

Supporting Information

In-situ sulfation of Cu/TiO₂ catalysts for catalytic combustion of dichloromethane

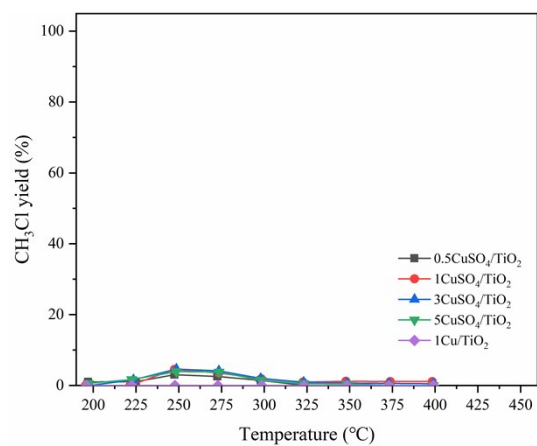
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24 **Fig. S1.** MCM yield during the catalytic oxidation process. Gas composition: 1000

25 ppm DCM, 10% O₂ and N₂ balance; GHSV = 30,000 mL g⁻¹ h⁻¹.

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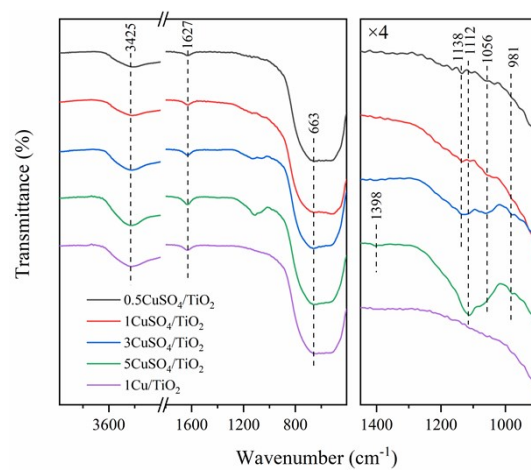


Fig. S2. FT-IR spectra of 1Cu/TiO₂ and *x*CuSO₄/TiO₂ samples.

