.27	0.29	0.28	0.27	0.27	0.28
Cl	0.02	0.03	0.03	0.03	0.04	0.03
Total	100	100	100	100	100	100

3.2 Characterization by X-ray Photoelectron Spectroscopy (XPS)

Characterization by X-ray photoelectron spectroscopy was performed on an ESCALab Xi (Thermo Scientific) spectrometer with a monochromatic Al K α source. The pressure in the analysis chamber during measurements was $< 10^{-8}$ mbar. The pass energy and step size for narrow scans were 20 eV and 0.1 eV, respectively. The take-off angle was normal to the sample surface. Spectral analysis was performed using Avantage 4.73 software and curve fitting was carried out with a mixture of Gaussian-Lorentzian functions. Peaks were calibrated to the carbon black C–C at 284.5 eV.

XPS survey and narrow band scans are shown in Figures S12-S18 and determined elemental compositions are reported in Tables S12-S18. Narrow band scans were conducted for the elements O, N, C, Rh or Ir and for a number of measurements also including for Cl and F. We note that inclusion of Cl and F in the elemental composition has only a subtle effect on the

overall metal concentration, it varied from 0.01 to 0.05 at% for the analyzed samples. For the sake of consistency, all metal concentration are reported without inclusion of Cl and F.

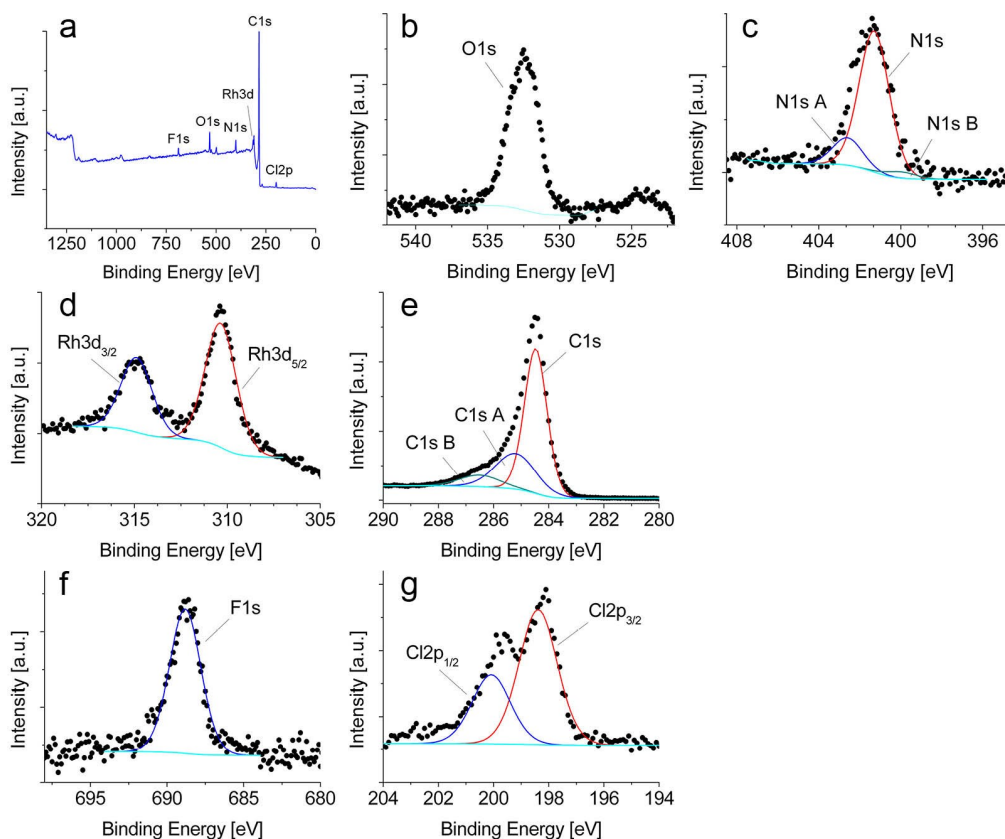


Figure S12. X-ray photoelectron spectra of the hybrid catalyst $\text{Rh}^{\text{III}}\text{C}_{11}\text{CB}$ optimized (Loading 4). (a) Survey scan, narrow scans of the (b) O1s signal, (b) N1s signals, (d) Rh3d_{3/2}/Rh3d_{5/2} signals, (e) C1s signals as well as the signals originating from the counterions, (f) F1s signal, (g) Cl 2p_{1/2}/2p_{3/2} signals.

Table S12. Determined elemental composition of $\text{Rh}^{\text{III}}\text{C}_{11}\text{CB}$ in at% by XPS.

Element	Binding Energy [eV]	FWHM [eV]	Area [CPS × eV]	Atomic % excluding F, Cl	Atomic % including F, Cl
C1s	284.47	0.97	21917.55	55.95	54.33
C1s A	285.20	1.73	9708.56	24.79	24.07
C1s B	286.50	1.73	3161.21	8.07	7.84
O1s	532.40	2.54	5978.17	5.46	5.30
N1s	401.30	1.73	2497.24	3.85	3.74
N1s A	402.60	1.73	486.88	0.75	0.73
N1s B	400.00	1.73	111.94	0.17	0.17
Rh3d _{5/2}	310.35	1.97	3754.37	0.96	0.93
Rh3d _{3/2}	314.89	1.97	2260.26	0	0
F1s	688.82	2.42	2510.55	0	1.69
Cl2p _{3/2}	198.38	1.64	905.45	0	1.22
Cl2p _{1/2}	200.08	1.64	464.72	0	0

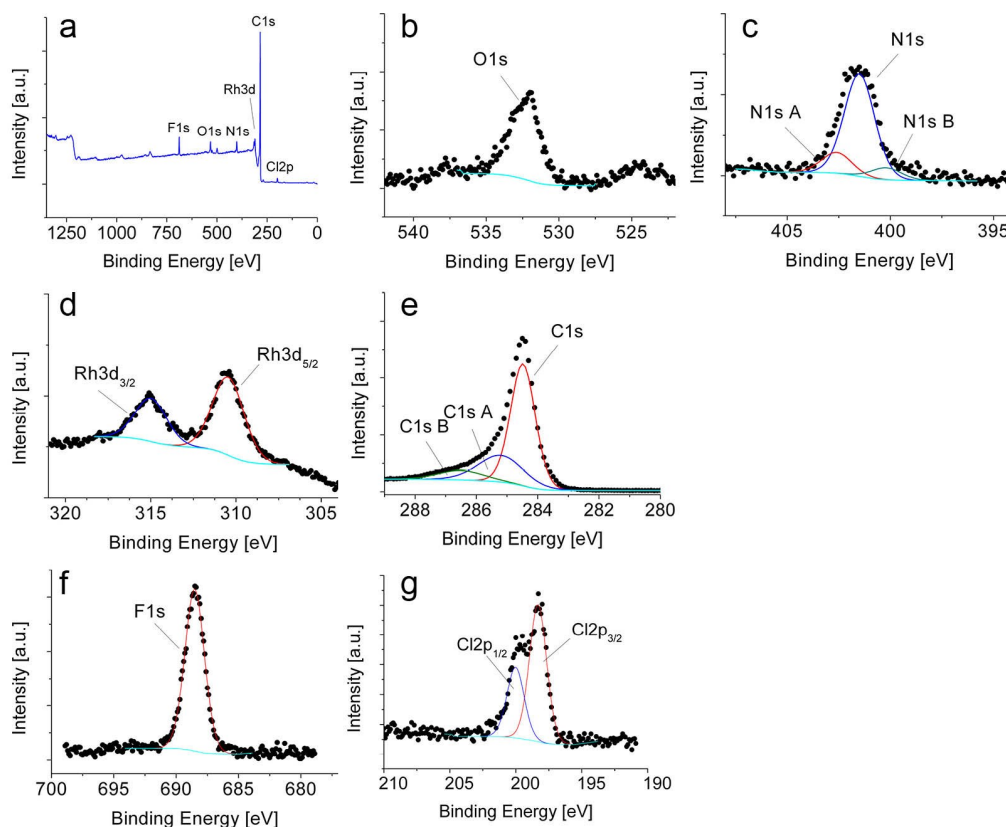


Figure S13. X-ray photoelectron spectra of the hybrid catalyst $\text{Rh}^{\text{III}}\text{C}_6\text{CB}$. (a) Survey scan, narrow scans of the (b) O1s signal, (b) N1s signals, (d) Rh3d_{3/2}/Rh3d_{5/2} signals, (e) C1s signals as well as the signals originating from the counterions, (f) F1s signal, (g) Cl 2p_{1/2}/2p_{3/2} signals.

Table S13. Determined elemental composition of $\text{Rh}^{\text{III}}\text{C}_6\text{CB}$ in at% by XPS.

Signal	Binding Energy [eV]	FWHM [eV]	Area [CPS × eV]	Atomic % excluding F, Cl	Atomic % including F, Cl
C1s	284.48	0.95	21940.43	58.55	55.53
C1s A	285.20	1.73	8793.80	23.47	22.26
C1s B	286.50	1.73	3092.25	8.25	7.83
O1s	532.21	2.30	3893.77	3.71	3.52
N1s	401.50	1.73	2373.69	3.82	3.63
N1s A	402.60	1.73	482.54	0.78	0.74
N1s B	400.20	1.73	275.37	0.44	0.42
Rh3d _{5/2}	310.45	2.18	3639.73	0.97	0.92
Rh3d _{3/2}	314.99	2.18	2020.63	0	0
F1s	688.82	2.42	2510.55	0	4.11
Cl2p _{3/2}	198.38	1.64	905.45	0	1.06
Cl2p _{1/2}	200.08	1.64	464.72	0	0

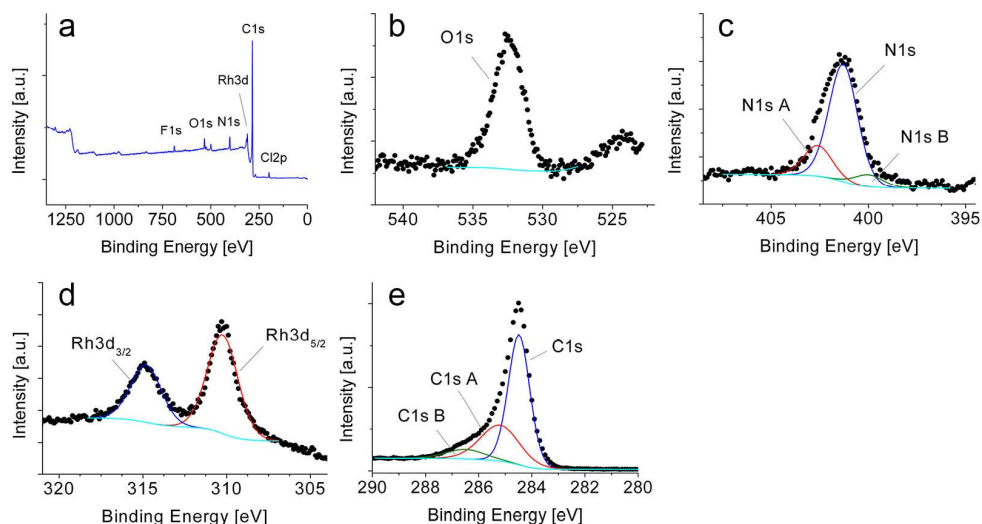


Figure S14. X-ray photoelectron spectra of the hybrid catalyst **Rh^{III}C₀CB** (Loading 4). (a) Survey scan, narrow scans of the (b) O1s signal, (c) N1s signals, (d) Rh3d_{3/2}/Rh3d_{5/2} signals, and (e) C1s signal.

Table S14. Determined elemental composition of **Rh^{III}C₀CB** in at% by XPS.

Signal	Binding Energy [eV]	FWHM [eV]	Area [CPS × eV]	Atomic % <i>excluding F, Cl</i>
C1s	284.48	0.96	32264.29	54.54
C1s A	285.20	1.73	16096.87	27.21
C1s B	286.50	1.73	4246.83	7.18
O1s	532.43	2.51	5650.62	3.41
N1s	401.30	1.73	4627.76	4.72
N1s A	402.60	1.73	1210.41	1.24
N1s B	400.00	1.73	476.04	0.49
Rh3d _{5/2}	310.24	2.15	7204.84	1.22
Rh3d _{3/2}	314.81	2.15	4180.44	

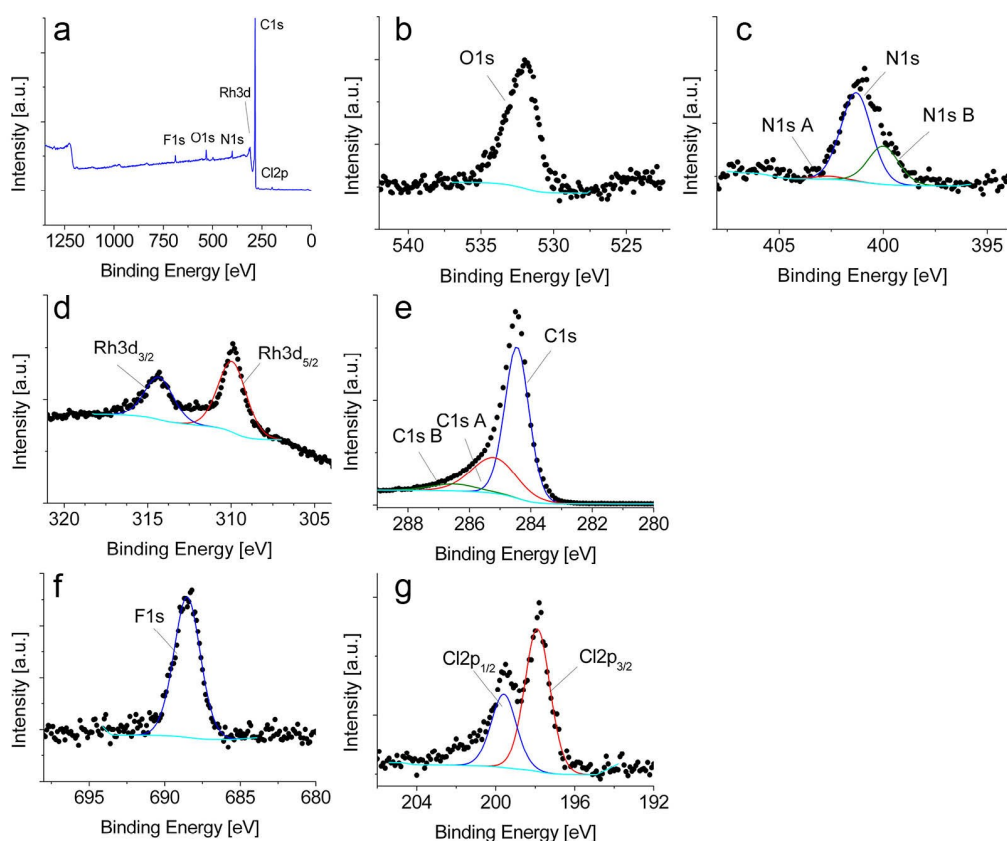


Figure S15. X-ray photoelectron spectra of the hybrid catalyst $\text{Rh}^{\text{III}}\text{Cp}^*\text{C}_{11}\text{CB}$. (a) Survey scan, narrow scans of the (b) O1s signal, (b) N1s signals, (d) Rh3d_{3/2}/Rh3d_{5/2} signals, (e) C1s signals as well as (f) F1s signal and (g) Cl 2p_{1/2}/2p_{3/2} signals.

Table S15. Determined elemental composition of $\text{Rh}^{\text{III}}\text{Cp}^*\text{C}_{11}\text{CB}$ in at% by XPS.

Signal	Binding Energy [eV]	FWHM [eV]	Area [CPS × eV]	Atomic % <i>excluding F, Cl</i>	Atomic % <i>including F, Cl</i>
C1s	284.45	0.92	28983.96	61.18	59.81
C1s A	285.20	1.73	12815.33	27.05	26.45
C1s B	286.50	1.73	2470.49	5.22	5.10
O1s	532.03	2.44	4000.37	3.02	2.95
N1s	401.30	1.73	1562.33	1.99	1.95
N1s A	402.60	1.73	53.18	0.07	0.07
N1s B	400.00	1.73	669.23	0.85	0.83
Rh3d _{5/2}	309.93	2.02	2966.93	0.63	0.61
Rh3d _{3/2}	314.32	2.02	1793.74	0	0
F1s	688.51	2.06	3005.94	0	1.68
Cl2p _{3/2}	197.89	1.51	494.82	0	0.56
Cl2p _{1/2}	199.59	1.51	253.97	0	0

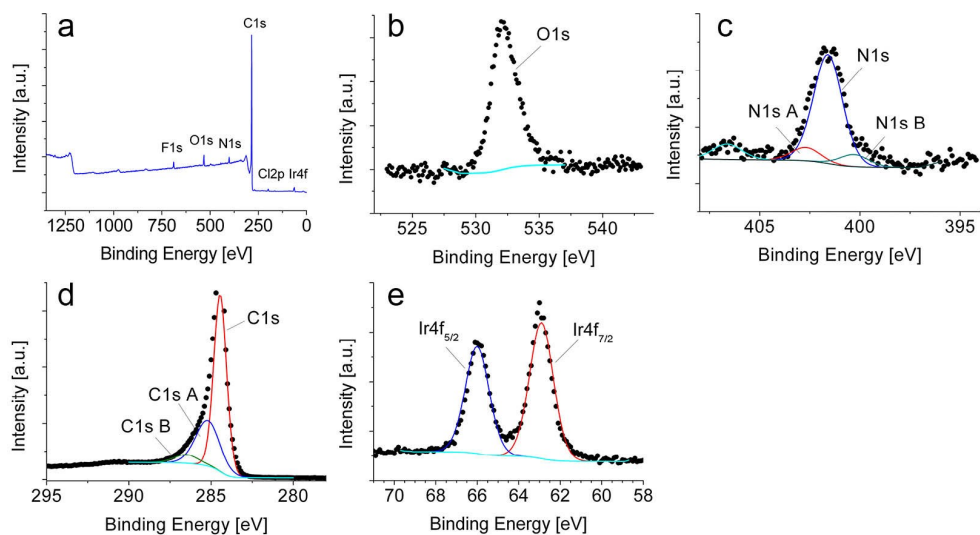


Figure S16. X-ray photoelectron spectra of the hybrid catalyst **Ir^{III}Cp*₁₁CB**. (a) Survey scan, (b) O1s signal, (c) N1s signal, (d) C1s signal and (e) Ir4f_{5/2}/Ir4f_{7/2} signals.

Table S16. Determined elemental composition of **Ir^{III}Cp*₁₁CB** in at% by XPS.

Signal	Binding Energy [eV]	FWHM [eV]	Area [CPS × eV]	Atomic % <i>excluding F, Cl</i>
C1s	284.45	0.92	62624.64	59.51
C1s A	285.20	1.73	30486.37	28.97
C1s B	286.50	1.73	5722.30	5.44
O1s	532.12	2.30	8879.90	3.02
N1s	401.40	1.70	3561.49	2.04
N1s A	402.50	1.70	434.50	0.25
N1s B	400.10	1.70	393.68	0.23
N1s C	406.43	1.73	508.65	0.29
Ir4f _{7/2}	62.91	1.36	2024.78	0.24
Ir4f _{5/2}	66.01	1.36	1592.76	

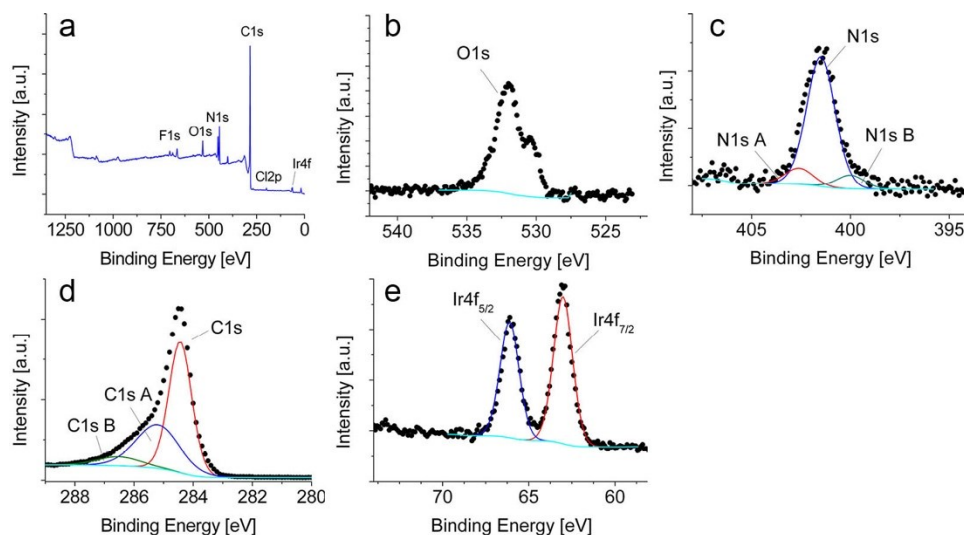


Figure S17. X-ray photoelectron spectra of the hybrid catalyst $\text{Ir}^{\text{III}}\text{Cp}^*\text{C}_6\text{CB}$. (a) Survey scan, (b) O1s signal, (c) N1s signal, (d) C1s signal and (e) Ir4f_{5/2}/Ir4f_{7/2} signals.

Table S17. Determined elemental composition of $\text{Ir}^{\text{III}}\text{Cp}^*\text{C}_6\text{CB}$ in at% by XPS.

Signal	Binding Energy [eV]	FWHM [eV]	Area [CPS × eV]	Atomic % <i>excluding F, Cl</i>
C1s	284.44	0.91	48255.82	52.12
C1s A	285.20	1.73	30374	32.81
C1s B	286.50	1.73	6330.84	6.84
O1s	532.03	1.98	11979.28	4.63
N1s	401.50	1.69	4144.49	2.70
N1s A	402.60	1.69	513.4	0.33
N1s B	400.00	1.69	433.05	0.28
Ir4f _{7/2}	63.02	1.30	2046.81	0.28
Ir4f _{5/2}	66.12	1.30	1610.09	

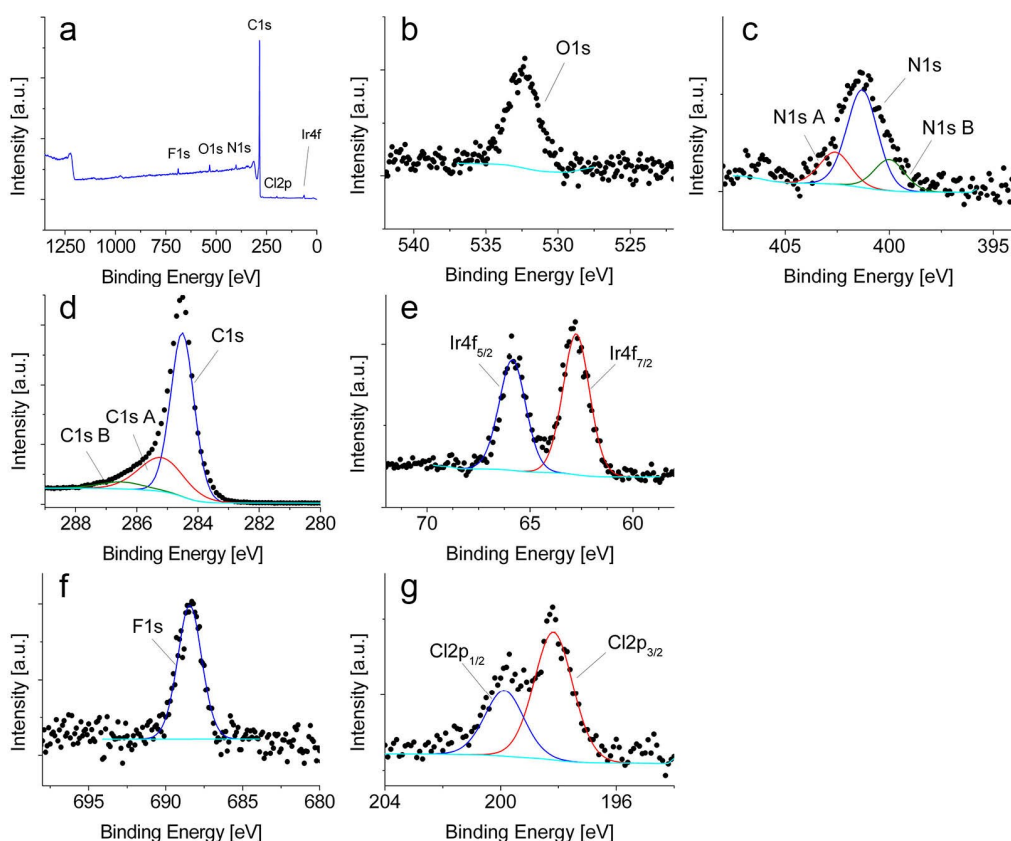


Figure S18. X-ray photoelectron spectra of the hybrid catalyst $\text{Ir}^{\text{III}}\text{Cp}^*\text{C}_0\text{CB}$. (a) Survey scan, narrow scans of the (b) O1s signal, (b) N1s signals, (d) C1s signal, (e) Ir4f_{5/2}/Ir4f_{7/2} signals and (f) F1s signal and (g) Cl 2p_{1/2}/2p_{3/2} signals.

Table S18. Determined elemental composition of $\text{Ir}^{\text{III}}\text{Cp}^*\text{C}_0\text{CB}$ in at% by XPS.

Signal	Binding Energy [eV]	FWHM [eV]	Area [CPS × eV]	Atomic % excluding F, Cl	Atomic % including F, Cl
C1s	284.51	0.90	26844.16	64.55	63.52
C1s A	285.20	1.73	10609.55	25.51	25.10
C1s B	286.50	1.73	2159.06	5.19	5.11
O1s	532.41	2.36	2646.25	2.27	2.24
N1s	401.30	1.73	942.41	1.37	1.35
N1s A	402.60	1.73	308.46	0.45	0.44
N1s B	400.00	1.73	298.74	0.43	0.43
Ir4f _{7/2}	62.76	1.52	711.56	0.22	0.21
Ir4f _{5/2}	65.86	1.52	559.73	0	0
F1s	688.44	1.86	1775.74	0	1.14
Cl2p _{3/2}	198.17	1.58	361.26	0	0.47
Cl2p _{1/2}	199.87	1.58	185.41	0	0

3.2. Optimization of Grafting Conditions

Optimization of Grafting for Rh^IC₁₁ on CB

To optimize the catalyst loading on CB, the amount of catalyst used in the grafting process was varied from 0.7 - 14.0 μ moles per mg CB. The obtained hybrid catalysts of the series with varying metal concentration (Loadings 1, 2, 3 and 5) were analyzed by EDX (Figure S19) and XPS (Figure S20). Data of loading 4, the optimized loading, is shown in Figure 3 (EDX) and Figures S12 (XPS). The determined elemental compositions are summarized in Tables S20 (EDX) and S21 (XPS); and for the loading 4 in Tables S4 (EDX) and S12 (XPS). All samples show an even distribution of Rh along the surface (EDX, Figures S19, S12). The Rh concentration increases with increasing amount of catalyst used. In the regime of lower loading a doubling of the amount of catalyst gives nearly twice as much metal on the surface (Loading 1 and 2). The surface bound amount becomes less proportional when increasing the amount further and reaches saturation (Loading 5) as doubling the amount from sample 4 to 5 gives only a slightly higher Rh reading using EDX and no increase using XPS.

Overall, the metal loading obtained by XPS are slightly higher than those obtained by EDX. This is likely due to differences in both elemental detection and data evaluation. While for instance the N1s signals are well resolved in XPS, N is not detectable by EDX in the analyzed Rh^I samples. Likewise, sulfur as impurity in the CB gives a distinct signal in EDX (Figure S5) but not in XPS. In addition, we note that all samples contain Cl, which is expressed in both EDX and XPS. We included Cl in the EDX data evaluation, but not in the XPS. As noted above, we tested inclusion of Cl and F for several samples and note that this results only in subtle differences for the Rh concentration.

Table S19. Summary of determined metal concentrations by EDX and XPS for $\text{Rh}^{\text{I}}\text{C}_{11}$ on CB.

Loading	$\mu\text{moles/mg}^1$	wt % Rh (EDX)	at% Rh (EDX)	at% Rh (XPS)
1	0.07	1.13 ± 0.11	0.15 ± 0.03	0.27
2	0.14	2.02 ± 0.16	0.25 ± 0.02	0.53
3	0.28	2.89 ± 0.15	0.36 ± 0.02	0.69
4	0.70	3.93 ± 0.22	0.49 ± 0.03	0.96
5	1.40	5.32 ± 0.45	0.68 ± 0.06	0.94

¹ Moles of catalyst per mg of carbon black applied in the grafting process.

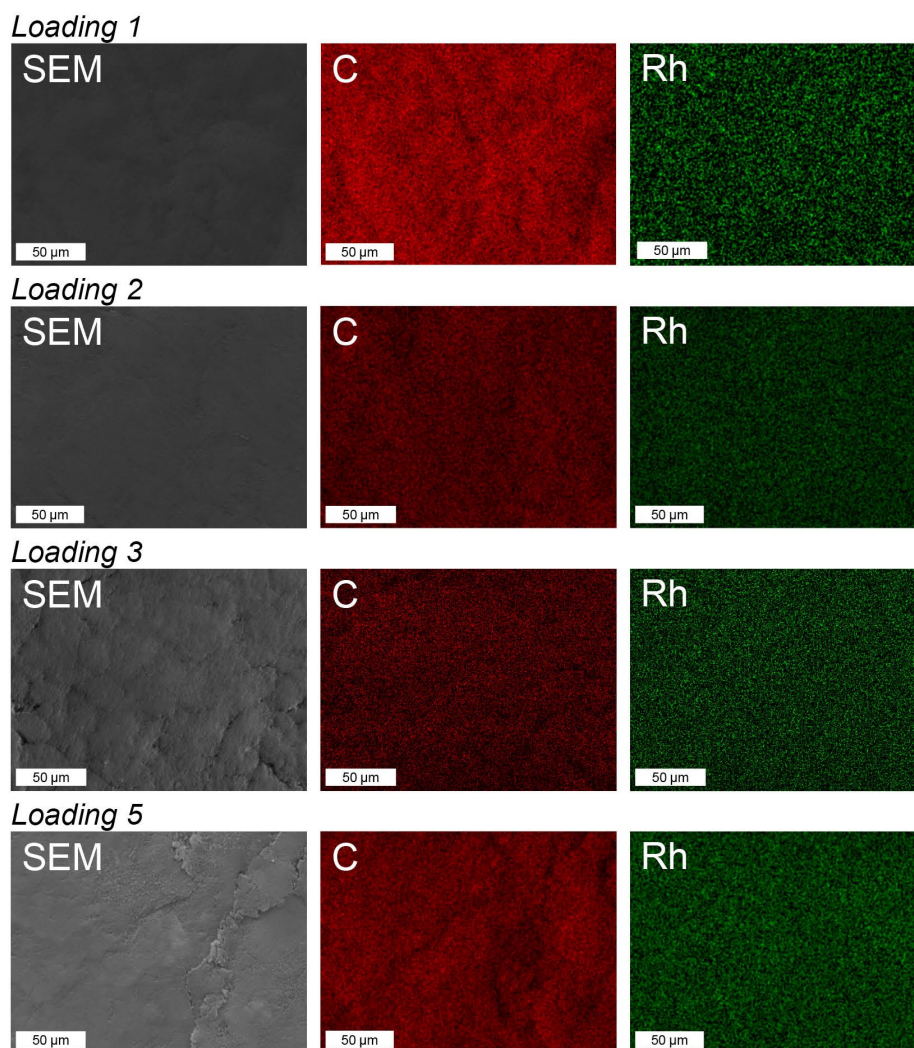


Figure S19. (a) SEM images and EDX maps for carbon (C) and rhodium (Rh) for the $\text{Rh}^{\text{III}}\text{C}_{11}\text{CB}$ samples 1, 2, 3 and 5 with varying catalyst loadings. Data of Loading 4 (optimized conditions) is shown in the main manuscript, Figure 2.

Table S20. Determined elemental compositions of the optimization series for $\text{Rh}^{\text{III}}\text{C}_{11}\text{CB}$ in wt% by EDX. The determined values for Loading 4 are reported on page S20.

Loading 1

Element	Area 1	Area 2	Area 3	Area 4	Area 5	Average
C	92.78	92.84	92.97	94.55	93.33	93.29
O	3.64	3.72	3.53	1.31	3.43	3.13
S	1.12	1.1	1.08	0.96	1.06	1.06
Cl	1.24	1.25	1.13	2.16	1.13	1.38
Rh	1.21	1.09	1.29	1.01	1.04	1.13
Total	100	100	100	100	100	100

Loading 2

Element	Area 1	Area 2	Area 3	Area 4	Area 5	Average
C	92.07	92.07	92.52	92.53	92.11	92.26
O	3.56	3.29	3.13	2.84	3.98	3.36
S	1.09	1.06	1.07	1.04	0.97	1.05
Cl	1.38	1.38	1.21	1.43	1.16	1.31
Rh	1.90	2.19	2.07	2.15	1.77	2.02
Total	100	100	100	100	100	100

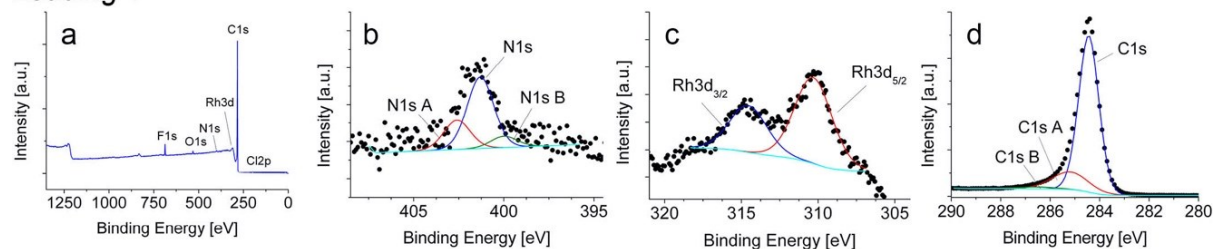
Loading 3

Element	Area 1	Area 2	Area 3	Area 4	Area 5	Average
C	89.59	88.72	89.02	89.01	88.39	88.95
O	4.73	5.10	4.79	4.62	5.09	4.87
S	1.07	1.06	1.11	1.05	1.14	1.09
Cl	2.00	2.26	2.15	2.33	2.29	2.21
Rh	2.60	2.86	2.92	2.99	3.09	2.89
Total	100	100	100	100	100	100

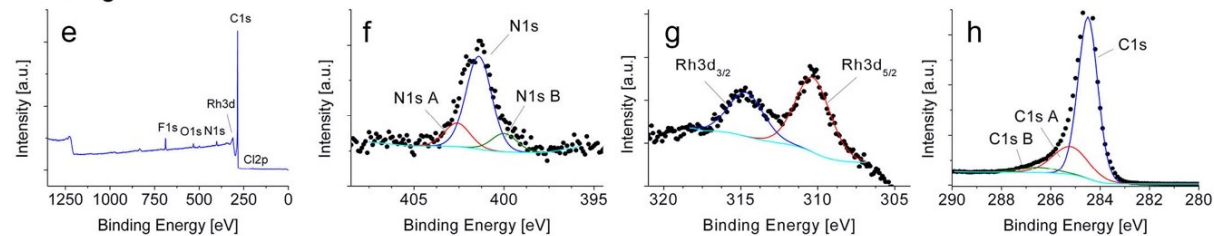
Loading 5

Element	Area 1	Area 2	Area 3	Area 4	Area 5	Area 6	Average
C	86.49	86.91	87.78	88.25	86.83	87.38	87.27
O	3.21	2.62	2.81	3.59	4.03	3.92	3.36
S	1.13	1.08	0.96	0.87	0.93	0.95	0.99
Cl	3.47	3.43	3.27	2.63	2.75	2.83	3.06
Rh	5.71	5.96	5.18	4.66	5.46	4.92	5.32
Total	100	100	100	100	100	100	100

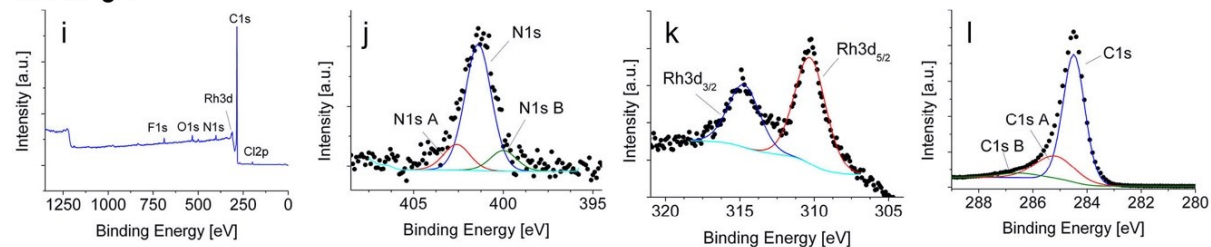
Loading 1



Loading 2



Loading 3



Loading 5

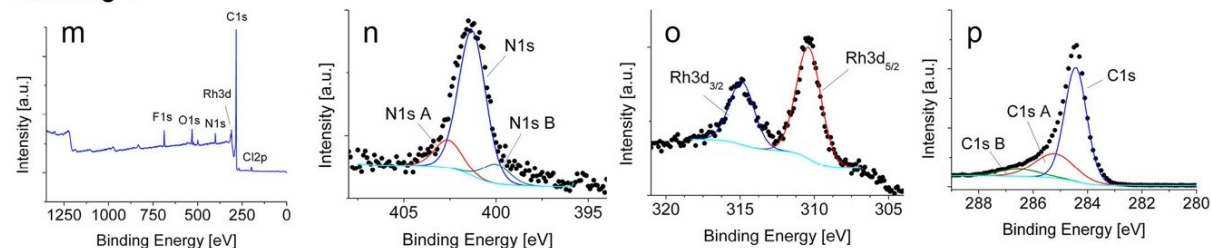


Figure S20. X-ray photoelectron spectra of the hybrid catalyst $\text{Rh}^{\text{III}}\text{C}_{11}\text{CB}$ with different catalyst loadings (Loadings 1, 2, 3, and 5). Survey scans (a, e, i, m), N1s signals (b, f, j, n), Rh3d_{3/2}/Rh3d_{5/2} signals (c, g, k, o), and (e) C1s signals (d, h, l, p). The spectra of loading 4 is shown on p S26.

Table S21. Determined elemental composition of the optimization series for **Rh^{III}C₁₁CB** on CB in at% by XPS. The determined values for Loading 4 are reported on p 26.

<i>Loading 1</i>				
Signal	Binding Energy [eV]	FWHM [eV]	Area [CPS × eV]	Atomic % <i>excluding F, Cl</i>
C1s	284.44	0.96	70590.51	79.96
C1s A	285.20	1.73	14245.06	16.14
C1s B	286.50	1.73	1418.26	1.61
O1s	532.14	2.23	3070.51	1.24
N1s	401.30	1.73	726.98	0.50
N1s A	402.60	1.73	295.89	0.20
N1s B	400.00	1.73	114.61	0.08
Rh3d _{5/2}	310.31	2.87	2414.98	0.27
Rh3d _{3/2}	314.55	2.87	1313.82	
<i>Loading 2</i>				
Signal	Binding Energy [eV]	FWHM [eV]	Area [CPS × eV]	Atomic % <i>excluding F, Cl</i>
C1s	284.50	0.94	65731.38	69.19
C1s A	285.20	1.73	21063.82	22.17
C1s B	286.50	1.73	4145.68	4.36
O1s	532.22	2.52	4105.75	1.55
N1s	401.40	1.67	2384.37	1.52
N1s A	402.60	1.67	601.37	0.38
N1s B	400.00	1.67	463.55	0.29
Rh3d _{5/2}	310.32	2.68	5040.46	0.53
Rh3d _{3/2}	314.65	2.68	2979.34	
<i>Loading 3</i>				
Signal	Binding Energy [eV]	FWHM [eV]	Area [CPS × eV]	Atomic % <i>excluding F, Cl</i>
C1s	284.49	0.96	37373.22	67.48
C1s A	285.20	1.73	12379.17	22.35
C1s B	286.50	1.73	2382.39	4.30
O1s	532.17	2.57	3445.08	2.22
N1s	401.40	1.73	1981.32	2.16
N1s A	402.60	1.73	400.52	0.44
N1s B	400.00	1.73	323.43	0.35
Rh3d _{5/2}	310.27	2.48	3840.58	0.69
Rh3d _{3/2}	314.72	2.48	2259.90	
<i>Loading 5</i>				
Signal	Binding Energy [eV]	FWHM [eV]	Area [CPS × eV]	Atomic % <i>excluding F, Cl</i>
C1s	284.43	0.96	20085.07	58.80
C1s A	285.20	1.73	7837.23	22.95
C1s B	286.50	1.73	2365.94	6.93
O1s	532.23	2.51	5065.69	5.30
N1s	401.30	1.73	2170.43	3.84
N1s A	402.60	1.73	416.49	0.74
N1s B	400.00	1.73	289.10	0.51
Rh3d _{5/2}	310.39	1.96	3195.20	0.94
Rh3d _{3/2}	314.88	1.96	1811.01	0

Optimization of Grafting for $\text{Rh}^{\text{I}}\text{C}_0$ on CB

The optimization was carried out analogously to the above described optimization for $\text{Rh}^{\text{I}}\text{C}_{11}$.

The obtained hybrid catalysts of the series with varying metal concentration (Loadings 1,2,3 and 5) were analyzed by EDX (Figure S21) and XPS (Figure S22). Data of loading 4, the optimized loading, is shown in Figure S6 (EDX) and Figures S14 (XPS). The determined elemental compositions are summarized in Tables S23 (EDX) and S24 (XPS); and for the loading 4 in Tables S6 (EDX) and S14 (XPS). The observed trend is in line with the above described while the obtained overall Rh loadings are slightly higher for the $\text{Rh}^{\text{III}}\text{C}_0\text{CB}$ derivatives compared to the $\text{Rh}^{\text{III}}\text{C}_{11}\text{CB}$ derivatives, which is likely due to the size of the molecule. The smaller molecule may bind at sites, which are less accessible to the larger analogues, explaining the observed differences.

Table S22. Summary of determined metal concentrations for $\text{Rh}^{\text{III}}\text{C}_0\text{CB}$ on CB:

Sample	$\mu\text{moles/mg}^1$	wt % Rh (EDX)	at % Rh (EDX)	at% Rh (XPS)
1	0.07	1.32 ± 0.25	0.16 ± 0.03	0.45
2	0.14	1.91 ± 0.32	0.23 ± 0.04	0.62
3	0.28	2.58 ± 0.30	0.32 ± 0.03	0.70
4	0.70	6.37 ± 0.28	0.82 ± 0.04	1.22
5	1.40	8.79 ± 1.13	1.17 ± 0.04	1.25

¹ Moles of catalyst per mg of carbon black applied in the grafting process.

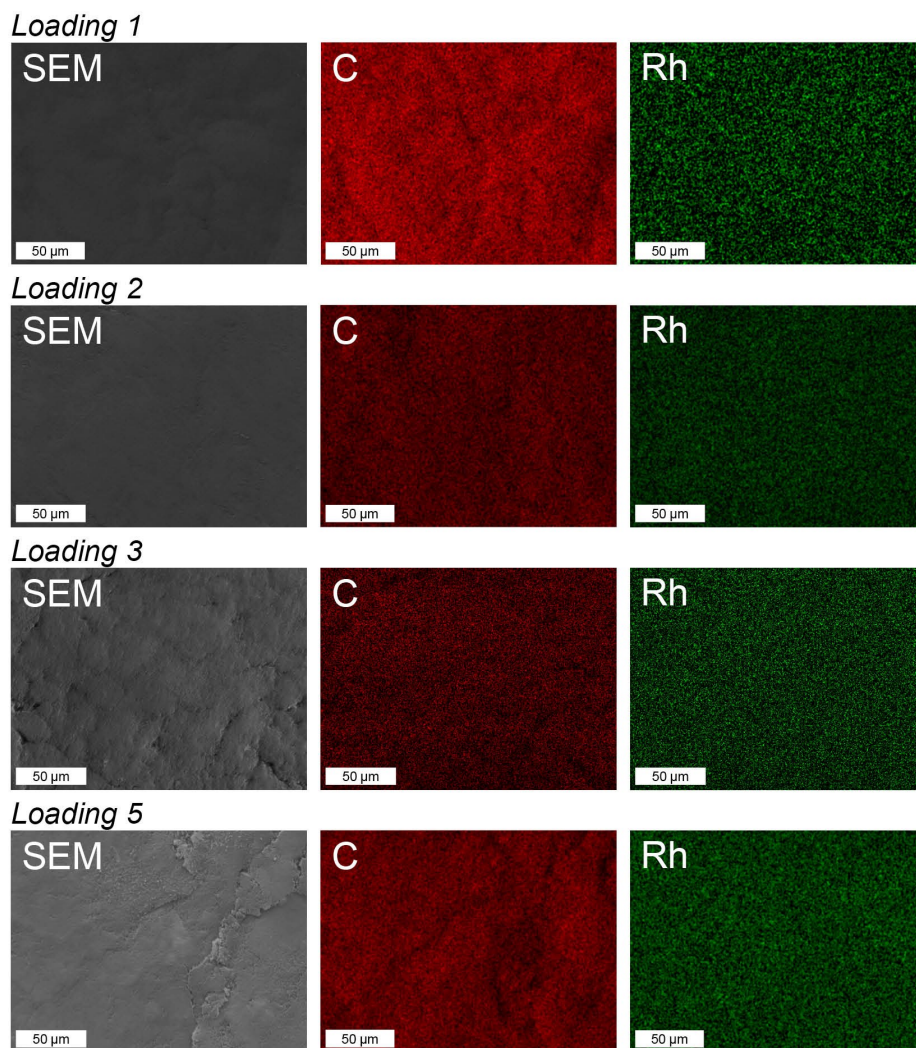
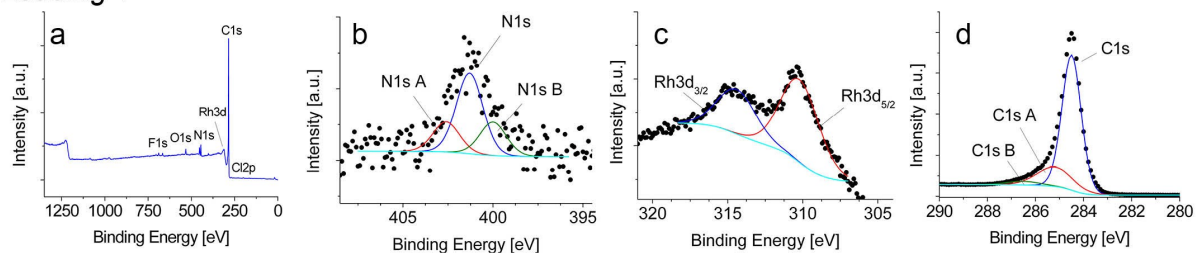


Figure S21. (a) SEM images and EDX maps for carbon (C) and rhodium (Rh) for the $\text{Rh}^{\text{III}}\text{C}_0\text{CB}$ samples 1, 2, 3 and 5 with varying catalyst loadings. Data of loading 4 (optimized conditions) is shown in Figure S6.

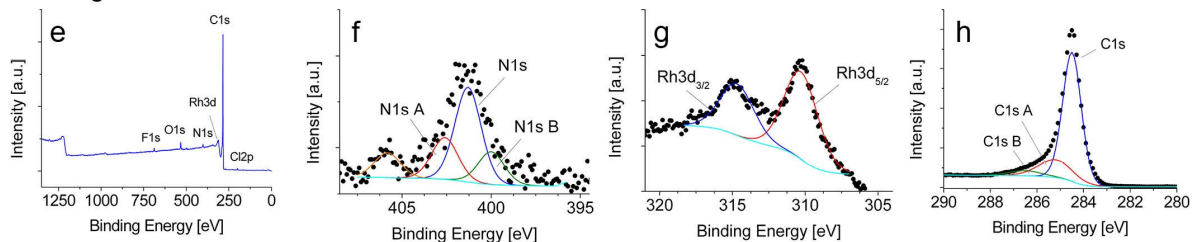
Table S23. Determined elemental compositions of the optimization series for $\text{Rh}^{\text{III}}\text{C}_0\text{CB}$ in wt% by EDX. The determined values for Loading 4 are reported on p S21.

<i>Loading 1</i>							
Element	Area 1	Area 2	Area 3	Area 4	Area 5	Area 6	Average
C	93.73	92.74	92.96	92.84	94.61	92.91	93.30
O	3.27	3.79	3.61	3.78	2.62	3.22	3.38
S	1.03	1.04	1.14	1.03	1.10	1.17	1.09
Cl	0.78	1.05	0.89	0.90	0.78	1.09	0.92
Rh	1.19	1.38	1.39	1.44	0.89	1.60	1.32
Total	100	100	100	100	100	100	100
<i>Loading 2</i>							
Element	Area 1	Area 2	Area 3	Area 4	Area 5	Area 6	Average
C	91.71	91.93	91.76	92.54	92.2	92.71	92.14
O	3.35	3.93	4.28	3.05	3.33	2.82	3.46
S	1.08	1.11	1.16	1.11	1.09	1.14	1.12
Cl	1.44	1.38	1.30	1.38	1.42	1.36	1.38
Rh	2.42	1.65	1.50	1.93	1.96	1.97	1.91
Total	100	100	100	100	100	100	100
<i>Loading 3</i>							
Element	Area 1	Area 2	Area 3	Area 4	Area 5	Area 6	Average
C	89.92	90.84	90.66	92.5	91.31	91.86	91.18
O	4.94	3.34	3.70	2.11	3.63	3.05	3.46
S	0.89	0.96	1.02	0.88	0.93	0.95	0.94
Cl	1.88	1.90	1.77	1.77	1.88	1.81	1.84
Rh	2.38	2.95	2.85	2.73	2.25	2.33	2.58
Total	100	100	100	100	100	100	100
<i>Loading 5</i>							
Element	Area 1	Area 2	Area 3	Area 4	Area 5	Area 6	Average
C	80.63	83.85	82.23	85.19	83.5	79.15	82.43
O	3.11	3.54	3.16	2.73	2.63	4.56	3.29
S	0.87	0.79	0.96	0.86	0.94	0.89	0.89
Cl	5.34	4.09	4.89	3.75	4.35	5.24	4.61
Rh	10.05	7.73	8.77	7.47	8.57	10.15	8.79
Total	100	100	100	100	100	100	100

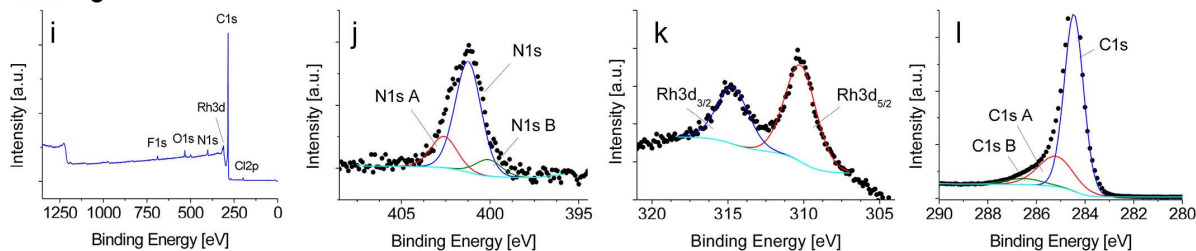
Loading 1



Loading 2



Loading 3



Loading 5

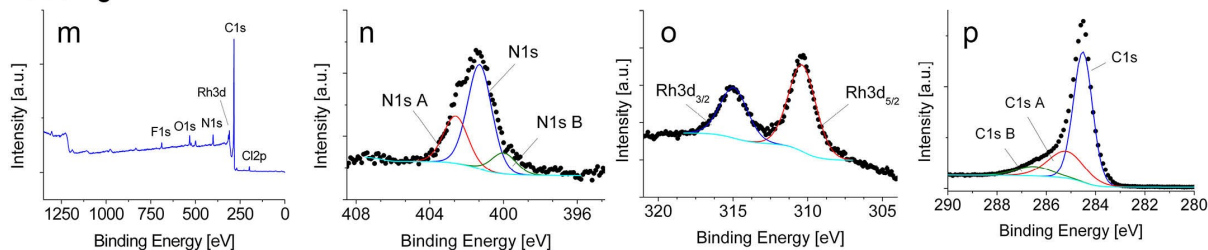


Figure S22. X-ray photoelectron spectra of the hybrid catalyst $\text{Rh}^{\text{III}}\text{C}_0\text{CB}$ with different catalyst loadings (Loadings 1, 2, 3, and 5). Survey scans (a, e, i, m), N1s signals (b, f, j, n), Rh3d_{3/2}/Rh3d_{5/2} signals (c, g, k, o), and (e) C1s signals (d, h, l, p). The spectra of loading 4 is shown on p S14.

Table S24. Determined elemental composition of the optimization series for **Rh^{III}C₀CB** on CB in at% by XPS. The determined values for Loading 4 are reported on p S14.

<i>Loading 1</i>				
Signal	Binding Energy [eV]	FWHM [eV]	Area [CPS × eV]	Atomic % <i>excluding F, Cl</i>
C1s	284.50	0.93	66769.23	73.33
C1s A	285.20	1.73	17666.29	19.40
C1s B	286.50	1.61	3068.72	3.37
O1s	532.35	2.37	5610.24	2.20
N1s	401.30	1.73	1054.13	0.70
N1s A	402.60	1.73	393.11	0.26
N1s B	400.00	1.73	437.05	0.29
Rh3d _{5/2}	310.21	3.01	4056.33	0.45
Rh3d _{3/2}	314.41	3.01	2184.51	

<i>Loading 2</i>				
Signal	Binding Energy [eV]	FWHM [eV]	Area [CPS × eV]	Atomic % <i>excluding F, Cl</i>
C1s	284.50	0.93	66181.53	71.30
C1s A	285.20	1.73	16641.32	17.93
C1s B	286.50	1.61	4471.69	4.82
O1s	532.59	2.34	7210.08	2.78
N1s	401.30	1.73	1926.69	1.25
N1s A	402.60	1.73	844.82	0.55
N1s B	400.00	1.73	662.98	0.43
N1s C	405.82	1.73	500.92	0.33
Rh3d _{5/2}	310.29	2.76	5736.71	0.62
Rh3d _{3/2}	314.69	2.76	3393.44	

<i>Loading 3</i>				
Signal	Binding Energy [eV]	FWHM [eV]	Area [CPS × eV]	Atomic % <i>excluding F, Cl</i>
C1s	284.46	0.93	66680.44	68.39
C1s A	285.20	1.73	21080.56	21.62
C1s B	286.50	1.73	4239.97	4.35
O1s	532.51	2.43	5496.64	2.02
N1s	401.24	1.73	3294.51	2.04
N1s A	402.60	1.73	927.89	0.57
N1s B	400.10	1.73	488.82	0.30
Rh3d _{5/2}	310.21	2.46	6816.78	0.70
Rh3d _{3/2}	314.68	2.46	4029.66	

<i>Loading 5</i>				
Signal	Binding Energy [eV]	FWHM [eV]	Area [CPS × eV]	Atomic % <i>excluding F, Cl</i>
C1s	284.45	0.91	25326.36	57.95
C1s A	285.20	1.73	9938.73	22.74
C1s B	286.50	1.73	3459.75	7.92
O1s	532.47	2.41	4639.02	3.79
N1s	401.30	1.56	2785.79	3.85
N1s A	402.60	1.56	1260.48	1.74
N1s B	400.00	1.56	551.44	0.76
Rh3d _{5/2}	310.18	2.17	5460.04	1.25
Rh3d _{3/2}	314.71	2.17	3182.5	0

3.4 Characterization by X-ray Absorption Spectroscopy (XAS)

XAS measurements were performed at the Australian Synchrotron. XANES and EXAFS data was recorded of the Rh and Ir precursors and the Rh and Ir hybrid catalysts the XAS Beamline in transmission mode, at the Rh K-edge (~ 23.2 keV) and the Ir L3 edge (~ 11.2 keV). Samples were measured using Hutch B under ambient temperature and pressure.

Sample preparation

The Rh or Ir complex (~ 70 mg) was ground to a fine powder and mixed with cellulose (~ 5 mg) as binding agent. The mixture was pressed into a 7 mm pellet using a handheld mechanical press. The hybrid catalyst samples were treated analogously or pressed into pellets under absence of the binding agent using a hydraulic press applying a force of 3t. The pellets were mounted on 7 mm sample holders with Kapton tape, and the sample holders were mounted on a revolving holder plate on the XAS spectrometer and subsequently measured.

Measurements and data analysis

We conducted radiation hardness tests for a number of samples. The Rh samples were measured twice for each sample and the reported data is the average of two individual measurements. The Ir samples were measured once. Recorded spectra were aligned to an internal reference standard of Rh or Ir foil. Data processing was performed with the Demeter software package using Athena.⁷

Radial distributions

To compare experimental radial distributions originating in the combined scattering paths attributed to molecular structure to predicted ones, we modelled individual and combined scattering paths of the relevant functional groups and structure elements of the catalyst by employing Feff calculations using Artemis.⁷ As reasonable approximation, we used the known

molecular structures obtained from X-ray single crystal diffraction^{8, 9} for the three different cationic fragments of the precursor head groups in this study, i.e. $[\text{Rh}(\text{CO})_2(\text{PyTPh})]^+$, $[\text{RhCp}^*\text{Cl}(\text{PyTPh})]^+$, $[\text{IrCp}^*\text{Cl}(\text{PyTPh})]^+$, to calculate expected scattering paths with weighted contributions. Figure S23 shows experimental (Exp.) and calculated (Calc.) radial distributions for the selected scattering paths.

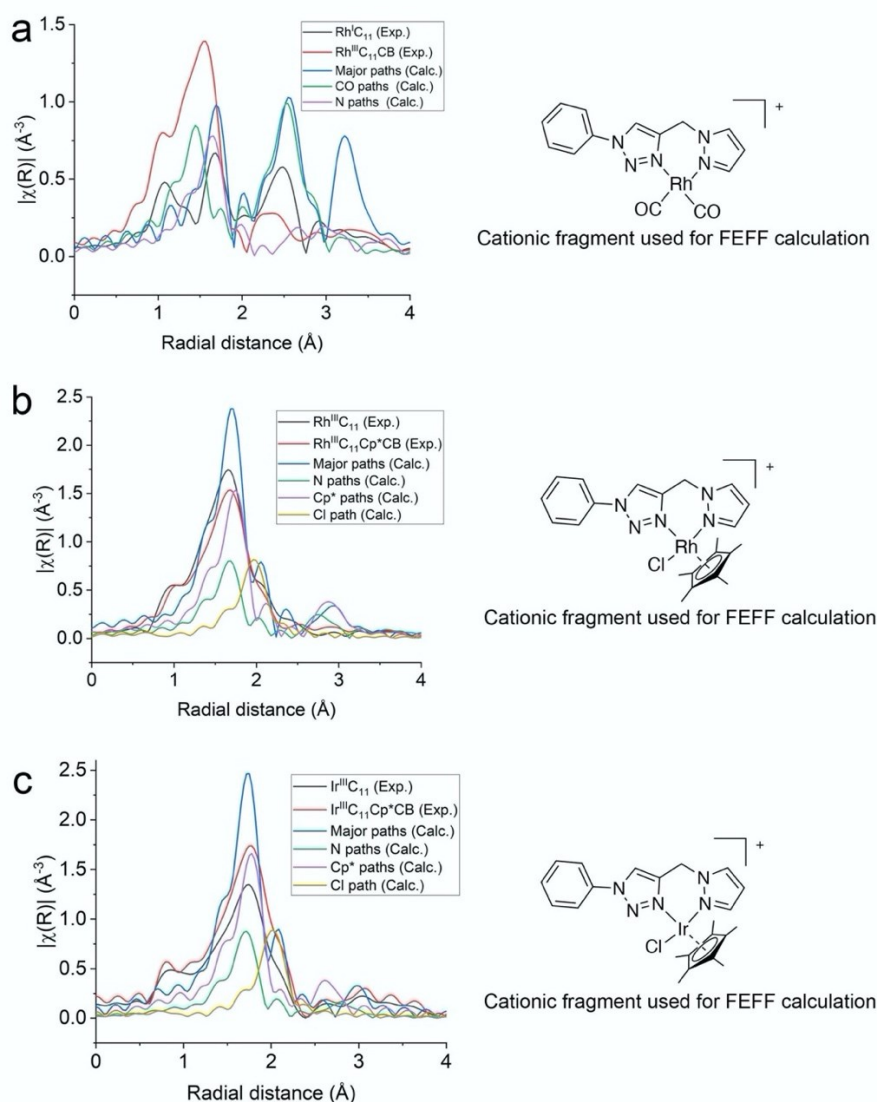


Figure S23. Comparison of experimental radial distributions for compounds before and after immobilization vs calculated scattering paths for the depicted cationic fragments. (a) $\text{Rh}^{\text{I}}\text{C}_{11}$ and $\text{Rh}^{\text{III}}\text{C}_{11}\text{CB}$ vs calculated combined major scattering paths and scattering paths attributed to CO ligands and N atoms of PyT ligands. (b) $\text{Rh}^{\text{III}}\text{C}_{11}$ and $\text{Rh}^{\text{III}}\text{C}_{11}\text{Cp}^*\text{CB}$ vs calculated combined major scattering paths and scattering paths attributed to N atoms of PyT ligands, Cp^* ligands and Cl atoms. (c) $\text{Ir}^{\text{III}}\text{C}_{11}$ and $\text{Ir}^{\text{III}}\text{C}_{11}\text{Cp}^*\text{CB}$ vs calculated combined major scattering paths and scattering paths attributed to N atoms of PyT ligands, Cp^* ligands and Cl atoms.

Curve Fitting

We conducted fitting of the experimental data versus data obtained from FEFF calculations involving the X-ray crystallography data of the matching cationic fragments mentioned above, using the Artemis software. For the set of Ir^{III}Cp**C_n* complexes and the CB immobilised hybrid catalysts Ir^{III}Cp**C_n*CB, as well as **Rh^{III}Cp**C₁₁*** and **Rh^{III}Cp**C₁₁*CB**, obtained fits match the experiment which further supports that the immobilised catalysts do not undergo structural changes during immobilisation on the head-group. Starting guess values for the refinement for selected scattering paths were: $\sigma^2 = 0.003$; Amplitude factor = 1, $\Delta E_0 = 0$. We implemented in certain circumstance restraints of σ^2 to obtain reasonable fits. The calculated bond distance for the first coordination shell, i.e. the two binding N atoms, five C atoms of the Cp* ring and, the Cl atoms are reported below. N denotes the coordination number for each atom in the relevant scattering path. Fourier transformed EXAFS data (k^2 weighted) are shown in Figures S24-27.

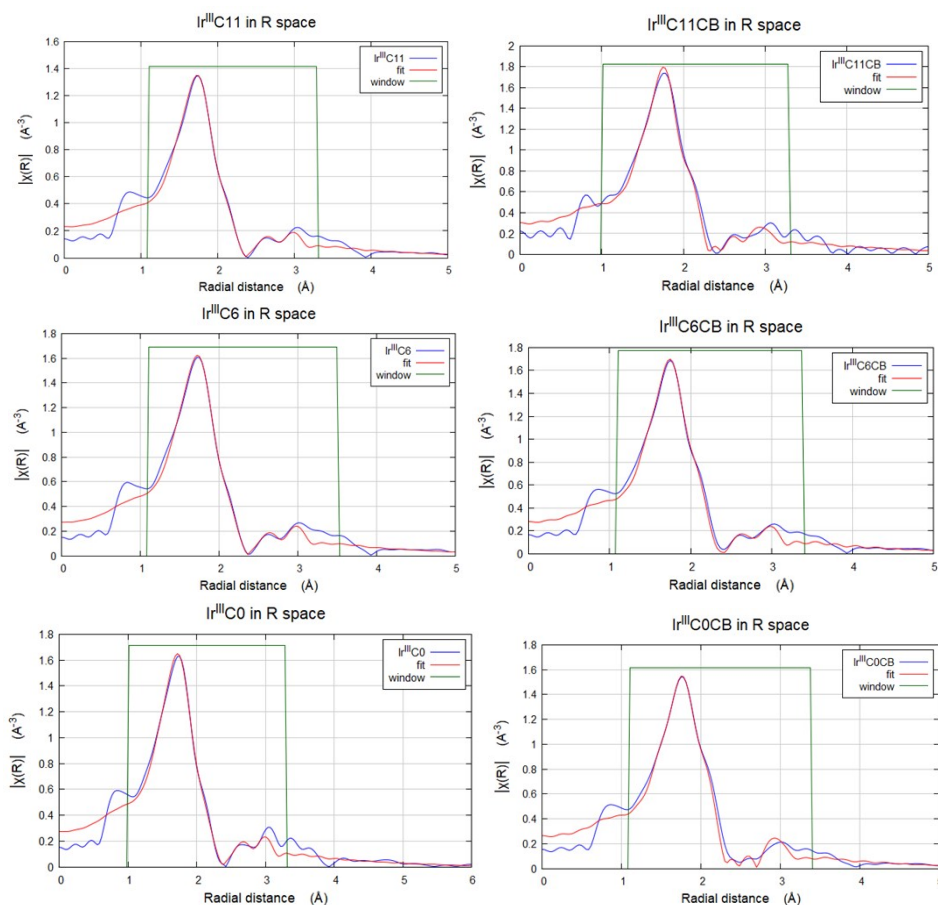


Figure S24. Fourier transformed EXAFS data (k^2 weighted) for $\text{Ir}^{\text{III}}\text{Cp}^*\text{C}_n$ and $\text{Ir}^{\text{III}}\text{Cp}^*\text{C}_n\text{CB}$ ($n = 11, 6, 0$) showing the experimental data and obtained fit, and the applied fitting window.

Table S25. Summary of refined parameters for the first coordination shell from the fit for $\text{Ir}^{\text{III}}\text{Cp}^*\text{C}_n$ and $\text{Ir}^{\text{III}}\text{Cp}^*\text{C}_n\text{CB}$ ($n = 11, 6, 0$) involving the major scattering paths.

Compound	Bond	N	Bond length [$\text{\AA}$]	σ^2	E_0 [eV]	R-factor
$\text{Ir}^{\text{III}}\text{Cp}^*\text{C}_{11}$	Ir-N (PyT, 2N)	2	2.094	0.005(2)	9.03 +/- 1.07	0.005
	Ir-C (Cp*, 5C)	5	2.166	0.006(1)		
	Ir-Cl	1	2.400	0.007(2)		
$\text{Ir}^{\text{III}}\text{Cp}^*\text{C}_{11}\text{CB}$	Ir-N (PyT, 2N)	2	2.091	0.005(3)	12.00 +/- 0.84	0.013
	Ir-C (Cp*, 5C)	5	2.163	0.003(1)		
	Ir-Cl	1	2.396	0.003(1)		
$\text{Ir}^{\text{III}}\text{Cp}^*\text{C}_6$	Ir-N (PyT, 2N)	2	2.101	0.005(3)	8.53 +/- 1.64	0.007
	Ir-C (Cp*, 5C)	5	2.174	0.007(2)		
	Ir-Cl	1	2.408	0.008(3)		
$\text{Ir}^{\text{III}}\text{Cp}^*\text{C}_6\text{CB}$	Ir-N (PyT, 2N)	2	2.091	0.006(3)	9.05 +/- 0.89	0.007
	Ir-C (Cp*, 5C)	5	2.163	0.004(1)		
	Ir-Cl	1	2.396	0.004(1)		
$\text{Ir}^{\text{III}}\text{Cp}^*\text{C}_0$	Ir-N (PyT, 2N)	2	2.097	0.006(3)	7.75 +/- 1.36	0.007
	Ir-C (Cp*, 5C)	5	2.169	0.007(2)		
	Ir-Cl	1	2.403	0.008(3)		
$\text{Ir}^{\text{III}}\text{Cp}^*\text{C}_0\text{CB}$	Ir-N (PyT, 2N)	2	2.091	0.009(6)	9.27 +/- 1.04	0.011
	Ir-C (Cp*, 5C)	5	2.163	0.004(1)		
	Ir-Cl	1	2.396	0.003(1)		

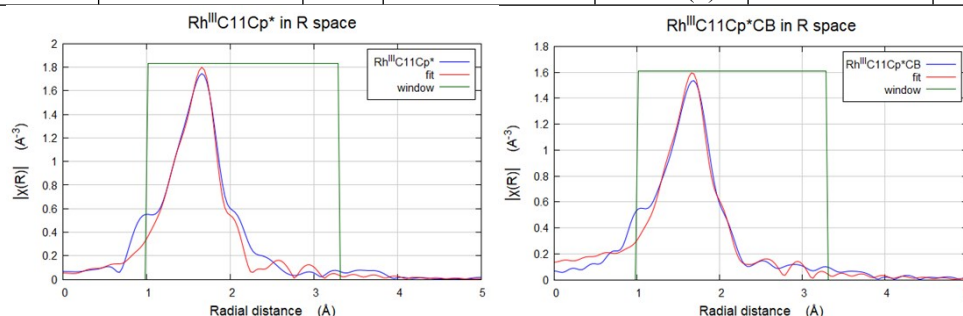


Figure S25. Fourier transformed EXAFS data (k^2 weighted) for $\text{RhCp}^*\text{C}_{11}$ and $\text{RhCp}^*\text{C}_{11}\text{CB}$ showing the experimental data and obtained fit, and the applied fitting window.

Table S26. Summary of refined parameters for the first coordination shell from the fit for $\text{Rh}^{\text{III}}\text{Cp}^*\text{C}_{11}$ and $\text{Rh}^{\text{III}}\text{Cp}^*\text{C}_{11}\text{CB}$ involving the major scattering paths.

Compound	Bond	N	Bond length [$\text{\AA}$]	σ^2	E_0 [eV]	R-factor
$\text{Rh}^{\text{III}}\text{Cp}^*\text{C}_{11}$	Rh-N (PyT, 2N)	2	2.093	0.004 ^a	-9.14 +/- 1.69	0.020
	Rh -C (Cp*, 5C)	5	2.153	0.003(1)		
	Rh-Cl	1	2.377	0.003(2)		
$\text{Rh}^{\text{III}}\text{Cp}^*\text{C}_{11}\text{CB}$	Rh-N (PyT, 2N)	2	2.094	0.003 ^a	-0.33 +/- 1.26	0.018
	Rh-C (Cp*, 5C)	5	2.154	0.003(1)		
	Rh-Cl	1	2.378	0.002(1)		

^a These were set to the reported value for the refinement

Regarding the series of homogeneous Rh^I complexes, Rh^IC_n, all fits contained significant portions of misfit in the range of ~ 1-1.3 Å, which is attributed to a distinct peak at ~ 1.1 Å for all samples which appears below the first coordination shell. It is very unlikely that this low-R value represents a real bond distance as it is significantly below the first coordination shell. Thus, we constrained the fitting window to exclude this non-physical interatomic distance, which resulted in a more reasonable result at higher radial distance albeit still high R-values.

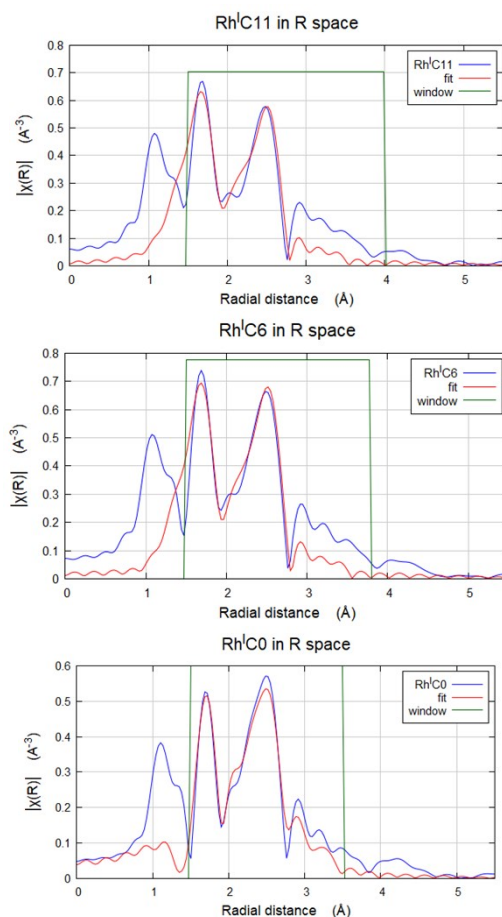


Figure S26. Fourier transformed EXAFS data (k^2 weighted) for $\text{Rh}^{\text{I}}\text{C}_n$ ($n = 11, 6, 0$) showing the experimental data and obtained fit, and the applied fitting window.

Table S27. Summary of refined parameters for the first coordination shell from the fit for $\text{Rh}^{\text{I}}\text{C}_n$ ($n = 11, 6, 0$) involving the major scattering paths.

Compound	Bond	N	Bond length [$\text{\AA}$]	σ^2	E_0 [eV]	R-factor
Rh^IC₁₁	Rh-N (PyT, 2N)	2	2.097	0.045(38)	-0.50 +/- 2.63	0.097
	Rh-C (CO, 2C)	2	1.888	0.001(3)		
Rh^IC₆	Rh-N (PyT, 2N)	2	2.099	0.039(31)	0.87 +/- 24.75	0.081
	Rh-C (CO, 2C)	2	1.890	0.003(2)		
Rh^IC₀	Rh-N (PyT, 2N)	2	2.089	0.009(7)	-3.38 +/- 2.82	0.056
	Rh-C (CO, 2C)	2	1.880	0.002(1)		

Lastly, we consider the hybrid catalysts $\text{Rh}^{\text{III}}\text{C}_n\text{CB}$. Based on our experimental results of the control experiments, CO ligands are absent in the hybrid catalysts. The samples are composed of two PyT-ligands and two co-ligands (Cl, NO_2 , NO_3 or OH) coordinated to one Rh. We have conducted a first shell fitting for the samples, using only the first shell N scattering path with a set amplitude factor of 1.2 for **Rh^{III}C₁₁CB** and 1.4 for **Rh^{III}C₆CB** and **Rh^{III}C₀CB**, and the

following guess starting values: $\sigma^2 = 0.003$, $E_0 = 0$, $\text{delr} = 0$ in order to calculate the coordination number. This model ignores the co-ligands X_1/X_2 and only focuses on the first shell possibly explaining the high R factors, while the obtained coordination numbers are 6, as expected.

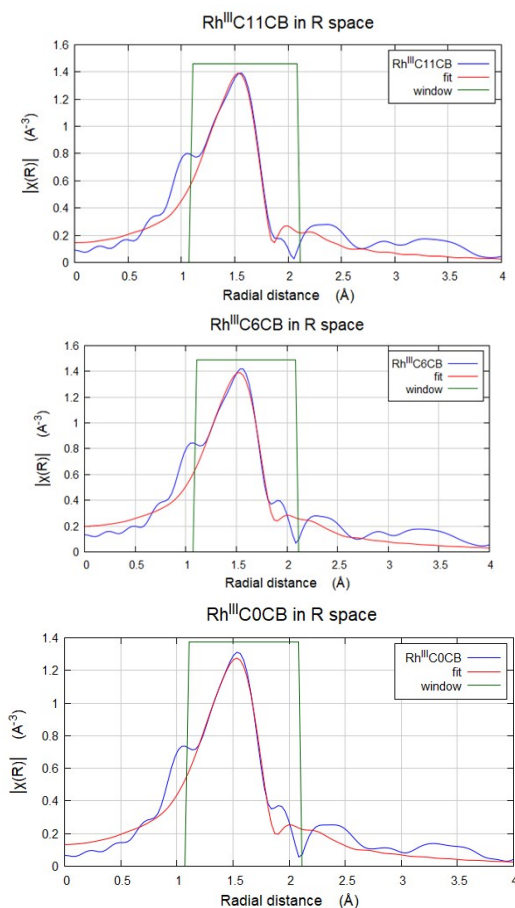


Figure S27. Fourier transformed EXAFS data (k^2 weighted) for $\text{Rh}^{\text{III}}\text{C}_n\text{CB}$ ($n = 11, 6, 0$) showing the experimental data and obtained fit, and the applied fitting window.

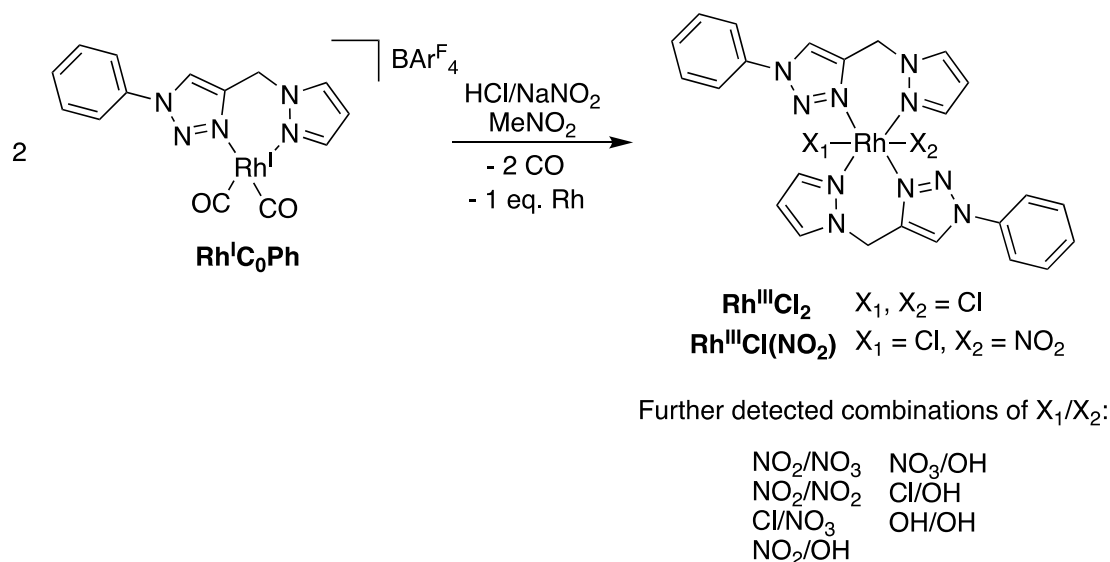
Table S28. Summary of refined parameters for the first coordination shell from the fit for $\text{Rh}^{\text{III}}\text{C}_n\text{CB}$ ($n = 11, 6, 0$) involving the major scattering paths.

Compound	Bond	Coordination Number	Bond length [Å]	σ^2	E_0 [eV]	S_0^2	R-factor
$\text{Rh}^{\text{III}}\text{C}_{11}\text{CB}$	Rh-N (PyT, 2N)	6.30 +/- 0.81	2.035	0.006(2)	-2.93 +/- 2.04	1.2	0.023
$\text{Rh}^{\text{III}}\text{C}_6\text{CB}$	Rh-N (PyT, 2N)	6.31 +/- 0.97	2.040	0.008(2)	-1.00 +/- 2.25	1.4	0.040
$\text{Rh}^{\text{III}}\text{C}_0\text{CB}$	Rh-N (PyT, 2N)	5.8 +/- 0.91	2.093	0.007(2)	-2.70 +/- 2.35	1.4	0.041

4. Control Experiments

Rh^IC₀, **Rh^IC₁₁** and **Rh^IC₀Ph**, prepared analogously to the **Rh^IC_n** according to the literature⁸, were treated with HCl/NaNO₂ in MeNO₂ under absence of CB. The reaction is shown for

RhC₀Ph in Scheme S2.



Scheme S2. Treatment of the **Rh^IC₀Ph** compound with HCl/NaNO₂ in MeNO₂ under absence of CB.

Exemplary procedure for **Rh^IC₀Ph**: The title compound (20 mg, 0.16 mmol) was dissolved in nitromethane (4 mL) in a Schlenk flask and cooled to 0 °C. HCl (0.2 mL) was added dropwise, which was followed by addition of NaNO₂ (2.2 mg, 0.32 mmol, 2 eq.). The mixture was stirred at room temperature for 16 h. The organic layer phase for separated using a pipette and the solvent was evaporated to dryness using a rotary evaporator, which was followed by drying under vacuum. The obtained light brown solids were taken up in DCM and filtered through celite. Solvent evaporation and drying under vacuum gave the Rh^{III} based material (21 mg) as light brown solid.

Both **Rh^IC₀** and **Rh^IC₁₁** were treated analogously. The obtained products are poorly soluble in DCM. Thus, the no filtration through celite was carried out.

IR spectra showed absence of the CO vibrational bands for all three examples (Figures 24-26). For **Rh^IC₀Ph**, we stopped the reaction as well at time points of 1.5 h and 4 h and note that some starting material is still present at these time points, and that the ¹H NMR spectra gain complexity over time (Figure S24) while the IR spectra did not show major changes apart from low intensity CO vibrations at 1.5 h and 4 h due to remaining **Rh^IC₀Ph**, which is absent after 16 h reaction time (Figure S25).

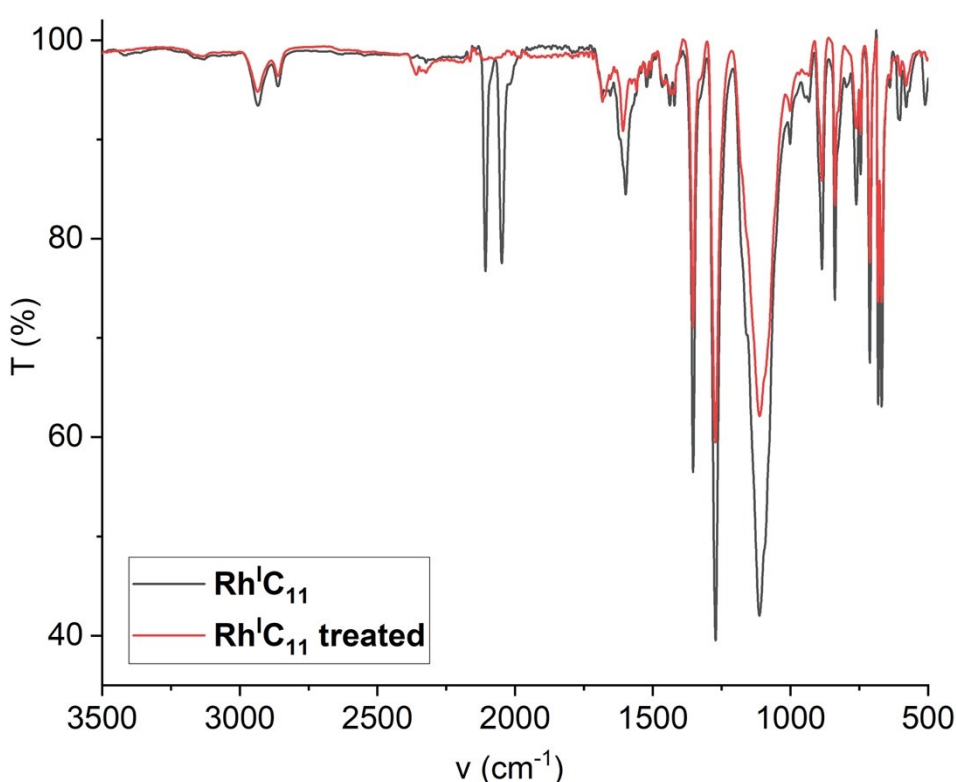


Figure S28. IR spectrum of **Rh^IC₁₁** before and after treatment with HCl/NaNO₂ in MeNO₂.

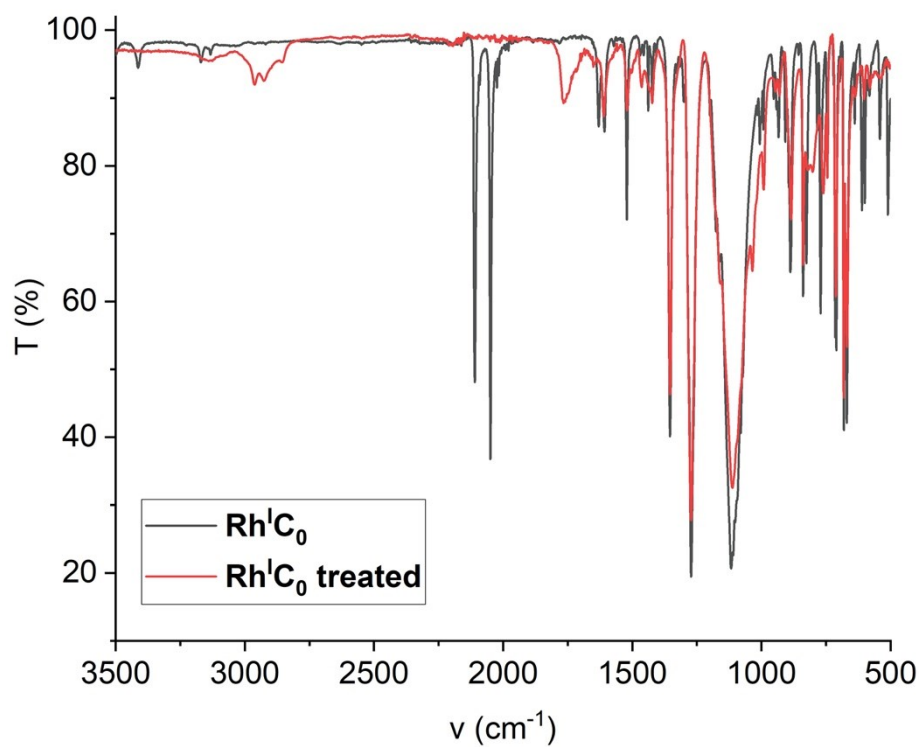


Figure S29. IR spectrum of $\text{Rh}^{\text{I}}\text{C}_0$ before and after treatment with HCl/NaNO_2 in MeNO_2 .

