

Supporting Information

Mn-decorated CeO₂ nanorod supported iron-based catalyst for high-temperature Fischer–Tropsch synthesis of light olefins

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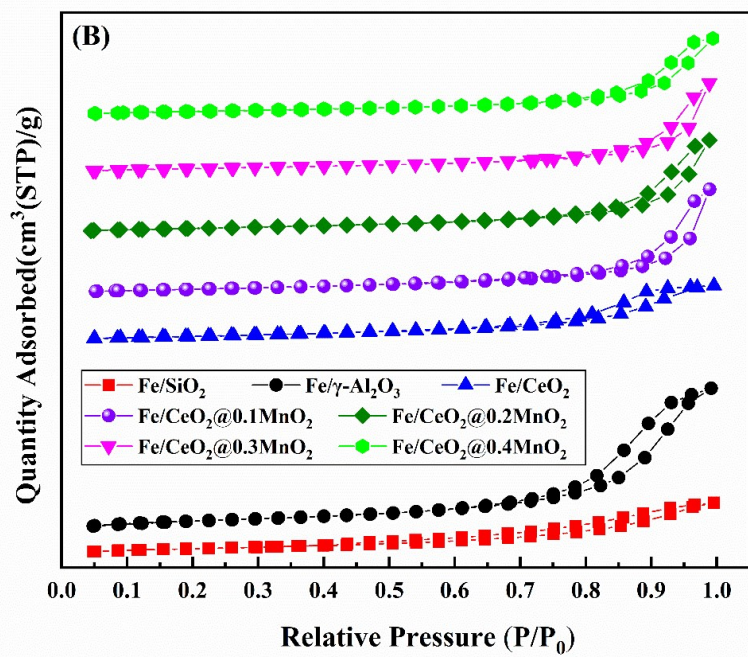
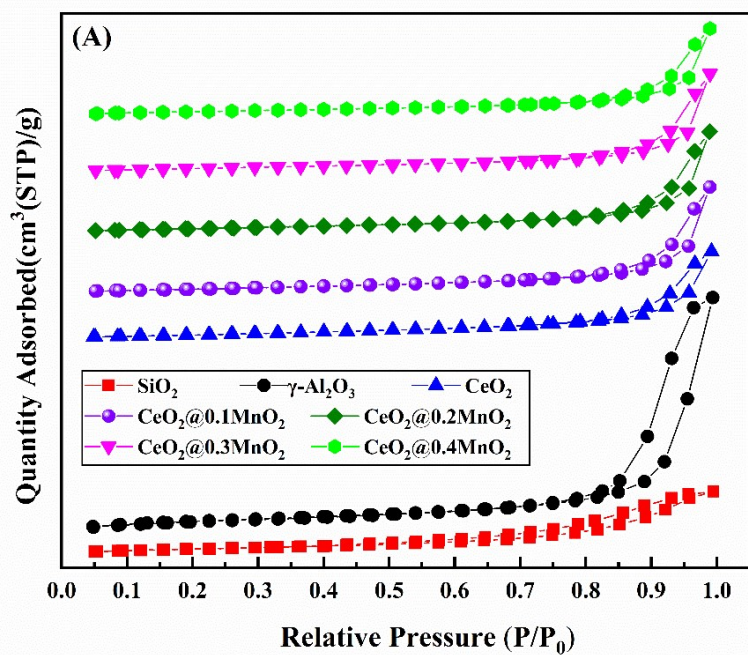


Fig. S1. Ar adsorption-desorption isotherms of the samples: (A) supports (B) Fe-based catalysts.