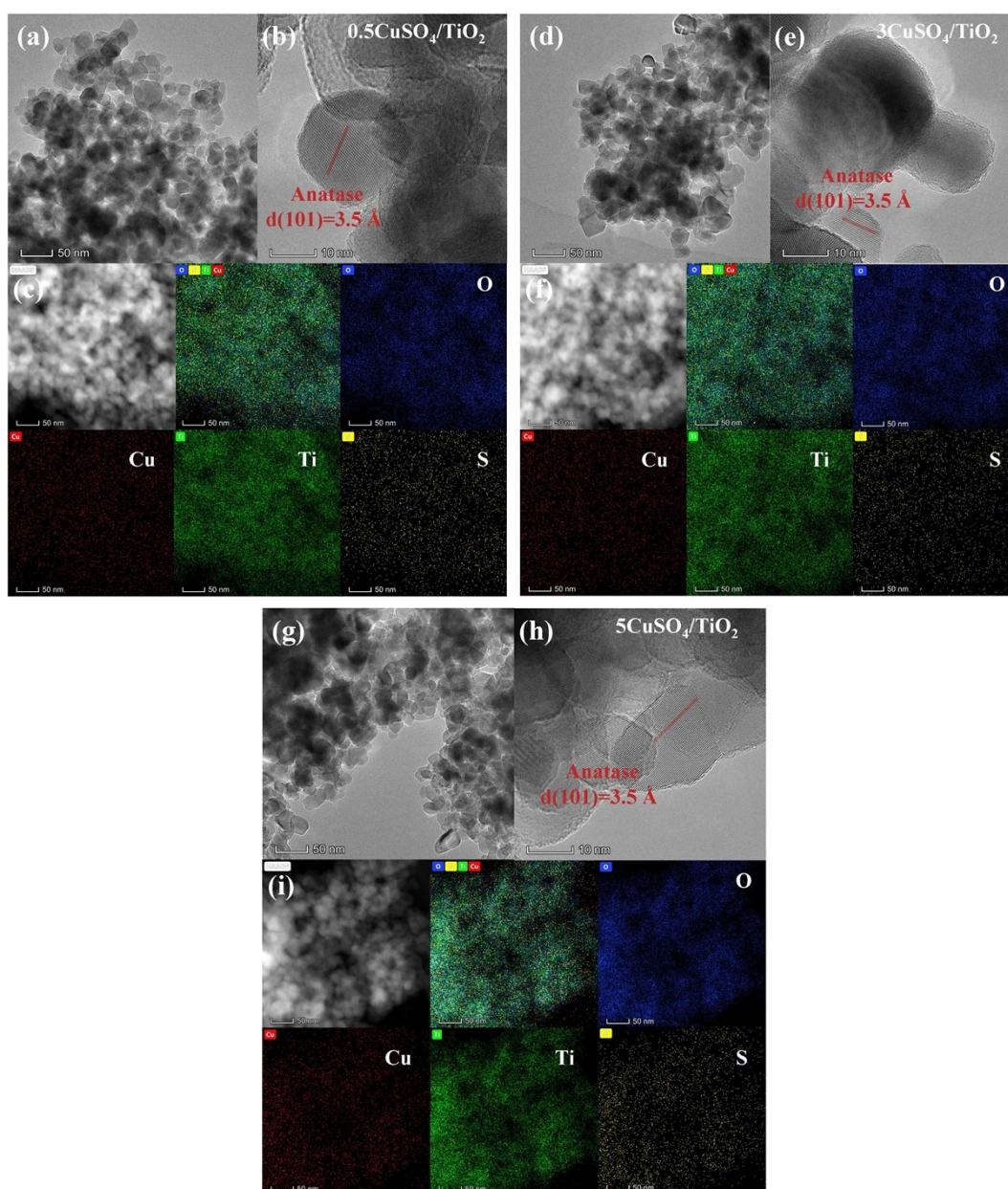
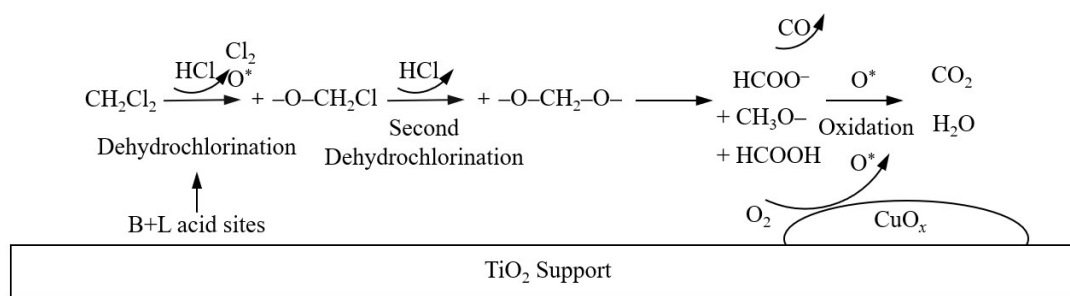


Fig. S3. TEM, HRTEM, HAADF-STEM, and EDS-Mapping images of $0.5\text{CuSO}_4/\text{TiO}_2$ (a, b, c), $3\text{CuSO}_4/\text{TiO}_2$ (d, e, f), and $5\text{CuSO}_4/\text{TiO}_2$ (g, h, i).



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36 **Scheme S1.** The possible reaction mechanism for the catalytic combustion of DCM

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over the $\text{CuSO}_4/\text{TiO}_2$ catalyst.

Table S1. Catalytic activity in DCM oxidation over various catalysts.

Entry	Catalyst	Reaction conditions	T ₉₀	Reference
1	1CuSO ₄ /TiO ₂	1000 ppm DCM+10% O ₂ , 30000 ml g _{cat} ⁻¹ h ⁻¹	270 °C	This work
2	Ce/TiO ₂	1000 ppm DCM+10% O ₂ , GHSV = 30000 h ⁻¹	320 °C	26
3	Cr _{0.1} Ti _{0.9}	1000 ppm DCM, 30000 ml g _{cat} ⁻¹ h ⁻¹	320 °C	52
4	SO ₄ ²⁻ /Fe ₂ O ₃ -II	750 ppm DCM+10% O ₂ , GHSV = 30000 h ⁻¹	305 °C	21
5	PO _x -CeO ₂ - 0.05	500 ppm DCM, 15000 ml g _{cat} ⁻¹ h ⁻¹	275 °C	19
6	Ru/Ti _{0.6} Sn _{0.4}	1000 ppm DCM+10% O ₂ , 30000 ml g _{cat} ⁻¹ h ⁻¹	295 °C	43
7	Ru/Ti	1000 ppm DCM+10% O ₂ , GHSV = 30000 h ⁻¹	305 °C	53