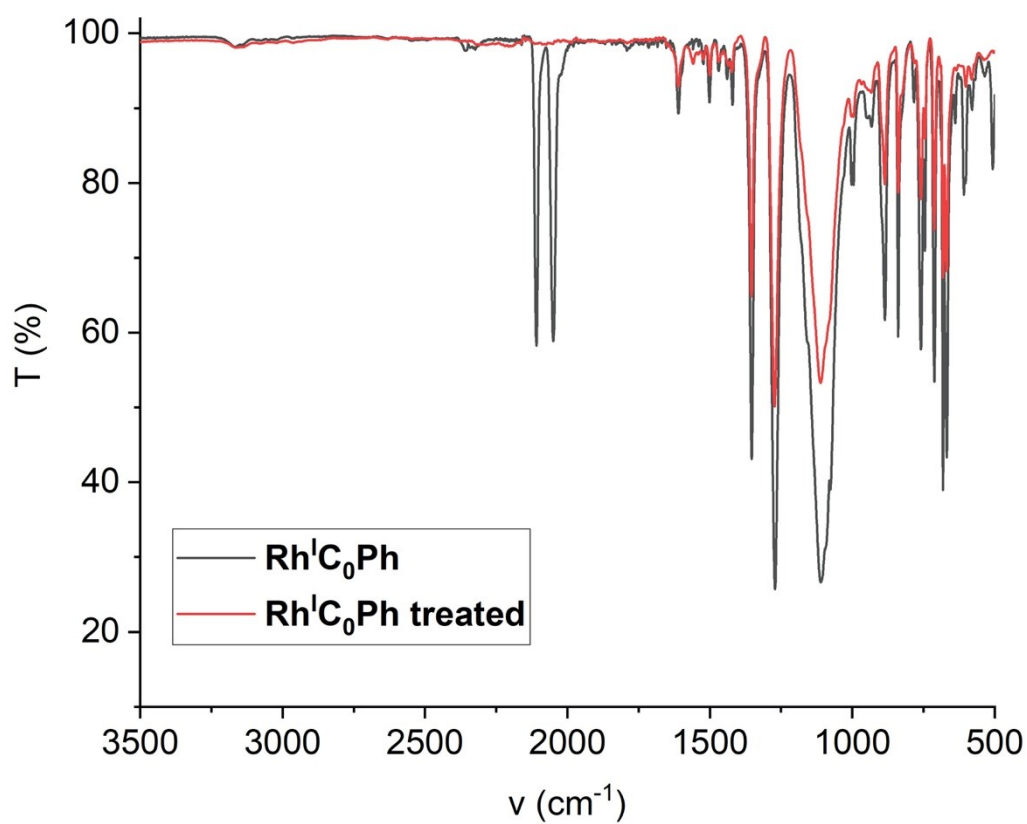


Figure S30. IR spectrum of $\text{Rh}^{\text{I}}\text{C}_0\text{Ph}$ before and after treatment with HCl/NaNO_2 in MeNO_2 .

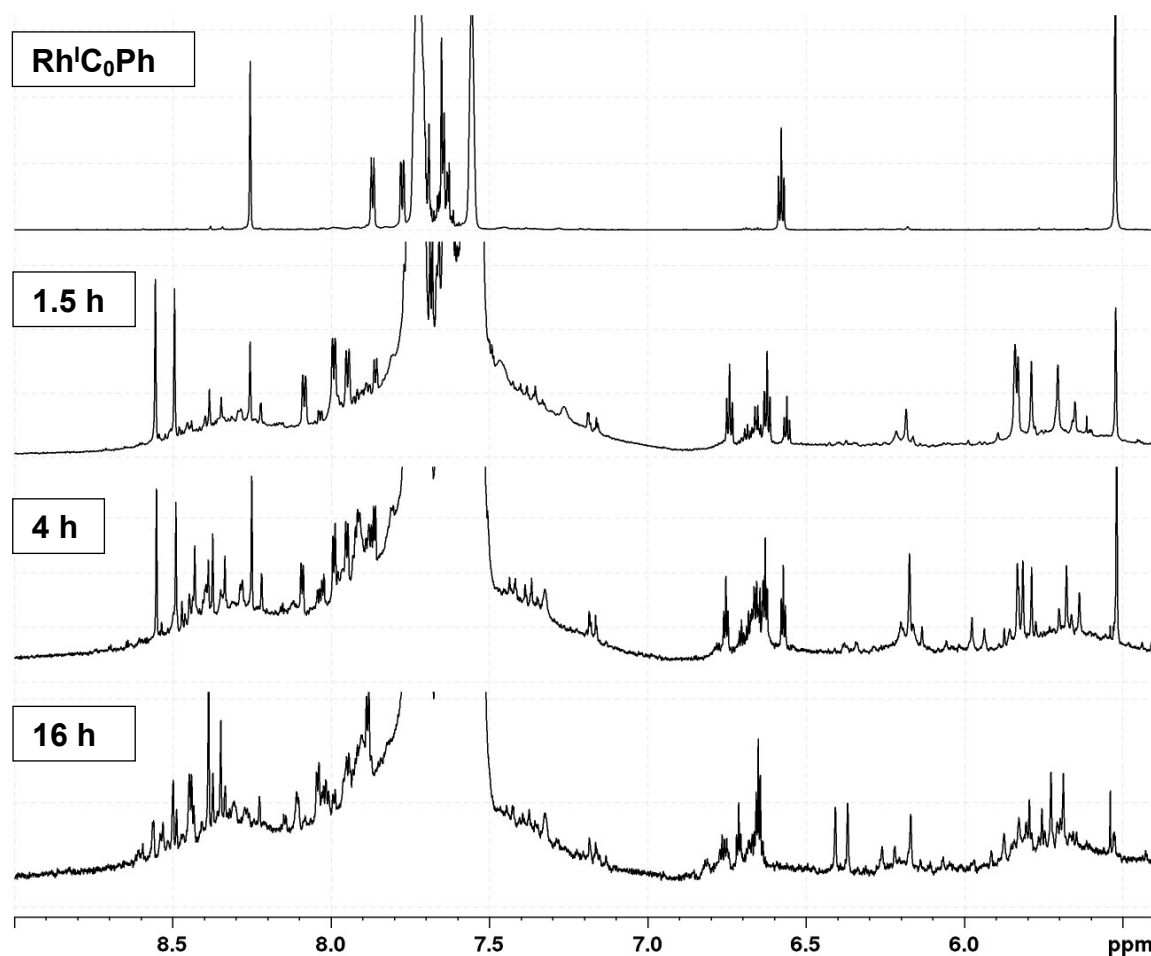


Figure S31. Selected ranges of ^1H NMR spectra of the parent **Rh^IC₀Ph** and after treatment with HCl/NaNO₂ after 1.5 h, 4 h and 16 h.

Analysis by MS/HRMS (ESI⁺) of the material obtained from treatment of **Rh^IC₀Ph** showed a number of fragments, all bearing two PyT-ligands per Rh atom, i.e. Rh(PyT)₂ as described in the main manuscript. The spectra are shown in Figures S28-S36.

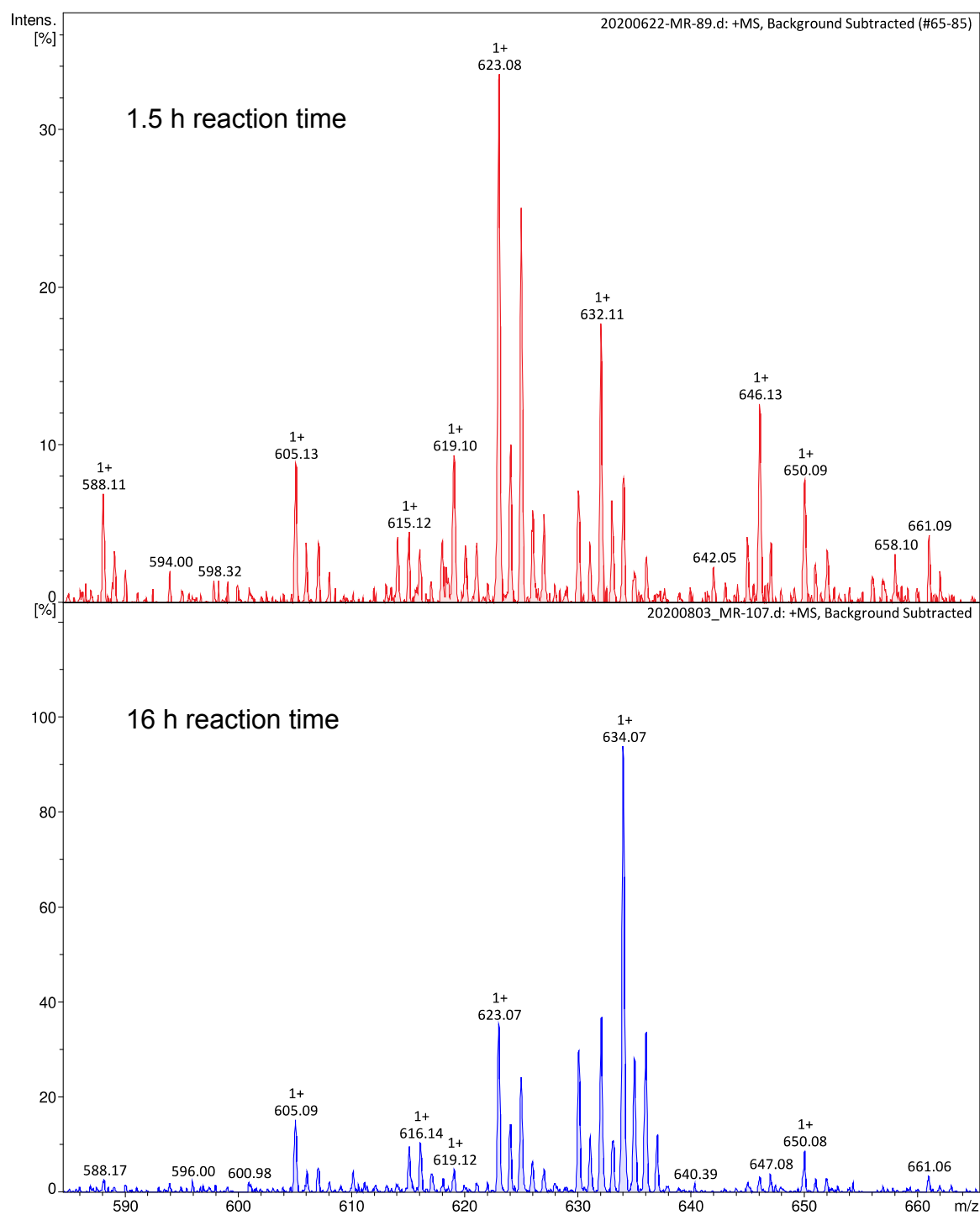


Figure S33. ESI⁺ mass spectra in range of ~ 580-670 Da of the material obtained from treatment of $\text{Rh}^1\text{C}_0\text{Ph}$ with HCl/NaNO_2 after work-up at 1.5 h and 16 h.

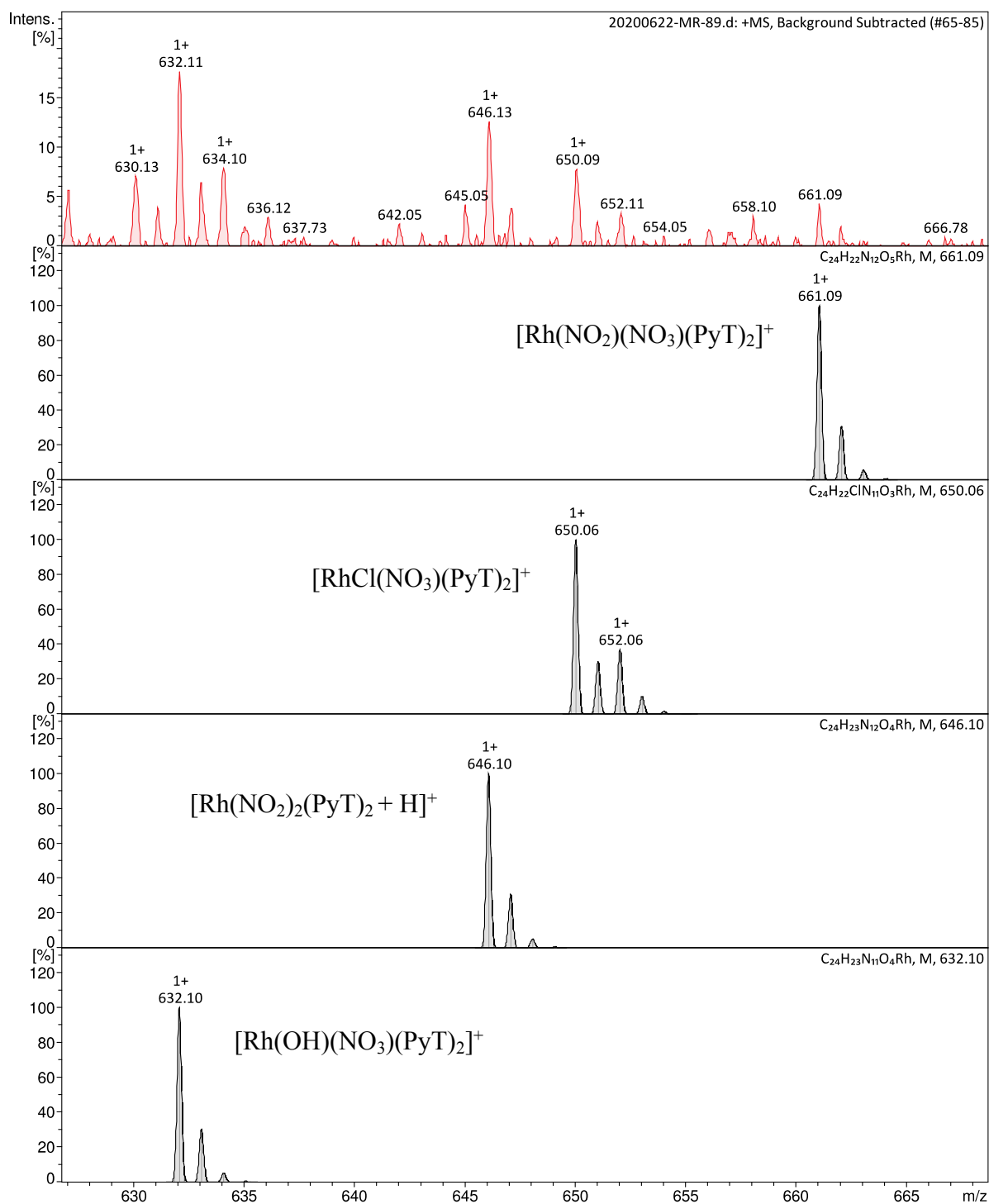


Figure S34. Experimental and simulated mass patterns for the materials obtained from treatment of $\text{Rh}^{\text{I}}\text{C}_0\text{Ph}$ with HCl/NaNO_2 after work-up at 1.5 h.

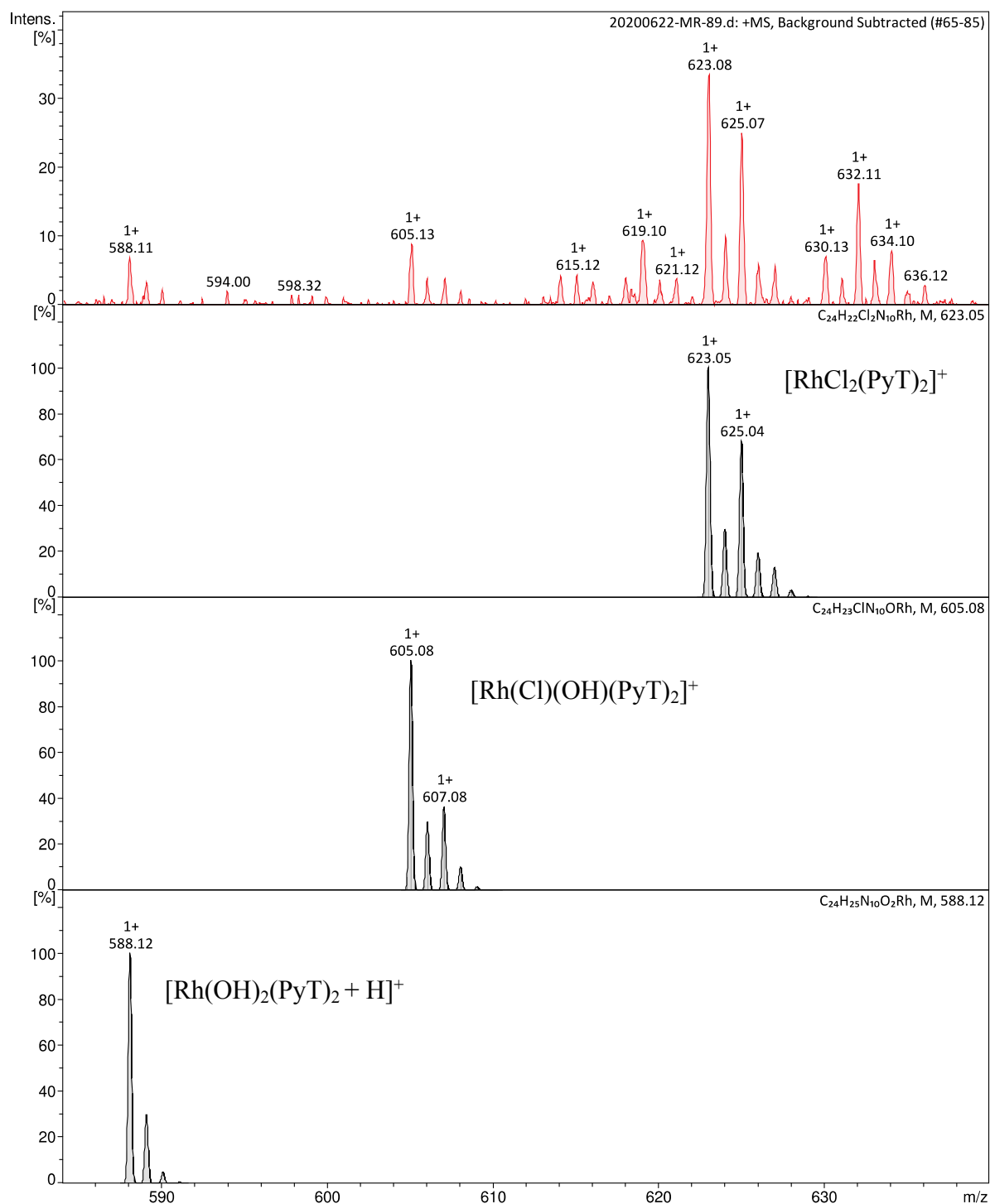


Figure S35. Experimental and simulated mass patterns for the materials obtained from treatment of $Rh^I C_0Ph$ with $HCl/NaNO_2$ after work-up at 1.5 h.

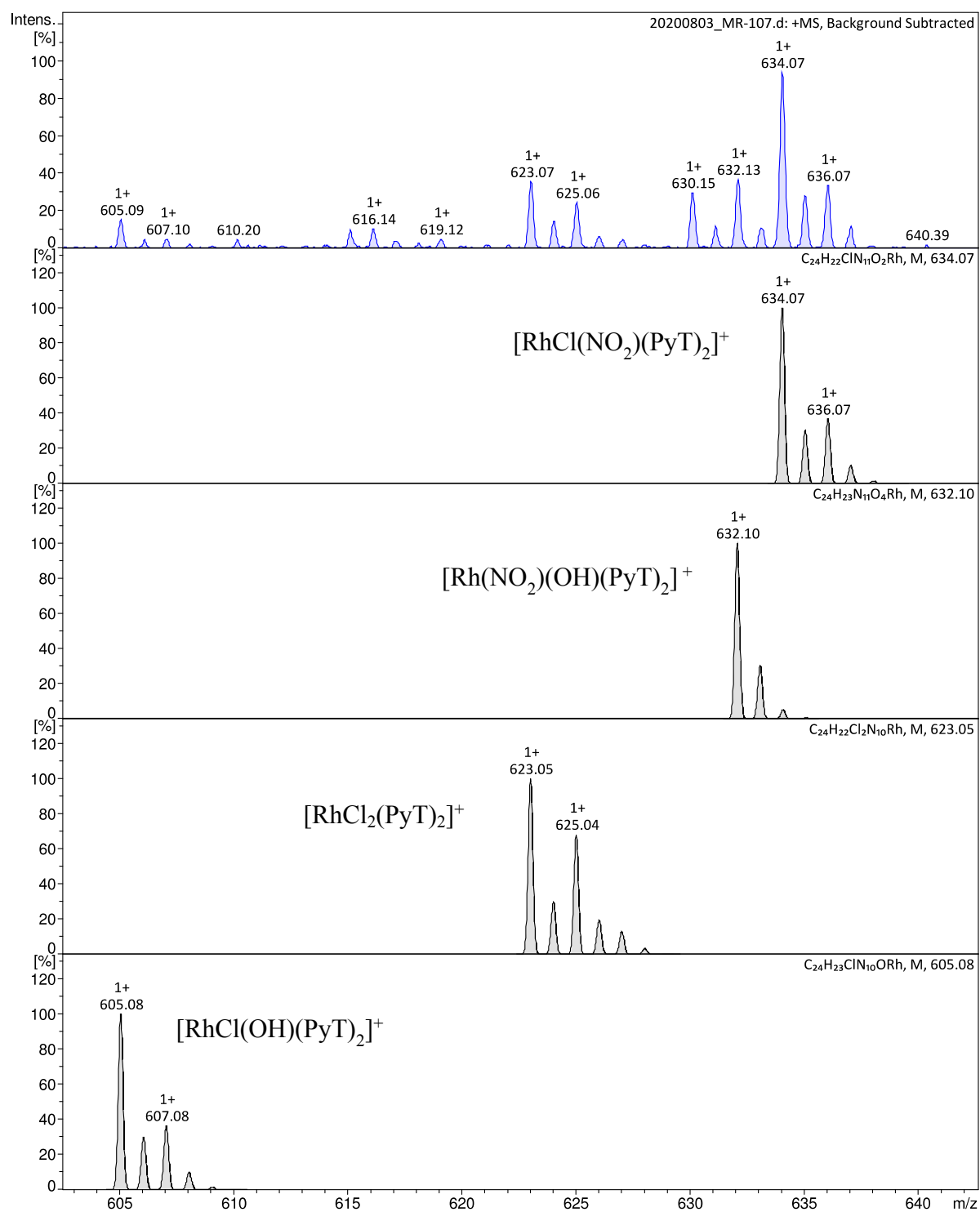


Figure S36. Experimental and simulated mass patterns for the materials obtained from treatment of $\text{Rh}^{\text{I}}\text{C}_0\text{Ph}$ with HCl/NaNO_2 after work-up at 16 h.

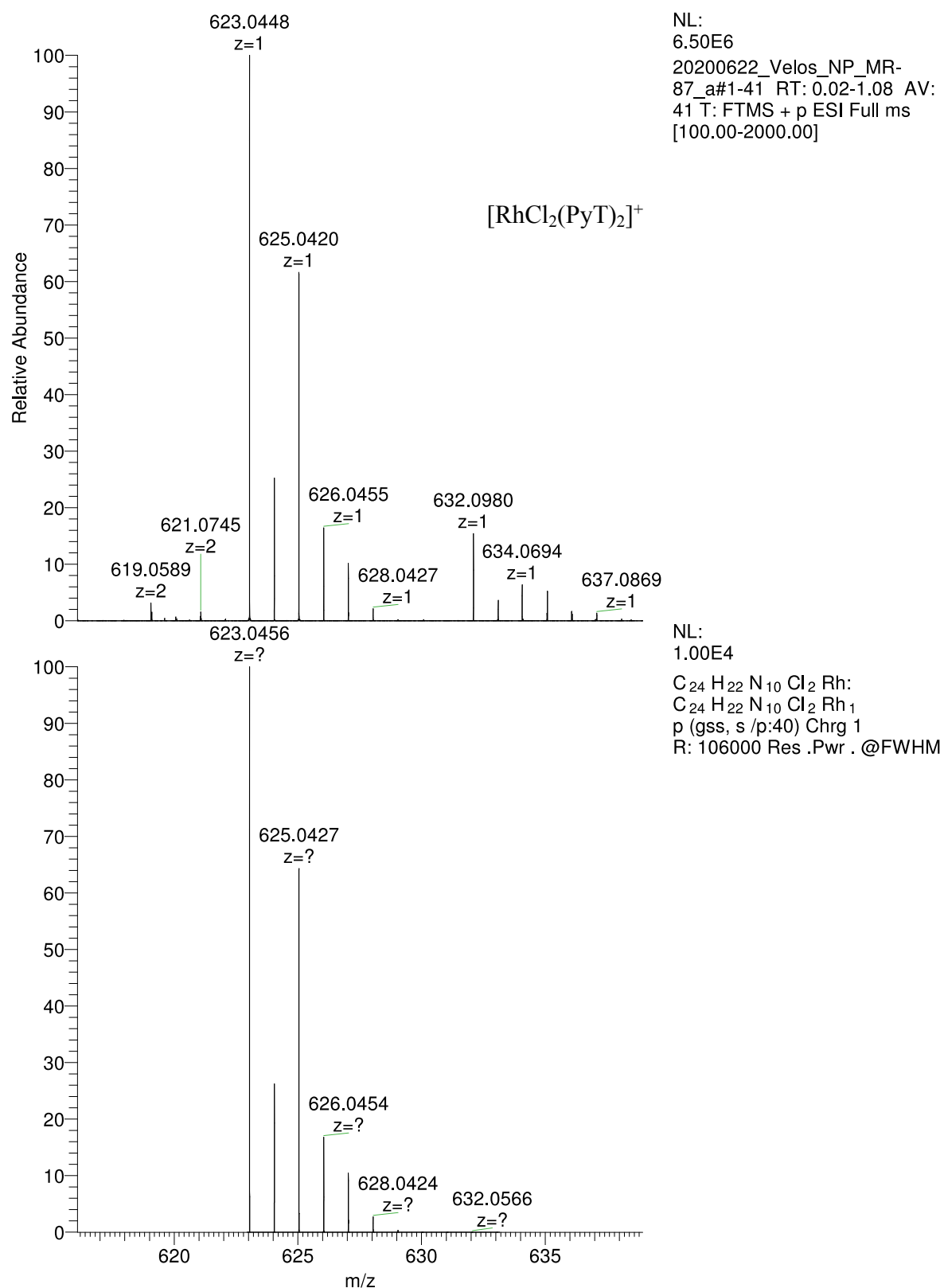


Figure S37. Experimental (top) and calculated (bottom) high resolution mass spectrum of the peak assigned to $[(\text{PhPyT})_2\text{RhCl}_2]^+$.

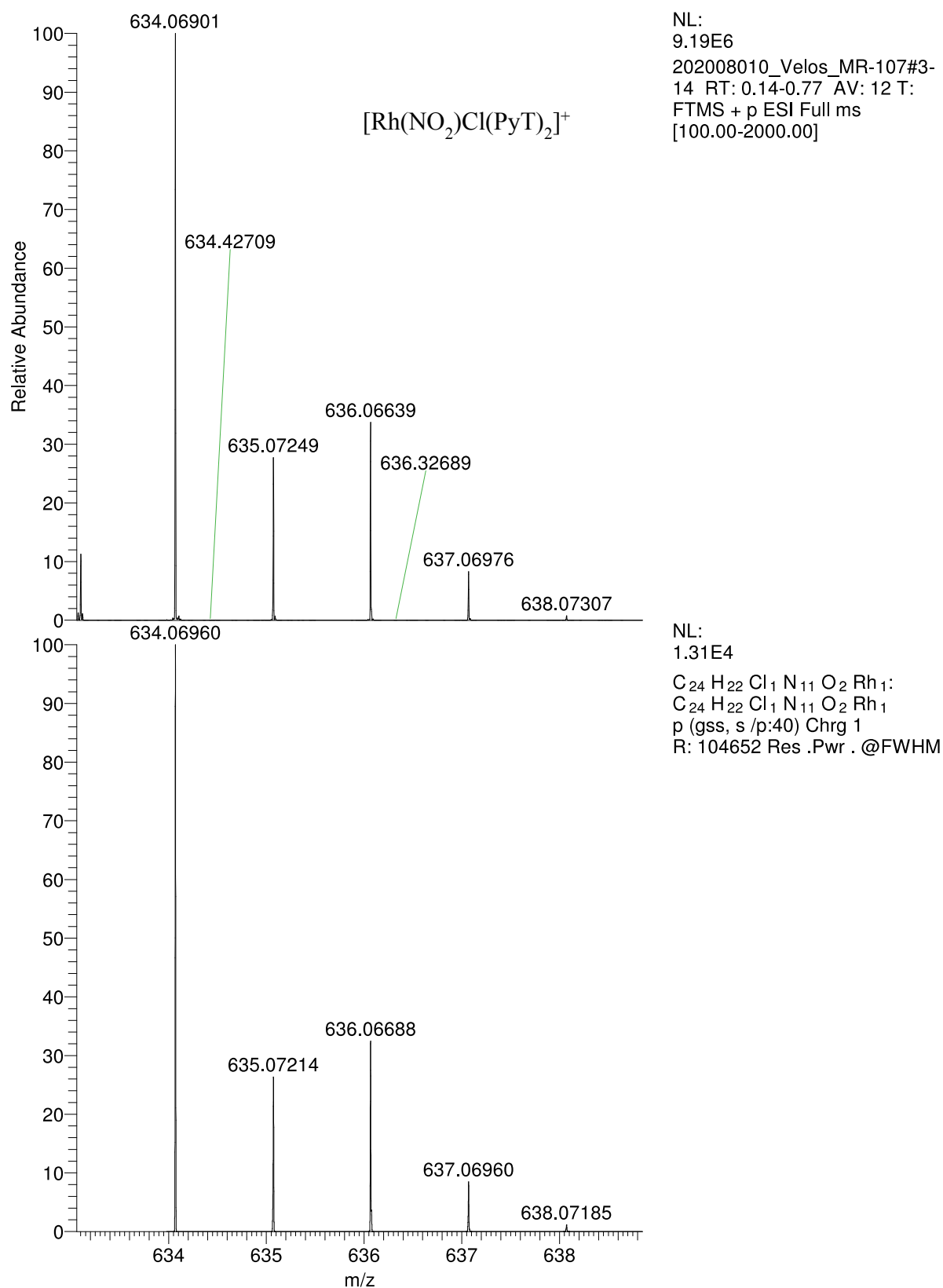


Figure S38. Experimental (top) and calculated (bottom) high resolution mass spectrum of the peak assigned to $[\text{Rh}(\text{NO}_2)\text{Cl}(\text{PyT})_2]^+$.

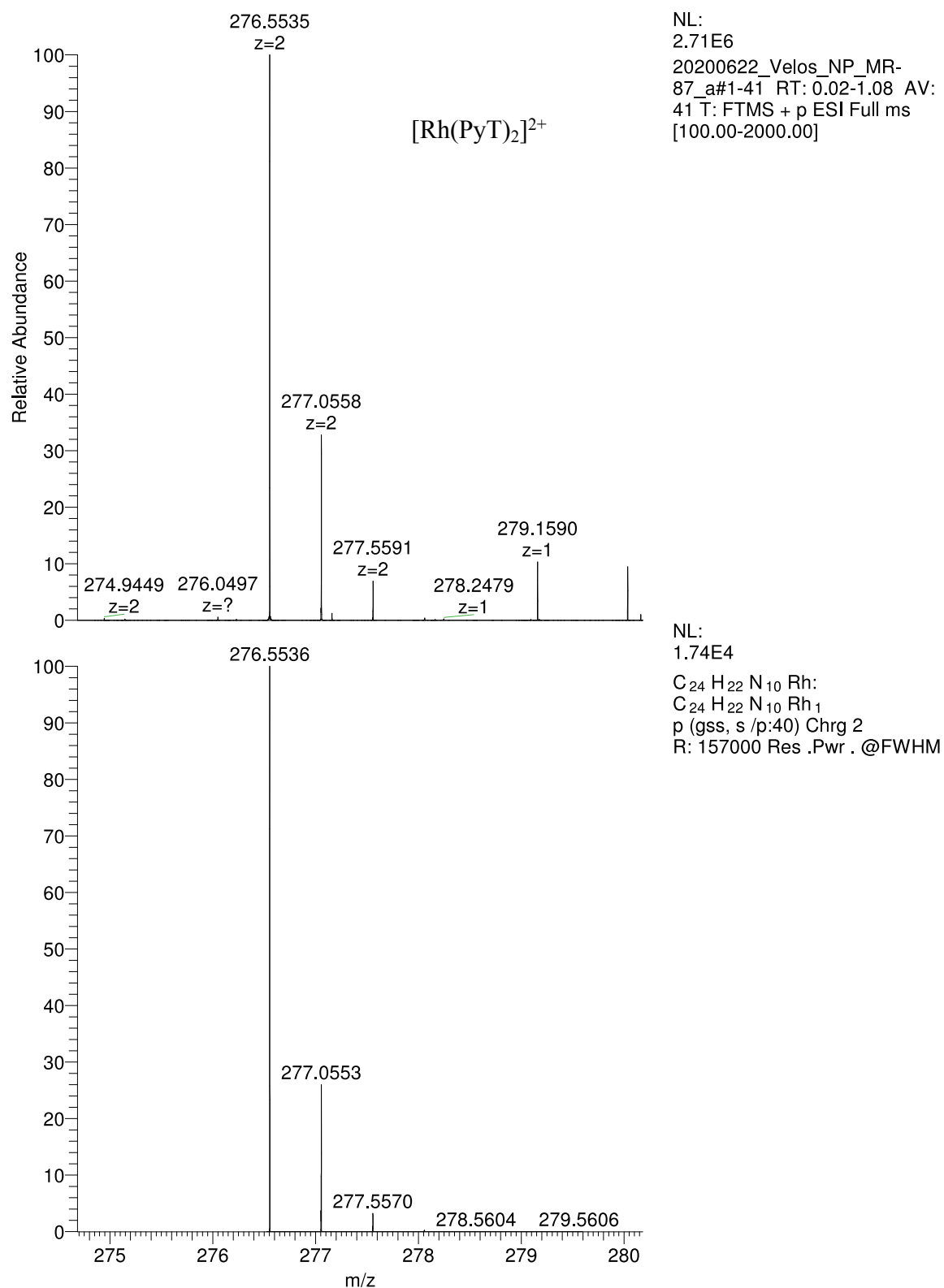


Figure S39. Experimental (top) and calculated (bottom) high resolution mass spectrum of the peak assigned to $[\text{Rh}(\text{PyT})_2]^{2+}$.

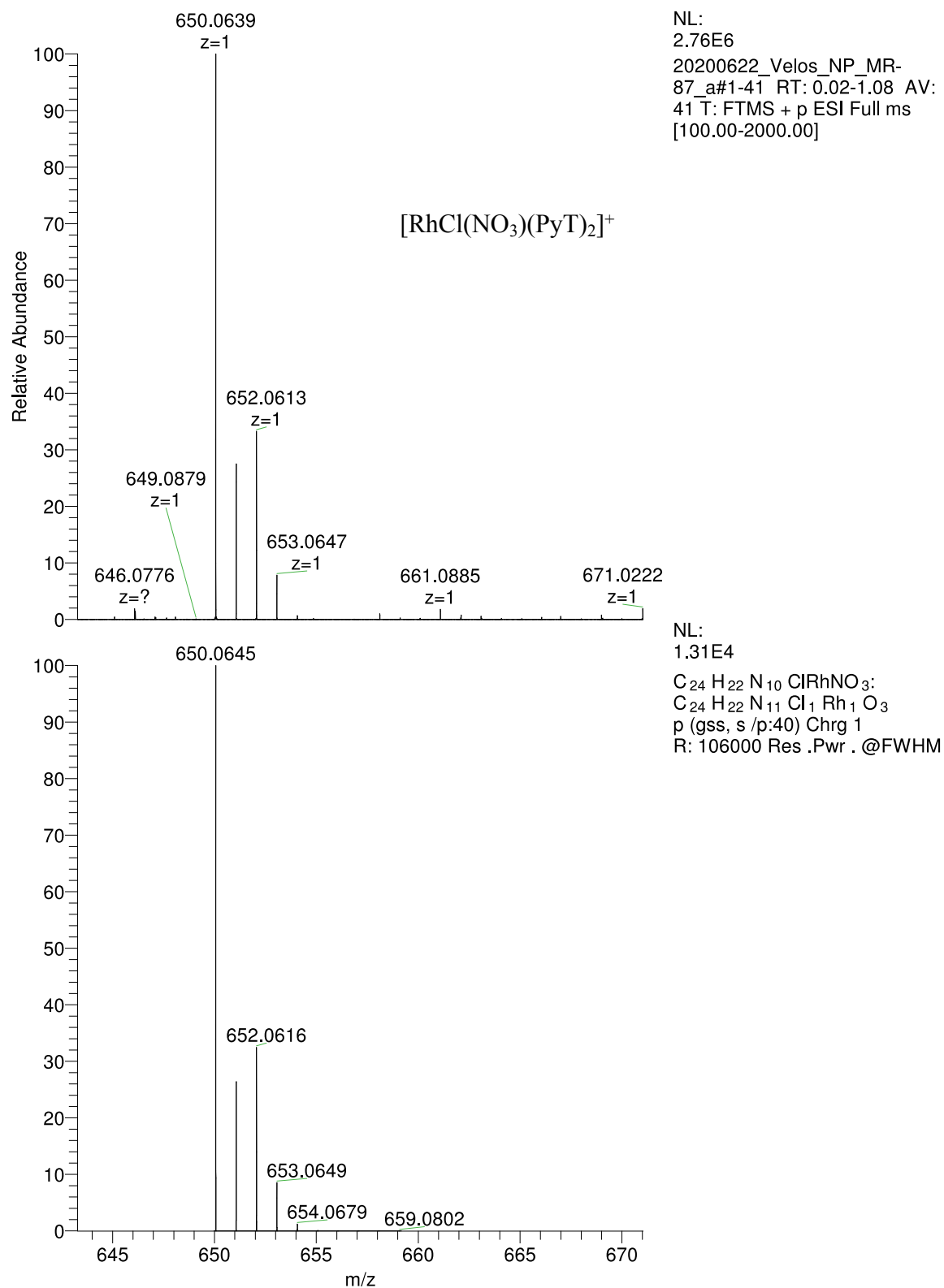


Figure S40. Experimental (top) and calculated (bottom) high resolution mass spectrum of the peak assigned to $[\text{Rh}(\text{NO}_3)\text{Cl}(\text{PyT})_2]^+$.

5. Thermogravimetric Analyses

To test thermal stability of the hybrid catalysts, we conducted thermogravimetric measurements. Figure 37 shows weight loss as function of temperature for the hybrid catalysts. Experiments were conducted on a TA Instruments Discovery Thermogravimetric analyzer. Approximately 0.5 mg of the hybrid catalyst was placed into a flame cleaned Pt pan and placed into the analyser. Weight loss was monitored while ramping up the temperature from room temperature to 700 °C with a heating rate of 20°C/min. The obtained TGA traces for the hybrid catalysts and a control sample of CB are shown in Figure S37. Up to ~ 300 °C the curves are essentially featureless with the $\text{Rh}^{\text{III}}\text{C}_0\text{CB}$ sample showing a slightly higher weight loss than all other examples. At ~ 380 °C, all catalysts exhibit a first process, which is followed for all $\text{M}^{\text{III}}\text{Cp}^*\text{C}_n\text{CB}$ examples by another at ~ 490 °C while the $\text{Rh}^{\text{III}}\text{C}_n\text{CB}$ hybrid catalysts exhibit a more gradual weight decrease.

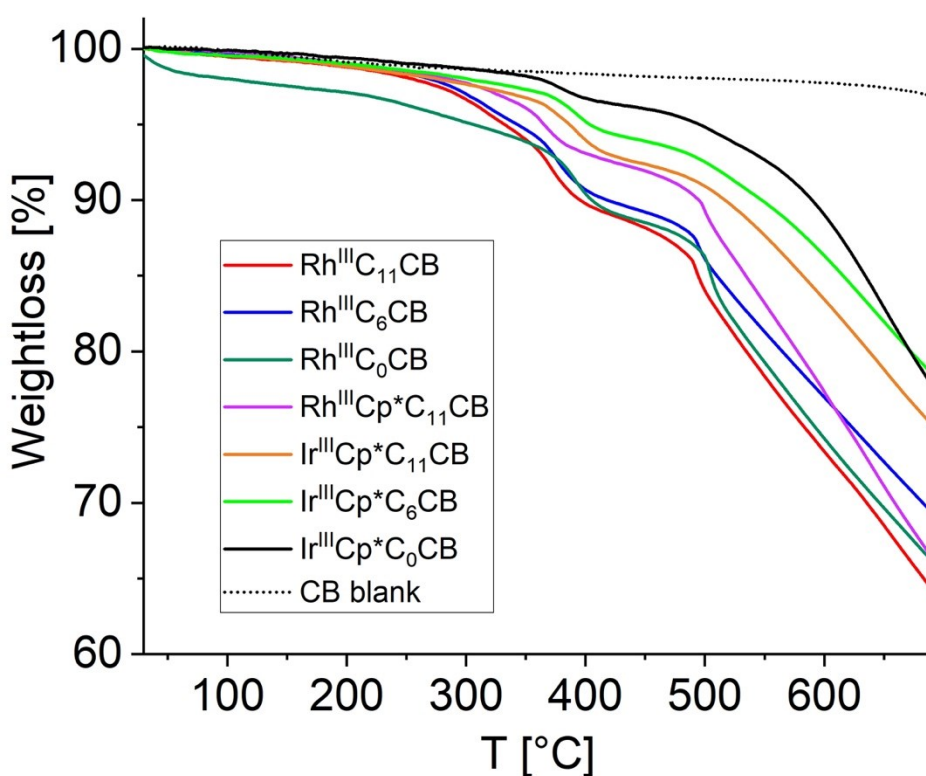


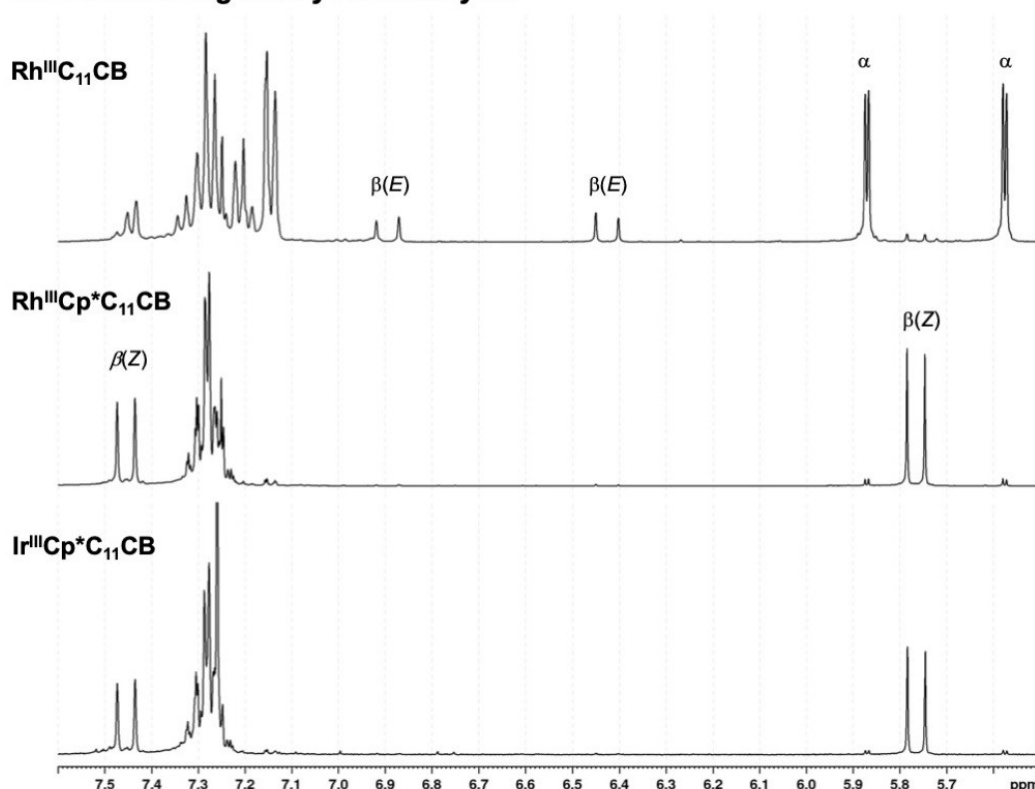
Figure S41. Thermogravimetric analysis of the hybrid catalysts and a control sample of CB treated under grafting conditions.

6. Catalysis Experiments

6.1 Hydrosilylation of Phenylacetylene

A flame dried 4 mL Schlenk tube, equipped with a PTFE stopcock and magnetic stirring bar, was charged with the hybrid catalyst and THF (1 mL). Phenylacetylene (9 mg, 0.09 mmol) and triethylsilane (18 mg, 0.16 mmol) were added. The mixture was heated while stirring for the specified amount of time using an oil bath. The headspace was fully submerged in the oil. After the reaction was completed, the mixture was allowed to cool down to room temperature and transferred into a V-shaped centrifuge vial and centrifuged at 4000 rpm for 20 min. The solution was removed and concentrated to dryness using rotary evaporator and briefly dried under high vacuum. The product mixture was re-dissolved in chloroform-*d*, filtered through a

Reactions using the Hybrid Catalysts



PTFE-syringe filter (45 μm) and analyzed by ¹H NMR spectroscopy.

Figure S42. Selected ranges of the ^1H NMR spectra of the reaction mixtures from hydrosilylation of phenylacetylene using the hybrid catalysts $\text{Rh}^{\text{III}}\text{C}_{11}\text{CB}$, $\text{Rh}^{\text{III}}\text{Cp}^*\text{C}_{11}\text{CB}$ and $\text{Ir}^{\text{III}}\text{Cp}^*\text{C}_{11}\text{CB}$.

The products were identified by comparison of the vinylic ^1H NMR resonances of the $\beta(\text{E})$ -, $\beta(\text{Z})$ - and α -isomers to the literature.¹⁰ Conversions were calculated by integration of combined vinylic proton resonances of all isomers and comparison to the proton signal of unreacted terminal alkyne.

Table S29. Summary of the hydrosilylation reactions of phenylacetylene using the different Rh- and Ir-catalysts.

No.	Catalyst	Metal [mol%] ^a	Loading [wt%] ^b	Et_3SiH [eq.]	Solvent	T [°C]	Time [min]	Conversion [%]	Ratio $\beta(\text{E}) : \beta(\text{Z}) : \alpha$
<i>Heterogeneous reactions</i>									
1	$\text{Rh}^{\text{III}}\text{C}_{11}\text{CB}$	1.1	3.9	4.8	THF	75	15	97	16 : 4 : 80
2	$\text{Rh}^{\text{III}}\text{C}_{11}\text{CB}$	1.3	3.9	4.8	THF	70	15	97	13 : 3 : 84
3	$\text{Rh}^{\text{III}}\text{C}_{11}\text{CB}$	2.0	3.9	3.5	THF	60	60	100	12 : 4 : 84
4	$\text{Rh}^{\text{III}}\text{C}_{11}\text{CB}$	1.3	3.9	4.8	THF	60	60	100	15 : 6 : 80
5	$\text{Rh}^{\text{III}}\text{C}_{11}\text{CB}$	1.3	3.9	1.8	THF	50	60	99	11 : 7 : 82
6	$\text{Rh}^{\text{III}}\text{C}_{11}\text{CB}$	1.1	3.9	1.8	DCM	40	30	24	13 : 19 : 68
7	$\text{Rh}^{\text{III}}\text{C}_{11}\text{CB}$	1.1	3.9	1.8	DCM	40	120	95	6 : 15 : 79
8	$\text{Rh}^{\text{III}}\text{C}_{11}\text{CB}$	1.0	3.9	1.8	THF	60	60	96	11 : 4 : 85
9	$\text{Rh}^{\text{III}}\text{C}_6\text{CB}$	1.0	4.1	1.8	THF	60	60	95	11 : 4 : 85
10	$\text{Rh}^{\text{III}}\text{C}_0\text{CB}$	1.0	6.4	1.8	THF	60	60	56	10 : 8 : 82
11	$\text{Rh}^{\text{III}}\text{C}_6\text{CB}$	1.4	4.1	1.8	THF	60	60	85	11 : 5 : 84
12	$\text{Rh}^{\text{III}}\text{C}_0\text{CB}$	2.1	6.4	1.8	THF	60	60	95	10 : 7 : 83
13	$\text{Ir}^{\text{III}}\text{Cp}^*\text{C}_{11}\text{CB}$	0.4	2.1	4.8	THF	70	60	60	8 : 72 : 20
14	$\text{Ir}^{\text{III}}\text{Cp}^*\text{C}_{11}\text{CB}$	0.5	2.1	1.8	THF	60	60	98	2 : 95 : 3
15	$\text{Ir}^{\text{III}}\text{Cp}^*\text{C}_6\text{CB}$	0.5	1.9	1.8	THF	60	60	60	2 : 95 : 3
16	$\text{Ir}^{\text{III}}\text{Cp}^*\text{C}_0\text{CB}$	0.5	1.3	1.8	THF	60	60	85	2 : 95 : 3
17	$\text{Ir}^{\text{III}}\text{Cp}^*\text{C}_6\text{CB}$	0.3	1.9	1.8	THF	70	60	98	2 : 94 : 4
18	$\text{Ir}^{\text{III}}\text{Cp}^*\text{C}_0\text{CB}$	0.2	1.3	1.8	THF	70	60	97	3 : 94 : 3
19	$\text{Rh}^{\text{III}}\text{Cp}^*\text{C}_{11}\text{CB}$	0.7	2.0	1.8	THF	70	60	99	1 : 94 : 4
<i>Control</i>									
20 ^c	-	1.0	-	-	THF	70	180	0	-

Reaction conditions: 1 mL solvent in sealed Schlenk tube under Ar on a 0.1 mmol scale.

^a Calculated molar amount of metal used in the reaction.

^b Determined by EDX.

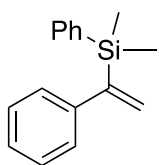
^c Control experiment under absence of catalyst. CB (3 mg) treated under the applied grafting conditions was added to the reaction.

Preparative Reaction on a ~ 0.1 g Scale

A glass ampule equipped with PTFE stop cock and magnetic stirring bar was filled with **Rh^{III}C₁₁CB** hybrid catalyst (23 mg, ~ 1 mol%) and THF (10 mL). Phenylacetylene (90 mg, 0.9 mmol) and triethylsilane (183 mg, 15.8 mmol) were added, the ampule was sealed and placed into an oil bath with the sealed volume fully submerged up to the stop cock and heated to 60 °C for 1h. After that, the mixture was cooled, transferred into two 5 ml glass centrifuge tubes, and centrifuged for 20 min at 4000 rpm. The solution was taken up with a syringe and filtered through a PTFE syringe filter (45 µm) into a round bottom flask and concentrated using a rotary evaporator. The remaining hybrid catalyst was washed once with pentane (8 mL) and centrifuged for 5 min at 4000 rpm. The pentane was also filtered through the syringe filter, combined with the content of the round bottom flask, concentrated again using a rotary evaporator and dried under high vacuum ($\sim 5.0 \times 10^{-1}$ mbar). Subsequently, the mixture was flash-filtered through a silica column (10 cm $\times$ 1 cm) eluting with pentane, giving after solvent evaporation and drying under vacuum the isomeric product mixture (176 mg, 0.8 mmol, 92 %). It contained mostly the α -isomer (88 %), minor amounts of the β (Z)-isomer (10 %), and small amounts of the β (E)-isomer (2 %), as determined by ¹H NMR spectroscopy (Fig. S111-114).

Reaction with Phenyldimethylsilane

Furthermore, we tested compatibility of **Rh^{III}C₁₁CB** for the hydrosilylation of phenylacetylene with phenyldimethylsilane. Under the same applied conditions using 1.8 eq. of dimethylphenylsilane, the reaction affords dimethyl(phenyl)(1-phenylvinyl)silane accompanied by the *E* isomer (ratio 84/16) with a conversion of 94%.



Dimethyl(phenyl)(1-phenylvinyl)silane: ¹H NMR (300 MHz, CDCl₃): δ = 7.59-7.55 (m, 2H), 7.39-7.36 (m, 3H), 7.28-7.19 (m, 3H), 7.15-7.12 (m, 2H), 6.01

(d, $^2J = 2.9$ Hz, 1H), 5.70 (d, $^2J = 2.9$ Hz, 1H), 0.44 (s, 6H, Si(CH₃)₂). The data is in agreement with the literature.¹¹

Recyclability

To test recyclability, the **Rh^{III}C₁₁CB** hybrid catalyst was recycled in several consecutive runs. Reactions were conducted using custom built glass centrifuge tubes equipped with PTFE taps (Figure S43) in a similar fashion as described above, but at 70 °C for 1 h, and using a catalyst loading of ~ 2 mol %.



Figure S43. Custom built centrifuge tube with PTFE tap.

After each run, the reaction mixture was centrifuged for 20 min at 4000 rpm, the solution was removed with a syringe equipped with a PTFE-syringe filter, concentrated to dryness and briefly dried under vacuum, followed by analysis with ¹H NMR spectroscopy. The remaining black solids were washed with pentane (1 mL), centrifuged at 4000 rpm for 1 min and the clear solution was discarded. This was repeated once before the catalyst was dried under high-vacuum and the vial was refilled with Ar. This was followed by addition of THF (1 mL), phenylacetylene and triethylsilane and the reaction was repeated.

A sample of hybrid catalyst was analyzed by XPS after five runs. The corresponding XP spectra are shown in Figure S44, and the determined elemental composition is reported in Table S30.

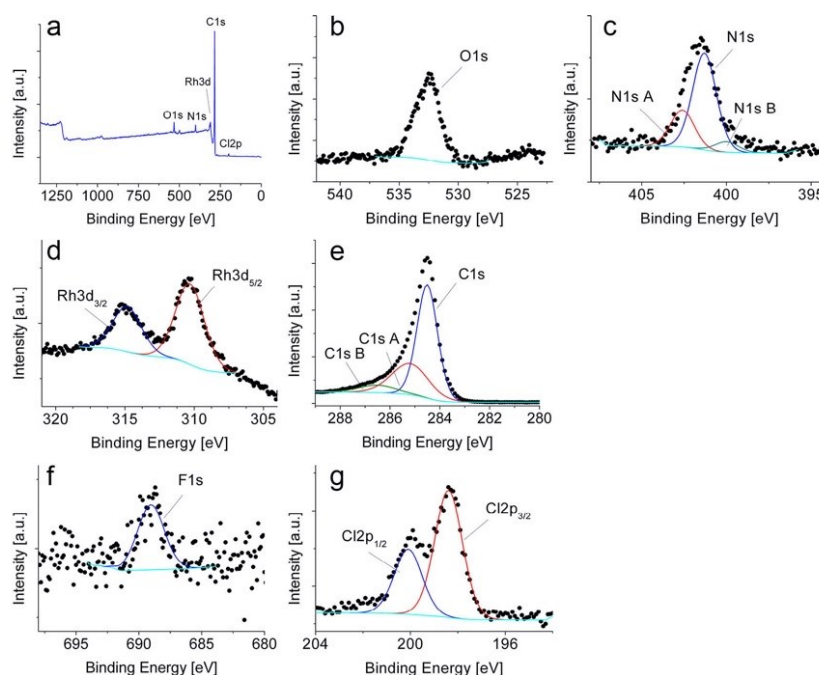


Figure S44. X-ray photoelectron spectra of the hybrid catalyst $\text{Rh}^{\text{III}}\text{C}_{11}\text{CB}$ recovered after five subsequent reaction runs. (a) Survey scan, narrow scans of the (b) O1s signal, (b) N1s signals, (d) C1s signal, (e) Rh3d_{3/2}/Rh3d_{5/2} signals and (f) F1s signal and (g) Cl 2p_{1/2}/2p_{3/2} signals.

Table S30. Determined elemental composition of $\text{Rh}^{\text{III}}\text{C}_{11}\text{CB}$, recovered after five subsequent runs, in at% by XPS. a) Calculated elemental composition with inclusion of F and Cl, b) Calculated elemental composition under absence of F and Cl.

Signal	Binding Energy [eV]	FWHM [eV]	Area [CPS × eV]	Atomic % <i>excluding F, Cl</i>	Atomic % <i>including F, Cl</i>
C1s	284.50	0.97	25546.79	56.48	55.84
C1s A	285.20	1.73	12526.38	27.70	27.38
C1s B	286.50	1.73	2992.59	6.62	6.54
O1s	532.44	2.35	5316.71	4.20	4.15
N1s	401.30	1.65	2088.59	2.79	2.76
N1s A	402.60	1.65	793.92	1.06	1.05
N1s B	400.00	1.65	226.44	0.30	0.30
Rh3d _{5/2}	310.30	2.45	3876.51	0.86	0.85
Rh3d _{3/2}	314.83	2.45	2172.66	0	0
F1s	689.04	2.51	643.47	-	0.38
Cl2p _{3/2}	198.39	1.38	631.47	-	0.75
Cl2p _{1/2}	200.09	1.38	324.10	-	0

The determined elemental composition indicates that only minor metal leaching has occurred. The Rh content was 0.86 at% (0.85 at%) for the recycled, compared to 0.96 at% (0.92%) for a freshly prepared sample. Furthermore, the change of the N/Rh ratio was only subtle, from 4.97

at% to 4.84 at%. This in turn suggests that small amounts of physisorbed material dissolve and are subsequently removed from the reaction mixture during the runs. Furthermore, the F signal decreased for the recycled sample which underwent five consecutive runs, from 1.69 at% to 0.38 at%, while Cl decreased from 1.22 at% to 0.75 at%. This may indicate also removal of some of the remaining counterions, $\text{BAr}^{\text{F}_4^-}$ and Cl^- . The weakly coordinating $\text{BAr}^{\text{F}_4^-}$ is more readily removed compared to the stronger coordinating Cl^- .

Test for Heterogeneity

We tested catalytic activity of a reaction solution by hot filtering a suspension of $\text{Rh}^{\text{III}}\text{C}_{11}\text{CB}$ and adding further substrate. The experiment was conducted as described in the following: A centrifuge tube with Schlenk tap was charged with $\text{Rh}^{\text{III}}\text{C}_{11}\text{CB}$ (2.3 mg, ~ 1.0 mol%), THF (1 mL), phenylacetylene (9 mg, 0.09 mmol) and triethylsilane (18 mg, 0.16 mmol) and sealed. The mixture was heated under stirring to 60 °C for 1 h in an oil bath, centrifuged at 4000 rpm for 20 min and the solution was taken up with a syringe and discarded. The black solids were washed twice with pentane (1 mL) followed by centrifugation for 1 min at 4000 rpm. The pentane washings were discarded, and the catalyst was dried under vacuum. THF (1 mL) was added and the suspension was stirred for 15 min at 60 °C, after which the tube was placed into a pre-heated metal centrifuge bucket (60 °C), briefly centrifuged at 4000 rpm and placed back into the hot oil bath. The suspension was quickly drawn up with a pre-heated syringe (60 °C), flushed with Ar and equipped with two pre-heated PTFE filters (0.45 μm) with a small of pad celite / cotton wool in-between the two filters. The obtained filtrate was a clear solution, which was transferred into an oven dried Schlenk tube under Ar. Phenylacetylene (9 mg, 0.09 mmol) and triethylsilane (18 mg, 0.16 mmol) were added. The tube was sealed and heated under stirring for one hour. Analysis by ^1H NMR spectroscopy showed no reaction had occurred as only starting material was detected.

Homogeneous Reactions

An oven dried 4 mL Schlenk tube was charged with **Rh^IC_n** (n = 0, 6 or 11; 0.9 μmol, 1 %), THF (1 mL), phenylacetylene (9 mg, 0.09 mmol) and triethylsilane (18 mg, 0.16 mmol) and heated to 60 °C for 1 h. The mixture was allowed to cool down to room temperature and the solvent was removed under reduced pressure, which was followed by drying under high vacuum and analysis by ¹H NMR spectroscopy (Figure S45).

Homogeneous Reactions

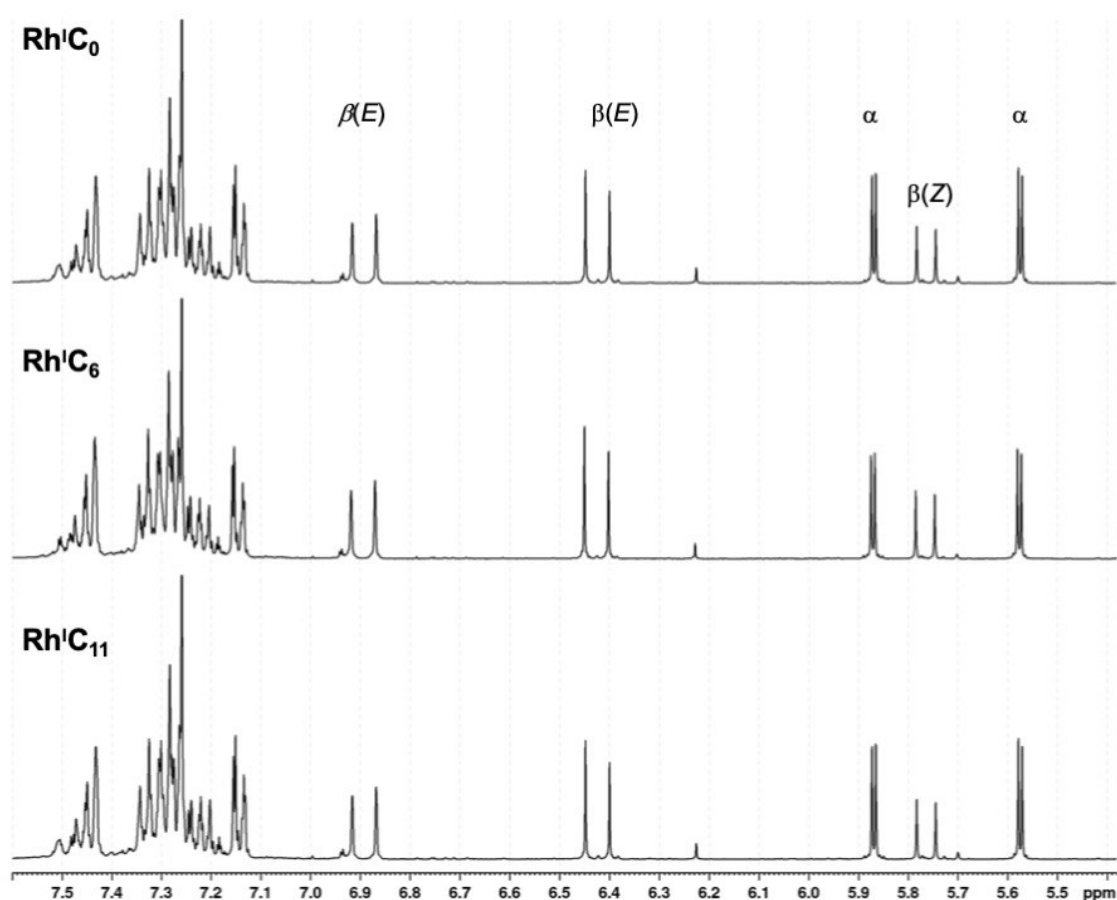


Figure S45. Selected ranges of the ¹H NMR spectra of the crude reaction mixtures using the catalysts **Rh^IC₀**, **Rh^IC₆** and **Rh^IC₁₁**. The vinylic protons of each of the three different isomers give rise to two doublets. The d of the β (Z)-isomer at 7.45 ppm is buried under aromatic resonances.

The control reactions were performed analogously to the above described procedure using the catalysts **Rh^IC₀Ph** and **Rh^{III}C₀Ph** without NH₂-groups and **Rh^{III}Cl₂** and **Rh^{III}C₁₁Cl₂**,

which were prepared as described in section 5. For the latter two, a composition of $[(\text{PhPyT})_2\text{RhCl}_2]\text{BAr}^{\text{F}}_4$ and $[(\text{PhCO}(\text{CH}_2)_{10}\text{PyT})_2\text{RhCl}_2]\text{BAr}^{\text{F}}_4$ was used as approximation for calculation of the catalyst amount applied. ^1H NMR spectra of the reaction mixtures are shown in the Figure S46.

Homogeneous Control Reactions

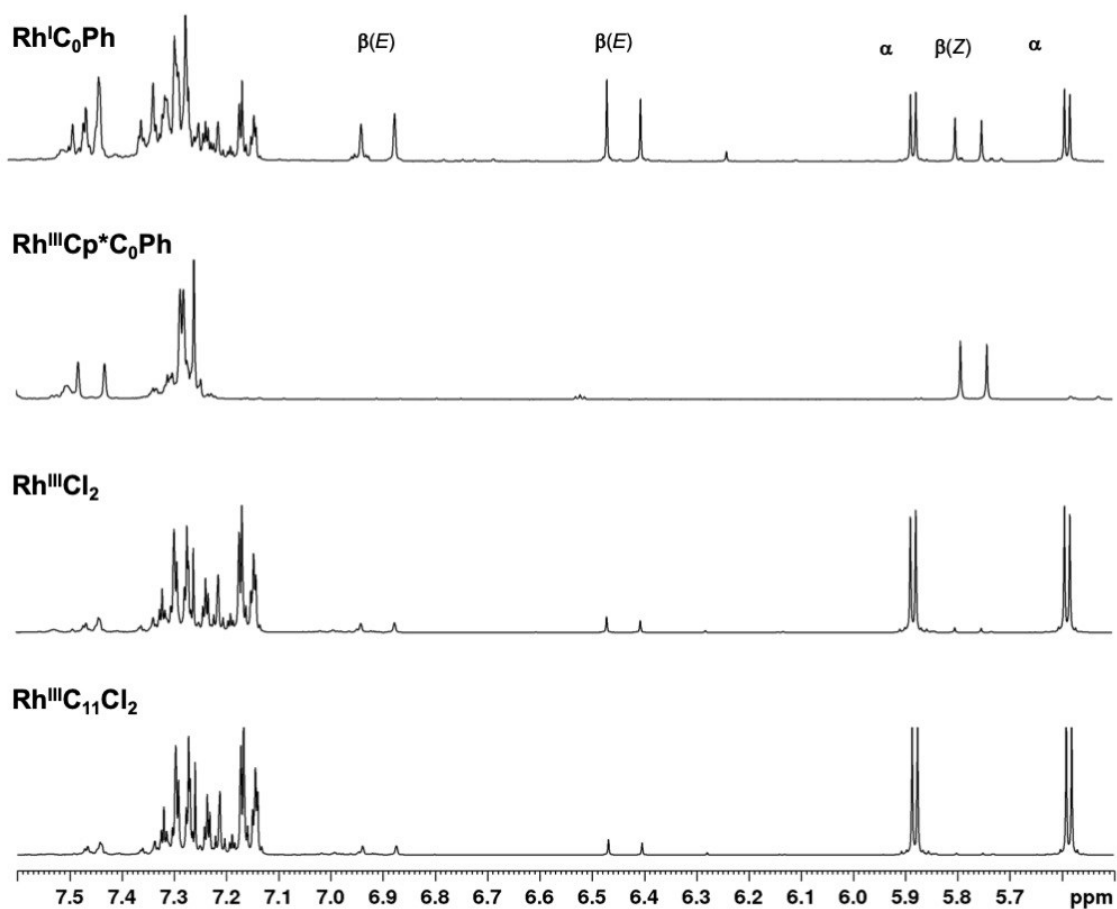
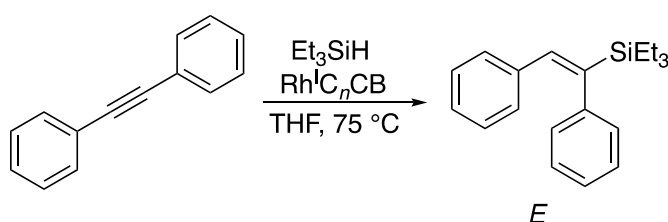


Figure S46. Selected ranges of the ^1H NMR spectra of the reaction mixtures from hydrosilylation of phenylacetylene using the control Rh catalysts **Rh^IC₀Ph** and **Rh^{III}Cp^{*}C₀Ph** as well as the materials **Rh^{III}Cl₂** and **Rh^{III}C₁₁Cl₂** obtained from treatment of the **Rh^IC₀Ph** and **Rh^IC₁₁** with NaNO_2/HCl under absence of CB.

6.2 Hydrosilylation of Diphenylacetylene

We tested the series of catalysts for the hydrosilylation of diphenylacetylene (Scheme S2). This reaction affords for all Rh^I catalysts the *Z*-isomer with the Rh^I catalysts not being active towards this reaction under the applied reaction conditions as only traces of the product formed over the same period of time, which led to near quantitative conversion for the Rh^I derivatives (Table S31).



Scheme S3. Hydrosilylation of diphenylacetylene.

Table S31. Summary of the hydrosilylation reactions of diphenylacetylene using the Rh and Ir based catalysts.

Entry	Catalyst	mol[%] cat.	T [°C]	t [min]	Conversion [%]
1	Rh ^{III} C ₁₁ CB	1.0	75	15	80
2	Rh ^{III} C ₁₁ CB	1.0	75	30	99
3	Rh ^{III} C ₁₁ CB ¹	1.0	75	60	100 (100 ²)
4	Rh ^{III} C ₆ CB	1.0	75	15	27
5	Rh ^{III} C ₆ CB	1.0	75	30	96
6	Rh ^{III} C ₀ CB	1.0	75	15	12
7	Rh ^{III} C ₀ CB	1.0	75	30	25
8	Rh ^{III} Cp [*] C ₁₁ C B	1.0	75	30	Traces (< 1)
9	Ir ^{III} Cp [*] C ₁₁ CB	0.5	75	30	Traces (< 1)
10	Ir ^{III} Cp [*] C ₆ CB	0.5	75	30	Traces (< 1)
11	Ir ^{III} Cp [*] C ₀ CB	0.5	75	30	Traces (< 1)

Reaction were conducted on a 15 mg scale in 1 mL THF using the specified amount of catalyst and 1.8 eq. of Et₃SiH. ¹ This reaction was conducted a preparative 60 mg scale. ² Isolated yield.

Preparative scale reaction:

(*E*)-(1,2-Diphenylvinyl)triethylsilane: A Schlenk tube was charged with diphenylacetylene (60 mg, 0.32 mmol), THF (4 mL), triethylsilane (72 mg, 0.60 mmol) and Rh^{III}C₁₁CB (8.8 mg, ~ 1.0 mol %). The tube was sealed with PTFE stopcock and the mixture was heated under

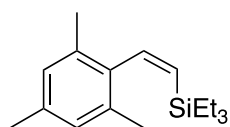
stirring for 1 h to 75 °C using an oil bath. After cooling down to room temperature, the mixture was transferred to glass centrifuge tubes as centrifuged for 20 min at 4000 rpm. The solution was removed and the remaining catalyst was washed with pentane (4 mL) and centrifuged again for 2 min. This was repeated one more time and the pentane from washings was combined with the reaction solution, filtered through a PTFE syringe filter and concentrated to dryness using a rotary evaporator. This was followed by drying under vacuum, affording the product as colorless oil (99 mg, 0.32 mmol; > 99%).

6.3 Scope: Hydrosilylation of Terminal Alkynes

Typical procedure for the hydrosilylation reactions: An oven dried Schlenk tube equipped with magnetic stirring bar and PTFE stopcock was charged with hybrid catalyst (~1.0 mol %), THF (1 mL), 0.009 mmol of the terminal alkyne and 18 mg (0.16 mmol) of triethylsilane. The mixture was heated to 75 °C for 1 h or 4 h while stirring using an oil bath with the head space fully submerged in the oil. After the specified time, the Schlenk tube was allowed to cool down to room temperature and the black suspension was transferred to a glass centrifuge tube and centrifuged for 20 min at 4000 rpm. The clear solution was transferred to a round bottom flask and the solvent was evaporated using a rotary evaporator. The product was briefly dried under vacuum. The product composition was analyzed by ¹H NMR spectroscopy. Vinylic proton resonances of formed isomeric products were integrated to each other and to unreacted terminal alkyne, allowing determination of isomeric ratios and conversions.

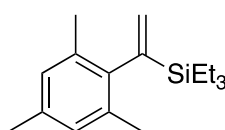
For preparative runs, the remaining black solids from the centrifugation were washed with pentane (2 mL) and centrifuged again for 5 min at 4000 rpm. The pentane was combined with the extract obtained from the first centrifugation and concentrated to dryness. This was followed by flash chromatography on silica using hexane as eluent. All silanes were obtained as colorless oils after solvent evaporation and drying under vacuum. Most reactions afforded

isomeric mixtures of the silanes, which did not separate on the column. We characterized these as isomeric mixtures. Reported ^1H - and ^{13}C NMR chemical shifts are assigned to the major components of the isomeric mixture.



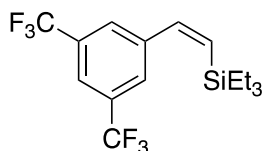
(Z)-triethyl(2,4,6-trimethylstyryl)silane (7b): 2-Ethynyl-1,3,5-

trimethylbenzene (13 mg, 0.09 mmol) afforded the title compound as mixture with **7c** (9 mg, 0.03 mmol, 38 %). ^1H NMR (300 MHz, CDCl_3): δ = 7.22 (d, $^3J_{\text{H-H}}$ = 15.4 Hz, 1H), 6.80 (2H), 5.97 (d, $^3J_{\text{H-H}}$ = 15.4 Hz, 1H), 2.27 (s, 3H), 2.18 (s, 6H), 0.81 (apparent t, $^3J_{\text{H-H}}$ $\approx$ 8.0 Hz, 9H, $\text{Si}(\text{CH}_2\text{CH}_3)_3$, 0.33 (apparent q, $^3J_{\text{H-H}}$ $\approx$ 8.0 Hz, $\text{Si}(\text{CH}_2\text{CH}_3)_3$) ppm. ^{13}C NMR (75 MHz, CDCl_3): δ = 146.4, 137.5, 136.0, 135.2, 131.7, 127.8, 21.1, 20.6, 7.6, 3.7 ppm. MS (APCI) m/z : $[\text{M} + \text{H}]^+$ 261 (32%), 249 (18%), $[\text{M-Et} + \text{H}]^+$ 231 (100%).



Triethyl(1-mesitylvinyl)silane (7c): 2-Ethynyl-1,3,5-trimethylbenzene

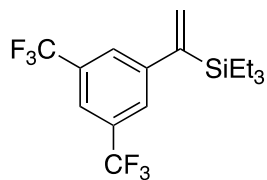
(26 mg, 0.18 mmol) afforded the title compound as mixture with **7a/b** (27 mg, 0.10 mmol, 57 %). ^1H NMR (300 MHz, CDCl_3): δ = 6.84 (m, 2H), 5.78 (d, $^2J_{\text{H-H}}$ = 3.4 Hz, 1H), 5.67 (d, $^2J_{\text{H-H}}$ = 3.4 Hz, 1H), 2.27 (s, CH_3 , 3H), (s, CH_3 , 6H), 0.93 (apparent t, $^3J_{\text{H-H}}$ $\approx$ 8.0 Hz, 9H, $\text{Si}(\text{CH}_2\text{CH}_3)_3$, 0.63 (apparent q, $^3J_{\text{H-H}}$ $\approx$ 8.0 Hz, $\text{Si}(\text{CH}_2\text{CH}_3)_3$) ppm. ^{13}C NMR (75 MHz, CDCl_3): δ = 151.1, 141.8, 134.7, 134.1, 129.5, 128.1, 21.0, 7.4, 4.4 ppm. HRMS (ESI) m/z : $[\text{M} + \text{H}]^+$ calcd for $\text{C}_{17}\text{H}_{29}\text{Si}_1$: 261.20330; found: 261.20318.



(Z)-(3,5-bis(trifluoromethyl)styryl)triethylsilane (8b): 1-Ethynyl-

3,5-bis(trifluoromethyl)benzene (21 mg, 0.09 mmol) afforded the title compound as mixture with **8c** (7 mg, 0.02 mmol, 22 %). ^1H NMR (400 MHz, CDCl_3): δ = 7.77 (m, 1H), 7.73 (m, 2H), 7.46 (d, $^3J_{\text{H-H}}$ = 15.3 Hz, 1H), 6.03 (d, $^3J_{\text{H-H}}$ = 15.3 Hz, 1H), 0.87 (apparent t, $^3J_{\text{H-H}}$ $\approx$ 8.0 Hz, 9H, $\text{Si}(\text{CH}_2\text{CH}_3)_3$, 0.54 (apparent q, $^3J_{\text{H-H}}$ $\approx$ 8.0

Hz, Si(CH₂CH₃)₃). ¹³C NMR (101 MHz, CDCl₃): δ = 144.3, 142.4, 134.6, 131.5 (q, ²J_{C-F} = 33 Hz), 128.1 (m), 123.5, (q, ¹J_{C-F} = 273 Hz), 121.1 (m), 7.4, 4.7 ppm. MS (GC/EI) m/z: [M - Et]⁺ 325 (60%), [M - Et - C₂H₄]⁺ 297 (100%), [M - 2 Et - C₂H₃]⁺ 269 (42%), 230 (64%).

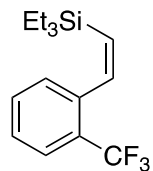


(1-(3,5-bis(trifluoromethyl)phenyl)vinyl)triethylsilane (8c): 1-

Ethynyl-3,5-bis(trifluoromethyl)benzene (40 mg, 0.17 mmol) afforded

the title compound as mixture with **8a** (49 mg, 0.14 mmol, 82 %). ¹H

NMR (300 MHz, CDCl₃): δ = 7.73 (m, 1H), 7.56 (m, 2H), 5.94 (d, ²J_{H-H} = 2.5 Hz, 1H), 5.74 (d, ²J_{H-H} = 2.5 Hz, 1H), 0.93 (apparent t, ³J_{H-H} ≈ 8.0 Hz, 9H, Si(CH₂CH₃)₃), 0.67 (apparent q, ³J_{H-H} ≈ 8.0 Hz, 6H, Si(CH₂CH₃)₃). ¹³C NMR (101 MHz, CDCl₃): δ = 150.0, 147.8, 131.62, 131.59 (q, J_{C-F} = 33 Hz), 126.9 (m), 123.60, (q, ²J_{C-F} = 273 Hz), 120.1 (m), 7.3, 3.4 ppm. MS (GC/EI) m/z: [M - Et]⁺ 325 (58%), [M - Et - C₂H₄]⁺ 297 (100%), [M - 2 Et - C₂H₃]⁺ 269 (42%), 230 (64%).

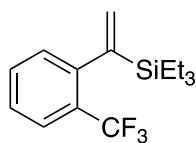


(Z)-Triethyl(2-(trifluoromethyl)styryl)silane (9b): 1-Ethynyl-2-

(trifluoromethyl)benzene (15 mg, 0.09 mmol) afforded the title compound as

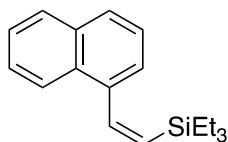
mixture with **9c** (22 mg, 0.08 mmol, 87 %). ¹H NMR (300 MHz, CDCl₃): δ =

7.68-7.31 (m, 5H), 5.94 (²J_{H-H} = 15.2 Hz, 1H), 0.83 (apparent t, ³J_{H-H} ≈ 8.0 Hz, 9H, Si(CH₂CH₃)₃), 0.42 (apparent q, ³J_{H-H} ≈ 8.0 Hz, 6H, Si(CH₂CH₃)₃) ppm. ¹³C NMR (101 MHz, CDCl₃): δ = 144.3, 139.8 (m), 132.8, 131.3 (m), 130.6, 128.1 (q, ²J_{C-F} = 30 Hz), 127.5, 125.6 (q, ³J_{C-F} = 5 Hz), 124.4 (q, ¹J_{C-F} = 274 Hz), 7.4, 4.6 ppm. MS (GC/EI) m/z: [M - Et]⁺ 257 (4%), [M - SiEt₃ - HF]⁺ 151 (22%), 143 (35%), [M - SiEt₃ - 2F]⁺ 133 (50%), [M - C₂SiEt₃ - F]⁺ 128 (100%), [M - SiEt₃ - 3F]⁺ 115 (22%).



Triethyl(1-(2-(trifluoromethyl)phenyl)vinyl)silane (9c): 1-Ethynyl-2-(trifluoromethyl)benzene (30 mg, 0.21 mmol) afforded the title compound as

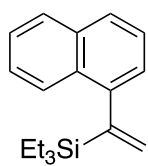
mixture with **9a/b** (51 mg, 0.18 mmol, 86 %). ^1H NMR (300 MHz, CDCl_3): δ = 7.63 (m, 2H), 7.42 (m, 2H), 7.28 (m, 2H), 7.02 (m, 2H), 5.76 (m, 1H), 5.59 (dm, $^2J_{\text{H-H}} = 2.9$ Hz, 1H), 0.90 (apparent t, $^3J_{\text{H-H}} \approx 8.0$ Hz, 9H, $\text{Si}(\text{CH}_2\text{CH}_3)_3$), 0.61 (apparent q, 6H, $^3J_{\text{H-H}} \approx 8.0$ Hz, $\text{Si}(\text{CH}_2\text{CH}_3)_3$) ppm. ^{13}C NMR (75 MHz, CDCl_3): δ = 148.4, 144.8 (m), 140.8 (m), 130.9, 129.9 (m), 129.3, 127.2 (q, $^2J_{\text{C-F}} = 29$ Hz), 126.3 (q, $^3J_{\text{C-F}} = 5$ Hz), 125.8, ~ 124.5, (q, $^1J_{\text{C-F}} = 273$ Hz), 7.2, 3.5 ppm. MS (APCI) m/z: $[\text{M-Et} + \text{H}]^+$ 257 (100%), 285 (45%).



(Z)-Triethyl(2-(naphthalen-1-yl)vinyl)silane (10b): 2-

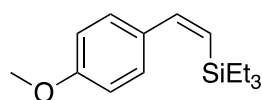
Ethynyl-naphthalene (13 mg, 0.09 mmol) afforded the title compound as mixture with **10c** (20 mg, 0.07 mmol, 87 %). ^1H NMR (300 MHz, CDCl_3):

δ = 8.0-7.3 (m, 7H), 7.91 (d, 1H, $^3J = 15.2$ Hz, 1H), 6.06 (d, $^3J = 15.2$ Hz, 1H), 0.79 (apparent t, $^3J_{\text{H-H}} \approx 8.0$ Hz, 9H, $\text{Si}(\text{CH}_2\text{CH}_3)_3$), 0.38 (apparent q, $^3J_{\text{H-H}} \approx 8.0$ Hz, 6H, $\text{Si}(\text{CH}_2\text{CH}_3)_3$) ppm. ^{13}C NMR (75 MHz, CDCl_3): δ = 146.2, 138.5, 133.3, 131.9, 131.8, 128.4, 128.0, 125.9, 125.9, 125.6, 125.3, 125.2, 29.9, 7.5, 4.7 ppm. MS (APCI) m/z $[\text{M} + \text{H}]^+$ 269 (100%).



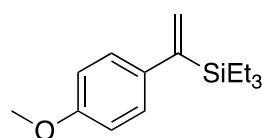
Triethyl(1-(naphthalen-1-yl)vinyl)silane (10c): 2-Ethynyl-naphthalene (26 mg, 0.17 mmol) afforded the title compound as mixture with **10a/b** (31 mg, 0.01 mmol, 68 %). ^1H NMR (300 MHz, CDCl_3): δ = 7.97 (m, 1H), 7.84 (m, 1H), 7.72

(m, 1H), 7.50-7.39 (m, 3H), 7.09 (m, 1H), 5.95 (d, $^2J_{\text{H-H}} = 3.5$ Hz, 1H), 5.88 (d, $^2J_{\text{H-H}} = 3.5$ Hz, 1H), 0.93 (apparent t, $^3J_{\text{H-H}} \approx 8.0$ Hz, 9H, $\text{Si}(\text{CH}_2\text{CH}_3)_3$), 0.63 (apparent q, $^3J_{\text{H-H}} \approx 8.0$ Hz, $\text{Si}(\text{CH}_2\text{CH}_3)_3$) ppm. ^{13}C NMR (75 MHz, CDCl_3): δ = 150.1, 143.5, 133.9, 131.5, 130.5, 128.3, 126.4, 126.1, 125.7, 125.3, 125.3, 123.8, 7.4, 3.4 ppm. The data is in agreement with the literature.¹²



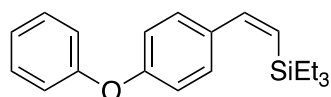
(Z)-triethyl(4-methoxystyryl)silane (11b): 1-Ethynyl-4-methoxybenzene (12 mg, 0.09 mmol) afforded the title compound as

mixture with **11c**. We analyzed the product mixture by ^1H NMR spectroscopy without isolating the compound. ^1H NMR (300 MHz, CDCl_3): δ = 7.39 (d, $^3J_{\text{H-H}}$ = 15.2 Hz, 1H), 7.23 (m, 2H), 6.85 (m, 2H), 3.82 (s, 3H, OCH_3), 0.89 (apparent t, $^3J_{\text{H-H}}$ $\approx$ 8.0 Hz, 9H, $\text{Si}(\text{CH}_2\text{CH}_3)_3$), 0.58 (apparent q, $^3J_{\text{H-H}}$ $\approx$ 8.0 Hz, 6H, $\text{Si}(\text{CH}_2\text{CH}_3)_3$). The data is in agreement with the literature.¹³



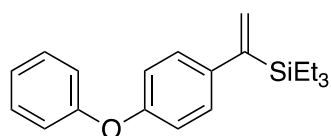
Triethyl(1-(4-methoxyphenyl)vinyl)silane (11c): 1-Ethynyl-4-methoxybenzene (23 mg, 0.17 mmol) afforded the title compound as

mixture with **11b** (38 mg, 0.15 mmol, 88 %). ^1H NMR (300 MHz, CDCl_3): δ = 7.13 (dm, $^3J_{\text{H-H}}$ = 8.8 Hz, 2H), 6.86 (dm, $^3J_{\text{H-H}}$ = 8.8 Hz, 2H), 5.88 (d, $^3J_{\text{H-H}}$ = 3.1 Hz, 1H), 5.54 (d, $^3J_{\text{H-H}}$ = 3.1 Hz, 1H), 3.81 (s, 3H), 0.95 (apparent t, $^3J_{\text{H-H}}$ $\approx$ 8.0 Hz, 9H, $\text{Si}(\text{CH}_2\text{CH}_3)_3$), 0.70 (apparent q, $^3J_{\text{H-H}}$ $\approx$ 8.0 Hz, 6H, $\text{Si}(\text{CH}_2\text{CH}_3)_3$). ^{13}C NMR (75 MHz, CDCl_3): δ = 158.2, 149.5, 137.9, 127.9, 127.7, 113.5, 55.2, 7.3, 3.4 ppm. MS (APCI) m/z $[\text{M} + \text{H}]^+$ 249 (100%).