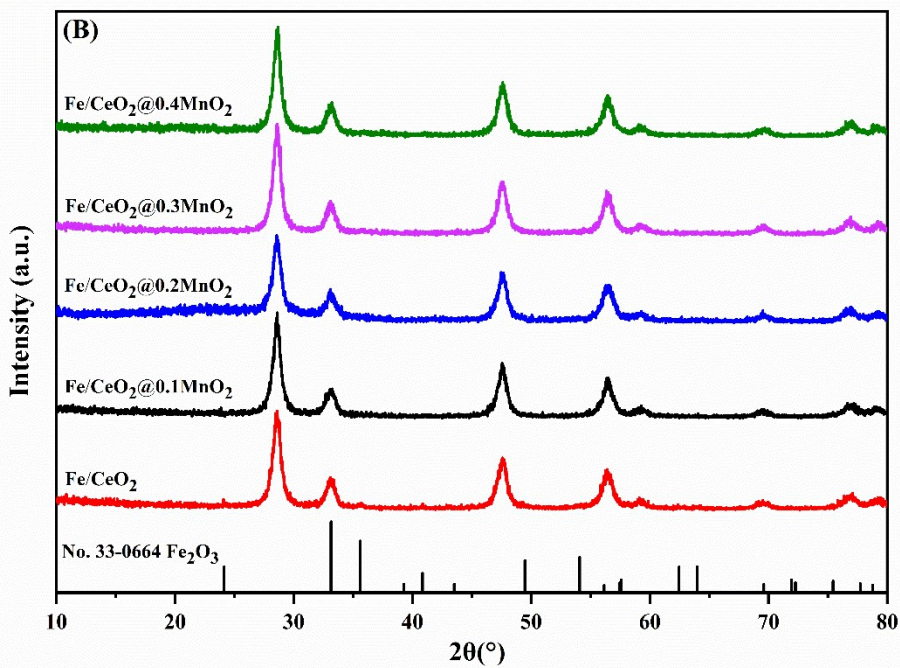
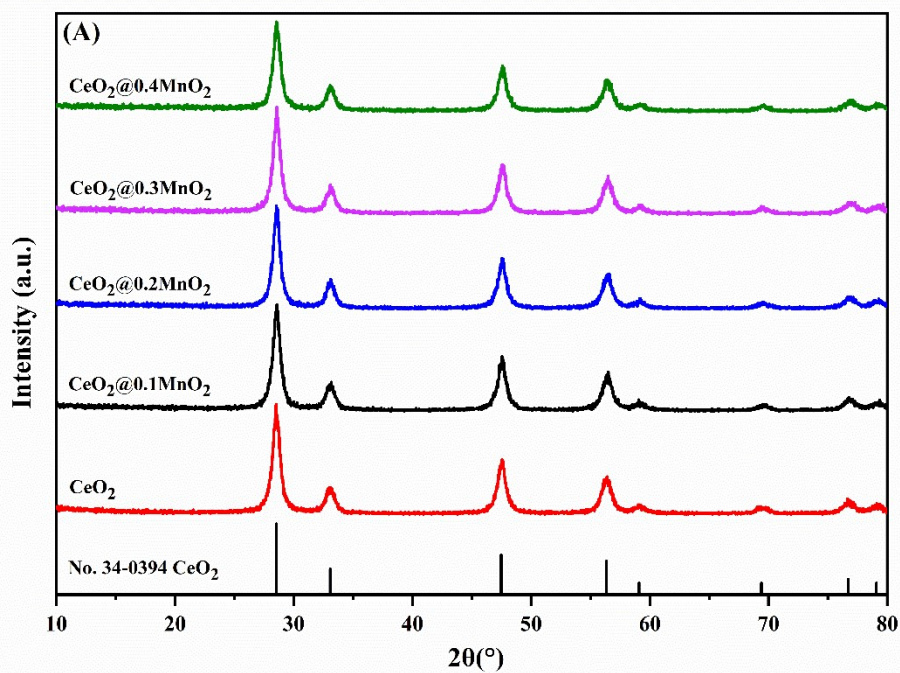


Fig. S2. XRD patterns of the samples: (A) Mn-doped CeO₂ (B) Mn-doped CeO₂ supported Fe-based catalysts.

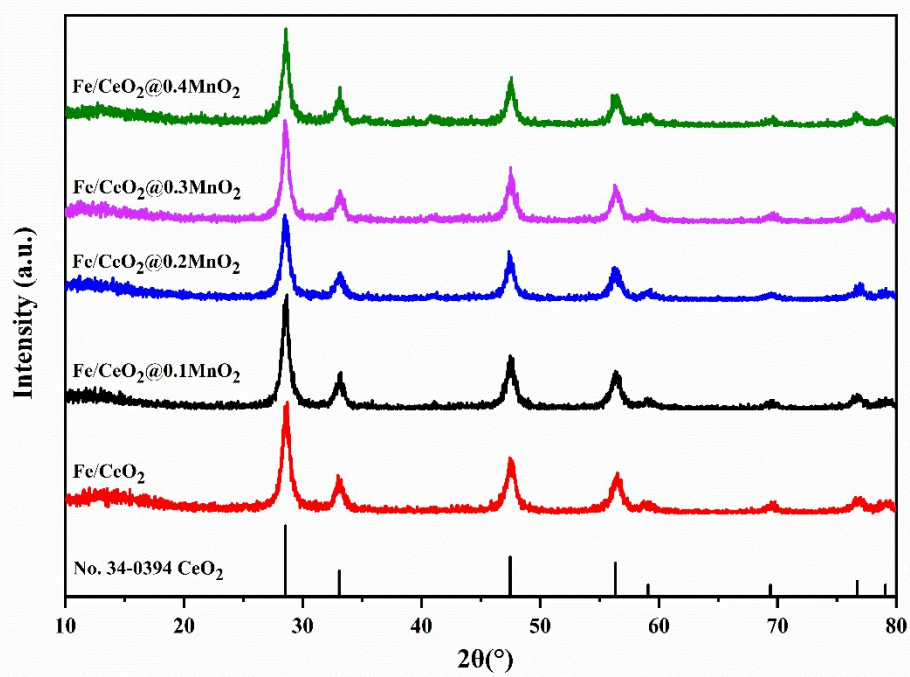


Fig. S3. XRD patterns of the samples after reaction of 48 h.

Table S1. Textural properties of the fresh catalysts

Catalyst	Content (wt.%) ^a	
	Na	K
Fe/SiO ₂	< 10 ⁻⁵	< 10 ⁻⁵
Fe/ γ -Al ₂ O ₃	< 10 ⁻⁵	< 10 ⁻⁵
Fe/CeO ₂	0.12	< 10 ⁻⁵
Fe/CeO ₂ @0.1MnO ₂	0.05	0.09
Fe/CeO ₂ @0.2MnO ₂	0.10	0.15
Fe/CeO ₂ @0.3MnO ₂	0.09	0.22
Fe/CeO ₂ @0.4MnO ₂	0.07	0.17

^a Measured by ICP-AES.