(Z)-Triethyl(4-phenoxystyryl)silane (12b): 1-Ethynyl-4-phenoxybenzene (18 mg, 0.09 mmol) afforded the title compound

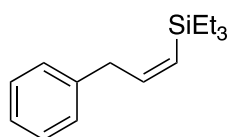
as mixture with **12c** (26 mg, 0.08 mmol, 90 %). ^1H NMR (300 MHz, CDCl_3): δ = 7.42 (d, 1H, $^3J_{\text{H-H}}$ = 15.0 Hz, 1H), 5.73 (d, $^3J_{\text{H-H}}$ = 15.0 Hz, 1H), 0.89 (apparent t, $^3J_{\text{H-H}}$ $\approx$ 8.0 Hz, 9H, $\text{Si}(\text{CH}_2\text{CH}_3)_3$), 0.58 (apparent q, $^3J_{\text{H-H}}$ $\approx$ 8.0 Hz, 6H, $\text{Si}(\text{CH}_2\text{CH}_3)_3$) ppm. ^{13}C NMR (75 MHz, CDCl_3): δ = 157.4, 156.7, 147.0, 135.9, 129.9, 129.4, 129.0, 123.4, 119.0, 118.5, 7.6, 4.9 ppm. MS (APCI) m/z $[\text{M} + \text{H}]^+$ 310 (100%).



Triethyl(1-(4-phenoxyphenyl)vinyl)silane (12c): 1-Ethynyl-4-phenoxybenzene (36 mg, 0.19 mmol) afforded the title compound

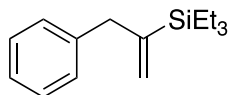
as mixture with **12a/b** (48 mg, 0.15 mmol, 83 %). ^1H NMR (300 MHz, CDCl_3): δ = 7.35 (m,

2H), 7.14 (m, 3H), 7.04 (m, 2H), 6.96 (m, 2H), 5.91 (d, $^2J_{\text{H-H}} = 3.1$ Hz, 1H), 5.59 (d, $^2J_{\text{H-H}} = 3.1$ Hz, 1H), 0.93 (apparent t, $^3J_{\text{H-H}} \approx 8.0$ Hz, 9H, Si(CH₂CH₃)₃), 0.70 (apparent q, $^3J_{\text{H-H}} \approx 8.0$ Hz, 6H, Si(CH₂CH₃)₃) ppm. ¹³C NMR (75 MHz, CDCl₃): δ = 157.6, 155.8, 149.7, 140.7, 129.9, 128.7, 128.1, 123.2, 118.9, 118.7, 7.5, 3.5 ppm. HRMS (ESI) *m/z*: [M + H]⁺ calcd for C₂₀H₂₇O₁Si₁: 311.18257; found: 311.18238.



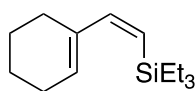
(Z)-Triethyl(3-phenylprop-1-en-1-yl)silane (13b): Prop-2-yn-1-ylbenzene (10 mg, 0.09 mmol) afforded the title compound as mixture with

13c. We analyzed the product mixture by ¹H NMR spectroscopy without isolating the compound. ¹H NMR (400 MHz, CDCl₃): δ = 7.34 – 7.19 (m, 5H, Ph-*H*), 6.51 (dt, $^3J_{\text{H-H}} = 14.0$ Hz, $^3J_{\text{H-H}} = 7.3$ Hz, 1H), 5.57 (dt, $^3J = 14.1$ Hz, $^4J = 1.5$ Hz, 1H), 3.49 (m, 2H, PhCH₂), 1.00 (apparent t, $^3J_{\text{H-H}} \approx 8.0$ Hz, 9H, Si(CH₂CH₃)₃), 0.69 (apparent q, $^3J_{\text{H-H}} \approx 8.0$ Hz, 6H, Si(CH₂CH₃)₃). The data is in agreement with the literature.¹⁴



Triethyl(3-phenylprop-1-en-2-yl)silane (13c): Prop-2-yn-1-ylbenzene (20 mg, 0.17 mmol) afforded the title compound as mixture with **13a/b** (26

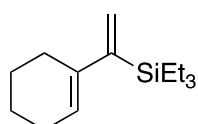
mg, 0.1 mmol, 65 %). ¹H NMR (300 MHz, CDCl₃): δ = 7.31 – 7.15 (m, 5H, Ph-*H*), 5.51 (m, 1H), 5.41 (m, 1H), 3.43 (m, 2H, PhCH₂), 0.89 (apparent t, $^3J_{\text{H-H}} \approx 8.0$ Hz, 9H, Si(CH₂CH₃)₃), 0.54 (apparent q, $^3J_{\text{H-H}} \approx 8.0$ Hz, 6H, Si(CH₂CH₃)₃). ¹³C NMR (75 MHz, CDCl₃): δ = 148.4, 140.2, 129.5, 128.2, 127.4, 126.0, 43.2, 7.4, 3.0 ppm. The data is in agreement with the literature.¹⁵



(Z)-(2-(Cyclohex-1-en-1-yl)vinyl)triethylsilane (14b): 1-Ethynylcyclohex-1-ene (9 mg, 0.09 mmol) afforded the title compound as mixture with **14c** (14

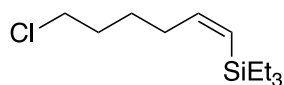
mg, 0.06 mmol, 74 %). We analyzed the product mixture by ¹H NMR spectroscopy without

isolating the compound. ^1H NMR (400 MHz, CDCl_3): δ = 6.74 (d, $^3J_{\text{H-H}}$ = 14.8 Hz, 1H), 5.65 (m, 1H), 5.32 (d, $^3J_{\text{H-H}}$ = 14.8 Hz, 1H), 2.07 (m, 4H), 1.6 (m, 4H), 0.93 (apparent t, $^3J_{\text{H-H}}$ $\approx$ 8.0 Hz, 9H, $\text{Si}(\text{CH}_2\text{CH}_3)_3$), 0.59 (apparent q, $^3J_{\text{H-H}}$ $\approx$ 8.0 Hz, 6H, $\text{Si}(\text{CH}_2\text{CH}_3)_3$). The data is in agreement with the literature.¹⁶



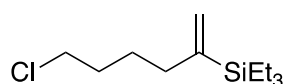
(1-(Cyclohex-1-en-1-yl)vinyl)triethylsilane (14c): 1-Ethynylcyclohex-1-ene (20 mg, 0.19 mmol) afforded the title compound as mixture with **14a/b**

(35 mg, 0.15 mmol, 80 %). ^1H NMR (300 MHz, CDCl_3): δ = 5.71 (d, $^2J_{\text{H-H}}$ = 2.9 Hz, 1H), 5.60 (m, 1H), 5.27 (d, $^2J_{\text{H-H}}$ = 2.9 Hz, 1H), 2.14 (m, 4H), 1.6 (m, 4H), 0.93 (apparent t, $^3J_{\text{H-H}}$ $\approx$ 8.0 Hz, 9H, $\text{Si}(\text{CH}_2\text{CH}_3)_3$), 0.64 (apparent q, $^3J_{\text{H-H}}$ $\approx$ 8.0 Hz, 6H, $\text{Si}(\text{CH}_2\text{CH}_3)_3$). The data is in agreement with the literature.¹⁷



(Z)-(6-Chlorohex-1-en-1-yl)triethylsilane (15b): 6-Chlorohex-1-yne (10 mg, 0.09 mmol) afforded the title compound isometrically

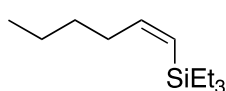
pure. We analyzed the product mixture by ^1H NMR spectroscopy without isolating the compound. ^1H NMR (300 MHz, CDCl_3): δ = 6.35 (dt, $^3J_{\text{H-H}}$ = 14.1 Hz, $^3J_{\text{H-H}}$ = 7.3 Hz 1H), 5.44 (dt, $^3J_{\text{H-H}}$ = 14.1 Hz, $^4J_{\text{H-H}}$ = 1.4 Hz 1H), 3.54 (t, $^3J_{\text{H-H}}$ = 6.7 Hz, 2H, CH_2Cl), 2.14 (m, 2H), 1.80 (m, 2H), 1.53 (m, 2H), 0.95 (apparent t, $^3J_{\text{H-H}}$ $\approx$ 8.0 Hz, 9H, $\text{Si}(\text{CH}_2\text{CH}_3)_3$), 0.61 (apparent q, $^3J_{\text{H-H}}$ $\approx$ 8.0 Hz, 6H, $\text{Si}(\text{CH}_2\text{CH}_3)_3$). The data is in agreement with the literature.¹³



(6-Chlorohex-1-en-2-yl)triethylsilane (15c): 6-Chlorohex-1-yne (20 mg, 0.17 mmol) afforded the title compound as mixture with **15a/b** (39

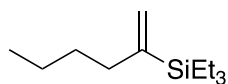
mg, 0.17 mmol, 98 %). ^1H NMR (300 MHz, CDCl_3): δ = 5.94 (m, 1H), 5.32 (m, 1H), 3.55 (t, $^3J_{\text{H-H}}$ = 6.8 Hz, 2H), 2.11 (m, 2H), 1.79 (m, 2H), 1.56 (m, 2H), 0.92 (apparent t, $^3J_{\text{H-H}}$ $\approx$ 8.0 Hz, 9H, $\text{Si}(\text{CH}_2\text{CH}_3)_3$), 0.60 (apparent q, $^3J_{\text{H-H}}$ $\approx$ 8.0 Hz, 6H, $\text{Si}(\text{CH}_2\text{CH}_3)_3$) ppm. ^{13}C NMR (75

MHz, CDCl₃): δ = 148.5, 125.6, 45.1, 35.5, 32.6, 26.2, 7.5, 3.1 ppm. MS (GC/EI) m/z: [M - Et]⁺ 203 (5%), [M - Et - C₂H₄]⁺ 175 (10%), [M - 2 Et - C₂H₃]⁺ 147 (9%), [M - SiEt₂ - C₂H₂]⁺ 121 (100%), [C₄H₁₀Cl]⁺ 92 (48%).



(Z)-Triethyl(hex-1-en-1-yl)silane (16b): Hex-1-yne (7 mg, 0.09 mmol)

afforded the title compound isometrically pure. We analyzed the product mixture by ¹H NMR spectroscopy without isolating the compound. ¹H NMR (300 MHz, CDCl₃): δ = 6.34 (dt, ³J_{H-H} = 14.1 Hz, ³J = 7.3 Hz 1H), 5.44 (dt, ³J_{H-H} = 14.1 Hz, ⁴J_{H-H} = 1.3 Hz 1H), 2.10 (m, 2H), 1.80 (m, 2H), 1.35 (m, 4H), 0.94 (apparent t, ³J_{H-H} $\approx$ 8.0 Hz, 9H, Si(CH₂CH₃)₃), 0.93 (m, 3H, overlapping), 0.61 (apparent q, ³J_{H-H} $\approx$ 8.0 Hz, 6H, Si(CH₂CH₃)₃). The data is in agreement with the literature.¹³



Triethyl(hex-1-en-2-yl)silane (16c): Hex-1-yne (14 mg, 0.17 mmol)

afforded the title compound as mixture with **16a/b** (32 mg, 0.16 mmol, 95 %). ¹H NMR (300 MHz, CDCl₃): δ = 5.63 (m, 1H), 5.28 (m, 1H), 2.08 (m, 2H), 1.35 (m, 4H), 0.92 (apparent t, ³J_{H-H} $\approx$ 8.0 Hz, 9H, Si(CH₂CH₃)₃), 0.91 (m, 3H, overlapping), 0.60 (apparent q, ³J_{H-H} $\approx$ 8.0 Hz, 6H, Si(CH₂CH₃)₃). ¹³C NMR (75 MHz, CDCl₃): δ = 149.4, 125.1, 36.2, 31.3, 22.8, 14.2, 7.5, 3.1 ppm. MS (GC/EI) m/z: [M - Et]⁺ 169 (97%), [M - Et - C₂H₄]⁺ 141 (100%), [M - 2 Et - C₂H₃]⁺ 113 (29%).

7. Computational details

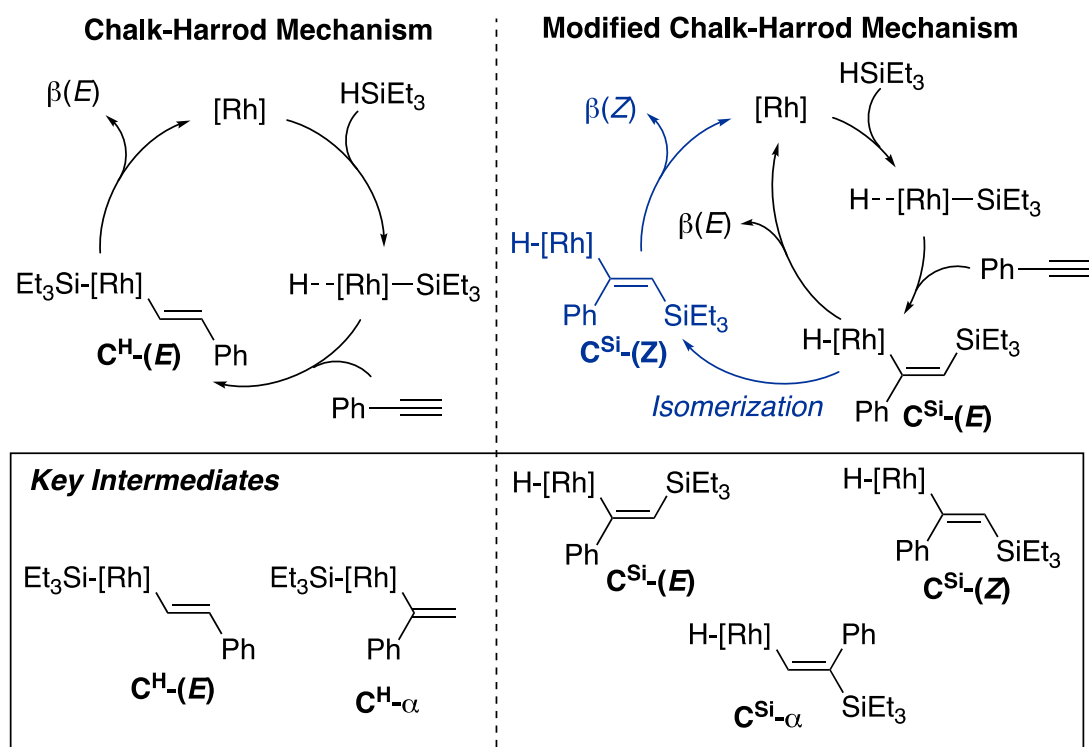
All calculations were performed using the Gaussian 16 software¹⁸ on the Australian National Computational Infrastructure, with access provided through the Macquarie University Partner Scheme (project ID gy32). All images of the calculated structures were generated using CYLView.¹⁹ The frequency calculations and structure optimizations were performed at 25 °C using ω B97XD with the 6-31G(d) basis set for C, O, H, Cl and N, and the SDD ECP for Rh and Si. Single point energy calculations were performed using the M06L method, and the def2TZVP basis set. The CPCM solvation model was used to incorporate the solvent effects of tetrahydrofuran. Frequency calculations were performed to confirm whether the calculated structure is a local minimum or a transition state. For transition state structures, intrinsic reaction coordinate (IRC) calculations were performed to confirm whether the located transition state connects the correct minima.

Table S32. Summary of data obtained from the DFT calculations.^a

Rh^I(CO)₂	Gibbs free energy (kcal mol⁻¹)	
H-Rh-Si intermediate (+ free phenylacetylene)	18.0	
	<i>Intermediate</i>	<i>Reductive Elimination</i>
C ^{Si} -(<i>E</i>)	17.0	35.3
C ^{Si} -(<i>Z</i>)	14.0	26.8
C ^{Si} - α	11.4	23.0
C ^H -(<i>E</i>)	4.5	14.6
C ^H - α	7.2	30.2
Rh^{III}ClCp*		
Rh-H-Si intermediate (+ free phenylacetylene)	13.5	
	<i>Intermediate</i>	<i>Reductive Elimination</i>
C ^{Si} -(<i>E</i>), two N coordinated	61.0 (<i>C-Cl bond formed</i>)	
C ^{Si} -(<i>E</i>), one N coordinated	65.6	72.5
C ^{Si} -(<i>Z</i>) two N coordinated	50.2	
C ^{Si} -(<i>Z</i>) one N coordinated	43.8	45.4
C ^{Si} - α two N coordinated	52.8	
C ^{Si} - α one N coordinated	44.4	42.2
Rh^{III}Cl₂		
Rh-H-Si intermediate (+ free	-7.0	

phenylacetylene)		
	<i>Intermediate</i>	<i>Reductive Elimination</i>
$C^{Si-\alpha}$	6.8	18.8
$\beta(E)$ product (+ free catalyst)	-20.9	
$\beta(Z)$ product (+ free catalyst)	-25.0	
α product (+ free catalyst)	-21.5	

^a All energies are relative to the sum of the energies of the free catalyst, phenylacetylene and triethylsilane.



Scheme S4. Chalk-Harrod and modified Chalk-Harrod mechanism for the hydrosilylation of phenylacetylene using the Rh catalysts.

The most commonly reported mechanism for Rh catalyzed hydrosilylation is a modified Chalk-Harrod mechanism, which involves 4 key steps: 1) oxidative addition of the silane to Rh; 2) silyl insertion into the alkyne to give the $C^{Si}-(E)$ intermediate; 3)

potential isomerization of $C^{Si}-(E)$ to $C^{Si}-(Z)$; and 4) reductive elimination to generate the product(s) (Scheme 3).²⁰⁻²² This mechanism is consistent with the commonly observed $\beta(E)$ selectivity for cationic Rh catalysts and the products observed here on application of the homogeneous controls. The **Rh^IC₀Ph** catalyst gave a mixture of $\beta(E)$, $\beta(Z)$ and α products in our experimental analyses, in analogy to the series Rh^IC_n ($n = 0, 6, 11$). If a step-wise Chalk-Harrod type mechanism is operative, the reaction should proceed via a H-Rh-Si intermediate (Scheme 3) - this species was readily located (18.0 kcal mol⁻¹) confirming that a Chalk-Harrod type mechanism is feasible. Alkyne insertion to H-Rh-Si could either occur at the Rh-H bond (Chalk Harrod) or at the Rh-Si bond (modified Chalk-Harrod), followed by potential isomerisation, resulting in five possible intermediates. All five intermediates were located, with subsequent reductive elimination from each intermediate feasible (see SI). The $C^{Si}-(E)$ intermediate was the least stable (17.0 kcal mol⁻¹), with the corresponding reductive elimination transition state highest in energy (35.3 kcal mol⁻¹), suggesting that isomerisation to the thermodynamically favored $C^{Si}-(Z)$ intermediate (14.0 kcal mol⁻¹) is likely. Reductive elimination from $C^{Si}-(Z)$ proceeds *via* a lower energy transition state (26.8 kcal mol⁻¹), accounting for the experimental formation of $\beta(Z)$. While it is unclear whether $C^{Si}-\alpha$ or $C^H-\alpha$ form directly in the insertion step, or result from isomerisation, the experimental observation of the α -product indicates that a pathway involving $C^{Si}-\alpha$ or $C^H-\alpha$ formation is likely.

Cartesian Coordinates

1. Different isomers of RhCl₂PyT₂

C	2.40146100	-1.76701300	0.86061400
C	3.75348700	-1.81869300	0.64606900
H	4.50969600	-2.53277700	0.93034200
N	2.95418500	0.04792000	-0.23185500
N	1.96895200	-0.60318800	0.30732600
N	4.04139100	-0.67579500	-0.02309500
C	5.31669100	-0.21192200	-0.47799800
C	5.65122100	1.12514200	-0.29291100
C	6.18454000	-1.11243200	-1.08530800
C	6.89379400	1.56643500	-0.73339400
H	4.95322000	1.79851500	0.19278300
C	7.42938800	-0.65811500	-1.50824900
H	5.88593200	-2.14340800	-1.24815300
C	7.78270900	0.67771200	-1.33467300
H	7.17149800	2.60610300	-0.59640800
H	8.11678600	-1.34782200	-1.98610600
H	8.75284000	1.02797800	-1.67113900
C	1.47609000	-2.69698100	1.57476600
H	1.97838600	-3.64328300	1.77982300
H	1.15055600	-2.24882700	2.51695600
N	0.30901800	-2.99811200	0.76208900
C	-0.15544700	-4.21622900	0.41838400
C	-1.25198500	-4.03369100	-0.39590700
H	0.32573000	-5.11250700	0.78111100
C	-1.38696900	-2.64747100	-0.51231700
H	-1.86177700	-4.79490500	-0.85588600
H	-2.09347200	-2.06321800	-1.07913300
N	-0.44824300	-2.02926000	0.20380000
Rh	-0.00000600	-0.00006600	0.27001100
Cl	-0.00007300	0.00007700	-2.06820800
Cl	0.00001000	-0.00017400	2.66466600
N	-1.96897000	0.60309300	0.30742200
C	-2.40141600	1.76694900	0.86071200
C	-3.75344300	1.81868600	0.64618100
H	-4.50964100	2.53279300	0.93043200
N	0.44824300	2.02917300	0.20397400
C	0.15557600	4.21612800	0.41884200
C	1.25211400	4.03363100	-0.39546900
H	-0.32558500	5.11238200	0.78164900
H	1.86195700	4.79486800	-0.85534300
C	-1.47604000	2.69684200	1.57496600
H	-1.15059200	2.24859500	2.51714400
H	-1.97831900	3.64314700	1.78006300
N	-2.95422300	-0.04795900	-0.23178400
N	-4.04139100	0.67580400	-0.02299200
C	-5.31671100	0.21204100	-0.47794800
C	-6.18417200	1.11248500	-1.08591100
C	-5.65164400	-1.12483200	-0.29224800
C	-7.42905200	0.65829300	-1.50888900
H	-5.88522400	2.14329000	-1.24922100
C	-6.89424400	-1.56601400	-0.73277900
H	-4.95392800	-1.79815400	0.19392800
C	-7.78278100	-0.67735200	-1.33470100
H	-8.11616300	1.34793800	-1.98724900
H	-7.17225500	-2.60553800	-0.59532200
H	-8.75294000	-1.02751300	-1.67119600
N	-0.30894600	2.99799200	0.76240400
C	1.38700600	2.64742100	-0.51206900
H	2.09345400	2.06319200	-1.07898100

Thermal correction to Gibbs Free Energy= 0.394575

E(RM06L) = -2506.64671226

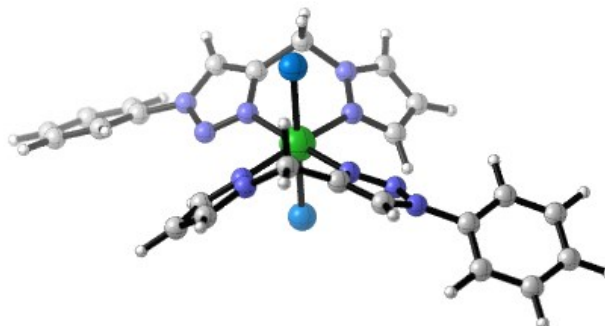


Figure S47. Calculated structure of isomer 1.

C	2.78315600	1.29229100	0.45096400
C	4.08960900	0.88185100	0.47565200
H	5.01033900	1.39653500	0.69845000
N	2.80736400	-0.81462500	-0.15348100
N	2.04922100	0.21809000	0.05173700
N	4.04579800	-0.41961500	0.09284800
C	5.13340300	-1.33691800	-0.07445700
C	5.04317300	-2.30833700	-1.06582500
C	6.24291100	-1.23230500	0.75735300
C	6.10359400	-3.19298800	-1.22508600
H	4.15717600	-2.37232100	-1.68769500
C	7.30036500	-2.11716900	0.57493200
H	6.27531000	-0.49552900	1.55401600
C	7.23194400	-3.09529600	-0.41380400
H	6.04642200	-3.95929500	-1.99066500
H	8.17092500	-2.04969100	1.21857500
H	8.05657400	-3.78758100	-0.54731600
C	2.14529600	2.61434700	0.73725100
H	2.88158800	3.30949200	1.14292600
H	1.73806600	3.04598700	-0.18251800
N	1.07365400	2.49965000	1.71585200
C	0.93106600	3.16442900	2.88267800
C	-0.22898900	2.71614300	3.47879200
H	1.65820500	3.89896000	3.19639500
C	-0.73466500	1.75177900	2.59956800
H	-0.64038000	3.02661300	4.42623400
H	-1.59626600	1.10745400	2.69346200
N	0.05275700	1.63915900	1.53168300
Rh	-0.00000500	0.18441600	-0.00000600
Cl	-0.01925300	-1.40915000	-1.68108100
N	-2.04921600	0.21809200	-0.05174400
C	-2.78315700	1.29229500	-0.45095400
C	-4.08961000	0.88184900	-0.47564500
H	-5.01034200	1.39653100	-0.69843300
N	-0.05276800	1.63917200	-1.53167300
C	-0.93106200	3.16446800	-2.88264700
C	0.22897700	2.71616200	-3.47877800
H	-1.65818700	3.89902000	-3.19634700
H	0.64036300	3.02663500	-4.42622100
C	-2.14530300	2.61435600	-0.73722900
H	-1.73807100	3.04598600	0.18254300
H	-2.88159500	3.30950300	-1.14289800
N	-2.80735400	-0.81463200	0.15346100
N	-4.04579100	-0.41962000	-0.09285500
C	-5.13339300	-1.33692500	0.07445400
C	-6.24290900	-1.23231000	-0.75734500
C	-5.04315300	-2.30834900	1.06581800
C	-7.30036000	-2.11717700	-0.57491800
H	-6.27531500	-0.49553100	-1.55400500
C	-6.10357000	-3.19300200	1.22508400
H	-4.15714900	-2.37233400	1.68767900
C	-7.23192800	-3.09530800	0.41381300
H	-8.17092500	-2.04969700	-1.21855300
H	-6.04639000	-3.95931300	1.99065900
H	-8.05655600	-3.78759500	0.54733000
N	-1.07365700	2.49967300	-1.71583100
C	0.73464600	1.75178300	-2.59956600
H	1.59623000	1.10743900	-2.69347700
Cl	0.01924600	-1.40916900	1.68105300

Thermal correction to Gibbs Free Energy= 0.393204

E(RM06L) = -2506.64070055

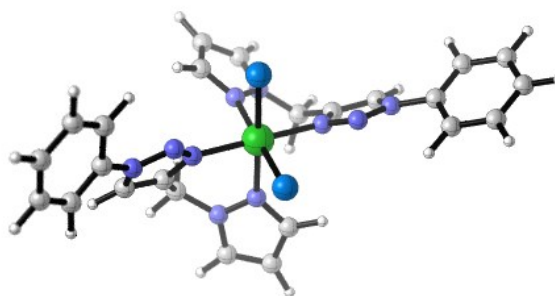


Figure S48. Calculated structure of isomer 2.

C	0.39970100	2.71640000	0.78849200
C	-0.79109900	3.39380100	0.79267100
H	-1.05543800	4.39459000	1.09411700
N	-1.11155400	1.37461100	-0.05827700
N	0.14055300	1.49040900	0.25622200
N	-1.68702200	2.51838200	0.26963600
C	-3.09216800	2.67724000	0.05101100
C	-3.67179600	2.04590700	-1.04608000
C	-3.84094200	3.43339400	0.94589800
C	-5.04181200	2.18119500	-1.24267000
H	-3.05802200	1.46511100	-1.72603800
C	-5.20770800	3.57131600	0.72491100
H	-3.37594900	3.88795700	1.81514700
C	-5.80832700	2.94519100	-0.36433500
H	-5.50670400	1.69897700	-2.09629300
H	-5.80371700	4.15917800	1.41477000
H	-6.87512100	3.05427200	-0.52967700
C	1.76838900	3.10020100	1.25239500
H	1.80993700	4.16938800	1.46403400
H	2.02905400	2.54210400	2.15393700
N	2.75449400	2.83050700	0.21529200
C	3.56710500	3.73109700	-0.37422800
C	4.17765000	3.11350000	-1.44304900
H	3.64551000	4.73741900	0.00988400
C	3.65397100	1.81889100	-1.44857900
H	4.88714600	3.54025700	-2.13385500
H	3.79466500	1.03568800	-2.17463400
N	2.81071600	1.64414600	-0.43119100
Rh	1.54663900	0.01790900	-0.00795300
Cl	1.10454300	-0.11430700	-2.29960500
Cl	1.99866900	0.08483700	2.34349600
N	2.98956600	-1.49252600	-0.19470200
C	3.93557600	-3.48501800	0.06246900
H	4.03184200	-4.48201000	0.46613800
N	0.22482500	-1.50902200	0.30533700
C	0.50872000	-2.71323500	0.86497500
C	-0.65637100	-3.43377900	0.84278700
H	-0.90805200	-4.41644700	1.20751400
C	1.86881500	-3.02308600	1.39072500
H	2.05408900	-2.45789800	2.30778200
H	1.95762700	-4.08990200	1.60015300
N	-1.01374500	-1.45076000	-0.07344100
N	-1.56103500	-2.60445100	0.26138300
C	-2.95623200	-2.80899200	0.01603400
C	-3.42564700	-4.08809400	-0.25804700
C	-3.80481600	-1.70627300	0.05545400
C	-4.78735100	-4.26656100	-0.48194200
H	-2.74414000	-4.93052900	-0.32240100
C	-5.15942200	-1.90006700	-0.18488400
H	-3.40534200	-0.71993100	0.26399000
C	-5.65312700	-3.17664900	-0.44667600
H	-5.16710400	-5.25943000	-0.69791300
H	-5.82978800	-1.04706500	-0.15883800
H	-6.71294900	-3.32250300	-0.62739500
N	2.90176600	-2.69543200	0.41770400
C	4.08478100	-1.54493300	-0.94966000
C	4.72340600	-2.77950400	-0.81868800
H	4.36488500	-0.70724100	-1.56169200
H	5.62648700	-3.10956000	-1.30662500

Thermal correction to Gibbs Free Energy= 0.394262
E(RM06L) = -2506.64141586

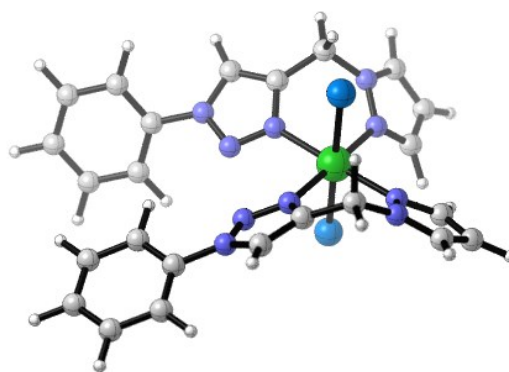


Figure S49. Calculated structure of isomer 3.

C	-4.65758600	-1.51875100	-1.58548900
H	-5.66703200	-1.15705900	-1.45487800
N	-2.49731400	-1.69836300	-1.11126800
C	-3.75370400	-0.20692000	0.36576800
H	-4.77574900	0.17280100	0.40050300
H	-3.57367600	-0.78464800	1.27913400
C	-2.94284300	2.28745500	0.34455700
H	-3.81373900	2.91917800	0.41867500
N	-1.45240800	0.70080500	0.13472000
Rh	-0.63679000	-1.14863000	-0.38194100
Cl	0.22167700	-3.22822700	-0.97018700
N	1.20886200	-0.63910400	0.32859900
C	1.48794800	-0.41814100	1.64083800
C	2.84040500	-0.22074600	1.72135000
H	3.49388300	0.01672500	2.54517800
N	-1.04490900	-1.95837300	1.51758200
C	-0.72729600	-2.40219000	3.66724800
C	-1.57171000	-3.37452800	3.17391100
H	-0.32299200	-2.24135200	4.65586200
H	-1.99702500	-4.20695400	3.71191100
C	0.41125800	-0.37682700	2.67687300
H	-0.22127800	0.50412100	2.52775600
H	0.84954700	-0.31313500	3.67377200
N	2.28653000	-0.60693500	-0.39091500
N	3.27962800	-0.34530500	0.44331700
C	4.61571400	-0.21849500	-0.05578200
C	5.67904500	-0.62669000	0.74202600
C	4.80825600	0.30789000	-1.32873900
C	6.97382000	-0.48435000	0.25442700
H	5.50538100	-1.07580600	1.71498900
C	6.10830000	0.42982100	-1.80578400
H	3.95556200	0.60327100	-1.92996800
C	7.18901500	0.04255500	-1.01637100
H	7.81277100	-0.80182400	0.86444000
H	6.27497100	0.83334100	-2.79880700
H	8.20055700	0.14428800	-1.39530000
N	-0.42605700	-1.56741700	2.64921300
C	-1.73055800	-3.05909200	1.81959600
H	-2.26677800	-3.57966600	1.03978200
Cl	-0.19137300	-0.23454100	-2.46940600
C	-2.78254800	0.92729100	0.28162500
N	-0.78827800	1.81813200	0.11602500
N	-1.68543800	2.78544900	0.24393800
C	-1.27411300	4.15623600	0.20998300
C	-0.30586100	4.53866200	-0.71258000
C	-1.86053900	5.06220500	1.08701600
C	0.08322700	5.87308200	-0.74814100
H	0.11957300	3.80515000	-1.38924800
C	-1.46903300	6.39574500	1.02863000
H	-2.59571800	4.73250700	1.81505000
C	-0.49826600	6.79967600	0.11510400
H	0.83487200	6.19063200	-1.46291000
H	-1.91670600	7.11584700	1.70519700
H	-0.19435000	7.84045900	0.07544900
N	-3.66014400	-1.09502600	-0.77800000
C	-4.11767300	-2.41533700	-2.48072600
C	-2.75915300	-2.48792000	-2.15053700
H	-4.62888900	-2.93900300	-3.27281600
H	-1.95272900	-3.04433000	-2.60465000

Thermal correction to Gibbs Free Energy= 0.393126
E(RM06L) = -2506.64200104

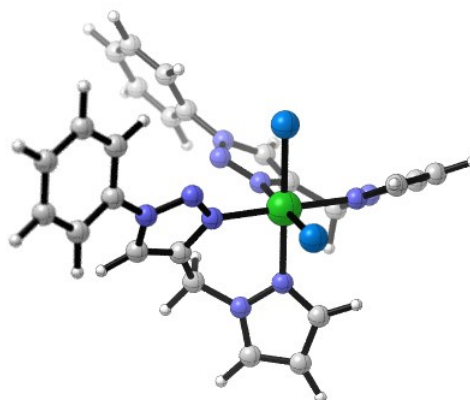


Figure S50. Calculated structure of isomer 4.

C	-0.74750400	0.82726100	4.08346500
H	-0.64515100	0.20875500	4.96306500
N	-0.63081100	1.29587500	1.91887500
C	0.05722000	-0.97215100	2.51954600
H	0.21902600	-1.51322100	3.45290600
H	-0.74267600	-1.47918600	1.96859800
C	2.52309400	-1.58502100	1.87049200
H	2.92470400	-2.20869900	2.65322900
N	1.39567100	-0.25668500	0.54972600
Rh	-0.00016500	1.16662800	-0.04367700
Cl	-1.58957100	2.74761700	-0.68347800
N	0.63017900	1.14718500	-2.01011500
C	0.74825500	0.51450400	-4.13244000
H	0.64712500	-0.16948200	-4.96229500
N	-1.39528300	-0.29882500	-0.52809900
C	-2.52091400	-1.72464900	-1.74472000
H	-2.92142800	-2.40611000	-2.47830700
C	-0.05428900	-1.16121000	-2.43572600
H	0.74547300	-1.62333400	-1.84647900
H	-0.21409500	-1.77245200	-3.32510100
Cl	1.58816600	2.79198200	0.47473000
C	-1.31166000	-1.08911500	-1.62836700
N	-2.55756900	-0.41347500	0.04265900
N	-3.24626400	-1.27718600	-0.68961400
C	-4.60192600	-1.58654800	-0.34860200
C	-5.43238000	-0.55694800	0.08179300
C	-5.04922100	-2.89795900	-0.46765800
C	-6.75037700	-0.86089300	0.40375800
H	-5.05091600	0.45609000	0.15200400
C	-6.37442900	-3.18167000	-0.15440800
H	-4.37339100	-3.68822900	-0.78065600
C	-7.22219500	-2.16642000	0.28221300
H	-7.41303000	-0.07040200	0.73944400
H	-6.73980500	-4.19928100	-0.24216400
H	-8.25412400	-2.39351700	0.52853600
N	0.39003100	0.14462800	-2.88266900
C	1.11599700	2.16384900	-2.71625400
C	1.21690000	1.80725000	-4.06655500
H	1.34773700	3.09110400	-2.21442900
H	1.57308700	2.41776000	-4.88095900
N	2.55752100	-0.41388300	-0.01168400
N	3.24716100	-1.21917600	0.78352800
C	1.31346100	-0.96071400	1.70734700
C	4.60232700	-1.55421000	0.46564000
C	5.43320800	-0.56089500	-0.04210100
C	5.04854000	-2.85329000	0.68328700
C	6.75072300	-0.88961400	-0.34100800
H	5.05255400	0.44426900	-0.18853100
C	6.37327800	-3.16112100	0.39155000
H	4.37222700	-3.61699400	1.05558700
C	7.22155300	-2.18257900	-0.12126200
H	7.41379800	-0.12740200	-0.73603900
H	6.73784200	-4.16951300	0.55580600
H	8.25310500	-2.42857100	-0.35042000
N	-0.38872400	0.36348900	2.86559600
C	-1.11862900	2.36263400	2.54522200
C	-1.21870500	2.11023700	3.91890000
H	-1.35206200	3.24839100	1.97402000
H	-1.57605100	2.78058800	4.68429700

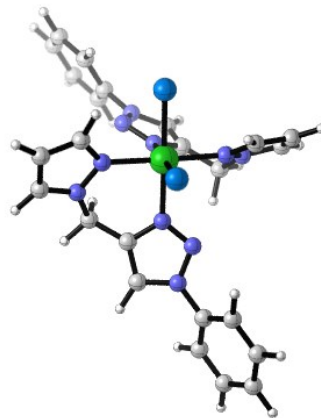


Figure S51. Calculated structure of isomer 5.

Thermal correction to Gibbs Free Energy = 0.392701
E(RM06L) = -2506.64295929

2. Reactants and Products

Phenylacetylene

C	2.20667200	0.00001000	-0.00003300
C	1.50913800	1.20556000	-0.00001600
C	0.11970600	1.20948400	0.00001600
C	-0.58768800	-0.00002300	0.00003300
C	0.11972300	-1.20949400	0.00001600
C	1.50917400	-1.20553900	-0.00001500
H	3.29252500	0.00003400	-0.00005800
H	2.04958000	2.14729700	-0.00002800
H	-0.42952800	2.14546700	0.00003200
H	-0.42946400	-2.14550500	0.00003100
H	2.04961400	-2.14727600	-0.00002800
C	-2.02175700	-0.00000900	0.00008400
C	-3.22798600	0.00001100	0.00003200
H	-4.29462300	-0.00002600	-0.00064600

Thermal correction to Gibbs Free Energy= 0.080603
E(RM06L) = -308.462724253

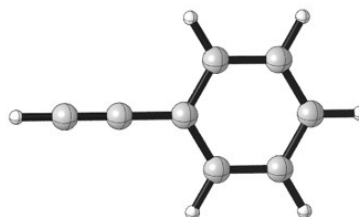


Figure S52. Calculated structure of phenylacetylene.

Triethylsilane

Si	0.00008000	0.00004200	0.27961800
H	0.00025100	0.00037700	1.78763300
C	1.72842500	0.54786200	-0.31881200
H	1.70787700	0.62937600	-1.41393700
H	1.92333500	1.56076700	0.05611500
C	-0.38954600	-1.77091400	-0.31814700
H	0.38968200	-2.44608400	0.05798200
H	-0.30763200	-1.79455600	-1.41319000
C	-1.33870000	1.22270300	-0.31860300
H	-2.31320800	0.88501400	0.05666100
H	-1.39925800	1.16400400	-1.41371300
C	2.85973400	-0.39291300	0.11431700
H	2.90425000	-0.48591500	1.20563100
H	2.72222700	-1.39959900	-0.29648600
H	3.83534000	-0.02915200	-0.22685600
C	-1.77043100	-2.27973700	0.11402000
H	-2.57293700	-1.65731600	-0.29787000
H	-1.94319800	-3.30670000	-0.22661400
H	-1.87421600	-2.27112600	1.20523400
C	-1.08962400	2.67288500	0.11431500
H	-0.14870800	3.05683200	-0.29599500
H	-1.89211400	3.33603200	-0.22742900
H	-1.03195000	2.75814500	1.20564500

Thermal correction to Gibbs Free Energy= 0.171568
E(RM06L) = -527.846054215

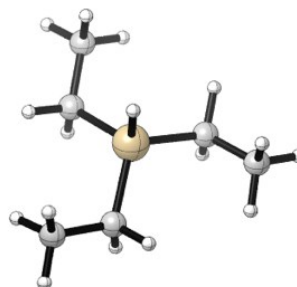


Figure S53. Calculated structure of triethylsilane.

$\beta(E)$ -Vinylsilane

Si	2.02325500	0.21999500	-0.19492000
C	3.03370300	-0.15584600	1.38902000
H	4.08740800	-0.26679900	1.09927500
H	2.99366600	0.72088200	2.04779800
C	2.74622700	-0.76151400	-1.66838300
H	3.75252300	-0.37367600	-1.87792200
H	2.13992600	-0.53217300	-2.55440300
C	2.10495400	2.08341100	-0.60924300
H	1.50154100	2.26528800	-1.50890400
H	3.13854700	2.31827600	-0.89686900
C	2.57504700	-1.39335100	2.17240900
H	1.55560700	-1.26281200	2.55175500
H	2.57994700	-2.29624700	1.55114400
H	3.22577200	-1.58307200	3.03339300
C	2.80889000	-2.27910200	-1.45912500
H	1.81894700	-2.69345400	-1.23482800
H	3.18536700	-2.79196100	-2.35135000

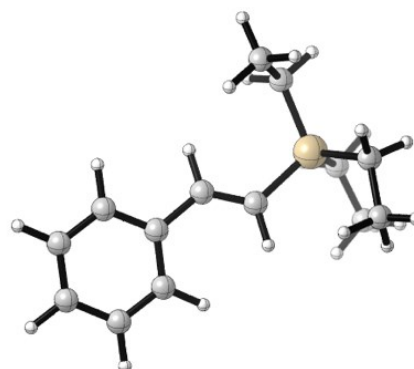


Figure S54. Calculated structure of the $\beta(E)$ -vinylsilane.

H	3.47332000	-2.53937500	-0.62724700
C	1.65680700	3.01791300	0.52129400
H	2.30338600	2.91785000	1.40016200
H	1.68934300	4.06771600	0.20889400
H	0.63371100	2.79695800	0.84511900
C	0.22383200	-0.30054000	0.06850000
H	0.04110200	-1.27756900	0.51994600
C	-0.83545100	0.44446900	-0.27774500
H	-0.66394800	1.43239600	-0.70975000
C	-2.26302100	0.09923700	-0.13097700
C	-2.70520600	-1.19106800	0.18997900
C	-3.22407600	1.09934700	-0.32192700
C	-4.05990900	-1.46468500	0.32861400
H	-1.98419900	-1.99265900	0.32043200
C	-4.58145500	0.82883700	-0.18235200
H	-2.89882800	2.10505700	-0.57800100
C	-5.00446700	-0.45570200	0.14550000
H	-4.38230900	-2.47212600	0.57540400
H	-5.30854200	1.62176300	-0.33128700
H	-6.06310400	-0.67263800	0.25273000

Thermal correction to Gibbs Free Energy= 0.278641
E(RM06L) = -836.372149089

β (Z)-Vinylsilane

Si	-1.40191300	-0.21396400	-0.28265200
C	-3.09441100	-1.09482900	-0.13316300
H	-3.42267500	-1.37931300	-1.14196800
H	-2.94368900	-2.03682400	0.41075300
C	-1.65263800	1.52282900	-1.05426300
H	-2.44973300	1.44609200	-1.80628000
H	-0.74186100	1.78336000	-1.60887400
C	-0.26927200	-1.26610700	-1.40429100
H	0.66041500	-0.71741400	-1.59328000
H	-0.77177700	-1.36616500	-2.37654400
C	-4.19049000	-0.26885000	0.55063700
H	-3.88840100	0.05535500	1.55397100
H	-4.42580900	0.63249600	-0.02673200
H	-5.11867200	-0.84160500	0.65766900
C	-1.98645500	2.63494300	-0.05238200
H	-1.19969500	2.73832800	0.70401700
H	-2.10136900	3.60361100	-0.55191700
H	-2.91825700	2.42708200	0.48485800
C	0.05248300	-2.65289500	-0.83660900
H	-0.85742400	-3.24245100	-0.67294100
H	0.69765400	-3.22165800	-1.51550700
H	0.57178700	-2.57088200	0.12443600
C	-0.72279600	-0.11989900	1.49416300
H	-1.45860200	-0.25529200	2.28962800
C	0.55398000	-0.03040600	1.88989700
H	0.79441200	-0.12862400	2.95057000
C	1.71594900	0.16682700	0.99206000
C	2.79795100	-0.71986200	1.03088500
C	1.75475700	1.23112400	0.08574300
C	3.86750700	-0.57358800	0.15426100
H	2.78604800	-1.54387800	1.73995600
C	2.82766600	1.38387100	-0.78776200
H	0.93973300	1.94857800	0.07759400
C	3.88311100	0.47655600	-0.76186400
H	4.69100300	-1.28107300	0.18480800
H	2.84024000	2.21575400	-1.48601600
H	4.71940600	0.59173500	-1.44494700

Thermal correction to Gibbs Free Energy= 0.281897
E(RM06L) = -836.368811139

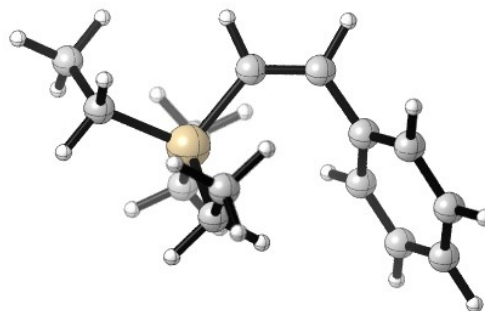


Figure S55. Calculated structure of the β (Z)-vinylsilane.

α -Vinylsilane

Si	-1.40622100	-0.02111400	0.17399400
C	-1.06184800	-0.57123400	-1.62089400
H	-0.46509100	0.20664000	-2.11433300
H	-0.42726000	-1.46550700	-1.61459900
C	-2.18618600	-1.45001700	1.17338300
H	-3.12249100	-1.73918900	0.67710400
H	-2.47462100	-1.05610300	2.15682000
C	-2.62624700	1.44729700	0.15019100
H	-2.82934900	1.75889400	1.18313200
H	-3.58567800	1.09030500	-0.24732600
C	-2.34086700	-0.84475100	-2.42145400
H	-2.94796800	-1.62886500	-1.95329100
H	-2.96638600	0.05246200	-2.49712300
H	-2.11190400	-1.17252900	-3.44145300
C	-1.28644600	-2.67714300	1.35069900
H	-0.37371700	-2.41922900	1.90011100
H	-1.79401600	-3.47163600	1.90895600
H	-0.98691600	-3.09799200	0.38290000
C	-2.12705400	2.64223600	-0.66955600
H	-1.96331300	2.36519500	-1.71777800
H	-2.84517700	3.46964200	-0.65526400
H	-1.17569000	3.01765800	-0.27611400
C	0.21648100	0.57190600	1.00271200
C	0.14799600	1.31941900	2.11204400
H	-0.80233700	1.59930100	2.56279600
C	1.54003000	0.25990200	0.40091000
C	1.88712600	-1.04705500	0.03697500
C	2.47499500	1.27857600	0.17867200
C	3.13365600	-1.33010200	-0.51090100
H	1.17956600	-1.85467600	0.20280900
C	3.72060200	0.99865800	-0.37492100
H	2.20903600	2.30127700	0.43131300
C	4.05572600	-0.30755900	-0.72053700
H	3.38521500	-2.35321500	-0.77513700
H	4.42884200	1.80522300	-0.54207700
H	5.02640200	-0.52693000	-1.15524300
H	1.04217200	1.67406600	2.62394300

Thermal correction to Gibbs Free Energy= 0.280634
E(RM06L) = -836.368446189

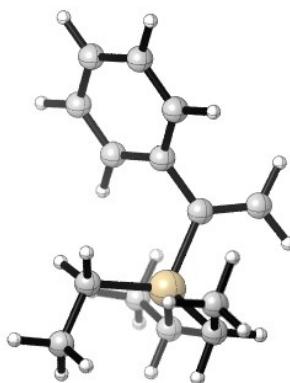


Figure S56. Calculated structure of the α -vinylsilane.

3. Rh(CO)₂ based compounds

Catalyst

C	-0.31534300	1.24738300	0.85745900
C	-1.65452700	1.53502900	0.81933800
H	-2.23248100	2.37474900	1.17121700
N	-1.31945400	-0.46540500	-0.07321500
N	-0.16649600	0.01469800	0.30571600
N	-2.22384000	0.44830900	0.23996000
C	-3.61343100	0.22154400	-0.03180100
C	-4.14700300	-1.04223400	0.19585300
C	-4.38436500	1.27580400	-0.50989700
C	-5.49685600	-1.24819900	-0.06557500
H	-3.51923500	-1.84256600	0.57150400
C	-5.73572900	1.05572600	-0.75379900
H	-3.93723900	2.24403100	-0.71349000
C	-6.29029900	-0.20276800	-0.53342700
H	-5.93008500	-2.22772000	0.10540700
H	-6.35081500	1.86636400	-1.12915900
H	-7.34389800	-0.37036500	-0.73017800
C	0.86642200	1.98894400	1.39648800
H	0.57866600	2.98968600	1.72086200
H	1.28623800	1.45844500	2.25828800
N	1.90662900	2.13561600	0.38968000
C	2.49459700	3.26862900	-0.04524200
C	3.46311200	2.91484300	-0.96157100
H	2.18952300	4.23352400	0.33259400
C	3.40089500	1.52088700	-1.02466200
H	4.12241900	3.57170000	-1.50651200
H	3.98169200	0.83736400	-1.62657100
N	2.45827800	1.05270200	-0.20063200
Rh	1.69077000	-0.91253500	-0.00016600
C	0.91725400	-2.62320300	0.13635100
C	3.40534700	-1.66226100	-0.20107800
O	0.42178100	-3.64525500	0.21898300
O	4.45625700	-2.08891200	-0.31244100

Thermal correction to Gibbs Free Energy= 0.194631
E(RM06L) = -1075.11385799

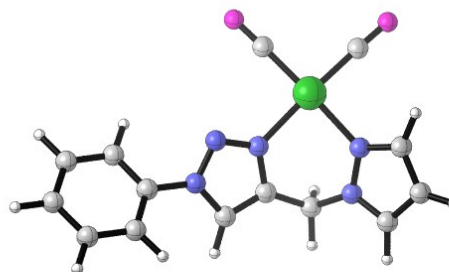


Figure S57. Calculated structure of Rh^IC₀Ph.

H-Rh-Si intermediate

C	-0.65535700	2.06206300	0.92779400
C	-2.02205200	2.18730600	0.93604700
H	-2.68371400	2.94665400	1.32176700
N	-1.47069800	0.26142500	-0.00642400
N	-0.37400000	0.86462700	0.35086700
N	-2.47462900	1.05164700	0.35262500
C	-3.82907200	0.65173600	0.11498300
C	-4.17929500	-0.68118500	0.30265300
C	-4.75341200	1.60592900	-0.29632300
C	-5.49605200	-1.06161100	0.07048400
H	-3.43537800	-1.40056200	0.62681500
C	-6.06984500	1.21153700	-0.51126200
H	-4.45009100	2.63376100	-0.47052800
C	-6.44041800	-0.11862900	-0.32996800
H	-5.78561600	-2.09736400	0.21193600
H	-6.80188500	1.94399100	-0.83393700
H	-7.46735300	-0.42191400	-0.50419300
C	0.42543700	2.95700300	1.44788100
H	-0.00784400	3.87268700	1.85310600
H	0.96867700	2.46065800	2.25978000
N	1.37924400	3.34408600	0.42035100
C	1.75752200	4.59314800	0.06565500
C	2.76518500	4.47136700	-0.86604200
H	1.29207200	5.46361700	0.50455400
C	2.94409200	3.08944500	-1.01638200
H	3.29702700	5.26642400	-1.36457000
H	3.63376000	2.55117800	-1.65062000

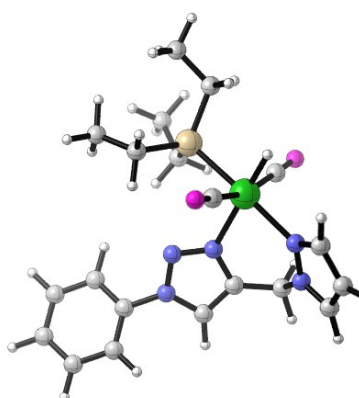


Figure S58. Calculated structure of the H-Rh-Si intermediate.

N	2.10605600	2.41421100	-0.23276100
Rh	1.63915700	0.19516400	-0.17266000
C	1.28113700	0.08909300	-2.06766100
O	1.14519300	0.04764900	-3.19553000
Si	1.03724700	-2.21084400	-0.05496800
C	0.00171900	-2.43221900	1.54968300
H	0.32589000	-1.71786000	2.31574000
H	-1.03334500	-2.15901500	1.30906600
C	2.65664100	-3.22551200	-0.00842200
H	3.36817700	-2.75666500	-0.69907000
H	3.09986300	-3.12255700	0.99149900
C	-0.04959200	-2.67204400	-1.56330400
H	-0.72152200	-1.83275500	-1.77920100
H	0.59572200	-2.79519300	-2.44275600
C	0.05126500	-3.84986900	2.13781200
H	-0.56782700	-3.91836000	3.03841600
H	-0.31460800	-4.59926100	1.42880900
H	1.07220600	-4.12966000	2.41840600
C	-0.89659700	-3.93653600	-1.35614100
H	-1.59932500	-3.81197200	-0.52478000
H	-1.48465100	-4.15588100	-2.25320500
H	-0.28164400	-4.81623000	-1.14262200
C	2.50724700	-4.71074200	-0.36796600
H	2.12731300	-4.83618400	-1.38763400
H	3.47535900	-5.21887500	-0.31427100
H	1.82430400	-5.23253800	0.30973500
C	2.35088400	-0.10694000	1.58652200
O	2.84703800	-0.29640900	2.59385400
H	3.07463300	-0.24862800	-0.59844600

Thermal correction to Gibbs Free Energy= 0.391147
E(RM06L) = -1602.95322855

C^{Si}-(E) intermediate

C	-2.34985000	-1.70823400	0.96676600
C	-3.70357500	-1.48672200	0.98808500
H	-4.50630700	-1.90747800	1.57260200
N	-2.77684500	-0.23347500	-0.58778900
N	-1.83594200	-0.92564500	-0.01476500
N	-3.91431100	-0.57350300	0.00928300
C	-5.15724600	0.00414300	-0.40701800
C	-5.39276800	0.19168300	-1.76471700
C	-6.09356700	0.35688900	0.55916300
C	-6.60435200	0.74855100	-2.15837500
H	-4.64295800	-0.09495400	-2.49355400
C	-7.30611800	0.90010400	0.14788800
H	-5.87373400	0.23425700	1.61531900
C	-7.56089300	1.09675100	-1.20710300
H	-6.80319000	0.90238400	-3.21355800
H	-8.04547100	1.18176100	0.88989900
H	-8.50613500	1.52575100	-1.52248300
C	-1.46684300	-2.61852300	1.75759800
H	-2.04322900	-3.13459700	2.52660800
H	-1.02758600	-3.38333900	1.10738700
N	-0.39304900	-1.90553900	2.43309500
C	-0.08242900	-1.95317400	3.74680400
C	1.05672800	-1.20206500	3.93585700
H	-0.68813100	-2.51909900	4.43948200
C	1.38416200	-0.72280100	2.66366200
H	1.57689200	-1.01973300	4.86284800
H	2.19539600	-0.08427300	2.34774400
N	0.50923900	-1.15721600	1.75722700
Rh	0.33451800	-0.58460100	-0.30280800
C	0.24015600	0.23149600	-1.98914300
C	0.61891000	-2.56173600	-1.02879000
O	0.18063200	0.73537100	-3.00719700
O	0.71001700	-3.60681800	-1.45633100
H	0.12031100	0.84773200	0.23526700
C	2.34633700	-0.12296500	-0.34329900
C	3.17040000	-1.29649700	-0.75233500
C	3.58682400	-1.42156200	-2.08340400
C	3.50846700	-2.30517600	0.15901100
C	4.30934500	-2.53669700	-2.49697800
H	3.34314800	-0.63461700	-2.79268800

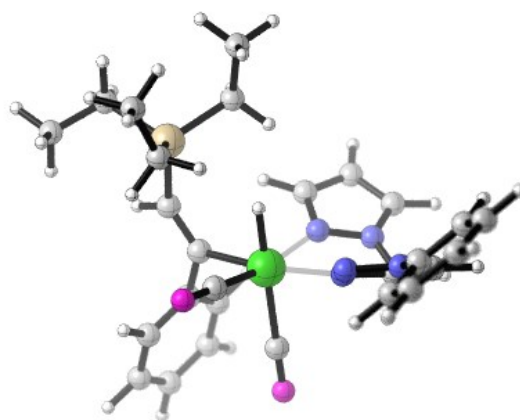


Figure S59. Calculated structure of the C^{Si}-(E) intermediate.

C	4.23677300	-3.41631200	-0.25502500
H	3.20006800	-2.21700500	1.19701700
C	4.63123300	-3.53974900	-1.58533700
H	4.62684500	-2.61951200	-3.53184900
H	4.50044500	-4.18641700	0.46364800
H	5.19553600	-4.40880200	-1.90807200
C	2.90169700	1.05857700	-0.05003100
H	3.99708800	1.04699900	-0.08564500
Si	2.18950500	2.77253000	0.40074600
C	3.69864800	3.93409300	0.46714600
H	3.36136700	4.93365600	0.76982900
H	4.36859600	3.58997000	1.26633000
C	1.35735400	2.71107200	2.12490500
H	0.61200800	1.90533300	2.14366600
H	2.12851700	2.43060700	2.85594000
C	4.46925400	4.02569900	-0.85687000
H	5.32032000	4.70938100	-0.77509400
H	4.86183400	3.04970100	-1.16478400
H	3.83146900	4.39025700	-1.67034600
C	0.69484100	4.02593300	2.55603800
H	0.32080600	3.96295200	3.58356200
H	1.39570900	4.86689500	2.51365200
H	-0.15593900	4.27101000	1.91115900
C	0.97051600	3.40568700	-0.93653100
H	1.32117300	3.05184600	-1.91434500
H	-0.01546300	2.94880100	-0.78000300
C	0.82299000	4.93399600	-0.97857000
H	1.77791800	5.42255800	-1.19788000
H	0.11343700	5.23648900	-1.75587000
H	0.46207700	5.33755900	-0.02739100
Thermal correction to Gibbs Free Energy= 0.495328			
E(RM06L) = -1911.43812412			

C^{Si}-(Z) intermediate

C	-3.10057800	-1.52347700	0.23425000
C	-4.44459900	-1.32628000	0.42557900
H	-5.24810400	-1.98415800	0.71660300
N	-3.51049200	0.56850500	-0.24567200
N	-2.58188300	-0.33972700	-0.17981500
N	-4.64531800	-0.02298400	0.11314800
C	-5.87641800	0.70908100	0.12859300
C	-6.13747400	1.61184700	-0.89648400
C	-6.77651800	0.49202800	1.16656800
C	-7.33770100	2.31324700	-0.87484500
H	-5.41546200	1.75790600	-1.69206200
C	-7.97863500	1.19159700	1.16606000
H	-6.53588900	-0.19117000	1.97536700
C	-8.25852200	2.10079200	0.14892100
H	-7.55594000	3.02210100	-1.66638100
H	-8.68971700	1.03435700	1.96994300
H	-9.19519200	2.64814100	0.15607000
C	-2.23373800	-2.73482400	0.36794700
H	-2.81090700	-3.57602600	0.75422900
H	-1.83593700	-3.03038000	-0.60912100
N	-1.12077200	-2.52143100	1.28049300
C	-0.80821800	-3.24280600	2.37877100
C	0.34580900	-2.71227500	2.91315000
H	-1.42265000	-4.07527700	2.68920600
C	0.67692900	-1.64885000	2.06723700
H	0.87356900	-3.04229300	3.79393500
H	1.50117400	-0.95199900	2.11392900
N	-0.20920100	-1.54431700	1.07632400
Rh	-0.40028100	0.04060700	-0.36059000
C	-0.45657700	1.68093300	-1.27111700
C	-0.28098300	-1.19354200	-2.06200800
O	-0.48091000	2.69772600	-1.78065000
O	-0.28043000	-1.80879700	-3.01361500
H	-0.50368700	0.95591200	0.89004800
C	1.64193800	0.34450200	-0.26445100
C	2.40506400	-0.47257200	-1.24602600
C	2.94082100	0.13861500	-2.38568300
C	2.59511800	-1.84878400	-1.06522200
C	3.66071800	-0.60567600	-3.31639900

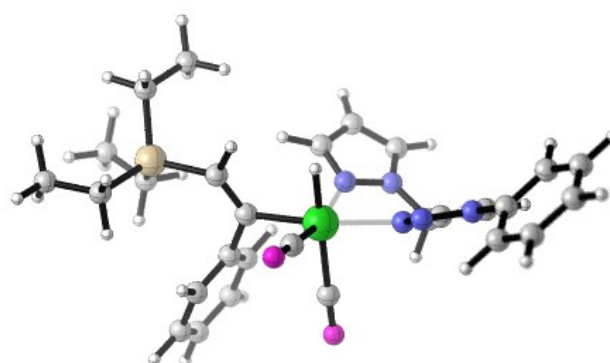


Figure S60. Calculated structure of the C^{Si}-(Z) intermediate.

H	2.79591100	1.20558700	-2.53243200
C	3.31946100	-2.59004400	-1.99203600
H	2.19083200	-2.33371800	-0.18088200
C	3.85271900	-1.97112900	-3.12122000
H	4.07577000	-0.11693700	-4.19237700
H	3.47585700	-3.65218700	-1.83004200
H	4.41796100	-2.55058100	-3.84411700
C	2.22619600	1.18439100	0.59669900
H	1.58666500	1.79520200	1.23537600
Si	4.10469800	1.42657600	0.88310800
C	4.24012400	1.99156000	2.70128700
H	5.29775900	2.15171700	2.94500000
H	3.75556900	2.97077000	2.80846900
C	4.73169900	2.79130200	-0.29461900
H	4.54945000	2.46699000	-1.32698700
H	4.10672400	3.68097000	-0.14032200
C	3.62589500	0.99253400	3.69017200
H	3.75335400	1.31989900	4.72730300
H	2.54836900	0.87318700	3.51703200
H	4.08828800	0.00184200	3.59893700
C	6.21187800	3.15641400	-0.12113100
H	6.48869500	3.99731000	-0.76562000
H	6.44067500	3.44478100	0.91099800
H	6.86269400	2.31569100	-0.38369100
C	5.06594500	-0.21328300	0.66618200
H	4.43364000	-1.03771700	1.02067700
H	5.21672700	-0.39863400	-0.40452700
C	6.41196500	-0.24952800	1.40343800
H	6.28136700	-0.14113700	2.48594700
H	6.92910600	-1.19884000	1.22810500
H	7.08066600	0.55289200	1.07551400

Thermal correction to Gibbs Free Energy= 0.494467
E(RM06L) = -1911.44193550

C^{Si}- α intermediate

C	-3.14757600	-1.64932600	-0.71201500
C	-4.47359200	-1.34612700	-0.53413700
H	-5.37795600	-1.92121100	-0.65469400
N	-3.25029700	0.45841300	-0.13549600
N	-2.44719200	-0.51134700	-0.46452800
N	-4.48170200	-0.03755500	-0.18226900
C	-5.61205600	0.79171800	0.11191300
C	-5.62033900	2.10620300	-0.34208500
C	-6.66950100	0.25624200	0.83985700
C	-6.72462100	2.90081900	-0.05539000
H	-4.78027800	2.49362900	-0.90761900
C	-7.77246500	1.06048500	1.10629900
H	-6.62591500	-0.76184100	1.21463800
C	-7.80001300	2.37980000	0.66102200
H	-6.74650000	3.92863200	-0.40143700
H	-8.60406200	0.65741000	1.67423700
H	-8.66038300	3.00471100	0.87600700
C	-2.46497100	-2.91206600	-1.13681400
H	-3.19526200	-3.71361000	-1.25728700
H	-1.96882700	-2.76907900	-2.10318200
N	-1.47515400	-3.36682800	-0.17247000
C	-1.41018400	-4.56355100	0.45269500
C	-0.26276100	-4.57070800	1.21599800
H	-2.17242500	-5.31403500	0.30224300
C	0.32178900	-3.31804500	0.98985100
H	0.10055100	-5.36684900	1.84651100
H	1.23129700	-2.90045600	1.39890600
N	-0.41357700	-2.59801800	0.14410600
Rh	-0.25039700	-0.42401900	-0.38701800
C	-0.22764600	0.17386500	1.46960300
C	0.06310200	-0.56906900	-2.28313600
O	-0.18832000	0.55319200	2.53725800
O	0.30031800	-0.57383500	-3.39454400
H	-0.15809400	1.09056200	-0.69414900
C	1.80024800	-0.49512600	-0.26280900
C	2.67730700	0.49984700	-0.12419400
Si	4.55269800	0.11487000	0.06364400
C	5.04473800	0.57204800	1.84929100

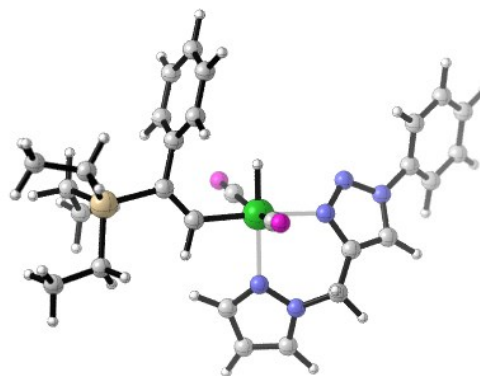


Figure S61. Calculated structure of the C^{Si}- α intermediate.

H	6.12472700	0.41271700	1.96204700
H	4.88575800	1.64836400	1.99259100
C	5.51350900	1.17180500	-1.20041700
H	5.48813900	0.65957300	-2.17135000
H	4.96992600	2.11489400	-1.34089900
C	4.28800900	-0.22615300	2.91660100
H	4.56559100	0.08991200	3.92750700
H	3.20168500	-0.10240000	2.81670900
H	4.50301800	-1.29839600	2.83778300
C	6.96476800	1.47630100	-0.80199700
H	7.46770300	2.06639400	-1.57504400
H	7.00761900	2.04977700	0.13020300
H	7.55075400	0.56300500	-0.65237600
C	4.82830700	-1.75397300	-0.21370700
H	4.23672300	-2.31314100	0.52439600
H	4.43170700	-2.03144200	-1.19947600
C	6.30067500	-2.17499500	-0.11170500
H	6.73270800	-1.89863100	0.85691200
H	6.41348100	-3.25824000	-0.22512300
H	6.90495000	-1.69955700	-0.89149200
C	2.28479900	1.93383600	-0.08209800
C	2.02135800	2.63101800	-1.26708100
C	2.18583900	2.61415000	1.13581300
C	1.64411900	3.97037400	-1.23284200
H	2.12099900	2.11567100	-2.21906300
C	1.80242800	3.95284800	1.16995900
H	2.40589800	2.08980200	2.06213300
C	1.52862300	4.63381000	-0.01319900
H	1.44619800	4.49821600	-2.16094500
H	1.72454500	4.46561200	2.12393200
H	1.23743000	5.67910300	0.01388300
H	2.14447300	-1.53014100	-0.32048600

Thermal correction to Gibbs Free Energy= 0.492320
E(RM06L) = -1911.44390772

C^H-(E) intermediate

C	-2.36689000	1.94375300	0.82324500
C	-3.70219400	1.84906200	0.52161900
H	-4.53940200	2.50534100	0.69945900
N	-2.67180500	0.00627800	-0.14351400
N	-1.78627800	0.78567700	0.40771900
N	-3.83637200	0.63729000	-0.06861200
C	-5.02851800	0.01213800	-0.55695400
C	-5.21491800	-1.34681700	-0.32625100
C	-5.96417000	0.78011000	-1.24139500
C	-6.37606400	-1.94741000	-0.79931400
H	-4.46672200	-1.91510400	0.21580500
C	-7.12666500	0.16650200	-1.69631200
H	-5.78121100	1.83206100	-1.43827800
C	-7.33170700	-1.19356100	-1.47755000
H	-6.53748500	-3.00639600	-0.62860500
H	-7.86554600	0.75131100	-2.23351700
H	-8.23787700	-1.66809000	-1.83903200
C	-1.58696900	3.03071700	1.49528100
H	-2.24338700	3.87042800	1.72898000
H	-1.16511400	2.66974300	2.43986100
N	-0.50610500	3.53818200	0.66647400
C	-0.25398300	4.81815500	0.30807200
C	0.93791800	4.82468700	-0.38367800
H	-0.92957900	5.61744800	0.57626700
C	1.35243600	3.48491300	-0.39404200
H	1.43752800	5.67568300	-0.81938700
H	2.23338900	3.03318900	-0.82928700
N	0.47601500	2.71635400	0.24763900
Rh	0.39345400	0.42888200	0.34155400
C	0.25495100	0.29457100	-1.58972000
C	0.55758100	0.28318300	2.25619100
O	0.13013500	0.28240600	-2.71909100
O	0.63817300	0.22903700	3.39032300
C	2.41974000	0.34350900	0.20422300
C	3.17374500	-0.00631600	-0.84692900
H	2.71074200	-0.42765100	-1.73947700
Si	0.50076000	-2.06185500	0.47743300

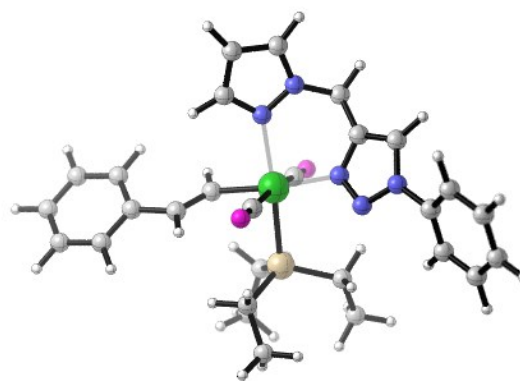


Figure S62. Calculated structure of the C^H-(E) intermediate.

C	1.80659500	-2.55514200	1.79362800
H	1.35685000	-2.44478500	2.78940200
H	2.63642200	-1.84128800	1.74408700
C	0.94763600	-2.83203900	-1.22019600
H	2.03289600	-2.74469400	-1.35466700
H	0.48967900	-2.25629100	-2.03145600
C	-1.24167900	-2.63232400	1.04673300
H	-1.90295700	-2.63673700	0.17177900
H	-1.65626700	-1.87894000	1.72866100
C	-1.25869700	-4.00299200	1.73957100
H	-2.27596000	-4.27162700	2.04458500
H	-0.89263000	-4.79871000	1.08358000
H	-0.63624400	-4.00451800	2.64068900
C	2.36783000	-3.97594300	1.64237000
H	3.10772400	-4.17942300	2.42313300
H	1.58663000	-4.73868000	1.72325900
H	2.86730300	-4.10644500	0.67662700
C	0.51491100	-4.29835500	-1.37185200
H	-0.57295600	-4.40164200	-1.29687600
H	0.81493100	-4.68581600	-2.35091900
H	0.96565000	-4.94314300	-0.61186100
H	2.93376400	0.69461600	1.10105100
C	4.64768500	0.06601700	-0.91348200
C	5.31933600	-0.73716300	-1.84271400
C	5.40369400	0.90600900	-0.08577400
C	6.70740000	-0.72320500	-1.92711800
H	4.74610700	-1.38597100	-2.50116400
C	6.79054300	0.92103900	-0.16937000
H	4.90561900	1.56739000	0.61822600
C	7.44778300	0.10426000	-1.08764900
H	7.21064900	-1.35741500	-2.65033900
H	7.36167400	1.57978800	0.47774000
H	8.53106500	0.12049600	-1.15389900

Thermal correction to Gibbs Free Energy= 0.497867
E(RM06L) = -1911.46047864

C^H- α intermediate

C	1.73760800	2.00125500	-0.89387300
C	3.10596800	2.02893300	-0.99117100
H	3.79532000	2.75783000	-1.38703300
N	2.47566400	0.10355000	-0.10024500
N	1.40428100	0.79677000	-0.35623600
N	3.50982300	0.83614500	-0.49188300
C	4.84051200	0.31906000	-0.38138700
C	5.06062900	-1.02601200	-0.65951400
C	5.87093500	1.17313000	-0.00394000
C	6.35490500	-1.52244900	-0.55439100
H	4.23406700	-1.66397400	-0.95363700
C	7.16219000	0.66326400	0.08336200
H	5.66970400	2.21142900	0.24151600
C	7.40383300	-0.68085200	-0.18964700
H	6.54434900	-2.56911000	-0.76719400
H	7.97658600	1.31626900	0.37816100
H	8.41206300	-1.07440900	-0.11501200
C	0.70172300	3.01701100	-1.26126900
H	1.17777700	3.90935100	-1.67066200
H	0.02777500	2.62028400	-2.02812400
N	-0.08855500	3.42834500	-0.11258900
C	-0.27492200	4.67420000	0.37954000
C	-1.14683900	4.57015000	1.44224200
H	0.21518900	5.52942500	-0.06265100
C	-1.44559000	3.20294000	1.52640900
H	-1.51602700	5.36825300	2.06702300
H	-2.09689400	2.67405200	2.20873900
N	-0.80419300	2.52140100	0.58013900
Rh	-0.65321500	0.24589800	0.30559400
C	0.08652300	0.07508400	2.09112600
C	-1.23380300	0.08453800	-1.53461100
O	0.59699500	0.03727700	3.10612500
O	-1.45527400	-0.00251300	-2.64780500
C	-2.56870500	0.00201700	1.00966300
C	-3.69864900	0.29230500	0.07947600
C	-4.74775900	-0.63233100	-0.02926200

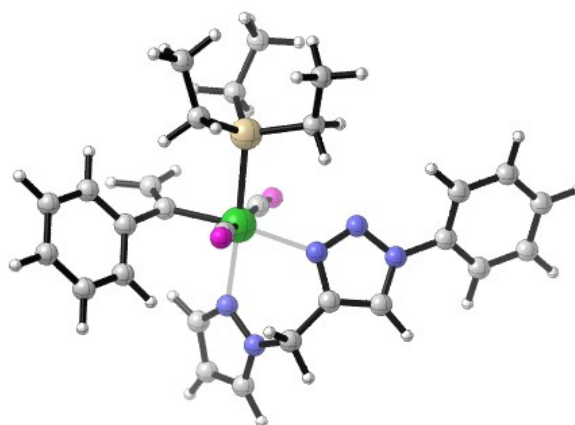


Figure S63. Calculated structure of the C^H- α intermediate.

C	-3.76834900	1.46327300	-0.68814300
C	-5.82172800	-0.40035800	-0.88228600
H	-4.70807700	-1.54708400	0.55565400
C	-4.84601300	1.69887700	-1.53568400
H	-2.98288500	2.20797300	-0.60477900
C	-5.87387400	0.76556200	-1.64099900
H	-6.61789900	-1.13470300	-0.95543300
H	-4.88579700	2.61839900	-2.11192500
H	-6.71102600	0.94730600	-2.30734100
C	-2.85778100	-0.33879300	2.27390000
H	-3.89055200	-0.33985500	2.61962200
H	-2.11976800	-0.64107300	3.00981400
Si	-0.59787400	-2.25075900	0.13467000
C	-2.01724700	-2.84646100	-1.00923300
H	-1.67328000	-2.76221300	-2.04867700
H	-2.87168800	-2.16808900	-0.91878600
C	-0.75671800	-3.05511200	1.86854700
H	-1.81983200	-3.05835900	2.13684500
H	-0.25963100	-2.43550400	2.62239800
C	1.07889900	-2.69392200	-0.68589100
H	1.87293000	-2.60560600	0.06530200
H	1.30422500	-1.93579200	-1.44641600
C	1.10805100	-4.08378900	-1.33974700
H	2.08392300	-4.27539100	-1.79938900
H	0.92500000	-4.88371700	-0.61627100
H	0.35367200	-4.17183700	-2.12879700
C	-2.49184400	-4.28177200	-0.74489700
H	-3.31124200	-4.54452300	-1.42187400
H	-1.69210100	-5.01351200	-0.89884900
H	-2.86024000	-4.40102800	0.27979900
C	-0.18258200	-4.47639100	1.96123700
H	0.89250500	-4.48716000	1.75205900
H	-0.32523500	-4.88019800	2.96889500
H	-0.66713900	-5.16395400	1.26172500
Thermal correction to Gibbs Free Energy= 0.500394			
E(RM06L) = -1911.45875492			

Reductive elimination from C^{Si}-(E)

C	-0.99387000	-2.87626000	-0.08973200
C	-2.30408100	-2.90552800	0.31950000
H	-2.87287200	-3.61444800	0.89936300
N	-1.94800400	-1.09953900	-0.91250100
N	-0.83193300	-1.76202400	-0.84242100
N	-2.84953500	-1.78545200	-0.21579900
C	-4.17194800	-1.25967200	-0.06406700
C	-4.68684500	-0.43943300	-1.06312000
C	-4.88371100	-1.53277400	1.09922300
C	-5.94631200	0.12068300	-0.88217400
H	-4.10787200	-0.23820700	-1.95687600
C	-6.14905500	-0.97790200	1.25701400
H	-4.45137700	-2.13774000	1.88973700
C	-6.67902100	-0.14917900	0.27162100
H	-6.35667000	0.76640200	-1.65115400
H	-6.71226200	-1.18156100	2.16139500
H	-7.66239100	0.28926300	0.40429900
C	0.18446000	-3.75338400	0.16487000
H	-0.11337100	-4.68442800	0.64863600
H	0.67164500	-4.00553100	-0.78313700
N	1.16186200	-3.11447100	1.04047600
C	1.70459700	-3.64148800	2.15829000
C	2.65546900	-2.76011600	2.62233400
H	1.37876700	-4.60156100	2.53111500
C	2.63460100	-1.70078500	1.71064300
H	3.27952300	-2.86492000	3.49560000
H	3.22489600	-0.79686800	1.69647700
N	1.73082300	-1.91766100	0.75409700
Rh	1.15086500	-0.63585200	-0.87726900
C	0.69823300	0.54821000	-2.25497000
C	2.36436500	-1.71258800	-2.09444700
O	0.46747800	1.23133600	-3.13622500
O	3.11810600	-2.33528000	-2.67574900
H	0.59380400	0.25214100	0.29378900
C	1.87071200	1.21848300	0.02834800

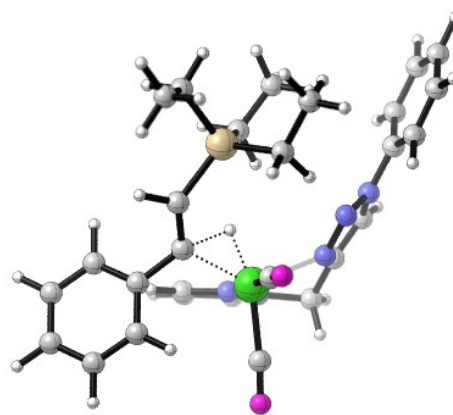


Figure S64. Calculated structure of the reductive elimination intermediate from C^{Si}-(E).

C	3.34443600	1.20242100	0.24057000
C	4.25314900	0.84705600	-0.76309300
C	3.84808600	1.59884700	1.48793000
C	5.62243400	0.88836800	-0.52989300
H	3.88535400	0.55954400	-1.74324800
C	5.21966100	1.62507000	1.72695500
H	3.15227300	1.87588200	2.27521800
C	6.11063400	1.26770200	0.71945600
H	6.31164700	0.62597000	-1.32648400
H	5.59015900	1.92806900	2.70137000
H	7.17999700	1.28821700	0.90335400
C	1.14394100	2.31535200	0.31604200
H	1.75060800	3.18195600	0.59885500
Si	-0.74364200	2.59028000	0.53104700
C	-0.90909100	4.33917600	1.26403600
H	-1.95932900	4.51061500	1.53288700
H	-0.34825800	4.38048600	2.20695700
C	-1.36143400	1.28942300	1.79957400
H	-1.12241300	0.27970900	1.43846900
H	-0.77291600	1.42649700	2.71669700
C	-0.42908700	5.44705000	0.31661300
H	-0.53891400	6.43563000	0.77351700
H	0.62769700	5.32489700	0.05170800
H	-1.00028000	5.45098000	-0.61880000
C	-2.85870700	1.36978100	2.12354600
H	-3.13088800	0.66464700	2.91709800
H	-3.14610200	2.37064300	2.46368000
H	-3.47087300	1.12914400	1.24846300
C	-1.72546700	2.49209000	-1.10702700
H	-1.13866700	2.98182500	-1.89357700
H	-1.82531300	1.43920000	-1.39720600
C	-3.11864000	3.13646000	-1.04487500
H	-3.06258300	4.19526800	-0.77056900
H	-3.61502400	3.07842200	-2.01984700
H	-3.76817000	2.64248400	-0.31553200

Thermal correction to Gibbs Free Energy= 0.494627

E(RM06L) = -1911.40821001

Reductive elimination from $C^{Si}-(Z)$

C	-3.27992200	-1.36411200	-0.61507500
C	-4.60891600	-1.08141900	-0.42831000
H	-5.50899000	-1.65664900	-0.57640300
N	-3.39833300	0.71097000	0.05917600
N	-2.58795200	-0.23922600	-0.30699700
N	-4.62709400	0.21023600	-0.01749900
C	-5.76446500	1.01683500	0.30946300
C	-5.78472100	2.34856000	-0.09084000
C	-6.81785300	0.44334600	1.01381500
C	-6.89617000	3.12129800	0.22620700
H	-4.94794100	2.76637100	-0.63932200
C	-7.92838100	1.22624000	1.31097500
H	-6.76497200	-0.58844800	1.34753800
C	-7.96755800	2.56214700	0.91949400
H	-6.92678900	4.16207900	-0.07790500
H	-8.75677800	0.79301400	1.86114700
H	-8.83366900	3.17017000	1.15855200
C	-2.55152100	-2.58782000	-1.06564400
H	-3.24063100	-3.42244200	-1.20055500
H	-2.05455500	-2.40822200	-2.02416400
N	-1.56090900	-3.00143200	-0.08498700
C	-1.55964500	-4.15323300	0.61912900
C	-0.51486300	-4.09722800	1.51308400
H	-2.29469600	-4.92068900	0.42539300
C	0.07475200	-2.84986400	1.29308800
H	-0.21531800	-4.84960700	2.22511500
H	0.93103400	-2.41143600	1.77934400
N	-0.55476100	-2.18657400	0.32220900
Rh	-0.37041100	-0.13987700	-0.35857100
C	-0.49964200	1.69840100	-0.66412400
C	-0.13925300	-0.59572000	-2.40799600
O	-0.66708400	2.81030700	-0.84990300
O	0.28929400	-0.61964200	-3.46165300
H	0.40761400	0.20993100	1.01744000

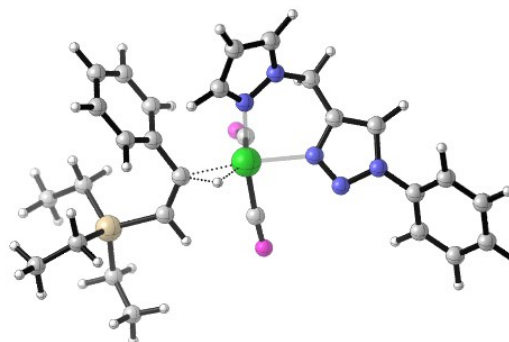


Figure S65. Calculated structure of the reductive elimination intermediate from $C^{Si}-(Z)$.

C	1.69633700	0.11725600	0.17411900
C	2.49125400	-1.13535200	0.29204000
C	2.62160700	-2.04759900	-0.75790900
C	3.15911000	-1.39394700	1.49536400
C	3.41174300	-3.18242100	-0.61609300
H	2.12075400	-1.85808100	-1.70183900
C	3.94404700	-2.53465200	1.64290300
H	3.05383600	-0.69260700	2.31841800
C	4.07231000	-3.43163500	0.58580600
H	3.51865100	-3.87209200	-1.44761800
H	4.45723400	-2.71901000	2.58165800
H	4.68712700	-4.31935500	0.69514200
C	2.30336100	1.31653300	0.23898600
H	1.67869900	2.20781500	0.28998500
Si	4.18336600	1.69940800	0.16946200
C	4.28985600	3.45458100	-0.57385300
H	5.35068600	3.69628400	-0.71825400
H	3.84833600	3.44354000	-1.57919500
C	5.09188200	0.44638500	-0.95159600
H	5.26824200	-0.47546600	-0.38377100
H	4.41794300	0.16790200	-1.77207100
C	3.63066500	4.54368400	0.28158700
H	3.74481100	5.53155700	-0.17649000
H	2.55535900	4.37024800	0.41047700
H	4.07534300	4.59170200	1.28213100
C	6.41405500	0.96809900	-1.53144300
H	6.89548200	0.20536600	-2.15244200
H	6.25703500	1.85218200	-2.15860300
H	7.12473800	1.24515100	-0.74562600
C	4.89414700	1.73472100	1.94474700
H	4.18036300	2.25955300	2.59323500
H	4.94904100	0.70543200	2.31944000
C	6.27495200	2.39589300	2.05130000
H	6.26017700	3.42687200	1.68065900
H	6.61757200	2.42563600	3.09091900
H	7.02828900	1.84903700	1.47467900

Thermal correction to Gibbs Free Energy= 0.490605

E(RM06L) = -1911.41768484

Reductive elimination from C^{Si}- α

C	3.10986100	1.78283100	0.02875400
C	4.43343800	1.43777900	0.13130300
H	5.31114800	1.99339200	0.42120400
N	3.28552700	-0.30207900	-0.61541500
N	2.45374900	0.68583200	-0.43849400
N	4.48728300	0.14737400	-0.27721900
C	5.63642900	-0.70212400	-0.37338200
C	5.75183900	-1.56152800	-1.46061100
C	6.60543100	-0.63858800	0.62243500
C	6.87461100	-2.37715900	-1.54617900
H	4.97905300	-1.58647700	-2.22075900
C	7.72883600	-1.45140500	0.51473200
H	6.47721600	0.01440300	1.48030900
C	7.86313000	-2.31940300	-0.56603600
H	6.97966900	-3.05361500	-2.38762300
H	8.49229400	-1.41396200	1.28435700
H	8.73894800	-2.95504200	-0.64300500
C	2.41936000	3.08539600	0.29270300
H	3.13027000	3.80626600	0.69990900
H	2.02464900	3.50152500	-0.64094100
N	1.32719100	2.95732000	1.23949500
C	1.12665800	3.63131100	2.39560700
C	-0.09597900	3.23388300	2.89240000
H	1.85467800	4.33784600	2.76696800
C	-0.57536900	2.30645800	1.95464200
H	-0.57336600	3.56773600	3.80034200
H	-1.49700200	1.74188200	1.94791900
N	0.28827800	2.15069800	0.95592600
Rh	0.26097300	0.43691300	-0.66743800
C	0.47339300	-0.94163700	0.72158100
C	0.36089600	1.49572200	-2.26915600
O	0.84585200	-1.67367900	1.50542200
O	0.48671400	2.09280300	-3.23022400
H	-0.63207700	-0.60392100	-1.51935900
C	-1.78771200	0.18878800	-0.99506000
C	-2.68953600	-0.54955800	-0.32673100
Si	-4.57719800	-0.27484300	-0.61955500
C	-5.37408800	0.20908200	1.04336500
H	-6.46080000	0.25050400	0.89568500
H	-5.20367300	-0.59812400	1.76678100
C	-5.28775400	-1.92934500	-1.25097100
H	-5.05717800	-2.01713000	-2.32077800
H	-4.74626200	-2.74564700	-0.75570400
C	-4.88015700	1.54726200	1.60223300
H	-5.33642600	1.77176700	2.57187600
H	-3.79111300	1.54887400	1.73917900
H	-5.11861700	2.37516700	0.92510300
C	-6.79505300	-2.10710200	-1.01893400
H	-7.14200400	-3.06359300	-1.42298300
H	-7.03666000	-2.09404000	0.04927100
H	-7.38160600	-1.31631300	-1.49850300
C	-4.79973200	1.14466600	-1.87332400
H	-4.26568400	2.02954800	-1.50083800
H	-4.31706200	0.86860300	-2.82023600
C	-6.26910400	1.50636800	-2.13264600
H	-6.78822200	1.77577900	-1.20576100
H	-6.35111500	2.35846200	-2.81499900
H	-6.81243300	0.67086000	-2.58602300
C	-2.32441600	-1.64761200	0.60059700
C	-1.91184700	-2.88746500	0.09464200
C	-2.42917000	-1.48282200	1.98733100
C	-1.58219300	-3.92750900	0.95814000
H	-1.85388900	-3.02971300	-0.98135200
C	-2.09507700	-2.52289500	2.84876000
H	-2.76988500	-0.53236300	2.38931300
C	-1.66617200	-3.74554000	2.33660200
H	-1.26060600	-4.88204100	0.55334300
H	-2.17193800	-2.37957800	3.92215100
H	-1.40854200	-4.55710500	3.00948800
H	-2.15263700	0.90864800	-1.72427700

Thermal correction to Gibbs Free Energy= 0.490476
E(RM06L) = -1911.42368487

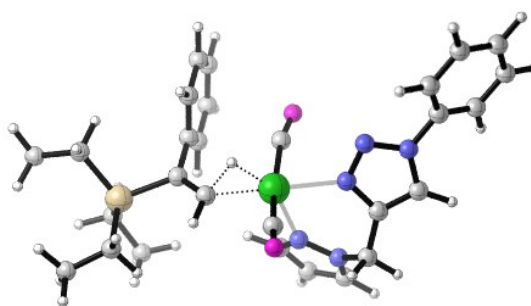


Figure S66. Calculated structure of the reductive elimination intermediate from C^{Si}- α .

Reductive elimination from C^H-(E)

C	-2.38272900	-2.09048800	-0.73791000
C	-3.74951300	-1.97889600	-0.68362400
H	-4.54379700	-2.67411500	-0.90504900
N	-2.84774400	-0.03769500	-0.12662500
N	-1.87989900	-0.87307500	-0.39382000
N	-3.98230300	-0.69932600	-0.30615100
C	-5.24192300	-0.04940500	-0.10380700
C	-5.39808400	1.26641200	-0.52641500
C	-6.27412700	-0.75447500	0.50602600
C	-6.62656600	1.88678200	-0.32928800
H	-4.57449600	1.78770000	-1.00146700
C	-7.50100100	-0.12383400	0.68369200
H	-6.11939300	-1.76960200	0.85856100
C	-7.67707800	1.19378200	0.26856100
H	-6.76456200	2.91282300	-0.65318700
H	-8.31478200	-0.66078500	1.15914100
H	-8.63458500	1.68276300	0.41376400
C	-1.53095700	-3.26201600	-1.13330100
H	-2.15505400	-4.15484700	-1.21613000
H	-1.06874000	-3.08616100	-2.10860500
N	-0.45856900	-3.52479300	-0.19963500
C	-0.46470100	-4.31339200	0.90249600
C	0.76687100	-4.16764900	1.50366800
H	-1.31972900	-4.92330100	1.15656800
C	1.45263000	-3.25067100	0.68566600
H	1.12467000	-4.66756400	2.39059600
H	2.45517600	-2.85755000	0.78858000
N	0.70326600	-2.86007600	-0.34326000
Rh	0.15835800	-0.22841700	-0.09516500
C	-0.20676700	-0.22264100	1.80893700
C	0.30719800	0.00092600	-2.00619900
O	-0.46413500	-0.22238300	2.91824200
O	0.30314100	0.09103600	-3.14257200
C	2.13249500	0.25895100	0.33939800
C	3.11438600	-0.00392900	-0.56368000
Si	1.27605900	2.25422500	0.19071100
C	1.66247500	2.88250800	-1.57460100
H	0.71865200	2.92952500	-2.13273600
H	2.29622200	2.16368900	-2.10428700
C	2.55142300	2.85021700	1.49683100
H	3.55286000	2.69815300	1.07431800
H	2.48843100	2.17504500	2.36085200
C	-0.45608500	2.90280100	0.73494300
H	-0.52925100	2.80773100	1.82692400
H	-1.24412900	2.26774000	0.31925400
C	-0.71415200	4.35916100	0.31671200
H	-1.69527200	4.68832700	0.67510000
H	0.03003000	5.05138900	0.72095000
H	-0.71036000	4.46679700	-0.77354100
C	2.34813700	4.25690800	-1.60382800
H	2.55164000	4.56198100	-2.63511600
H	1.72482300	5.03291400	-1.14881900
H	3.30346100	4.23860800	-1.06889200
C	2.38845900	4.29173700	1.99382300
H	1.42355200	4.43771700	2.49032100
H	3.16982700	4.53178300	2.72235200
H	2.46317700	5.02086500	1.18148900
H	2.41407600	0.13027000	1.38387700
H	2.90965500	0.14608200	-1.62345700
C	4.47015900	-0.47404300	-0.28794700
C	4.92610000	-0.78619700	1.00376000
C	5.35311900	-0.62579500	-1.36769800
C	6.22528600	-1.22933800	1.20306900
H	4.26572000	-0.68198600	1.85942900
C	6.65368600	-1.07130900	-1.16757200
H	5.01278300	-0.38995300	-2.37296100
C	7.09199800	-1.37308600	0.11890900
H	6.56686000	-1.46518000	2.20587500
H	7.32445000	-1.18246500	-2.01325500
H	8.10738400	-1.72138500	0.27938400

Thermal correction to Gibbs Free Energy= 0.494426
E(RM06L) = -1911.44090421

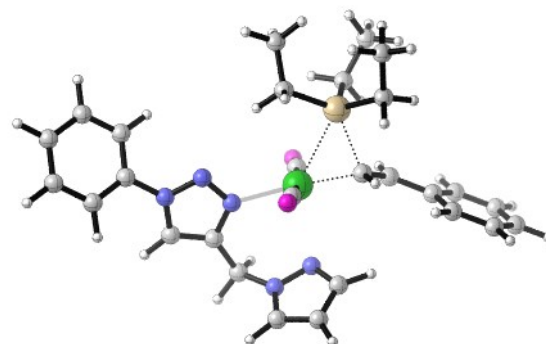


Figure S67. Calculated structure of the reductive elimination intermediate from C^H-(E).

Reductive elimination from C^H-α

C	-2.27279900	-1.91691200	-0.63973200
C	-3.60235500	-1.62037300	-0.80764600
H	-4.44692900	-2.22139200	-1.10568500
N	-2.52452300	0.22616200	-0.24053300
N	-1.65949700	-0.75100700	-0.29824400
N	-3.70229200	-0.29368100	-0.55375800
C	-4.87229800	0.53143600	-0.59491100
C	-4.77693200	1.81041300	-1.13265500
C	-6.07030500	0.02688200	-0.10035600
C	-5.91955400	2.60145100	-1.17242200
H	-3.82872000	2.17247400	-1.51418700
C	-7.20703200	0.82623000	-0.15969200
H	-6.11390300	-0.96249400	0.34457700
C	-7.13213500	2.11063200	-0.69283200
H	-5.86248100	3.60139900	-1.58896300
H	-8.14836200	0.44709300	0.22327600
H	-8.02073600	2.73174600	-0.73253900
C	-1.54629700	-3.21879600	-0.79828600
H	-2.26973400	-4.01730100	-0.98665600
H	-0.86048200	-3.17973100	-1.64818700
N	-0.73665900	-3.54709800	0.35486600
C	-1.14953300	-3.86405000	1.60817300
C	-0.01366600	-4.03884300	2.36704000
H	-2.19835200	-3.95460000	1.85261500
C	1.04720300	-3.80091900	1.47016800
H	0.04128100	-4.31924400	3.40784200
H	2.11223000	-3.84165100	1.65243100
N	0.60445800	-3.49923500	0.25520400
Rh	0.36759500	-0.33780800	0.17969800
C	-0.15856300	-0.46203800	2.04062600
C	0.65602800	-0.19399900	-1.74492900
O	-0.55527600	-0.52843700	3.10549600
O	0.68588000	-0.14013800	-2.88038300
C	2.48703600	-0.11623400	0.58761400
C	3.40977000	-0.49014500	-0.54167400
C	3.25211700	-1.72665200	-1.18188400
C	4.46632800	0.33789100	-0.93389700
C	4.11904600	-2.11049800	-2.20105500
H	2.44668000	-2.38991900	-0.87497600
C	5.32923100	-0.04632000	-1.95626000
H	4.62228200	1.29291000	-0.44182900
C	5.15584000	-1.26984400	-2.59649600
H	3.98460000	-3.07418600	-2.68292000
H	6.14165100	0.61201400	-2.24777300
H	5.82953600	-1.56920800	-3.39300100
C	2.82747300	-0.58997300	1.80846000
H	3.71621600	-1.20522600	1.94402000
H	2.27173000	-0.36329400	2.71428500
Si	1.79668400	1.96693400	0.63309800
C	2.02680200	2.59008800	-1.15944200
H	1.02586800	2.71947900	-1.59154300
H	2.52764400	1.82173200	-1.75621200
C	3.18286100	2.53277400	1.84031100
H	4.15700200	2.29024600	1.40039900
H	3.09670700	1.89265500	2.72744400
C	0.14414400	2.66283300	1.35016600
H	0.17220800	2.57301800	2.44427900
H	-0.71467500	2.07949600	1.00589100
C	-0.08222500	4.12945200	0.94413700
H	-1.00246900	4.50741600	1.40137700
H	0.73335200	4.78711100	1.25627200
H	-0.19035400	4.22700800	-0.14134200
C	2.81757200	3.90006600	-1.27624700
H	2.93155900	4.18602600	-2.32651000
H	2.31784000	4.72682700	-0.76155300
H	3.82220100	3.80054900	-0.85043600
C	3.15421700	3.99939000	2.28461000
H	2.24118200	4.23217600	2.84175700
H	4.00027500	4.20728600	2.94774800
H	3.22270500	4.69280200	1.44133500

Thermal correction to Gibbs Free Energy= 0.496989
E(RM06L) = -1911.41872491

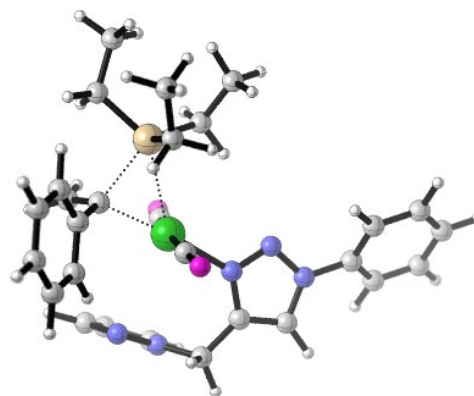


Figure S68. Calculated structure of the reductive elimination from C^H-α intermediate.

RhClCp* based compounds

Catalyst

C	1.05422800	1.80356800	-0.63736100
C	2.42138900	1.90158600	-0.58197400
H	3.10548200	2.70392300	-0.80856600
N	1.81444600	-0.14930500	-0.01462200
N	0.73941100	0.53103100	-0.28628900
N	2.84040200	0.67199300	-0.19938000
C	4.18114300	0.21234800	-0.00316200
C	4.54404100	-1.03642300	-0.49662600
C	5.08380000	1.02926600	0.66913800
C	5.84963300	-1.47432200	-0.30469200
H	3.81936100	-1.64243300	-1.02948700
C	6.38997500	0.58272100	0.84096000
H	4.76889100	1.98808900	1.06951500
C	6.77166100	-0.66650600	0.35779300
H	6.14959900	-2.44482600	-0.68537900
H	7.10543900	1.20863800	1.36345100
H	7.79061300	-1.01169000	0.49786400
C	0.01777700	2.78346900	-1.07486700
H	0.45046500	3.78277800	-1.14173900
H	-0.36864200	2.48852800	-2.05527900
N	-1.10157300	2.86457000	-0.14660500
C	-1.70498200	3.99445300	0.27995600
C	-2.82794600	3.62915600	0.98934300
H	-1.29617300	4.96508200	0.04080900
C	-2.83559900	2.23210500	0.93545400
H	-3.54012900	4.27793000	1.47414300
H	-3.54475000	1.54024000	1.35979200
N	-1.79187300	1.77404200	0.24533700
Rh	-1.23168600	-0.27380000	-0.15568000
C	-2.94654600	-1.45251000	0.40807300
C	-2.14170700	-2.22889000	-0.49515900
C	-2.15534300	-1.21276300	1.59961200
C	-0.83530400	-2.37196200	0.07423500
C	-0.85982400	-1.76150700	1.39151400
C	-4.38708300	-1.10999500	0.18498400
H	-5.01057200	-2.00430300	0.29385600
H	-4.74652200	-0.36817000	0.90149500
H	-4.53782200	-0.70692400	-0.81983300
C	-2.60115300	-2.79512200	-1.79585400
H	-3.00554900	-3.79858900	-1.61429700
H	-3.38108600	-2.18030400	-2.24737000
H	-1.78241200	-2.87346800	-2.51251500
C	-2.60841000	-0.56613200	2.87237700
H	-2.55741100	-1.29545700	3.68793200
H	-1.97543800	0.28386600	3.14300800
H	-3.64201100	-0.22077700	2.80921000
C	0.25279700	-1.78032700	2.38834900
H	0.24573500	-2.73569900	2.92612100
H	1.22370100	-1.67059700	1.90063600
H	0.14301200	-0.97735200	3.12131300
C	0.30902600	-3.12731100	-0.52251400
H	0.23814600	-4.18804400	-0.25585500
H	0.30519200	-3.04597500	-1.61195000
H	1.26270000	-2.74161600	-0.15664400
Cl	-1.47818300	0.21367300	-2.48889300

Thermal correction to Gibbs Free Energy= 0.396597
E(RM06L) = -1698.81492915

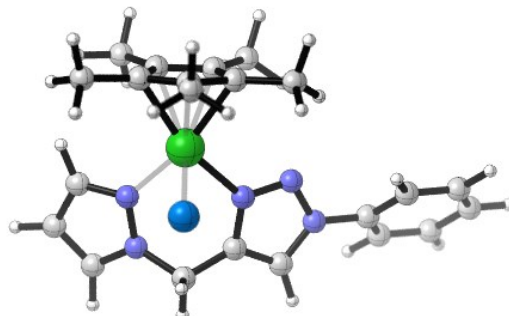


Figure S69. Calculated structure of Rh^{III}C₀Ph.

Rh-H-Si intermediate

C	2.61045200	0.21129200	0.15508400
C	3.81317800	0.85237400	0.32249100
H	4.08571900	1.80641400	0.74578800
N	4.12620000	-1.13925300	-0.60292100
N	2.85304400	-1.00204300	-0.40961700
N	4.72552700	-0.02154200	-0.16452200
C	6.14419300	0.12394200	-0.24110800
C	6.94910900	-0.99460800	-0.04866700
C	6.68910900	1.37710100	-0.50356700
C	8.32963200	-0.84623100	-0.11992300
H	6.49550100	-1.95905100	0.14887400
C	8.07246600	1.51285500	-0.55662700
H	6.04591100	2.23196700	-0.68852900
C	8.89272800	0.40397400	-0.36638500
H	8.96711400	-1.71192200	0.02597400
H	8.50704300	2.48567300	-0.76114800
H	9.97112600	0.51316700	-0.41474900
C	1.22319400	0.64136900	0.48979000
H	1.19084100	1.71059500	0.72350600
H	0.55906700	0.46241700	-0.35739100
N	0.67719700	-0.10908900	1.61769900
C	1.28147700	-0.33492800	2.80295300
C	0.35538900	-0.91008000	3.64657900
H	2.31795100	-0.07031100	2.95136500
C	-0.81706300	-0.99048400	2.88729800
H	0.50379800	-1.22537800	4.66739000
H	-1.79664900	-1.36163200	3.15109400
N	-0.61526500	-0.50010300	1.66378000
Rh	-1.96070400	-0.74244100	0.02313100
H	-1.67682100	0.96211900	-0.42290000
Cl	-3.73142400	-0.19666300	1.52488200
C	-1.04309500	-1.67157100	-1.72805500
C	-1.03484300	-2.59885300	-0.64127900
C	-2.43037800	-1.33613300	-2.00848000
C	-2.39880700	-2.81382200	-0.21784700
C	-3.26221000	-2.06186500	-1.10950200
C	-2.86412500	-3.74140900	0.85695500
H	-2.06844300	-3.94760900	1.57616800
H	-3.71278700	-3.31431900	1.39682900
H	-3.17805500	-4.69466100	0.41609700
C	0.17158700	-3.23143900	-0.02818800
H	0.30896300	-4.23015600	-0.45919500
H	1.07784800	-2.64960200	-0.21406800
H	0.05101500	-3.34652800	1.05220000
C	0.13612100	-1.26596500	-2.55362500
H	0.22251500	-1.94042200	-3.41389800
H	0.02533500	-0.25173900	-2.94432600
H	1.06991500	-1.32543800	-1.98903400
C	-2.89834700	-0.43417600	-3.10549100
H	-3.84105800	0.05203300	-2.84507200
H	-2.16221100	0.34194600	-3.32537400
H	-3.05777700	-1.01857200	-4.01908800
C	-4.75437000	-2.08446400	-1.09259100
H	-5.10607300	-2.93939900	-1.68216400
H	-5.13619400	-2.17871600	-0.07465000
H	-5.17480900	-1.17355200	-1.52301800
Si	-2.11865000	2.44788600	-0.13804800
C	-1.90030300	2.81415300	1.70899300
H	-2.79049700	2.45931900	2.23755400
H	-1.06987100	2.20816000	2.09329600
C	-3.87199700	2.61998800	-0.83118300
H	-3.83648800	2.47370200	-1.91857600
H	-4.48231600	1.81129900	-0.41397300
C	-0.83804500	3.38819100	-1.18955400
H	-1.14961100	4.44145700	-1.21119800
H	0.13137000	3.38068200	-0.67521600
C	-0.68175000	2.86125000	-2.61830100
H	-1.63483100	2.87102500	-3.15909600
H	0.02708200	3.46474600	-3.19356800
H	-0.30902300	1.83000900	-2.61574900
C	-1.64668100	4.30277700	1.99360800
H	-1.54078200	4.47479500	3.06910700
H	-0.73140000	4.66179800	1.50940100

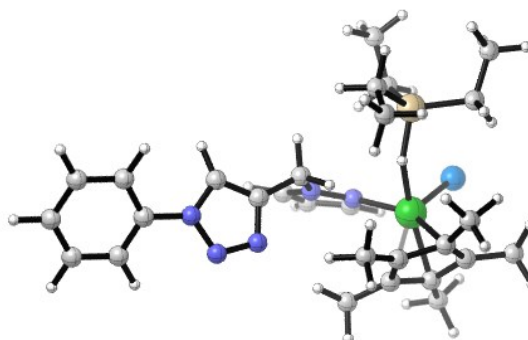


Figure S70. Calculated structure of the Rh-H-Si intermediate.

H	-2.47278000	4.93093600	1.64335600
C	-4.50224600	3.98250400	-0.50240600
H	-5.49859300	4.06454400	-0.94696900
H	-4.61409400	4.11608800	0.57877800
H	-3.90311600	4.81659500	-0.88581100

Thermal correction to Gibbs Free Energy= 0.587028
E(RM06L) = -2226.65529716

C^{Si}-(E) intermediate, two N coordinated

C	-2.62347500	-1.17088600	1.56970000
C	-3.97083200	-0.92342900	1.61661100
H	-4.72347300	-1.09373100	2.36990400
N	-3.14079800	-0.18536000	-0.30038100
N	-2.16310000	-0.70858400	0.37815800
N	-4.24146600	-0.31898000	0.43508800
C	-5.49949200	0.16304300	-0.04539800
C	-5.54325000	1.37843500	-0.72043700
C	-6.64471600	-0.59339200	0.18008200
C	-6.77104200	1.84176200	-1.17949000
H	-4.63274600	1.94644400	-0.87620600
C	-7.86759200	-0.10934700	-0.27302500
H	-6.58235100	-1.55491200	0.68051300
C	-7.93131100	1.10431900	-0.95278100
H	-6.82118200	2.78787400	-1.70793600
H	-8.76847900	-0.68970800	-0.10461900
H	-8.88668500	1.47522700	-1.30878900
C	-1.65743500	-1.66551300	2.59189400
H	-2.17119400	-2.03477700	3.48030400
H	-1.04224600	-2.47474500	2.18893200
N	-0.79447000	-0.56922400	3.01058600
C	-0.67615800	-0.03545300	4.24466700
C	0.17144100	1.04628900	4.15118500
H	-1.19719500	-0.46700100	5.08677800
C	0.52621300	1.10783100	2.79984300
H	0.49643900	1.69763700	4.94698800
H	1.19801100	1.78190100	2.29373800
N	-0.06453400	0.13130600	2.11091600
Rh	0.07636100	-0.41164100	0.06343500
H	-0.25444900	0.99675200	-0.32160800
C	2.01158500	0.18202100	0.25423700
C	2.93852900	-0.87594200	0.77386600
C	4.26204400	-0.95875200	0.31243100
C	2.55102700	-1.78305600	1.76928000
C	5.14582200	-1.91122400	0.80383400
H	4.60506000	-0.26203600	-0.44415400
C	3.43272400	-2.73906300	2.26618000
H	1.55071100	-1.73657200	2.18456700
C	4.73352500	-2.81476900	1.78052400
H	6.16180100	-1.94771000	0.42175300
H	3.10054400	-3.42271300	3.04191500
H	5.42155200	-3.56104400	2.16501800
C	2.48610900	1.43377100	0.11015700
H	3.54381100	1.53370500	0.38610300
Si	1.84508500	3.18054800	-0.27531500
C	2.33401800	4.25414900	1.23647400
H	2.21732800	5.30554100	0.94385200
H	1.60126600	4.09492700	2.04001300
C	-0.04079200	3.41879200	-0.50835300
H	-0.41414100	2.79734200	-1.33024400
H	-0.55445000	3.07258800	0.39911600
C	3.75272300	4.02578500	1.77113600
H	4.00029600	4.74101000	2.56245100
H	3.86794700	3.01928100	2.18905700
H	4.50477300	4.13855000	0.98096300
C	-0.41626800	4.88308500	-0.78342600
H	-1.50208300	5.00381100	-0.86228800
H	-0.06937800	5.54824200	0.01463300
H	0.02156500	5.23571900	-1.72343200
C	2.80054900	3.81095000	-1.81825900
H	3.71898000	3.21765400	-1.91886800
H	2.20467300	3.59240100	-2.71507200
C	3.15972600	5.30282500	-1.79812400

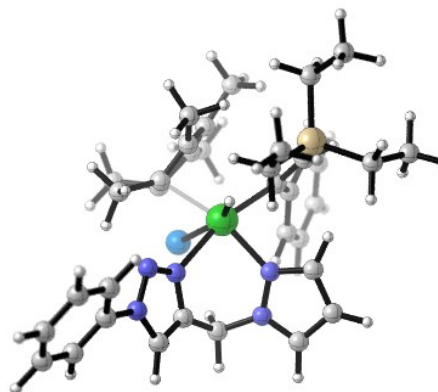


Figure S71. Calculated structure of the C^{Si}-(E) intermediate, two N coordinated.

H	3.82309500	5.54168000	-0.96008600
H	3.67416000	5.59670400	-2.71904900
H	2.27125000	5.93587200	-1.70047600
Cl	0.26015700	-3.36625400	-0.03124300
C	0.58199100	-2.78706100	-1.75658500
C	-0.12593700	-1.44122700	-1.94649000
C	2.05182200	-2.48013800	-1.95467800
C	0.85680300	-0.53217400	-2.30618200
C	2.18142600	-1.19937900	-2.34701000
C	0.63757500	0.77185200	-2.99841600
H	-0.38197600	1.14082100	-2.87924700
H	1.33633600	1.52781900	-2.63348700
H	0.83417400	0.63613600	-4.06991700
C	-1.53274600	-1.41678300	-2.48161200
H	-1.52546500	-1.74571100	-3.52843800
H	-2.19051400	-2.08479400	-1.92207600
H	-1.96746900	-0.41840100	-2.44381200
C	0.09065300	-3.87838900	-2.70057900
H	0.26916100	-3.56223700	-3.73381800
H	0.63599100	-4.80864500	-2.52447400
H	-0.97601600	-4.07179200	-2.56724300
C	3.10831700	-3.50594300	-1.72574700
H	4.09715100	-3.12663000	-1.98847100
H	3.13426400	-3.78633000	-0.66538200
H	2.92537600	-4.41661100	-2.30738300
C	3.39669600	-0.48062600	-2.83696500
H	3.29230400	-0.23721900	-3.90138800
H	3.53168400	0.46321700	-2.29840100
H	4.30060600	-1.08134900	-2.72108500

Thermal correction to Gibbs Free Energy= 0.703207
E(RM06L) = -2535.07488222

C^{Si}-(E) intermediate, one N coordinated

C	-1.83057900	-2.24258400	1.99890700
C	-2.56116900	-1.18043200	1.52181800
H	-2.47915900	-0.11430900	1.65487800
N	-3.36054700	-3.09209800	0.72906200
N	-2.35438200	-3.38873000	1.49123000
N	-3.49893600	-1.75556800	0.73302200
C	-4.45622300	-1.11095100	-0.10428100
C	-4.84776000	-1.73429100	-1.28643800
C	-4.96096800	0.13464600	0.25885200
C	-5.75934500	-1.09047000	-2.11643100
H	-4.45113300	-2.71175700	-1.53623500
C	-5.86290400	0.77096700	-0.58771800
H	-4.67490100	0.59676000	1.19854900
C	-6.26245200	0.16253400	-1.77481200
H	-6.07557000	-1.57288600	-3.03545100
H	-6.26626600	1.73870800	-0.30763500
H	-6.97214000	0.66004400	-2.42744000
C	-0.65226400	-2.27034400	2.92307900
H	-0.91046000	-2.75753800	3.86659600
H	0.17424500	-2.83540200	2.48089500
N	-0.20491700	-0.92790000	3.25188900
C	-0.24980300	-0.32860800	4.45910700
C	0.21853300	0.95922700	4.30888800
H	-0.60630500	-0.86419800	5.32679000
C	0.52938400	1.07018100	2.95163500
H	0.32820400	1.71220800	5.07318700
H	0.93819700	1.90372200	2.40191300
N	0.26979300	-0.07468400	2.31271400
Rh	0.42608400	-0.58055400	0.27591700
H	-0.69891400	0.37171200	-0.00348200
C	1.68487500	0.95975400	0.09222600
C	3.10185600	0.54775200	0.31899200
C	4.12312400	1.06542700	-0.49264000
C	3.47005300	-0.30625900	1.36712700
C	5.45285300	0.72597800	-0.28007100
H	3.86977700	1.74851200	-1.29550400
C	4.80227800	-0.64974100	1.58007100
H	2.71538700	-0.69769000	2.04087100
C	5.79894200	-0.14232500	0.75339100
H	6.22240300	1.14270000	-0.92286500

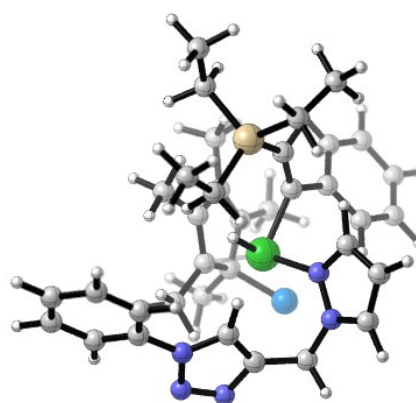


Figure S72. Calculated structure of the C^{Si}-(E) intermediate, one N coordinated.

H	5.05985100	-1.31030100	2.40218800
H	6.83750900	-0.41034200	0.91914500
C	1.34673000	2.23840500	-0.12076900
H	2.21750000	2.90596800	-0.12649400
Si	-0.21733000	3.31867200	-0.22649100
C	-0.02573900	4.60098600	1.18200000
H	-0.77728900	5.38421200	1.01988200
H	-0.30107400	4.12208300	2.13195500
C	-1.91493500	2.47249900	0.04742400
H	-2.08473800	1.66517900	-0.67518600
H	-1.90562100	2.00969300	1.04430900
C	1.36247000	5.24110800	1.30250300
H	1.36776900	6.03888800	2.05204900
H	2.12048100	4.50765800	1.60013300
H	1.68899900	5.68204600	0.35310800
C	-3.07690900	3.47249200	-0.04656000
H	-4.03395300	2.98173800	0.15846600
H	-2.96646200	4.29483200	0.66807500
H	-3.14447900	3.90742200	-1.04920000
C	-0.17959900	4.21165000	-1.92396900
H	0.85093100	4.19287600	-2.30212500
H	-0.77209600	3.62606000	-2.64024200
C	-0.68825900	5.65981500	-1.90462000
H	-0.07192200	6.29073900	-1.25562200
H	-0.66485200	6.09743500	-2.90784100
H	-1.71882800	5.72665000	-1.54009700
Cl	1.94367600	-3.00867200	0.16997100
C	1.55730600	-2.60698000	-1.59475800
C	0.27532700	-1.77946000	-1.58647500
C	2.60146700	-1.67540700	-2.17649200
C	0.56136600	-0.58051300	-2.21178700
C	1.99622400	-0.54692800	-2.59341700
C	-0.41506500	0.39623900	-2.78000900
H	-1.43182200	0.22943400	-2.41993300
H	-0.11827000	1.42146100	-2.54342900
H	-0.41466000	0.30109200	-3.87301700
C	-1.06785700	-2.44697900	-1.49577600
H	-1.25859300	-3.02295400	-2.41014900
H	-1.12068000	-3.14601900	-0.65680000
H	-1.87582600	-1.71903200	-1.39458200
C	1.43430800	-3.91821700	-2.35877000
H	1.17856700	-3.69734800	-3.40078100
H	2.38095700	-4.46363700	-2.34175000
H	0.65840600	-4.55588500	-1.92827600
C	4.04433900	-2.04366400	-2.22414200
H	4.63413900	-1.28692300	-2.74409800
H	4.44139900	-2.13161300	-1.20507300
H	4.20254100	-3.00454200	-2.72665500
C	2.56299900	0.60016200	-3.36588300
H	2.15494800	0.61581300	-4.38382100
H	2.29730900	1.55346700	-2.89632200
H	3.65011800	0.54303600	-3.44440200

Thermal correction to Gibbs Free Energy= 0.701686

E(RM06L) = -2535.06609306

C^{Si}-(Z) intermediate, two N coordinated

C	-2.69598500	-0.29768400	1.91776000
C	-3.97003000	0.19189000	2.06138400
H	-4.59403900	0.37989300	2.92068100
N	-3.39844600	0.20350000	-0.07736500
N	-2.39500300	-0.26647800	0.59413700
N	-4.36137400	0.48267400	0.79739100
C	-5.59647000	1.04145500	0.34502200
C	-5.56177600	2.02479000	-0.63907900
C	-6.79201300	0.59073300	0.89440400
C	-6.76432100	2.56460800	-1.08160500
H	-4.60900000	2.36269000	-1.03294000
C	-7.98547700	1.15130600	0.45065000
H	-6.79404400	-0.19640200	1.64241200
C	-7.97236700	2.13349900	-0.53675400
H	-6.75496800	3.33487600	-1.84552300
H	-8.92607200	0.81083700	0.87041600
H	-8.90612800	2.56491500	-0.88195300
C	-1.65949400	-0.71433100	2.90941400
H	-2.07488200	-0.77602600	3.91628100
H	-1.24971200	-1.69476800	2.64522000
N	-0.57561400	0.25319800	2.94179400
C	-0.19258400	1.04576100	3.96740300
C	0.76856000	1.90853600	3.48874900
H	-0.63267100	0.93545200	4.94772400
C	0.90520700	1.58193500	2.13420000
H	1.29781200	2.67120700	4.03769900
H	1.54723600	2.00393700	1.37676900
N	0.08871000	0.58198200	1.81334800
Rh	-0.14931400	-0.36564900	-0.09424500
H	0.09636000	-1.65046000	0.75305000
C	1.90285500	-0.28621000	-0.18760200
C	2.68749100	-1.31351600	0.55528700
C	3.75115800	-1.94750100	-0.10178900
C	2.45562200	-1.65412700	1.89546800
C	4.56645900	-2.86512300	0.55269800
H	3.93559800	-1.71444200	-1.14606900
C	3.27123900	-2.56822900	2.55461900
H	1.65298400	-1.17318100	2.44343700
C	4.33144500	-3.17827400	1.88780100
H	5.38463700	-3.33588400	0.01598700
H	3.08462700	-2.79705800	3.59983800
H	4.96682700	-3.89023800	2.40493000
C	2.49999200	0.73865500	-0.80752400
Cl	-0.59588800	1.89319500	-0.91282400
C	-1.59003000	-2.74007000	-1.02648600
C	-2.50294200	-2.07299900	-1.81019600
C	-0.20417700	-2.19096400	-1.25613800
C	-1.79890100	-1.00099400	-2.42499200
C	-0.36034000	-1.08067900	-2.12978300
C	-2.38475600	0.07071600	-3.23215900
H	-2.18797800	1.00926200	-2.67877500
H	-1.84787700	0.17671100	-4.18258500
H	-3.45347700	-0.05073600	-3.40850000
C	-3.97827600	-2.30629100	-1.91987900
H	-4.24552000	-2.62201700	-2.93411400
H	-4.31257900	-3.09017500	-1.23610700
H	-4.53830300	-1.39465300	-1.69213500
C	-1.84390900	-3.84581100	-0.07210400
H	-1.60674000	-4.80427900	-0.55427000
H	-1.18827200	-3.76026600	0.79970400
H	-2.88492700	-3.88572500	0.25552500
C	0.92261000	-3.18472700	-1.32142600
H	1.84600100	-2.71643300	-1.65587200
H	1.11979700	-3.66021600	-0.35766700
H	0.65566700	-3.96321900	-2.04813300
C	0.62085800	-0.57638200	-3.15513200
H	0.42702800	-1.05413000	-4.12398700
H	0.53055900	0.50698500	-3.27301100
H	1.64711200	-0.79272000	-2.86193100
H	1.85870600	1.42138800	-1.36423700
Si	4.30489400	1.36543200	-0.68001600
C	4.14296600	3.20939100	-1.17137400
H	3.18456000	3.57162600	-0.77376700

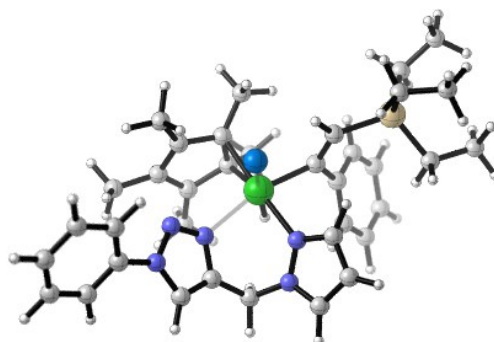


Figure S73. Calculated structure of the C^{Si}-(Z) intermediate, two N coordinated.

H	4.04775800	3.26971600	-2.26419800
C	5.26970000	4.13303400	-0.69227100
H	5.09077800	5.16805800	-1.00294300
H	5.35029700	4.12828400	0.40073700
H	6.24320200	3.83361600	-1.09423600
C	4.93744800	1.22782800	1.12260900
H	4.57114600	0.29437200	1.56392000
H	4.47876700	2.04070000	1.70284100
C	6.46450200	1.28754600	1.26061600
H	6.76992800	1.21402400	2.30997800
H	6.94167300	0.46117500	0.72171300
H	6.87586300	2.22154100	0.86293100
C	5.47385800	0.45055400	-1.89556700
H	4.87563500	0.10489900	-2.74984800
H	5.84909600	-0.45378100	-1.40004400
C	6.65733700	1.28292800	-2.40546800
H	6.31785800	2.17835000	-2.93724600
H	7.30281800	1.61115300	-1.58380300
H	7.27692600	0.70336600	-3.09835900

Thermal correction to Gibbs Free Energy= 0.697948

E(RM06L) = -2535.08686280

C^{Si}-(Z) intermediate, one N coordinated

C	3.18044400	1.60139400	1.48078000
C	3.36970100	0.26324500	1.23792300
H	2.69162700	-0.57661000	1.19196700
N	5.28861200	1.37102600	1.07277700
N	4.37348500	2.23935700	1.37255700
N	4.69680200	0.16507600	0.98859200
C	5.44531700	-0.99698500	0.63534700
C	6.65783100	-0.84337800	-0.03228700
C	4.93927700	-2.25953400	0.93449400
C	7.37017600	-1.97856300	-0.40317200
H	7.03370100	0.15062900	-0.24454600
C	5.65812700	-3.38496400	0.54459500
H	4.00419000	-2.37429100	1.47304000
C	6.87239600	-3.24878300	-0.12272300
H	8.31890200	-1.86589500	-0.91761200
H	5.26884200	-4.37086800	0.77649100
H	7.43214000	-4.13011000	-0.41801600
C	1.92344700	2.36825900	1.74655900
H	2.12628800	3.19607200	2.42932500
H	1.50495000	2.79832800	0.83026200
N	0.90100800	1.53753400	2.34716700
C	0.63115800	1.41448900	3.66263300
C	-0.23038800	0.34974400	3.82052700
H	1.08434400	2.08295800	4.38001200
C	-0.42376700	-0.14909700	2.53107000
H	-0.66028500	-0.01825700	4.73857900
H	-1.03567800	-0.96980000	2.19208000
N	0.27272300	0.56761000	1.64406000
Rh	0.13519400	0.31632700	-0.46475700
H	-0.33197900	1.79047000	-0.31181200
C	-1.92166200	0.11283500	-0.19493500
C	-2.68544200	1.37731500	-0.02118300
C	-3.76098200	1.63670700	-0.88249000
C	-2.40121700	2.32075800	0.97903500
C	-4.53187000	2.78804400	-0.74650000
H	-3.99825000	0.91896200	-1.66000300
C	-3.17284500	3.46898700	1.11657100
H	-1.59376900	2.14234600	1.68031400
C	-4.23993700	3.71026700	0.25315200
H	-5.36254500	2.96038700	-1.42394200
H	-2.94693700	4.17409900	1.91063900
H	-4.84036700	4.60744800	0.36340500
C	-2.46424200	-1.10230200	-0.25284800
Cl	0.68301200	-1.95972300	0.13385700
C	0.48042600	1.83940500	-2.03867200
C	1.89789800	1.44175800	-1.85976500
C	-0.23525200	0.70471900	-2.58881600
C	1.99101700	0.09113900	-2.11592700
C	0.64351700	-0.39649800	-2.50949700

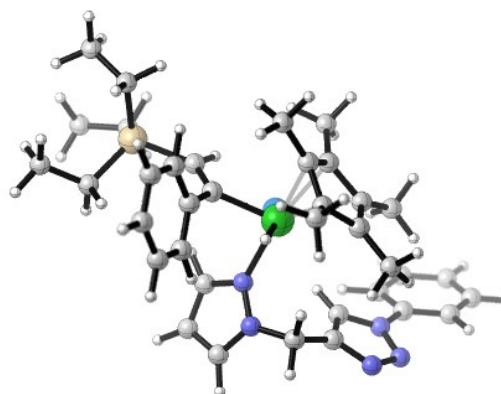


Figure S74. Calculated structure of the C^{Si}-(Z) intermediate, one N coordinated.

C	3.19038500	-0.78836900	-2.07441700
H	3.03224100	-1.60729300	-1.36241500
H	3.36301900	-1.23864500	-3.05856100
H	4.08984100	-0.24355700	-1.78390000
C	3.01015200	2.40820200	-1.61869200
H	3.23208100	2.93188400	-2.55715300
H	2.76142100	3.17321200	-0.88026600
H	3.92295200	1.91198900	-1.28721500
C	0.02145100	3.26305500	-2.14041200
H	0.25360700	3.63198700	-3.14669900
H	-1.05569000	3.35277500	-1.97937200
H	0.53537700	3.90510000	-1.42115500
C	-1.55802900	0.73704200	-3.28104000
H	-2.11989800	-0.18383700	-3.11269600
H	-2.16797100	1.57622300	-2.94537800
H	-1.38741500	0.84728500	-4.35864800
C	0.36072000	-1.75603400	-3.05168500
H	0.52740300	-1.73957300	-4.13658900
H	1.01029100	-2.50902600	-2.60528900
H	-0.67687600	-2.04540600	-2.86982700
H	-1.80664900	-1.94617600	-0.45743500
Si	-4.25686200	-1.63531100	0.19603500
C	-4.04594900	-3.48405500	0.63709300
H	-3.08341700	-3.59199000	1.15589000
H	-3.94431000	-4.05428900	-0.29638200
C	-5.15732300	-4.09049200	1.50347600
H	-4.95671300	-5.14545300	1.71746100
H	-5.24033100	-3.57099300	2.46467000
H	-6.13473900	-4.03525000	1.01325300
C	-4.88335200	-0.65041100	1.71006800
H	-4.54637300	0.38943400	1.63241500
H	-4.39746500	-1.06301500	2.60471000
C	-6.40836600	-0.67598300	1.88413300
H	-6.71235400	-0.10943400	2.77049400
H	-6.91309900	-0.22913300	1.02015200
H	-6.79098600	-1.69565900	1.99885100
C	-5.42604200	-1.42173000	-1.30844000
H	-4.82978100	-1.51112700	-2.22707600
H	-5.82040900	-0.39765600	-1.29773000
C	-6.59280600	-2.41733100	-1.35824200
H	-6.23542800	-3.45151900	-1.40557100
H	-7.23506700	-2.32891200	-0.47566000
H	-7.21950300	-2.24421900	-2.23938900

Thermal correction to Gibbs Free Energy= 0.699540
 E(RM06L) = -2535.09861785

C^{Si}- α intermediate, two N coordinated

C	-1.50670700	1.44590500	-1.31013400
C	-2.72615300	1.90677900	-1.73847800
H	-3.02374900	2.56034600	-2.54318000
N	-3.00786500	0.51814300	-0.04001500
N	-1.73334800	0.60189800	-0.26908600
N	-3.62040200	1.30582200	-0.91768700
C	-5.04627400	1.39712500	-0.92040200
C	-5.78829300	0.22768800	-0.78536400
C	-5.65191700	2.64117800	-1.05703100
C	-7.17584900	0.31614700	-0.78194200
H	-5.28027600	-0.72740300	-0.70408300
C	-7.04154000	2.71147600	-1.06595300
H	-5.05012400	3.54143000	-1.13694800
C	-7.80175700	1.55311300	-0.92508400
H	-7.76939400	-0.58659100	-0.68260900
H	-7.52853700	3.67497300	-1.17250500
H	-8.88502100	1.61400700	-0.92925200
C	-0.12610500	1.66177200	-1.83748500
H	-0.12279300	2.44005300	-2.60219800
H	0.55146600	1.96859200	-1.03695800
N	0.39759500	0.45445300	-2.45246400
C	0.86839100	0.30596300	-3.71049400
C	1.21299500	-1.01699100	-3.87363600
H	0.91607800	1.14814800	-4.38520100
C	0.91224900	-1.61909500	-2.64668600
H	1.63005400	-1.48524300	-4.75075200
H	1.02895600	-2.64503200	-2.33354500
N	0.41804700	-0.72384900	-1.79767300
Rh	-0.21193100	-1.12807000	0.21272500
H	0.81925000	-2.23274600	0.28299500
C	1.35677700	0.15796300	0.58244500
C	2.64325200	0.09782600	0.22275800
Cl	-1.73777100	-2.58257200	-0.99855600
C	-0.96686900	0.06032400	2.80934000
C	-2.31674500	-0.23241500	2.68636900
C	-0.13198700	-1.10778300	2.39983400
C	-2.40348600	-1.52191900	2.13163400
C	-1.04283300	-2.09758000	1.96115700
C	-3.62006300	-2.25133700	1.76033100
H	-3.54737500	-2.46904600	0.68010500
H	-3.63839900	-3.23105400	2.25431900
H	-4.53749300	-1.70641200	1.98241800
C	-3.44065400	0.71197600	2.98573500
H	-3.61529800	0.78772600	4.06503800
H	-3.21638200	1.71551100	2.61078600
H	-4.37057100	0.39337600	2.51173600
C	-0.42391100	1.36093700	3.26108200
H	0.60330900	1.28405900	3.62197600
H	-0.42939800	2.06056200	2.41051600
H	-1.05429200	1.80450000	4.03761800
C	1.15038700	-1.38391100	3.13721900
H	1.58984600	-2.32769600	2.81209700
H	1.89239000	-0.59742100	2.98330300
H	0.94057600	-1.46330800	4.21167500
C	-0.84805100	-3.58767600	2.05396800
H	-1.17421500	-3.94404500	3.03983900
H	-1.42211600	-4.10343200	1.28135000
H	0.20098400	-3.85893300	1.92354100
Si	3.74640400	1.65302600	0.50079400
C	5.57280300	1.10022500	0.48962200
H	5.87881400	0.92588400	-0.54942600
H	5.62180300	0.11575600	0.97325300
C	6.55706000	2.05862400	1.17240900
H	7.57941300	1.67057200	1.11818000
H	6.55718500	3.04872800	0.70388200
H	6.31449200	2.19329800	2.23237500
C	3.36378000	2.87912300	-0.92757400
H	2.45545100	3.43846000	-0.65845600
H	3.10457200	2.29493300	-1.82152300
C	4.49334100	3.86030400	-1.26557800
H	4.21428800	4.52299800	-2.09188200
H	4.74909200	4.49254300	-0.40896700
H	5.40301200	3.32723800	-1.56258500

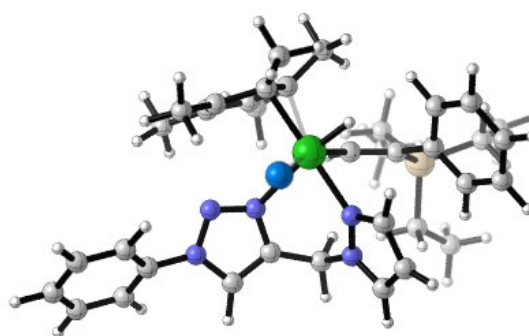


Figure S75. Calculated structure of the C^{Si}- α intermediate, two N coordinated.

C	3.28254400	2.46140900	2.17301700
H	3.72377300	1.85914500	2.97871900
H	2.19510500	2.39072600	2.30484600
C	3.70272700	3.93003700	2.32243700
H	4.78459100	4.05711700	2.21824800
H	3.22361100	4.55954800	1.56399500
H	3.41966600	4.32505700	3.30399200
C	3.28739800	-1.05662300	-0.45640100
C	3.88400800	-0.88938700	-1.71151400
C	3.34817300	-2.32075600	0.13921600
C	4.47967600	-1.95920000	-2.37063600
H	3.85643400	0.08524000	-2.19213400
C	3.94468800	-3.39456000	-0.51630000
H	2.93878900	-2.45689700	1.13462400
C	4.50587400	-3.21942000	-1.77843100
H	4.92464600	-1.80790000	-3.34957500
H	3.98082500	-4.36702800	-0.03433300
H	4.97391800	-4.05488700	-2.28938500
H	1.00261500	1.07032300	1.06757400

Thermal correction to Gibbs Free Energy= 0.700306
E(RM06L) = -2535.08505259

C^{Si}- α intermediate, one N coordinated

C	2.36191500	-2.45676100	-0.41447900
C	3.01141400	-1.32126000	-0.83372700
H	2.66167500	-0.37180900	-1.21109500
N	4.46691400	-2.83708200	-0.11163100
N	3.28459100	-3.35474900	0.01455300
N	4.31918700	-1.60184800	-0.62555600
C	5.44505600	-0.74975500	-0.82909900
C	6.60877000	-0.98140800	-0.09999800
C	5.35088300	0.31376700	-1.72325700
C	7.69329500	-0.12887600	-0.27554700
H	6.65813700	-1.82196200	0.58232600
C	6.44012700	1.16525800	-1.87674200
H	4.44929400	0.47556400	-2.30520800
C	7.61140100	0.94704500	-1.15621800
H	8.60563500	-0.30736300	0.28411400
H	6.37295300	1.99450500	-2.57331600
H	8.46032900	1.61032200	-1.28531900
C	0.89804600	-2.73827600	-0.28018900
H	0.70489000	-3.80424200	-0.41631100
H	0.52692900	-2.46084700	0.71256300
N	0.11792200	-2.01008400	-1.25773100
C	-0.31193500	-2.46756300	-2.45113800
C	-0.72294800	-1.38208200	-3.19533500
H	-0.27028500	-3.52176000	-2.68269700
C	-0.48054300	-0.27591500	-2.37715400
H	-1.13348300	-1.38629800	-4.19270700
H	-0.67206600	0.77000200	-2.55957800
N	0.04090300	-0.66193800	-1.21060100
Rh	0.20820300	0.65684400	0.49536700
H	-0.84723200	1.65193800	-0.00780200
C	-1.75418700	-0.09819300	0.56403400
C	-2.88442900	0.33216200	0.00908000
Cl	1.35742000	1.98903200	-1.11682700
C	0.28175500	-0.10404600	2.63660500
C	1.67817600	-0.23278600	2.29928100
C	-0.08860100	1.28996600	2.55835000
C	2.13451300	1.02110300	1.87388100
C	1.02619500	1.98088900	2.00459200
C	3.51572400	1.39356100	1.45901200
H	3.50753400	1.86056400	0.46782000
H	3.93054900	2.12264400	2.16456800
H	4.18128500	0.52930500	1.42773800
C	2.50515400	-1.44678900	2.57258000
H	2.76336000	-1.45280100	3.63920300
H	1.97628300	-2.37941200	2.36725100
H	3.43743300	-1.45678200	2.00754900
C	-0.53090700	-1.17299400	3.28856300
H	-1.58467500	-0.89764600	3.36044600
H	-0.45482300	-2.13044900	2.76287500
H	-0.15591000	-1.33482100	4.30640500

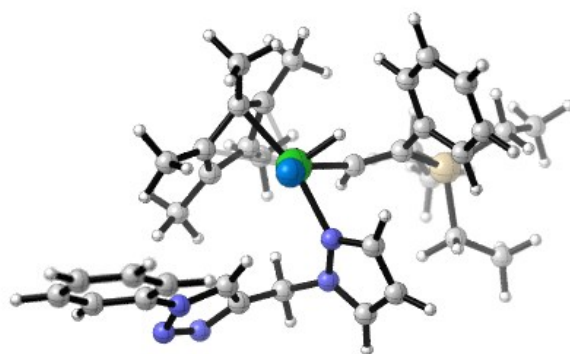


Figure S76. Calculated structure of the C^{Si}- α intermediate, one N coordinated.

C	-1.33358800	1.89046800	3.12896900
H	-1.44539000	2.93279400	2.82485800
H	-2.22689500	1.34036700	2.82257600
H	-1.27162500	1.86734800	4.22302300
C	1.16096500	3.46092700	1.84600400
H	1.49230200	3.89571500	2.79675900
H	1.88954800	3.70887300	1.07301300
H	0.20987000	3.92030600	1.56772200
Si	-4.40327300	-0.86478300	0.08085300
C	-5.95770900	0.12437100	-0.40706800
H	-5.95709400	0.26273100	-1.49569400
H	-5.86443700	1.13301900	0.01623500
C	-7.28509700	-0.50231300	0.04170600
H	-8.13314700	0.10912300	-0.28252100
H	-7.42556200	-1.50571500	-0.37437700
H	-7.33912900	-0.58492900	1.13281200
C	-4.01748900	-2.26959000	-1.16004700
H	-3.30094900	-2.95295300	-0.68180800
H	-3.47948300	-1.83201100	-2.01125000
C	-5.23890100	-3.04787100	-1.66553000
H	-4.94507400	-3.83343700	-2.36965900
H	-5.78502200	-3.52785100	-0.84687900
H	-5.94065100	-2.38638300	-2.18525200
C	-4.53782000	-1.54825100	1.86185800
H	-4.99420300	-0.77326200	2.49201900
H	-3.52465700	-1.69939400	2.25811400
C	-5.32238200	-2.86137700	1.99046700
H	-6.34866100	-2.76034300	1.62364500
H	-4.84601800	-3.66591000	1.41924100
H	-5.37805400	-3.18648900	3.03449700
C	-3.00578800	1.59223300	-0.76131600
C	-3.31343000	1.55447700	-2.12775700
C	-2.80469300	2.83862400	-0.15515800
C	-3.37432800	2.72599000	-2.87306500
H	-3.48906100	0.59610900	-2.60997100
C	-2.87189500	4.01362200	-0.90166700
H	-2.62469400	2.88549000	0.91388600
C	-3.14759700	3.95949400	-2.26322500
H	-3.60378600	2.67868000	-3.93310300
H	-2.71856800	4.97142200	-0.41406500
H	-3.19947300	4.87299300	-2.84653200
H	-1.70869500	-1.06670800	1.05961500

Thermal correction to Gibbs Free Energy= 0.699677

E(RM06L) = -2535.09787152

Reductive elimination from C^{Si}-(E)

C	-2.74011700	-1.90503100	0.80035000
C	-3.54419100	-1.18939700	-0.05149600
H	-3.34171100	-0.57025600	-0.91012900
N	-4.77348800	-2.26304700	1.45338300
N	-3.53476300	-2.54746900	1.69964000
N	-4.79714000	-1.43983800	0.38942400
C	-6.02911300	-0.94226100	-0.12659500
C	-7.14130000	-1.77720700	-0.15495800
C	-6.09294800	0.36918000	-0.58862700
C	-8.33713800	-1.28315500	-0.66445300
H	-7.06292000	-2.79117400	0.22040800
C	-7.29096500	0.84509400	-1.11065300
H	-5.22591900	1.01910500	-0.51453800
C	-8.41278400	0.02094900	-1.14884600
H	-9.21170800	-1.92480100	-0.68890300
H	-7.35000700	1.86642000	-1.47240200
H	-9.34812600	0.39705400	-1.55014400
C	-1.25473400	-2.00010200	0.87726700
H	-0.95045600	-3.02913200	1.08770900
H	-0.79142100	-1.68331100	-0.06008400
N	-0.71231200	-1.12373200	1.92132400
C	-1.21170100	-0.93400000	3.15888000
C	-0.43433300	0.02027500	3.78378400
H	-2.08535500	-1.48230400	3.48149100
C	0.53290700	0.36904800	2.83908800
H	-0.55713300	0.41596900	4.77947900
H	1.32973800	1.09449600	2.90600900
N	0.36242400	-0.33014000	1.71332000
Rh	1.60881900	-0.48053600	0.03892900
H	0.99360500	0.62493600	-0.86271000
C	1.67419300	1.59909000	0.10289800
C	3.06190200	1.97833600	0.50872700
C	3.65332800	3.12131600	-0.04833500
C	3.77562000	1.27525300	1.49151600
C	4.90849700	3.55361800	0.36965800
H	3.11766300	3.67417000	-0.81498500
C	5.02664600	1.71217200	1.91501100
H	3.34906600	0.38300100	1.94141900
C	5.59781100	2.85273400	1.35640700
H	5.34614300	4.44201100	-0.07485400
H	5.55445800	1.16003400	2.68642900
H	6.57380200	3.19307000	1.68711100
C	0.69798000	2.52209000	0.06809900
H	1.00291800	3.51190800	0.42012100
Si	-1.11614900	2.42978400	-0.50321300
C	-2.24269200	1.97362600	0.96516700
H	-3.28712900	1.99578500	0.62445400
H	-2.04626100	0.93960300	1.26646800
C	-1.29541600	1.15923300	-1.92908400
H	-0.43339300	1.27373800	-2.60142900
H	-1.21609900	0.14423500	-1.51730500
C	-2.06999200	2.90728600	2.16931300
H	-2.72118600	2.61018100	2.99764900
H	-1.03805700	2.88791900	2.53729100
H	-2.31060100	3.94548400	1.91384000
C	-2.58400900	1.30297500	-2.75081000
H	-2.67868000	0.50264800	-3.49265800
H	-3.48318100	1.27936700	-2.12292400
H	-2.60081400	2.25487100	-3.29061900
C	-1.56052500	4.18379800	-1.11492100
H	-1.05543100	4.90876500	-0.46301500
H	-1.12611800	4.33033600	-2.11272700
C	-3.06326600	4.49632400	-1.14494500
H	-3.50739100	4.41957300	-0.14629200
H	-3.24515800	5.51340000	-1.50648800
H	-3.61231200	3.81304300	-1.80162000
Cl	2.12789300	-2.88962300	0.56827900
C	3.07109700	-2.68293800	-1.06109800
C	2.32853800	-1.52952700	-1.74361400
C	4.43215700	-2.12838700	-0.72020300
C	3.20013000	-0.44990400	-1.75604900
C	4.50275300	-0.85234100	-1.14223800
C	3.10597400	0.73599900	-2.66807900

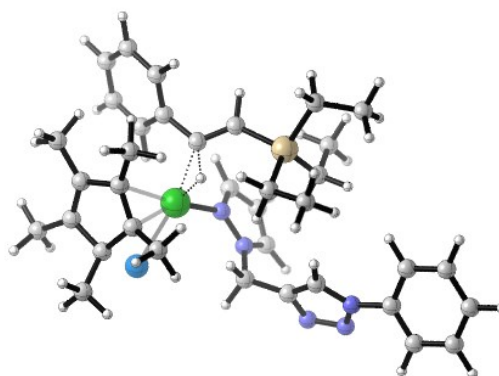


Figure S77. Calculated structure of the reductive elimination intermediate from C^{Si}-(E).

H	2.07987000	0.93758200	-2.98242300
H	3.50880000	1.63641900	-2.20066100
H	3.70421200	0.53390400	-3.56555900
C	1.18134000	-1.80070600	-2.67691600
H	1.53897200	-2.32248200	-3.57404600
H	0.42202000	-2.43294500	-2.20533200
H	0.69749300	-0.87527100	-2.99407200
C	3.06529800	-4.01203500	-1.78304200
H	3.59718600	-3.88821800	-2.73314900
H	3.58368200	-4.77479400	-1.19642300
H	2.04976600	-4.35672700	-1.99117800
C	5.46010500	-2.95866600	-0.02571800
H	6.32410000	-2.36039200	0.27043900
H	5.04531100	-3.42088000	0.87821100
H	5.82232800	-3.76904700	-0.66976200
C	5.68106900	0.06509600	-1.12675100
H	6.10977300	0.14624300	-2.13334000
H	5.39836400	1.07202300	-0.81041700
H	6.46659800	-0.29228600	-0.45792500

Thermal correction to Gibbs Free Energy= 0.698879
E(RM06L) = -2535.05228471

Reductive elimination from $C^{Si}-(Z)$

C	3.35052500	1.73045100	1.30947600
C	3.43211300	0.36568500	1.18136800
H	2.68892000	-0.41830600	1.19524300
N	5.43490000	1.29861400	0.94514200
N	4.59156100	2.25953200	1.16248500
N	4.74741900	0.14107100	0.95318200
C	5.39626900	-1.10171500	0.68632600
C	6.63270300	-1.09508400	0.04524000
C	4.77097700	-2.29535200	1.03897700
C	7.24782500	-2.30843000	-0.24302200
H	7.10112100	-0.15219300	-0.21122000
C	5.39448200	-3.50034700	0.73186700
H	3.81499400	-2.29698200	1.55235500
C	6.63132400	-3.51129500	0.09272400
H	8.21421400	-2.30984900	-0.73655000
H	4.91174500	-4.43276000	1.00567900
H	7.11578800	-4.45434300	-0.13811000
C	2.15172900	2.61265900	1.46255200
H	2.40861800	3.50260400	2.04090800
H	1.76881500	2.94920300	0.49415500
N	1.06887600	1.93281600	2.14089100
C	0.81220900	1.95545000	3.46458200
C	-0.09925800	0.95727900	3.73564300
H	1.30919200	2.66739900	4.10726100
C	-0.33098800	0.34227200	2.50334600
H	-0.53554200	0.70476300	4.68906400
H	-0.98390600	-0.47991900	2.25542800
N	0.38589900	0.93194700	1.54169000
Rh	0.17828300	0.39429100	-0.53378500
H	-0.90944300	1.46353000	-0.36354400
C	-2.01373300	0.22480600	-0.18423500
C	-2.90025000	1.41071200	-0.01033200
C	-4.00724200	1.55884700	-0.85710900
C	-2.67304500	2.38601900	0.97245000
C	-4.88565800	2.62654200	-0.70110800
H	-4.17720400	0.82535100	-1.63793100
C	-3.55326400	3.45097700	1.12709000
H	-1.82410300	2.29503200	1.64252100
C	-4.66394400	3.57354700	0.29435000
H	-5.74233600	2.71809900	-1.36149100
H	-3.37606700	4.18406700	1.90778000
H	-5.34961400	4.40532800	0.41917500
C	-2.44058700	-1.03555000	-0.22535200
Cl	0.66410900	-1.81651800	0.35592500
C	0.62702100	1.83353900	-2.12986600
C	1.95522100	1.28027600	-1.95015500
C	-0.23552000	0.77479200	-2.62623400
C	1.87196600	-0.10739900	-2.12446500
C	0.49387500	-0.43783200	-2.52015100
C	2.97734800	-1.10329900	-2.05715600

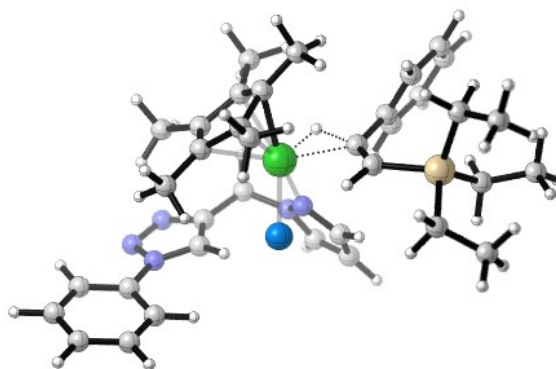


Figure S78. Calculated structure of the reductive elimination intermediate from $C^{Si}-(Z)$.

H	2.72639200	-1.90422700	-1.35355800
H	3.12463800	-1.55838400	-3.04355300
H	3.91962500	-0.65110700	-1.74407100
C	3.20539300	2.08534400	-1.81174500
H	3.50530000	2.41148800	-2.81595400
H	3.07556600	2.98681500	-1.21127400
H	4.03121200	1.51238400	-1.39019000
C	0.30059700	3.29420400	-2.16520900
H	0.54842700	3.70134500	-3.15275600
H	-0.76078300	3.47416400	-1.97865700
H	0.87535500	3.85427100	-1.42368800
C	-1.58023700	0.95966700	-3.24851600
H	-2.19386800	0.06260300	-3.14906400
H	-2.12511500	1.79583800	-2.80562800
H	-1.44455900	1.16911800	-4.31609800
C	0.03855500	-1.78812000	-2.96415800
H	0.30098100	-1.92792000	-4.01988200
H	0.51591600	-2.57387600	-2.37605100
H	-1.04396200	-1.89485300	-2.86496400
H	-1.72150500	-1.82644800	-0.42641300
Si	-4.17644300	-1.71970400	0.26320800
C	-3.77637700	-3.53145600	0.72167800
H	-2.80138800	-3.53671100	1.22803900
H	-3.63141900	-4.10050000	-0.20684400
C	-4.81229800	-4.23234200	1.61013400
H	-4.50557300	-5.25923900	1.83410900
H	-4.93403600	-3.71084600	2.56602100
H	-5.79627800	-4.28039800	1.13237600
C	-4.86863400	-0.77688800	1.77435800
H	-4.63413500	0.28930400	1.68111400
H	-4.33373200	-1.13177600	2.66561400
C	-6.38200100	-0.94650100	1.97029100
H	-6.72575200	-0.40328300	2.85659600
H	-6.93901800	-0.55749500	1.11050300
H	-6.66388600	-1.99686700	2.09905400
C	-5.37467200	-1.63876000	-1.23060300
H	-4.78238900	-1.67843900	-2.15528100
H	-5.86935500	-0.65930300	-1.22555300
C	-6.43553600	-2.74740900	-1.25606400
H	-5.97639800	-3.74093800	-1.29507600
H	-7.07525700	-2.71280300	-0.36799200
H	-7.08449600	-2.64892900	-2.13246700

Thermal correction to Gibbs Free Energy= 0.697073
E(RM06L) = -2535.09357752

Reductive elimination from C^{Si}- α

C	2.53859100	-2.43306800	-0.66219900
C	3.08487800	-1.21044800	-0.96828000
H	2.65028600	-0.25776700	-1.23239500
N	4.67474100	-2.67265600	-0.44765500
N	3.54065500	-3.29486800	-0.35387200
N	4.41637100	-1.40513400	-0.81995100
C	5.46757700	-0.45006900	-0.95816800
C	6.68750100	-0.69125600	-0.33105900
C	5.24740600	0.71618800	-1.68703300
C	7.69968800	0.25581000	-0.44150700
H	6.83529100	-1.61063000	0.22317300
C	6.26622800	1.65906300	-1.77662400
H	4.30216000	0.89185600	-2.19016400
C	7.49223200	1.43261100	-1.15693100
H	8.65476100	0.07091200	0.03923300
H	6.09996900	2.56854500	-2.34461100
H	8.28553900	2.16860400	-1.23609200
C	1.10521500	-2.84307000	-0.53059500
H	0.98797000	-3.89664300	-0.79230100
H	0.74423200	-2.71689500	0.49500200
N	0.24356400	-2.06461900	-1.39342800
C	-0.17284100	-2.40531600	-2.63037300
C	-0.67819300	-1.27273800	-3.23163200
H	-0.05722400	-3.41763000	-2.98900300
C	-0.50329300	-0.25966600	-2.28577400
H	-1.10729900	-1.18693000	-4.21744000
H	-0.77274300	0.78387400	-2.34046200

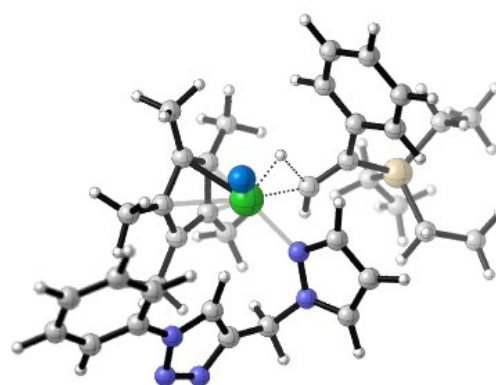


Figure S79. Calculated structure of the reductive elimination intermediate from C^{Si}- α .

N	0.06580200	-0.74184300	-1.17862400
Rh	0.21686700	0.43504600	0.64737700
H	-0.95288300	1.38895200	0.30033800
C	-1.84662000	-0.10430600	0.53996500
C	-2.93885500	0.41284300	-0.01864300
Cl	1.22759800	1.99844700	-0.85198600
C	0.29052400	-0.58232800	2.64617100
C	1.69033400	-0.60721600	2.29489700
C	-0.12017700	0.79607200	2.76462600
C	2.09747400	0.71184500	2.02967500
C	0.96460300	1.60070000	2.30155600
C	3.46143000	1.18277600	1.65380200
H	3.42144700	1.80082800	0.75059300
H	3.87623300	1.79707900	2.46123200
H	4.14477700	0.35184600	1.47119100
C	2.57890600	-1.80076500	2.42208900
H	2.88397100	-1.87695300	3.47377200
H	2.08036600	-2.73759600	2.16784800
H	3.48494400	-1.72124900	1.82179300
C	-0.49352400	-1.76200300	3.12470100
H	-1.55852400	-1.53396100	3.20717300
H	-0.37629000	-2.62806700	2.46598000
H	-0.14133500	-2.05861900	4.11966800
C	-1.39193300	1.27848600	3.38626400
H	-1.56176800	2.33579300	3.17501600
H	-2.26024200	0.71482800	3.03700000
H	-1.31947700	1.16231200	4.47372900
C	1.03792900	3.09340500	2.31989100
H	1.37491900	3.42846300	3.30808400
H	1.73723100	3.45850500	1.56628600
H	0.06421300	3.54392200	2.11498400
Si	-4.51204100	-0.71123400	-0.07647300
C	-6.00816800	0.38403100	-0.51607700
H	-5.97921600	0.60291400	-1.59096900
H	-5.87926600	1.35266100	-0.01579700
C	-7.37087500	-0.21418700	-0.13961400
H	-8.18431000	0.45830400	-0.42969500
H	-7.54689700	-1.17520500	-0.63473900
H	-7.45080100	-0.37796400	0.94070500
C	-4.15287400	-2.02703200	-1.41753700
H	-3.46980800	-2.77165700	-0.98430900
H	-3.58313100	-1.54380100	-2.22232300
C	-5.39123200	-2.71706400	-2.00370800
H	-5.11087200	-3.45627200	-2.76137200
H	-5.97096200	-3.23821800	-1.23482000
H	-6.05789800	-1.99191700	-2.48317800
C	-4.72044500	-1.51953400	1.64461600
H	-5.17232900	-0.77804200	2.31686700
H	-3.72559000	-1.73268800	2.05927200
C	-5.54970200	-2.81118900	1.65166700
H	-6.56054900	-2.64740000	1.26506000
H	-5.08231900	-3.58642000	1.03440400
H	-5.64779400	-3.21118600	2.66613100
C	-2.96423100	1.74143700	-0.67279100
C	-3.26632400	1.86035400	-2.03559000
C	-2.66164600	2.90036300	0.05250000
C	-3.22300600	3.09948200	-2.66283000
H	-3.51910100	0.97123200	-2.60801900
C	-2.62366900	4.14321300	-0.57625500
H	-2.48657800	2.82820600	1.12243100
C	-2.89567400	4.24384300	-1.93574100
H	-3.44808800	3.17535300	-3.72205400
H	-2.39092500	5.03282100	0.00064900
H	-2.86581100	5.21053000	-2.42769900
H	-1.85921200	-1.09176900	0.99618700

Thermal correction to Gibbs Free Energy= 0.695658

E(RM06L) = -2535.09736703

RhCl₂ based compounds

Catalyst

C	0.15296000	1.24321100	-0.68126200
C	1.48157600	1.58121400	-0.66559700
H	2.00682500	2.47468300	-0.96409500
N	1.28381000	-0.50439600	0.06375800
N	0.10932000	-0.03572400	-0.23239200
N	2.12448100	0.47495500	-0.21016600
C	3.53623500	0.29466100	-0.01598100
C	4.12034700	-0.90234600	-0.41449600
C	4.26722000	1.32840300	0.55925800
C	5.48862600	-1.06103600	-0.22704000
H	3.51999800	-1.68642500	-0.86284500
C	5.63676100	1.15588000	0.72746500
H	3.77876200	2.23777300	0.89596000
C	6.24467300	-0.03505100	0.33657800
H	5.96462300	-1.98717900	-0.52998200
H	6.22457800	1.94815400	1.17791000
H	7.31249600	-0.16593600	0.47572800
C	-1.08678600	1.93569200	-1.14934600
H	-0.91444900	3.00764400	-1.24912900
H	-1.39437200	1.52963000	-2.11895200
N	-2.17860600	1.76154000	-0.19759600
C	-3.05101000	2.68733800	0.25096200
C	-3.98175200	2.05360100	1.04864700
H	-2.94854200	3.72359000	-0.03613500
C	-3.61142700	0.70874100	1.05189400
H	-4.81096600	2.50229200	1.57233600
H	-4.03821800	-0.13336900	1.57658000
N	-2.53362300	0.54873300	0.27861400
Rh	-1.50615600	-1.08839300	-0.02769400
Cl	-1.29188100	-1.51313300	2.19879300
Cl	-1.86964500	-1.20044200	-2.28897900

Thermal correction to Gibbs Free Energy= 0.181984

E(RM06L) = -1768.76458763

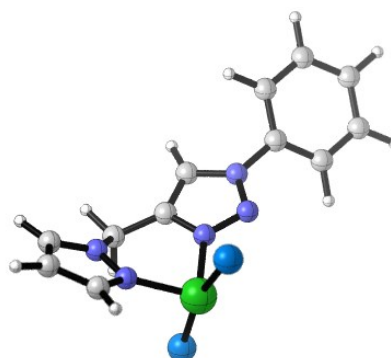


Figure S80. Calculated structure of Rh^{III}Cl₂.

Rh-H-Si intermediate

C	1.22969300	-0.16270900	1.47600800
C	2.54954400	0.12891000	1.70357400
H	3.13407200	0.22602300	2.60455500
N	2.18806700	0.05316400	-0.48310700
N	1.07151400	-0.19924900	0.13109600
N	3.08882400	0.25733900	0.46473100
C	4.45206700	0.53623200	0.11849600
C	5.02846500	-0.15654400	-0.94067000
C	5.15305900	1.48422900	0.85601100
C	6.35272700	0.11631600	-1.26467300
H	4.45251500	-0.89183000	-1.49207700
C	6.48070300	1.73454500	0.52610200
H	4.66916500	2.03581300	1.65646200
C	7.07824800	1.05355100	-0.53194500
H	6.81984100	-0.41310700	-2.08802700
H	7.04256700	2.47210900	1.08889900
H	8.11266500	1.25636400	-0.78847300
C	0.05327100	-0.36501800	2.37313800
H	0.37222200	-0.49309100	3.40805200
H	-0.60709400	0.50855700	2.32750100
N	-0.71022300	-1.55048300	2.01168300
C	-1.15541400	-2.52894100	2.82850300
C	-1.93353400	-3.38969300	2.08437300
H	-0.89118700	-2.53242400	3.87585100
C	-1.92710900	-2.86868200	0.78895800
H	-2.43265000	-4.28291400	2.42505000
H	-2.38154700	-3.22990500	-0.12128400
N	-1.20195000	-1.74769000	0.76556300
Rh	-0.67222300	-0.73535500	-0.83283200
H	-1.38828600	0.69571100	-0.17246900
Cl	-2.56618700	-1.17226800	-2.05504600

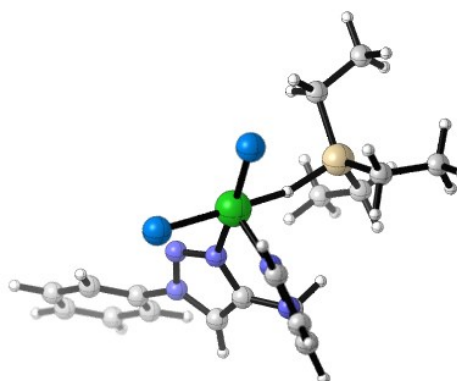


Figure S81. Calculated structure of the Rh-H-Si intermediate.

Si	-2.59950300	1.71893500	0.00391200
C	-4.03193000	0.70291600	0.70540200
H	-4.50732800	0.15787300	-0.11658600
H	-3.62094300	-0.06132200	1.37864400
C	-2.83199800	2.51708100	-1.69112300
H	-1.89871700	3.01737900	-1.97779900
H	-3.00584100	1.72456500	-2.42680900
C	-1.81489900	2.88313000	1.28566400
H	-2.47137000	3.76063800	1.36544400
H	-1.84646500	2.41188000	2.27710200
C	-0.39063900	3.33177100	0.93716300
H	-0.35850700	3.84907300	-0.02731400
H	0.00197500	4.02019500	1.69162000
H	0.29658000	2.47894700	0.87105800
C	-5.05559300	1.56983700	1.45808100
H	-5.86805800	0.94864200	1.84595300
H	-4.60188300	2.09036600	2.30872700
H	-5.50650700	2.32593800	0.80712100
C	-3.99807900	3.52095600	-1.70143500
H	-4.07742200	3.99647600	-2.68306700
H	-4.95402700	3.02638100	-1.50033500
H	-3.86510100	4.31818400	-0.96129800
Cl	0.42143900	-2.51770500	-1.73632700

Thermal correction to Gibbs Free Energy= 0.373837
E(RM06L) = -2296.63908594

C^{Si}- α intermediate

C	-3.08075900	-1.59926100	-0.55171300
C	-4.38550400	-1.27324700	-0.28673700
H	-5.30188500	-1.84072800	-0.31773200
N	-3.11844900	0.53053700	-0.05093900
N	-2.35344500	-0.46212600	-0.39751100
N	-4.35459000	0.04905800	0.01248600
C	-5.45170000	0.90368500	0.34963900
C	-5.48747100	2.19281300	-0.17096000
C	-6.45483400	0.41868500	1.18225100
C	-6.56143000	3.01290200	0.15622700
H	-4.69130200	2.53934100	-0.82027500
C	-7.52948500	1.24681200	1.48877200
H	-6.38938000	-0.57988100	1.60340400
C	-7.58261200	2.54145300	0.97848200
H	-6.60368200	4.02104400	-0.24208800
H	-8.31861400	0.88178500	2.13745200
H	-8.42063000	3.18530900	1.22404400
C	-2.43089900	-2.88079200	-0.96790600
H	-3.14479200	-3.70450100	-0.92822200
H	-2.04596900	-2.79186600	-1.98749300
N	-1.32594200	-3.22134300	-0.08058800
C	-1.19140300	-4.32942900	0.67688200
C	-0.02870100	-4.19792900	1.40648700
H	-1.92349100	-5.12248500	0.63600700
C	0.48237600	-2.95234400	1.02823300
H	0.38760200	-4.89969900	2.11178200
H	1.36942700	-2.44683300	1.37782800
N	-0.30194700	-2.36891500	0.12202300
Rh	-0.19952500	-0.30646000	-0.74079300
C	1.51845000	0.30555700	-1.07610500
C	2.13676600	0.92765300	0.06428000
Si	3.78974100	-0.22288700	0.39103200
C	4.12952000	-0.40581100	2.25433800
H	5.17088600	-0.75674400	2.29767100
H	4.14932600	0.59469300	2.70192200
C	5.16247400	0.79326000	-0.44597700
H	4.79634400	1.15461800	-1.41570400
H	5.34874500	1.69125100	0.15414400
C	3.24848600	-1.33994100	3.09046400
H	3.61674000	-1.38861900	4.11981100
H	2.21055000	-0.99856300	3.13039500
H	3.25691500	-2.36234400	2.69849800
C	6.46404400	0.00304500	-0.64415600
H	7.23690300	0.64286600	-1.08029800
H	6.85768200	-0.38337400	0.30286500
H	6.32196500	-0.84644100	-1.32050500

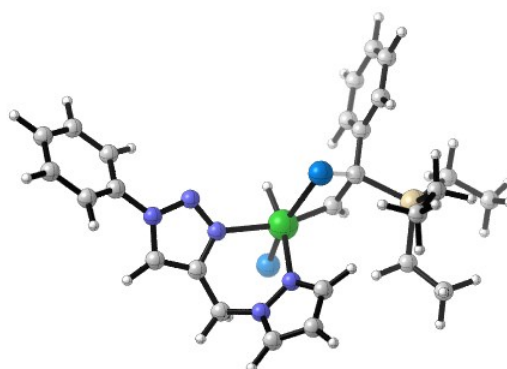


Figure S82. Calculated structure of the C^{Si}- α intermediate.