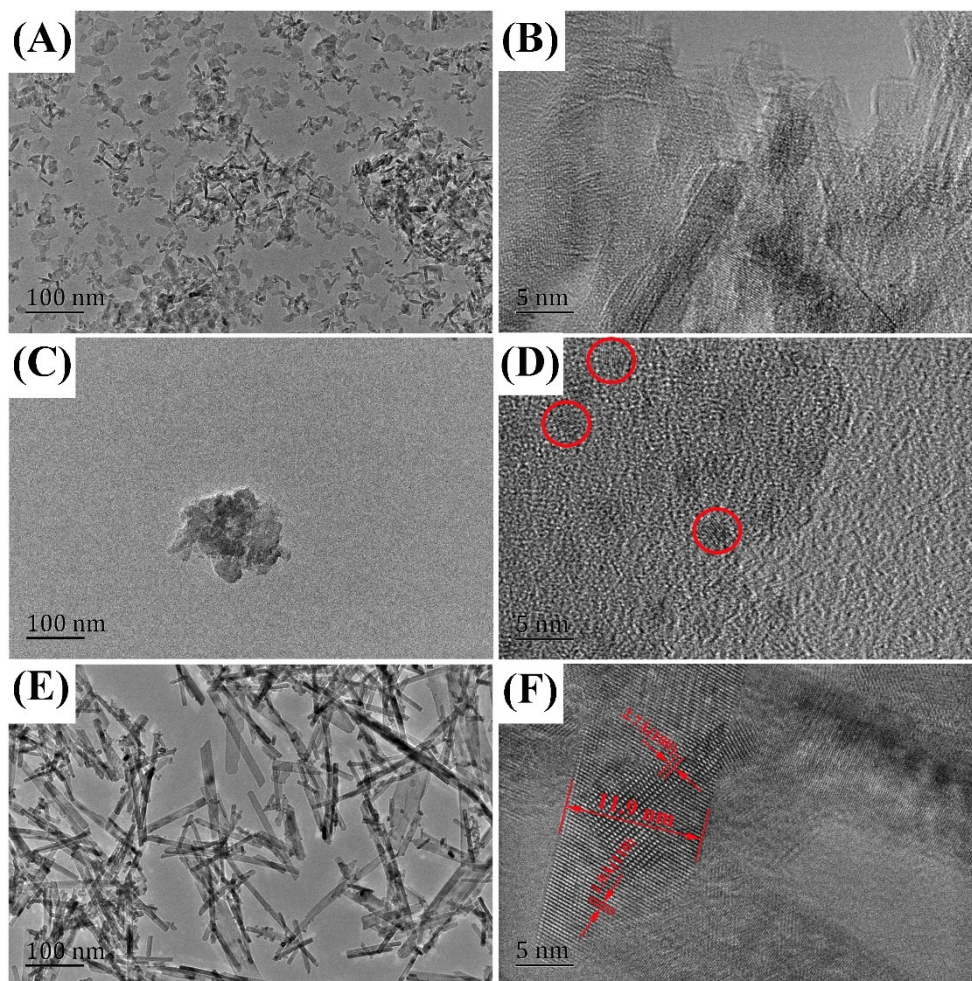


Fig. S4. HRTEM images of the supports: (A, B) SiO₂; (C, D) γ -Al₂O₃; (E, F) CeO₂.