C	3.55606000	-1.87519300	-0.53823700
H	2.51353800	-2.20848800	-0.52220800
H	3.77807100	-1.67528100	-1.59548300
C	4.46845800	-3.00508500	-0.03556700
H	4.21952700	-3.30356600	0.98752300
H	4.36426200	-3.88953200	-0.67103400
H	5.52433800	-2.71667100	-0.04521000
C	2.49848100	2.38828700	0.01278100
C	2.36222100	3.11011900	-1.17570400
C	3.00102000	3.02801200	1.15066900
C	2.75078000	4.44610900	-1.23015200
H	1.95130700	2.64193000	-2.06530500
C	3.37367100	4.36424600	1.09583900
H	3.08391600	2.48583400	2.08823400
C	3.25793700	5.07440800	-0.09821000
H	2.64428700	4.99619200	-2.15937300
H	3.75646100	4.85226400	1.98622500
H	3.55413400	6.11738300	-0.14115200
H	2.03574800	0.31488200	-2.03902100
Cl	-0.43584400	-0.98047600	-2.92887700
Cl	0.78263800	0.65702200	1.36484300
H	-0.45721600	1.12748500	-1.27872200

Thermal correction to Gibbs Free Energy= 0.484258
E(RM06L) = -2605.10654682

Reductive elimination from C^{Si}- α

C	1.06630900	0.00461300	1.73797600
C	2.26633100	-0.54408800	1.29721300
H	2.64666400	-1.55362700	1.25970700
N	2.37564900	1.67055100	1.29393100
N	1.18509800	1.36077000	1.70863400
N	3.04328000	0.54334000	1.05280800
C	4.39711400	0.57237000	0.59281500
C	5.24372500	1.56520900	1.07530000
C	4.82749200	-0.39141900	-0.31250900
C	6.56123700	1.58686000	0.63327100
H	4.87407100	2.30241800	1.77902500
C	6.15243500	-0.36151200	-0.73376800
H	4.13607400	-1.12678800	-0.71165800
C	7.01741100	0.62363800	-0.26427700
H	7.23415300	2.35545200	0.99844000
H	6.50147800	-1.10265900	-1.44481300
H	8.04842400	0.64420500	-0.60193300
C	-0.14580000	-0.59580000	2.38709900
H	0.03586200	-0.75430200	3.45377300
H	-0.97522000	0.11099100	2.28717600
N	-0.51486900	-1.87861600	1.82217700
C	-1.02547300	-2.95976600	2.44413400
C	-1.30971000	-3.90293200	1.47713200
H	-1.14139100	-2.97884800	3.51789700
C	-0.93469300	-3.30723700	0.26990800
H	-1.71244900	-4.89221000	1.62563600
H	-0.94800400	-3.70558500	-0.73273900
N	-0.47441500	-2.07635200	0.48805900
Rh	0.53572200	-0.72594300	-0.73796000
H	-0.31177500	-1.03776800	-2.02534200
C	-0.97746500	0.41303900	-1.41825000
C	-2.13808500	0.35728700	-0.76433200
Si	-2.70698400	2.21205100	-0.23428300
C	-4.57142200	2.18532800	-0.63194300
H	-4.73410200	1.58814400	-1.53756800
H	-5.08779100	1.65319600	0.17658000
C	-1.74705400	3.48760000	-1.26166700
H	-1.79008800	3.22234400	-2.32619700
H	-2.32324300	4.41955700	-1.18093800
C	-5.17293100	3.58615000	-0.82139600
H	-6.25073700	3.51565200	-0.99590700
H	-5.02318100	4.22293700	0.05656800
H	-4.73300400	4.09753700	-1.68334700
C	-0.30026500	3.74910100	-0.81544000
H	0.15081700	4.53176900	-1.43215000
H	-0.25696000	4.08229400	0.22629600
H	0.34056800	2.86507300	-0.89632000

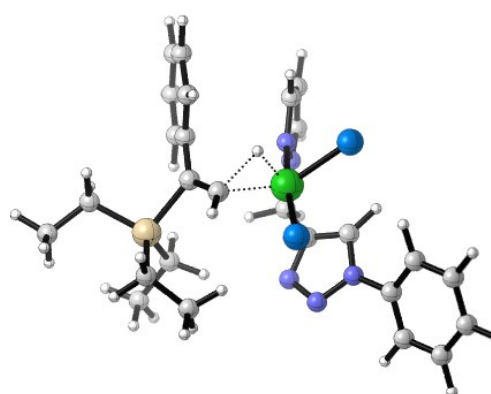


Figure S83. Calculated structure of the reductive elimination intermediate from C^{Si}- α .

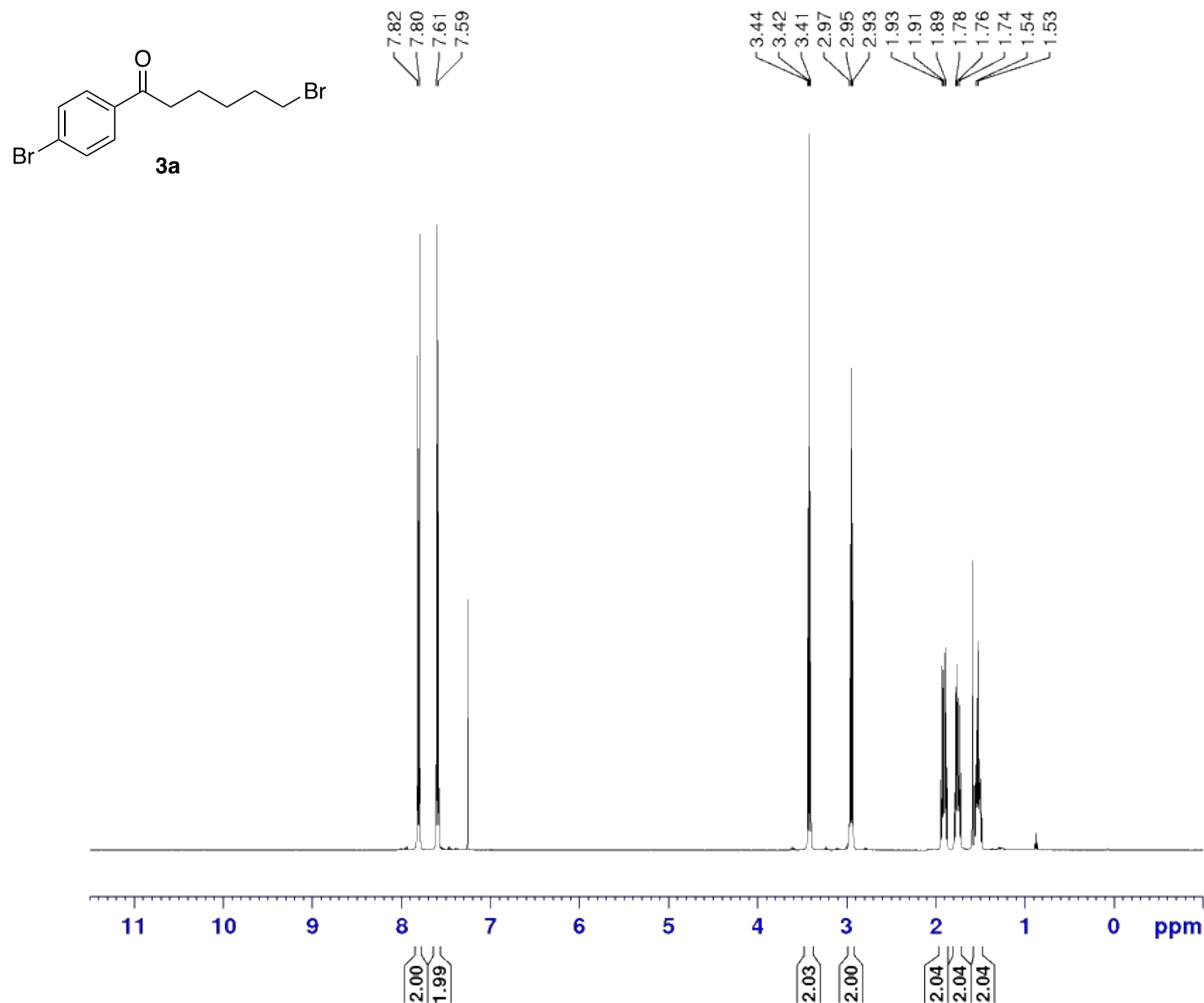
C	-2.28812300	2.43641300	1.60693400
H	-2.76215400	1.66045100	2.22028300
H	-1.20183900	2.32696000	1.72211000
C	-2.72841800	3.81880300	2.11878600
H	-2.29464300	4.62866000	1.52241500
H	-3.81727800	3.93023400	2.09730900
H	-2.40175700	3.96490100	3.15249300
C	-2.99175900	-0.78531700	-0.48285600
C	-3.17997300	-1.75288800	-1.48582900
C	-3.67534000	-0.91731400	0.73521000
C	-4.02941100	-2.82832500	-1.26862800
H	-2.67728200	-1.64227400	-2.44249000
C	-4.49096200	-2.01693300	0.96160600
H	-3.55168100	-0.16950300	1.51116100
C	-4.67539500	-2.96820400	-0.04121500
H	-4.18560500	-3.55974900	-2.05468100
H	-5.00037200	-2.12387500	1.91350200
H	-5.33305200	-3.81418100	0.13070800
H	-0.60092500	1.15335100	-2.11819100
Cl	1.88046200	0.70106800	-1.96400200
Cl	1.80982500	-2.51177100	-1.43303100

Thermal correction to Gibbs Free Energy= 0.483481
E(RM06L) = -2605.08667647

8. References

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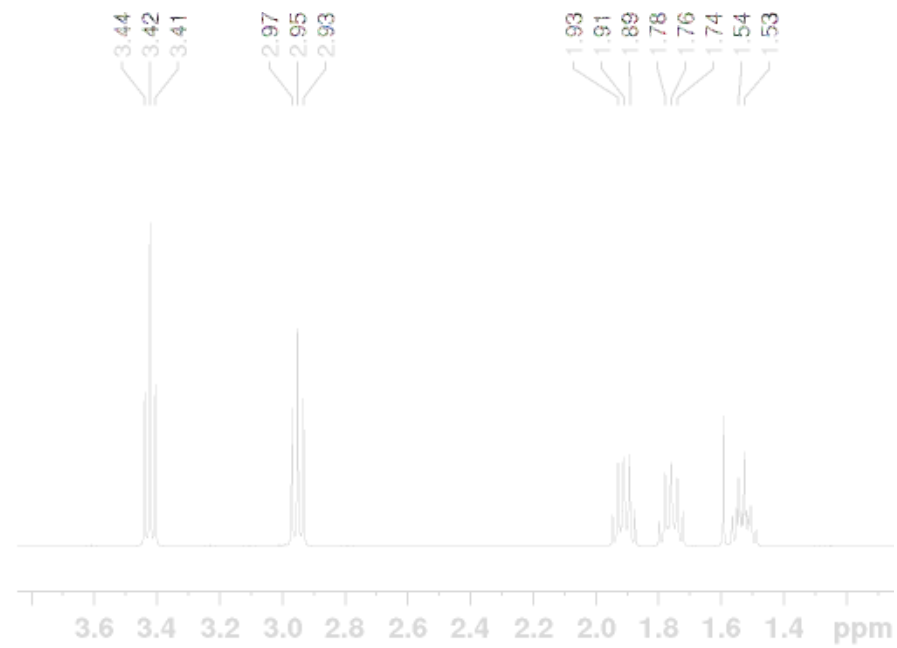
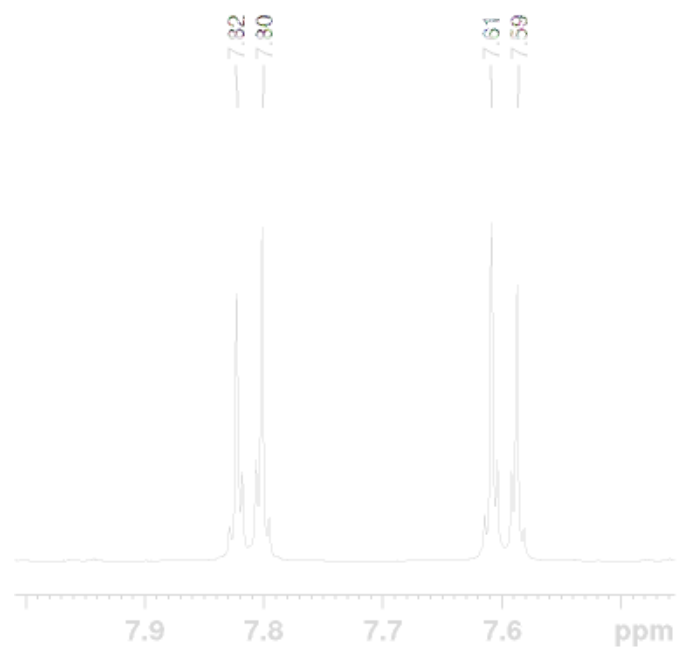
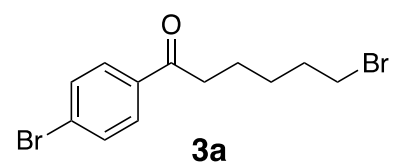
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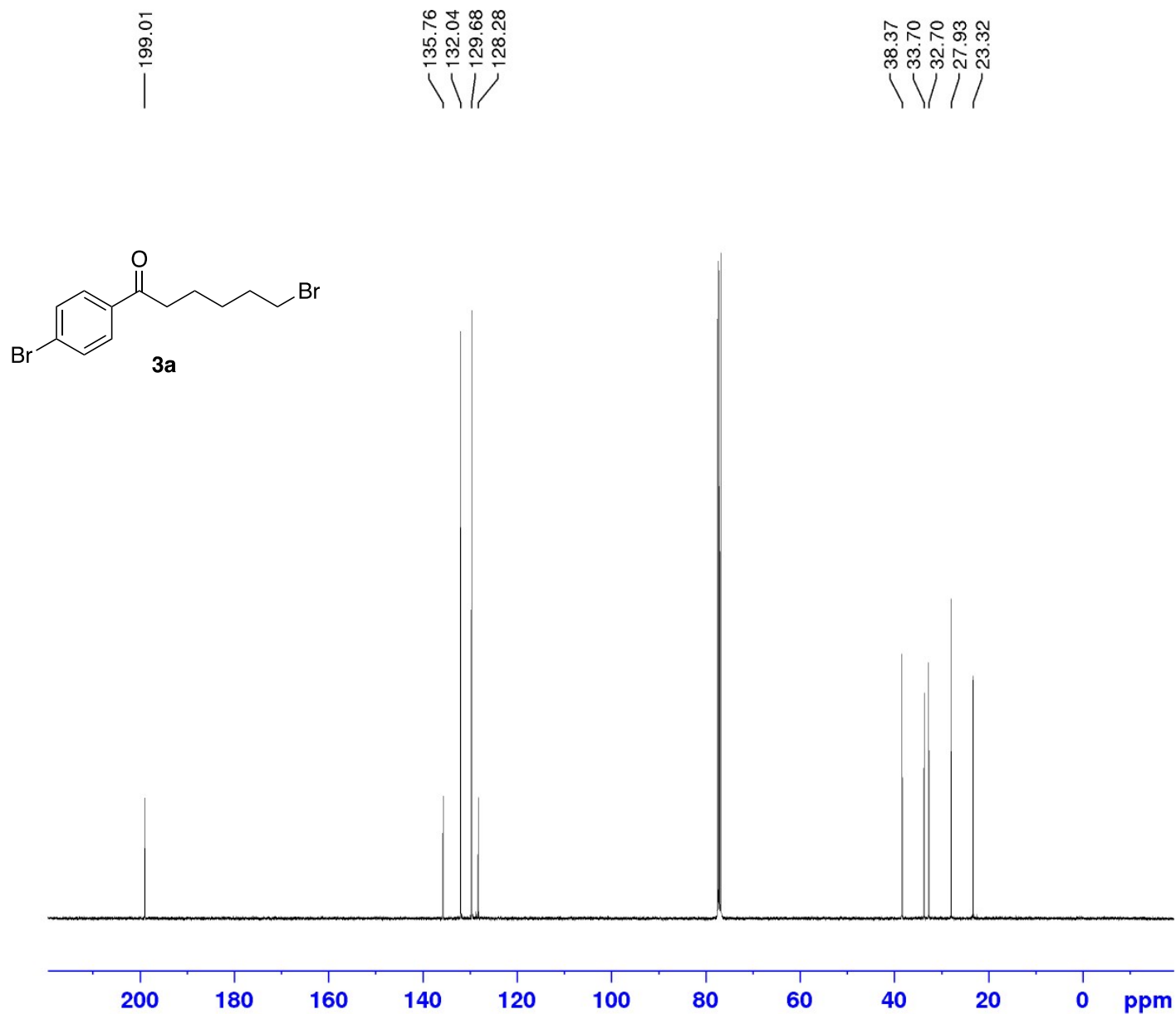


Current Data Parameters
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 PROCNO 1

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 Time 8.48 h
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 PROBHD Z116098_0573 {
 PULPROG zg30
 TD 65536
 SOLVENT CDCl3
 NS 16
 DS 2
 SWH 8012.820 Hz
 FIDRES 0.244532 Hz
 AQ 4.0894465 sec
 RG 87.74
 DW 62.400 usec
 DE 12.00 usec
 TE 298.0 K
 D1 2.50000000 sec
 TD0 1
 SFO1 400.1524709 MHz
 NUC1 1H
 P1 10.00 usec
 PLW1 15.75399971 W

F2 - Processing parameters
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 SF 400.1500097 MHz
 WDW EM
 SSB 0
 LB 0.30 Hz
 GB 0
 PC 1.00

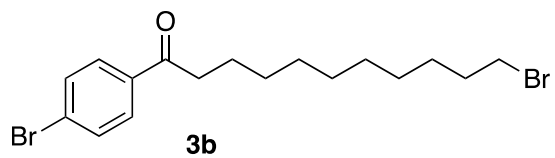




Current Data Parameters
NAME MR-204
EXPNO 5
PROCNO 1

F2 - Acquisition Parameters
Date_ 20190726
Time 23.13 h
INSTRUM spect
PROBHD Z116098_0573 (
PULPROG zgpg30
TD 65536
SOLVENT CDCl3
NS 3000
DS 4
SWH 24038.461 Hz
FIDRES 0.733596 Hz
AQ 1.3631488 sec
RG 198.41
DW 20.800 usec
DE 6.50 usec
TE 298.0 K
D1 2.00000000 sec
D11 0.03000000 sec
TD0 1
SFO1 100.6278593 MHz
NUC1 13C
P1 10.00 usec
PLW1 80.53199768 W
SFO2 400.1516006 MHz
NUC2 1H
CPDPRG[2] waltz16
PCPD2 90.00 usec
PLW2 15.75399971 W
PLW12 0.19449000 W
PLW13 0.09782900 W

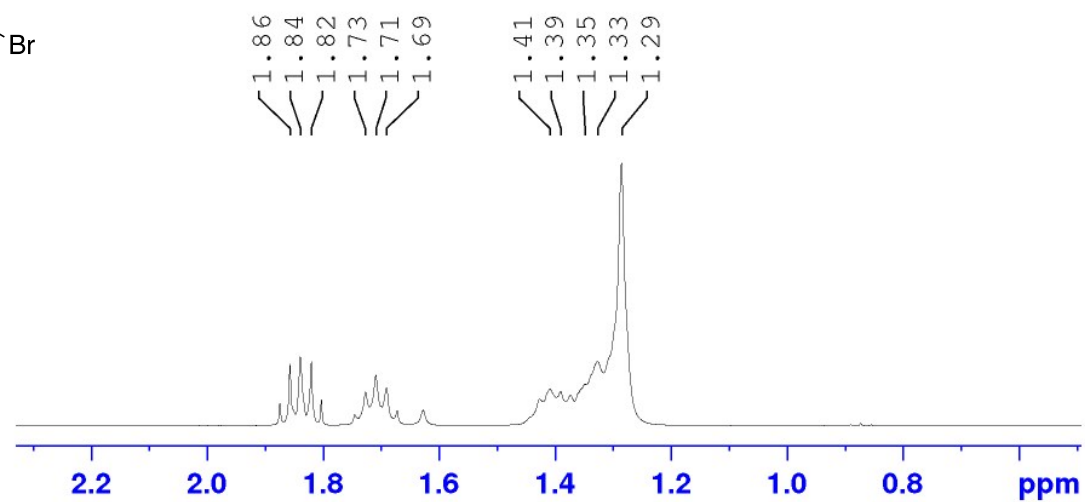
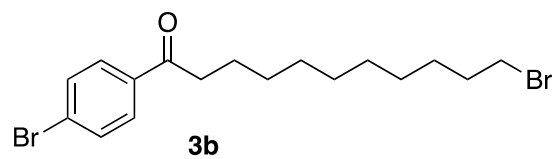
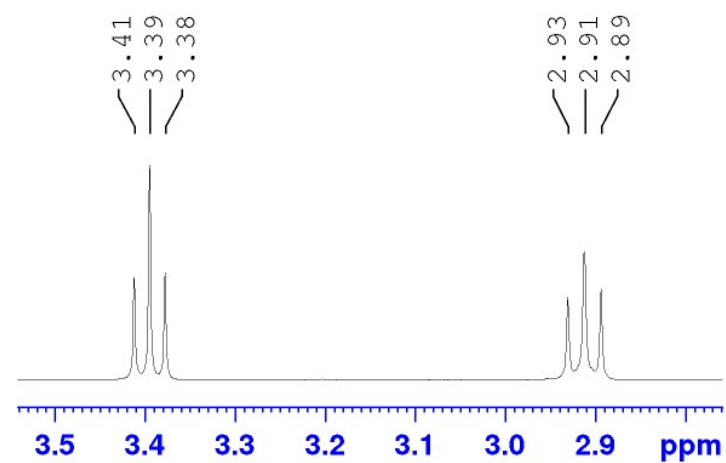
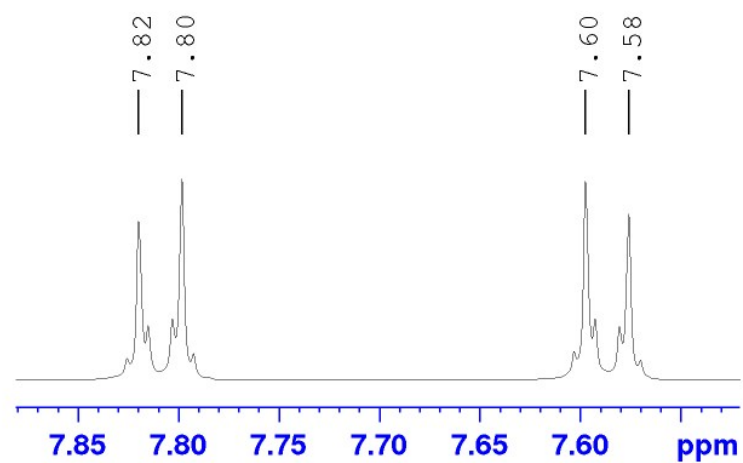
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WDW EM
SSB 0
LB 1.00 Hz
GB 0
PC 1.40

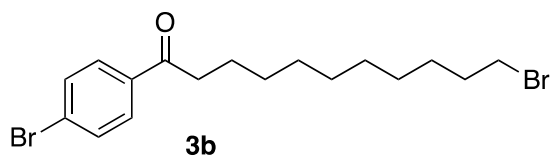


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Current Data Parameters
NAME MR-212
EXPNO 1



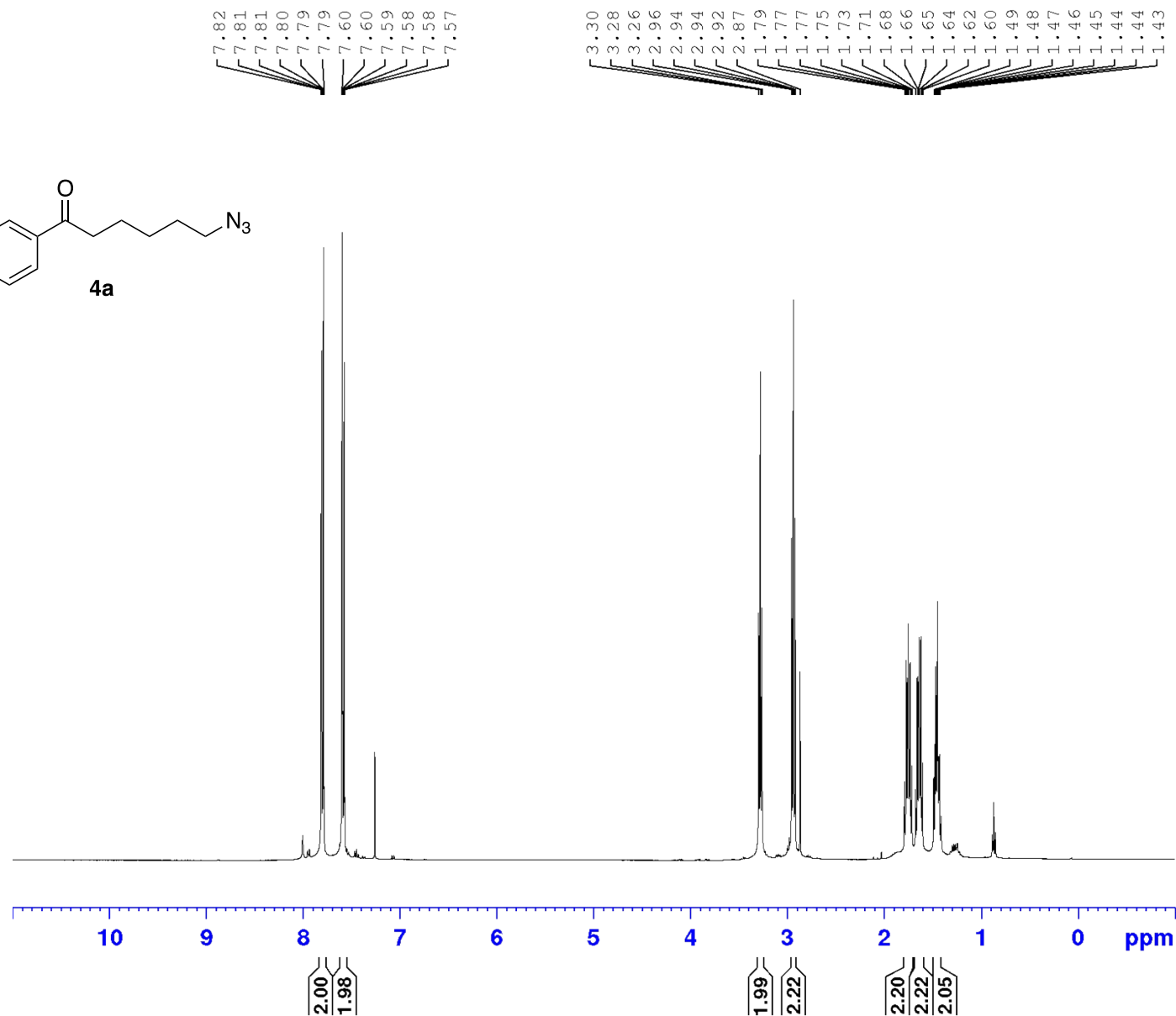
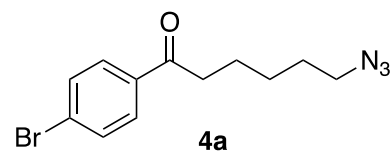


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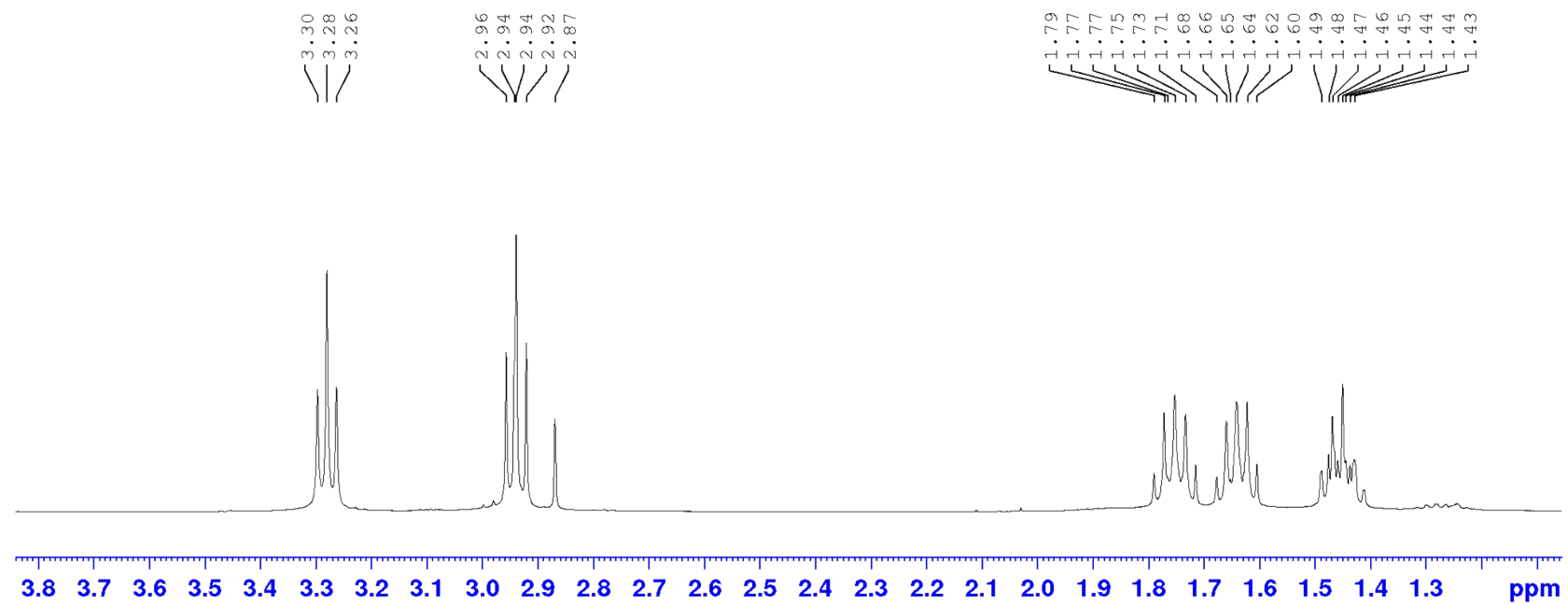
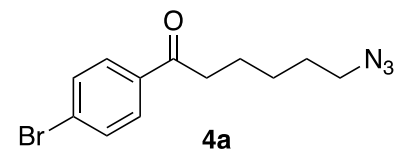
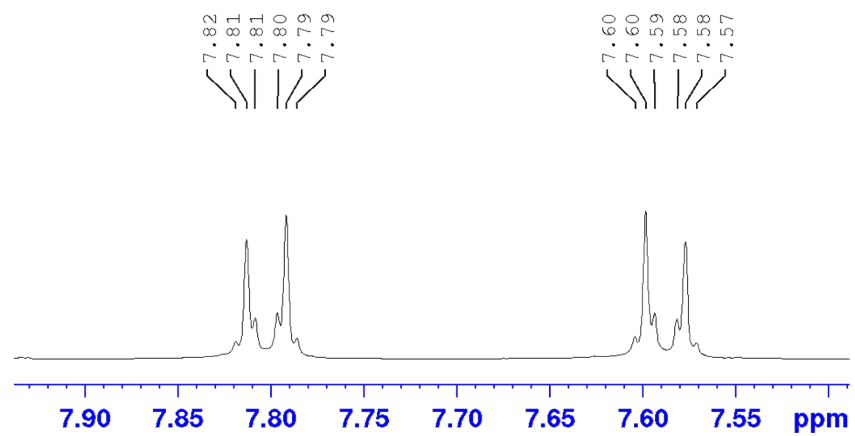
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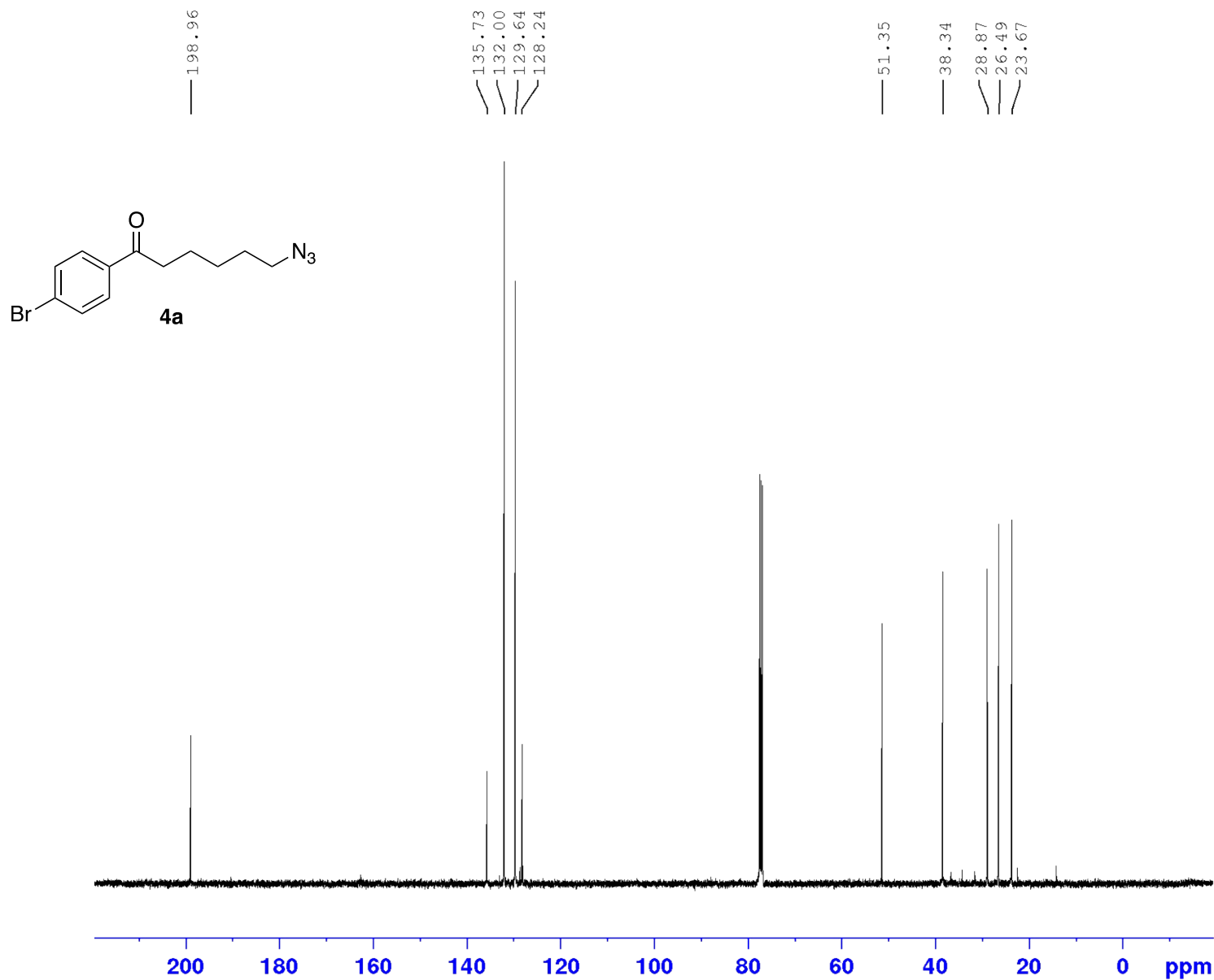


Current Data Parameters
NAME MR-94
EXPNO 1
PROCNO 1

F2 - Acquisition Parameters
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Time 15.55 h
INSTRUM spect
PROBHD Z116098_0573 {
PULPROG zg30
TD 65536
SOLVENT CDCl3
NS 16
DS 2
SWH 8012.820 Hz
FIDRES 0.244532 Hz
AQ 4.0894465 sec
RG 55.9
DW 62.400 usec
DE 12.00 usec
TE 298.0 K
D1 2.50000000 sec
TD0 1
SFO1 400.1524709 MHz
NUC1 1H
P1 10.00 usec
PLW1 15.7539971 W

F2 - Processing parameters
SI 65536
SF 400.1500098 MHz
WDW EM
SSB 0
LB 0.30 Hz
GB 0
PC 1.00

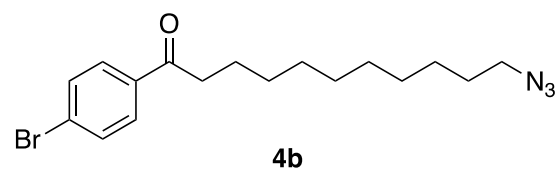




Current Data Parameters
NAME MR-94
EXPNO 2
PROCNO 1

F2 - Acquisition Parameters
Date_ 20190226
Time_ 16.17 h
INSTRUM spect
PROBHD Z116098_0573 (
PULPROG zgpg30
TD 65536
SOLVENT CDCl3
NS 350
DS 4
SWH 24038.461 Hz
FIDRES 0.733596 Hz
AQ 1.3631488 sec
RG 198.41
DW 20.800 usec
DE 6.50 usec
TE 298.0 K
D1 2.00000000 sec
D11 0.03000000 sec
TD0 1
SFO1 100.6278593 MHz
NUC1 13C
P1 10.00 usec
PLW1 80.53199768 W
SFO2 400.1516006 MHz
NUC2 1H
CPDPRG[2] waltz16
PCPD2 90.00 usec
PLW2 15.75399971 W
PLW12 0.19449000 W
PLW13 0.09782900 W

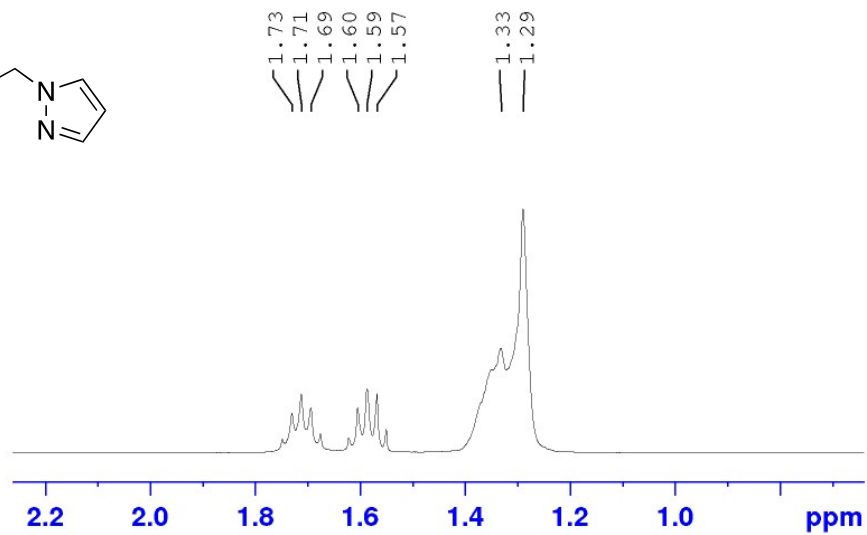
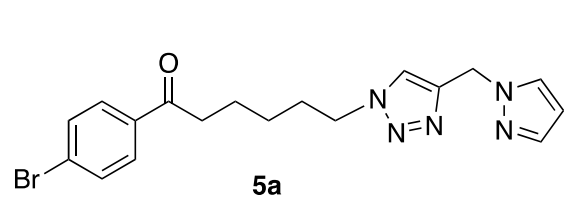
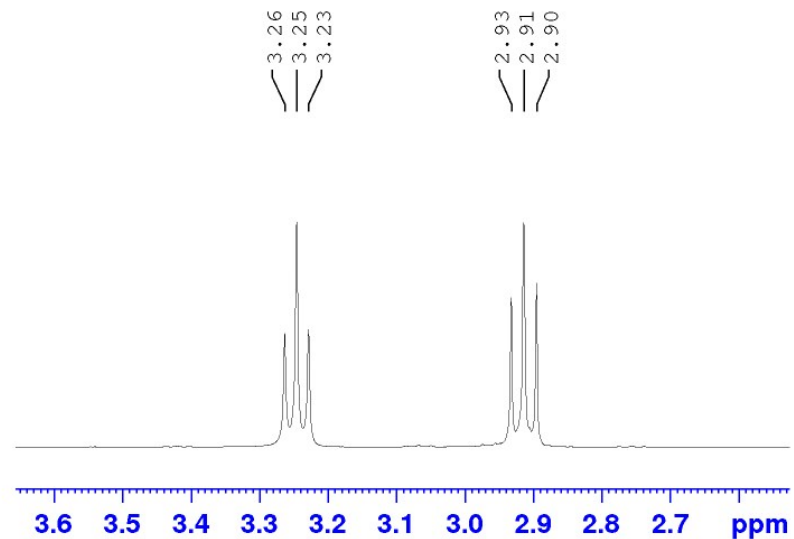
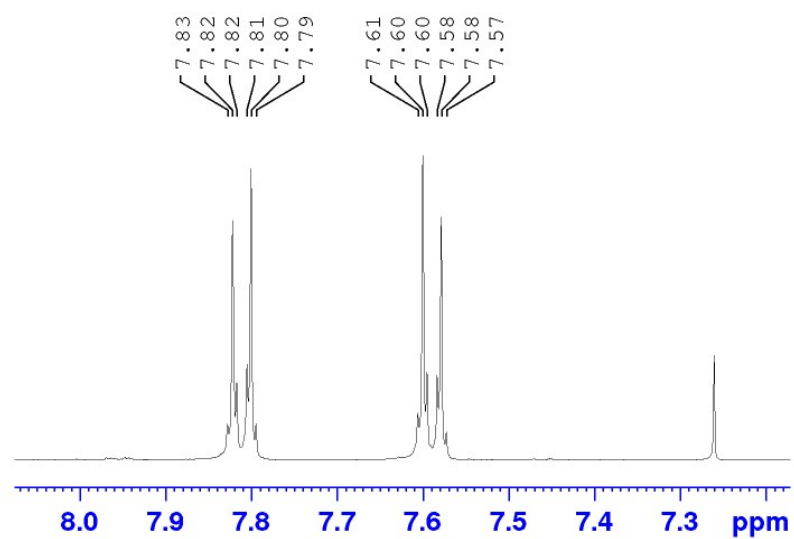
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SF 100.6177887 MHz
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LB 1.00 Hz
GB 0
PC 1.40

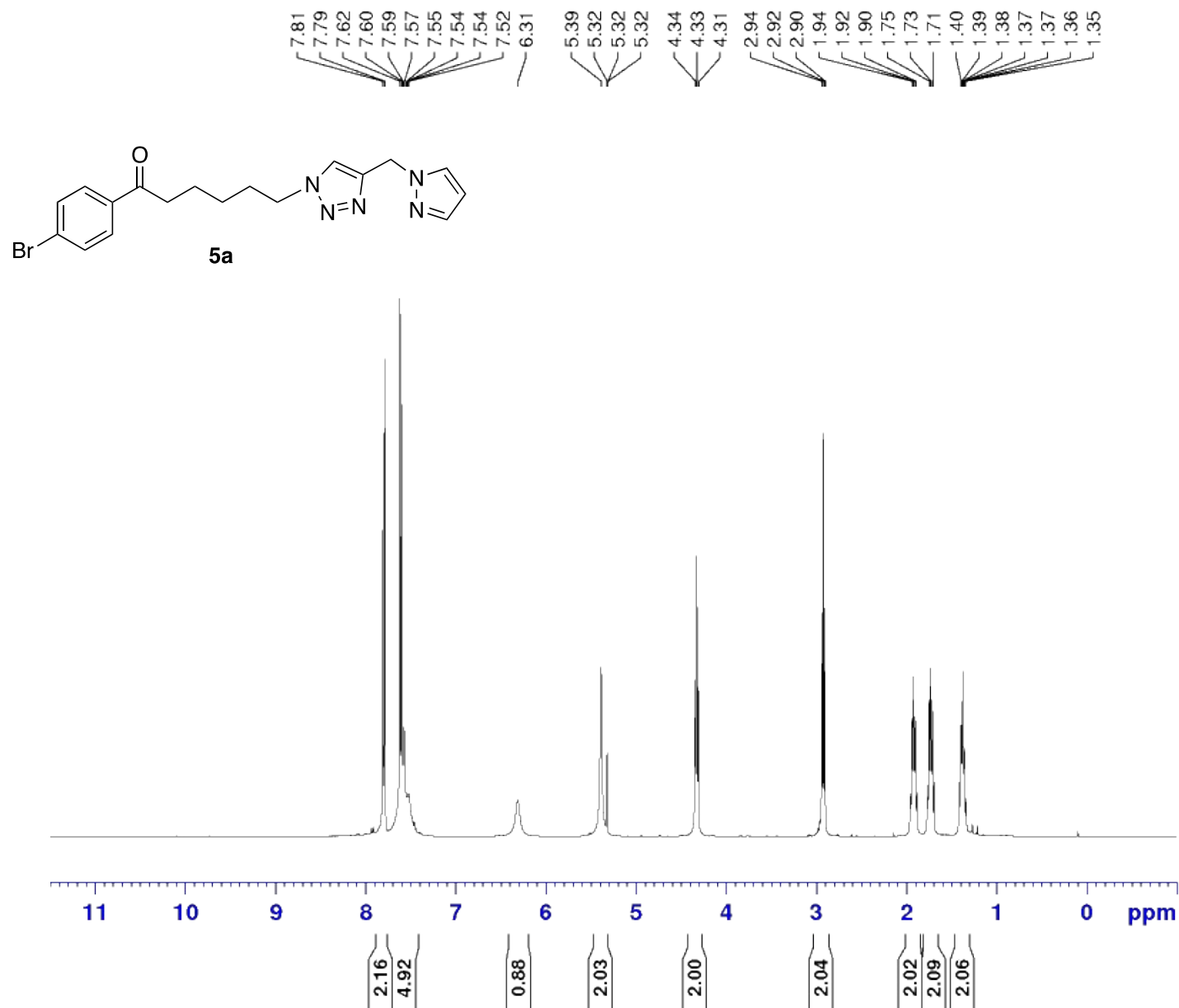


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Current Data Parameters

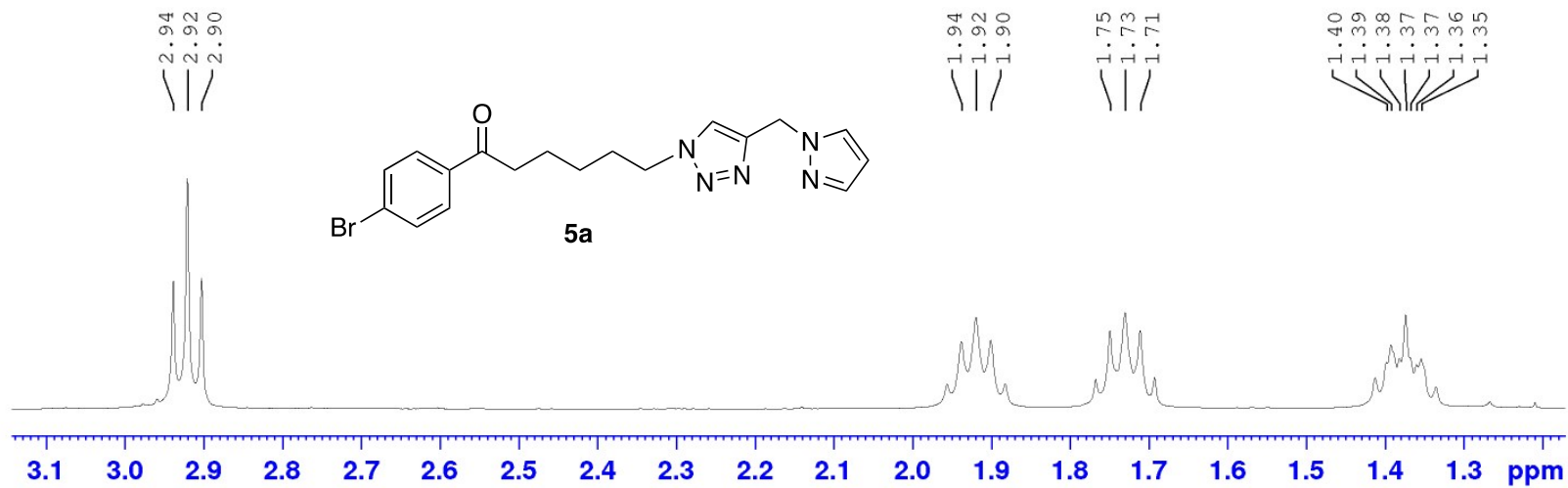
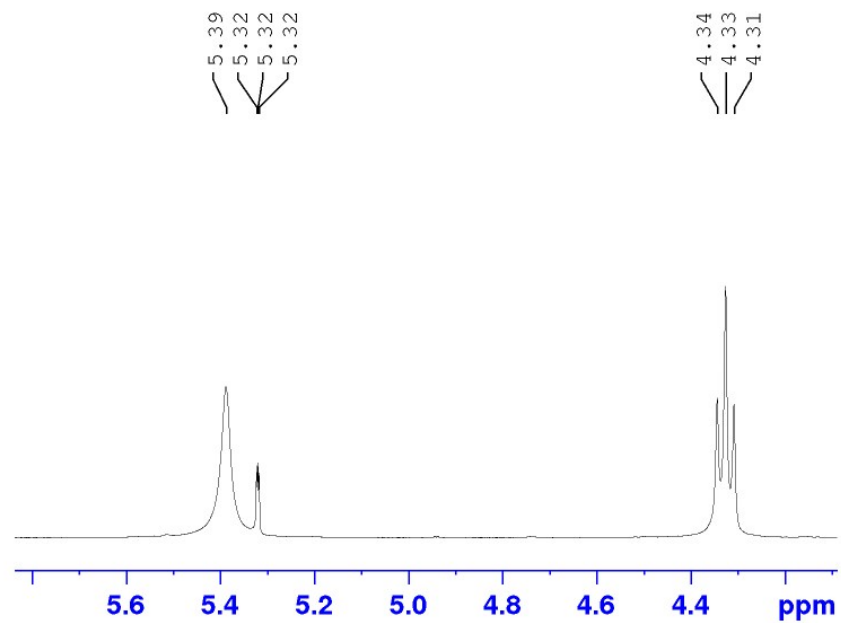
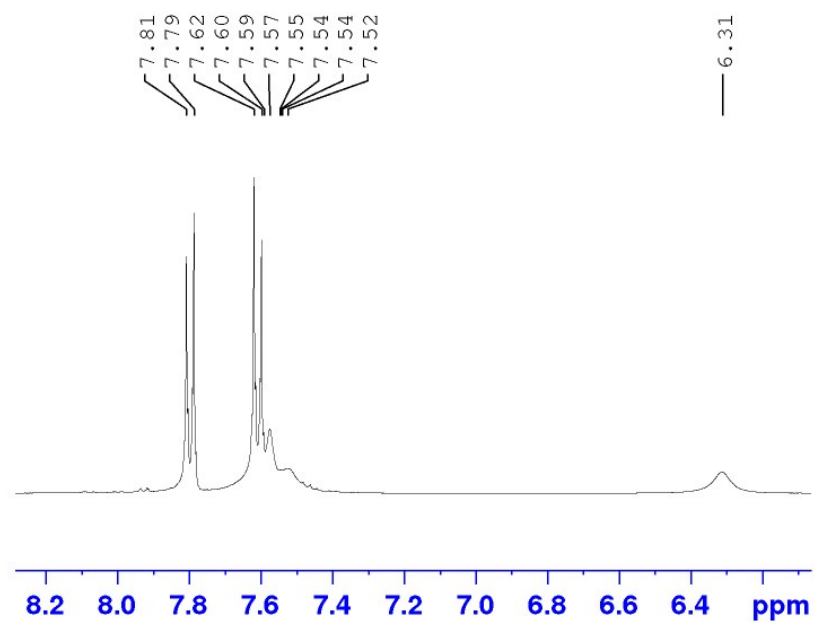


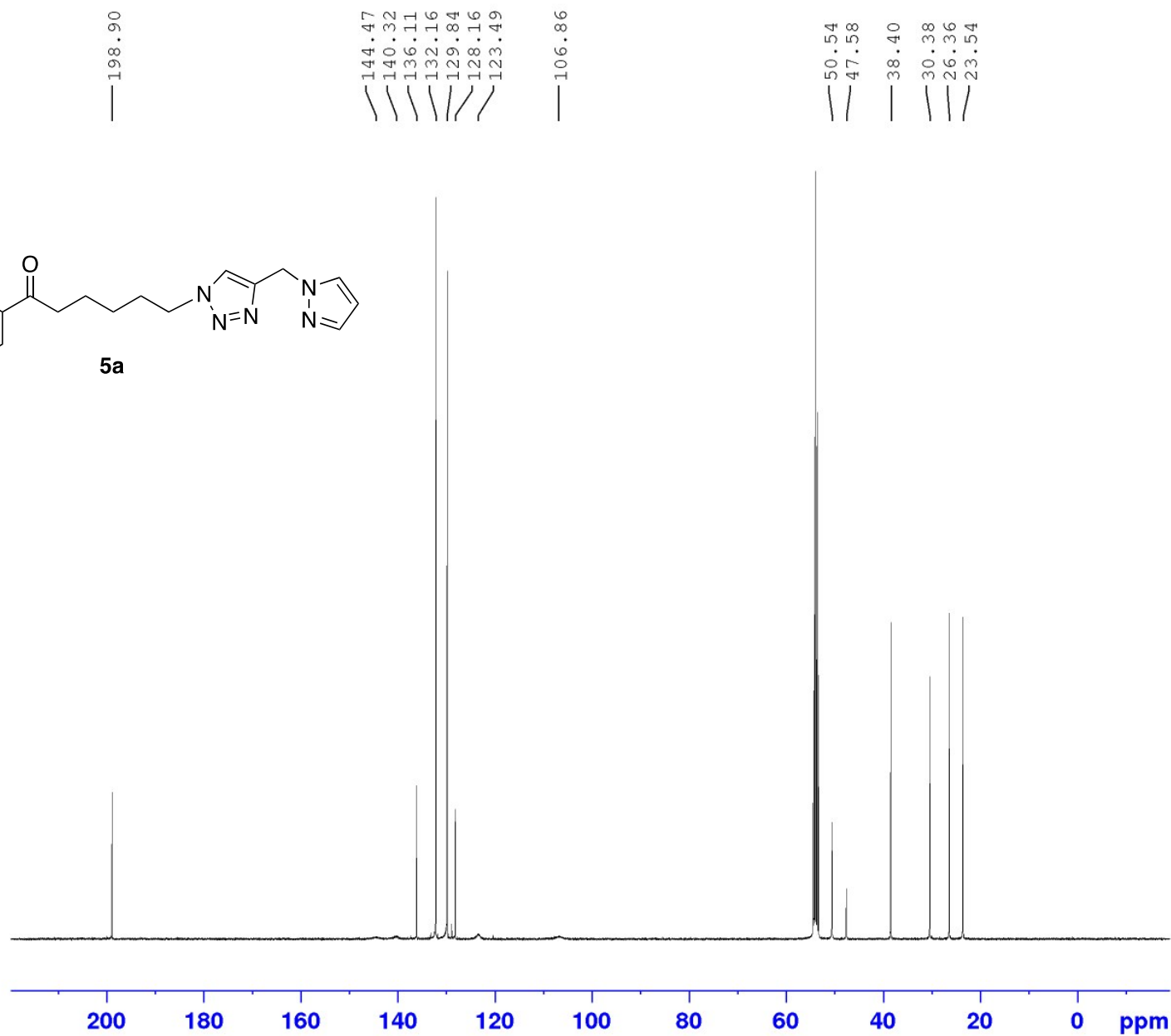
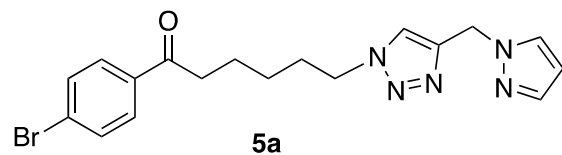


Current Data Parameters
 NAME MR-95_2
 EXPNO 1
 PROCNO 1

F2 - Acquisition Parameters
 Date_ 20190808
 Time 9.09 h
 INSTRUM spect
 PROBHD Z116096_0573 {
 PULPROG zg30
 TD 65536
 SOLVENT CD2Cl2
 NS 16
 DS 2
 SWH 8012.820 Hz
 FIDRES 0.244532 Hz
 AQ 4.0894465 sec
 RG 31.83
 DW 62.400 usec
 DE 12.00 usec
 TE 298.0 K
 D1 2.50000000 sec
 TD0 1
 SFO1 400.1524709 MHz
 NUC1 1H
 P1 10.00 usec
 PLW1 15.75399971 W

F2 - Processing parameters
 SI 65536
 SF 400.1500154 MHz
 WDW EM
 SSB 0
 LB 0.30 Hz
 GB 0
 PC 1.00



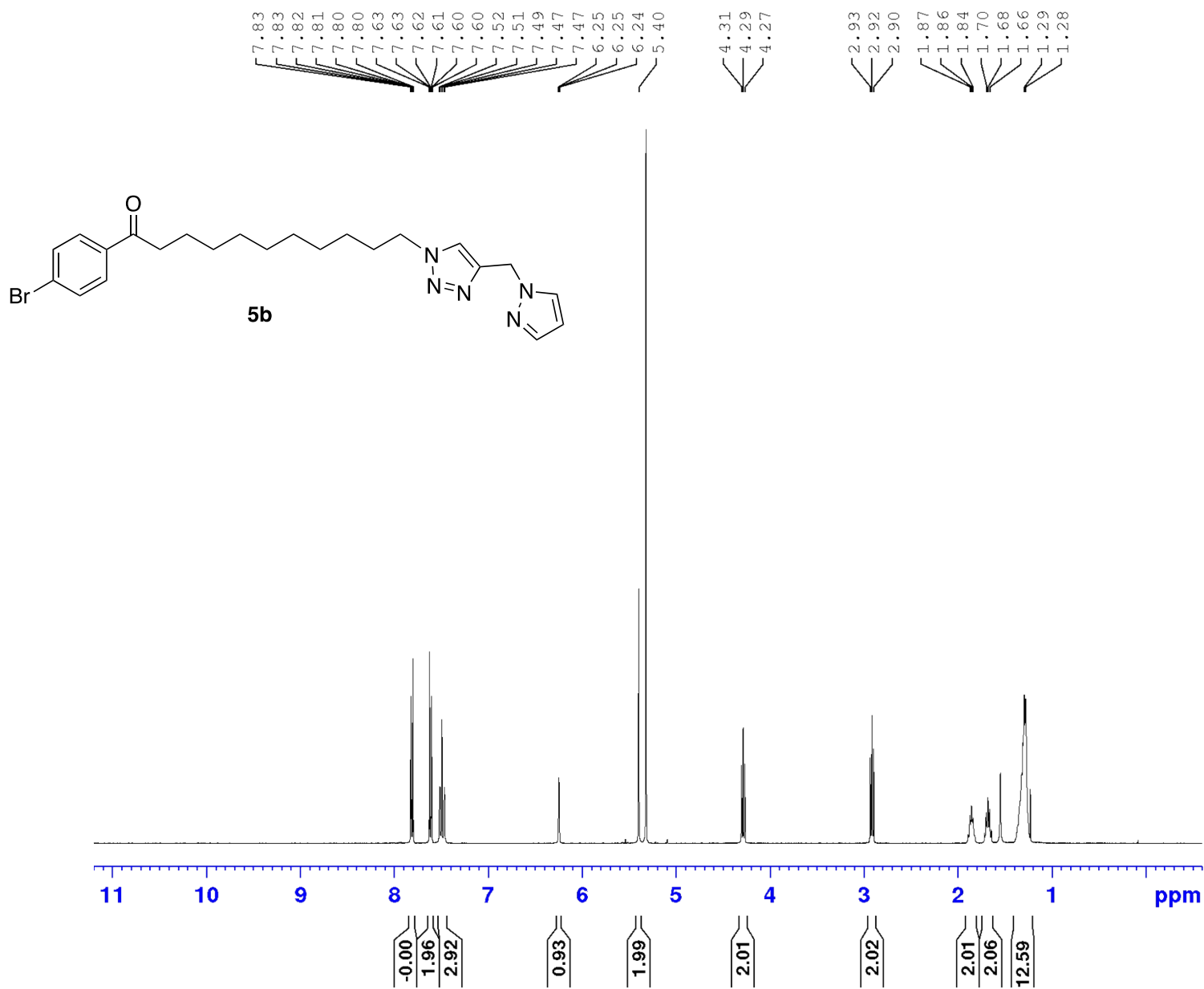


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Current Data Parameters
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EXPNO            3
PROCNO           1

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Time             2.04 h
INSTRUM          spect
PROBHD           Z116098_0573 (
PULPROG          zgpg30
TD               65536
SOLVENT          CD2Cl2
NS               4000
DS               4
SWH              24038.461 Hz
FIDRES           0.733596 Hz
AQ               1.3631488 sec
RG               198.41
DW               20.800 usec
DE               6.50 usec
TE               298.0 K
D1               2.00000000 sec
D11              0.03000000 sec
TD0              1
SFO1             100.6278593 MHz
NUC1             13C
P1               10.00 usec
PLW1             80.53199768 W
SFO2             400.1516006 MHz
NUC2             1H
CPDPRG[2]        waltz16
PCPD2            90.00 usec
PLW2             15.75399971 W
PLW12            0.19449000 W
PLW13            0.09782900 W

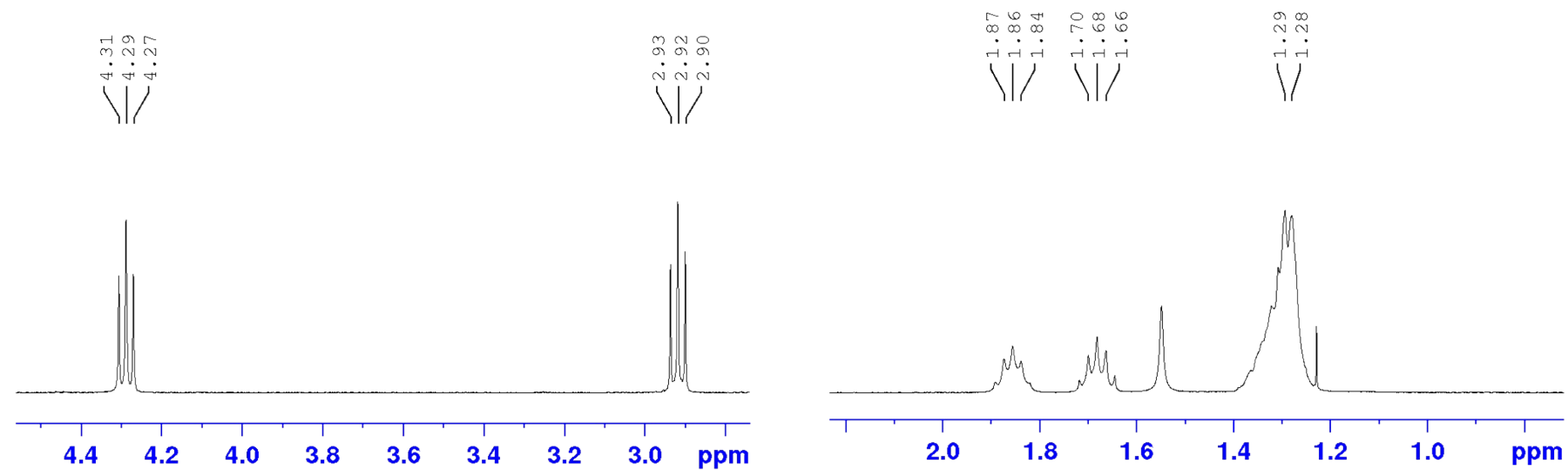
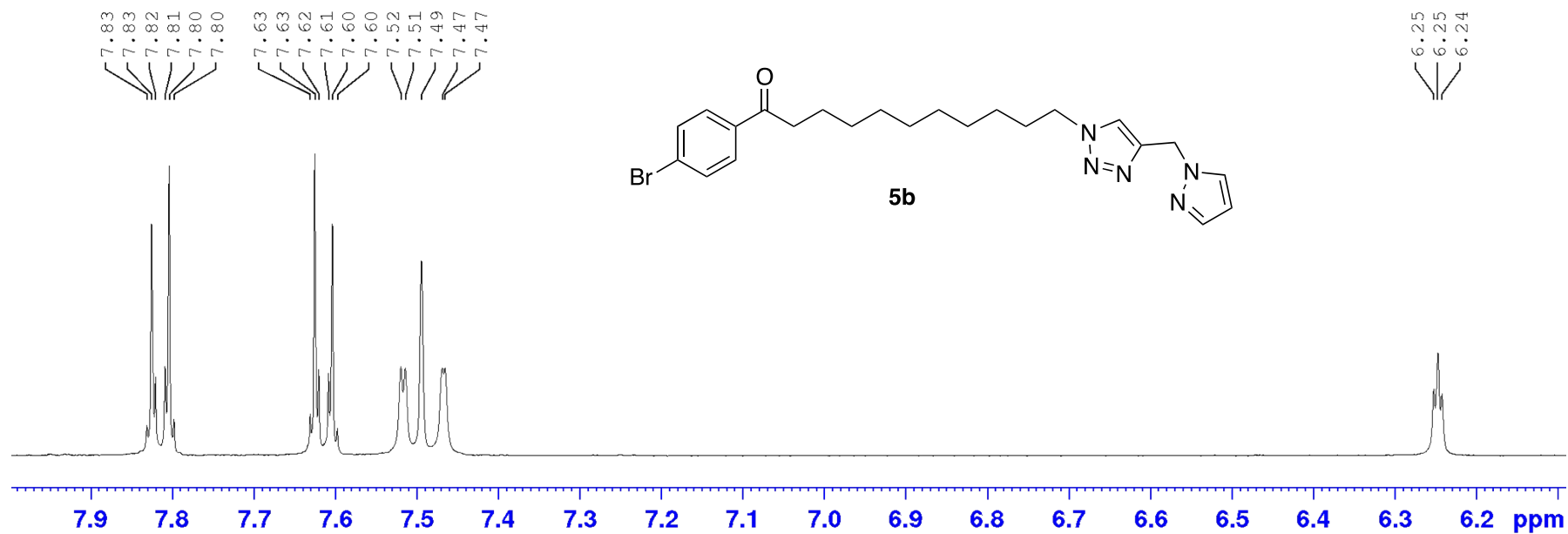
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WDW              EM
SSB              0
LB               1.00 Hz
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PC               1.40
  
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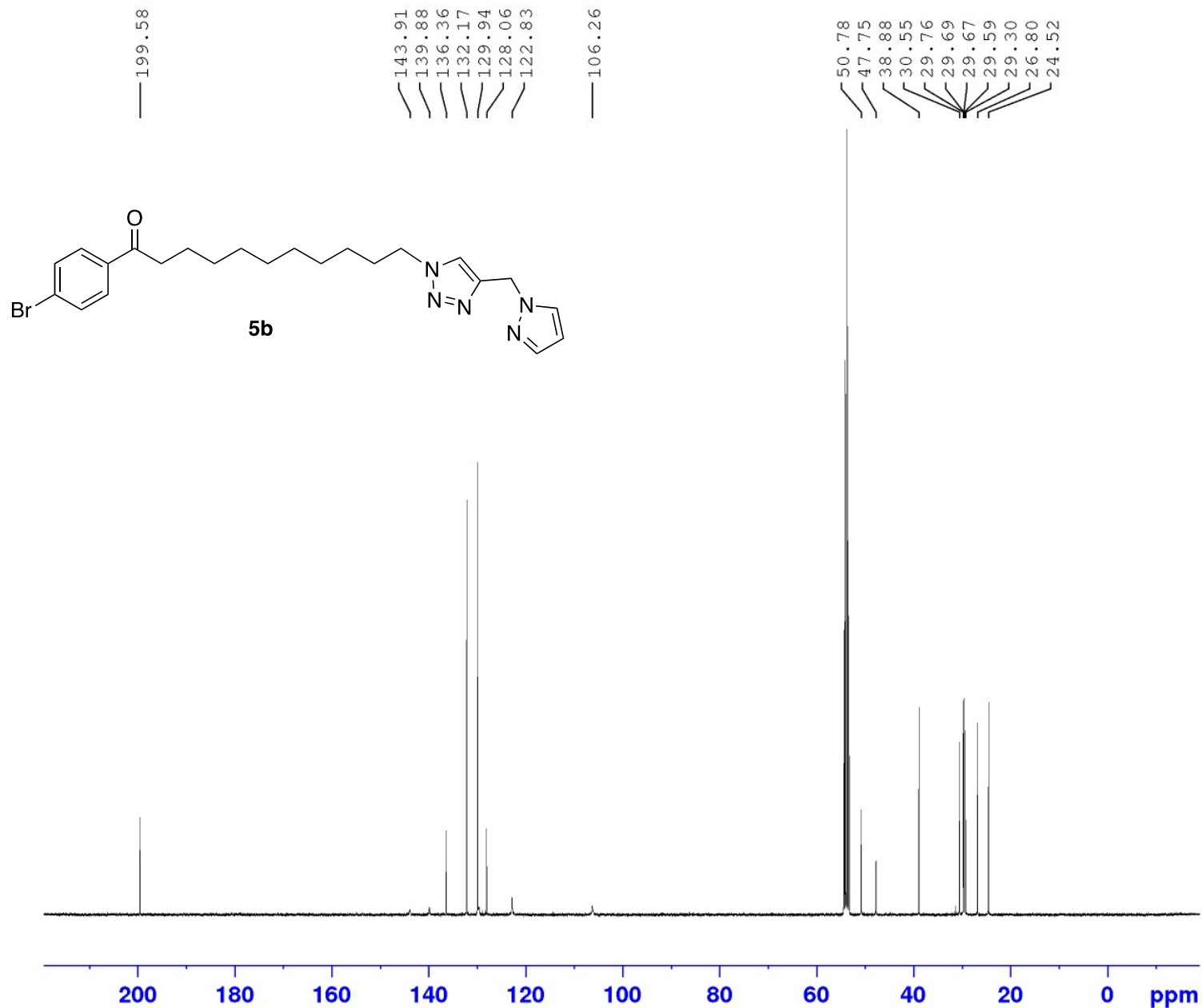


Current Data Parameters
 NAME MR-80
 EXPNO 1
 PROCNO 1

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 Time 9.06 h
 INSTRUM spect
 PROBHD Z116098_0573 (
 PULPROG zg30
 TD 65536
 SOLVENT CD2Cl2
 NS 16
 DS 2
 SWH 8012.820 Hz
 FIDRES 0.244532 Hz
 AQ 4.0894465 sec
 RG 198.41
 DW 62.400 usec
 DE 12.00 usec
 TE 298.0 K
 D1 2.50000000 sec
 TD0 1
 SFO1 400.1524709 MHz
 NUC1 1H
 P1 10.00 usec
 PLW1 15.75399971 W

F2 - Processing parameters
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 SF 400.1500155 MHz
 WDW EM
 SSB 0
 LB 0.30 Hz
 GB 0
 PC 1.00

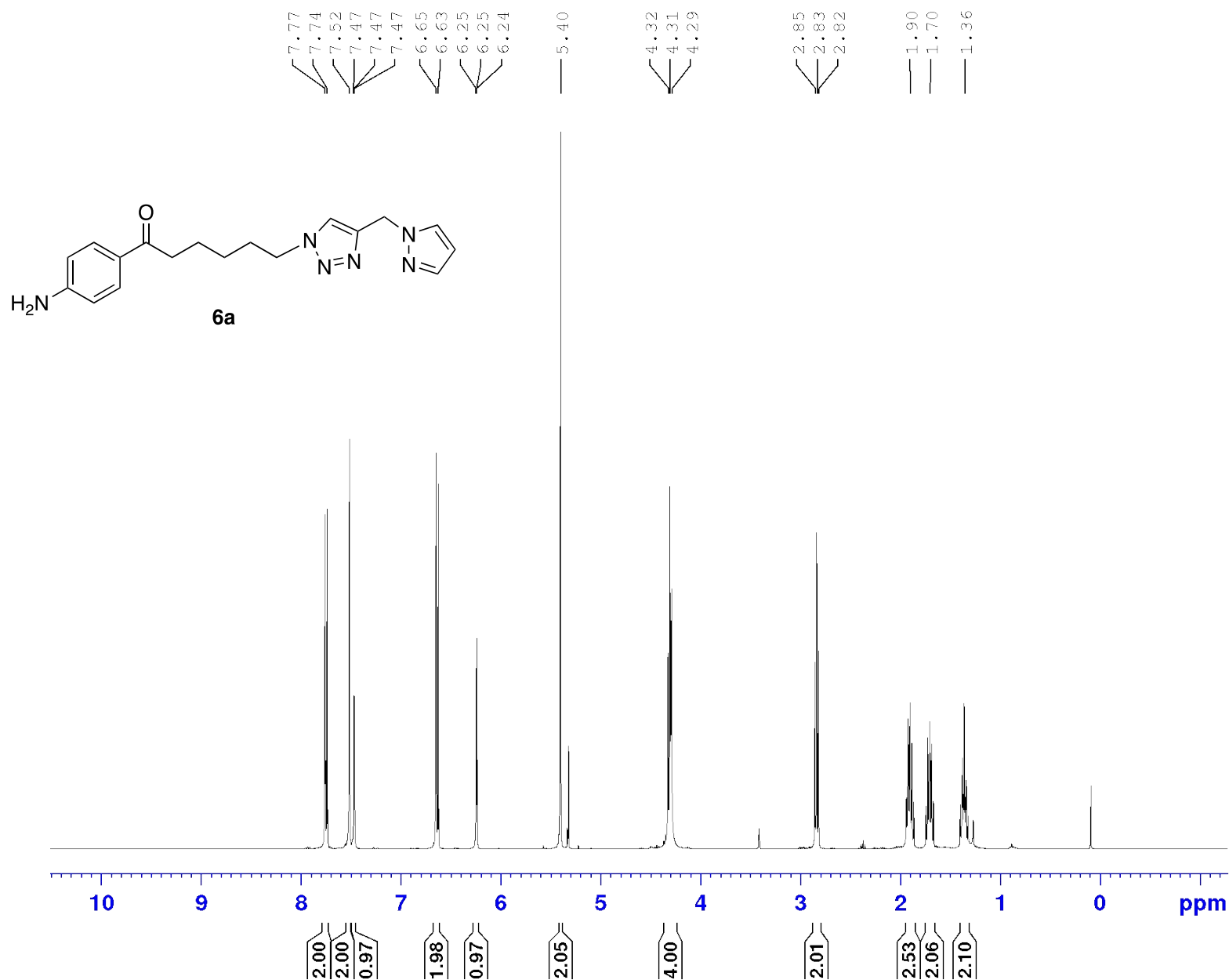




Current Data Parameters
 NAME MR-80
 EXPNO 3
 PROCNO 1

F2 - Acquisition Parameters
 Date_ 20190315
 Time 22.59 h
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 PROBHD Z116098_0573 (
 PULPROG zgpg30
 TD 65536
 SOLVENT CD2Cl2
 NS 3000
 DS 4
 SWH 24038.461 Hz
 FIDRES 0.733596 Hz
 AQ 1.3631488 sec
 RG 198.41
 DW 20.800 usec
 DE 6.50 usec
 TE 298.0 K
 D1 2.00000000 sec
 D11 0.03000000 sec
 TD0 1
 SFO1 100.6278593 MHz
 NUC1 13C
 P1 10.00 usec
 PLW1 80.53199768 W
 SFO2 400.1516006 MHz
 NUC2 1H
 CPDPRG[2] waltz16
 PCPD2 90.00 usec
 PLW2 15.75399971 W
 PLW12 0.19449000 W
 PLW13 0.09782900 W

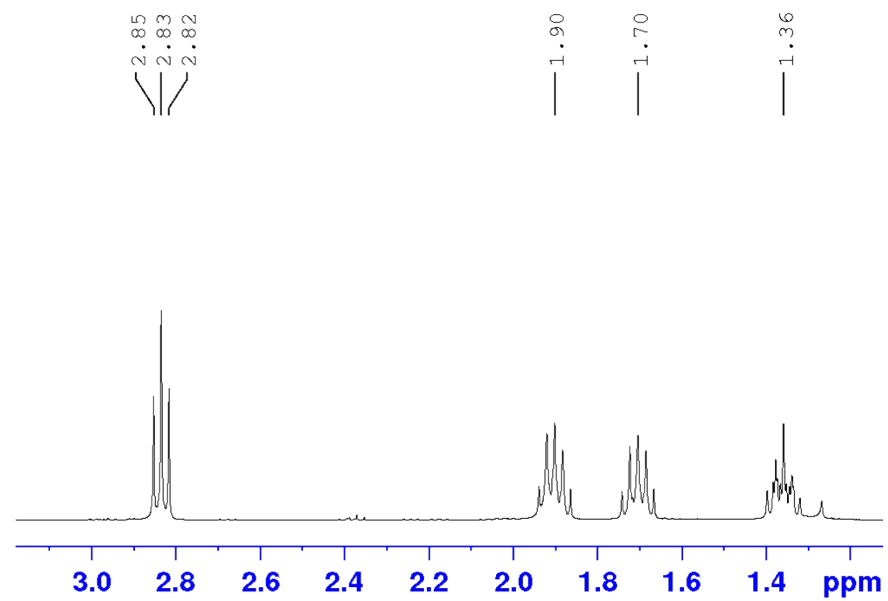
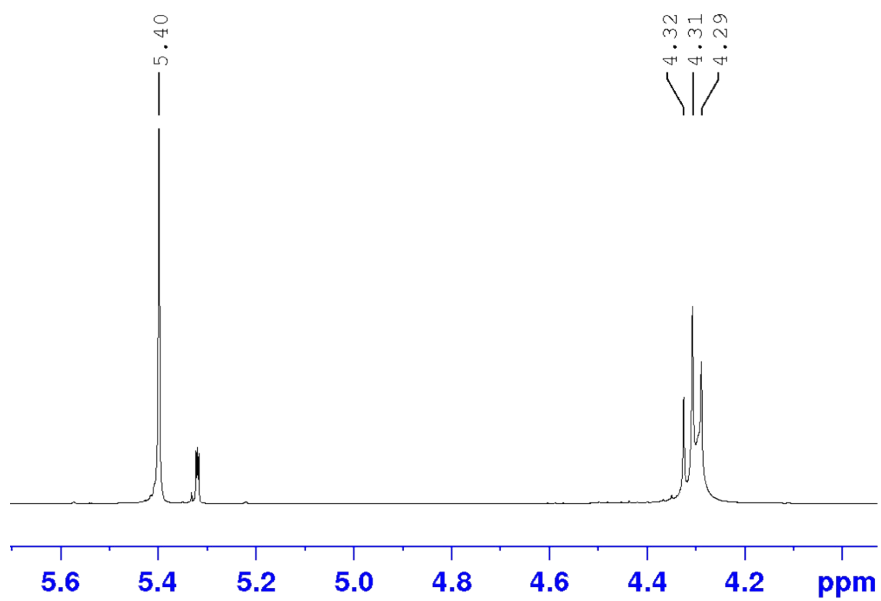
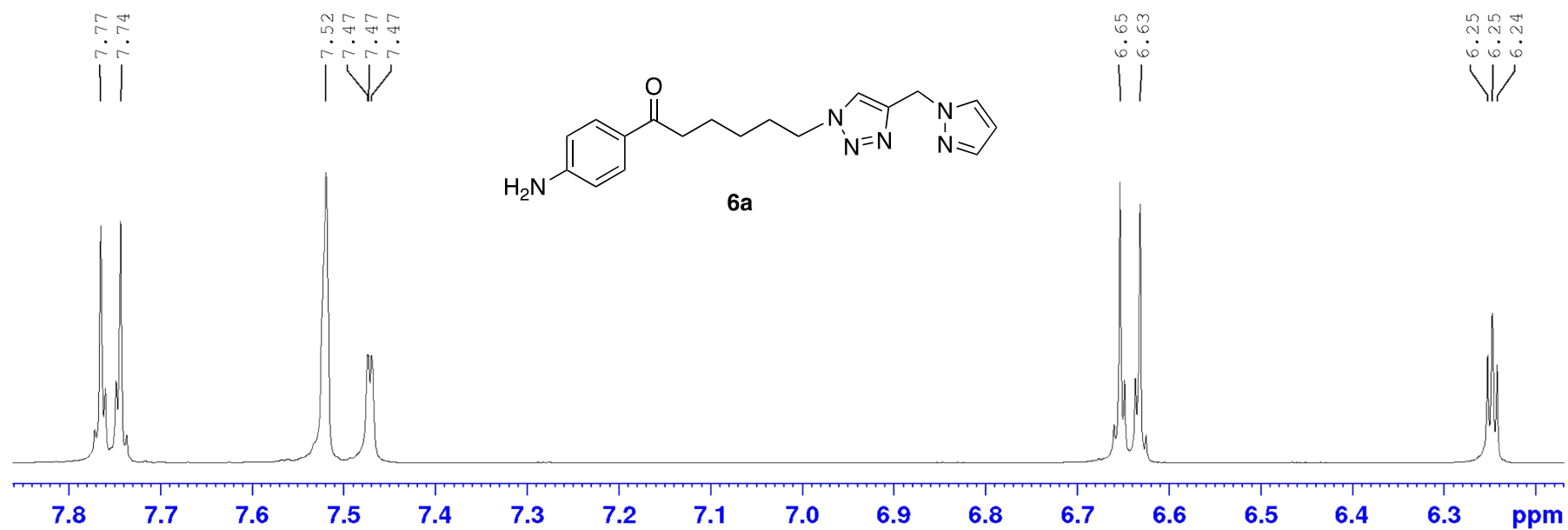
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 GB 0
 PC 1.40

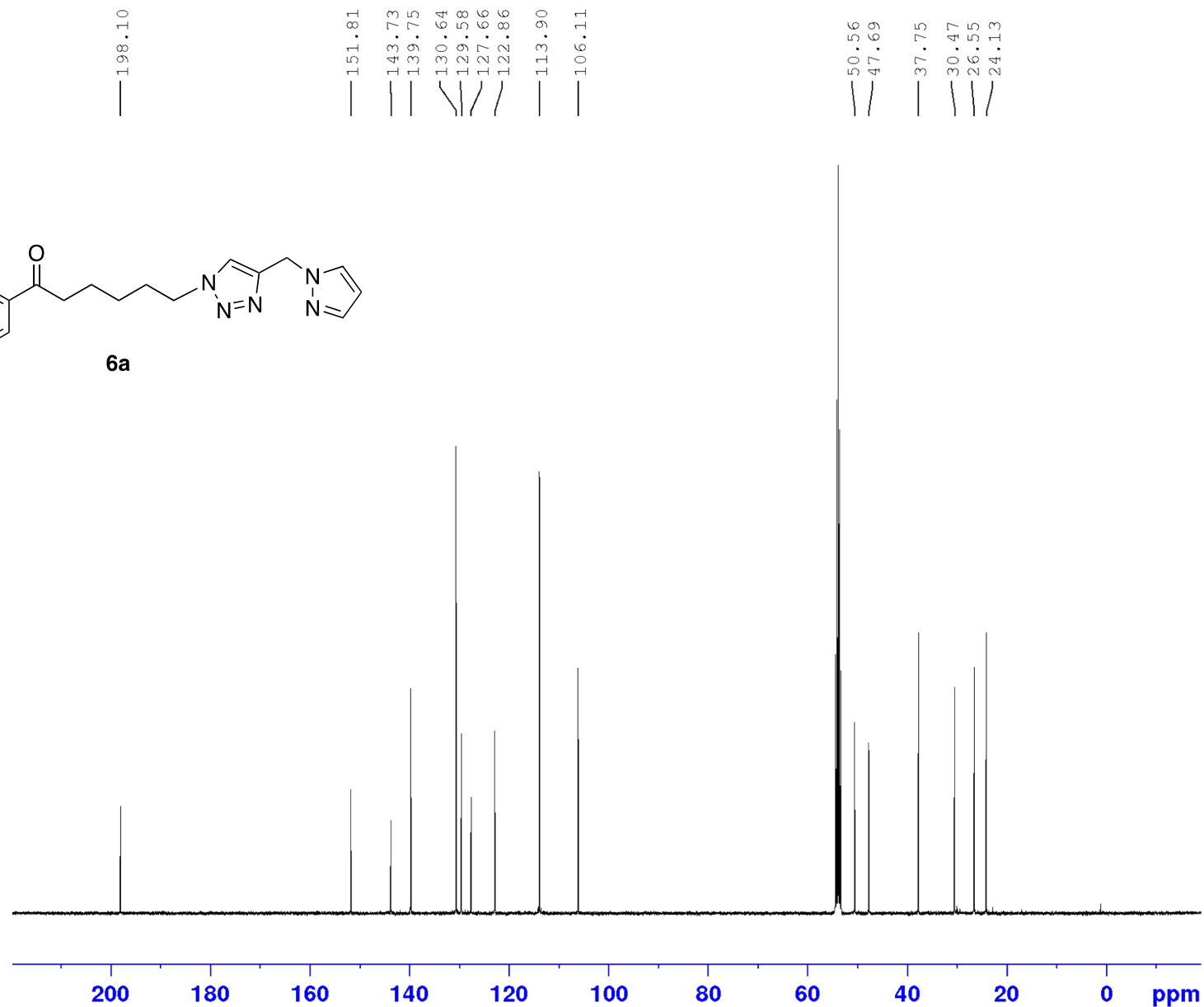
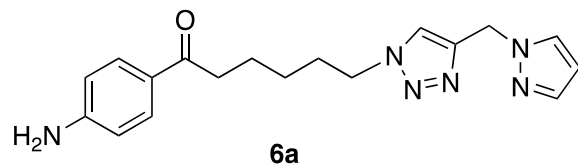


Current Data Parameters
NAME MR-132
EXPNO 2
PROCNO 1

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Time 11.46 h
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PULPROG zg30
TD 65536
SOLVENT CD2Cl2
NS 16
DS 2
SWH 8012.820 Hz
FIDRES 0.244532 Hz
AQ 4.0894465 sec
RG 63.59
DW 62.400 usec
DE 12.00 usec
TE 298.0 K
D1 2.50000000 sec
TD0 1
SF01 400.1524709 MHz
NUC1 1H
P1 10.00 usec
PLW1 15.75399971 W

F2 - Processing parameters
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SF 400.1500154 MHz
WDW EM
SSB 0
LB 0.30 Hz
GB 0
PC 1.00

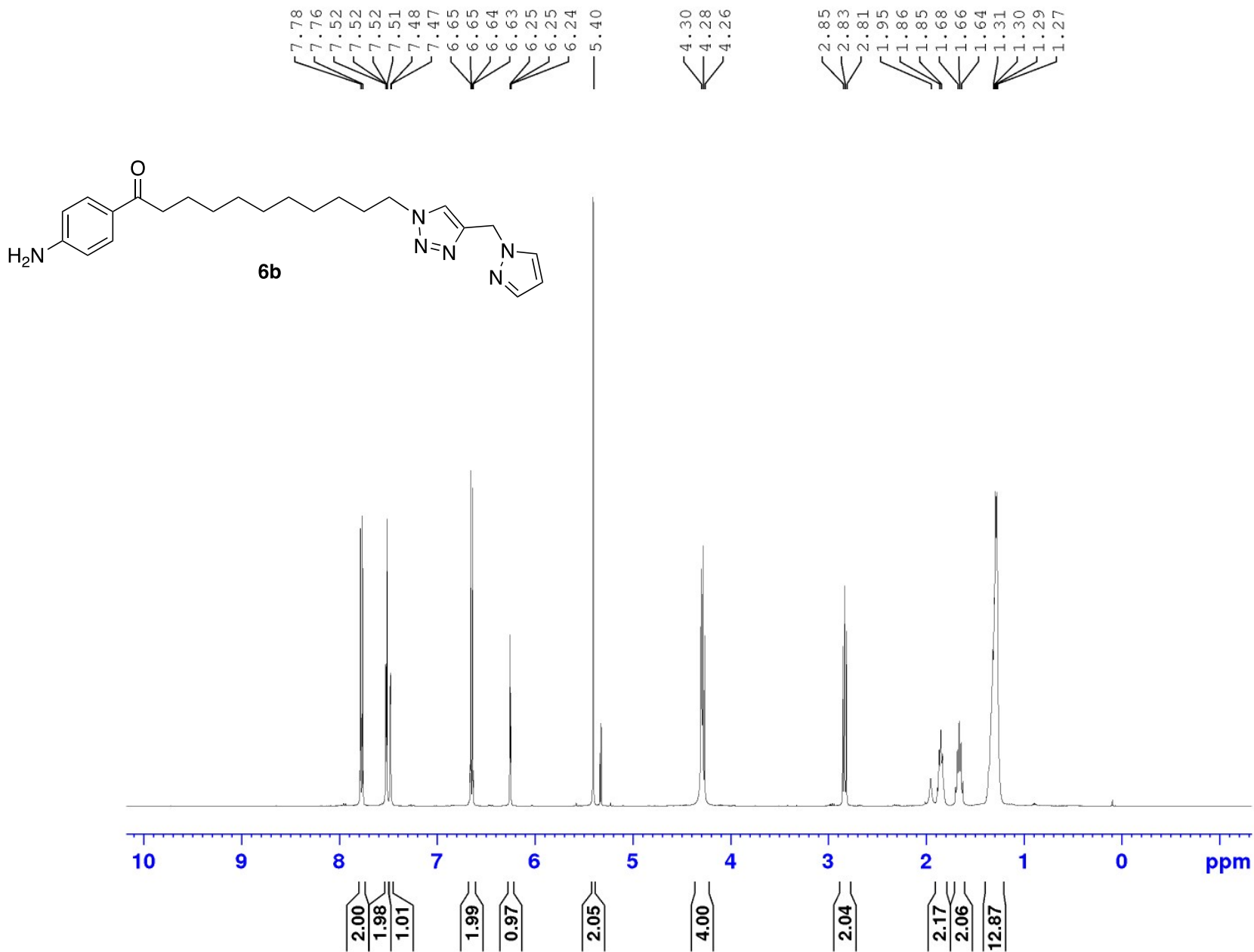




Current Data Parameters
NAME MR-132
EXPNO 3
PROCNO 1

F2 - Acquisition Parameters
Date_ 20190409
Time 23.00 h
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PROBHD Z116098_0573 (
PULPROG zgpg30
TD 65536
SOLVENT CD2Cl2
NS 3000
DS 4
SWH 24038.461 Hz
FIDRES 0.733596 Hz
AQ 1.3631488 sec
RG 198.41
DW 20.800 usec
DE 6.50 usec
TE 298.0 K
D1 2.00000000 sec
D11 0.03000000 sec
TD0 1
SFO1 100.6278593 MHz
NUC1 13C
P1 10.00 usec
PLW1 80.53199768 W
SFO2 400.1516006 MHz
NUC2 1H
CPDPRG[2] waltz16
PCPD2 90.00 usec
PLW2 15.75399971 W
PLW12 0.19449000 W
PLW13 0.09782900 W

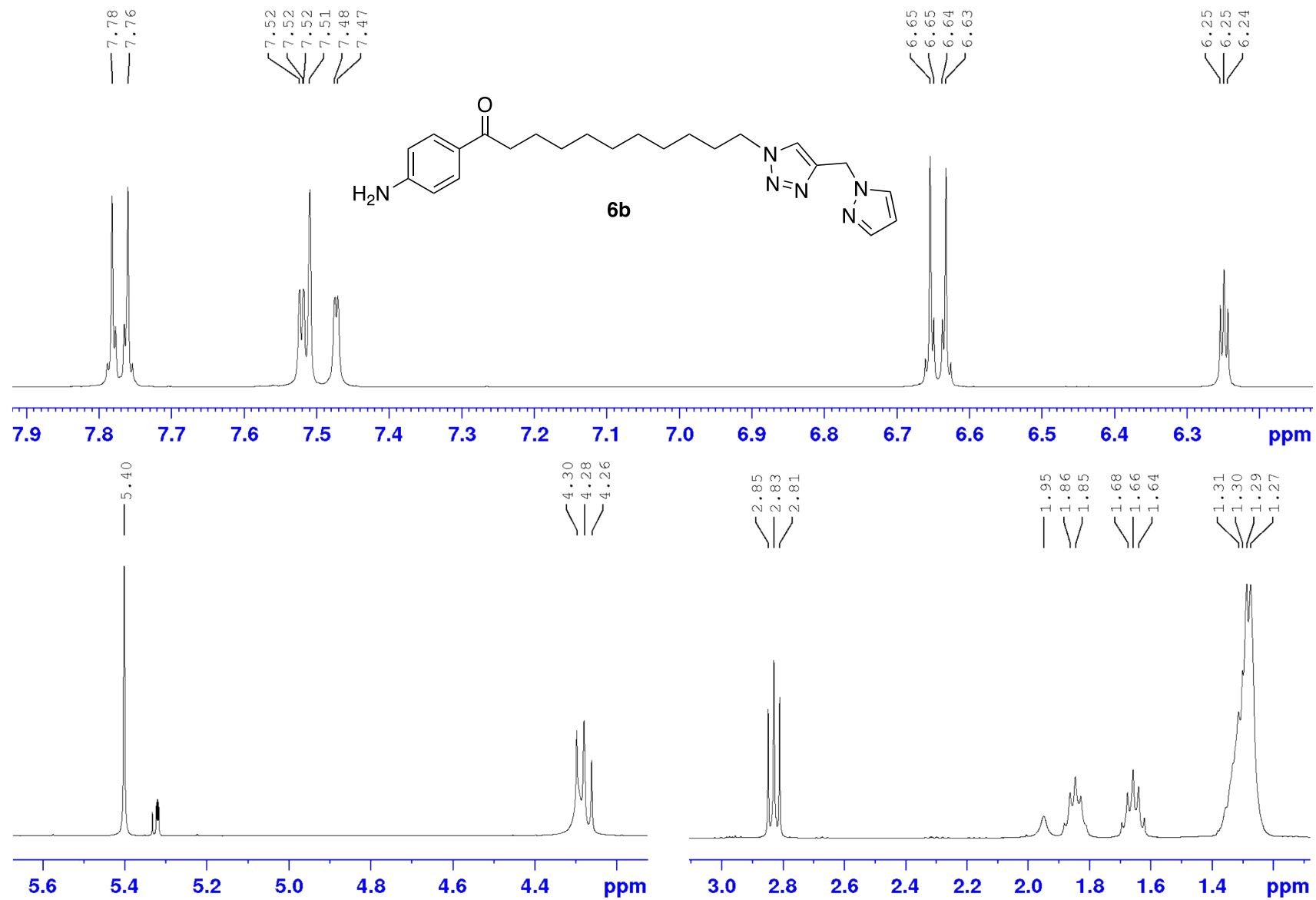
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SSB 0
LB 1.00 Hz
GB 0
PC 1.40

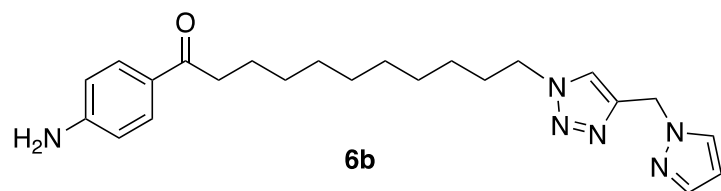


Current Data Parameters
NAME MR-129
EXPNO 2
PROCNO 1

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Time 18.20 h
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PROBHD Z116098_0573 (
PULPROG zg30
TD 65536
SOLVENT CD2Cl2
NS 16
DS 2
SWH 8012.820 Hz
FIDRES 0.244532 Hz
AQ 4.0894465 sec
RG 31.83
DW 62.400 usec
DE 12.00 usec
TE 298.0 K
D1 2.50000000 sec
TD0 1
SFO1 400.1524709 MHz
NUC1 1H
P1 10.00 usec
PLW1 15.75399971 W

F2 - Processing parameters
SI 65536
SF 400.1500154 MHz
WDW EM
SSB 0
LB 0.30 Hz
GB 0
PC 1.00





98.79

51.72

43.73

39.77

30.68

29.58

27.83

22.77

13.90

06.12

0.76

7.70

8.23

0.54

9.76

9.73

9.68

9.65

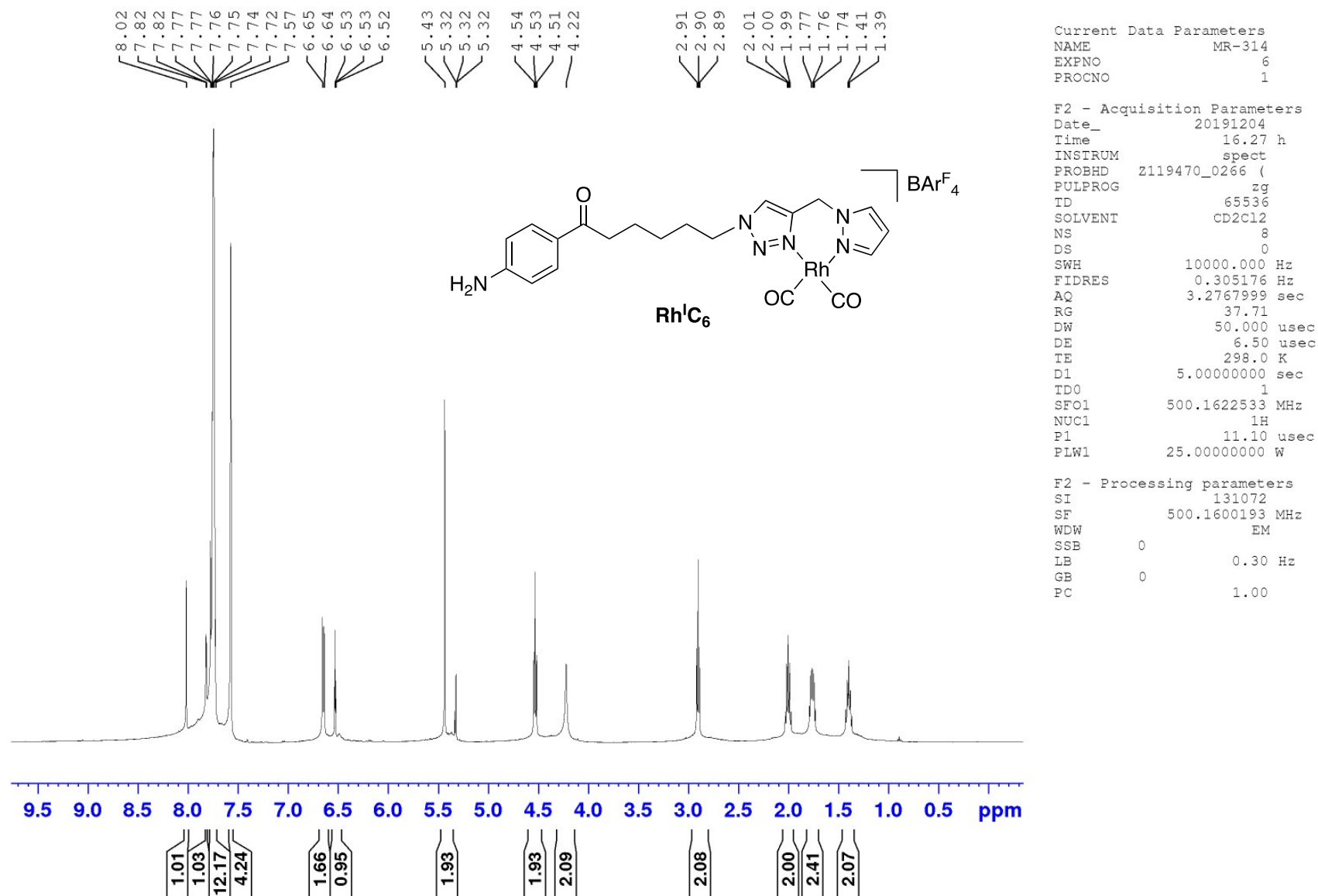
9.27

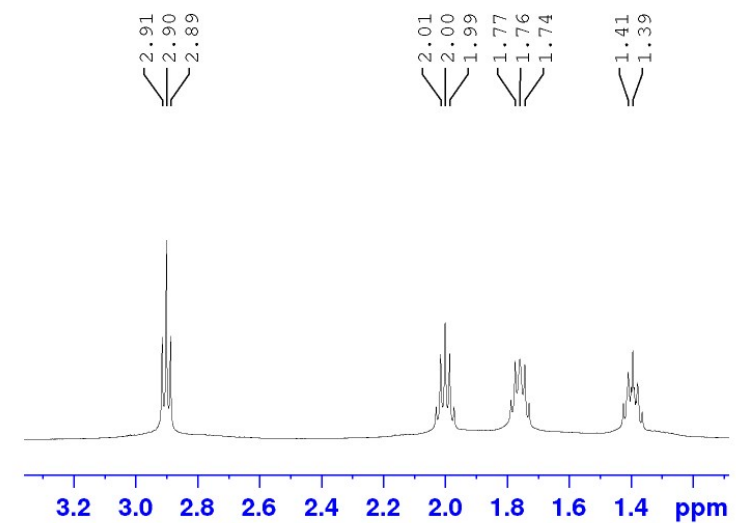
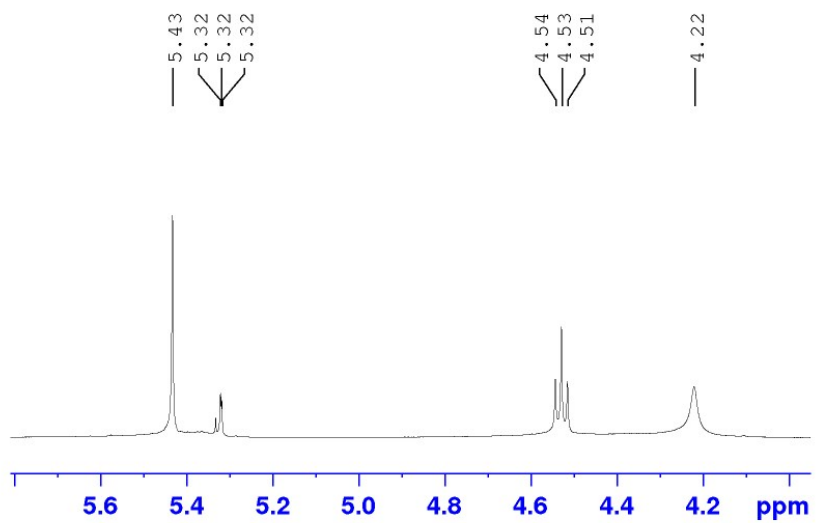
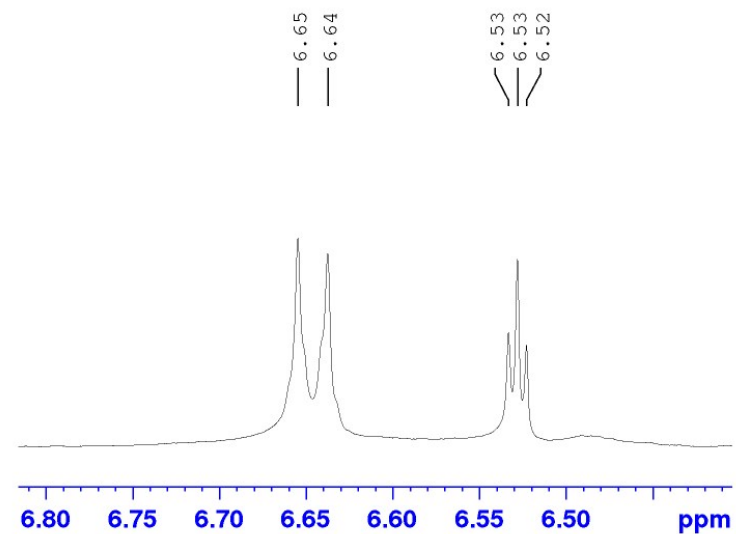
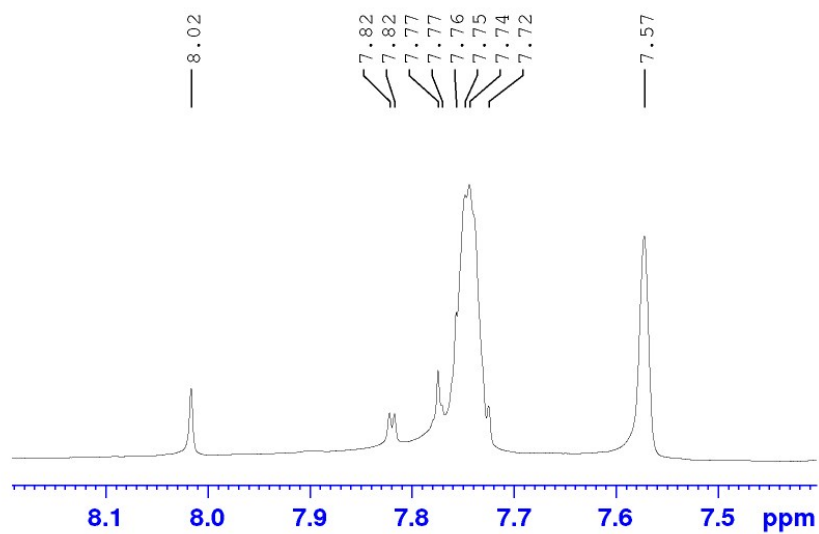
6.78

6.12

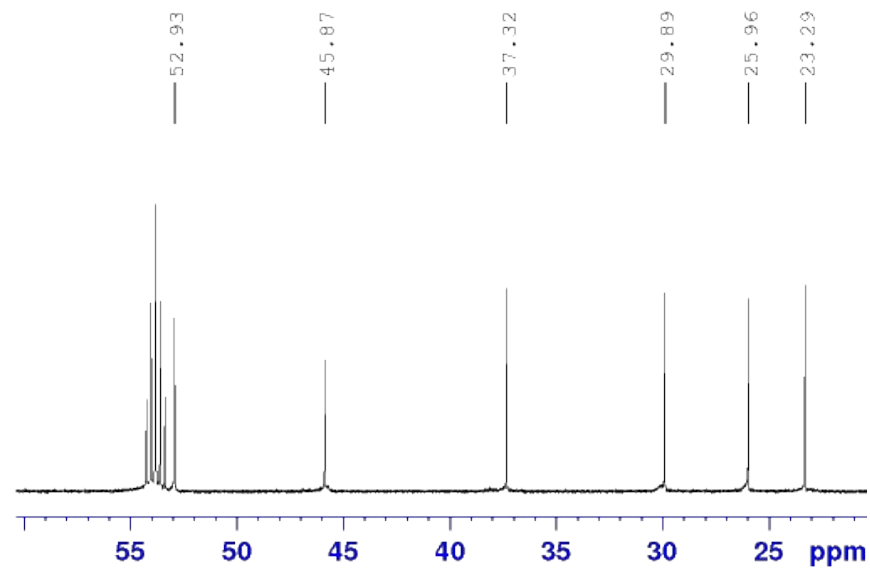
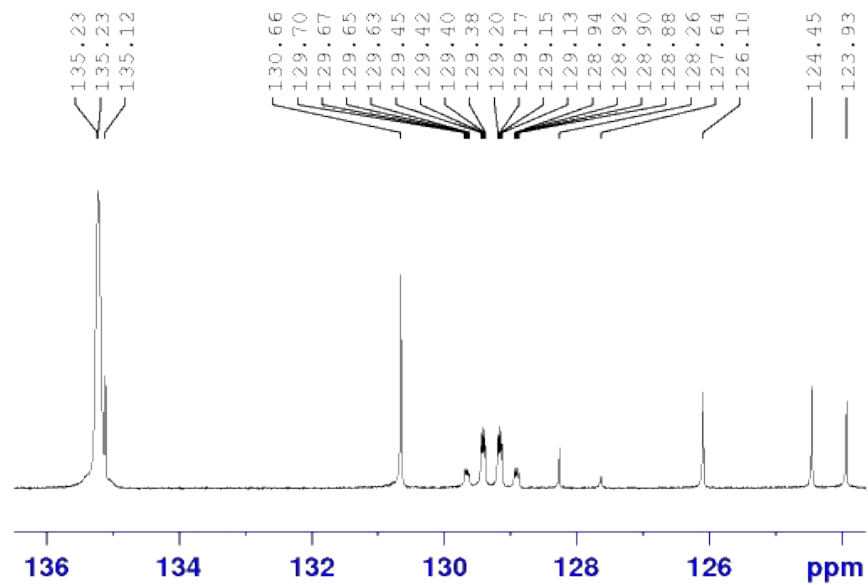
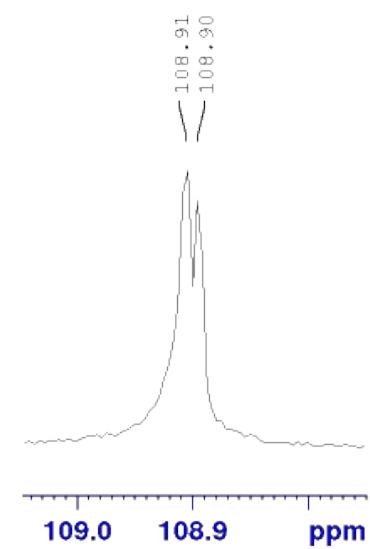
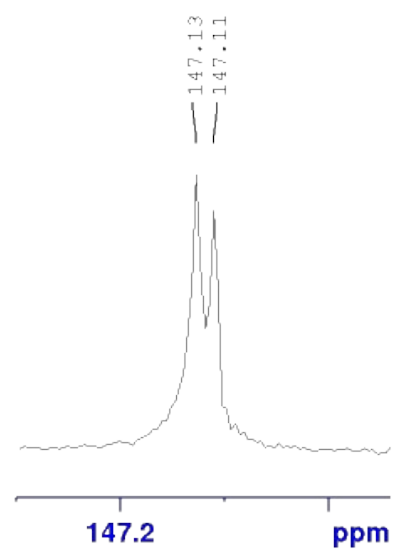
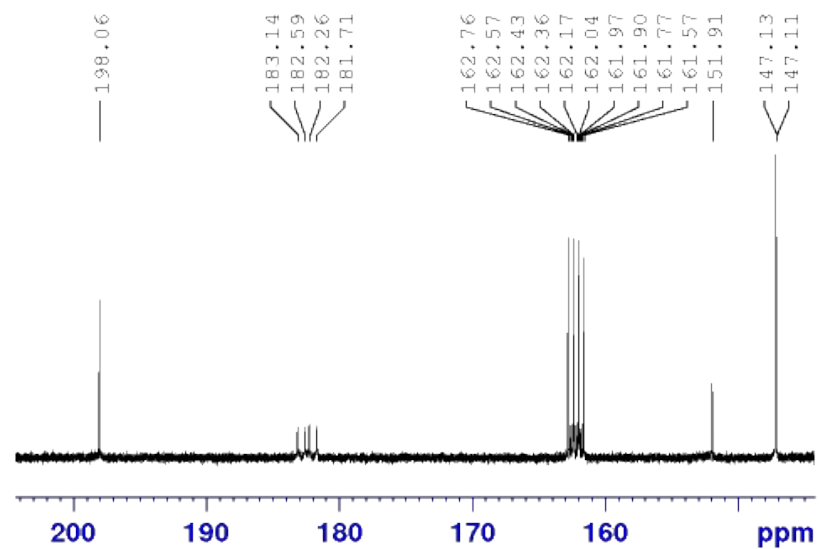
Current Data Parameters
NAME MR-129
EXPNO 3

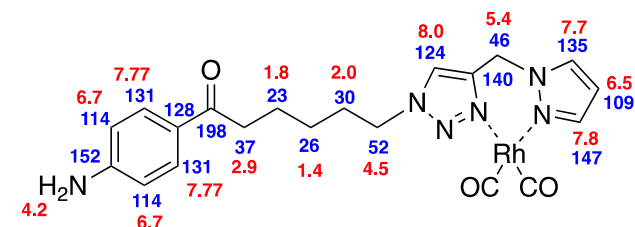
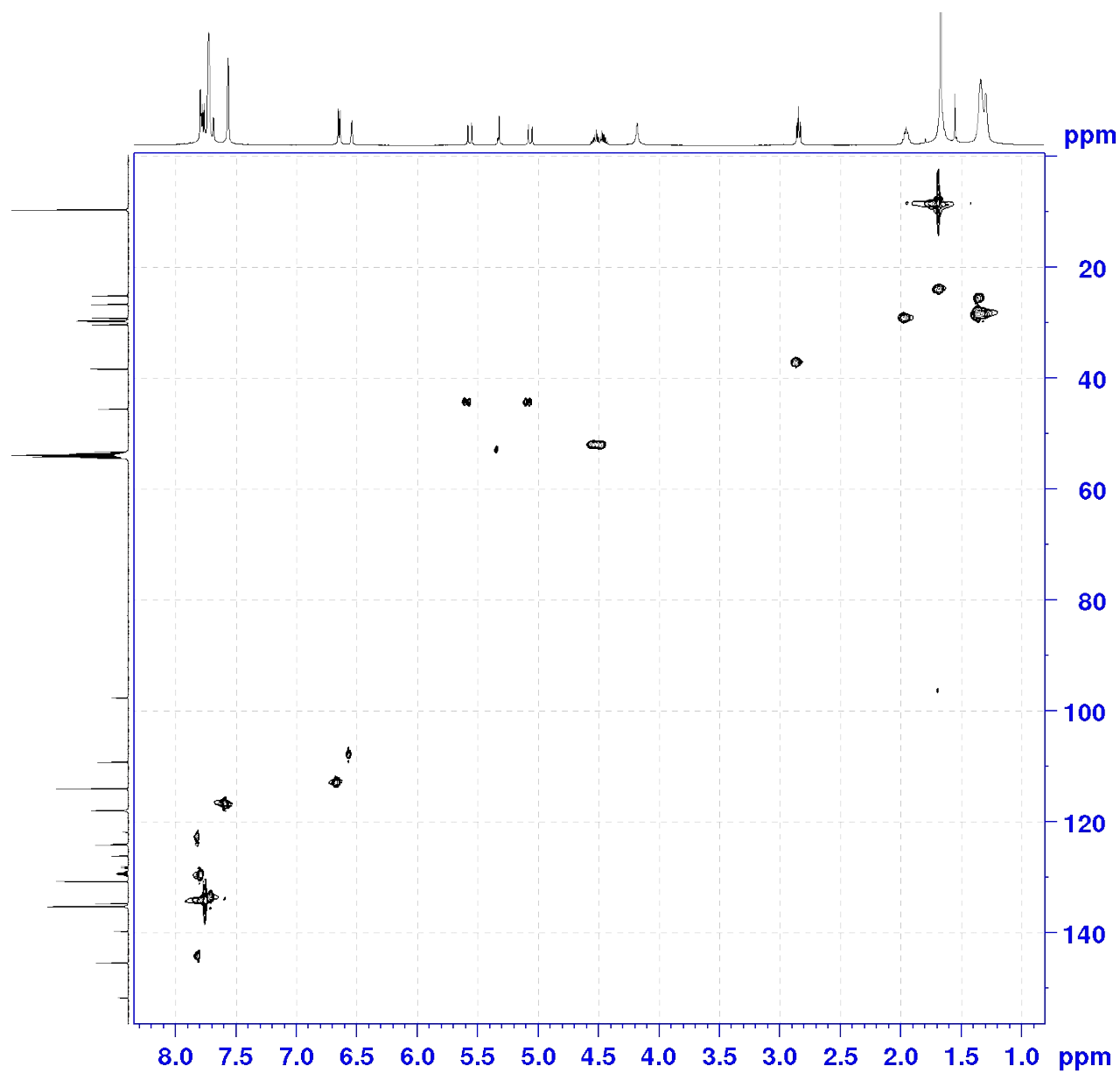
NMR spectra of metal complexes



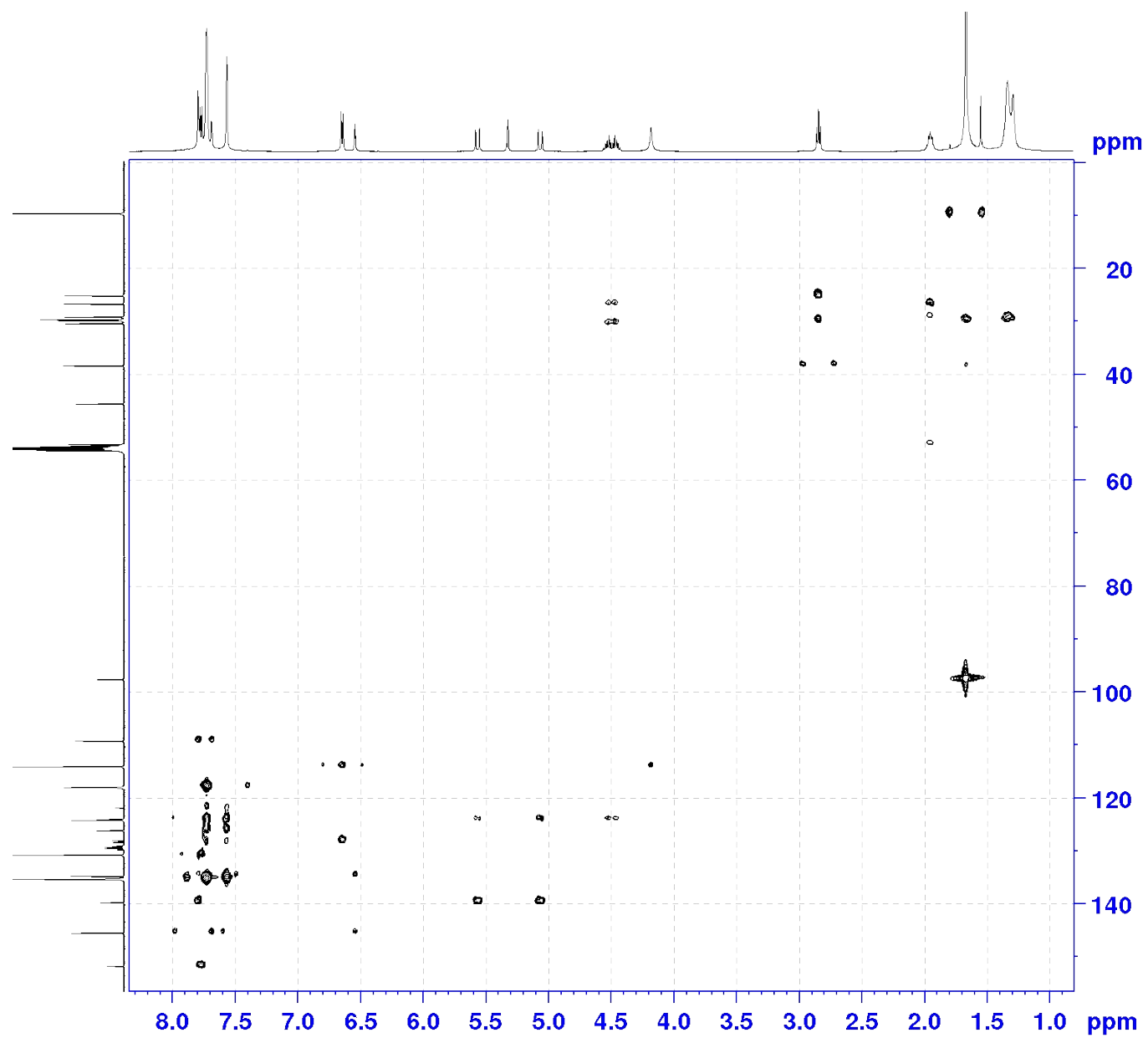




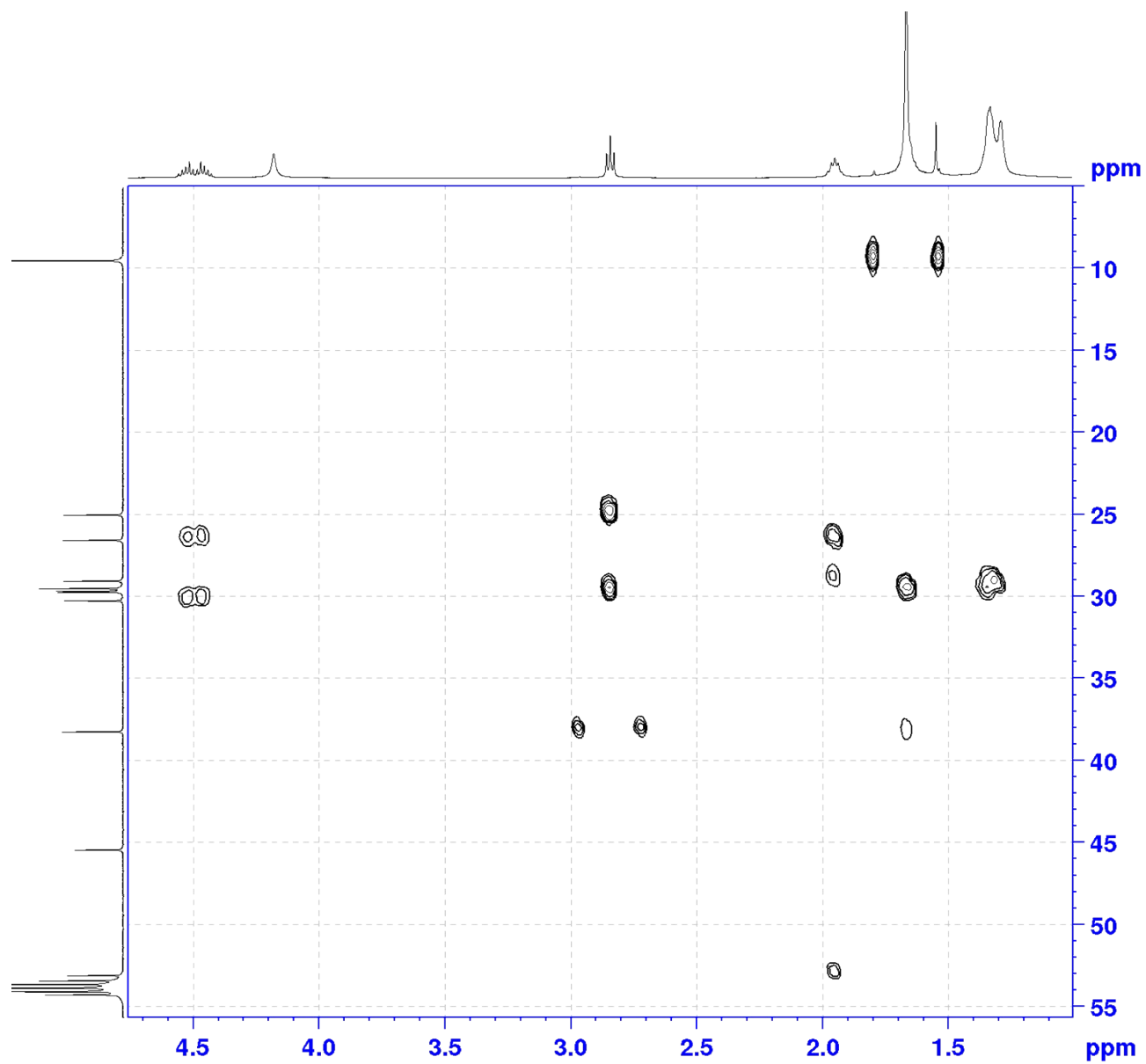




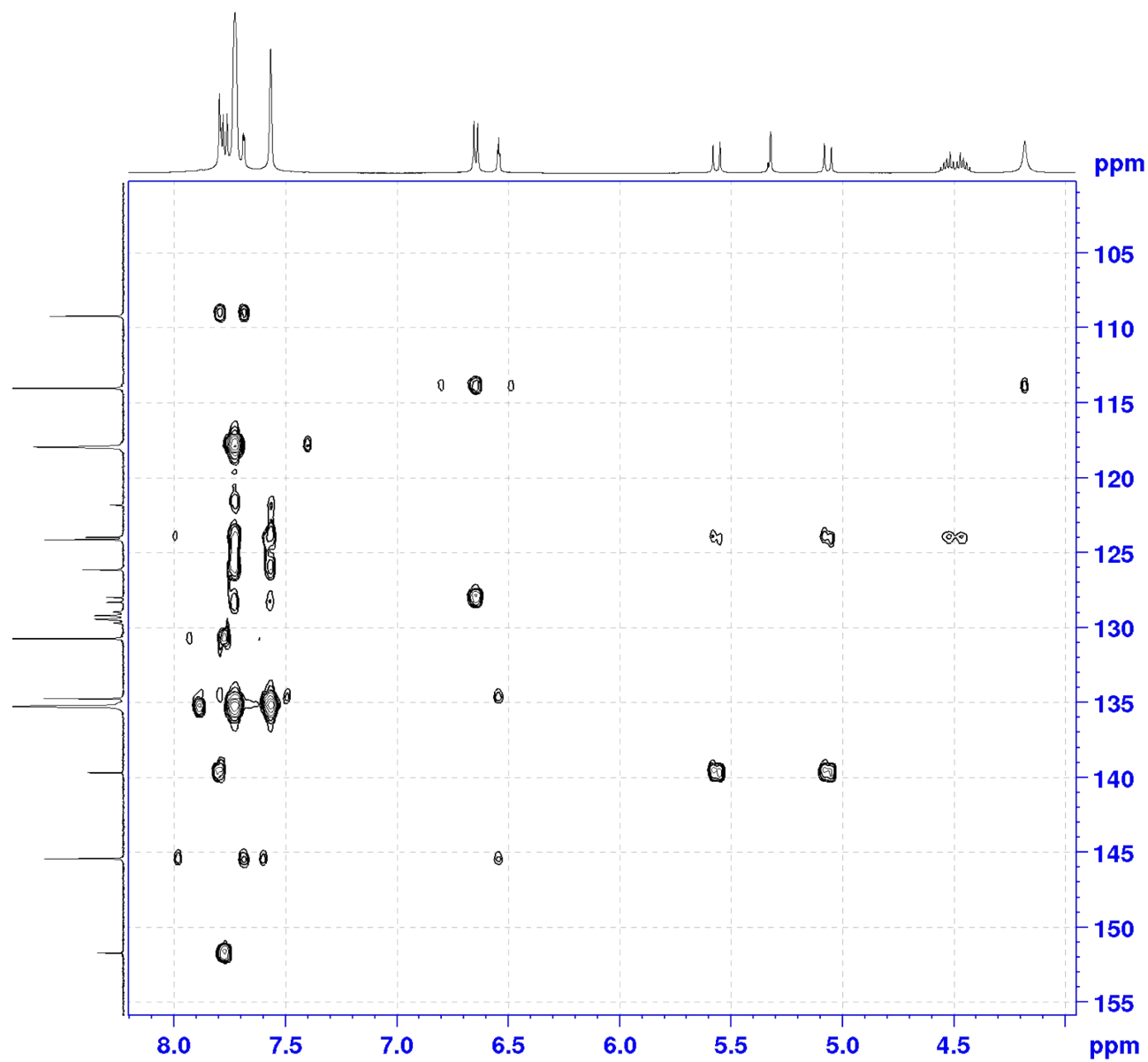
2D HSQC NMR spectrum of compound **Rh^IC₆** in CD₂Cl₂ and assignment of ¹H (red) and ¹³C (blue) NMR resonances.



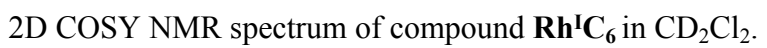
2D HMBC NMR spectrum of compound **Rh^IC₆** in CD_2Cl_2 .

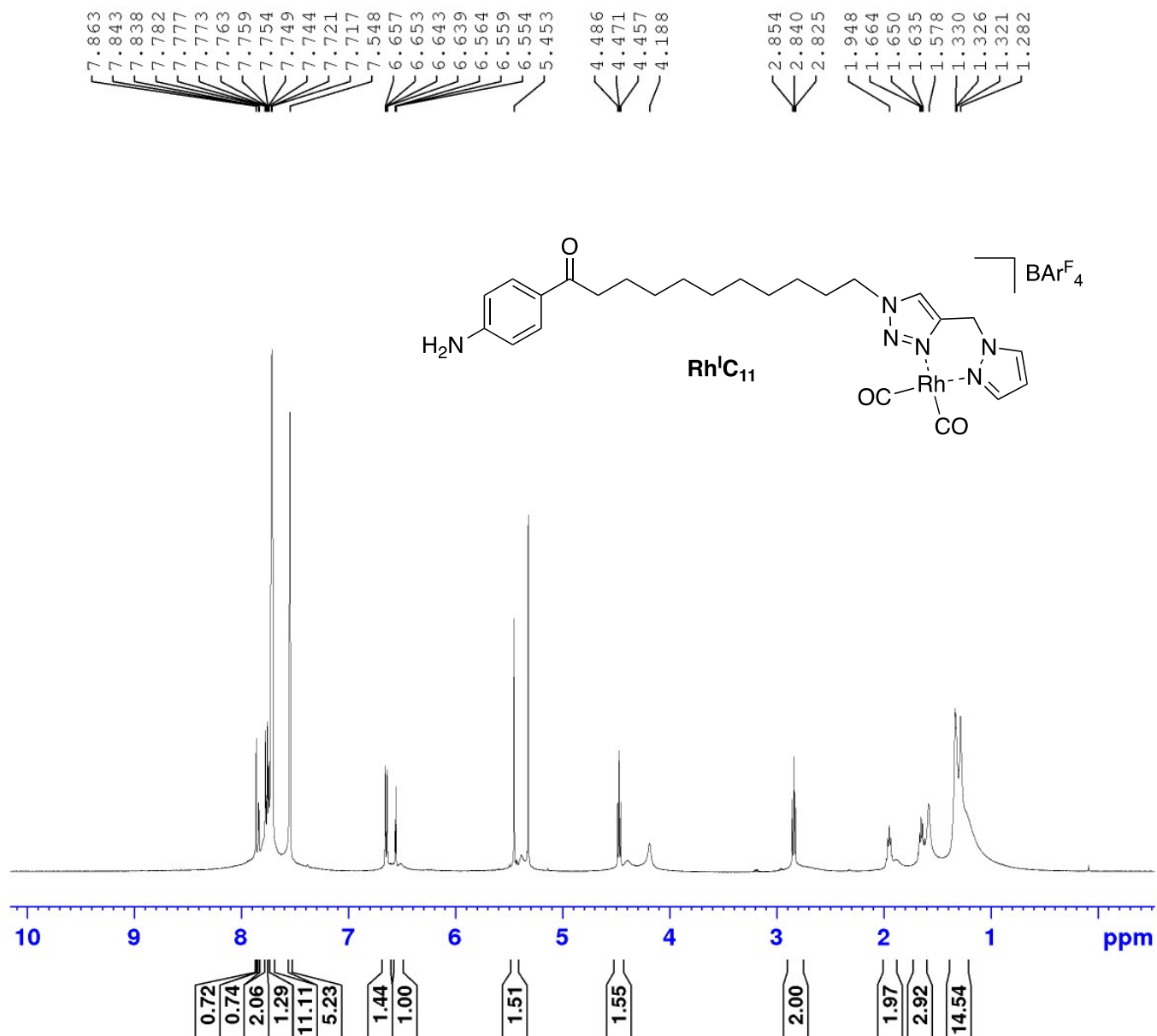


Selected range of the 2D HMBC NMR spectrum of compound **Rh^IC₆** in CD₂Cl₂.



Selected range of the 2D HMBC NMR spectrum of compound **Rh^IC₆** in CD_2Cl_2 .

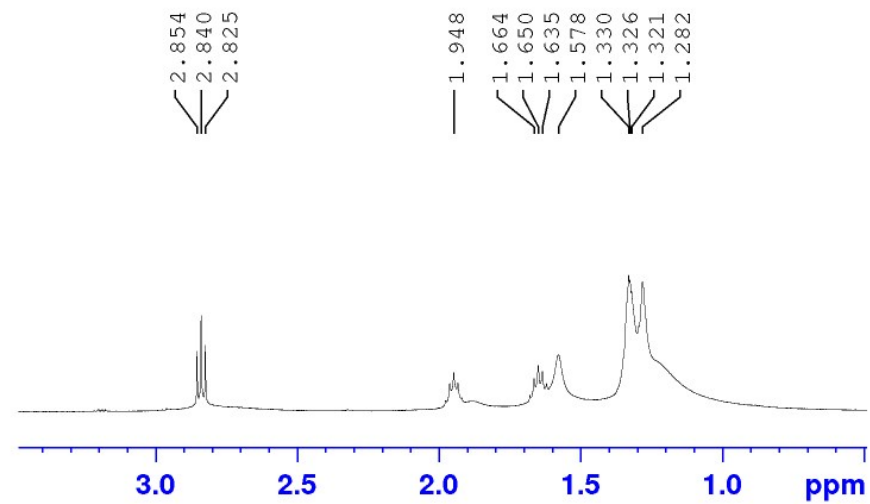
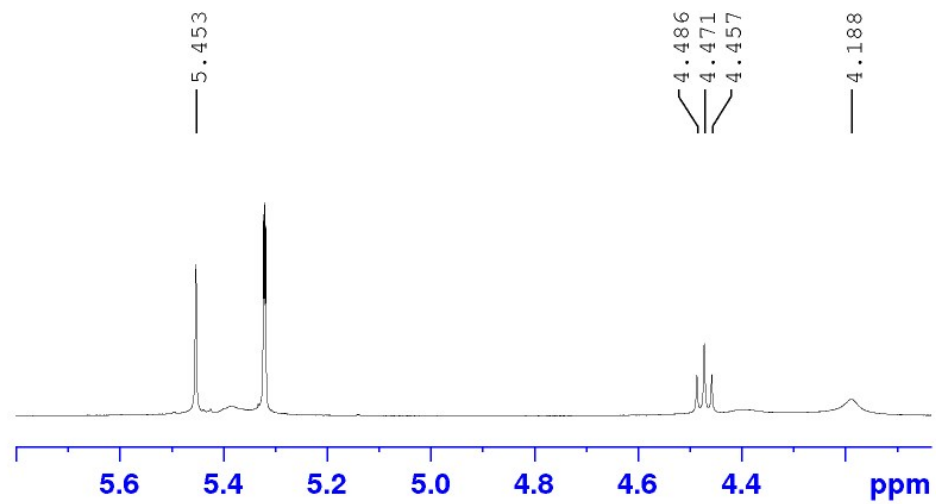
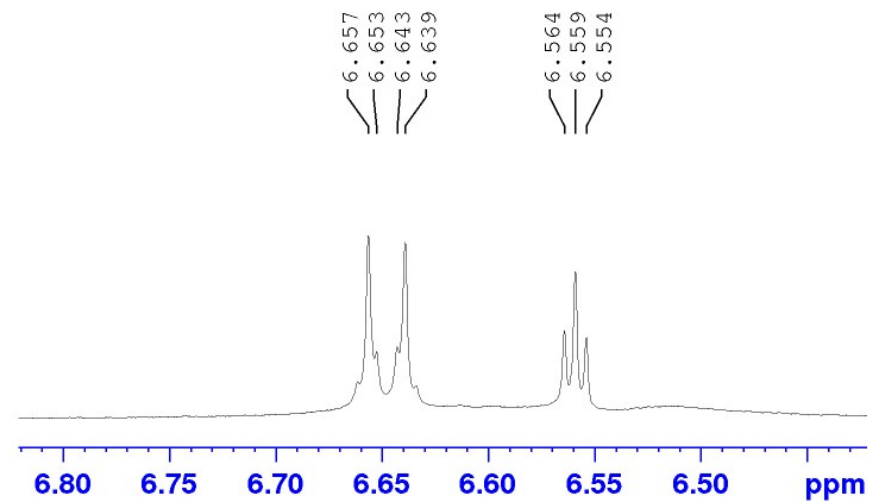
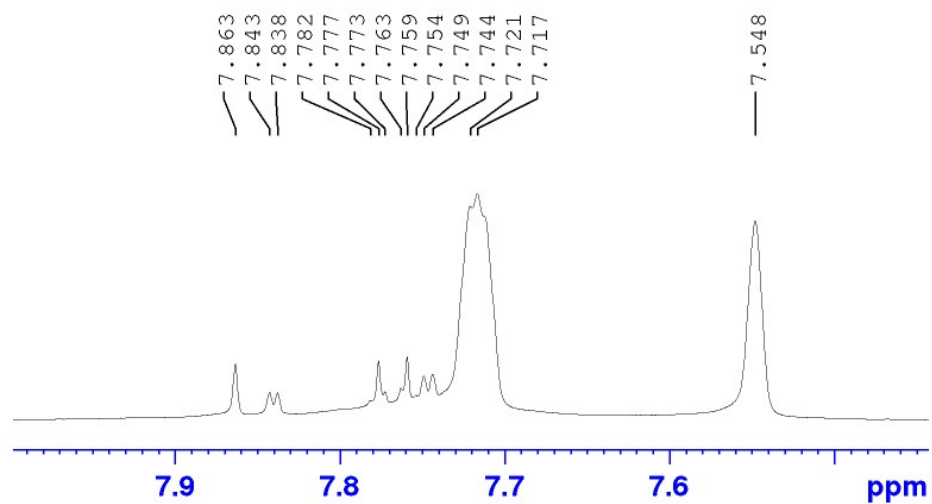


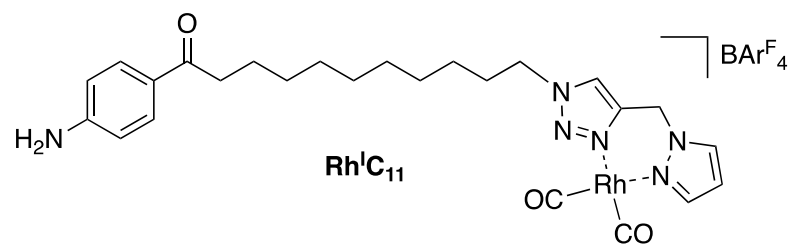


Current Data Parameters
 NAME MR-104
 EXPNO 7
 PROCNO 1

F2 - Acquisition Parameters
 Date_ 20200116
 Time 15.52 h
 INSTRUM spect
 PROBHD Z119470_0266 (
 PULPROG zg
 TD 65536
 SOLVENT CD2Cl2
 NS 8
 DS 0
 SWH 10000.000 Hz
 FIDRES 0.305176 Hz
 AQ 3.276799 sec
 RG 65.43
 DW 50.000 usec
 DE 6.50 usec
 TE 298.0 K
 D1 5.00000000 sec
 TD0 1
 SFO1 500.1622533 MHz
 NUC1 1H
 P1 11.10 usec
 PLW1 25.00000000 W

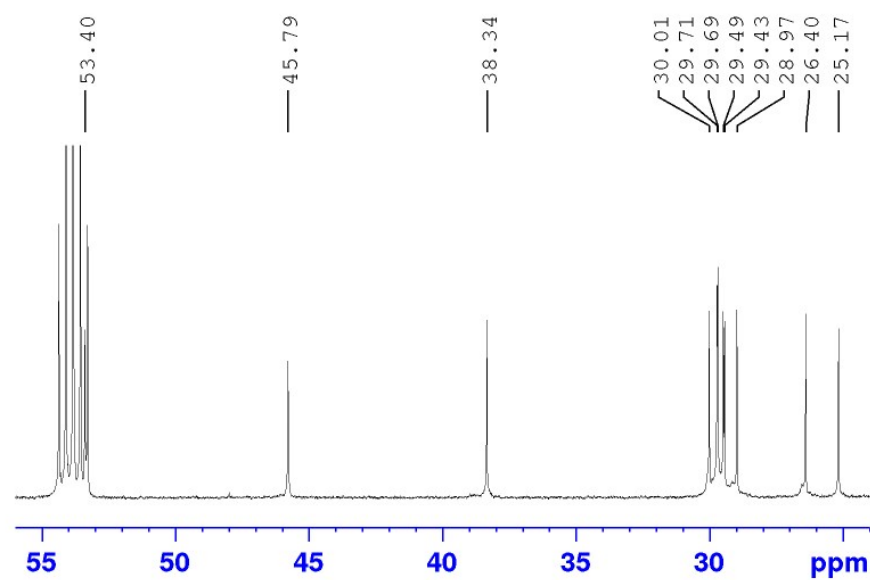
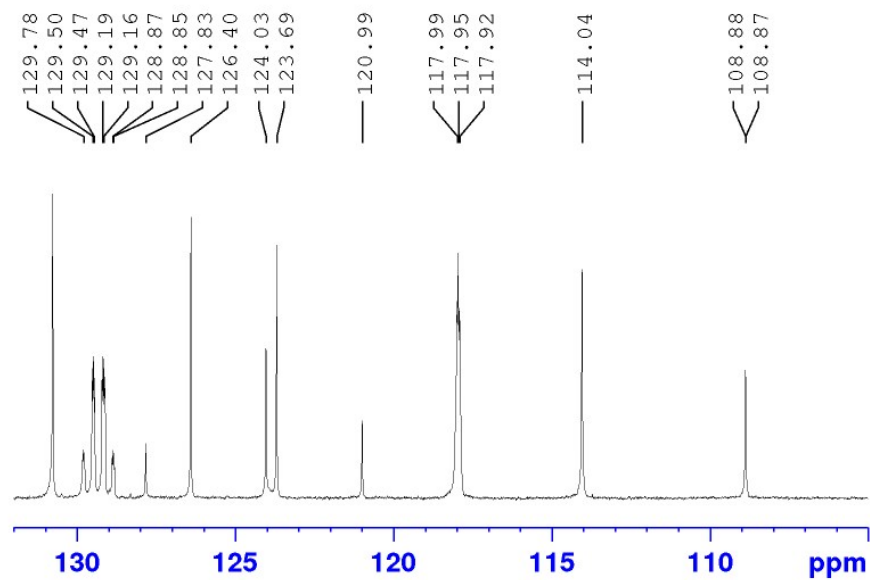
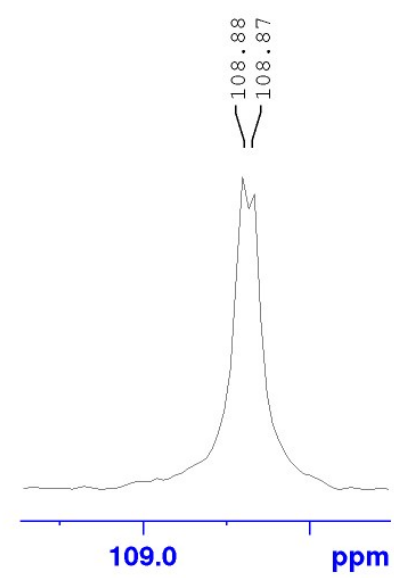
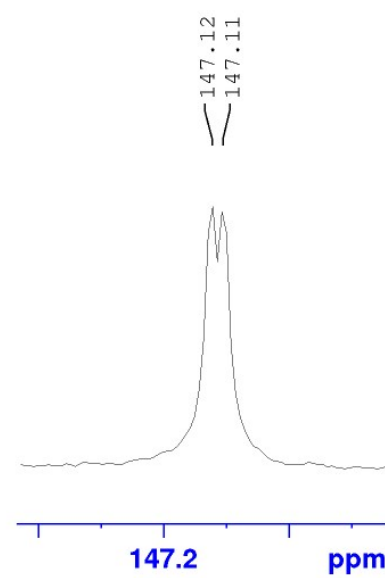
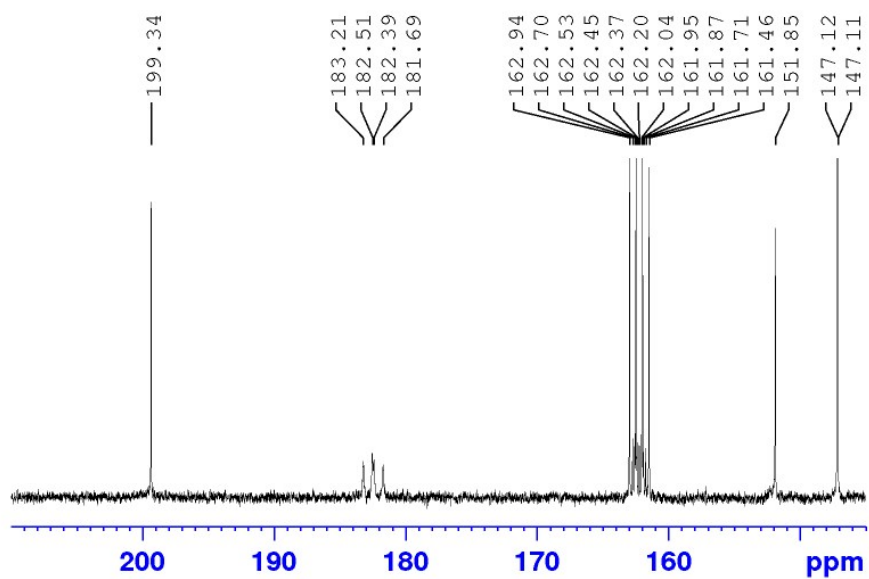
F2 - Processing parameters
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 SF 500.1600194 MHz
 WDW EM
 SSB 0
 LB 0.30 Hz
 GB 0
 PC 1.00

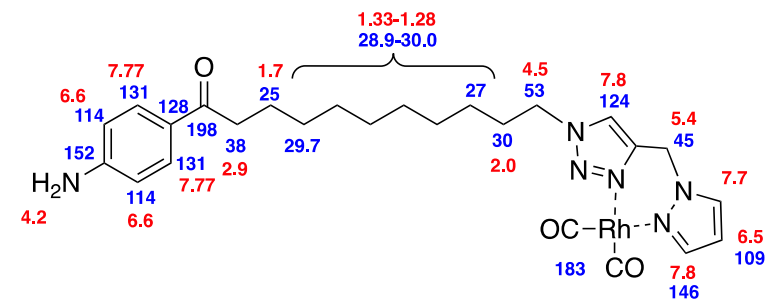
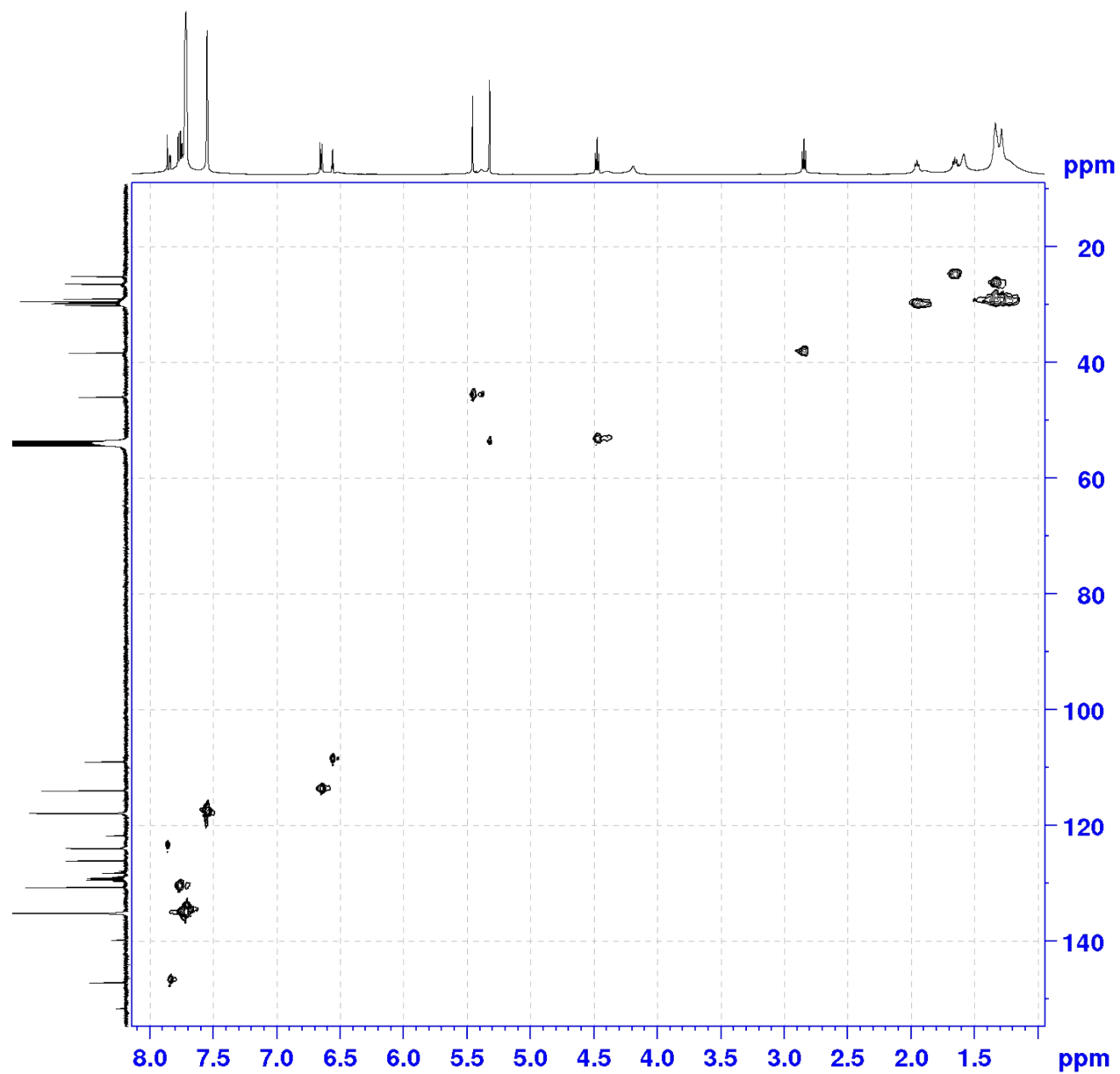




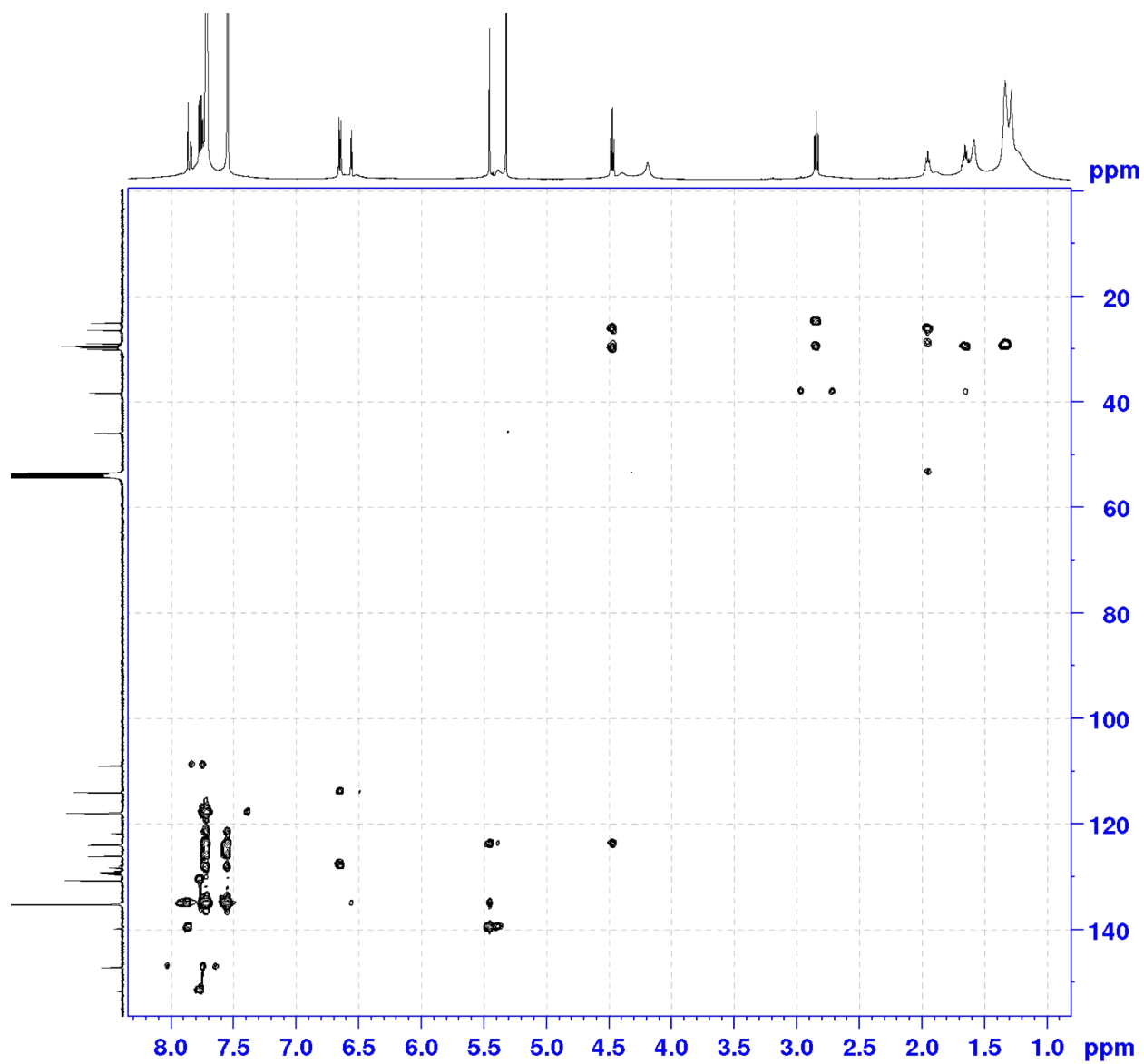
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2.94
2.70
2.53
2.45
2.37
2.20
2.04
1.95
1.87
1.71
1.46
1.85
7.12
7.11
9.88
5.26
5.16
0.77
9.81
9.78
9.50
9.47
9.19
9.16
8.87
8.85
7.83
6.40
4.03
3.69
0.99
7.99
7.95
7.92
4.04
8.88
8.87
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.34
.01
.71
.69
.49
.43
.97
.40
.17

Current Data Parameters
NAME MR-104

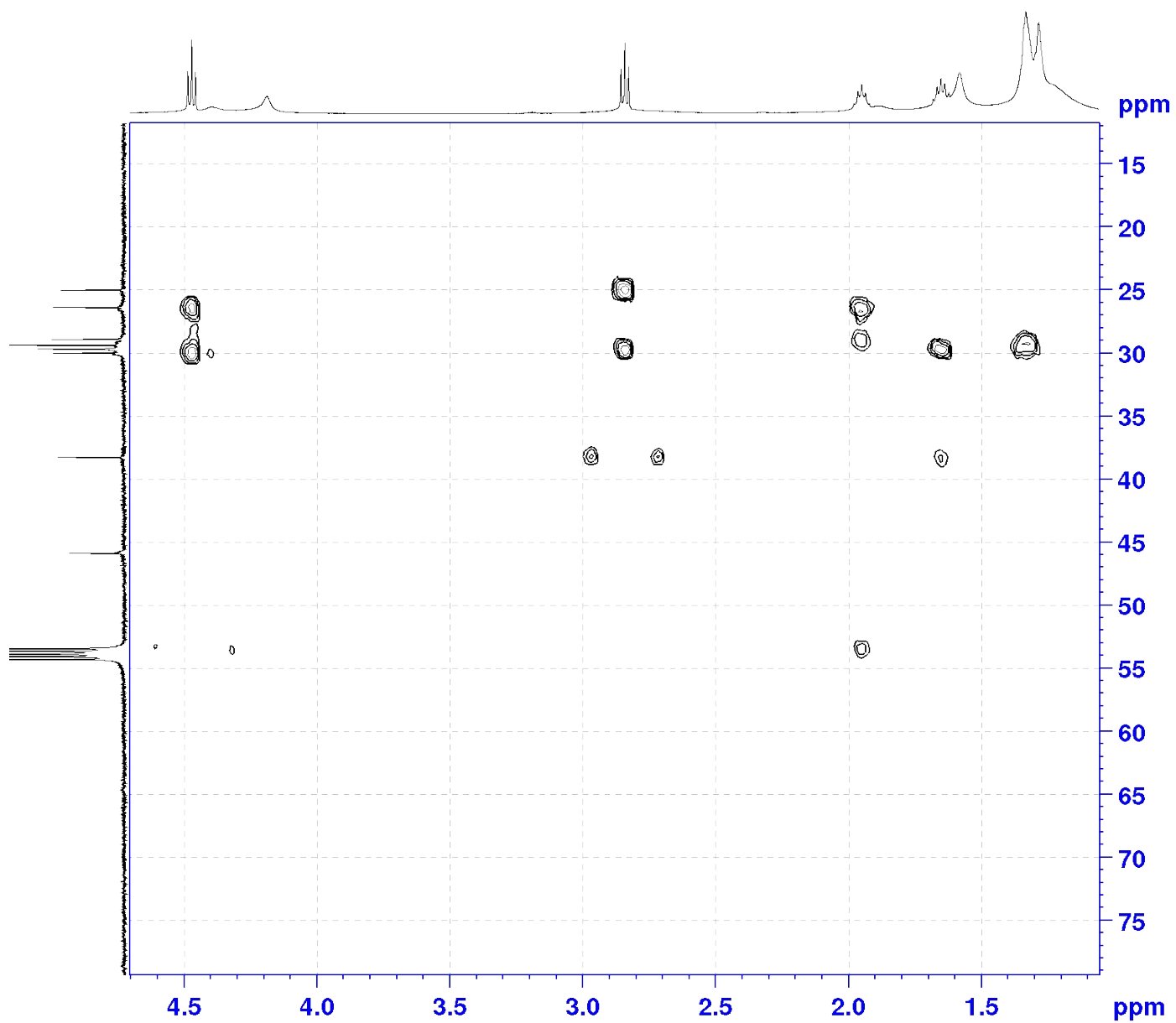




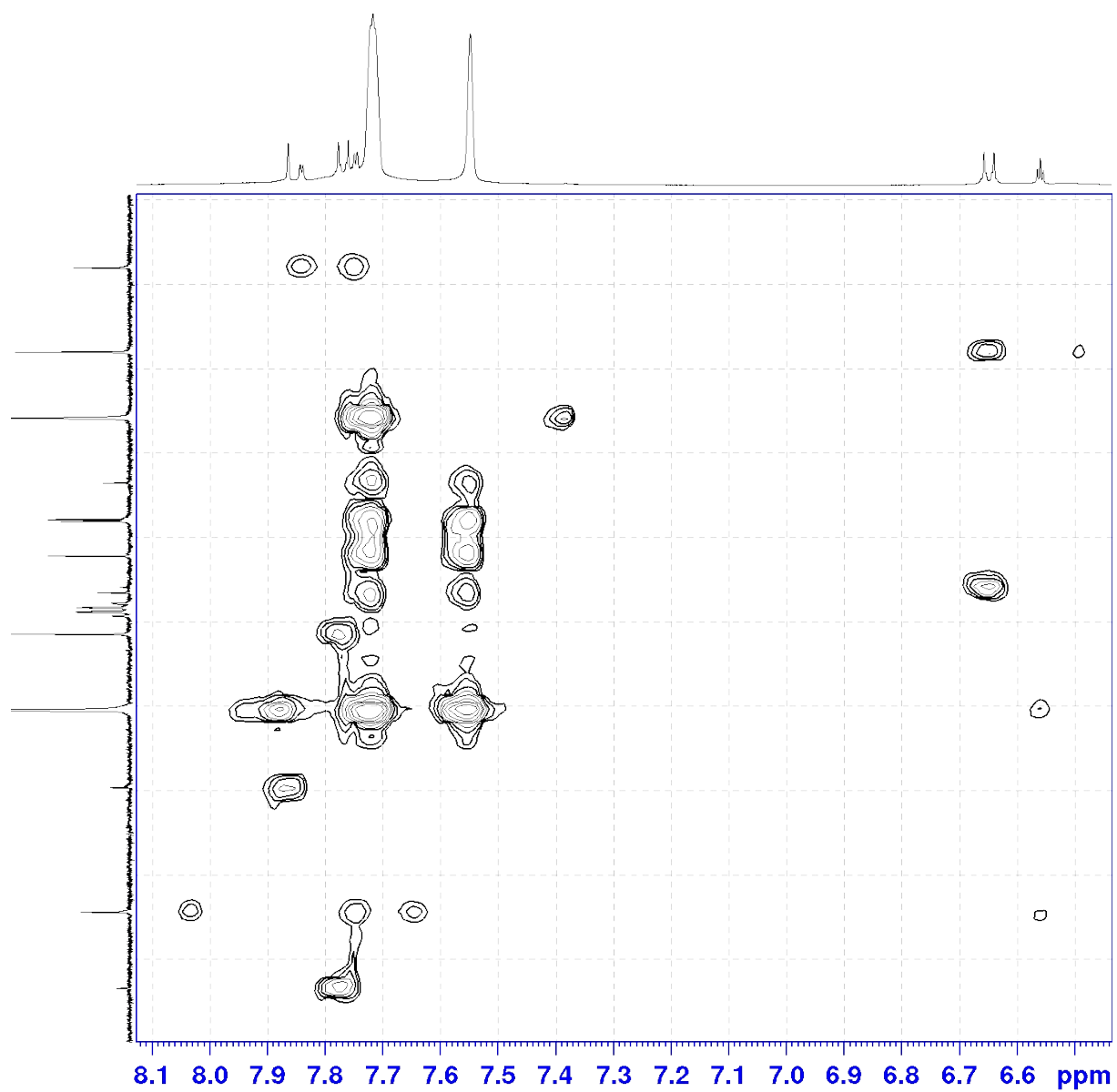
2D HSQC NMR spectrum of compound **Rh^IC₁₁** in CD₂Cl₂ assignment of ¹H (red) and ¹³C (blue) NMR resonances.



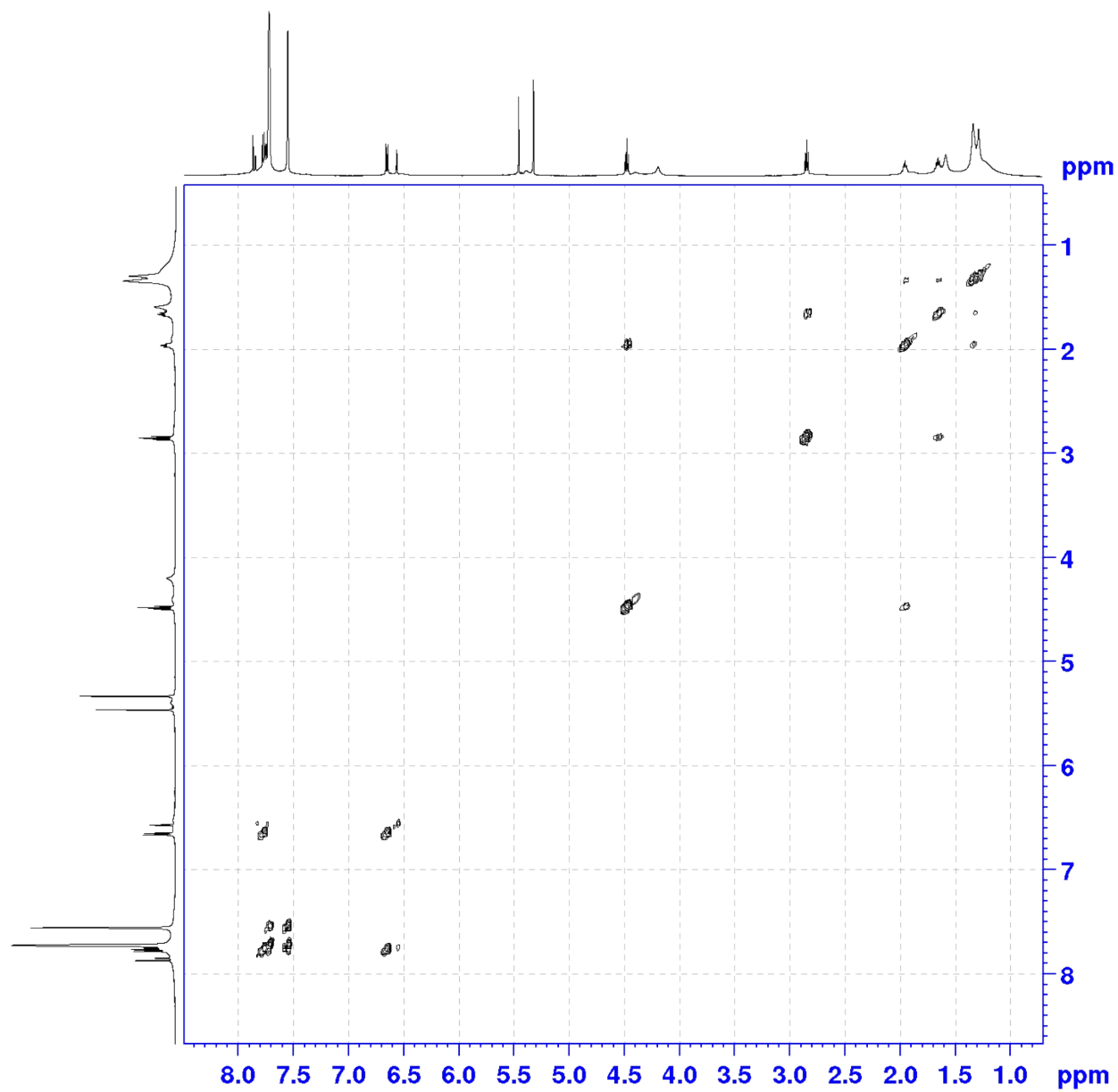
2D HMBC NMR spectrum of compound **Rh^IC₁₁** in CD_2Cl_2 .



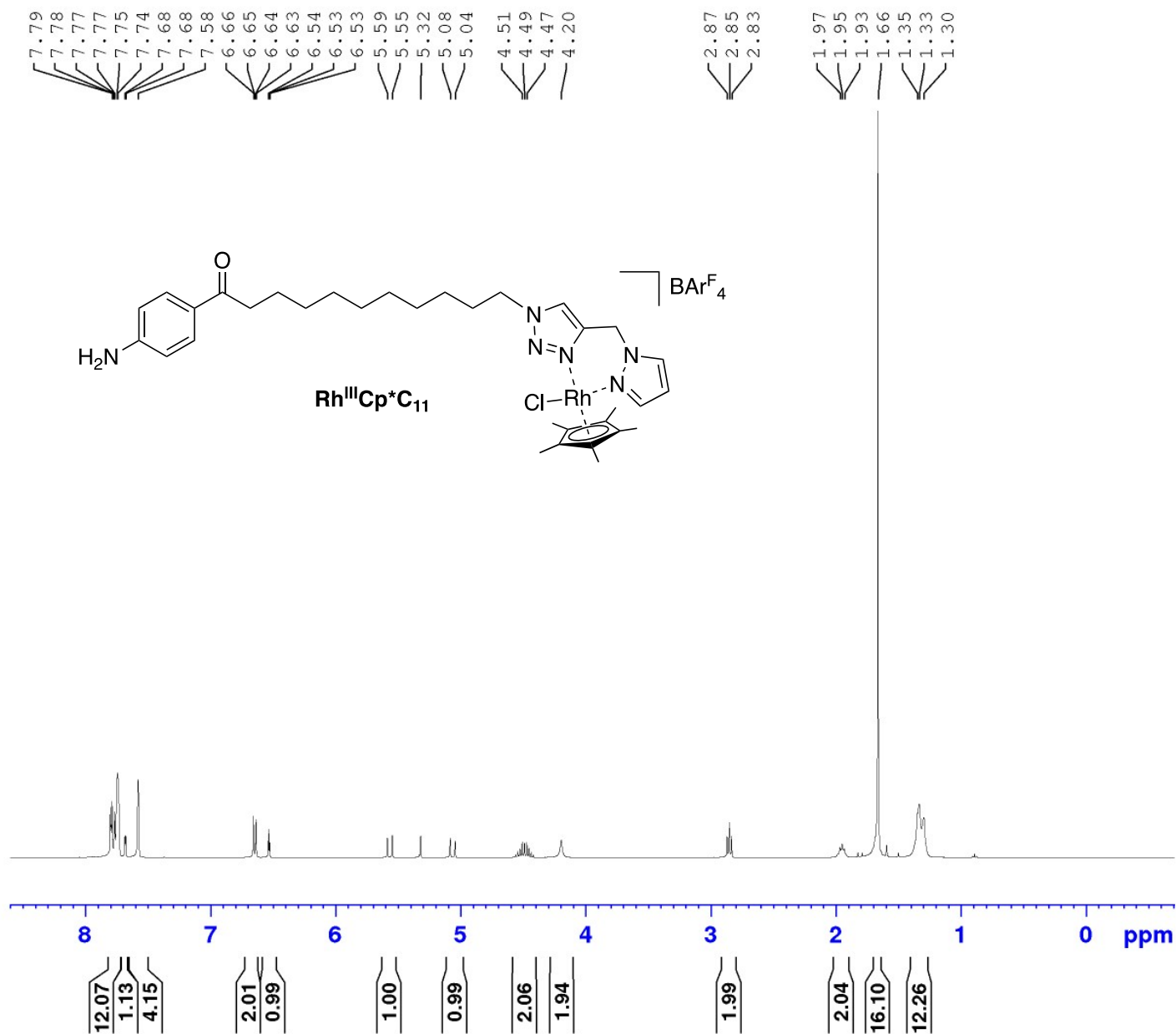
Selected range of the 2D HMBC NMR spectrum of compound **Rh^IC₁₁** in CD₂Cl₂.



Selected range of the 2D HMBC NMR spectrum of compound **Rh^IC₁₁** in CD₂Cl₂.



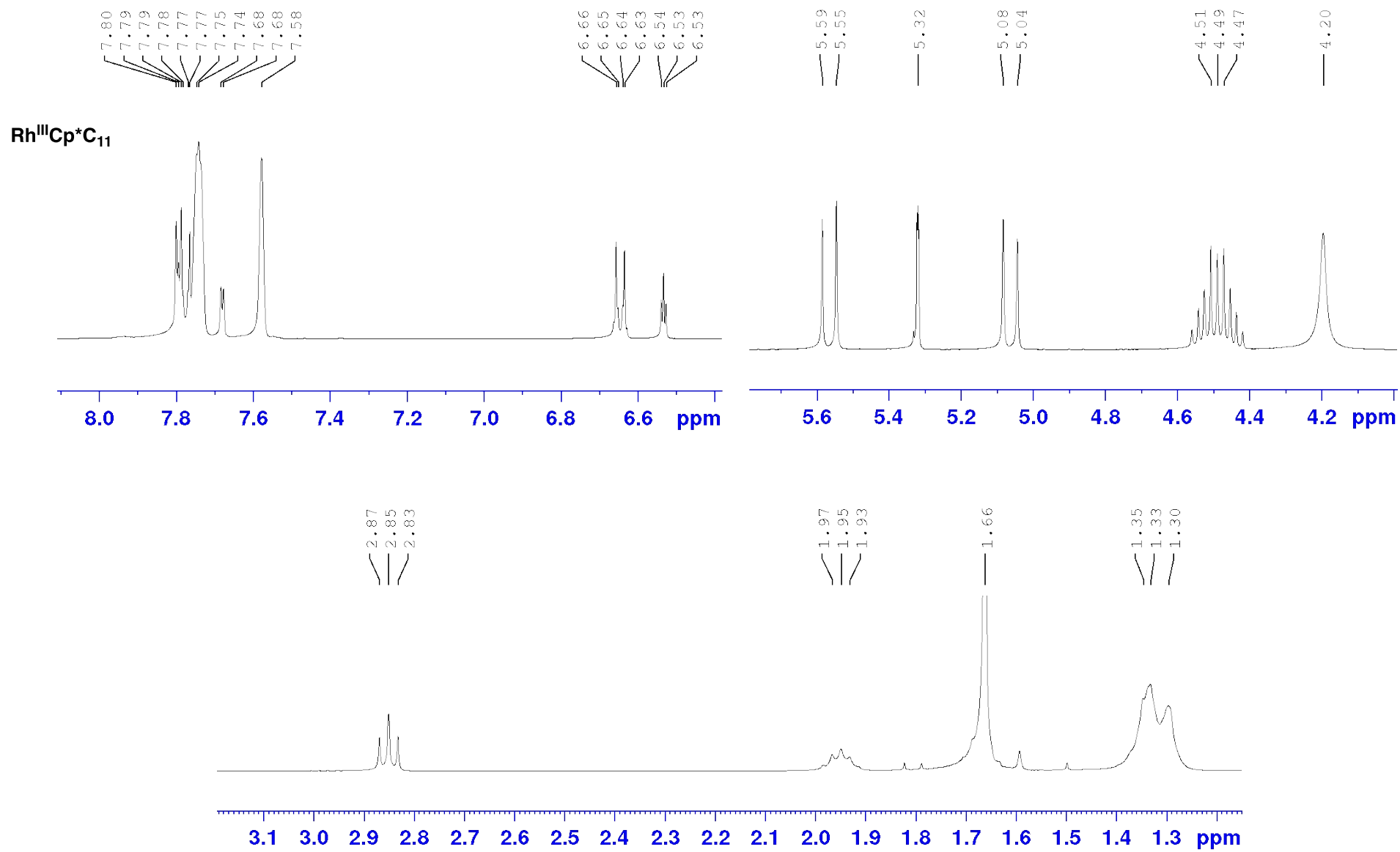
2D COSY NMR spectrum of compound **Rh^IC₁₁** in CD₂Cl₂.