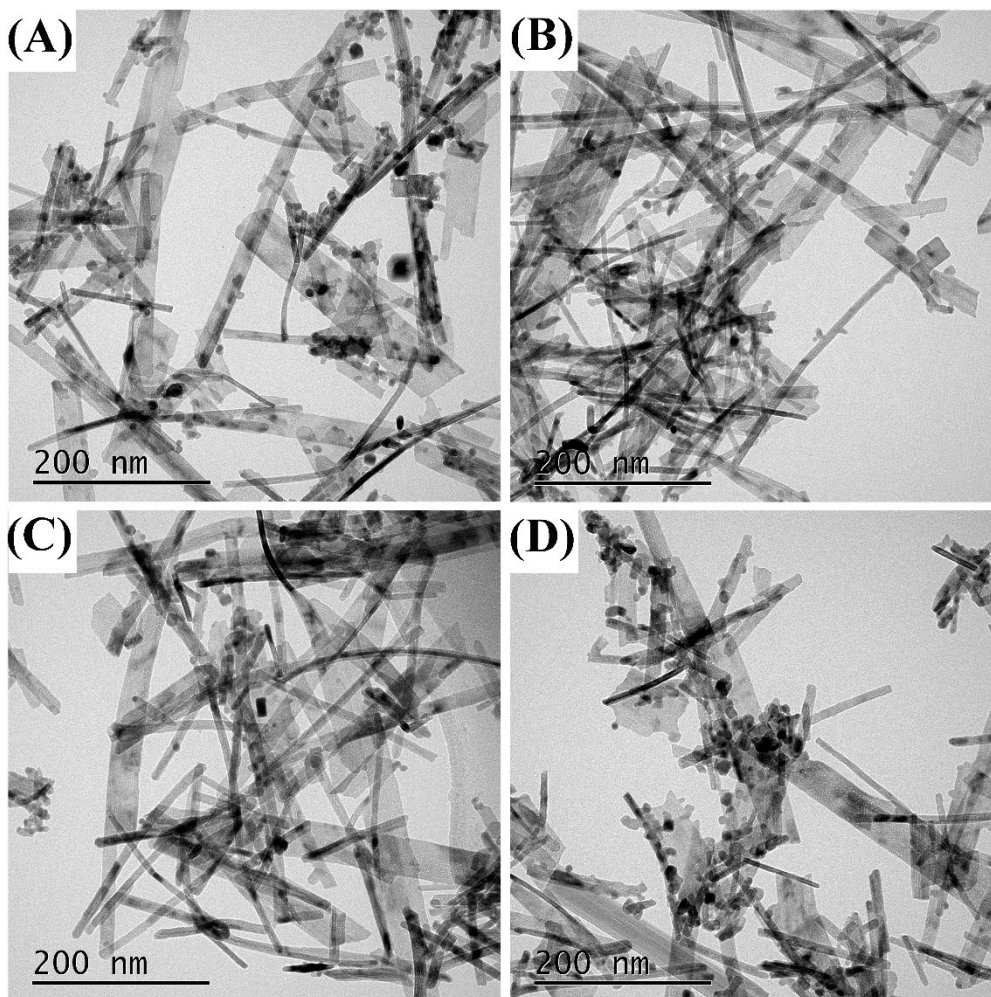
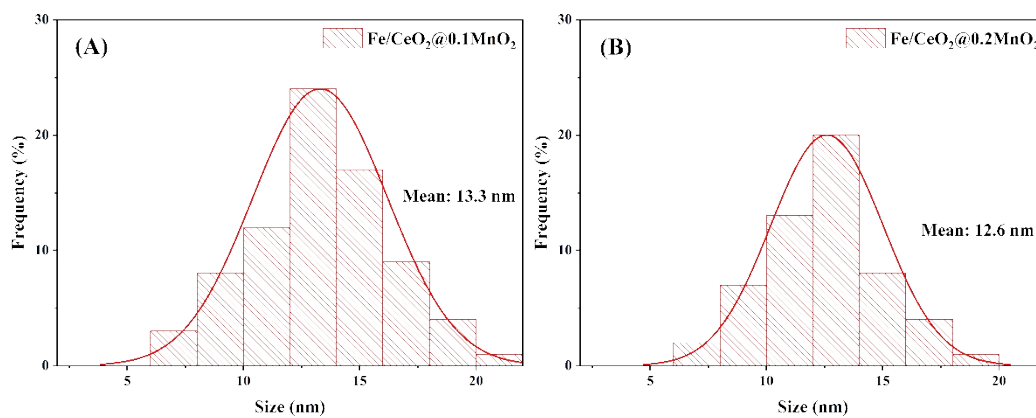


Fig. S5. HRTEM images of Mn-doped CeO₂ supports: (A) CeO₂@0.1MnO₂, (B) CeO₂@0.2MnO₂, (C) CeO₂@0.3MnO₂, (D) CeO₂@0.4MnO₂.