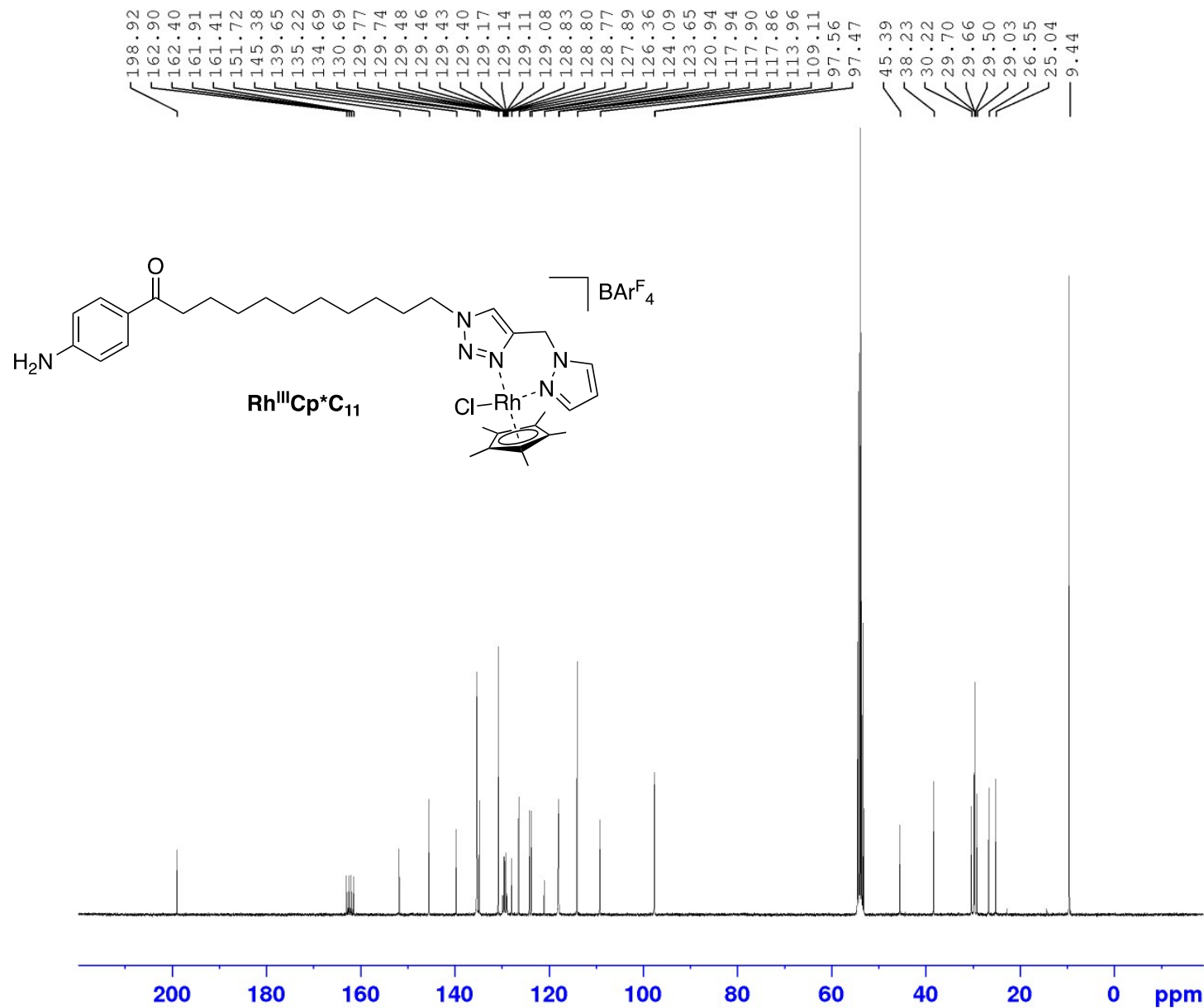


Current Data Parameters
NAME MR-230
EXPNO 2
PROCNO 1

F2 - Acquisition Parameters
Date_ 20190819
Time 16.32 h
INSTRUM spect
PROBHD Z116098_0573
PULPROG zg30
TD 65536
SOLVENT CD2Cl2
NS 16
DS 2
SWH 8012.820 Hz
FIDRES 0.244532 Hz
AQ 4.0894465 sec
RG 55.9
DW 62.400 usec
DE 12.00 usec
TE 298.0 K
D1 2.50000000 sec
TD0 1
SFO1 400.1524709 MHz
NUC1 1H
P1 10.00 usec
PLW1 15.75399971 W

F2 - Processing parameters
SI 65536
SF 400.1500154 MHz
WDW EM
SSB 0
LB 0.30 Hz
GB 0
PC 1.00

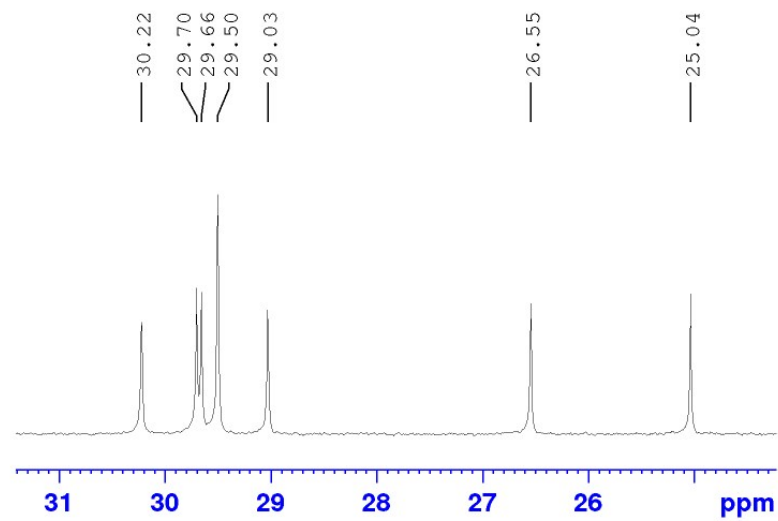
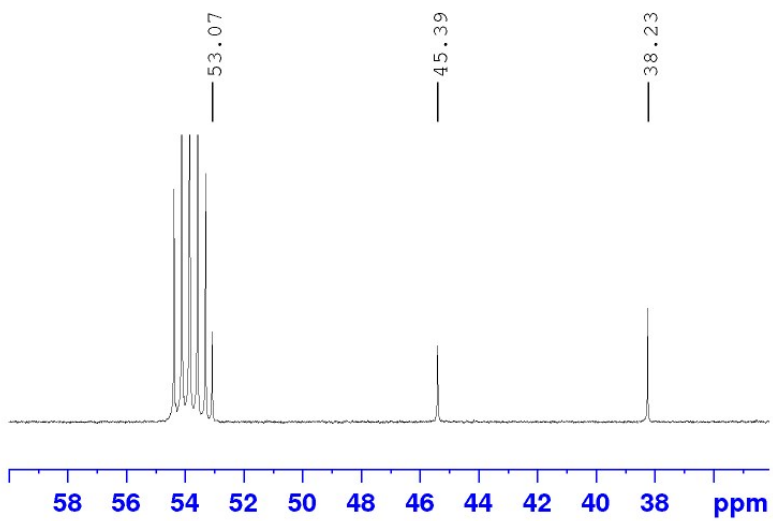
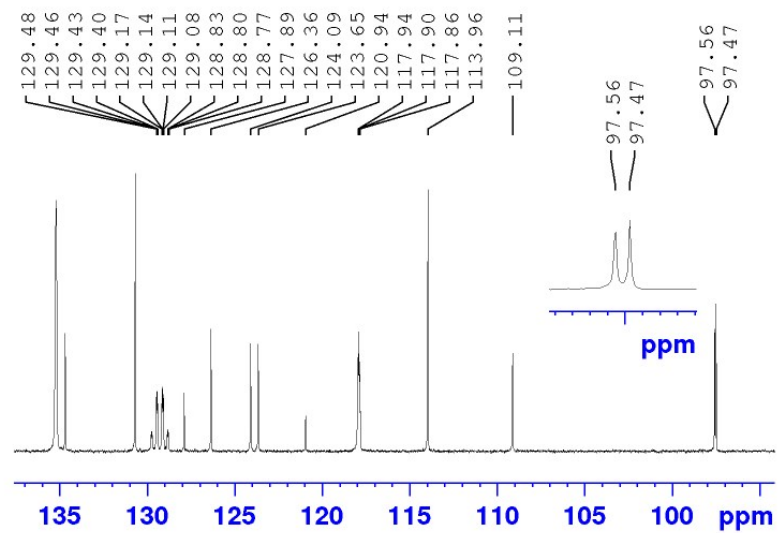
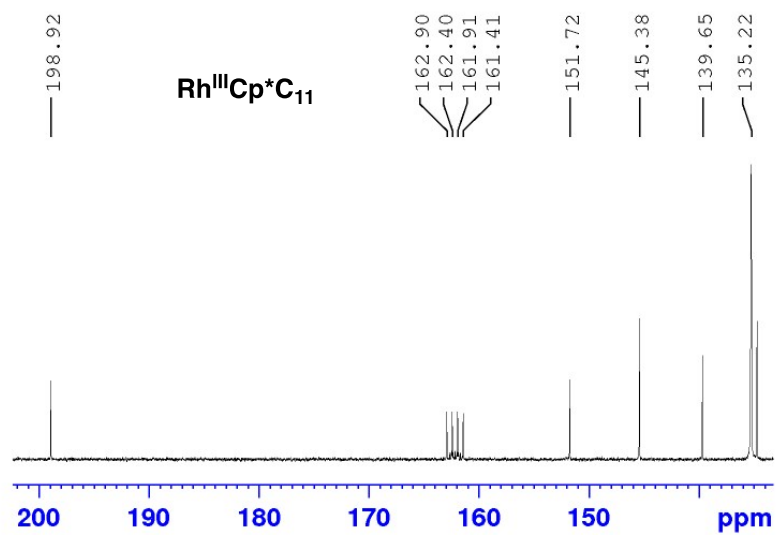


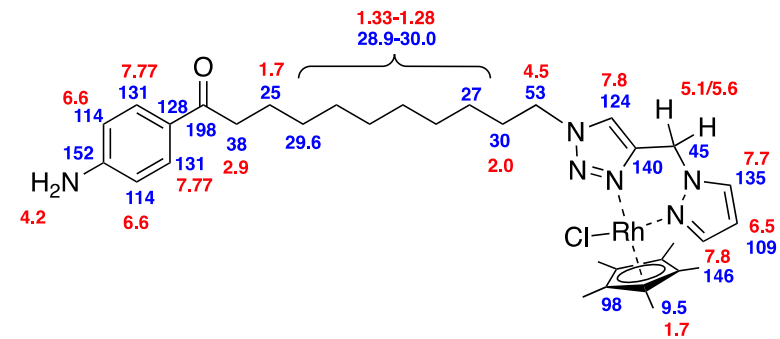
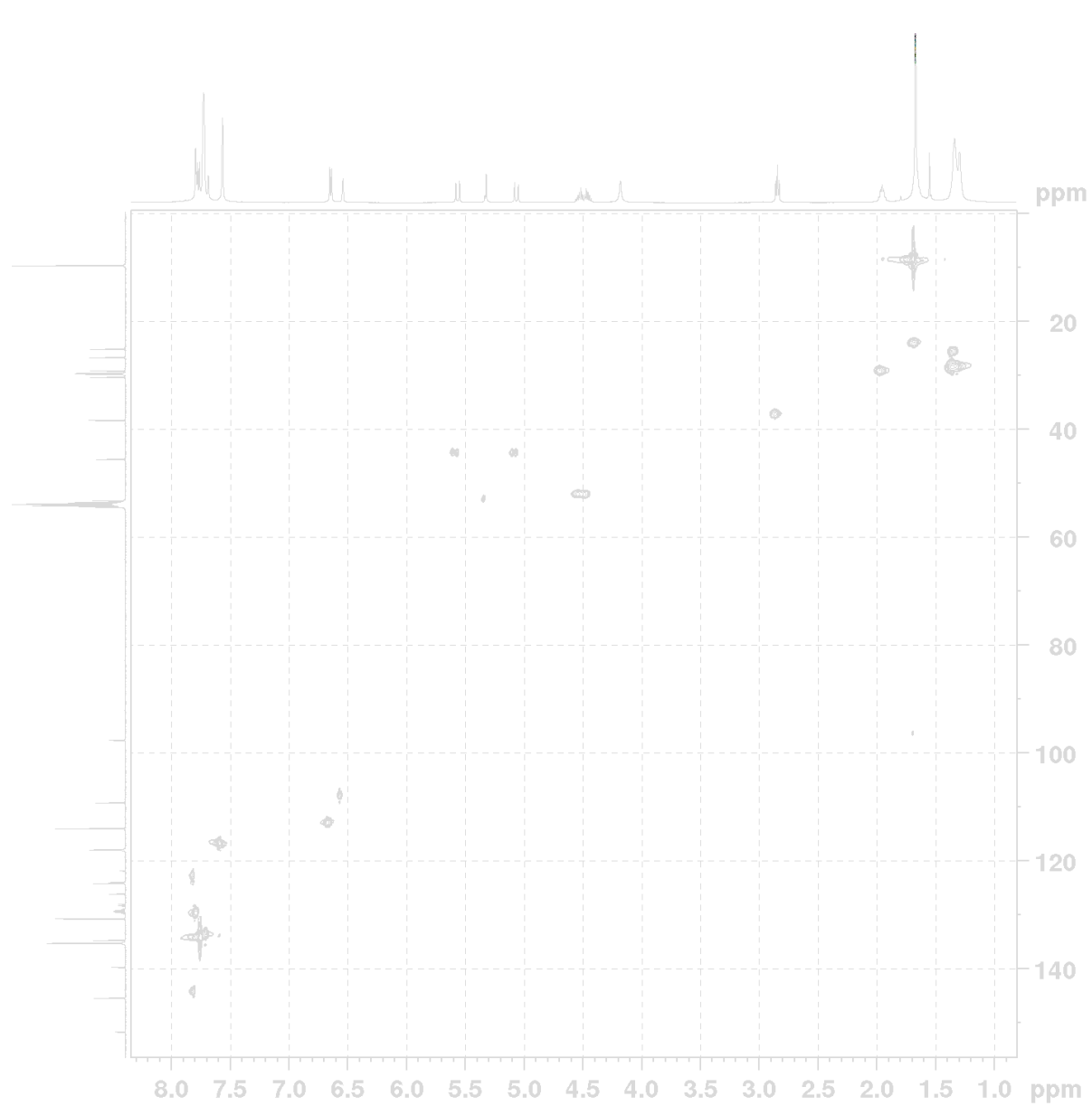


Current Data Parameters
NAME MR-230
EXPNO 3
PROCNO 1

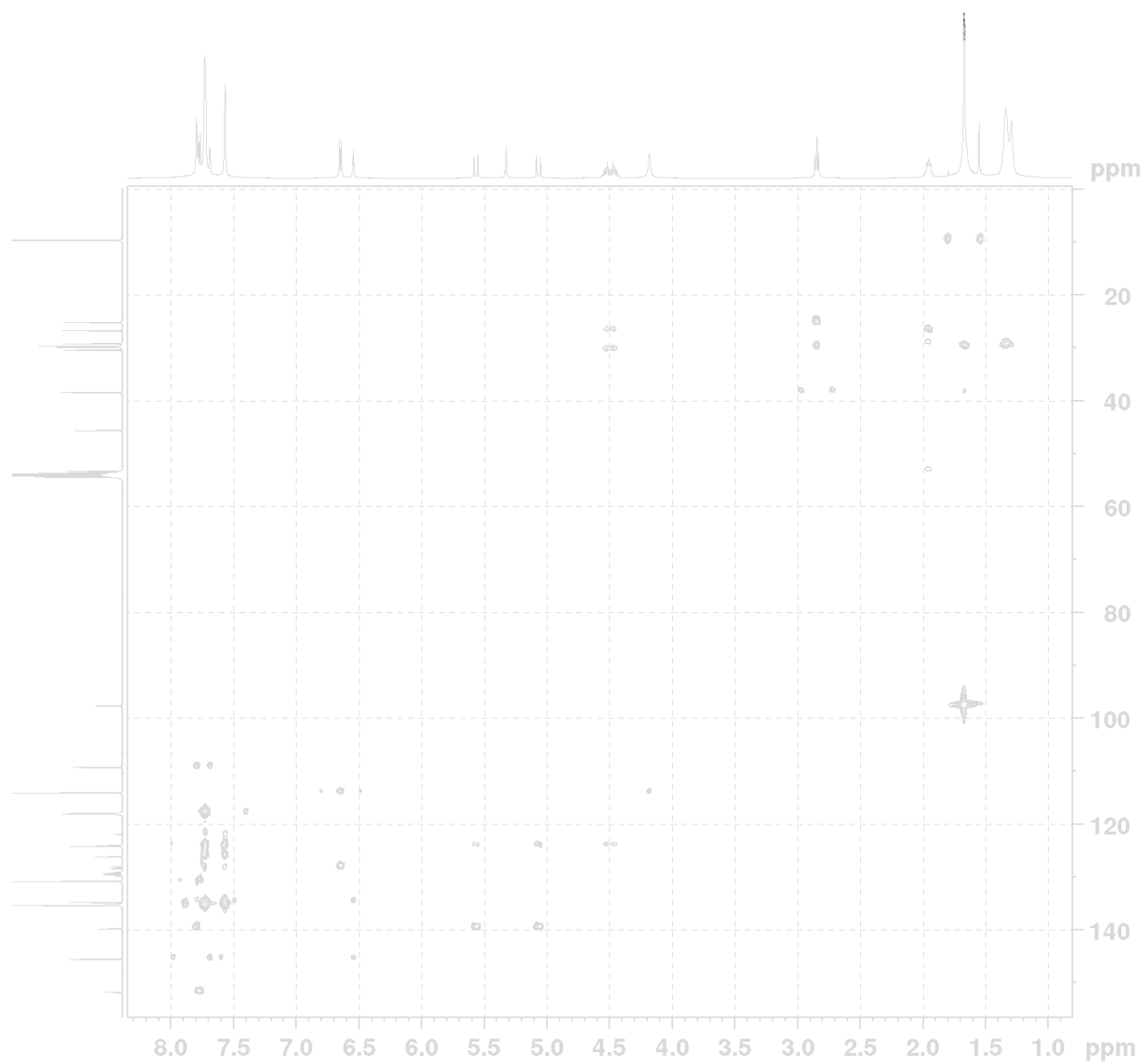
F2 - Acquisition Parameters
Date_ 20190820
Time 0.48 h
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PROBHD Z116098_0573 (
PULPROG zgpg30
TD 65536
SOLVENT CD2Cl2
NS 6000
DS 4
SWH 24038.461 Hz
FIDRES 0.733596 Hz
AQ 1.3631488 sec
RG 198.41
DW 20.800 usec
DE 6.50 usec
TE 298.0 K
D1 2.00000000 sec
D11 0.03000000 sec
TD0 1
SFO1 100.6278593 MHz
NUC1 13C
P1 10.00 usec
PLW1 80.53199768 W
SFO2 400.1516006 MHz
NUC2 1H
CPDPRG[2] waltz16
PCPD2 90.00 usec
PLW2 15.75399971 W
PLW12 0.19449000 W
PLW13 0.09782900 W

F2 - Processing parameters
SI 32768
SF 100.6177559 MHz
WDW EM
SSB 0
LB 1.00 Hz
GB 0
PC 1.40

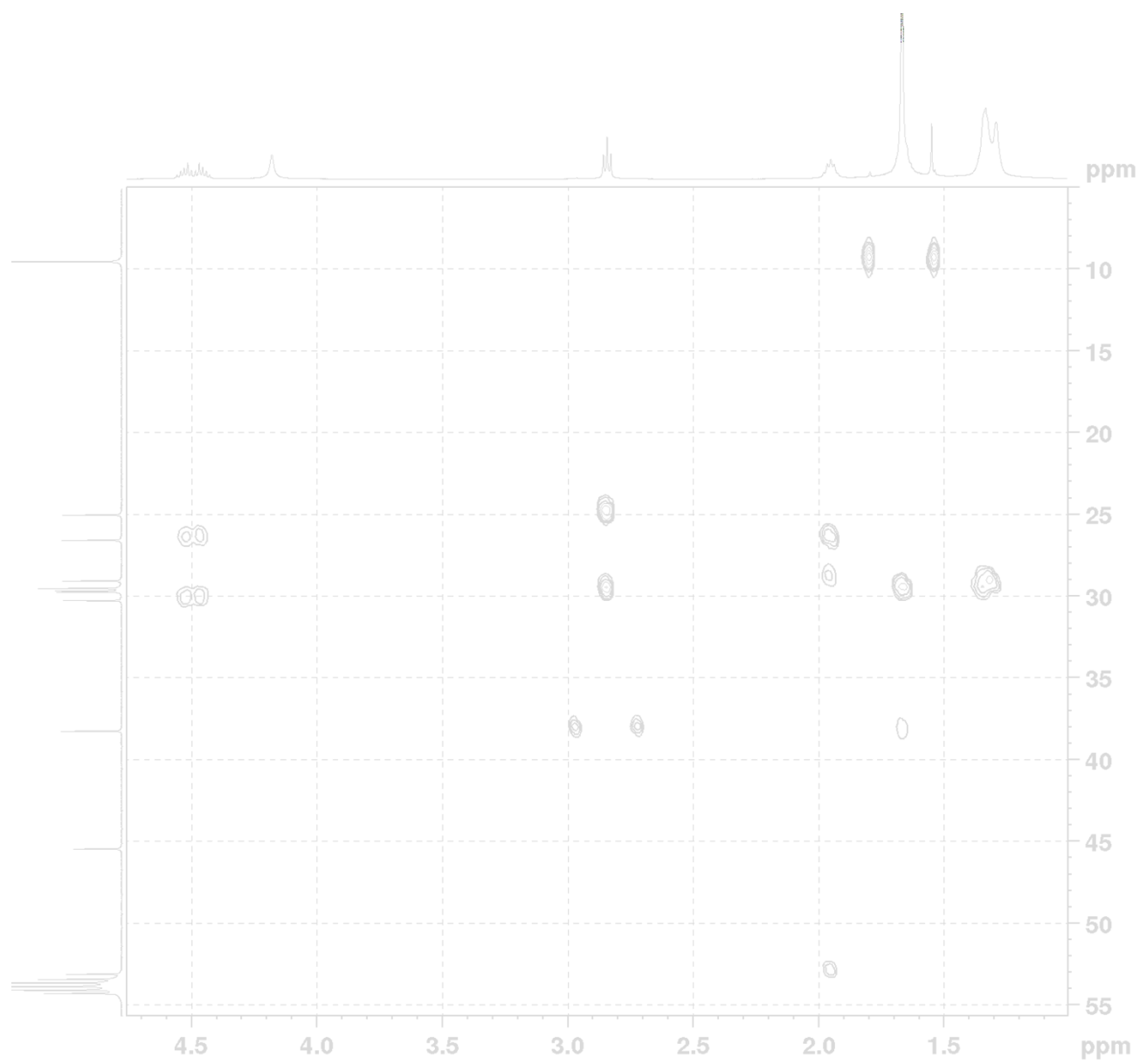




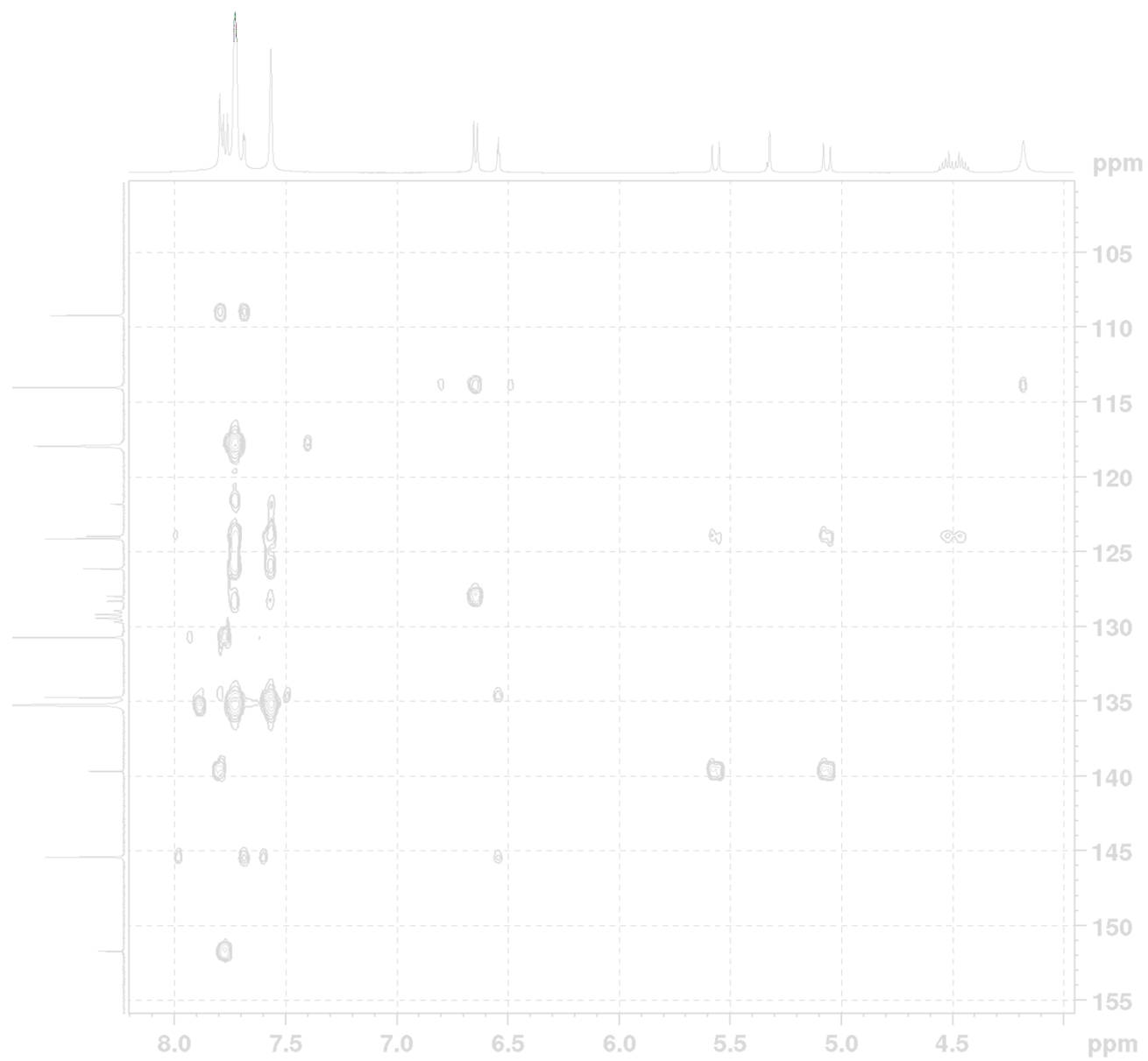
2D HSQC NMR spectrum of compound **Rh^{III}C₁₁** in CD₂Cl₂ and assignment of ¹H (red) and ¹³C (blue) NMR resonances.



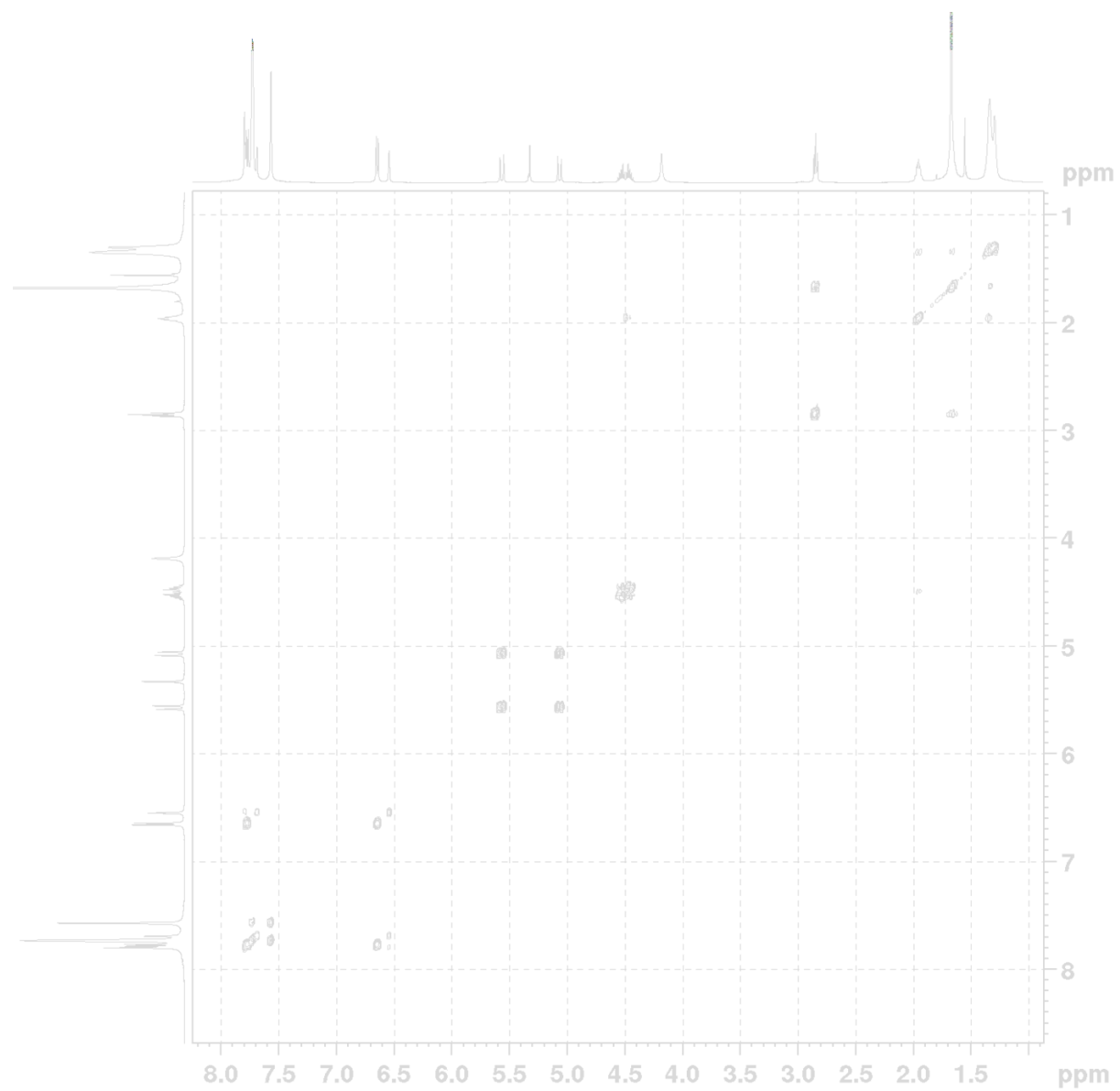
2D HMBC NMR spectrum of compound $\text{Rh}^{\text{III}}\text{C}_{11}$ in CD_2Cl_2 .



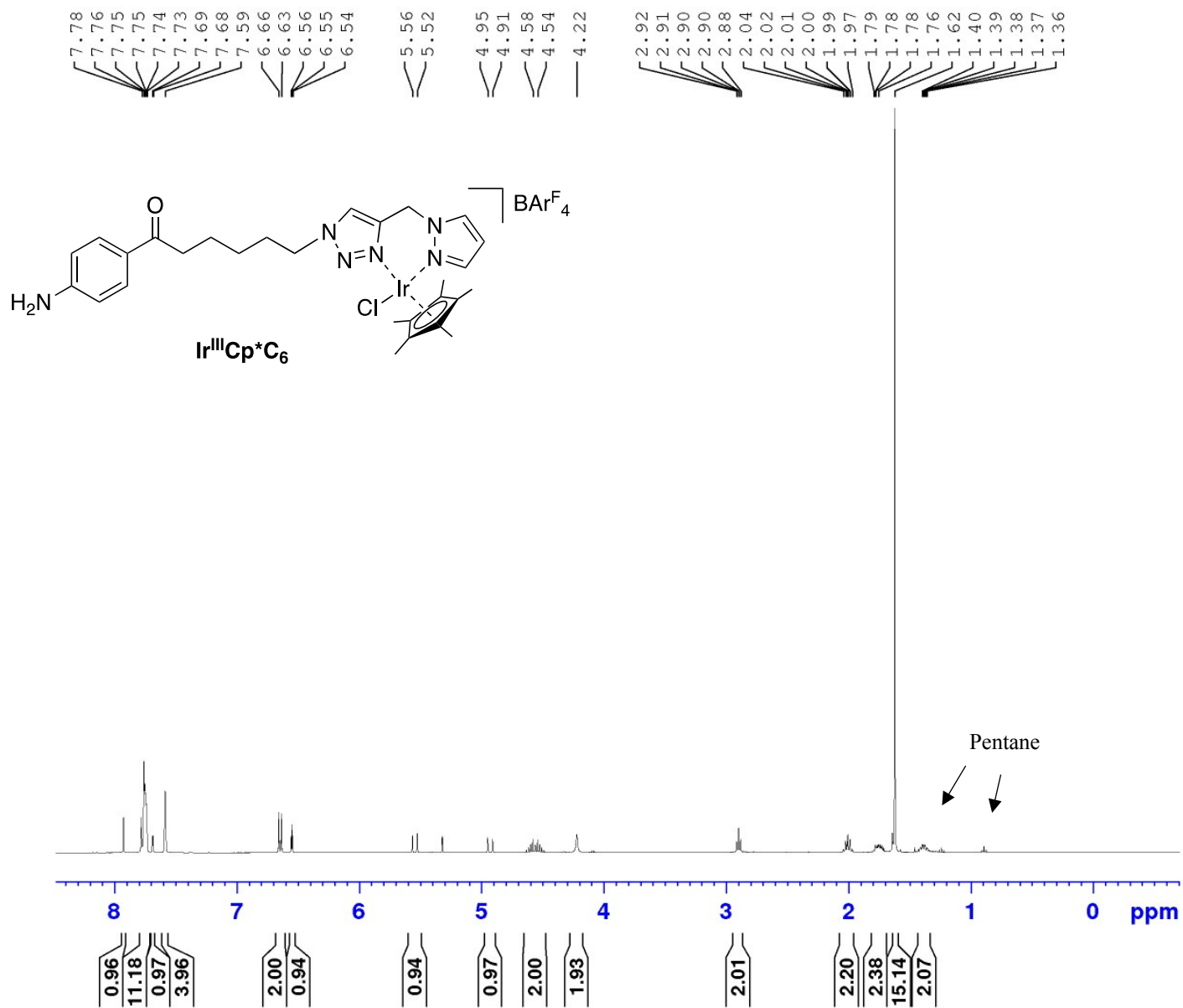
Selected range of the 2D HMBC NMR spectrum of compound $\text{Rh}^{\text{III}}\text{C}_{11}$ in CD_2Cl_2 .



Selected range of the 2D HMBC NMR spectrum of compound **Rh^{III}C₁₁** in CD₂Cl₂.



2D COSY NMR spectrum of compound **Rh^{III}C₁₁** in CD₂Cl₂.



Current Data Parameters

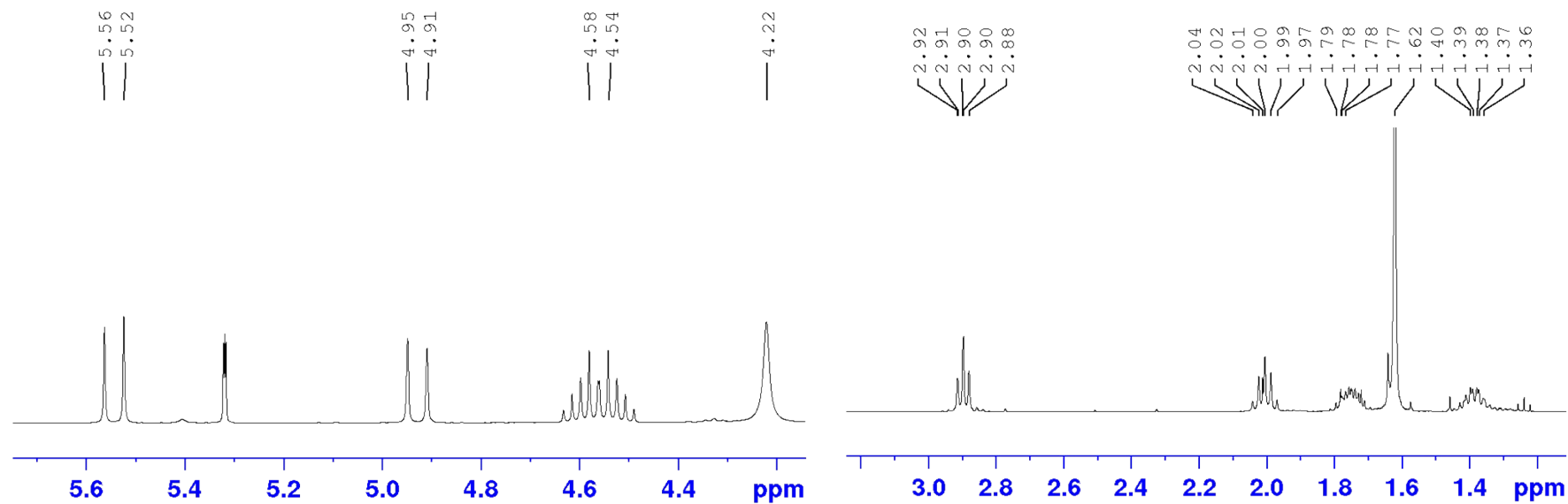
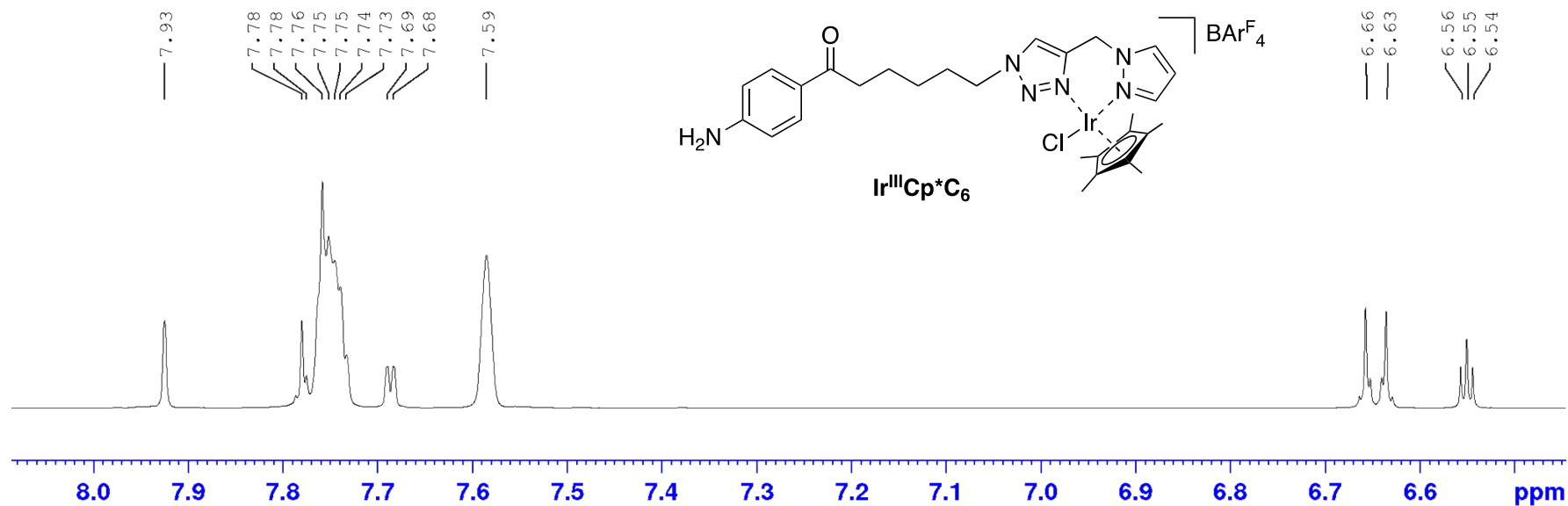
NAME	MR-97-3
EXPNO	1
PROCNO	1

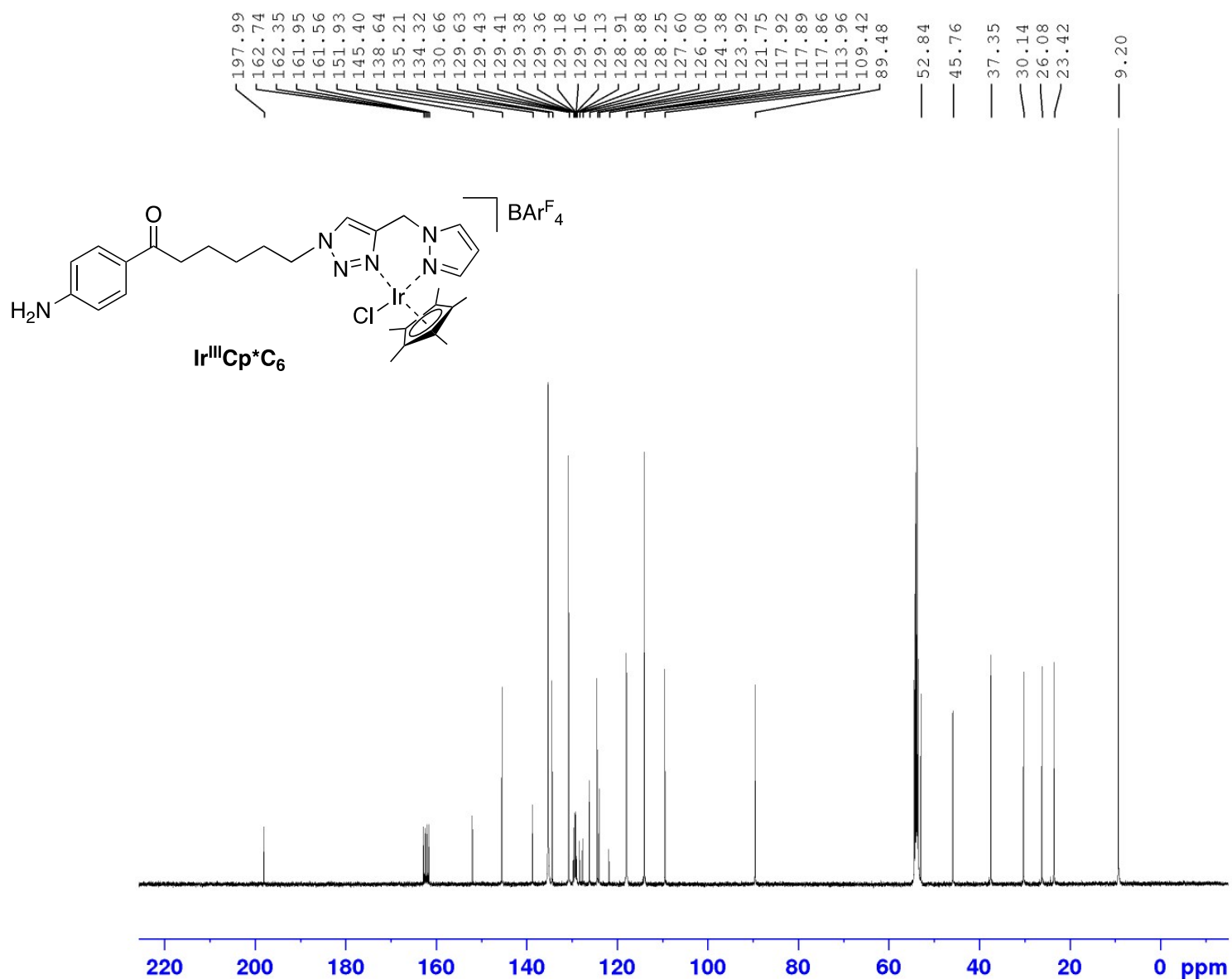
F2 - Acquisition Parameters

Date_	20190305
Time	9.43 h
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PROBHD	Z116098_0573 (
PULPROG	zg30
TD	65536
SOLVENT	CD2Cl2
NS	16
DS	2
SWH	8012.820 Hz
FIDRES	0.244532 Hz
AQ	4.0894465 sec
RG	49.89
DW	62.400 usec
DE	12.00 usec
TE	298.0 K
D1	2.50000000 sec
TD0	1
SFO1	400.1524709 MHz
NUC1	1H
P1	10.00 usec
PLW1	15.75399971 W

F2 - Processing parameters

SI	65536
SF	400.1500154 MHz
WDW	EM
SSB	0
LB	0.30 Hz
GB	0
PC	1.00



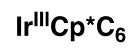


Current Data Parameters
NAME MR-97
EXPNO 2
PROCNO 1

F2 - Acquisition Parameters
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Time 21.32 h
INSTRUM spect
PROBHD Z119470_0266 (
PULPROG zgpg
TD 65536
SOLVENT CD2Cl2
NS 6000
DS 2
SWH 34090.910 Hz
FIDRES 1.040372 Hz
AQ 0.9611947 sec
RG 187.37
DW 14.667 usec
DE 7.59 usec
TE 298.2 K
D1 3.00000000 sec
D11 0.03000000 sec
TD0 1
SFO1 125.7804239 MHz
NUC1 13C
P1 10.00 usec
PLW1 98.23000336 W
SFO2 500.1632510 MHz
NUC2 1H
CPDPRG[2] waltz16
PCPD2 80.00 usec
PLW2 25.00000000 W
PLW12 0.48129001 W
PLW13 0.24208000 W

F2 - Processing parameters
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WDW EM
SSB 0
LB 1.00 Hz
GB 0
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74
35
95
56

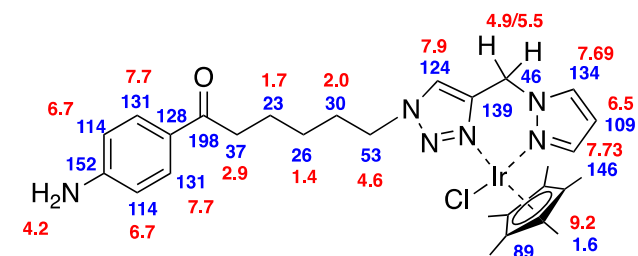
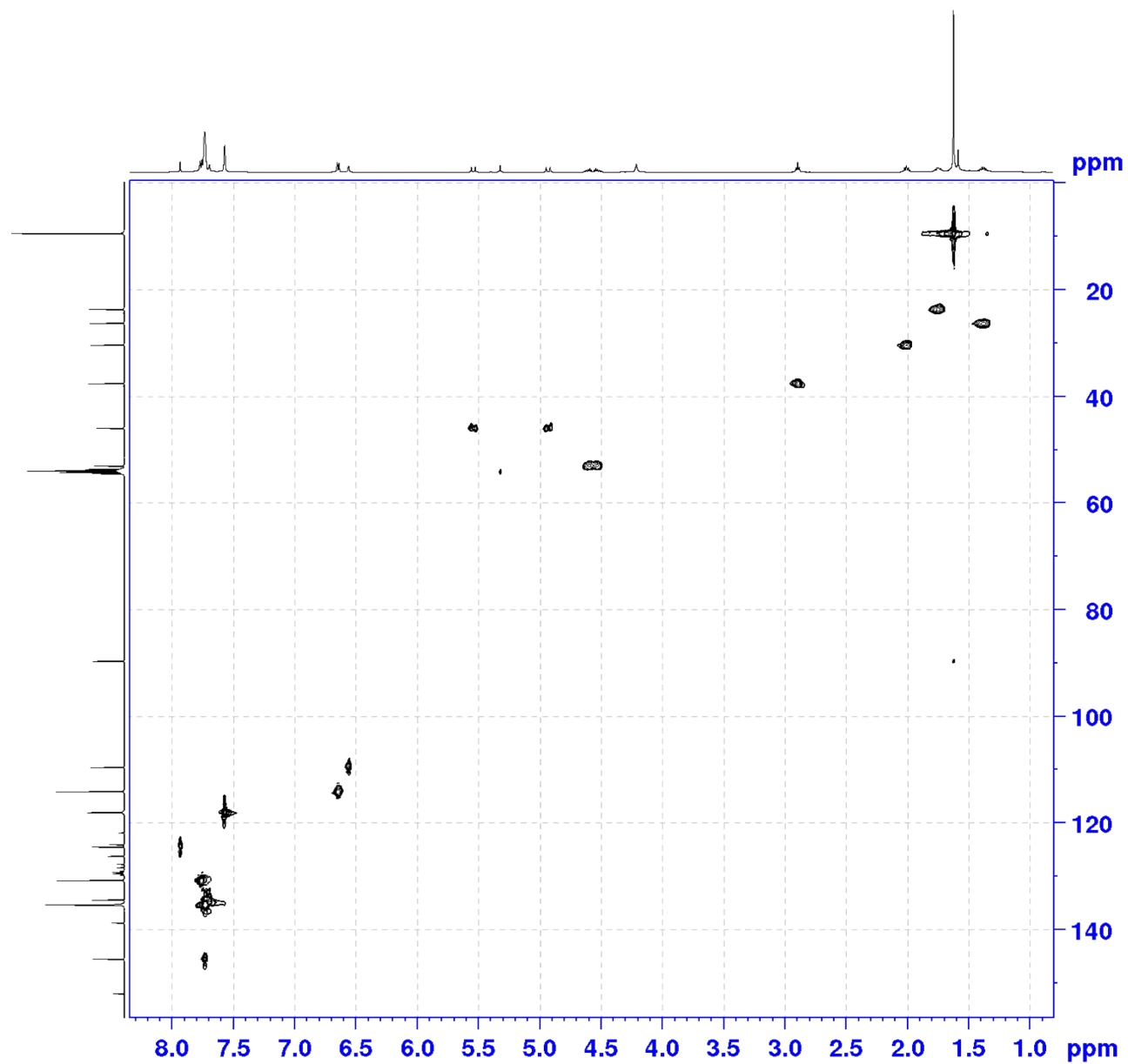


93

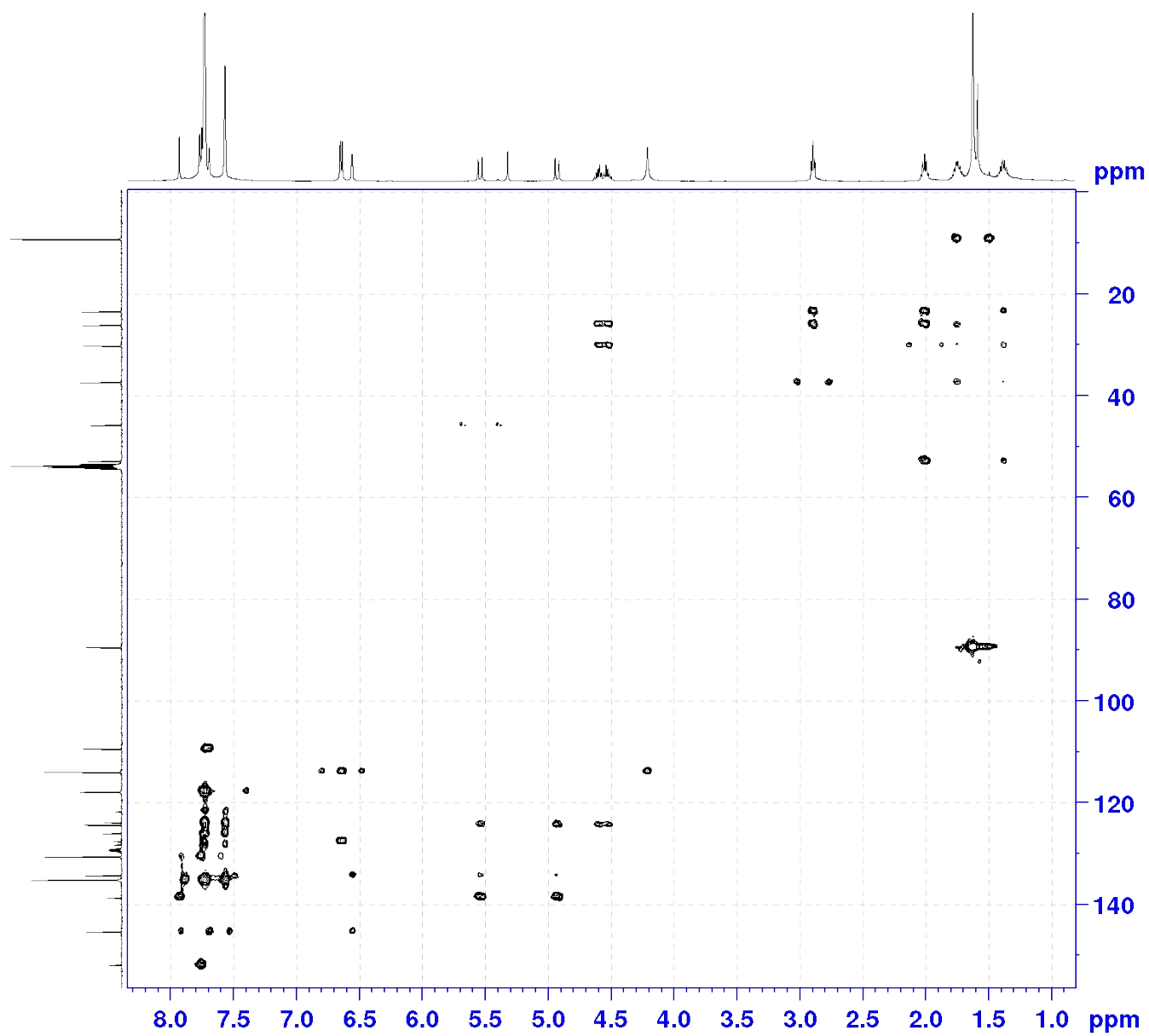
40

64

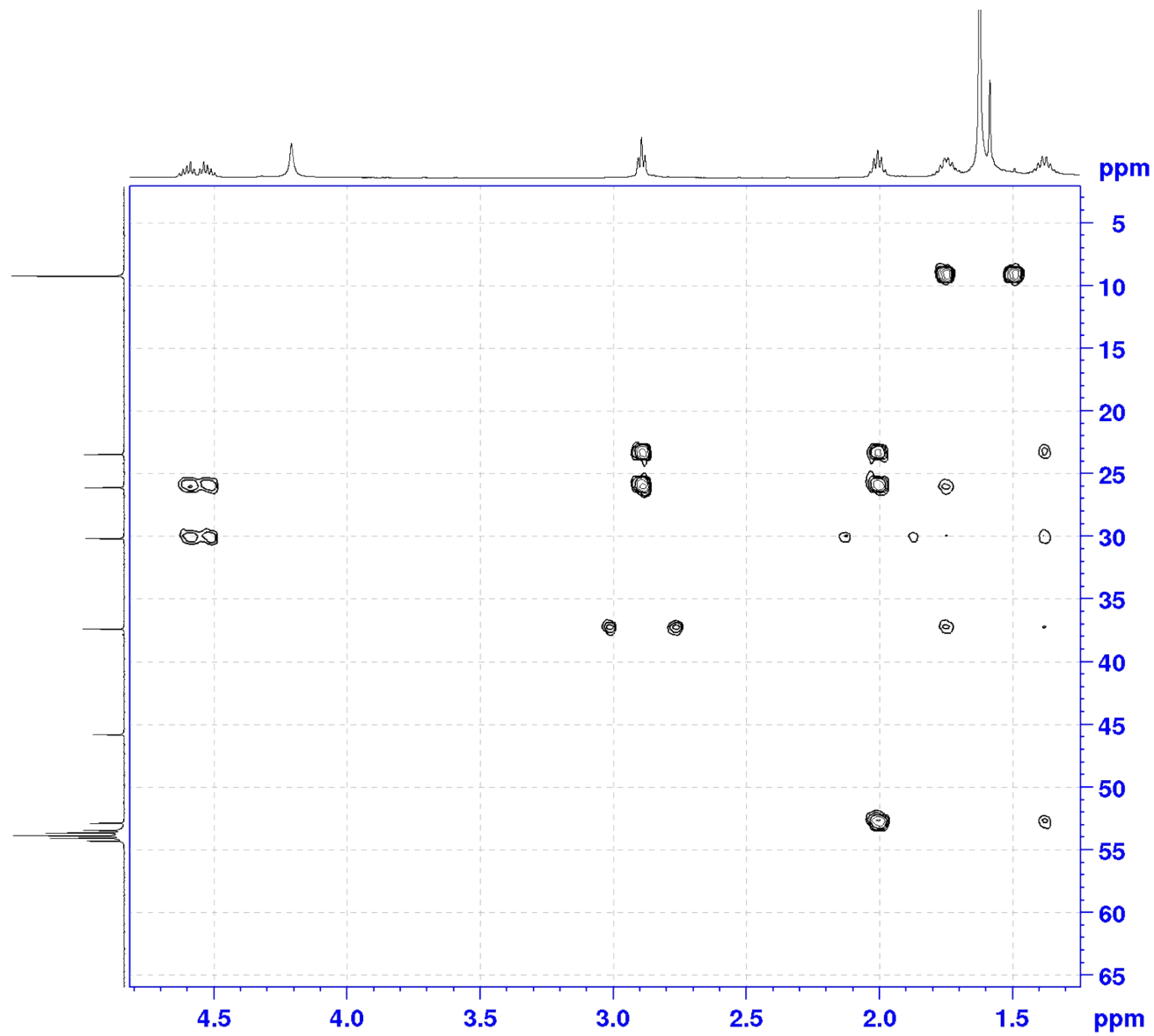
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08
38
92
75



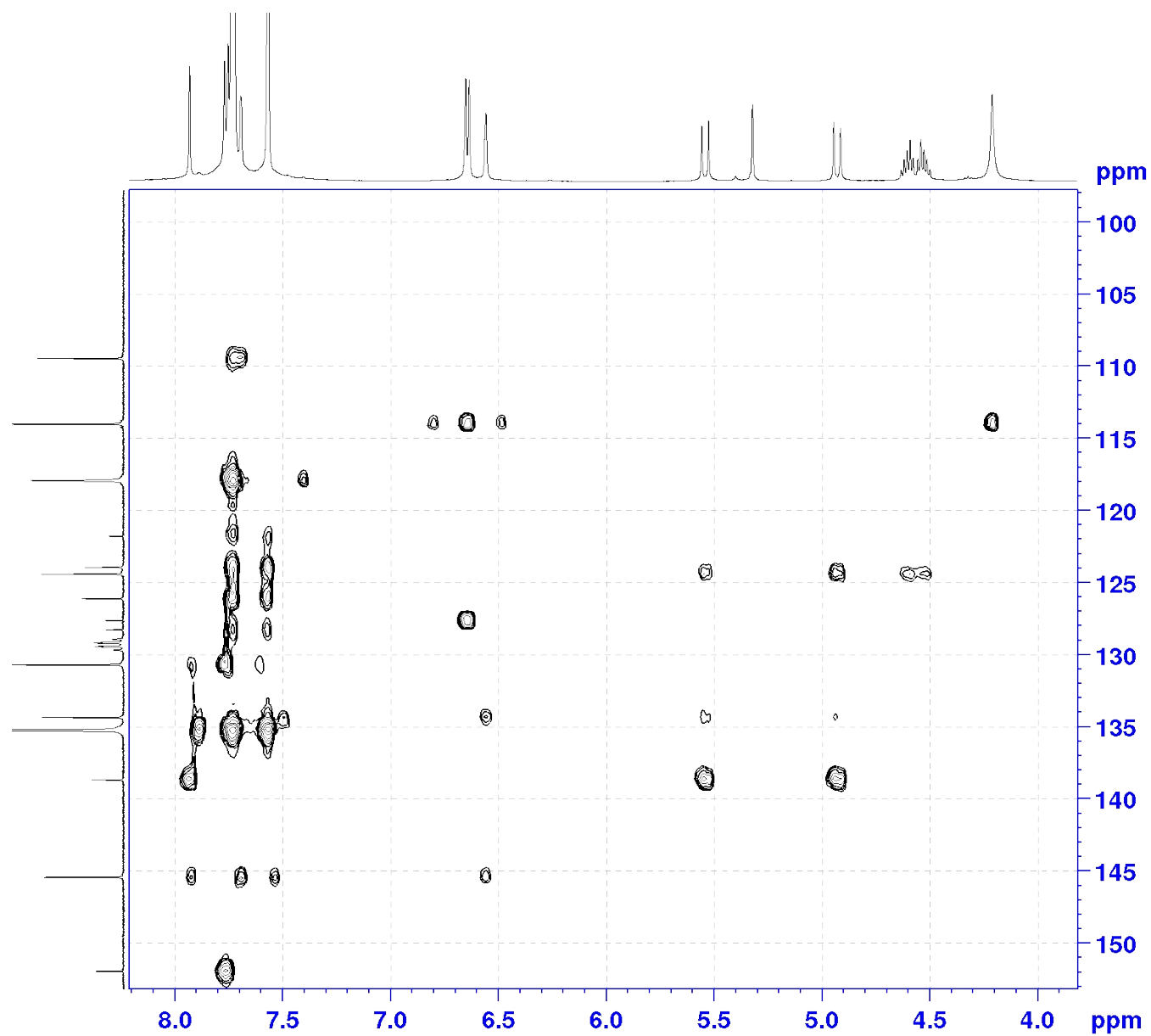
2D HSQC NMR spectrum of compound **Ir^{III}C₆** in CD₂Cl₂ and assignment of ¹H (red) and ¹³C (blue) NMR resonances.



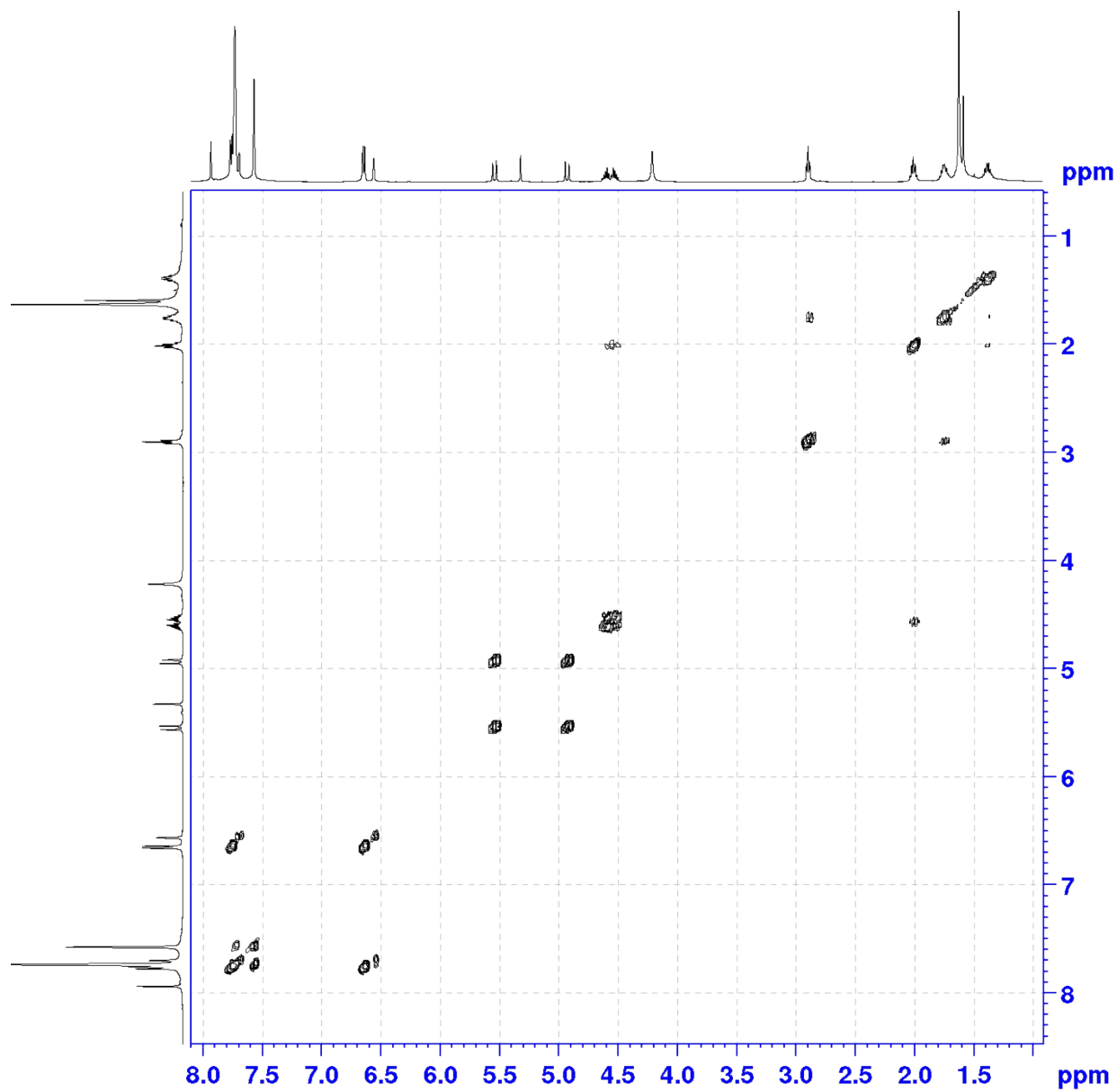
2D HMBC NMR spectrum of compound $\text{Ir}^{\text{III}}\text{C}_6$ in CD_2Cl_2 .



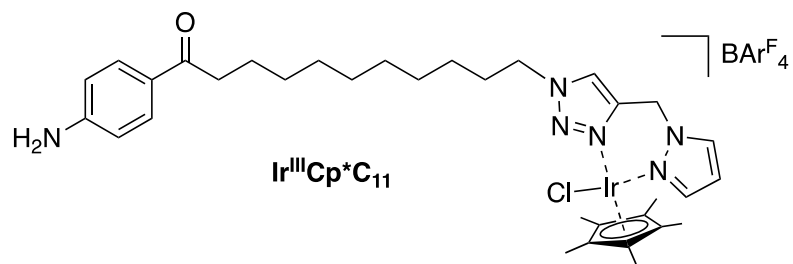
Selected ranges of the 2D HSQC NMR spectrum of compound $\text{Ir}^{\text{III}}\text{C}_6$ in CD_2Cl_2 .



Selected ranges of the 2D HSQC NMR spectrum of compound $\text{Ir}^{\text{III}}\text{C}_6$ in CD_2Cl_2 .



2D COSY NMR spectrum of compound **Ir^{III}C₆** in CD₂Cl₂.



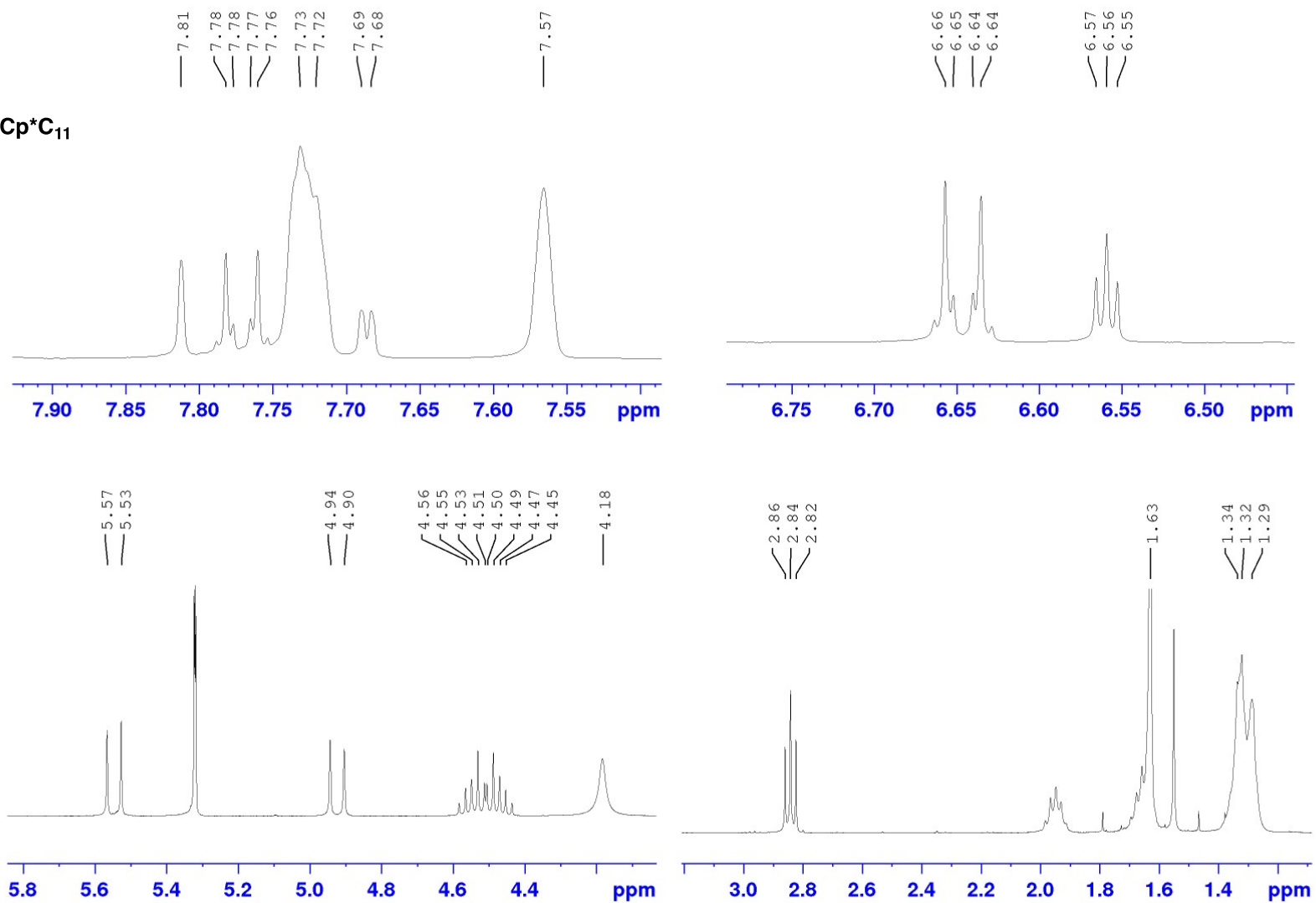
7.81
7.78
7.78
7.77
7.76
7.73
7.72
7.69
7.68
7.57
5.66
5.65
5.64
5.64
5.57
5.56
5.55
5.57
5.53
4.94
4.90
4.56
4.55
4.53
4.51
4.50
4.49
4.47
4.45
4.18

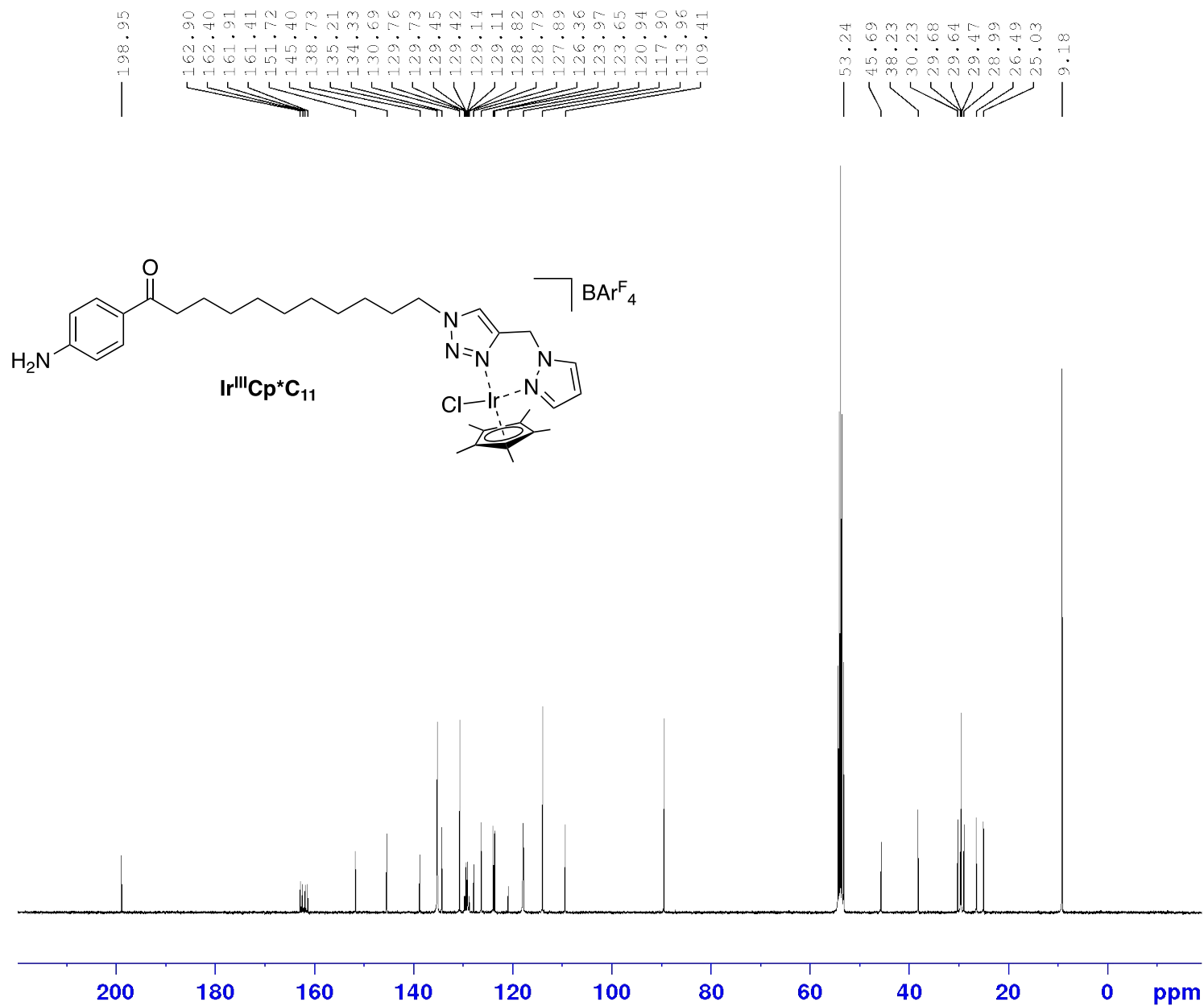
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2.84
2.82

1.63
1.34
1.32
1.29

Current Data Parameters
NAME MBL131

$\text{Ir}^{\text{III}}\text{Cp}^*\text{C}_{11}$

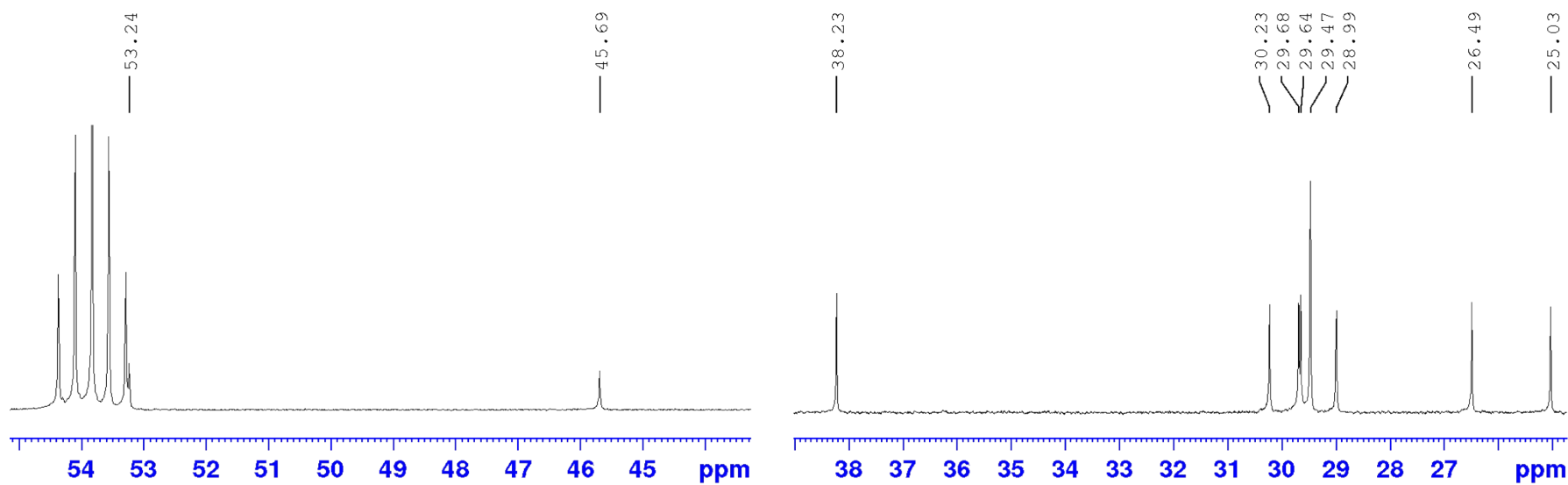
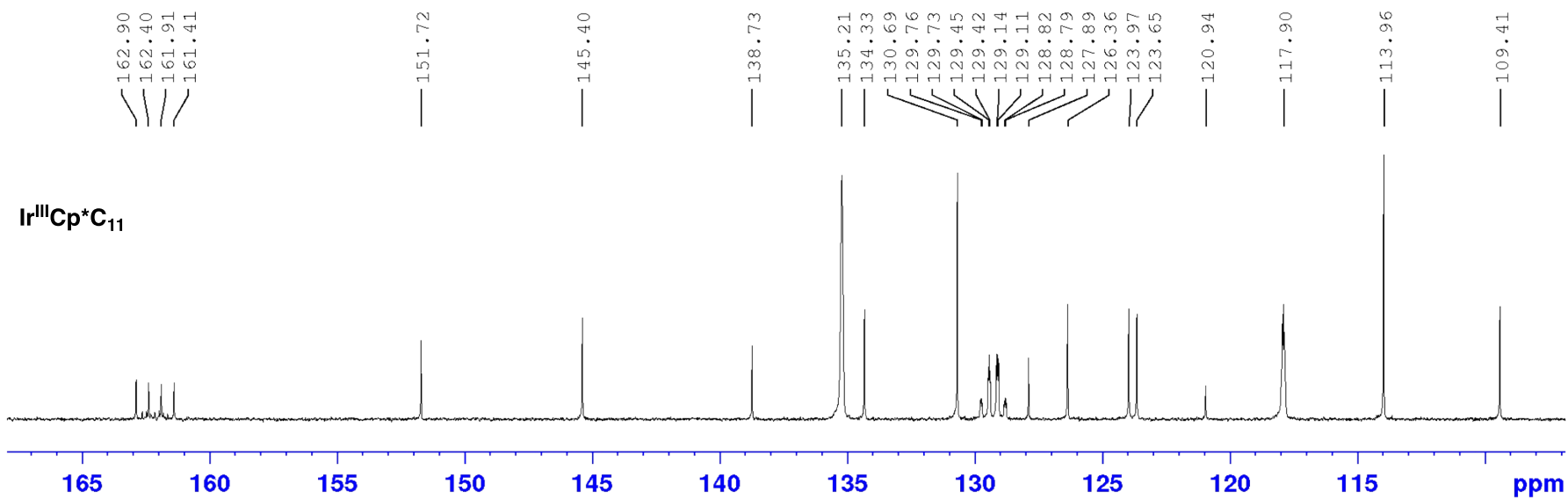




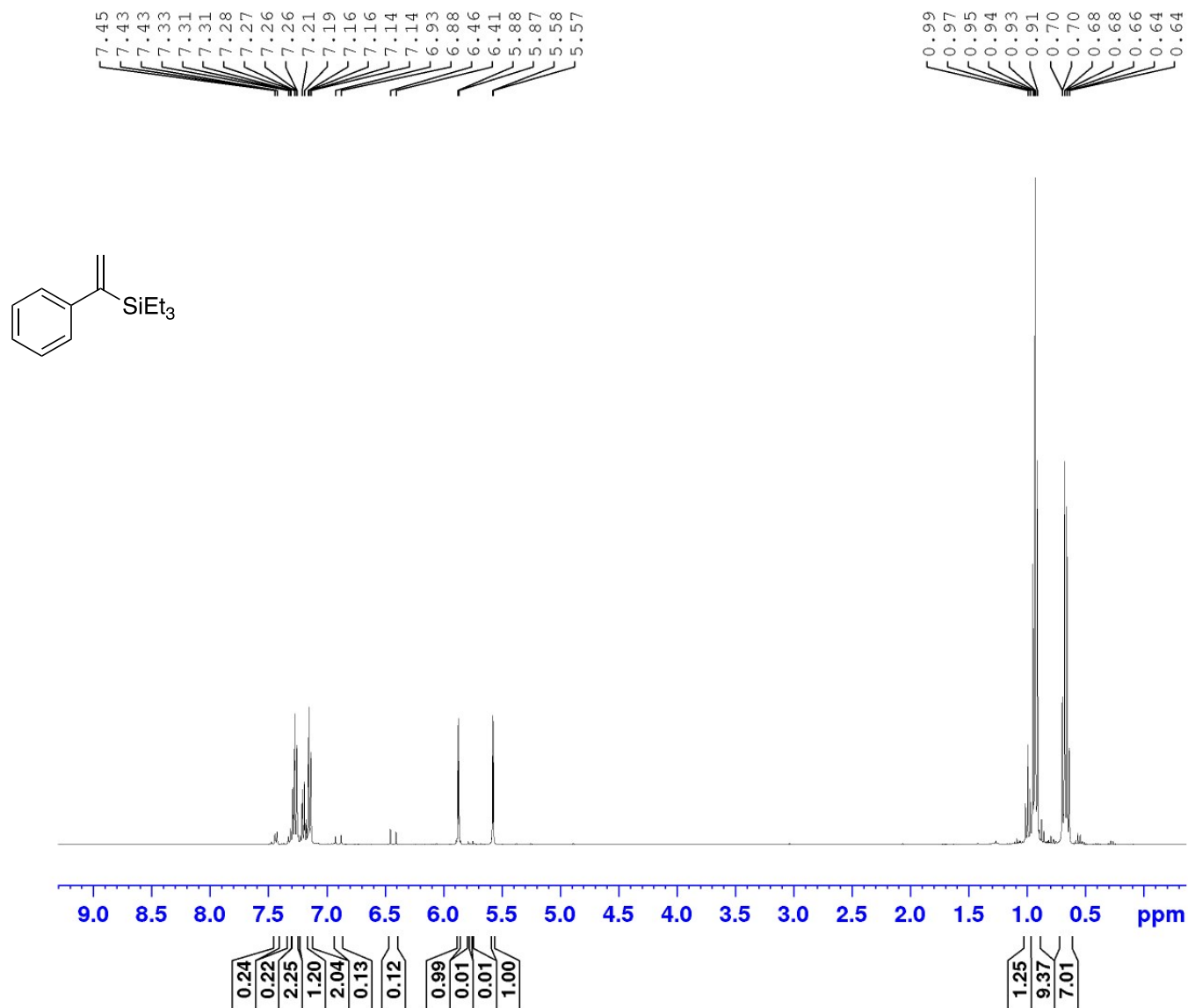
Current Data Parameters
NAME MR-131
EXPNO 3
PROCNO 1

F2 - Acquisition Parameters
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Time 3.09 h
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TD 65536
SOLVENT CD2Cl2
NS 6250
DS 4
SWH 24038.461 Hz
FIDRES 0.733596 Hz
AQ 1.3631488 sec
RG 193.41
DW 20.800 usec
DE 6.50 usec
TE 298.0 K
D1 2.00000000 sec
D11 0.03000000 sec
TD0 1
SFO1 100.6278593 MHz
NUC1 13C
P1 10.00 usec
PLW1 80.53199768 W
SFO2 400.1516006 MHz
NUC2 1H
CPDPRG[2] waltz16
PCPD2 90.00 usec
PLW2 15.75399971 W
PLW12 0.19449000 W
PLW13 0.09782900 W

F2 - Processing parameters
SI 32768
SF 100.6177560 MHz
WDW EM
SSE 0
LB 1.00 Hz
GB 0
PC 1.40



NMR Spectra of Catalysis Experiments



Current Data Parameters

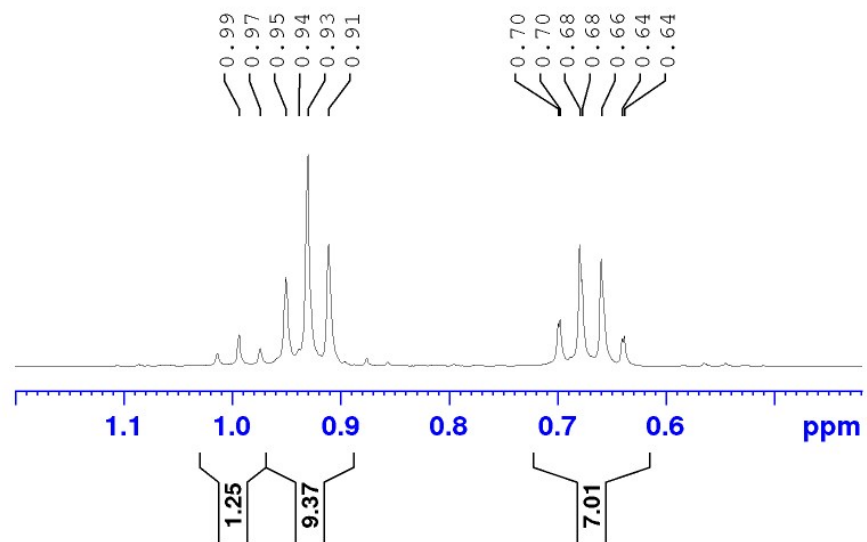
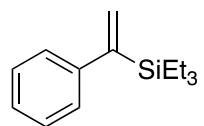
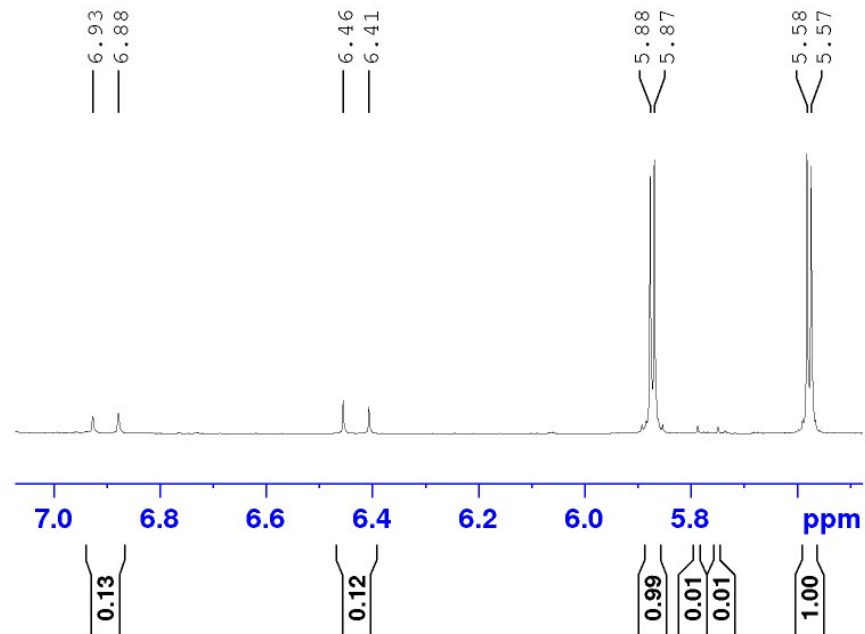
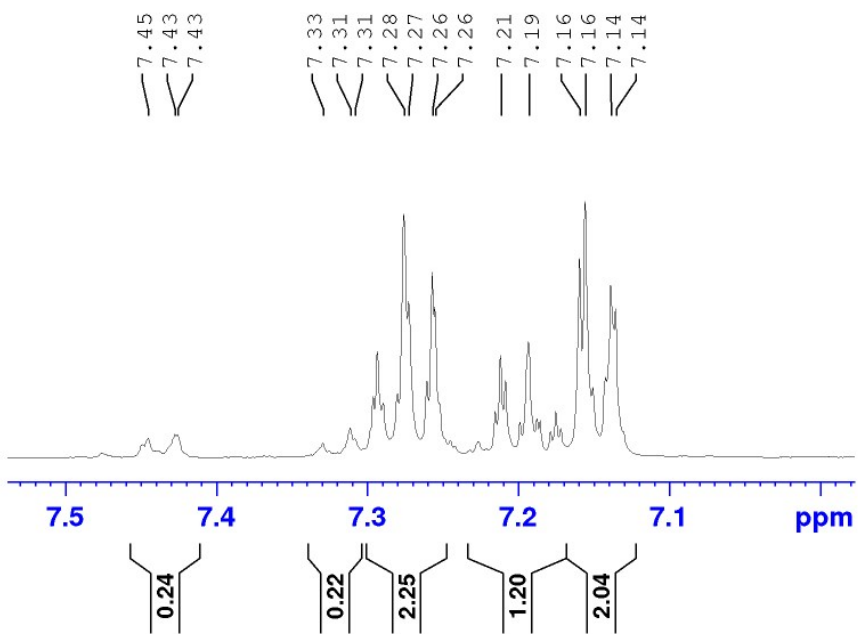
NAME MR-316
EXPNO 1
PROCNO 1

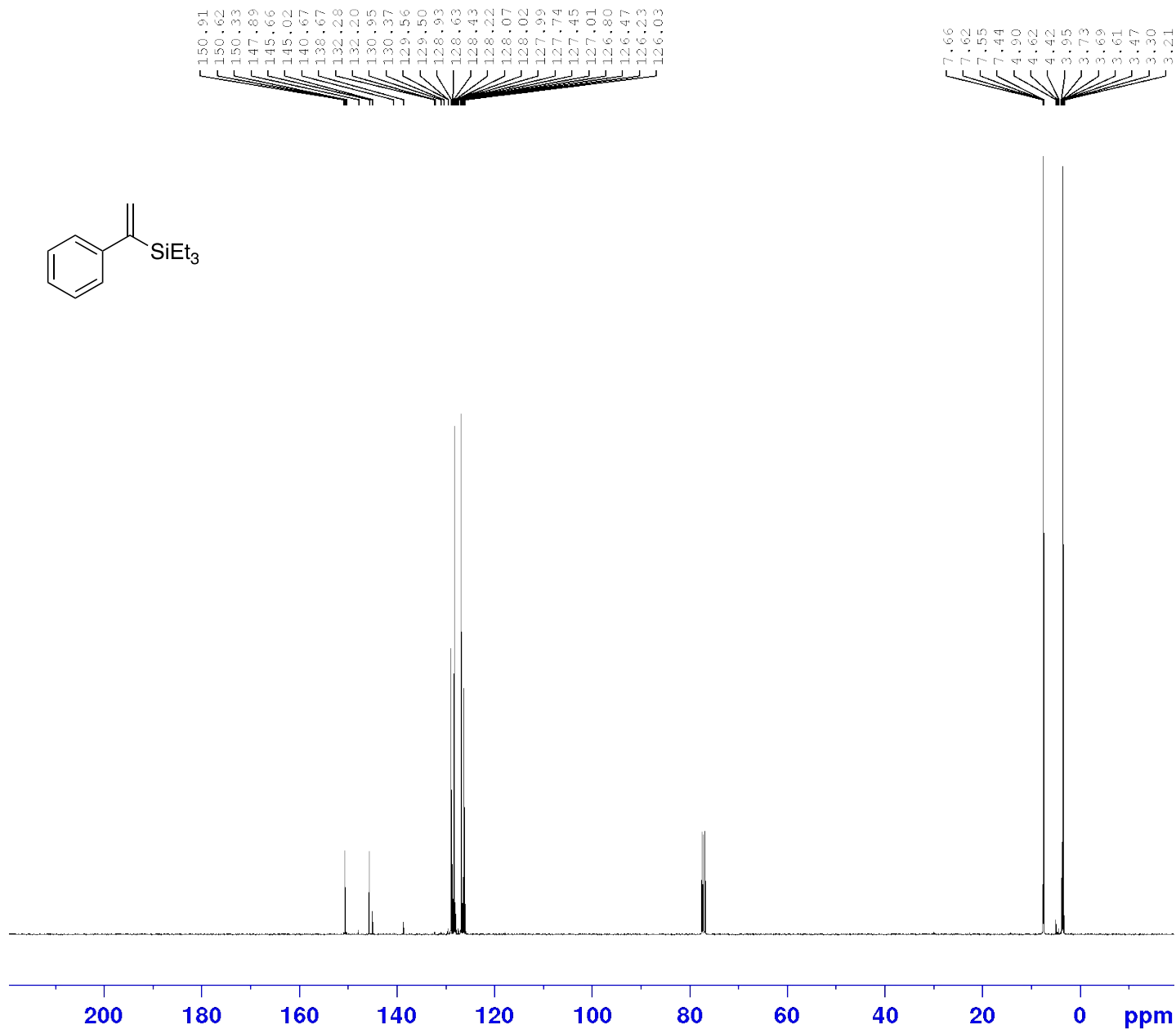
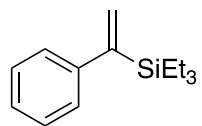
F2 - Acquisition Parameters

Date_ 20191113
Time 14.40 h
INSTRUM spect
PROBHD Z116098_0573 (
PULPROG zg30
TD 65536
SOLVENT CDCl3
NS 8
DS 2
SWH 8012.820 Hz
FIDRES 0.244532 Hz
AQ 4.0894465 sec
RG 16.03
DW 62.400 usec
DE 12.00 usec
TE 298.0 K
D1 2.50000000 sec
TD0 1
SFO1 400.1524709 MHz
NUC1 1H
P1 10.00 usec
PLW1 15.75399971 W

F2 - Processing parameters

SI 65536
SF 400.1500395 MHz
WDW EM
SSB 0
LB 0.30 Hz
GB 0
PC 1.00

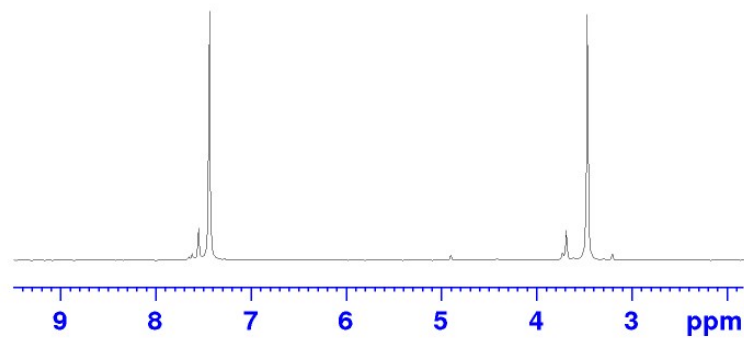
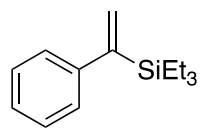
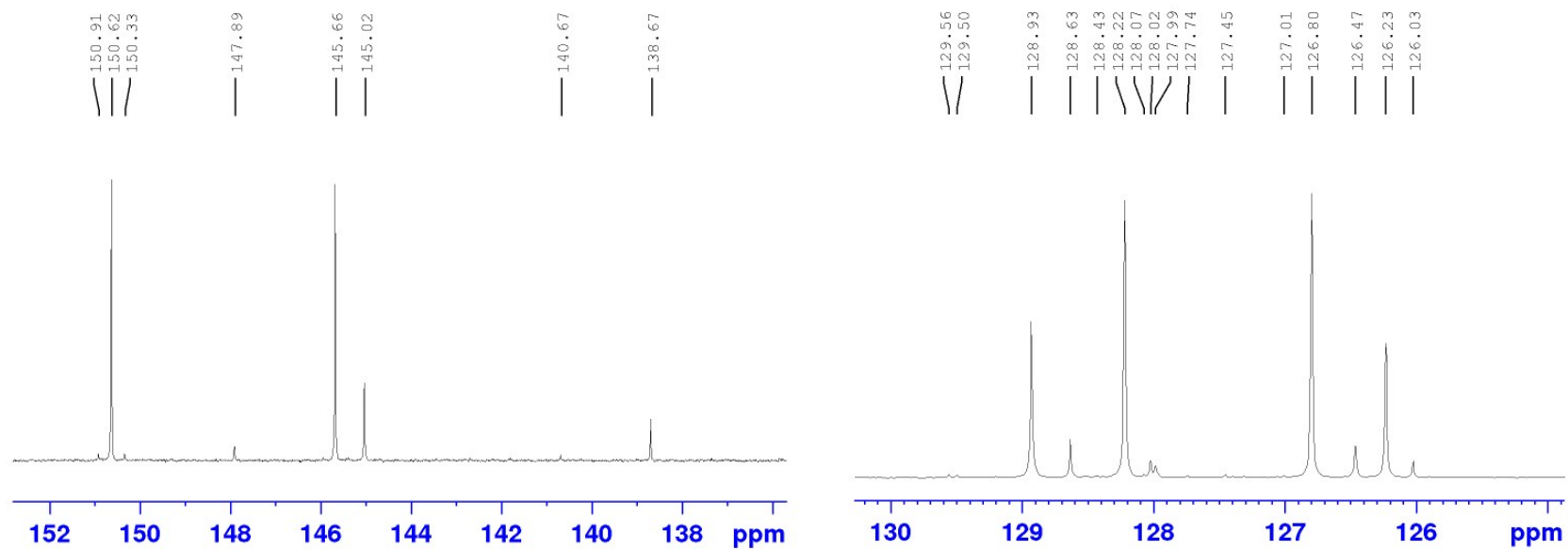


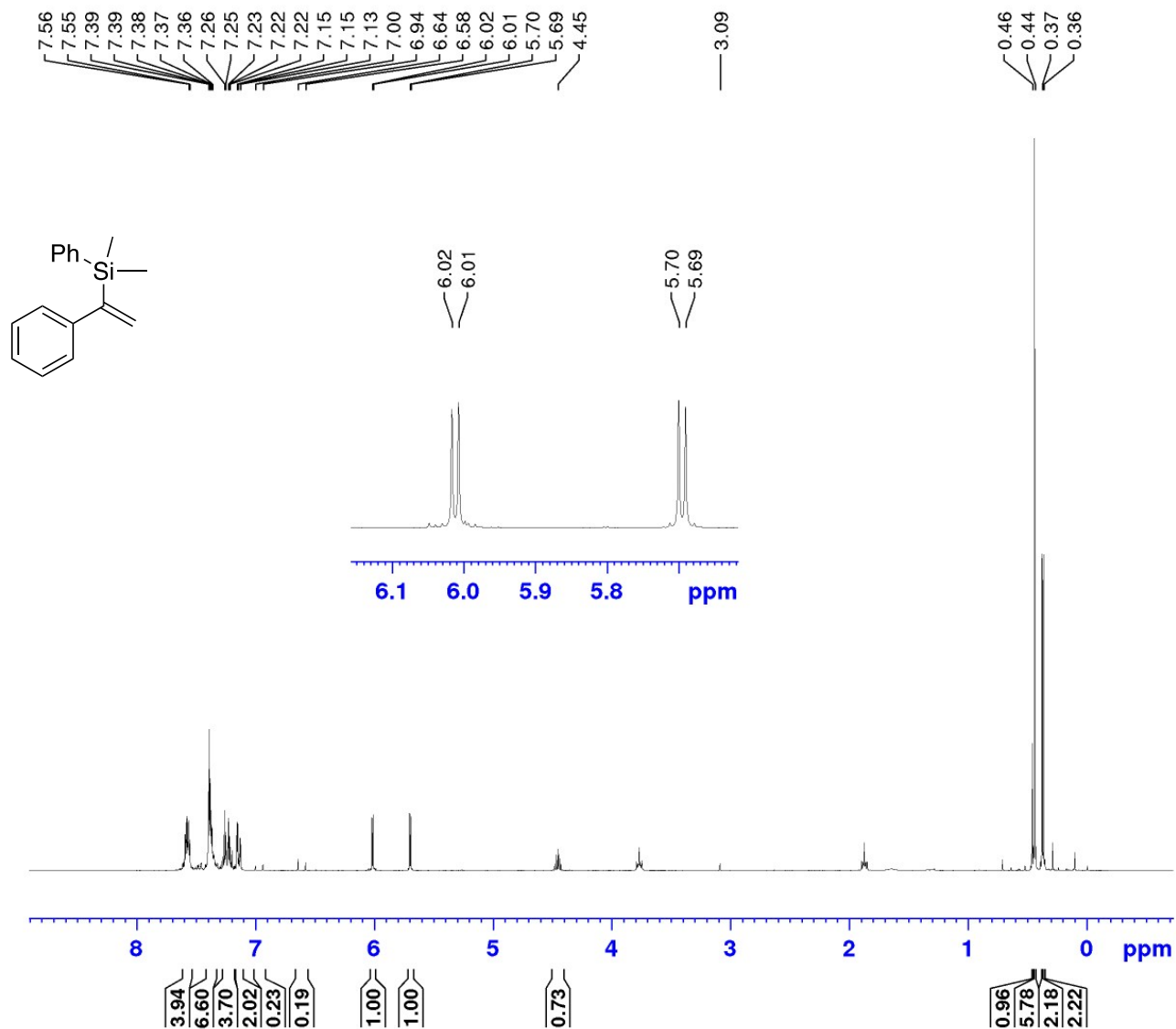


Current Data Parameters
NAME MR-316
EXPNO 2
PROCNO 1

F2 - Acquisition Parameters
Date_ 20191114
Time 2.27 h
INSTRUM spect
PROBHD Z116098_0573 (
PULPROG zgpg30
TD 65536
SOLVENT CDCl3
NS 2000
DS 4
SWH 24038.461 Hz
FIDRES 0.733596 Hz
AQ 1.3631488 sec
RG 198.41
DW 20.800 usec
DE 6.50 usec
TE 298.0 K
D1 2.00000000 sec
D11 0.03000000 sec
TD0 1
SFO1 100.6278593 MHz
NUC1 13C
P1 10.00 usec
PLW1 80.53199768 W
SFO2 400.1516006 MHz
NUC2 1H
CPDPRG[2] waltz16
PCPD2 90.00 usec
PLW2 15.75399971 W
PLW12 0.19449000 W
PLW13 0.09782900 W

F2 - Processing parameters
SI 32768
SF 100.6177888 MHz
WDW EM
SSB 0
LB 1.00 Hz
GB 0
PC 1.40

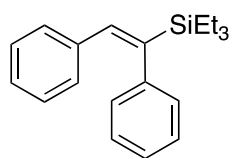




Current Data Parameters
NAME mr-66
EXPNO 1
PROCNO 1

F2 - Acquisition Parameters
Date_ 20200604
Time 17.10 h
INSTRUM NEO_SPECT
PROBHD Z104275_0460 (
PULPROG zg30
TD 65536
SOLVENT CDCl3
NS 16
DS 2
SWH 5882.353 Hz
FIDRES 0.179515 Hz
AQ 5.5705600 sec
RG 101
DW 85.000 usec
DE 8.24 usec
TE -7.8 K
D1 2.00000000 sec
TD0 1
SFO1 300.1319508 MHz
NUC1 1H
P0 4.67 usec
P1 14.00 usec
PLW1 8.82489967 W

F2 - Processing parameters
SI 65536
SF 300.1300073 MHz
WDW EM
SSB 0
LB 0.30 Hz
GB 0
PC 1.00



7.40
7.37
7.19
7.19
7.18
7.17
7.14
7.14
7.11
7.11
7.09
6.90

1.11
1.08
1.06
0.79
0.78
0.76

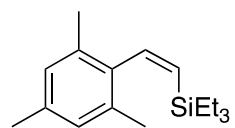
Current Data Parameters
NAME: 88-55



Current Data Parameters
NAME mr-55
EXPNO 3
PROCNO 1

F2 - Acquisition Parameters
Date_ 20200526
Time 16.55 h
INSTRUM NEO_SPECT
PROBHD Z104275_0460 (
PULPROG zgpg30
TD 65536
SOLVENT CDCl3
NS 256
DS 2
SWH 20000.000 Hz
FIDRES 0.610352 Hz
AQ 1.6384000 sec
RG 101
DW 25.000 usec
DE 7.67 usec
TE 298.1 K
D1 2.00000000 sec
D11 0.03000000 sec
TD0 1
SFO1 75.4768047 MHz
NOC1 13C
P0 3.33 usec
P1 10.00 usec
PLW1 34.53799820 W
SFO2 300.1319508 MHz
NOC2 1H
CPDPRG[2] waltz16
PCPD2 90.00 usec
PLW2 8.82489967 W
PLW12 0.21354000 W
PLW13 0.10741000 W

F2 - Processing parameters
SI 65536
SF 75.4677485 MHz
WDW EM
SSB 0
LB 1.00 Hz
GB 0
PC 1.40

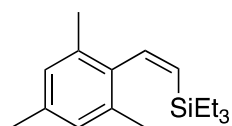


7b

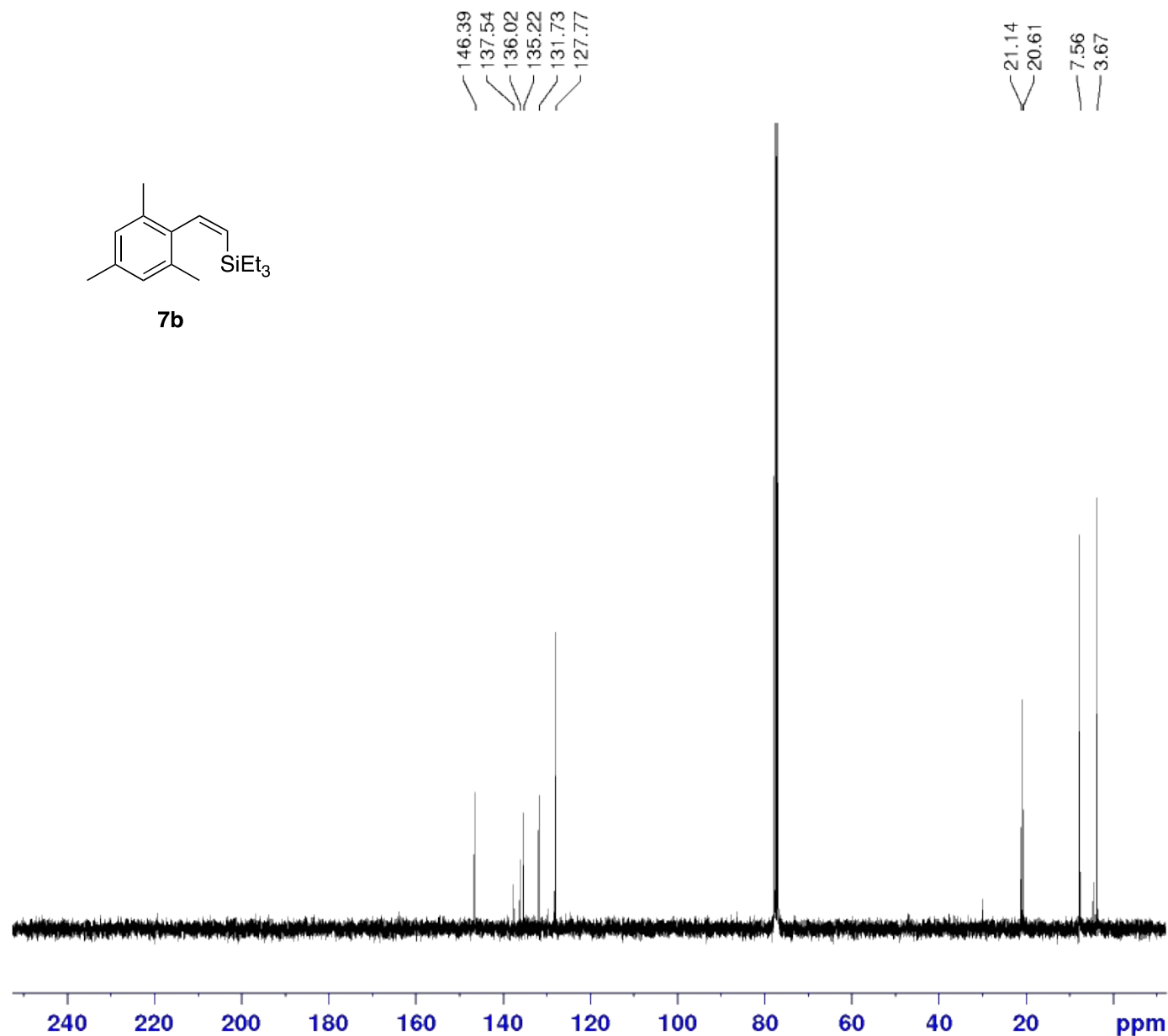
7.24
7.19
6.81
5.99
5.94
5.77
5.76
5.65
5.64

2.34
2.27
2.26
2.18
2.12
1.55
1.26
0.94
0.91
0.88
0.85
0.84
0.83
0.82
0.81
0.80
0.80
0.79
0.62

Current Data Parameters
NAME: mpr-80_2



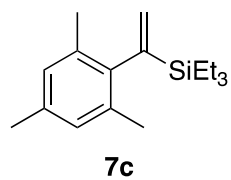
7b



Current Data Parameters
 NAME mr-90_2
 EXPNO 2
 PROCNO 1

F2 - Acquisition Parameters
 Date_ 20200624
 Time 14.45 h
 INSTRUM NEO_SPECT
 PROBHD Z104275_0460 (
 PULPROG zgpg30
 TD 65536
 SOLVENT CDCl3
 NS 450
 DS 2
 SWH 20000.000 Hz
 FIDRES 0.610352 Hz
 AQ 1.6384000 sec
 RG 101
 DW 25.000 usec
 DE 7.67 usec
 TE -7.8 K
 D1 2.00000000 sec
 D11 0.03000000 sec
 TD0 1
 SFO1 75.4768047 MHz
 NUC1 13C
 P0 3.33 usec
 P1 10.00 usec
 PLW1 34.53799820 W
 SFO2 300.1319508 MHz
 NUC2 1H
 CPDPRG[2] waltz65
 PCPD2 90.00 usec
 PLW2 8.82489967 W
 PLW12 0.21354000 W
 PLW13 0.10741000 W

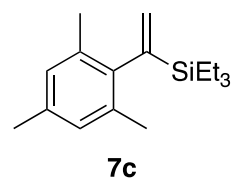
F2 - Processing parameters
 SI 65536
 SF 75.4677377 MHz
 WDW EM
 SSB 0
 LB 1.00 Hz
 GB 0
 PC 1.40



6.84
6.84
5.79
5.78
5.67
5.66

2.27
2.14
-0.96
-0.95
-0.93
-0.93
-0.90
-0.67
-0.67
-0.64
-0.64
-0.62
-0.59
-0.59

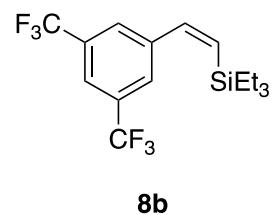
Current Data Parameters
NAME mp-79



51.11
41.85
34.75
34.06
29.49
28.12

1.01
.45
.41

Current Data Parameters
NAME mr-79

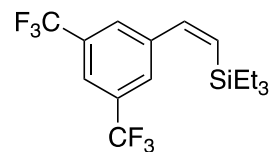


7.77
7.73
7.46
7.42

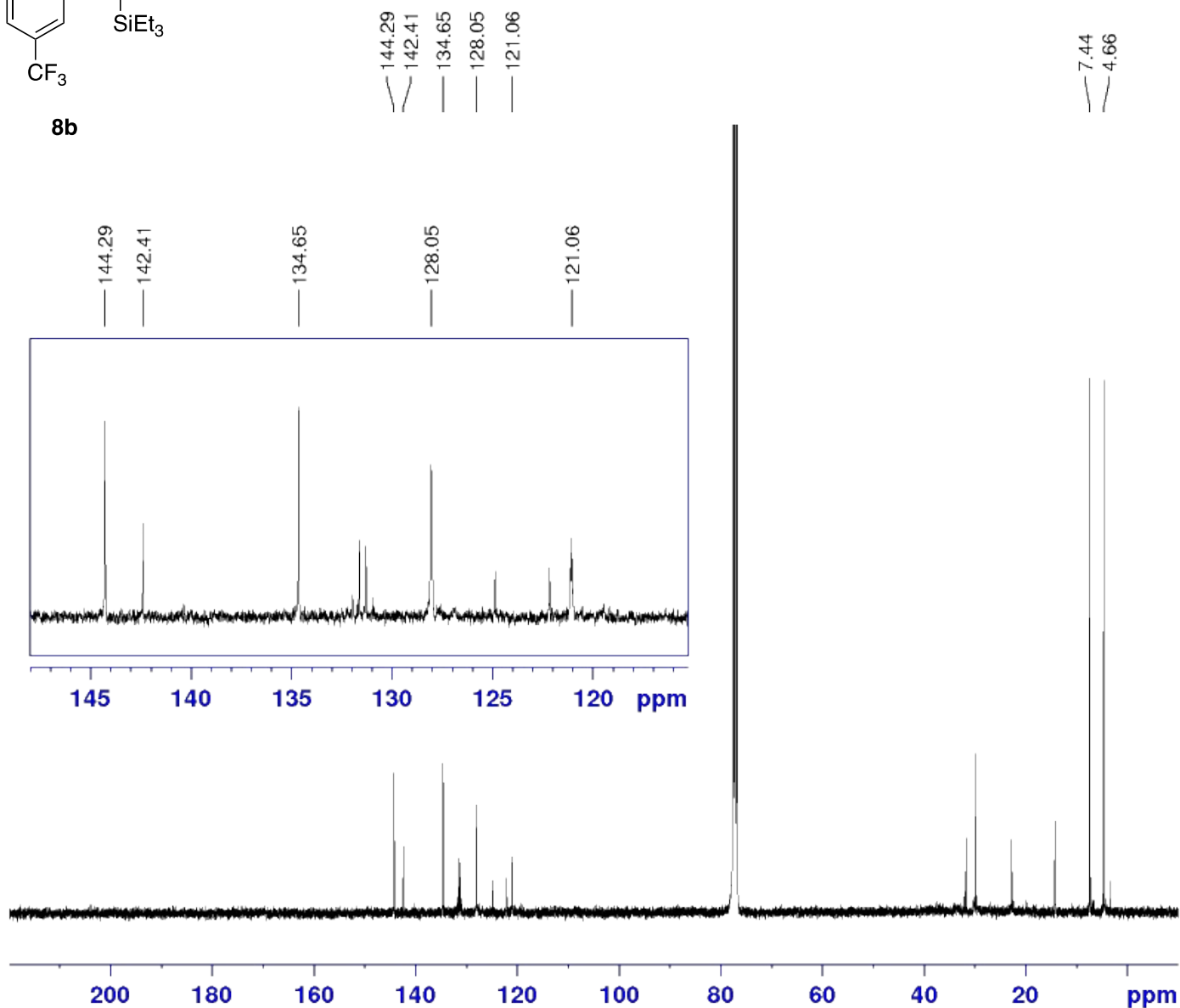
6.05
6.01

1.54
1.26
0.89
0.88
0.87
0.85
0.55
0.54

Current Data Parameters



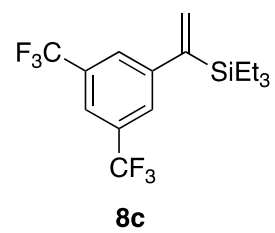
8b



Current Data Parameters
 NAME mr-103
 EXPNO 3
 PROCNO 1

F2 - Acquisition Parameters
 Date_ 20200707
 Time 6.41 h
 INSTRUM spect
 PROBHD Z108618_0516 ()
 PULPROG zgpg30
 TD 65536
 SOLVENT CDCl3
 NS 13312
 DS 2
 SWH 27173.912 Hz
 FIDRES 0.829282 Hz
 AQ 1.2058624 sec
 RG 203
 DW 18.400 usec
 DE 8.65 usec
 TE 300.1 K
 D1 2.00000000 sec
 D11 0.03000000 sec
 TD0 13
 SFO1 100.6248421 MHz
 NUC1 13C
 P0 3.27 usec
 P1 9.80 usec
 PLW1 77.98300171 W
 SFO2 400.1326008 MHz
 NUC2 1H
 CPDPRG[2] waltz16
 PCPD2 90.00 usec
 PLW2 11.99499989 W
 PLW12 0.32877001 W
 PLW13 0.16537000 W

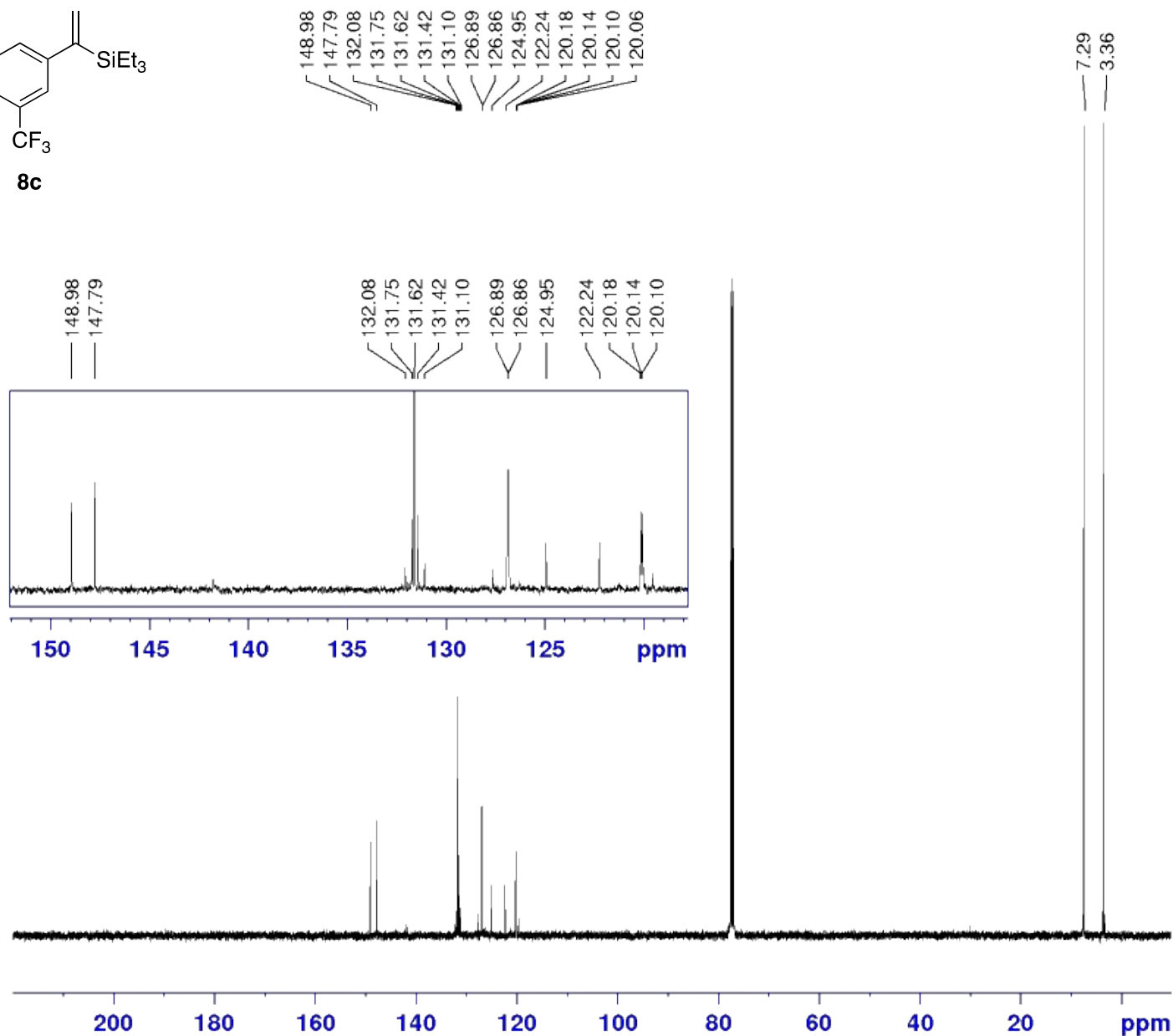
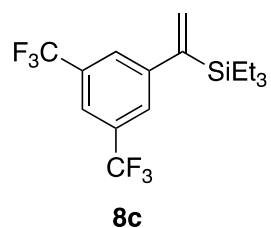
F2 - Processing parameters
 SI 65536
 SF 100.6127518 MHz
 WDW EM
 SSB 0
 LB 1.00 Hz
 GB 0
 PC 1.40



7.84
7.73
7.56
5.97
5.90
5.66
5.60
5.95
5.95
5.74
5.73

0.96
0.94
0.91
0.71
0.71
0.69
0.68
0.66

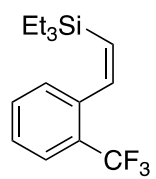
Current Data Parameters



Current Data Parameters
 NAME mr-94_2
 EXPNO 1
 PROCNO 1

F2 - Acquisition Parameters
 Date_ 20200625
 Time 15.58 h
 INSTRUM spect
 PROBHD Z108618_0516 {
 PULPROG zgpg30
 TD 65536
 SOLVENT CDCl3
 NS 650
 DS 2
 SWH 27173.912 Hz
 FIDRES 0.829282 Hz
 AQ 1.2058624 sec
 RG 203
 DW 18.400 usec
 DE 8.65 usec
 TE 300.2 K
 D1 2.00000000 sec
 D11 0.03000000 sec
 TD0 1
 SFO1 100.6248421 MHz
 NUC1 13C
 P0 3.27 usec
 P1 9.80 usec
 PLW1 77.98300171 W
 SFO2 400.1326008 MHz
 NUC2 1H
 CPDPRG[2] waltz65
 PCPD2 90.00 usec
 PLW2 11.99499989 W
 PLW12 0.32877001 W
 PLW13 0.16537000 W

F2 - Processing parameters
 SI 65536
 SF 100.6127504 MHz
 WDW EM
 SSB 0
 LB 1.00 Hz
 GB 0
 PC 1.40

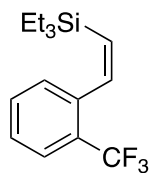


9b

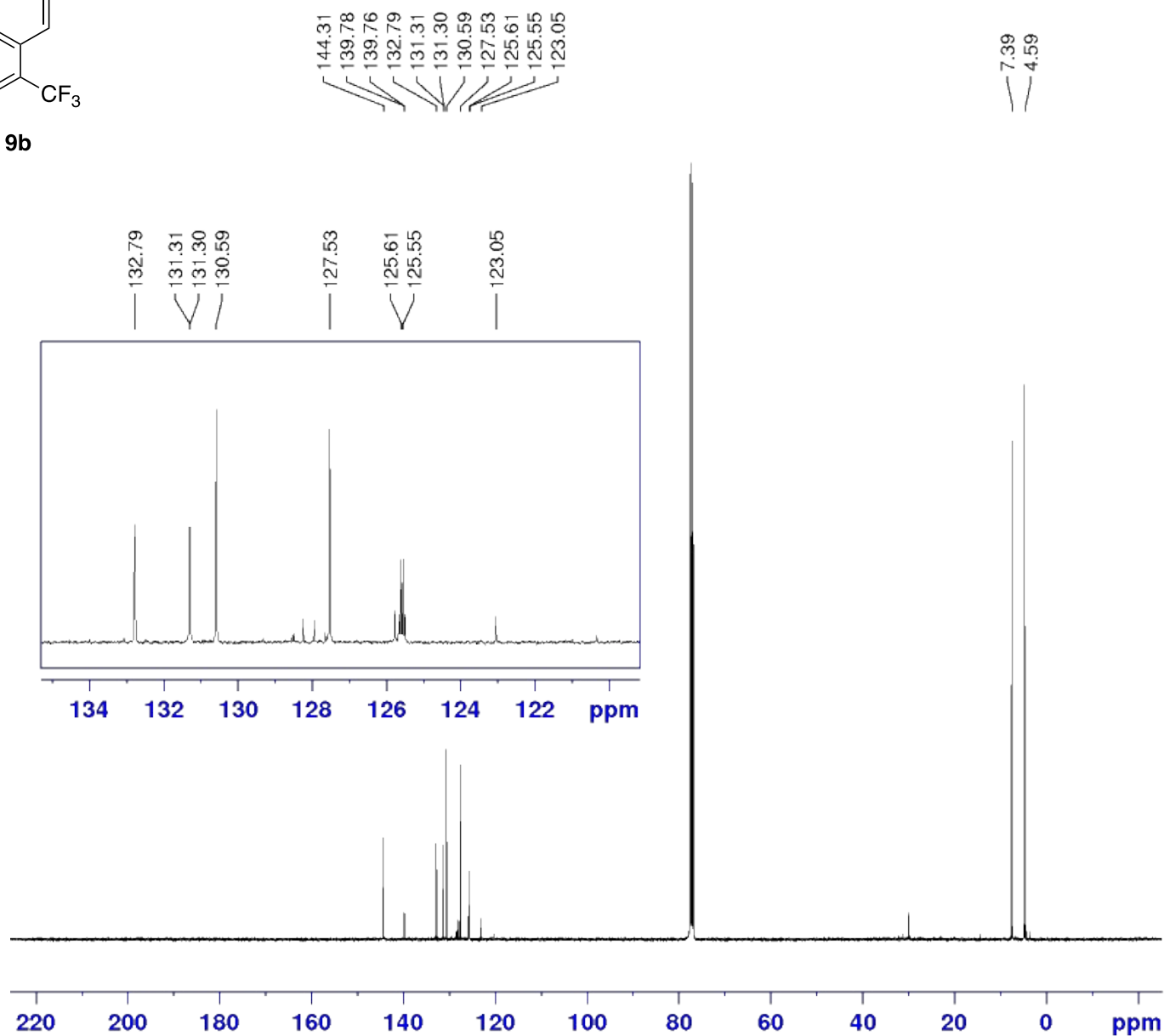
7.65
7.64
7.63
7.63
7.63
7.61
7.60
7.48
7.48
7.48
7.45
7.45
7.45
7.40
7.40
7.37
7.37
7.34
7.33
7.31
7.31
7.26
5.97
5.92

0.86
0.84
0.81
0.46
0.46
0.43
0.41
0.38
0.38

Current Data Parameters



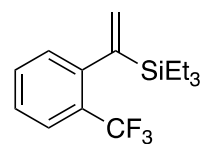
9b



Current Data Parameters
 NAME MR-98
 EXPNO 2
 PROCNO 1

F2 - Acquisition Parameters
 Date_ 20200630
 Time 1.37 h
 INSTRUM spect
 PROBHD Z108618_0516 {
 PULPROG zgpg30
 TD 65536
 SOLVENT CDCl3
 NS 7168
 DS 4
 SWH 25252.525 Hz
 FIDRES 0.770646 Hz
 AQ 1.2976128 sec
 RG 203
 DW 19.800 usec
 DE 6.50 usec
 TE 300.1 K
 D1 2.00000000 sec
 D11 0.03000000 sec
 TD0 7
 SFO1 100.6228303 MHz
 NUC1 13C
 P0 3.27 usec
 P1 9.80 usec
 PLW1 77.98300171 W
 SFO2 400.1316005 MHz
 NUC2 1H
 CPDPRG[2] waltz16
 PCPD2 90.00 usec
 PLW2 11.99499989 W
 PLW12 0.32877001 W
 PLW13 0.16537000 W

F2 - Processing parameters
 SI 65536
 SF 100.6127521 MHz
 WDW EM
 SSB 0
 LB 1.00 Hz
 GB 0
 PC 1.40

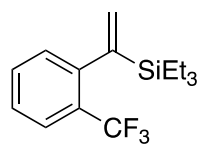


9c

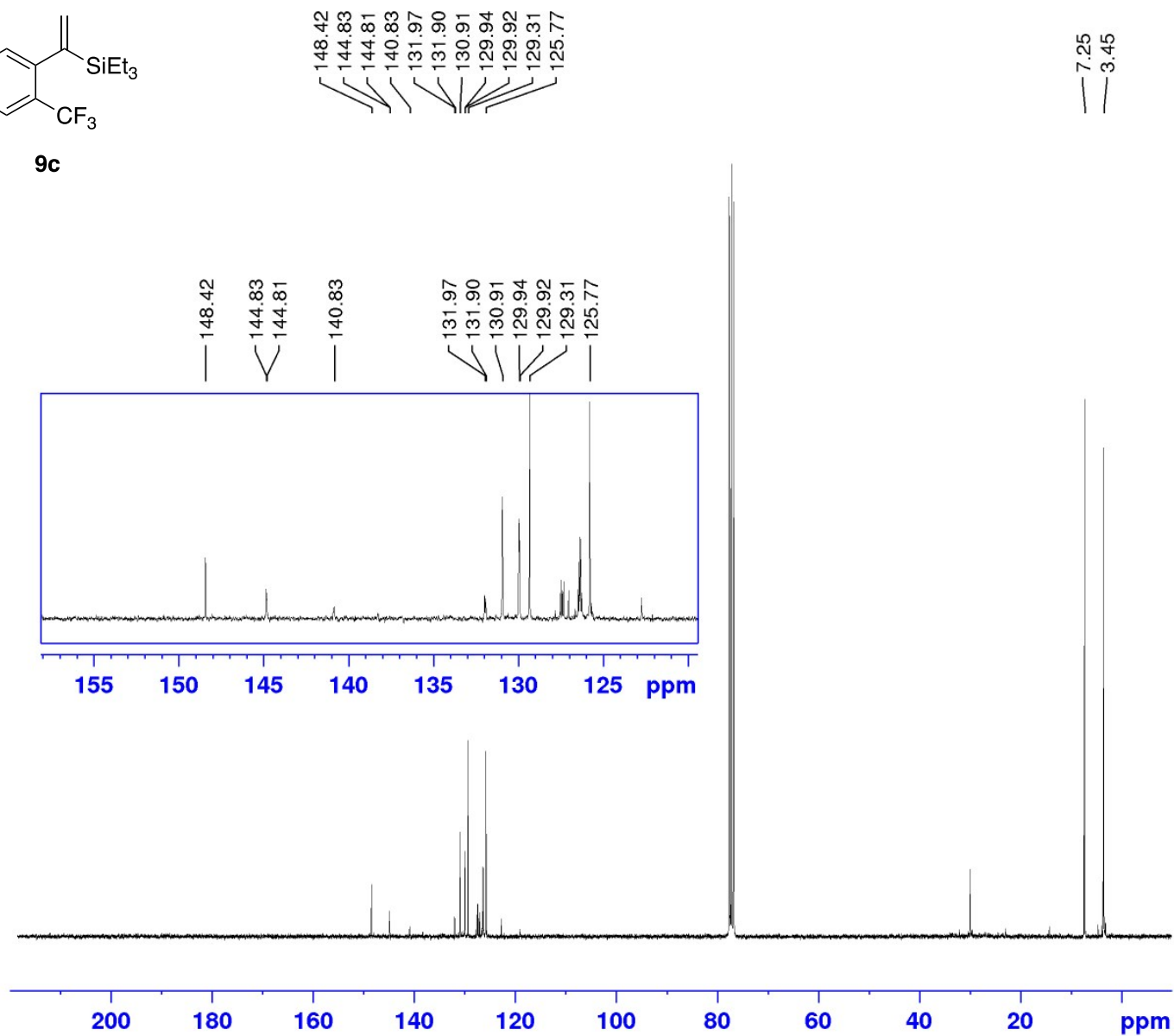
7.64
 7.62
 7.45
 7.45
 7.42
 7.42
 7.40
 7.40
 7.30
 7.28
 7.03
 7.01
 5.76
 5.76
 5.75
 5.75
 5.75
 5.72
 5.71

0.93
 0.90
 0.87
 0.67
 0.65
 0.65
 0.63
 0.60
 0.57
 0.57

Current Data Parameters
 NAME mr-83
 8/1/2015



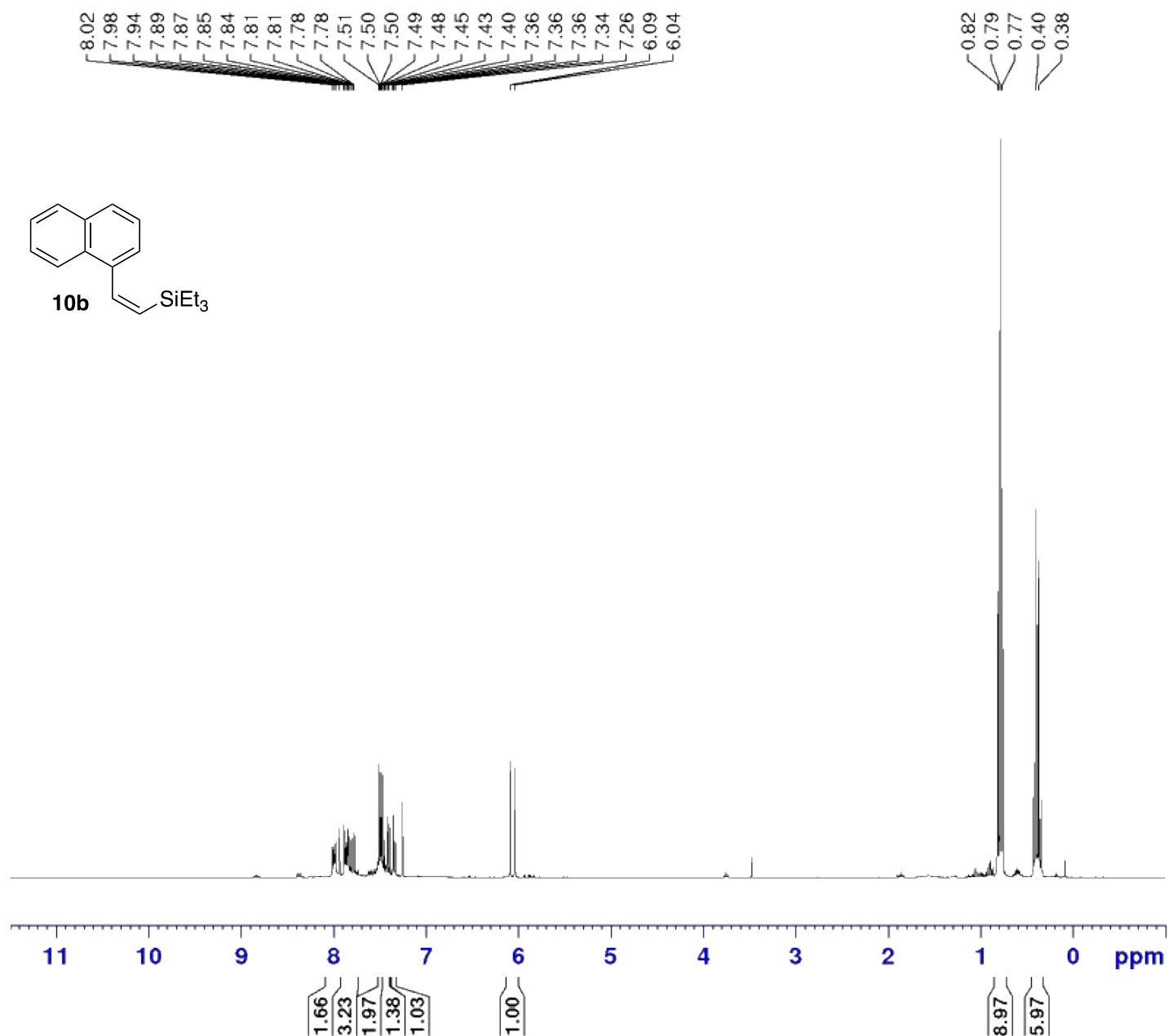
9c



Current Data Parameters
NAME mr-83_2
EXPNO 3
PROCNO 1

F2 - Acquisition Parameters
Date_ 20200618
Time 0.30 h
INSTRUM NEO_SPECT
PROBHD Z104275_0460 (
PULPROG zgpg30
TD 65536
SOLVENT CDCl3
NS 6000
DS 4
SWH 17857.143 Hz
FIDRES 0.544957 Hz
AQ 1.8350080 sec
RG 101
DW 28.000 usec
DE 6.50 usec
TE -7.8 K
D1 2.00000000 sec
D11 0.03000000 sec
TD0 1
SFO1 75.4752953 MHz
NUC1 13C
P0 3.33 usec
P1 10.00 usec
PLW1 34.53799820 W
SFO2 300.1312005 MHz
NUC2 1H
CPDPRG[2] waltz65
PCPD2 90.00 usec
PLW2 8.82489967 W
PLW12 0.21354000 W
PLW13 0.10741000 W

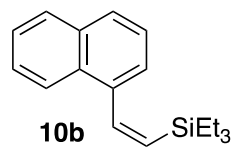
F2 - Processing parameters
SI 32768
SF 75.4677371 MHz
WDW EM
SSB 0
LB 1.00 Hz
GB 0
PC 1.40



Current Data Parameters
 NAME mr-91
 EXPNO 1
 PROCNO 1

F2 - Acquisition Parameters
 Date_ 20200622
 Time 15.50 h
 INSTRUM NEO_SPECT
 PROBHD Z104275_0460 {
 PULPROG zg30
 TD 65536
 SOLVENT CDCl3
 NS 16
 DS 2
 SWH 5882.353 Hz
 FIDRES 0.179515 Hz
 AQ 5.5705600 sec
 RG 101
 DW 85.000 usec
 DE 8.24 usec
 TE -7.8 K
 D1 2.00000000 sec
 TD0 1
 SFO1 300.1319508 MHz
 NUC1 1H
 P0 4.67 usec
 P1 14.00 usec
 PLW1 8.82489967 W

F2 - Processing parameters
 SI 65536
 SF 300.1300073 MHz
 WDW EM
 SSB 0
 LB 0.30 Hz
 GB 0
 PC 1.00

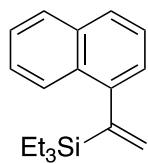


3.19
3.52
3.33
1.93
1.76
3.39
7.96
5.93
5.89
5.65
5.32
5.19

87

3
8

Current Data Parameters

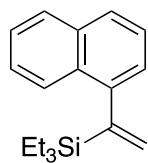


10c

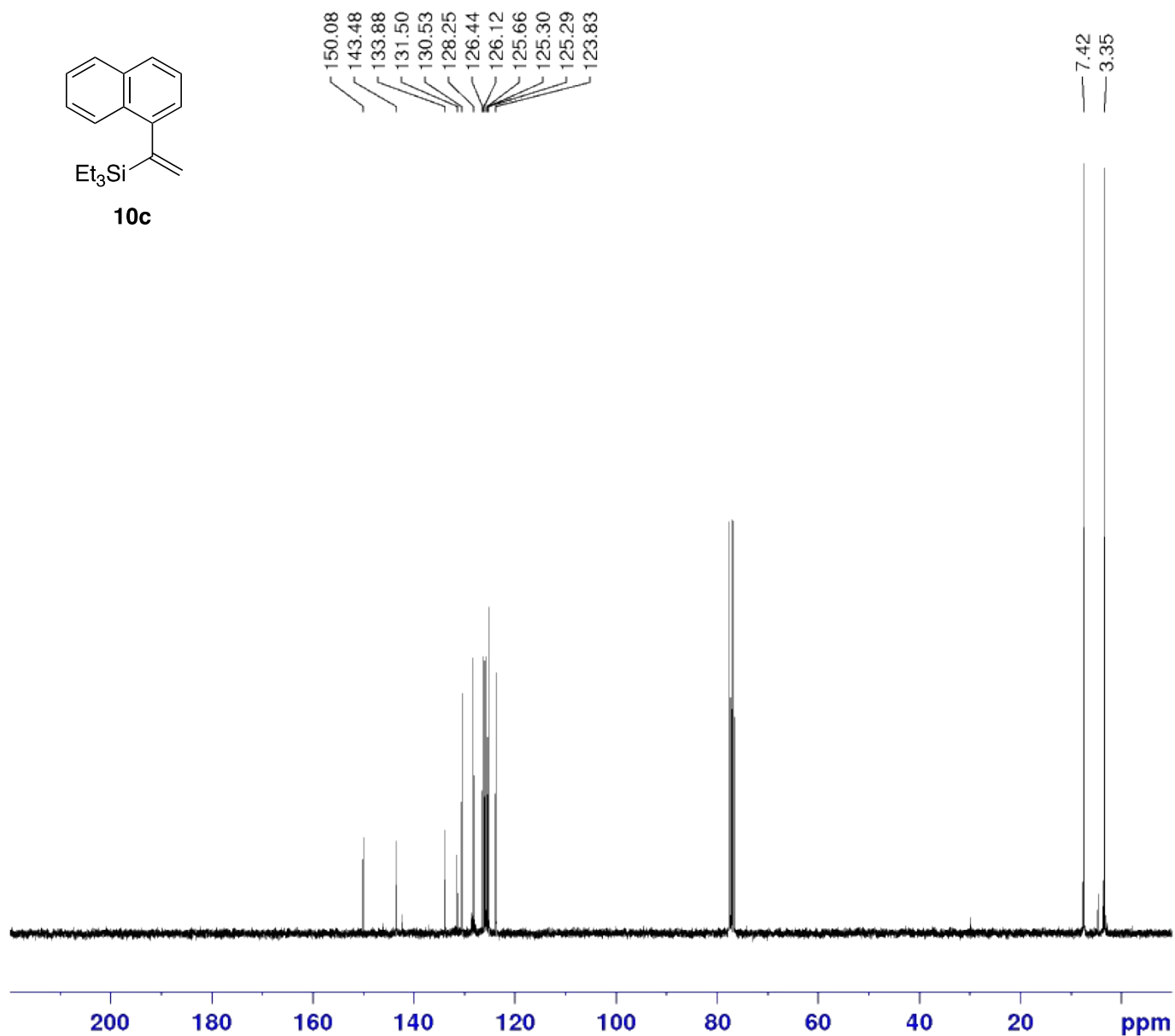
7.98
7.98
7.96
7.95
7.86
7.85
7.83
7.73
7.70
7.48
7.47
7.46
7.44
7.42
7.42
7.39
7.10
7.10
7.08
7.07
6.55
6.48
6.10
6.05
5.95
5.94
5.89
5.88

1.10
1.07
1.05
0.95
0.93
0.90
0.80
0.77
0.75
0.67
0.66
0.64
0.61
0.59

Current Data Parameters



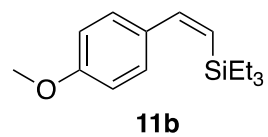
10c



Current Data Parameters
NAME mr-77
EXPNO 2
PROCNO 1

F2 - Acquisition Parameters
Date_ 20200622
Time 10.36 h
INSTRUM NEO_SPECT
PROBHD 2104275_0460 (
PULPROG zgpg30
TD 65536
SOLVENT CDCl3
NS 480
DS 2
SWH 20000.000 Hz
FIDRES 0.610352 Hz
AQ 1.6384000 sec
RG 101
DW 25.000 usec
DE 7.67 usec
TE -7.8 K
D1 2.00000000 sec
D11 0.03000000 sec
TD0 1
SFO1 75.4768047 MHz
NUC1 13C
P0 3.33 usec
P1 10.00 usec
PLW1 34.53799820 W
SFO2 300.1319508 MHz
NUC2 1H
CPDPRG[2] waltz16
PCPD2 90.00 usec
PLW2 8.82489967 W
PLW12 0.21354000 W
PLW13 0.10741000 W

F2 - Processing parameters
SI 65536
SF 75.4677400 MHz
WDW EM
SSB 0
LB 1.00 Hz
GB 0
PC 1.40

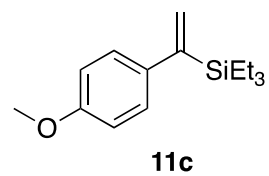


7.41
7.36
7.24
7.21
6.88
6.86
6.83
6.81
6.29
6.22
5.68
5.63

3.82
3.82

1.02
1.01
0.99
0.96
0.92
0.92
0.90
0.89
0.89
0.86
0.67
0.67
0.64
0.62
0.62
0.59

Current Data Parameters

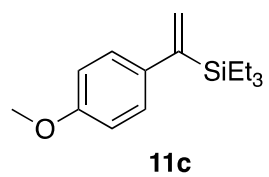


7.41
7.38
7.13
7.10
6.86
6.83
6.29
6.23
5.87
5.86
5.53
5.52

3.82
3.81

0.96
0.94
0.91
0.72
0.71
0.69
0.66
0.64
0.63

Current Data Parameters
NAME mr-83

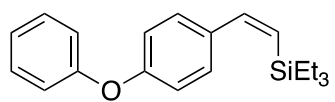


9.66
3.36
9.63
4.35
3.00
3.07
7.81
7.67
3.18
4.03
3.67

35

6
6
5
5

Current Data Parameters
11/11/2016 10:00:00

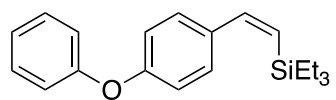


12b

7.45
7.40
7.37
7.35
7.32
7.28
7.25
7.14
7.11
7.09
7.04
7.04
7.01
7.01
6.98
6.95
6.37
5.76
5.71

0.92
0.89
0.87
0.62
0.59
0.57

Current Data Parameters
NAME mr-100

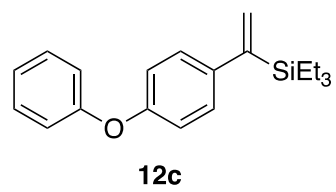


12b

157.42
156.72
147.01
135.90
129.89
129.45
128.96
127.64
123.38
119.03
119.00
118.97
118.46

7.64
4.93

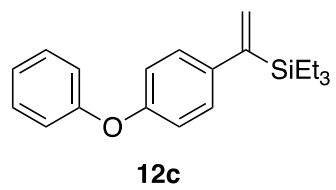
Current Data Parameters
NAME mr-100

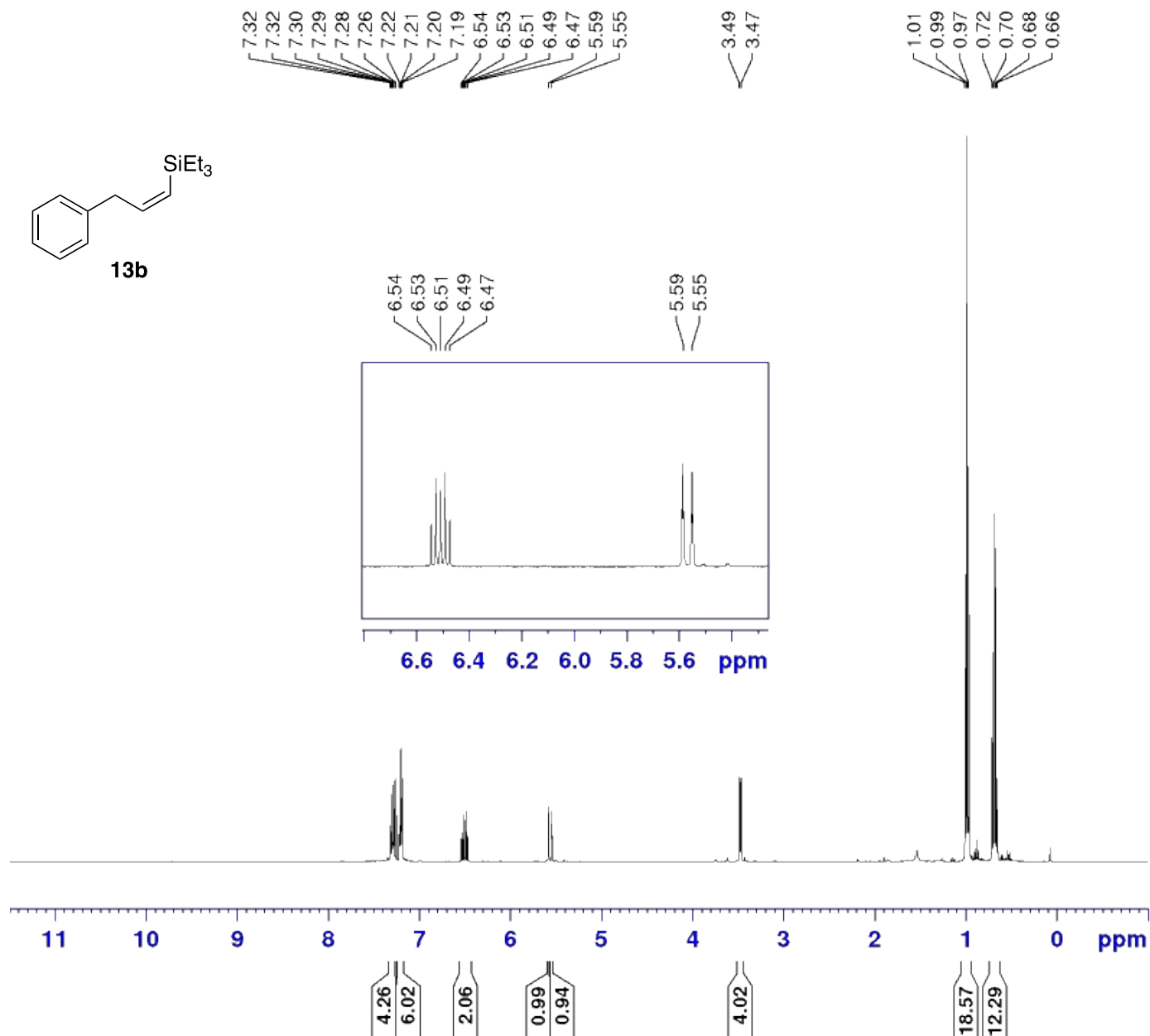


7.38
7.35
7.35
7.32
7.17
7.15
7.14
7.13
7.11
7.06
7.05
7.03
7.03
7.02
6.98
6.97
6.95
6.39
6.32
5.92
5.91
5.59
5.58

1.29
0.99
0.96
0.94
0.74
0.74
0.72
0.71
0.69
0.66
0.66
0.11

Current Data Parameters
NAME: 12c

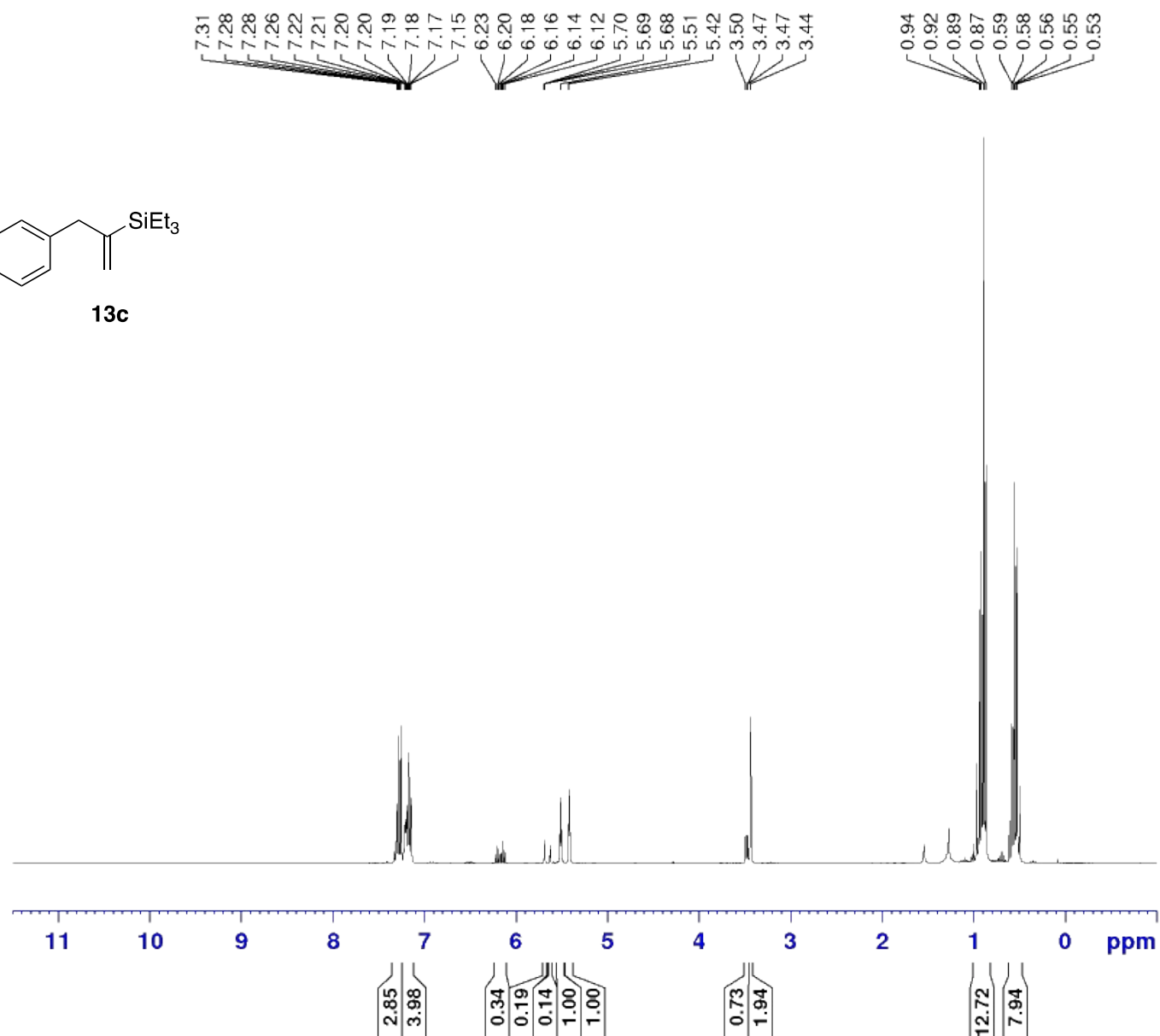
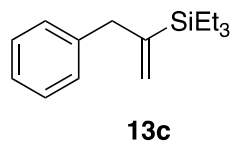




Current Data Parameters
 NAME mr-104
 EXPNO 1
 PROCNO 1

F2 - Acquisition Parameters
 Date_ 20200706
 Time 18.24 h
 INSTRUM spect
 PROBHD Z106618_0516 (
 PULPROG zg30
 TD 65536
 SOLVENT CDCl3
 NS 16
 DS 2
 SWH 8012.820 Hz
 FIDRES 0.244532 Hz
 AQ 4.0894465 sec
 RG 203
 DW 62.400 usec
 DE 16.93 usec
 TE 300.0 K
 D1 2.00000000 sec
 TDO 1
 SFO1 400.1326008 MHz
 NUC1 1H
 P0 4.97 usec
 P1 14.90 usec
 PLW1 11.99499989 W

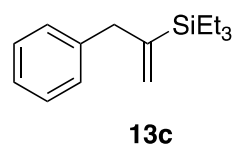
F2 - Processing parameters
 SI 65536
 SF 400.1300089 MHz
 WDW EM
 SSB 0
 LB 0.30 Hz
 GB 0
 PC 1.00



Current Data Parameters
NAME mr-95
EXPNO 2
PROCNO 1

F2 - Acquisition Parameters
Date_ 20200625
Time 10.07 h
INSTRUM NEO_SPECT
PROBHD Z104275_0460 (
PULPROG zg30
TD 65536
SOLVENT CDCl3
NS 16
DS 2
SWH 5882.353 Hz
FIDRES 0.179515 Hz
AQ 5.5705600 sec
RG 101
DW 85.000 usec
DE 8.24 usec
TE -7.8 K
D1 2.00000000 sec
TD0 1
SFO1 300.1319508 MHz
NUC1 1H
PQ 4.67 usec
P1 14.00 usec
PLW1 8.82489967 W

F2 - Processing parameters
SI 65536
SF 300.1300072 MHz
WDW EM
SSB 0
LB 0.30 Hz
GB 0
PC 1.00

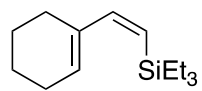


48.36
46.53
40.32
40.18
29.47
28.80
28.51
28.24
27.97
27.40
26.09
26.03

33.70
33.16

7.52
7.39
3.65
2.99

Current Data Parameters
NAME mr-95
EXPNO 2

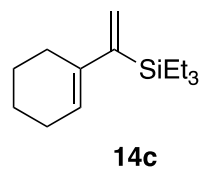


14b

6.76
6.76
6.72
6.72
5.65
5.65
5.35
5.31

2.09
2.04
1.65
1.64
1.64
1.63
1.62
1.58
1.57
1.57
1.54
0.95
0.93
0.91
0.61
0.59

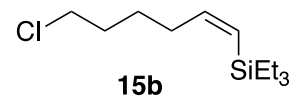
Current Data Parameters
NAME: 14b
EXP: 105



57
51
80
72
71
66
66
60
27
26

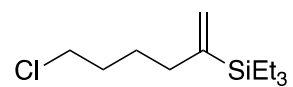
16
59
54
97
95
95
94
93
91
90
68
66
65
56
53

Support Data Parameters



6.40
6.38
6.35
6.33
6.30
5.46
5.46
5.46
5.42
5.41
5.41
3.56
3.54
3.52
2.15
2.15
2.13
2.12
1.83
1.82
1.80
1.80
1.77
1.56
1.53
1.51
0.97
0.95
0.92
0.62
0.62
0.60

Current Data Parameters
NAME: 15b



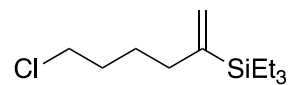
15c

3.53
7.72

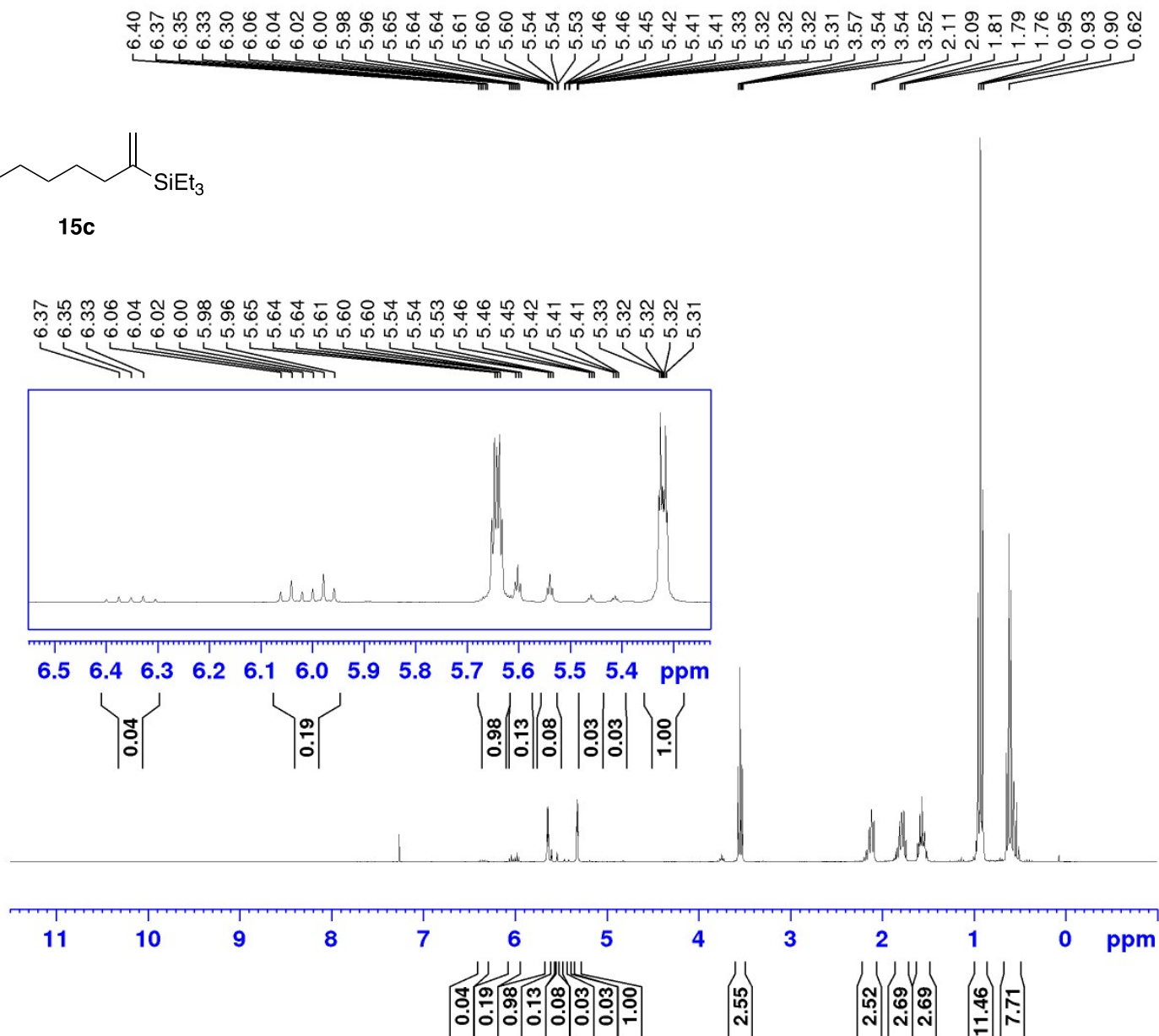
3.76
5.64

13
26
55
58
18
18
11
4
9
5
5
7

Current Data Parameters



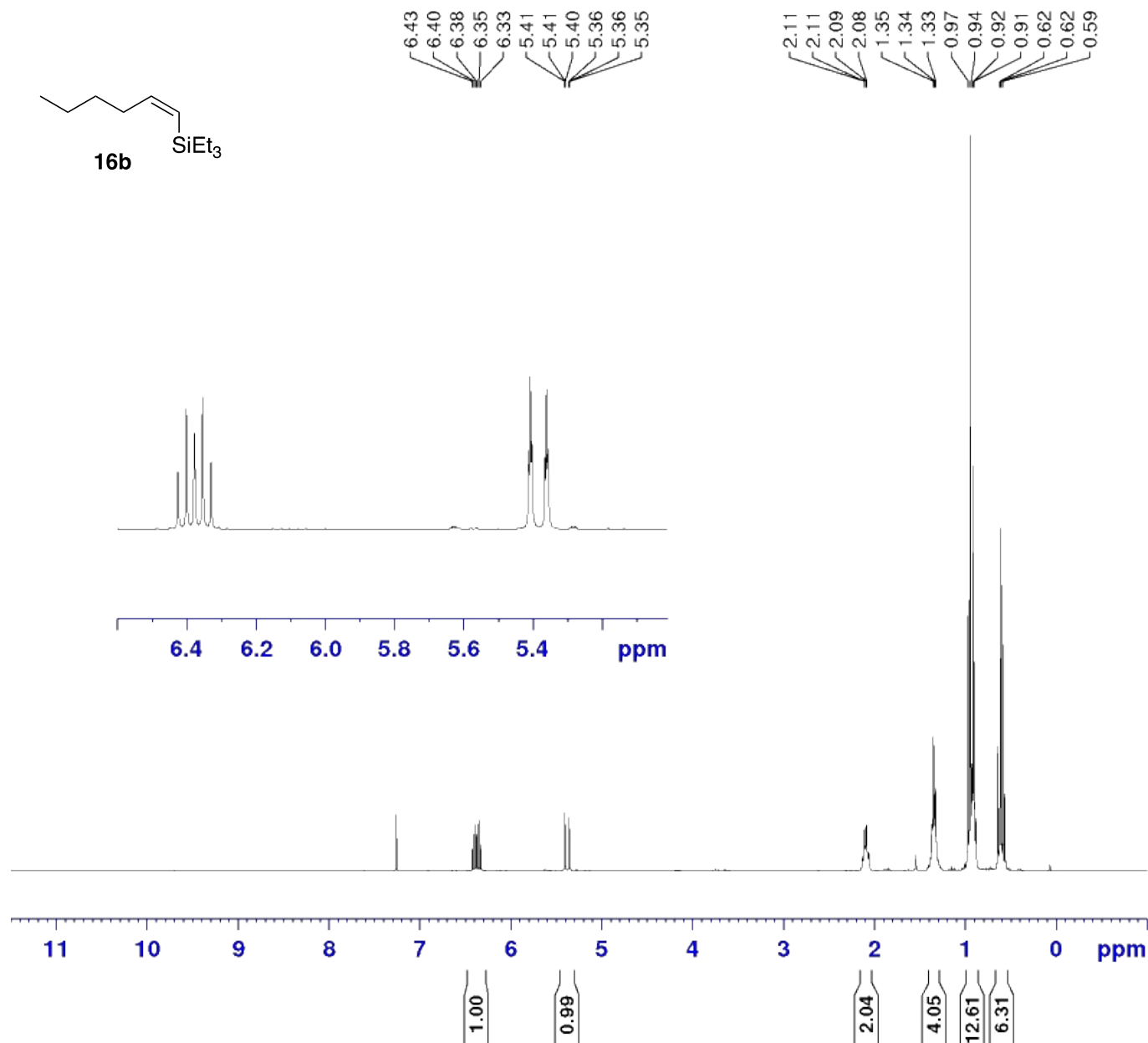
15c



Current Data Parameters
 NAME mr-101
 EXPNO 1
 PROCNO 1

F2 - Acquisition Parameters
 Date_ 20200701
 Time 16.01 h
 INSTRUM NEO_SPECT
 PROBHD Z104275_0460 (
 PULPROG zg30
 TD 65536
 SOLVENT CDCl3
 NS 16
 DS 2
 SWH 5882.353 Hz
 FIDRES 0.179515 Hz
 AQ 5.5705600 sec
 RG 47.4347
 DW 85.000 usec
 DE 8.24 usec
 TE -7.8 K
 D1 2.00000000 sec
 TD0 1
 SFO1 300.1319508 MHz
 NUC1 1H
 P0 4.67 usec
 P1 14.00 usec
 PLW1 8.82489967 W

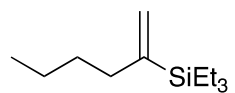
F2 - Processing parameters
 SI 65536
 SF 300.1300072 MHz
 WDW EM
 SSB 0
 LB 0.30 Hz
 GB 0
 PC 1.00



Current Data Parameters
NAME mr-62
EXPNO 1
PROCNO 1

F2 - Acquisition Parameters
Date_ 20200601
Time 17.08 h
INSTRUM NEO_SPECT
PROBHD Z104275_0460 (
PULPROG zg30
TD 65536
SOLVENT CDCl3
NS 16
DS 2
SWH 5882.353 Hz
FIDRES 0.179515 Hz
AQ 5.5705600 sec
RG 90.5
DW 85.000 usec
DE 8.24 usec
TE 296.0 K
D1 2.00000000 sec
TDO 1
SFO1 300.1319508 MHz
NUC1 1H
P0 4.67 usec
P1 14.00 usec
PLW1 8.82489967 W

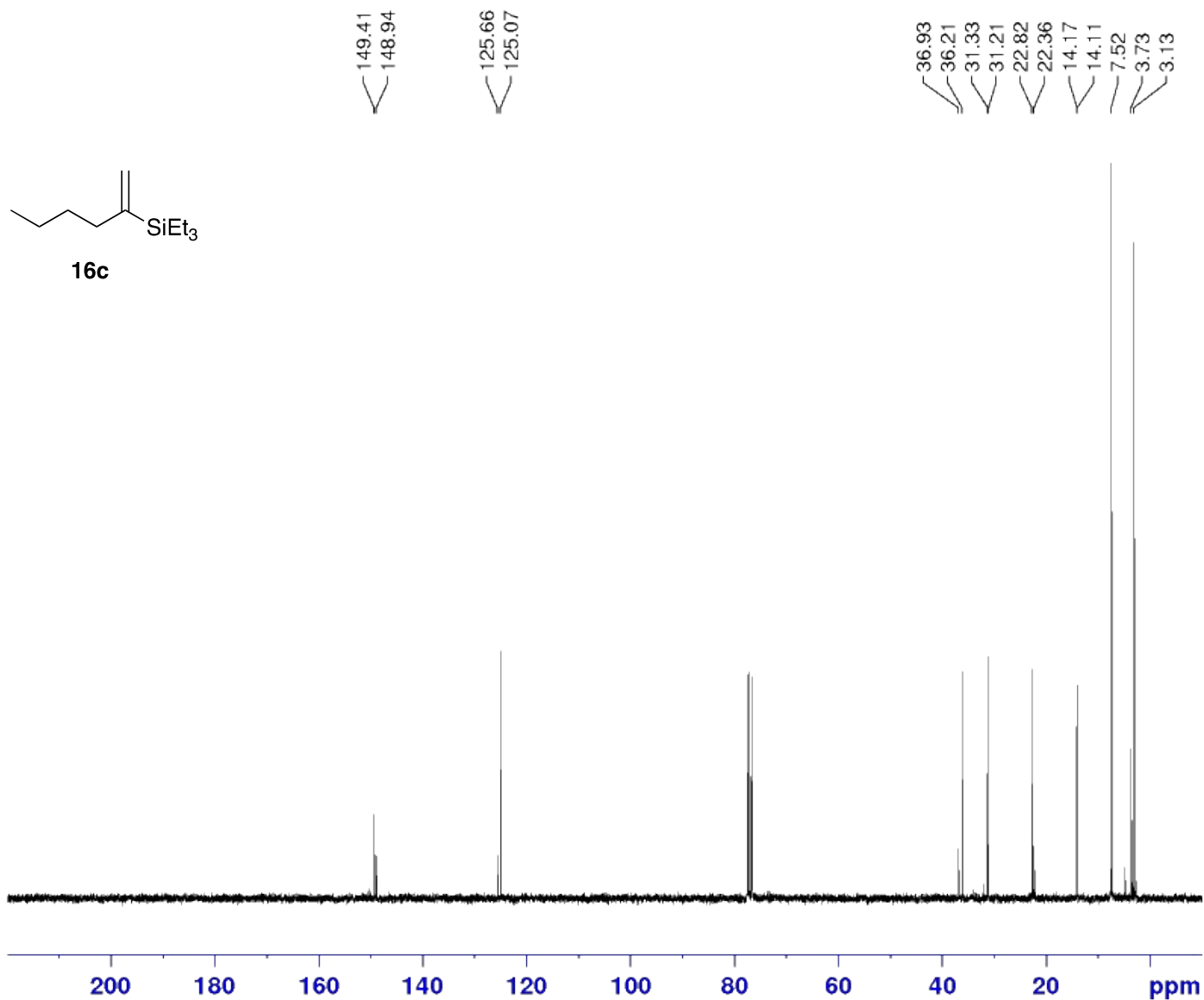
F2 - Processing parameters
SI 65536
SF 300.1300076 MHz
WDW EM
SSB 0
LB 0.30 Hz
GB 0
PC 1.00



16c

.41
.38
.36
.33
.09
.07
.05
.03
.01
.98
.64
.64
.63
.63
.62
.57
.51
.41
.37
.30
.29
.29
.29
.28
.28
.09
.37
.35
.35
.96
.93
.91
.91
.65
.64
.62
.62
.59

Support Data Page 10



Current Data Parameters
NAME mr-102
EXPNO 4
PROCNO 1

F2 - Acquisition Parameters
Date_ 20200701
Time 16.25 h
INSTRUM NEO_SPECT
PROBHD Z104275_D460 (
PULPROG zgpg30
TD 65536
SOLVENT CDC13
NS 151
DS 2
SWH 20000.000 Hz
FIDRES 0.610352 Hz
AQ 1.6384000 sec
RG 101
DW 25.000 usec
DE 7.67 usec
TE -7.8 K
D1 2.00000000 sec
D11 0.03000000 sec
TD0 1
SFO1 75.4768047 MHz
NUC1 13C
P0 3.33 usec
P1 10.00 usec
PLW1 34.53799820 W
SFO2 300.1319508 MHz
NUC2 1H
CPDPRG[2] waltz16
PCPD2 90.00 usec
PLW2 8.82489967 W
PLW12 0.21354000 W
PLW13 0.10741000 W

F2 - Processing parameters
SI 65536
SF 75.4677360 MHz
WDW EM
SSB 0
LB 1.00 Hz
GB 0
PC 1.40