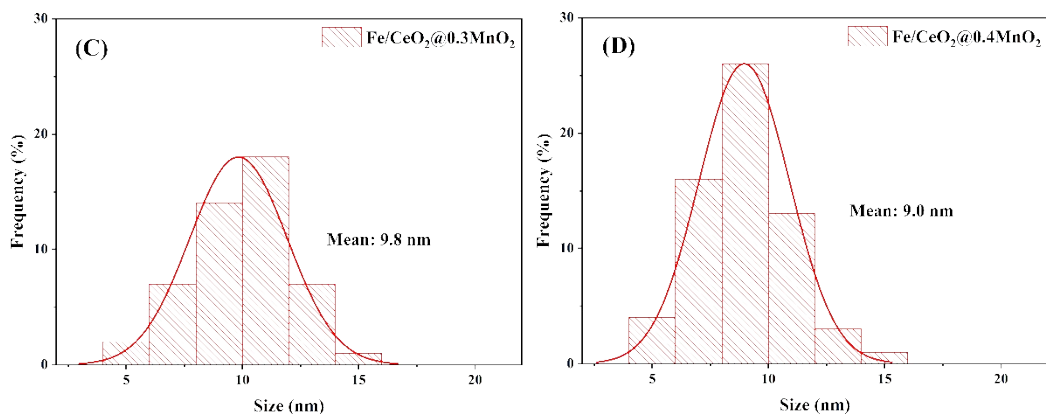


Fig. S6. Particle size distribution histograms: (A) $\text{CeO}_2@0.1\text{MnO}_2$, (B) $\text{CeO}_2@0.2\text{MnO}_2$, (C) $\text{CeO}_2@0.3\text{MnO}_2$, (D) $\text{CeO}_2@0.4\text{MnO}_2$.

Table S2. XPS peak positions of the spent catalysts

Catalyst	Ce 3d				Fe 2p	
	Peak 1 ^a	Peak 2 ^a	Peak 3 ^a	Peak 4 ^a	Peak 1 ^a	Peak 2 ^a
Fe/CeO ₂	916.7	900.9	898.4	882.4	733.8	717.7
Fe/CeO ₂ @0.1MnO ₂	917.0	901.1	898.7	882.7	733.5	717.5
Fe/CeO ₂ @0.2MnO ₂	917.1	901.3	898.8	882.8	732.8	717.1
Fe/CeO ₂ @0.3MnO ₂	916.6	900.8	898.3	882.3	733.6	717.7
Fe/CeO ₂ @0.4MnO ₂	916.7	901.1	898.5	882.6	733.6	717.5

^a Binding Energy (eV).

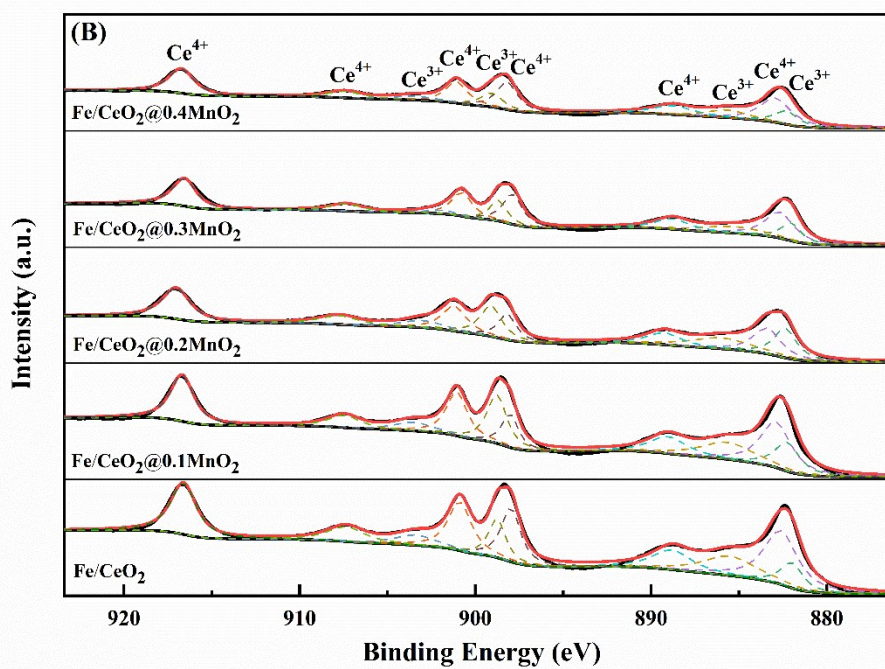
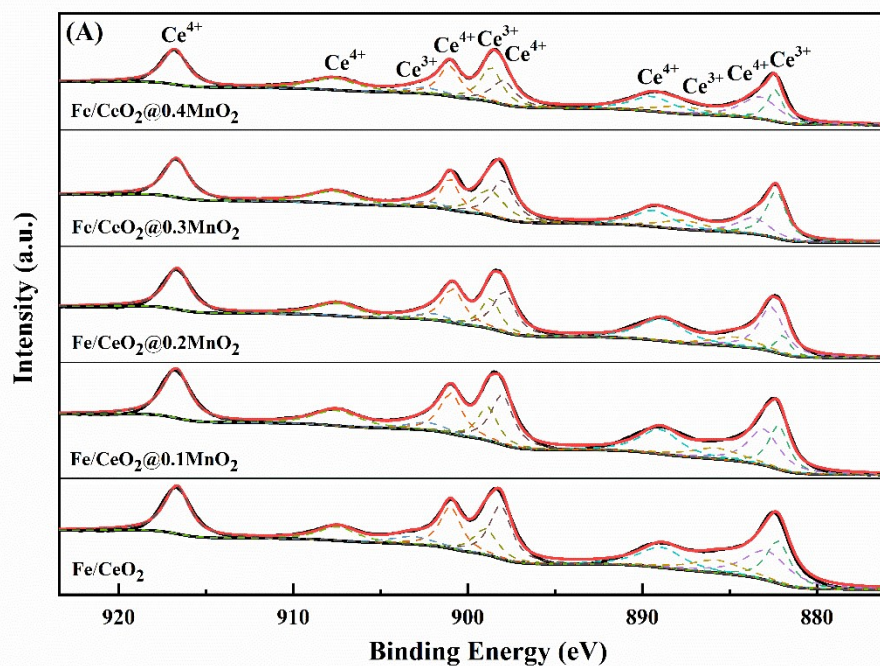


Fig. S7. Fitting results of Ce 3d XPS spectra: (A) fresh samples, (B) spent samples.

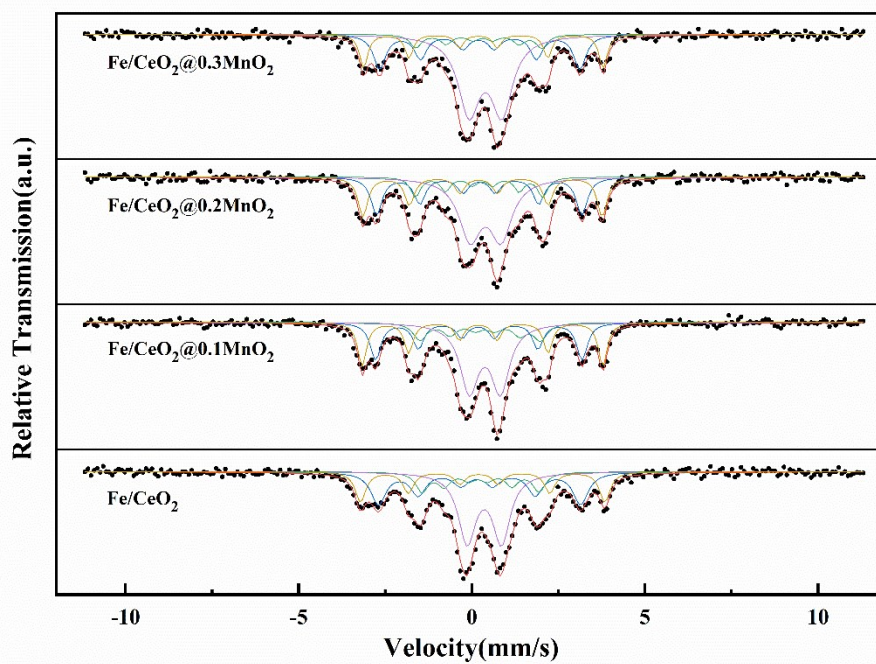


Fig. S8. Mössbauer spectrum of the samples after reaction. Reaction condition: 320 °C, $H_2/CO=2$, 1.5 MPa, 8000 mL/(h·g_{Cat}), 48 h.